

Supporting Information for

Synthesis of (Arylimido)vanadium(V) Complexes Containing (2-Anilidomethyl)pyridine Ligands and their Use as the Catalyst Precursors for Olefin Polymerization

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V(N-2,6-Me₂C₆H₃)(CH₂SiMe₃)₂[2-(2,6-ⁱPr₂C₆H₃)NCH₂(C₅H₃N)] (**2b**) in C₆D₆ at 80 °C in the
presence/absence of PMe₃.
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V(N-2,6-Me₂C₆H₃)Cl₂[2-(2,6-R₂C₆H₃)NCH₂(C₅H₃N)] [R = ⁱPr (**1b**), F (**1c**)] and
V(N-2,6-Me₂C₆H₃)(CH₂SiMe₃)₂[2-(2,6-R₂C₆H₃)NCH₂(C₅H₃N)] [R = Me (**2a**), ⁱPr (**2b**)].
- 5) Crystal structure reports and CIF files for **1b**, **1c**, **2a** and **2b**.

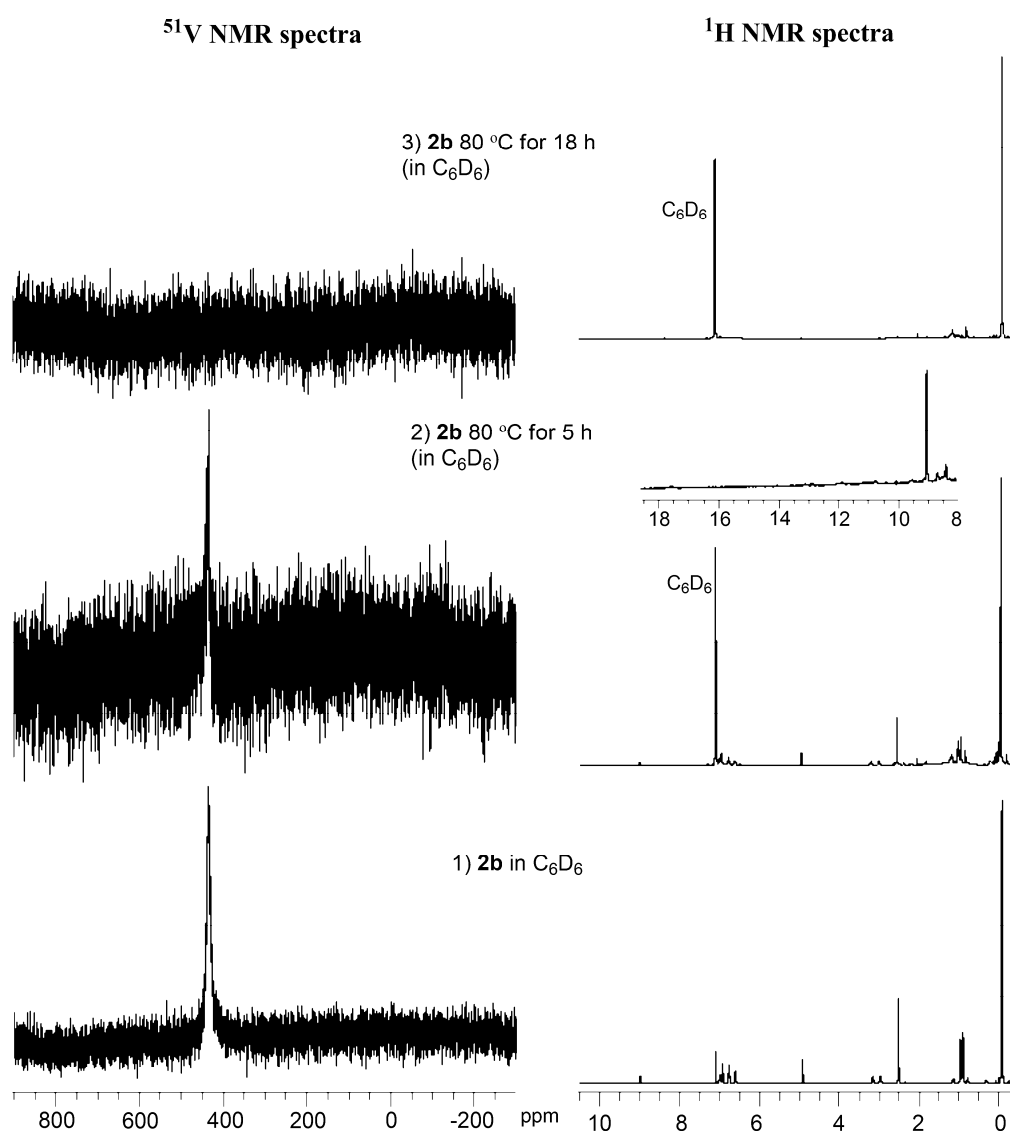
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1. ^1H and ^{51}V NMR spectra for monitoring reactions of $\text{V}(\text{N-2,6-Me}_2\text{C}_6\text{H}_3)(\text{CH}_2\text{SiMe}_3)_2$ - $[\text{2-(2,6-}^i\text{Pr}_2\text{C}_6\text{H}_3)\text{NCH}_2(\text{C}_5\text{H}_3\text{N})]$ (**2b**) in C_6D_6 at 80 °C in the presence/absence of PMe_3 .

1-1. Reaction (thermolysis) of **2b** in C_6D_6 at 80 °C

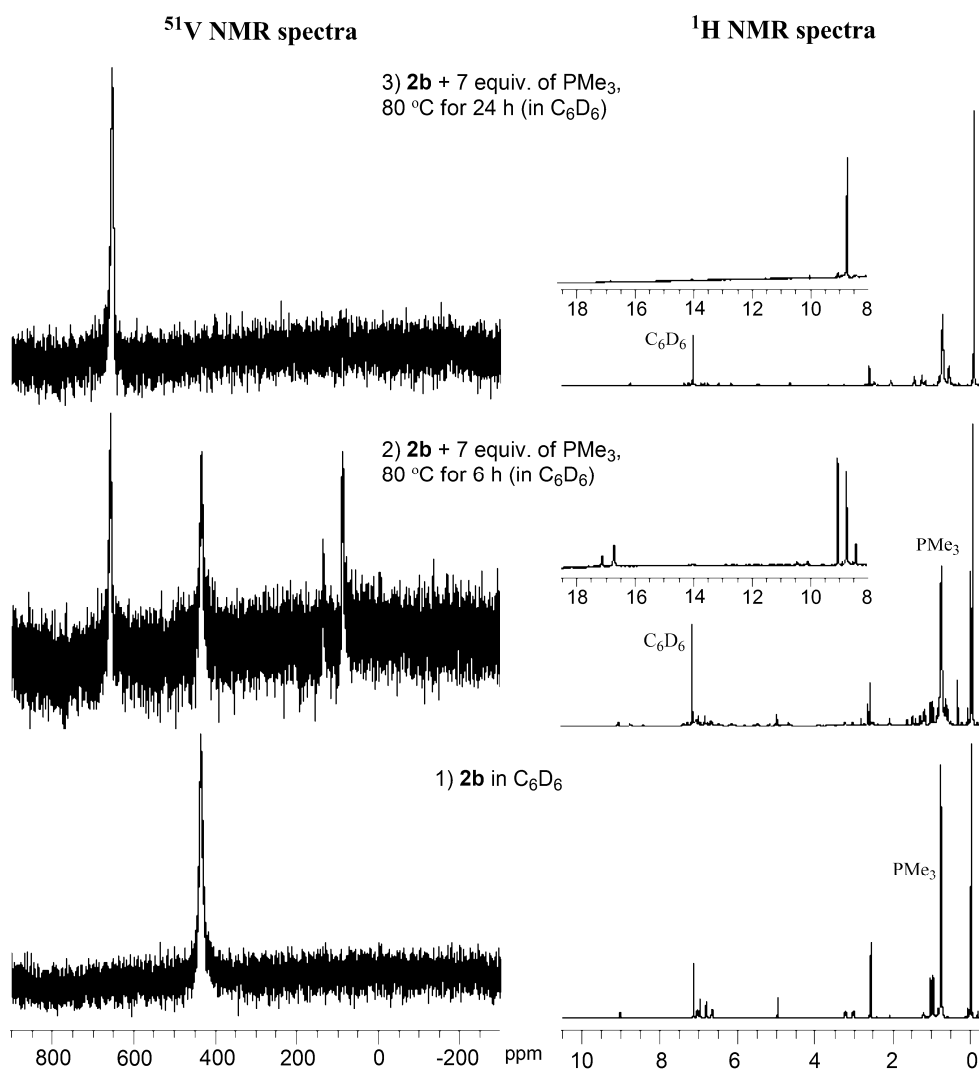
Procedure: The C_6D_6 solution (0.5 mL) containing **2b** (31 mg) was heated at 80 °C, and the reaction was monitored by both ^1H and ^{51}V NMR spectra.



Additional note: No resonances in the ^{51}V NMR spectrum were seen after 18 h, probably suggesting that the reaction may give certain paramagnetic species [vanadium(III) or vanadium(IV)].

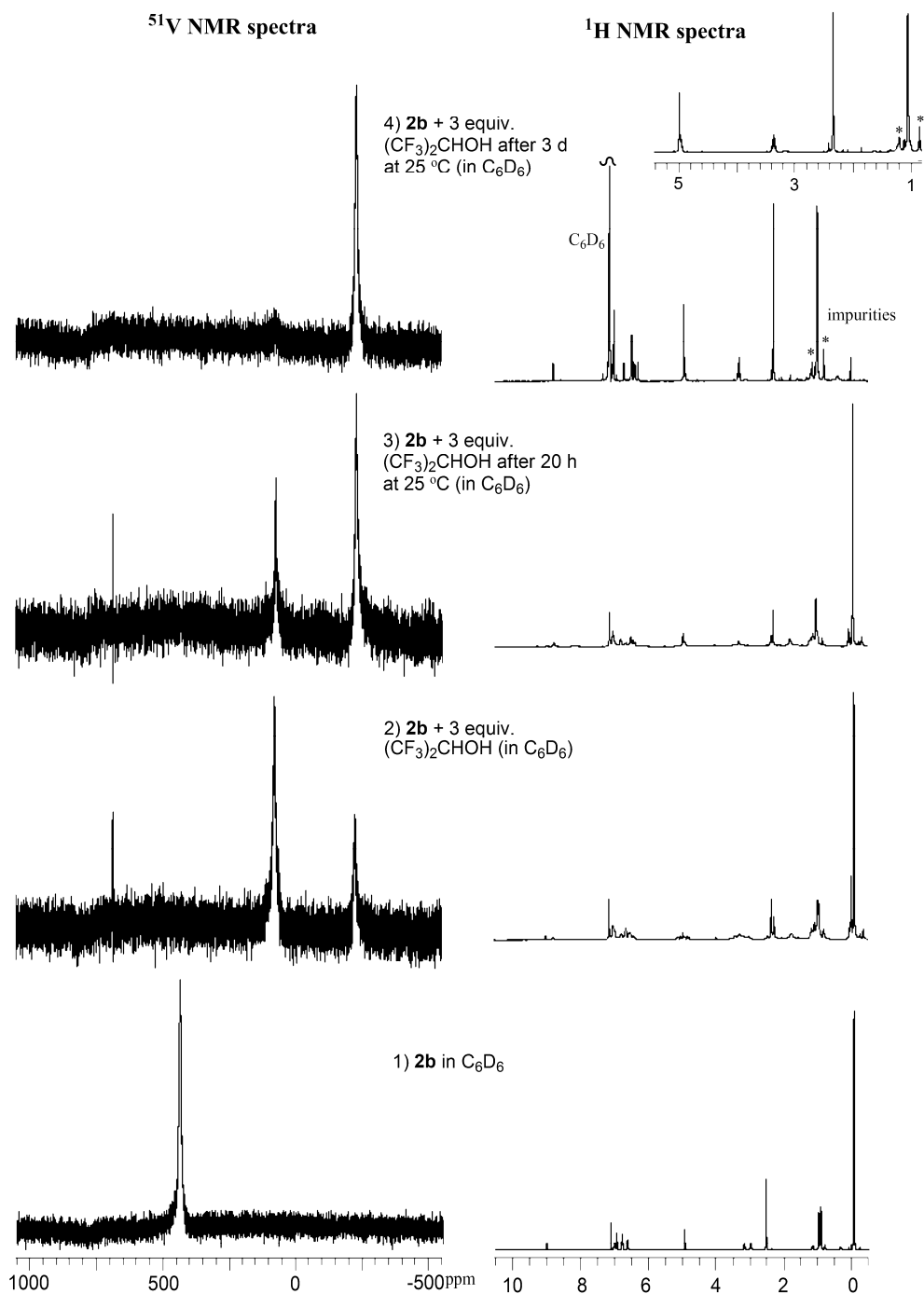
1-2. Reaction of **2b** in C₆D₆ in the presence of PMe₃ (7 equiv.).

Procedure: The C₆D₆ solution (0.5 mL) containing complex **2b** (31 mg) and PMe₃ (27 mg, 7 equiv.) was heated at 80 °C, and the reaction was monitored by ¹H and ⁵¹V NMR spectra.

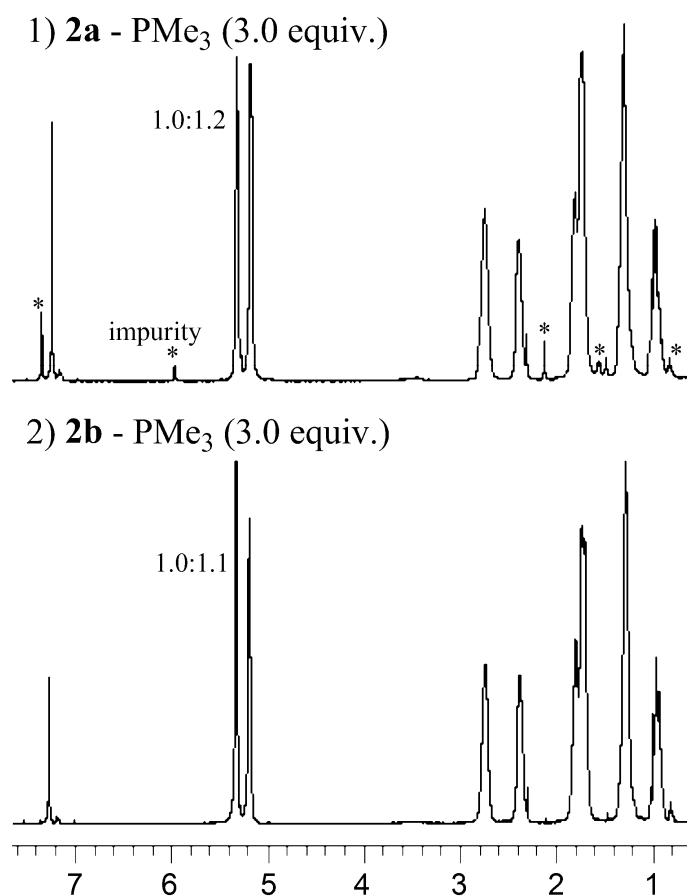


Additional note: Resonances seen at δ 16.7 and 17.1 ppm (after 6 h, in the ¹H NMR spectrum) presumably ascribed to the protons of the vanadium(V)-alkylidene(s) disappeared after 24 h to afford another (unidentified) species. For comparison, δ 14.52 ppm in V(CHSiMe₃)(NAr)(N=C^tBu₂)(PMe₃) (reference 14a,c), δ 16.09 and 16.38 ppm in V(CHSiMe₃)(NAr)(O-2,6-Me₂C₆H₃)(PMe₃)_{0.89} (reference 14b). Although only one resonance was seen in the ⁵¹V NMR spectrum after 24 h, the isolation from the mixture was not successful (due to a difficulty to isolate as microcrystals from the mixture probably containing certain amount of decomposed and/or paramagnetic products.)

2. ^1H and ^{51}V NMR spectra for monitoring reactions of **2b** with 3 equiv. of $(\text{CF}_3)_2\text{CHOH}$ in C_6D_6 at 25 °C. Experimental details are shown in the Experimental Section.



3. ^1H NMR spectra in CDCl_3 for ring-opened poly(norbornene) prepared by **2b** in the presence of PMe_3 (Table 4).



Additional note: The spectra clearly indicates that the resultant polymer possessed ring-opened structure containing a mixture of *cis/trans* olefinic double bonds.

4) Crystal data and collection parameters of complexes

$V(N\text{-}2,6\text{-Me}_2C_6H_3)Cl_2[2\text{-(}2,6\text{-R}_2C_6H_3)NCH_2(C_5H_3N)]$ [R = *i*Pr (**1b**), F (**1c**)] and

$V(N\text{-}2,6\text{-Me}_2C_6H_3)(CH_2SiMe_3)_2[2\text{-(}2,6\text{-R}_2C_6H_3)NCH_2(C_5H_3N)]$ [R = Me (**2a**), *i*Pr (**2b**)].

	1b	1c	2a	2b
Formula	C ₂₆ H ₃₂ Cl ₂ N ₃ V	C ₂₃ H ₁₈ Cl ₂ F ₂ N ₃ V	C ₃₀ H ₄₆ N ₃ Si ₂ V	C ₃₄ H ₅₄ N ₃ Si ₂ V
Formula Weight	508.41	496.26	555.83	611.93
Crystal color, Habit	red, block	red, block	red, block	red, block
Crystal size (mm)	0.25 × 0.25 × 0.10	0.40 × 0.30 × 0.20	0.50 × 0.40 × 0.40	0.20 × 0.15 × 0.10
Crystal system	monoclinic	triclinic	monoclinic	triclinic
Space group	<i>P</i> 2 ₁ / <i>n</i> (#14)	<i>P</i> -1 (#2)	<i>P</i> 2 ₁ / <i>c</i> (#14)	<i>P</i> -1 (#2)
<i>a</i> (Å)	11.3290(7)	7.3016(4)	15.9173(4)	10.1893(9)
<i>b</i> (Å)	16.2926(7)	10.2681(6)	11.9182(3)	11.0729(9)
<i>c</i> (Å)	14.8872(8)	15.9162(12)	17.7030(4)	16.2502(16)
α (deg)		99.261(3)		97.395(3)
β (deg)	106.8908(18)	92.447(3)	103.6830(7)	95.290(3)
γ (deg)		97.1083(15)		96.446(2)
<i>V</i> (Å ³)	2629.3(2)	1166.29(12)	3263.05(13)	1796.1(3)
<i>Z</i> value	4	2	4	2
<i>D</i> _{calcd} (g/cm ³)	1.284	1.413	1.131	1.131
<i>F</i> 000	1064.00	504.00	1192.00	660.00
Temp (K)	193	193	193	153
μ (MoK α) (cm ⁻¹)	5.980	6.843	3.985	3.677
No. of reflections measured	Total: 25405	Total: 9482	Total: 26522	Total: 13412
	Unique: 5988	Unique: 4207	Unique: 5936	Unique: 5866
No. of observations (<i>I</i> > 2.00 σ (<i>I</i>))	3126	4207	5233	5866
No. of Variables	321	281	371	362
<i>R</i> 1 (<i>I</i> > 2.00 σ (<i>I</i>))	0.0398	0.0404	0.0372	0.0637
<i>wR</i> 2 (<i>I</i> > 2.00 σ (<i>I</i>))	0.0954	0.1316	0.1323	0.2134
Goodness of Fit	1.003	1.154	1.008	1.181

X-ray Structure Report for $\text{V}(\text{N}=2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Cl}_2[2\text{-(2,6-}^i\text{Pr}_2\text{C}_6\text{H}_3)\text{CH}_2(\text{C}_5\text{H}_4\text{N})]$ (**1b**)

July 14, 2009

Experimental

Data Collection

A red block crystal of $C_{26}H_{32}Cl_2N_3V$ having approximate dimensions of 0.25 x 0.25 x 0.10 mm was mounted on a glass fiber. All measurements were made on a Rigaku RAXIS RAPID imaging plate area detector with graphite monochromated Mo-K α radiation.

Indexing was performed from 3 oscillations that were exposed for 420 seconds. The crystal-to-detector distance was 127.40 mm.

Cell constants and an orientation matrix for data collection corresponded to a primitive monoclinic cell with dimensions:

$$\begin{aligned} a &= 11.3290(7) \text{ \AA} \\ b &= 16.2926(7) \text{ \AA} \\ c &= 14.8872(8) \text{ \AA} \\ V &= 2629.3(2) \text{ \AA}^3 \end{aligned} \quad \beta = 106.8908(18)^\circ$$

For $Z = 4$ and F.W. = 508.41, the calculated density is 1.284 g/cm³. The systematic absences of:

$$\begin{aligned} h0l: & h+l \pm 2n \\ 0k0: & k \pm 2n \end{aligned}$$

uniquely determine the space group to be:

$$P2_1/n \text{ (\#14)}$$

The data were collected at a temperature of $-80 \pm 1^\circ\text{C}$ to a maximum 2θ value of 54.9° . A total of 55 oscillation images were collected. A sweep of data was done using ω scans from 130.0 to 190.0° in 4.0° step, at $\chi=45.0^\circ$ and $\phi = 0.0^\circ$. The exposure rate was 240.0 [sec./ $^\circ$]. A second sweep was performed using ω scans from 0.0 to 160.0° in 4.0° step, at $\chi=45.0^\circ$ and $\phi = 180.0^\circ$. The exposure rate was 240.0 [sec./ $^\circ$]. The crystal-to-detector distance was 127.40 mm. Readout was performed in the 0.100 mm pixel mode.

Data Reduction

Of the 25405 reflections that were collected, 5988 were unique ($R_{\text{int}} = 0.062$); equivalent reflections were merged.

The linear absorption coefficient, μ , for Mo-K α radiation is 5.980 cm⁻¹. An empirical absorption correction was applied which resulted in transmission factors ranging from 0.568 to 0.942. The data were corrected for Lorentz and polarization effects.

Structure Solution and Refinement

The structure was solved by direct methods¹ and expanded using Fourier techniques². The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were refined using the riding model. The final cycle of full-matrix least-squares refinement³ on F^2 was based on 3126 observed reflections and 321 variable parameters and converged (largest parameter shift was 0.00 times its esd) with unweighted and weighted agreement factors of:

$$R1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o| = 0.0398$$

$$wR2 = [\Sigma (w (F_o^2 - F_c^2)^2) / \Sigma w(F_o^2)^2]^{1/2} = 0.0954$$

The standard deviation of an observation of unit weight⁴ was 1.00. A Sheldrick weighting scheme was used. Plots of $\Sigma w (|F_o| - |F_c|)^2$ versus $|F_o|$, reflection order in data collection, $\sin \theta/\lambda$ and various classes of indices showed no unusual trends. The maximum and minimum peaks on the final difference Fourier map corresponded to 0.70 and -0.27 e⁻/Å³, respectively.

Neutral atom scattering factors were taken from Cromer and Waber⁵. Anomalous dispersion effects were included in F_{calc} ⁶; the values for $\Delta f'$ and $\Delta f''$ were those of Creagh and McAuley⁷. The values for the mass attenuation coefficients are those of Creagh and Hubbell⁸. All calculations were performed using the CrystalStructure^{9,10} crystallographic software package.

References

(1) SIR92: Altomare, A., Casciarano, G., Giacovazzo, C., Guagliardi, A., Burla, M., Polidori, G., and Camalli, M. (1994) J. Appl. Cryst., 27, 435.

(2) DIRDIF99: Beurskens, P.T., Admiraal, G., Beurskens, G., Bosman, W.P., de Gelder, R., Israel, R. and Smits, J.M.M.(1999). The DIRDIF-99 program system, Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands.

(3) Least Squares function minimized:

$$\sum w(F_o^2 - F_c^2)^2 \quad \text{where } w = \text{Least Squares weights.}$$

(4) Standard deviation of an observation of unit weight:

$$[\sum w(F_o^2 - F_c^2)^2 / (N_o - N_v)]^{1/2}$$

where: N_o = number of observations
 N_v = number of variables

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(7) Creagh, D. C. & McAuley, W.J. ; "International Tables for Crystallography", Vol C, (A.J.C. Wilson, ed.), Kluwer Academic Publishers, Boston, Table 4.2.6.8, pages 219-222 (1992).

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EXPERIMENTAL DETAILS

A. Crystal Data

Empirical Formula	C ₂₆ H ₃₂ Cl ₂ N ₃ V
Formula Weight	508.41
Crystal Color, Habit	red, block
Crystal Dimensions	0.25 X 0.25 X 0.10 mm
Crystal System	monoclinic
Lattice Type	Primitive
Indexing Images	3 oscillations @ 420.0 seconds
Detector Position	127.40 mm
Pixel Size	0.100 mm
Lattice Parameters	a = 11.3290(7) Å b = 16.2926(7) Å c = 14.8872(8) Å β = 106.8908(18) ° V = 2629.3(2) Å ³
Space Group	P2 ₁ /n (#14)
Z value	4
D _{calc}	1.284 g/cm ³
F ₀₀₀	1064.00
μ(MoKα)	5.980 cm ⁻¹

B. Intensity Measurements

Diffractometer	Rigaku RAXIS-RAPID
Radiation	MoK α (λ = 0.71075 Å) graphite monochromated
Detector Aperture	280 mm x 256 mm
Data Images	55 exposures
ω oscillation Range (χ =45.0, ϕ =0.0)	130.0 - 190.0°
Exposure Rate	240.0 sec./°
ω oscillation Range (χ =45.0, ϕ =180.0)	0.0 - 160.0°
Exposure Rate	240.0 sec./°
Detector Position	127.40 mm
Pixel Size	0.100 mm
$2\theta_{\max}$	54.9°
No. of Reflections Measured	Total: 25405 Unique: 5988 (R_{int} = 0.062)
Corrections	Lorentz-polarization Absorption (trans. factors: 0.568 - 0.942)

C. Structure Solution and Refinement

Structure Solution	Direct Methods (SIR92)
Refinement	Full-matrix least-squares on F^2
Function Minimized	$\sum w (F_o^2 - F_c^2)^2$
Least Squares Weights	$1/[0.0008F_o^2 + 1.0000\sigma(F_o^2)]/(4F_o^2)$
$2\theta_{\text{max}}$ cutoff	54.9°
Anomalous Dispersion	All non-hydrogen atoms
No. Observations ($I > 2.00\sigma(I)$)	3126
No. Variables	321
Reflection/Parameter Ratio	9.74
Residuals: R1 ($I > 2.00\sigma(I)$)	0.0398
Residuals: wR2 ($I > 2.00\sigma(I)$)	0.0954
Goodness of Fit Indicator	1.003
Max Shift/Error in Final Cycle	0.000
Maximum peak in Final Diff. Map	0.70 e ⁻ /Å ³
Minimum peak in Final Diff. Map	-0.27 e ⁻ /Å ³

Table 1. Atomic coordinates and $B_{\text{iso}}/B_{\text{eq}}$

atom	x	y	z	B_{eq}
V(1)	0.52187(5)	0.21060(3)	0.71452(3)	2.464(12)
Cl(1)	0.38770(10)	0.10409(5)	0.68731(6)	4.38(2)
Cl(2)	0.72897(8)	0.19223(5)	0.78155(6)	4.12(2)
N(1)	0.4744(2)	0.29724(13)	0.63156(16)	2.29(5)
N(2)	0.5608(2)	0.16298(14)	0.58699(16)	2.52(5)
N(3)	0.4926(2)	0.25352(14)	0.80834(16)	2.60(5)
C(1)	0.4497(3)	0.28513(17)	0.52960(19)	2.74(6)
C(2)	0.5066(2)	0.20718(17)	0.5105(2)	2.32(6)
C(3)	0.5023(3)	0.18079(19)	0.4210(2)	2.95(7)
C(4)	0.5570(3)	0.1063(2)	0.4115(2)	3.33(8)
C(5)	0.6160(3)	0.06240(19)	0.4903(2)	3.45(8)
C(6)	0.6168(3)	0.09144(18)	0.5763(2)	3.12(7)
C(7)	0.4608(2)	0.38220(17)	0.65577(18)	2.30(6)
C(8)	0.5618(2)	0.43534(17)	0.66305(19)	2.57(6)
C(9)	0.5517(3)	0.51566(19)	0.6937(2)	3.44(7)
C(10)	0.4474(3)	0.5413(2)	0.7150(2)	3.86(8)
C(11)	0.3475(3)	0.4894(2)	0.7030(2)	3.63(8)
C(12)	0.3515(3)	0.40864(18)	0.6720(2)	2.86(7)
C(13)	0.6762(2)	0.41004(19)	0.6363(2)	2.98(7)
C(14)	0.7947(3)	0.4224(2)	0.7147(2)	4.55(9)
C(15)	0.6848(3)	0.4569(2)	0.5498(2)	4.94(10)
C(16)	0.2372(3)	0.3551(2)	0.6505(2)	3.51(8)
C(17)	0.1659(3)	0.3635(2)	0.7213(2)	5.24(11)
C(18)	0.1501(3)	0.3753(2)	0.5523(2)	4.62(9)
C(19)	0.4720(3)	0.28444(18)	0.8893(2)	2.76(7)
C(20)	0.5513(3)	0.3462(2)	0.9415(2)	3.43(8)
C(21)	0.5279(4)	0.3753(2)	1.0229(2)	4.93(10)
C(22)	0.4328(4)	0.3456(2)	1.0521(2)	5.41(11)
C(23)	0.3564(3)	0.2843(2)	1.0020(2)	4.30(9)
C(24)	0.3749(3)	0.2528(2)	0.9203(2)	3.20(7)
C(25)	0.6558(3)	0.3783(2)	0.9107(2)	4.69(9)
C(26)	0.2940(3)	0.1857(2)	0.8679(2)	4.45(9)

$$B_{\text{eq}} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 2. Atomic coordinates and B_{iso} involving hydrogens/B_{eq}

atom	x	y	z	B _{eq}
H(1)	0.4835	0.3297	0.5042	3.21
H(2)	0.3631	0.2829	0.5009	3.20
H(3)	0.4633	0.2175	0.3618	3.52
H(4)	0.5486	0.0872	0.3413	4.13
H(5)	0.6527	0.0073	0.4893	4.24
H(6)	0.6522	0.0562	0.6442	3.87
H(7)	0.6267	0.5587	0.7012	3.82
H(8)	0.4399	0.6013	0.7425	4.36
H(9)	0.2705	0.5069	0.7309	4.36
H(10)	0.6614	0.3465	0.6160	3.53
H(11)	0.8080	0.4794	0.7269	5.38
H(12)	0.8618	0.4002	0.6965	5.38
H(13)	0.7887	0.3955	0.7697	5.37
H(14)	0.7527	0.4368	0.5309	6.05
H(15)	0.6108	0.4495	0.5002	6.06
H(16)	0.6964	0.5137	0.5643	6.05
H(17)	0.2665	0.2923	0.6506	4.39
H(18)	0.1285	0.4162	0.7155	6.54
H(19)	0.1039	0.3224	0.7102	6.55
H(20)	0.2205	0.3572	0.7828	6.55
H(21)	0.1922	0.3668	0.5064	5.56
H(22)	0.0798	0.3406	0.5393	5.56
H(23)	0.1247	0.4310	0.5507	5.55
H(24)	0.5928	0.4221	1.0641	5.81
H(25)	0.4186	0.3671	1.1073	6.51
H(26)	0.2749	0.2613	1.0254	5.18
H(27)	0.6890	0.3353	0.8823	5.56
H(28)	0.6274	0.4213	0.8666	5.58
H(29)	0.7179	0.3987	0.9635	5.57
H(30)	0.3415	0.1372	0.8704	5.62
H(31)	0.2298	0.1754	0.8956	5.62
H(32)	0.2592	0.2017	0.8043	5.61

$$B_{eq} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 3. Anisotropic displacement parameters

atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
V(1)	0.0379(3)	0.0249(2)	0.0293(2)	-0.0014(2)	0.0075(2)	-0.0009(2)
Cl(1)	0.0764(7)	0.0417(4)	0.0584(5)	-0.0265(4)	0.0356(5)	-0.0169(4)
Cl(2)	0.0454(5)	0.0499(5)	0.0499(5)	0.0140(4)	-0.0044(4)	-0.0086(4)
N(1)	0.0284(14)	0.0271(12)	0.0293(12)	0.0013(11)	0.0052(10)	-0.0021(10)
N(2)	0.0343(15)	0.0266(12)	0.0347(14)	-0.0004(11)	0.0098(12)	0.0002(11)
N(3)	0.0378(16)	0.0274(12)	0.0311(13)	-0.0020(11)	0.0061(12)	0.0009(11)
C(1)	0.040(2)	0.0331(16)	0.0287(15)	-0.0006(15)	0.0057(14)	-0.0027(13)
C(2)	0.0274(17)	0.0306(15)	0.0305(16)	-0.0012(14)	0.0086(13)	-0.0023(13)
C(3)	0.040(2)	0.0405(17)	0.0318(17)	-0.0023(15)	0.0098(15)	-0.0028(14)
C(4)	0.046(2)	0.0452(19)	0.0394(19)	-0.0032(17)	0.0186(16)	-0.0127(15)
C(5)	0.050(2)	0.0354(18)	0.049(2)	0.0014(16)	0.0192(18)	-0.0086(16)
C(6)	0.045(2)	0.0331(17)	0.0433(19)	0.0042(15)	0.0177(16)	0.0003(14)
C(7)	0.0355(19)	0.0267(14)	0.0228(15)	0.0053(14)	0.0045(13)	0.0007(12)
C(8)	0.0331(19)	0.0284(15)	0.0310(16)	0.0003(14)	0.0014(14)	0.0024(13)
C(9)	0.046(2)	0.0313(17)	0.0437(19)	-0.0012(16)	-0.0025(16)	-0.0023(14)
C(10)	0.059(2)	0.0319(17)	0.047(2)	0.0081(18)	0.0006(18)	-0.0111(15)
C(11)	0.053(2)	0.0422(19)	0.0422(19)	0.0123(18)	0.0137(17)	-0.0041(16)
C(12)	0.039(2)	0.0370(17)	0.0325(17)	0.0058(15)	0.0105(15)	0.0012(14)
C(13)	0.0330(19)	0.0395(18)	0.0389(18)	-0.0057(15)	0.0076(15)	-0.0008(14)
C(14)	0.040(2)	0.076(2)	0.054(2)	-0.0002(19)	0.0088(18)	0.006(2)
C(15)	0.060(2)	0.077(2)	0.055(2)	0.001(2)	0.023(2)	0.013(2)
C(16)	0.036(2)	0.0423(19)	0.060(2)	0.0041(16)	0.0221(17)	0.0033(16)
C(17)	0.053(2)	0.080(2)	0.074(2)	0.013(2)	0.032(2)	0.020(2)
C(18)	0.044(2)	0.069(2)	0.063(2)	-0.0143(19)	0.0164(19)	-0.0063(19)
C(19)	0.041(2)	0.0352(16)	0.0262(15)	0.0043(15)	0.0055(14)	0.0010(14)
C(20)	0.044(2)	0.0487(19)	0.0350(18)	-0.0021(17)	0.0069(16)	-0.0058(15)
C(21)	0.058(2)	0.082(2)	0.044(2)	-0.006(2)	0.009(2)	-0.027(2)
C(22)	0.055(2)	0.109(3)	0.042(2)	0.005(2)	0.015(2)	-0.025(2)
C(23)	0.045(2)	0.080(2)	0.0384(19)	0.004(2)	0.0129(17)	-0.0090(19)
C(24)	0.040(2)	0.0450(19)	0.0353(18)	0.0014(16)	0.0088(15)	0.0039(15)
C(25)	0.067(2)	0.061(2)	0.048(2)	-0.022(2)	0.012(2)	-0.0202(18)
C(26)	0.063(2)	0.058(2)	0.057(2)	-0.014(2)	0.031(2)	-0.0035(18)

