
S U P P L E M E N T A R Y M A T E R I A L

Table S1 - Crystal Data and Structure Refinement for **2**.

Table S2 - Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **2**.

Table S3 - Bond lengths (\AA) and angles (degrees) for **2**.

Table S4 - Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for **2**.

Table S5 - Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **2**.

Table S1' - Crystal Data and Structure Refinement for **4C**.

Table S2' - Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4C**.

Table S3' - Bond lengths (\AA) and angles (degrees) for **4C**.

Table S4' - Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for **4C**.

Table S5' - Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4C**.

Table S6 - Distances and angles for the intramolecular hydrogen bonds found in complexes **2** and **4C**.

Table S7 - Complete catalytic data for the evolution with time of the transfer hydrogenation of benzophenone (1000 equivalents) in 2-propanol, using as precatalyst $[\text{RuCl}(\text{Arene})(\text{NN})][\text{BPh}_4]$, without the addition of base.

Table S8 - Complete catalytic data for the evolution with time of the transfer hydrogenation of benzophenone (1000 equivalents) in 2-propanol, using as precatalyst $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{m})][\text{X}]$, without the addition of base.

Table S9 - Complete catalytic data for the evolution with time of the transfer hydrogenation of different carbonyl compounds (1000 equivalents) in 2-propanol, using as precatalyst $[\text{RuCl}(p\text{-cymene})(\text{NN})][\text{BPh}_4]$, without the addition of base.

Table S10 - Transfer hydrogenation of benzophenone with different changes in the reaction media, using as precatalyst $[\text{RuCl}(p\text{-cymene})(\text{NN})][\text{BPh}_4]$.

Table S11 - Evolution with time of the NMR signals obtained for each precatalyst in refluxing 2-propanol.

Figure S1 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when $[\text{RuCl}(\text{benzene})(\text{NN})][\text{BPh}_4]$ complexes are used.

Figure S2 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when $[\text{RuCl}(p\text{-cymene})(\text{NN})][\text{BPh}_4]$ complexes are used.

Figure S3 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone. Comparison between the results obtained for $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mArOH})][\text{BPh}_4]$ complex (**2**) and when the catalyst is generated in situ in the reaction medium.

Figure S4 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone. Comparison between the results obtained for $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mArOH})][\text{BPh}_4]$ complex (**2**) and when the catalyst is generated in situ in the reaction medium.

cymene)(bpz*mArNO₂)] [BPh₄] complex (**14**) and when the catalyst is generated in situ in the reaction medium.

Figure S5 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone. Comparison between the results obtained for [RuCl(*p*-cymene)(bpz*m)][BPh₄] complex (**8**) and when the catalyst is generated in situ in the reaction medium.

Figure S6 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*m)][X] complexes are used.

Figure S7 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mPh)][BPh₄] complex (**10**) is used as precatalyst, with different incubation times.

Figure S8 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mArOH)][BPh₄] complex (**2**) is used as precatalyst, with different incubation times.

Figure S9 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*m)][BPh₄] complex (**8**) is used as precatalyst, with different incubation times.

Figure S10 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mArOH)][BPh₄] (**2**) is used as precatalyst in two catalytic cycles.

Figure S11 - Evolution with time of the yield of product for the transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mArOH)][BPh₄] (**2**) is used as precatalyst and a excess of *p*-cymene is added.

Figure S12. ¹H NMR spectra (hydride region) obtained after a treatment of 4 hours at 82°C of complex **2** in 2-propanol.

Figure S13. ^1H NMR spectra (hydride region) obtained after a treatment of 16 hours at 82°C of complex **14** in 2-propanol.

Figure S14 - Representation of $\text{Ln} (\text{K})$ versus $\text{Ln} [\text{Ru}]$ for the transfer hydrogenation of acetophenone with $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mArOH})][\text{BPh}_4]$ (**2**) as precatalyst.

Table S1 - Crystal data and structure refinement for **2**.

Empirical formula	[C ₂₇ H ₃₄ ClN ₄ ORu][C ₂₄ H ₂₀ B]	
Formula weight	886.31	
Temperature	150(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2 ₁ /c	
Unit cell dimensions	a = 13.9184(7) Å	α= 90.0°
	b = 16.6271(8) Å	β= 104.071(2)°.
	c = 20.0114(9) Å	γ= 90.0°
Volume	4492(2) Å ³	
Z	4	
Density (calculated)	1.311 Mg/m ³	
Absorption coefficient	0.450 mm ⁻¹	
F(000)	1848	
Crystal size	0.46 x 0.19 x 0.04 mm ³	
Theta range for data collection	1.51 to 25.00°.	
Index ranges	-16 ≤ h ≤ 16, -19 ≤ k ≤ 19, -23 ≤ l ≤ 23	
Reflections collected	31454	
Independent reflections	7885 [R(int) = 0.0888]	
Completeness to theta = 25.00°	99.7 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.932 and 0.783	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	7885 / 0 / 540	
Goodness-of-fit on F ²	1.037	
Final R indices [I>2σ(I)]	R1 = 0.0541, wR2 = 0.1334	
R indices (all data)	R1 = 0.1053, wR2 = 0.1760	
Largest diff. peak and hole	0.733 and -0.478 e.Å ⁻³	

Table 2 - Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **2**. U(eq) is defined as one third of the trace of the orthogonalized U_{ij}^{ij} tensor.

	x	y	z	U(eq)
Ru(1)	1637(1)	3651(1)	1354(1)	41(1)
Cl(1)	1244(1)	3975(1)	156(1)	62(1)
O(13)	2655(3)	781(2)	2009(2)	63(1)
N(1)	279(3)	2999(3)	1066(2)	45(1)
N(2)	231(3)	2187(2)	952(2)	46(1)
N(3)	2322(3)	2635(3)	1034(2)	46(1)
N(4)	1898(3)	1887(3)	951(2)	47(1)
B(1)	7017(5)	3090(4)	2888(4)	60(2)
C(1)	-707(4)	1947(3)	667(3)	55(2)
C(2)	-1278(4)	2620(4)	589(3)	59(2)
C(3)	-659(4)	3262(3)	826(3)	50(1)
C(4)	-980(4)	1103(4)	495(4)	72(2)
C(5)	-954(4)	4125(3)	835(4)	71(2)
C(6)	1071(3)	1674(3)	1238(3)	45(1)
C(7)	2384(4)	1390(3)	606(3)	57(2)
C(8)	3132(4)	1837(4)	472(3)	64(2)
C(9)	3064(4)	2596(4)	727(3)	59(2)
C(10)	2078(5)	550(4)	420(4)	79(2)
C(11)	3716(5)	3294(4)	670(4)	85(2)
C(12)	1274(3)	1591(3)	2009(3)	46(1)
C(13)	612(4)	1893(3)	2366(3)	53(1)
C(14)	726(4)	1749(4)	3058(4)	68(2)
C(15)	1507(5)	1278(4)	3403(3)	68(2)
C(16)	2147(4)	952(4)	3056(3)	63(2)
C(17)	2041(4)	1108(3)	2371(3)	49(1)
C(18)	2500(4)	3635(3)	2457(3)	49(1)
C(19)	1530(4)	3886(3)	2409(3)	52(1)
C(20)	1112(4)	4550(4)	1988(3)	58(2)
C(21)	1640(4)	4959(3)	1597(3)	57(2)
C(22)	2612(4)	4685(4)	1612(3)	59(2)
C(23)	3028(4)	4051(4)	2042(3)	52(1)
C(24)	1205(5)	5651(4)	1140(3)	77(2)

C(25)	2971(4)	2996(4)	2960(3)	60(2)
C(26)	3330(6)	3394(6)	3669(3)	106(3)
C(27)	3798(4)	2541(4)	2752(4)	76(2)
C(28)	8153(4)	3417(3)	3179(3)	53(2)
C(29)	8803(4)	3035(4)	3745(3)	67(2)
C(30)	9767(4)	3282(5)	4011(3)	76(2)
C(31)	10132(5)	3928(5)	3735(4)	73(2)
C(32)	9532(5)	4331(4)	3193(4)	73(2)
C(33)	8569(4)	4066(4)	2922(3)	62(2)
C(34)	6393(4)	3763(3)	2343(4)	61(2)
C(35)	6518(5)	3795(4)	1673(4)	74(2)
C(36)	6048(6)	4371(5)	1194(4)	92(2)
C(37)	5448(6)	4934(5)	1389(6)	102(3)
C(38)	5294(5)	4928(5)	2029(5)	94(3)
C(39)	5744(4)	4349(4)	2498(4)	73(2)
C(40)	6578(4)	2935(4)	3572(3)	63(2)
C(41)	6549(4)	3559(4)	4036(4)	74(2)
C(42)	6246(5)	3462(6)	4638(4)	97(3)
C(43)	5984(5)	2726(7)	4810(5)	101(3)
C(44)	6011(5)	2086(5)	4386(5)	89(2)
C(45)	6306(4)	2192(4)	3769(4)	71(2)
C(46)	6987(4)	2264(3)	2442(3)	56(2)
C(47)	6106(5)	1904(4)	2106(4)	78(2)
C(48)	6029(6)	1216(5)	1713(4)	92(2)
C(49)	6894(8)	859(4)	1638(4)	93(2)
C(50)	7776(7)	1194(5)	1956(4)	86(2)
C(51)	7828(5)	1879(4)	2345(3)	67(2)

Table S3 - Bond lengths [\AA] and angles [$^\circ$] for **2**.

Ru(1)-N(3)	2.113(4)
Ru(1)-N(1)	2.133(4)
Ru(1)-C(22)	2.173(5)
Ru(1)-C(19)	2.187(6)
Ru(1)-C(23)	2.189(5)
Ru(1)-C(20)	2.198(5)
Ru(1)-C(21)	2.228(6)
Ru(1)-C(18)	2.241(5)
Ru(1)-Cl(1)	2.3889(16)
O(13)-C(17)	1.360(6)
N(1)-C(3)	1.350(6)
N(1)-N(2)	1.368(5)
N(2)-C(1)	1.352(6)
N(2)-C(6)	1.448(6)
N(3)-C(9)	1.325(6)
N(3)-N(4)	1.370(6)
N(4)-C(7)	1.358(7)
N(4)-C(6)	1.450(6)
B(1)-C(46)	1.633(9)
B(1)-C(28)	1.639(8)
B(1)-C(40)	1.649(10)
B(1)-C(34)	1.654(9)
C(1)-C(2)	1.360(8)
C(1)-C(4)	1.472(8)
C(2)-C(3)	1.381(7)
C(3)-C(5)	1.493(8)
C(6)-C(12)	1.506(7)
C(7)-C(8)	1.358(8)
C(7)-C(10)	1.481(8)
C(8)-C(9)	1.374(8)
C(9)-C(11)	1.495(8)
C(12)-C(13)	1.389(7)
C(12)-C(17)	1.390(7)
C(13)-C(14)	1.376(8)
C(14)-C(15)	1.381(9)
C(15)-C(16)	1.367(8)

