

Supporting Information for:

“Hydration Energies of Zinc (II): Threshold Collision-induced Dissociation Experiments and Theoretical Studies” by Theresa Cooper, D. R. Carl, and P. B. Armentrout

Theoretical geometries of zinc water clusters. As described the text, geometry optimizations and frequency calculations were performed at the B3LYP/6-311+G(d,p) level of theory for $\text{Zn}^{2+}(\text{H}_2\text{O})_n(\text{H}_2\text{O})_m$, where n and m are the number of waters in the inner and outer shells, respectively, and complexes considered included $n = 1 - 4$ and $m = 0$, $n = 4$ and $m = 1 - 6$, $n = 5$ and $m = 0 - 5$, and $n = 6$ and $m = 0 - 4$. Single point energies including zero point energy corrections are calculated at four different levels of theory with a 6-311+G(2d,2p) basis set and relative energies are given for the lowest energy isomers in Table 3. Figures S1 and S2 shows the low-lying isomers of $\text{Zn}^{2+}(\text{H}_2\text{O})_9$ and $\text{Zn}^{2+}(\text{H}_2\text{O})_{10}$. In addition, a number of high energy isomers were investigated for all inner solvent shell sizes and this section details the geometries and relative energies of all isomers calculated for the zinc water complexes that are not explicitly described in the text. Tables S1 and S2 provide relative energies and critical bond angles and lengths, respectively, of all complexes. Figure S3 shows their geometries. Table S3 provides the experimental and theoretical 0 K bond energies for all reactant and product isomers analyzed in Table 3. Finally, Table S4 is a conversion of the 0 K thresholds in Table S3 to 298 K enthalpies and free energies.

Figures S3a and S3b show the structures of $\text{Zn}^{2+}(\text{H}_2\text{O})_5$, which are described in the text. Figures S3c and S3d exhibit the geometries of the $\text{Zn}^{2+}(\text{H}_2\text{O})_6$ complexes. In contrast to the (4,2)_2AA GS, the (4,2)_AA,A isomer has one second solvent shell water forming only one H-bond, thereby raising its energy by 4 – 8 kJ/mol, Table S1. In the (5,1)_A_aA_b and (5,1)_A_aA_bsym structures, where A_aA_b denotes a second water binding to both apex and base inner shell waters. In both of these structures, the sixth water binds in a similar fashion, such that (5,1)_A_aA_bsym is only slightly higher in energy (3 - 4 kJ/mol), Table S1. The (5,1)_A_aA_bsym structure maintains the symmetry of the inner solvent water molecules in the base found in both the (5,0) complexes, requiring that the apex water rotates to accommodate binding the sixth water bonding in a bridging site. The (5,1)_A_a isomer has the sixth water singly hydrogen bound to the apex of the square pyramid inner shell, whereas in (5,1)_A_bA_b, this water forms two H-bonds to the downward angled waters of the square pyramid base of (5,0).

Figures S3e and S3f show the structures of the $\text{Zn}^{2+}(\text{H}_2\text{O})_7$ complexes. The (4,3) complexes include three isomers higher in energy than the (4,3)_3D,DD_2AA,A GS. The (4,3)_4D_AA,2A structure has two of the three second shell waters forming single H-bonds with different inner shell waters. Slightly higher in energy, (4,3)_2D,2DD_AA,2A is favored by free energy at the Bx levels. As discussed in the text, the (4,3)_2D,2DD_3AA isomer is highest in energy because of distortions in the inner shell tetrahedron as detailed in Table S2. From the (5,2)_4D_2A_bA_b GS, four high energy isomers were located, 1 - 12 kJ/mol higher in energy. The (5,2)_2D,DD_2A_aA_b structure allows both second shell waters to form two H-bonds to the inner shell, but are now bound between inner shell waters at the apex and base of the square

pyramid. The (5,2)_2D,DD_2A_bA_b structure is similar to (5,2)_4D_2A_bA_b GS except the second shell waters both H-bond to one inner shell water on the same side of the square pyramid. The (5,2)_3D_A_bA_b,A_{a,trans} takes the (5,2)_4D_2A_bA_b GS structure and moves one of the second shell waters so that it is singly bound to the apex of the square pyramid and oriented trans to the AA water. Similar to this structure is the (5,2)_3D_A_bA_b,A_{a,cis} complex, where now the A_a water is on the same side of the square pyramidal inner shell as the A_bA_b formation. This is the highest energy (5,2) complex.

We located six different (4,4) isomers, Figure S3g, each of which is progressively 1 – 4 kJ/mol higher in energy, Table S1. The (4,4)_2D,2DD_2AA,2A GS structure adds two single acceptor water molecules to the (4,2) GS or one to (4,3) GS. In (4,4)_3D,DD_AA,3A, only one second shell water forms two H-bonds with the inner shell, and three second shell waters form only one H-bond to the inner shell, with one of these sharing one inner shell water with the doubly H-bonded second shell water. Similar to the (4,4)_2D,2DD_2AA,2A GS, the (4,4)_2D,2DD_2AA,2A_share structure, two second shell waters are double H-bonded but share one inner shell water, with the remaining two second shell waters both singly H-bonded. The (4,4)_3D,DD_AA,A,AD,AA_p complex is similar to (4,4)_3D,DD_AA,3A except that two of the outer shell waters (those across from the double H-bonded second shell water) H-bond to one another, thereby forming a “pseudo third shell” and a ring-like series of H-bonds. This pseudo formation is denoted AD,AA_p, where the AD is both a single acceptor and a donor water molecule, and the AA_p is a double acceptor water that H-bonds to both the first and second solvent shell. The (4,4)_3D,DD_AA,A,AD,AA_p_share is again similar to the previous structure except now the doubly bound second shell water shares the inner shell water with the pseudo shell. The highest energy (4,4) isomer, (4,4)_2D,2DD_2AA,AD,AA_p, has two second shell waters forming two H-bonds to the inner shell and the remaining two in the H-bonded pseudo third shell. The (5,3) complex can form three isomers, the (5,3)_3AA GS, (5,3)_2AA,A, and (5,3)_AA,AAD,AA_p structures, Figure S3h, where each are ~ 7 kJ/mol higher in energy than each other at the Bx levels and 10 - 16 kJ/mol higher in energy than the GS at the Mx level. The (5,3)_2AA,A adds a water by a single H-bond to the apex of the square pyramid inner solvent shell water of (5,2)_2D,DD_2A_bA_b. The (5,3)_AA,AAD,AA_p structure H-bonds the third second shell water to the apex site and to one of the other second shell waters of (5,2), again creating this “pseudo third shell”. The (6,2)_2D,DD_2AA GS and (6,2)_4D_2AA structures of Figure S3i are discussed previously in the text.

For the Zn²⁺(H₂O)₉ complex, there are five high energy isomers located with a 4-coordinate inner shell, Figure S3j. The (4,5)_2D,2DD_AA,2A,AD,AA_{p,trans} again demonstrates the ability of two second shell waters to H-bond to one another, with the remaining three waters forming a single, double, single H-bond pattern on the opposite (trans) side of the tetrahedral inner shell. Similar in energy and geometry is the (4,5)_2D,2DD_AA,2A,AD,AA_{p,cis} where now the H-bond pattern is single, single, double opposite of the pseudo water formation. The (4,5)_D,3DD_2AA,A,AD,AA_p complex now has a double, single, double pattern for the remaining three second shell waters, where two of these share inner shell waters with the pseudo

formation. The remaining isomers are very similar in energy and involve the formation of a true third solvent shell, (4,4,1). Both (4,4,1)_3D,DD_AA,A,2AD_AA and (4,4,1)_4D_2A,2AD_AA are 20 kJ/mol higher in energy than the (4,5)_D,3DD_2AA,3A GS at the Bx levels and 37 – 57 kJ/mol higher at the Mx levels. In (4,4,1) _3D,DD_AA,A,2AD_AA, one of the second shell waters is doubly hydrogen bound to the inner shell, whereas in (4,4,1)_ 4D_2A,2AD_AA , all second solvent shell waters are singly bound to the first shell. Close in energy to the (5,4)_4A_bA_b GS is the (5,4)_2A_bA_b,2A_aA_b structure, Figure S3k, where two waters H-bond between the apex and base of the square pyramid and two waters form two H-bonds at the base of the pyramid maximizing the distance between each other. The remaining isomers have a similar naming scheme as discussed above where two of the complexes form a single H-bond to the apex of the pyramid, (5,4)_3A_bA_b,A_a and (5,4)_2A_bA_b,A_aA_b,A_a. The highest energy isomers both form a pseudo third shell with one water bound to both the first and second shell, (5,4)_AA,A,AAD,AA_p and (5,4)_2AA,AD,AA_p. The only high energy six coordinate isomer, (6,3)_4D,DD_3AA, is ~3 kJ/mol higher in energy than the (6,3)_6D_3AA GS because one inner shell water is shared between two second shell waters, Figure S3l.

Figures S3m, S3n, and S3o show the 4-, 5-, and 6-coordinate structures of Zn²⁺(H₂O)₁₀, respectively. The (4,6)_2AA,4A GS forms a highly symmetric structure and is 10 and 20 kJ/mol lower in energy than the (4,5,1)_2AA,A,2AD_AA and (4,5,1)_AA,2A,AAD,AD_AA structures, where the tenth water forms a third solvent shell. The highest energy isomer is a (4,4,2)_4AD_2AA structure where two waters are both doubly H-bonded to pairs of waters in the second solvent shell. The (5,5)_4A_bA_b,A_a GS structure forms two higher energy isomers, similar in energy to each other. The (5,4,1)_3A_bA_b,A_bA_bD_A structure is similar to the (5,5) GS except the tenth water is now singly bound in the third shell at the base of the square pyramid. The (5,5)_3A_bA_b,A_bA_bD,AA_p allows the water hydrogen bound to the apex site in (5,5) GS to hydrogen bind to one of the other second shell waters. Slightly higher in energy than the (6,4)_4D,2DD_4AA GS, the (6,4)_4D,2DD_4AA_share structure has two inner solvent shell waters sharing with different second shell waters. Similar in energy is the (6,4)_5D,DD_3AA,A complex which takes (6,3) GS and singly H-bonds the tenth water molecule to the inner shell. The (6,4)_5D,DD_2AA,AD,AA_p complex is the highest energy isomer and H-bonds the singly bound second shell water of (6,4)_5D,DD_3AA,A with another second shell water.

