

Molecular Dynamics of Poly(L-lysine) Dendrimers with Naphthalene Disulfonate Caps

Benjamin P. Roberts,¹ Martin J. Scanlon,¹ Guy Y. Krippner² and David K. Chalmers¹

1. Medicinal Chemistry and Drug Action, Monash Institute of Pharmaceutical Sciences. 381 Royal Parade, Parkville, Victoria 3052, Australia.
2. Verva Pharmaceuticals Ltd. PO Box 1069, Grovedale, Victoria 3216, Australia.



Figure S1. (a) The aryl sulfonate group. The atoms for which we created parameters for are shown as **A**, **B** and **C**. (b) The benzhydrylamide group. The atoms for which we created parameters are shown as **D**, **E** and **F**.

Table S1: Symbols used to represent atom types other than the atoms **A–F**.

Symbol	Description
C _{am}	an amide carbonyl carbon atom
C _{ar}	an aromatic carbon atom
N _{am}	an amide nitrogen atom
O _{am}	an amide carbonyl oxygen atom
R	an aliphatic carbon atom
X	any atom

Table S2. Non-bonding parameters for the aromatic sulfonate atoms **A–C**. The charges $q(\text{Imp})$ are those we have implemented in the OPLS-AA force field.

Atom	$q(\text{MMFF94})$ (e)	$q(\text{RESP})$ (e)	$q(\text{Imp})$ (e)	$\sigma(\text{\AA})$	$\varepsilon(\text{kcal mol}^{-1})$
A	-0.009 ^a	-0.238	0.000	3.550	0.070
B	1.459	1.632	1.400	3.550	0.250
C	-0.817	-0.798	-0.800	2.960	0.170

^a -0.008 was used in verification steps due to rounding.

Table S3. Non-bonding parameters for the benzhydrylamide parameters **D–F**.

Atom	$q(\text{IMP})$ (e)	$\sigma(\text{\AA})$	$\varepsilon(\text{kcal mol}^{-1})$
D	0.200	3.500	0.066
E	0.100	2.420	0.015
F	-0.050	3.550	0.070

Table S4: Bond stretching parameters for the sulfonate and benzhydrylamide groups.

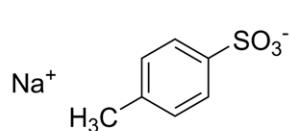
Bond	K_r (kcal mol ⁻¹ Å ⁻²)	r_0 (Å)
A-C_{ar}	469	1.40
A-B	340	1.77
B-C	700	1.44
D-E	340	1.09
D-F	317	1.51
F-C_{ar}	469	1.40

Table S5: Angle bending parameters for the sulfonate and benzhydrylamide groups.

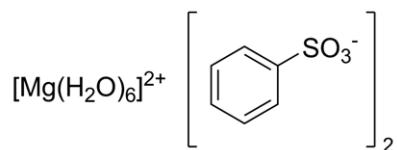
Angle	K_θ (kcal mol ⁻¹ ° ⁻²)	θ_0 (°)
C_{ar}-A-B	85.0	119.40
A-B-C	74.0	107.20
E-D-N_{am}	35.0	109.50
F-D-E	35.0	109.50
F-D-N_{am}	80.0	111.20
C_{ar}-F-D	70.0	120.00

Table S6: Fourier coefficients (kcal mol⁻¹) for torsions in the sulfonate and benzhydrylamide groups.

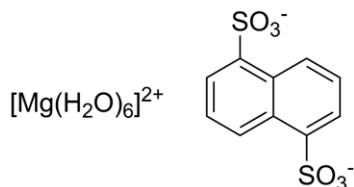
Torsion	<i>V</i> ₁	<i>V</i> ₂	<i>V</i> ₃	<i>V</i> ₄
X-C _{ar} -C _{ar} - A	0.000	7.250	0.000	0.000
X-C _{ar} - A - B	0.000	7.250	0.000	0.000
C _{ar} - A - B -C	0.000	0.000	0.000	0.000
C _{ar} -F-D-F	0.000	0.000	0.000	0.000
C _{ar} -F-D-E	0.000	0.000	0.000	0.000
C _{ar} -F-D-N _{am}	0.000	0.000	0.000	0.000
X-C _{ar} -F-D	0.000	7.250	0.000	0.000
X-C _{ar} -C _{ar} -F	0.000	7.250	0.000	0.000
H-N _{am} -D-E	0.000	0.000	0.000	0.000
H-N _{am} -D-F	0.000	0.000	0.000	0.000
C _{am} -N _{am} -D-E	0.000	0.000	-0.139	0.000
C _{am} -N _{am} -D-F	-2.365	0.912	-0.850	0.000
O _{am} -C _{am} -N _{am} -D	0.000	6.089	0.000	0.000
R-C _{am} -N _{am} -D	2.300	6.089	0.000	0.000