C(16)-C(17)	1.367(8)
C(18)-C(19)	1.394(7)
C(18)-C(23)	1.416(8)
C(18)-C(25)	1.499(8)
C(19)-C(20)	1.424(8)
C(20)-C(21)	1.377(8)
C(21)-C(22)	1.420(8)
C(21)-C(24)	1.503(8)
C(22)-C(23)	1.395(8)
C(25)-C(27)	1.518(8)
C(25)-C(26)	1.535(8)
C(28)-C(33)	1.382(8)
C(28)-C(29)	1.416(8)
C(29)-C(30)	1.380(8)
C(30)-C(31)	1.362(10)
C(31)-C(32)	1.371(9)
C(32)-C(33)	1.390(8)
C(34)-C(35)	1.394(10)
C(34)-C(39)	1.413(8)
C(35)-C(36)	1.399(10)
C(36)-C(37)	1.373(11)
C(37)-C(38)	1.350(11)
C(38)-C(39)	1.383(10)
C(40)-C(45)	1.378(9)
C(40)-C(41)	1.401(9)
C(41)-C(42)	1.379(10)
C(42)-C(43)	1.344(11)
C(43)-C(44)	1.368(11)
C(44)-C(45)	1.404(10)
C(46)-C(47)	1.383(8)
C(46)-C(51)	1.389(8)
C(47)-C(48)	1.377(10)
C(48)-C(49)	1.383(10)
C(49)-C(50)	1.358(10)
C(50)-C(51)	1.372(9)
N(3)-Ru(1)-N(1)	86.99(16)
N(3)-Ru(1)-C(22)	113.7(2)

N(1)-Ru(1)-C(22)	158.0(2)
N(3)-Ru(1)-C(19)	125.31(19)
N(1)-Ru(1)-C(19)	95.06(19)
C(22)-Ru(1)-C(19)	79.3(2)
N(3)-Ru(1)-C(23)	92.21(19)
N(1)-Ru(1)-C(23)	156.2(2)
C(22)-Ru(1)-C(23)	37.3(2)
C(19)-Ru(1)-C(23)	66.2(2)
N(3)-Ru(1)-C(20)	163.0(2)
N(1)-Ru(1)-C(20)	96.01(19)
C(22)-Ru(1)-C(20)	66.5(2)
C(19)-Ru(1)-C(20)	37.9(2)
C(23)-Ru(1)-C(20)	78.3(2)
N(3)-Ru(1)-C(21)	150.3(2)
N(1)-Ru(1)-C(21)	120.62(19)
C(22)-Ru(1)-C(21)	37.6(2)
C(19)-Ru(1)-C(21)	67.2(2)
C(23)-Ru(1)-C(21)	67.2(2)
C(20)-Ru(1)-C(21)	36.2(2)
N(3)-Ru(1)-C(18)	96.70(18)
N(1)-Ru(1)-C(18)	119.20(18)
C(22)-Ru(1)-C(18)	67.8(2)
C(19)-Ru(1)-C(18)	36.68(19)
C(23)-Ru(1)-C(18)	37.3(2)
C(20)-Ru(1)-C(18)	67.3(2)
C(21)-Ru(1)-C(18)	79.9(2)
N(3)-Ru(1)-Cl(1)	83.44(13)
N(1)-Ru(1)-Cl(1)	82.61(12)
C(22)-Ru(1)-Cl(1)	92.12(17)
C(19)-Ru(1)-Cl(1)	151.12(15)
C(23)-Ru(1)-Cl(1)	120.98(16)
C(20)-Ru(1)-Cl(1)	113.49(17)
C(21)-Ru(1)-Cl(1)	89.20(17)
C(18)-Ru(1)-Cl(1)	158.19(14)
C(3)-N(1)-N(2)	104.9(4)
C(3)-N(1)-Ru(1)	130.4(4)
N(2)-N(1)-Ru(1)	123.4(3)
C(1)-N(2)-N(1)	111.4(4)

C(1)-N(2)-C(6)	126.7(4)
N(1)-N(2)-C(6)	120.9(4)
C(9)-N(3)-N(4)	105.1(4)
C(9)-N(3)-Ru(1)	129.8(4)
N(4)-N(3)-Ru(1)	123.3(3)
C(7)-N(4)-N(3)	111.2(4)
C(7)-N(4)-C(6)	126.5(5)
N(3)-N(4)-C(6)	122.3(4)
C(46)-B(1)-C(28)	111.5(5)
C(46)-B(1)-C(40)	110.9(5)
C(28)-B(1)-C(40)	106.1(5)
C(46)-B(1)-C(34)	105.6(5)
C(28)-B(1)-C(34)	108.4(5)
C(40)-B(1)-C(34)	114.3(5)
N(2)-C(1)-C(2)	106.4(5)
N(2)-C(1)-C(4)	123.2(5)
C(2)-C(1)-C(4)	130.3(5)
C(1)-C(2)-C(3)	107.4(5)
N(1)-C(3)-C(2)	109.9(5)
N(1)-C(3)-C(5)	123.5(5)
C(2)-C(3)-C(5)	126.6(5)
N(2)-C(6)-N(4)	110.0(4)
N(2)-C(6)-C(12)	113.0(4)
N(4)-C(6)-C(12)	117.3(4)
N(4)-C(7)-C(8)	105.6(5)
N(4)-C(7)-C(10)	123.6(6)
C(8)-C(7)-C(10)	130.8(6)
C(7)-C(8)-C(9)	107.7(5)
N(3)-C(9)-C(8)	110.5(5)
N(3)-C(9)-C(11)	123.5(5)
C(8)-C(9)-C(11)	126.0(6)
C(13)-C(12)-C(17)	117.5(5)
C(13)-C(12)-C(6)	120.7(4)
C(17)-C(12)-C(6)	121.0(5)
C(14)-C(13)-C(12)	121.7(5)
C(13)-C(14)-C(15)	119.0(6)
C(16)-C(15)-C(14)	120.2(6)
C(15)-C(16)-C(17)	120.6(6)

O(13)-C(17)-C(16)	122.0(5)
O(13)-C(17)-C(12)	117.1(5)
C(16)-C(17)-C(12)	120.9(5)
C(19)-C(18)-C(23)	116.5(5)
C(19)-C(18)-C(25)	121.1(5)
C(23)-C(18)-C(25)	122.3(5)
C(19)-C(18)-Ru(1)	69.6(3)
C(23)-C(18)-Ru(1)	69.4(3)
C(25)-C(18)-Ru(1)	135.3(4)
C(18)-C(19)-C(20)	121.5(5)
C(18)-C(19)-Ru(1)	73.8(3)
C(20)-C(19)-Ru(1)	71.5(3)
C(21)-C(20)-C(19)	121.4(5)
C(21)-C(20)-Ru(1)	73.1(3)
C(19)-C(20)-Ru(1)	70.6(3)
C(20)-C(21)-C(22)	117.8(6)
C(20)-C(21)-C(24)	122.2(6)
C(22)-C(21)-C(24)	119.9(6)
C(20)-C(21)-Ru(1)	70.7(3)
C(22)-C(21)-Ru(1)	69.1(3)
C(24)-C(21)-Ru(1)	129.5(4)
C(23)-C(22)-C(21)	120.4(6)
C(23)-C(22)-Ru(1)	72.0(3)
C(21)-C(22)-Ru(1)	73.3(3)
C(22)-C(23)-C(18)	122.2(5)
C(22)-C(23)-Ru(1)	70.7(3)
C(18)-C(23)-Ru(1)	73.4(3)
C(18)-C(25)-C(27)	114.0(5)
C(18)-C(25)-C(26)	107.9(5)
C(27)-C(25)-C(26)	111.7(5)
C(33)-C(28)-C(29)	113.8(5)
C(33)-C(28)-B(1)	125.6(5)
C(29)-C(28)-B(1)	120.6(5)
C(30)-C(29)-C(28)	123.2(6)
C(31)-C(30)-C(29)	120.0(6)
C(30)-C(31)-C(32)	119.4(6)
C(31)-C(32)-C(33)	119.8(6)
C(28)-C(33)-C(32)	123.7(6)

C(35)-C(34)-C(39)	114.9(6)
C(35)-C(34)-B(1)	119.4(6)
C(39)-C(34)-B(1)	125.7(6)
C(34)-C(35)-C(36)	122.6(7)
C(37)-C(36)-C(35)	119.1(8)
C(38)-C(37)-C(36)	120.9(8)
C(37)-C(38)-C(39)	119.9(8)
C(38)-C(39)-C(34)	122.7(8)
C(45)-C(40)-C(41)	114.8(6)
C(45)-C(40)-B(1)	124.1(6)
C(41)-C(40)-B(1)	120.7(6)
C(42)-C(41)-C(40)	123.9(7)
C(43)-C(42)-C(41)	119.3(8)
C(42)-C(43)-C(44)	120.0(9)
C(43)-C(44)-C(45)	120.4(8)
C(40)-C(45)-C(44)	121.6(7)
C(47)-C(46)-C(51)	114.2(6)
C(47)-C(46)-B(1)	122.1(5)
C(51)-C(46)-B(1)	123.6(5)
C(48)-C(47)-C(46)	125.0(7)
C(47)-C(48)-C(49)	118.0(7)
C(50)-C(49)-C(48)	119.0(7)
C(49)-C(50)-C(51)	121.6(7)
C(50)-C(51)-C(46)	122.1(7)

Symmetry transformations used to generate equivalent atoms:

Table S4 - Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **2**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^*{}^2 U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U11	U22	U33	U23	U13	U12
Ru(1)	40(1)	41(1)	42(1)	-4(1)	8(1)	1(1)
Cl(1)	81(1)	60(1)	45(1)	3(1)	14(1)	4(1)
O(13)	63(2)	65(3)	61(3)	11(2)	14(2)	27(2)
N(1)	43(2)	40(3)	49(3)	4(2)	5(2)	2(2)
N(2)	41(2)	36(2)	53(3)	2(2)	-4(2)	-2(2)
N(3)	44(2)	46(3)	48(3)	-6(2)	10(2)	1(2)
N(4)	47(2)	44(3)	48(3)	2(2)	10(2)	8(2)
B(1)	49(3)	54(4)	76(5)	-3(4)	11(3)	0(3)
C(1)	51(3)	52(4)	55(4)	1(3)	2(3)	-9(3)
C(2)	39(3)	62(4)	69(4)	7(3)	1(3)	-2(3)
C(3)	42(3)	49(3)	56(4)	7(3)	4(3)	3(3)
C(4)	60(4)	63(4)	80(5)	1(4)	-6(3)	-13(3)
C(5)	57(3)	55(4)	92(5)	7(4)	4(3)	18(3)
C(6)	44(3)	39(3)	48(3)	1(3)	0(2)	8(2)
C(7)	72(4)	49(4)	47(4)	-7(3)	10(3)	13(3)
C(8)	68(4)	72(4)	59(4)	0(4)	27(3)	20(3)
C(9)	56(3)	64(4)	62(4)	-8(3)	25(3)	3(3)
C(10)	105(5)	55(4)	75(5)	-9(4)	22(4)	17(4)
C(11)	79(4)	82(5)	110(6)	-13(5)	57(4)	-11(4)
C(12)	42(3)	44(3)	51(4)	4(3)	9(2)	2(2)
C(13)	44(3)	54(4)	59(4)	7(3)	10(3)	2(3)
C(14)	60(4)	76(5)	72(5)	9(4)	25(3)	3(3)
C(15)	85(4)	75(5)	44(4)	16(3)	17(3)	3(4)
C(16)	68(4)	68(4)	51(4)	13(3)	8(3)	13(3)
C(17)	46(3)	47(3)	51(4)	3(3)	8(3)	2(2)
C(18)	48(3)	54(3)	39(3)	-7(3)	0(2)	-8(3)
C(19)	50(3)	58(4)	45(3)	-5(3)	7(3)	3(3)
C(20)	58(3)	60(4)	50(4)	-18(3)	5(3)	9(3)
C(21)	70(4)	43(3)	51(4)	-14(3)	1(3)	4(3)
C(22)	69(4)	51(4)	56(4)	-7(3)	12(3)	-15(3)
C(23)	43(3)	62(4)	50(4)	-8(3)	6(3)	-6(3)
C(24)	111(5)	42(4)	69(5)	-5(3)	3(4)	12(3)
C(25)	56(3)	69(4)	48(4)	4(3)	-2(3)	4(3)