Table S1. Relative calculated enthalpy (ΔH_0) and free energies (ΔG_{298})^a (kJ/mol) of $Zn^{2+}(H_2O)_x(H_2O)_y$ where x is the number of waters in the inner solvent shell and y is the number of waters in the second shell.^b

Complex (x,y)		B3LYP	B3P86	M06	MP2(full)
$Zn^{2+}(H_2O)_5$	(4,1)_AA	0.0 (2.6)	0.0 (2.9)	6.5 (8.2)	3.3 (4.3)
	(4,1)_A	1.1 (0.0)	0.8 (0.0)	14.9 (12.9)	8.4 (5.7)
	(5,0)	4.2 (5.8)	4.6 (6.4)	0.0 (0.6)	0.0 (0.0)
	(5,0)_switch	5.8 (5.4)	6.0 (5.8)	1.4 (0.0)	2.4 (0.3)
$Zn^{2+}(H_2O)_6$	(4,2)_2AA	0.0 (0.0)	0.0 (0.0)	13.1 (12.8)	3.8 (7.5)
	(4,2)_AA,A	4.5 (0.5)	4.4 (0.4)	26.1 (21.8)	11.7 (11.4)
	(5,1)_A _a A _b	8.2 (0.3)	8.6 (0.6)	12.2 (4.1)	4.2 (0.0)
	(5,1)_A _a A _b _sym	11.7 (8.0)	12.3 (8.6)	16.6 (12.8)	7.7 (7.8)
	(5,1)_A _a	17.1 (3.1)	17.4 (3.3)	26.6 (12.4)	16.9 (6.6)
	(5,1)_A _b A _b	18.9 (14.8)	19.2 (15.1)	21.9 (17.5)	15.3 (14.9)
	(6,0)	13.7 (14.0)	14.6 (14.8)	0.0 (0.0)	0.0 (3.9)
	(4,3)_3D,DD_2AA,A	0.0 (6.5)	0.0 (6.5)	14.9 (9.6)	10.8 (5.5)
$Zn^{2+}(H_2O)_7$	(4,3)_4D_AA,2A	1.9 (5.4)	1.6 (5.0)	21.6 (13.2)	16.1 (7.7)
	(4,3)_2D,DD_AA,2A	4.1 (0.0)	4.1 (0.0)	23.6 (7.7)	18.0 (2.1)
	(4,3)_2D,2DD_3AA	10.1 (23.0)	10.2 (23.0)	17.9 (18.9)	18.1 (19.1)
	(5,2)_4D_2A _b A _b	0.2 (12.0)	0.1 (12.0)	0.0 (0.0)	0.0 (0.0)
	(5,2)_2D,2DD_2A _a A _b	4.6 (14.6)	5.0 (15.0)	5.4 (3.6)	4.2 (2.4)
	(5,2)_2D,DD_2A _b A _b	7.3 (16.9)	7.4 (17.0)	8.8 (6.6)	7.0 (4.8)
	(5,2)_3D_A _b A _b ,A _{a,trans}	7.5 (11.8)	7.4 (11.7)	11.5 (4.0)	10.4 (2.9)
	(5,2)_3D_A _b A _b ,A _{a,cis}	8.4 (13.2)	8.4 (13.1)	15.7 (8.6)	11.5 (4.4)
	(6,1)_AA	18.4 (31.3)	18.7 (31.6)	3.2 (4.3)	8.7 (9.7)
	$Zn^{2+}(H_2O)_8$	(4,4)_2D,2DD_2AA,2A	1.8 (0.0)	2.4 (0.1)	21.4 (12.5)

	(4,4)_3D,DD_AA,3A	4.2 (0.0)	4.8 (0.0)	28.2 (16.9)	20.7 (9.4)
	(4,4)_2D,2DD_2AA,2A_share	7.2 (3.0)	7.7 (2.9)	27.3 (15.9)	20.8 (9.4)
	(4,4)_3D,DD_AA,A,AD,AA _p	8.9 (6.6)	9.6 (6.6)	31.5 (21.0)	23.0 (13.5)
	(4,4)_3D,DD_AA,A,AD,AA _p	12.7 (8.1)	13.5 (8.4)	35.0 (23.3)	26.7 (15.0)
	(4,4)_2D,2DD_2AA,AD,AA _p	16.5 (19.2)	17.1 (19.4)	32.6 (28.3)	27.5 (23.1)
	(5,3)_3AA	0.0 (7.1)	0.0 (6.6)	0.0 (0.0)	0.0 (0.0)
	(5,3)_2AA,A	7.6 (7.9)	7.9 (7.6)	15.5 (8.6)	10.6 (3.7)
	(5,3)_AA,AAD,AA _p	14.0 (21.4)	14.3 (21.2)	12.4 (12.7)	13.2 (13.5)
	(6,2)_2D,DD_2AA	16.8 (23.5)	17.7 (23.9)	2.1 (1.7)	6.7 (6.3)
	(6,2)_4D_2AA	17.7 (25.0)	17.8 (24.6)	2.0 (2.2)	7.3 (7.5)
Zn ²⁺ (H ₂ O) ₉	(4,5)_D,3DD_2AA,3A	5.8 (0.0)	6.3 (0.0)	30.7 (14.6)	22.1 (6.0)
	(4,5)_2D,2DD_AA,2A,AD,AA _{p,trans}	11.9 (6.5)	12.5 (6.6)	40.0 (24.3)	28.8 (13.1)
	(4,5)_2D,2DD_AA,2A,AD,AA _{p,cis}	13.9 (9.8)	14.4 (9.7)	41.8 (27.4)	30.6 (16.2)
	(4,5)_D,3DD_2AA,A,AD,AA _p	15.0 (16.9)	15.5 (16.9)	37.5 (29.1)	28.9 (20.5)
	(4,4,1)_3D,DD_AA,A,2AD_AA	18.7 (23.0)	18.8 (22.6)	49.1 (43.1)	37.0 (30.9)
	(4,4,1)_4D_2A,2AD_AA	19.9 (10.4)	19.7 (9.6)	57.1 (37.2)	42.1 (22.2)
	(5,4)_4A _b A _b	0.0 (10.3)	0.0 (9.8)	0.0 (0.0)	0.0 (0.0)
	(5,4)_2A _b A _b ,2A _a A _b	1.25 (12.8)	0.8 (11.8)	3.8 (5.0)	1.9 (3.1)
	(5,4)_3A _b A _b ,A _a	7.0 (9.5)	7.1 (9.0)	12.7 (4.9)	10.2 (2.5)
	(5,4)_2A _b A _b ,A _a A _b ,A _a	9.2 (12.9)	9.1 (12.4)	16.7 (10.2)	12.5 (6.0)
	(5,4)_AA,A,AAD,AA _p	11.4 (19.6)	11.4 (19.1)	12.1 (10.0)	11.1 (9.0)
	(5,4)_2AA,AD,AA _p	14.9 (19.3)	14.8 (18.7)	23.6 (17.7)	19.1 (13.2)
	(6,3)_6D_3AA	13.5 (27.9)	13.1 (26.9)	2.8 (7.0)	3.9 (8.0)
	(6,3)_4D,DD_3AA	16.9 (26.0)	16.7 (25.3)	4.5 (3.4)	7.2 (6.0)
Zn ²⁺ (H ₂ O) ₁₀	(4,6)_2AA,4A	3.5 (0.0)	4.4 (0.0)	26.9 (16.3)	18.6 (8.0)
	(4,5,1)_2AA,A,2AD_AA	13.7 (18.7)	14.1 (18.2)	38.4 (36.2)	27.7 (25.6)

(4,5,1)_AA,A,AAD,AD_AA	22.1 (24.7)	21.9 (23.7)	44.6 (40.2)	35.8 (31.5)
(4,4,2)_4AD_2AA	22.9 (24.6)	22.5 (23.3)	58.1 (52.6)	41.7 (36.3)
(5,5)_4A _b A _b ,A _a	0.0 (7.1)	0.0 (6.2)	0.0 (0.0)	0.0 (0.0)
(5,4,1)_3A _b A _b ,AAD_A	5.6 (15.8)	4.4 (13.7)	5.4 (8.5)	6.7 (9.8)
(5,5)_3A _b A _b ,A _b A _b D,AA _p	6.0 (21.7)	6.0 (20.9)	0.6 (9.2)	2.2 (10.9)
(6,4)_4D,2DD_4AA	18.1 (35.5)	16.9 (33.5)	0.0 (10.4)	5.5 (15.9)
(6,4)_4D,2DD_4AA_share	19.6 (34.2)	19.0 (32.8)	3.2 (10.7)	6.8 (14.3)
(6,4)_5D,DD_3AA,A	19.8 (29.3)	19.7 (28.4)	4.0 (6.5)	8.7 (11.2)
(6,4)_5D,DD_2AA,AAD,AA _p	20.3 (37.1)	19.6 (35.6)	0.1 (9.9)	6.6 (16.4)

^a ΔG_{298} values given in parentheses.

^b Values are single point energies calculated at the level shown using a 6-311+G(2d,2p) basis set with geometries calculated at the B3LYP/6-311+G(d,p) level. Zero point energy corrections are included.

Table S2. B3LYP/6-311+G(d,p) geometry optimized structures for the ground state of all inner shell sizes of $\text{Zn}^{2+}(\text{H}_2\text{O})_{5-10}$ ^a

Complex	Sym	r(ZnO) (Å)	∠OZnO (°) ^b	∠ZnOH (°)	r(OH) (Å) ^c	r(O••H) (Å) ^d	∠HOH (°)
(4,1)_AA	C ₂	1.984 (2)	95.0, 104.5, 111.3 (2), 117.6 (2)	117.8 (2), 130.8 (2)	0.988 (2), 0.970 (2)		110.4 (2)
		2.008 (2)		125.0 (2), 126.8 (2)	0.970 (4)		107.6 (2)
		3.642		127.0	0.974 (2)	1.812 (2)	106.0
(4,1)_A	C ₁	1.943	103.9, 104.6, 110.6 (2), 111.4, 115.8	127.0, 124.3	1.034, 0.969		108.6
		2.014		126.1, 125.8	0.969 (2)		107.8
		2.020		124.7, 127.7	0.970 (2)		107.5
		2.022		124.6, 127.7	0.970 (2)		107.5
		3.993		126.7, 126.4	0.967 (2)	1.486	106.9
(5,0)	C _{2v}	2.038 (2)	88.0 (4), 95.9 (2), 109.8 (2), 140.4,	126.0 (4)	0.968 (4)		108.0 (2)
		2.055	168.1	126.6 (2)	0.965 (2)		106.8
		2.127 (2)		124.6 (2), 128.4 (2)	0.969 (4)		106.9 (2)
(5,0)_switch	C _{2v}	2.042	87.6 (4), 98.9 (2), 105.9 (2), 148.3,	127.5 (2)	0.953 (2)		104.9
		2.065 (2)	162.3	126.4 (4)	0.967 (4)		107.1 (2)
		2.107 (2)		126.9 (4)	0.969 (4)		107.1 (2)
(4,2)_2AA	D _{2d}	1.992 (4)	93.0 (2), 118.3 (4)	119.1 (4), 130.3 (4)	0.986(4), 0.967 (4)		110.6 (4)
		3.688 (2)		127.4 (4)	0.970 (4)	1.816 (4)	105.2 (2)
(4,2)_AA,A	C ₁	1.950	93.4, 104.9, 107.4, 112.9, 116.7, 121.4	126.8, 124.0	1.026, 0.966		108.6
		1.996		118.9, 130.6	0.985, 0.967		110.5
		2.002		118.6, 130.5	0.985, 0.967		110.3
		2.028		126.4, 125.7	0.969 (2)		107.9
		3.689		126.6, 128.3	0.970 (2)	1.823 (2)	105.2
		4.011		125.1, 128.1	0.966 (2)	1.513	106.8
(5,1)_A _a A _b	C ₁	2.020	86.2, 86.6, 89.2, 90.9, 93.3, 93.7,	122.1, 127.9	0.986, 0.966		110.0
		2.051	114.0, 121.8, 124.2, 179.9	124.3, 127.7	0.968 (2)		108.0
		2.084		124.7, 126.5	0.968 (2)		107.2
		2.087		121.0, 129.1	0.982, 0.966		109.2
		2.123		123.2, 128.7	0.968 (2)		107.2

