

Characterization of β -B-Agostic Isomers in Zirconocene Amidoborane Complexes

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Experimental Section

General considerations. All operations were carried out with careful exclusion of air and moisture using standard Schlenk and glove box techniques. The solvents were dried and deoxygenated prior to use. Starting materials were purchased from commercial suppliers and used as received. All NMR spectra were run on either a Bruker Avance DRX-400 or a DRY-400 instrument and chemical shifts are reported in δ units (ppm) using the solvent as an internal reference: $\text{C}_6\text{D}_5\text{H}$ (7.15 ppm, ^1H) and C_6D_6 (128.39, ^{13}C); THF- d_7 (3.58 ppm, ^1H) and THF- d_8 (67.57 ppm, ^{13}C); toluene- d_7 (2.09 ppm, ^1H), and toluene- d_8 (20.4 ppm, ^{13}C). $\text{BF}_3\cdot\text{Et}_2\text{O}$ (0 ppm) was used as reference for the ^{11}B NMR measurements. The infrared spectra were recorded using KBr pellets on a Nicolet Nexus 470 CSI optics FT-IR instrument while the Raman spectra were recorded using a Bruker Vortex 70 RT-DLaTGS RAM II instrument. The powder X-ray diffraction pattern was recorded on a Rigaku multiflex diffractometer.

Synthesis of $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**

THF (30 mL) was condensed over a mixture of ammonia borane (0.302 g, 9.8 mmol) and Cp_2ZrCl_2 (1.43 g, 4.9 mmol). The reactants dissolved upon warming up to room temperature and the solution was subsequently cooled to $-78\text{ }^\circ\text{C}$. $n\text{BuLi}$ (1.6 M in hexanes, 6.07 mL, 9.7 mmol) was added, the mixture was stirred at $-78\text{ }^\circ\text{C}$ for 2 hours and then allowed to warm to room temperature. The solvent was removed in vacuum, benzene (25 mL) was added to the remaining solid and the slurry was filtered. $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$ (0.98 g, 3.5 mmol, 71%) was isolated upon removal of benzene in vacuum. X-ray quality single crystals were obtained by cooling a saturated toluene solution to $-35\text{ }^\circ\text{C}$. The solid compound could be handled in the air for short periods of time, but showed signs of hydrolysis with formation of ammonia borane after exposure to the air for a few hours.

EA: Calcd.: C 47.60, H 6.41, N, 5.55. Found: C 47.49, H 6.45, N 5.42.

IR (KBr): $\nu(\text{cm}^{-1}) = 3377$ (w, NH), 3305 (m, NH), 3081 (w, CH), 2376 (s, BH_{term}), 2304 (m, BH_{term}), 1899 (w, $\text{BH}_{\text{bridging}}$), 1824 (m, $\text{BH}_{\text{bridging}}$), 1553 (s, ZrH).

Raman: $\nu(\text{cm}^{-1}) = 3115$ (m, CH), 3085 (m, CH), 2380 (m, BH_{term}), 2308 (w, BH_{term}).

^1H NMR (C_6D_6 , 400 MHz, $20\text{ }^\circ\text{C}$): **Isomer 1** (~50%) δ (ppm) = 0.15 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 91.0$ Hz), 0.49, (s, br, 2H, NH_2), 3.58 (s, 1H, ZrH), 5.43 (s, 10H, C_5H_5); **Isomer 2** (~50%) δ (ppm) =

0.10, (s, br, 2H, NH_2), 0.15 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 91.0$ Hz), 3.61 (s, 1H, ZrH), 5.29 (s, 10H, C_5H_5).

^1H NMR (THF- d_8 , 400 MHz, 20 °C): **Isomer 1** (~66%) δ (ppm) = -0.44 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 88.0$ Hz), 1.41 (s, br, 2H, NH_2), 3.12 (s, 1H, ZrH), 5.71 (s, 10H, C_5H_5); **Isomer 2** (~33%) δ (ppm) = -0.44 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 88.0$ Hz), 1.41 (s, br, 2H, NH_2), 3.49 (s, 1H, ZrH), 5.76 (s, 10H, C_5H_5).

^1H NMR (toluene- d_8 , 400 MHz, 20 °C): **Isomer 1** (identified as **1a**) (~50%) δ (ppm) = -0.34 (q, br, 3H, BH_3 , 0.44, (s, br, 2H, NH_2), 3.51 (s, 1H, ZrH), 5.41 (s, 10H, C_5H_5); **Isomer 2** (identified as **1b**) (~50%) δ (ppm) = 0.06, (s, br, 2H, NH_2), 0.06 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 91.0$ Hz), 3.51 (s, 1H, ZrH), 5.28 (s, 10H, C_5H_5).

^1H NMR (toluene- d_8 , 400 MHz, 60 °C): **Isomer 1** (identified as **1a**) (~50%) δ (ppm) = -0.38 (q, br, 3H, BH_3 , 0.45, (s, br, 2H, NH_2), 3.56 (s, 1H, ZrH), 5.43 (s, 10H, C_5H_5); **Isomer 2** (identified as **1b**) (~50%) δ (ppm) = 0.01 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 91.0$ Hz), 0.09, (s, br, 2H, NH_2), 3.53 (s, 1H, ZrH), 5.31 (s, 10H, C_5H_5).

^1H NMR (toluene- d_8 , 400 MHz, -60 °C): **Isomer 1** (identified as **1a**) (~50%) δ (ppm) = -2.88 (q, br, 1H, $\text{B}-\mu\text{H}-\text{Zr}$), 0.53, (s, br, 2H, NH_2), 1.11 (q, br, 3H, BH_2 , 3.49 (s, 1H, ZrH), 5.36 (s, 10H, C_5H_5); **Isomer 2** (identified as **1b**) (~50%) δ (ppm) = 0.16, (s, br, 2H, NH_2), 0.26 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 91.0$ Hz), 3.40 (s, 1H, ZrH), 5.21 (s, 10H, C_5H_5).

^{11}B NMR (C_6D_6 , 128.4 MHz, 20 °C): **Isomer 1** δ (ppm) = -33.9 (q, $^1J_{\text{BH}} = 91.0$ Hz); **Isomer 2** δ (ppm) = -33.5 (q, $^1J_{\text{BH}} = 91.0$ Hz).

^{11}B NMR (THF- d_8 , 128.4 MHz, 20 °C): **Isomer 1** δ (ppm) = -34.0 (q, $^1J_{\text{BH}} = 88.0$ Hz); **Isomer 2** δ (ppm) = -34.3 (q, $^1J_{\text{BH}} = 88.0$ Hz).

^{11}B NMR (toluene- d_8 , 128.4 MHz, 20 °C): **Isomer 1** δ (ppm) = -33.5 (q, $^1J_{\text{BH}} = 88.0$ Hz); **Isomer 2** δ (ppm) = -33.9 (q, $^1J_{\text{BH}} = 88.0$ Hz).

^{13}C NMR (C_6D_6 , 100.6 MHz, 20 °C): **Isomer 1** δ 103.88 (C_5H_5); **Isomer 2** δ 103.91 (C_5H_5).

^{13}C NMR (THF- d_8 , 100.6 MHz, 20 °C): **Isomer 1** δ (ppm) = 104.25 (C_5H_5); **Isomer 2** δ (ppm) = 104.16 (C_5H_5).

^{13}C NMR (toluene- d_8 , 100.6 MHz, 20 °C): **Isomer 1** δ (ppm) = 103.84 (C_5H_5); **Isomer 2** δ (ppm) = 103.87 (C_5H_5).

Synthesis of $\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**

THF (30 mL) was condensed over a mixture of ammonia borane (0.1 g, 3.2 mmol) and $\text{Cp}^*_2\text{ZrCl}_2$ (0.7 g, 1.6 mmol). The reactants dissolved upon warming up to room temperature and the solution was subsequently cooled to -78°C . $n\text{BuLi}$ (1.6 M in hexanes, 2.02 mL, 3.2 mmol) was added, the mixture was stirred at -78°C for 2 hours and then allowed to warm to room temperature. The solvent was removed in vacuum, benzene (25 mL) was added to the remaining solid and the slurry was filtered. $\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$ (0.52 g, 2.06 mmol, 64%) was isolated upon removal of benzene in vacuum. X-ray quality single crystals were obtained by slow evaporation of a toluene solution.

IR (KBr): $\nu(\text{cm}^{-1}) = 3430$ (w, NH), 3366 (w, NH), 2903 (m, CH), 2390 (m, BH_{term}), 2318 (m, BH_{term}), 1892 (w, $\text{BH}_{\text{bridging}}$), 1819 (w, $\text{BH}_{\text{bridging}}$), 1531 (m ZrH).

Raman: $\nu(\text{cm}^{-1}) = 2904$ (vs, CH), 2397 (w, BH_{term}), 2321 (w, BH_{term}).

^1H NMR (C_6D_6 , 400 MHz, 20°C): **Isomer 1** (~80): ^1H NMR δ (ppm) = -0.10, (s, br, 2H, NH_2), -0.05 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 90$ Hz), 1.76 (s, 30H, $\text{C}_5(\text{CH}_3)_5$), 4.15 (s, 1H, ZrH); **Isomer 2** (~20%) δ (ppm) = -0.05 (q, br, 3H, BH_3 , $^1J_{\text{BH}} = 90$ Hz), 0.40, (s, br, 2H, NH_2), 1.84 (s, 30H, $\text{C}_5(\text{CH}_3)_5$), 4.08 (s, 1H, ZrH).

^{11}B NMR (C_6D_6 , 128.4 MHz, 20°C): **Isomer 1** δ (ppm) = -29.6 (q, $^1J_{\text{BH}} = 90$ Hz); **Isomer 2** δ (ppm) = -31.9 (q, $^1J_{\text{BH}} = 90$ Hz).

^{13}C NMR (C_6D_6 , 100.6 MHz, 20°C): **Isomer 1** δ (ppm) = 11.76 ($\text{C}_5(\underline{\text{C}}\text{H}_3)_5$), 114.06 ($\underline{\text{C}}_5(\text{CH}_3)_5$); **Isomer 2** δ (ppm) = 12.26 ($\text{C}_5(\underline{\text{C}}\text{H}_3)_5$), 115.09 ($\underline{\text{C}}_5(\text{CH}_3)_5$).

Synthesis of $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**

THF (30 mL) was condensed over a mixture of ammonia borane (0.3 g, 9.7 mmol) and Cp_2ZrCl_2 (2.84 g, 9.7 mmol). The reactants dissolved upon warming up to room temperature and the solution was subsequently cooled to -78°C . $n\text{BuLi}$ (1.6 M in hexanes, 6.07 mL, 9.7 mmol) was added, the mixture was stirred at -78°C for 2 hours and then allowed to warm to room temperature. The solvent was removed in vacuum, benzene (25 mL) was added to the remaining solid and the slurry was filtered. $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$ (1.48 g, 5.9 mmol, 60%) was isolated upon removal of benzene in vacuum. X-ray quality single crystals were obtained by cooling a saturated toluene solution to -35°C .

EA: Calcd.: C 41.88, H 5.28, N, 4.89. Found: C 41.92, H 5.34, N 3.95.

IR (KBr): $\nu(\text{cm}^{-1})$ = 3384 (m, NH), 3280 (s, NH), 3097 (m, CH), 2393 (s, BH_{term}), 2320 (m, BH_{term}), 1868 (m, BH_{bridging}), 1792 (m, BH_{bridging}).

Raman: $\nu(\text{cm}^{-1})$ = 3115 (m, CH), 3091 (m, CH), 2395 (s, BH_{term}), 2323 (m, BH_{term}).

¹H NMR (C₆D₆, 400 MHz, 20 °C): **Isomer 1** (~80%) δ (ppm) = 0.36 (q, br, 3H, BH₃, ¹J_{BH} = 93.7 Hz), 1.46, (s, br, 2H, NH₂), 5.61 (s, 10H, C₅H₅); **Isomer 2** (~20%) δ (ppm) = 0.36 (q, br, 3H, BH₃, ¹J_{BH} = 93.7 Hz), 1.46, (s, br, 2H, NH₂), 6.01 (s, 10H, C₅H₅).

¹H NMR (THF-d₈, 400 MHz, 20 °C): **Isomer 1** (~85%) ¹H NMR δ (ppm) = 0.20 (q, br, 3H, BH₃, ¹J_{BH} = 94.4 Hz), 2.76, (s, br, 2H, NH₂), 6.08 (s, 10H, C₅H₅); **Isomer 2** (~15%) δ (ppm) = 0.20 (q, br, 3H, BH₃, ¹J_{BH} = 94.4 Hz), 2.76, (s, br, 2H, NH₂), 6.37 (s, 10H, C₅H₅);

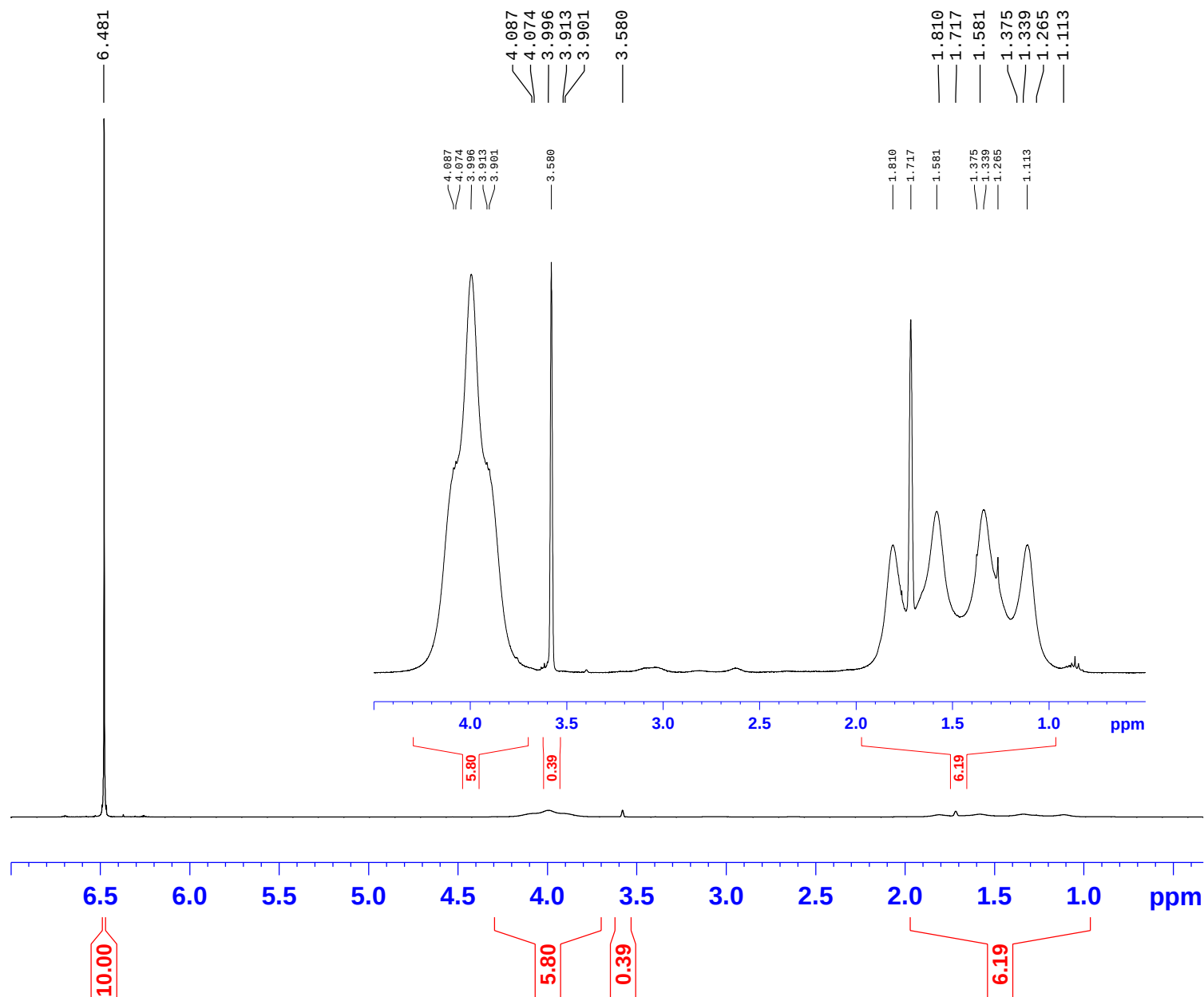
¹¹B NMR (C₆D₆, 128.4 MHz, 20 °C): **Isomer 1** δ (ppm) = -22.6 (q, ¹J_{BH} = 93.7 Hz); **Isomer 2** δ (ppm) = -21.2 (q, ¹J_{BH} = 93.7 Hz).

¹¹B NMR (THF-d₈, 128.4 MHz, 20 °C): **Isomer 1** δ (ppm) = -23.0 (q, ¹J_{BH} = 94.4 Hz); **Isomer 2** δ (ppm) = -22.4 (q, ¹J_{BH} = 94.4 Hz).

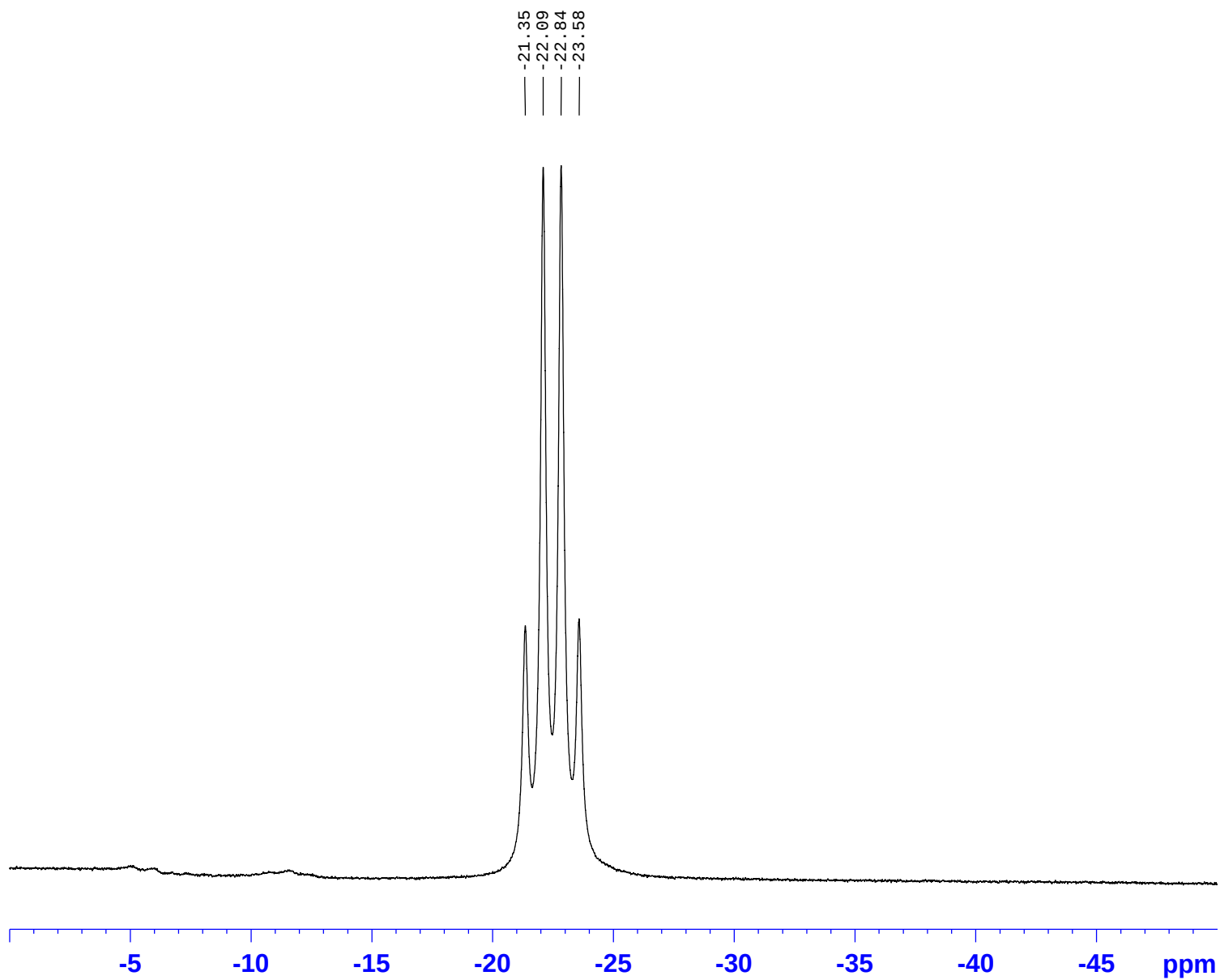
¹³C NMR (C₆D₆, 100.6 MHz, 20 °C): **Isomer 1** δ (ppm) = 111.72 (C₅H₅); **Isomer 2** δ (ppm) = 114.50 (C₅H₅).

¹³C NMR (THF-d₈, 100.6 MHz, 20 °C): **Isomer 1** δ (ppm) = 112.20 (C₅H₅); δ (ppm) = 115.07 (C₅H₅).

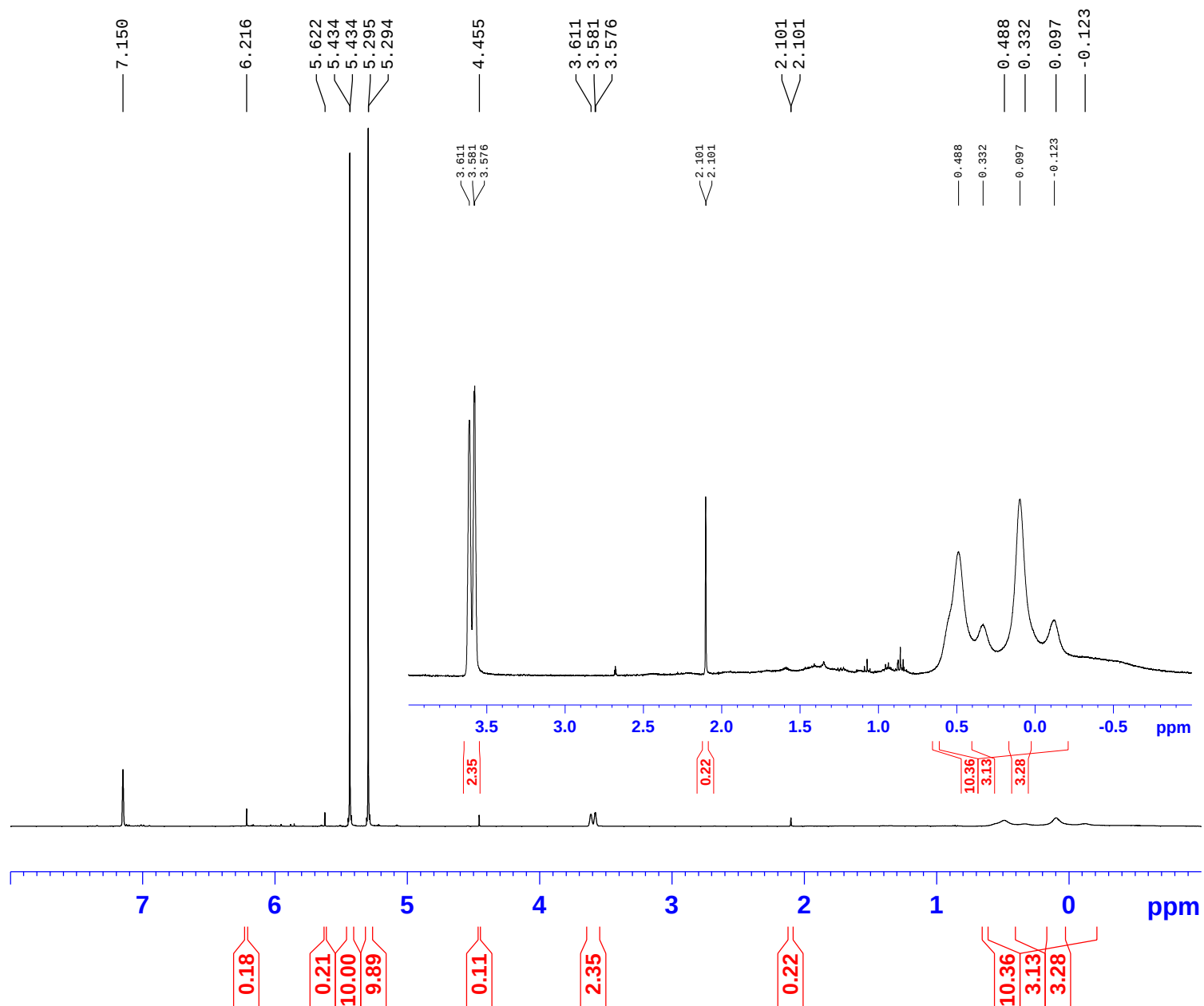
$\text{Cp}_2\text{ZrCl}_2 - \text{NH}_3\text{BH}_3$ 1 :1 mixture, ^1H NMR in THF-d_8 , 20 °C



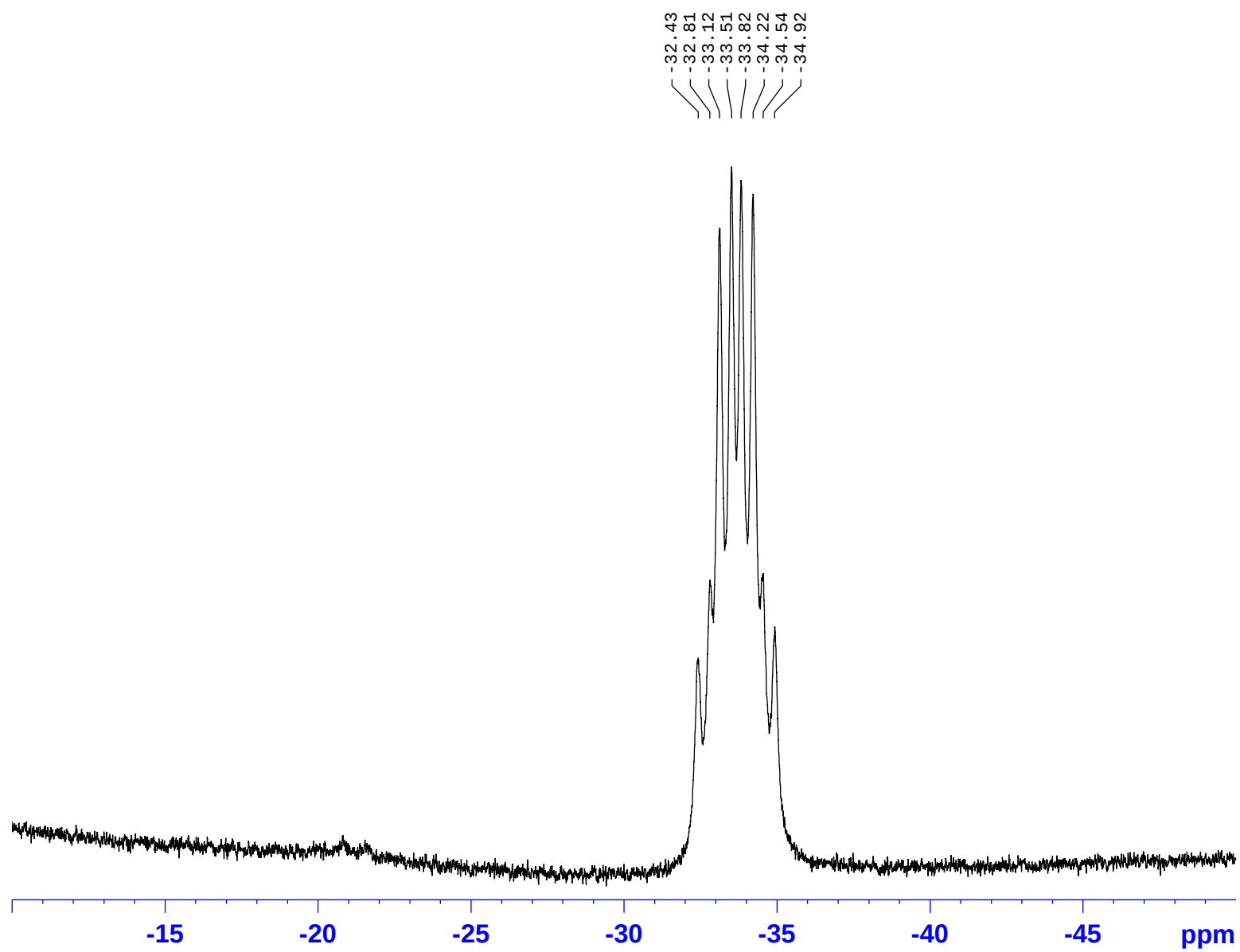
$\text{Cp}_2\text{ZrCl}_2 - \text{NH}_3\text{BH}_3$ 1 :1 mixture, ^{11}B NMR in THF-d_8 , 20 °C



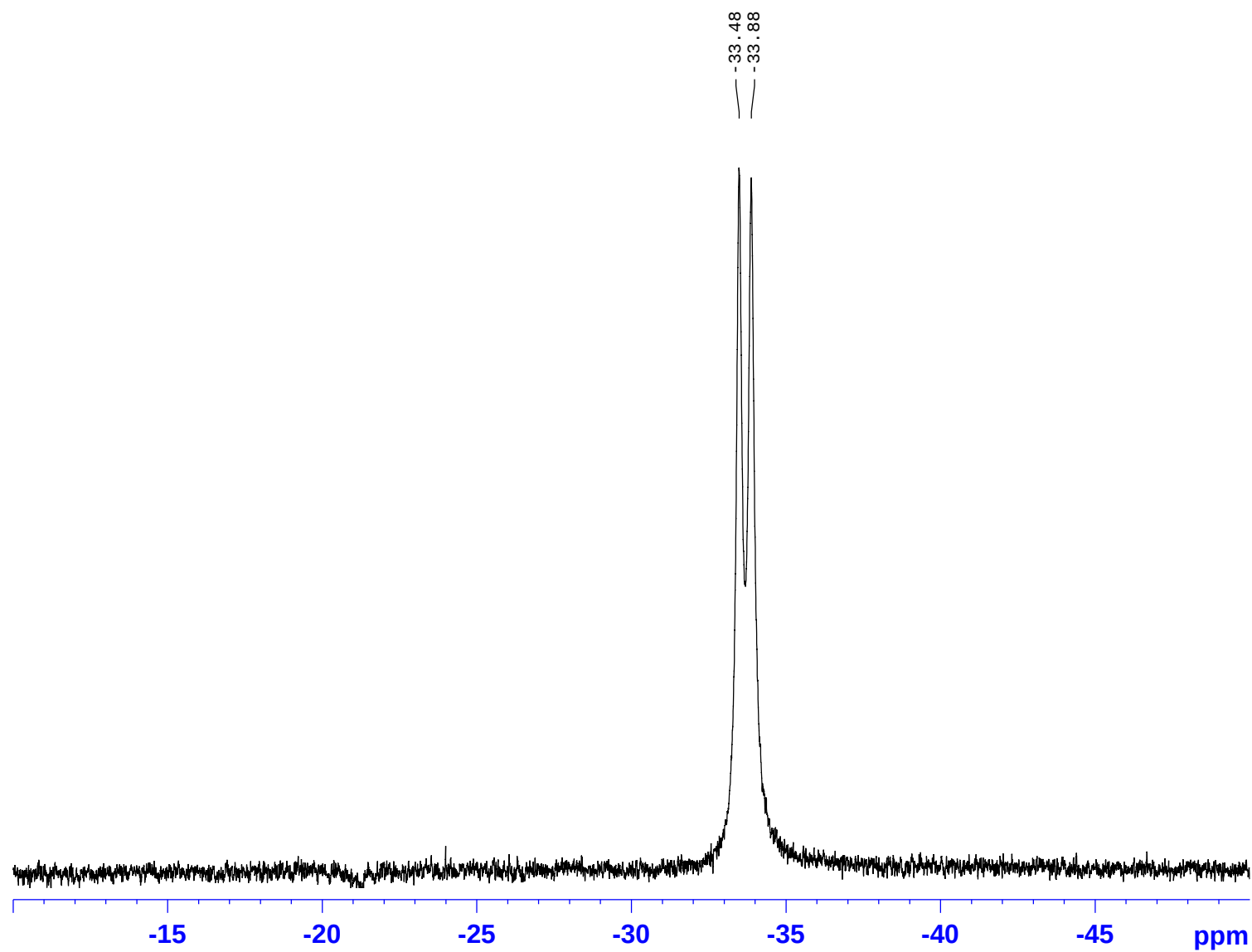
15 mg $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, ^1H NMR in C_6D_6



