

Supporting Information for:

Oxidative Addition of Iodomethane to Charge-tuned Rh^I Complexes.

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Table S1. Mulliken populations in gas phase and solution (PCM, CH_2Cl_2) for **1**, **2** and **3**. Only the relevant atoms are reported. For compound **2**, the protonation is on the atom N1 [B3LYP/BS-I//B3LYP/BS-I; BS-I = 6-31g** + DGDZVP (Rh)]

	Gas									
	Rh	S1	S2	N1	N2	N3	P1	P2	C	O
1	-0,217	-0,093	-0,093	-0,382	-0,382	-0,833	0,926	0,926	0,341	-0,289
2	-0,191	0,062	-0,039	-0,413	-0,355	-0,838	0,897	0,925	0,340	-0,249
3	-0,151	0,101	0,101	-0,406	-0,406	-0,846	0,894	0,894	0,335	-0,215
	CH_2Cl_2									
	Rh	S1	S2	N1	N2	N3	P1	P2	C	O
1	-0,229	-0,164	-0,165	-0,364	-0,364	-0,858	0,927	0,927	0,347	-0,327
2	-0,193	0,046	-0,112	-0,408	-0,360	-0,861	0,896	0,926	0,343	-0,295
3	-0,158	0,079	0,078	-0,401	-0,406	-0,862	0,896	0,896	0,338	-0,264

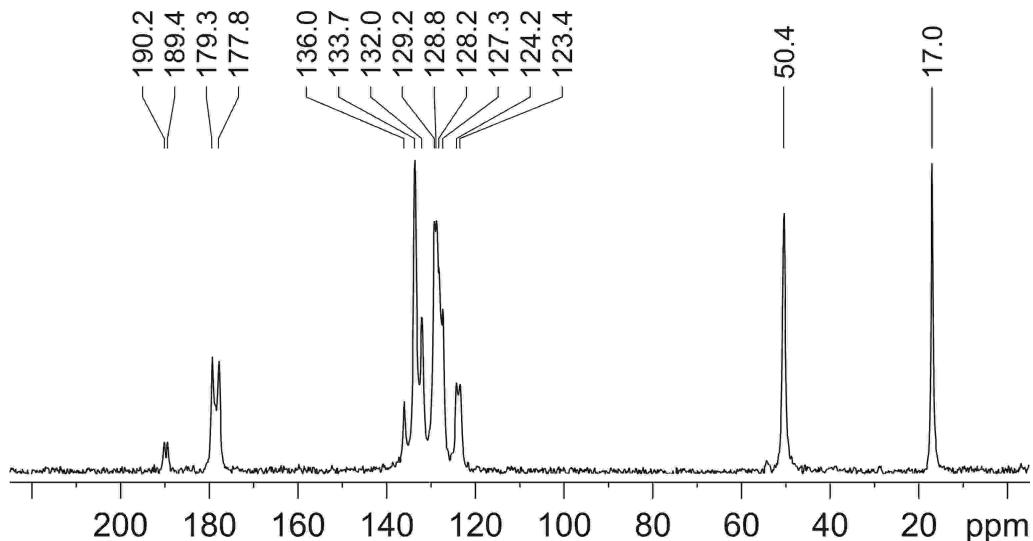


Figure S1. CP-MAS ^{13}C NMR of **1**.

CP-MAS ^{13}C NMR: $\delta = 189.8$ (d, $^1J_{\text{C},\text{Rh}} = 80.5$ Hz, CO), 178.5 (d, $^1J_{\text{C},\text{P}} = 151$ Hz, C=N), 136.0-123.4 (m, Ph), 50.4 (s, $-\text{CH}_2\text{CH}_3$), 17.0 (s, $-\text{CH}_2\text{CH}_3$) ppm.