(5,1)_A _a A _b _sym	C ₁	3.847 2.021 2.048 2.051 2.102 2.148 3.819 2.002 2.044 2.086 2.123 (2) 4.073 2.034 2.041 (2)	86.8, 87.2, 88.1, 90.1, 90.8, 97.3, 117.4, 118.3, 124.3, 174.6 88.6 (2), 89.1 (2), 92.0 (2), 111.1, 119.2, 129.7, 175.9 131.9 (2), 169.0	129.0, 125.7 119.6, 131.2 125.9, 125.2 125.7, 125.2 119.3, 131.9 125.3, 128.2 127.2, 127.4 128.9, 124.7 125.4 (2) 126.7, 126.1 123.8 (2), 127.8 (2) 126.8 (2) 125.6 (2) 116.0 (2), 130.5 (2)	0.969 (2) 0.969 (2) 0.967 (2) 0.967 (2) 0.982, 0.966 0.968 (2) 0.969 (2) 1.016, 0.966 0.970 (2) 0.968 (2) 0.967 (4) 0.965 (2) 0.968 (2) 0.984 (2), 0.967 (2)	1.786, 1.850 1.847, 1.774 1.016, 0.966 1.556 1.851 (2)	105.3 107.9 107.9 107.7 108.8 106.5 105.3 108.4 108.9 107.3 107.2 (2) 106.3 108.7 109.5 (2)
(5,1)_A _a	C _s	2.002 2.044 2.086 2.123 (2)	88.6 (2), 89.1 (2), 92.0 (2), 111.1, 119.2, 129.7, 175.9 131.9 (2), 169.0	128.9, 124.7 125.4 (2) 126.7, 126.1 123.8 (2), 127.8 (2)	1.016, 0.966 0.970 (2) 0.968 (2) 0.967 (4)		108.4 108.9 107.3 107.2 (2)
(5,1)_A _b A _b	C ₂	2.034 2.041 (2)	85.5 (2), 88.9 (2), 96.2, 98.4 (2), 131.9 (2), 169.0	125.6 (2) 116.0 (2), 130.5 (2)	0.968 (2) 0.984 (2), 0.967 (2)		108.7 109.5 (2)
(6,0)	T _h	2.128 (6)	90 (12), 180 (3)	122.0 (2), 129.1 (2)	0.968 (4)		107.1 (2)
(4,3)_3D,DD _2AA,A	C _s	1.948 2.000 2.004 (2)	91.9, 96.0, 114.6 (2), 120.7 (2)	127.6 (2) 118.3, 130.6 118.0, 131.2 119.7 (2), 129.8 (2)	0.970 (2) 0.971 (12) 0.976, 1.011 0.988, 0.966 0.984 (2), 0.966 (2)	1.851 (2)	104.9 108.1 (6) 111.1 110.8 110.5 (2)
(4,3)_4D_AA,2A	C ₂	1.959 (2) 2.022 (2)	91.3, 109.4 (2), 112.7, 116.3 (2)	128.9 (2), 122.4 (2) 119.6 (2), 129.6 (2)	1.017 (2), 0.966 (2) 0.984 (2), 0.966 (2)		105.2 107.0 105.2 108.6 (2) 110.1 (2)
(4,3)_2D,DD _AA,A	C ₁	3.348 4.095 (2)	96.6 (2), 101.4, 107.8, 113.3, 117.8, 120.0	127.3 (2) 125.4 (2), 127.9 (2) 118.2, 130.7 126.6, 123.2	0.966 (4) 0.970 (2) 0.976, 1.010 1.018, 0.966	1.547 (2) 1.828 (2)	106.7 (2) 105.4

			2.034	116.0, 117.8, 126.2, 167.9	123.1, 126.9	0.984, 0.965	109.9	
			2.054		123.6, 127.8	0.967 (2)	108.5	
			2.111		120.2, 126.2	0.982, 0.966	108.4	
			2.147		129.7, 122.9	0.967 (2)	107.4	
			2.893		128.9, 125.8	0.969 (2)	1.801, 1.849	105.3
			4.135		123.5, 130.0	0.969 (2)	1.576	106.4
(5,2)_3D_A _b A _b ,A _a	C ₁	2.002	86.0, 86.4, 89.3, 90.6, 100.6, 111.8,		123.6, 127.6	0.965, 1.011		108.2
		,cis						
			2.026	123.6, 124.5, 168.6, 87.8	122.6, 127.3	0.985, 0.965		110.0
			2.060		127.9, 123.6	0.967 (2)		108.4
			2.113		120.2, 126.2	0.981, 0.966		108.4
			2.154		130.1, 122.6	0.967 (2)		107.3
			3.881		125.1, 129.6	0.969 (2)	1.866, 1.793	105.3
			4.123		127.2, 126.3	0.966 (2)	1.573	108.2
(6,1)_AA	C ₁	2.102	84.0 (2), 85.7, 86.6, 88.5, 89.0, 90.1,		119.7/129.5	0.981, 0.965		108.1
		2.107	92.7, 93.0, 94.7, 96.0 (2), 167.9, 178.3,		122.0, 128.7	0.981, 0.965		109.2
		2.135		179.6	124.9, 125.8	0.967 (2)		108.2
		2.147			123.7, 124.2	0.966, 0.967		107.0
		2.149			124.4, 124.7	0.966, 0.967		107.1
		2.150			120.6, 131.8	0.967 (2)		107.2
		3.92			126.2, 128.7	0.969 (2)	1.844, 1.845	105.1
(4,4)_2D,2DD _2AA,2A	C ₂	1.958 (2)	94.7 (2), 111.0, 116.6 (2), 124.4		118.5 (2), 130.5 (2)	0.975 (2), 1.005 (2)		111.0 (2)
		2.009 (2)			118.7 (2), 130.7 (2)	0.986 (2), 0.965 (2)		110.6 (2)
		3.724 (2)			127.3 (2), 127.4 (2)	0.968 (2), 0.969 (2)	1.816 (2), 1.943 (2)	105.2 (2)
		4.141 (2)			126.2 (2), 126.5 (2)	0.965 (4)	1.587 (2)	106.8 (2)
(4,4)_3D,DD _AA,2A	C ₁	1.971	94.7, 107.1, 110.6, 111.5, 113.1, 119.7		118.8, 129.7	0.974, 1.004		110.8
		1.972			127.1, 124.2	1.010, 0.965		108.7
		1.979			128.3, 123.1	1.009, 0.965		108.6
		2.016			118.4, 131.0	0.986, 0.965		110.5
		3.723			128.1, 126.7	0.968 (2)	1.787, 1.971	105.2

				4.078		128.9, 124.6	0.966 (2)	1.571	106.5
				4.118		128.3, 125.2	0.966 (2)	1.578	106.5
				4.126		118.8, 129.7	0.965 (2)	1.595	106.5
(4,4)_2D,2DD _2AA,2A_share	C ₁	1.964	95.1, 98.9, 108.0, 109.0, 121.3, 125.0	116.7, 131.9		0.976, 1.005			111.1
		1.967		128.2, 122.4		1.012, 0.965			108.7
		2.010		118.1, 131.0		0.985, 0.965			110.7
		2.011		116.6, 119.3		0.981, 0.976			110.9
		3.649		132.6, 122.5		0.968 (2)	1.865, 1.953		105.0
		3.715		130.4, 124.5		0.968 (2)	1.971, 1.798		105.1
		4.098		124.2, 129.1		0.966 (2)	1.567		106.6
		4.154		129.4, 123.4		0.965 (2)	1.586		106.7
(4,4)_3D,DD _AA,A,AD,AA _p	C ₁	1.958	94.4, 102.7, 111.5, 112.7, 113.2, 122.7	124.0, 125.9		1.077, 0.965			110.2
		1.963		118.6, 129.7		0.975, 1.007			110.8
		2.004		128.0, 123.4		0.994, 0.965			108.6
		2.020		118.5, 130.8		0.986, 0.965			110.6
		3.716		126.6, 128.1		0.969 (2)	1.777, 1.954		105.2
		3.898		98.2, 152.9		0.969, 0.965	1.532		108.9
		4.127		129.9, 122.6		0.965 (2)	1.580		106.8
		4.241		119.9, 129.6		0.968, 0.969	1.715, 2.056		105.4
(4,4)_3D,DD _AA,A,AD,AA _p	C ₁	1.961	95.4, 108.4, 109.6, 110.3, 115.3, 117.4	119.7, 128.3		1.013, 0.964			110.8
		1.969		127.0, 124.2		1.013, 0.965			108.8
		2.001		117.7, 126.0		0.977, 0.989			110.5
		2.014		118.1, 130.4		0.985, 0.966			110.4
		3.705		128.0, 126.8		0.969 (2)	1.798, 1.949		105.1
		3.780		94.8, 156.1		0.970, 0.964	1.553		108.7
		4.076		125.7, 127.2		0.965 (2)	1.558		106.8
		4.166		132.0, 119.1		0.968 (2)	1.761, 2.040		105.1
(4,4)_2D,2DD _2AA,AD,AA _p	C ₁	1.957	94.7, 98.9, 106.0, 109.0, 123.3, 126.4	120.8, 128.1		1.015, 0.965			111.0
		1.990		116.0, 128.0		0.978, 0.989			110.8
		2.000		116.8, 118.6		0.981, 0.978			110.8
		2.013		118.1, 129.6		0.984, 0.966			110.4
		3.649		124.5, 130.5		0.969 (2)	1.877, 1.926		105.0
		3.709		131.5, 123.5		0.969 (2)	1.821, 1.926		105.1

(5,3)_3AA	C _s	3.796 4.207 2.04 2.089 2.144 2.039 (2)	86.6, 87.7 (2), 91.9 (2), 94.3, 116.2 (2), 127.6, 179.0	96.3, 154.5 129.8, 128.9 123.4, 126.3 119.2 (2) 120.0, 131.0 121.2 (2), 127.3 (2)	0.970, 0.965 0.968 (2) 0.981, 0.965 0.977 (2) 0.965, 0.981 0.984 (2), 0.965 (2)	1.541 1.756, 2.026	108.8 105.2 110.3 109.3 109 209.8 (2)	
(5,3)_2AA,A	C ₁	3.875 (2) 3.884 2.012 2.020 2.067 2.118 2.126 3.896 (2)	86.2, 87.2, 87.3, 87.4, 95.9, 96.7, 108.9, 124.3, 126.8, 167.1	124.8 (2), 129.9 (2) 127.4 (2) 128.3, 123.3 122.8, 122.7 124.4, 125.1 119.1, 125.7 118.6, 124.9 128.4 (2), 126.3 (2)	0.968 (4) 0.968 (2) 1.006, 0.965 0.978 (2) 0.966 (2) 0.980, 0.965 0.980, 0.966 0.968 (4)	1.799 (2), 1.926 (2) 1.836, 1.856 108.2 111.3 108.7 108.4 108.0 1.870 (2), 1.851 (2)	105.2 (2) 105.3 108.2 111.3 108.7 108.4 108.0 105.3	
(6,2)_2D,DD _2AA	C _{2v}	4.162 2.080 2.109 (2)	82.5 (2), 86.3 (2), 90.5 (4), 93.7 (2), 97.5 (2), 165.0, 172.0, 180.0	128.9, 124.7 123.9 (2) 121.3 (2), 129.2 (2)	0.966 (2) 0.980 (2) 0.981 (2), 0.965 (2)	1.597	106.2 112.2 109.5 (2)	
(6,2)_4D_2AA	C ₂	2.125 2.171 (2) 3.935 (2)		125.7 (2) 123.6 (4) 127.5 (4)	0.966 (2) 0.967 (4) 0.968 (4)	1.928 (2), 1.838 (2)	108.6 107.0 (2) 105.1 (2)	
		2.105 (2) 2.128 (2) 2.157 (2) 3.960 (2)	84.5 (2), 84.8 (2), 87.3 (2), 89.3 (2), 89.7, 99.4 (2), 102.6, 169.3 (2), 170.6	122.4 (2), 124.5 (2) 121.8 (2), 124.2 (2) 116.7 (2), 131.9 (2) 127.3 (2), 127.5 (2)	0.980 (2), 0.965 (2) 0.980 (2), 0.965 (2) 0.967 (2), 0.966 (2) 0.968 (4)	109.1 (2) 108.9 (2) 107.0 (2) 1.860 (2), 1.869 (2)		

