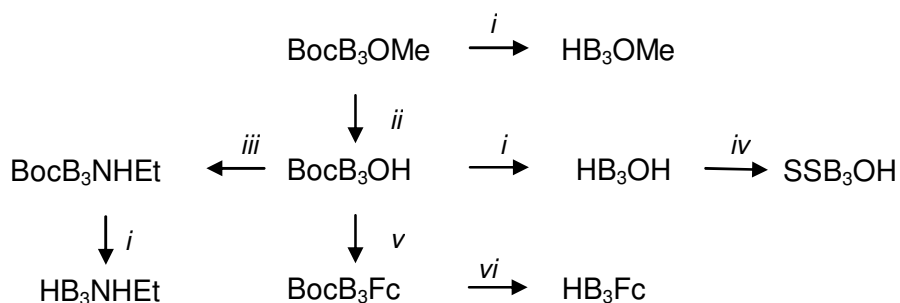


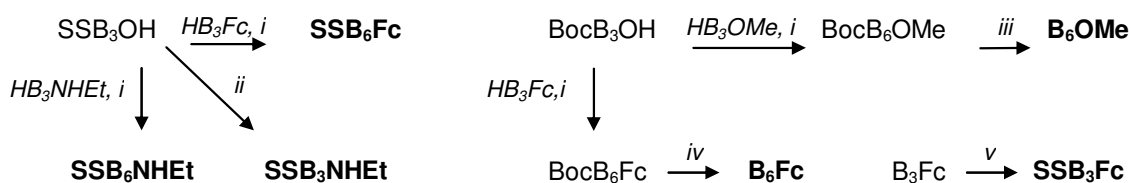
Supporting Information:

Electrochemistry of Ferrocenoyl β -Peptide Monolayers on Gold

Paula A. Brooksby, Kelly H. Anderson, Alison J. Downard, Andrew D. Abell.

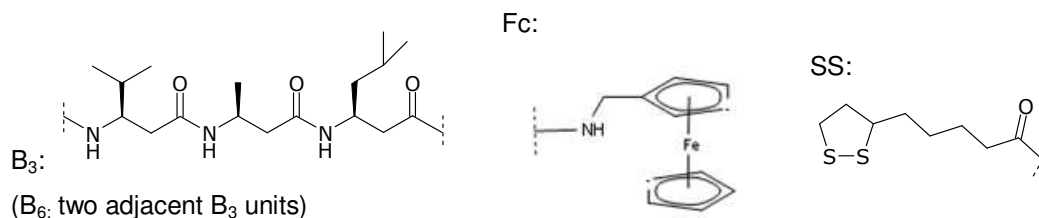


Scheme S1. Synthesis of β -tripeptides: (i) TFA, DCM, (ii) LiOH, THF, (iii) ethyl amine hydrochloride, HATU, DIEA, DMF, (iv) N-hydroxysuccinimide-activated lipoic acid, Et₃N, THF, H₂O, (v) ferrocene methyl amine, HATU, DIEA, DMF, (vii) 4M HCl in dioxane.



Scheme S2. Synthesis of β -peptides, with peptides mentioned directly in the manuscript highlighted in bold: (i) HATU, DIEA, DMF, (ii), ethyl amine hydrochloride,

HATU, DIEA, DMF, (iii)TFA, DCM, (iv) 4M HCl in dioxane, (v) N-hydroxysuccinimide-activated lipoic acid, THF, H₂O, DIEA.

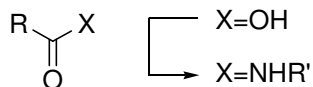


Scheme S3. Synthesis of β -tripeptides, with peptides mentioned directly in the manuscript highlighted in bold: (i) HATU, DIEA, DMF, (ii), ethyl amine hydrochloride, HATU, DIEA, DMF, (iii)TFA, DCM, (iv) 4M HCl in dioxane, (v) N-hydroxysuccinimide-activated lipoic acid, THF, H₂O, DIEA.

All NMR spectra are given at the end of this document. The synthesis of HB3OMe, BocB3OH, HB3OH, BocB6OMe, B6OMe^{S1}; ferrocene methyl amine^{S2,S3}; and N-hydroxysuccinimide-activated lipoic acid^{S4} are published. Tetrahydrofuran was distilled from sodium benzophenone ketyl under an inert atmosphere immediately prior to use. Dichloromethane and triethylamine were distilled from calcium hydride under an inert atmosphere. Anhydrous dimethylformamide and 2,2,2-trifluoroethanol were obtained from commercial sources. Flash chromatography was carried out on Fluka Silica gel 100 C₁₈-reversed phase (fully endcapped) or lipophilic sephadex (Aldrich).