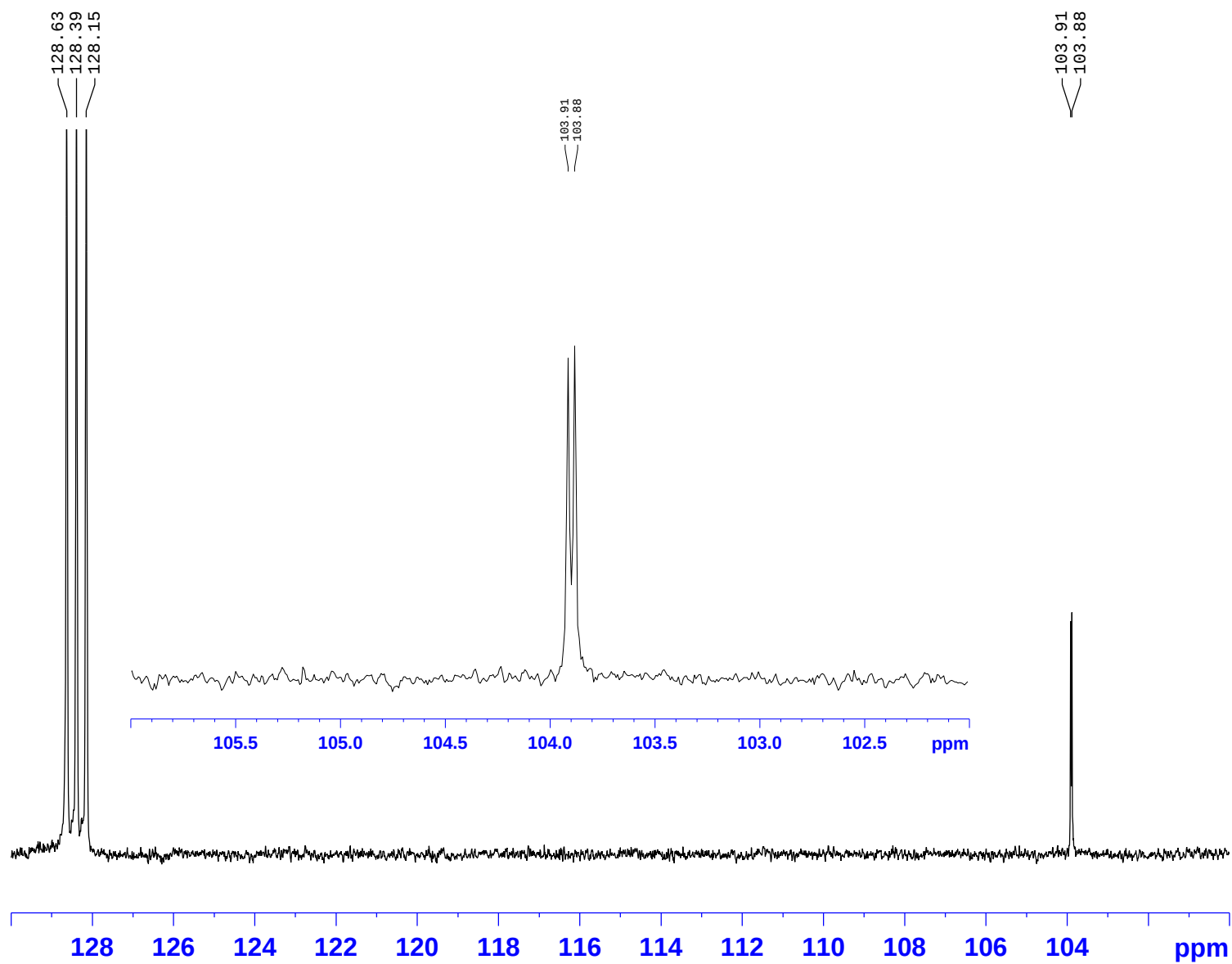
15 mg $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, ^{11}B NMR in C_6D_6



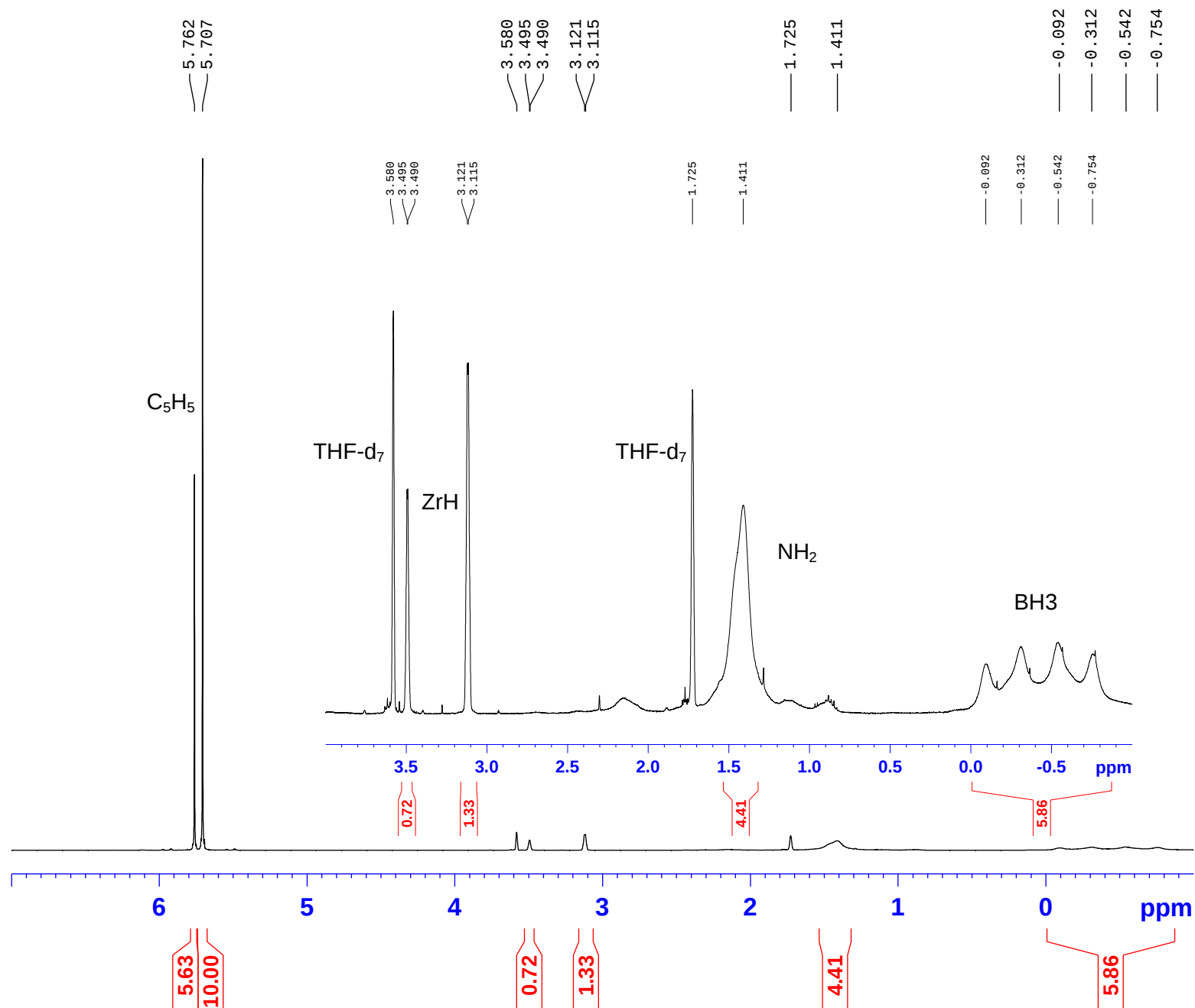
15 mg $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^{11}\text{B}\{^1\text{H}\}$ NMR in C_6D_6



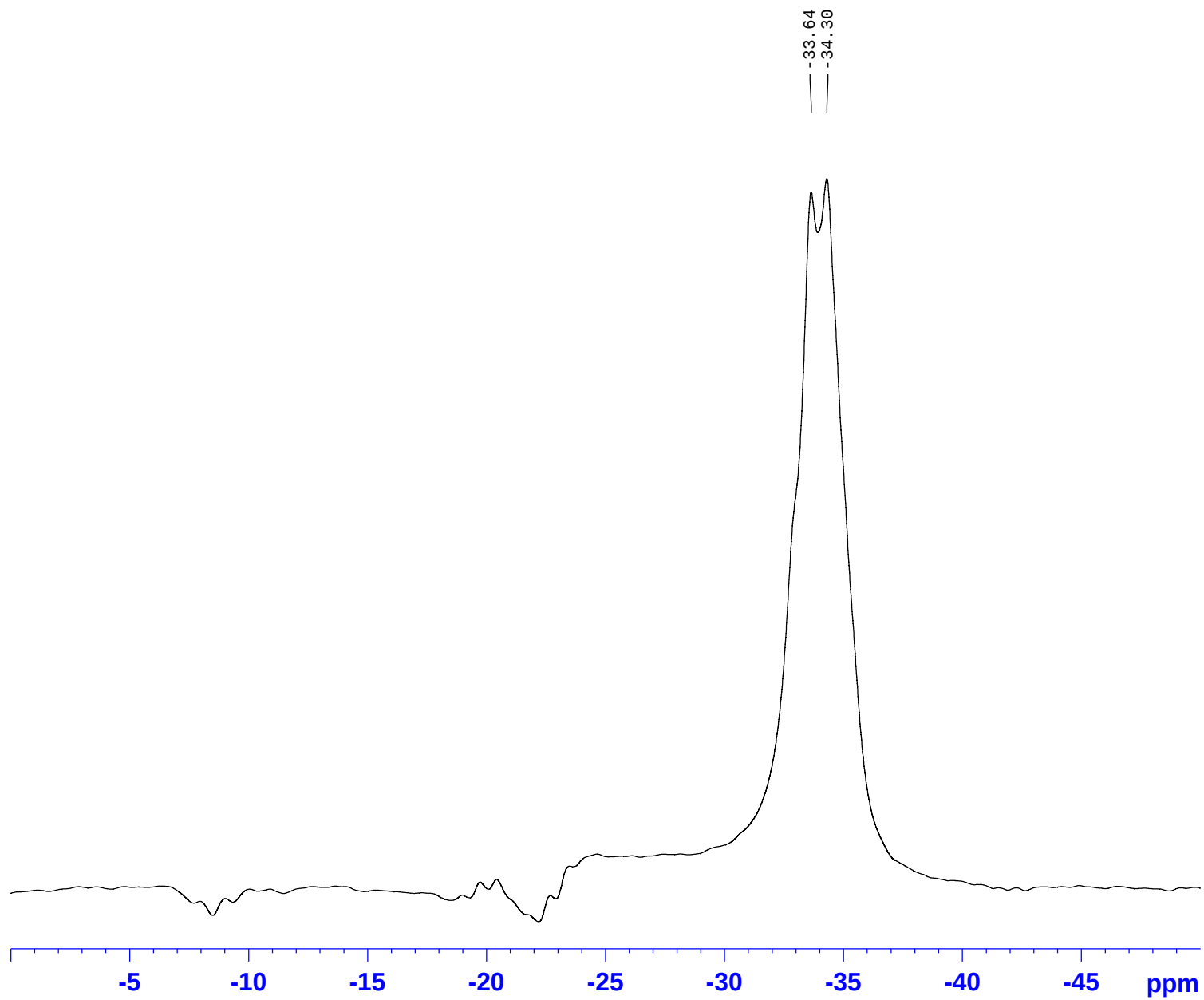
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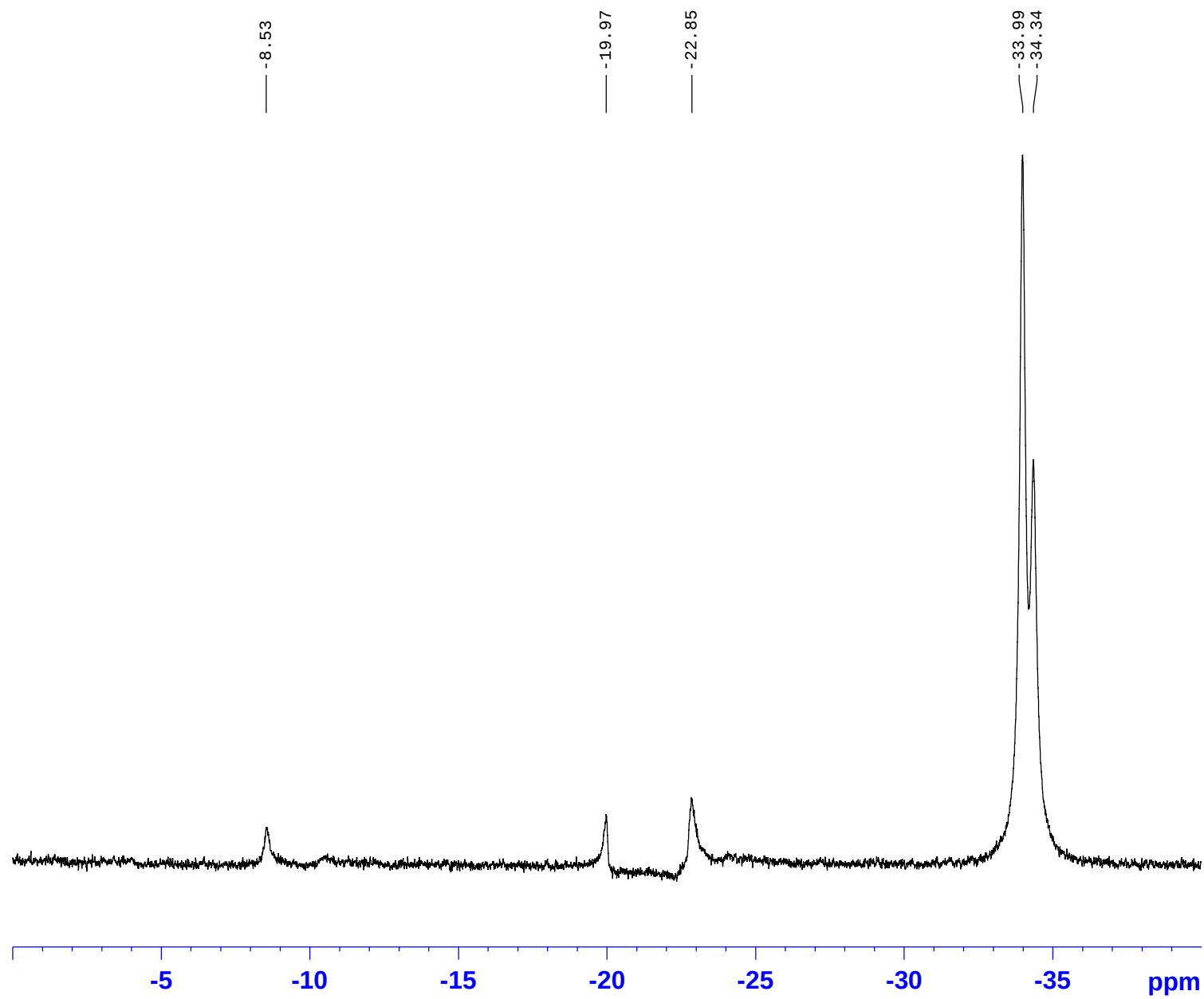
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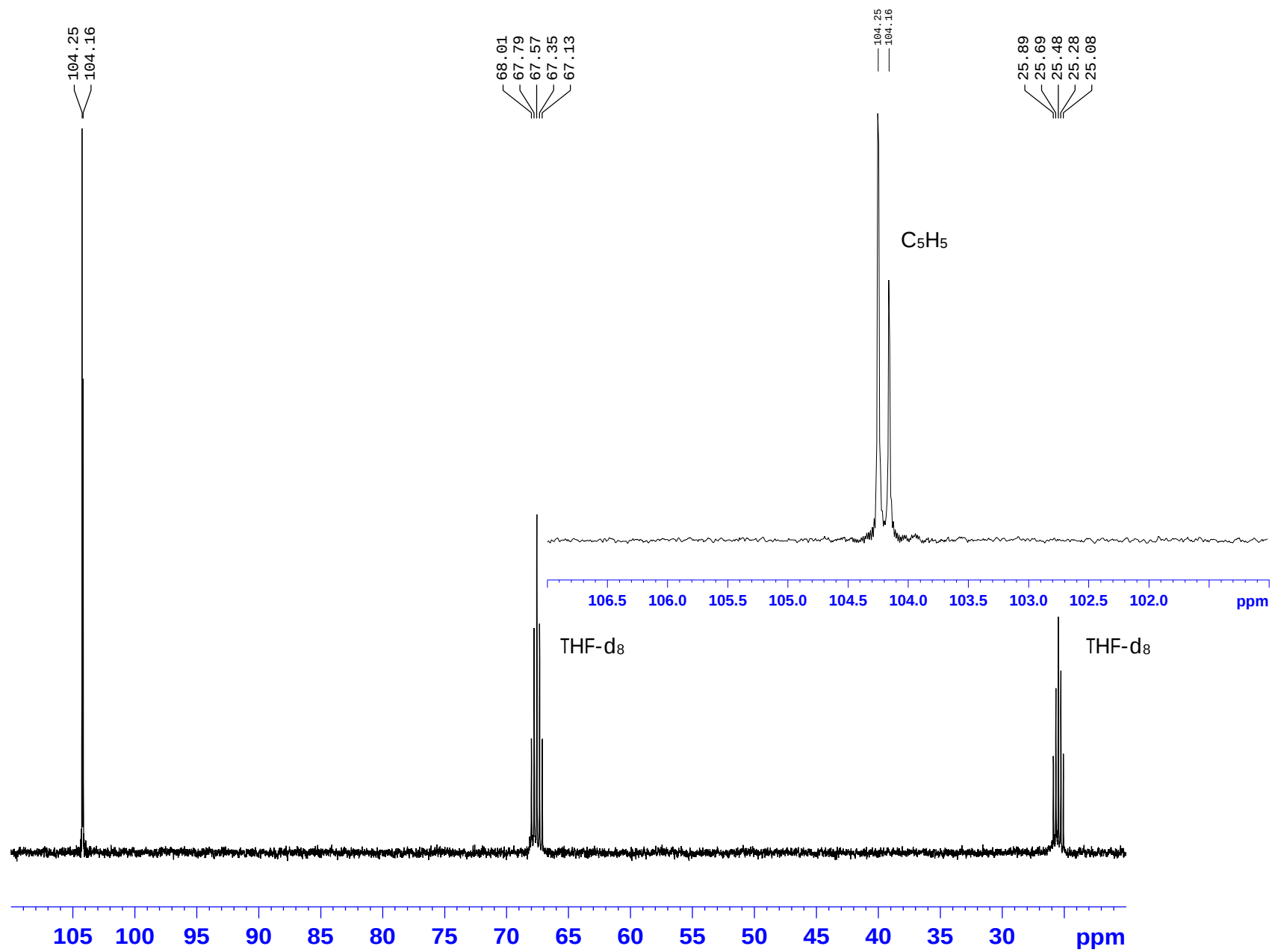
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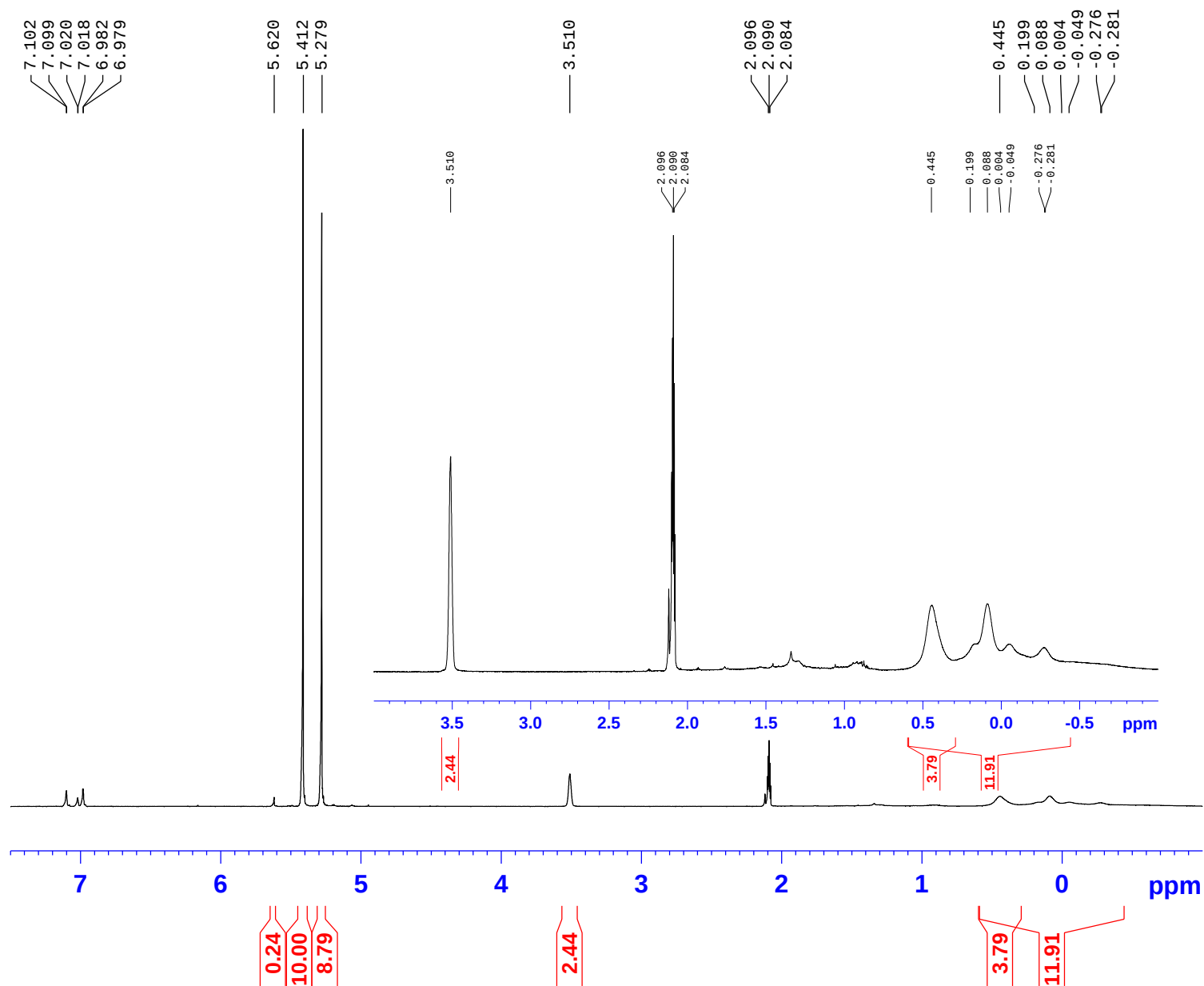
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^{11}\text{B}\{^1\text{H}\}$ NMR in THF-d_8 , 20 °C



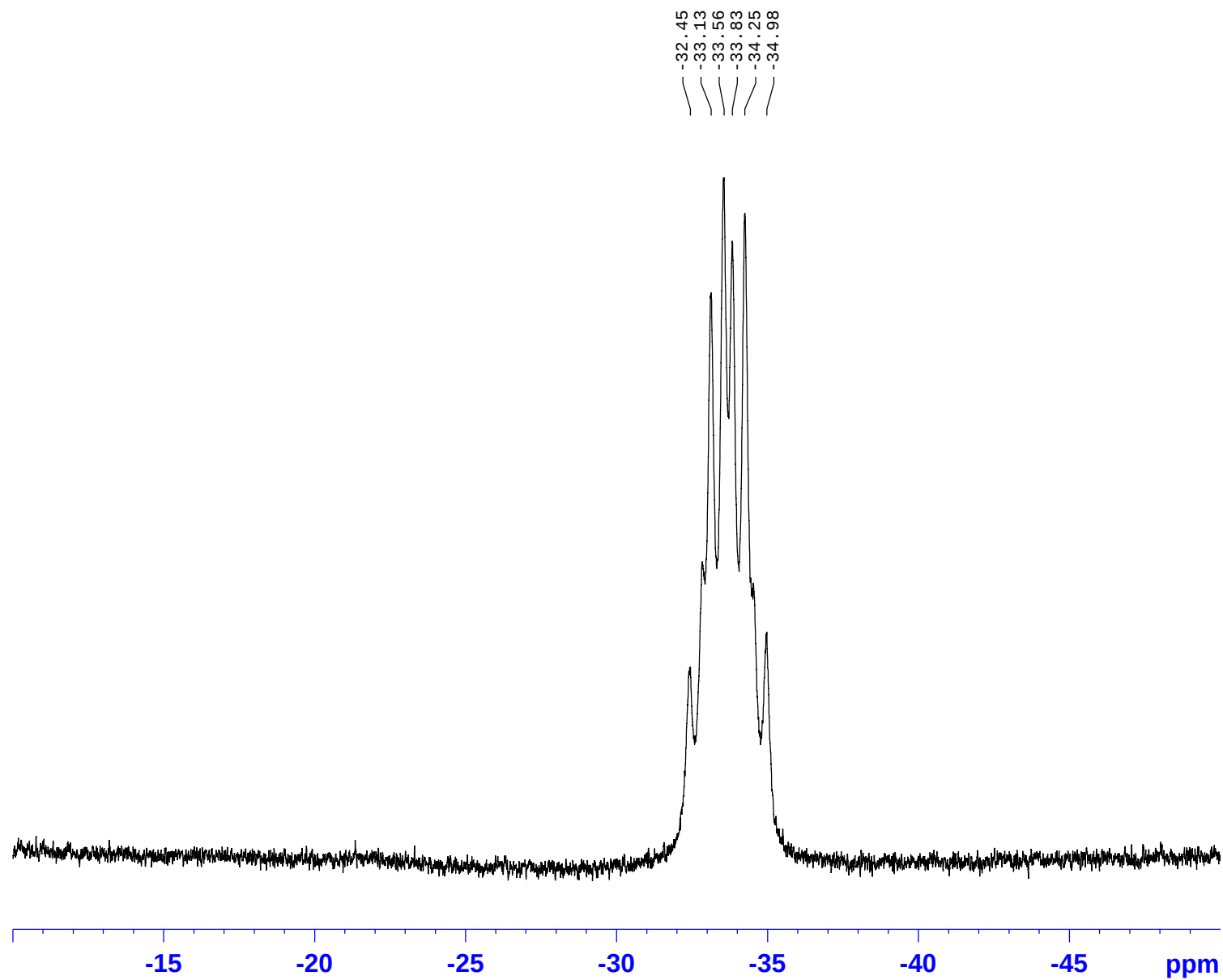
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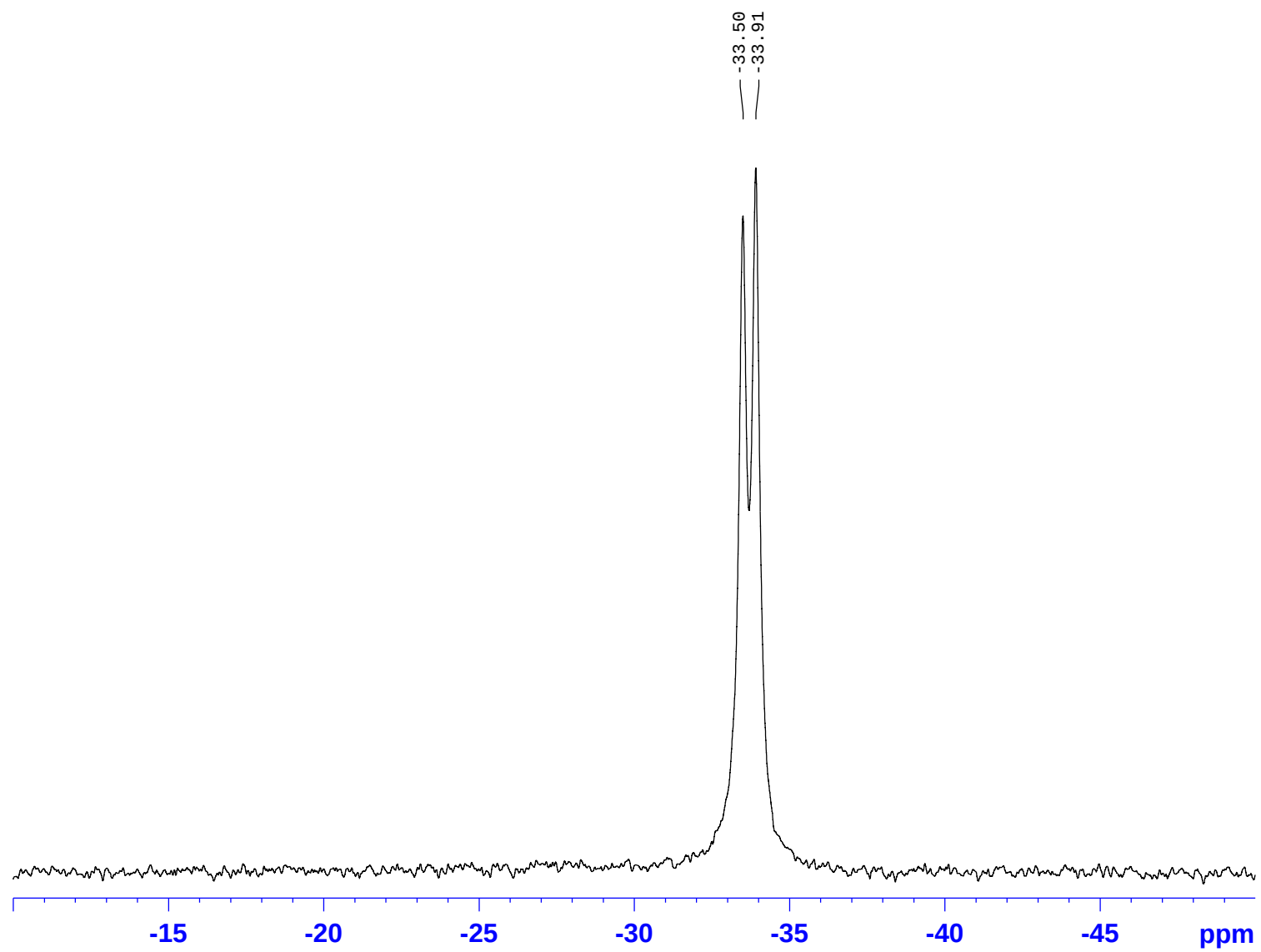
15 mg $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, ^1H NMR in tol-d_8 , 20 °C