(4,5)_D,3DD _2AA,3A	C _s	1.969 1.972 (2)	93.3, 94.4, 113.7 (2), 121.5 (2)	119.3, 130.0 117.7 (2), 131.1 (2)	0.973, 1.001 0.977 (2), 1.000 (2)	110.8 111.1 (2)
		2.027		119.2, 130.1	0.983, 0.963	110.7
		3.695		126.3, 127.9	0.968 (2)	1.851 (2)
		3.752		123.9 (2)	1.030 (2)	1.794, 1.960
		4.155		126.2 (2)	0.974 (2)	1.606
		4.178 (2)		127.8 (2), 125.2 (2)	0.965 (4)	1.675 (2)
(4,5)_2D,2DD _AA,2A,AD,AA _{p,t}	C ₁	1.970 1.973	91.1, 101.3, 109.8, 113.8, 115.7, 123.0	124.2, 125.4 117.3, 130.4	1.012, 0.964 0.977, 1.000	110.1 110.8
trans						
		1.974		117.5, 131.5	0.977, 1.000	110.9
		2.018		128.3, 123.0	0.990, 0.965	108.5
		3.689		126.4, 128.4	0.968 (2)	1.898 (2)
		3.927		98.6, 152.4	0.969, 0.964	1.551
		4.169		131.1, 121.7	0.965 (2)	1.612
		4.191		125.8, 127.4	0.965 (2)	1.614
		4.285		118.8, 130.5	0.968 (2)	1.744, 2.048
(4,5)_2D,2DD _AA,2A,AD,AA _{p,c}	C ₁	1.964 1.978	97.9, 100.8, 107.9, 115.1, 115.8, 120.0	117.7, 131.0 129.7, 121.3	0.976, 1.002 1.006, 0.965	111.3 109.0
is						
		1.987		118.3, 126.8	1.009, 0.964	110.1
		2.006		116.8, 127.4	0.978, 0.985	111.0
		3.678		125.4, 129.4	0.968 (2)	1.921, 1.889
		3.775		95.2, 155.4	0.971, 0.964	1.568
		4.151		125.9, 127.4	0.965 (2)	1.589
		4.158		126.4, 126.0	0.965 (2)	1.603
		4.218		129.4, 121.7	0.968 (2)	1.791, 2.011
(4,5)_D,3DD _2AA,A,AD,AA _p	C ₁	1.962 1.964 1.997 2.024 3.686	95.0, 97.3, 110.2, 109.7, 119.0, 127.6	118.0, 130.2 118.0, 120.2 116.8, 127.2 118.2, 130.5 124.8, 130.0	0.976, 1.003 0.975, 1.005 0.979, 0.986 0.984, 0.965 0.968 (2)	110.8 110.7 111.2 110.5 105.2

(4,4,1)_3D,DD _AA,A,2AD_AA	C ₁	3.720 3.841 4.138 4.211 1.964 1.965 1.981 2.031 3.702 3.966 4.011 4.127 5.557	96.5, 102.8, 108.8, 110.0, 118.2, 120.5	130.6, 124.2 95.2, 154.1 130.3, 122.4 116.5, 133.8 119.0, 126.0 124.3, 125.5 128.8, 121.9 116.8, 130.0 130.1, 124.7 114.4, 136.9 113.3, 136.9 131.2, 122.2 123.0, 131.7	0.968 (2) 0.969, 0.964 0.965 (2) 0.968 (2) 0.973, 1.009 1.017, 0.964 1.009, 0.965 0.986, 0.965 0.968 (2) 0.977, 0.964 0.975, 0.964 0.965 (2) 0.968 (2)	1.800, 1.968 1.583 1.598 1.783, 2.055 1.785, 2.013 1.535 1.563 1.577 1.875, 1.895	105.2 108.8 106.7 105.0 111.4 109.9 108.6 110.3 105.2 108.1 108.1 106.6 105.3
(4,4,1)_4D _2A,2AD_AA	C ₁	1.975 1.981 1.993 1.979 4.044 4.045 4.114 4.135 5.563	107.2, 108.0, 108.7, 108.9, 110.1, 112.9	126.3, 124.5 126.3, 123.1 127.5, 124.2 128.0, 127.8 112.7, 138.4 114.5, 140.1 124.6, 128.7 124.4, 128.9 115.9, 138.9	1.016, 0.964 1.017, 0.964 1.008, 0.965 1.007, 0.965 0.967, 0.964 0.976, 0.964 0.965 (2) 0.965 (2) 0.968 (2)	1.536 1.544 1.585 1.582 1.881, 1.885	109.2 108.8 108.3 108.9 108.2 107.9 106.6 106.6 105.2
(5,4)_4A _b A _b	C _{2v}	2.030 (2) 2.076 2.083 (2) 3.860 (4)	89.5 (4), 91.3 (2), 112.7 (2), 134.6, 177.5	123.1 (4) 125.7 (2) 119.3 (4) 125.2 (4), 129.7 (4)	0.980 (4) 0.965 (2) 0.974 (4) 0.968 (8)	1.912 (4), 1.880 (4)	112.9 (2) 108.6 109.7 (2) 105.2 (4)
(5,4)_3A _b	C ₂	2.028 2.045 (2) 2.093 (2) 3.865 (2)	89.0 (2), 89.2 (2), 91.8 (2), 113.6, 123.2 (2), 178.6	123.6 (2) 122.3 (2), 126.9 (2) 118.8 (2), 118.1 (2) 131.7 (2), 123.1 (2)	0.979 (2) 0.981 (2), 0.964 (2) 0.976 (2), 0.977 (2) 0.968 (4)	1.830 (2), 105.2 (2)	112.8 110.6 (2) 109.1 (2) 105.2 (2)

			3.870 (2)		130.7 (2), 124.1 (2)	0.968 (4)	1.945 (2)	
(5,4)_3A _b A _b ,A _a	C ₁	2.010	82.3, 85.2, 88.5, 91.0, 92.9, 96.5,		128.3, 123.2	1.00, 0.964	1.869 (2),	105.2 (2)
		2.034	113.7, 118.1, 128.2, 170.4		122.6, 125.4	0.978, 0.976	1.918 (2)	
		2.051			121.5, 128.1	0.982, 0.964		108.4
		2.092			119.4, 121.1	0.976, 0.975		112.0
		2.132			120.5, 130.4	0.980, 0.964		110.0
		3.878			131.5, 123.3	0.968 (2)	1.859, 1.921	109.7
		3.897			131.5, 123.2	0.968 (2)	1.808, 1.949	105.2
		3.937			126.2, 128.4	0.968 (2)	1.886, 1.846	105.3
		4.207			129.7, 123.9	0.965 (2)	1.626	105.4
(5,4) _2A _b A _b ,A _a A _b ,A _a	C _s	1.995	86.5 (2), 88.5, 89.2 (2), 101.2,		120.9, 129.4	0.974, 0.996		106.4
		2.056 (2)	116.2 (2), 127.4, 170.3		121.9 (2), 126.3 (2)	0.981 (2),		109.7
						0.964 (2)		
		2.088			120.8 (2)	0.975 (2)		109.2
		2.133			119.6, 131.1	0.981, 0.964		109.4
		3.881			127.3 (2)	0.968 (2)		105.3
		3.911 (2)			129.5 (2), 125.1 (2)	0.968 (4)	1.826, 1.929	105.3 (2)
							1.811 (2),	
		4.240			126.8 (2)	0.962 (2)	1.931 (2)	106.0
(5,4) _AA,A,AAD,AA _p	C ₁	2.030	82.2, 85.4, 88.9, 93.1, 93.6, 94.2,		122.5, 124.8	0.979, 0.977		111.9
		2.050	100.1, 125.5, 134.4, 171.4		118.0, 130.8	0.983, 0.964		110.7
		2.059			122.1, 126.5	0.984, 0.964		108.9
		2.081			122.2, 119.6	0.975 (2)		111.3
		2.097			121.8, 128.1	0.980, 0.964		109.4
		3.660			96.1, 156.7	0.975, 0.966		107.2
		3.894			126.8, 127.9	0.968 (2)		105.3
		3.922			128.2, 126.4	0.938 (2)		105.4
		4.150			126.3, 126.9	0.938 (2)		105.0
(5,4) _2AA,AD,AA _p	C ₁	1.993	85.6, 86.7, 88.5, 88.9, 95.8, 96.3,		126.7, 122.7	1.007, 0.964		109.7
		2.055	114.8, 115.8, 129.3, 167.9		120.8, 125.4	0.983, 0.964		109.5
		2.057			121.9, 125.8	0.982, 0.964		109.6

			2.105	120.5, 121.4	0.976 (2)	109.4
			2.122	130.0, 122.5	0.986, 0.965	107.5
			3.901	123.8, 130.8	0.968 (2)	1.929, 1.808
			3.928	125.3, 129.3	0.968 (2)	1.923, 1.805
			4.021	99.7, 151.7	0.970, 0.964	1.577
			4.466	120.1, 129.5	0.968 (2)	1.796, 2.876
(6,3)_6D_3AA	D ₃	2.127 (6)	53.2, 83.2 (2), 88.0 (6), 102.5 (3), 166.0 (3)	123.8 (6), 123.3 (6)	0.964 (6), 0.978 (6)	109.2 (6)
(6,3)_4D,DD _3AA	C _s	4.000 (3)		127.4 (6)	0.966 (6)	1.870 (6)
		2.099	82.6, 83.3 (2), 85.1, 85.9 (2), 90.2, 95.0	125.9, 122.9	0.975 (2)	110.4
		2.110	(2), 97.7 (2), 102.1, 162.6, 167.7, 172.8	125.3, 124.8	0.978, 0.964	109.9
		2.121		124.5, 126.0	0.977, 0.963	109.5
		2.137 (2)		119.8 (2), 124.6 (2)	0.979 (2), 0.964 (2)	108.4 (2)
		2.161		124.3 (2)	0.965 (2)	107.9
		3.958 (2)		127.7 (2), 127.1 (2)	0.968 (4)	1.862 (2), 1.910 (2)
(4,6)_2AA,4A	D _{2d}	4.012		127.5 (2)	0.965 (2)	1.867, 1.885
		1.979 (4)	95.8 (2), 116.7 (4)	118.3 (4), 130.8 (4)	0.975 (4), 0.995 (4)	110.9 (4)
		3.733 (2)		126.6 (4)	0.979 (4)	1.913 (2)
		4.202 (4)		125.6 (8)	0.988 (8)	1.648 (4)
(4,5,1)_2AA,A, 2AD_AA	C ₁	1.968	94.5, 97.9, 110.4, 113.4, 118.0, 124.3	119.0, 124.4	0.973, 1.006	111.1
		1.968		118.1, 125.8	0.975, 1.004	111.8
		1.969		117.4, 131.1	0.976, 0.999	111.3
		2.025		118.8, 130.5	0.983, 0.964	110.7
		3.686		129.3, 125.5	0.967 (2)	1.900 (2)
		3.741		129.4, 125.2	0.968 (2)	1.988, 1.798
		3.993		112.5, 137.4	0.975, 0.964	1.577
		4.015		112.5, 138.1	0.975, 0.964	1.588
		4.178		124.4, 128.4	0.965 (2)	1.618
		5.541		126.0, 128.8	0.967 (2)	1.906, 1.898
(4,5,1)_AA,A,	C ₁	1.971	94.0, 97.2, 114.7, 115.1, 118.2, 119.2	119.1, 128.1	0.974, 1.001	105.2
						110.7

