Supporting Information

Photoswitching of Dextro-/Levo- Rotation with Axially Chiral Binaphthyls linked to an Azobenzene

Kazuto Takaishi,*,† Masuki Kawamoto,† Kazunori Tsubaki,‡ and Tatsuo Wada*,†

†Supramolecular Science Laboratory, RIKEN (The Institute of Physical and Chemical Research), Hirosawa 2-1, Wako, Saitama 351-0198, Japan and †Graduate School of Life and Environmental Science, Kyoto Prefectural University, Shimogamo Hangi-cho, Sakyo, Kyoto 606-8522, Japan

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1. UV-vis and CD spectra of (R)-4-9 and (S)-4.

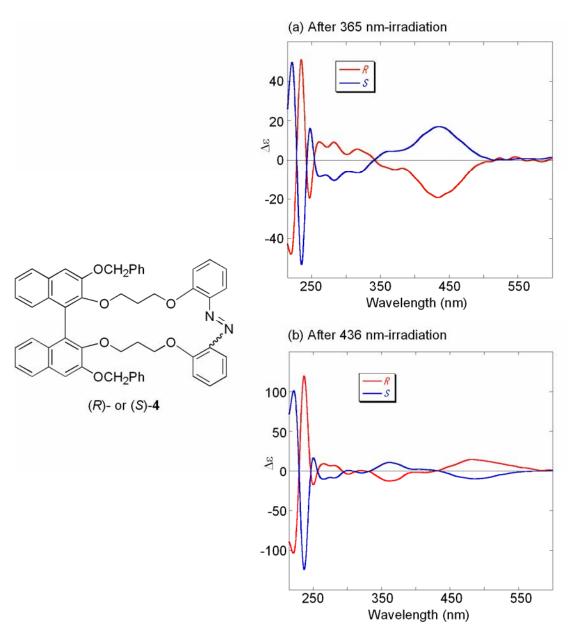


Figure S-1. CD spectra of (R)-4 (red line) and (S)-4 (blue line) after (a) 365 nm-irradiation and (b) 436 nm-irradiation. Conditions: 1,4-Dioxane, $1.0 \times 10^{-5} \text{ M}$, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm², 100 s), 436 nm (10 mW/cm², 100 s).

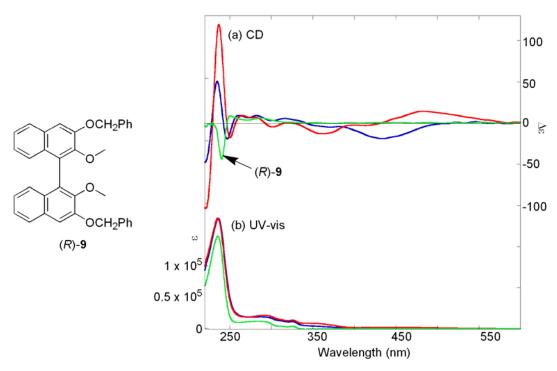


Figure S-2. CD and UV-vis spectra of (R)-4 and (R)-9^{s1}. ((R)-4 after 365 nm-irradiation (blue line), (R)-4 after 436 nm-irradiation (red line), (R)-9 (green line)). Conditions: 1,4-Dioxane, 1.0 x 10⁻⁵ M, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm², 100 s), 436 nm (10 mW/cm², 100 s).

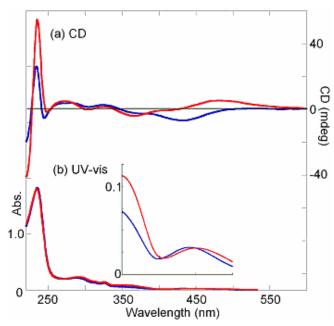


Figure S-3. (a) CD spectra of (*R*)-4, (b) Absorption spectra of (*R*)-4; after 365 nm-irradiation (blue line), after 436 nm-irradiation (red line). Conditions: Methanol, saturated concentration, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm², 100 s), 436 nm (10 mW/cm², 100 s).

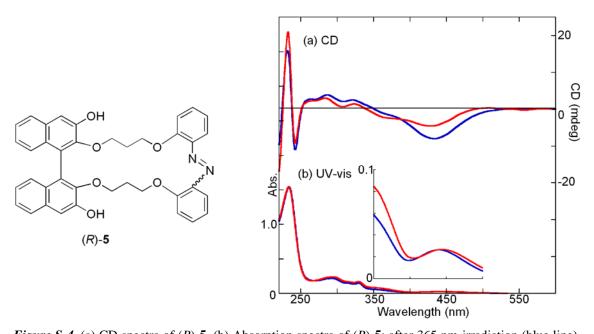


Figure S-4. (a) CD spectra of (R)-5, (b) Absorption spectra of (R)-5; after 365 nm-irradiation (blue line), after 436 nm-irradiation (red line). Conditions: Methanol, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm², 100 s), 436 nm (10 mW/cm², 100 s).

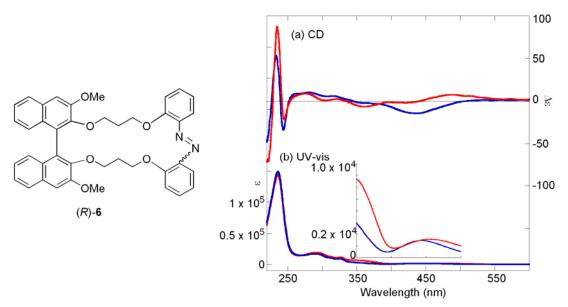


Figure S-5. (a) CD spectra of (R)-6, (b) Absorption spectra of (R)-6; after 365 nm-irradiation (blue line), after 436 nm-irradiation (red line). Conditions: 1,4-Dioxane, 1.0 x 10^{-5} M, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm^2 , 100 s), 436 nm (10 mW/cm^2 , 100 s).

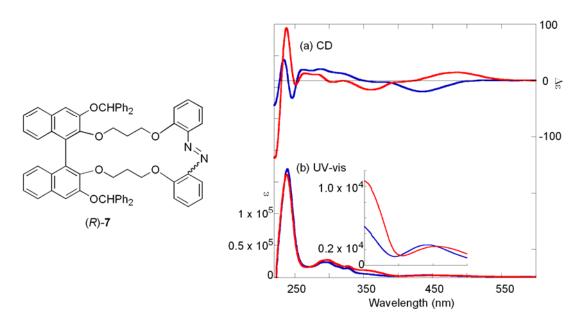


Figure S-6. (a) CD spectra of (R)-7, (b) Absorption spectra of (R)-7; after 365 nm-irradiation (blue line), after 436 nm-irradiation (red line). Conditions: 1,4-Dioxane, 1.0 x 10^{-5} M, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm^2 , 100 s), 436 nm (10 mW/cm^2 , 100 s).