The general temperature factor expression: $\exp(-2\pi^2(a^2U_{11}h^2 + b^2U_{22}k^2 + c^2U_{33}l^2 + 2a*b*U_{12}hk + 2a*c*U_{13}hl + 2b*c*U_{23}kl))$

Table 4. Bond lengths (Å)

atom	atom	distance	atom	atom	distance
V(1)	Cl(1)	2.2645(10)	V(1)	Cl(2)	2.2868(9)
V(1)	N(1)	1.850(2)	V(1)	N(2)	2.211(2)
V(1)	N(3)	1.679(2)	N(1)	C(1)	1.475(3)
N(1)	C(7)	1.450(3)	N(2)	C(2)	1.336(3)
N(2)	C(6)	1.358(4)	N(3)	C(19)	1.387(4)
C(1)	C(2)	1.488(4)	C(2)	C(3)	1.387(4)
C(3)	C(4)	1.389(4)	C(4)	C(5)	1.371(4)
C(5)	C(6)	1.362(4)	C(7)	C(8)	1.414(4)
C(7)	C(12)	1.397(4)	C(8)	C(9)	1.401(4)
C(8)	C(13)	1.519(4)	C(9)	C(10)	1.374(5)
C(10)	C(11)	1.382(5)	C(11)	C(12)	1.400(4)
C(12)	C(16)	1.517(4)	C(13)	C(14)	1.515(4)
C(13)	C(15)	1.525(5)	C(16)	C(17)	1.509(6)
C(16)	C(18)	1.543(4)	C(19)	C(20)	1.421(4)
C(19)	C(24)	1.409(5)	C(20)	C(21)	1.397(5)
C(20)	C(25)	1.483(5)	C(21)	C(22)	1.362(6)
C(22)	C(23)	1.389(5)	C(23)	C(24)	1.390(5)
C(24)	C(26)	1.494(4)			

Table 5. Bond lengths involving hydrogens (Å)

atom	atom	distance	atom	atom	distance
C(1)	H(1)	0.950	C(1)	H(2)	0.950
C(3)	H(3)	1.050	C(4)	H(4)	1.067
C(5)	H(5)	0.992	C(6)	H(6)	1.131
C(9)	H(7)	1.082	C(10)	H(8)	1.073
C(11)	H(9)	1.106	C(13)	H(10)	1.078
C(14)	H(11)	0.950	C(14)	H(12)	0.950
C(14)	H(13)	0.950	C(15)	H(14)	0.950
C(15)	H(15)	0.950	C(15)	H(16)	0.950
C(16)	H(17)	1.075	C(17)	H(18)	0.950
C(17)	H(19)	0.950	C(17)	H(20)	0.950
C(18)	H(21)	0.950	C(18)	H(22)	0.950
C(18)	H(23)	0.950	C(21)	H(24)	1.112
C(22)	H(25)	0.950	C(23)	H(26)	1.141
C(25)	H(27)	0.950	C(25)	H(28)	0.950
C(25)	H(29)	0.950	C(26)	H(30)	0.950
C(26)	H(31)	0.950	C(26)	H(32)	0.950

Table 6. Bond angles (°)

atom	atom	atom	angle	atom	atom	atom	angle
Cl(1)	V(1)	Cl(2)	121.87(3)	Cl(1)	V(1)	N(1)	114.55(7)
Cl(1)	V(1)	N(2)	82.45(7)	Cl(1)	V(1)	N(3)	100.56(9)
Cl(2)	V(1)	N(1)	116.98(8)	Cl(2)	V(1)	N(2)	83.87(6)
Cl(2)	V(1)	N(3)	96.74(8)	N(1)	V(1)	N(2)	77.66(9)
N(1)	V(1)	N(3)	98.52(11)	N(2)	V(1)	N(3)	175.93(10)
V(1)	N(1)	C(1)	121.10(17)	V(1)	N(1)	C(7)	126.44(17)
C(1)	N(1)	C(7)	112.4(2)	V(1)	N(2)	C(2)	112.7(2)
V(1)	N(2)	C(6)	128.2(2)	C(2)	N(2)	C(6)	118.5(2)
V(1)	N(3)	C(19)	176.2(2)	N(1)	C(1)	C(2)	110.3(2)
N(2)	C(2)	C(1)	114.5(2)	N(2)	C(2)	C(3)	122.0(2)
C(1)	C(2)	C(3)	123.5(2)	C(2)	C(3)	C(4)	118.5(2)
C(3)	C(4)	C(5)	119.3(3)	C(4)	C(5)	C(6)	119.5(3)
N(2)	C(6)	C(5)	122.1(2)	N(1)	C(7)	C(8)	117.5(2)
N(1)	C(7)	C(12)	120.1(2)	C(8)	C(7)	C(12)	122.4(2)
C(7)	C(8)	C(9)	117.1(3)	C(7)	C(8)	C(13)	123.1(2)
C(9)	C(8)	C(13)	119.7(2)	C(8)	C(9)	C(10)	121.1(3)
C(9)	C(10)	C(11)	120.7(3)	C(10)	C(11)	C(12)	120.9(3)
C(7)	C(12)	C(11)	117.6(3)	C(7)	C(12)	C(16)	122.1(2)
C(11)	C(12)	C(16)	120.2(3)	C(8)	C(13)	C(14)	113.3(2)
C(8)	C(13)	C(15)	110.8(2)	C(14)	C(13)	C(15)	108.9(2)
C(12)	C(16)	C(17)	113.4(2)	C(12)	C(16)	C(18)	110.4(2)
C(17)	C(16)	C(18)	108.5(2)	N(3)	C(19)	C(20)	119.6(3)
N(3)	C(19)	C(24)	119.7(2)	C(20)	C(19)	C(24)	120.7(3)
C(19)	C(20)	C(21)	117.7(3)	C(19)	C(20)	C(25)	121.0(3)
C(21)	C(20)	C(25)	121.3(3)	C(20)	C(21)	C(22)	121.6(3)
C(21)	C(22)	C(23)	120.8(3)	C(22)	C(23)	C(24)	120.4(3)
C(19)	C(24)	C(23)	118.8(2)	C(19)	C(24)	C(26)	121.1(3)
C(23)	C(24)	C(26)	120.1(3)				

Table 7. Bond angles involving hydrogens (°)

atom	atom	atom	angle	atom	atom	atom	angle
N(1)	C(1)	H(1)	109.3	N(1)	C(1)	H(2)	109.3
C(2)	C(1)	H(1)	109.3	C(2)	C(1)	H(2)	109.3
H(1)	C(1)	H(2)	109.5	C(2)	C(3)	H(3)	121.2
C(4)	C(3)	H(3)	120.2	C(3)	C(4)	H(4)	116.1
C(5)	C(4)	H(4)	124.6	C(4)	C(5)	H(5)	124.1
C(6)	C(5)	H(5)	116.2	N(2)	C(6)	H(6)	112.8
C(5)	C(6)	H(6)	124.6	C(8)	C(9)	H(7)	120.2
C(10)	C(9)	H(7)	118.8	C(9)	C(10)	H(8)	122.3
C(11)	C(10)	H(8)	117.0	C(10)	C(11)	H(9)	120.0
C(12)	C(11)	H(9)	118.0	C(8)	C(13)	H(10)	105.0
C(14)	C(13)	H(10)	112.0	C(15)	C(13)	H(10)	106.7
C(13)	C(14)	H(11)	109.5	C(13)	C(14)	H(12)	109.5
C(13)	C(14)	H(13)	109.5	H(11)	C(14)	H(12)	109.5
H(11)	C(14)	H(13)	109.5	H(12)	C(14)	H(13)	109.5
C(13)	C(15)	H(14)	109.5	C(13)	C(15)	H(15)	109.5
C(13)	C(15)	H(16)	109.5	H(14)	C(15)	H(15)	109.5
H(14)	C(15)	H(16)	109.5	H(15)	C(15)	H(16)	109.5
C(12)	C(16)	H(17)	107.6	C(17)	C(16)	H(17)	108.2
C(18)	C(16)	H(17)	108.5	C(16)	C(17)	H(18)	109.5
C(16)	C(17)	H(19)	109.5	C(16)	C(17)	H(20)	109.5
H(18)	C(17)	H(19)	109.5	H(18)	C(17)	H(20)	109.5
H(19)	C(17)	H(20)	109.5	C(16)	C(18)	H(21)	109.5
C(16)	C(18)	H(22)	109.5	C(16)	C(18)	H(23)	109.5
H(21)	C(18)	H(22)	109.5	H(21)	C(18)	H(23)	109.5
H(22)	C(18)	H(23)	109.5	C(20)	C(21)	H(24)	116.7
C(22)	C(21)	H(24)	121.7	C(21)	C(22)	H(25)	119.6
C(23)	C(22)	H(25)	119.6	C(22)	C(23)	H(26)	120.3
C(24)	C(23)	H(26)	119.2	C(20)	C(25)	H(27)	109.5
C(20)	C(25)	H(28)	109.5	C(20)	C(25)	H(29)	109.5
H(27)	C(25)	H(28)	109.5	H(27)	C(25)	H(29)	109.5
H(28)	C(25)	H(29)	109.5	C(24)	C(26)	H(30)	109.5
C(24)	C(26)	H(31)	109.5	C(24)	C(26)	H(32)	109.5
H(30)	C(26)	H(31)	109.5	H(30)	C(26)	H(32)	109.5
H(31)	C(26)	H(32)	109.5				

Table 8. Torsion Angles(°)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
Cl(1)	V(1)	N(1)	C(1)	-57.6(2)	Cl(1)	V(1)	N(1)	C(7)	124.6(2)
Cl(1)	V(1)	N(2)	C(2)	102.2(2)	Cl(1)	V(1)	N(2)	C(6)	-68.5(2)
Cl(1)	V(1)	N(3)	C(19)	70(3)	Cl(2)	V(1)	N(1)	C(1)	94.6(2)
Cl(2)	V(1)	N(1)	C(7)	-83.3(2)	Cl(2)	V(1)	N(2)	C(2)	-134.5(2)
Cl(2)	V(1)	N(2)	C(6)	54.8(2)	Cl(2)	V(1)	N(3)	C(19)	-54(3)
N(1)	V(1)	N(2)	C(2)	-15.1(2)	N(1)	V(1)	N(2)	C(6)	174.2(2)
N(2)	V(1)	N(1)	C(1)	18.1(2)	N(2)	V(1)	N(1)	C(7)	-159.8(2)
N(1)	V(1)	N(3)	C(19)	-173(3)	N(3)	V(1)	N(1)	C(1)	-163.3(2)
N(3)	V(1)	N(1)	C(7)	18.8(2)	N(2)	V(1)	N(3)	C(19)	-152(2)
N(3)	V(1)	N(2)	C(2)	-35.7(15)	N(3)	V(1)	N(2)	C(6)	153.7(14)
V(1)	N(1)	C(1)	C(2)	-18.5(3)	V(1)	N(1)	C(7)	C(8)	95.6(3)
V(1)	N(1)	C(7)	C(12)	-83.6(2)	C(1)	N(1)	C(7)	C(8)	-82.4(2)
C(1)	N(1)	C(7)	C(12)	98.3(3)	C(7)	N(1)	C(1)	C(2)	159.7(2)
V(1)	N(2)	C(2)	C(1)	9.3(3)	V(1)	N(2)	C(2)	C(3)	-169.4(2)
V(1)	N(2)	C(6)	C(5)	168.1(2)	C(2)	N(2)	C(6)	C(5)	-2.1(4)
C(6)	N(2)	C(2)	C(1)	-179.0(2)	C(6)	N(2)	C(2)	C(3)	2.2(4)
V(1)	N(3)	C(19)	C(20)	101(3)	V(1)	N(3)	C(19)	C(24)	-77(3)
N(1)	C(1)	C(2)	N(2)	3.5(3)	N(1)	C(1)	C(2)	C(3)	-177.8(2)
N(2)	C(2)	C(3)	C(4)	-0.4(4)	C(1)	C(2)	C(3)	C(4)	-179.0(3)
C(2)	C(3)	C(4)	C(5)	-1.7(5)	C(3)	C(4)	C(5)	C(6)	1.8(5)
C(4)	C(5)	C(6)	N(2)	0.1(4)	N(1)	C(7)	C(8)	C(9)	-175.1(2)
N(1)	C(7)	C(8)	C(13)	7.1(3)	N(1)	C(7)	C(12)	C(11)	174.6(2)
N(1)	C(7)	C(12)	C(16)	-9.9(3)	C(8)	C(7)	C(12)	C(11)	-4.6(4)
C(8)	C(7)	C(12)	C(16)	170.9(2)	C(12)	C(7)	C(8)	C(9)	4.1(3)
C(12)	C(7)	C(8)	C(13)	-173.7(2)	C(7)	C(8)	C(9)	C(10)	-0.4(4)
C(7)	C(8)	C(13)	C(14)	-125.0(3)	C(7)	C(8)	C(13)	C(15)	112.3(3)
C(9)	C(8)	C(13)	C(14)	57.2(3)	C(9)	C(8)	C(13)	C(15)	-65.5(3)
C(13)	C(8)	C(9)	C(10)	177.6(2)	C(8)	C(9)	C(10)	C(11)	-2.8(4)
C(9)	C(10)	C(11)	C(12)	2.3(4)	C(10)	C(11)	C(12)	C(7)	1.3(4)
C(10)	C(11)	C(12)	C(16)	-174.3(2)	C(7)	C(12)	C(16)	C(17)	142.8(2)
C(7)	C(12)	C(16)	C(18)	-95.2(3)	C(11)	C(12)	C(16)	C(17)	-41.8(3)
C(11)	C(12)	C(16)	C(18)	80.2(3)	N(3)	C(19)	C(20)	C(21)	-179.6(2)
N(3)	C(19)	C(20)	C(25)	0.0(3)	N(3)	C(19)	C(24)	C(23)	179.6(2)
N(3)	C(19)	C(24)	C(26)	0.6(4)	C(20)	C(19)	C(24)	C(23)	1.3(4)
C(20)	C(19)	C(24)	C(26)	-177.7(2)	C(24)	C(19)	C(20)	C(21)	-1.3(4)
C(24)	C(19)	C(20)	C(25)	178.4(2)	C(19)	C(20)	C(21)	C(22)	0.2(5)
C(25)	C(20)	C(21)	C(22)	-179.4(3)	C(20)	C(21)	C(22)	C(23)	0.8(5)

Table 8. Torsion angles ($^{\circ}$) (continued)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
C(21)	C(22)	C(23)	C(24)	-0.8(5)	C(22)	C(23)	C(24)	C(19)	-0.2(5)
C(22)	C(23)	C(24)	C(26)	178.7(3)					

The sign is positive if when looking from atom 2 to atom 3 a clock-wise motion of atom 1 would superimpose it on atom 4.

Table 9. Distances beyond the asymmetric unit out to 3.60 Å

atom	atom	distance	atom	atom	distance
C(4)	C(14) ¹⁾	3.546(4)	C(4)	C(17) ²⁾	3.436(6)
C(5)	C(5) ³⁾	3.400(5)	C(5)	C(6) ³⁾	3.564(4)
C(6)	C(5) ³⁾	3.564(4)	C(10)	C(26) ⁴⁾	3.546(5)
C(11)	C(26) ⁴⁾	3.595(4)	C(14)	C(4) ⁵⁾	3.546(4)
C(17)	C(4) ⁶⁾	3.436(6)	C(26)	C(10) ⁷⁾	3.546(5)
C(26)	C(11) ⁷⁾	3.595(4)			

Symmetry Operators:

- | | |
|---------------------------------|-------------------------------|
| (1) $X+1/2, -Y+1/2, Z+1/2-1$ | (2) $X+1/2, -Y+1/2, Z+1/2-1$ |
| (3) $-X+1, -Y, -Z+1$ | (4) $-X+1/2, Y+1/2, -Z+1/2+1$ |
| (5) $X+1/2, -Y+1/2, Z+1/2$ | (6) $X+1/2-1, -Y+1/2, Z+1/2$ |
| (7) $-X+1/2, Y+1/2-1, -Z+1/2+1$ | |

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens

atom	atom	distance	atom	atom	distance
Cl(1)	H(4) ¹⁾	3.254	Cl(1)	H(5) ¹⁾	3.116
Cl(1)	H(9) ²⁾	2.911	Cl(1)	H(18) ²⁾	3.414
Cl(1)	H(24) ³⁾	3.338	Cl(1)	H(29) ³⁾	3.328
Cl(2)	H(2) ⁴⁾	3.202	Cl(2)	H(3) ⁴⁾	2.966
Cl(2)	H(7) ⁵⁾	2.689	Cl(2)	H(11) ⁵⁾	3.490
N(2)	H(26) ⁶⁾	3.088	C(2)	H(26) ⁶⁾	3.027
C(2)	H(29) ³⁾	3.583	C(3)	H(12) ³⁾	3.515
C(3)	H(13) ³⁾	3.050	C(3)	H(26) ⁶⁾	3.171
C(3)	H(27) ³⁾	3.437	C(3)	H(31) ⁶⁾	3.583
C(4)	H(6) ¹⁾	3.489	C(4)	H(11) ³⁾	3.600
C(4)	H(12) ³⁾	3.322	C(4)	H(13) ³⁾	3.157
C(4)	H(18) ⁶⁾	3.268	C(4)	H(19) ⁶⁾	3.396
C(4)	H(20) ⁶⁾	3.084	C(4)	H(26) ⁶⁾	3.345
C(5)	H(5) ¹⁾	3.344	C(5)	H(25) ⁶⁾	3.555
C(5)	H(26) ⁶⁾	3.350	C(5)	H(29) ⁵⁾	3.221
C(6)	H(5) ¹⁾	3.337	C(6)	H(11) ⁵⁾	3.345
C(6)	H(25) ⁶⁾	3.382	C(6)	H(26) ⁶⁾	3.213
C(7)	H(15) ⁷⁾	3.532	C(8)	H(15) ⁷⁾	3.239
C(9)	H(6) ⁸⁾	3.579	C(9)	H(15) ⁷⁾	2.993
C(9)	H(25) ⁹⁾	3.458	C(10)	H(15) ⁷⁾	3.082
C(10)	H(24) ⁹⁾	3.499	C(10)	H(25) ⁹⁾	3.032
C(10)	H(30) ¹⁰⁾	3.515	C(10)	H(31) ¹⁰⁾	3.097
C(10)	H(32) ¹⁰⁾	3.464	C(11)	H(14) ⁷⁾	3.545
C(11)	H(15) ⁷⁾	3.346	C(11)	H(30) ¹⁰⁾	3.196
C(11)	H(31) ¹⁰⁾	3.370	C(12)	H(15) ⁷⁾	3.568
C(13)	H(26) ⁶⁾	3.584	C(14)	H(3) ⁴⁾	3.349
C(14)	H(4) ⁴⁾	2.950	C(14)	H(6) ⁸⁾	2.965
C(15)	H(15) ⁷⁾	3.557	C(15)	H(21) ⁷⁾	3.401
C(15)	H(23) ⁷⁾	3.479	C(15)	H(31) ⁶⁾	3.293
C(17)	H(4) ¹¹⁾	2.643	C(17)	H(12) ¹²⁾	3.409
C(18)	H(14) ⁷⁾	3.593	C(18)	H(16) ⁷⁾	3.326
C(19)	H(21) ⁴⁾	3.579	C(19)	H(22) ⁴⁾	3.007
C(20)	H(22) ⁴⁾	3.349	C(21)	H(8) ⁹⁾	3.425
C(21)	H(22) ⁴⁾	3.563	C(22)	H(8) ⁹⁾	3.101
C(22)	H(22) ⁴⁾	3.492	C(23)	H(22) ⁴⁾	3.168
C(24)	H(22) ⁴⁾	2.907	C(25)	H(2) ⁴⁾	3.520
C(25)	H(5) ⁸⁾	3.076	C(26)	H(8) ²⁾	3.013

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(26)	H(9) ²⁾	3.249	C(26)	H(14) ¹¹⁾	3.278
C(26)	H(22) ⁴⁾	3.518	H(1)	H(7) ⁷⁾	3.463
H(1)	H(16) ⁷⁾	3.240	H(1)	H(26) ⁶⁾	3.548
H(2)	Cl(2) ³⁾	3.202	H(2)	C(25) ³⁾	3.520
H(2)	H(16) ⁷⁾	3.464	H(2)	H(27) ³⁾	2.950
H(2)	H(29) ³⁾	3.354	H(3)	Cl(2) ³⁾	2.966
H(3)	C(14) ³⁾	3.349	H(3)	H(12) ³⁾	3.068
H(3)	H(13) ³⁾	2.758	H(3)	H(19) ⁶⁾	3.184
H(3)	H(27) ³⁾	3.323	H(3)	H(31) ⁶⁾	3.395
H(4)	Cl(1) ¹⁾	3.254	H(4)	C(14) ³⁾	2.950
H(4)	C(17) ⁶⁾	2.643	H(4)	H(6) ¹⁾	3.310
H(4)	H(11) ³⁾	2.969	H(4)	H(12) ³⁾	2.554
H(4)	H(13) ³⁾	2.839	H(4)	H(18) ⁶⁾	2.306
H(4)	H(19) ⁶⁾	2.662	H(4)	H(20) ⁶⁾	2.522
H(5)	Cl(1) ¹⁾	3.116	H(5)	C(5) ¹⁾	3.344
H(5)	C(6) ¹⁾	3.337	H(5)	C(25) ⁵⁾	3.076
H(5)	H(5) ¹⁾	3.575	H(5)	H(24) ⁵⁾	3.493
H(5)	H(27) ⁵⁾	3.567	H(5)	H(28) ⁵⁾	3.109
H(5)	H(29) ⁵⁾	2.274	H(6)	C(4) ¹⁾	3.489
H(6)	C(9) ⁵⁾	3.579	H(6)	C(14) ⁵⁾	2.965
H(6)	H(4) ¹⁾	3.310	H(6)	H(7) ⁵⁾	2.867
H(6)	H(11) ⁵⁾	2.223	H(6)	H(12) ⁵⁾	3.511
H(6)	H(13) ⁵⁾	2.906	H(6)	H(25) ⁶⁾	3.454
H(6)	H(28) ⁵⁾	3.365	H(6)	H(29) ⁵⁾	3.562
H(7)	Cl(2) ⁸⁾	2.689	H(7)	H(1) ⁷⁾	3.463
H(7)	H(6) ⁸⁾	2.867	H(7)	H(15) ⁷⁾	3.400
H(7)	H(25) ⁹⁾	3.271	H(8)	C(21) ⁹⁾	3.425
H(8)	C(22) ⁹⁾	3.101	H(8)	C(26) ¹⁰⁾	3.013
H(8)	H(15) ⁷⁾	3.587	H(8)	H(24) ⁹⁾	3.031
H(8)	H(25) ⁹⁾	2.398	H(8)	H(30) ¹⁰⁾	3.196
H(8)	H(31) ¹⁰⁾	2.662	H(8)	H(32) ¹⁰⁾	2.708
H(9)	Cl(1) ¹⁰⁾	2.911	H(9)	C(26) ¹⁰⁾	3.249
H(9)	H(24) ⁹⁾	3.210	H(9)	H(30) ¹⁰⁾	2.699
H(9)	H(31) ¹⁰⁾	3.329	H(9)	H(32) ¹⁰⁾	3.219
H(10)	H(26) ⁶⁾	2.751	H(11)	Cl(2) ⁸⁾	3.490
H(11)	C(4) ⁴⁾	3.600	H(11)	C(6) ⁸⁾	3.345
H(11)	H(4) ⁴⁾	2.969	H(11)	H(6) ⁸⁾	2.223

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(12)	C(3) ⁴⁾	3.515	H(12)	C(4) ⁴⁾	3.322
H(12)	C(17) ¹³⁾	3.409	H(12)	H(3) ⁴⁾	3.068
H(12)	H(4) ⁴⁾	2.554	H(12)	H(6) ⁸⁾	3.511
H(12)	H(18) ¹³⁾	2.963	H(12)	H(19) ¹³⁾	2.973
H(12)	H(26) ⁶⁾	3.594	H(13)	C(3) ⁴⁾	3.050
H(13)	C(4) ⁴⁾	3.157	H(13)	H(3) ⁴⁾	2.758
H(13)	H(4) ⁴⁾	2.839	H(13)	H(6) ⁸⁾	2.906
H(14)	C(11) ⁷⁾	3.545	H(14)	C(18) ⁷⁾	3.593
H(14)	C(26) ⁶⁾	3.278	H(14)	H(21) ⁷⁾	3.337
H(14)	H(23) ⁷⁾	3.005	H(14)	H(26) ⁶⁾	3.240
H(14)	H(30) ⁶⁾	3.094	H(14)	H(31) ⁶⁾	2.675
H(15)	C(7) ⁷⁾	3.532	H(15)	C(8) ⁷⁾	3.239
H(15)	C(9) ⁷⁾	2.993	H(15)	C(10) ⁷⁾	3.082
H(15)	C(11) ⁷⁾	3.346	H(15)	C(12) ⁷⁾	3.568
H(15)	C(15) ⁷⁾	3.557	H(15)	H(7) ⁷⁾	3.400
H(15)	H(8) ⁷⁾	3.587	H(15)	H(15) ⁷⁾	3.001
H(15)	H(16) ⁷⁾	3.385	H(15)	H(31) ⁶⁾	3.098
H(16)	C(18) ⁷⁾	3.326	H(16)	H(1) ⁷⁾	3.240
H(16)	H(2) ⁷⁾	3.464	H(16)	H(15) ⁷⁾	3.385
H(16)	H(21) ⁷⁾	2.696	H(16)	H(23) ⁷⁾	3.142
H(18)	Cl(1) ¹⁰⁾	3.414	H(18)	C(4) ¹¹⁾	3.268
H(18)	H(4) ¹¹⁾	2.306	H(18)	H(12) ¹²⁾	2.963
H(19)	C(4) ¹¹⁾	3.396	H(19)	H(3) ¹¹⁾	3.184
H(19)	H(4) ¹¹⁾	2.662	H(19)	H(12) ¹²⁾	2.973
H(20)	C(4) ¹¹⁾	3.084	H(20)	H(4) ¹¹⁾	2.522
H(21)	C(15) ⁷⁾	3.401	H(21)	C(19) ³⁾	3.579
H(21)	H(14) ⁷⁾	3.337	H(21)	H(16) ⁷⁾	2.696
H(22)	C(19) ³⁾	3.007	H(22)	C(20) ³⁾	3.349
H(22)	C(21) ³⁾	3.563	H(22)	C(22) ³⁾	3.492
H(22)	C(23) ³⁾	3.168	H(22)	C(24) ³⁾	2.907
H(22)	C(26) ³⁾	3.518	H(22)	H(30) ³⁾	3.130
H(23)	C(15) ⁷⁾	3.479	H(23)	H(14) ⁷⁾	3.005
H(23)	H(16) ⁷⁾	3.142	H(23)	H(23) ¹⁴⁾	3.583
H(23)	H(30) ¹⁰⁾	3.543	H(24)	Cl(1) ⁴⁾	3.338
H(24)	C(10) ⁹⁾	3.499	H(24)	H(5) ⁸⁾	3.493
H(24)	H(8) ⁹⁾	3.031	H(24)	H(9) ⁹⁾	3.210
H(24)	H(24) ⁹⁾	3.491	H(25)	C(5) ¹¹⁾	3.555

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(25)	C(6) ¹¹⁾	3.382	H(25)	C(9) ⁹⁾	3.458
H(25)	C(10) ⁹⁾	3.032	H(25)	H(6) ¹¹⁾	3.454
H(25)	H(7) ⁹⁾	3.271	H(25)	H(8) ⁹⁾	2.398
H(25)	H(28) ⁹⁾	3.525	H(26)	N(2) ¹¹⁾	3.088
H(26)	C(2) ¹¹⁾	3.027	H(26)	C(3) ¹¹⁾	3.171
H(26)	C(4) ¹¹⁾	3.345	H(26)	C(5) ¹¹⁾	3.350
H(26)	C(6) ¹¹⁾	3.213	H(26)	C(13) ¹¹⁾	3.584
H(26)	H(1) ¹¹⁾	3.548	H(26)	H(10) ¹¹⁾	2.751
H(26)	H(12) ¹¹⁾	3.594	H(26)	H(14) ¹¹⁾	3.240
H(27)	C(3) ⁴⁾	3.437	H(27)	H(2) ⁴⁾	2.950
H(27)	H(3) ⁴⁾	3.323	H(27)	H(5) ⁸⁾	3.567
H(28)	H(5) ⁸⁾	3.109	H(28)	H(6) ⁸⁾	3.365
H(28)	H(25) ⁹⁾	3.525	H(29)	Cl(1) ⁴⁾	3.328
H(29)	C(2) ⁴⁾	3.583	H(29)	C(5) ⁸⁾	3.221
H(29)	H(2) ⁴⁾	3.354	H(29)	H(5) ⁸⁾	2.274
H(29)	H(6) ⁸⁾	3.562	H(30)	C(10) ²⁾	3.515
H(30)	C(11) ²⁾	3.196	H(30)	H(8) ²⁾	3.196
H(30)	H(9) ²⁾	2.699	H(30)	H(14) ¹¹⁾	3.094
H(30)	H(22) ⁴⁾	3.130	H(30)	H(23) ²⁾	3.543
H(31)	C(3) ¹¹⁾	3.583	H(31)	C(10) ²⁾	3.097
H(31)	C(11) ²⁾	3.370	H(31)	C(15) ¹¹⁾	3.293
H(31)	H(3) ¹¹⁾	3.395	H(31)	H(8) ²⁾	2.662
H(31)	H(9) ²⁾	3.329	H(31)	H(14) ¹¹⁾	2.675
H(31)	H(15) ¹¹⁾	3.098	H(32)	C(10) ²⁾	3.464
H(32)	H(8) ²⁾	2.708	H(32)	H(9) ²⁾	3.219

Symmetry Operators:

- | | |
|-------------------------------|-----------------------------|
| (1) -X+1,-Y,-Z+1 | (2) -X+1/2,Y+1/2-1,-Z+1/2+1 |
| (3) X+1/2-1,-Y+1/2,Z+1/2-1 | (4) X+1/2,-Y+1/2,Z+1/2 |
| (5) -X+1/2+1,Y+1/2-1,-Z+1/2+1 | (6) X+1/2,-Y+1/2,Z+1/2-1 |
| (7) -X+1,-Y+1,-Z+1 | (8) -X+1/2+1,Y+1/2,-Z+1/2+1 |
| (9) -X+1,-Y+1,-Z+2 | (10) -X+1/2,Y+1/2,-Z+1/2+1 |
| (11) X+1/2-1,-Y+1/2,Z+1/2 | (12) X-1,Y,Z |
| (13) X+1,Y,Z | (14) -X,-Y+1,-Z+1 |

X-ray Structure Report for $\text{V}(\text{N}=2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Cl}_2[2\text{-(2,6-F}_2\text{C}_6\text{H}_3)\text{CH}_2(\text{C}_5\text{H}_4\text{N})]$ (**1c**)

June 24, 2009

Experimental

Data Collection

A red block crystal of $C_{23}H_{18}Cl_2F_2N_3V$ having approximate dimensions of 0.40 x 0.30 x 0.20 mm was mounted on a glass fiber. All measurements were made on a Rigaku RAXIS RAPID imaging plate area detector with graphite monochromated Mo-K α radiation.

Indexing was performed from 3 oscillations that were exposed for 240 seconds. The crystal-to-detector distance was 127.40 mm.