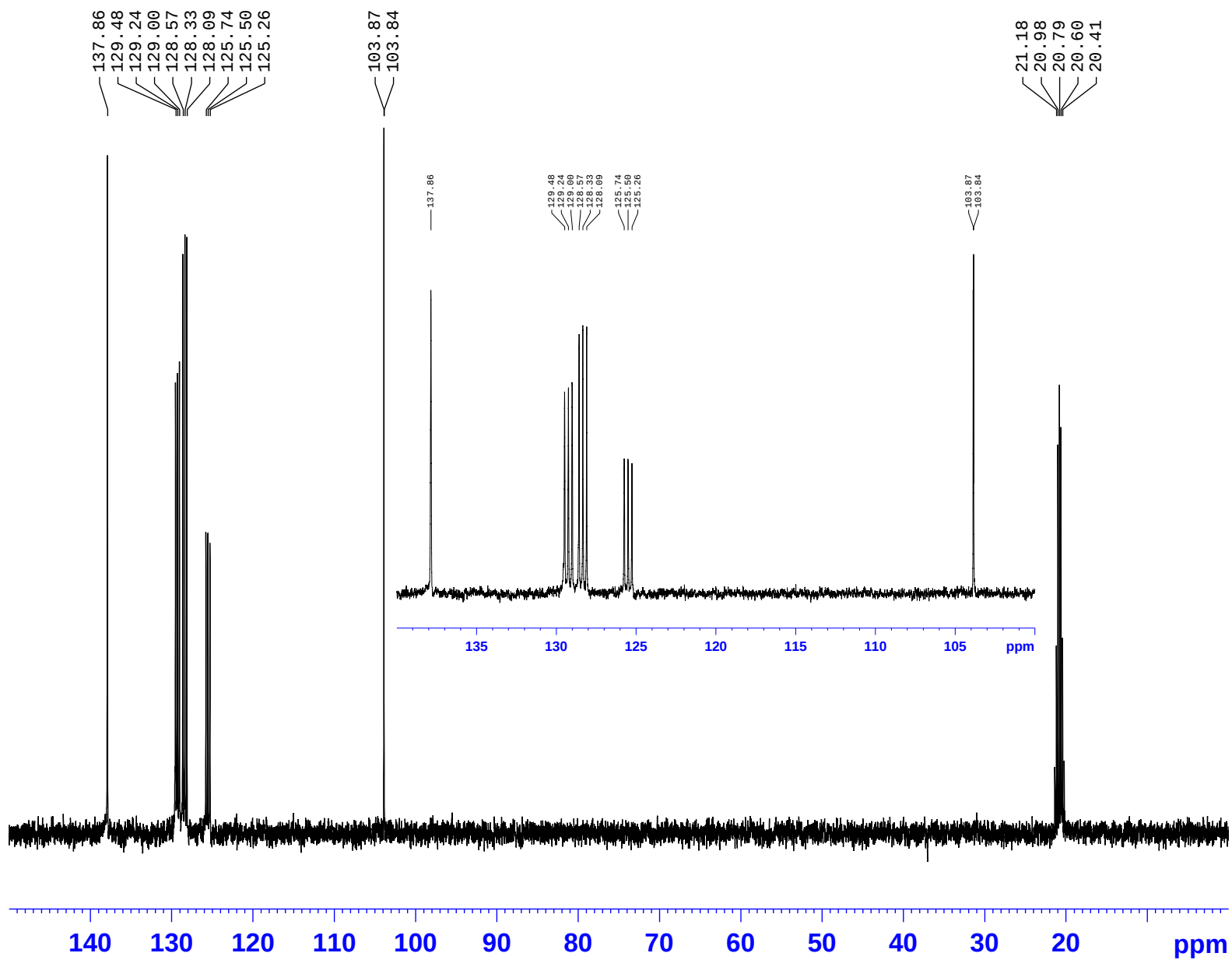
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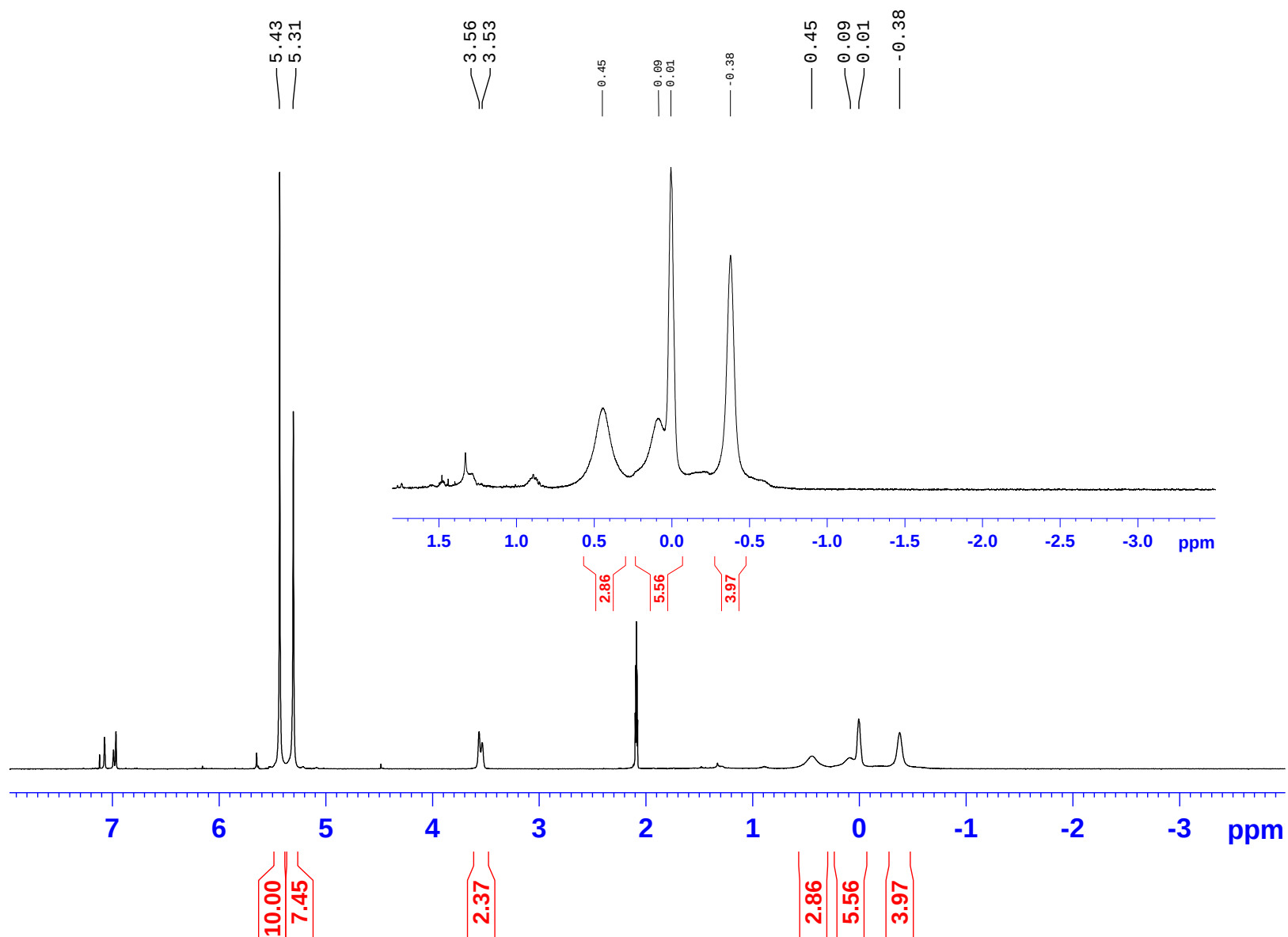
15 mg $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^{11}\text{B}\{^1\text{H}\}$ NMR in tol-d_8 , 20 °C



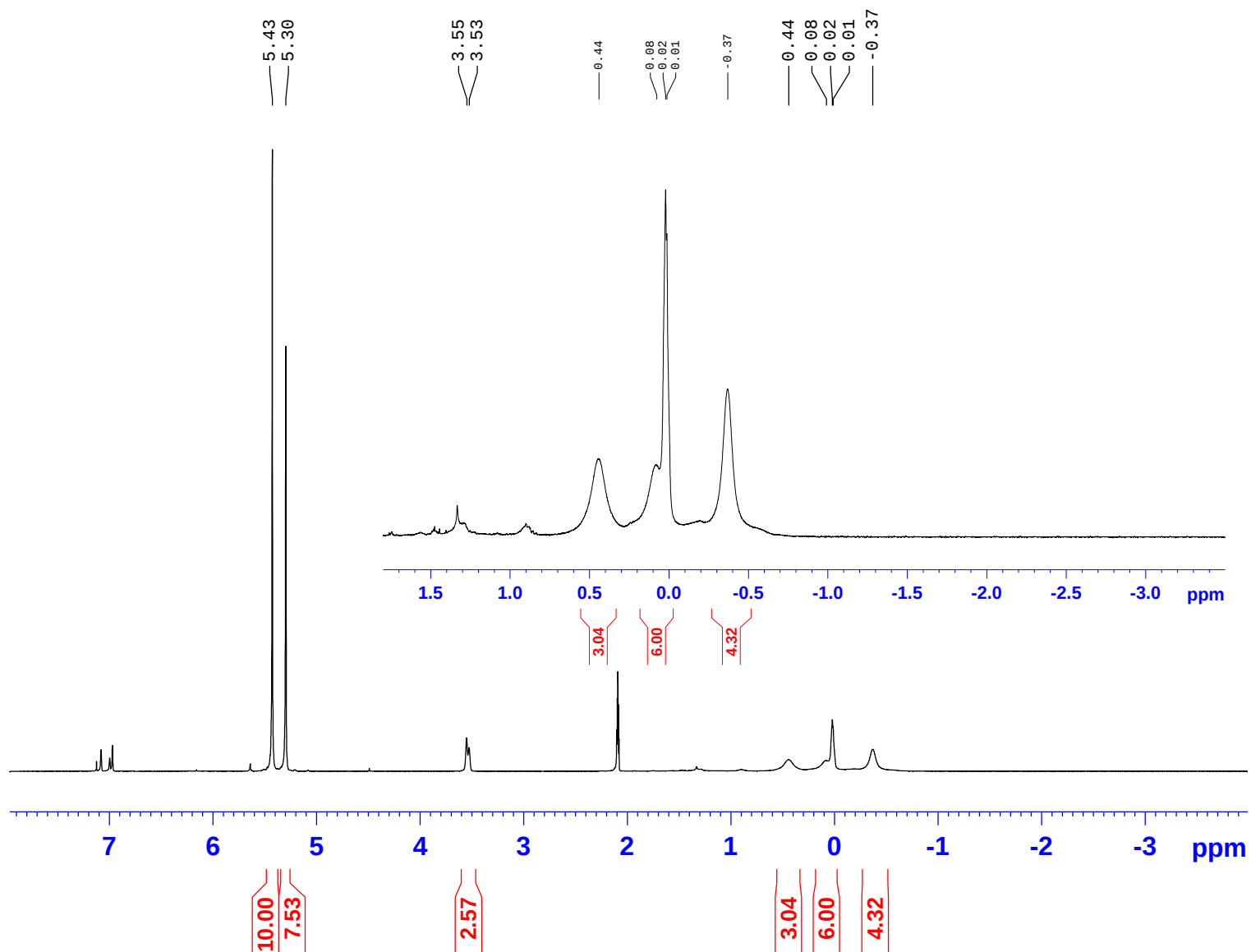
15 mg $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^{13}\text{C}\{^1\text{H}\}$ NMR in tol-d_8 , 20 °C



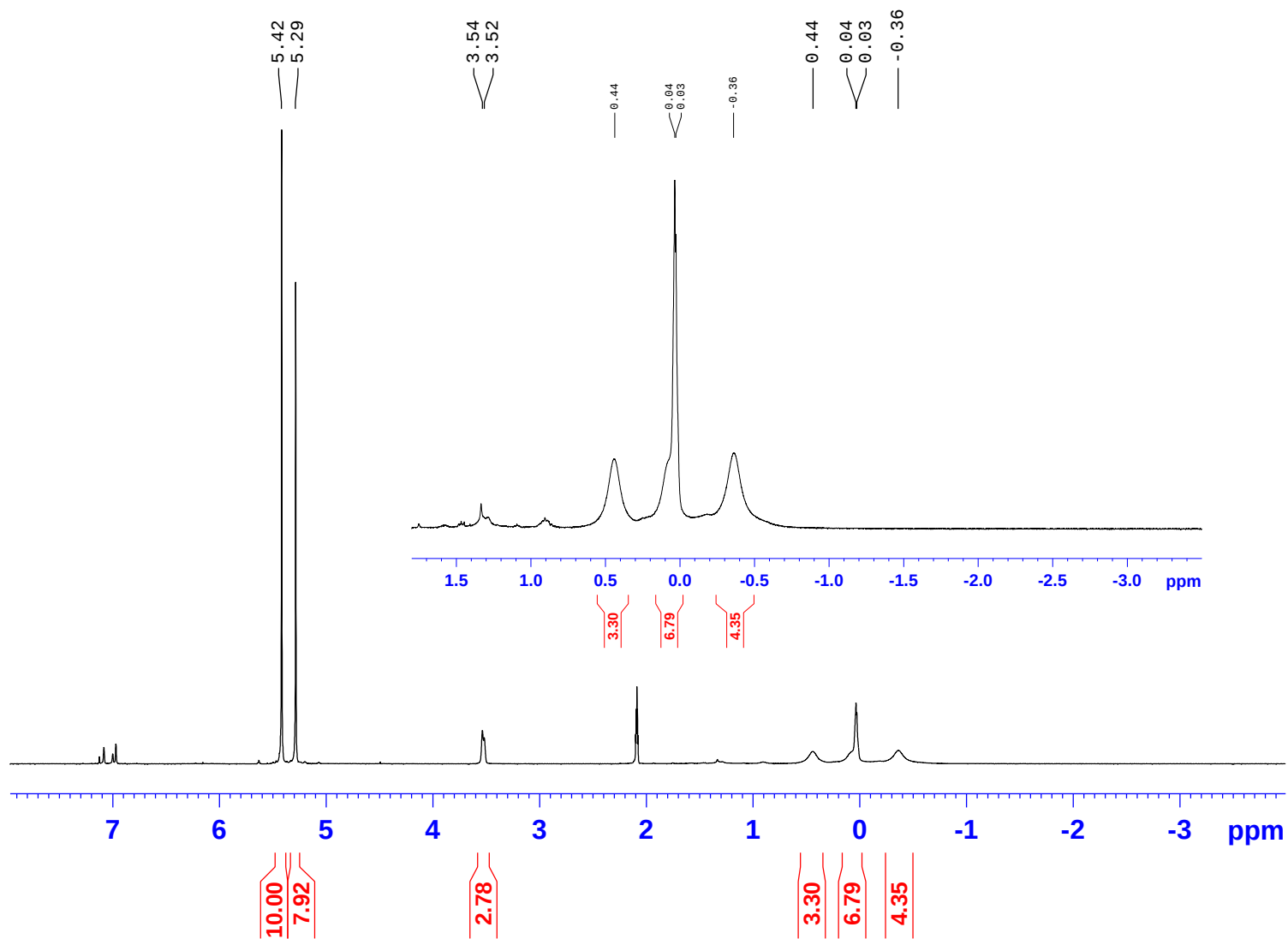
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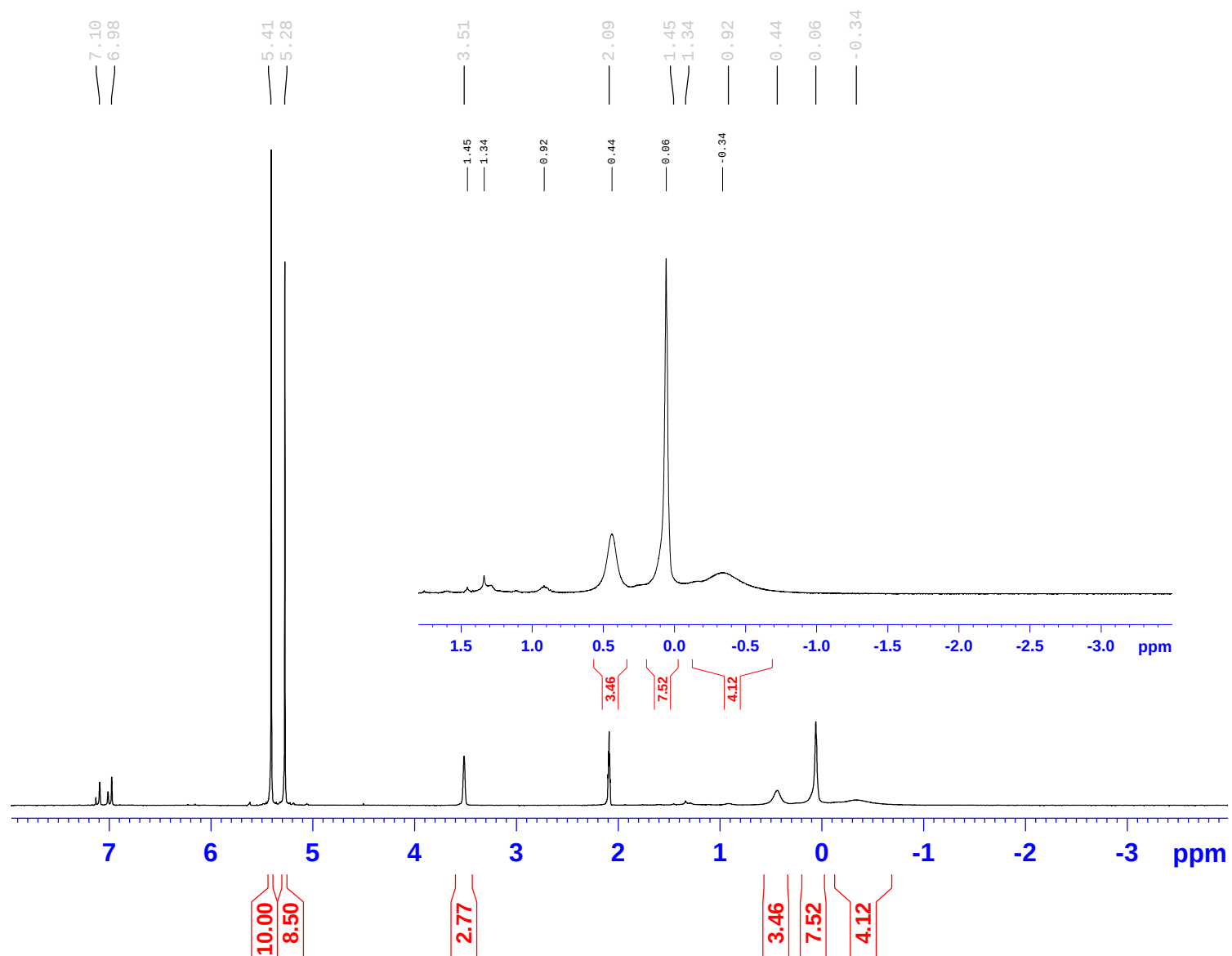
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^1\text{H}\{^{11}\text{B}\}$ NMR in tol-d_8 , 50 °C



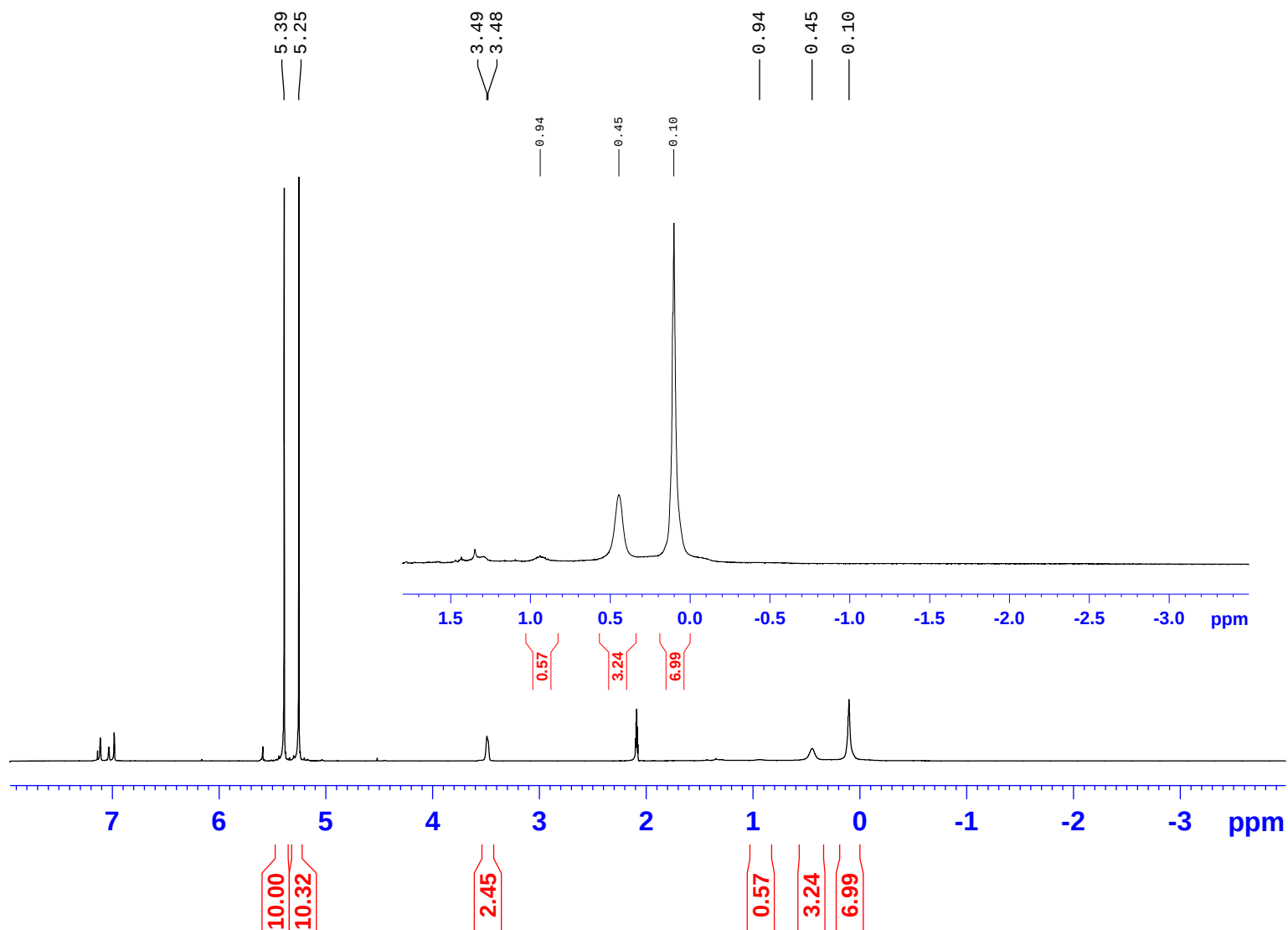
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^1\text{H}\{^{11}\text{B}\}$ NMR in tol-d_8 , 40 °C



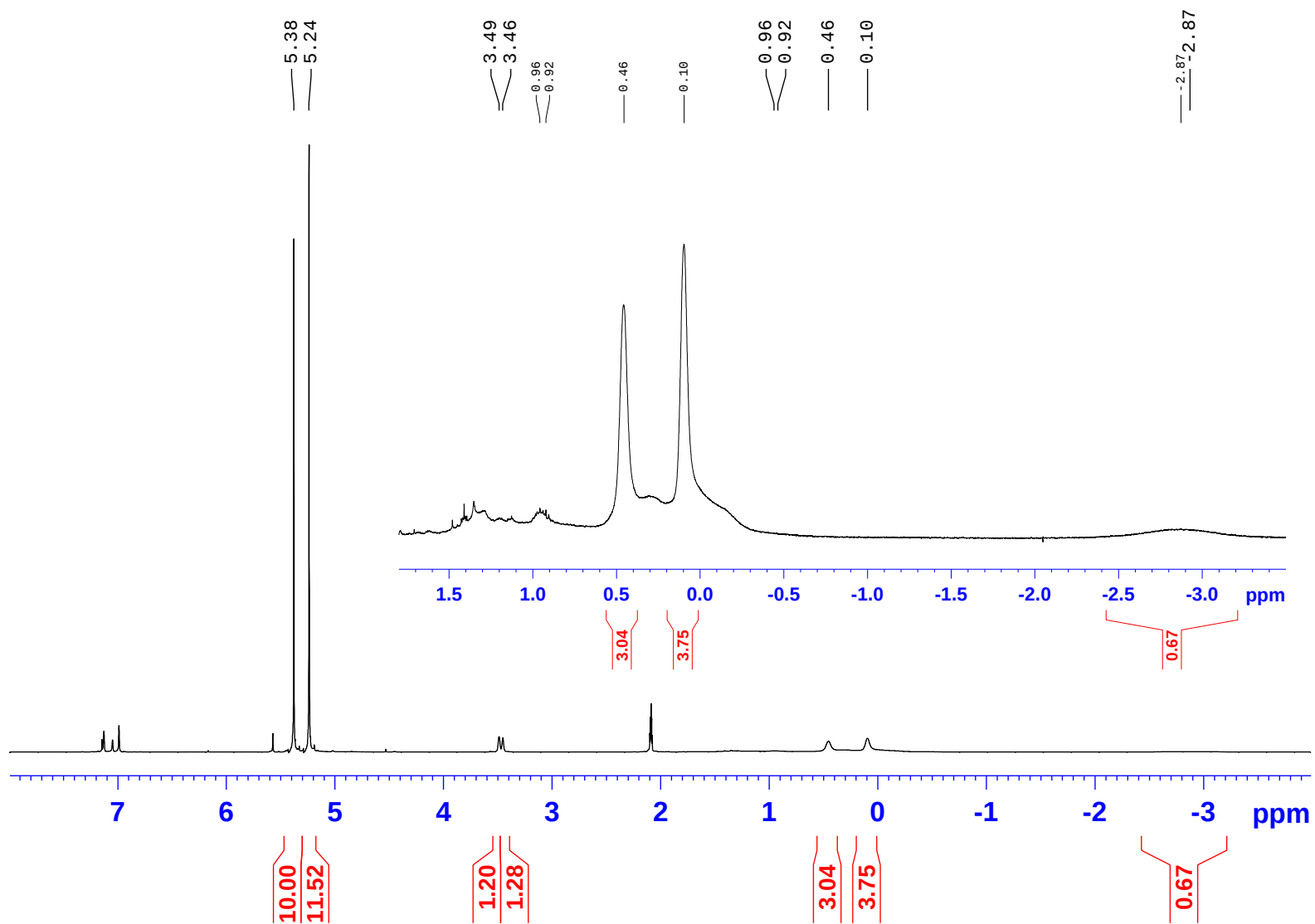
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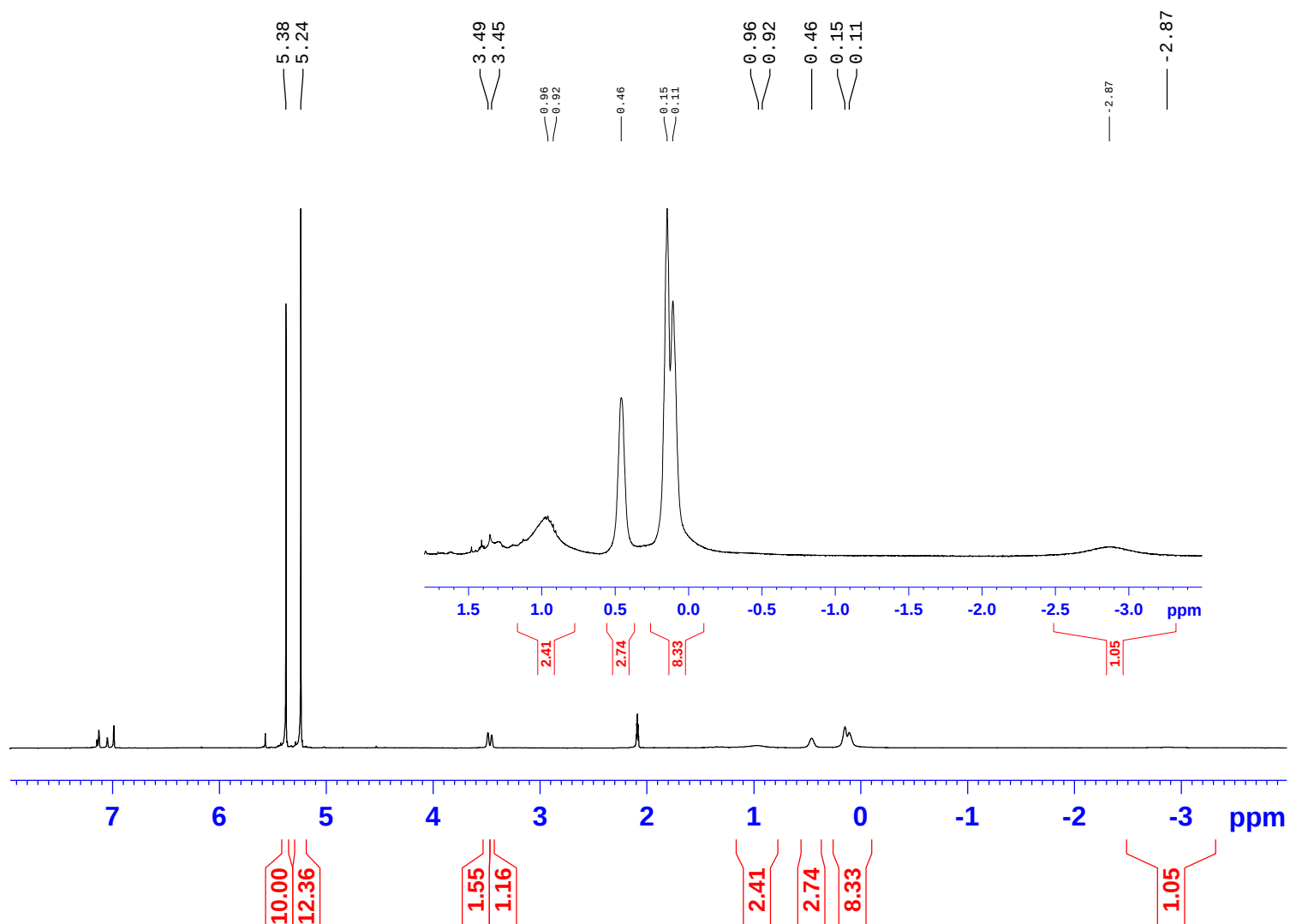
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^1\text{H}\{^{11}\text{B}\}$ NMR in tol-d_8 , 0 °C



