## Malachite nanoparticle: a new basic hydrophilic surface for pH controlled adsorption of BSA with a high loading capacity

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**Supporting information** 

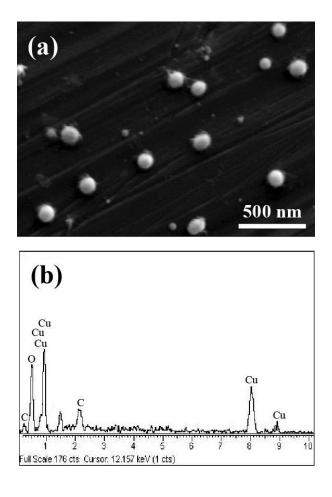


Figure S1. (a) SEM images of malachite NPs and (b) EDX spectra of malachite NPs showing the

presence of Cu, C and O.

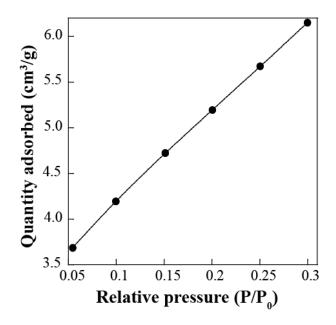


Figure S2. N<sub>2</sub> adsorption/desorption isotherms of malachite NPs measured from BET analysis.

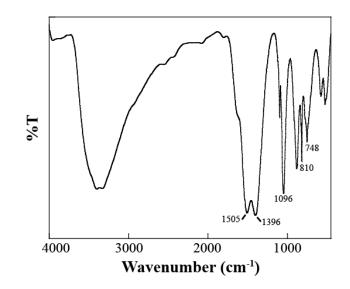
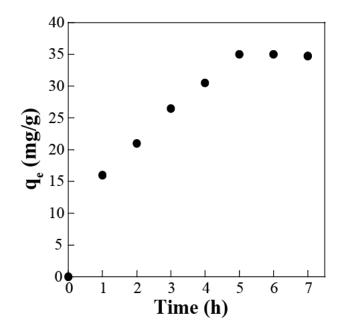
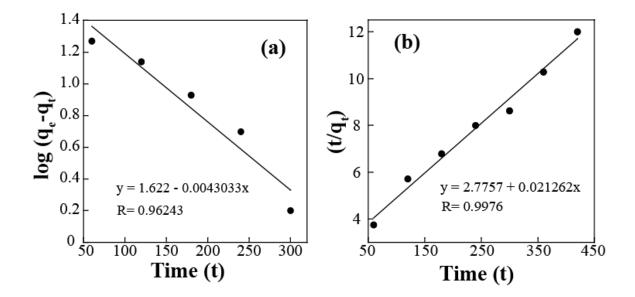


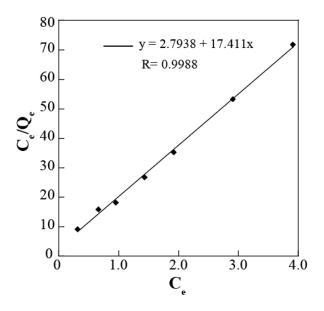
Figure S3. FT-IR spectra of malachite NPs.



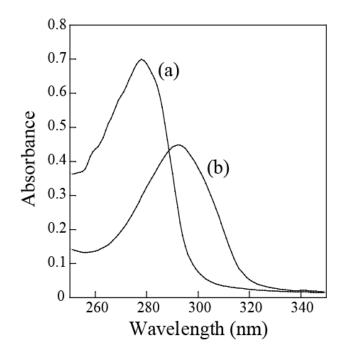
**Figure S4.** Adsorption kinetics with 1 mg/L BSA and 20 g/L malachite NPs at pH ~5.0. The steady state reaches within 5 h.



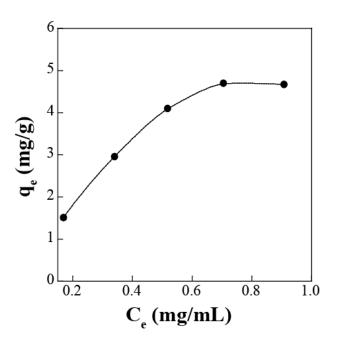
**Figure S5.** (a) Pseudo first order and (b) second order kinetics model of BSA adsorption studies on malachite performed with 1 mg/mL BSA and 20 g/L of malachite at pH ~5.0.