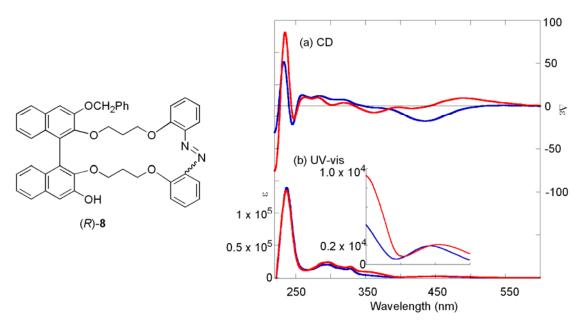


Figure S-7. (a) CD spectra of (R)-8, (b) Absorption spectra of (R)-8; after 365 nm-irradiation (blue line), after 436 nm-irradiation (red line). Conditions: 1,4-Dioxane, 1.0 x 10^{-5} M, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm^2 , 100 s), 436 nm (10 mW/cm^2 , 100 s).

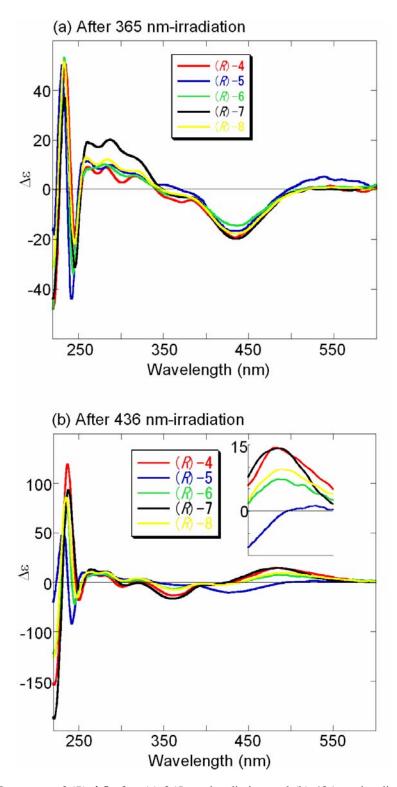


Figure S-8. CD spectra of (R)-4-8 after (a) 365 nm-irradiation and (b) 436 nm-irradiation. Conditions: 1,4-Dioxane, 1.0 x 10⁻⁵ M, 20 °C, light-path length = 10 mm, irradiation wavelength: 365 nm (10 mW/cm², 100 s), 436 nm (10 mW/cm², 100 s).

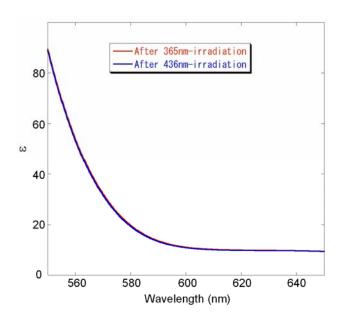


Figure S-9. Absorption spectra of concentrated (R)-4 after photo-irradiation. Conditions: CHCl₃, 0.0126 M (0.01 g/dl), light path length = 10 mm, 20 °C, Irradiation wavelength: 365 nm (10 mW/cm², 500 s), 436 nm (10 mW/cm², 500 s).

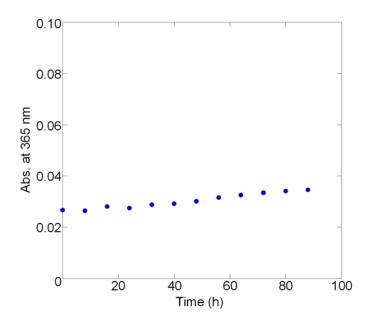


Figure S-10. Time-dependent change in absorbance of (R)-4 after 365 nm-irradiation. Conditions: 1,4-Dioxane, light path length = 10 mm, 20 °C in the dark, Monitoring wavelength: 365 nm, Irradiation wavelength: 365 nm (10 mW/cm², 100 s).

2. Thermodynamic parameters of cis-trans isomerization process of (R)-4 and (R)-7.

For the thermal *cis-trans* isomerization,

$$\ln\frac{\left[cis\right]_{t}}{\left[cis\right]_{0}} = -kt \qquad (1)$$

Where $[cis]_t$ and $[cis]_0$ are the concentrations of the cis-azobenzene at time t and time zero, respectively, and k is the rate constant for the thermal cis-trans isomerization. The first-order rate constant was determined by fitting the experimental data to the equation,

$$\ln \frac{A_{\infty} - A_t}{A_{\infty} - A_0} = -kt \qquad (2)$$

where A_t , A_0 and A_{∞} are the absorbance at 365 nm at time t, time zero and infinite time, respectively. The first order plots according to equation 2 for the *cis-trans* thermal isomerization of (R)-4 and (R)-7 at various temperatures are shown in Figure S-11.

Furthermore thermodynamic parameters such as enthalpy of activation (ΔH^{\ddagger}) and entropy of activation (ΔS^{\ddagger}) were determined according to the Eyring equation^{S2}:

$$k = \frac{k_B T}{h} \exp\left(\frac{\Delta S^{\ddagger}}{R}\right) \exp\left(-\frac{\Delta H^{\ddagger}}{RT}\right)$$
 (3)

where R, k_B , and h are gas constant, Boltzmann constant, and Planck's constant, respectively. Substituting into equation 3,

$$\ln\left(\frac{kh}{k_BT}\right) = -\frac{\Delta H^{\ddagger}}{RT} + \frac{\Delta S^{\ddagger}}{R} \qquad (4).$$

Figure S-12 shows Eyring plots for *cis* to *trans* thermal isomerization of (*R*)-4 and (*R*)-7. The values of ΔH^{\ddagger} and ΔS^{\ddagger} were obtained from intersect and aslope, respectively, of the linear plot of $\ln(hk/k_BT)$ versus 1/T extrapolated to $T \to \infty$.