C(26)	129(7)	119(7)	52(5)	-13(4)	-12(5)	27(5)
C(27)	61(4)	74(4)	86(5)	4(4)	2(3)	10(3)
C(28)	52(3)	52(3)	57(4)	-13(3)	15(3)	1(3)
C(29)	61(4)	69(4)	69(5)	-4(4)	11(3)	-3(3)
C(30)	59(4)	100(6)	61(4)	-18(4)	-2(3)	12(4)
C(31)	49(3)	99(6)	74(5)	-27(4)	21(3)	-18(4)
C(32)	67(4)	77(5)	81(5)	-15(4)	31(4)	-18(4)
C(33)	55(3)	67(4)	67(4)	-2(4)	21(3)	1(3)
C(34)	46(3)	54(4)	81(5)	-6(3)	10(3)	-7(3)
C(35)	72(4)	60(4)	83(6)	-3(4)	3(4)	-4(3)
C(36)	95(5)	80(6)	86(6)	9(5)	-6(4)	-20(5)
C(37)	75(5)	58(5)	144(9)	7(6)	-29(6)	-2(4)
C(38)	51(4)	61(5)	157(9)	-21(6)	1(5)	-7(3)
C(39)	45(3)	60(4)	110(6)	-7(4)	10(3)	-6(3)
C(40)	45(3)	63(4)	76(5)	-9(4)	8(3)	-13(3)
C(41)	58(4)	77(5)	88(6)	-12(4)	22(4)	-7(3)
C(42)	66(4)	142(9)	86(6)	-29(6)	21(4)	-14(5)
C(43)	59(4)	160(9)	79(6)	-7(7)	7(4)	-19(5)
C(44)	71(4)	102(6)	87(6)	25(5)	5(4)	-21(4)
C(45)	59(4)	85(5)	66(5)	-3(4)	8(3)	-17(3)
C(46)	60(3)	48(3)	57(4)	6(3)	9(3)	-6(3)
C(47)	75(4)	72(5)	86(5)	-19(4)	20(4)	-7(4)
C(48)	107(6)	88(6)	80(6)	-22(5)	23(5)	-29(5)
C(49)	155(8)	54(5)	69(5)	-8(4)	26(5)	-12(5)
C(50)	116(6)	74(5)	76(5)	6(4)	38(5)	24(5)
C(51)	79(4)	61(4)	59(4)	-2(3)	16(3)	8(3)

Table S5 - Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **2**.

	x	y	z	U(eq)
H(13)	3011	444	2247	95
H(2)	-1960	2644	410	71
H(4A)	-656	918	152	107
H(4B)	-1685	1063	320	107
H(4C)	-779	778	902	107
H(5A)	-937	4283	1299	106
H(5B)	-1612	4193	551	106
H(5C)	-502	4452	660	106
H(6)	870	1136	1058	54
H(8)	3606	1660	248	77
H(10A)	2156	235	831	118
H(10B)	2484	330	139	118
H(10C)	1397	541	168	118
H(11A)	3321	3772	563	127
H(11B)	4050	3193	312	127
H(11C)	4197	3367	1100	127
H(13A)	78	2201	2131	63
H(14)	283	1966	3291	81
H(15)	1597	1181	3872	81
H(16)	2659	622	3288	76
H(19)	1095	3547	2608	62
H(20)	398	4648	1905	69
H(22)	2953	4891	1273	71
H(23)	3648	3816	1984	63
H(24A)	502	5669	1092	116
H(24B)	1338	5582	694	116
H(24C)	1497	6145	1340	116
H(25)	2455	2607	2991	72
H(26A)	3779	3823	3640	158
H(26B)	3662	3001	3997	158
H(26C)	2772	3605	3815	158
H(27A)	3567	2345	2290	114

H(27B)	4000	2095	3059	114
H(27C)	4352	2894	2777	114
H(29)	8569	2597	3947	80
H(30)	10168	3007	4379	91
H(31)	10781	4095	3912	87
H(32)	9770	4780	3007	88
H(33)	8183	4340	2546	75
H(35)	6929	3419	1540	89
H(36)	6142	4372	750	110
H(37)	5143	5324	1075	122
H(38)	4886	5313	2155	112
H(39)	5614	4345	2932	88
H(41)	6745	4069	3932	88
H(42)	6223	3901	4922	117
H(43)	5785	2653	5217	121
H(44)	5833	1577	4507	107
H(45)	6318	1750	3486	85
H(47)	5521	2144	2150	93
H(48)	5414	999	1504	110
H(49)	6872	396	1373	111
H(50)	8360	953	1910	104
H(51)	8447	2093	2550	80

Table S1' - Crystal data and structure refinement for **4C**.

Empirical formula	[C ₂₃ H ₂₅ N ₄ ORu][BC ₂₄ H ₂₀] x C ₂ H ₄ Cl ₂	
Formula weight	892.70	
Temperature	180(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2 ₁ /c	
Unit cell dimensions	a = 12.0756(10) Å	α= 90°.
	b = 14.3033(11) Å	β= 93.8760(10)°.
	c = 24.6336(19) Å	γ= 90°.
Volume	4245.0(6) Å ³	
Z	4	
Density (calculated)	1.397 Mg/m ³	
Absorption coefficient	0.538 mm ⁻¹	
F(000)	1848	
Crystal size	0.52 x 0.28 x 0.21 mm ³	
Theta range for data collection	2.18 to 28.29°.	
Index ranges	-15 ≤ h ≤ 16, -18 ≤ k ≤ 18, -32 ≤ l ≤ 32	
Reflections collected	32582	
Independent reflections	10273 [R(int) = 0.0778]	
Completeness to theta = 28.29°	97.5 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.8954 and 0.7673	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	10273 / 0 / 526	
Goodness-of-fit on F ²	0.899	
Final R indices [I>2σ(I)]	R1 = 0.0506, wR2 = 0.1048	
R indices (all data)	R1 = 0.0809, wR2 = 0.1140	
Largest diff. peak and hole	1.217 and -1.181 e.Å ⁻³	

Table S2' - Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4C**. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
Ru(1)	-1950(1)	-406(1)	3280(1)	29(1)
O(1)	-1628(2)	-1459(2)	3834(1)	38(1)
N(1)	-469(2)	-767(2)	2925(1)	30(1)
N(2)	540(2)	-591(2)	3185(1)	32(1)
N(3)	93(2)	119(2)	4013(1)	30(1)
N(4)	-956(2)	372(2)	3843(1)	33(1)
C(1)	1346(3)	-669(2)	2835(2)	40(1)
C(2)	864(3)	-881(2)	2341(2)	42(1)
C(3)	-261(3)	-940(2)	2411(1)	34(1)
C(4)	-1207(3)	1056(2)	4180(1)	37(1)
C(5)	-354(3)	1238(2)	4559(1)	37(1)
C(6)	461(3)	633(2)	4448(1)	36(1)
C(7)	675(2)	-655(2)	3778(1)	32(1)
C(8)	382(2)	-1600(2)	3987(1)	30(1)
C(9)	-723(3)	-1913(2)	4011(1)	31(1)
C(10)	-865(3)	-2780(2)	4261(1)	39(1)
C(11)	15(3)	-3314(2)	4466(1)	45(1)
C(12)	1098(3)	-3012(3)	4423(1)	44(1)
C(13)	1266(3)	-2163(2)	4190(1)	36(1)
C(14)	-2722(3)	693(3)	2734(2)	44(1)
C(15)	-2723(3)	-134(3)	2473(2)	53(1)
C(16)	-3156(3)	-957(3)	2673(2)	55(1)
C(17)	-3649(3)	-963(3)	3204(2)	51(1)
C(18)	-3637(2)	-81(2)	3482(1)	31(1)
C(19)	-3182(2)	700(2)	3235(1)	34(1)
C(20)	-4059(3)	-1814(3)	3465(2)	81(2)
C(21)	-2228(3)	1583(3)	2493(2)	63(1)
C(22)	-3101(3)	1937(4)	2052(2)	76(2)
B(1)	3480(3)	2026(3)	3621(1)	28(1)
C(23)	-1131(3)	1502(3)	2317(2)	77(2)
C(24)	3181(2)	1622(2)	2997(1)	29(1)
C(25)	2170(2)	1774(2)	2694(1)	32(1)

C(26)	1949(3)	1424(3)	2173(1)	41(1)
C(27)	2730(3)	897(3)	1931(2)	47(1)
C(28)	3734(3)	728(3)	2211(2)	46(1)
C(29)	3958(3)	1093(2)	2727(1)	39(1)
C(30)	2632(2)	2866(2)	3761(1)	31(1)
C(31)	2932(3)	3810(2)	3820(1)	36(1)
C(32)	2178(3)	4499(3)	3940(1)	47(1)
C(33)	1089(3)	4279(3)	4013(1)	49(1)
C(34)	760(3)	3367(3)	3963(1)	46(1)
C(35)	1513(3)	2676(2)	3840(1)	37(1)
C(36)	4765(2)	2418(2)	3676(1)	30(1)
C(37)	5250(3)	2883(2)	3251(2)	39(1)
C(38)	6303(3)	3273(2)	3302(2)	45(1)
C(39)	6936(3)	3207(3)	3786(2)	49(1)
C(40)	6504(3)	2752(3)	4220(2)	43(1)
C(41)	5441(2)	2364(2)	4163(1)	34(1)
C(42)	3391(2)	1177(2)	4069(1)	30(1)
C(43)	3153(2)	1368(3)	4607(1)	36(1)
C(44)	3169(3)	695(3)	5010(2)	46(1)
C(45)	3435(3)	-210(3)	4898(2)	53(1)
C(46)	3681(3)	-438(3)	4376(2)	48(1)
C(47)	3648(2)	242(2)	3971(1)	35(1)
C(51)	4131(6)	3785(5)	5401(3)	149(3)
C(52)	3830(6)	4703(5)	5536(3)	142(3)
Cl(1)	2370(2)	4894(1)	5365(1)	108(1)
Cl(2)	5537(2)	3543(2)	5596(1)	130(1)

Table S3' - Bond lengths [\AA] and angles [$^\circ$] for **4C**.