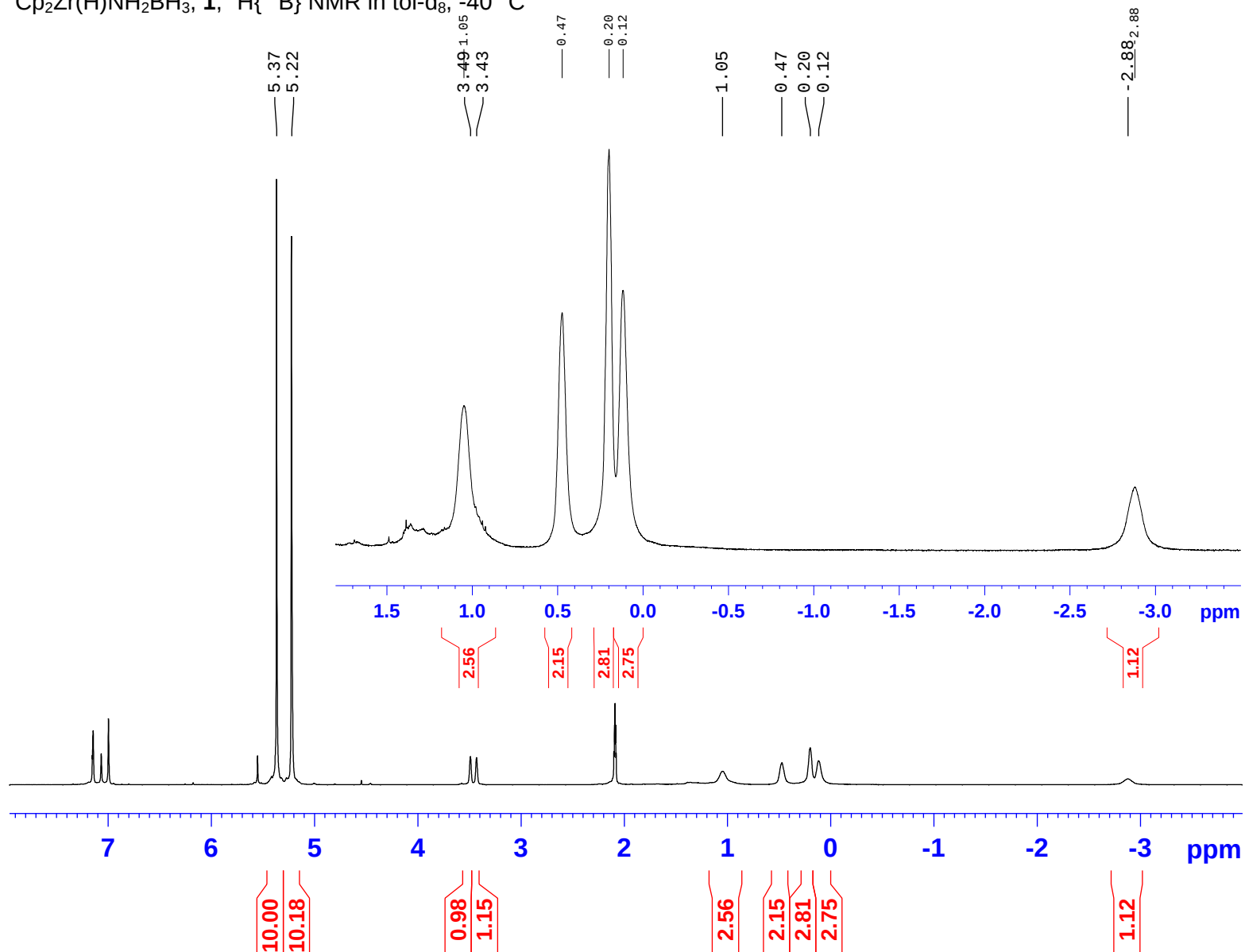
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, ^1H NMR in tol-d_8 , $-20\text{ }^\circ\text{C}$



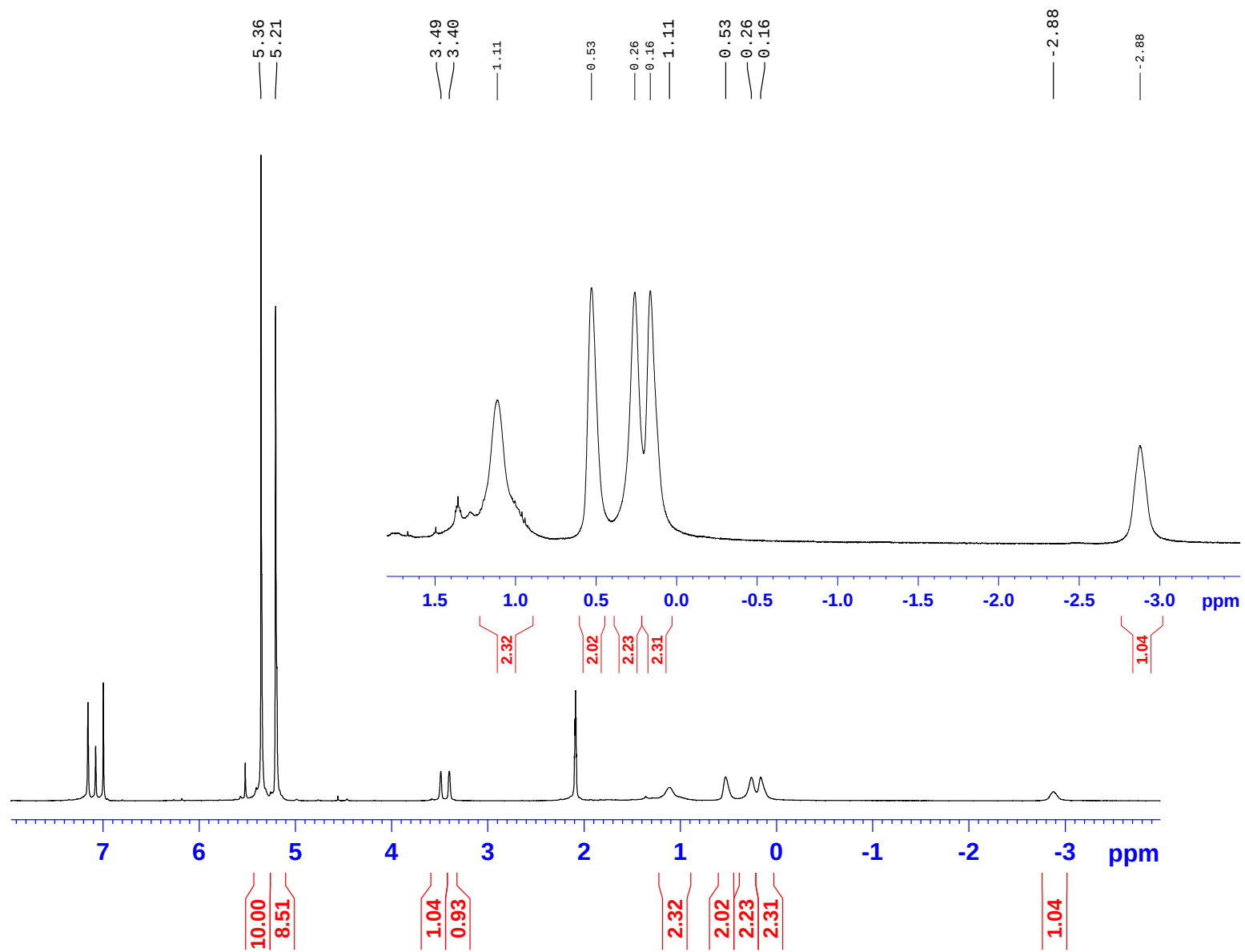
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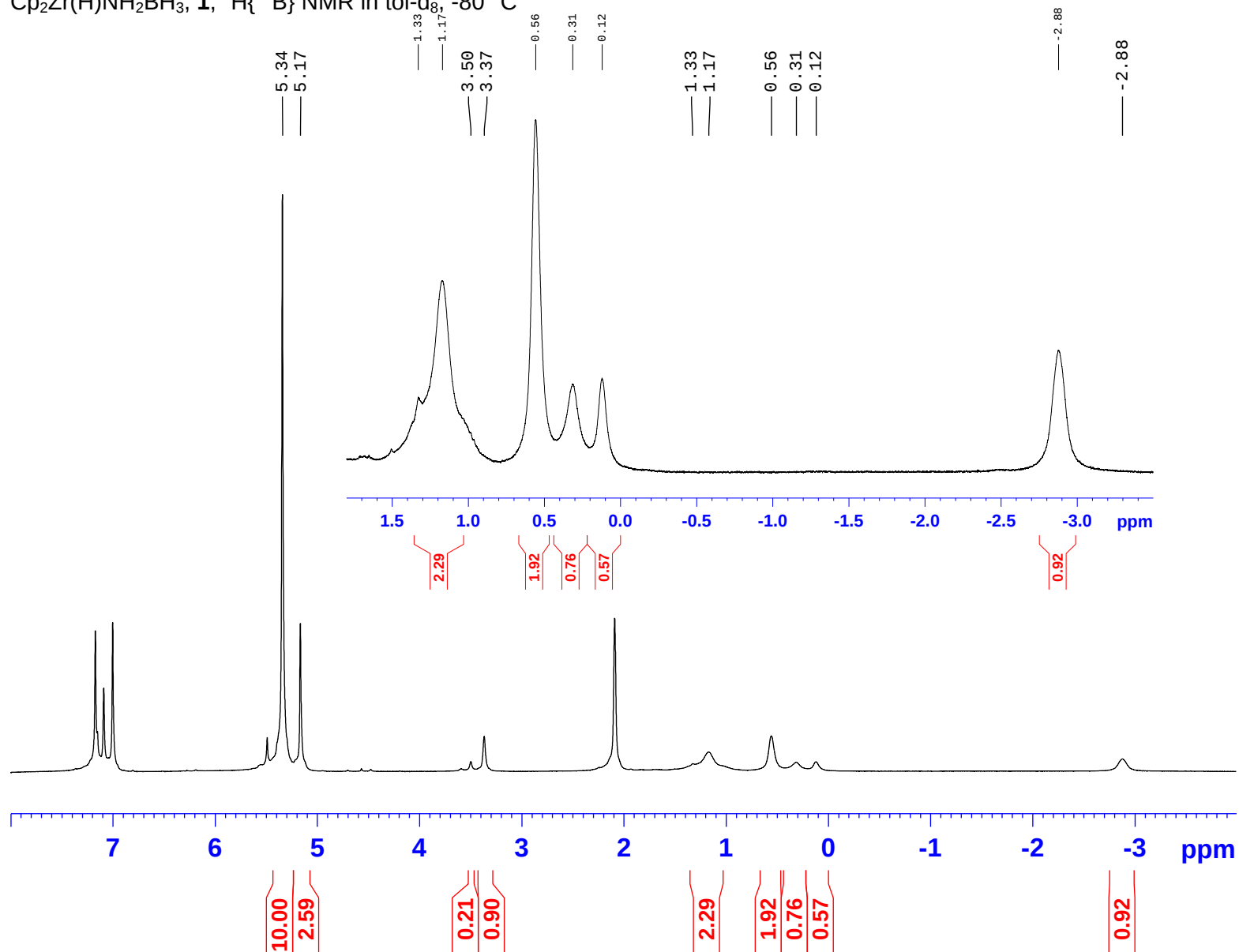
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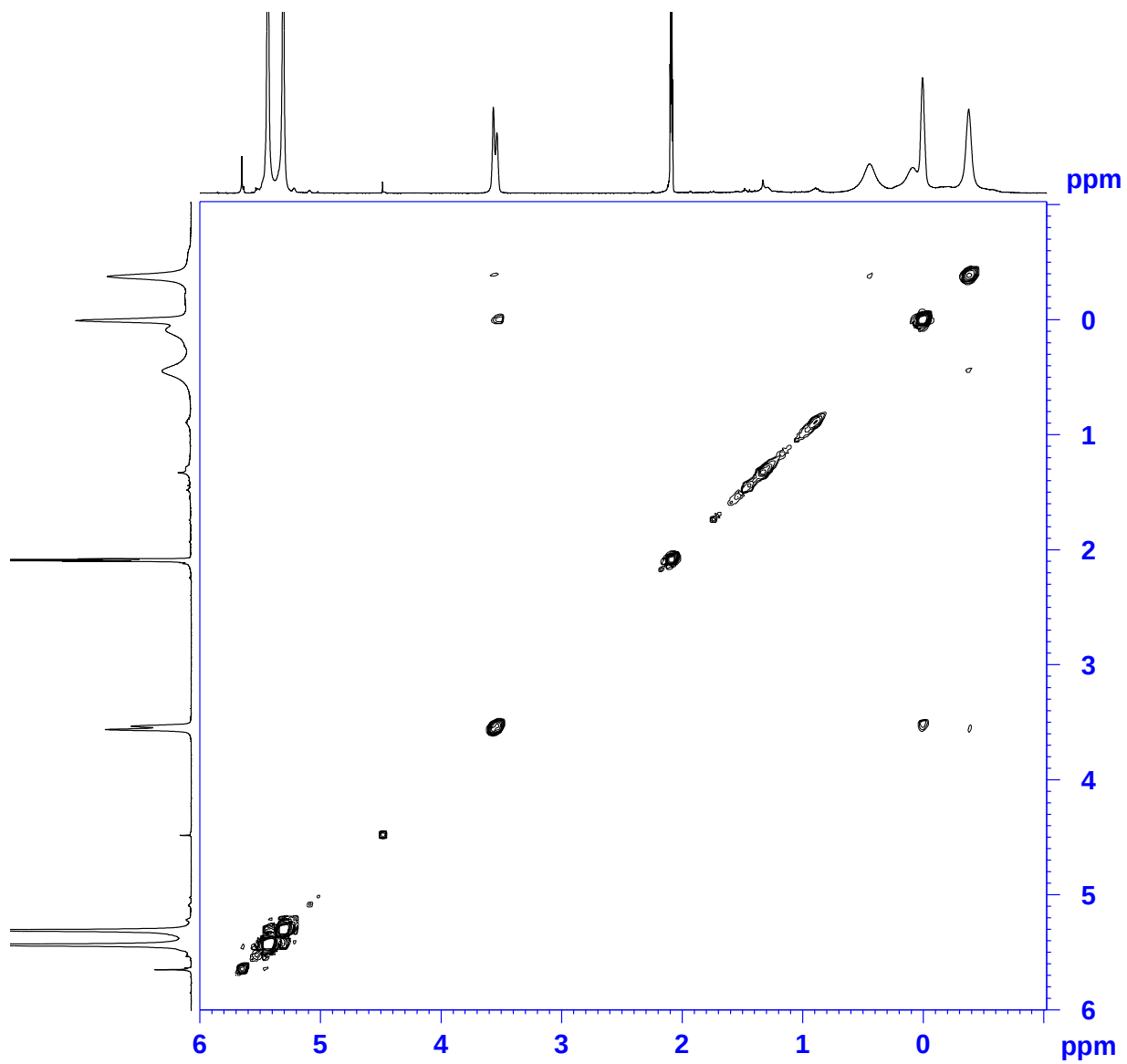
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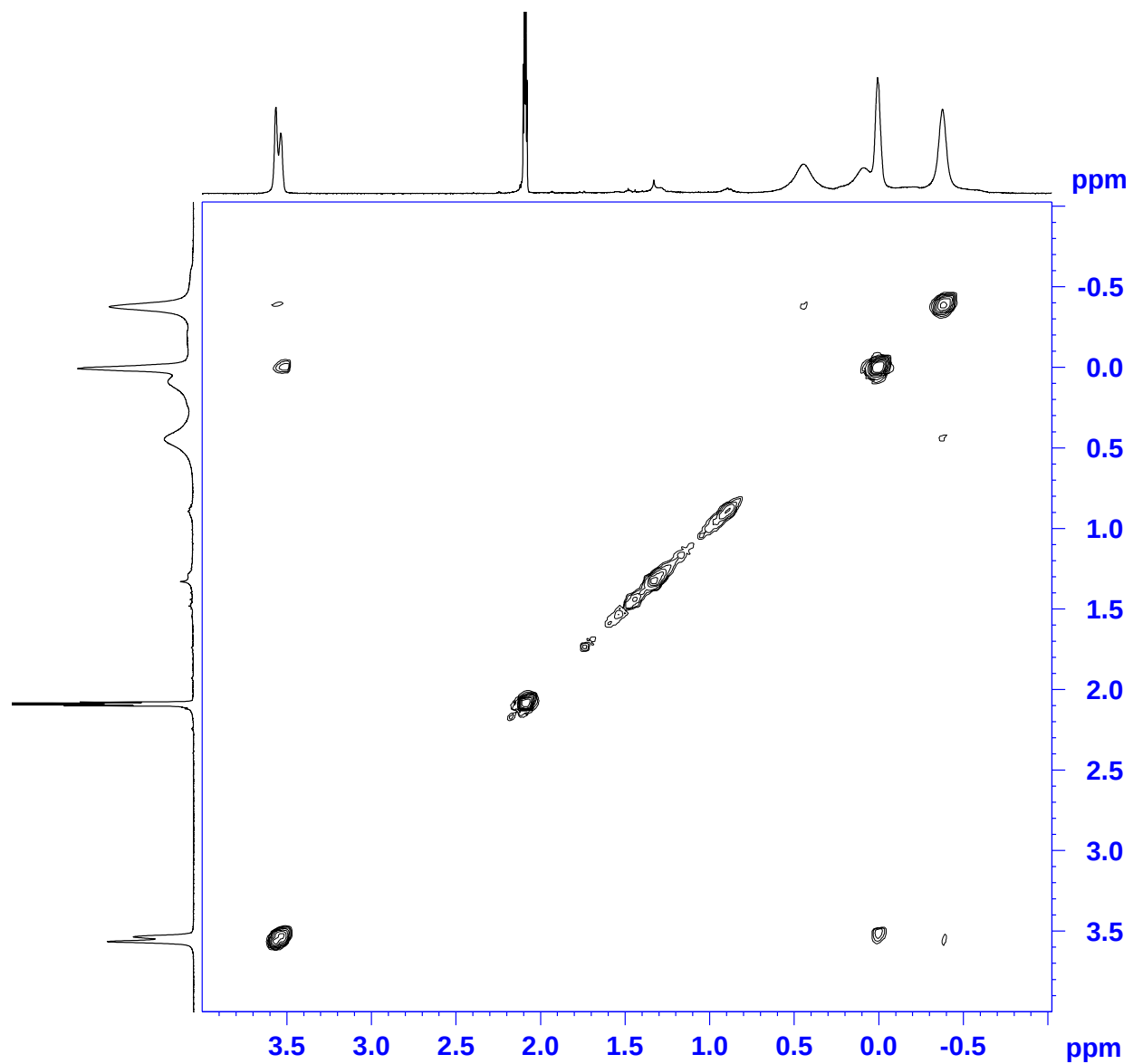
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, $^1\text{H}\{^{11}\text{B}\}$ NMR in tol-d_8 , -80°C



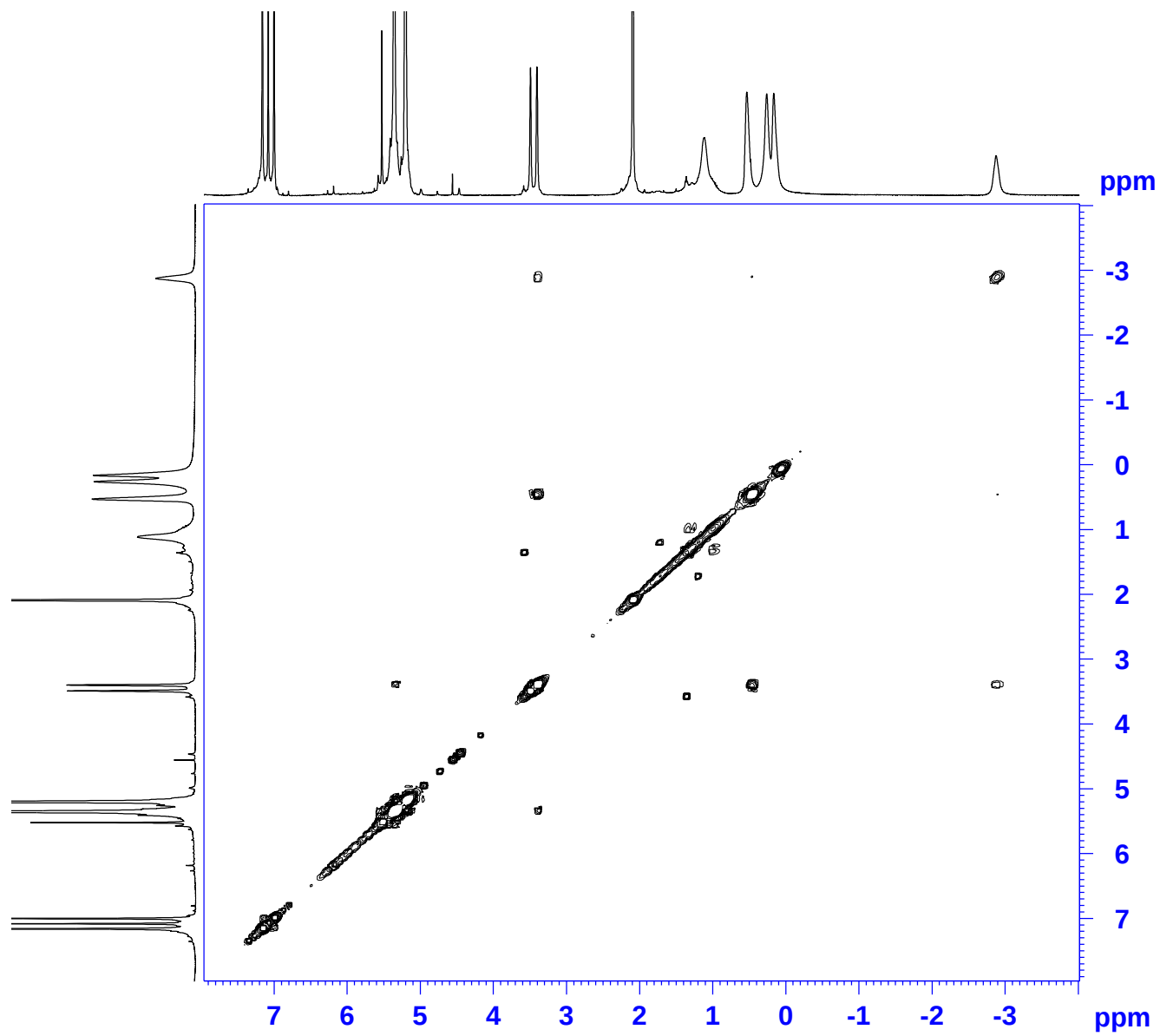
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, COSY $\{^{11}\text{B}\}$ in tol-d_8 , 60 °C



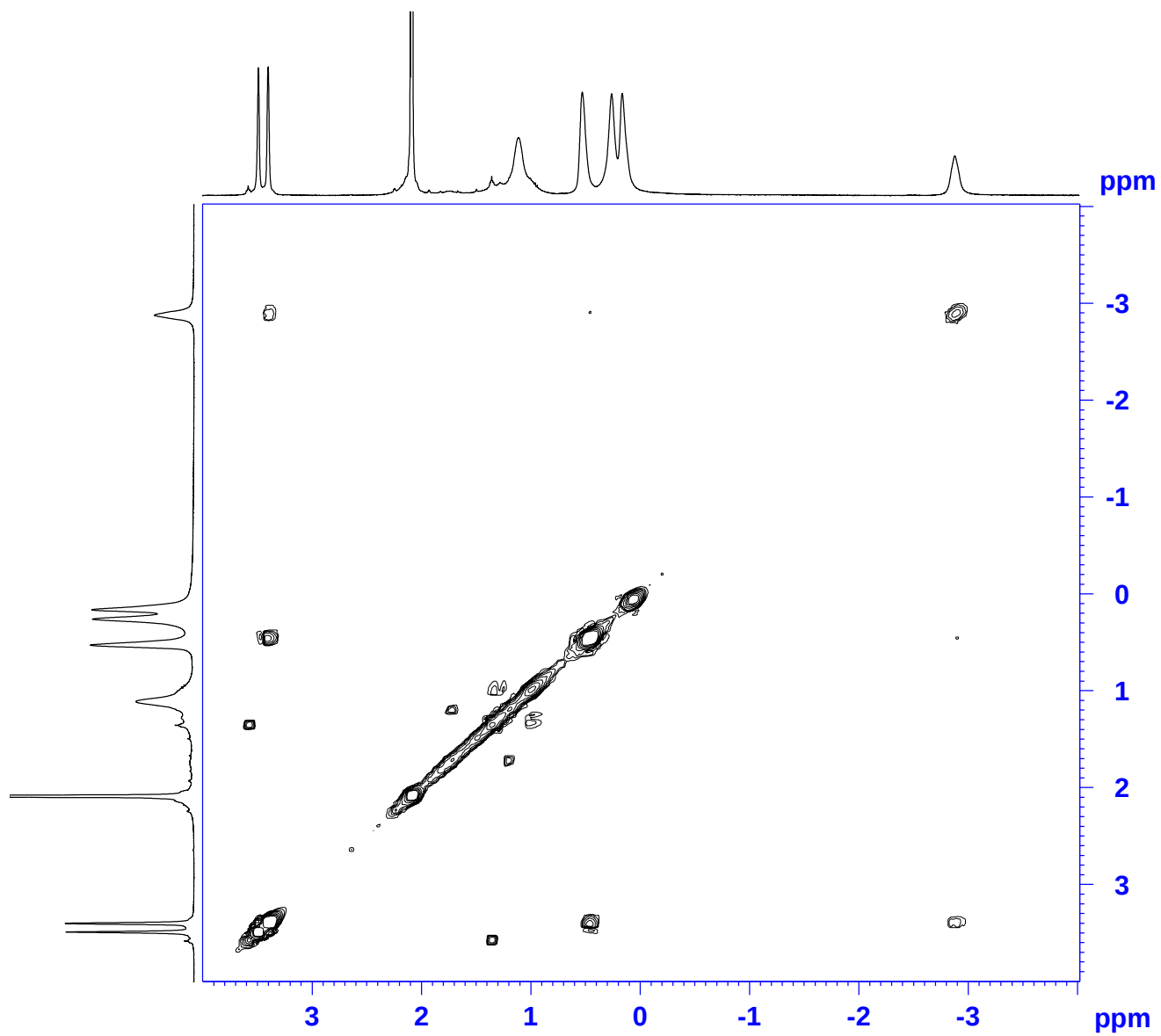
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, COSY $\{^{11}\text{B}\}$ in tol-d_8 , 60 °C, detail



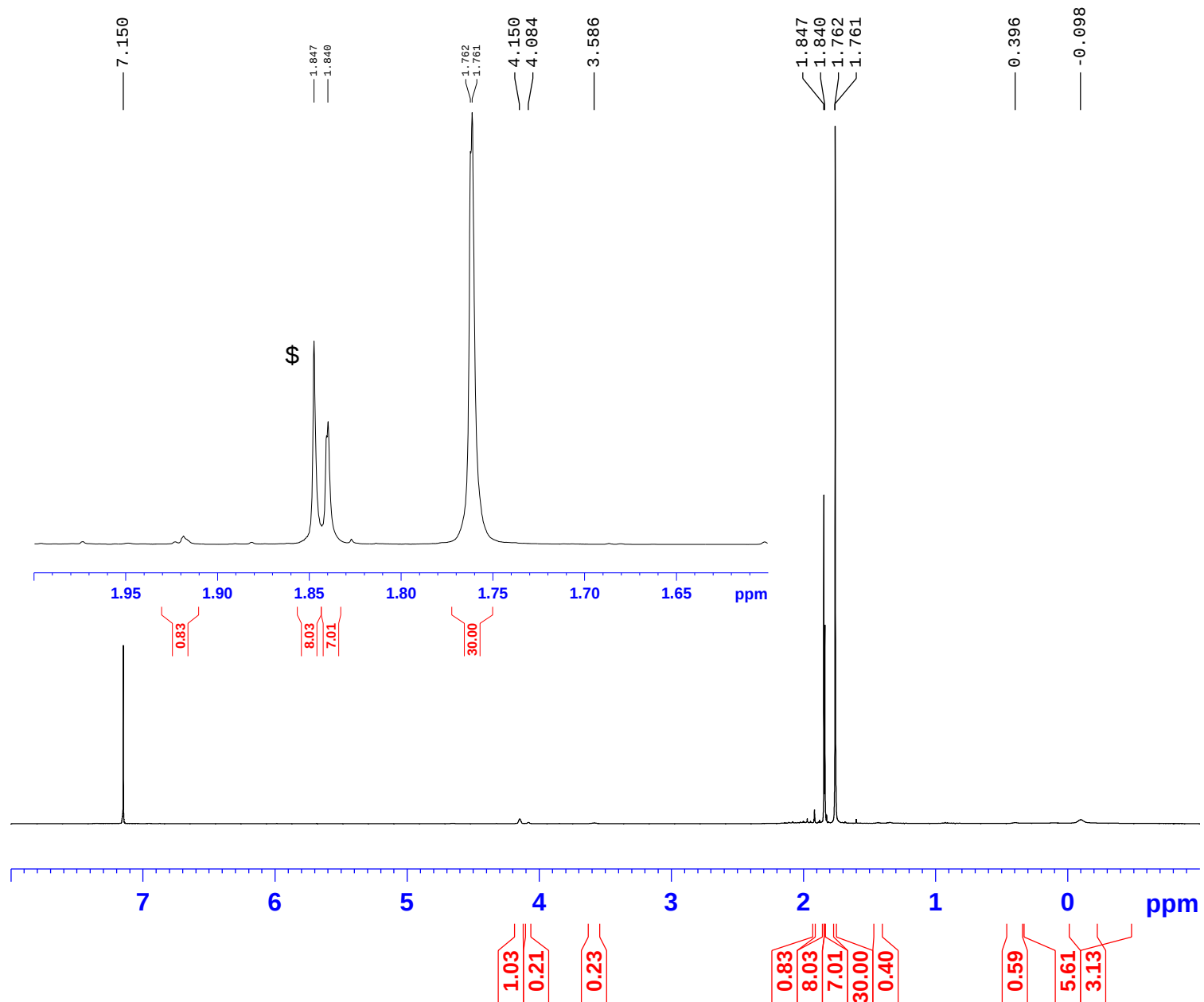
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, COSY $\{^{11}\text{B}\}$ in tol-d_8 , -60°C