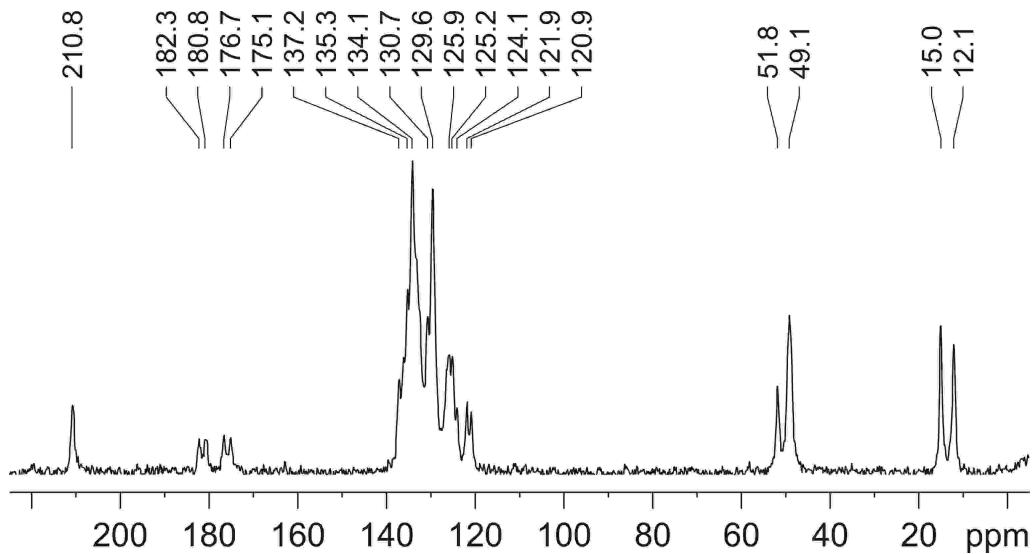


Figure S2. CP-MAS ^{13}C NMR of **4**.