Ru(1)-O(1)	2.053(2)
Ru(1)-N(4)	2.093(3)
Ru(1)-N(1)	2.107(2)
Ru(1)-C(16)	2.164(3)
Ru(1)-C(19)	2.169(3)
Ru(1)-C(15)	2.172(4)
Ru(1)-C(18)	2.180(3)
Ru(1)-C(17)	2.198(3)
Ru(1)-C(14)	2.230(4)
O(1)-C(9)	1.320(4)
N(1)-C(3)	1.331(4)
N(1)-N(2)	1.362(3)
N(2)-C(1)	1.347(4)
N(2)-C(7)	1.463(4)
N(3)-C(6)	1.349(4)
N(3)-N(4)	1.356(3)
N(3)-C(7)	1.452(4)
N(4)-C(4)	1.332(4)
C(1)-C(2)	1.349(5)
C(2)-C(3)	1.383(4)
C(4)-C(5)	1.367(4)
C(5)-C(6)	1.353(4)
C(7)-C(8)	1.498(4)
C(8)-C(13)	1.402(4)
C(8)-C(9)	1.412(4)
C(9)-C(10)	1.399(4)
C(10)-C(11)	1.376(5)
C(11)-C(12)	1.388(5)
C(12)-C(13)	1.365(5)
C(14)-C(15)	1.346(5)
C(14)-C(19)	1.386(5)
C(14)-C(21)	1.542(5)
C(15)-C(16)	1.393(6)
C(16)-C(17)	1.473(6)
C(17)-C(18)	1.435(5)
C(17)-C(20)	1.477(5)

C(18)-C(19)	1.401(4)
C(21)-C(23)	1.426(5)
C(21)-C(22)	1.546(5)
B(1)-C(30)	1.630(5)
B(1)-C(36)	1.648(4)
B(1)-C(42)	1.648(5)
B(1)-C(24)	1.658(5)
C(24)-C(25)	1.405(4)
C(24)-C(29)	1.407(4)
C(25)-C(26)	1.385(4)
C(26)-C(27)	1.375(5)
C(27)-C(28)	1.374(5)
C(28)-C(29)	1.385(5)
C(30)-C(31)	1.403(5)
C(30)-C(35)	1.405(4)
C(31)-C(32)	1.388(5)
C(32)-C(33)	1.376(5)
C(33)-C(34)	1.367(5)
C(34)-C(35)	1.389(4)
C(36)-C(37)	1.402(4)
C(36)-C(41)	1.406(4)
C(37)-C(38)	1.387(4)
C(38)-C(39)	1.375(5)
C(39)-C(40)	1.385(5)
C(40)-C(41)	1.396(4)
C(42)-C(47)	1.397(4)
C(42)-C(43)	1.404(4)
C(43)-C(44)	1.380(4)
C(44)-C(45)	1.366(5)
C(45)-C(46)	1.379(5)
C(46)-C(47)	1.392(5)
C(51)-C(52)	1.408(8)
C(51)-Cl(2)	1.768(8)
C(52)-Cl(1)	1.806(8)
O(1)-Ru(1)-N(4)	82.78(10)
O(1)-Ru(1)-N(1)	88.17(9)
N(4)-Ru(1)-N(1)	86.61(10)

O(1)-Ru(1)-C(16)	106.25(15)
N(4)-Ru(1)-C(16)	168.79(13)
N(1)-Ru(1)-C(16)	100.08(12)
O(1)-Ru(1)-C(19)	131.84(10)
N(4)-Ru(1)-C(19)	90.42(11)
N(1)-Ru(1)-C(19)	139.15(11)
C(16)-Ru(1)-C(19)	78.63(13)
O(1)-Ru(1)-C(15)	142.20(14)
N(4)-Ru(1)-C(15)	134.69(15)
N(1)-Ru(1)-C(15)	89.37(12)
C(16)-Ru(1)-C(15)	37.48(15)
C(19)-Ru(1)-C(15)	64.65(14)
O(1)-Ru(1)-C(18)	98.16(10)
N(4)-Ru(1)-C(18)	103.77(11)
N(1)-Ru(1)-C(18)	168.39(11)
C(16)-Ru(1)-C(18)	68.85(13)
C(19)-Ru(1)-C(18)	37.59(11)
C(15)-Ru(1)-C(18)	79.65(13)
O(1)-Ru(1)-C(17)	85.61(12)
N(4)-Ru(1)-C(17)	137.67(14)
N(1)-Ru(1)-C(17)	133.60(14)
C(16)-Ru(1)-C(17)	39.46(15)
C(19)-Ru(1)-C(17)	68.07(13)
C(15)-Ru(1)-C(17)	69.30(15)
C(18)-Ru(1)-C(17)	38.27(13)
O(1)-Ru(1)-C(14)	166.02(11)
N(4)-Ru(1)-C(14)	102.88(13)
N(1)-Ru(1)-C(14)	104.79(11)
C(16)-Ru(1)-C(14)	66.83(15)
C(19)-Ru(1)-C(14)	36.70(12)
C(15)-Ru(1)-C(14)	35.57(14)
C(18)-Ru(1)-C(14)	68.17(12)
C(17)-Ru(1)-C(14)	81.64(14)
C(9)-O(1)-Ru(1)	134.13(19)
C(3)-N(1)-N(2)	105.1(2)
C(3)-N(1)-Ru(1)	131.3(2)
N(2)-N(1)-Ru(1)	121.15(19)
C(1)-N(2)-N(1)	110.3(3)

C(1)-N(2)-C(7)	126.7(3)
N(1)-N(2)-C(7)	119.6(2)
C(6)-N(3)-N(4)	110.5(3)
C(6)-N(3)-C(7)	126.1(3)
N(4)-N(3)-C(7)	123.2(2)
C(4)-N(4)-N(3)	104.6(2)
C(4)-N(4)-Ru(1)	131.3(2)
N(3)-N(4)-Ru(1)	123.0(2)
N(2)-C(1)-C(2)	108.0(3)
C(1)-C(2)-C(3)	105.6(3)
N(1)-C(3)-C(2)	111.0(3)
N(4)-C(4)-C(5)	111.8(3)
C(6)-C(5)-C(4)	105.4(3)
N(3)-C(6)-C(5)	107.7(3)
N(3)-C(7)-N(2)	109.0(2)
N(3)-C(7)-C(8)	114.8(3)
N(2)-C(7)-C(8)	112.7(3)
C(13)-C(8)-C(9)	120.1(3)
C(13)-C(8)-C(7)	116.6(3)
C(9)-C(8)-C(7)	123.2(3)
O(1)-C(9)-C(10)	117.2(3)
O(1)-C(9)-C(8)	126.4(3)
C(10)-C(9)-C(8)	116.3(3)
C(11)-C(10)-C(9)	122.6(3)
C(10)-C(11)-C(12)	120.5(3)
C(13)-C(12)-C(11)	118.5(3)
C(12)-C(13)-C(8)	122.0(3)
C(15)-C(14)-C(19)	116.4(4)
C(15)-C(14)-C(21)	121.9(4)
C(19)-C(14)-C(21)	121.8(4)
C(15)-C(14)-Ru(1)	69.9(2)
C(19)-C(14)-Ru(1)	69.3(2)
C(21)-C(14)-Ru(1)	131.1(2)
C(14)-C(15)-C(16)	124.2(4)
C(14)-C(15)-Ru(1)	74.6(2)
C(16)-C(15)-Ru(1)	70.9(2)
C(15)-C(16)-C(17)	120.2(3)
C(15)-C(16)-Ru(1)	71.6(2)

C(17)-C(16)-Ru(1)	71.52(18)
C(18)-C(17)-C(16)	115.2(3)
C(18)-C(17)-C(20)	120.7(4)
C(16)-C(17)-C(20)	124.0(4)
C(18)-C(17)-Ru(1)	70.19(17)
C(16)-C(17)-Ru(1)	69.02(19)
C(20)-C(17)-Ru(1)	126.9(3)
C(19)-C(18)-C(17)	119.0(3)
C(19)-C(18)-Ru(1)	70.79(17)
C(17)-C(18)-Ru(1)	71.54(18)
C(14)-C(19)-C(18)	125.0(3)
C(14)-C(19)-Ru(1)	74.1(2)
C(18)-C(19)-Ru(1)	71.62(18)
C(23)-C(21)-C(14)	116.2(4)
C(23)-C(21)-C(22)	114.7(4)
C(14)-C(21)-C(22)	106.2(3)
C(30)-B(1)-C(36)	109.5(3)
C(30)-B(1)-C(42)	109.4(2)
C(36)-B(1)-C(42)	107.4(2)
C(30)-B(1)-C(24)	110.6(2)
C(36)-B(1)-C(24)	109.8(2)
C(42)-B(1)-C(24)	110.1(3)
C(25)-C(24)-C(29)	114.5(3)
C(25)-C(24)-B(1)	124.5(3)
C(29)-C(24)-B(1)	121.0(3)
C(26)-C(25)-C(24)	122.9(3)
C(27)-C(26)-C(25)	120.3(3)
C(28)-C(27)-C(26)	119.1(3)
C(27)-C(28)-C(29)	120.3(3)
C(28)-C(29)-C(24)	122.8(3)
C(31)-C(30)-C(35)	114.6(3)
C(31)-C(30)-B(1)	124.7(3)
C(35)-C(30)-B(1)	120.7(3)
C(32)-C(31)-C(30)	122.5(3)
C(33)-C(32)-C(31)	120.8(4)
C(34)-C(33)-C(32)	118.7(3)
C(33)-C(34)-C(35)	120.5(3)
C(34)-C(35)-C(30)	122.9(3)

C(37)-C(36)-C(41)	114.5(3)
C(37)-C(36)-B(1)	122.7(3)
C(41)-C(36)-B(1)	122.7(3)
C(38)-C(37)-C(36)	123.3(3)
C(39)-C(38)-C(37)	120.3(3)
C(38)-C(39)-C(40)	119.0(3)
C(39)-C(40)-C(41)	120.0(3)
C(40)-C(41)-C(36)	122.8(3)
C(47)-C(42)-C(43)	114.3(3)
C(47)-C(42)-B(1)	124.4(3)
C(43)-C(42)-B(1)	121.0(3)
C(44)-C(43)-C(42)	123.3(3)
C(45)-C(44)-C(43)	120.5(4)
C(44)-C(45)-C(46)	118.7(4)
C(45)-C(46)-C(47)	120.4(4)
C(46)-C(47)-C(42)	122.7(3)
C(52)-C(51)-Cl(2)	112.2(6)
C(51)-C(52)-Cl(1)	110.5(5)

Symmetry transformations used to generate equivalent atoms:

Table S4' - Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4C**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^*{}^2 U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U11	U22	U33	U23	U13	U12
Ru(1)	23(1)	39(1)	24(1)	-7(1)	-5(1)	6(1)
O(1)	26(1)	49(2)	38(1)	6(1)	-3(1)	0(1)
N(1)	30(1)	33(2)	27(1)	0(1)	0(1)	4(1)
N(2)	25(1)	37(2)	32(2)	2(1)	0(1)	2(1)
N(3)	27(1)	30(2)	31(2)	0(1)	-7(1)	1(1)
N(4)	28(1)	40(2)	30(2)	-4(1)	-6(1)	8(1)
C(1)	28(2)	43(2)	49(2)	6(2)	11(2)	3(2)
C(2)	45(2)	45(2)	39(2)	-1(2)	17(2)	6(2)
C(3)	42(2)	30(2)	30(2)	-4(2)	3(2)	5(2)
C(4)	44(2)	38(2)	30(2)	-6(2)	-2(2)	8(2)
C(5)	52(2)	32(2)	25(2)	-5(2)	-5(2)	-7(2)
C(6)	42(2)	33(2)	30(2)	0(2)	-14(2)	-13(2)
C(7)	22(2)	38(2)	34(2)	1(2)	-7(1)	5(1)
C(8)	32(2)	30(2)	26(2)	-2(1)	-7(1)	3(1)
C(9)	35(2)	35(2)	23(2)	-5(2)	-1(1)	-3(2)
C(10)	42(2)	37(2)	37(2)	-2(2)	-3(2)	-6(2)
C(11)	72(3)	26(2)	37(2)	-3(2)	0(2)	1(2)
C(12)	51(2)	38(2)	42(2)	-4(2)	-5(2)	15(2)
C(13)	35(2)	36(2)	35(2)	-7(2)	-4(2)	7(2)
C(14)	33(2)	57(3)	42(2)	4(2)	-5(2)	12(2)
C(15)	39(2)	77(3)	43(2)	5(2)	-2(2)	6(2)
C(16)	33(2)	62(3)	64(3)	-28(2)	-7(2)	6(2)
C(17)	26(2)	35(2)	87(3)	2(2)	-15(2)	-2(2)
C(18)	19(2)	41(2)	32(2)	4(2)	-3(1)	3(1)
C(19)	25(2)	42(2)	33(2)	1(2)	-5(1)	6(2)
C(20)	46(2)	49(3)	143(5)	20(3)	-21(3)	-13(2)
C(21)	58(3)	74(3)	57(3)	29(2)	11(2)	7(2)
C(22)	62(3)	109(4)	57(3)	42(3)	1(2)	21(3)
B(1)	30(2)	27(2)	28(2)	5(2)	-2(2)	2(2)
C(23)	60(3)	47(3)	123(4)	22(3)	-6(3)	-5(2)
C(24)	29(2)	30(2)	28(2)	7(1)	4(1)	1(1)
C(25)	30(2)	32(2)	34(2)	4(2)	-1(1)	1(1)
C(26)	36(2)	47(2)	39(2)	6(2)	-7(2)	-4(2)

