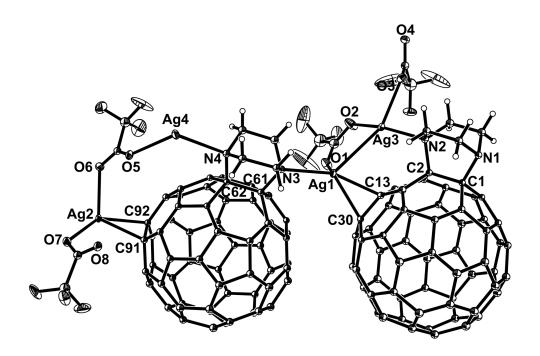
## **Supporting Information for**

**Title**: Formation of Crystalline Polymers from the Reaction of Amine Functionalized  $C_{60}$  with Silver Salts

Authors: Christopher J. Chancellor, Marilyn M. Olmstead, Alan L. Balch



**Figure SI-1**.Structure of  $\{[C_{60}(N(CH_2CH_2)_2N)][Ag(O_2CCF_3)]_2\}$  •  $CS_2$  with the asymmetric unit shown. Thermal ellipsoids are shown at 50% probability level. The carbon disulfide molecules were omitted for clarity.

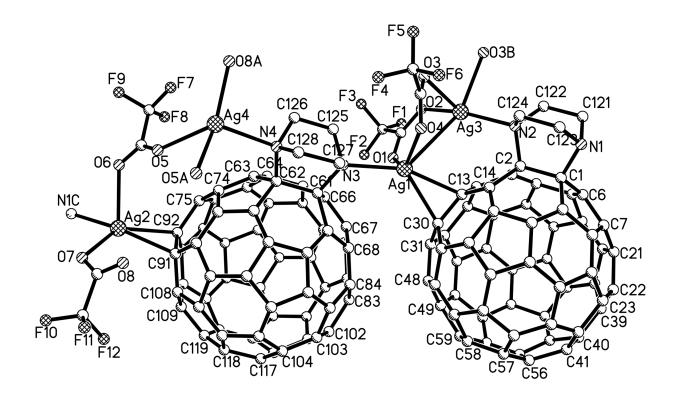
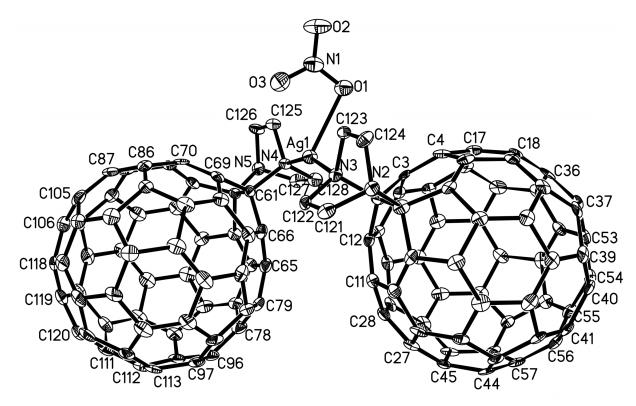
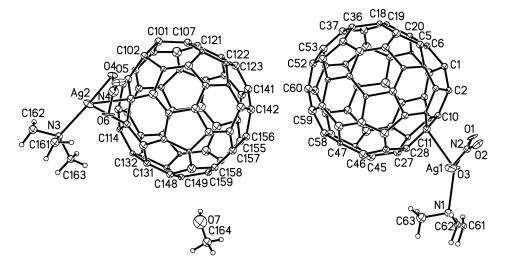


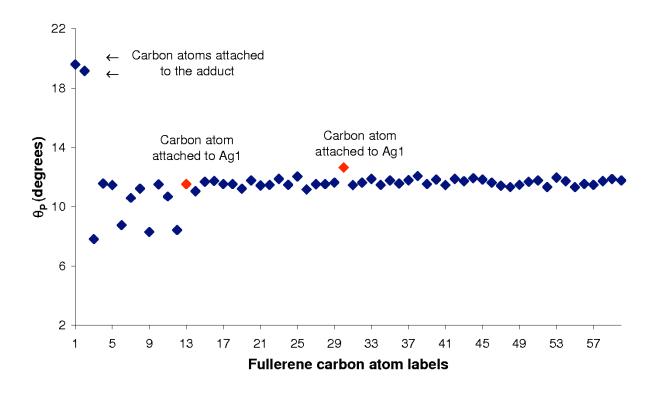
Figure SI-2. Numbering for  $\{[C_{60}(N(CH_2CH_2)_2N)][Ag(O_2CCF_3)]_2\} \bullet CS_2$ .

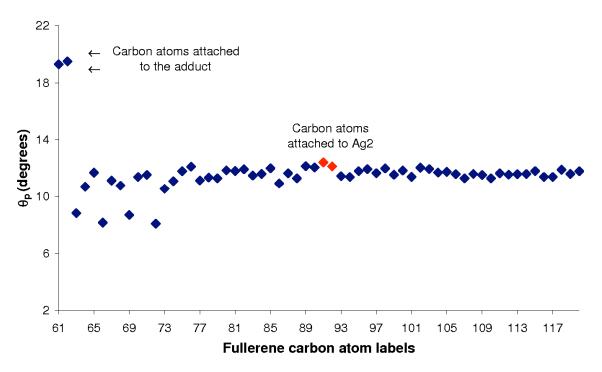


**Figure SI-3**. Numbering for  $[C_{60}(N(CH_2CH_2)_2N)]_2Ag(NO_3) \bullet 0.5CH_3OH \bullet CH_2Cl_2$  with 50 % thermal contours.

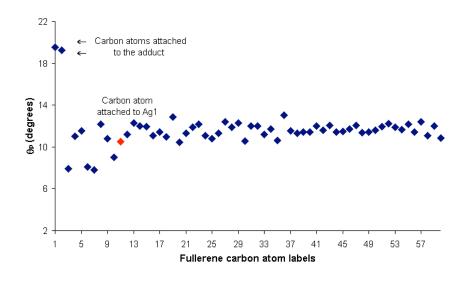


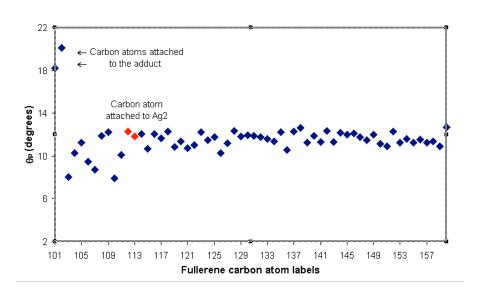
**Figure SI-4**. Numbering for  $\{[C_{60}(CH_2N(CH_3)CH_2)]Ag(NO_3)\}$  • 0.25CH<sub>3</sub>OH with 50 % thermal contours.





**Figure SI-5**. Pyramidalization angles for fullerene carbon atoms in  $\{[C_{60}(N(CH_2CH_2)_2N)][Ag(O_2CCF_3)]_2\} \bullet CS_2$ 





**Figure SI-6**. Pyramidalization angles for fullerene carbon atoms in  $\{[C_{60}(CH_2N(CH_3)CH_2)]Ag(NO_3)\} \bullet 0.25CH_3OH.$