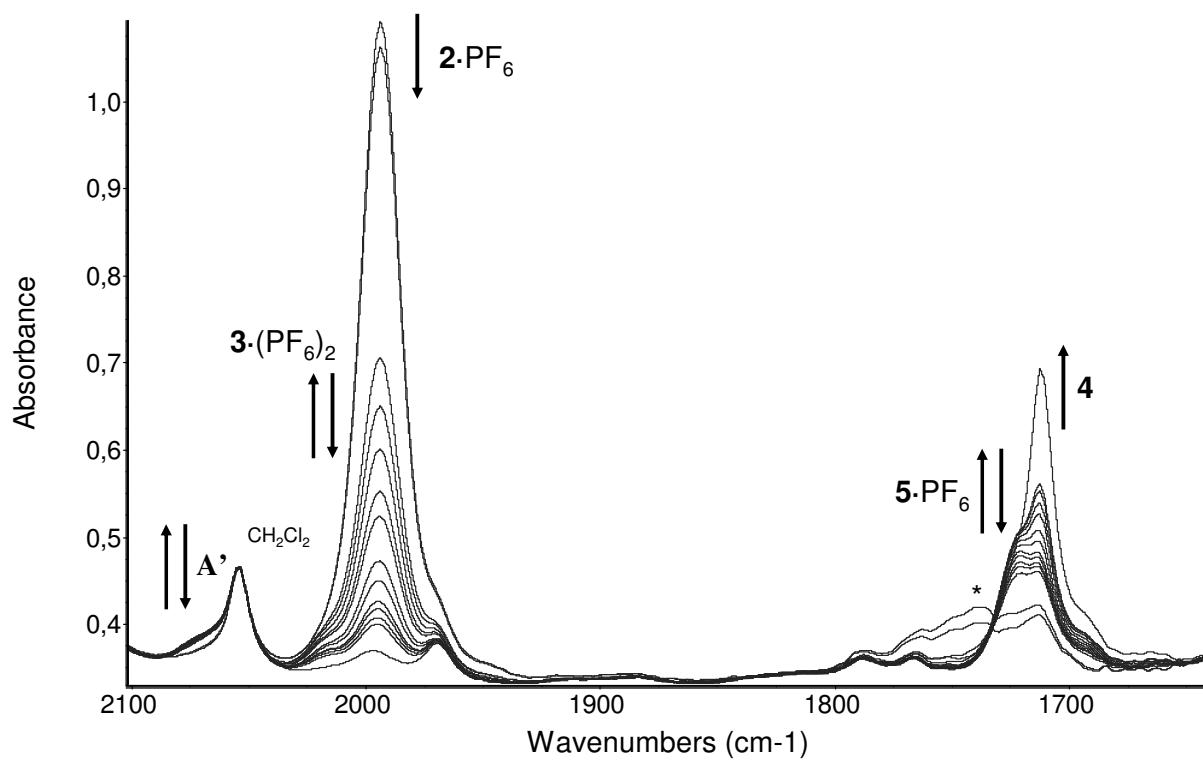


Figure S3. Superimposed FTIR spectra in CH_2Cl_2 solution of $\mathbf{2}\cdot\text{PF}_6 + \text{CH}_3\text{I}$. (* = possible isomer of $\mathbf{5}\cdot\text{PF}_6$)

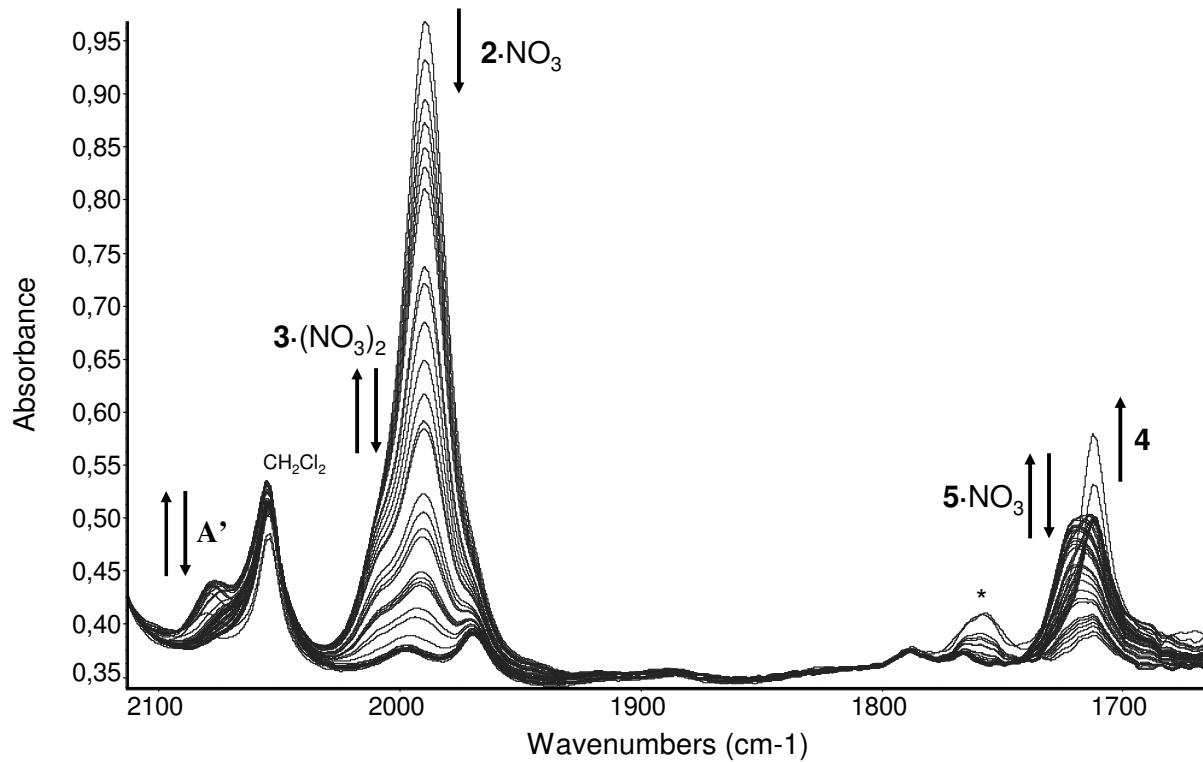


Figure S4. Superimposed FTIR spectra in CH_2Cl_2 solution of $2\cdot\text{NO}_3 + \text{CH}_3\text{I}$. (* = possible isomer of $5\cdot\text{NO}_3$)