C(27)	48(2)	61(3)	32(2)	-7(2)	1(2)	-10(2)
C(28)	44(2)	52(3)	42(2)	-7(2)	7(2)	6(2)
C(29)	36(2)	45(2)	34(2)	3(2)	-1(2)	4(2)
C(30)	36(2)	33(2)	23(2)	7(1)	0(1)	5(2)
C(31)	47(2)	34(2)	28(2)	2(2)	3(2)	5(2)
C(32)	79(3)	32(2)	31(2)	0(2)	3(2)	10(2)
C(33)	65(3)	47(3)	35(2)	1(2)	10(2)	23(2)
C(34)	44(2)	59(3)	35(2)	11(2)	11(2)	14(2)
C(35)	40(2)	33(2)	37(2)	7(2)	7(2)	8(2)
C(36)	31(2)	23(2)	36(2)	2(1)	-2(1)	2(1)
C(37)	32(2)	37(2)	49(2)	13(2)	2(2)	3(2)
C(38)	37(2)	37(2)	61(3)	12(2)	12(2)	0(2)
C(39)	34(2)	40(2)	72(3)	-12(2)	8(2)	-8(2)
C(40)	35(2)	43(2)	49(2)	-13(2)	-6(2)	-1(2)
C(41)	33(2)	33(2)	35(2)	-5(2)	1(2)	0(2)
C(42)	23(2)	33(2)	32(2)	6(2)	-7(1)	-1(1)
C(43)	33(2)	40(2)	35(2)	9(2)	-5(2)	1(2)
C(44)	33(2)	66(3)	38(2)	20(2)	-5(2)	0(2)
C(45)	33(2)	69(3)	57(3)	40(2)	-5(2)	-5(2)
C(46)	31(2)	32(2)	78(3)	20(2)	-10(2)	-1(2)
C(47)	25(2)	38(2)	42(2)	5(2)	-5(1)	-4(2)
C(51)	217(9)	92(5)	138(6)	7(5)	25(6)	-64(6)
C(52)	179(7)	131(6)	128(6)	-61(5)	97(5)	-97(5)
Cl(1)	173(2)	82(1)	64(1)	11(1)	-24(1)	-48(1)
Cl(2)	124(1)	194(2)	75(1)	-7(1)	17(1)	-68(1)

Table S5' - Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4C**.

	x	y	z	U(eq)
H(1)	2119	-590	2922	48
H(2)	1221	-970	2013	51
H(3)	-811	-1084	2129	40
H(4)	-1896	1379	4160	45
H(5)	-337	1694	4840	45
H(6)	1166	580	4641	43
H(7)	1483	-560	3881	38
H(10)	-1597	-3008	4290	47
H(11)	-121	-3894	4637	54
H(12)	1708	-3388	4554	53
H(13)	2005	-1947	4163	43
H(15)	-2408	-157	2130	64
H(16)	-3134	-1517	2467	66
H(18)	-3933	-27	3828	37
H(19)	-3190	1277	3425	40
H(20A)	-4018	-1728	3861	121
H(20B)	-3600	-2349	3375	121
H(20C)	-4831	-1929	3334	121
H(21)	-2192	2061	2789	75
H(22A)	-2808	2484	1870	114
H(22B)	-3781	2111	2223	114
H(22C)	-3267	1441	1785	114
H(23A)	-623	1313	2625	116
H(23B)	-895	2106	2177	116
H(23C)	-1121	1031	2029	116
H(25)	1613	2132	2852	39
H(26)	1255	1549	1982	49
H(27)	2578	653	1575	56
H(28)	4276	359	2049	55
H(29)	4667	982	2907	46
H(31)	3680	3984	3775	44
H(32)	2417	5131	3973	57

H(33)	575	4752	4097	58
H(34)	11	3204	4012	55
H(35)	1259	2049	3807	44
H(37)	4836	2934	2910	47
H(38)	6589	3587	3002	54
H(39)	7660	3471	3822	58
H(40)	6930	2703	4557	51
H(41)	5162	2052	4465	40
H(43)	2971	1992	4700	43
H(44)	2993	863	5367	55
H(45)	3451	-673	5175	64
H(46)	3873	-1063	4292	57
H(47)	3808	63	3613	42
H(51A)	4002	3697	5003	178
H(51B)	3653	3337	5582	178
H(52A)	4271	5155	5336	170
H(52B)	3995	4807	5931	170

Table S6 - Distances and angles for the intramolecular hydrogen bonds found in complexes **2** and **4C**.

Complex	Hydrogen bond	H···A (Å)	D···A (Å)	DHA angle (°)
	D–H···A			
2	C10– H10A···O13	2.463(4)	3.109(8)	124.5(3)
2	C24– H24B···Cl1	2.872(2)	3.420(7)	117.3(4)
2	C27– H27A···O13	2.889(4)	3.487(8)	121.4(4)
2	C5– H5C···Cl1	2.956(2)	3.645(7)	129.7(4)
2	C11– H11A···Cl1	2.827(2)	3.530(7)	130.8(4)
4C	C20– H20A···O1	2.916(2)	3.056(4)	88.7(2)

Table S7 - Complete catalytic data for the evolution with time of the transfer hydrogenation of benzophenone (1000 equivalents) in 2-propanol, using as precatalyst [RuCl(Arene)(NN)][BPh₄], without the addition of base.

Run	Arene	NN-Ligand (complex)	t = 1.5 h		t = 3 h		t = 6 h		t = 9 h		t = 24 h	
			Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹
1a	-	-	0	-	0	-	0	-	0	-	0	-
1b	benzene ^b	-	-	-	-	-	-	-	-	-	12	5
1c	p-cymene ^c	-	-	-	-	-	-	-	-	-	34	14
2a	benzene	bpz*m (7)	5	28	10	33	18	30	29	32	45	19
3a	p-cymene	bpz*m (8)	0	0	21	69	85	142	94	104	96	40
4a	benzene	bpz*mPh (9)	2	16	5	17	6	10	7	8	20	8
5a	p-cymene	bpz*mPh (10)	10	67	19	62	49	81	61	68	96	40
6a	benzene	bpzmPh (11)	22	147	35	117	42	64	44	48	54	22
7a	p-cymene	bpzmPh (12)	32	214	39	130	51	85	58	65	87	36
8a	benzene	bpz*mArNO ₂ (13)	4	24	12	41	23	38	27	30	41	17
9a	p-cymene	bpz*mArNO ₂ (14)	45	300	61	204	86	143	94	104	96	40
10a	benzene	bpzmArNO ₂ (15)	11	75	30	99	52	86	61	68	65	27
11a	p-cymene	bpzmArNO ₂ (16)	47	313	65	215	78	130	95	106	95	39
12a	benzene	bpz*mArNH ₂ (17)	10	67	16	53	21	36	22	24	25	10
13a	p-cymene	bpz*mArNH ₂ (18)	20	133	34	112	55	91	75	84	91	38
14a	benzene	bpz*mArOH (1)	5	34	6	20	7	10	8	9	9	4
15a	p-cymene	bpz*mArOH (2)	45	300	65	215	86	143	94	104	96	40
16a	benzene	bpzmArOH (3)	6	39	7	23	8	13	9	9	31	13
17a	p-cymene	bpzmArOH (4)	2	13	2	7	3	5	6	7	14	6

^a Determined by ¹H-NMR. ^b 0.002 mmol of [RuCl₂(C₆H₆)(CH₃CN)] added. ^c 0.001 mmol of [RuCl₂(p-cymene)]₂ added.

Table S8 - Complete catalytic data for the evolution with time of the transfer hydrogenation of benzophenone (1000 equivalents) in 2-propanol, using as precatalyst $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{m})][\text{X}]$, without the addition of base.

Run	Anion (complex)	t = 1.5 h		t = 3 h		t = 6 h		t = 9 h		t = 24 h	
		Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹	Yield ^a /%	TOF / h ⁻¹
3a	BPh ₄ (8)	0	0	21	69	85	142	94	104	96	40
21a	BF ₄ (5)	4	24	27	88	78	129	85	94	94	39
22a	Cl (6)	20	133	35	113	55	92	72	83	90	38

^a Determined by ¹H-NMR.

Table S9 - Complete catalytic data for the evolution with time of the transfer hydrogenation of different carbonyl compounds (1000 equivalents) in 2-propanol, using as precatalyst [RuCl(*p*-cymene)(NN)][BPh₄], without the addition of base.

Run	NN-Ligand (complex)	Substrate	Susbstrate/Catalyst Ratio ^a	t = 1.5 h		t = 3 h		t = 6 h		t = 9 h		t = 24 h	
				Yield ^d /%	TOF / h ⁻¹								
23a	bpz*mArOH (2)	Acetofenone	1000/1	43	287	69	230	89	148	95	106	98	41
24a	bpz*mArNO ₂ (14)	Acetofenone	1000/1	46	307	71	237	91	152	96	107	>99	41
25a	bpz*mArOH (2)	Benzylphenylketone	1000/1	-	-	-	-	-	-	-	-	15	6
26a	bpz*mArNO ₂ (14)	Benzylphenylketone	1000/1	-	-	-	-	-	-	-	-	34	14
27a	bpz*mArOH (2)	Benzylphenylketone	167/1 ^b	8	9	19	11	25	7	36	7	60	4
28a	bpz*mArNO ₂ (14)	Benzylphenylketone	167/1 ^b	15	17	30	17	59	16	84	16	98	7
29a	bpz*mArOH (2)	Cyclohexanone	1000/1	-	-	-	-	-	-	-	-	9	4
30a	bpz*mArNO ₂ (14)	Cyclohexanone	1000/1	-	-	-	-	-	-	-	-	11	5
31a	bpz*mArOH (2)	Cyclohexanone	250/1 ^c	0	-	11	9	30	12	52	14	74	8
32a	bpz*mArNO ₂ (14)	Cyclohexanone	250/1 ^c	38	63	67	56	86	36	>99	28	>99	10
33a	bpz*mArOH (2)	Benzaldehyde	1000/1	19	127	81	270	85	142	97	108	-	-
34a	bpz*mArNO ₂ (14)	Benzaldehyde	1000/1	56	373	85	285	93	155	96	107	-	-
35a	bpz*mArOH (2)	2-Phenylpropanal	1000/1	-	-	-	-	-	-	-	-	10	4
36a	bpz*mArNO ₂ (14)	2-Phenylpropanal	1000/1	-	-	-	-	-	-	-	-	13	5
37a	bpz*mArOH (2)	2-Phenylpropanal	125/1 ^c	31	26	45	19	84	17	91	5	>99	5
38a	bpz*mArNO ₂ (14)	2-Phenylpropanal	125/1 ^c	46	38	80	33	84	17	87	5	>99	5

^aReaction conditions: solvent: 10 mL of 2-propanol, 0.002 mmol of precatalyst, no base. ^bPrecatalyst: 0.006 mmol used. ^cPrecatalyst: 0.004 mmol used.