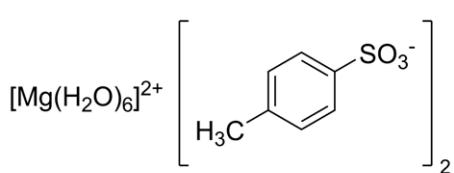
1



2



3



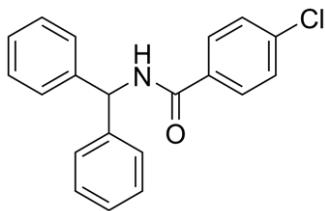
4

Figure S2: Compounds used to verify aromatic sulfonate parameters.

Table 1: Verification of aromatic sulfonate parameters. Comparison of unit cell dimensions and root mean squared deviations of the compounds **1–4** from their reference geometries using MMFF94 charges, RESP charges and implemented charges.

Compound	Parameter set	Unit Cell Dimensions ^a					RMSD
		<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	β (°)	<i>V</i> (Å ³)	
1	<i>Reference structure</i>	15.4030	6.1090	8.5360	92.7400	802.293	
	Implemented	14.8554	5.9027	8.1009	93.5165	709.006	0.2288
	(% diff.)	3.56%	3.38%	5.10%	0.837%	11.63%	
	MMFF94	14.8685	5.8982	8.0961	93.4950	708.686	0.2273
	(% diff.)	3.47%	3.45%	5.15%	0.814%	11.67%	
	RESP	14.5522	6.0190	8.1653	95.3347	712.098	0.2959
2	(% diff.)	5.52%	1.47%	4.34%	2.80%	11.24%	
	<i>Reference structure</i>	22.6000	6.3200	6.9400	93.6000	989.298	
	Implemented	22.0598	6.0002	6.5992	92.1517	872.875	0.2366
	(% diff.)	2.39%	5.06%	4.91%	1.55%	11.77%	
	MMFF94	21.8889	5.9553	6.6970	90.7697	872.908	0.6411
	(% diff.)	3.15%	5.77%	3.50%	3.02%	11.76%	
3	RESP	21.7133	6.0638	6.6813	90.8101	879.606	0.3315
	(% diff.)	3.92%	4.05%	3.73%	2.98%	11.09%	
	<i>Reference structure</i>	13.1955	6.6981	9.6616	92.3040	853.248	
	Implemented	12.4912	6.3055	10.9927	84.0788	861.202	0.5710
	(% diff.)	5.34%	5.86%	13.8%	8.91%	0.9322%	
	MMFF94	12.4959	6.2957	11.0051	84.0200	861.065	0.5753
4	(% diff.)	5.30%	6.01%	13.91%	8.97%	0.9161%	
	RESP	11.4532	6.7637	11.5757	83.0403	890.116	0.9697
	(% diff.)	13.2%	0.979%	19.8%	10.0%	4.321%	
	<i>Reference structure</i>	6.9481	6.2809	25.3600	91.7720	1106.19	
	Implemented	6.5821	5.9996	24.7537	89.7213	977.511	0.2648
	(% diff.)	5.27%	4.48%	2.39%	2.23%	11.63%	
4	MMFF94	6.5839	5.9888	24.7482	89.7266	975.802	0.2665
	(% diff.)	5.24%	4.65%	2.41%	2.23%	11.79%	
	RESP	6.6810	6.0667	24.4491	87.7918	990.226	0.4017
	(% diff.)	3.84%	3.41%	3.59%	4.34%	10.48%	

^a The internal angles α and γ remain constant at 90°.



5

Figure S3: The compound used to verify parameters for the benzhydrylamide group.

Table 2: Verification of parameters for the benzhydrylamide group. Alteration of unit cell dimensions and root mean squared deviation of the compound **5** from its reference geometry using implemented charges.

Compound	Parameter set	Unit Cell Dimensions ^a					RMSD
		<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	β	<i>V</i> (Å ³)	
5	<i>Reference structure</i>	26.3540	5.1630	23.6540	91.2000	3217.79	
	Implemented (% diff.)	26.1000 0.964%	4.9710 3.72%	23.9485 1.25%	89.2608 2.13%	3106.89 3.45%	0.2509

^a The internal angles α and γ remain constant at 90°.