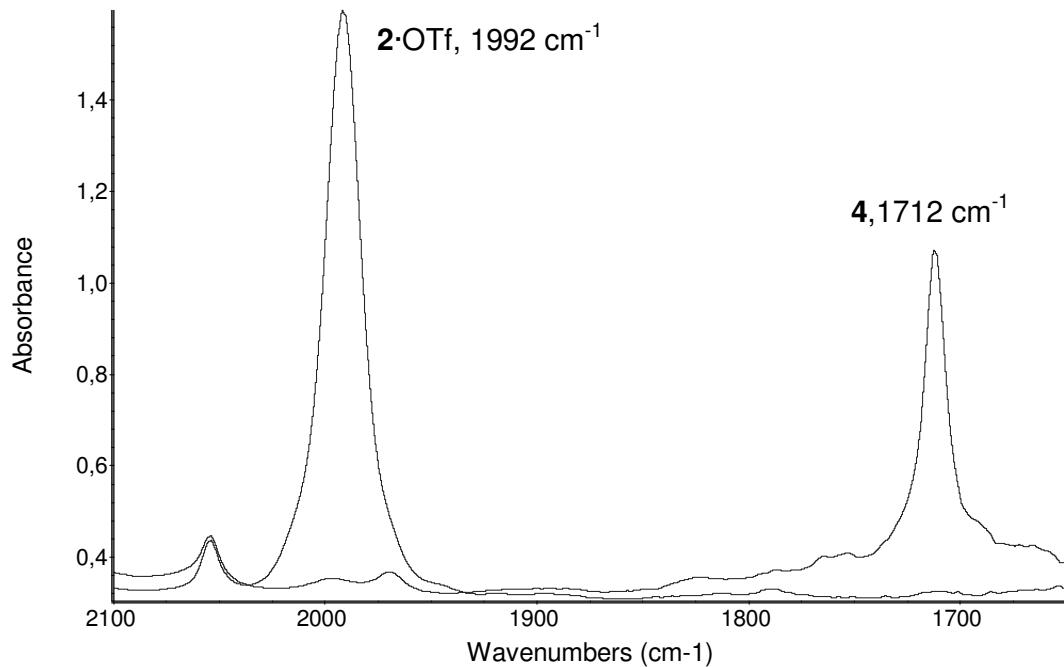


Figure S5. Superimposed FTIR spectra in CH_2Cl_2 solution of $2\cdot\text{OTf} + \text{CH}_3\text{I}$ at the beginning (1992 cm^{-1}) and at the end (1712 cm^{-1}) of the reaction.