AAD,AD_AA		1.975 1.977 2.023 3.660 3.740 3.996 4.119 4.193 5.355	116.7, 131.5 111.1, 120.7 118.9, 130.4 118.2, 134.8 127.7, 127.0 115.3, 136.7 131.0, 120.6 124.7, 128.1 115.2, 139.8	0.978, 0.998 0.981, 1.001 0.984, 0.964 0.980, 0.966 0.968 (2) 0.973, 0.964 0.965 (2) 0.965 (2) 0.967 (2)	1.893, 1.860 1.794, 1.970 1.630 1.609 1.627 1.907, 2.003	111.4 104.8 110.7 106.4 105.4 107.6 106.7 106.7 104.9
(4,4,2)_4AD_2AA	C ₁	1.976 1.978 1.980 1.985 3.978 4.009 4.052 4.083 5.608 5.620	104.5, 108.2 (2), 109.1, 111.8, 115.0 124.8, 124.6 126.2, 124.2 127.5, 122.3 124.6, 126.0 116.8, 134.4 114.8, 136.2 113.8, 136.7 113.0, 137.9 130.0, 124.7 113.8, 141.0	1.012, 0.964 1.014, 0.964 1.015, 0.964 1.014, 0.964 0.976, 0.964 0.976, 0.964 0.976, 0.964 0.976, 0.964 0.968 (2) 0.967 (2)	1.553 1.550 1.548 1.544 1.885, 1.883 1.885, 1.886	109.9 108.9 108.7 109.4 108.0 108.0 108.0 108.1 105.3 105.2
(5,5)_4A _b A _b ,A _a	C _s	2.018 2.045 (2) 2.082 2.111 3.894 (2) 3.909 (2)	87.2 (2), 89.0 (2), 90.9 98.6, 114.0 (2), 132.1, 170.5 131.5,120.0 123.5 (2),123.8 (2) 122.7 (2) 121.6 (2) 130.3 (2),124.4 (2) 130.4 (2),124.4 (2)	0.996,0.964 0.976 (4) 0.939 (2) 0.944 (2) 0.967 (4) 0.967 (4)	1.888 (2),1.920 (2) 1.979 (2),1.921 (2)	108.5 112.3 (2) 104.6 105.4 105.3 (2) 105.2 (2)
(5,4,1) _3A _b A _b ,AAD_A	C ₁	4.305 2.020 2.030 2.076 2.080 2.089	127.9 (2) 124.0, 120.9 122.8, 123.5 116.4, 119.0 125.0, 126.1 118.2, 118.9	0.941 (2) 0.974, 0.985 0.977, 0.976 0.984, 0.975 0.965 (2) 0.977, 0.976	1.648 113.3 112.7 110.2 108.9 109.9	104.1 113.3 112.7 110.2 108.9 109.9

			3.757	132.2, 122.3	0.993, 0.966	1.783, 1.812	105.5
			3.854	129.1, 126.7	0.967 (2)	1.954, 1.869	105.2
			3.857	130.0, 124.8	0.967 (2)	1.903, 1.924	105.2
			3.868	129.4, 125.4	0.967 (2)	1.886, 1.922	105.2
			5.911	120.9, 131.7	0.964 (2)	1.692	106.1
(5,5)_3A _b A _b , A _b A _b D_AA _p	C ₁	2.036	87.4, 87.6, 88.8, 89.3, 92.9, 97.4,	123.1, 124.0	0.977, 0.976		112.2
		2.047	103.5, 116.0, 140.4, 169.8	118.3, 125.6	0.977, 0.975		112.0
		2.059		126.6, 125.6	0.982, 0.964		109.4
		2.076		123.0, 119.9	0.975 (2)		112.0
		2.083		119.4, 120.6	0.977, 0.976		110.2
		3.660		95.6, 157.3	0.974, 0.966	1.905	107.1
		3.880		126.9, 128.0	0.967 (2)	1.905, 1.910	105.2
		3.902		127.6, 127.0	0.967 (2)	1.869, 1.924	105.4
		3.906		129.8, 125.0	0.967 (2)	1.881, 1.908	105.3
		4.157		119.7, 131.6	0.968 (2)	1.834, 2.033	104.9
(6,4)_4D,2DD _4AA	C ₂	2.106 (2)	84.2, 84.4 (2), 87.2 (2), 89.5 (2), 92.5 (2), 96.3 (2), 98.1, 168.9 (2), 175.5	123.3 (2), 118.9 (2)	0.975 (2), 0.976 (2)		110.3 (2)
		2.142 (2)		115.6 (4), 119 (4)	0.978 (2), 0.965 (2)		107.3 (2)
		2.143 (2)		122.4 (2), 125.5 (2)	0.975 (2), 0.964 (2)		108.9 (2)
		3.885 (2)		118.6 (2), 136.2 (2)	0.967 (4)	1.897 (2), 1.907 (2)	105.3 (2)
		4.000 (2)		129.1 (2), 125.8 (2)	0.967 (2)	1.925 (2), 1.876 (2)	105.2 (2)
(6,4)_4D,2DD _4AA_share	C ₁	2.111	82.6, 84.1, 84.9, 85.0, 85.4, 87.4, 88.4,	118.8, 115.9	0.975 (2)		107.7
		2.114	89.6, 91.6, 100.1, 100.2, 105.7, 165.7,	122.3, 125.9	0.974, 0.973		111.4
		2.126	166.4, 167.1	122.5, 122.3	0.978, 0.964		109.2
		2.136		121.9, 123.3	0.978, 0.964		109.1
		2.139		122.4, 125.2	0.977, 0.963		109.0
		2.150		123.3, 123.9	0.977, 0.964		109.5
		3.920		120.4, 134.6	0.967 (2)	1.950, 1.917	105.0
		3.999		132.9, 121.6	0.967 (2)	1.882, 1.868	105.4

(6,4)_5D,DD _3AA,A	C ₁	4.002 4.009 2.079 2.121 2.132 2.143 2.143 2.146 4.001 4.014 4.019 4.310	83.1 (2), 84.6, 84.9, 85.6, 87.3, 87.4, 89.4, 94.1, 98.0, 102.8, 103.9, 164.1, 165.5, 169.7	124.1, 130.8 126.8, 129.1 128.6, 121.5 124.1, 124.8 123.5, 127.3 122.7, 122.4 122.9, 123.3 122.3, 120.4 127.4 (2) 127.4, 127.3 129.2, 125.5 126.3, 128.0	0.967 (2) 0.967 (2) 0.988, 0.973 0.976, 0.963 0.978, 0.964 0.977, 0.964 0.977, 0.964 0.977, 0.964 0.967 (2) 0.968, 0.967 0.967 (2) 0.965 (2)	1.940, 1.872 1.936, 1.872 105.1 105.1 109.1 109.4 110.0 109.0 109.1 108.8 105.2 105.2 105.3 105.7		105.1 105.1 109.1 109.4 110.0 109.0 109.1 108.8 105.2 105.2 105.3 105.7
(6,4)_5D,DD _2AA,AAD,AA _p	C ₁	2.117 2.118 2.126 2.128 2.134 2.134 3.792 3.987 3.997 4.329	83.7, 84.1, 85.3, 86.2, 87.5, 88.7, 89.0, 89.8, 92.2, 97.6, 99.3, 99.4, 167.6, 170.1, 170.7	120.1, 122.0 121.6, 128.1 126.1, 123.3 117.9, 122.0 122.5, 120.6 122.7, 123.0 97.4, 155.0 127.3, 127.5 130.6, 124.2 119.7, 131.0	0.977, 0.975 0.979, 0.963 0.977, 0.964 0.980, 0.964 0.977, 0.965 0.977, 0.964 0.976, 0.966 0.967 (2) 0.967 (2) 0.968 (2)	1.832, 1.848 1.938, 1.867 1.879, 1.893 1.936, 1.959		108.8 110.1 109.5 108.9 108.9 109.4 107.7 105.2 105.2 104.8

^a Values in parentheses indicate degeneracies. ^b Inner shell water molecules only. ^c Covalent bond length. ^d Hydrogen bond length between the first and second shell.

Table S3. Comparison of experimental 0 K bond energies (kJ/mol) to theoretical values. The reactant and product complex is the GS of each inner shell size, unless otherwise noted.

<i>n</i>	Reactant	Product	Experiment	B3LYP ^a	B3P86 ^a	MP2(full) ^a
6	(4,2)	(4,1)	98.4 ^{b,c} (3.9)	97.0	100.6	95.7
			109.0 ^c (4.8)			
	(6,0)	(5,0)	94.6 ^{b,c} (3.9)	87.1	90.1	93.6
			105.8 ^c (4.8)			
7	(5,1)	(5,0)	90.7 ^{b,c} (3.9)	93.0	96.5	92.0
			99.3 ^c (4.8)			
	(4,3)	(4,2)	78.2 ^b (4.8)	79.8	82.5	76.5
			92.6 ^c (5.8)			
7	(4,3)_2D,DD _AA,2A	(4,2)	71.4 ^b (4.8)	75.7	78.4	69.3
	(5,2)	(6,0)	79.2 ^b (4.8)	93.5	97.0	83.3
	(5,2)	(5,1)	82.0 ^b (4.8)	87.9	91.0	87.5
	8	(5,3)	70.4 ^b (5.8)	75.8	78.6	86.3
			71.4 ^c (8.7)			
		(5,3)	67.5 ^b (5.8)	76.0	78.8	75.4
		(4,4)	59.8 ^b (5.8)	74.9	77.1	61.8
9	(4,4)	(4,3)	63.7 ^b (5.8)	74.8	77.0	72.7
	(5,4)	(5,3)	53.1 ^d (7.7)	70.8	72.5	71.6
			70.4 ^c (2.9)			
	(4,5)	(5,3)	45.3 ^d (7.7)	65.6	66.8	50.9
10	(4,5)	(4,4)	47.3 ^d (7.7)	67.4	69.1	65.8
			63.7 ^c (2.9)			
	(5,5)	(5,4)	42.5 ^d (2.9)	60.6	62.4	59.0
	(4,6)	(5,4)	36.7 ^d (2.9)	57.2	58.1	40.2

^a Geometry optimizations calculated using a B3LYP/6-311+G(d,p) level and SPE values calculated with a 6-311+G(2d,2p) basis set for each level. ZPE and cp corrected. ^b Values including competitive shift reported in following publication.²⁰ ^c Sequential dissociation modeled using eqs 4 and 6. ^d Single channel model using eq 4.

Table S4. Conversion of 0 K thresholds to 298 K enthalpies and free energies for water loss from $\text{Zn}^{2+}(\text{H}_2\text{O})_n$. The reactant and product complex is the GS of each inner shell size, unless otherwise noted. All values in kJ/mol with uncertainties in parentheses.

<i>n</i>	Dissociation	ΔH_0^{a}	$\Delta H_{298} - \Delta H_0^{\text{b}}$	ΔH_{298}	$T\Delta S_{298}^{\text{b}}$	ΔG_{298}
6	(4,2) → (4,1)	98.4 (3.9)	4.7 (0.4)	103.1 (3.9)	46.6 (1.0)	56.5 (4.1)
	(6,0) → (5,0)	94.6 (3.9)	2.1 (0.5)	96.7 (3.9)	45.5 (1.4)	51.2 (4.1)
	(5,1) → (5,0)	90.7 (3.9)	4.0 (0.4)	94.7 (3.9)	42.7 (1.1)	52.0 (4.1)
7	(4,3) → (4,2)	78.2 (4.8)	1.6 (0.3)	79.8 (4.8)	31.1 (1.2)	48.7 (4.9)
7	(4,3)_2D,DD_AA,2A → (4,2)	71.4 (4.8)	-1.0 (0.3)	70.4 (4.8)	17.9 (1.4)	52.5 (5.0)
	(5,2) → (6,0)	79.2 (4.8)	6.2 (0.3)	85.4 (4.8)	39.1 (1.2)	46.3 (4.9)
	(5,2) → (5,1)	82.0 (4.8)	4.3 (0.4)	86.3 (4.8)	47.1 (1.0)	39.2 (4.9)
8	(5,3) → (4,3)	70.4 (5.8)	3.6 (0.6)	74.0 (5.8)	46.0 (1.2)	28.0 (5.9)
	(5,3) → (5,2)	67.5 (5.8)	4.2 (0.4)	71.7 (5.8)	41.2 (1.1)	30.5 (5.9)
	(4,4) → (5,2)	59.8 (5.8)	1.6 (0.2)	59.8 (5.8)	29.8 (1.1)	30.0 (5.9)
	(4,4) → (4,3)	63.7 (5.8)	1.1 (0.3)	64.8 (5.8)	34.6 (1.6)	30.2 (6.0)
9	(5,4) → (5,3)	53.1 (7.7)	2.6 (0.4)	55.7 (6.8)	40.6 (1.1)	15.1 (7.8)
	(4,5) → (5,3)	45.3 (7.7)	-1.4 (0.1)	43.9 (7.7)	20.7 (1.6)	23.2 (7.9)
	(4,5) → (4,4)	47.3 (7.7)	1.2 (0.3)	48.4 (7.7)	32.1 (1.6)	16.3 (7.9)
10	(5,5) → (5,4)	42.5 (2.9)	1.1 (0.3)	43.6 (2.9)	29.2 (1.0)	14.4 (3.1)
	(4,6) → (5,4)	36.7 (2.9)	-3.2 (0.04)	33.8 (2.9)	10.9 (1.4)	22.9 (3.2)

^a Experimental values from Table 3.