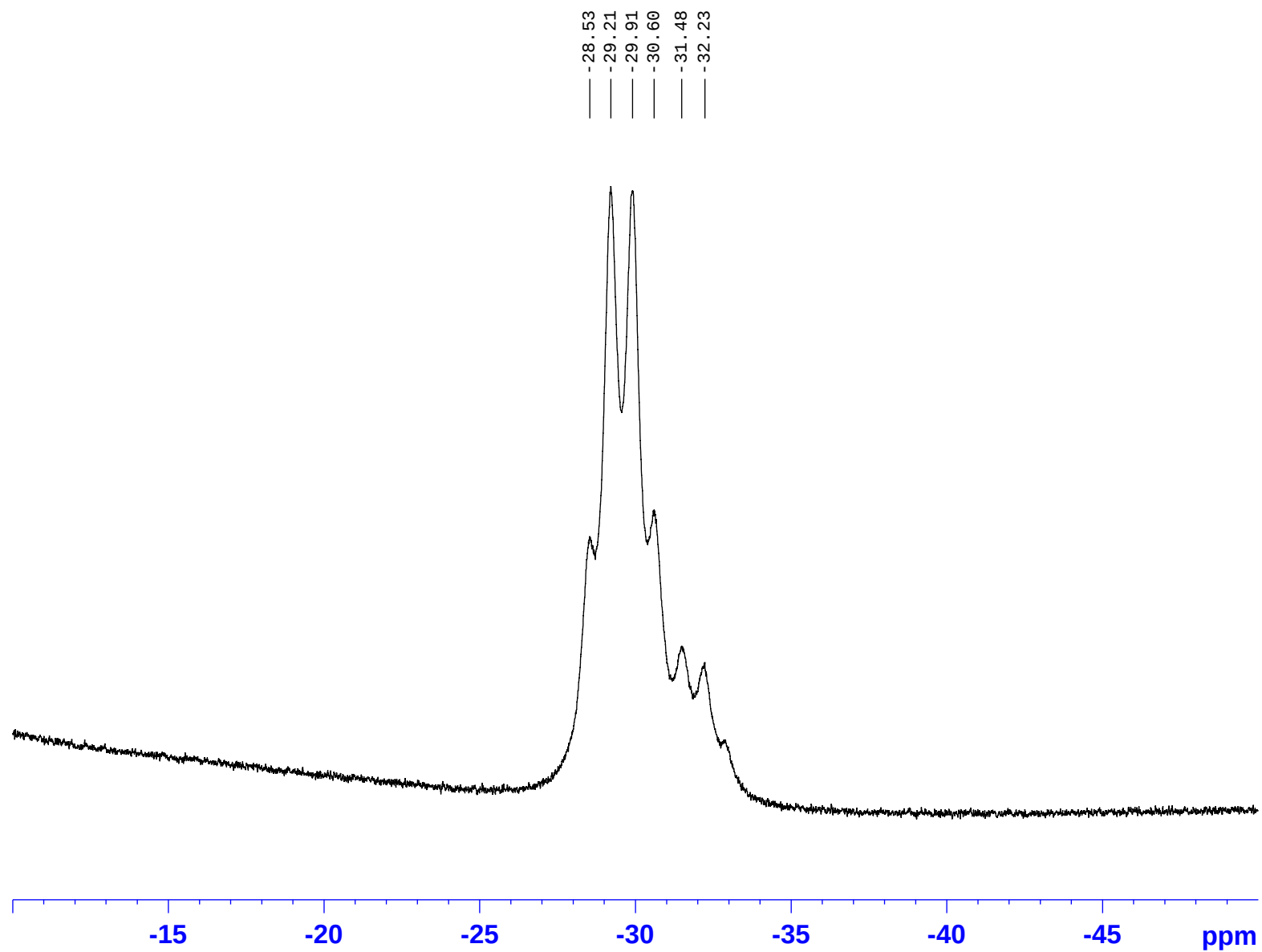
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, COSY $\{^{11}\text{B}\}$ in tol-d_8 , -60°C , detail



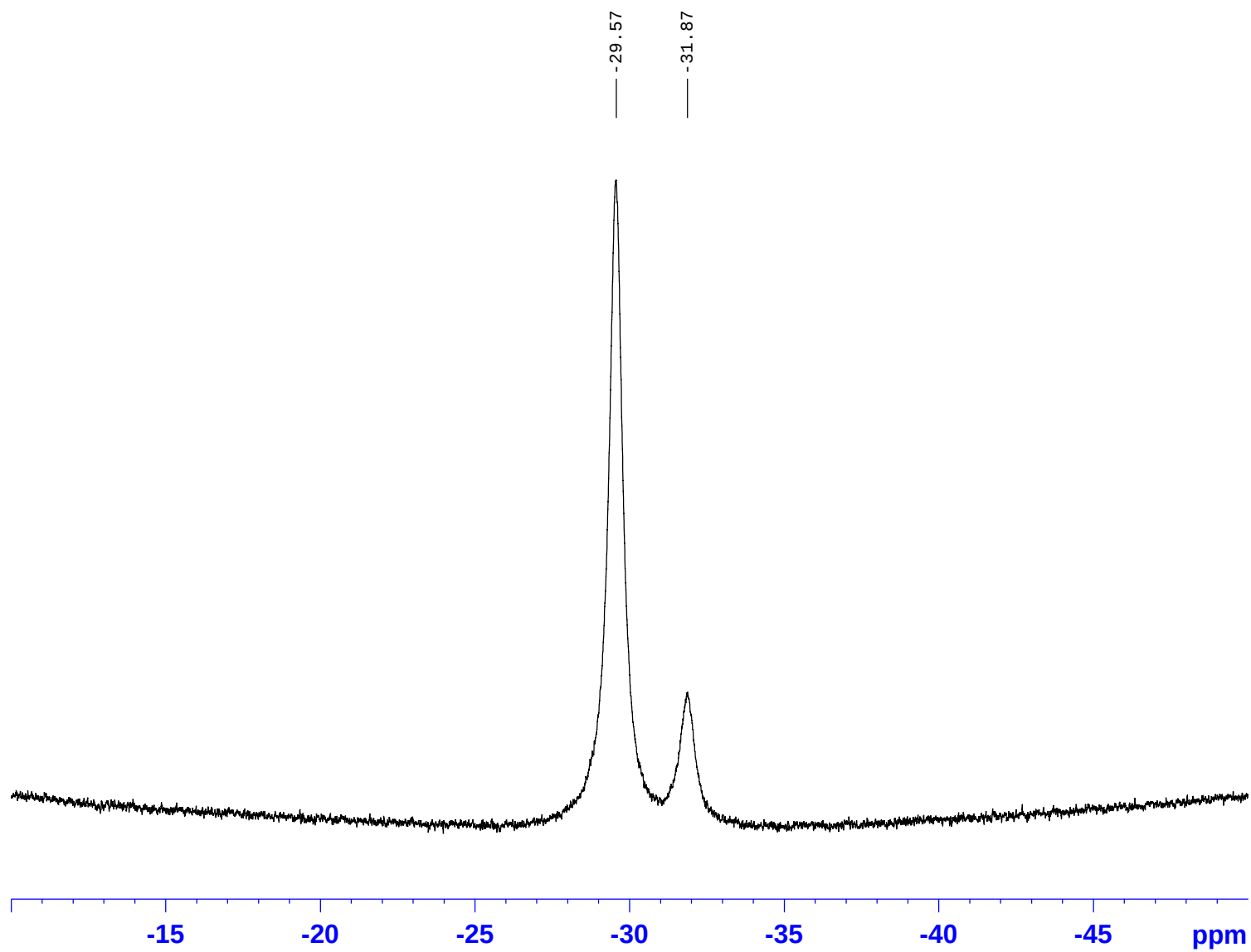
$\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**, ^1H NMR in C_6D_6 , 20 °C (\$ indicates $\text{Cp}^*_2\text{ZrCl}_2$ impurity)



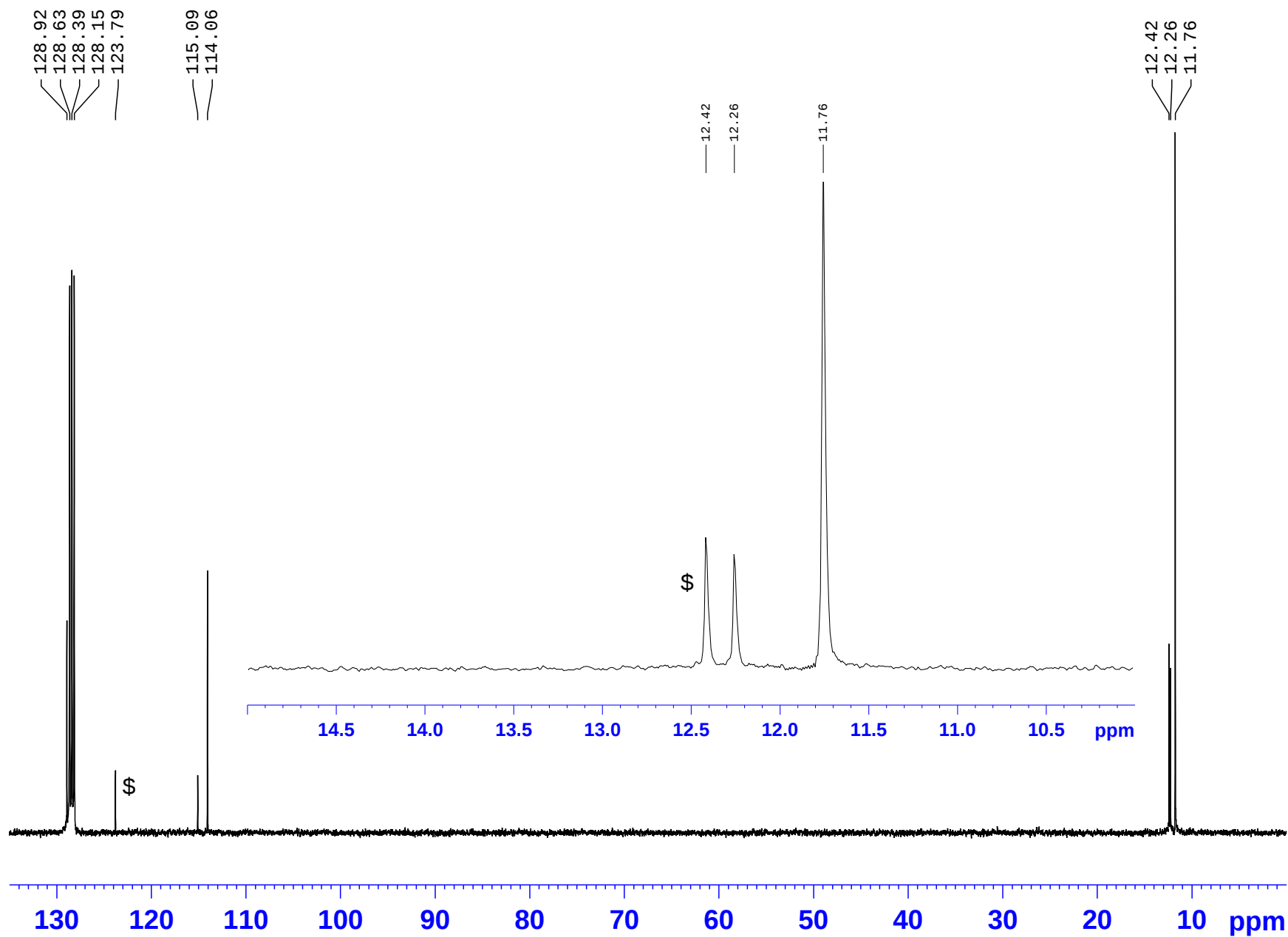
$\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**, ^{11}B NMR in C_6D_6 , 20 °C



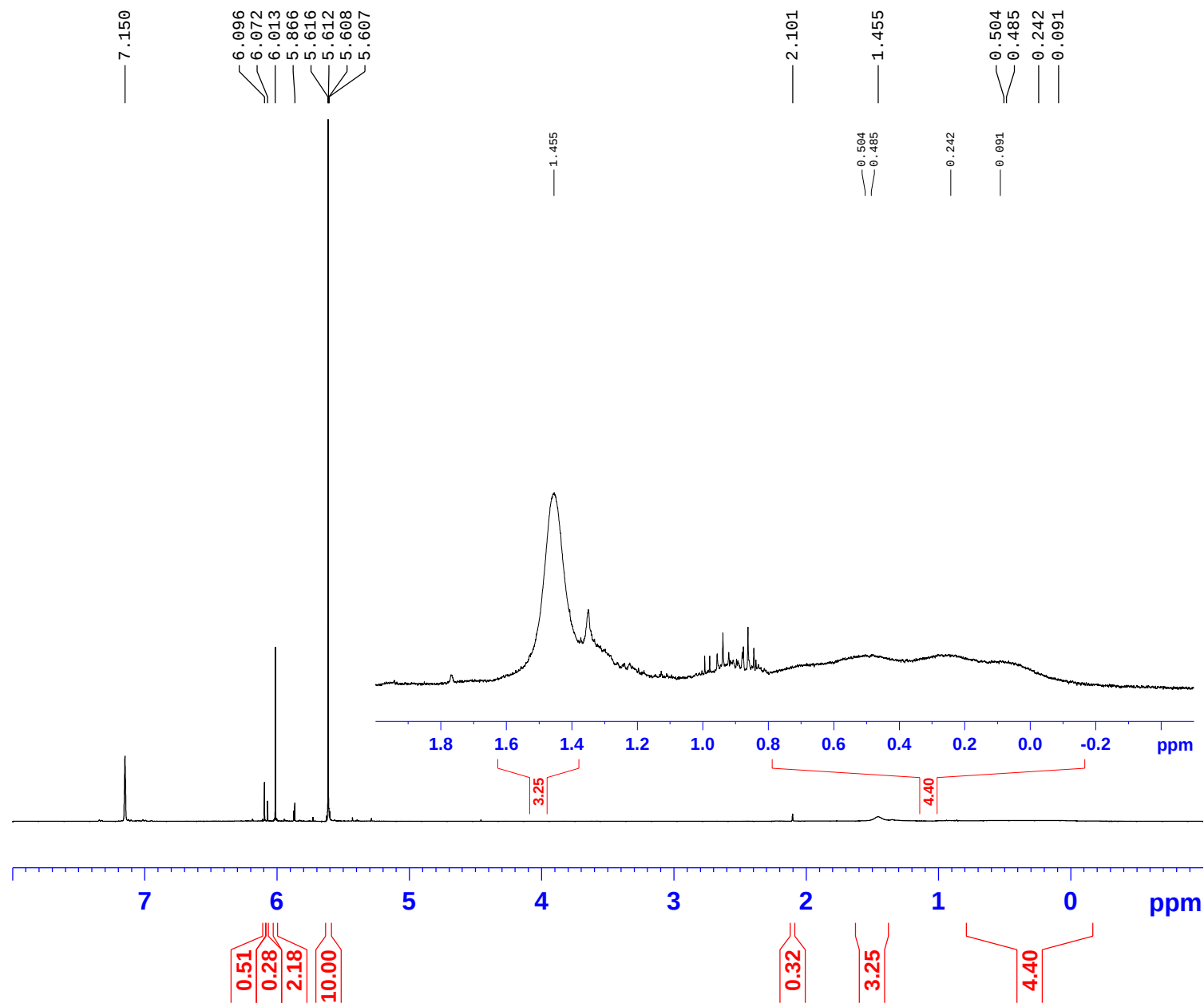
$\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**, $^{11}\text{B}\{^1\text{H}\}$ NMR in C_6D_6 , 20 °C



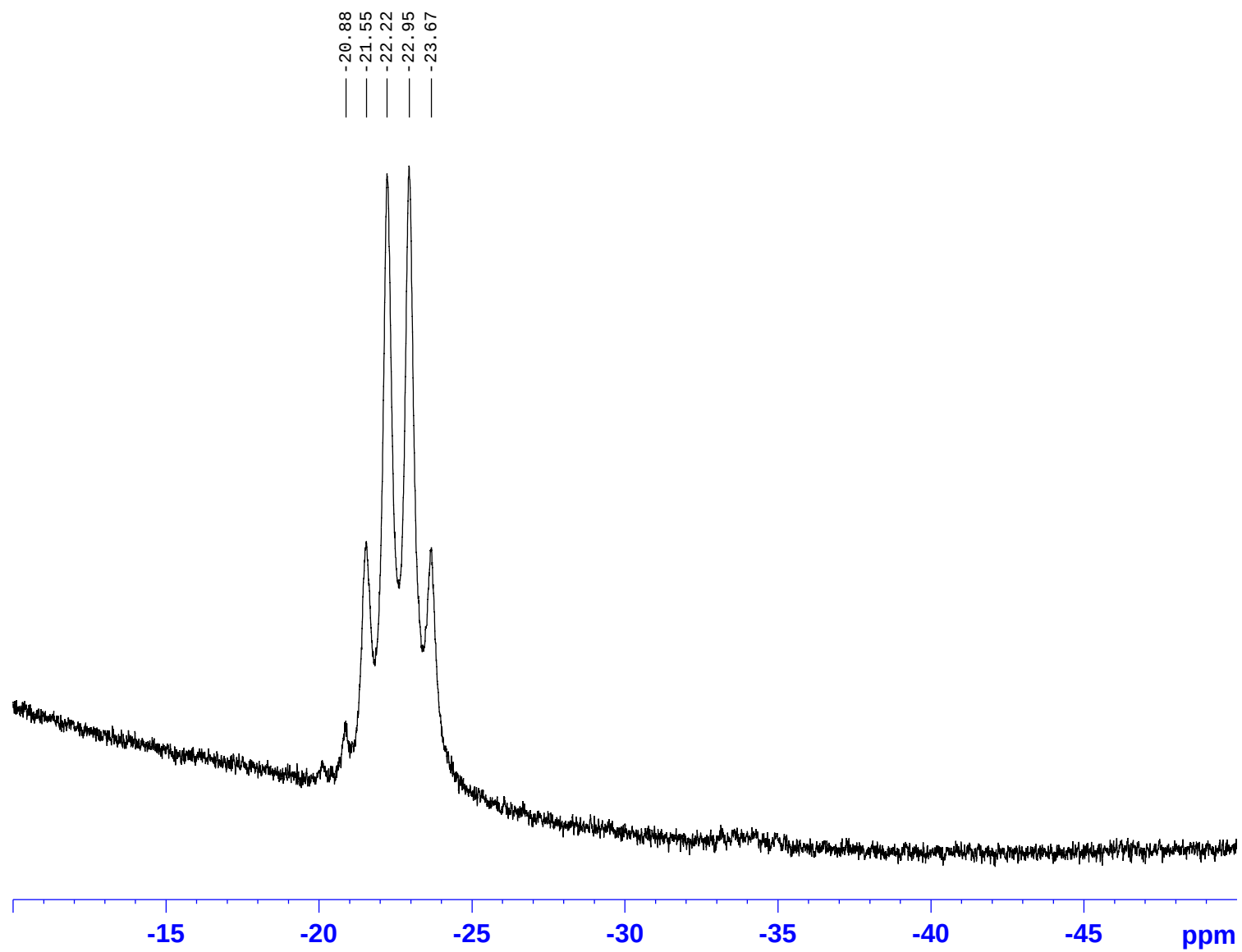
$\text{Cp}^*\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**, $^{13}\text{C}\{^1\text{H}\}$ NMR in C_6D_6 , 20 °C (\$ indicates Cp^*ZrCl_2 impurity)



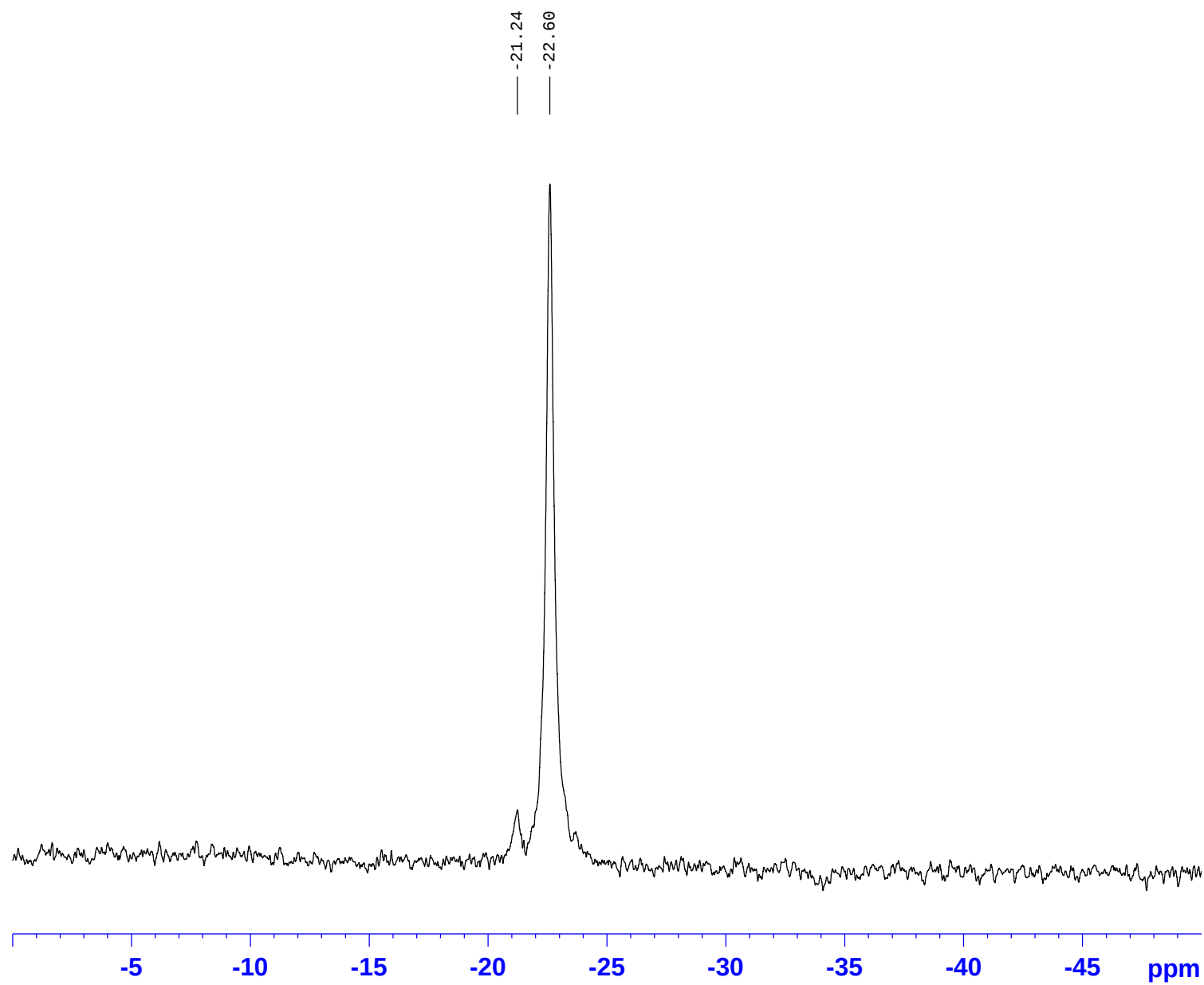
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, ^1H NMR in C_6D_6 , 20 °C



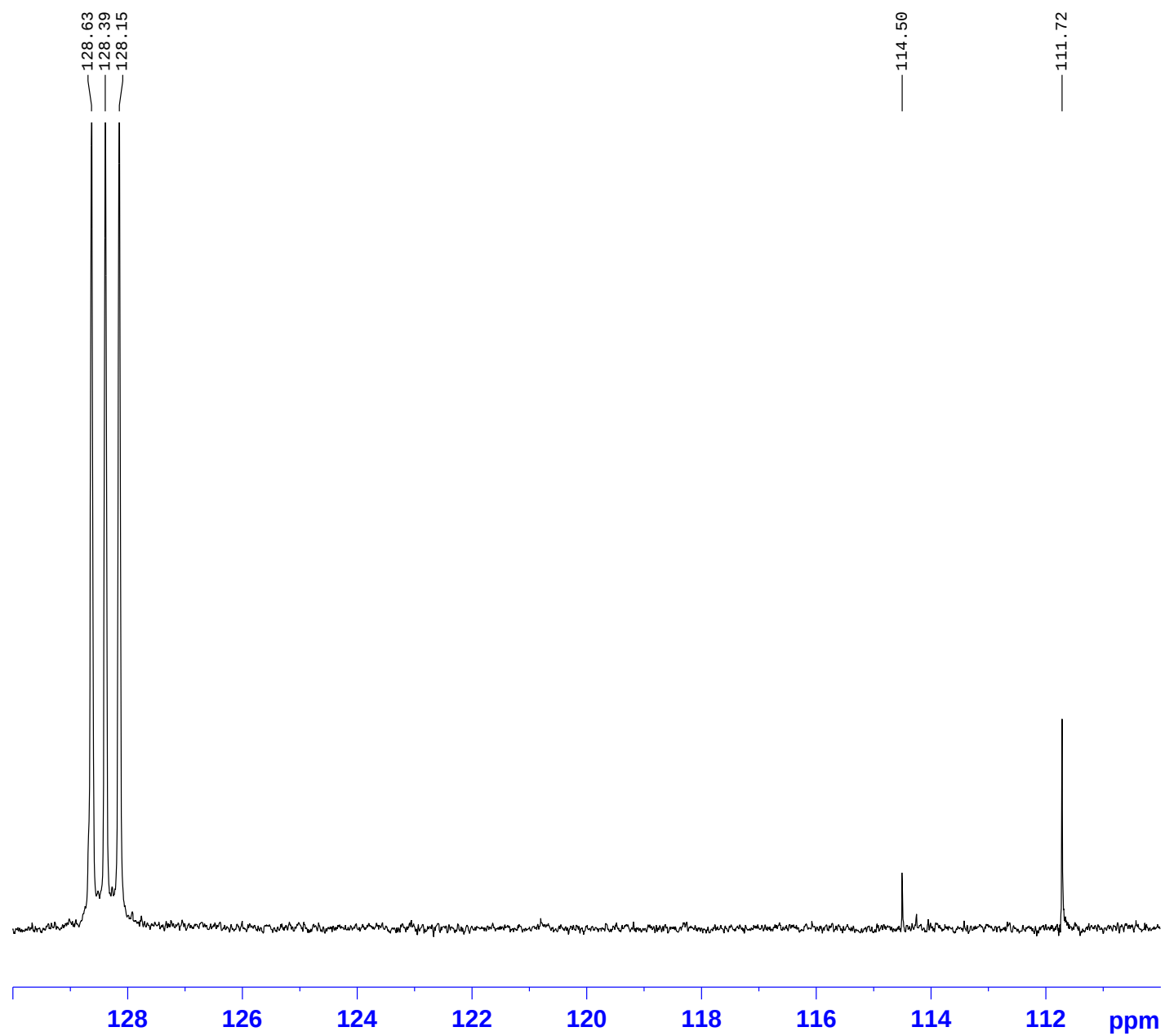
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, ^{11}B NMR in C_6D_6 , 20 °C



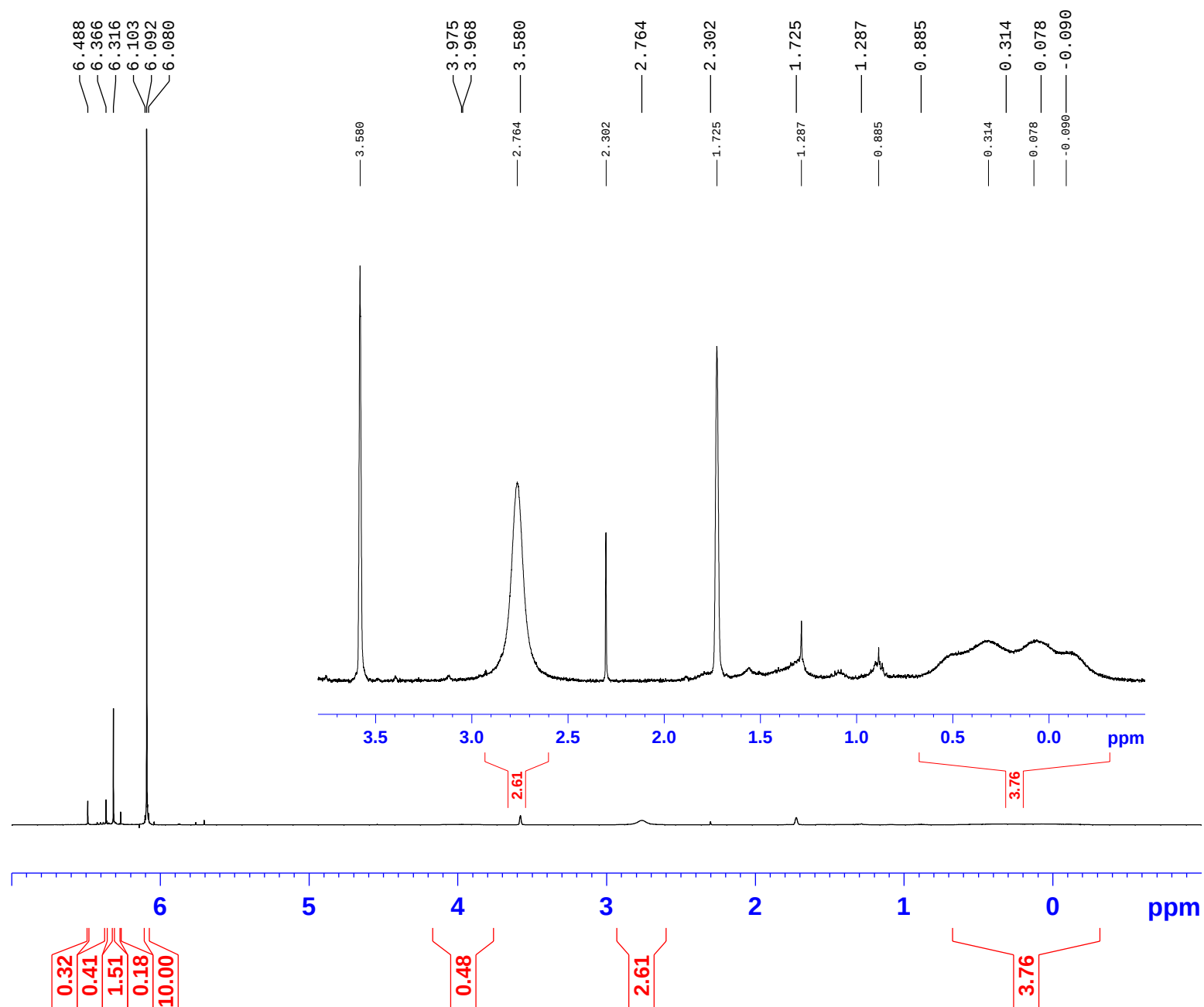
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, $^{11}\text{B}\{^1\text{H}\}$ NMR in C_6D_6 , 20 °C



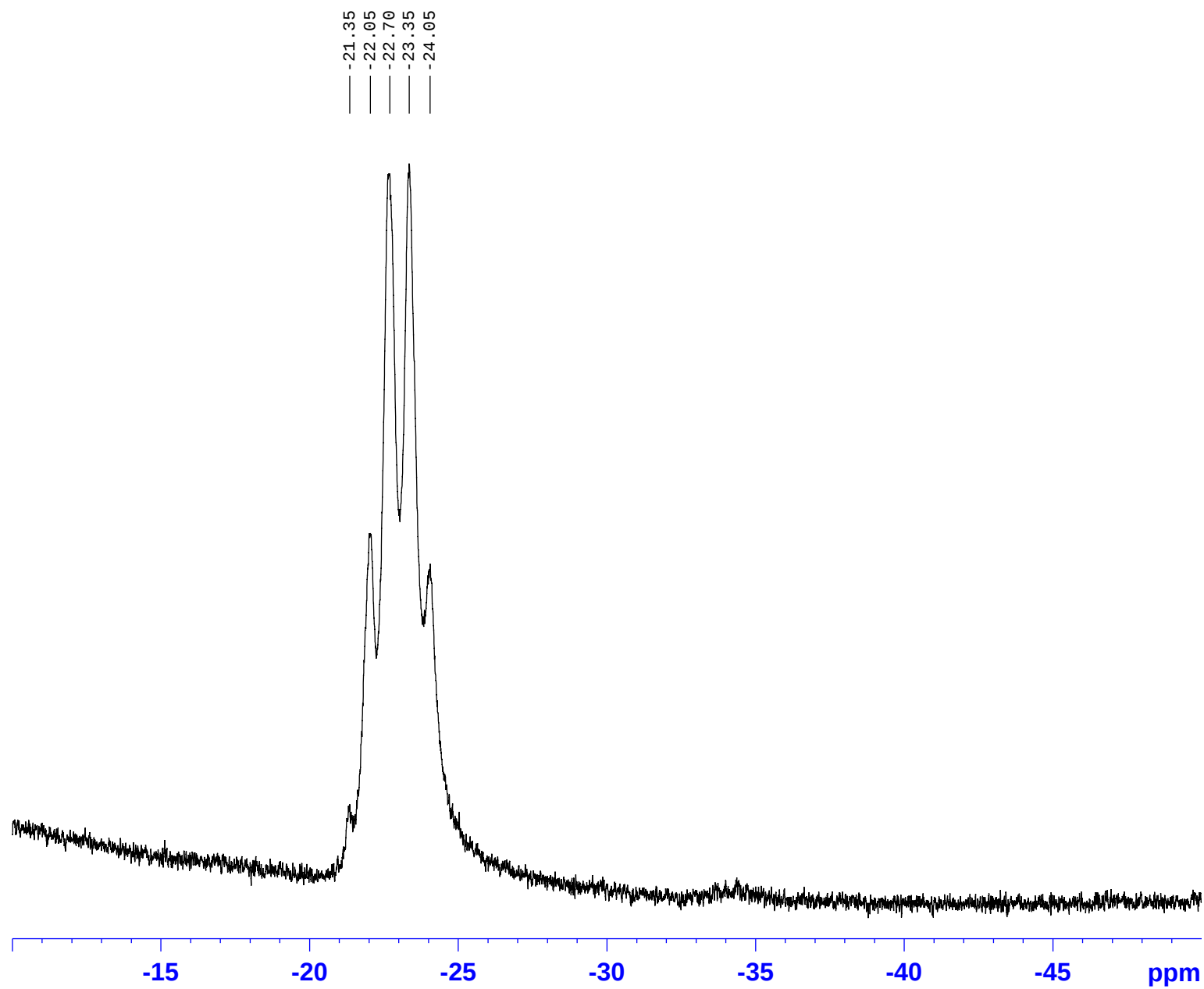
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, ^{11}C NMR in C_6D_6 , 20 °C