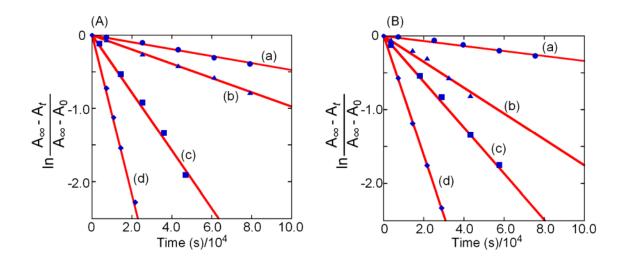


Figure S-11. First-order plots for *cis* to *trans* thermal isomerization of (*R*)-4 (A) and (*R*)-7 (B). (a) 35 °C, (b) 45 °C, (c) 55 °C, (d) 65 °C. Conditions: 1,4-Dioxane, 1.0×10^{-5} M.

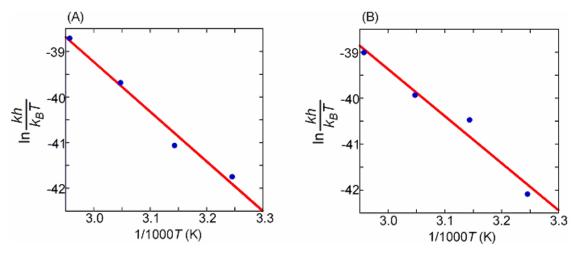


Figure S-12. Eyring plots for *cis* to *trans* thermal isomerization of (*R*)-4 and (*R*)-7. Conditions: 1,4-Dioxane, 1.0×10^{-5} M.

Table S-1. Rate coefficients k (s⁻¹) of cis to trans isomerization and the values of ΔH^{\ddagger} and ΔS^{\ddagger} of (R)-4 and (R)-7.

	<i>k</i> at 35 °C	k at 45 °C	k at 55 °C	<i>k</i> at 65 °C	ΔH^{\ddagger}	ΔS^{\ddagger}
(R)- 4	4.7 x 10 ⁻⁶	9.8 x 10 ⁻⁶	4.0 x 10 ⁻⁵	1.1 x 10 ⁻⁴	22 kcal/mol	-13 cal/mol·K
(R)- 7	3.4 x 10 ⁻⁶	1.8 x 10 ⁻⁵	3.1 x 10 ⁻⁵	8.1 x 10 ⁻⁵	20 kcal/mol	-17 cal/mol·K

Conditions: 1,4-Dioxane, 1.0 x 10⁻⁵ M.

3. Synthesis

Synthetic route of binaphthyl with azobenzene **4-8** is shown below. Binaphthyl 2^{S3} and dibromide 3^{S4} was prepared by following the reported procedure.

3,3'-Dibenzyloxybinaphthyl with azobenzene (S)-4

A suspension of (*S*)-2 (100 mg, 0.201 mmol), NaH (10.6 mg, 0.442 mmol, 2.2 eq.) and **3** (91 mg, 0.201 mmol, 1.0 eq.) in DMF (15 ml) was stirred for 12 hours at room temperature. The reaction mixture was poured into the mixed solvent of chloroform and water. The organic layer was separated and washed successively with 0.1 N hydrochloric acid solution, water (twice) and brine. After dried over sodium sulfate, the solvent was evaporated in vacuo to give a residue. The residue was purified by column chromatography (SiO₂; *n*-hexane/chloroform = 1/2) and GPC to afford (*S*)-4 (57 mg, 0.0719 mmol, 36%). Red amorphous; IR (KBr) 2959, 1439, 1281, 1250, 1017, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.2-2.1 (m, 4H), 3.2-4.3 (m, 8H), 4.88, 5.19 (ABq, v_{AB} = 61.7 Hz, J_{AB} = 11.2 Hz; s, 4H), 6.4-7.5 (m, 26H), 7.73, 7.78 (two d, J = 8.4 Hz, 8.4 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 29.5, 29.7, 64.8, 65.1, 68.9, 70.1, 70.5, 70.6, 108.4, 108.4, 113.1, 113.9, 119.4, 120.2, 124.0, 124.2, 125.0, 125.6, 125.7, 125.2, 126.3, 127.8, 127.9, 128.2, 128.5, 128.8, 128.8, 130.5, 130.6, 130.8, 136.4, 136.5, 142.6, 143.7, 145.9, 146.4, 151.1, 151.3, 152.8 (Some peaks overlapped); HRMS (ESI⁺) Calcd for $C_{52}H_{44}N_2O_6Na$ (M+Na)⁺: 815.3097. Found: 815.3076; LR MS (FAB⁺) 792.4 (M)⁺.

3,3'-Dimethoxybinaphthyl with azobenzene (R)-6

A suspension of (*R*)-**5** (30 mg, 0.0490 mmol), K_2CO_3 (34 mg, 0.245 mmol, 5 eq.) and MeI (30 μl, 0.490 mmol, 10 eq.) in DMF (1.5 ml) was stirred for 12 hours at room temperature. The reaction mixture was poured into the mixed solvent of chloroform and water. The organic layer was separated and washed successively with 0.1 N hydrochloric acid solution, water (twice) and brine. After dried over sodium sulfate, the solvent was evaporated in vacuo to give a residue. The residue was purified by column chromatography (SiO₂; *n*-hexane/chloroform/ethyl acetate = 4/3/1, then 2/3/1) and GPC to afford (*R*)-**6** (20 mg, 0.0312 mmol, 64%). Red amorphous; IR (KBr) 2959, 1281, 1250, 1022, 748 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.2-2.1 (m, 4H), 3.2-4.2 (m, 8H), 3.65, 3.83 (two s, 6H), 6.52, 6.61 (two d, *J* = 8.1 Hz, 8.1 Hz, 2H), 6.6-7.5 (m, 14H), 7.66, 7.71 (two d, *J* = 8.1 Hz, 8.1 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 29.7, 29.9, 55.1, 55.3, 64.9, 65.5, 69.1, 70.4, 107.0, 107.2, 113.1, 114.6, 119.4, 120.5, 120.5, 121.1, 123.9, 124.1, 125.1, 125.2, 125.6, 125.9, 126.2, 126.3, 126.5, 126.7, 128.1, 128.6, 130.5, 130.8, 130.9, 131.1, 142.7, 144.2, 145.9, 146.2, 151.8, 152.1, 152.3, 152.9; HRMS (ESI⁺) Calcd for $C_{40}H_{36}N_2O_6Na$ (M+Na)⁺: 663.2471. Found: 663.2464; LR MS (FAB⁺) 640.3 (M)⁺.