Cell constants and an orientation matrix for data collection corresponded to a primitive triclinic cell with dimensions:

$$\begin{array}{ll} a = 7.3016(4) \text{ \AA} & \alpha = 99.261(3)^\circ \\ b = 10.2681(6) \text{ \AA} & \beta = 92.447(3)^\circ \\ c = 15.9162(12) \text{ \AA} & \gamma = 97.1083(15)^\circ \\ V = 1166.29(12) \text{ \AA}^3 & \end{array}$$

For $Z = 2$ and F.W. = 496.26, the calculated density is 1.413 g/cm³. Based on a statistical analysis of intensity distribution, and the successful solution and refinement of the structure, the space group was determined to be:

P-1 (#2)

The data were collected at a temperature of $-80 \pm 1^\circ\text{C}$ to a maximum 2θ value of 50.7° . A total of 55 oscillation images were collected. A sweep of data was done using ω scans from 130.0 to 190.0° in 4.0° step, at $\chi=45.0^\circ$ and $\phi = 30.0^\circ$. The exposure rate was 140.0 [sec./ $^\circ$]. A second sweep was performed using ω scans from 0.0 to 160.0° in 4.0° step, at $\chi=45.0^\circ$ and $\phi = 180.0^\circ$. The exposure rate was 140.0 [sec./ $^\circ$]. The crystal-to-detector distance was 127.40 mm. Readout was performed in the 0.100 mm pixel mode.

Data Reduction

Of the 9482 reflections that were collected, 4207 were unique ($R_{\text{int}} = 0.022$); equivalent reflections were merged.

The linear absorption coefficient, μ , for Mo-K α radiation is 6.843 cm⁻¹. An empirical absorption correction was applied which resulted in transmission factors ranging from 0.432 to 0.872. The data were corrected for Lorentz and polarization effects.

Structure Solution and Refinement

The structure was solved by direct methods¹ and expanded using Fourier techniques². The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were refined using the riding model. The final cycle of full-matrix least-squares refinement³ on F^2 was based on 4207 observed reflections and 281 variable parameters and converged (largest parameter shift was 0.00 times its esd) with unweighted and weighted agreement factors of:

$$R1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o| = 0.0404$$

$$wR2 = [\Sigma (w (F_o^2 - F_c^2)^2) / \Sigma w(F_o^2)^2]^{1/2} = 0.1316$$

The standard deviation of an observation of unit weight⁴ was 1.15. Unit weights were used. The maximum and minimum peaks on the final difference Fourier map corresponded to 0.75 and -0.42 e-/Å³, respectively.

Neutral atom scattering factors were taken from Cromer and Waber⁵. Anomalous dispersion effects were included in F_{calc} ⁶; the values for $\Delta f'$ and $\Delta f''$ were those of Creagh and McAuley⁷. The values for the mass attenuation coefficients are those of Creagh and Hubbell⁸. All calculations were performed using the CrystalStructure⁹ crystallographic software package except for refinement, which was performed using SHELXL-97¹⁰.

References

(1) SIR92: Altomare, A., Casciarano, G., Giacovazzo, C., Guagliardi, A., Burla, M., Polidori, G., and Camalli, M. (1994) J. Appl. Cryst., 27, 435.

(2) DIRDIF99: Beurskens, P.T., Admiraal, G., Beurskens, G., Bosman, W.P., de Gelder, R., Israel, R. and Smits, J.M.M.(1999). The DIRDIF-99 program system, Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands.

(3) Least Squares function minimized: (SHELXL97)

$$\sum w(F_o^2 - F_c^2)^2 \quad \text{where } w = \text{Least Squares weights.}$$

(4) Standard deviation of an observation of unit weight:

$$[\sum w(F_o^2 - F_c^2)^2 / (N_o - N_v)]^{1/2}$$

where: N_o = number of observations
 N_v = number of variables

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(10) SHELX97: Sheldrick, G.M. (1997).

EXPERIMENTAL DETAILS

A. Crystal Data

Empirical Formula	C ₂₃ H ₁₈ Cl ₂ F ₂ N ₃ V
Formula Weight	496.26
Crystal Color, Habit	red, block
Crystal Dimensions	0.40 X 0.30 X 0.20 mm
Crystal System	triclinic
Lattice Type	Primitive
Indexing Images	3 oscillations @ 240.0 seconds
Detector Position	127.40 mm
Pixel Size	0.100 mm
Lattice Parameters	a = 7.3016(4) Å b = 10.2681(6) Å c = 15.9162(12) Å α = 99.261(3) ° β = 92.447(3) ° γ = 97.1083(15) ° V = 1166.29(12) Å ³
Space Group	P-1 (#2)
Z value	2
D _{calc}	1.413 g/cm ³
F ₀₀₀	504.00
μ (MoK α)	6.843 cm ⁻¹

B. Intensity Measurements

Diffractometer	Rigaku RAXIS-RAPID
Radiation	MoK α (λ = 0.71075 Å) graphite monochromated
Detector Aperture	280 mm x 256 mm
Data Images	55 exposures
ω oscillation Range (χ =45.0, ϕ =30.0)	130.0 - 190.0°
Exposure Rate	140.0 sec./°
ω oscillation Range (χ =45.0, ϕ =180.0)	0.0 - 160.0°
Exposure Rate	140.0 sec./°
Detector Position	127.40 mm
Pixel Size	0.100 mm
$2\theta_{\max}$	50.7°
No. of Reflections Measured	Total: 9482 Unique: 4207 (R_{int} = 0.022)
Corrections	Lorentz-polarization Absorption (trans. factors: 0.432 - 0.872)

C. Structure Solution and Refinement

Structure Solution	Direct Methods (SIR92)
Refinement	Full-matrix least-squares on F^2
Function Minimized	$\sum w (F_o^2 - F_c^2)^2$
Least Squares Weights	$w = 1 / [\sigma^2(F_o^2) + (0.0619 \cdot P)^2 + 1.1383 \cdot P]$ where $P = (\text{Max}(F_o^2, 0) + 2F_c^2)/3$
$2\theta_{\text{max}}$ cutoff	50.7°
Anomalous Dispersion	All non-hydrogen atoms
No. Observations (All reflections)	4207
No. Variables	281
Reflection/Parameter Ratio	14.97
Residuals: R1 ($I > 2.00\sigma(I)$)	0.0404
Residuals: R (All reflections)	0.0467
Residuals: wR2 (All reflections)	0.1316
Goodness of Fit Indicator	1.154
Max Shift/Error in Final Cycle	0.001
Maximum peak in Final Diff. Map	0.75 e ⁻ /Å ³
Minimum peak in Final Diff. Map	-0.42 e ⁻ /Å ³

Table 1. Atomic coordinates and $B_{\text{iso}}/B_{\text{eq}}$

atom	x	y	z	B_{eq}
V(1)	0.12474(6)	0.27661(4)	0.80762(3)	2.266(12)
Cl(1)	0.00338(10)	0.29282(8)	0.93700(5)	3.345(16)
Cl(2)	0.00762(12)	0.40235(7)	0.72110(5)	3.639(17)
F(1)	0.4885(3)	0.0546(2)	0.89366(12)	4.62(4)
F(2)	0.3193(3)	0.1292(2)	0.61993(12)	4.34(4)
N(1)	0.3618(3)	0.2269(2)	0.79371(15)	2.56(4)
N(2)	0.3088(3)	0.4595(2)	0.86434(15)	2.68(4)
N(3)	0.0156(3)	0.1254(2)	0.76558(14)	2.40(3)
C(1)	0.3983(3)	0.0971(2)	0.75840(18)	2.64(4)
C(2)	0.4585(4)	0.0098(3)	0.8083(2)	3.42(5)
C(3)	0.4849(4)	-0.1176(3)	0.7763(2)	4.11(6)
C(4)	0.4503(4)	-0.1626(3)	0.6898(2)	4.22(6)
C(5)	0.3960(4)	-0.0807(3)	0.6372(2)	4.07(6)
C(6)	0.3720(4)	0.0475(3)	0.6718(2)	3.14(5)
C(7)	0.5335(3)	0.3184(3)	0.8197(2)	3.12(5)
C(8)	0.4900(3)	0.4483(3)	0.86354(18)	2.72(4)
C(9)	0.2567(4)	0.5735(2)	0.90314(19)	2.93(5)
C(10)	0.3817(4)	0.6801(3)	0.94179(19)	3.15(5)
C(11)	0.5682(4)	0.6682(3)	0.9418(2)	3.27(5)
C(12)	0.6245(4)	0.5517(3)	0.9020(2)	3.22(5)
C(13)	-0.0416(3)	-0.0060(2)	0.73089(19)	2.66(4)
C(14)	-0.0319(4)	-0.1050(3)	0.7814(2)	3.31(5)
C(15)	-0.0680(5)	-0.2368(3)	0.7420(2)	4.23(6)
C(16)	-0.1177(5)	-0.2694(3)	0.6563(2)	4.79(8)
C(17)	-0.1384(4)	-0.1710(3)	0.6076(2)	4.03(6)
C(18)	-0.1032(4)	-0.0375(3)	0.6434(2)	3.23(5)
C(19)	0.0138(5)	-0.0698(3)	0.8762(2)	4.12(6)
C(20)	-0.1336(5)	0.0706(3)	0.5933(2)	3.88(6)
C(21)	-0.4369(16)	-0.4494(7)	0.4422(5)	11.5(3)
C(22)	-0.319(2)	-0.5362(9)	0.4753(7)	17.6(5)
C(23)	-0.364(2)	-0.6115(8)	0.5454(9)	27.0(11)

$$B_{\text{eq}} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 2. Atomic coordinates and B_{iso} involving hydrogens/B_{eq}

atom	x	y	z	B _{eq}
H(1)	0.5262	-0.1742	0.8128	4.93
H(2)	0.4647	-0.2518	0.6669	5.06
H(3)	0.3751	-0.1112	0.5776	4.89
H(4)	0.6024	0.3314	0.7688	3.75
H(5)	0.6127	0.2796	0.8584	3.75
H(6)	0.1285	0.5811	0.9041	3.52
H(7)	0.3406	0.7602	0.9679	3.78
H(8)	0.6569	0.7397	0.9691	3.93
H(9)	0.7521	0.5421	0.9007	3.86
H(10)	-0.0583	-0.3056	0.7747	5.07
H(11)	-0.1381	-0.3603	0.6300	5.75
H(12)	-0.1772	-0.1955	0.5489	4.83
H(13)	-0.0772	-0.0159	0.9024	4.95
H(14)	0.1376	-0.0190	0.8871	4.95
H(15)	0.0105	-0.1515	0.9009	4.95
H(16)	-0.1686	0.0312	0.5336	4.66
H(17)	-0.0193	0.1327	0.5963	4.66
H(18)	-0.2326	0.1185	0.6171	4.66

$$B_{eq} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 3. Anisotropic displacement parameters

atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
V(1)	0.0252(2) 0.00248(19)	0.0323(2)	0.0280(2)	0.00555(19)	-0.00032(18)	
Cl(1)	0.0360(3)	0.0539(4)	0.0335(3)	0.0008(3)	0.0070(2)	-0.0010(3)
Cl(2)	0.0518(4)	0.0378(4)	0.0485(4)	0.0081(3)	-0.0140(3)	0.0094(3)
F(1)	0.0683(13)	0.0688(13)	0.0427(11)	0.0195(11)	-0.0041(9)	0.0164(9)
F(2)	0.0706(13)	0.0610(12)	0.0366(10)	0.0233(10)	-0.0019(9)	0.0079(8)
N(1)	0.0274(11)	0.0370(12)	0.0324(12)	0.0076(9)	0.0014(9)	0.0023(9)
N(2)	0.0305(11)	0.0407(13)	0.0290(11)	0.0031(10)	0.0009(9)	0.0023(9)
N(3)	0.0286(11)	0.0361(12)	0.0274(11)	0.0064(9)	0.0009(8)	0.0060(9)
C(1)	0.0259(13)	0.0387(14)	0.0368(15)	0.0090(11)	0.0034(11)	0.0060(12)
C(2)	0.0377(16)	0.0518(18)	0.0424(17)	0.0109(14)	0.0022(13)	0.0096(14)
C(3)	0.0434(18)	0.0467(18)	0.072(2)	0.0151(15)	0.0065(16)	0.0206(17)
C(4)	0.0484(19)	0.0396(17)	0.072(2)	0.0138(15)	0.0080(17)	0.0013(16)
C(5)	0.0497(19)	0.0515(19)	0.0499(19)	0.0137(16)	0.0022(15)	-0.0072(15)
C(6)	0.0350(15)	0.0457(16)	0.0392(16)	0.0119(13)	0.0029(12)	0.0042(13)
C(7)	0.0258(13)	0.0467(17)	0.0433(16)	0.0044(12)	0.0022(11)	-0.0007(13)
C(8)	0.0307(14)	0.0427(15)	0.0302(14)	0.0043(12)	0.0012(11)	0.0080(11)
C(9)	0.0345(14)	0.0387(15)	0.0380(15)	0.0088(12)	0.0021(12)	0.0029(12)
C(10)	0.0442(16)	0.0375(15)	0.0359(15)	0.0027(13)	0.0008(12)	0.0020(12)
C(11)	0.0405(16)	0.0394(16)	0.0405(16)	-0.0060(13)	-0.0045(12)	0.0052(12)
C(12)	0.0287(14)	0.0452(16)	0.0468(17)	-0.0006(12)	-0.0009(12)	0.0080(13)
C(13)	0.0277(13)	0.0337(14)	0.0392(15)	0.0048(11)	0.0023(11)	0.0033(11)
C(14)	0.0343(15)	0.0402(16)	0.0535(19)	0.0080(12)	0.0060(13)	0.0108(13)
C(15)	0.0509(19)	0.0349(16)	0.077(2)	0.0106(15)	0.0085(17)	0.0112(16)
C(16)	0.058(2)	0.0330(17)	0.085(2)	0.0033(15)	0.0077(19)	-0.0067(17)
C(17)	0.0463(18)	0.0448(18)	0.055(2)	0.0042(15)	-0.0014(15)	-0.0103(15)
C(18)	0.0369(15)	0.0436(16)	0.0385(16)	0.0026(13)	0.0019(12)	-0.0021(13)
C(19)	0.058(2)	0.0514(19)	0.052(2)	0.0088(16)	0.0049(16)	0.0227(16)
C(20)	0.0539(19)	0.055(2)	0.0346(16)	0.0046(16)	-0.0069(14)	0.0013(14)
C(21)	0.241(10)	0.062(3)	0.111(5)	-0.027(4)	-0.078(6)	0.002(3)
C(22)	0.363(17)	0.094(6)	0.154(9)	-0.081(8)	-0.067(10)	-0.046(5)
C(23)	0.54(2)	0.073(5)	0.342(19)	-0.015(10)	-0.35(2)	-0.030(8)

The general temperature factor expression: $\exp(-2\pi^2(a^2U_{11}h^2 + b^2U_{22}k^2 + c^2U_{33}l^2 + 2a*b*U_{12}hk + 2a*c*U_{13}hl + 2b*c*U_{23}kl))$

Table 4. Bond lengths (Å)

atom	atom	distance	atom	atom	distance
V(1)	Cl(1)	2.2661(9)	V(1)	Cl(2)	2.2449(9)
V(1)	N(1)	1.877(2)	V(1)	N(2)	2.208(2)
V(1)	N(3)	1.676(2)	F(1)	C(2)	1.362(3)
F(2)	C(6)	1.346(4)	N(1)	C(1)	1.422(3)
N(1)	C(7)	1.471(3)	N(2)	C(8)	1.343(3)
N(2)	C(9)	1.339(3)	N(3)	C(13)	1.378(3)
C(1)	C(2)	1.387(4)	C(1)	C(6)	1.386(4)
C(2)	C(3)	1.366(4)	C(3)	C(4)	1.382(5)
C(4)	C(5)	1.359(5)	C(5)	C(6)	1.378(4)
C(7)	C(8)	1.481(4)	C(8)	C(12)	1.395(3)
C(9)	C(10)	1.378(3)	C(10)	C(11)	1.382(4)
C(11)	C(12)	1.379(4)	C(13)	C(14)	1.401(4)
C(13)	C(18)	1.418(4)	C(14)	C(15)	1.385(4)
C(14)	C(19)	1.507(4)	C(15)	C(16)	1.374(6)
C(16)	C(17)	1.386(5)	C(17)	C(18)	1.385(4)
C(18)	C(20)	1.500(5)	C(21)	C(22)	1.456(16)
C(21)	C(23) ⁽¹⁾	1.66(2)	C(22)	C(23)	1.484(18)

Symmetry Operators:

(1) -X-1,-Y-1,-Z+1

Table 5. Bond lengths involving hydrogens (Å)

atom	atom	distance	atom	atom	distance
C(3)	H(1)	0.950	C(4)	H(2)	0.950
C(5)	H(3)	0.950	C(7)	H(4)	0.990
C(7)	H(5)	0.990	C(9)	H(6)	0.950
C(10)	H(7)	0.950	C(11)	H(8)	0.950
C(12)	H(9)	0.950	C(15)	H(10)	0.950
C(16)	H(11)	0.950	C(17)	H(12)	0.950
C(19)	H(13)	0.980	C(19)	H(14)	0.980
C(19)	H(15)	0.980	C(20)	H(16)	0.980
C(20)	H(17)	0.980	C(20)	H(18)	0.980

Table 6. Bond angles (°)

atom	atom	atom	angle	atom	atom	atom	angle
Cl(1)	V(1)	Cl(2)	114.03(3)	Cl(1)	V(1)	N(1)	120.40(8)
Cl(1)	V(1)	N(2)	85.09(6)	Cl(1)	V(1)	N(3)	97.67(8)
Cl(2)	V(1)	N(1)	120.51(8)	Cl(2)	V(1)	N(2)	85.86(7)
Cl(2)	V(1)	N(3)	100.97(8)	N(1)	V(1)	N(2)	76.93(9)
N(1)	V(1)	N(3)	94.10(10)	N(2)	V(1)	N(3)	170.71(10)
V(1)	N(1)	C(1)	124.67(16)	V(1)	N(1)	C(7)	123.54(18)
C(1)	N(1)	C(7)	111.8(2)	V(1)	N(2)	C(8)	114.57(18)
V(1)	N(2)	C(9)	126.56(19)	C(8)	N(2)	C(9)	118.7(2)
V(1)	N(3)	C(13)	169.3(2)	N(1)	C(1)	C(2)	122.4(2)
N(1)	C(1)	C(6)	122.4(2)	C(2)	C(1)	C(6)	115.1(2)
F(1)	C(2)	C(1)	117.3(2)	F(1)	C(2)	C(3)	119.3(3)
C(1)	C(2)	C(3)	123.4(3)	C(2)	C(3)	C(4)	118.7(3)
C(3)	C(4)	C(5)	120.7(3)	C(4)	C(5)	C(6)	118.8(3)
F(2)	C(6)	C(1)	117.8(2)	F(2)	C(6)	C(5)	119.0(2)
C(1)	C(6)	C(5)	123.2(3)	N(1)	C(7)	C(8)	110.1(2)
N(2)	C(8)	C(7)	114.7(2)	N(2)	C(8)	C(12)	121.9(2)
C(7)	C(8)	C(12)	123.5(2)	N(2)	C(9)	C(10)	122.6(2)
C(9)	C(10)	C(11)	118.7(2)	C(10)	C(11)	C(12)	119.5(2)
C(8)	C(12)	C(11)	118.6(2)	N(3)	C(13)	C(14)	119.7(2)
N(3)	C(13)	C(18)	118.6(2)	C(14)	C(13)	C(18)	121.7(2)
C(13)	C(14)	C(15)	118.1(3)	C(13)	C(14)	C(19)	121.2(2)
C(15)	C(14)	C(19)	120.7(3)	C(14)	C(15)	C(16)	120.9(3)
C(15)	C(16)	C(17)	120.7(3)	C(16)	C(17)	C(18)	121.0(3)
C(13)	C(18)	C(17)	117.4(3)	C(13)	C(18)	C(20)	120.6(2)
C(17)	C(18)	C(20)	122.0(2)	C(22)	C(21)	C(23) ¹⁾	143.2(9)
C(21)	C(22)	C(23)	124.6(12)	C(21) ¹⁾	C(23)	C(22)	92.2(10)

Symmetry Operators:

(1) -X-1,-Y-1,-Z+1

Table 7. Bond angles involving hydrogens (°)

atom	atom	atom	angle	atom	atom	atom	angle
C(2)	C(3)	H(1)	120.7	C(4)	C(3)	H(1)	120.7
C(3)	C(4)	H(2)	119.6	C(5)	C(4)	H(2)	119.6
C(4)	C(5)	H(3)	120.6	C(6)	C(5)	H(3)	120.6
N(1)	C(7)	H(4)	109.6	N(1)	C(7)	H(5)	109.6
C(8)	C(7)	H(4)	109.6	C(8)	C(7)	H(5)	109.6
H(4)	C(7)	H(5)	108.2	N(2)	C(9)	H(6)	118.7
C(10)	C(9)	H(6)	118.7	C(9)	C(10)	H(7)	120.6
C(11)	C(10)	H(7)	120.6	C(10)	C(11)	H(8)	120.3
C(12)	C(11)	H(8)	120.3	C(8)	C(12)	H(9)	120.7
C(11)	C(12)	H(9)	120.7	C(14)	C(15)	H(10)	119.5
C(16)	C(15)	H(10)	119.5	C(15)	C(16)	H(11)	119.7
C(17)	C(16)	H(11)	119.7	C(16)	C(17)	H(12)	119.5
C(18)	C(17)	H(12)	119.5	C(14)	C(19)	H(13)	109.5
C(14)	C(19)	H(14)	109.5	C(14)	C(19)	H(15)	109.5
H(13)	C(19)	H(14)	109.5	H(13)	C(19)	H(15)	109.5
H(14)	C(19)	H(15)	109.5	C(18)	C(20)	H(16)	109.5
C(18)	C(20)	H(17)	109.5	C(18)	C(20)	H(18)	109.5
H(16)	C(20)	H(17)	109.5	H(16)	C(20)	H(18)	109.5
H(17)	C(20)	H(18)	109.5				

Table 8. Torsion Angles(°)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
Cl(1)	V(1)	N(1)	C(1)	100.8(2)	Cl(1)	V(1)	N(1)	C(7)	-78.2(2)
Cl(1)	V(1)	N(2)	C(8)	121.6(2)	Cl(1)	V(1)	N(2)	C(9)	-53.7(2)
Cl(1)	V(1)	N(3)	C(13)	-115.1(11)	Cl(2)	V(1)	N(1)	C(1)	-105.9(2)
Cl(2)	V(1)	N(1)	C(7)	75.1(2)	Cl(2)	V(1)	N(2)	C(8)	-123.8(2)
Cl(2)	V(1)	N(2)	C(9)	60.9(2)	Cl(2)	V(1)	N(3)	C(13)	128.5(11)
N(1)	V(1)	N(2)	C(8)	-1.2(2)	N(1)	V(1)	N(2)	C(9)	-176.5(2)
N(2)	V(1)	N(1)	C(1)	177.0(2)	N(2)	V(1)	N(1)	C(7)	-2.0(2)
N(1)	V(1)	N(3)	C(13)	6.3(11)	N(3)	V(1)	N(1)	C(1)	-0.6(2)
N(3)	V(1)	N(1)	C(7)	-179.6(2)	N(2)	V(1)	N(3)	C(13)	-8.4(16)
N(3)	V(1)	N(2)	C(8)	13.9(7)	N(3)	V(1)	N(2)	C(9)	-161.4(5)
V(1)	N(1)	C(1)	C(2)	-104.3(2)	V(1)	N(1)	C(1)	C(6)	74.1(3)
V(1)	N(1)	C(7)	C(8)	4.5(3)	C(1)	N(1)	C(7)	C(8)	-174.6(2)
C(7)	N(1)	C(1)	C(2)	74.7(3)	C(7)	N(1)	C(1)	C(6)	-106.8(3)
V(1)	N(2)	C(8)	C(7)	3.9(3)	V(1)	N(2)	C(8)	C(12)	-175.6(2)
V(1)	N(2)	C(9)	C(10)	175.6(2)	C(8)	N(2)	C(9)	C(10)	0.5(4)
C(9)	N(2)	C(8)	C(7)	179.6(2)	C(9)	N(2)	C(8)	C(12)	0.1(3)
V(1)	N(3)	C(13)	C(14)	70.7(12)	V(1)	N(3)	C(13)	C(18)	-107.2(11)
N(1)	C(1)	C(2)	F(1)	-2.2(4)	N(1)	C(1)	C(2)	C(3)	176.4(2)
N(1)	C(1)	C(6)	F(2)	3.6(4)	N(1)	C(1)	C(6)	C(5)	-176.0(2)
C(2)	C(1)	C(6)	F(2)	-177.9(2)	C(2)	C(1)	C(6)	C(5)	2.5(4)
C(6)	C(1)	C(2)	F(1)	179.2(2)	C(6)	C(1)	C(2)	C(3)	-2.1(4)
F(1)	C(2)	C(3)	C(4)	178.7(2)	C(1)	C(2)	C(3)	C(4)	0.1(4)
C(2)	C(3)	C(4)	C(5)	1.8(5)	C(3)	C(4)	C(5)	C(6)	-1.4(5)
C(4)	C(5)	C(6)	F(2)	179.6(3)	C(4)	C(5)	C(6)	C(1)	-0.8(5)
N(1)	C(7)	C(8)	N(2)	-5.2(3)	N(1)	C(7)	C(8)	C(12)	174.3(2)
N(2)	C(8)	C(12)	C(11)	0.1(3)	C(7)	C(8)	C(12)	C(11)	-179.4(3)
N(2)	C(9)	C(10)	C(11)	-1.1(4)	C(9)	C(10)	C(11)	C(12)	1.3(4)
C(10)	C(11)	C(12)	C(8)	-0.8(5)	N(3)	C(13)	C(14)	C(15)	-172.0(2)
N(3)	C(13)	C(14)	C(19)	8.9(4)	N(3)	C(13)	C(18)	C(17)	172.3(2)
N(3)	C(13)	C(18)	C(20)	-9.4(4)	C(14)	C(13)	C(18)	C(17)	-5.5(4)
C(14)	C(13)	C(18)	C(20)	172.8(2)	C(18)	C(13)	C(14)	C(15)	5.9(4)
C(18)	C(13)	C(14)	C(19)	-173.2(3)	C(13)	C(14)	C(15)	C(16)	-2.0(5)
C(19)	C(14)	C(15)	C(16)	177.1(3)	C(14)	C(15)	C(16)	C(17)	-2.0(5)
C(15)	C(16)	C(17)	C(18)	2.3(5)	C(16)	C(17)	C(18)	C(13)	1.4(5)
C(16)	C(17)	C(18)	C(20)	-176.9(3)	C(22)	C(21)	C(23) ¹⁾	C(22) ¹⁾	-1.3(15)
C(23) ¹⁾	C(21)	C(22)	C(23)	1.6(18)	C(21)	C(22)	C(23)	C(21) ¹⁾	-1.0(11)

Symmetry Operators:

(1) $-X-1,-Y-1,-Z+1$

The sign is positive if when looking from atom 2 to atom 3 a clock-wise motion of atom 1 would superimpose it on atom 4.

Table 9. Distances beyond the asymmetric unit out to 3.60 Å

atom	atom	distance	atom	atom	distance
V(1)	F(2)	3.560(2)	V(1)	C(1)	2.929(2)
V(1)	C(7)	2.956(2)	V(1)	C(8)	3.024(2)
V(1)	C(9)	3.192(2)	V(1)	C(13)	3.041(2)
Cl(1)	N(2)	3.025(2)	Cl(1)	N(3)	2.993(2)
Cl(1)	C(9)	3.359(3)	Cl(1)	C(9) ¹⁾	3.449(3)
Cl(1)	C(10) ¹⁾	3.491(3)	Cl(1)	C(11) ²⁾	3.549(3)
Cl(2)	N(1)	3.583(2)	Cl(2)	N(2)	3.033(2)
Cl(2)	N(3)	3.046(2)	Cl(2)	C(9)	3.451(2)
F(1)	N(1)	2.775(3)	F(1)	C(1)	2.347(3)
F(1)	C(3)	2.353(4)	F(1)	C(6)	3.583(3)
F(1)	C(7)	3.110(4)	F(1)	C(10) ²⁾	3.469(3)
F(1)	C(19)	3.526(4)	F(2)	V(1)	3.560(2)
F(2)	N(1)	2.777(2)	F(2)	N(3)	3.277(3)
F(2)	C(1)	2.339(3)	F(2)	C(2)	3.571(4)
F(2)	C(4)	3.581(4)	F(2)	C(5)	2.347(4)
F(2)	C(13)	3.503(3)	F(2)	C(18)	3.404(3)
F(2)	C(20)	3.284(4)	F(2)	C(21) ³⁾	3.597(8)
N(1)	Cl(2)	3.583(2)	N(1)	F(1)	2.775(3)
N(1)	F(2)	2.777(2)	N(1)	N(2)	2.554(3)
N(1)	N(3)	2.604(3)	N(1)	C(2)	2.462(4)
N(1)	C(6)	2.461(3)	N(1)	C(8)	2.419(3)
N(1)	C(13)	3.554(3)	N(2)	Cl(1)	3.025(2)
N(2)	Cl(2)	3.033(2)	N(2)	N(1)	2.554(3)
N(2)	C(7)	2.378(3)	N(2)	C(10)	2.383(3)
N(2)	C(11)	2.765(3)	N(2)	C(12)	2.394(3)
N(3)	Cl(1)	2.993(2)	N(3)	Cl(2)	3.046(2)
N(3)	F(2)	3.277(3)	N(3)	N(1)	2.604(3)
N(3)	C(1)	2.850(3)	N(3)	C(6)	3.178(3)
N(3)	C(14)	2.404(4)	N(3)	C(18)	2.404(3)
N(3)	C(19)	2.872(4)	N(3)	C(20)	2.850(3)
C(1)	V(1)	2.929(2)	C(1)	F(1)	2.347(3)
C(1)	F(2)	2.339(3)	C(1)	N(3)	2.850(3)
C(1)	C(3)	2.423(4)	C(1)	C(4)	2.792(4)
C(1)	C(5)	2.431(4)	C(1)	C(7)	2.396(3)
C(1)	C(13)	3.243(3)	C(2)	F(2)	3.571(4)
C(2)	N(1)	2.462(4)	C(2)	C(4)	2.363(4)
C(2)	C(5)	2.732(4)	C(2)	C(6)	2.341(4)

Table 9. Distances beyond the asymmetric unit out to 3.60 Å (continued)

atom	atom	distance	atom	atom	distance
C(2)	C(7)	3.122(4)	C(2)	C(19)	3.509(4)
C(3)	F(1)	2.353(4)	C(3)	C(1)	2.423(4)
C(3)	C(5)	2.382(5)	C(3)	C(6)	2.730(5)
C(3)	C(14) ⁴⁾	3.512(4)	C(4)	F(2)	3.581(4)
C(4)	C(1)	2.792(4)	C(4)	C(2)	2.363(4)
C(4)	C(6)	2.357(5)	C(4)	C(16) ⁴⁾	3.499(5)
C(4)	C(17) ⁴⁾	3.331(5)	C(4)	C(18) ⁴⁾	3.506(4)
C(5)	F(2)	2.347(4)	C(5)	C(1)	2.431(4)
C(5)	C(2)	2.732(4)	C(5)	C(3)	2.382(5)
C(6)	F(1)	3.583(3)	C(6)	N(1)	2.461(3)
C(6)	N(3)	3.178(3)	C(6)	C(2)	2.341(4)
C(6)	C(3)	2.730(5)	C(6)	C(4)	2.357(5)
C(6)	C(7)	3.393(4)	C(6)	C(13)	3.209(4)
C(6)	C(18)	3.463(4)	C(7)	V(1)	2.956(2)
C(7)	F(1)	3.110(4)	C(7)	N(2)	2.378(3)
C(7)	C(1)	2.396(3)	C(7)	C(2)	3.122(4)
C(7)	C(6)	3.393(4)	C(7)	C(12)	2.533(4)
C(8)	V(1)	3.024(2)	C(8)	N(1)	2.419(3)
C(8)	C(9)	2.307(4)	C(8)	C(10)	2.726(4)
C(8)	C(11)	2.385(4)	C(8)	C(11) ²⁾	3.520(4)
C(9)	V(1)	3.192(2)	C(9)	Cl(1)	3.359(3)
C(9)	Cl(1) ¹⁾	3.449(3)	C(9)	Cl(2)	3.451(2)
C(9)	C(8)	2.307(4)	C(9)	C(11)	2.374(4)
C(9)	C(12)	2.723(4)	C(10)	Cl(1) ¹⁾	3.491(3)
C(10)	F(1) ²⁾	3.469(3)	C(10)	N(2)	2.383(3)
C(10)	C(8)	2.726(4)	C(10)	C(12)	2.384(4)
C(11)	Cl(1) ²⁾	3.549(3)	C(11)	N(2)	2.765(3)
C(11)	C(8)	2.385(4)	C(11)	C(8) ²⁾	3.520(4)
C(11)	C(9)	2.374(4)	C(12)	N(2)	2.394(3)
C(12)	C(7)	2.533(4)	C(12)	C(9)	2.723(4)
C(12)	C(10)	2.384(4)	C(13)	V(1)	3.041(2)
C(13)	F(2)	3.503(3)	C(13)	N(1)	3.554(3)
C(13)	C(1)	3.243(3)	C(13)	C(6)	3.209(4)
C(13)	C(15)	2.389(4)	C(13)	C(16)	2.752(4)
C(13)	C(17)	2.394(4)	C(13)	C(19)	2.533(4)
C(13)	C(20)	2.535(4)	C(14)	N(3)	2.404(4)
C(14)	C(3) ⁵⁾	3.512(4)	C(14)	C(16)	2.400(4)