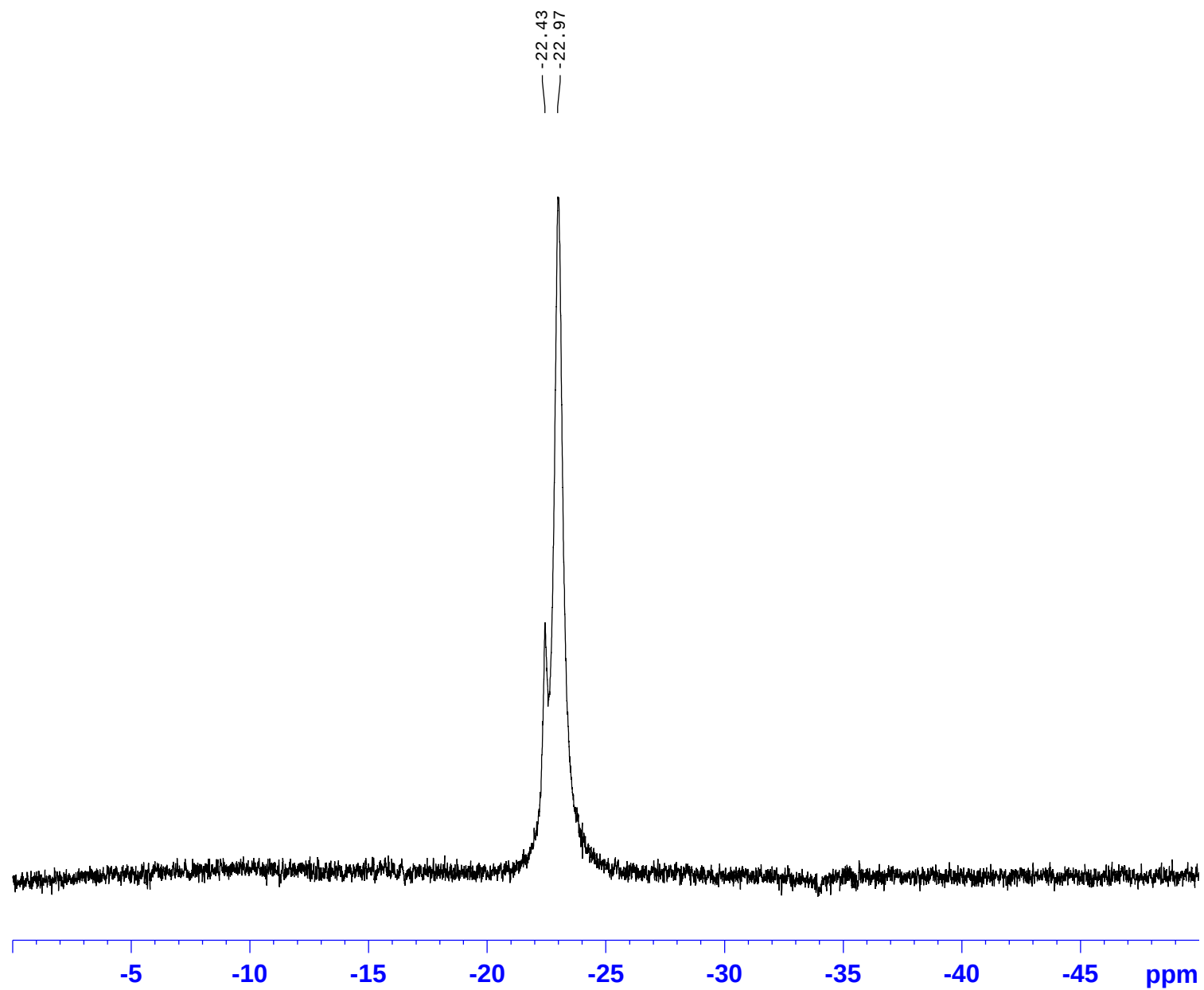
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, ^1H NMR in THF-d_8 , 20 °C



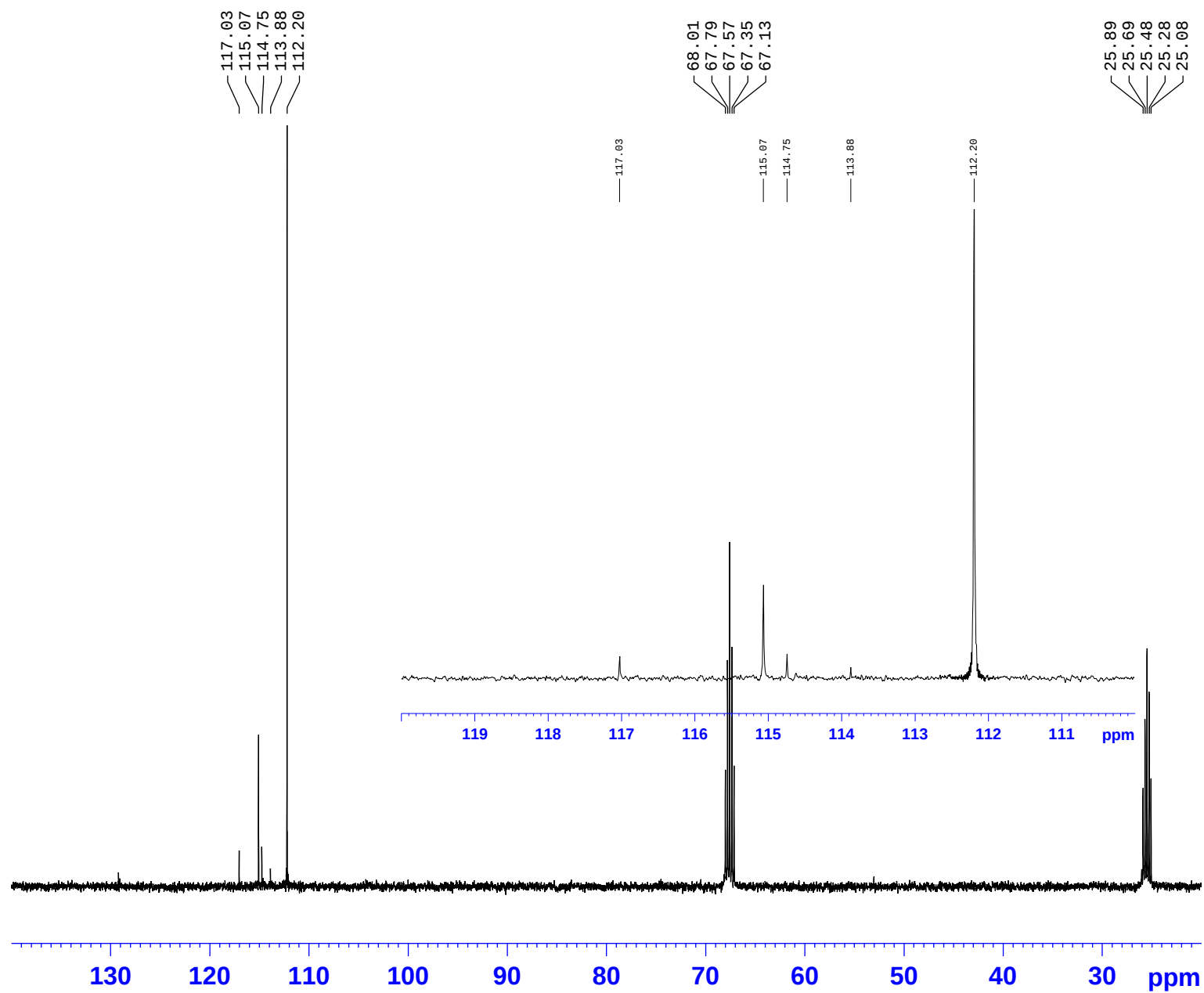
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, ^{11}B NMR in THF-d_8 , 20 °C



$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, $^{11}\text{B}\{^1\text{H}\}$ NMR in THF- d_8 , 20 °C



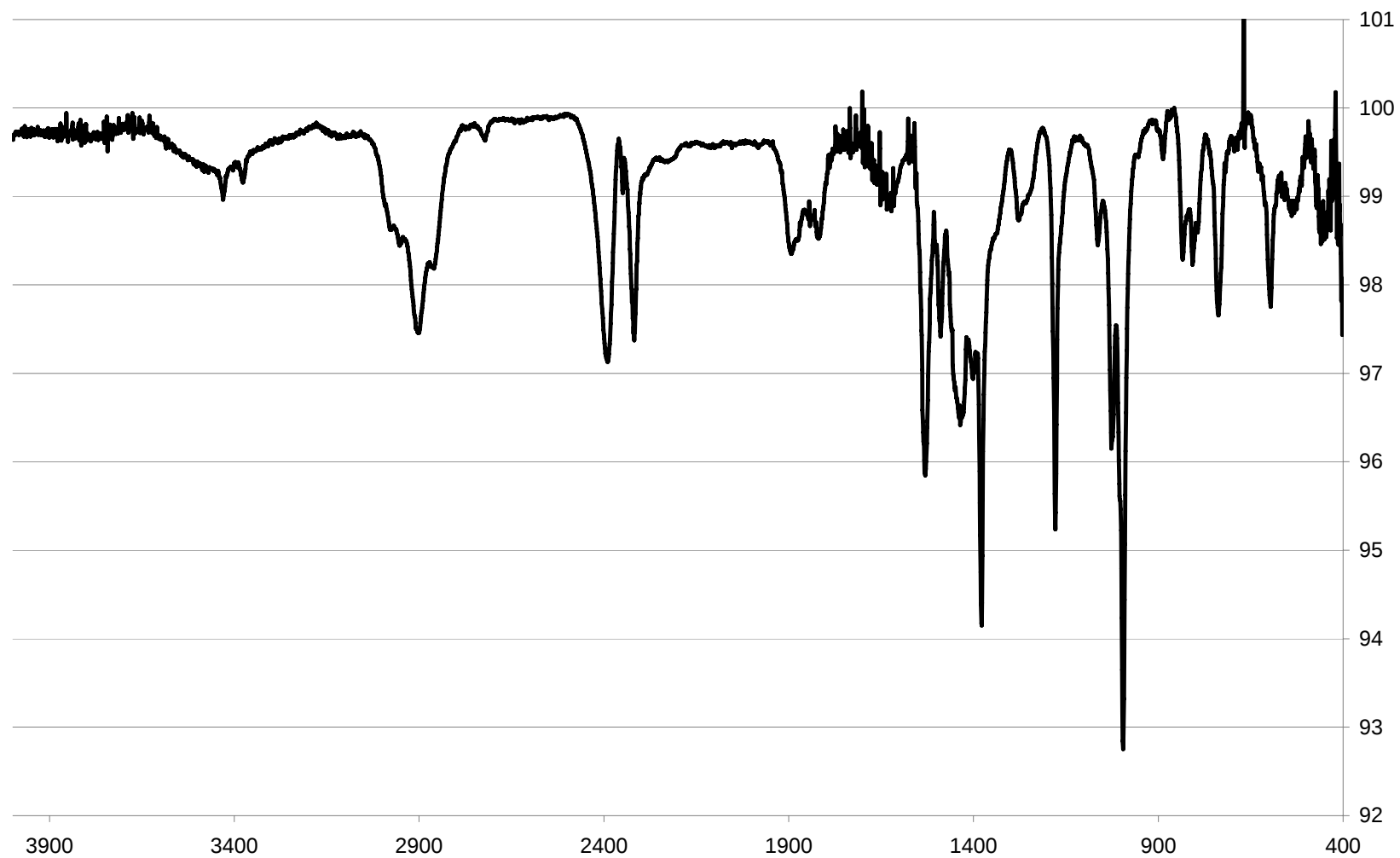
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, $^{13}\text{C}\{^1\text{H}\}$ NMR in THF-d_8 , 20 °C



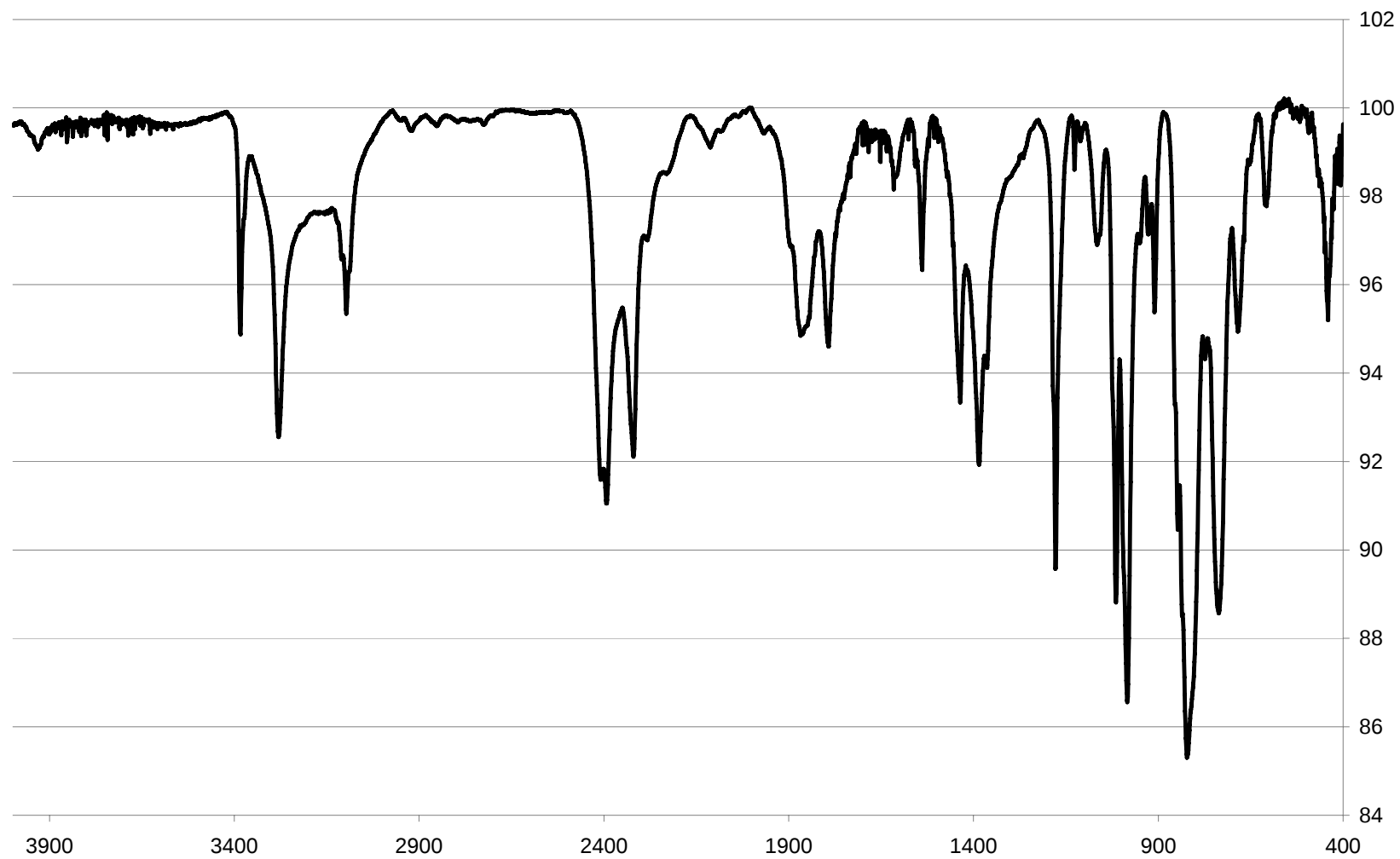
$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, IR (KBr)

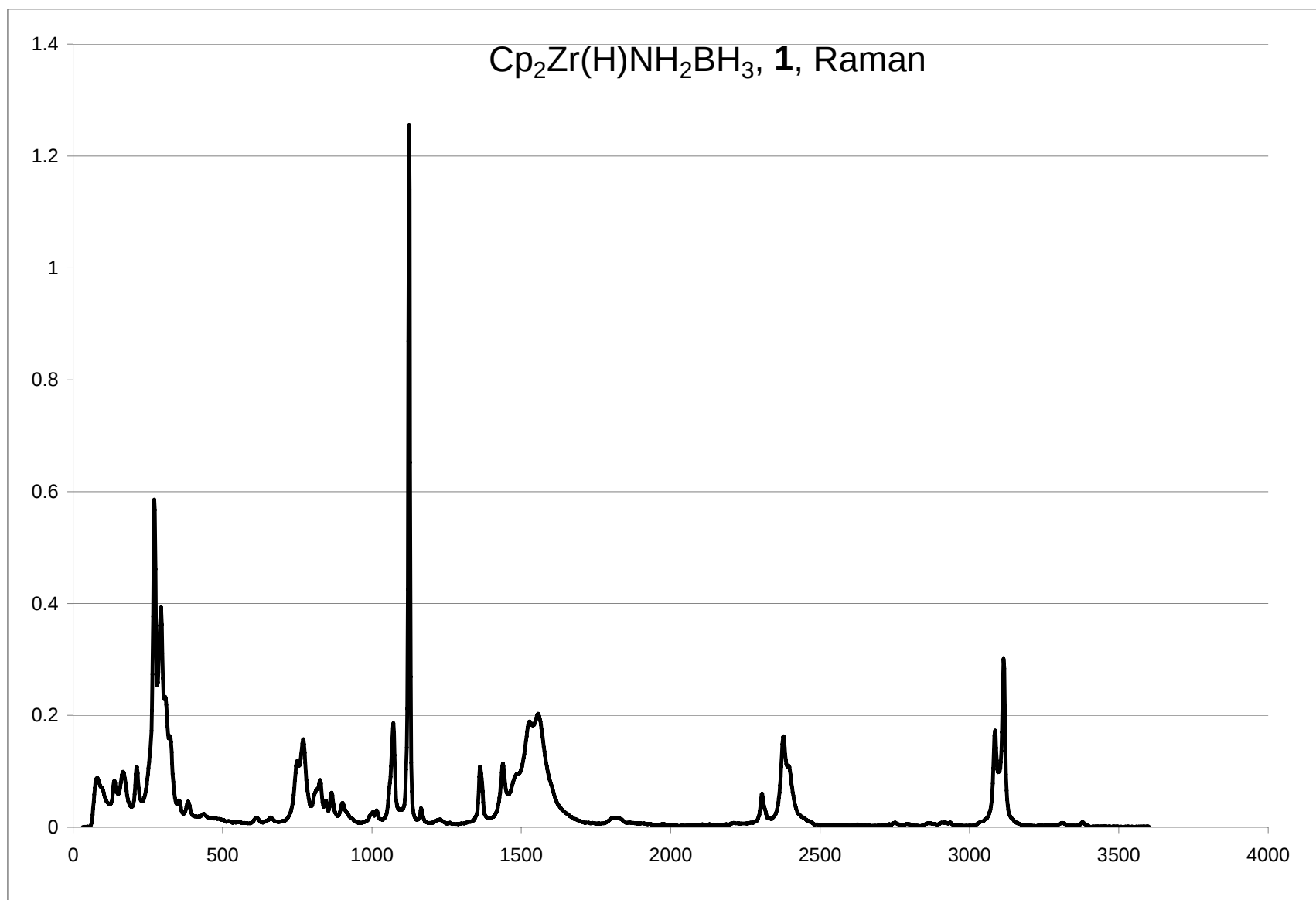


$\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**, IR (KBr)

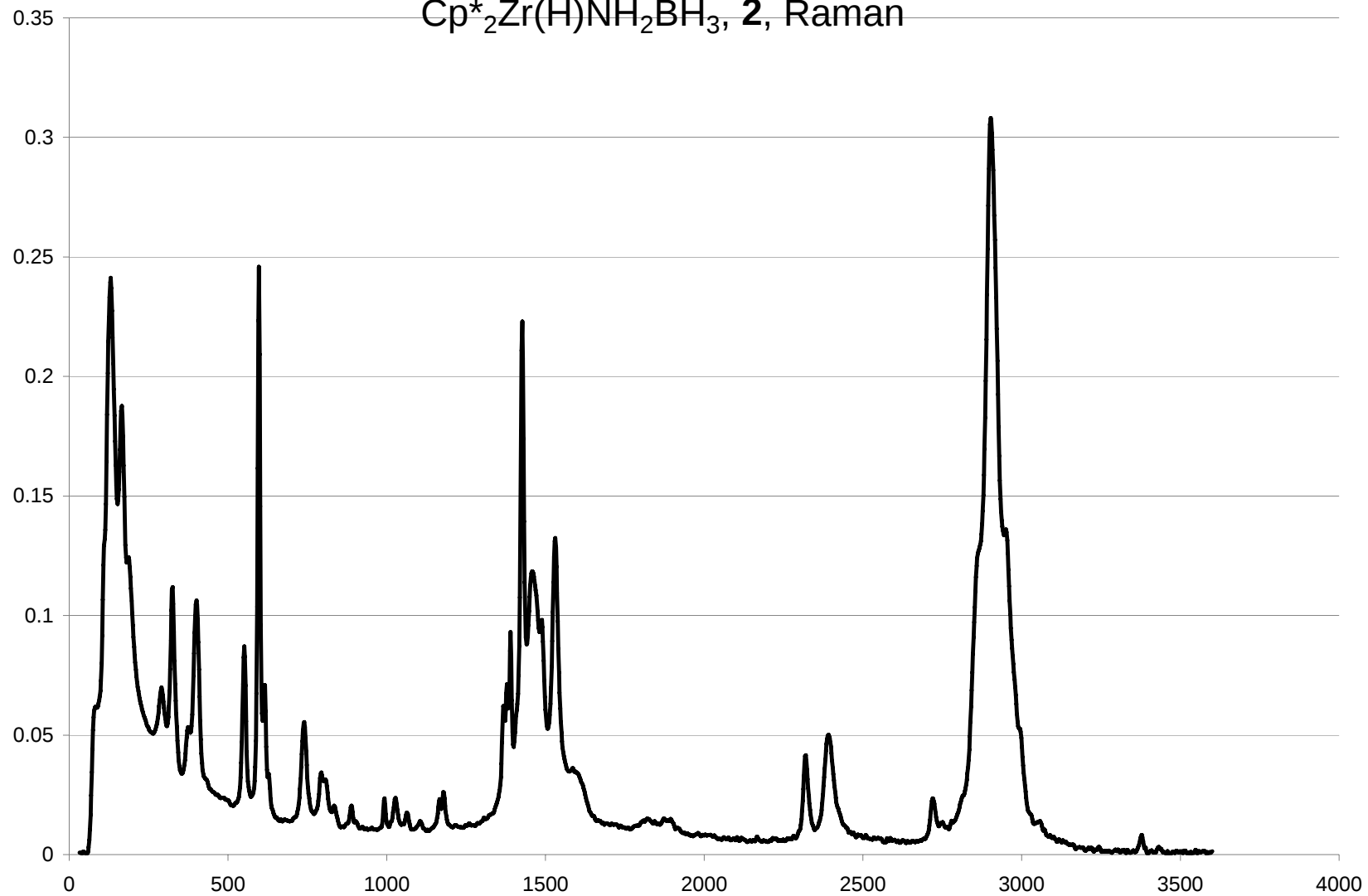


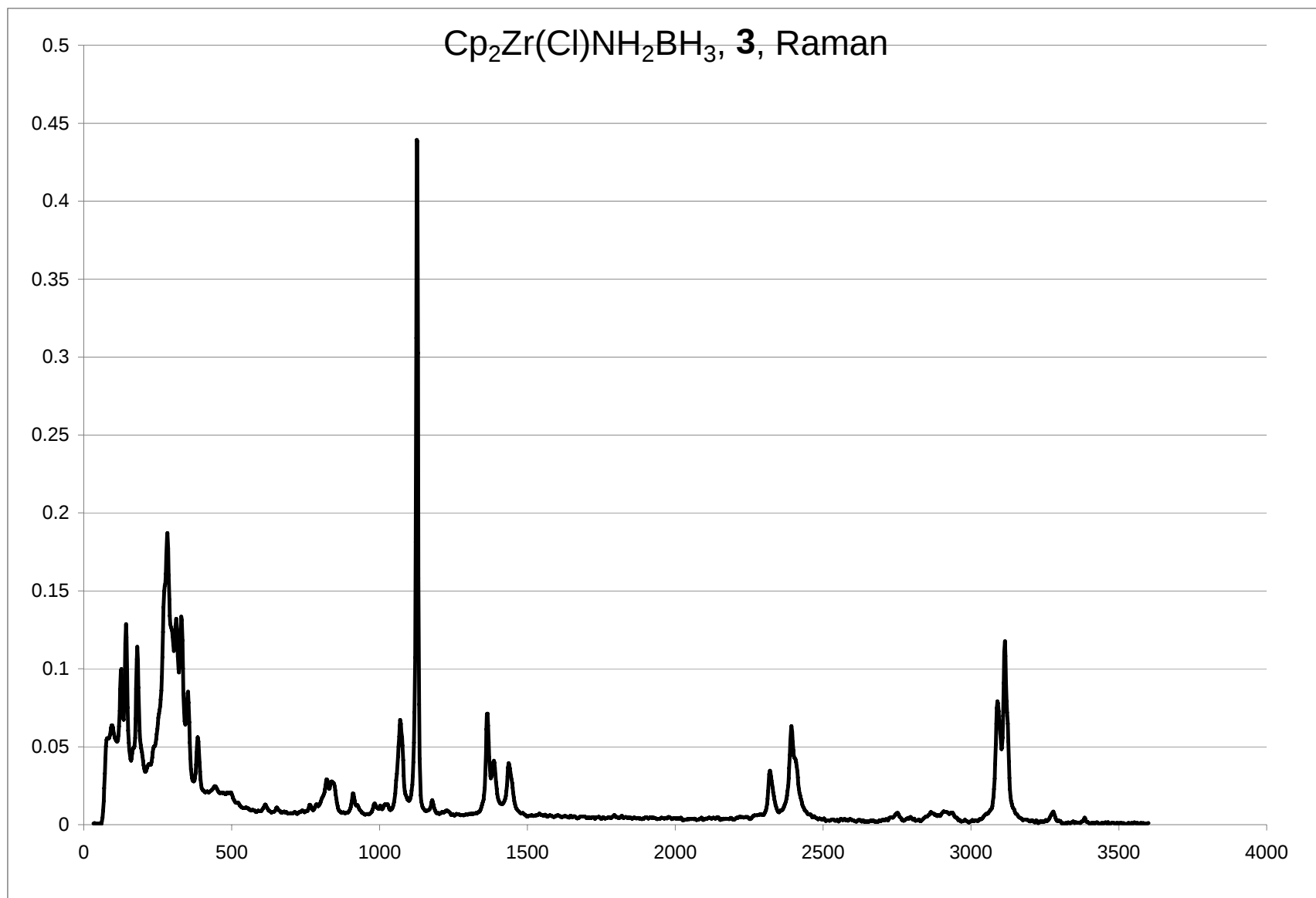
$\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3**, IR (KBr)



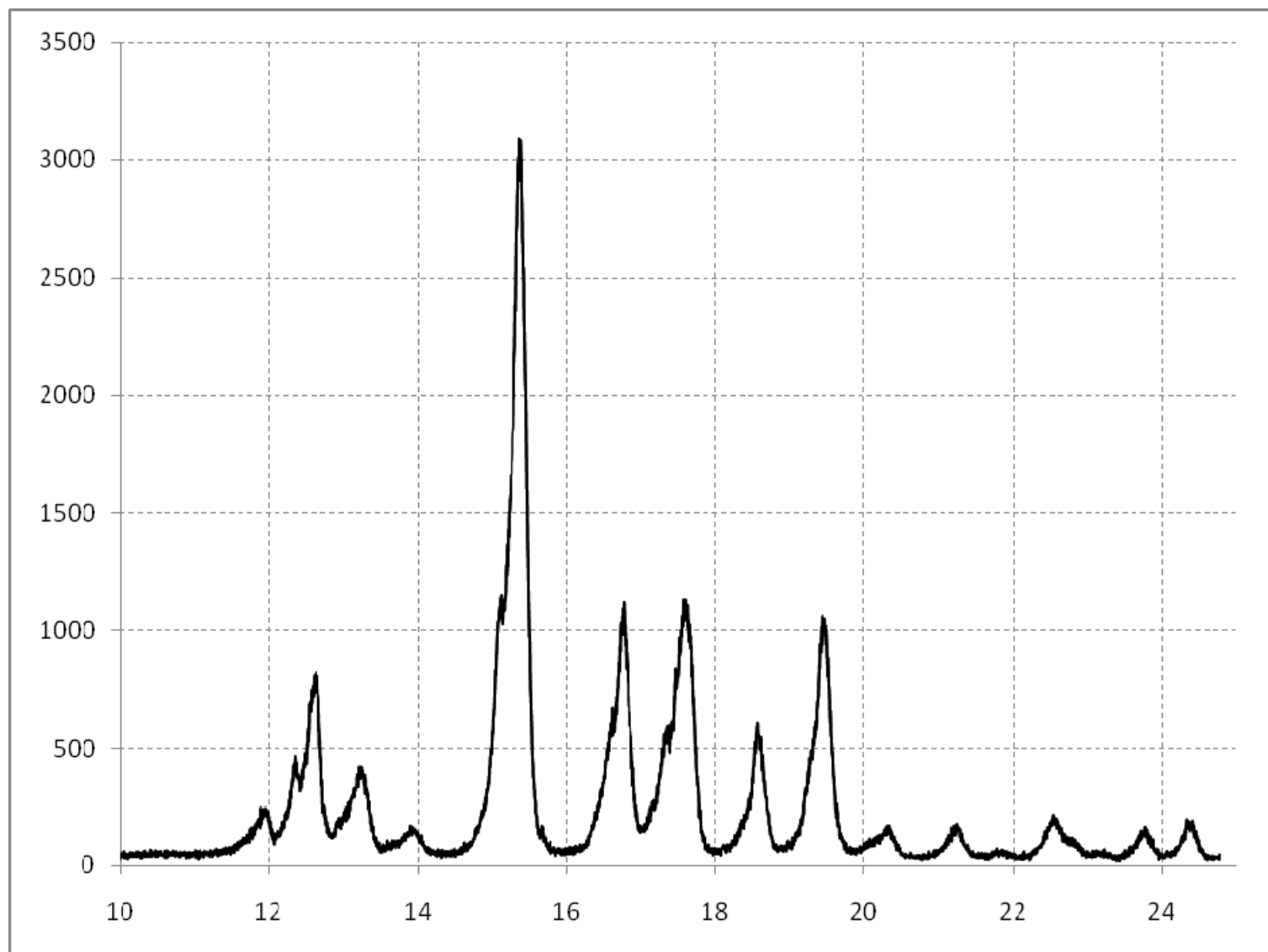


$\text{Cp}^*_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **2**, Raman





$\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1**, powder X-ray pattern



Single crystal X-ray diffraction data for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bII**

A colorless needle crystal of **1bII**, was coated with Paratone 8277 oil (Exxon) and mounted on a glass fiber. All measurements were made on a Nonius KappaCCD diffractometer with graphite monochromated Mo-K_α radiation. Details of crystal data, data collection^{1,2} and structure refinement have been provided in Table 1. The data were corrected for Lorentz and polarization effects and for absorption using multi-scan method¹.