3,3'-Bis(diphenylmethyloxy)binaphthyl with azobenzene (R)-7

A suspension of (*R*)-**5** (64 mg, 0.104 mmol), K_2CO_3 (72 mg, 0.522 mmol, 5 eq.) and α-bromodiphenylmethane (257 mg, 1.04 mmol, 10 eq.) in DMF (4.0 ml) was stirred for 14 hours at 50 °C. The reaction mixture was poured into the mixed solvent of chloroform and water. The organic layer was separated and washed successively with 0.1 N hydrochloric acid solution, water (twice) and brine. After dried over sodium sulfate, the solvent was evaporated in vacuo to give a residue. The residue was purified by GPC to afford (*R*)-**7** (36 mg, 0.0381 mmol, 37%). Red amorphous; IR (KBr) 2960, 1283, 1248, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.2-2.0 (m, 4H), 3.2-4.4 (m, 8H), 6.07, 6.43 (two s, 2H), 6.4-7.5 (m, 36H), 7.56, 7.61 (two d, J = 8.4 Hz, 8.0 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 29.2, 29.7, 64.5, 65.5, 69.1, 70.8, 82.3, 82.9, 111.4, 111.7, 112.9, 114.3, 119.4, 120.3, 120.6, 122.2, 124.1, 124.3, 124.8, 125.6, 125.8, 126.1, 126.6, 126.7, 126.8, 127.0, 127.4, 127.6, 127.8, 127.9, 128.2, 128.6, 128.7, 128.9, 130.5, 130.6, 140.8, 140.9, 140.9, 141.1, 142.9, 143.8, 146.4, 147.1, 150.1, 150.4, 152.9; HRMS (FAB⁺) Calcd for $C_{64}H_{53}N_2O_6$ (M+H)⁺: 945.3907. Found: 945.3893; LR MS (FAB⁺) 944.4 (M)⁺.

3-Benzyloxy-3'-hydroxy-binaphthyl with azobenzene (R)-8

A suspension of (R)-5 (40 mg, 0.0653 mmol), K_2CO_3 (10.8 mg, 0.0783 mmol, 1.2 eq.) and benzyl bromide (8.2 μ l, 0.0686 mmol, 1.05 eq.) in DMF (4.0 ml) was stirred for 24 hours at 60 °C. The reaction mixture was poured into the mixed solvent of chloroform and water. The organic layer was separated and washed successively with 0.1 N hydrochloric acid solution, water (twice) and brine. After dried over sodium sulfate, the solvent was evaporated in vacuo to give a residue. The residue was

purified by GPC to afford (R)-**8** (16 mg, 0.0228 mmol, 35%), (R)-**4** (15 mg, 0.0189 mmol, 29%), unreacted (R)-**5** (10 mg, 0.0163 mmol, 25%). Red amorphous; IR (KBr) 3447, 3065, 1283, 1250, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.2-2.1 (m, 4H), 3.0-4.6 (m, 8H), 5.19, 5.30 (AB q, ν_{AB} = 17.2 Hz, J_{AB} = 11.2 Hz; s, 2H), 6.5-7.8 (m, 23H); ¹³C NMR (150 MHz, CDCl₃) δ 28.9, 29.2, 29.9, 30.1, 65.3, 66.0, 67.1, 68.7, 68.7, 69.1, 69.6, 70.4, 70.7, 71.8, 108.9, 109.3, 110.5, 111.7, 113.6, 114.0, 114.4, 116.6, 118.8, 119.1, 119.3, 120.6, 121.2, 121.5, 121.7, 122.2, 123.4, 123.7, 124.0, 124.4, 124.7, 124.8, 125.1, 125.2, 125.4, 125.8, 125.8, 125.9, 126.1, 126.2, 126.4, 126.5, 126.6, 126.6, 126.9, 127.5, 127.6, 128.0, 128.1, 128.2, 128.4, 128.5, 128.7, 128.8, 129.0, 129.0, 129.0, 130.4, 130.8, 131.0, 131.1, 131.2, 131.5, 131.6, 136.6, 136.7, 142.4, 144.4, 144.5, 144.6, 144.9, 146.3, 146.4, 146.6, 148.3, 151.5, 151.6, 152.8, 153.6 (Some peaks overlapped); HRMS (FAB⁺) Calcd for C₄₅H₃₈N₂O₆ (M)⁺: 702.2730. Found: 702.2729; LR MS (FAB⁺) 702.3 (M)⁺.

4. Determination of configuration of (R)-4.

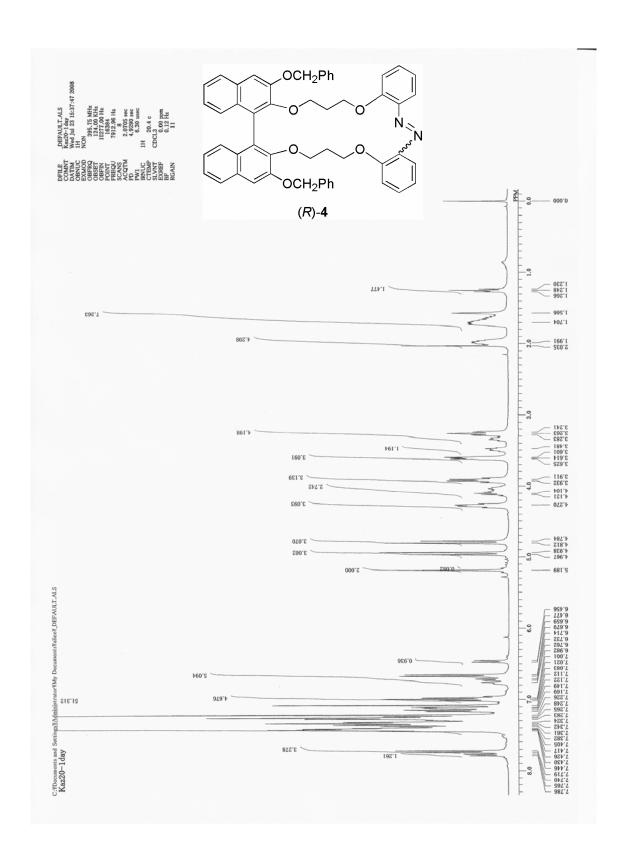
As a precautionary measure, **4** was derivatized to known compound $\mathbf{11}^{S5}$ to certify that inversion did not occur during the synthesis of **4** from (R)-**2**. Below is a scheme that outlined the derivatization of **4** synthesized from (R)-**2** to **11**. The specific optical rotation of derivatized **11** was +49°, whereas that of known compound (S)-**11** was -52°. Thus, the absolute configuration of **4** synthesized from (R)-**2** was (R) without inversion or racemization.