Table 9. Distances beyond the asymmetric unit out to 3.60 Å (continued)

atom	atom	distance	atom	atom	distance
C(14)	C(17)	2.793(4)	C(14)	C(18)	2.461(4)
C(15)	C(13)	2.389(4)	C(15)	C(17)	2.398(5)
C(15)	C(18)	2.800(5)	C(15)	C(19)	2.514(4)
C(16)	C(4) ⁵⁾	3.499(5)	C(16)	C(13)	2.752(4)
C(16)	C(14)	2.400(4)	C(16)	C(18)	2.412(4)
C(17)	C(4) ⁵⁾	3.331(5)	C(17)	C(13)	2.394(4)
C(17)	C(14)	2.793(4)	C(17)	C(15)	2.398(5)
C(17)	C(20)	2.523(5)	C(18)	F(2)	3.404(3)
C(18)	N(3)	2.404(3)	C(18)	C(4) ⁵⁾	3.506(4)
C(18)	C(6)	3.463(4)	C(18)	C(14)	2.461(4)
C(18)	C(15)	2.800(5)	C(18)	C(16)	2.412(4)
C(19)	F(1)	3.526(4)	C(19)	N(3)	2.872(4)
C(19)	C(2)	3.509(4)	C(19)	C(13)	2.533(4)
C(19)	C(15)	2.514(4)	C(20)	F(2)	3.284(4)
C(20)	N(3)	2.850(3)	C(20)	C(13)	2.535(4)
C(20)	C(17)	2.523(5)	C(21)	F(2) ³⁾	3.597(8)
C(21)	C(21) ⁶⁾	2.417(12)	C(21)	C(22) ⁶⁾	2.266(17)
C(21)	C(23)	2.603(16)	C(22)	C(21) ⁶⁾	2.266(17)
C(22)	C(22) ⁶⁾	2.94(2)	C(22)	C(23) ⁶⁾	2.95(2)
C(23)	C(21)	2.603(16)	C(23)	C(22) ⁶⁾	2.95(2)

Symmetry Operators:

- | | |
|------------------|--------------------|
| (1) -X,-Y+1,-Z+2 | (2) -X+1,-Y+1,-Z+2 |
| (3) -X,-Y,-Z+1 | (4) X+1,Y,Z |
| (5) X-1,Y,Z | (6) -X-1,-Y-1,-Z+1 |

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens

atom	atom	distance	atom	atom	distance
V(1)	H(4)	3.560	V(1)	H(6)	3.250
V(1)	H(14)	3.482	V(1)	H(17)	3.510
Cl(1)	H(5) ¹⁾	3.044	Cl(1)	H(6)	3.121
Cl(1)	H(6) ²⁾	2.909	Cl(1)	H(7) ²⁾	3.024
Cl(1)	H(8) ³⁾	2.924	Cl(1)	H(9) ¹⁾	3.435
Cl(1)	H(9) ³⁾	3.191	Cl(1)	H(13)	3.106
Cl(1)	H(14)	3.445	Cl(1)	H(15) ⁴⁾	3.157
Cl(2)	H(4) ¹⁾	3.114	Cl(2)	H(6)	3.207
Cl(2)	H(10) ⁵⁾	3.086	Cl(2)	H(11) ⁵⁾	3.290
Cl(2)	H(17)	3.117	Cl(2)	H(18)	3.354
F(1)	H(1)	2.544	F(1)	H(5)	2.532
F(1)	H(7) ⁶⁾	3.486	F(1)	H(7) ³⁾	2.803
F(1)	H(8) ³⁾	3.103	F(1)	H(13) ⁷⁾	3.341
F(1)	H(14)	2.573	F(2)	H(3)	2.540
F(2)	H(4)	3.336	F(2)	H(12) ⁸⁾	3.047
F(2)	H(16) ⁸⁾	2.815	F(2)	H(17)	2.490
F(2)	H(18) ⁷⁾	3.289	N(1)	H(4)	2.030
N(1)	H(5)	2.030	N(1)	H(14)	3.415
N(2)	H(4)	2.996	N(2)	H(5)	3.054
N(2)	H(6)	1.979	N(2)	H(7)	3.234
N(2)	H(9)	3.246	N(3)	H(13)	2.864
N(3)	H(14)	2.799	N(3)	H(17)	2.710
N(3)	H(18)	2.902	C(1)	H(1)	3.276
C(1)	H(3)	3.283	C(1)	H(4)	2.645
C(1)	H(5)	2.569	C(1)	H(14)	3.131
C(1)	H(18) ⁷⁾	3.591	C(2)	H(1)	2.022
C(2)	H(2)	3.222	C(2)	H(4)	3.502
C(2)	H(5)	2.831	C(2)	H(14)	2.712
C(3)	H(2)	2.027	C(3)	H(3)	3.243
C(3)	H(14)	3.318	C(4)	H(1)	2.038
C(4)	H(3)	2.016	C(5)	H(1)	3.240
C(5)	H(2)	2.007	C(5)	H(16) ⁸⁾	3.267
C(5)	H(18) ⁷⁾	3.247	C(6)	H(2)	3.217
C(6)	H(3)	2.033	C(6)	H(4)	3.290
C(6)	H(16) ⁸⁾	3.463	C(6)	H(17)	3.322
C(6)	H(18) ⁷⁾	3.086	C(7)	H(9)	2.729
C(8)	H(4)	2.039	C(8)	H(5)	2.039

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(8)	H(6)	3.162	C(8)	H(8)	3.250
C(8)	H(9)	2.050	C(9)	H(1) ⁵⁾	3.582
C(9)	H(7)	2.033	C(9)	H(8)	3.237
C(9)	H(9) ³⁾	3.514	C(9)	H(10) ⁵⁾	3.484
C(9)	H(15) ⁵⁾	3.534	C(10)	H(1) ⁵⁾	2.890
C(10)	H(5) ³⁾	3.137	C(10)	H(6)	2.014
C(10)	H(8)	2.033	C(10)	H(9)	3.248
C(10)	H(15) ⁵⁾	3.483	C(11)	H(1) ⁵⁾	2.841
C(11)	H(5) ³⁾	3.482	C(11)	H(6)	3.228
C(11)	H(7)	2.037	C(11)	H(9)	2.035
C(12)	H(1) ⁵⁾	3.486	C(12)	H(4)	2.824
C(12)	H(5)	2.758	C(12)	H(7)	3.247
C(12)	H(8)	2.031	C(12)	H(10) ⁹⁾	3.483
C(13)	H(10)	3.251	C(13)	H(12)	3.259
C(13)	H(13)	2.770	C(13)	H(14)	2.790
C(13)	H(15)	3.324	C(13)	H(16)	3.330
C(13)	H(17)	2.757	C(13)	H(18)	2.798
C(14)	H(1) ¹⁾	3.294	C(14)	H(10)	2.029
C(14)	H(11)	3.253	C(14)	H(13)	2.053
C(14)	H(14)	2.053	C(14)	H(15)	2.053
C(15)	H(1) ¹⁾	3.315	C(15)	H(2) ¹⁾	3.544
C(15)	H(11)	2.020	C(15)	H(12)	3.249
C(15)	H(13)	3.139	C(15)	H(14)	3.130
C(15)	H(15)	2.555	C(16)	H(2) ¹⁾	3.087
C(16)	H(10)	2.019	C(16)	H(12)	2.030
C(17)	H(2) ¹⁾	3.141	C(17)	H(10)	3.251
C(17)	H(11)	2.032	C(17)	H(16)	2.574
C(17)	H(16) ⁸⁾	3.580	C(17)	H(17)	3.168
C(17)	H(17) ⁸⁾	3.554	C(17)	H(18)	3.115
C(18)	H(11)	3.264	C(18)	H(12)	2.029
C(18)	H(16)	2.047	C(18)	H(16) ⁸⁾	3.520
C(18)	H(17)	2.047	C(18)	H(18)	2.047
C(19)	H(7) ⁶⁾	3.525	C(19)	H(8) ¹⁰⁾	3.578
C(19)	H(10)	2.665	C(19)	H(13) ⁴⁾	3.488
C(20)	H(3) ⁸⁾	3.288	C(20)	H(12)	2.685
C(20)	H(12) ⁸⁾	3.545	C(20)	H(16) ⁸⁾	3.211
C(21)	H(2) ¹¹⁾	3.251	C(21)	H(11)	3.550

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(21)	H(11) ¹²⁾	3.499	C(21)	H(12)	3.218
C(22)	H(2) ¹¹⁾	2.937	C(22)	H(3) ¹¹⁾	3.550
C(22)	H(11)	2.961	C(22)	H(12)	3.516
C(23)	H(2) ¹¹⁾	3.454	C(23)	H(3) ¹¹⁾	3.175
C(23)	H(11)	2.972	C(23)	H(18) ⁶⁾	3.392
H(1)	F(1)	2.544	H(1)	C(1)	3.276
H(1)	C(2)	2.022	H(1)	C(4)	2.038
H(1)	C(5)	3.240	H(1)	C(9) ⁶⁾	3.582
H(1)	C(10) ⁶⁾	2.890	H(1)	C(11) ⁶⁾	2.841
H(1)	C(12) ⁶⁾	3.486	H(1)	C(14) ⁷⁾	3.294
H(1)	C(15) ⁷⁾	3.315	H(1)	H(2)	2.336
H(1)	H(7) ⁶⁾	2.997	H(1)	H(8) ⁶⁾	2.932
H(1)	H(10) ⁷⁾	3.507	H(1)	H(13) ⁷⁾	3.286
H(1)	H(14)	3.576	H(2)	C(2)	3.222
H(2)	C(3)	2.027	H(2)	C(5)	2.007
H(2)	C(6)	3.217	H(2)	C(15) ⁷⁾	3.544
H(2)	C(16) ⁷⁾	3.087	H(2)	C(17) ⁷⁾	3.141
H(2)	C(21) ¹¹⁾	3.251	H(2)	C(22) ¹¹⁾	2.937
H(2)	C(23) ¹¹⁾	3.454	H(2)	H(1)	2.336
H(2)	H(3)	2.312	H(2)	H(11) ⁷⁾	3.276
H(2)	H(12) ⁷⁾	3.330	H(3)	F(2)	2.540
H(3)	C(1)	3.283	H(3)	C(3)	3.243
H(3)	C(4)	2.016	H(3)	C(6)	2.033
H(3)	C(20) ⁸⁾	3.288	H(3)	C(22) ¹¹⁾	3.550
H(3)	C(23) ¹¹⁾	3.175	H(3)	H(2)	2.312
H(3)	H(12) ⁷⁾	3.513	H(3)	H(16) ⁸⁾	2.574
H(3)	H(18) ⁷⁾	3.444	H(3)	H(18) ⁸⁾	3.212
H(4)	V(1)	3.560	H(4)	Cl(2) ⁷⁾	3.114
H(4)	F(2)	3.336	H(4)	N(1)	2.030
H(4)	N(2)	2.996	H(4)	C(1)	2.645
H(4)	C(2)	3.502	H(4)	C(6)	3.290
H(4)	C(8)	2.039	H(4)	C(12)	2.824
H(4)	H(5)	1.603	H(4)	H(9)	2.836
H(4)	H(18) ⁷⁾	3.355	H(5)	Cl(1) ⁷⁾	3.044
H(5)	F(1)	2.532	H(5)	N(1)	2.030
H(5)	N(2)	3.054	H(5)	C(1)	2.569
H(5)	C(2)	2.831	H(5)	C(8)	2.039

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(5)	C(10) ³⁾	3.137	H(5)	C(11) ³⁾	3.482
H(5)	C(12)	2.758	H(5)	H(4)	1.603
H(5)	H(7) ³⁾	2.870	H(5)	H(8) ³⁾	3.462
H(5)	H(9)	2.726	H(6)	V(1)	3.250
H(6)	Cl(1)	3.121	H(6)	Cl(1) ²⁾	2.909
H(6)	Cl(2)	3.207	H(6)	N(2)	1.979
H(6)	C(8)	3.162	H(6)	C(10)	2.014
H(6)	C(11)	3.228	H(6)	H(7)	2.318
H(6)	H(9) ¹⁾	2.724	H(6)	H(10) ⁵⁾	2.894
H(6)	H(15) ⁵⁾	2.984	H(7)	Cl(1) ²⁾	3.024
H(7)	F(1) ⁵⁾	3.486	H(7)	F(1) ³⁾	2.803
H(7)	N(2)	3.234	H(7)	C(9)	2.033
H(7)	C(11)	2.037	H(7)	C(12)	3.247
H(7)	C(19) ⁵⁾	3.525	H(7)	H(1) ⁵⁾	2.997
H(7)	H(5) ³⁾	2.870	H(7)	H(6)	2.318
H(7)	H(8)	2.345	H(7)	H(14) ⁵⁾	3.255
H(7)	H(15) ⁵⁾	2.898	H(8)	Cl(1) ³⁾	2.924
H(8)	F(1) ³⁾	3.103	H(8)	C(8)	3.250
H(8)	C(9)	3.237	H(8)	C(10)	2.033
H(8)	C(12)	2.031	H(8)	C(19) ⁹⁾	3.578
H(8)	H(1) ⁵⁾	2.932	H(8)	H(5) ³⁾	3.462
H(8)	H(7)	2.345	H(8)	H(9)	2.343
H(8)	H(13) ⁹⁾	3.320	H(8)	H(14) ³⁾	3.499
H(8)	H(15) ⁹⁾	3.002	H(9)	Cl(1) ⁷⁾	3.435
H(9)	Cl(1) ³⁾	3.191	H(9)	N(2)	3.246
H(9)	C(7)	2.729	H(9)	C(8)	2.050
H(9)	C(9) ³⁾	3.514	H(9)	C(10)	3.248
H(9)	C(11)	2.035	H(9)	H(4)	2.836
H(9)	H(5)	2.726	H(9)	H(6) ⁷⁾	2.724
H(9)	H(8)	2.343	H(9)	H(10) ⁹⁾	3.016
H(9)	H(15) ⁹⁾	3.462	H(10)	Cl(2) ⁶⁾	3.086
H(10)	C(9) ⁶⁾	3.484	H(10)	C(12) ¹⁰⁾	3.483
H(10)	C(13)	3.251	H(10)	C(14)	2.029
H(10)	C(16)	2.019	H(10)	C(17)	3.251
H(10)	C(19)	2.665	H(10)	H(1) ¹⁾	3.507
H(10)	H(6) ⁶⁾	2.894	H(10)	H(9) ¹⁰⁾	3.016
H(10)	H(11)	2.313	H(10)	H(13)	3.343

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(10)	H(14)	3.309	H(10)	H(15)	2.341
H(11)	Cl(2) ⁶⁾	3.290	H(11)	C(14)	3.253
H(11)	C(15)	2.020	H(11)	C(17)	2.032
H(11)	C(18)	3.264	H(11)	C(21)	3.550
H(11)	C(21) ¹²⁾	3.499	H(11)	C(22)	2.961
H(11)	C(23)	2.972	H(11)	H(2) ¹⁾	3.276
H(11)	H(10)	2.313	H(11)	H(12)	2.325
H(12)	F(2) ⁸⁾	3.047	H(12)	C(13)	3.259
H(12)	C(15)	3.249	H(12)	C(16)	2.030
H(12)	C(18)	2.029	H(12)	C(20)	2.685
H(12)	C(20) ⁸⁾	3.545	H(12)	C(21)	3.218
H(12)	C(22)	3.516	H(12)	H(2) ¹⁾	3.330
H(12)	H(3) ¹⁾	3.513	H(12)	H(11)	2.325
H(12)	H(16)	2.372	H(12)	H(16) ⁸⁾	3.300
H(12)	H(17)	3.376	H(12)	H(17) ⁸⁾	2.889
H(12)	H(18)	3.312	H(13)	Cl(1)	3.106
H(13)	F(1) ¹⁾	3.341	H(13)	N(3)	2.864
H(13)	C(13)	2.770	H(13)	C(14)	2.053
H(13)	C(15)	3.139	H(13)	C(19) ⁴⁾	3.488
H(13)	H(1) ¹⁾	3.286	H(13)	H(8) ¹⁰⁾	3.320
H(13)	H(10)	3.343	H(13)	H(13) ⁴⁾	3.212
H(13)	H(14)	1.600	H(13)	H(14) ⁴⁾	3.366
H(13)	H(15)	1.600	H(13)	H(15) ⁴⁾	3.306
H(14)	V(1)	3.482	H(14)	Cl(1)	3.445
H(14)	F(1)	2.573	H(14)	N(1)	3.415
H(14)	N(3)	2.799	H(14)	C(1)	3.131
H(14)	C(2)	2.712	H(14)	C(3)	3.318
H(14)	C(13)	2.790	H(14)	C(14)	2.053
H(14)	C(15)	3.130	H(14)	H(1)	3.576
H(14)	H(7) ⁶⁾	3.255	H(14)	H(8) ³⁾	3.499
H(14)	H(10)	3.309	H(14)	H(13)	1.600
H(14)	H(13) ⁴⁾	3.366	H(14)	H(15)	1.600
H(15)	Cl(1) ⁴⁾	3.157	H(15)	C(9) ⁶⁾	3.534
H(15)	C(10) ⁶⁾	3.483	H(15)	C(13)	3.324
H(15)	C(14)	2.053	H(15)	C(15)	2.555
H(15)	H(6) ⁶⁾	2.984	H(15)	H(7) ⁶⁾	2.898
H(15)	H(8) ¹⁰⁾	3.002	H(15)	H(9) ¹⁰⁾	3.462

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(15)	H(10)	2.341	H(15)	H(13)	1.600
H(15)	H(13) ⁴⁾	3.306	H(15)	H(14)	1.600
H(16)	F(2) ⁸⁾	2.815	H(16)	C(5) ⁸⁾	3.267
H(16)	C(6) ⁸⁾	3.463	H(16)	C(13)	3.330
H(16)	C(17)	2.574	H(16)	C(17) ⁸⁾	3.580
H(16)	C(18)	2.047	H(16)	C(18) ⁸⁾	3.520
H(16)	C(20) ⁸⁾	3.211	H(16)	H(3) ⁸⁾	2.574
H(16)	H(12)	2.372	H(16)	H(12) ⁸⁾	3.300
H(16)	H(16) ⁸⁾	2.831	H(16)	H(17)	1.600
H(16)	H(17) ⁸⁾	2.947	H(16)	H(18)	1.600
H(17)	V(1)	3.510	H(17)	Cl(2)	3.117
H(17)	F(2)	2.490	H(17)	N(3)	2.710
H(17)	C(6)	3.322	H(17)	C(13)	2.757
H(17)	C(17)	3.168	H(17)	C(17) ⁸⁾	3.554
H(17)	C(18)	2.047	H(17)	H(12)	3.376
H(17)	H(12) ⁸⁾	2.889	H(17)	H(16)	1.600
H(17)	H(16) ⁸⁾	2.947	H(17)	H(18)	1.600
H(18)	Cl(2)	3.354	H(18)	F(2) ¹⁾	3.289
H(18)	N(3)	2.902	H(18)	C(1) ¹⁾	3.591
H(18)	C(5) ¹⁾	3.247	H(18)	C(6) ¹⁾	3.086
H(18)	C(13)	2.798	H(18)	C(17)	3.115
H(18)	C(18)	2.047	H(18)	C(23) ⁵⁾	3.392
H(18)	H(3) ¹⁾	3.444	H(18)	H(3) ⁸⁾	3.212
H(18)	H(4) ¹⁾	3.355	H(18)	H(12)	3.312
H(18)	H(16)	1.600	H(18)	H(17)	1.600

Symmetry Operators:

- | | |
|--------------------|---------------------|
| (1) X-1,Y,Z | (2) -X,-Y+1,-Z+2 |
| (3) -X+1,-Y+1,-Z+2 | (4) -X,-Y,-Z+2 |
| (5) X,Y+1,Z | (6) X,Y-1,Z |
| (7) X+1,Y,Z | (8) -X,-Y,-Z+1 |
| (9) X+1,Y+1,Z | (10) X-1,Y-1,Z |
| (11) -X,-Y-1,-Z+1 | (12) -X-1,-Y-1,-Z+1 |

X-ray Structure Report for $\text{V}(\text{N}=2,6\text{-Me}_2\text{C}_6\text{H}_3)(\text{CH}_2\text{SiMe}_3)_2[2\text{-(}2,6\text{-Me}_2\text{C}_6\text{H}_3\text{)CH}_2(\text{C}_5\text{H}_4\text{N})]$ (**2a**)

December 8, 2008

Experimental

Data Collection

A red block crystal of $C_{30}H_{46}N_3Si_2V$ having approximate dimensions of 0.50 x 0.40 x 0.40 mm was mounted on a glass fiber. All measurements were made on a Rigaku RAXIS RAPID imaging plate area detector with graphite monochromated Mo-K α radiation.

Indexing was performed from 3 oscillations that were exposed for 140 seconds. The crystal-to-detector distance was 127.40 mm.

Cell constants and an orientation matrix for data collection corresponded to a primitive monoclinic cell with dimensions:

$$\begin{aligned} a &= 15.9173(4) \text{ \AA} \\ b &= 11.9182(3) \text{ \AA} & \beta &= 103.6830(7)^\circ \\ c &= 17.7030(4) \text{ \AA} \\ V &= 3263.05(13) \text{ \AA}^3 \end{aligned}$$

For $Z = 4$ and F.W. = 555.83, the calculated density is 1.131 g/cm³. The systematic absences of:

$$\begin{aligned} h0l: & l \pm 2n \\ 0k0: & k \pm 2n \end{aligned}$$

uniquely determine the space group to be:

$$P2_1/c \text{ (\#14)}$$

The data were collected at a temperature of $-80 \pm 1^\circ\text{C}$ to a maximum 2θ value of 50.7° . A total of 112 oscillation images were collected. A sweep of data was done using ω scans from 130.0 to 190.0° in 2.0° step, at $\chi=45.0^\circ$ and $\phi = 0.0^\circ$. The exposure rate was 190.0 [sec./ $^\circ$]. A second sweep was performed using ω scans from 0.0 to 164.0° in 2.0° step, at $\chi=45.0^\circ$ and $\phi = 210.0^\circ$. The exposure rate was 190.0 [sec./ $^\circ$]. The crystal-to-detector distance was 127.40 mm. Readout was performed in the 0.100 mm pixel mode.

Data Reduction

Of the 26522 reflections that were collected, 5936 were unique ($R_{\text{int}} = 0.022$); equivalent reflections were merged.

The linear absorption coefficient, μ , for Mo-K α radiation is 3.985 cm⁻¹. An empirical absorption correction was applied which resulted in transmission factors ranging from 0.581 to 0.853. The data were corrected for Lorentz and polarization effects.

Structure Solution and Refinement

The structure was solved by direct methods¹ and expanded using Fourier techniques². The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were refined using the riding model. The final cycle of full-matrix least-squares refinement³ on F^2 was based on 5233 observed reflections and 371 variable parameters and converged (largest parameter shift was 0.00 times its esd) with unweighted and weighted agreement factors of:

$$R1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o| = 0.0372$$

$$wR2 = [\Sigma (w (F_o^2 - F_c^2)^2) / \Sigma w(F_o^2)^2]^{1/2} = 0.1323$$

The standard deviation of an observation of unit weight⁴ was 1.01. A Sheldrick weighting scheme was used. Plots of $\Sigma w (|F_o| - |F_c|)^2$ versus $|F_o|$, reflection order in data collection, $\sin \theta/\lambda$ and various classes of indices showed no unusual trends. The maximum and minimum peaks on the final difference Fourier map corresponded to 0.51 and -0.30 e⁻/Å³, respectively.

Neutral atom scattering factors were taken from Cromer and Waber⁵. Anomalous dispersion effects were included in F_{calc} ⁶; the values for $\Delta f'$ and $\Delta f''$ were those of Creagh and McAuley⁷. The values for the mass attenuation coefficients are those of Creagh and Hubbell⁸. All calculations were performed using the CrystalStructure^{9,10} crystallographic software package.

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(3) Least Squares function minimized:

$$\sum w(F_o^2 - F_c^2)^2 \quad \text{where } w = \text{Least Squares weights.}$$

(4) Standard deviation of an observation of unit weight:

$$[\sum w(F_o^2 - F_c^2)^2 / (N_o - N_v)]^{1/2}$$

where: N_o = number of observations
 N_v = number of variables

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EXPERIMENTAL DETAILS

A. Crystal Data

Empirical Formula	C ₃₀ H ₄₆ N ₃ Si ₂ V
Formula Weight	555.83
Crystal Color, Habit	red, block
Crystal Dimensions	0.50 X 0.40 X 0.40 mm
Crystal System	monoclinic
Lattice Type	Primitive
Indexing Images	3 oscillations @ 140.0 seconds
Detector Position	127.40 mm
Pixel Size	0.100 mm
Lattice Parameters	a = 15.9173(4) Å b = 11.9182(3) Å c = 17.7030(4) Å β = 103.6830(7) ° V = 3263.05(13) Å ³
Space Group	P2 ₁ /c (#14)
Z value	4
D _{calc}	1.131 g/cm ³
F ₀₀₀	1192.00
μ(MoKα)	3.985 cm ⁻¹

B. Intensity Measurements

Diffractometer	Rigaku RAXIS-RAPID
Radiation	MoK α (λ = 0.71075 Å) graphite monochromated
Detector Aperture	280 mm x 256 mm
Data Images	112 exposures
ω oscillation Range (χ =45.0, ϕ =0.0)	130.0 - 190.0 $^\circ$
Exposure Rate	190.0 sec./ $^\circ$
ω oscillation Range (χ =45.0, ϕ =210.0)	0.0 - 164.0 $^\circ$
Exposure Rate	190.0 sec./ $^\circ$
Detector Position	127.40 mm
Pixel Size	0.100 mm
$2\theta_{\max}$	50.7 $^\circ$
No. of Reflections Measured	Total: 26522 Unique: 5936 (R_{int} = 0.022)
Corrections	Lorentz-polarization Absorption (trans. factors: 0.581 - 0.853)

C. Structure Solution and Refinement

Structure Solution	Direct Methods (SIR92)
Refinement	Full-matrix least-squares on F^2
Function Minimized	$\sum w (F_o^2 - F_c^2)^2$
Least Squares Weights	$1/[0.0035F_o^2 + 1.0000\sigma(F_o^2)]/(4F_o^2)$
$2\theta_{\max}$ cutoff	50.7°
Anomalous Dispersion	All non-hydrogen atoms
No. Observations ($I > 2.00\sigma(I)$)	5233
No. Variables	371
Reflection/Parameter Ratio	14.11
Residuals: R1 ($I > 2.00\sigma(I)$)	0.0372
Residuals: wR2 ($I > 2.00\sigma(I)$)	0.1323
Goodness of Fit Indicator	1.008
Max Shift/Error in Final Cycle	0.000
Maximum peak in Final Diff. Map	0.51 e ⁻ /Å ³
Minimum peak in Final Diff. Map	-0.30 e ⁻ /Å ³

Table 1. Atomic coordinates and $B_{\text{iso}}/B_{\text{eq}}$

atom	x	y	z	B_{eq}
V(1)	0.73681(2)	0.05304(2)	0.716240(10)	2.142(7)
Si(1)	0.85832(4)	-0.05171(4)	0.58422(3)	3.300(12)
Si(2)	0.82580(4)	-0.02203(5)	0.90525(3)	3.642(12)
N(1)	0.62918(9)	0.08907(12)	0.64756(8)	2.56(2)
N(2)	0.68027(9)	-0.11734(11)	0.67387(8)	2.68(2)
N(3)	0.77042(9)	0.18497(12)	0.73959(8)	2.61(2)
C(1)	0.57561(12)	0.00630(15)	0.59725(12)	3.47(4)
C(2)	0.60810(12)	-0.10997(14)	0.61729(11)	2.99(3)
C(3)	0.56784(14)	-0.20393(16)	0.57782(13)	4.06(4)
C(4)	0.60300(15)	-0.30765(17)	0.59744(13)	4.25(4)
C(5)	0.67739(15)	-0.31690(16)	0.65716(12)	3.93(4)
C(6)	0.71307(12)	-0.22076(15)	0.69357(11)	3.22(3)
C(7)	0.58860(11)	0.19635(14)	0.63618(11)	2.88(3)
C(8)	0.52098(14)	0.22050(19)	0.67268(15)	4.73(5)
C(9)	0.48100(18)	0.3241(2)	0.65976(19)	6.44(7)
C(10)	0.50636(17)	0.4038(2)	0.61308(18)	5.73(6)
C(11)	0.57207(16)	0.38000(16)	0.57820(13)	4.40(4)
C(12)	0.61365(12)	0.27666(15)	0.58836(10)	3.08(3)
C(13)	0.4930(2)	0.1363(2)	0.7252(2)	7.69(9)
C(14)	0.68454(16)	0.25185(18)	0.54750(13)	4.35(4)
C(15)	0.79475(13)	0.29628(15)	0.75570(10)	3.28(3)
C(16)	0.75058(18)	0.36307(18)	0.80007(12)	4.81(5)
C(17)	0.7774(2)	0.4756(2)	0.81391(15)	6.53(7)
C(18)	0.8442(2)	0.5210(2)	0.7853(2)	7.27(7)
C(19)	0.88455(19)	0.4549(2)	0.7432(2)	6.54(7)
C(20)	0.86270(14)	0.34339(18)	0.72756(16)	4.77(5)
C(21)	0.6810(2)	0.3172(2)	0.83238(18)	7.63(9)
C(22)	0.91254(18)	0.2735(2)	0.6825(2)	7.10(8)
C(23)	0.84375(11)	-0.01064(15)	0.68256(10)	2.81(3)
C(24)	0.75669(15)	-0.1002(2)	0.51494(12)	4.31(4)
C(25)	0.90224(19)	0.0675(2)	0.53573(16)	5.49(6)
C(26)	0.93643(17)	-0.1718(2)	0.59607(16)	5.37(6)
C(27)	0.72712(11)	0.00143(16)	0.82648(10)	3.01(3)
C(28)	0.7956(2)	-0.0434(2)	1.00012(14)	6.51(7)
C(29)	0.88107(16)	-0.1567(2)	0.88725(13)	5.18(5)
C(30)	0.9009(2)	0.0979(3)	0.9135(2)	9.97(10)

$$B_{\text{eq}} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 2. Atomic coordinates and B_{iso} involving hydrogens/ B_{eq}

atom	x	y	z	B_{eq}
H(1)	0.5131	0.0122	0.6037	3.89
H(2)	0.5765	0.0228	0.5403	3.83
H(3)	0.5128	-0.1955	0.5348	4.53
H(4)	0.5756	-0.3760	0.5693	4.87
H(5)	0.7044	-0.3925	0.6732	4.64
H(6)	0.7650	-0.2273	0.7360	3.77
H(7)	0.4301	0.3427	0.6861	8.00
H(8)	0.4768	0.4784	0.6049	6.71
H(9)	0.5910	0.4391	0.5440	5.08
H(10)	0.5388	0.0835	0.7436	10.04
H(11)	0.4796	0.1745	0.7687	10.06
H(12)	0.4426	0.0972	0.6969	10.05
H(13)	0.6828	0.1747	0.5338	5.43
H(14)	0.6765	0.2964	0.5018	5.44
H(15)	0.7390	0.2689	0.5811	5.45
H(16)	0.7471	0.5247	0.8456	7.26
H(17)	0.8635	0.6083	0.7967	7.67
H(18)	0.9347	0.4894	0.7214	7.22
H(19)	0.6717	0.3659	0.8736	9.76
H(20)	0.6282	0.3125	0.7917	9.77
H(21)	0.6970	0.2429	0.8532	9.79
H(22)	0.8728	0.2269	0.6455	8.86
H(23)	0.9438	0.3223	0.6549	8.90
H(24)	0.9532	0.2261	0.7181	8.89
H(25)	0.8901	0.0476	0.7012	3.45
H(26)	0.8585	-0.0811	0.7146	3.42
H(27)	0.7149	-0.0384	0.5045	5.24
H(28)	0.7700	-0.1243	0.4662	5.21
H(29)	0.7320	-0.1632	0.5379	5.17
H(30)	0.9591	0.0920	0.5691	7.05
H(31)	0.9103	0.0423	0.4840	7.09
H(32)	0.8607	0.1317	0.5282	7.09
H(33)	0.9137	-0.2360	0.6217	6.42
H(34)	0.9437	-0.1958	0.5438	6.48
H(35)	0.9936	-0.1476	0.6290	6.46
H(36)	0.6912	0.0616	0.8465	3.74
H(37)	0.6936	-0.0733	0.8190	3.72