^dDetermined by ¹H-NMR.

Table S10 - Transfer hydrogenation of benzophenone with different changes in the reaction media, using as precatalyst [RuCl(*p*-cymene)(NN)][BPh₄].

Run ^a	NN- Ligand (complex)	Special fact	Yield ^b (%)	TOF (h ⁻¹) ^c	Sample (Time in hours / Yield in %)		
9c	bpz*mArNO ₂ ^e (14)	2 equivalents of precatalyst used	96	192	0.5 / 40	1.5 / 56	5 / 71 8 / 92
15c	bpz*mArOH ^e (2)	2 equivalents of precatalyst used	97	417	0.5 / 49	1.5 / 69	5 / 87 8 / 97
18a	bpz*mArOH (2)	“In situ” precatalyst	47	-	1.5 / 15	3 / 26	6 / 37 9 / 43
19a	bpz*mArNO ₂ (14)	“In situ” precatalyst	69	172	1.5 / 34	3 / 52	6 / 63 9 / 67
20a	bpz*m (8)	“In situ” precatalyst	75	62	1.5 / 16	3 / 30	6 / 45 9 / 57
39a	bpz*mArOH (2)	300 equivalents of Hg added	74	51	1 / 12	4 / 35	10 / 53
40a	bpz*mArOH ^d (2)	2 hours of incubation	57	119	1 / 16	2 / 30	4 / 46
41a		16 hours of incubation	19	-	1 / 6	2 / 11	4 / 16
42a		24 hours of incubation	4	-	1 / 1	2 / 2	4 / 3
43a	bpz*m ^d (8)	1.5 hours of incubation	22	-	1 / 4	2 / 9	4 / 16
44a		2 hours of incubation	13	-	1 / 5	2 / 5	4 / 9
45a		16 hours of incubation	6	-	1 / 1	2 / 2	4 / 4
46a	bpz*mPh ^d (10)	1 hour of incubation	44	-	1 / 25	2 / 36	4 / 43
47a		24 hours of incubation	15	-	1 / 4	2 / 7	4 / 15
48a	bpz*mArOH ^f (2)	4 mmols of ketone added	66	40	24 / 48	26 / 51	32 / 64 48 / 66
49a	bpz*mArOH (2)	20 equivalents of <i>p</i> -cymene added	54	48	1.5 / 18	3 / 28	6 / 37 9 / 45
50a	bpz*mArNO ₂ (14)	20 equivalents of <i>p</i> -cymene added	32	-	1.5 / 20	3 / 29	6 / 29 9 / 31

^aReaction conditions: benzophenone/catalyst = 1000/1, solvent: 2-propanol, no base. ^bDetermined by ¹H-NMR at t = 24 hours. ^cTOF calculated at Yield = 50 %. ^dYield determined at t = 6 hours. ^eYield determined at t = 8 hours. ^fAfter 24 hours of normal conditions, 2 mmols of ketone are added. Yield over 4 mmol of ketone calculated at t = 48 hours. TOF calculated at 32 hours.

Table S11 - Evolution with time of the NMR signals obtained for each precatalyst in refluxing 2-propanol.

Precatalyst	Time (h)	NMR signals (higher to lower intensity order) (ppm)			
$[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mArNO}_2)][\text{BPh}_4]$ 14	2	-10.27	-13.46		
	4	-13.46	-10.27	-8.84	
	16	-13.46			
$[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mArOH})][\text{BPh}_4]$ 2	2	-8.84			
	4	-8.84	-12.07		
	16	-12.07	-8.84		
$[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mPh})][\text{BPh}_4]$ 12	2	-8.84	-9.64	-9.62	
	4	-8.84			
	6	-9.77			
$[\text{RuCl}_2(p\text{-cymene})]_2$	16	-10.04	-8.81	-13.41	-13.44
	16	-10.26			

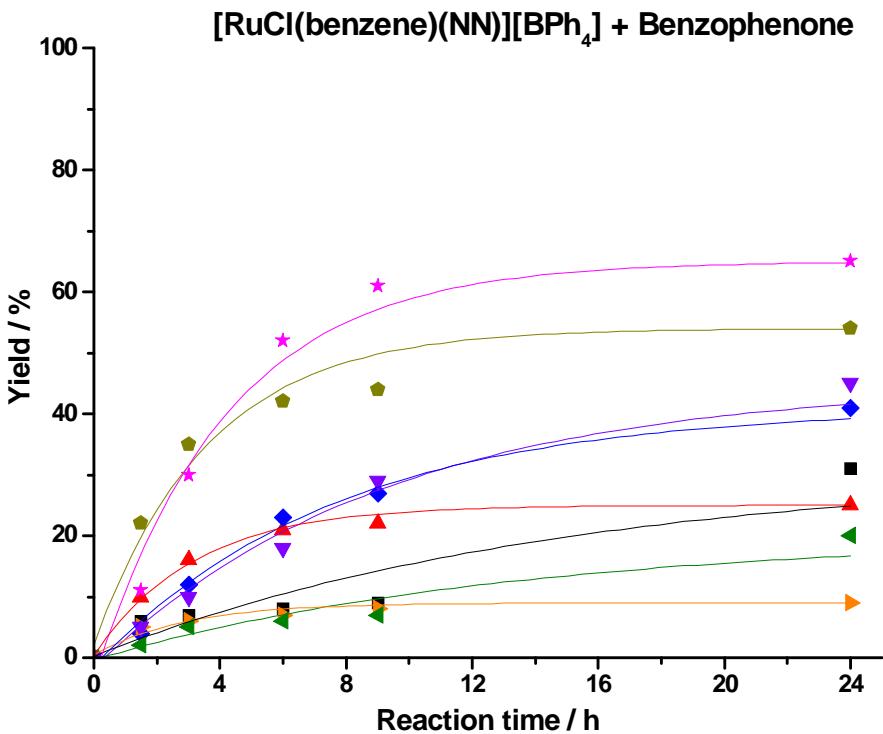


Figure S1. Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when [RuCl(benzene)(NN)][BPh₄] complexes are used.

▼ 7 (bpz*m) ◆ 11 (bpzmPh) ◀ 9 (bpz*mPh) ■ 3 (bpzmArOH) ▶ 1 (bpz*mArOH)
 ★ 15 (bpzmArNO₂) ♦ 13 (bpz*mArNO₂) ▲ 17 (bpz*mArNH₂)

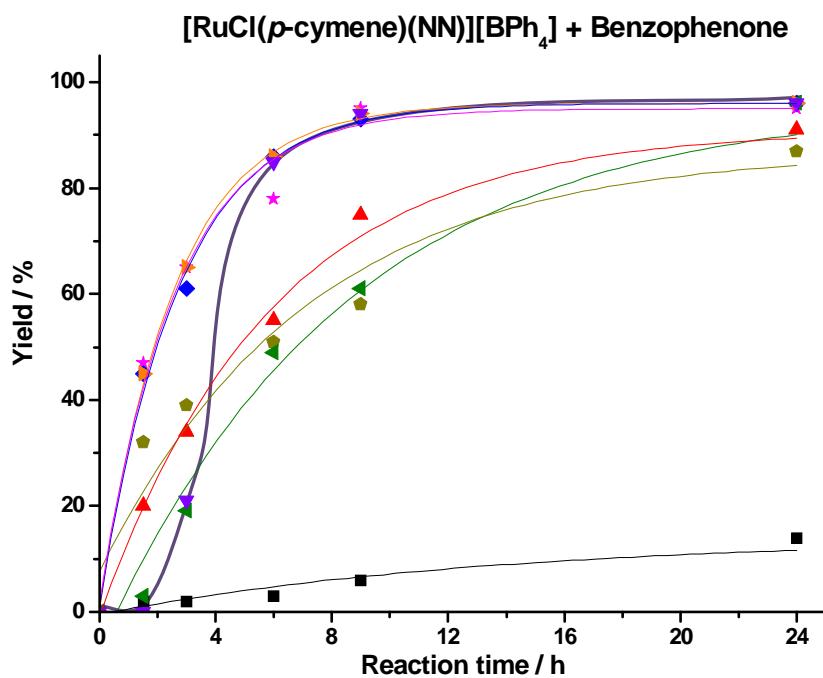


Figure S2. Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(NN)][BPh₄] complexes are used. ▼ 8 (bpz*^m) ◉ 12 (bpzmPh) ▲ 10 (bpz*^mPh) ■ 4 (bpzmArOH) ▶ 2 (bpz*^mArOH) ★ 16 (bpzmArNO₂) ♦ 14 (bpz*^mArNO₂) ▲ 18 (bpz*^mArNH₂)

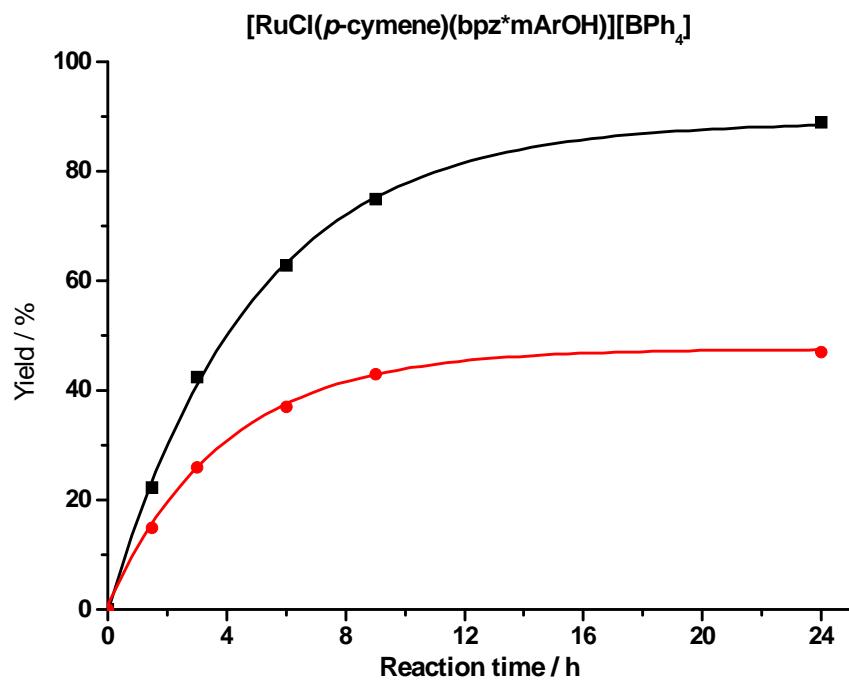


Figure S3 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone. Comparison between the results obtained for [RuCl(*p*-

cymene)(bpz^{*}mArOH)][BPh₄] complex (**2**) and when the catalyst is generated in situ in the reaction medium. ■ Normal test ● “In situ” precatalyst