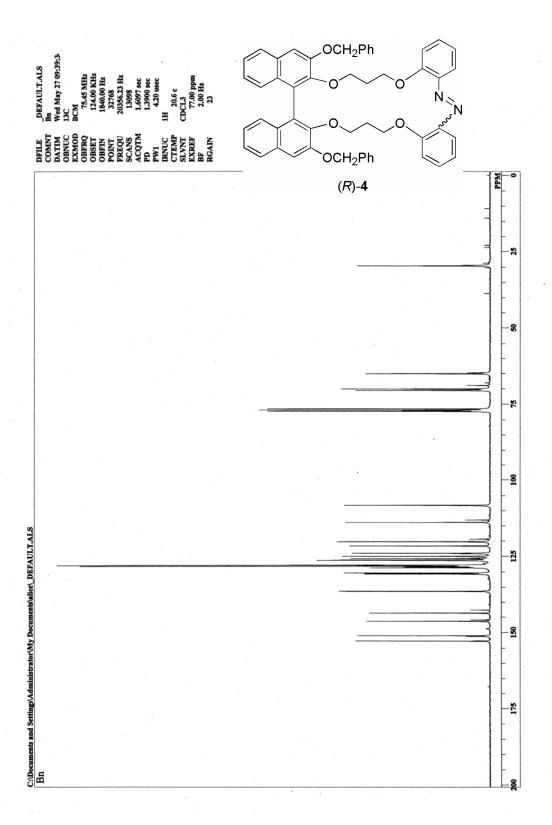
4 (from (R)-2)
BBr₃
OH
OH
$$K_2CO_3$$
 25% (2 steps)
OMe
OMe
OMe
OMe
 (R) -11
$$[\alpha]_D^{20} = +49 \text{ (CHCl}_3, c 0.10)}$$

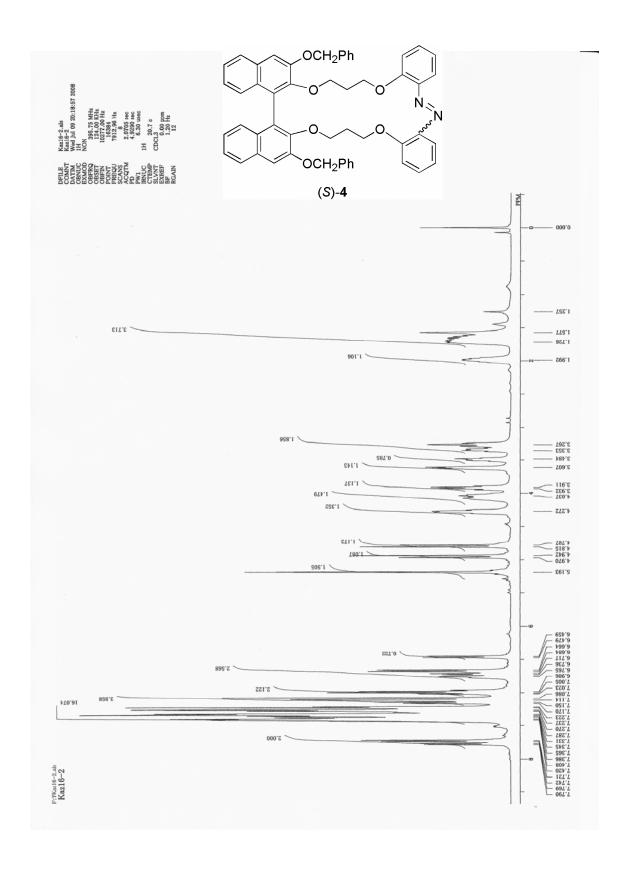
$$(S)$$
-11 is known,
$$[\alpha]_D^{20} = -52 \text{ (CHCl}_3, c 1.0)}$$

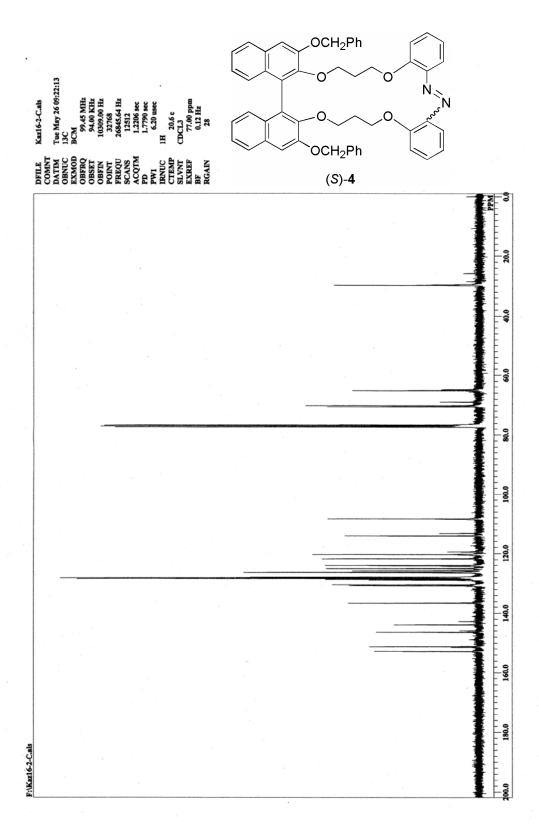
Tetramethoxybinaphthyl (R)-11.