^b Values calculated from the vibrations and rotations calculated at the B3LYP/6-311+G(d,p) level. Uncertainties found by scaling the frequencies up and down by 10%.

Figure Captions

Figure S1. Low energy isomers of $\text{Zn}^{2+}(\text{H}_2\text{O})_9$ calculated at B3LYP/6-311+G(d,p).

Figure S2. Low energy isomers of $\text{Zn}^{2+}(\text{H}_2\text{O})_{10}$ calculated at B3LYP/6-311+G(d,p).

Figure S3a-o. Geometries for all low and high energy isomers of the $\text{Zn}^{2+}(\text{H}_2\text{O})_x(\text{H}_2\text{O})_y$ complexes, where x and y are the number of waters in the inner shell and outer shell and complexes considered include $x = 4$ and $y = 1 - 6$, $x = 5$ and $y = 0 - 5$, and $x = 6$ and $y = 0 - 4$.

Figures

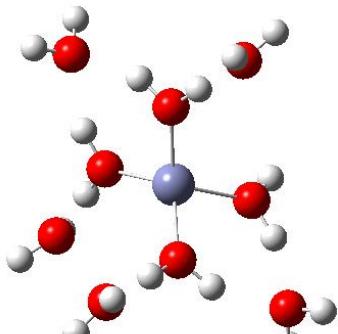
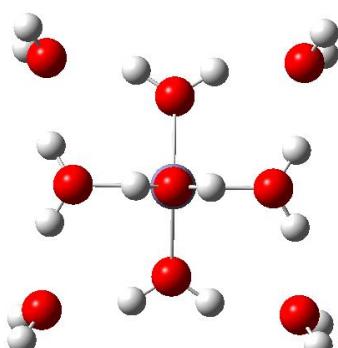
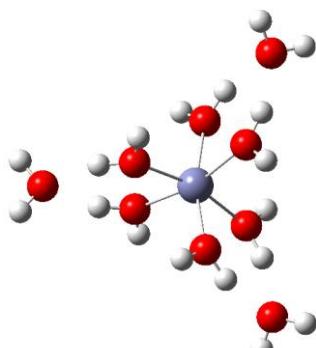
(4,5)_{D,3D}_2AA,3A(5,4)_{4AbAb}(6,3)_{6D}_3AA

Figure S1

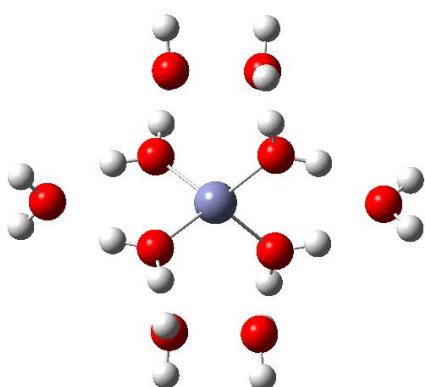
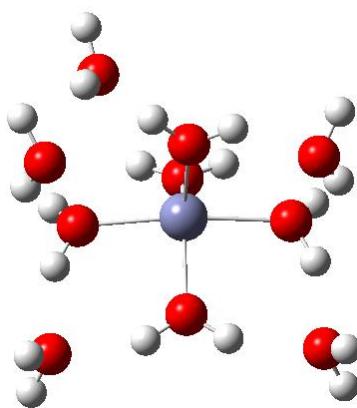
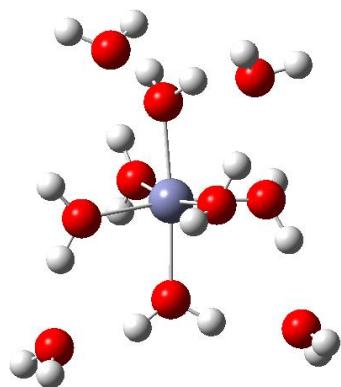
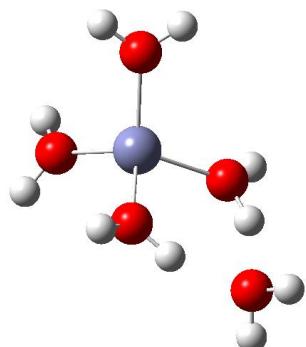
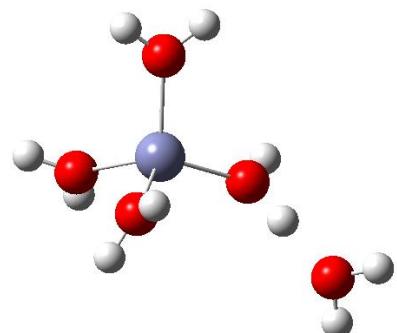
(4,6)_{2AA},4A(5,5)_{4AbAb,Aa}(6,4)_{4D,2DD}_4AA

Figure S2

a)

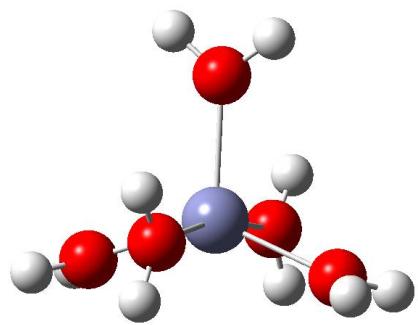


(4,1)_AA

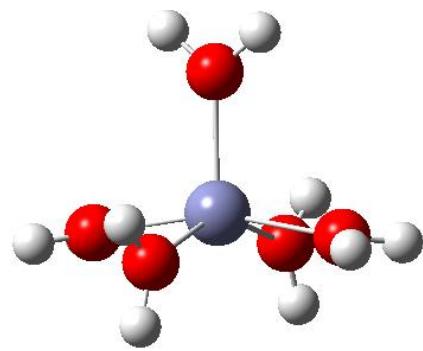


(4,1)_A

b)

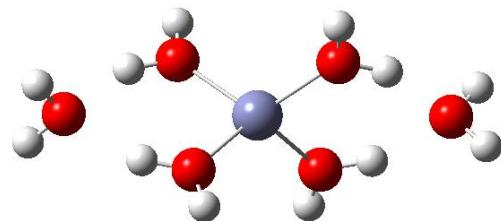


(5,0)

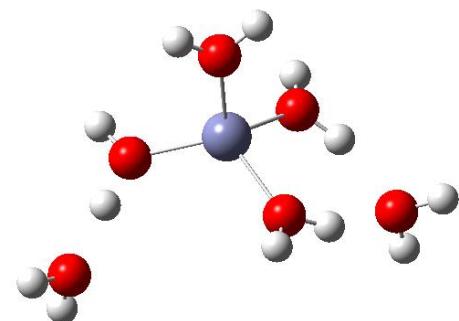


(5,0)_switch

c)

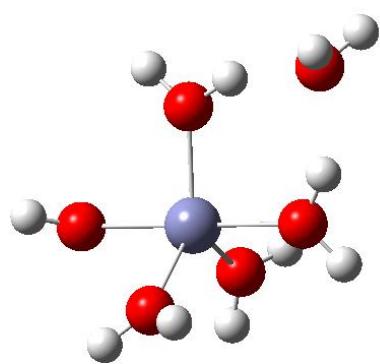
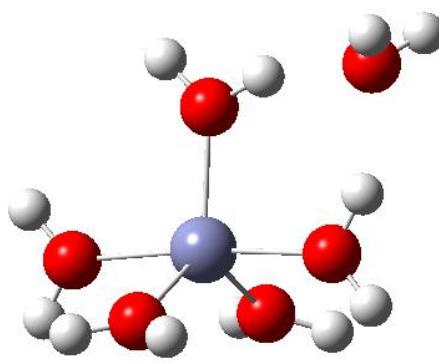
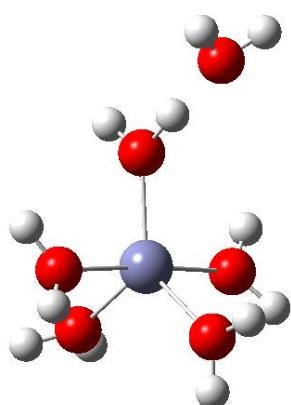
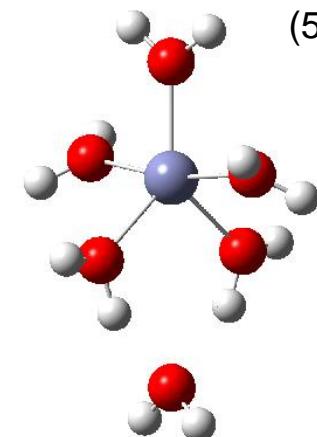


(4,2)_2AA

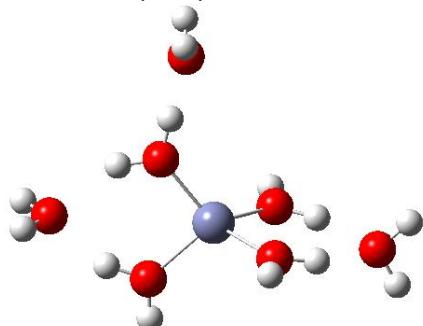


(4,2)_AA,A

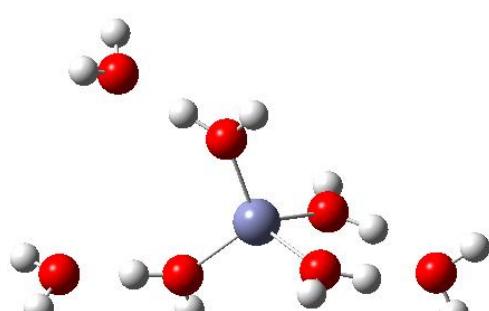
d)

(5,1)_A_aA_b(5,1)_A_aA_b_sym(5,1)_A_a(5,1)_A_bA_b

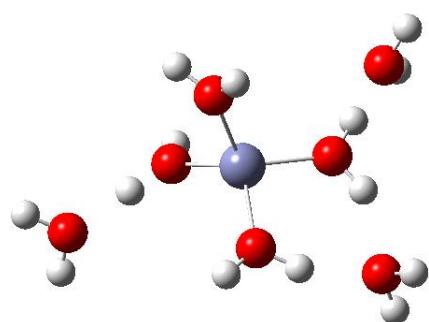
e)