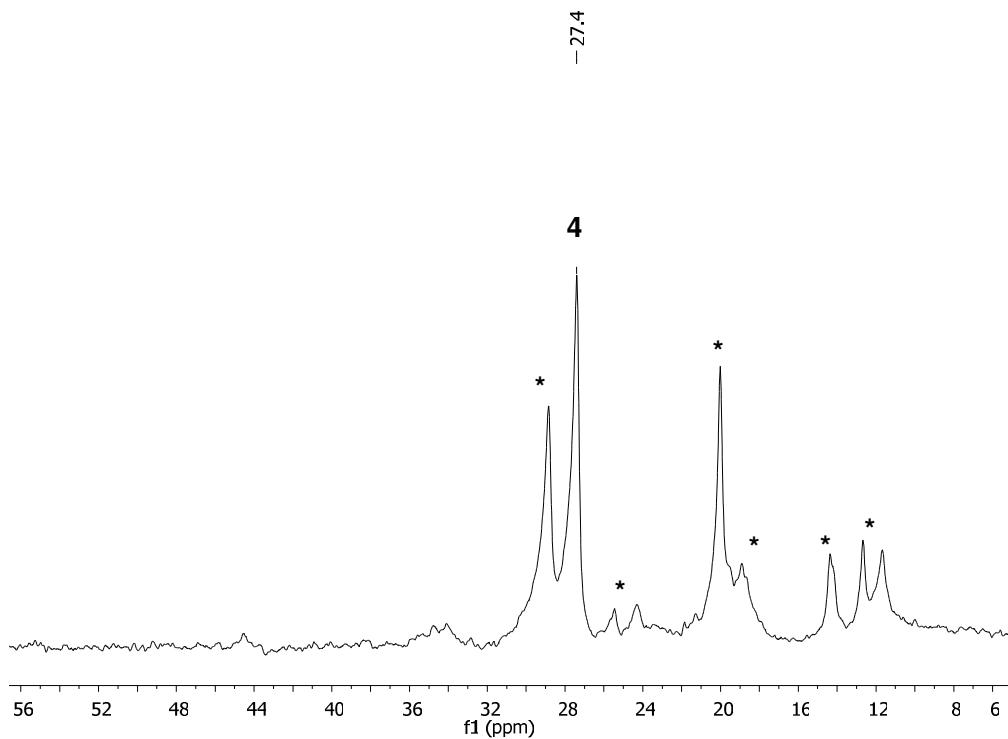
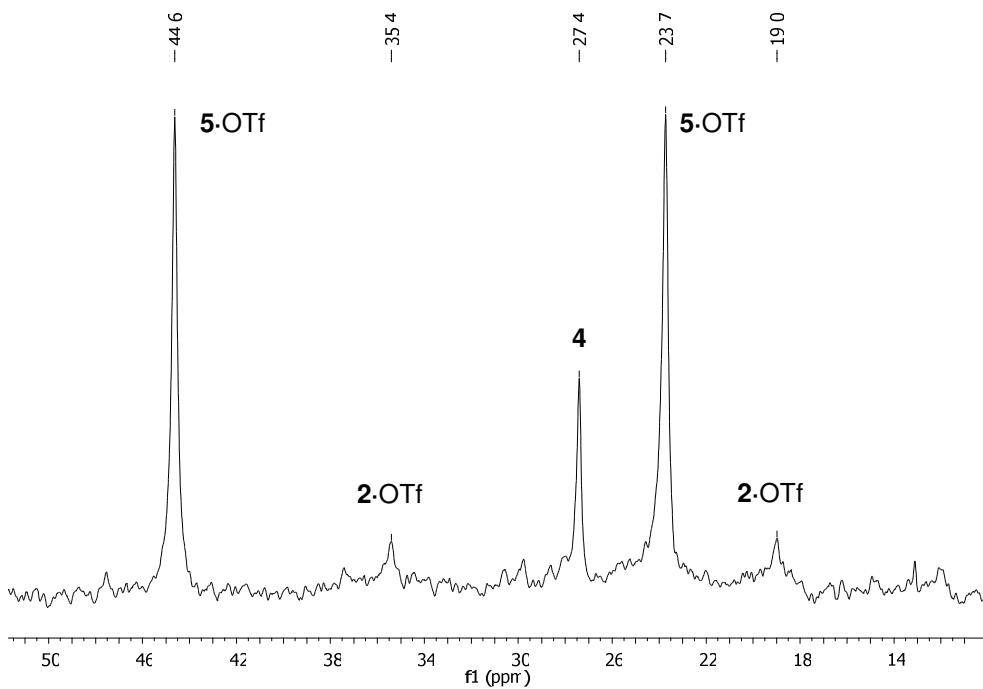


Figure S7. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of the reaction of **2**·OTf + CH₃I in CD₂Cl₂ after 2 days; * corresponds to unidentified decomposition products.

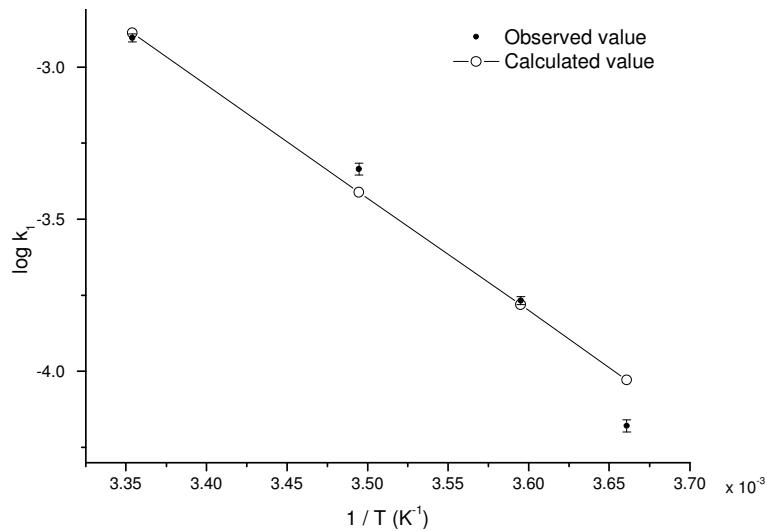


Figure S8. Linearized Eyring plot [$\log(k_i/T)$ vs. $1/T$] for the reaction of **1** with CH_3I

Table S2. Second-order rate constants for the reaction **2**- $\text{PF}_6 + \text{CH}_3\text{I}$ at different temperatures (standard deviations are given in parentheses). The corresponding least-square sample standard deviations (σ) are also reported.