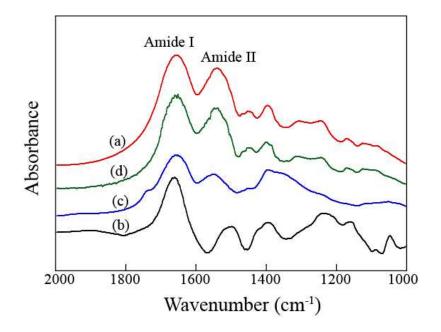
**Figure S4.** Scatchard plot of BSA adsorption on malachite NPs at different protein concentration. Correlation coefficient (R) value of 0.998 implies no significant co-operative effect.



**Figure S5.** UV spectra of BSA in (a) native and (b) after desorption from malachite surface. Desorbed spectra shows a ~10 nm shift which is due to alteration in protein conformation upon adsorption-desorption cycle.



**Figure S6.** Steady state adsorption isotherm of BSA on stearic acid coated malachite NPs at pH ~5.0. Maximum adsorption capacity was very much lower (~5 mg/g) compared to that on bare malachite surface (~50 mg/g).



**Figure S7.** FTIR of lyophilized BSA in (a) native, (b) adsorbed (after subtraction of only malachite spectra), (c) desorbed and (d) after interaction with SA-malachite NPs.

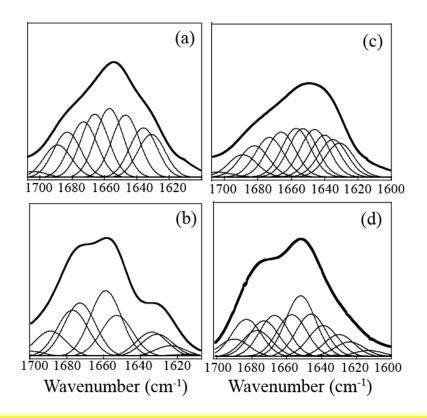
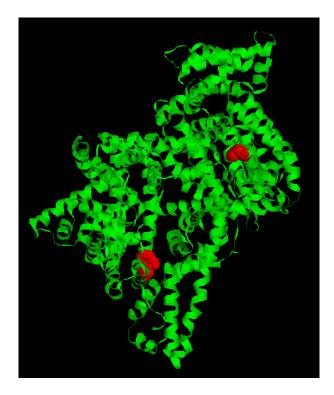


Figure S8. Gaussian distribution plot of amide I peak of FT-IR spectra of lyophilized BSA at (a) native,

(b) adsorbed, (c) desorbed and (d) after interaction with SA-malachite NPs.

Area (%) of Gaussian bands of BSA					
orbed of	d on	after		interacted wit	h assignment
state malachite NI		s desorption		SA-malachite N	<b>JPs</b>
2 ± 1	1	4 ± 1		3 ± 1	Un-ordered
7 ± 2	2	$12 \pm 2$	2	6 ± 1	β-sheet
5 ± 1	1	4 ± 1		$12 \pm 2$	Un-ordered
7 ± 1	1	8 ± 2		$10 \pm 1$	Un-ordered
$1 \pm 2$	2	8 ± 1		$6 \pm 2$	Un-ordered
5 ± 1	1	$23 \pm 2$	2	$30 \pm 1$	α-helix
↓±1	1	8 ± 2		8 ± 2	Un-ordered
2 ± 1	1	$11 \pm 2$	2	6 ± 1	β-sheet
$3 \pm 2$	2	9±1		6 ± 1	Un-ordered
5±1	1	$9 \pm 2$		9 ± 1	β-sheet
$3 \pm 1$	1	4 ± 1		$4 \pm 1$	Un-ordered

**TABLE S1.** Secondary structural elements of lyophilized BSA at different stages as calculated fromGaussian distribution of amide I peak of FT-IR spectra.



**Figure S9.** PDB structure of 'BSA-like' human serum albumin dimer showing the tryptophan residue (red) (PDB ID 1bm0).