The structure was solved by the direct methods³ and expanded using Fourier techniques⁴. The non-hydrogen atoms were refined anisotropically. The hydrogen atoms on nitrogen, boron, and zirconium were refined isotropically, and the cyclopentadienyl hydrogen atoms were included at geometrically idealized positions and were not refined. The final cycle of full-matrix least-squares refinement using SHELXL97⁵ converged with unweighted and weighted agreement factors, $R = 0.0323$ and $wR = 0.0665$ (all data), respectively, and goodness of fit, $S = 1.048$. The weighting scheme was based on counting statistics and the final difference map had no chemically significant features.

Table 1. Crystal data and structure refinement for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bII**.

Identification code	tf840	
Empirical formula	$\text{C}_{20}\text{H}_{32}\text{B}_2\text{N}_2\text{Zr}_2$	
Formula weight	504.54	
Temperature	173(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	C 1 2/c 1	
Unit cell dimensions	$a = 16.4630(6)$ Å	$\alpha = 90^\circ$.
	$b = 15.4550(5)$ Å	$\beta = 123.341(2)^\circ$.
	$c = 10.1380(3)$ Å	$\gamma = 90^\circ$.
Volume	$2154.92(12)$ Å ³	
Z	4	
Density (calculated)	1.555 Mg/m ³	
Absorption coefficient	0.973 mm ⁻¹	
F(000)	1024	
Crystal size	0.14 x 0.08 x 0.07 mm ³	
Theta range for data collection	2.41 to 27.48°.	
Index ranges	-21 ≤ h ≤ 21, -19 ≤ k ≤ 20, -13 ≤ l ≤ 13	
Reflections collected	9977	
Independent reflections	2459 [R(int) = 0.0682]	
Completeness to theta = 27.48°	99.6 %	
Absorption correction	'Multi-scan'	
Max. and min. transmission	0.9350 and 0.8758	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2459 / 0 / 143	
Goodness-of-fit on F ²	1.048	
Final R indices [I > 2sigma(I)]	R1 = 0.0268, wR2 = 0.0640	
R indices (all data)	R1 = 0.0323, wR2 = 0.0665	
Extinction coefficient	0.0025(4)	
Largest diff. peak and hole	0.484 and -0.524 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bII**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	$U(\text{eq})$
C(6)	3955(2)	1574(2)	8690(4)	48(1)
C(7)	3671(2)	1094(2)	7319(3)	41(1)
C(8)	3620(2)	227(2)	7650(3)	35(1)
C(9)	3842(2)	181(2)	9192(3)	39(1)
C(10)	4054(2)	1013(2)	9824(3)	45(1)
N(1)	1460(1)	-61(1)	5894(2)	23(1)
B(1)	1337(2)	-445(2)	7166(3)	27(1)
C(1)	1032(2)	1875(2)	5284(3)	39(1)
C(2)	626(2)	1690(2)	6162(3)	38(1)
C(3)	1154(2)	2154(2)	7591(3)	33(1)
C(4)	1898(2)	2605(2)	7610(3)	33(1)
C(5)	1814(2)	2441(2)	6174(3)	37(1)
Zr(1)	2288(1)	1031(1)	7653(1)	19(1)

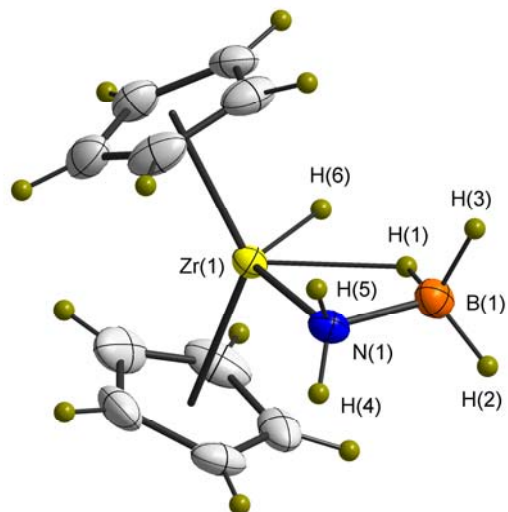


Figure 1. Solid-state molecular structures of **1bII** with thermal ellipsoids at 50% probability.

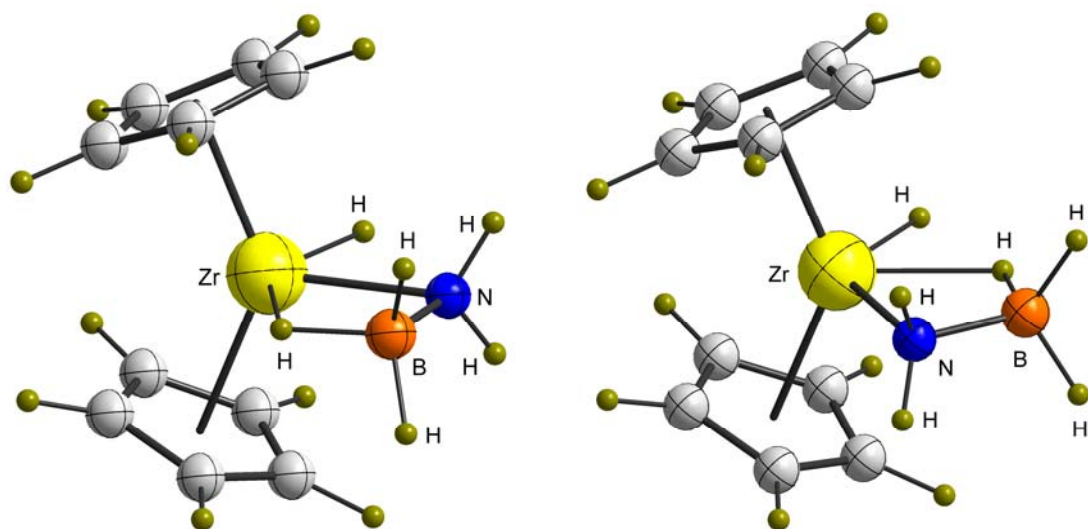


Figure 2. Optimized molecular structures of **1a** (left) and **1b** (right).

Table 3. Bond lengths [Å] and angles [°] for Cp₂Zr(H)NH₂BH₃, **1bII**.

N(1)-Zr(1)	2.2843(18)	N(1)-B(1)	1.531(3)
N(1)-H(4)	0.86(3)	N(1)-H(5)	0.84(3)
B(1)-Zr(1)	2.657(2)	B(1)-H(3)	1.11(3)
B(1)-H(1)	1.24(2)	B(1)-H(2)	1.11(3)
C-C	1.376(4) - 1.409(4)	C-Zr	2.480(2) - 2.524(2)
B(1)-N(1)-Zr(1)	85.86(12)	H(4)-N(1)-H(5)	111(2)
B(1)-N(1)-H(4)	116.5(16)	B(1)-N(1)-H(5)	112.7(17)
Zr(1)-N(1)-H(4)	116.7(17)	Zr(1)-N(1)-H(5)	112.6(17)
N(1)-B(1)-Zr(1)	59.05(10)	Zr(1)-B(1)-H(1)	45.0(11)
Zr(1)-B(1)-H(3)	117.0(13)	Zr(1)-B(1)-H(2)	124.8(17)
N(1)-B(1)-H(1)	103.9(11)	N(1)-B(1)-H(2)	115.3(16)
H(1)-B(1)-H(2)	104(2)	N(1)-B(1)-H(3)	112.9(12)
H(1)-B(1)-H(3)	104.6(17)	H(2)-B(1)-H(3)	115(2)
N(1)-Zr(1)-B(1)	35.09(7)	N(1)-Zr(1)-H(1)	61.4(7)
B(1)-Zr(1)-H(1)	26.3(7)	N(1)-Zr(1)-H(6)	123.0(8)
B(1)-Zr(1)-H(6)	87.9(8)	H(1)-Zr(1)-H(6)	61.8(10)
C-C-C	107.5(2) - 108.3(2)		

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr(H)NH}_2\text{BH}_3$, **1bII**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
C(6)	30(1)	40(1)	79(2)	-19(1)	32(1)	-15(1)
C(7)	34(1)	57(2)	46(1)	14(1)	31(1)	7(1)
C(8)	20(1)	41(1)	44(1)	-13(1)	18(1)	1(1)
C(9)	22(1)	51(2)	44(1)	19(1)	19(1)	11(1)
C(10)	18(1)	81(2)	29(1)	-16(1)	8(1)	-3(1)
N(1)	20(1)	26(1)	22(1)	-5(1)	12(1)	0(1)
B(1)	25(1)	27(1)	29(1)	-4(1)	16(1)	-4(1)
C(1)	46(2)	34(1)	23(1)	2(1)	10(1)	16(1)
C(2)	25(1)	29(1)	44(1)	-3(1)	10(1)	6(1)
C(3)	34(1)	29(1)	39(1)	-1(1)	23(1)	10(1)
C(4)	41(1)	24(1)	33(1)	-4(1)	18(1)	2(1)
C(5)	51(2)	27(1)	36(1)	7(1)	27(1)	9(1)
Zr(1)	18(1)	21(1)	17(1)	-2(1)	10(1)	0(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bII**.

	x	y	z	U(eq)
H(12)	4060	2182	8810	58
H(13)	3538	1319	6348	49
H(14)	3462	-247	6953	42
H(15)	3847	-330	9718	46
H(16)	4236	1167	10860	55
H(1)	1810(18)	52(16)	8310(30)	32(6)
H(2)	1690(20)	-1084(18)	7630(40)	49(9)
H(3)	577(19)	-399(16)	6860(30)	31(6)
H(4)	1779(19)	-367(16)	5620(30)	27(6)
H(5)	932(19)	105(16)	5100(30)	25(6)
H(6)	2450(20)	1082(15)	9500(30)	32(7)
H(7)	815	1655	4266	47
H(8)	90	1318	5847	45
H(9)	1027	2159	8399	39
H(10)	2376	2960	8445	40
H(11)	2218	2674	5862	44

Single crystal X-ray diffraction data for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bI**

A colorless needle crystal of **1bI**, was coated with Paratone 8277 oil (Exxon) and mounted on a glass fiber. All measurements were made on a Nonius KappaCCD diffractometer with graphite monochromated Mo-K α radiation. Details of crystal data, data collection^{1,2} and structure refinement have been provided in Table 6. The data were corrected for Lorentz and polarization effects and for absorption using multi-scan method¹.

The structure was solved by the direct methods³ and expanded using Fourier techniques⁴. The non-hydrogen atoms were refined anisotropically. All hydrogen atoms were refined isotropically. The final cycle of full-matrix least-squares refinement using SHELXL97⁵ converged with unweighted and weighted agreement factors, $R = 0.0819$ and $wR = 0.0841$ (all data), respectively, and goodness of fit, $S = 1.144$. The weighting scheme was based on counting statistics and the final difference map had no chemically significant features.

Table 6. Crystal data and structure refinement for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bI**.

Identification code	tf903	
Empirical formula	$\text{C}_{10} \text{H}_{16} \text{B N Zr}$	
Formula weight	252.27	
Temperature	173(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	$P 1 2_1/c 1$	
Unit cell dimensions	$a = 7.9840(5) \text{ Å}$	$\alpha = 90.000(3)^\circ$.
	$b = 14.8370(5) \text{ Å}$	$\beta = 117.830(3)^\circ$.
	$c = 10.3050(5) \text{ Å}$	$\gamma = 90.000(2)^\circ$.
Volume	$1079.52(9) \text{ Å}^3$	
Z	4	
Density (calculated)	1.552 Mg/m^3	
Absorption coefficient	0.971 mm^{-1}	
F(000)	512	
Crystal size	$0.06 \times 0.02 \times 0.02 \text{ mm}^3$	
Theta range for data collection	2.62 to 27.58° .	
Index ranges	$-10 \leq h \leq 10$, $-19 \leq k \leq 19$, $-13 \leq l \leq 13$	
Reflections collected	16239	
Independent reflections	2488 [$R(\text{int}) = 0.1289$]	
Completeness to $\theta = 27.58^\circ$	99.3 %	
Absorption correction	Multi-scan	
Max. and min. transmission	0.9808 and 0.9440	
Refinement method	Full-matrix least-squares on F^2	
Data / restraints / parameters	2488 / 0 / 182	
Goodness-of-fit on F^2	1.144	
Final R indices [$I > 2\sigma(I)$]	$R1 = 0.0536$, $wR2 = 0.0762$	
R indices (all data)	$R1 = 0.0819$, $wR2 = 0.0841$	
Largest diff. peak and hole	0.636 and -0.648 e.Å^{-3}	

Table 7. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bI**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
C(1)	3168(7)	2210(4)	2651(7)	41(1)
C(2)	3272(7)	1302(4)	2956(6)	43(1)
C(3)	2701(7)	836(4)	1637(7)	38(1)
C(4)	2258(6)	1465(4)	540(5)	36(1)
C(5)	2557(7)	2319(4)	1172(6)	38(1)
C(6)	7233(9)	238(4)	3802(7)	47(2)
C(7)	8607(8)	908(4)	4308(6)	39(1)
C(8)	9119(8)	1079(4)	3217(6)	39(1)
C(9)	8087(8)	513(4)	2029(7)	45(2)
C(10)	6913(9)	7(4)	2396(7)	50(2)
B(1)	7095(9)	3225(4)	2207(6)	33(1)
N(1)	7161(6)	2822(3)	3607(5)	27(1)
Zr(1)	5760(1)	1579(1)	2210(1)	20(1)

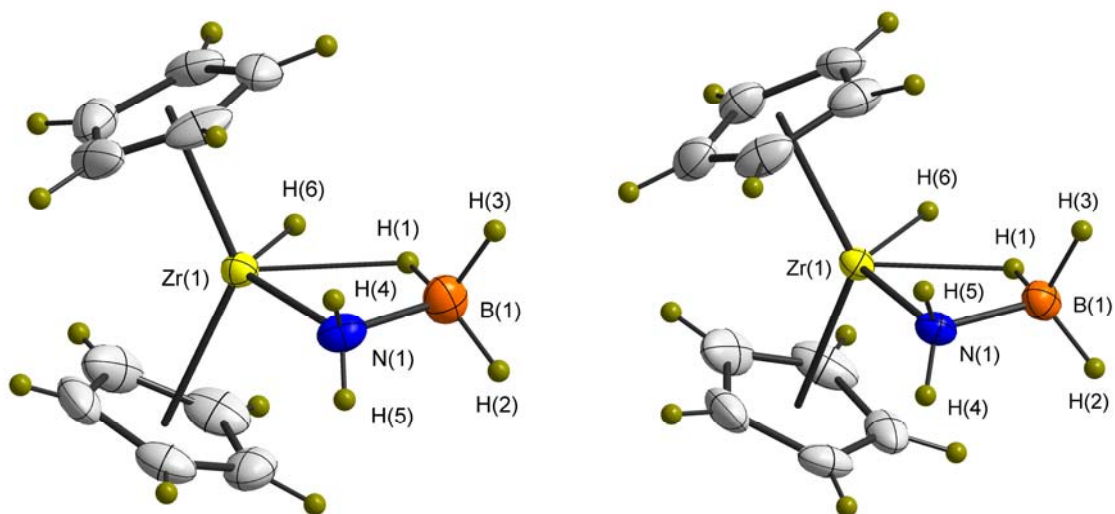


Figure 3. Solid-state molecular structures of **1bI** (left) and **1bII** (right) with thermal ellipsoids at 50% probability.

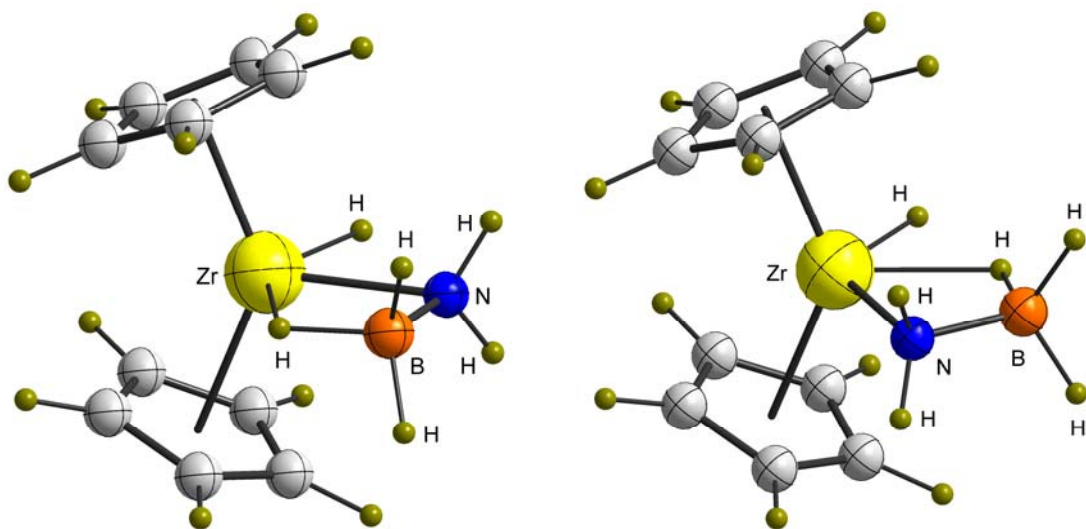


Figure 4. Optimized molecular structures of **1a** (left) and **1b** (right).

Table 8. Bond lengths [Å] and angles [°] for Cp₂Zr(H)NH₂BH₃, **1bI**.

B(1)-N(1)	1.539(7)	N(1)-Zr(1)	2.286(4)
B(1)-Zr(1)	2.665(5)	B(1)-H(1)	1.22(4)
N(1)-H(4)	0.81(5)	B(1)-H(2)	1.09(5)
N(1)-H(5)	0.81(6)	B(1)-H(3)	1.10(4)
Zr(1)-H(6)	1.71(4)	Zr(1)-H(1)	2.02(4)
C-Zr	2.477(5) - 2.517(5)	C-C	1.375(8) - 1.398(8)
N(1)-Zr(1)-H(1)	60.9(13)	B(1)-Zr(1)-H(1)	25.7(13)
N(1)-Zr(1)-H(6)	125.1(14)	B(1)-Zr(1)-H(6)	90.0(14)
B(1)-N(1)-Zr(1)	86.0(3)	H(1)-Zr(1)-H(6)	64.4(19)
H(1)-B(1)-H(3)	104(3)	H(2)-B(1)-H(3)	111(3)
Zr(1)-N(1)-H(4)	116(3)	Zr(1)-N(1)-H(5)	119(4)
B(1)-N(1)-H(4)	115(3)	H(4)-N(1)-H(5)	110(5)
B(1)-N(1)-H(5)	108(4)	N(1)-B(1)-Zr(1)	58.8(2)
Zr(1)-B(1)-H(1)	46(2)	Zr(1)-B(1)-H(3)	122(2)
Zr(1)-B(1)-H(2)	125(3)	H(1)-B(1)-H(2)	109(3)
N(1)-B(1)-H(1)	105(2)	N(1)-B(1)-H(3)	115(2)
N(1)-B(1)-H(2)	112(3)	C(10)-C(9)-C(8)	106.6(6) - 109.4(6)

Table 9. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bI**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
C(1)	21(3)	59(4)	48(4)	-24(3)	20(3)	-3(2)
C(2)	24(3)	78(4)	32(3)	18(3)	17(2)	5(3)
C(3)	30(3)	30(3)	59(4)	3(3)	24(3)	1(2)
C(4)	19(2)	59(4)	26(3)	-8(3)	7(2)	-2(2)
C(5)	23(3)	34(3)	51(4)	9(3)	14(2)	7(2)
C(6)	43(4)	41(3)	61(4)	30(3)	26(3)	18(3)
C(7)	36(3)	44(3)	33(3)	3(3)	12(3)	19(2)
C(8)	27(3)	41(3)	52(4)	1(3)	20(3)	5(2)
C(9)	46(4)	50(4)	42(3)	0(3)	24(3)	25(3)
C(10)	42(4)	26(3)	57(4)	-6(3)	2(3)	3(3)
B(1)	38(3)	28(3)	30(3)	-4(2)	13(3)	-5(2)
N(1)	23(2)	32(2)	27(2)	-7(2)	12(2)	1(2)
Zr(1)	19(1)	24(1)	19(1)	1(1)	9(1)	1(1)

Table 10. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$)
for $\text{Cp}_2\text{Zr}(\text{H})\text{NH}_2\text{BH}_3$, **1bI**.

	x	y	z	U(eq)
H(1)	6370(60)	2640(30)	1280(50)	35(13)
H(2)	8510(70)	3370(30)	2340(50)	50(15)
H(3)	6160(60)	3810(30)	1750(50)	34(13)
H(4)	6550(70)	3090(30)	3930(50)	34(15)
H(5)	8260(80)	2790(40)	4230(60)	54(19)
H(6)	5330(60)	1430(30)	430(50)	35(13)
H(7)	3360(80)	2600(40)	3230(60)	52(19)
H(8)	3650(70)	1010(30)	3830(50)	38(14)
H(9)	2520(60)	240(30)	1480(50)	27(13)
H(10)	1790(60)	1340(30)	-400(50)	32(13)
H(11)	2410(70)	2830(30)	720(50)	41(15)
H(12)	6700(90)	50(40)	4300(70)	90(30)
H(13)	9070(70)	1160(30)	5140(50)	34(15)
H(14)	9970(70)	1480(40)	3270(50)	49(16)
H(15)	8200(90)	510(40)	1220(70)	70(20)
H(16)	6210(70)	-350(30)	1880(50)	32(15)

Single crystal X-ray diffraction data for Cp*₂Zr(H)NH₂BH₃, **2**

Severe disorder of the Cp* groups prevented the location of the hydrogen atoms on the Fourier map and the structure was not included in this study.

Table 11. Crystal data and structure refinement for Cp*₂Zr(H)NH₂BH₃, **2**.

Identification code	tf909
Empirical formula	C ₂₀ H ₃₅ B N Zr
Formula weight	391.52
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	Orthorhombic
Space group	P 2 ₁ 2 ₁ 2 ₁
Unit cell dimensions	a = 8.1330(2) Å α = 90°. b = 11.1820(4) Å β = 90°. c = 22.3460(9) Å γ = 90°.
Volume	2032.22(12) Å ³
Z	8
Density (calculated)	2.559 Mg/m ³
Absorption coefficient	1.082 mm ⁻¹
F(000)	1656

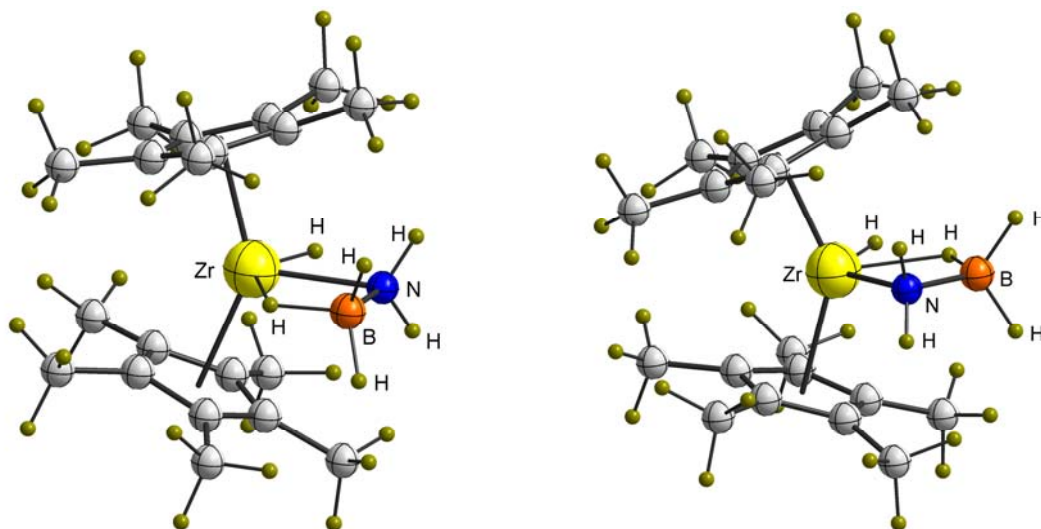


Figure 5. Optimized molecular structures of **2a** (left) and **2b** (right).

Single crystal X-ray diffraction data for $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3a**

A colorless needle crystal of **3a** was coated with Paratone 8277 oil (Exxon) and mounted on a glass fiber. All measurements were made on a Nonius KappaCCD diffractometer with graphite monochromated Mo-K α radiation. Details of crystal data, data collection^{1,2} and structure refinement have been provided in Table 12. The data were corrected for Lorentz and polarization effects and for absorption using multi-scan method¹.

The structure was solved by the direct methods³ and expanded using Fourier techniques⁴. The non-hydrogen atoms were refined anisotropically. All hydrogen atoms were refined isotropically. The final cycle of full-matrix least-squares refinement using SHELXL97⁵ converged with unweighted and weighted agreement factors, $R = 0.0624$ and $wR = 0.0781$ (all data), respectively, and goodness of fit, $S = 1.042$. The weighting scheme was based on counting statistics and the final difference map had no chemically significant features.

Table 12. Crystal data and structure refinement for $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3a**.

Identification code	tf823	
Empirical formula	$\text{C}_{10} \text{H}_{15} \text{B Cl N Zr}$	
Formula weight	286.71	
Temperature	173(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	$P 1 2_1/c 1$	
Unit cell dimensions	$a = 8.1000(4) \text{ Å}$	$\alpha = 90^\circ$.
	$b = 10.7920(6) \text{ Å}$	$\beta = 119.120(3)^\circ$.
	$c = 15.0870(6) \text{ Å}$	$\gamma = 90^\circ$.
Volume	$1152.14(10) \text{ Å}^3$	
Z	4	
Density (calculated)	1.653 Mg/m^3	
Absorption coefficient	1.145 mm^{-1}	
F(000)	576	
Crystal size	$0.10 \times 0.05 \times 0.04 \text{ mm}^3$	
Theta range for data collection	3.03 to 27.45° .	
Index ranges	$-9 \leq h \leq 10$, $-14 \leq k \leq 13$, $-19 \leq l \leq 19$	
Reflections collected	10749	
Independent reflections	2626 [$R(\text{int}) = 0.0704$]	
Completeness to $\theta = 27.45^\circ$	99.7 %	
Absorption correction	Multi-scan	
Max. and min. transmission	0.9556 and 0.8941	
Refinement method	Full-matrix least-squares on F^2	
Data / restraints / parameters	2626 / 0 / 188	
Goodness-of-fit on F^2	1.042	
Final R indices [$I > 2\sigma(I)$]	$R1 = 0.0354$, $wR2 = 0.0691$	
R indices (all data)	$R1 = 0.0624$, $wR2 = 0.0781$	
Extinction coefficient	$0.0071(9)$	
Largest diff. peak and hole	0.636 and -0.983 e.Å^{-3}	

Table 13. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3a**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
C(1)	7946(5)	7732(4)	2618(2)	37(1)
C(2)	7050(5)	6591(4)	2246(3)	37(1)
C(3)	5192(5)	6823(4)	1499(3)	36(1)
C(4)	4943(5)	8119(4)	1394(3)	34(1)
C(5)	6633(5)	8668(3)	2110(3)	35(1)
C(6)	9866(5)	7423(3)	324(3)	31(1)
C(7)	8834(5)	6332(3)	27(3)	33(1)
C(8)	9099(6)	5706(3)	906(3)	40(1)
C(9)	10348(5)	6405(4)	1748(3)	43(1)
C(10)	10789(5)	7469(4)	1394(3)	35(1)
B(1)	4322(6)	6884(4)	-894(3)	32(1)
N(1)	5165(4)	8177(3)	-755(2)	31(1)
Zr(1)	7344(1)	7608(1)	839(1)	21(1)
Cl(1)	8244(1)	9890(1)	810(1)	31(1)

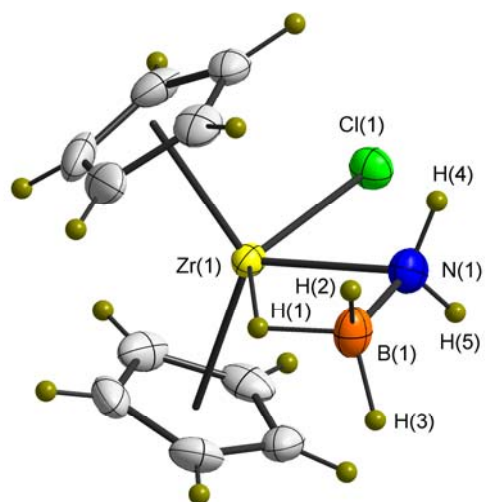


Figure 6. Solid-state molecular structures of **3a** with thermal ellipsoids at 50% probability.

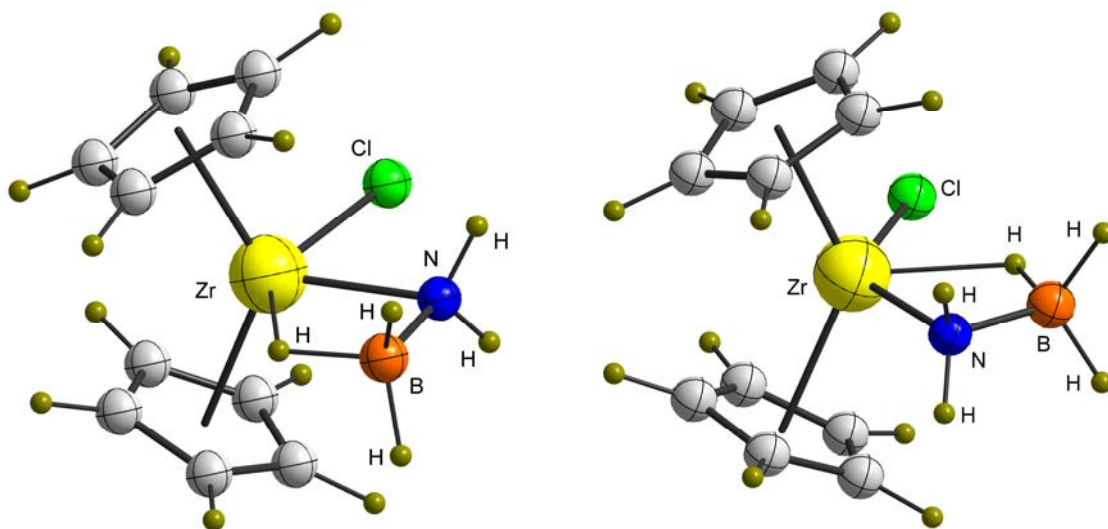


Figure 7. Optimized molecular structures of **3a** (left) and **3b** (right).