Table 2. Atomic coordinates and B_{iso} involving hydrogens/ B_{eq} (continued)

atom	x	y	z	B_{eq}
H(38)	0.8524	-0.0535	1.0450	7.95
H(39)	0.7606	0.0273	1.0123	7.94
H(40)	0.7566	-0.1162	0.9968	7.95
H(41)	0.9325	-0.1702	0.9312	6.19
H(42)	0.8397	-0.2208	0.8836	6.22
H(43)	0.9001	-0.1503	0.8375	6.17
H(44)	0.8693	0.1726	0.9258	10.45
H(45)	0.9567	0.0826	0.9594	10.49
H(46)	0.9201	0.1082	0.8599	10.49

$$B_{\text{eq}} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 3. Anisotropic displacement parameters

atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
V(1)	0.02481(19)	0.0288(2)	0.0272(2)	0.00015(10)	0.00512(13)	-0.00051(9)
Si(1)	0.0439(3)	0.0446(3)	0.0411(3)	0.0060(2)	0.0184(2)	-0.0033(2)
Si(2)	0.0540(3)	0.0465(3)	0.0325(2)	-0.0058(2)	-0.0006(2)	0.0065(2)
N(1)	0.0289(7)	0.0273(7)	0.0389(7)	-0.0005(5)	0.0037(5)	0.0025(5)
N(2)	0.0330(7)	0.0300(7)	0.0366(7)	0.0003(5)	0.0040(6)	0.0024(5)
N(3)	0.0348(7)	0.0345(7)	0.0279(6)	-0.0003(6)	0.0037(5)	-0.0030(5)
C(1)	0.0321(9)	0.0335(9)	0.0564(12)	-0.0039(7)	-0.0092(8)	0.0044(8)
C(2)	0.0343(9)	0.0304(8)	0.0449(9)	-0.0044(7)	0.0010(7)	0.0044(7)
C(3)	0.0502(11)	0.0364(10)	0.0561(11)	-0.0110(8)	-0.0104(9)	0.0039(8)
C(4)	0.0625(13)	0.0313(9)	0.0603(12)	-0.0102(8)	-0.0006(10)	-0.0006(8)
C(5)	0.0619(12)	0.0305(8)	0.0537(11)	0.0053(8)	0.0072(9)	0.0049(8)
C(6)	0.0416(9)	0.0349(9)	0.0426(10)	0.0052(7)	0.0038(8)	0.0058(7)
C(7)	0.0315(8)	0.0308(8)	0.0440(9)	0.0030(6)	0.0030(7)	0.0015(7)
C(8)	0.0443(11)	0.0533(12)	0.0887(17)	0.0166(9)	0.0289(11)	0.0174(11)
C(9)	0.0647(16)	0.0710(17)	0.118(2)	0.0343(13)	0.0389(16)	0.0136(15)
C(10)	0.0726(16)	0.0422(11)	0.098(2)	0.0245(11)	0.0093(14)	0.0073(12)
C(11)	0.0734(14)	0.0288(9)	0.0581(12)	0.0005(9)	0.0017(11)	0.0033(8)
C(12)	0.0452(10)	0.0297(8)	0.0382(9)	-0.0040(7)	0.0018(7)	-0.0006(6)
C(13)	0.0775(19)	0.090(2)	0.150(3)	0.0252(16)	0.079(2)	0.039(2)
C(14)	0.0769(15)	0.0453(11)	0.0493(11)	-0.0057(10)	0.0271(11)	0.0052(9)
C(15)	0.0492(10)	0.0325(9)	0.0330(9)	0.0016(7)	-0.0100(7)	-0.0046(6)
C(16)	0.1005(18)	0.0437(11)	0.0306(9)	0.0184(11)	-0.0006(10)	-0.0052(8)
C(17)	0.134(2)	0.0444(12)	0.0501(14)	0.0293(16)	-0.0176(16)	-0.0165(10)
C(18)	0.100(2)	0.0384(12)	0.105(2)	0.0009(14)	-0.042(2)	-0.0071(14)
C(19)	0.0605(16)	0.0430(13)	0.124(2)	-0.0148(11)	-0.0204(16)	0.0020(13)
C(20)	0.0424(11)	0.0415(11)	0.0863(16)	-0.0130(9)	-0.0072(10)	-0.0012(10)
C(21)	0.164(3)	0.0727(17)	0.0721(18)	0.0323(19)	0.065(2)	-0.0040(14)
C(22)	0.0532(14)	0.0659(16)	0.162(3)	-0.0205(12)	0.0481(17)	-0.0037(18)
C(23)	0.0318(8)	0.0391(9)	0.0362(9)	0.0018(7)	0.0089(7)	-0.0025(7)
C(24)	0.0645(13)	0.0644(13)	0.0354(10)	0.0030(10)	0.0129(9)	-0.0051(9)
C(25)	0.0805(18)	0.0758(16)	0.0678(15)	-0.0079(13)	0.0481(14)	0.0008(12)
C(26)	0.0672(15)	0.0695(15)	0.0675(15)	0.0219(12)	0.0164(12)	-0.0171(12)
C(27)	0.0372(9)	0.0446(9)	0.0343(9)	0.0010(7)	0.0119(7)	0.0039(7)
C(28)	0.138(2)	0.0758(17)	0.0373(12)	0.0444(16)	0.0275(15)	0.0095(10)
C(29)	0.0566(13)	0.0940(19)	0.0451(12)	0.0266(13)	0.0097(10)	0.0132(11)
C(30)	0.130(3)	0.101(2)	0.101(2)	-0.069(2)	-0.067(2)	0.045(2)

The general temperature factor expression: $\exp(-2\pi^2(a^2U_{11}h^2 + b^2U_{22}k^2 + c^2U_{33}l^2 + 2a*b*U_{12}hk + 2a*c*U_{13}hl + 2b*c*U_{23}kl))$

Table 4. Bond lengths (Å)

atom	atom	distance	atom	atom	distance
V(1)	N(1)	1.9006(13)	V(1)	N(2)	2.2752(13)
V(1)	N(3)	1.6807(14)	V(1)	C(23)	2.0758(19)
V(1)	C(27)	2.0864(18)	Si(1)	C(23)	1.8749(19)
Si(1)	C(24)	1.875(2)	Si(1)	C(25)	1.878(2)
Si(1)	C(26)	1.875(2)	Si(2)	C(27)	1.8584(16)
Si(2)	C(28)	1.871(3)	Si(2)	C(29)	1.894(2)
Si(2)	C(30)	1.847(3)	N(1)	C(1)	1.461(2)
N(1)	C(7)	1.425(2)	N(2)	C(2)	1.337(2)
N(2)	C(6)	1.352(2)	N(3)	C(15)	1.392(2)
C(1)	C(2)	1.492(2)	C(2)	C(3)	1.393(2)
C(3)	C(4)	1.367(2)	C(4)	C(5)	1.393(2)
C(5)	C(6)	1.370(2)	C(7)	C(8)	1.410(3)
C(7)	C(12)	1.397(2)	C(8)	C(9)	1.384(3)
C(8)	C(13)	1.506(4)	C(9)	C(10)	1.380(4)
C(10)	C(11)	1.364(4)	C(11)	C(12)	1.390(2)
C(12)	C(14)	1.507(3)	C(15)	C(16)	1.416(3)
C(15)	C(20)	1.410(3)	C(16)	C(17)	1.411(3)
C(16)	C(21)	1.468(4)	C(17)	C(18)	1.391(5)
C(18)	C(19)	1.348(5)	C(19)	C(20)	1.385(3)
C(20)	C(22)	1.503(4)			

Table 5. Bond lengths involving hydrogens (Å)

atom	atom	distance	atom	atom	distance
C(1)	H(1)	1.030	C(1)	H(2)	1.030
C(3)	H(3)	1.020	C(4)	H(4)	1.000
C(5)	H(5)	1.010	C(6)	H(6)	0.980
C(9)	H(7)	1.050	C(10)	H(8)	1.000
C(11)	H(9)	1.020	C(13)	H(10)	0.960
C(13)	H(11)	0.960	C(13)	H(12)	0.960
C(14)	H(13)	0.950	C(14)	H(14)	0.950
C(14)	H(15)	0.950	C(17)	H(16)	1.010
C(18)	H(17)	1.090	C(19)	H(18)	1.050
C(21)	H(19)	0.970	C(21)	H(20)	0.970
C(21)	H(21)	0.970	C(22)	H(22)	0.970
C(22)	H(23)	0.970	C(22)	H(24)	0.970
C(23)	H(25)	1.010	C(23)	H(26)	1.010
C(24)	H(27)	0.980	C(24)	H(28)	0.980
C(24)	H(29)	0.980	C(25)	H(30)	1.000
C(25)	H(31)	1.000	C(25)	H(32)	1.000
C(26)	H(33)	1.000	C(26)	H(34)	1.000
C(26)	H(35)	1.000	C(27)	H(36)	1.030
C(27)	H(37)	1.030	C(28)	H(38)	1.060
C(28)	H(39)	1.060	C(28)	H(40)	1.060
C(29)	H(41)	1.000	C(29)	H(42)	1.000
C(29)	H(43)	1.000	C(30)	H(44)	1.070
C(30)	H(45)	1.070	C(30)	H(46)	1.070

Table 6. Bond angles (°)

atom	atom	atom	angle	atom	atom	atom	angle
N(1)	V(1)	N(2)	76.37(5)	N(1)	V(1)	N(3)	97.60(6)
N(1)	V(1)	C(23)	125.11(6)	N(1)	V(1)	C(27)	114.21(6)
N(2)	V(1)	N(3)	173.78(5)	N(2)	V(1)	C(23)	82.32(6)
N(2)	V(1)	C(27)	86.23(6)	N(3)	V(1)	C(23)	100.24(7)
N(3)	V(1)	C(27)	97.78(7)	C(23)	V(1)	C(27)	114.08(6)
C(23)	Si(1)	C(24)	114.25(9)	C(23)	Si(1)	C(25)	111.59(10)
C(23)	Si(1)	C(26)	108.65(10)	C(24)	Si(1)	C(25)	106.91(10)
C(24)	Si(1)	C(26)	106.54(10)	C(25)	Si(1)	C(26)	108.66(13)
C(27)	Si(2)	C(28)	110.05(12)	C(27)	Si(2)	C(29)	110.19(9)
C(27)	Si(2)	C(30)	111.02(12)	C(28)	Si(2)	C(29)	105.30(12)
C(28)	Si(2)	C(30)	109.51(15)	C(29)	Si(2)	C(30)	110.61(15)
V(1)	N(1)	C(1)	123.04(11)	V(1)	N(1)	C(7)	127.08(10)
C(1)	N(1)	C(7)	109.88(12)	V(1)	N(2)	C(2)	113.03(10)
V(1)	N(2)	C(6)	129.07(10)	C(2)	N(2)	C(6)	117.67(13)
V(1)	N(3)	C(15)	176.96(11)	N(1)	C(1)	C(2)	111.26(14)
N(2)	C(2)	C(1)	115.30(14)	N(2)	C(2)	C(3)	122.34(15)
C(1)	C(2)	C(3)	122.33(15)	C(2)	C(3)	C(4)	119.15(17)
C(3)	C(4)	C(5)	119.19(18)	C(4)	C(5)	C(6)	118.31(17)
N(2)	C(6)	C(5)	123.32(15)	N(1)	C(7)	C(8)	119.21(17)
N(1)	C(7)	C(12)	121.04(17)	C(8)	C(7)	C(12)	119.74(17)
C(7)	C(8)	C(9)	118.5(2)	C(7)	C(8)	C(13)	121.1(2)
C(9)	C(8)	C(13)	120.4(2)	C(8)	C(9)	C(10)	121.7(2)
C(9)	C(10)	C(11)	119.5(2)	C(10)	C(11)	C(12)	121.2(2)
C(7)	C(12)	C(11)	119.36(19)	C(7)	C(12)	C(14)	120.66(16)
C(11)	C(12)	C(14)	119.98(18)	N(3)	C(15)	C(16)	119.90(19)
N(3)	C(15)	C(20)	120.43(18)	C(16)	C(15)	C(20)	119.67(18)
C(15)	C(16)	C(17)	117.5(2)	C(15)	C(16)	C(21)	121.8(2)
C(17)	C(16)	C(21)	120.8(2)	C(16)	C(17)	C(18)	122.2(2)
C(17)	C(18)	C(19)	118.6(2)	C(18)	C(19)	C(20)	122.8(3)
C(15)	C(20)	C(19)	119.3(2)	C(15)	C(20)	C(22)	120.8(2)
C(19)	C(20)	C(22)	119.9(2)	V(1)	C(23)	Si(1)	130.70(8)
V(1)	C(27)	Si(2)	120.65(9)				

Table 7. Bond angles involving hydrogens (°)

atom	atom	atom	angle	atom	atom	atom	angle
N(1)	C(1)	H(1)	109.0	N(1)	C(1)	H(2)	109.0
C(2)	C(1)	H(1)	109.0	C(2)	C(1)	H(2)	109.0
H(1)	C(1)	H(2)	109.5	C(2)	C(3)	H(3)	120.4
C(4)	C(3)	H(3)	120.4	C(3)	C(4)	H(4)	120.4
C(5)	C(4)	H(4)	120.4	C(4)	C(5)	H(5)	120.8
C(6)	C(5)	H(5)	120.8	N(2)	C(6)	H(6)	118.3
C(5)	C(6)	H(6)	118.3	C(8)	C(9)	H(7)	119.2
C(10)	C(9)	H(7)	119.2	C(9)	C(10)	H(8)	120.3
C(11)	C(10)	H(8)	120.3	C(10)	C(11)	H(9)	119.4
C(12)	C(11)	H(9)	119.4	C(8)	C(13)	H(10)	109.5
C(8)	C(13)	H(11)	109.5	C(8)	C(13)	H(12)	109.5
H(10)	C(13)	H(11)	109.5	H(10)	C(13)	H(12)	109.5
H(11)	C(13)	H(12)	109.5	C(12)	C(14)	H(13)	109.5
C(12)	C(14)	H(14)	109.5	C(12)	C(14)	H(15)	109.5
H(13)	C(14)	H(14)	109.5	H(13)	C(14)	H(15)	109.5
H(14)	C(14)	H(15)	109.5	C(16)	C(17)	H(16)	118.9
C(18)	C(17)	H(16)	118.9	C(17)	C(18)	H(17)	120.7
C(19)	C(18)	H(17)	120.7	C(18)	C(19)	H(18)	118.6
C(20)	C(19)	H(18)	118.6	C(16)	C(21)	H(19)	109.5
C(16)	C(21)	H(20)	109.5	C(16)	C(21)	H(21)	109.5
H(19)	C(21)	H(20)	109.5	H(19)	C(21)	H(21)	109.5
H(20)	C(21)	H(21)	109.5	C(20)	C(22)	H(22)	109.5
C(20)	C(22)	H(23)	109.5	C(20)	C(22)	H(24)	109.5
H(22)	C(22)	H(23)	109.5	H(22)	C(22)	H(24)	109.5
H(23)	C(22)	H(24)	109.5	V(1)	C(23)	H(25)	103.9
V(1)	C(23)	H(26)	103.9	Si(1)	C(23)	H(25)	103.9
Si(1)	C(23)	H(26)	103.9	H(25)	C(23)	H(26)	109.5
Si(1)	C(24)	H(27)	109.5	Si(1)	C(24)	H(28)	109.5
Si(1)	C(24)	H(29)	109.5	H(27)	C(24)	H(28)	109.5
H(27)	C(24)	H(29)	109.5	H(28)	C(24)	H(29)	109.5
Si(1)	C(25)	H(30)	109.5	Si(1)	C(25)	H(31)	109.5
Si(1)	C(25)	H(32)	109.5	H(30)	C(25)	H(31)	109.5
H(30)	C(25)	H(32)	109.5	H(31)	C(25)	H(32)	109.5
Si(1)	C(26)	H(33)	109.5	Si(1)	C(26)	H(34)	109.5
Si(1)	C(26)	H(35)	109.5	H(33)	C(26)	H(34)	109.5
H(33)	C(26)	H(35)	109.5	H(34)	C(26)	H(35)	109.5
V(1)	C(27)	H(36)	106.6	V(1)	C(27)	H(37)	106.6

Table 7. Bond angles involving hydrogens ($^{\circ}$) (continued)

atom	atom	atom	angle	atom	atom	atom	angle
Si(2)	C(27)	H(36)	106.6	Si(2)	C(27)	H(37)	106.6
H(36)	C(27)	H(37)	109.5	Si(2)	C(28)	H(38)	109.5
Si(2)	C(28)	H(39)	109.5	Si(2)	C(28)	H(40)	109.5
H(38)	C(28)	H(39)	109.5	H(38)	C(28)	H(40)	109.5
H(39)	C(28)	H(40)	109.5	Si(2)	C(29)	H(41)	109.5
Si(2)	C(29)	H(42)	109.5	Si(2)	C(29)	H(43)	109.5
H(41)	C(29)	H(42)	109.5	H(41)	C(29)	H(43)	109.5
H(42)	C(29)	H(43)	109.5	Si(2)	C(30)	H(44)	109.5
Si(2)	C(30)	H(45)	109.5	Si(2)	C(30)	H(46)	109.5
H(44)	C(30)	H(45)	109.5	H(44)	C(30)	H(46)	109.5
H(45)	C(30)	H(46)	109.5				

Table 8. Torsion Angles(°)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
N(1)	V(1)	N(2)	C(2)	7.37(13)	N(1)	V(1)	N(2)	C(6)	-178.41(17)
N(2)	V(1)	N(1)	C(1)	-9.50(14)	N(2)	V(1)	N(1)	C(7)	170.87(16)
N(1)	V(1)	N(3)	C(15)	-12(2)	N(3)	V(1)	N(1)	C(1)	168.94(15)
N(3)	V(1)	N(1)	C(7)	-10.68(16)	N(1)	V(1)	C(23)	Si(1)	-3.02(16)
C(23)	V(1)	N(1)	C(1)	60.70(17)	C(23)	V(1)	N(1)	C(7)	-118.92(15)
N(1)	V(1)	C(27)	Si(2)	-173.09(9)	C(27)	V(1)	N(1)	C(1)	-88.95(16)
C(27)	V(1)	N(1)	C(7)	91.43(16)	N(2)	V(1)	N(3)	C(15)	2(2)
N(3)	V(1)	N(2)	C(2)	-7.0(7)	N(3)	V(1)	N(2)	C(6)	167.2(5)
N(2)	V(1)	C(23)	Si(1)	64.30(11)	C(23)	V(1)	N(2)	C(2)	-121.68(14)
C(23)	V(1)	N(2)	C(6)	52.55(17)	N(2)	V(1)	C(27)	Si(2)	113.68(10)
C(27)	V(1)	N(2)	C(2)	123.40(14)	C(27)	V(1)	N(2)	C(6)	-62.38(17)
N(3)	V(1)	C(23)	Si(1)	-109.95(12)	C(23)	V(1)	N(3)	C(15)	115(2)
N(3)	V(1)	C(27)	Si(2)	-71.09(11)	C(27)	V(1)	N(3)	C(15)	-128(2)
C(23)	V(1)	C(27)	Si(2)	33.83(13)	C(27)	V(1)	C(23)	Si(1)	146.66(11)
C(24)	Si(1)	C(23)	V(1)	-28.67(16)	C(25)	Si(1)	C(23)	V(1)	92.75(14)
C(26)	Si(1)	C(23)	V(1)	-147.46(13)	C(28)	Si(2)	C(27)	V(1)	170.38(11)
C(29)	Si(2)	C(27)	V(1)	-73.93(14)	C(30)	Si(2)	C(27)	V(1)	48.97(18)
V(1)	N(1)	C(1)	C(2)	10.2(2)	V(1)	N(1)	C(7)	C(8)	-100.68(18)
V(1)	N(1)	C(7)	C(12)	80.6(2)	C(1)	N(1)	C(7)	C(8)	79.7(2)
C(1)	N(1)	C(7)	C(12)	-99.07(18)	C(7)	N(1)	C(1)	C(2)	-170.13(17)
V(1)	N(2)	C(2)	C(1)	-4.2(2)	V(1)	N(2)	C(2)	C(3)	173.50(17)
V(1)	N(2)	C(6)	C(5)	-172.20(16)	C(2)	N(2)	C(6)	C(5)	1.8(3)
C(6)	N(2)	C(2)	C(1)	-179.15(18)	C(6)	N(2)	C(2)	C(3)	-1.4(3)
V(1)	N(3)	C(15)	C(16)	82(2)	V(1)	N(3)	C(15)	C(20)	-98(2)
N(1)	C(1)	C(2)	N(2)	-2.5(2)	N(1)	C(1)	C(2)	C(3)	179.8(2)
N(2)	C(2)	C(3)	C(4)	0.1(2)	C(1)	C(2)	C(3)	C(4)	177.6(2)
C(2)	C(3)	C(4)	C(5)	1.0(3)	C(3)	C(4)	C(5)	C(6)	-0.7(3)
C(4)	C(5)	C(6)	N(2)	-0.8(3)	N(1)	C(7)	C(8)	C(9)	-178.66(19)
N(1)	C(7)	C(8)	C(13)	1.9(2)	N(1)	C(7)	C(12)	C(11)	179.48(15)
N(1)	C(7)	C(12)	C(14)	-0.1(2)	C(8)	C(7)	C(12)	C(11)	0.8(2)
C(8)	C(7)	C(12)	C(14)	-178.78(17)	C(12)	C(7)	C(8)	C(9)	0.1(2)
C(12)	C(7)	C(8)	C(13)	-179.4(2)	C(7)	C(8)	C(9)	C(10)	-0.7(3)
C(13)	C(8)	C(9)	C(10)	178.8(2)	C(8)	C(9)	C(10)	C(11)	0.5(4)
C(9)	C(10)	C(11)	C(12)	0.4(3)	C(10)	C(11)	C(12)	C(7)	-1.0(2)
C(10)	C(11)	C(12)	C(14)	178.5(2)	N(3)	C(15)	C(16)	C(17)	-179.35(18)
N(3)	C(15)	C(16)	C(21)	2.2(2)	N(3)	C(15)	C(20)	C(19)	178.8(2)
N(3)	C(15)	C(20)	C(22)	-2.5(3)	C(16)	C(15)	C(20)	C(19)	-0.6(3)

Table 8. Torsion angles ($^{\circ}$) (continued)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
C(16)	C(15)	C(20)	C(22)	178.0(2)	C(20)	C(15)	C(16)	C(17)	0.1(2)
C(20)	C(15)	C(16)	C(21)	-178.4(2)	C(15)	C(16)	C(17)	C(18)	0.3(3)
C(21)	C(16)	C(17)	C(18)	178.8(2)	C(16)	C(17)	C(18)	C(19)	-0.2(3)
C(17)	C(18)	C(19)	C(20)	-0.4(4)	C(18)	C(19)	C(20)	C(15)	0.8(4)
C(18)	C(19)	C(20)	C(22)	-177.9(2)					

The sign is positive if when looking from atom 2 to atom 3 a clock-wise motion of atom 1 would superimpose it on atom 4.

Table 9. Distances beyond the asymmetric unit out to 3.60 Å

atom	atom	distance	atom	atom	distance
V(1)	Si(1)	3.5917(6)	V(1)	Si(2)	3.4294(5)
V(1)	C(1)	2.9621(17)	V(1)	C(2)	3.0565(16)
V(1)	C(6)	3.2986(18)	V(1)	C(7)	2.9849(16)
V(1)	C(15)	3.0721(18)	Si(1)	V(1)	3.5917(6)
Si(2)	V(1)	3.4294(5)	N(1)	N(2)	2.5982(19)
N(1)	N(3)	2.6985(18)	N(1)	C(2)	2.438(2)
N(1)	C(8)	2.445(2)	N(1)	C(12)	2.457(2)
N(1)	C(13)	2.884(4)	N(1)	C(14)	2.901(2)
N(1)	C(23)	3.530(2)	N(1)	C(27)	3.349(2)
N(2)	N(1)	2.5982(19)	N(2)	C(1)	2.391(2)
N(2)	C(3)	2.392(2)	N(2)	C(4)	2.775(2)
N(2)	C(5)	2.396(2)	N(2)	C(23)	2.868(2)
N(2)	C(24)	3.325(2)	N(2)	C(27)	2.984(2)
N(3)	N(1)	2.6985(18)	N(3)	C(7)	3.040(2)
N(3)	C(12)	3.381(2)	N(3)	C(14)	3.445(2)
N(3)	C(16)	2.431(2)	N(3)	C(20)	2.432(2)
N(3)	C(21)	2.884(4)	N(3)	C(22)	2.886(3)
N(3)	C(23)	2.894(2)	N(3)	C(27)	2.851(2)
N(3)	C(30)	3.441(3)	C(1)	V(1)	2.9621(17)
C(1)	N(2)	2.391(2)	C(1)	C(3)	2.528(2)
C(1)	C(7)	2.362(2)	C(1)	C(8)	3.099(3)
C(1)	C(12)	3.289(2)	C(1)	C(13)	3.267(4)
C(2)	V(1)	3.0565(16)	C(2)	N(1)	2.438(2)
C(2)	C(4)	2.381(2)	C(2)	C(5)	2.726(2)
C(2)	C(6)	2.300(2)	C(2)	C(24)	3.304(3)
C(3)	N(2)	2.392(2)	C(3)	C(1)	2.528(2)
C(3)	C(5)	2.381(2)	C(3)	C(6)	2.710(2)
C(4)	N(2)	2.775(2)	C(4)	C(2)	2.381(2)
C(4)	C(6)	2.372(2)	C(5)	N(2)	2.396(2)
C(5)	C(2)	2.726(2)	C(5)	C(3)	2.381(2)
C(6)	V(1)	3.2986(18)	C(6)	C(2)	2.300(2)
C(6)	C(3)	2.710(2)	C(6)	C(4)	2.372(2)
C(6)	C(23)	3.291(2)	C(6)	C(27)	3.514(2)
C(7)	V(1)	2.9849(16)	C(7)	N(3)	3.040(2)
C(7)	C(1)	2.362(2)	C(7)	C(9)	2.401(3)
C(7)	C(10)	2.782(2)	C(7)	C(11)	2.405(2)
C(7)	C(13)	2.540(4)	C(7)	C(14)	2.524(3)

Table 9. Distances beyond the asymmetric unit out to 3.60 Å (continued)

atom	atom	distance	atom	atom	distance
C(8)	N(1)	2.445(2)	C(8)	C(1)	3.099(3)
C(8)	C(10)	2.414(3)	C(8)	C(11)	2.778(3)
C(8)	C(12)	2.428(3)	C(8)	C(21)	3.521(3)
C(9)	C(7)	2.401(3)	C(9)	C(11)	2.371(4)
C(9)	C(12)	2.764(3)	C(9)	C(13)	2.508(4)
C(10)	C(7)	2.782(2)	C(10)	C(8)	2.414(3)
C(10)	C(12)	2.400(3)	C(11)	C(7)	2.405(2)
C(11)	C(8)	2.778(3)	C(11)	C(9)	2.371(4)
C(11)	C(14)	2.509(3)	C(12)	N(1)	2.457(2)
C(12)	N(3)	3.381(2)	C(12)	C(1)	3.289(2)
C(12)	C(8)	2.428(3)	C(12)	C(9)	2.764(3)
C(12)	C(10)	2.400(3)	C(13)	N(1)	2.884(4)
C(13)	C(1)	3.267(4)	C(13)	C(7)	2.540(4)
C(13)	C(9)	2.508(4)	C(14)	N(1)	2.901(2)
C(14)	N(3)	3.445(2)	C(14)	C(7)	2.524(3)
C(14)	C(11)	2.509(3)	C(15)	V(1)	3.0721(18)
C(15)	C(17)	2.416(3)	C(15)	C(18)	2.806(3)
C(15)	C(19)	2.411(3)	C(15)	C(21)	2.520(4)
C(15)	C(22)	2.533(4)	C(16)	N(3)	2.431(2)
C(16)	C(18)	2.453(4)	C(16)	C(19)	2.788(4)
C(16)	C(20)	2.444(3)	C(17)	C(15)	2.416(3)
C(17)	C(19)	2.355(5)	C(17)	C(20)	2.765(4)
C(17)	C(21)	2.503(4)	C(18)	C(15)	2.806(3)
C(18)	C(16)	2.453(4)	C(18)	C(20)	2.400(3)
C(19)	C(15)	2.411(3)	C(19)	C(16)	2.788(4)
C(19)	C(17)	2.355(5)	C(19)	C(22)	2.501(4)
C(20)	N(3)	2.432(2)	C(20)	C(16)	2.444(3)
C(20)	C(17)	2.765(4)	C(20)	C(18)	2.400(3)
C(21)	N(3)	2.884(4)	C(21)	C(8)	3.521(3)
C(21)	C(15)	2.520(4)	C(21)	C(17)	2.503(4)
C(22)	N(3)	2.886(3)	C(22)	C(15)	2.533(4)
C(22)	C(19)	2.501(4)	C(22)	C(23)	3.559(3)
C(22)	C(25)	3.550(4)	C(23)	N(1)	3.530(2)
C(23)	N(2)	2.868(2)	C(23)	N(3)	2.894(2)
C(23)	C(6)	3.291(2)	C(23)	C(22)	3.559(3)
C(23)	C(24)	3.149(2)	C(23)	C(25)	3.104(3)
C(23)	C(26)	3.046(3)	C(23)	C(27)	3.492(2)

Table 9. Distances beyond the asymmetric unit out to 3.60 Å (continued)

atom	atom	distance	atom	atom	distance
C(24)	N(2)	3.325(2)	C(24)	C(2)	3.304(3)
C(24)	C(23)	3.149(2)	C(24)	C(25)	3.016(3)
C(24)	C(26)	3.005(3)	C(25)	C(22)	3.550(4)
C(25)	C(23)	3.104(3)	C(25)	C(24)	3.016(3)
C(25)	C(26)	3.049(3)	C(26)	C(23)	3.046(3)
C(26)	C(24)	3.005(3)	C(26)	C(25)	3.049(3)
C(27)	N(1)	3.349(2)	C(27)	N(2)	2.984(2)
C(27)	N(3)	2.851(2)	C(27)	C(6)	3.514(2)
C(27)	C(23)	3.492(2)	C(27)	C(28)	3.056(2)
C(27)	C(29)	3.077(3)	C(27)	C(30)	3.054(4)
C(28)	C(27)	3.056(2)	C(28)	C(29)	2.993(4)
C(28)	C(30)	3.036(5)	C(29)	C(27)	3.077(3)
C(29)	C(28)	2.993(4)	C(29)	C(30)	3.076(4)
C(30)	N(3)	3.441(3)	C(30)	C(27)	3.054(4)
C(30)	C(28)	3.036(5)	C(30)	C(29)	3.076(4)

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens

atom	atom	distance	atom	atom	distance
V(1)	H(2)	3.547	V(1)	H(6)	3.378
V(1)	H(10)	3.320	V(1)	H(13)	3.458
V(1)	H(15)	3.519	V(1)	H(21)	3.483
V(1)	H(22)	3.439	V(1)	H(25)	2.518
V(1)	H(26)	2.518	V(1)	H(36)	2.577
V(1)	H(37)	2.577	V(1)	H(46)	3.449
Si(1)	H(22)	3.484	Si(1)	H(25)	2.334
Si(1)	H(26)	2.334	Si(1)	H(27)	2.388
Si(1)	H(28)	2.388	Si(1)	H(29)	2.388
Si(1)	H(30)	2.404	Si(1)	H(31)	2.404
Si(1)	H(32)	2.404	Si(1)	H(33)	2.401
Si(1)	H(34)	2.401	Si(1)	H(35)	2.401
Si(2)	H(36)	2.368	Si(2)	H(37)	2.368
Si(2)	H(38)	2.438	Si(2)	H(39)	2.438
Si(2)	H(40)	2.438	Si(2)	H(41)	2.418
Si(2)	H(42)	2.418	Si(2)	H(43)	2.418
Si(2)	H(44)	2.423	Si(2)	H(45)	2.423
Si(2)	H(46)	2.423	N(1)	H(1)	2.043
N(1)	H(2)	2.044	N(1)	H(10)	2.474
N(1)	H(12)	3.290	N(1)	H(13)	2.577
N(1)	H(15)	3.163	N(1)	H(27)	3.495
N(1)	H(36)	3.441	N(1)	H(37)	3.541
N(2)	H(1)	3.072	N(2)	H(2)	3.043
N(2)	H(3)	3.305	N(2)	H(5)	3.303
N(2)	H(6)	2.011	N(2)	H(7) ¹⁾	3.393
N(2)	H(26)	2.791	N(2)	H(27)	3.310
N(2)	H(29)	2.776	N(2)	H(37)	2.582
N(3)	H(13)	3.582	N(3)	H(15)	2.908
N(3)	H(20)	3.046	N(3)	H(21)	2.646
N(3)	H(22)	2.638	N(3)	H(24)	3.061
N(3)	H(25)	2.718	N(3)	H(26)	3.538
N(3)	H(36)	2.908	N(3)	H(44)	3.303
N(3)	H(46)	2.940	C(1)	H(1) ²⁾	3.511
C(1)	H(2) ²⁾	3.021	C(1)	H(3)	2.735
C(1)	H(3) ²⁾	3.313	C(1)	H(10)	2.936
C(1)	H(12)	3.246	C(1)	H(13)	3.018
C(1)	H(27)	3.101	C(1)	H(29)	3.552

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(2)	H(1)	2.071	C(2)	H(2)	2.071
C(2)	H(3)	2.103	C(2)	H(4)	3.292
C(2)	H(6)	3.182	C(2)	H(10)	3.564
C(2)	H(27)	3.035	C(2)	H(29)	2.753
C(2)	H(37)	3.536	C(3)	H(1)	2.791
C(3)	H(2)	2.794	C(3)	H(2) ²⁾	3.470
C(3)	H(4)	2.063	C(3)	H(5)	3.300
C(3)	H(11) ¹⁾	3.320	C(3)	H(27)	3.535
C(3)	H(29)	2.904	C(4)	H(3)	2.079
C(4)	H(5)	2.098	C(4)	H(6)	3.255
C(4)	H(8) ³⁾	3.268	C(4)	H(9) ³⁾	3.155
C(4)	H(11) ¹⁾	2.977	C(4)	H(29)	3.052
C(4)	H(40) ⁴⁾	3.469	C(5)	H(3)	3.311
C(5)	H(4)	2.086	C(5)	H(6)	2.028
C(5)	H(11) ¹⁾	3.087	C(5)	H(17) ³⁾	3.495
C(5)	H(29)	3.071	C(5)	H(40) ⁴⁾	3.461
C(6)	H(4)	3.283	C(6)	H(5)	2.078
C(6)	H(7) ¹⁾	3.549	C(6)	H(11) ¹⁾	3.519
C(6)	H(17) ³⁾	3.339	C(6)	H(26)	2.804
C(6)	H(29)	2.923	C(6)	H(37)	2.906
C(6)	H(42)	3.489	C(6)	H(43)	3.532
C(7)	H(1)	2.503	C(7)	H(2)	2.653
C(7)	H(3) ²⁾	3.076	C(7)	H(7)	3.353
C(7)	H(9)	3.327	C(7)	H(10)	2.601
C(7)	H(11)	3.242	C(7)	H(12)	3.023
C(7)	H(13)	2.625	C(7)	H(14)	3.257
C(7)	H(15)	2.921	C(7)	H(20)	3.012
C(8)	H(1)	2.757	C(8)	H(2)	3.578
C(8)	H(3) ²⁾	3.597	C(8)	H(7)	2.106
C(8)	H(8)	3.314	C(8)	H(10)	2.038
C(8)	H(11)	2.038	C(8)	H(12)	2.038
C(8)	H(20)	2.618	C(9)	H(8)	2.073
C(9)	H(9)	3.293	C(9)	H(10)	3.260
C(9)	H(10) ⁵⁾	3.583	C(9)	H(11)	2.631
C(9)	H(12)	2.883	C(9)	H(20)	2.894
C(9)	H(37) ⁵⁾	3.139	C(10)	H(3) ²⁾	3.569
C(10)	H(4) ⁶⁾	3.018	C(10)	H(4) ²⁾	3.200