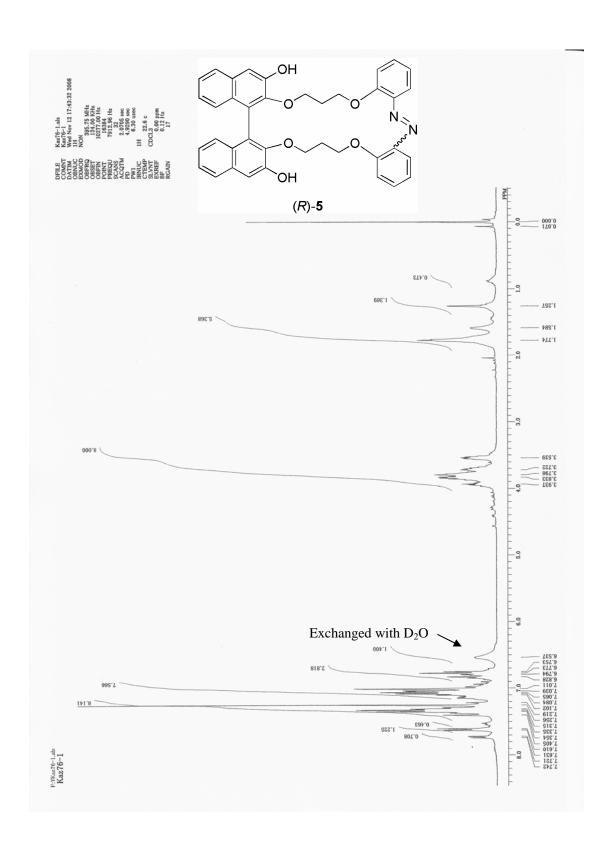
A suspension of (R)-4 (100 mg, 0.126 mmol) in CH₂Cl₂ (4.5 ml) containing BBr₃ (1.51 mmol, 12 eq.) was stirred for 24 hours at 0 °C. The reaction mixture was poured into the mixed solvent of chloroform and water. The organic layer was separated and washed successively with 0.1 N hydrochloric acid solution, water (twice) and brine. After dried over sodium sulfate, the solvent was evaporated in vacuo to afford crude (R)-10. Next, a suspension of crude (R)-10 (ca. 40 mg), K_2CO_3 (139 mg, 1.01 mmol, 8.0 eq.) and MeI (93 µl, 1.51 mmol, 12 eq.) in DMF (2.0 ml) was stirred for 24 hours at room temperature. The reaction mixture was poured into the mixed solvent of ethyl acetate and water. The organic layer was separated and washed successively with 0.1 N hydrochloric acid solution, water (twice) and brine. After dried over sodium sulfate, the solvent was evaporated in vacuo to give a residue. The residue was purified by preparative TLC (SiO₂; n-hexane/chloroform/ethyl acetate = 9/3/1) and GPC and preparative TLC (SiO₂; n-hexane/chloroform/ethyl acetate = 8/3/1) to afford (R)-11 (12 mg, 0.0321 mmol, 25% over 2 steps). Colorless powder; $\left[\alpha\right]_{D}^{20} = +49 (c \ 0.6, \text{CHCl}_3)$; IR (KBr) 2940, 1464, 1420, 1254, 1117 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 3.61 (s, 6H), 4.06 (s, 6H), 7.0-7.2 (m, 4H), 7.3-7.5 (m, 4H), 7.78 (d, J =8.1 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 55.6, 60.7, 107.2, 124.0, 125.2, 125.7, 125.9, 126.5, 128.9, 131.2, 147.3, 152.3; HR MS (ESI⁺) Calcd for C₂₄H₂₂O₄Na (M+Na)⁺: 397.1416. Found: 397.1411; LR MS (FAB^{+}) 374.2 $(M)^{+}$.

