

Supporting Information

Deposition of TiO₂ Nanoparticles onto Silica Measured Using a Quartz Crystal Microbalance with Dissipation Monitoring

Langmuir

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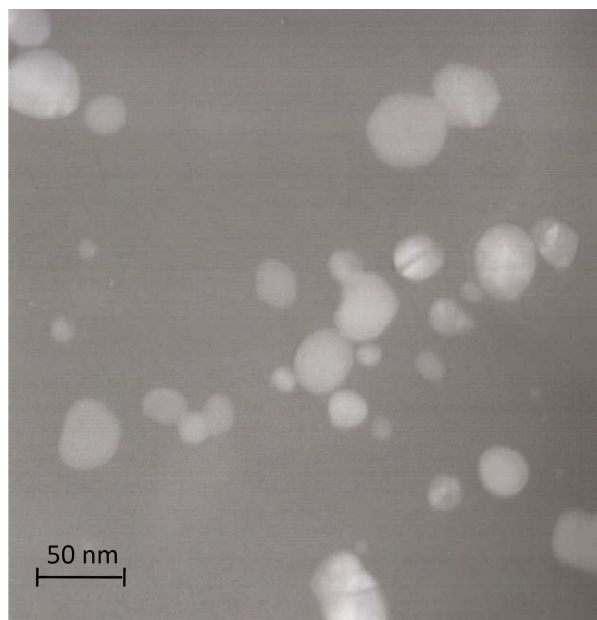


Figure S1. Representative TEM image taken at 230,000 $\times$ magnification of TiO₂ particles suspended in 10 mM NaNO₃ electrolyte at pH 3, deposited and dried on a formvar grid.