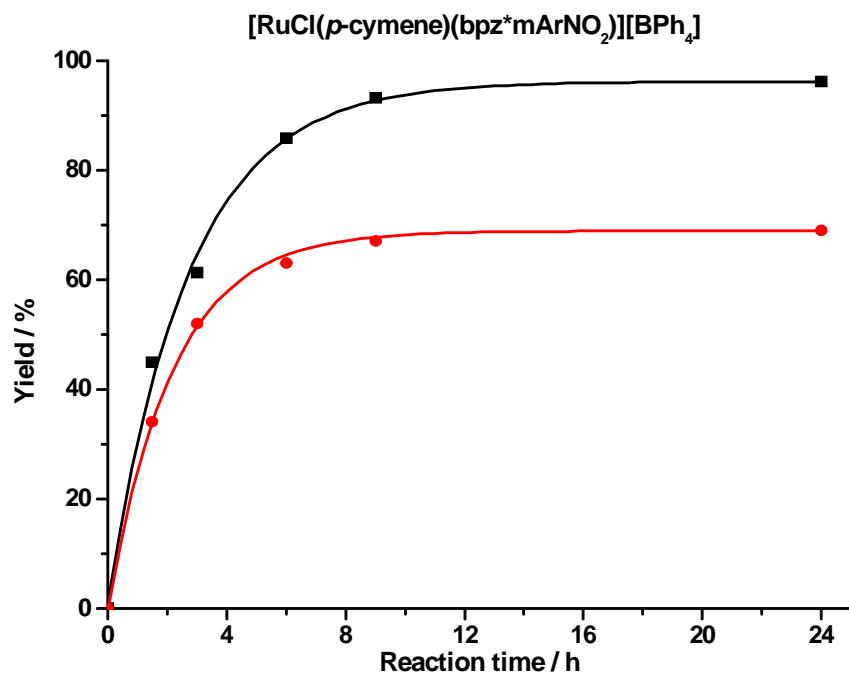


Figure S4 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone. Comparison between the results obtained for [RuCl(*p*-cymene)(bpz^{*}mArNO₂)][BPh₄] complex (**14**) and when the catalyst is generated in situ in the reaction medium. ■ Normal test ● “In situ” precatalyst

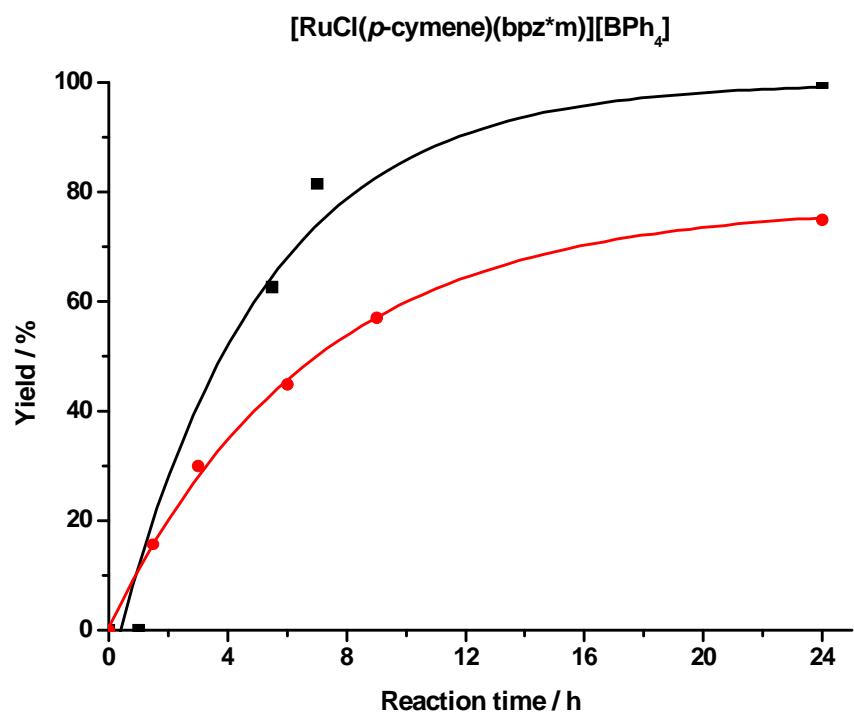


Figure S5 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone. Comparison between the results obtained for $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{m})][\text{BPh}_4]$ complex (**8**) and when the catalyst is generated *in situ* in the reaction medium. ■ Normal test ● “In situ” precatalyst

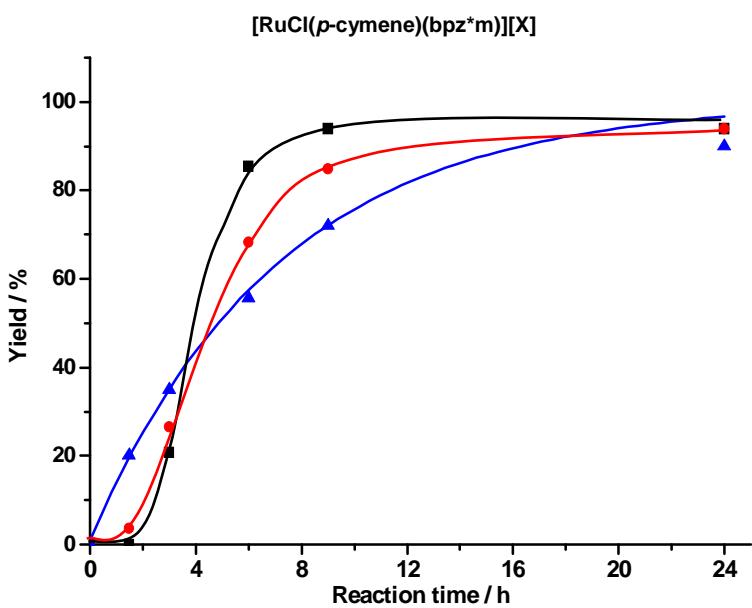


Figure S6 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{m})][\text{X}]$ complexes are used. $\text{X} = \blacksquare\text{BPh}_4$ (**8**), $\bullet\text{BF}_4$ (**5**), and $\blacktriangle\text{Cl}$ (**6**)

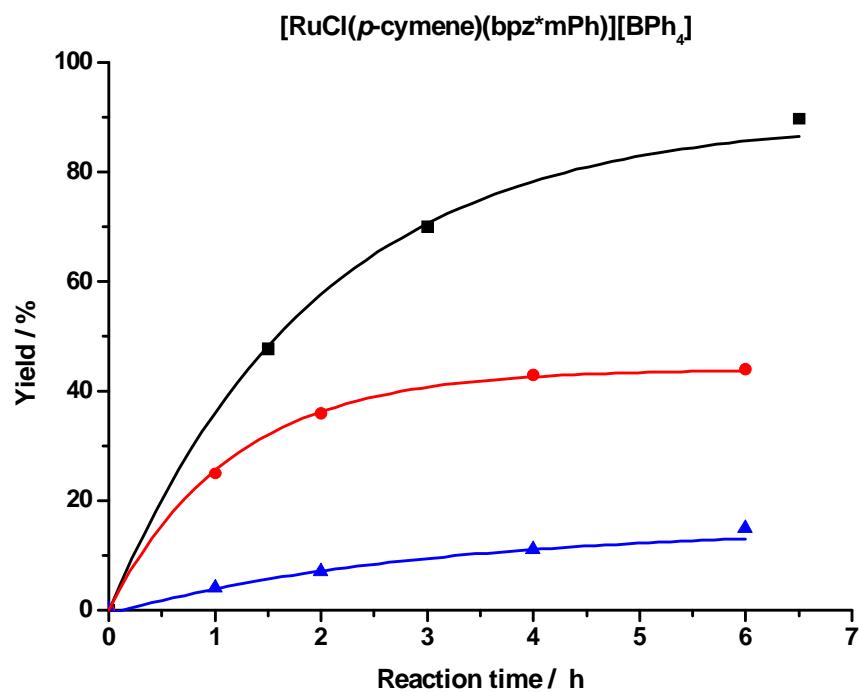


Figure S7 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mPh})][\text{BPh}_4]$ complex (**10**) is used as precatalyst, with different incubation times.

■ No incubation ● 1 hour of incubation ▲ 24 hours of incubation

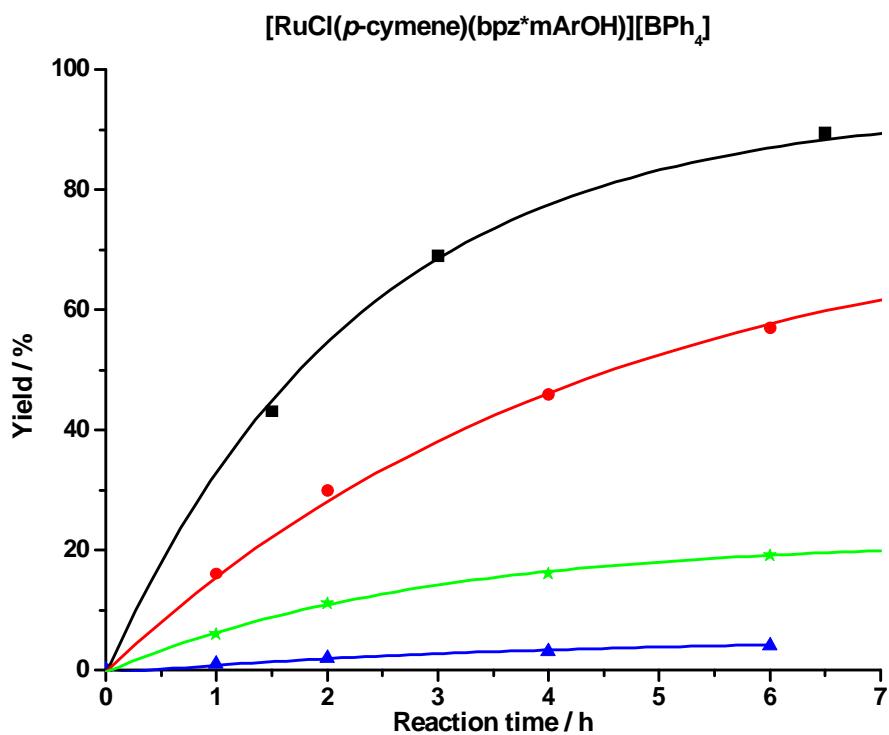


Figure S8 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mArOH)][BPh₄] complex (**2**) is used as precatalyst, with different incubation times.

■ No incubation ● 2 hours of incubation ★ 16 hours of incubation ▲ 24 hours of incubation

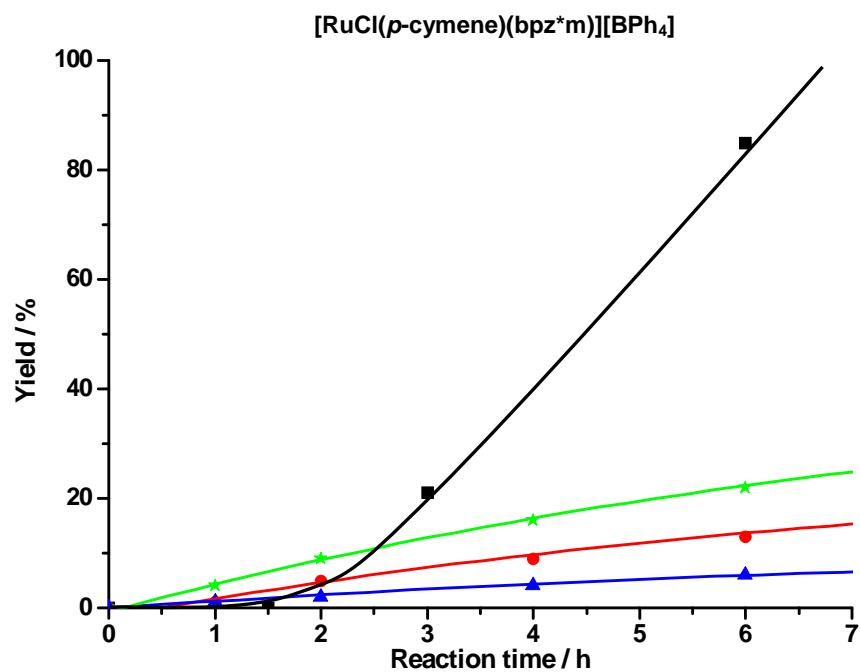


Figure S9 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*m)][BPh₄] complex (**8**) is used as precatalyst, with different incubation times.