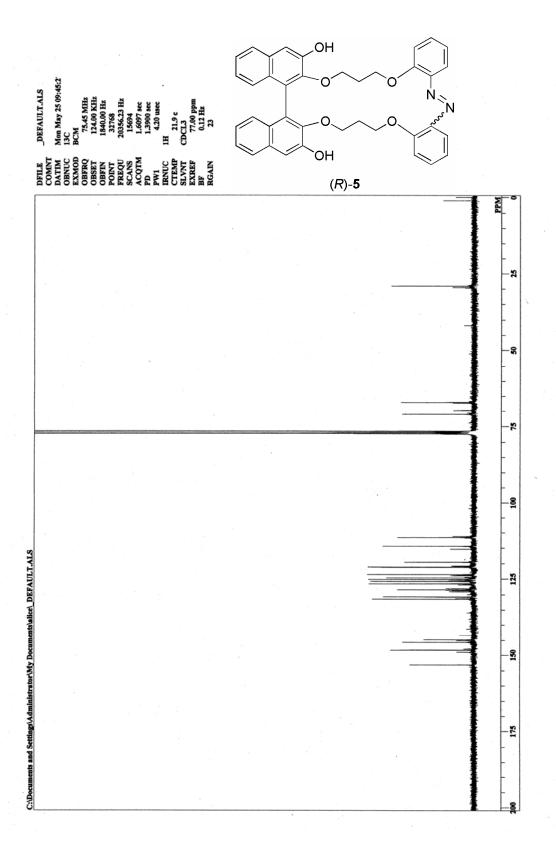


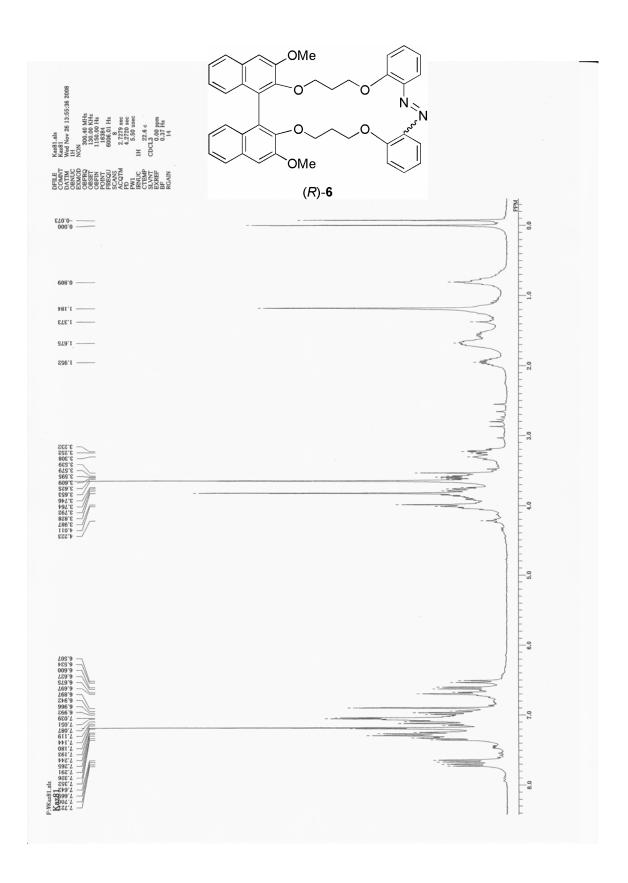


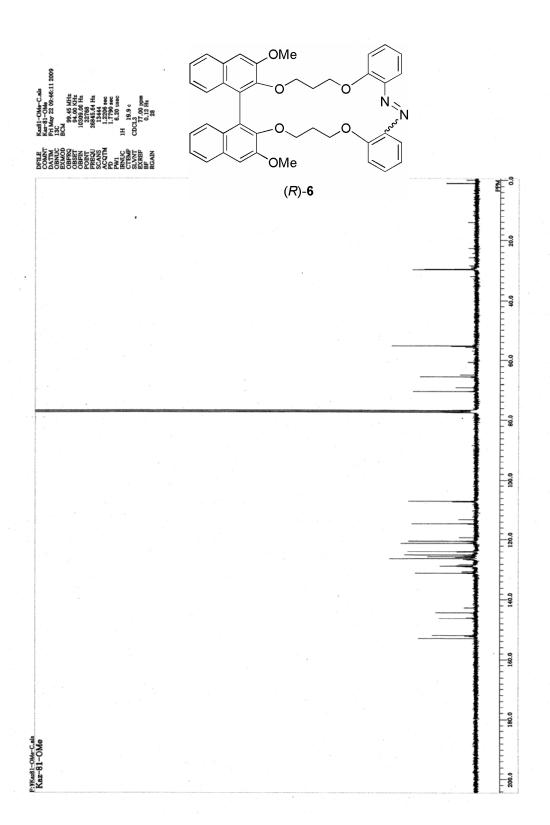


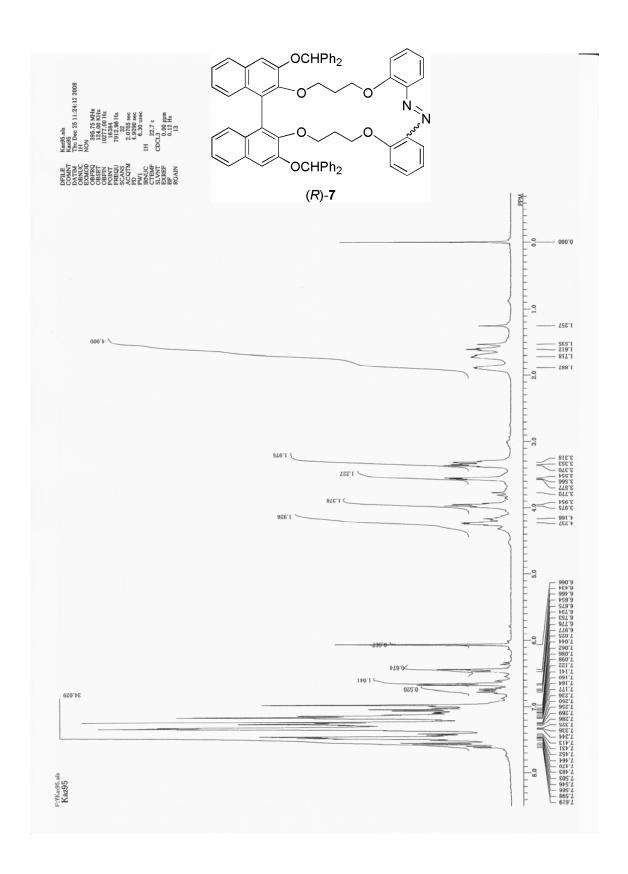


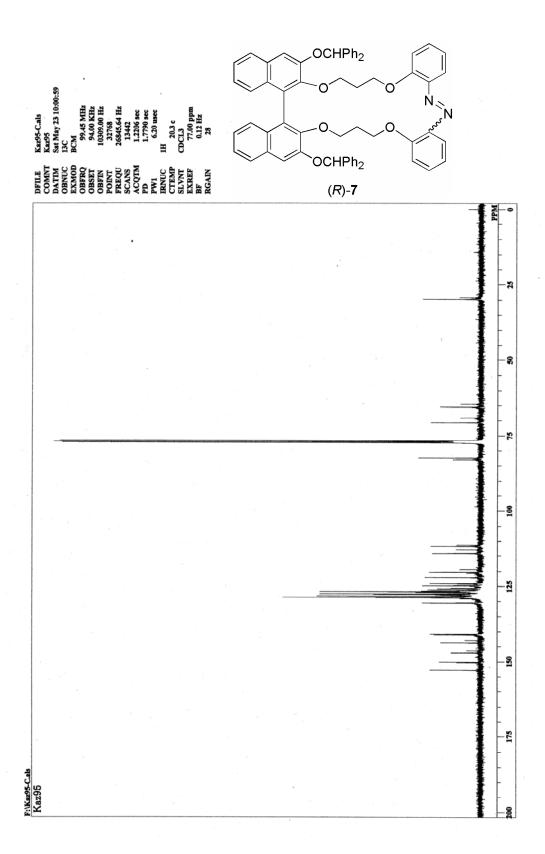


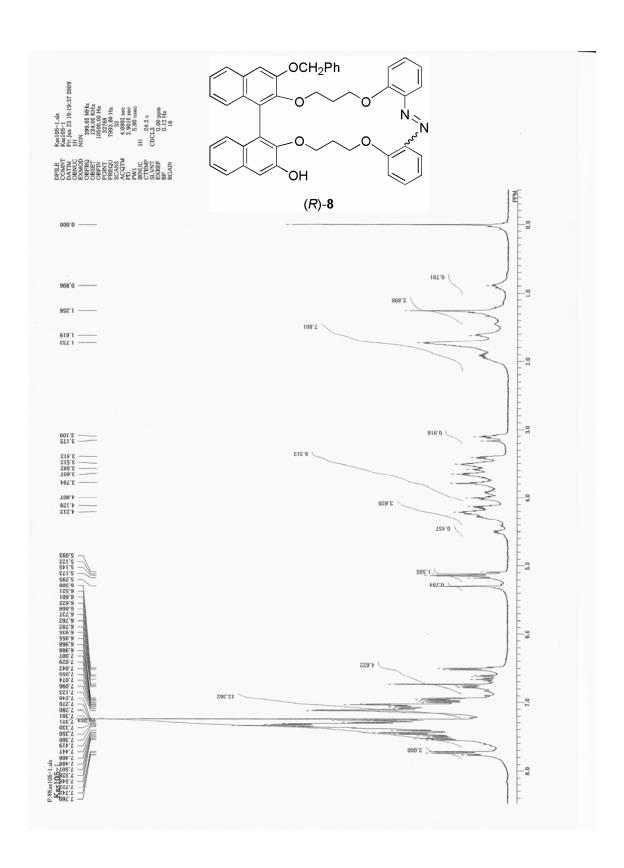


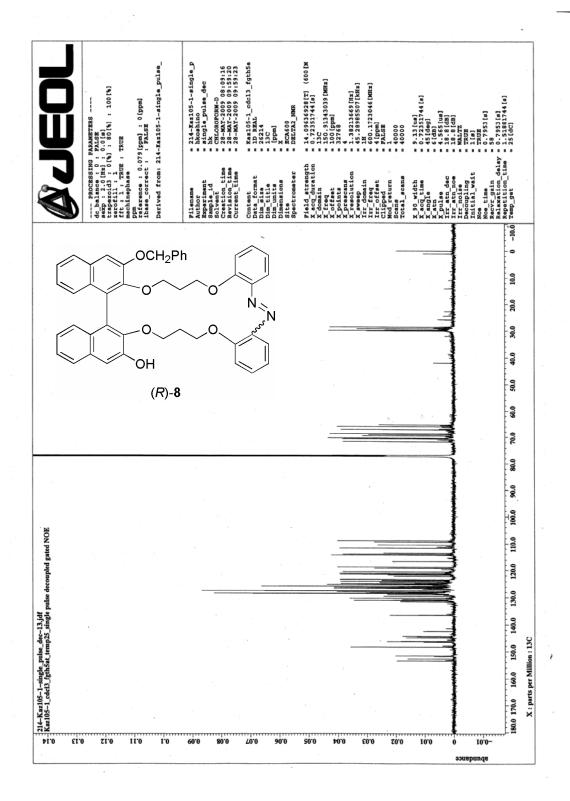


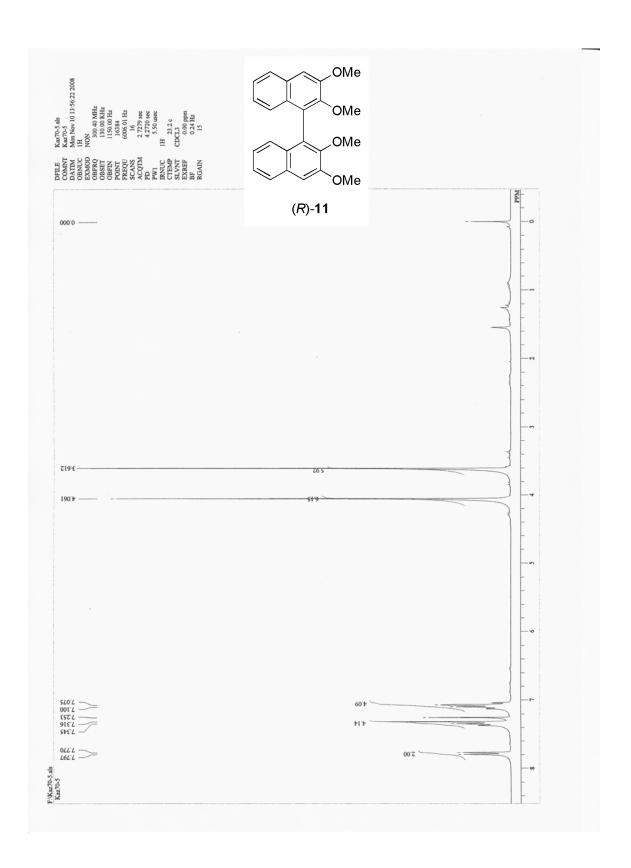


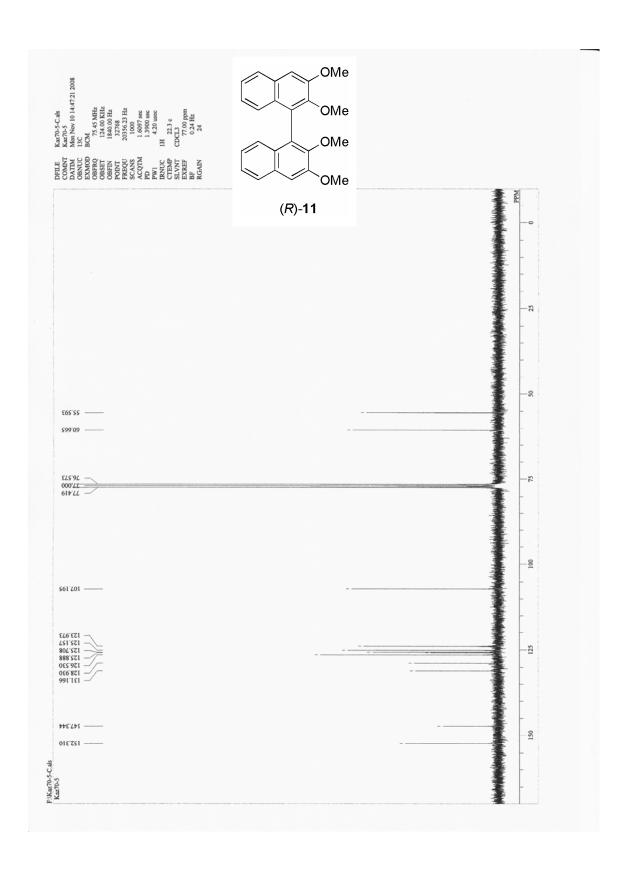












References

- (S1) Tsubaki, K.; Tanaka, H.; Furuta, T.; Tanaka, K.; Kinoshita, T.; Fuji, K. Tetrahedron, 2002, 58, 5611.
- (S2) Eyring, H. Chem. Rev. 1935, 17, 65.
- (S3) Tsubaki, K.; Morikawa, H.; Tanaka, H.; Fuji, K. Tetrahedron: Asymmetry 2003, 14, 1393-1396.
- (S4) Kawamoto, M.; Aoki, T.; Wada, T. Chem. Commun. 2007, 930.
- (S5) Furuta, T.; Tanaka, K.; Tsubaki, K.; Fuji, K. Tetrahedron 2004, 60, 4431.