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(10)	H(7)	2.103	C(10)	H(9)	2.066
C(10)	H(9) ⁷⁾	3.406	C(10)	H(10) ⁵⁾	3.522
C(10)	H(20)	3.470	C(11)	H(3) ²⁾	3.064
C(11)	H(4) ⁶⁾	2.913	C(11)	H(4) ²⁾	3.072
C(11)	H(5) ⁶⁾	3.597	C(11)	H(7)	3.317
C(11)	H(8)	2.058	C(11)	H(8) ⁷⁾	3.573
C(11)	H(13)	3.220	C(11)	H(14)	2.578
C(11)	H(15)	2.958	C(12)	H(1)	3.575
C(12)	H(2)	3.160	C(12)	H(3) ²⁾	2.769
C(12)	H(8)	3.303	C(12)	H(9)	2.089
C(12)	H(13)	2.032	C(12)	H(14)	2.032
C(12)	H(15)	2.031	C(12)	H(20)	3.578
C(13)	H(1)	2.692	C(13)	H(7)	2.684
C(13)	H(8) ¹⁾	3.483	C(13)	H(20)	3.038
C(13)	H(36)	3.486	C(14)	H(2)	3.212
C(14)	H(3) ²⁾	3.206	C(14)	H(9)	2.676
C(14)	H(19) ⁸⁾	3.346	C(14)	H(21) ⁸⁾	3.492
C(14)	H(22)	3.101	C(14)	H(27)	3.599
C(14)	H(32)	3.237	C(14)	H(39) ⁸⁾	3.023
C(15)	H(15)	3.025	C(15)	H(16)	3.331
C(15)	H(18)	3.357	C(15)	H(19)	3.288
C(15)	H(20)	2.876	C(15)	H(21)	2.660
C(15)	H(22)	2.679	C(15)	H(23)	3.304
C(15)	H(24)	2.880	C(15)	H(25)	3.566
C(15)	H(35) ⁹⁾	3.565	C(15)	H(44)	3.310
C(15)	H(46)	3.266	C(16)	H(16)	2.095
C(16)	H(17)	3.439	C(16)	H(19)	2.011
C(16)	H(20)	2.011	C(16)	H(21)	2.011
C(16)	H(44)	3.419	C(17)	H(5) ⁶⁾	2.942
C(17)	H(17)	2.161	C(17)	H(18)	3.305
C(17)	H(19)	2.549	C(17)	H(20)	3.022
C(17)	H(21)	3.199	C(17)	H(28) ¹⁰⁾	3.251
C(17)	H(31) ¹⁰⁾	3.253	C(18)	H(5) ⁶⁾	2.804
C(18)	H(6) ⁶⁾	3.290	C(18)	H(16)	2.077
C(18)	H(18)	2.067	C(18)	H(31) ¹⁰⁾	3.507
C(18)	H(35) ⁹⁾	3.340	C(18)	H(42) ⁶⁾	3.544
C(19)	H(5) ⁶⁾	3.375	C(19)	H(16)	3.263

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(19)	H(17)	2.122	C(19)	H(22)	3.202
C(19)	H(23)	2.555	C(19)	H(24)	3.009
C(19)	H(35) ⁹⁾	2.880	C(20)	H(15)	2.995
C(20)	H(17)	3.386	C(20)	H(18)	2.100
C(20)	H(22)	2.042	C(20)	H(23)	2.042
C(20)	H(24)	2.042	C(20)	H(25)	3.596
C(20)	H(35) ⁹⁾	2.990	C(21)	H(11)	3.567
C(21)	H(13) ¹⁰⁾	3.560	C(21)	H(14) ¹⁰⁾	3.307
C(21)	H(16)	2.677	C(21)	H(36)	3.058
C(21)	H(44)	3.513	C(22)	H(15)	2.917
C(22)	H(18)	2.665	C(22)	H(25)	2.746
C(22)	H(30)	3.157	C(22)	H(32)	3.151
C(22)	H(35) ⁹⁾	3.448	C(22)	H(41) ⁹⁾	3.596
C(22)	H(43) ⁹⁾	3.216	C(23)	H(6)	3.114
C(23)	H(18) ¹¹⁾	3.534	C(23)	H(22)	2.967
C(23)	H(24)	3.297	C(23)	H(27)	3.345
C(23)	H(29)	3.297	C(23)	H(30)	3.263
C(23)	H(32)	3.281	C(23)	H(33)	3.191
C(23)	H(35)	3.213	C(23)	H(43)	3.152
C(23)	H(46)	3.396	C(24)	H(2)	3.345
C(24)	H(13)	3.523	C(24)	H(16) ⁸⁾	3.098
C(24)	H(26)	3.530	C(24)	H(31)	3.130
C(24)	H(32)	3.201	C(24)	H(33)	3.194
C(24)	H(34)	3.114	C(24)	H(40) ⁴⁾	3.395
C(25)	H(22)	2.836	C(25)	H(25)	2.990
C(25)	H(27)	3.164	C(25)	H(28)	3.153
C(25)	H(31) ¹²⁾	3.351	C(25)	H(34)	3.203
C(25)	H(34) ¹²⁾	3.460	C(25)	H(35)	3.206
C(26)	H(24) ¹¹⁾	3.558	C(26)	H(25)	3.390
C(26)	H(26)	2.888	C(26)	H(28)	3.121
C(26)	H(29)	3.174	C(26)	H(30)	3.213
C(26)	H(31)	3.198	C(26)	H(31) ¹²⁾	3.461
C(26)	H(38) ⁴⁾	3.572	C(26)	H(41) ⁴⁾	3.462
C(26)	H(46) ¹¹⁾	3.444	C(27)	H(6)	3.289
C(27)	H(7) ¹⁾	3.103	C(27)	H(10)	3.164
C(27)	H(21)	2.974	C(27)	H(26)	3.351
C(27)	H(39)	3.220	C(27)	H(40)	3.257

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(27)	H(42)	3.224	C(27)	H(43)	3.262
C(27)	H(44)	3.239	C(27)	H(46)	3.246
C(28)	H(14) ¹⁰⁾	3.506	C(28)	H(36)	3.096
C(28)	H(37)	3.252	C(28)	H(41)	3.129
C(28)	H(42)	3.147	C(28)	H(44)	3.235
C(28)	H(45)	3.199	C(29)	H(6)	2.991
C(29)	H(17) ³⁾	3.205	C(29)	H(23) ¹¹⁾	3.061
C(29)	H(26)	3.125	C(29)	H(34) ¹³⁾	3.234
C(29)	H(37)	3.106	C(29)	H(38)	3.182
C(29)	H(40)	3.121	C(29)	H(45)	3.239
C(29)	H(45) ¹⁴⁾	3.391	C(29)	H(46)	3.276
C(30)	H(34) ⁹⁾	3.443	C(30)	H(36)	3.295
C(30)	H(38)	3.182	C(30)	H(39)	3.257
C(30)	H(41)	3.239	C(30)	H(41) ¹⁴⁾	3.448
C(30)	H(43)	3.249	C(30)	H(45) ¹⁴⁾	3.524
H(1)	N(1)	2.043	H(1)	N(2)	3.072
H(1)	C(1) ²⁾	3.511	H(1)	C(2)	2.071
H(1)	C(3)	2.791	H(1)	C(7)	2.503
H(1)	C(8)	2.757	H(1)	C(12)	3.575
H(1)	C(13)	2.692	H(1)	H(2)	1.682
H(1)	H(2) ²⁾	2.645	H(1)	H(3)	2.758
H(1)	H(3) ²⁾	3.238	H(1)	H(10)	2.558
H(1)	H(12)	2.426	H(1)	H(19) ¹⁾	3.523
H(2)	V(1)	3.547	H(2)	N(1)	2.044
H(2)	N(2)	3.043	H(2)	C(1) ²⁾	3.021
H(2)	C(2)	2.071	H(2)	C(3)	2.794
H(2)	C(3) ²⁾	3.470	H(2)	C(7)	2.653
H(2)	C(8)	3.578	H(2)	C(12)	3.160
H(2)	C(14)	3.212	H(2)	C(24)	3.345
H(2)	H(1)	1.682	H(2)	H(1) ²⁾	2.645
H(2)	H(2) ²⁾	2.573	H(2)	H(3)	2.785
H(2)	H(3) ²⁾	2.670	H(2)	H(13)	2.499
H(2)	H(27)	2.540	H(2)	H(29)	3.331
H(3)	N(2)	3.305	H(3)	C(1)	2.735
H(3)	C(1) ²⁾	3.313	H(3)	C(2)	2.103
H(3)	C(4)	2.079	H(3)	C(5)	3.311
H(3)	C(7) ²⁾	3.076	H(3)	C(8) ²⁾	3.597

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(3)	C(10) ²⁾	3.569	H(3)	C(11) ²⁾	3.064
H(3)	C(12) ²⁾	2.769	H(3)	C(14) ²⁾	3.206
H(3)	H(1)	2.758	H(3)	H(1) ²⁾	3.238
H(3)	H(2)	2.785	H(3)	H(2) ²⁾	2.670
H(3)	H(4)	2.390	H(3)	H(9) ²⁾	3.468
H(3)	H(13) ²⁾	3.074	H(3)	H(14) ²⁾	3.166
H(3)	H(29)	3.497	H(4)	C(2)	3.292
H(4)	C(3)	2.063	H(4)	C(5)	2.086
H(4)	C(6)	3.283	H(4)	C(10) ³⁾	3.018
H(4)	C(10) ²⁾	3.200	H(4)	C(11) ³⁾	2.913
H(4)	C(11) ²⁾	3.072	H(4)	H(3)	2.390
H(4)	H(5)	2.417	H(4)	H(8) ³⁾	2.520
H(4)	H(8) ²⁾	3.236	H(4)	H(9) ³⁾	2.272
H(4)	H(9) ²⁾	3.018	H(4)	H(11) ¹⁾	3.249
H(4)	H(40) ⁴⁾	3.420	H(5)	N(2)	3.303
H(5)	C(3)	3.300	H(5)	C(4)	2.098
H(5)	C(6)	2.078	H(5)	C(11) ³⁾	3.597
H(5)	C(17) ³⁾	2.942	H(5)	C(18) ³⁾	2.804
H(5)	C(19) ³⁾	3.375	H(5)	H(4)	2.417
H(5)	H(6)	2.353	H(5)	H(9) ³⁾	3.249
H(5)	H(11) ¹⁾	3.420	H(5)	H(16) ³⁾	3.125
H(5)	H(17) ³⁾	2.929	H(5)	H(39) ⁴⁾	3.564
H(5)	H(40) ⁴⁾	3.418	H(6)	V(1)	3.378
H(6)	N(2)	2.011	H(6)	C(2)	3.182
H(6)	C(4)	3.255	H(6)	C(5)	2.028
H(6)	C(18) ³⁾	3.290	H(6)	C(23)	3.114
H(6)	C(27)	3.289	H(6)	C(29)	2.991
H(6)	H(5)	2.353	H(6)	H(16) ³⁾	3.583
H(6)	H(17) ³⁾	2.579	H(6)	H(26)	2.380
H(6)	H(29)	3.506	H(6)	H(33)	3.461
H(6)	H(37)	2.759	H(6)	H(42)	2.603
H(6)	H(43)	2.620	H(7)	N(2) ⁵⁾	3.393
H(7)	C(6) ⁵⁾	3.549	H(7)	C(7)	3.353
H(7)	C(8)	2.106	H(7)	C(10)	2.103
H(7)	C(11)	3.317	H(7)	C(13)	2.684
H(7)	C(27) ⁵⁾	3.103	H(7)	H(8)	2.397
H(7)	H(10)	3.569	H(7)	H(10) ⁵⁾	3.121

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(7)	H(11)	2.497	H(7)	H(12)	2.936
H(7)	H(20)	3.283	H(7)	H(36) ⁵⁾	3.216
H(7)	H(37) ⁵⁾	2.192	H(8)	C(4) ⁶⁾	3.268
H(8)	C(8)	3.314	H(8)	C(9)	2.073
H(8)	C(11)	2.058	H(8)	C(11) ⁷⁾	3.573
H(8)	C(12)	3.303	H(8)	C(13) ⁵⁾	3.483
H(8)	H(4) ⁶⁾	2.520	H(8)	H(4) ²⁾	3.236
H(8)	H(7)	2.397	H(8)	H(9)	2.369
H(8)	H(9) ⁷⁾	2.781	H(8)	H(10) ⁵⁾	3.023
H(8)	H(11) ⁵⁾	3.196	H(8)	H(36) ⁵⁾	3.157
H(8)	H(37) ⁵⁾	3.359	H(9)	C(4) ⁶⁾	3.155
H(9)	C(7)	3.327	H(9)	C(9)	3.293
H(9)	C(10)	2.066	H(9)	C(10) ⁷⁾	3.406
H(9)	C(12)	2.089	H(9)	C(14)	2.676
H(9)	H(3) ²⁾	3.468	H(9)	H(4) ⁶⁾	2.272
H(9)	H(4) ²⁾	3.018	H(9)	H(5) ⁶⁾	3.249
H(9)	H(8)	2.369	H(9)	H(8) ⁷⁾	2.781
H(9)	H(9) ⁷⁾	3.286	H(9)	H(13)	3.497
H(9)	H(14)	2.405	H(9)	H(15)	3.060
H(9)	H(39) ⁸⁾	2.911	H(10)	V(1)	3.320
H(10)	N(1)	2.474	H(10)	C(1)	2.936
H(10)	C(2)	3.564	H(10)	C(7)	2.601
H(10)	C(8)	2.038	H(10)	C(9)	3.260
H(10)	C(9) ¹⁾	3.583	H(10)	C(10) ¹⁾	3.522
H(10)	C(27)	3.164	H(10)	H(1)	2.558
H(10)	H(7)	3.569	H(10)	H(7) ¹⁾	3.121
H(10)	H(8) ¹⁾	3.023	H(10)	H(11)	1.568
H(10)	H(12)	1.568	H(10)	H(20)	3.102
H(10)	H(21)	3.377	H(10)	H(36)	2.680
H(10)	H(37)	3.129	H(11)	C(3) ⁵⁾	3.320
H(11)	C(4) ⁵⁾	2.977	H(11)	C(5) ⁵⁾	3.087
H(11)	C(6) ⁵⁾	3.519	H(11)	C(7)	3.242
H(11)	C(8)	2.038	H(11)	C(9)	2.631
H(11)	C(21)	3.567	H(11)	H(4) ⁵⁾	3.249
H(11)	H(5) ⁵⁾	3.420	H(11)	H(7)	2.497
H(11)	H(8) ¹⁾	3.196	H(11)	H(10)	1.568
H(11)	H(12)	1.568	H(11)	H(20)	2.830

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens
(continued)

atom	atom	distance	atom	atom	distance
H(11)	H(21)	3.525	H(11)	H(36)	3.586
H(12)	N(1)	3.290	H(12)	C(1)	3.246
H(12)	C(7)	3.023	H(12)	C(8)	2.038
H(12)	C(9)	2.883	H(12)	H(1)	2.426
H(12)	H(7)	2.936	H(12)	H(10)	1.568
H(12)	H(11)	1.568	H(12)	H(16) ¹⁾	3.058
H(12)	H(19) ¹⁾	3.375	H(12)	H(20) ¹⁾	3.596
H(13)	V(1)	3.458	H(13)	N(1)	2.577
H(13)	N(3)	3.582	H(13)	C(1)	3.018
H(13)	C(7)	2.625	H(13)	C(11)	3.220
H(13)	C(12)	2.032	H(13)	C(21) ⁸⁾	3.560
H(13)	C(24)	3.523	H(13)	H(2)	2.499
H(13)	H(3) ²⁾	3.074	H(13)	H(9)	3.497
H(13)	H(14)	1.551	H(13)	H(15)	1.551
H(13)	H(19) ⁸⁾	2.841	H(13)	H(21) ⁸⁾	3.402
H(13)	H(22)	3.258	H(13)	H(27)	2.665
H(13)	H(32)	2.900	H(14)	C(7)	3.257
H(14)	C(11)	2.578	H(14)	C(12)	2.032
H(14)	C(21) ⁸⁾	3.307	H(14)	C(28) ⁸⁾	3.506
H(14)	H(3) ²⁾	3.166	H(14)	H(9)	2.405
H(14)	H(13)	1.551	H(14)	H(15)	1.551
H(14)	H(19) ⁸⁾	2.969	H(14)	H(21) ⁸⁾	2.766
H(14)	H(32)	3.466	H(14)	H(36) ⁸⁾	3.285
H(14)	H(39) ⁸⁾	2.475	H(15)	V(1)	3.519
H(15)	N(1)	3.163	H(15)	N(3)	2.908
H(15)	C(7)	2.921	H(15)	C(11)	2.958
H(15)	C(12)	2.031	H(15)	C(15)	3.025
H(15)	C(20)	2.995	H(15)	C(22)	2.917
H(15)	H(9)	3.060	H(15)	H(13)	1.551
H(15)	H(14)	1.551	H(15)	H(22)	2.222
H(15)	H(23)	3.275	H(15)	H(32)	2.856
H(15)	H(39) ⁸⁾	2.775	H(16)	C(15)	3.331
H(16)	C(16)	2.095	H(16)	C(18)	2.077
H(16)	C(19)	3.263	H(16)	C(21)	2.677
H(16)	C(24) ¹⁰⁾	3.098	H(16)	H(5) ⁶⁾	3.125
H(16)	H(6) ⁶⁾	3.583	H(16)	H(12) ⁵⁾	3.058
H(16)	H(17)	2.436	H(16)	H(19)	2.356

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(16)	H(20)	3.168	H(16)	H(21)	3.462
H(16)	H(27) ¹⁰⁾	2.980	H(16)	H(28) ¹⁰⁾	2.392
H(16)	H(31) ¹⁰⁾	3.223	H(16)	H(42) ⁶⁾	3.370
H(17)	C(5) ⁶⁾	3.495	H(17)	C(6) ⁶⁾	3.339
H(17)	C(16)	3.439	H(17)	C(17)	2.161
H(17)	C(19)	2.122	H(17)	C(20)	3.386
H(17)	C(29) ⁶⁾	3.205	H(17)	H(5) ⁶⁾	2.929
H(17)	H(6) ⁶⁾	2.579	H(17)	H(16)	2.436
H(17)	H(18)	2.407	H(17)	H(24) ⁹⁾	3.304
H(17)	H(30) ⁹⁾	3.237	H(17)	H(41) ⁶⁾	3.549
H(17)	H(42) ⁶⁾	2.632	H(17)	H(43) ⁶⁾	2.989
H(18)	C(15)	3.357	H(18)	C(17)	3.305
H(18)	C(18)	2.067	H(18)	C(20)	2.100
H(18)	C(22)	2.665	H(18)	C(23) ⁹⁾	3.534
H(18)	H(17)	2.407	H(18)	H(22)	3.454
H(18)	H(23)	2.335	H(18)	H(24)	3.153
H(18)	H(24) ⁹⁾	3.375	H(18)	H(25) ⁹⁾	2.886
H(18)	H(26) ⁹⁾	3.325	H(18)	H(35) ⁹⁾	3.091
H(18)	H(38) ⁸⁾	3.180	H(18)	H(43) ⁹⁾	3.473
H(18)	H(46) ⁹⁾	3.315	H(19)	C(14) ¹⁰⁾	3.346
H(19)	C(15)	3.288	H(19)	C(16)	2.011
H(19)	C(17)	2.549	H(19)	H(1) ⁵⁾	3.523
H(19)	H(12) ⁵⁾	3.375	H(19)	H(13) ¹⁰⁾	2.841
H(19)	H(14) ¹⁰⁾	2.969	H(19)	H(16)	2.356
H(19)	H(20)	1.584	H(19)	H(21)	1.584
H(19)	H(27) ¹⁰⁾	3.053	H(19)	H(32) ¹⁰⁾	3.557
H(20)	N(3)	3.046	H(20)	C(7)	3.012
H(20)	C(8)	2.618	H(20)	C(9)	2.894
H(20)	C(10)	3.470	H(20)	C(12)	3.578
H(20)	C(13)	3.038	H(20)	C(15)	2.876
H(20)	C(16)	2.011	H(20)	C(17)	3.022
H(20)	H(7)	3.283	H(20)	H(10)	3.102
H(20)	H(11)	2.830	H(20)	H(12) ⁵⁾	3.596
H(20)	H(16)	3.168	H(20)	H(19)	1.584
H(20)	H(21)	1.584	H(20)	H(36)	3.229
H(21)	V(1)	3.483	H(21)	N(3)	2.646
H(21)	C(14) ¹⁰⁾	3.492	H(21)	C(15)	2.660

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(21)	C(16)	2.011	H(21)	C(17)	3.199
H(21)	C(27)	2.974	H(21)	H(10)	3.377
H(21)	H(11)	3.525	H(21)	H(13) ¹⁰⁾	3.402
H(21)	H(14) ¹⁰⁾	2.766	H(21)	H(16)	3.462
H(21)	H(19)	1.584	H(21)	H(20)	1.584
H(21)	H(36)	2.165	H(21)	H(44)	2.864
H(22)	V(1)	3.439	H(22)	Si(1)	3.484
H(22)	N(3)	2.638	H(22)	C(14)	3.101
H(22)	C(15)	2.679	H(22)	C(19)	3.202
H(22)	C(20)	2.042	H(22)	C(23)	2.967
H(22)	C(25)	2.836	H(22)	H(13)	3.258
H(22)	H(15)	2.222	H(22)	H(18)	3.454
H(22)	H(23)	1.584	H(22)	H(24)	1.584
H(22)	H(25)	2.342	H(22)	H(30)	2.681
H(22)	H(32)	2.334	H(23)	C(15)	3.304
H(23)	C(19)	2.555	H(23)	C(20)	2.042
H(23)	C(29) ⁹⁾	3.061	H(23)	H(15)	3.275
H(23)	H(18)	2.335	H(23)	H(22)	1.584
H(23)	H(24)	1.584	H(23)	H(25)	3.528
H(23)	H(30)	3.174	H(23)	H(32)	3.246
H(23)	H(38) ⁸⁾	3.487	H(23)	H(41) ⁹⁾	2.760
H(23)	H(43) ⁹⁾	2.477	H(24)	N(3)	3.061
H(24)	C(15)	2.880	H(24)	C(19)	3.009
H(24)	C(20)	2.042	H(24)	C(23)	3.297
H(24)	C(26) ⁹⁾	3.558	H(24)	H(17) ¹¹⁾	3.304
H(24)	H(18)	3.153	H(24)	H(18) ¹¹⁾	3.375
H(24)	H(22)	1.584	H(24)	H(23)	1.584
H(24)	H(25)	2.341	H(24)	H(30)	3.105
H(24)	H(32)	3.520	H(24)	H(33) ⁹⁾	3.146
H(24)	H(35) ⁹⁾	3.037	H(24)	H(43) ⁹⁾	3.111
H(24)	H(46)	3.030	H(25)	V(1)	2.518
H(25)	Si(1)	2.334	H(25)	N(3)	2.718
H(25)	C(15)	3.566	H(25)	C(20)	3.596
H(25)	C(22)	2.746	H(25)	C(25)	2.990
H(25)	C(26)	3.390	H(25)	H(18) ¹¹⁾	2.886
H(25)	H(22)	2.342	H(25)	H(23)	3.528
H(25)	H(24)	2.341	H(25)	H(26)	1.649

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(25)	H(30)	2.858	H(25)	H(32)	3.149
H(25)	H(35)	3.279	H(25)	H(43)	3.351
H(25)	H(46)	2.830	H(26)	V(1)	2.518
H(26)	Si(1)	2.334	H(26)	N(2)	2.791
H(26)	N(3)	3.538	H(26)	C(6)	2.804
H(26)	C(24)	3.530	H(26)	C(26)	2.888
H(26)	C(27)	3.351	H(26)	C(29)	3.125
H(26)	H(6)	2.380	H(26)	H(18) ¹¹⁾	3.325
H(26)	H(25)	1.649	H(26)	H(29)	3.438
H(26)	H(33)	2.751	H(26)	H(35)	3.015
H(26)	H(37)	3.552	H(26)	H(42)	3.498
H(26)	H(43)	2.274	H(26)	H(46)	3.387
H(27)	Si(1)	2.388	H(27)	N(1)	3.495
H(27)	N(2)	3.310	H(27)	C(1)	3.101
H(27)	C(2)	3.035	H(27)	C(3)	3.535
H(27)	C(14)	3.599	H(27)	C(23)	3.345
H(27)	C(25)	3.164	H(27)	H(2)	2.540
H(27)	H(13)	2.665	H(27)	H(16) ⁸⁾	2.980
H(27)	H(19) ⁸⁾	3.053	H(27)	H(28)	1.600
H(27)	H(29)	1.600	H(27)	H(31)	3.357
H(27)	H(32)	3.034	H(28)	Si(1)	2.388
H(28)	C(17) ⁸⁾	3.251	H(28)	C(25)	3.153
H(28)	C(26)	3.121	H(28)	H(16) ⁸⁾	2.392
H(28)	H(27)	1.600	H(28)	H(29)	1.600
H(28)	H(31)	2.948	H(28)	H(32)	3.442
H(28)	H(33)	3.406	H(28)	H(34)	2.908
H(28)	H(40) ⁴⁾	3.156	H(28)	H(42) ⁴⁾	2.747
H(29)	Si(1)	2.388	H(29)	N(2)	2.776
H(29)	C(1)	3.552	H(29)	C(2)	2.753
H(29)	C(3)	2.904	H(29)	C(4)	3.052
H(29)	C(5)	3.071	H(29)	C(6)	2.923
H(29)	C(23)	3.297	H(29)	C(26)	3.174
H(29)	H(2)	3.331	H(29)	H(3)	3.497
H(29)	H(6)	3.506	H(29)	H(26)	3.438
H(29)	H(27)	1.600	H(29)	H(28)	1.600
H(29)	H(33)	3.047	H(29)	H(34)	3.368
H(29)	H(40) ⁴⁾	2.780	H(30)	Si(1)	2.404

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(30)	C(22)	3.157	H(30)	C(23)	3.263
H(30)	C(26)	3.213	H(30)	H(17) ¹¹⁾	3.237
H(30)	H(22)	2.681	H(30)	H(23)	3.174
H(30)	H(24)	3.105	H(30)	H(25)	2.858
H(30)	H(31)	1.633	H(30)	H(31) ¹²⁾	2.948
H(30)	H(32)	1.633	H(30)	H(34)	3.460
H(30)	H(34) ¹²⁾	3.064	H(30)	H(35)	3.051
H(30)	H(41) ⁹⁾	3.318	H(31)	Si(1)	2.404
H(31)	C(17) ⁸⁾	3.253	H(31)	C(18) ⁸⁾	3.507
H(31)	C(24)	3.130	H(31)	C(25) ¹²⁾	3.351
H(31)	C(26)	3.198	H(31)	C(26) ¹²⁾	3.461
H(31)	H(16) ⁸⁾	3.223	H(31)	H(27)	3.357
H(31)	H(28)	2.948	H(31)	H(30)	1.633
H(31)	H(30) ¹²⁾	2.948	H(31)	H(31) ¹²⁾	2.953
H(31)	H(32)	1.633	H(31)	H(34)	3.031
H(31)	H(34) ¹²⁾	3.088	H(31)	H(35)	3.443
H(31)	H(35) ¹²⁾	3.061	H(31)	H(44) ⁸⁾	3.566
H(32)	Si(1)	2.404	H(32)	C(14)	3.237
H(32)	C(22)	3.151	H(32)	C(23)	3.281
H(32)	C(24)	3.201	H(32)	H(13)	2.900
H(32)	H(14)	3.466	H(32)	H(15)	2.856
H(32)	H(19) ⁸⁾	3.557	H(32)	H(22)	2.334
H(32)	H(23)	3.246	H(32)	H(24)	3.520
H(32)	H(25)	3.149	H(32)	H(27)	3.034
H(32)	H(28)	3.442	H(32)	H(30)	1.633
H(32)	H(31)	1.633	H(32)	H(44) ⁸⁾	2.977
H(33)	Si(1)	2.401	H(33)	C(23)	3.191
H(33)	C(24)	3.194	H(33)	H(6)	3.461
H(33)	H(24) ¹¹⁾	3.146	H(33)	H(26)	2.751
H(33)	H(28)	3.406	H(33)	H(29)	3.047
H(33)	H(34)	1.633	H(33)	H(35)	1.633
H(33)	H(38) ⁴⁾	2.909	H(33)	H(40) ⁴⁾	3.409
H(33)	H(45) ¹¹⁾	3.520	H(33)	H(46) ¹¹⁾	3.186
H(34)	Si(1)	2.401	H(34)	C(24)	3.114
H(34)	C(25)	3.203	H(34)	C(25) ¹²⁾	3.460
H(34)	C(29) ⁴⁾	3.234	H(34)	C(30) ¹¹⁾	3.443
H(34)	H(28)	2.908	H(34)	H(29)	3.368

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(34)	H(30)	3.460	H(34)	H(30) ¹²⁾	3.064
H(34)	H(31)	3.031	H(34)	H(31) ¹²⁾	3.088
H(34)	H(33)	1.633	H(34)	H(35)	1.633
H(34)	H(38) ⁴⁾	3.325	H(34)	H(41) ⁴⁾	2.529
H(34)	H(42) ⁴⁾	3.091	H(34)	H(44) ¹¹⁾	3.295
H(34)	H(45) ¹¹⁾	3.088	H(34)	H(46) ¹¹⁾	3.365
H(35)	Si(1)	2.401	H(35)	C(15) ¹¹⁾	3.565
H(35)	C(18) ¹¹⁾	3.340	H(35)	C(19) ¹¹⁾	2.880
H(35)	C(20) ¹¹⁾	2.990	H(35)	C(22) ¹¹⁾	3.448
H(35)	C(23)	3.213	H(35)	C(25)	3.206
H(35)	H(18) ¹¹⁾	3.091	H(35)	H(24) ¹¹⁾	3.037
H(35)	H(25)	3.279	H(35)	H(26)	3.015
H(35)	H(30)	3.051	H(35)	H(31)	3.443
H(35)	H(31) ¹²⁾	3.061	H(35)	H(33)	1.633
H(35)	H(34)	1.633	H(35)	H(44) ¹¹⁾	3.361
H(35)	H(46) ¹¹⁾	3.205	H(36)	V(1)	2.577
H(36)	Si(2)	2.368	H(36)	N(1)	3.441
H(36)	N(3)	2.908	H(36)	C(13)	3.486
H(36)	C(21)	3.058	H(36)	C(28)	3.096
H(36)	C(30)	3.295	H(36)	H(7) ¹⁾	3.216
H(36)	H(8) ¹⁾	3.157	H(36)	H(10)	2.680
H(36)	H(11)	3.586	H(36)	H(14) ¹⁰⁾	3.285
H(36)	H(20)	3.229	H(36)	H(21)	2.165
H(36)	H(37)	1.682	H(36)	H(39)	2.910
H(36)	H(40)	3.368	H(36)	H(44)	3.142
H(37)	V(1)	2.577	H(37)	Si(2)	2.368
H(37)	N(1)	3.541	H(37)	N(2)	2.582
H(37)	C(2)	3.536	H(37)	C(6)	2.906
H(37)	C(9) ¹⁾	3.139	H(37)	C(28)	3.252
H(37)	C(29)	3.106	H(37)	H(6)	2.759
H(37)	H(7) ¹⁾	2.192	H(37)	H(8) ¹⁾	3.359
H(37)	H(10)	3.129	H(37)	H(26)	3.552
H(37)	H(36)	1.682	H(37)	H(39)	3.543
H(37)	H(40)	3.111	H(37)	H(42)	2.924
H(37)	H(43)	3.353	H(38)	Si(2)	2.438
H(38)	C(26) ¹³⁾	3.572	H(38)	C(29)	3.182
H(38)	C(30)	3.182	H(38)	H(18) ¹⁰⁾	3.180