Temperature (K)	k ($\text{M}^{-1} \text{s}^{-1}$)	σ
280.16	$0.64(3) \times 10^{-5}$	1.02×10^{-3}
291.16	$1.71(2) \times 10^{-5}$	3.07×10^{-3}
298.16	$2.56(6) \times 10^{-5}$	7.66×10^{-3}

^a $\sigma = [\sum_i (I_i^o - I_i^c)^2 / (n - m)]^{1/2}$, where I_i^o , I_i^c are the observed and calculated integral values, i is the number of the spectrum, n is the number of observations and m is the number of parameters refined.

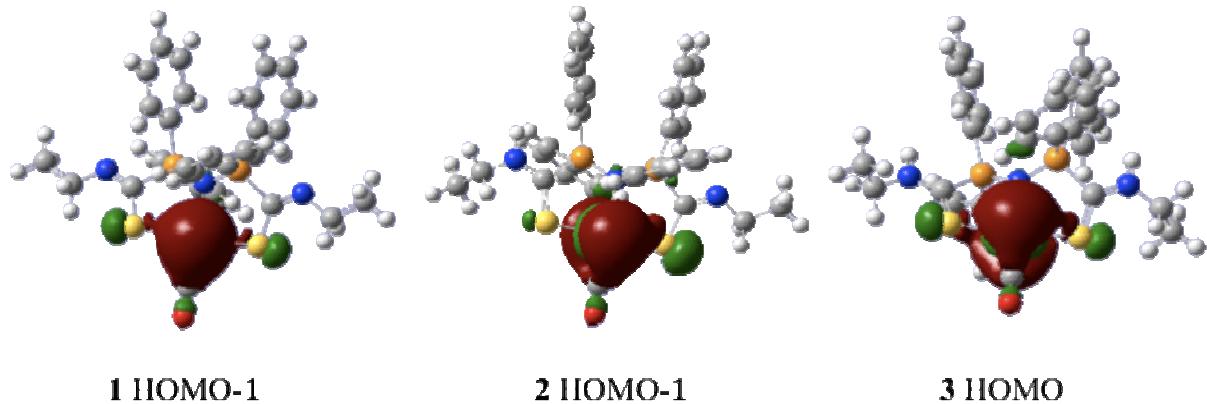


Figure S10. Drawings of the highest energy σ -symmetry MOs for **1**, **2** and **3**.

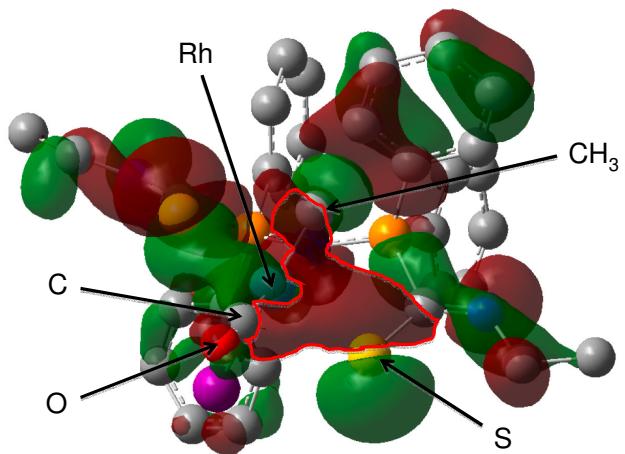


Figure S11. Drawing of the optimized structure of the $[(\text{EtSNS})\text{Rh}(\text{CO})\text{I}\text{CH}_3]$ complex (**A**) with continuous in-phase MO. The MO involves the CH_3 group, the Rh atom, one S atom of the ligand and the C atom of the CO group.