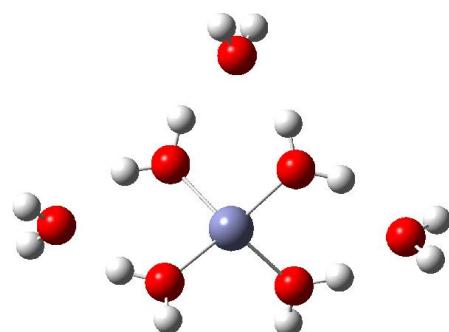
(4,3)_3D,DD_2AA,A



(4,3)_4D_AA,2A

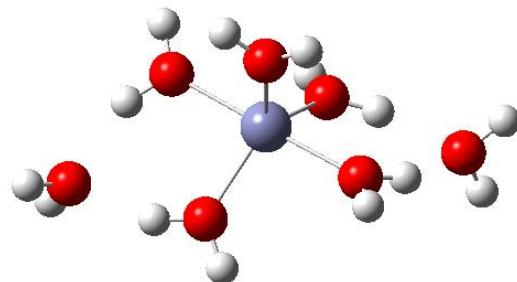
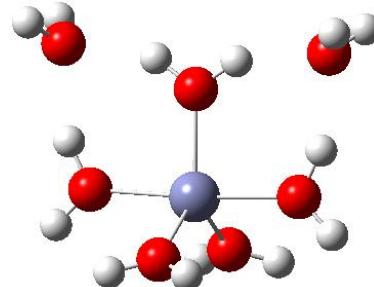
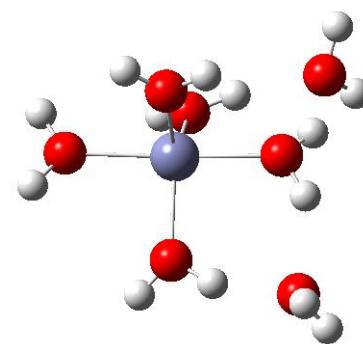
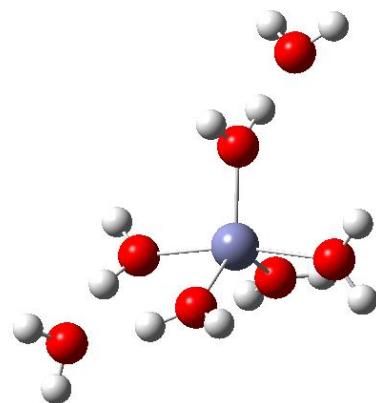
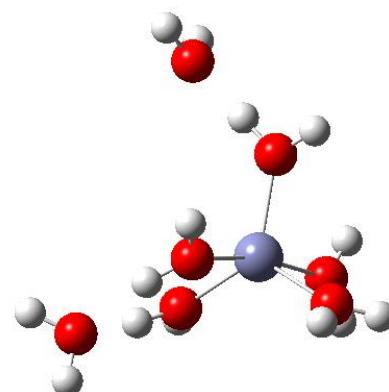


(4,3)_2D,DD_AA,2A



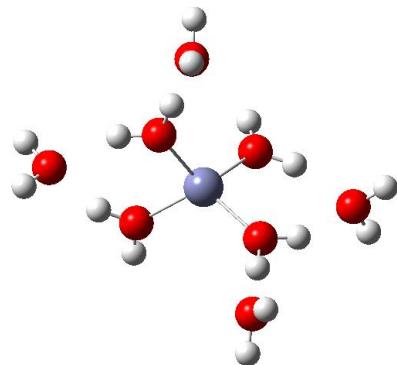
(4,3)_3D,2DD_3AA

f)

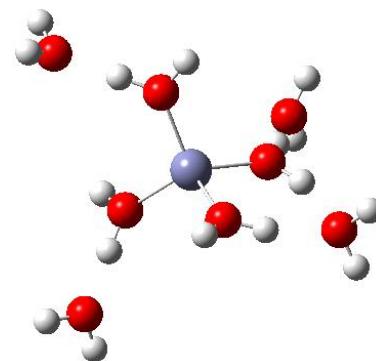
(5,2)_4D_2A_bA_b(5,2)_2D,2DD_2A_a(5,2)_2D,DD_2A_bA_b(5,2)_3D_A_bA_b,A_{a,trans}(5,2)_3D_A_bA_b,A_{a,cis}

g)

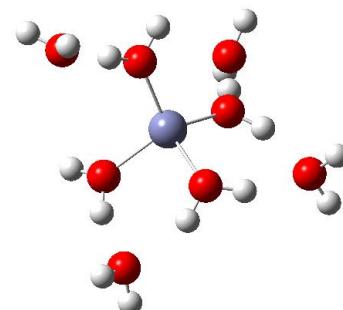
27



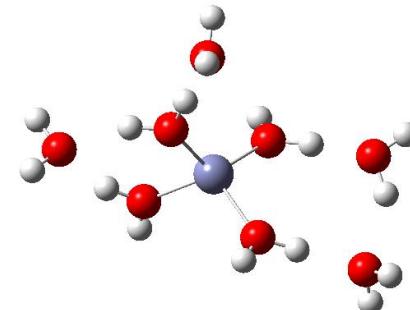
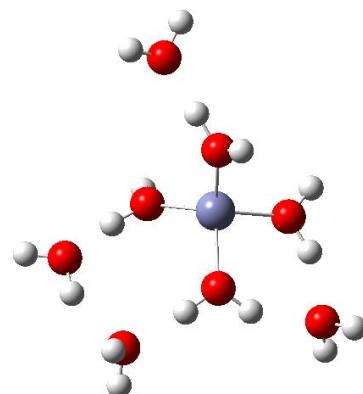
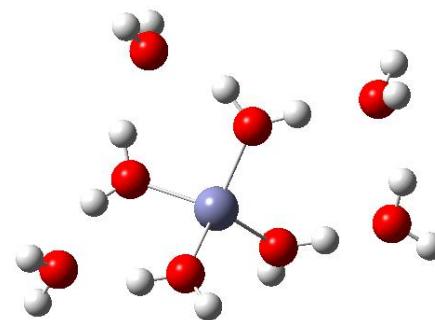
(4,4)_2D,2DD_2AA,2A



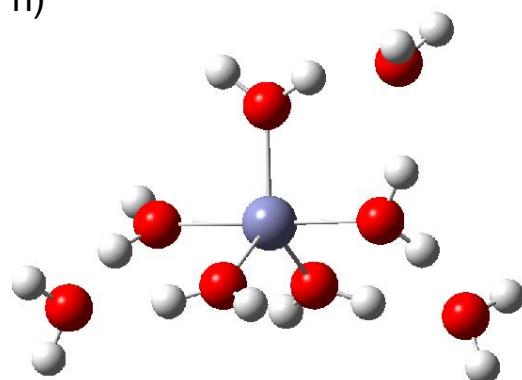
(4,4)_3D,DD_AA,3A



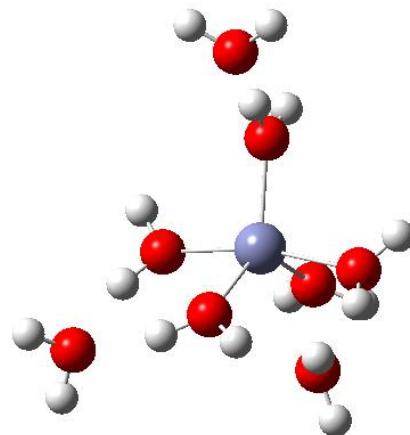
(4,4)_2D,2DD_2AA,2A

(4,4)_3D,DD_AA,A,AD,AA_p(4,4)_3D,DD_AA,A,AD,AA_p(4,4)_2D,2DD_2AA,AD,AA_p

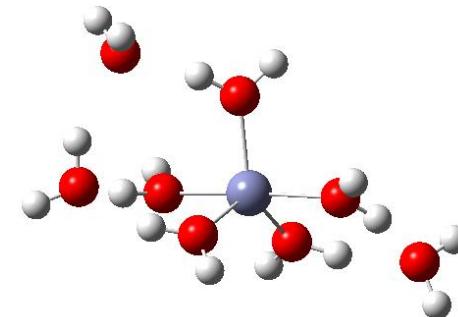
h)



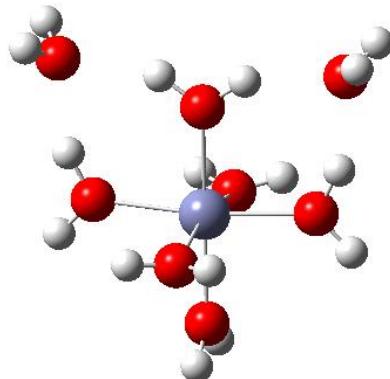
(5,3)_3AA



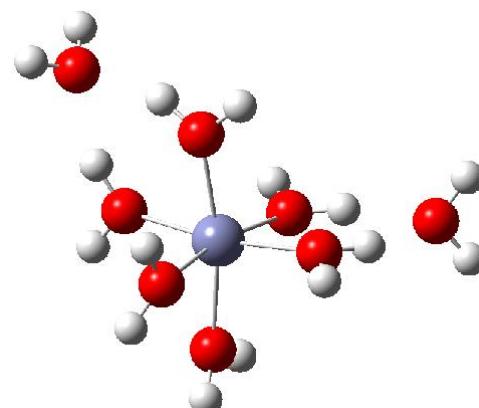
(5,3)_2AA,A

(5,3)_AA,AAD,AA_p

i)

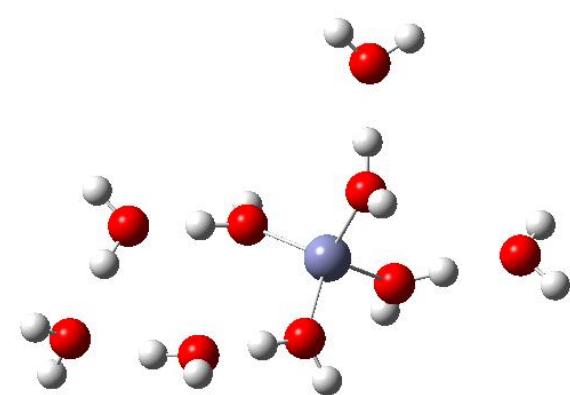
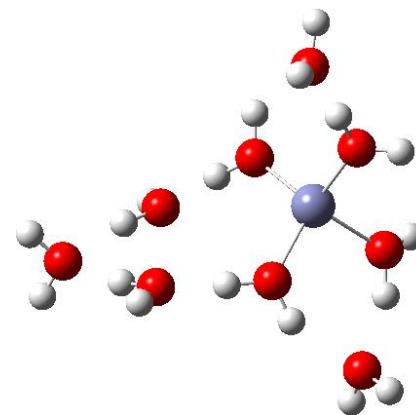
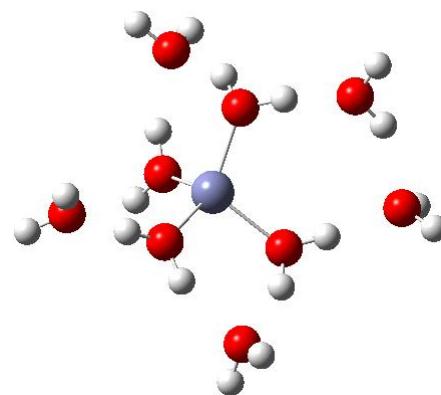
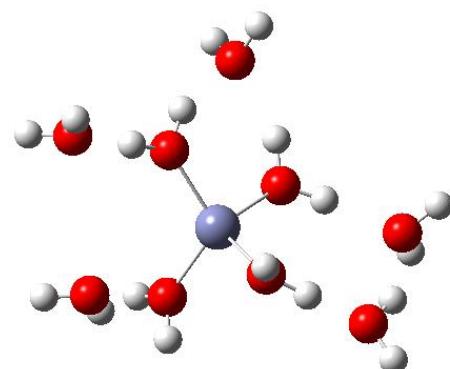
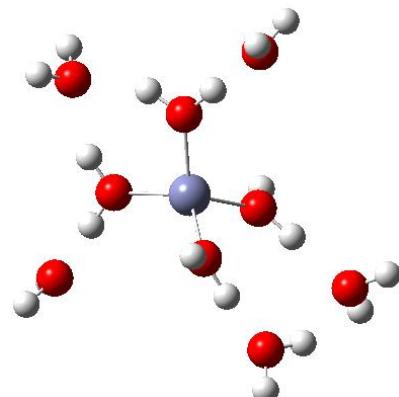
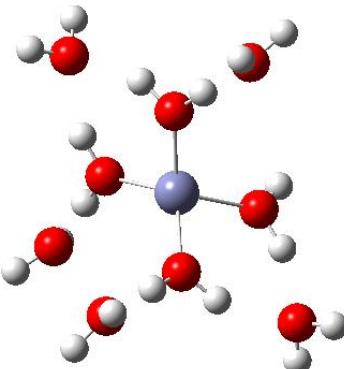