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(38)	H(23) ¹⁰⁾	3.487	H(38)	H(33) ¹³⁾	2.909
H(38)	H(34) ¹³⁾	3.325	H(38)	H(39)	1.731
H(38)	H(40)	1.731	H(38)	H(41)	2.974
H(38)	H(42)	3.451	H(38)	H(44)	3.472
H(38)	H(45)	2.979	H(38)	H(45) ¹⁴⁾	3.078
H(39)	Si(2)	2.438	H(39)	C(14) ¹⁰⁾	3.023
H(39)	C(27)	3.220	H(39)	C(30)	3.257
H(39)	H(5) ¹³⁾	3.564	H(39)	H(9) ¹⁰⁾	2.911
H(39)	H(14) ¹⁰⁾	2.475	H(39)	H(15) ¹⁰⁾	2.775
H(39)	H(36)	2.910	H(39)	H(37)	3.543
H(39)	H(38)	1.731	H(39)	H(40)	1.731
H(39)	H(44)	3.097	H(39)	H(45)	3.528
H(40)	Si(2)	2.438	H(40)	C(4) ¹³⁾	3.469
H(40)	C(5) ¹³⁾	3.461	H(40)	C(24) ¹³⁾	3.395
H(40)	C(27)	3.257	H(40)	C(29)	3.121
H(40)	H(4) ¹³⁾	3.420	H(40)	H(5) ¹³⁾	3.418
H(40)	H(28) ¹³⁾	3.156	H(40)	H(29) ¹³⁾	2.780
H(40)	H(33) ¹³⁾	3.409	H(40)	H(36)	3.368
H(40)	H(37)	3.111	H(40)	H(38)	1.731
H(40)	H(39)	1.731	H(40)	H(41)	3.339
H(40)	H(42)	2.928	H(41)	Si(2)	2.418
H(41)	C(22) ¹¹⁾	3.596	H(41)	C(26) ¹³⁾	3.462
H(41)	C(28)	3.129	H(41)	C(30)	3.239
H(41)	C(30) ¹⁴⁾	3.448	H(41)	H(17) ³⁾	3.549
H(41)	H(23) ¹¹⁾	2.760	H(41)	H(30) ¹¹⁾	3.318
H(41)	H(34) ¹³⁾	2.529	H(41)	H(38)	2.974
H(41)	H(40)	3.339	H(41)	H(42)	1.633
H(41)	H(43)	1.633	H(41)	H(44) ¹⁴⁾	3.548
H(41)	H(45)	3.064	H(41)	H(45) ¹⁴⁾	2.516
H(41)	H(46)	3.538	H(42)	Si(2)	2.418
H(42)	C(6)	3.489	H(42)	C(18) ³⁾	3.544
H(42)	C(27)	3.224	H(42)	C(28)	3.147
H(42)	H(6)	2.603	H(42)	H(16) ³⁾	3.370
H(42)	H(17) ³⁾	2.632	H(42)	H(26)	3.498
H(42)	H(28) ¹³⁾	2.747	H(42)	H(34) ¹³⁾	3.091
H(42)	H(37)	2.924	H(42)	H(38)	3.451
H(42)	H(40)	2.928	H(42)	H(41)	1.633

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens
(continued)

atom	atom	distance	atom	atom	distance
H(42)	H(43)	1.633	H(43)	Si(2)	2.418
H(43)	C(6)	3.532	H(43)	C(22) ¹¹⁾	3.216
H(43)	C(23)	3.152	H(43)	C(27)	3.262
H(43)	C(30)	3.249	H(43)	H(6)	2.620
H(43)	H(17) ³⁾	2.989	H(43)	H(18) ¹¹⁾	3.473
H(43)	H(23) ¹¹⁾	2.477	H(43)	H(24) ¹¹⁾	3.111
H(43)	H(25)	3.351	H(43)	H(26)	2.274
H(43)	H(37)	3.353	H(43)	H(41)	1.633
H(43)	H(42)	1.633	H(43)	H(45)	3.501
H(43)	H(46)	3.113	H(44)	Si(2)	2.423
H(44)	N(3)	3.303	H(44)	C(15)	3.310
H(44)	C(16)	3.419	H(44)	C(21)	3.513
H(44)	C(27)	3.239	H(44)	C(28)	3.235
H(44)	H(21)	2.864	H(44)	H(31) ¹⁰⁾	3.566
H(44)	H(32) ¹⁰⁾	2.977	H(44)	H(34) ⁹⁾	3.295
H(44)	H(35) ⁹⁾	3.361	H(44)	H(36)	3.142
H(44)	H(38)	3.472	H(44)	H(39)	3.097
H(44)	H(41) ¹⁴⁾	3.548	H(44)	H(45)	1.747
H(44)	H(46)	1.747	H(45)	Si(2)	2.423
H(45)	C(28)	3.199	H(45)	C(29)	3.239
H(45)	C(29) ¹⁴⁾	3.391	H(45)	C(30) ¹⁴⁾	3.524
H(45)	H(33) ⁹⁾	3.520	H(45)	H(34) ⁹⁾	3.088
H(45)	H(38)	2.979	H(45)	H(38) ¹⁴⁾	3.078
H(45)	H(39)	3.528	H(45)	H(41)	3.064
H(45)	H(41) ¹⁴⁾	2.516	H(45)	H(43)	3.501
H(45)	H(44)	1.747	H(45)	H(45) ¹⁴⁾	2.626
H(45)	H(46)	1.747	H(46)	V(1)	3.449
H(46)	Si(2)	2.423	H(46)	N(3)	2.940
H(46)	C(15)	3.266	H(46)	C(23)	3.396
H(46)	C(26) ⁹⁾	3.444	H(46)	C(27)	3.246
H(46)	C(29)	3.276	H(46)	H(18) ¹¹⁾	3.315
H(46)	H(24)	3.030	H(46)	H(25)	2.830
H(46)	H(26)	3.387	H(46)	H(33) ⁹⁾	3.186
H(46)	H(34) ⁹⁾	3.365	H(46)	H(35) ⁹⁾	3.205
H(46)	H(41)	3.538	H(46)	H(43)	3.113
H(46)	H(44)	1.747	H(46)	H(45)	1.747

Symmetry Operators:

- | | |
|--------------------------------|----------------------------|
| (1) $-X+1, Y+1/2-1, -Z+1/2+1$ | (2) $-X+1, -Y, -Z+1$ |
| (3) $X, Y-1, Z$ | (4) $X, -Y+1/2-1, Z+1/2-1$ |
| (5) $-X+1, Y+1/2, -Z+1/2+1$ | (6) $X, Y+1, Z$ |
| (7) $-X+1, -Y+1, -Z+1$ | (8) $X, -Y+1/2, Z+1/2-1$ |
| (9) $-X+2, Y+1/2, -Z+1/2+1$ | (10) $X, -Y+1/2, Z+1/2$ |
| (11) $-X+2, Y+1/2-1, -Z+1/2+1$ | (12) $-X+2, -Y, -Z+1$ |
| (13) $X, -Y+1/2-1, Z+1/2$ | (14) $-X+2, -Y, -Z+2$ |

X-ray Structure Report for $\text{V}(\text{N}=2,6\text{-Me}_2\text{C}_6\text{H}_3)(\text{CH}_2\text{SiMe}_3)_2[2\text{-(}2,6\text{-}^i\text{Pr}_2\text{C}_6\text{H}_3\text{)CH}_2(\text{C}_5\text{H}_4\text{N})]$ (**2b**)

June 24, 2009

Experimental

Data Collection

A red block crystal of $C_{34}H_{54}N_3Si_2V$ having approximate dimensions of 0.20 x 0.15 x 0.10 mm was mounted on a glass fiber. All measurements were made on a Rigaku RAXIS RAPID imaging plate area detector with graphite monochromated Mo-K α radiation.

Indexing was performed from 3 oscillations that were exposed for 320 seconds. The crystal-to-detector distance was 127.40 mm.

Cell constants and an orientation matrix for data collection corresponded to a primitive triclinic cell with dimensions:

$$\begin{array}{ll} a = 10.1893(9) \text{ \AA} & \alpha = 97.395(3)^\circ \\ b = 11.0729(9) \text{ \AA} & \beta = 95.290(3)^\circ \\ c = 16.2502(16) \text{ \AA} & \gamma = 96.446(2)^\circ \\ V = 1796.1(3) \text{ \AA}^3 & \end{array}$$

For $Z = 2$ and F.W. = 611.93, the calculated density is 1.131 g/cm³. Based on a statistical analysis of intensity distribution, and the successful solution and refinement of the structure, the space group was determined to be:

P-1 (#2)

The data were collected at a temperature of $-120 \pm 1^\circ\text{C}$ to a maximum 2θ value of 48.8° . A total of 74 oscillation images were collected. A sweep of data was done using ω scans from 130.0 to 190.0° in 3.0° step, at $\chi=45.0^\circ$ and $\phi = 0.0^\circ$. The exposure rate was 240.0 [sec./ $^\circ$]. A second sweep was performed using ω scans from 0.0 to 162.0° in 3.0° step, at $\chi=45.0^\circ$ and $\phi = 180.0^\circ$. The exposure rate was 240.0 [sec./ $^\circ$]. The crystal-to-detector distance was 127.40 mm. Readout was performed in the 0.100 mm pixel mode.

Data Reduction

Of the 13412 reflections that were collected, 5866 were unique ($R_{\text{int}} = 0.054$); equivalent reflections were merged.

The linear absorption coefficient, μ , for Mo-K α radiation is 3.677 cm⁻¹. An empirical absorption correction was applied which resulted in transmission factors ranging from 0.576 to 0.964. The data were corrected for Lorentz and polarization effects.

Structure Solution and Refinement

The structure was solved by direct methods¹ and expanded using Fourier techniques². The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were refined using the riding model. The final cycle of full-matrix least-squares refinement³ on F^2 was based on 5866 observed reflections and 362 variable parameters and converged (largest parameter shift was 0.00 times its esd) with unweighted and weighted agreement factors of:

$$R1 = \sum ||F_o| - |F_c|| / \sum |F_o| = 0.0637$$

$$wR2 = [\sum (w (F_o^2 - F_c^2)^2) / \sum w(F_o^2)^2]^{1/2} = 0.2134$$

The standard deviation of an observation of unit weight⁴ was 1.18. Unit weights were used. The maximum and minimum peaks on the final difference Fourier map corresponded to 0.66 and -1.09 e-/Å³, respectively.

Neutral atom scattering factors were taken from Cromer and Waber⁵. Anomalous dispersion effects were included in F_{calc} ⁶; the values for $\Delta f'$ and $\Delta f''$ were those of Creagh and McAuley⁷. The values for the mass attenuation coefficients are those of Creagh and Hubbell⁸. All calculations were performed using the CrystalStructure⁹ crystallographic software package except for refinement, which was performed using SHELXL-97¹⁰.

References

(1) SIR92: Altomare, A., Casciarano, G., Giacovazzo, C., Guagliardi, A., Burla, M., Polidori, G., and Camalli, M. (1994) J. Appl. Cryst., 27, 435.

(2) DIRDIF99: Beurskens, P.T., Admiraal, G., Beurskens, G., Bosman, W.P., de Gelder, R., Israel, R. and Smits, J.M.M.(1999). The DIRDIF-99 program system, Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands.

(3) Least Squares function minimized: (SHELXL97)

$$\sum w(F_o^2 - F_c^2)^2 \quad \text{where } w = \text{Least Squares weights.}$$

(4) Standard deviation of an observation of unit weight:

$$[\sum w(F_o^2 - F_c^2)^2 / (N_o - N_v)]^{1/2}$$

where: N_o = number of observations
 N_v = number of variables

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EXPERIMENTAL DETAILS

A. Crystal Data

Empirical Formula	C ₃₄ H ₅₄ N ₃ Si ₂ V
Formula Weight	611.93
Crystal Color, Habit	red, block
Crystal Dimensions	0.20 X 0.15 X 0.10 mm
Crystal System	triclinic
Lattice Type	Primitive
Indexing Images	3 oscillations @ 320.0 seconds
Detector Position	127.40 mm
Pixel Size	0.100 mm
Lattice Parameters	a = 10.1893(9) Å b = 11.0729(9) Å c = 16.2502(16) Å α = 97.395(3) ° β = 95.290(3) ° γ = 96.446(2) ° V = 1796.1(3) Å ³
Space Group	P-1 (#2)
Z value	2
D _{calc}	1.131 g/cm ³
F ₀₀₀	660.00
μ (MoK α)	3.677 cm ⁻¹

B. Intensity Measurements

Diffractometer	Rigaku RAXIS-RAPID
Radiation	MoK α (λ = 0.71075 Å) graphite monochromated
Detector Aperture	280 mm x 256 mm
Data Images	74 exposures
ω oscillation Range (χ =45.0, ϕ =0.0)	130.0 - 190.0°
Exposure Rate	240.0 sec./°
ω oscillation Range (χ =45.0, ϕ =180.0)	0.0 - 162.0°
Exposure Rate	240.0 sec./°
Detector Position	127.40 mm
Pixel Size	0.100 mm
$2\theta_{\max}$	48.8°
No. of Reflections Measured	Total: 13412 Unique: 5866 (R_{int} = 0.054)
Corrections	Lorentz-polarization Absorption (trans. factors: 0.576 - 0.964)

C. Structure Solution and Refinement

Structure Solution	Direct Methods (SIR92)
Refinement	Full-matrix least-squares on F^2
Function Minimized	$\sum w (F_o^2 - F_c^2)^2$
Least Squares Weights	$w = 1 / [\sigma^2(F_o^2) + (0.0585 \cdot P)^2 + 5.5510 \cdot P]$ where $P = (\text{Max}(F_o^2, 0) + 2F_c^2)/3$
$2\theta_{\text{max}}$ cutoff	48.8°
Anomalous Dispersion	All non-hydrogen atoms
No. Observations (All reflections)	5866
No. Variables	362
Reflection/Parameter Ratio	16.20
Residuals: R1 ($I > 2.00\sigma(I)$)	0.0637
Residuals: R (All reflections)	0.1097
Residuals: wR2 (All reflections)	0.2134
Goodness of Fit Indicator	1.181
Max Shift/Error in Final Cycle	0.000
Maximum peak in Final Diff. Map	0.66 e ⁻ /Å ³
Minimum peak in Final Diff. Map	-1.09 e ⁻ /Å ³

Table 1. Atomic coordinates and $B_{\text{iso}}/B_{\text{eq}}$

atom	x	y	z	B_{eq}
V(1)	0.30575(9)	0.17342(8)	0.22466(6)	2.45(2)
Si(1)	0.20614(17)	0.31577(15)	0.04828(10)	3.25(3)
Si(2)	0.38487(16)	-0.11858(13)	0.18168(10)	3.07(3)
N(1)	0.3381(4)	0.3111(3)	0.3099(2)	2.50(7)
N(2)	0.4946(4)	0.2792(3)	0.1956(2)	2.67(8)
N(3)	0.1675(3)	0.1034(3)	0.2526(2)	2.31(7)
C(1)	0.4320(5)	0.4189(4)	0.3045(3)	2.48(9)
C(2)	0.5248(5)	0.3907(4)	0.2403(3)	2.46(9)
C(3)	0.6320(5)	0.4711(4)	0.2284(3)	3.04(10)
C(4)	0.7129(5)	0.4376(5)	0.1684(3)	3.36(11)
C(5)	0.6821(5)	0.3207(5)	0.1220(3)	3.62(11)
C(6)	0.5739(5)	0.2449(4)	0.1371(3)	2.87(9)
C(7)	0.2930(5)	0.3176(4)	0.3924(3)	2.47(9)
C(8)	0.3813(5)	0.2971(4)	0.4596(3)	2.39(9)
C(9)	0.3384(5)	0.3052(4)	0.5385(3)	2.87(10)
C(10)	0.2110(5)	0.3303(5)	0.5516(3)	3.35(11)
C(11)	0.1265(5)	0.3497(5)	0.4851(3)	3.25(10)
C(12)	0.1630(5)	0.3448(4)	0.4045(3)	2.70(9)
C(13)	0.5224(5)	0.2652(4)	0.4491(3)	2.71(9)
C(14)	0.0675(5)	0.3712(5)	0.3342(3)	3.16(10)
C(15)	0.0507(5)	0.0450(4)	0.2778(3)	2.89(10)
C(16)	0.0579(5)	-0.0045(5)	0.3536(3)	3.12(10)
C(17)	-0.0607(6)	-0.0553(5)	0.3807(4)	3.98(12)
C(18)	-0.1820(6)	-0.0587(6)	0.3339(4)	4.36(13)
C(19)	-0.1865(5)	-0.0130(5)	0.2587(4)	4.03(12)
C(20)	-0.0716(5)	0.0386(5)	0.2290(3)	3.28(10)
C(21)	0.1864(5)	-0.0062(5)	0.4050(3)	3.80(12)
C(22)	-0.0806(6)	0.0846(6)	0.1458(4)	4.17(12)
C(23)	0.2462(5)	0.1808(4)	0.0983(3)	2.86(10)
C(24)	0.3405(8)	0.4443(6)	0.0645(5)	7.4(2)
C(25)	0.0549(10)	0.3772(9)	0.0831(6)	9.9(3)
C(26)	0.1711(9)	0.2706(6)	-0.0656(4)	6.6(2)
C(27)	0.4238(5)	0.0353(4)	0.2460(3)	2.71(9)
C(28)	0.4640(6)	-0.2359(5)	0.2357(4)	4.26(13)
C(29)	0.4563(8)	-0.1168(6)	0.0808(4)	5.59(17)
C(30)	0.2062(7)	-0.1738(6)	0.1588(5)	5.96(18)
C(31)	0.6287(5)	0.3621(5)	0.4979(3)	3.58(11)

Table 1. Atomic coordinates and $B_{\text{iso}}/B_{\text{eq}}$ (continued)

atom	x	y	z	B_{eq}
C(32)	0.5400(6)	0.1407(4)	0.4755(3)	3.56(11)
C(33)	0.0859(6)	0.5079(5)	0.3257(4)	4.03(12)
C(34)	-0.0775(5)	0.3295(5)	0.3441(3)	3.93(12)

$$B_{\text{eq}} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 2. Atomic coordinates and B_{iso} involving hydrogens/ B_{eq}

atom	x	y	z	B_{eq}
H(1)	0.4838	0.4465	0.3595	2.98
H(2)	0.3825	0.4864	0.2897	2.98
H(3)	0.6501	0.5495	0.2615	3.65
H(4)	0.7872	0.4920	0.1589	4.03
H(5)	0.7359	0.2940	0.0802	4.34
H(6)	0.5541	0.1656	0.1054	3.45
H(7)	0.3977	0.2933	0.5847	3.44
H(8)	0.1826	0.3340	0.6058	4.02
H(9)	0.0396	0.3672	0.4946	3.90
H(10)	0.5354	0.2608	0.3886	3.26
H(11)	0.0892	0.3250	0.2812	3.79
H(12)	-0.0575	-0.0880	0.4321	4.78
H(13)	-0.2614	-0.0921	0.3534	5.23
H(14)	-0.2698	-0.0168	0.2264	4.83
H(15)	0.2212	0.0775	0.4310	4.56
H(16)	0.2503	-0.0388	0.3692	4.56
H(17)	0.1723	-0.0584	0.4485	4.56
H(18)	-0.0435	0.1714	0.1531	5.01
H(19)	-0.1738	0.0757	0.1224	5.01
H(20)	-0.0303	0.0368	0.1078	5.01
H(21)	0.3167	0.1475	0.0675	3.43
H(22)	0.1664	0.1192	0.0836	3.43
H(23)	0.3241	0.5003	0.0237	8.91
H(24)	0.4254	0.4128	0.0574	8.91
H(25)	0.3436	0.4885	0.1211	8.91
H(26)	0.0614	0.3889	0.1442	11.93
H(27)	-0.0237	0.3191	0.0605	11.93
H(28)	0.0474	0.4561	0.0629	11.93
H(29)	0.1327	0.3364	-0.0906	7.87
H(30)	0.1081	0.1953	-0.0773	7.87
H(31)	0.2539	0.2561	-0.0894	7.87
H(32)	0.5164	0.0670	0.2394	3.25
H(33)	0.4217	0.0231	0.3052	3.25
H(34)	0.4515	-0.3145	0.1987	5.12
H(35)	0.4226	-0.2462	0.2869	5.12
H(36)	0.5592	-0.2086	0.2498	5.12
H(37)	0.4750	-0.1995	0.0600	6.71

Table 2. Atomic coordinates and B_{iso} involving hydrogens/ B_{eq} (continued)

atom	x	y	z	B_{eq}
H(38)	0.5389	-0.0600	0.0889	6.71
H(39)	0.3927	-0.0898	0.0401	6.71
H(40)	0.1922	-0.2394	0.1112	7.16
H(41)	0.1579	-0.1058	0.1455	7.16
H(42)	0.1733	-0.2054	0.2077	7.16
H(43)	0.6189	0.3672	0.5576	4.30
H(44)	0.6190	0.4419	0.4796	4.30
H(45)	0.7168	0.3397	0.4879	4.30
H(46)	0.6194	0.1118	0.4539	4.27
H(47)	0.4619	0.0819	0.4531	4.27
H(48)	0.5503	0.1481	0.5366	4.27
H(49)	0.1736	0.5301	0.3077	4.84
H(50)	0.0790	0.5563	0.3796	4.84
H(51)	0.0169	0.5249	0.2841	4.84
H(52)	-0.0881	0.2423	0.3507	4.71
H(53)	-0.1340	0.3416	0.2944	4.71
H(54)	-0.1036	0.3777	0.3935	4.71

$$B_{\text{eq}} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos \gamma + 2U_{13}(aa^*cc^*)\cos \beta + 2U_{23}(bb^*cc^*)\cos \alpha)$$

Table 3. Anisotropic displacement parameters

atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
V(1)	0.0329(5)	0.0281(4)	0.0322(5)	0.0040(3)	0.0049(3)	0.0043(3)
Si(1)	0.0451(9)	0.0407(8)	0.0384(9)	0.0050(7)	0.0024(7)	0.0098(7)
Si(2)	0.0454(9)	0.0300(7)	0.0399(9)	0.0082(6)	-0.0007(7)	0.0000(6)
N(1)	0.039(2)	0.032(2)	0.028(2)	0.0086(19)	0.0117(19)	0.0068(18)
N(2)	0.035(2)	0.034(2)	0.036(2)	0.0050(19)	0.011(2)	0.009(2)
N(3)	0.026(2)	0.027(2)	0.032(2)	-0.0040(17)	0.0014(18)	0.0026(18)
C(1)	0.031(2)	0.026(2)	0.038(3)	-0.002(2)	0.005(2)	0.009(2)
C(2)	0.036(2)	0.029(2)	0.028(2)	0.002(2)	0.004(2)	0.007(2)
C(3)	0.037(3)	0.033(2)	0.044(3)	-0.007(2)	0.006(2)	0.010(2)
C(4)	0.039(3)	0.047(3)	0.041(3)	-0.002(2)	0.001(2)	0.017(2)
C(5)	0.039(3)	0.058(3)	0.044(3)	0.008(2)	0.012(2)	0.014(3)
C(6)	0.035(3)	0.032(2)	0.041(3)	0.000(2)	0.010(2)	0.000(2)
C(7)	0.039(3)	0.018(2)	0.037(3)	0.002(2)	0.006(2)	0.003(2)
C(8)	0.035(2)	0.024(2)	0.030(2)	-0.001(2)	0.000(2)	0.003(2)
C(9)	0.048(3)	0.032(2)	0.027(2)	-0.001(2)	0.002(2)	0.003(2)
C(10)	0.049(3)	0.041(3)	0.035(3)	-0.002(2)	0.010(2)	0.001(2)
C(11)	0.042(3)	0.046(3)	0.037(3)	0.007(2)	0.015(2)	-0.000(2)
C(12)	0.032(2)	0.033(2)	0.037(3)	0.006(2)	0.006(2)	0.001(2)
C(13)	0.038(3)	0.028(2)	0.037(3)	0.005(2)	0.003(2)	0.005(2)
C(14)	0.038(3)	0.042(3)	0.040(3)	0.013(2)	0.003(2)	0.000(2)
C(15)	0.036(3)	0.032(2)	0.042(3)	0.010(2)	0.005(2)	0.003(2)
C(16)	0.037(3)	0.037(3)	0.044(3)	-0.000(2)	0.006(2)	0.008(2)
C(17)	0.046(3)	0.056(3)	0.048(3)	-0.007(2)	0.009(3)	0.011(3)
C(18)	0.043(3)	0.066(4)	0.054(4)	-0.006(3)	0.013(3)	0.008(3)
C(19)	0.028(3)	0.056(3)	0.063(4)	-0.005(2)	0.002(2)	0.000(3)
C(20)	0.035(3)	0.042(3)	0.049(3)	0.009(2)	0.005(2)	0.010(2)
C(21)	0.044(3)	0.051(3)	0.050(3)	-0.003(2)	0.005(2)	0.018(3)
C(22)	0.043(3)	0.058(3)	0.057(4)	0.002(3)	-0.003(3)	0.014(3)
C(23)	0.040(3)	0.037(2)	0.030(2)	0.009(2)	0.002(2)	-0.001(2)
C(24)	0.102(6)	0.059(4)	0.112(7)	-0.011(4)	-0.053(5)	0.041(4)
C(25)	0.143(9)	0.143(8)	0.152(9)	0.105(7)	0.102(7)	0.112(8)
C(26)	0.134(7)	0.056(4)	0.055(4)	0.006(4)	-0.018(4)	0.015(3)
C(27)	0.031(2)	0.032(2)	0.037(3)	-0.004(2)	-0.001(2)	0.007(2)
C(28)	0.073(4)	0.027(2)	0.057(4)	0.004(2)	-0.008(3)	0.001(2)
C(29)	0.112(6)	0.044(3)	0.056(4)	0.020(4)	0.009(4)	-0.001(3)
C(30)	0.066(4)	0.045(3)	0.104(6)	0.002(3)	-0.005(4)	-0.020(3)
C(31)	0.045(3)	0.037(3)	0.052(3)	0.002(2)	0.000(2)	0.003(2)

Table 3. Anisotropic displacement parameters (continued)

atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
C(32)	0.052(3)	0.030(2)	0.051(3)	0.007(2)	-0.003(2)	0.001(2)
C(33)	0.048(3)	0.054(3)	0.054(4)	0.020(3)	0.002(3)	0.010(3)
C(34)	0.043(3)	0.055(3)	0.049(3)	0.011(2)	0.003(2)	-0.003(3)

The general temperature factor expression: $\exp(-2\pi^2(a^2U_{11}h^2 + b^2U_{22}k^2 + c^2U_{33}l^2 + 2a*b*U_{12}hk + 2a*c*U_{13}hl + 2b*c*U_{23}kl))$

Table 4. Bond lengths (Å)

atom	atom	distance	atom	atom	distance
V(1)	N(1)	1.900(3)	V(1)	N(2)	2.259(4)
V(1)	N(3)	1.659(4)	V(1)	C(23)	2.101(5)
V(1)	C(27)	2.093(5)	Si(1)	C(23)	1.858(5)
Si(1)	C(24)	1.835(7)	Si(1)	C(25)	1.858(11)
Si(1)	C(26)	1.844(7)	Si(2)	C(27)	1.862(4)
Si(2)	C(28)	1.873(6)	Si(2)	C(29)	1.856(7)
Si(2)	C(30)	1.842(7)	N(1)	C(1)	1.460(6)
N(1)	C(7)	1.453(7)	N(2)	C(2)	1.337(6)
N(2)	C(6)	1.349(7)	N(3)	C(15)	1.410(6)
C(1)	C(2)	1.498(7)	C(2)	C(3)	1.374(7)
C(3)	C(4)	1.375(8)	C(4)	C(5)	1.399(8)
C(5)	C(6)	1.370(8)	C(7)	C(8)	1.407(7)
C(7)	C(12)	1.417(7)	C(8)	C(9)	1.389(7)
C(8)	C(13)	1.537(7)	C(9)	C(10)	1.387(8)
C(10)	C(11)	1.373(8)	C(11)	C(12)	1.390(8)
C(12)	C(14)	1.511(7)	C(13)	C(31)	1.522(7)
C(13)	C(32)	1.521(7)	C(14)	C(33)	1.530(8)
C(14)	C(34)	1.527(8)	C(15)	C(16)	1.411(8)
C(15)	C(20)	1.403(7)	C(16)	C(17)	1.407(8)
C(16)	C(21)	1.490(8)	C(17)	C(18)	1.385(8)
C(18)	C(19)	1.379(10)	C(19)	C(20)	1.399(8)
C(20)	C(22)	1.503(9)			

Table 5. Bond lengths involving hydrogens (Å)

atom	atom	distance	atom	atom	distance
C(1)	H(1)	0.990	C(1)	H(2)	0.990
C(3)	H(3)	0.950	C(4)	H(4)	0.950
C(5)	H(5)	0.950	C(6)	H(6)	0.950
C(9)	H(7)	0.950	C(10)	H(8)	0.950
C(11)	H(9)	0.950	C(13)	H(10)	1.000
C(14)	H(11)	1.000	C(17)	H(12)	0.950
C(18)	H(13)	0.950	C(19)	H(14)	0.950
C(21)	H(15)	0.980	C(21)	H(16)	0.980
C(21)	H(17)	0.980	C(22)	H(18)	0.980
C(22)	H(19)	0.980	C(22)	H(20)	0.980
C(23)	H(21)	0.990	C(23)	H(22)	0.990
C(24)	H(23)	0.980	C(24)	H(24)	0.980
C(24)	H(25)	0.980	C(25)	H(26)	0.980
C(25)	H(27)	0.980	C(25)	H(28)	0.980
C(26)	H(29)	0.980	C(26)	H(30)	0.980
C(26)	H(31)	0.980	C(27)	H(32)	0.990
C(27)	H(33)	0.990	C(28)	H(34)	0.980
C(28)	H(35)	0.980	C(28)	H(36)	0.980
C(29)	H(37)	0.980	C(29)	H(38)	0.980
C(29)	H(39)	0.980	C(30)	H(40)	0.980
C(30)	H(41)	0.980	C(30)	H(42)	0.980
C(31)	H(43)	0.980	C(31)	H(44)	0.980
C(31)	H(45)	0.980	C(32)	H(46)	0.980
C(32)	H(47)	0.980	C(32)	H(48)	0.980
C(33)	H(49)	0.980	C(33)	H(50)	0.980
C(33)	H(51)	0.980	C(34)	H(52)	0.980
C(34)	H(53)	0.980	C(34)	H(54)	0.980

Table 6. Bond angles (°)

atom	atom	atom	angle	atom	atom	atom	angle
N(1)	V(1)	N(2)	75.88(17)	N(1)	V(1)	N(3)	99.74(19)
N(1)	V(1)	C(23)	125.0(2)	N(1)	V(1)	C(27)	113.17(19)
N(2)	V(1)	N(3)	175.60(17)	N(2)	V(1)	C(23)	82.61(18)
N(2)	V(1)	C(27)	85.45(18)	N(3)	V(1)	C(23)	100.4(2)
N(3)	V(1)	C(27)	96.1(2)	C(23)	V(1)	C(27)	114.8(2)
C(23)	Si(1)	C(24)	114.3(3)	C(23)	Si(1)	C(25)	112.9(4)
C(23)	Si(1)	C(26)	109.6(3)	C(24)	Si(1)	C(25)	107.2(4)
C(24)	Si(1)	C(26)	106.4(4)	C(25)	Si(1)	C(26)	106.0(4)
C(27)	Si(2)	C(28)	110.1(2)	C(27)	Si(2)	C(29)	110.1(2)
C(27)	Si(2)	C(30)	114.5(2)	C(28)	Si(2)	C(29)	106.5(3)
C(28)	Si(2)	C(30)	107.3(3)	C(29)	Si(2)	C(30)	107.9(3)
V(1)	N(1)	C(1)	122.5(3)	V(1)	N(1)	C(7)	126.2(3)
C(1)	N(1)	C(7)	110.5(3)	V(1)	N(2)	C(2)	114.2(3)
V(1)	N(2)	C(6)	127.6(3)	C(2)	N(2)	C(6)	118.1(4)
V(1)	N(3)	C(15)	179.0(3)	N(1)	C(1)	C(2)	111.1(3)
N(2)	C(2)	C(1)	113.9(4)	N(2)	C(2)	C(3)	122.7(4)
C(1)	C(2)	C(3)	123.4(4)	C(2)	C(3)	C(4)	119.7(4)
C(3)	C(4)	C(5)	117.7(5)	C(4)	C(5)	C(6)	119.6(5)
N(2)	C(6)	C(5)	122.1(4)	N(1)	C(7)	C(8)	118.6(4)
N(1)	C(7)	C(12)	120.3(4)	C(8)	C(7)	C(12)	121.1(4)
C(7)	C(8)	C(9)	118.4(4)	C(7)	C(8)	C(13)	122.9(4)
C(9)	C(8)	C(13)	118.7(4)	C(8)	C(9)	C(10)	121.4(5)
C(9)	C(10)	C(11)	119.1(5)	C(10)	C(11)	C(12)	122.7(5)
C(7)	C(12)	C(11)	117.2(5)	C(7)	C(12)	C(14)	122.3(4)
C(11)	C(12)	C(14)	120.4(5)	C(8)	C(13)	C(31)	112.2(4)
C(8)	C(13)	C(32)	111.0(4)	C(31)	C(13)	C(32)	109.5(4)
C(12)	C(14)	C(33)	110.6(4)	C(12)	C(14)	C(34)	113.0(5)
C(33)	C(14)	C(34)	110.0(4)	N(3)	C(15)	C(16)	119.1(4)
N(3)	C(15)	C(20)	120.4(5)	C(16)	C(15)	C(20)	120.5(5)
C(15)	C(16)	C(17)	118.5(5)	C(15)	C(16)	C(21)	122.3(5)
C(17)	C(16)	C(21)	119.2(5)	C(16)	C(17)	C(18)	121.2(6)
C(17)	C(18)	C(19)	119.5(6)	C(18)	C(19)	C(20)	121.7(5)
C(15)	C(20)	C(19)	118.7(5)	C(15)	C(20)	C(22)	121.4(5)
C(19)	C(20)	C(22)	119.9(5)	V(1)	C(23)	Si(1)	128.4(2)
V(1)	C(27)	Si(2)	119.6(2)				