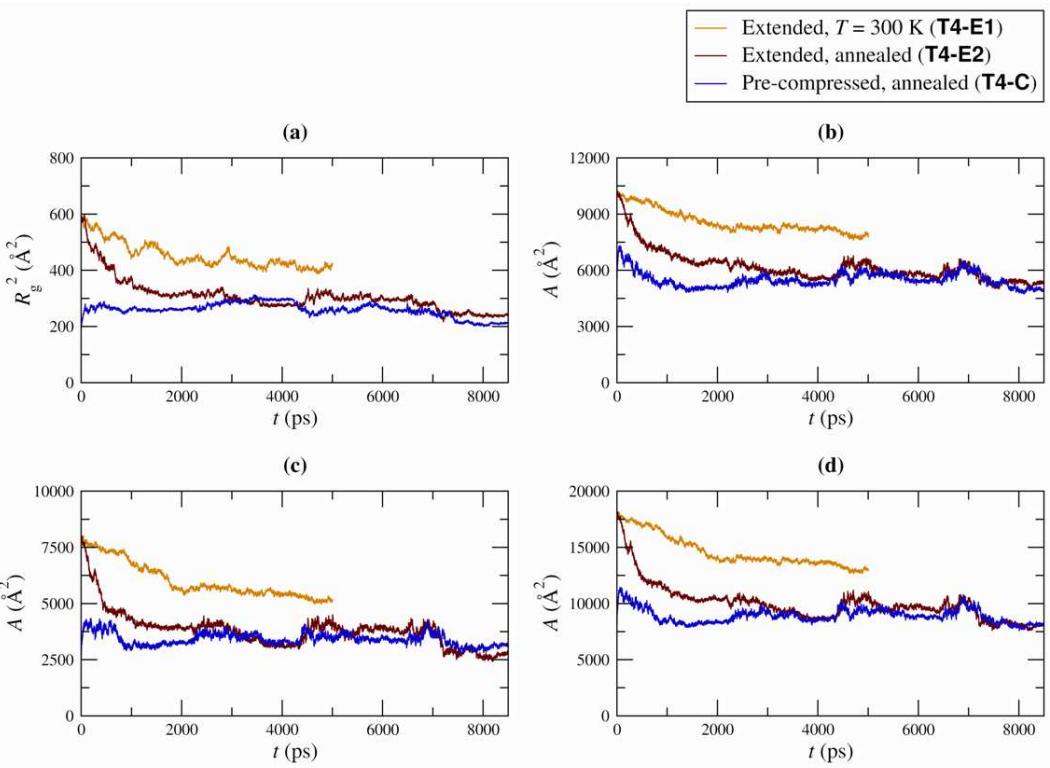


Figure S4: Comparison of dendrimer sizes over three different equilibration protocols. (a) Mean-squared radius of gyration. (b) Polar solvent-accessible surface area. (c) Non-polar solvent-accessible surface area. (d) Total solvent-accessible surface area.

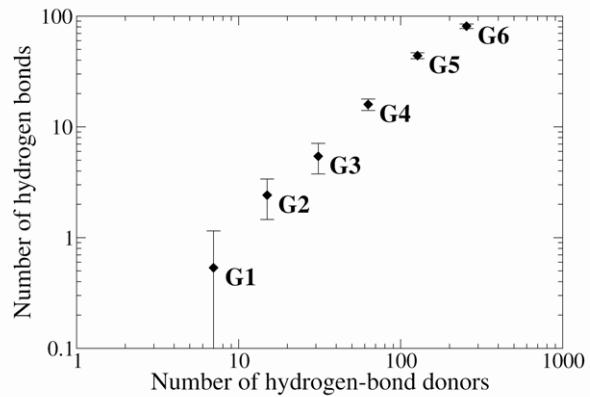


Figure S5: Number of hydrogen bonds in the simulated dendrimers. Error bars indicate one standard deviation each side of the mean.