(6,2)_2D,DD_2AA

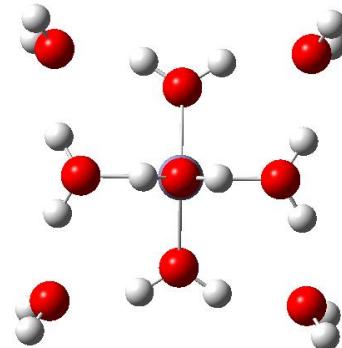
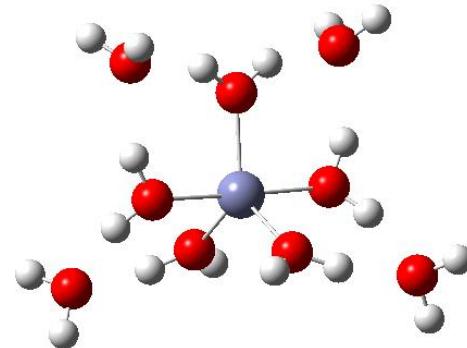
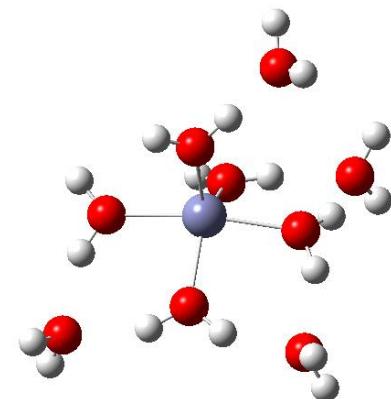
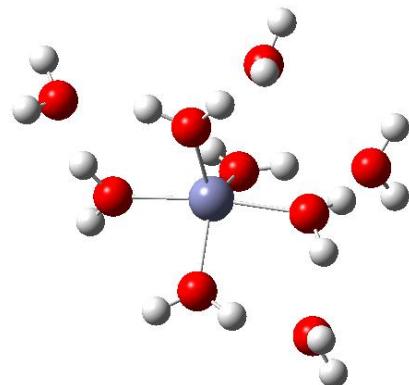
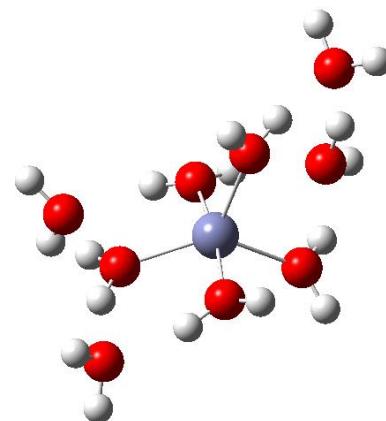
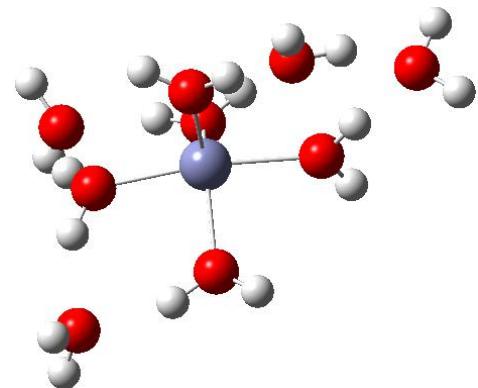


(6,2)_4D_2AA

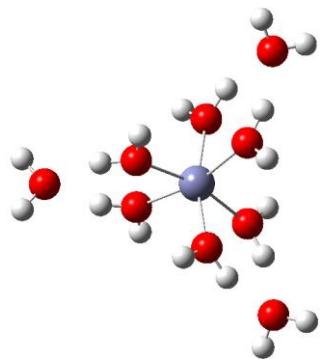
j)



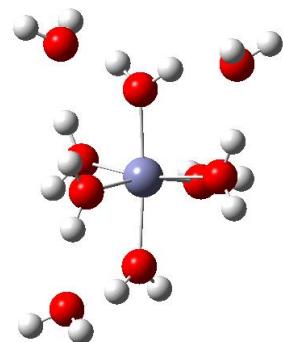
k)

(5,4)_4A_bA_b(5,4)_2A_bA_b,2A_aA_b(5,4)_3A_bA_b,A_a(5,4)_2A_bA_b,A_aA_b,A_a(5,4)_AA,A,AAD,AA_p(5,4)_2AA,AD,AA_p

l)

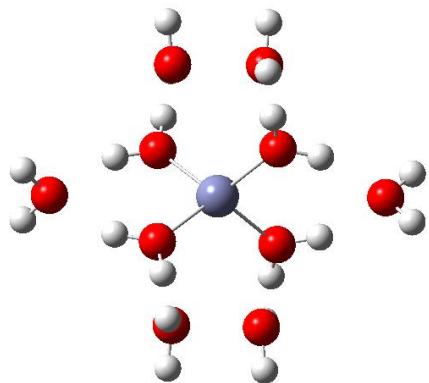


(6,3)_6D_3AA

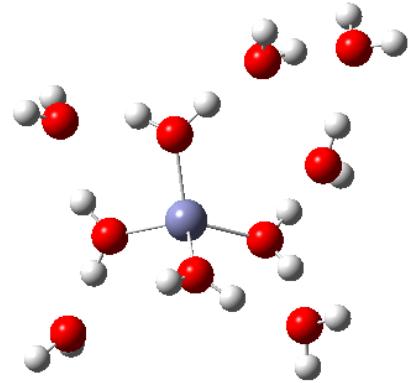


(6,3)_4D,DD_3AA

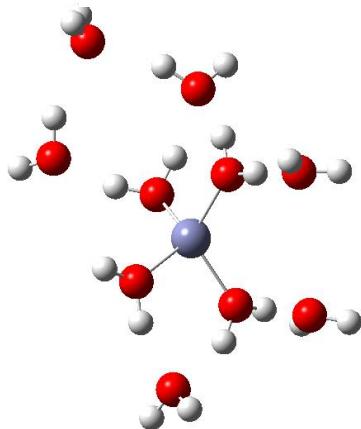
m)



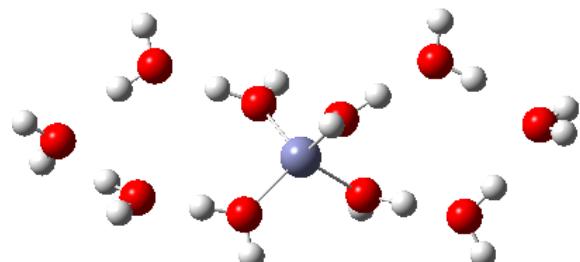
(4,6)_2AA,4A



(4,5,1)_2AA,A,2AD_AA

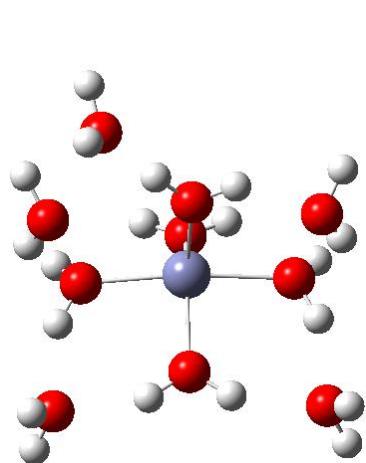
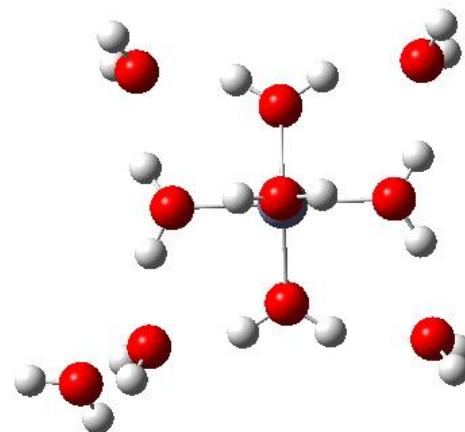
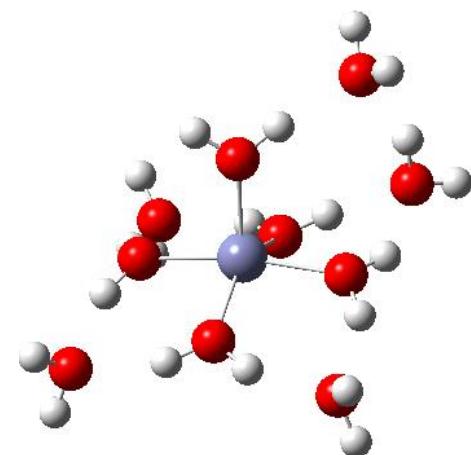


(4,5,1)_AA,A,AAD,AD_AA



(4,4,2)_4AD_2AA

n)

(5,5)₄A_bA_b,A_a(5,4,1)₃A_bA_b,AAD_A(5,5)₃A_bA_b,A_bA_bD,AA_p

o)

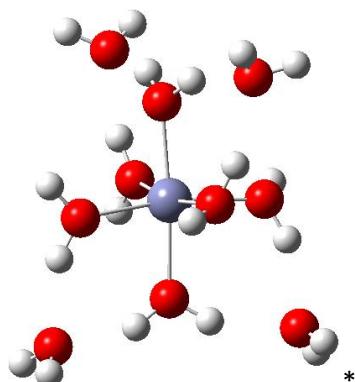
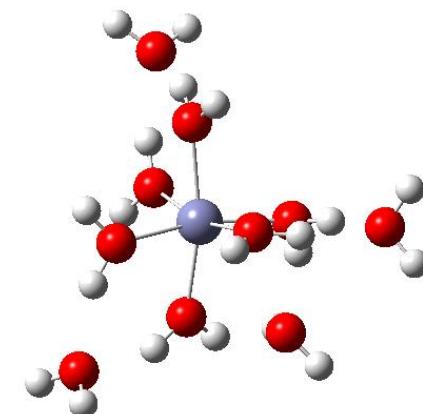
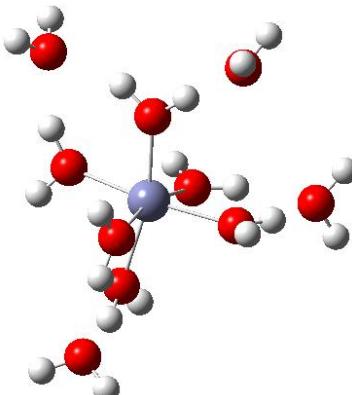
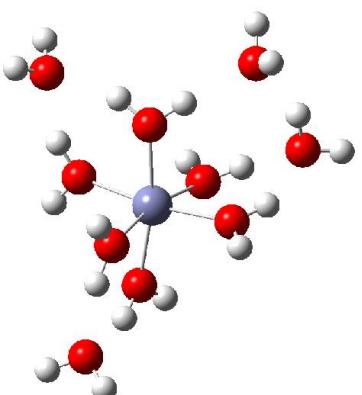
(6,4)₄D,2DD_4AA(6,4)₄D,2DD_4AA_share(6,4)₄D,2DD_4A(6,4)₅D,DD_2AA,AAD,AA_p

Figure S3