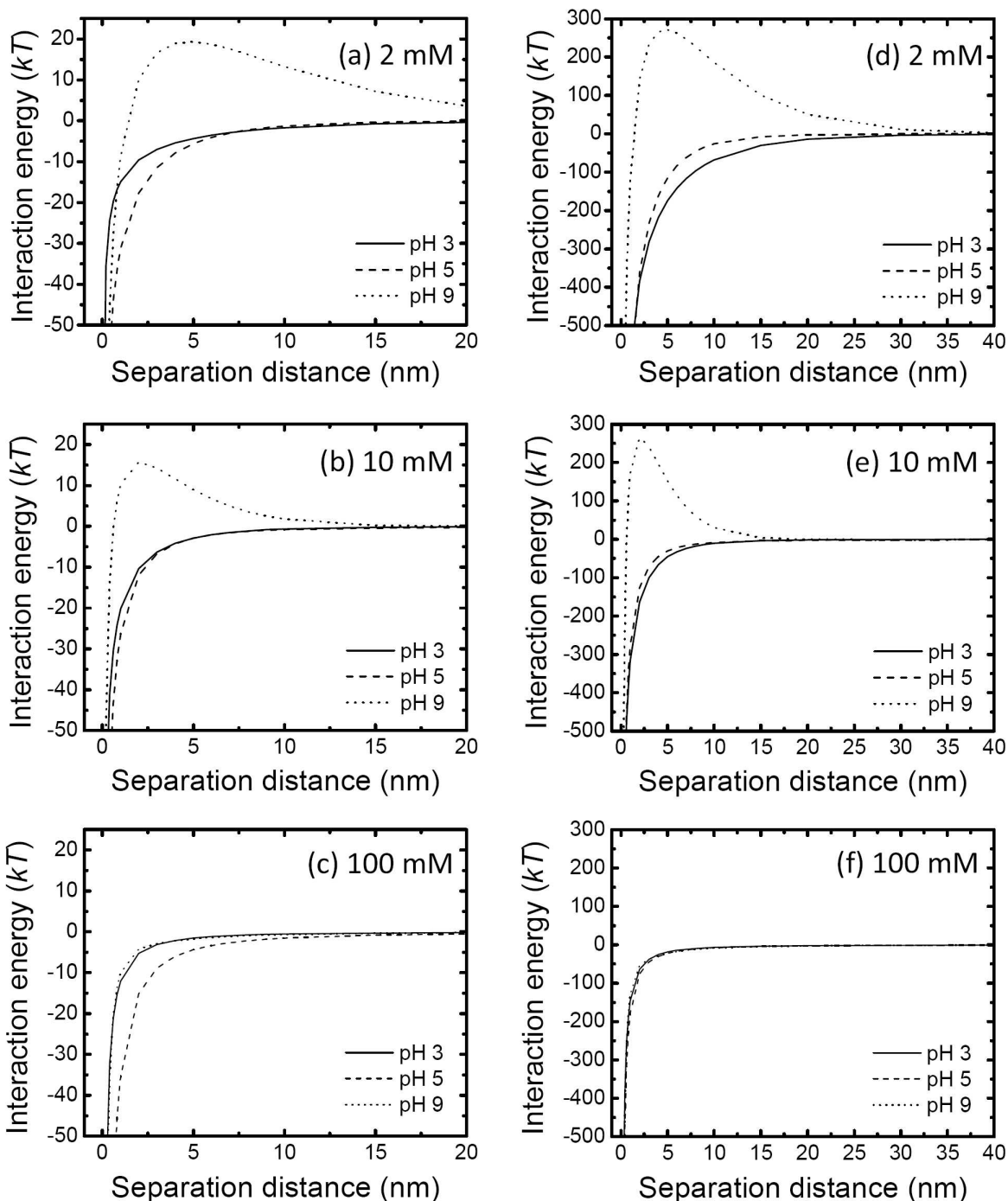


Figure S2. Calculated DLVO interaction energy profiles for a TiO_2 nanoparticle approaching a flat SiO_2 surface at (a) 2 mM, (b) 10 mM, and (c) 100 mM for different pH values: pH 3 (solid lines), pH 5 (dashed lines) and pH 9 (dotted lines). Nanoparticle sizes were taken from values obtained from the FCS measurements (Fig. 1b) and are listed in Table S2. Calculated DLVO interaction energy profiles for a TiO_2 particle approaching a flat SiO_2 surface at (d) 2 mM, (e) 10 mM, and (f) 100 mM for different pH values using sizes obtained from the DLS measurements (Fig. 1a) as listed in Table S2.

Table S1. TiO₂ deposition rates onto the silica surface estimated using Eq. 2.

pH	salt	<i>I</i> (mM)	<i>f</i> _{slope} (Hz/min)	deposition rate (ng/cm ² min)
3	NaNO ₃	2	-0.52	3.0
3	NaNO ₃	3	-0.36	2.1
3	NaNO ₃	4	-0.26	1.5
3	NaNO ₃	7	-0.33	1.9
3	NaNO ₃	11	-0.15	0.87
5	NaNO ₃	2.5	-0.31	5.4
5	NaNO ₃	3.5	-0.19	3.4
5	NaNO ₃	4.5	-0.019	0.34
5	NaNO ₃	7.5	-0.012	0.20
3	Ca(NO ₃) ₂	2.5	-0.60	3.5
3	Ca(NO ₃) ₂	4.0	-0.53	3.1
3	Ca(NO ₃) ₂	16	-0.39	2.3
3	Ca(NO ₃) ₂	31	-0.36	2.1
3	Ca(NO ₃) ₂	101	-0.07	0.43
9	Ca(NO ₃) ₂	3.5	-0.017	0.10
9	Ca(NO ₃) ₂	5.0	-0.022	0.13

Table S2. TiO₂ particle diameters (*d_p*) used in the calculation of the DLVO interaction energy profiles presented in Figures 6 and S2.

pH	<i>I</i> (mM)	<i>d_p</i> by FCS (nm)	<i>d_p</i> by DLS (nm)
3	2	9.43	378
3	10	27.1	428
3	100	44.7	545
5	2	22.8	467
5	10	56.3	592
5	100	128	638
9	2	25.5	358
9	10	33.4	565
9	100	53.6	688