## **Supporting Information**

## Mechanistic studies of photoinitiated free radical polymerization using a bifunctional thioxanthone acetic acid derivative as photoinitiator.

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## Detection of gaseous CO<sub>2</sub>, which is generated by photolysis of TX-Ct and TX-Ma.

1 mL of each solution of TX-Ma and TX-Ct (5 mM) in DMF was placed in a Pyrex tube which was connected to another tube containing an aqueous solution of  $Na_2CO_3$  (0.67 mM) and one drop of phenolphthalein solution (0.025 mM). Under our irradiation conditions (UVA: 12 W/m<sup>2</sup>, UVB: 50 W/m<sup>2</sup>) the pink color of the phenolphthalein solution disappeared after approximately 1 hour of irradiation of the two photoinitiators, which indicates  $CO_2$  generation.



**Figure S1:** Schematic representation of the apparatus to demonstrate decarboxyation of photoinitiators.

Determination of bi-molecular quenching rate constants by molecular oxygen.



**Figure S2:** Dependence of the pseudo-first-order rate constant of the transient absorption decays at 320 nm and 620 nm on the concentration of dissolved molecular oxygen in acetonitrile. The transients were generated by pulsed laser excitation (355 nm, 5 ns) of acetonitrile solutions of TX-Ma. Varying oxygen concentrations were achieved by bubbling the sample solutions with different  $O_2/N_2$  mixtures. Saturation concentration of acetonitrile after bubbling with  $O_2$  at 24°C:  $[O_2] = 9.1$  mM (Clark, W. D. K.; Steel, C. J. *Am. Chem. Soc.* **1971**, *93*, 6347-6355).

UV-Vis analysis of polymer generated by photopolymerization.