■ No incubation ● 1.5 hour of incubation ★ 2 hours of incubation ▲ 16 hours of incubation

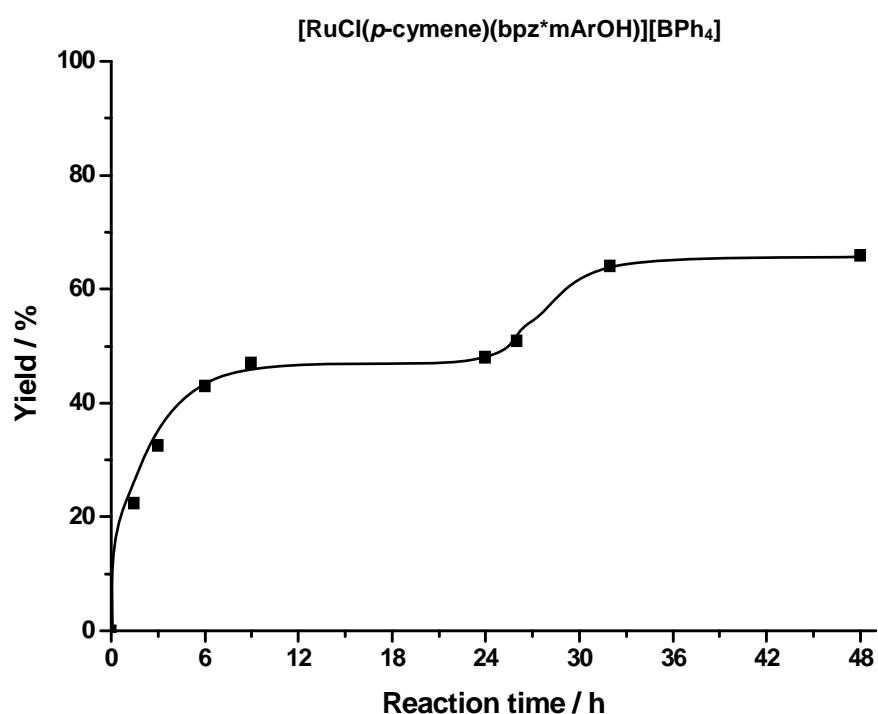


Figure S10 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mArOH)][BPh₄] (**2**) is used as precatalyst in two catalytic cycles. 2 additional mmol of benzophenone are added at 24 hours of reaction.

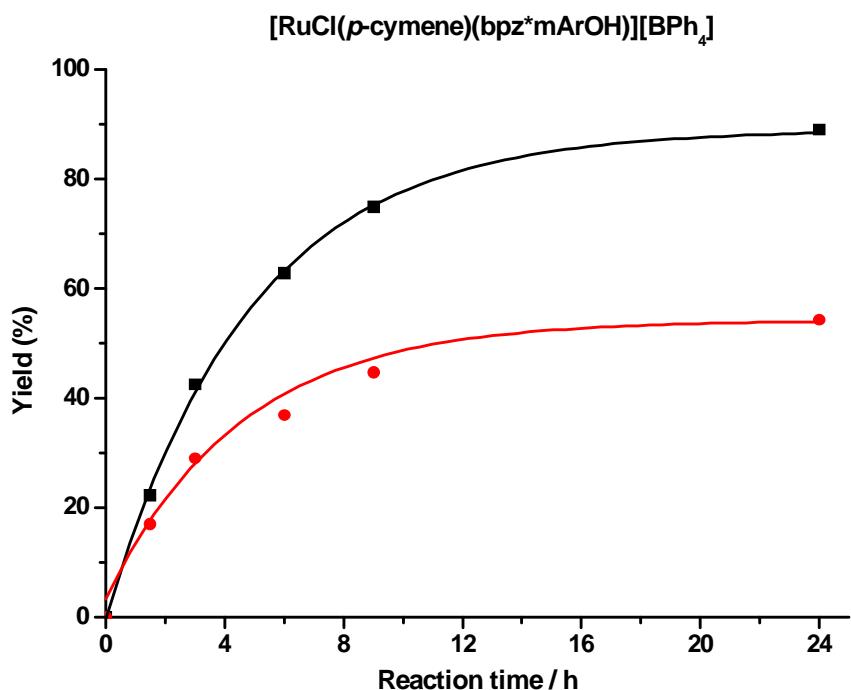


Figure S11 - Evolution with time of the yield of product for the reaction of transfer hydrogenation of benzophenone when [RuCl(*p*-cymene)(bpz*mArOH)][BPh₄] (**2**) is used as precatalyst and a excess of *p*-cymene is added. ■ Normal test ● 20 eq of *p*-cymene added.

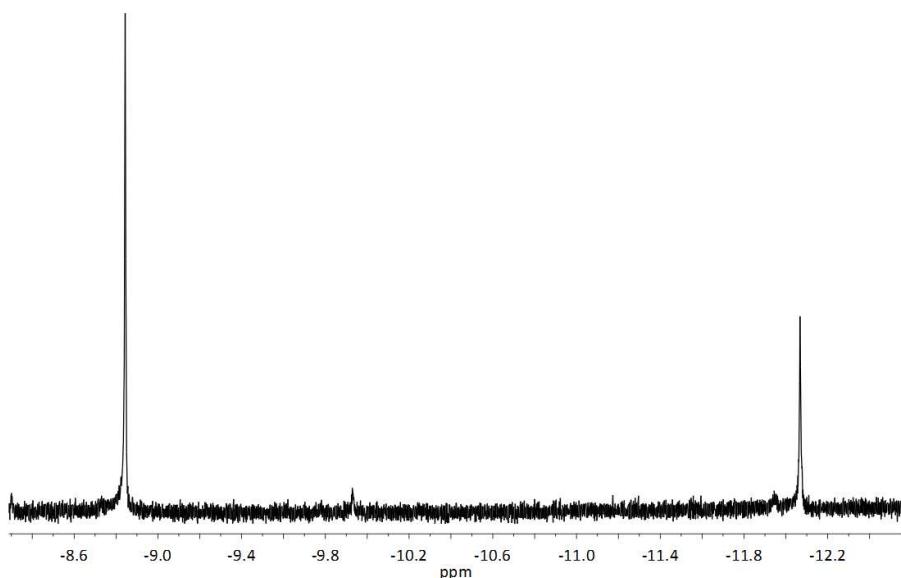


Figure S12. ¹H NMR spectra (hydride region) obtained after a treatment of 4 hours at 82°C of complex **2** in 2-propanol.

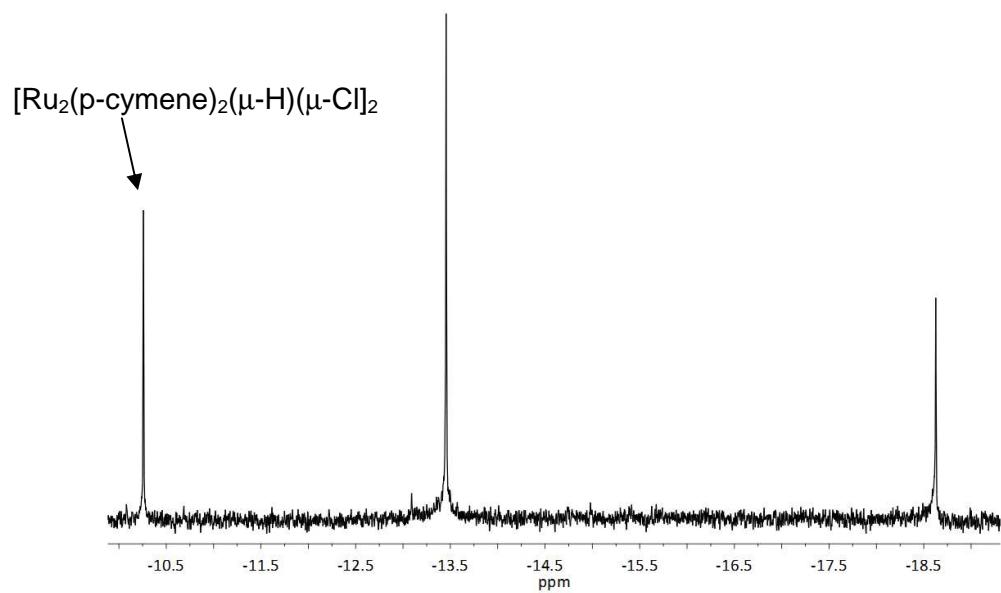


Figure S13. ¹H NMR spectra (hydride region) obtained after a treatment of 16 hours at 82°C of complex **14** in 2-propanol.

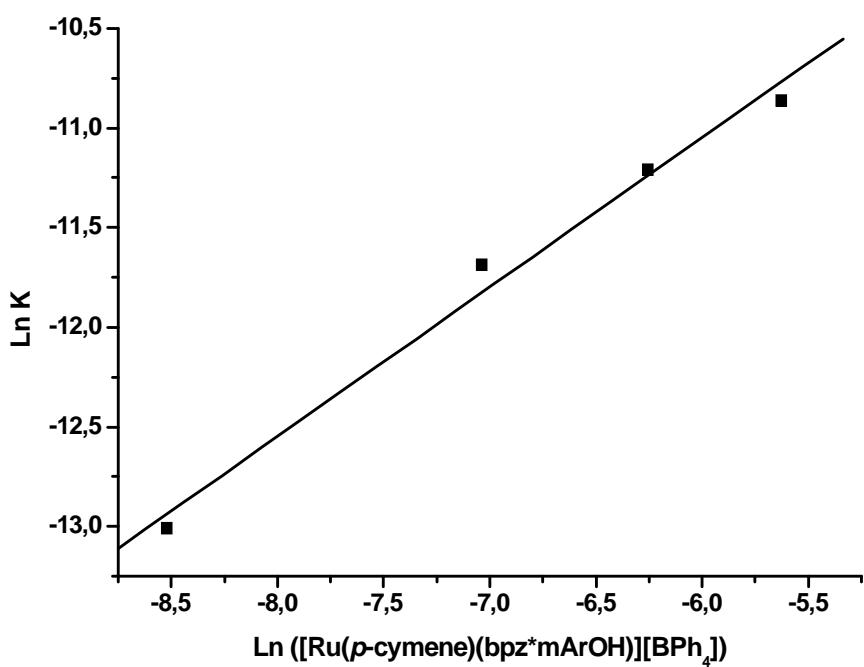


Figure S14 - Representation of $\ln (K)$ versus $\ln [\text{Ru}]$ for the reaction of transfer hydrogenation of acetophenone with $[\text{RuCl}(p\text{-cymene})(\text{bpz}^*\text{mArOH})][\text{BPh}_4]$ (**2**) as precatalyst.