Table 14. Bond lengths [Å] and angles [°] for Cp₂Zr(Cl)NH₂BH₃, **3a**

B(1)-Zr(1)	2.685(4)	B(1)-N(1)	1.523(5)
N(1)-Zr(1)	2.268(3)	Zr(1)-Cl(1)	2.5742(8)
N(1)-H(4)	0.98(4)	B(1)-H(1)	1.24(3)
N(1)-H(5)	0.93(5)	B(1)-H(3)	1.13(3)
Zr(1)-H(1)	2.02(3)	B(1)-H(2)	1.09(4)
C(6)-C(10)	1.384(5) - 1.412(5)	C(3)-Zr(1)	2.471(3) - 2.537(3)
N(1)-B(1)-Zr(1)	57.59(15)	B(1)-N(1)-Zr(1)	87.89(19)
N(1)-B(1)-H(1)	102.8(15)	N(1)-B(1)-H(3)	113(2)
Zr(1)-B(1)-H(1)	45.6(15)	Zr(1)-B(1)-H(3)	117.5(16)
H(1)-B(1)-H(3)	107(2)	Zr(1)-B(1)-H(2)	122.6(18)
N(1)-B(1)-H(2)	114.8(19)	H(1)-B(1)-H(2)	101(2)
H(3)-B(1)-H(2)	116(3)	Zr(1)-N(1)-H(4)	112(2)
B(1)-N(1)-H(4)	118(2)	B(1)-N(1)-H(5)	114(3)
Zr(1)-N(1)-H(5)	110(3)	N(1)-Zr(1)-Cl(1)	78.68(8)
H(4)-N(1)-H(5)	112(4)	N(1)-Zr(1)-H(1)	60.4(9)
Cl(1)-Zr(1)-H(1)	139.1(9)	C(7)-C(6)-C(10)	107.4(3) - 108.7(3)

Table 15. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3a**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
C(1)	26(2)	66(3)	18(2)	-2(2)	10(1)	-3(2)
C(2)	45(2)	42(2)	35(2)	17(2)	28(2)	12(2)
C(3)	33(2)	45(2)	37(2)	-5(2)	23(2)	-12(2)
C(4)	27(2)	48(2)	33(2)	7(2)	19(2)	7(2)
C(5)	51(2)	35(2)	39(2)	-7(2)	37(2)	-5(2)
C(6)	27(2)	40(2)	31(2)	4(2)	19(1)	7(2)
C(7)	37(2)	37(2)	33(2)	-6(2)	23(2)	6(2)
C(8)	45(2)	28(2)	60(3)	14(2)	36(2)	17(2)
C(9)	35(2)	62(3)	32(2)	18(2)	16(2)	29(2)
C(10)	20(2)	53(2)	30(2)	-6(2)	10(1)	5(2)
B(1)	29(2)	33(2)	24(2)	1(2)	4(2)	0(2)
N(1)	28(2)	36(2)	25(1)	2(1)	9(1)	1(1)
Zr(1)	18(1)	25(1)	19(1)	1(1)	9(1)	2(1)
Cl(1)	33(1)	26(1)	34(1)	-1(1)	17(1)	-2(1)

Table 16. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cp}_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **3a**.

	x	y	z	U(eq)
H(6)	9110(50)	7820(30)	3080(30)	34(10)
H(7)	7570(40)	5830(30)	2450(20)	21(8)
H(8)	4280(50)	6280(30)	1140(30)	35(10)
H(9)	3910(50)	8500(40)	980(30)	47(11)
H(10)	6880(50)	9510(30)	2220(20)	34(9)
H(11)	9840(50)	8040(40)	-140(30)	46(11)
H(12)	8050(40)	6090(30)	-650(20)	21(8)
H(13)	8490(40)	5020(30)	880(20)	19(8)
H(14)	10690(60)	6180(40)	2420(30)	60(12)
H(15)	11490(50)	8050(40)	1770(30)	46(12)
H(1)	5450(40)	6350(30)	-100(20)	36(9)
H(3)	2910(50)	6880(30)	-910(20)	37(9)
H(2)	4440(50)	6310(30)	-1460(30)	48(10)
H(4)	5750(50)	8410(40)	-1170(30)	47(11)
H(5)	4400(60)	8810(50)	-730(30)	72(15)

$\text{Cp}^*_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **4a** and **4b**

These compounds were not observed experimentally.

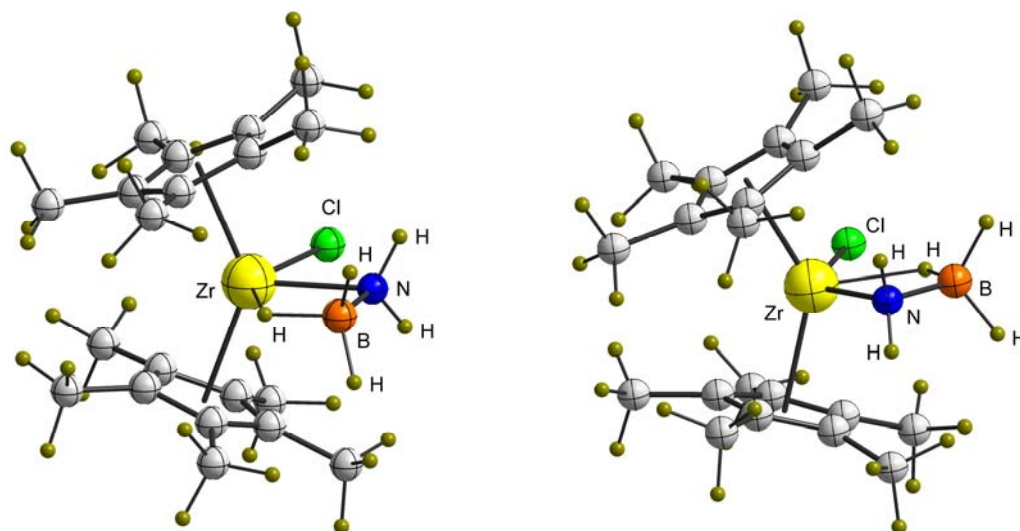


Figure 8. Optimized molecular structures of $\text{Cp}^*_2\text{Zr}(\text{Cl})\text{NH}_2\text{BH}_3$, **4a** (left) and **4b** (right). These compounds were not observed experimentally.

Computational Details

Molecular Structures were optimized using density functional theory. ThePBE1PBE hybrid functional⁶ was used in together with the def2-TZVP basis set.⁷ Frequency calculations were performed on the optimized structures to ensure that they represent stable minima on the potential energy surface. All calculations were performed with the Gaussian 03 program package.⁸

Optimized coordinates of 1a in mol2 format (Figure 2 and 4):

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NO_CHARGES
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  3  C      6.9875      6.3792      0.3272  C.3    1 RES1    0.0000
  4  C      6.8984      6.2147      1.7214  C.3    1 RES1    0.0000
  5  C      7.3892      7.3945      2.3325  C.3    1 RES1    0.0000
  6  ZR     5.2897      8.0486      1.1565  ZR     1 RES1    0.0000
  7  C      3.8235      9.1466      2.8762  C.3    1 RES1    0.0000
  8  C      4.3669      7.9897      3.4717  C.3    1 RES1    0.0000
  9  C      3.8341      6.8580      2.8056  C.3    1 RES1    0.0000
 10  C      2.9617      7.3181      1.8016  C.3    1 RES1    0.0000
 11  C      2.9664      8.7295      1.8326  C.3    1 RES1    0.0000
 12  H      5.7782      9.8035      1.4234  H      1 RES1    0.0000
 13  N      4.7907      9.1356     -0.8016  N.4    1 RES1    0.0000
 14  B      4.3019      7.7803     -1.3113  B      1 RES1    0.0000
 15  H      5.0518      7.9711      4.3062  H      1 RES1    0.0000
 16  H      4.0383      5.8245      3.0423  H      1 RES1    0.0000
 17  H      2.4058      6.7052      1.1087  H      1 RES1    0.0000
 18  H      2.3960      9.3755      1.1814  H      1 RES1    0.0000
 19  H      4.0170     10.1647      3.1717  H      1 RES1    0.0000
 20  H      7.7107      8.0817     -0.9044  H      1 RES1    0.0000
 21  H      6.6824      5.6646     -0.4217  H      1 RES1    0.0000
 22  H      6.5386      5.3350      2.2338  H      1 RES1    0.0000
 23  H      7.4701      7.5743      3.3940  H      1 RES1    0.0000
 24  H      8.2223      9.2602      1.4564  H      1 RES1    0.0000
 25  H      4.4865      6.9431     -0.3292  H      1 RES1    0.0000
 26  H      3.1078      7.7325     -1.4895  H      1 RES1    0.0000
 27  H      4.9867      7.3193     -2.1936  H      1 RES1    0.0000
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  5      2      3      1
  6      2      6      1
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19	6	10	1
20	6	11	1
21	7	8	1
22	7	11	1
23	7	19	1
24	8	9	1
25	8	15	1
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Optimized coordinates of 1b in mol2 format (Figure 2 and 4):

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 10  C       1.1110      1.3935      8.3971  C.3  1 RES1  0.0000
 11  C       1.5997      2.4533      7.6053  C.3  1 RES1  0.0000
 12  N      -0.8334     -0.1455      5.0563  N.4  1 RES1  0.0000
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15	H	0.9155	-0.7804	7.9627	H	1	RES1	0.0000
16	H	0.7280	1.4733	9.4011	H	1	RES1	0.0000
17	H	1.6515	3.4907	7.9002	H	1	RES1	0.0000
18	H	2.4838	2.4700	5.5561	H	1	RES1	0.0000
19	H	-1.3279	2.3187	3.4515	H	1	RES1	0.0000
20	H	-3.3350	2.0417	5.2240	H	1	RES1	0.0000
21	H	-2.7938	3.6081	7.3276	H	1	RES1	0.0000
22	H	-0.4175	4.7673	6.9077	H	1	RES1	0.0000
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24	H	-1.4200	-1.6404	6.7054	H	1	RES1	0.0000
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26	H	-1.7599	0.2601	7.1156	H	1	RES1	0.0000
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28	H	-0.0578	-0.7640	4.8777	H	1	RES1	0.0000
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34	13	24	1
35	13	25	1
36	13	26	1

@<TRIPOS>SUBSTRUCTURE

1	RES1	1
---	------	---

Optimized coordinates of 2a in mol2 format (Figure 5):

```
@<TRIPOS>MOLECULE
Molden generated mol2
59      66      1
```

SMALL

USER_CHARGES

```
@<TRIPOS>ATOM
```

1	C	-2.4815	-0.6762	-0.4963	C.3	1	RES1	0.0387
2	C	-2.1510	0.5368	-1.1628	C.3	1	RES1	0.1347
3	C	-1.7663	1.4800	-0.1775	C.3	1	RES1	-0.0026
4	C	-1.8587	0.8513	1.0986	C.3	1	RES1	0.0167
5	C	-2.3258	-0.4692	0.8993	C.3	1	RES1	0.0806
6	ZR	0.0086	-0.3416	-0.1723	ZR	1	RES1	-0.0947
7	C	1.7482	1.1343	0.9696	C.3	1	RES1	0.0192
8	C	1.8143	1.4279	-0.4220	C.3	1	RES1	0.0487
9	C	2.3007	0.2764	-1.0919	C.3	1	RES1	0.1681
10	C	2.5205	-0.7314	-0.1215	C.3	1	RES1	-0.0820
11	C	2.1733	-0.2034	1.1548	C.3	1	RES1	0.1272
12	N	0.0951	-2.5235	-0.8652	N.4	1	RES1	-0.3723
13	B	-0.0152	-2.9092	0.6104	B	1	RES1	-0.1983
14	H	0.0400	-0.2824	-2.0075	H	1	RES1	-0.0689
15	C	-1.5993	2.9441	-0.4171	C.3	1	RES1	-0.4407
16	C	-1.7938	1.5267	2.4296	C.3	1	RES1	-0.4399
17	C	-2.7707	-1.3862	1.9875	C.3	1	RES1	-0.3956
18	C	-3.0681	-1.8887	-1.1406	C.3	1	RES1	-0.5052
19	C	-2.3902	0.8367	-2.6050	C.3	1	RES1	-0.4214
20	C	2.7043	0.2177	-2.5272	C.3	1	RES1	-0.4292
21	C	3.2056	-2.0352	-0.3659	C.3	1	RES1	-0.4488
22	C	2.4086	-0.8813	2.4639	C.3	1	RES1	-0.4281
23	C	1.5328	2.1270	2.0630	C.3	1	RES1	-0.4460
24	C	1.6724	2.7690	-1.0624	C.3	1	RES1	-0.4632
25	H	-0.0688	-1.7941	1.2671	H	1	RES1	0.0949
26	H	-1.0508	-3.4687	0.8794	H	1	RES1	-0.0486
27	H	0.9628	-3.4616	1.0570	H	1	RES1	-0.0409
28	H	0.9572	-2.7620	-1.3318	H	1	RES1	0.2463
29	H	-0.6721	-2.7964	-1.4601	H	1	RES1	0.2515
30	H	2.7729	-2.8473	0.2209	H	1	RES1	0.1667
31	H	3.1724	-2.3152	-1.4214	H	1	RES1	0.1231
32	H	4.2648	-1.9649	-0.0958	H	1	RES1	0.1392
33	H	2.5738	-0.7795	-2.9486	H	1	RES1	0.1309
34	H	2.1143	0.9038	-3.1344	H	1	RES1	0.1368
35	H	3.7607	0.4899	-2.6398	H	1	RES1	0.1335
36	H	1.7451	-0.5031	3.2435	H	1	RES1	0.1228
37	H	2.2520	-1.9579	2.3899	H	1	RES1	0.1573
38	H	3.4375	-0.7138	2.8028	H	1	RES1	0.1350
39	H	1.1032	1.6731	2.9565	H	1	RES1	0.1343
40	H	2.4896	2.5733	2.3569	H	1	RES1	0.1452
41	H	0.8800	2.9448	1.7567	H	1	RES1	0.1154
42	H	1.1520	3.4791	-0.4218	H	1	RES1	0.1289
43	H	2.6608	3.1907	-1.2773	H	1	RES1	0.1437
44	H	1.1340	2.7180	-2.0116	H	1	RES1	0.1407
45	H	-3.8076	-1.1592	2.2615	H	1	RES1	0.1335
46	H	-2.7212	-2.4316	1.6879	H	1	RES1	0.1400
47	H	-2.1631	-1.2749	2.8869	H	1	RES1	0.1317

48	H	-2.8052	1.7540	2.7857	H	1	RES1	0.1418
49	H	-1.3230	0.9024	3.1920	H	1	RES1	0.1339
50	H	-1.2496	2.4692	2.3893	H	1	RES1	0.1270
51	H	-4.1615	-1.8265	-1.1682	H	1	RES1	0.1406
52	H	-2.7328	-1.9962	-2.1750	H	1	RES1	0.1392
53	H	-2.8118	-2.8016	-0.5983	H	1	RES1	0.1671
54	H	-3.4076	1.2186	-2.7526	H	1	RES1	0.1314
55	H	-1.6960	1.5881	-2.9820	H	1	RES1	0.1315
56	H	-2.2724	-0.0516	-3.2255	H	1	RES1	0.1343
57	H	-2.5789	3.4358	-0.4232	H	1	RES1	0.1428
58	H	-1.0064	3.4277	0.3600	H	1	RES1	0.1150
59	H	-1.1281	3.1560	-1.3770	H	1	RES1	0.1364

@<TRIPOS>BOND

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41	16	49	1
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46	18	51	1
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66	24	44	1

@<TRIPOS>SUBSTRUCTURE
1 RES1 1

Optimized coordinates of 2b in mol2 format (Figure 5):

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Molten generated mol2
59 66 1
SMALL
NO_CHARGES

@<TRIPOS>ATOM

1	C	-1.7021	1.4268	0.4758	C.3	1	RES1	0.0000
2	C	-1.9171	0.2769	1.2894	C.3	1	RES1	0.0000
3	C	-2.4326	-0.7474	0.4520	C.3	1	RES1	0.0000
4	C	-2.4742	-0.2605	-0.8753	C.3	1	RES1	0.0000
5	C	-2.0245	1.0869	-0.8612	C.3	1	RES1	0.0000
6	ZR	0.0000	-0.3194	-0.2585	ZR	1	RES1	0.0000
7	C	1.8377	1.4264	-0.4070	C.3	1	RES1	0.0000
8	C	1.7568	1.0790	0.9695	C.3	1	RES1	0.0000
9	C	2.1532	-0.2733	1.1020	C.3	1	RES1	0.0000
10	C	2.5106	-0.7558	-0.1880	C.3	1	RES1	0.0000
11	C	2.3103	0.2931	-1.1185	C.3	1	RES1	0.0000
12	N	0.0231	-2.4523	0.5699	N.4	1	RES1	0.0000
13	B	-0.0768	-2.8798	-0.9091	B	1	RES1	0.0000
14	C	-3.0009	-2.0465	0.9160	C.3	1	RES1	0.0000
15	C	-3.0202	-0.9793	-2.0624	C.3	1	RES1	0.0000
16	C	-2.0853	2.0351	-2.0114	C.3	1	RES1	0.0000
17	C	-1.5042	2.8225	0.9685	C.3	1	RES1	0.0000
18	C	-1.8372	0.2129	2.7809	C.3	1	RES1	0.0000
19	C	2.3579	-0.9774	2.4033	C.3	1	RES1	0.0000
20	C	3.1846	-2.0537	-0.4860	C.3	1	RES1	0.0000
21	C	2.6954	0.2672	-2.5596	C.3	1	RES1	0.0000
22	C	1.6561	2.7811	-1.0085	C.3	1	RES1	0.0000

23	C	1.5840	2.0102	2.1236	C.3	1	RES1	0.0000
24	H	-1.1528	-3.3584	-1.1856	H	1	RES1	0.0000
25	H	0.8590	-3.5387	-1.2926	H	1	RES1	0.0000
26	H	-0.0030	-1.8021	-1.6215	H	1	RES1	0.0000
27	H	0.8745	-2.7319	1.0330	H	1	RES1	0.0000
28	H	-0.7535	-2.7444	1.1439	H	1	RES1	0.0000
29	H	-0.0126	0.0593	-2.0498	H	1	RES1	0.0000
30	H	-2.4774	3.3207	1.0458	H	1	RES1	0.0000
31	H	-1.0501	2.8563	1.9574	H	1	RES1	0.0000
32	H	-0.8958	3.4246	0.2931	H	1	RES1	0.0000
33	H	-2.8329	0.2725	3.2345	H	1	RES1	0.0000
34	H	-1.3804	-0.7163	3.1328	H	1	RES1	0.0000
35	H	-1.2512	1.0373	3.1885	H	1	RES1	0.0000
36	H	-4.0947	-1.9961	0.9231	H	1	RES1	0.0000
37	H	-2.7243	-2.8788	0.2635	H	1	RES1	0.0000
38	H	-2.6954	-2.2820	1.9384	H	1	RES1	0.0000
39	H	-4.0799	-0.7394	-2.2060	H	1	RES1	0.0000
40	H	-2.4885	-0.6990	-2.9730	H	1	RES1	0.0000
41	H	-2.9308	-2.0600	-1.9514	H	1	RES1	0.0000
42	H	-3.0637	2.5281	-2.0548	H	1	RES1	0.0000
43	H	-1.3305	2.8188	-1.9322	H	1	RES1	0.0000
44	H	-1.9260	1.5194	-2.9585	H	1	RES1	0.0000
45	H	2.1476	1.0181	-3.1285	H	1	RES1	0.0000
46	H	3.7669	0.4694	-2.6748	H	1	RES1	0.0000
47	H	2.4884	-0.7013	-3.0160	H	1	RES1	0.0000
48	H	3.0427	-2.3521	-1.5238	H	1	RES1	0.0000
49	H	4.2619	-1.9668	-0.3050	H	1	RES1	0.0000
50	H	2.8223	-2.8785	0.1307	H	1	RES1	0.0000
51	H	1.0507	2.7445	-1.9172	H	1	RES1	0.0000
52	H	1.1816	3.4756	-0.3156	H	1	RES1	0.0000
53	H	2.6246	3.2129	-1.2828	H	1	RES1	0.0000
54	H	2.5190	-2.0488	2.2682	H	1	RES1	0.0000
55	H	3.2467	-0.5936	2.9168	H	1	RES1	0.0000
56	H	1.5137	-0.8492	3.0858	H	1	RES1	0.0000
57	H	1.1481	2.9619	1.8260	H	1	RES1	0.0000
58	H	0.9663	1.5873	2.9188	H	1	RES1	0.0000
59	H	2.5623	2.2285	2.5672	H	1	RES1	0.0000

@<TRIPOS>BOND

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6	2	6	1
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8	3	4	1
9	3	6	1
10	3	14	1
11	4	5	1
12	4	6	1
13	4	15	1
14	5	6	1
15	5	16	1
16	6	7	1
17	6	8	1
18	6	9	1
19	6	10	1

20	6	11	1
21	7	8	1
22	7	11	1
23	7	22	1
24	8	9	1
25	8	23	1
26	9	10	1
27	9	19	1
28	10	11	1
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59	21	46	1
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62	22	52	1
63	22	53	1
64	23	57	1
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66	23	59	1

@<TRIPOS>SUBSTRUCTURE
1 RES1 1

Optimized coordinates of 3a in mol2 format (Figure 7):

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Molden generated mol2
29 36 1
SMALL
NO_CHARGES

@<TRIPOS>ATOM

1	CL	6.0675	10.6352	1.0575	CL	1	RES1	0.0000
2	N	4.6739	8.8413	-0.9716	N.4	1	RES1	0.0000
3	B	4.1439	7.4219	-1.1507	B	1	RES1	0.0000
4	ZR	5.3449	8.2213	1.1090	ZR	1	RES1	0.0000
5	C	4.4993	8.3898	3.4588	C.3	1	RES1	0.0000
6	C	3.1065	7.3799	1.9556	C.3	1	RES1	0.0000
7	C	4.0529	7.1362	2.9645	C.3	1	RES1	0.0000
8	C	2.9756	8.7811	1.8083	C.3	1	RES1	0.0000
9	C	7.7855	8.0145	0.4404	C.3	1	RES1	0.0000
10	C	7.1607	6.8425	-0.0067	C.3	1	RES1	0.0000
11	C	3.8121	9.3980	2.7597	C.3	1	RES1	0.0000
12	C	6.7182	6.1193	1.1290	C.3	1	RES1	0.0000
13	C	7.1089	6.8406	2.2762	C.3	1	RES1	0.0000
14	C	7.7416	8.0247	1.8557	C.3	1	RES1	0.0000
15	H	5.2208	8.5468	4.2471	H	1	RES1	0.0000
16	H	4.3679	6.1637	3.3119	H	1	RES1	0.0000
17	H	2.5849	6.6352	1.3758	H	1	RES1	0.0000
18	H	2.3226	9.2851	1.1113	H	1	RES1	0.0000
19	H	3.9517	10.4589	2.8865	H	1	RES1	0.0000
20	H	8.1910	8.8014	-0.1765	H	1	RES1	0.0000
21	H	7.0005	6.5515	-1.0339	H	1	RES1	0.0000
22	H	6.2015	5.1713	1.1114	H	1	RES1	0.0000
23	H	6.9364	6.5455	3.3003	H	1	RES1	0.0000
24	H	8.1204	8.8104	2.4907	H	1	RES1	0.0000
25	H	4.4760	6.8152	-0.0427	H	1	RES1	0.0000
26	H	2.9390	7.3498	-1.1622	H	1	RES1	0.0000
27	H	4.7222	6.7781	-1.9923	H	1	RES1	0.0000
28	H	5.4514	9.1309	-1.5480	H	1	RES1	0.0000
29	H	4.0071	9.5991	-0.9533	H	1	RES1	0.0000

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24	8	11	1

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26	9	10	1
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29	10	12	1
30	10	21	1
31	11	19	1
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@<TRIPOS>SUBSTRUCTURE
1 RES1      1
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Optimized coordinates of 3b in mol2 format (Figure 7):

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Molden generated mol2
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SMALL
NO_CHARGES
****
****
@<TRIPOS>ATOM
  1  C      -0.1942      0.0155      0.8764  C.3  1 RES1  0.0000
  2  C       0.6316      0.1986      2.0035  C.3  1 RES1  0.0000
  3  C       1.9766      0.0889      1.5735  C.3  1 RES1  0.0000
  4  C       1.9761     -0.1384      0.1907  C.3  1 RES1  0.0000
  5  C       0.6340     -0.1683     -0.2486  C.3  1 RES1  0.0000
  6 ZR       1.0337      2.2043      0.5161  ZR   1 RES1  0.0000
  7  C      -0.9084      3.2213     -0.7217  C.3  1 RES1  0.0000
  8  C      -1.4137      2.8584      0.5450  C.3  1 RES1  0.0000
  9  C      -0.7830      3.6606      1.5137  C.3  1 RES1  0.0000
 10  C       0.0940      4.5458      0.8391  C.3  1 RES1  0.0000
 11  C       0.0064      4.2802     -0.5357  C.3  1 RES1  0.0000
 12  N       1.9195      2.8872      2.5135  N.4  1 RES1  0.0000
 13  B       3.1694      3.2933      1.6913  B    1 RES1  0.0000
 14  H       2.8594      0.1902      2.1881  H    1 RES1  0.0000
 15  H       2.8470     -0.2141     -0.4409  H    1 RES1  0.0000
 16  H       0.3100     -0.2979     -1.2695  H    1 RES1  0.0000
 17  H      -1.2728      0.0044      0.8766  H    1 RES1  0.0000
 18  H       0.2916      0.3661      3.0151  H    1 RES1  0.0000
 19  H      -0.9579      3.6171      2.5786  H    1 RES1  0.0000
 20  H       0.7397      5.2833      1.2938  H    1 RES1  0.0000
 21  H       0.5876      4.7522     -1.3114  H    1 RES1  0.0000
 22  H      -1.1665      2.7718     -1.6684  H    1 RES1  0.0000
 23  H      -2.1640      2.1081      0.7384  H    1 RES1  0.0000
 24  H       4.1331      2.6086      1.9348  H    1 RES1  0.0000
 25  H       3.3688      4.4839      1.6922  H    1 RES1  0.0000
 26  H       2.8864      3.0273      0.4722  H    1 RES1  0.0000
 27  H       1.4653      3.6600      2.9759  H    1 RES1  0.0000
 28  H       2.0946      2.1742      3.2051  H    1 RES1  0.0000
 29 CL       1.9877      2.2629     -1.8011  CL   1 RES1  0.0000
@<TRIPOS>BOND
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  2      1      5      1
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@<TRIPOS>SUBSTRUCTURE
1 RES1 1

Optimized coordinates of 4a in mol2 format (Figure 8):

@<TRIPOS>MOLECULE
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3	C	-2.5024	0.1902	-0.8224	C.3	1	RES1	0.0000
4	C	-1.9498	1.4320	-0.3905	C.3	1	RES1	0.0000
5	C	-1.6417	1.3084	0.9881	C.3	1	RES1	0.0000
6	ZR	-0.0073	-0.2210	-0.2725	ZR	1	RES1	0.0000
7	C	2.3388	-0.7341	0.7236	C.3	1	RES1	0.0000
8	C	1.8165	0.3730	1.4300	C.3	1	RES1	0.0000
9	C	1.7479	1.4745	0.5230	C.3	1	RES1	0.0000
10	C	2.2034	1.0290	-0.7386	C.3	1	RES1	0.0000

11	C	2.5285	-0.3511	-0.6296	C.3	1	RES1	0.0000
12	CL	-0.0706	-2.6240	0.5029	CL	1	RES1	0.0000
13	N	-0.0765	-1.3615	-2.2331	N.4	1	RES1	0.0000
14	B	0.0193	-0.0224	-2.9557	B	1	RES1	0.0000
15	C	1.6536	0.4208	2.9139	C.3	1	RES1	0.0000
16	C	1.6164	2.9190	0.8779	C.3	1	RES1	0.0000
17	C	2.5279	1.9345	-1.8777	C.3	1	RES1	0.0000
18	C	3.1658	-1.1935	-1.6850	C.3	1	RES1	0.0000
19	C	2.8082	-2.0019	1.3446	C.3	1	RES1	0.0000
20	C	-3.1690	-2.0258	0.3040	C.3	1	RES1	0.0000
21	C	-3.2115	-0.0628	-2.1110	C.3	1	RES1	0.0000
22	C	-1.9272	2.6943	-1.1883	C.3	1	RES1	0.0000
23	C	-1.4145	2.4487	1.9225	C.3	1	RES1	0.0000
24	C	-1.8947	-0.5624	2.7814	C.3	1	RES1	0.0000
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26	H	1.0657	0.1387	-3.5305	H	1	RES1	0.0000
27	H	-0.9483	0.2737	-3.6079	H	1	RES1	0.0000
28	H	-0.9284	-1.8952	-2.3317	H	1	RES1	0.0000
29	H	0.7010	-2.0001	-2.3245	H	1	RES1	0.0000
30	H	-4.2930	-0.0757	-1.9352	H	1	RES1	0.0000
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37	H	-1.1476	3.3783	-0.8483	H	1	RES1	0.0000
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39	H	-2.3786	2.7525	2.3461	H	1	RES1	0.0000
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41	H	-0.9962	3.3201	1.4229	H	1	RES1	0.0000
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46	H	1.0590	1.2750	3.2368	H	1	RES1	0.0000
47	H	1.1820	-0.4833	3.3031	H	1	RES1	0.0000
48	H	2.6045	3.3912	0.8383	H	1	RES1	0.0000
49	H	0.9781	3.4705	0.1849	H	1	RES1	0.0000
50	H	1.2309	3.0653	1.8844	H	1	RES1	0.0000
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55	H	2.9371	-2.2527	-1.5438	H	1	RES1	0.0000
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57	H	3.4695	2.4566	-1.6697	H	1	RES1	0.0000
58	H	2.6445	1.3895	-2.8119	H	1	RES1	0.0000
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Optimized coordinates of 4b in mol2 format (Figure 8):

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  7  C      -2.1504      1.0239      0.6666  C.3    1 RES1    0.0000
  8  C      -1.7033      0.0187      1.5539  C.3    1 RES1    0.0000
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 13  C      -1.6492     -2.5905      1.5029  C.3    1 RES1    0.0000
 14  C      -2.9132     -2.0313     -1.3190  C.3    1 RES1    0.0000
 15  C      -3.1805      1.0819     -1.7128  C.3    1 RES1    0.0000
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 19  C      2.1410     -2.7958      0.5188  C.3    1 RES1    0.0000
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 38  H     -2.5944     -3.0376     -1.0366  H      1 RES1    0.0000
 39  H     -4.2435      1.2666     -1.5219  H      1 RES1    0.0000
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46	H	4.1391	2.0174	-0.7947	H	1	RES1	0.0000
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50	H	2.8425	-1.9251	-2.2318	H	1	RES1	0.0000
51	H	1.3288	2.9247	1.3886	H	1	RES1	0.0000
52	H	1.2032	1.9273	2.8416	H	1	RES1	0.0000
53	H	2.7834	2.4511	2.2719	H	1	RES1	0.0000
54	H	2.2815	-3.2608	-0.4591	H	1	RES1	0.0000
55	H	3.0086	-3.0851	1.1229	H	1	RES1	0.0000
56	H	1.2634	-3.2502	0.9834	H	1	RES1	0.0000
57	H	1.1020	-0.1663	3.5621	H	1	RES1	0.0000
58	H	0.8457	-1.8108	2.9700	H	1	RES1	0.0000
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1 RES1 1

References:

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