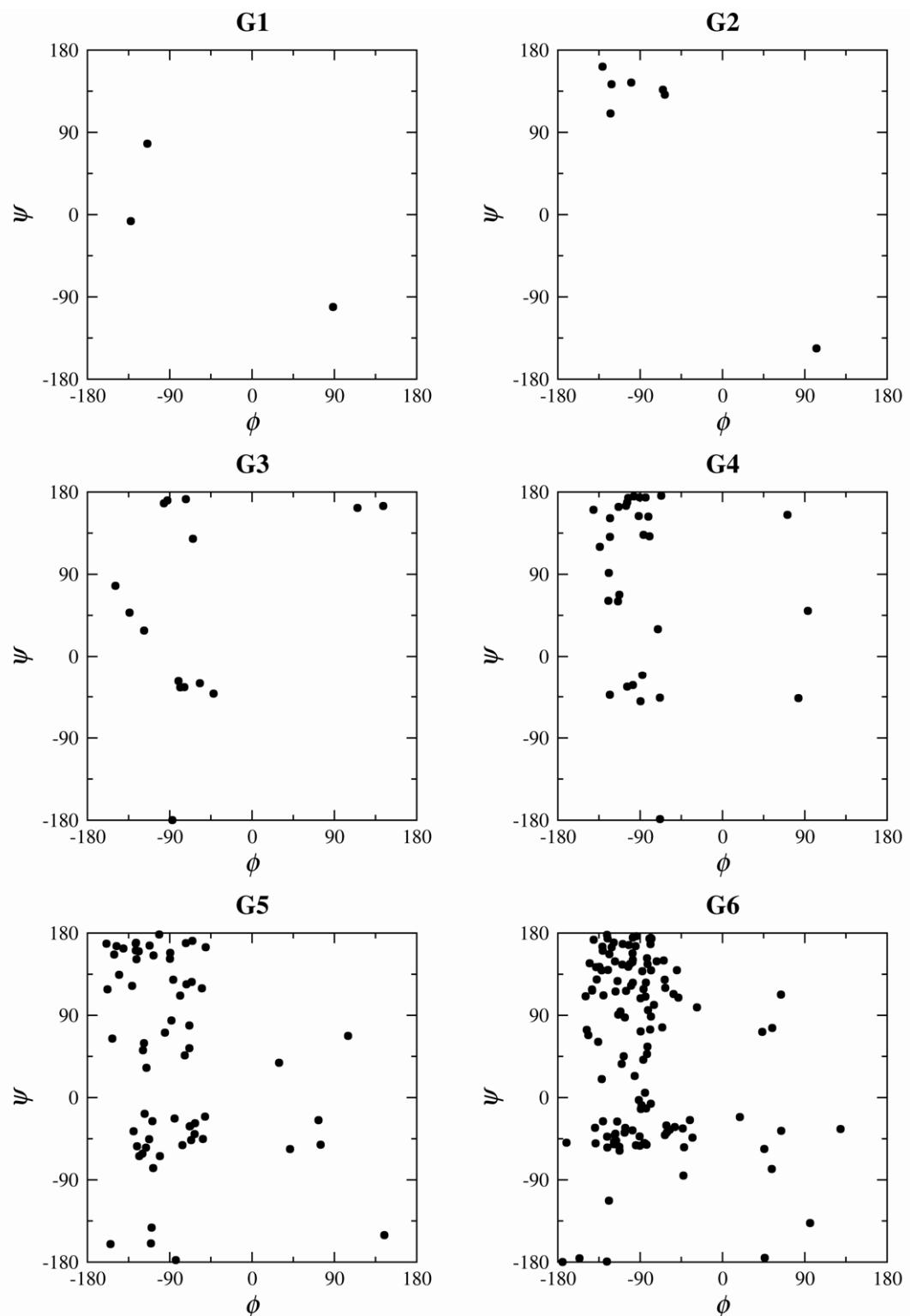


Figure S6: Ramachandran plots of dendrimers G1 to G6 at the final frames of the simulations.

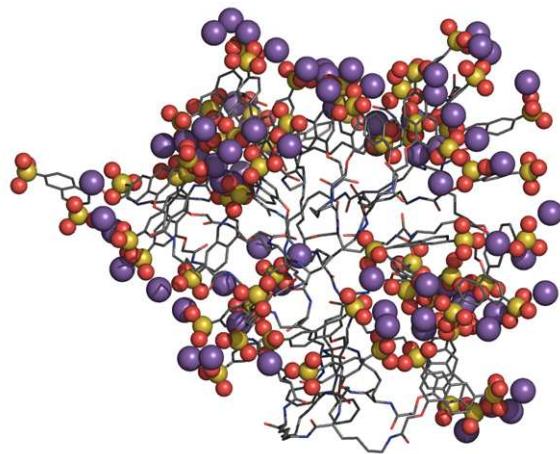


Figure S7: The final frame of the simulation of G4, showing all sodium ions within 4 Å of any dendrimer atom (except hydrogen atoms). Sodium ions are shown as purple spheres. Sulfonate groups are shown as yellow spheres (sulphur) and red spheres (oxygen).

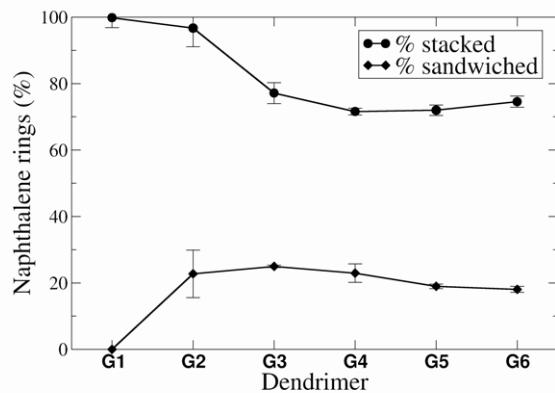


Figure S8: Average extent of stacking and sandwiching in the dendrimers **G1** to **G6**. Error bars are one standard deviation either side of the mean. The lines between points are shown to guide the eye.