Table 7. Bond angles involving hydrogens (°)

atom	atom	atom	angle	atom	atom	atom	angle
N(1)	C(1)	H(1)	109.4	N(1)	C(1)	H(2)	109.4
C(2)	C(1)	H(1)	109.4	C(2)	C(1)	H(2)	109.4
H(1)	C(1)	H(2)	108.0	C(2)	C(3)	H(3)	120.1
C(4)	C(3)	H(3)	120.1	C(3)	C(4)	H(4)	121.2
C(5)	C(4)	H(4)	121.2	C(4)	C(5)	H(5)	120.2
C(6)	C(5)	H(5)	120.2	N(2)	C(6)	H(6)	119.0
C(5)	C(6)	H(6)	119.0	C(8)	C(9)	H(7)	119.3
C(10)	C(9)	H(7)	119.3	C(9)	C(10)	H(8)	120.4
C(11)	C(10)	H(8)	120.4	C(10)	C(11)	H(9)	118.6
C(12)	C(11)	H(9)	118.6	C(8)	C(13)	H(10)	108.0
C(31)	C(13)	H(10)	108.0	C(32)	C(13)	H(10)	108.0
C(12)	C(14)	H(11)	107.7	C(33)	C(14)	H(11)	107.7
C(34)	C(14)	H(11)	107.7	C(16)	C(17)	H(12)	119.4
C(18)	C(17)	H(12)	119.4	C(17)	C(18)	H(13)	120.3
C(19)	C(18)	H(13)	120.3	C(18)	C(19)	H(14)	119.2
C(20)	C(19)	H(14)	119.2	C(16)	C(21)	H(15)	109.5
C(16)	C(21)	H(16)	109.5	C(16)	C(21)	H(17)	109.5
H(15)	C(21)	H(16)	109.5	H(15)	C(21)	H(17)	109.5
H(16)	C(21)	H(17)	109.5	C(20)	C(22)	H(18)	109.5
C(20)	C(22)	H(19)	109.5	C(20)	C(22)	H(20)	109.5
H(18)	C(22)	H(19)	109.5	H(18)	C(22)	H(20)	109.5
H(19)	C(22)	H(20)	109.5	V(1)	C(23)	H(21)	105.2
V(1)	C(23)	H(22)	105.2	Si(1)	C(23)	H(21)	105.2
Si(1)	C(23)	H(22)	105.2	H(21)	C(23)	H(22)	105.9
Si(1)	C(24)	H(23)	109.5	Si(1)	C(24)	H(24)	109.5
Si(1)	C(24)	H(25)	109.5	H(23)	C(24)	H(24)	109.5
H(23)	C(24)	H(25)	109.5	H(24)	C(24)	H(25)	109.5
Si(1)	C(25)	H(26)	109.5	Si(1)	C(25)	H(27)	109.5
Si(1)	C(25)	H(28)	109.5	H(26)	C(25)	H(27)	109.5
H(26)	C(25)	H(28)	109.5	H(27)	C(25)	H(28)	109.5
Si(1)	C(26)	H(29)	109.5	Si(1)	C(26)	H(30)	109.5
Si(1)	C(26)	H(31)	109.5	H(29)	C(26)	H(30)	109.5
H(29)	C(26)	H(31)	109.5	H(30)	C(26)	H(31)	109.5
V(1)	C(27)	H(32)	107.4	V(1)	C(27)	H(33)	107.4
Si(2)	C(27)	H(32)	107.4	Si(2)	C(27)	H(33)	107.4
H(32)	C(27)	H(33)	106.9	Si(2)	C(28)	H(34)	109.5
Si(2)	C(28)	H(35)	109.5	Si(2)	C(28)	H(36)	109.5

Table 7. Bond angles involving hydrogens ($^{\circ}$) (continued)

atom	atom	atom	angle	atom	atom	atom	angle
H(34)	C(28)	H(35)	109.5	H(34)	C(28)	H(36)	109.5
H(35)	C(28)	H(36)	109.5	Si(2)	C(29)	H(37)	109.5
Si(2)	C(29)	H(38)	109.5	Si(2)	C(29)	H(39)	109.5
H(37)	C(29)	H(38)	109.5	H(37)	C(29)	H(39)	109.5
H(38)	C(29)	H(39)	109.5	Si(2)	C(30)	H(40)	109.5
Si(2)	C(30)	H(41)	109.5	Si(2)	C(30)	H(42)	109.5
H(40)	C(30)	H(41)	109.5	H(40)	C(30)	H(42)	109.5
H(41)	C(30)	H(42)	109.5	C(13)	C(31)	H(43)	109.5
C(13)	C(31)	H(44)	109.5	C(13)	C(31)	H(45)	109.5
H(43)	C(31)	H(44)	109.5	H(43)	C(31)	H(45)	109.5
H(44)	C(31)	H(45)	109.5	C(13)	C(32)	H(46)	109.5
C(13)	C(32)	H(47)	109.5	C(13)	C(32)	H(48)	109.5
H(46)	C(32)	H(47)	109.5	H(46)	C(32)	H(48)	109.5
H(47)	C(32)	H(48)	109.5	C(14)	C(33)	H(49)	109.5
C(14)	C(33)	H(50)	109.5	C(14)	C(33)	H(51)	109.5
H(49)	C(33)	H(50)	109.5	H(49)	C(33)	H(51)	109.5
H(50)	C(33)	H(51)	109.5	C(14)	C(34)	H(52)	109.5
C(14)	C(34)	H(53)	109.5	C(14)	C(34)	H(54)	109.5
H(52)	C(34)	H(53)	109.5	H(52)	C(34)	H(54)	109.5
H(53)	C(34)	H(54)	109.5				

Table 8. Torsion Angles(°)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
N(1)	V(1)	N(2)	C(2)	-10.1(3)	N(1)	V(1)	N(2)	C(6)	173.2(4)
N(2)	V(1)	N(1)	C(1)	14.9(3)	N(2)	V(1)	N(1)	C(7)	-154.5(4)
N(1)	V(1)	N(3)	C(15)	-23(22)	N(3)	V(1)	N(1)	C(1)	-165.5(3)
N(3)	V(1)	N(1)	C(7)	25.1(4)	N(1)	V(1)	C(23)	Si(1)	-3.6(4)
C(23)	V(1)	N(1)	C(1)	-55.3(4)	C(23)	V(1)	N(1)	C(7)	135.2(4)
N(1)	V(1)	C(27)	Si(2)	168.1(2)	C(27)	V(1)	N(1)	C(1)	93.6(4)
C(27)	V(1)	N(1)	C(7)	-75.9(4)	N(2)	V(1)	N(3)	C(15)	-18(24)
N(3)	V(1)	N(2)	C(2)	-15(2)	N(3)	V(1)	N(2)	C(6)	169(2)
N(2)	V(1)	C(23)	Si(1)	-70.6(3)	C(23)	V(1)	N(2)	C(2)	118.8(3)
C(23)	V(1)	N(2)	C(6)	-57.9(4)	N(2)	V(1)	C(27)	Si(2)	-119.4(3)
C(27)	V(1)	N(2)	C(2)	-125.4(3)	C(27)	V(1)	N(2)	C(6)	57.9(4)
N(3)	V(1)	C(23)	Si(1)	106.2(3)	C(23)	V(1)	N(3)	C(15)	-151(22)
N(3)	V(1)	C(27)	Si(2)	64.8(3)	C(27)	V(1)	N(3)	C(15)	92(22)
C(23)	V(1)	C(27)	Si(2)	-39.7(3)	C(27)	V(1)	C(23)	Si(1)	-152.0(3)
C(24)	Si(1)	C(23)	V(1)	56.8(5)	C(25)	Si(1)	C(23)	V(1)	-66.1(4)
C(26)	Si(1)	C(23)	V(1)	176.1(4)	C(28)	Si(2)	C(27)	V(1)	-160.6(3)
C(29)	Si(2)	C(27)	V(1)	82.2(4)	C(30)	Si(2)	C(27)	V(1)	-39.6(4)
V(1)	N(1)	C(1)	C(2)	-17.4(5)	V(1)	N(1)	C(7)	C(8)	97.2(4)
V(1)	N(1)	C(7)	C(12)	-83.3(5)	C(1)	N(1)	C(7)	C(8)	-73.3(5)
C(1)	N(1)	C(7)	C(12)	106.3(4)	C(7)	N(1)	C(1)	C(2)	153.5(4)
V(1)	N(2)	C(2)	C(1)	4.0(5)	V(1)	N(2)	C(2)	C(3)	-176.5(4)
V(1)	N(2)	C(6)	C(5)	175.9(4)	C(2)	N(2)	C(6)	C(5)	-0.7(8)
C(6)	N(2)	C(2)	C(1)	-179.0(4)	C(6)	N(2)	C(2)	C(3)	0.5(7)
V(1)	N(3)	C(15)	C(16)	-44(22)	V(1)	N(3)	C(15)	C(20)	135(22)
N(1)	C(1)	C(2)	N(2)	6.7(6)	N(1)	C(1)	C(2)	C(3)	-172.8(4)
N(2)	C(2)	C(3)	C(4)	0.1(6)	C(1)	C(2)	C(3)	C(4)	179.5(5)
C(2)	C(3)	C(4)	C(5)	-0.5(8)	C(3)	C(4)	C(5)	C(6)	0.3(8)
C(4)	C(5)	C(6)	N(2)	0.3(7)	N(1)	C(7)	C(8)	C(9)	179.0(4)
N(1)	C(7)	C(8)	C(13)	-1.5(6)	N(1)	C(7)	C(12)	C(11)	-179.9(3)
N(1)	C(7)	C(12)	C(14)	-1.6(6)	C(8)	C(7)	C(12)	C(11)	-0.3(6)
C(8)	C(7)	C(12)	C(14)	178.0(4)	C(12)	C(7)	C(8)	C(9)	-0.5(6)
C(12)	C(7)	C(8)	C(13)	179.0(4)	C(7)	C(8)	C(9)	C(10)	1.3(7)
C(7)	C(8)	C(13)	C(31)	117.7(5)	C(7)	C(8)	C(13)	C(32)	-119.4(5)
C(9)	C(8)	C(13)	C(31)	-62.8(6)	C(9)	C(8)	C(13)	C(32)	60.1(5)
C(13)	C(8)	C(9)	C(10)	-178.2(4)	C(8)	C(9)	C(10)	C(11)	-1.2(7)
C(9)	C(10)	C(11)	C(12)	0.3(6)	C(10)	C(11)	C(12)	C(7)	0.5(7)
C(10)	C(11)	C(12)	C(14)	-177.9(4)	C(7)	C(12)	C(14)	C(33)	-87.9(6)

Table 8. Torsion angles ($^{\circ}$) (continued)

atom1	atom2	atom3	atom4	angle	atom1	atom2	atom3	atom4	angle
C(7)	C(12)	C(14)	C(34)	148.2(4)	C(11)	C(12)	C(14)	C(33)	90.3(6)
C(11)	C(12)	C(14)	C(34)	-33.5(6)	N(3)	C(15)	C(16)	C(17)	175.9(4)
N(3)	C(15)	C(16)	C(21)	-5.0(7)	N(3)	C(15)	C(20)	C(19)	-175.9(4)
N(3)	C(15)	C(20)	C(22)	4.8(8)	C(16)	C(15)	C(20)	C(19)	2.6(8)
C(16)	C(15)	C(20)	C(22)	-176.7(5)	C(20)	C(15)	C(16)	C(17)	-2.6(8)
C(20)	C(15)	C(16)	C(21)	176.4(5)	C(15)	C(16)	C(17)	C(18)	0.8(8)
C(21)	C(16)	C(17)	C(18)	-178.3(5)	C(16)	C(17)	C(18)	C(19)	1.0(9)
C(17)	C(18)	C(19)	C(20)	-1.0(9)	C(18)	C(19)	C(20)	C(15)	-0.7(8)
C(18)	C(19)	C(20)	C(22)	178.6(5)					

The sign is positive if when looking from atom 2 to atom 3 a clock-wise motion of atom 1 would superimpose it on atom 4.

Table 9. Distances beyond the asymmetric unit out to 3.60 Å

atom	atom	distance	atom	atom	distance
C(10)	C(17) ¹⁾	3.598(8)	C(17)	C(10) ¹⁾	3.598(8)
C(32)	C(32) ²⁾	3.345(7)			

Symmetry Operators:

(1) -X,-Y,-Z+1

(2) -X+1,-Y,-Z+1

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens

atom	atom	distance	atom	atom	distance
Si(2)	H(14) ¹⁾	3.551	C(1)	H(34) ²⁾	3.599
C(1)	H(43) ³⁾	3.166	C(2)	H(34) ²⁾	3.563
C(3)	H(8) ³⁾	3.483	C(3)	H(34) ²⁾	3.220
C(3)	H(53) ¹⁾	3.096	C(4)	H(23) ⁴⁾	3.280
C(4)	H(29) ⁴⁾	3.260	C(4)	H(51) ¹⁾	3.442
C(4)	H(53) ¹⁾	2.846	C(5)	H(18) ¹⁾	3.435
C(5)	H(19) ¹⁾	3.226	C(5)	H(23) ⁴⁾	3.279
C(5)	H(27) ¹⁾	3.248	C(5)	H(37) ⁵⁾	3.277
C(5)	H(39) ⁵⁾	3.393	C(5)	H(53) ¹⁾	3.191
C(6)	H(19) ¹⁾	3.354	C(6)	H(37) ⁵⁾	3.163
C(6)	H(39) ⁵⁾	3.225	C(7)	H(43) ³⁾	3.476
C(7)	H(44) ³⁾	3.149	C(8)	H(44) ³⁾	2.932
C(9)	H(1) ³⁾	3.277	C(9)	H(3) ³⁾	3.420
C(9)	H(13) ⁶⁾	3.190	C(9)	H(44) ³⁾	2.842
C(10)	H(3) ³⁾	3.277	C(10)	H(12) ⁶⁾	3.006
C(10)	H(13) ⁶⁾	3.288	C(10)	H(44) ³⁾	3.022
C(10)	H(50) ⁷⁾	3.555	C(10)	H(54) ⁷⁾	3.561
C(11)	H(12) ⁶⁾	3.381	C(11)	H(44) ³⁾	3.228
C(11)	H(45) ³⁾	3.581	C(11)	H(50) ⁷⁾	3.324
C(11)	H(54) ⁷⁾	3.430	C(12)	H(44) ³⁾	3.312
C(16)	H(12) ⁶⁾	3.502	C(17)	H(8) ⁶⁾	3.236
C(17)	H(12) ⁶⁾	3.298	C(17)	H(17) ⁶⁾	3.253
C(18)	H(8) ⁶⁾	3.319	C(18)	H(17) ⁶⁾	3.595
C(18)	H(36) ⁸⁾	3.057	C(18)	H(46) ⁸⁾	3.479
C(19)	H(30) ⁹⁾	3.559	C(19)	H(31) ⁹⁾	3.553
C(19)	H(32) ⁸⁾	3.247	C(19)	H(36) ⁸⁾	3.165
C(20)	H(30) ⁹⁾	3.301	C(21)	H(12) ⁶⁾	3.173
C(21)	H(46) ¹⁰⁾	3.304	C(21)	H(48) ¹⁰⁾	3.381
C(22)	H(5) ⁸⁾	3.355	C(22)	H(30) ⁹⁾	3.128
C(24)	H(24) ⁴⁾	3.598	C(24)	H(34) ²⁾	3.246
C(25)	H(4) ⁸⁾	3.402	C(25)	H(5) ⁸⁾	3.272
C(25)	H(28) ¹¹⁾	3.351	C(26)	H(4) ⁴⁾	3.207
C(26)	H(20) ⁹⁾	3.498	C(27)	H(14) ¹⁾	3.275
C(28)	H(2) ¹²⁾	3.341	C(28)	H(3) ¹²⁾	3.247
C(28)	H(7) ¹⁰⁾	3.291	C(28)	H(13) ¹⁾	3.354
C(28)	H(14) ¹⁾	3.459	C(28)	H(25) ¹²⁾	3.400
C(29)	H(5) ⁵⁾	3.377	C(29)	H(6) ⁵⁾	2.994

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
C(29)	H(14) ¹⁾	3.468	C(29)	H(21) ⁵⁾	3.503
C(29)	H(31) ⁵⁾	3.482	C(29)	H(38) ⁵⁾	3.582
C(29)	H(39) ⁵⁾	3.520	C(30)	H(30) ⁹⁾	3.321
C(31)	H(1) ³⁾	3.308	C(31)	H(44) ³⁾	3.523
C(31)	H(49) ³⁾	3.575	C(31)	H(50) ³⁾	3.389
C(31)	H(54) ¹⁾	3.342	C(32)	H(16) ¹⁰⁾	3.533
C(32)	H(17) ¹⁰⁾	3.342	C(32)	H(46) ¹⁰⁾	3.452
C(32)	H(47) ¹⁰⁾	2.854	C(32)	H(48) ¹⁰⁾	3.202
C(33)	H(8) ⁷⁾	3.595	C(33)	H(9) ⁷⁾	3.482
C(33)	H(43) ³⁾	3.449	C(33)	H(45) ³⁾	3.588
C(34)	H(45) ⁸⁾	3.279	H(1)	C(9) ³⁾	3.277
H(1)	C(31) ³⁾	3.308	H(1)	H(7) ³⁾	2.987
H(1)	H(43) ³⁾	2.681	H(1)	H(44) ³⁾	3.069
H(2)	C(28) ²⁾	3.341	H(2)	H(7) ³⁾	3.433
H(2)	H(34) ²⁾	2.868	H(2)	H(35) ²⁾	2.950
H(2)	H(43) ³⁾	2.786	H(3)	C(9) ³⁾	3.420
H(3)	C(10) ³⁾	3.277	H(3)	C(28) ²⁾	3.247
H(3)	H(7) ³⁾	2.968	H(3)	H(8) ³⁾	2.707
H(3)	H(34) ²⁾	2.847	H(3)	H(35) ²⁾	3.431
H(3)	H(36) ²⁾	2.954	H(3)	H(53) ¹⁾	3.415
H(4)	C(25) ¹⁾	3.402	H(4)	C(26) ⁴⁾	3.207
H(4)	H(23) ⁴⁾	3.093	H(4)	H(26) ¹⁾	3.152
H(4)	H(27) ¹⁾	3.247	H(4)	H(28) ¹⁾	3.234
H(4)	H(29) ⁴⁾	2.422	H(4)	H(31) ⁴⁾	3.196
H(4)	H(51) ¹⁾	2.914	H(4)	H(53) ¹⁾	3.031
H(5)	C(22) ¹⁾	3.355	H(5)	C(25) ¹⁾	3.272
H(5)	C(29) ⁵⁾	3.377	H(5)	H(18) ¹⁾	2.985
H(5)	H(19) ¹⁾	2.823	H(5)	H(23) ⁴⁾	3.086
H(5)	H(26) ¹⁾	3.389	H(5)	H(27) ¹⁾	2.489
H(5)	H(28) ¹⁾	3.520	H(5)	H(37) ⁵⁾	2.981
H(5)	H(39) ⁵⁾	2.894	H(5)	H(40) ⁵⁾	3.257
H(5)	H(53) ¹⁾	3.560	H(6)	C(29) ⁵⁾	2.994
H(6)	H(14) ¹⁾	3.514	H(6)	H(19) ¹⁾	3.050
H(6)	H(37) ⁵⁾	2.756	H(6)	H(38) ⁵⁾	3.237
H(6)	H(39) ⁵⁾	2.533	H(7)	C(28) ¹⁰⁾	3.291
H(7)	H(1) ³⁾	2.987	H(7)	H(2) ³⁾	3.433
H(7)	H(3) ³⁾	2.968	H(7)	H(13) ⁶⁾	2.837

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(7)	H(35) ¹⁰⁾	2.788	H(7)	H(36) ¹⁰⁾	2.972
H(7)	H(44) ³⁾	3.253	H(8)	C(3) ³⁾	3.483
H(8)	C(17) ⁶⁾	3.236	H(8)	C(18) ⁶⁾	3.319
H(8)	C(33) ⁷⁾	3.595	H(8)	H(3) ³⁾	2.707
H(8)	H(12) ⁶⁾	2.841	H(8)	H(13) ⁶⁾	3.016
H(8)	H(44) ³⁾	3.529	H(8)	H(50) ⁷⁾	3.067
H(8)	H(51) ⁷⁾	3.230	H(8)	H(54) ⁷⁾	3.378
H(9)	C(33) ⁷⁾	3.482	H(9)	H(9) ⁷⁾	3.123
H(9)	H(12) ⁶⁾	3.472	H(9)	H(45) ⁸⁾	3.259
H(9)	H(50) ⁷⁾	2.568	H(9)	H(54) ⁷⁾	3.123
H(12)	C(10) ⁶⁾	3.006	H(12)	C(11) ⁶⁾	3.381
H(12)	C(16) ⁶⁾	3.502	H(12)	C(17) ⁶⁾	3.298
H(12)	C(21) ⁶⁾	3.173	H(12)	H(8) ⁶⁾	2.841
H(12)	H(9) ⁶⁾	3.472	H(12)	H(12) ⁶⁾	2.819
H(12)	H(15) ⁶⁾	2.901	H(12)	H(17) ⁶⁾	2.792
H(13)	C(9) ⁶⁾	3.190	H(13)	C(10) ⁶⁾	3.288
H(13)	C(28) ⁸⁾	3.354	H(13)	H(7) ⁶⁾	2.837
H(13)	H(8) ⁶⁾	3.016	H(13)	H(15) ⁶⁾	3.472
H(13)	H(17) ⁶⁾	3.422	H(13)	H(32) ⁸⁾	3.553
H(13)	H(35) ⁸⁾	3.478	H(13)	H(36) ⁸⁾	2.486
H(13)	H(46) ⁸⁾	3.045	H(14)	Si(2) ⁸⁾	3.551
H(14)	C(27) ⁸⁾	3.275	H(14)	C(28) ⁸⁾	3.459
H(14)	C(29) ⁸⁾	3.468	H(14)	H(6) ⁸⁾	3.514
H(14)	H(30) ⁹⁾	3.567	H(14)	H(31) ⁹⁾	3.264
H(14)	H(32) ⁸⁾	2.478	H(14)	H(33) ⁸⁾	3.553
H(14)	H(36) ⁸⁾	2.682	H(14)	H(38) ⁸⁾	2.781
H(15)	H(12) ⁶⁾	2.901	H(15)	H(13) ⁶⁾	3.472
H(15)	H(46) ¹⁰⁾	3.427	H(16)	C(32) ¹⁰⁾	3.533
H(16)	H(46) ¹⁰⁾	3.288	H(16)	H(48) ¹⁰⁾	2.897
H(17)	C(17) ⁶⁾	3.253	H(17)	C(18) ⁶⁾	3.595
H(17)	C(32) ¹⁰⁾	3.342	H(17)	H(12) ⁶⁾	2.792
H(17)	H(13) ⁶⁾	3.422	H(17)	H(46) ¹⁰⁾	2.697
H(17)	H(48) ¹⁰⁾	3.099	H(18)	C(5) ⁸⁾	3.435
H(18)	H(5) ⁸⁾	2.985	H(19)	C(5) ⁸⁾	3.226
H(19)	C(6) ⁸⁾	3.354	H(19)	H(5) ⁸⁾	2.823
H(19)	H(6) ⁸⁾	3.050	H(19)	H(30) ⁹⁾	3.157
H(19)	H(38) ⁸⁾	3.103	H(19)	H(39) ⁹⁾	3.327

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens
(continued)

atom	atom	distance	atom	atom	distance
H(20)	C(26) ⁹⁾	3.498	H(20)	H(22) ⁹⁾	3.445
H(20)	H(30) ⁹⁾	2.573	H(21)	C(29) ⁵⁾	3.503
H(21)	H(37) ⁵⁾	3.159	H(21)	H(38) ⁵⁾	3.153
H(22)	H(20) ⁹⁾	3.445	H(23)	C(4) ⁴⁾	3.280
H(23)	C(5) ⁴⁾	3.279	H(23)	H(4) ⁴⁾	3.093
H(23)	H(5) ⁴⁾	3.086	H(23)	H(24) ⁴⁾	3.093
H(23)	H(34) ²⁾	3.346	H(23)	H(37) ²⁾	3.455
H(23)	H(40) ²⁾	3.507	H(24)	C(24) ⁴⁾	3.598
H(24)	H(23) ⁴⁾	3.093	H(24)	H(24) ⁴⁾	3.211
H(24)	H(34) ²⁾	3.519	H(24)	H(37) ⁵⁾	3.160
H(25)	C(28) ²⁾	3.400	H(25)	H(34) ²⁾	2.462
H(25)	H(40) ²⁾	3.550	H(26)	H(4) ⁸⁾	3.152
H(26)	H(5) ⁸⁾	3.389	H(27)	C(5) ⁸⁾	3.248
H(27)	H(4) ⁸⁾	3.247	H(27)	H(5) ⁸⁾	2.489
H(27)	H(28) ¹¹⁾	3.408	H(27)	H(40) ⁹⁾	3.107
H(28)	C(25) ¹¹⁾	3.351	H(28)	H(4) ⁸⁾	3.234
H(28)	H(5) ⁸⁾	3.520	H(28)	H(27) ¹¹⁾	3.408
H(28)	H(28) ¹¹⁾	2.534	H(28)	H(29) ¹¹⁾	3.111
H(28)	H(40) ²⁾	3.485	H(29)	C(4) ⁴⁾	3.260
H(29)	H(4) ⁴⁾	2.422	H(29)	H(28) ¹¹⁾	3.111
H(29)	H(40) ⁹⁾	3.337	H(29)	H(42) ⁹⁾	3.554
H(30)	C(19) ⁹⁾	3.559	H(30)	C(20) ⁹⁾	3.301
H(30)	C(22) ⁹⁾	3.128	H(30)	C(30) ⁹⁾	3.321
H(30)	H(14) ⁹⁾	3.567	H(30)	H(19) ⁹⁾	3.157
H(30)	H(20) ⁹⁾	2.573	H(30)	H(40) ⁹⁾	3.162
H(30)	H(41) ⁹⁾	2.849	H(30)	H(42) ⁹⁾	3.426
H(31)	C(19) ⁹⁾	3.553	H(31)	C(29) ⁵⁾	3.482
H(31)	H(4) ⁴⁾	3.196	H(31)	H(14) ⁹⁾	3.264
H(31)	H(36) ⁵⁾	3.391	H(31)	H(37) ⁵⁾	2.912
H(31)	H(38) ⁵⁾	3.195	H(32)	C(19) ¹⁾	3.247
H(32)	H(13) ¹⁾	3.553	H(32)	H(14) ¹⁾	2.478
H(33)	H(14) ¹⁾	3.553	H(33)	H(48) ¹⁰⁾	3.395
H(34)	C(1) ¹²⁾	3.599	H(34)	C(2) ¹²⁾	3.563
H(34)	C(3) ¹²⁾	3.220	H(34)	C(24) ¹²⁾	3.246
H(34)	H(2) ¹²⁾	2.868	H(34)	H(3) ¹²⁾	2.847
H(34)	H(23) ¹²⁾	3.346	H(34)	H(24) ¹²⁾	3.519
H(34)	H(25) ¹²⁾	2.462	H(35)	H(2) ¹²⁾	2.950

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(35)	H(3) ¹²⁾	3.431	H(35)	H(7) ¹⁰⁾	2.788
H(35)	H(13) ¹⁾	3.478	H(35)	H(43) ¹⁰⁾	3.050
H(35)	H(48) ¹⁰⁾	2.910	H(35)	H(49) ¹²⁾	3.414
H(36)	C(18) ¹⁾	3.057	H(36)	C(19) ¹⁾	3.165
H(36)	H(3) ¹²⁾	2.954	H(36)	H(7) ¹⁰⁾	2.972
H(36)	H(13) ¹⁾	2.486	H(36)	H(14) ¹⁾	2.682
H(36)	H(31) ⁵⁾	3.391	H(37)	C(5) ⁵⁾	3.277
H(37)	C(6) ⁵⁾	3.163	H(37)	H(5) ⁵⁾	2.981
H(37)	H(6) ⁵⁾	2.756	H(37)	H(21) ⁵⁾	3.159
H(37)	H(23) ¹²⁾	3.455	H(37)	H(24) ⁵⁾	3.160
H(37)	H(31) ⁵⁾	2.912	H(38)	C(29) ⁵⁾	3.582
H(38)	H(6) ⁵⁾	3.237	H(38)	H(14) ¹⁾	2.781
H(38)	H(19) ¹⁾	3.103	H(38)	H(21) ⁵⁾	3.153
H(38)	H(31) ⁵⁾	3.195	H(38)	H(38) ⁵⁾	3.398
H(38)	H(39) ⁵⁾	2.922	H(39)	C(5) ⁵⁾	3.393
H(39)	C(6) ⁵⁾	3.225	H(39)	C(29) ⁵⁾	3.520
H(39)	H(5) ⁵⁾	2.894	H(39)	H(6) ⁵⁾	2.533
H(39)	H(19) ⁹⁾	3.327	H(39)	H(38) ⁵⁾	2.922
H(39)	H(39) ⁵⁾	3.265	H(40)	H(5) ⁵⁾	3.257
H(40)	H(23) ¹²⁾	3.507	H(40)	H(25) ¹²⁾	3.550
H(40)	H(27) ⁹⁾	3.107	H(40)	H(28) ¹²⁾	3.485
H(40)	H(29) ⁹⁾	3.337	H(40)	H(30) ⁹⁾	3.162
H(41)	H(30) ⁹⁾	2.849	H(42)	H(29) ⁹⁾	3.554
H(42)	H(30) ⁹⁾	3.426	H(42)	H(49) ¹²⁾	3.528
H(43)	C(1) ³⁾	3.166	H(43)	C(7) ³⁾	3.476
H(43)	C(33) ³⁾	3.449	H(43)	H(1) ³⁾	2.681
H(43)	H(2) ³⁾	2.786	H(43)	H(35) ¹⁰⁾	3.050
H(43)	H(44) ³⁾	3.452	H(43)	H(49) ³⁾	2.919
H(43)	H(50) ³⁾	3.138	H(44)	C(7) ³⁾	3.149
H(44)	C(8) ³⁾	2.932	H(44)	C(9) ³⁾	2.842
H(44)	C(10) ³⁾	3.022	H(44)	C(11) ³⁾	3.228
H(44)	C(12) ³⁾	3.312	H(44)	C(31) ³⁾	3.523
H(44)	H(1) ³⁾	3.069	H(44)	H(7) ³⁾	3.253
H(44)	H(8) ³⁾	3.529	H(44)	H(43) ³⁾	3.452
H(44)	H(44) ³⁾	2.958	H(44)	H(54) ¹⁾	3.373
H(45)	C(11) ³⁾	3.581	H(45)	C(33) ³⁾	3.588
H(45)	C(34) ¹⁾	3.279	H(45)	H(9) ¹⁾	3.259

Table 10. Distances beyond the asymmetric unit out to 3.60 Å involving hydrogens (continued)

atom	atom	distance	atom	atom	distance
H(45)	H(49) ³⁾	3.476	H(45)	H(50) ³⁾	2.879
H(45)	H(52) ¹⁾	3.280	H(45)	H(54) ¹⁾	2.528
H(46)	C(18) ¹⁾	3.479	H(46)	C(21) ¹⁰⁾	3.304
H(46)	C(32) ¹⁰⁾	3.452	H(46)	H(13) ¹⁾	3.045
H(46)	H(15) ¹⁰⁾	3.427	H(46)	H(16) ¹⁰⁾	3.288
H(46)	H(17) ¹⁰⁾	2.697	H(46)	H(47) ¹⁰⁾	2.869
H(46)	H(48) ¹⁰⁾	3.215	H(47)	C(32) ¹⁰⁾	2.854
H(47)	H(46) ¹⁰⁾	2.869	H(47)	H(47) ¹⁰⁾	2.651
H(47)	H(48) ¹⁰⁾	2.563	H(48)	C(21) ¹⁰⁾	3.381
H(48)	C(32) ¹⁰⁾	3.202	H(48)	H(16) ¹⁰⁾	2.897
H(48)	H(17) ¹⁰⁾	3.099	H(48)	H(33) ¹⁰⁾	3.395
H(48)	H(35) ¹⁰⁾	2.910	H(48)	H(46) ¹⁰⁾	3.215
H(48)	H(47) ¹⁰⁾	2.563	H(48)	H(48) ¹⁰⁾	3.352
H(49)	C(31) ³⁾	3.575	H(49)	H(35) ²⁾	3.414
H(49)	H(42) ²⁾	3.528	H(49)	H(43) ³⁾	2.919
H(49)	H(45) ³⁾	3.476	H(50)	C(10) ⁷⁾	3.555
H(50)	C(11) ⁷⁾	3.324	H(50)	C(31) ³⁾	3.389
H(50)	H(8) ⁷⁾	3.067	H(50)	H(9) ⁷⁾	2.568
H(50)	H(43) ³⁾	3.138	H(50)	H(45) ³⁾	2.879
H(51)	C(4) ⁸⁾	3.442	H(51)	H(4) ⁸⁾	2.914
H(51)	H(8) ⁷⁾	3.230	H(52)	H(45) ⁸⁾	3.280
H(53)	C(3) ⁸⁾	3.096	H(53)	C(4) ⁸⁾	2.846
H(53)	C(5) ⁸⁾	3.191	H(53)	H(3) ⁸⁾	3.415
H(53)	H(4) ⁸⁾	3.031	H(53)	H(5) ⁸⁾	3.560
H(54)	C(10) ⁷⁾	3.561	H(54)	C(11) ⁷⁾	3.430
H(54)	C(31) ⁸⁾	3.342	H(54)	H(8) ⁷⁾	3.378
H(54)	H(9) ⁷⁾	3.123	H(54)	H(44) ⁸⁾	3.373
H(54)	H(45) ⁸⁾	2.528			

Symmetry Operators:

- | | |
|--------------------|-------------------|
| (1) X+1,Y,Z | (2) X,Y+1,Z |
| (3) -X+1,-Y+1,-Z+1 | (4) -X+1,-Y+1,-Z |
| (5) -X+1,-Y,-Z | (6) -X,-Y,-Z+1 |
| (7) -X,-Y+1,-Z+1 | (8) X-1,Y,Z |
| (9) -X,-Y,-Z | (10) -X+1,-Y,-Z+1 |
| (11) -X,-Y+1,-Z | (12) X,Y-1,Z |