The following abbreviations apply: HATU: *O*-(7-azabenzotriazole-1-yl)-*N,N,N',N'*-tetramethyl uranium hexafluorophosphate; TFE: 2,2,2-trifluoroethanol; TFA: trifluoroacetic acid; DCM: dichloromethane; DMF: N,N-dimethyl formamide; DIEA: diisopropylethylamine; THF: tetrahydrofuran

General Method A: Coupling with HATU

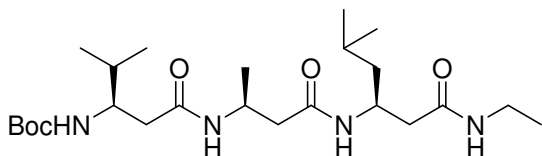


Primary amine (1.0 equiv) and carboxylic acid (1.0 equiv) were dissolved in dimethylformamide at rt, under N_2 . First diisopropylethylamine (4.0 equiv) and then HATU (1.0 equiv) were added, and the reaction stirred for 16 h at rt. The reaction mixture was diluted with ethyl acetate, washed with saturated aqueous NH_4Cl , $NaHCO_3$, and $NaCl$, dried ($MgSO_4$), and the solvent removed *in vacuo*. The crude mixture was purified by column chromatography, as specified for each compound.

General Method B: Coupling with HATU (dimethylformamide-insoluble product)

Primary amine (1.0 equiv) and carboxylic acid (1.0 equiv) were dissolved in dimethylformamide at rt, under N_2 . First diisopropylethylamine (4 equiv), and then HATU (1 equiv) were added, and the reaction stirred for 16 h. The solvent was removed *in vacuo* (high vacuum rotary evaporator), and the residue was sonicated in methanol for 10 min. The insoluble material was filtered, and washed with methanol and methanol/water (1:1) redissolved in 2,2,2-trifluoroethanol (to remove from the glass frit filter), and the solvent removed *in vacuo*. If necessary, the crude mixture was purified by column chromatography on lipophilic sephadex, as specified for each compound.

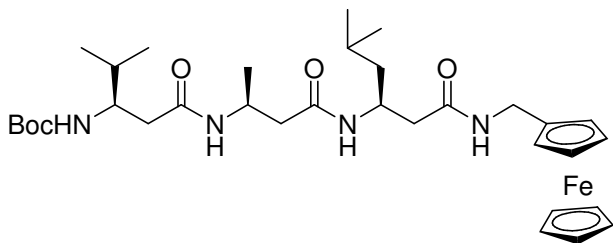
Boc-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-*N*-ethanamide (BocB₃NHET)



Boc- β^3 hVal- β^3 hAla- β^3 hLeu-OH (**BocB₃OH**) (76 mg, 0.17 mmol) and ethylamine hydrochloride (30 mg, 0.37 mmol) were coupled in dimethylformamide (5 mL) with HATU (65 mg, 0.17 mmol) and diisopropylethylamine (120 μ L, 0.69 mmol) according to general method A. The crude residue was suspended in ethyl acetate (5 mL), sonicated, and filtered to give the product as a white solid (60 mg, 74%). mp = 208-210 °C

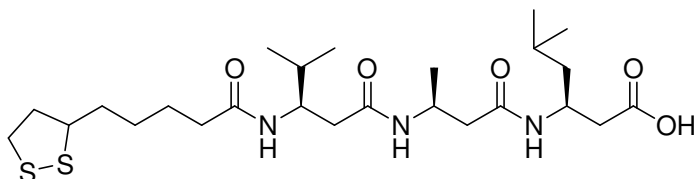
¹H NMR (TFE-*d*₃/CD₃OD 1:1) δ : 4.22 (m, 1H, NHCHCH₂CH), 4.10 (m, 1H, NHCHCH₃), 3.70 (m, 1H, NHCHCH), 3.13 (q, 2H, NHCH₂CH₃, *J*=7.2 Hz), 2.40-2.21 (m, 4H, NHCH(CH(CH₃)₂CHH, NHCH(CH₃)CHH, NHCH(CHCH₂(CH₃)₂CH₂), 2.17-2.11 (m, 2H, NHCH(CH(CH₃)₂CHH, NHCH(CH₃)CHH), 1.68 (m, 1H, NHCHCH), 1.55 (m, 1H, NHCHCH₂CH), 1.39 (m, 1H, NHCHCHH), 1.36 (s, 9H, C(CH₃)₃), 1.23 (m, 1H, NHCHCHH), 1.11 (d, 3H, NHCHCH₃, *J*=6.7 Hz), 1.06 (t, 3H, NHCH₂CH₃, *J*=7.3 Hz), 0.85 (m, 12H, 2 x CH(CH₃)₂). ¹³C NMR (TFE-*d*₃; CD₃OD 1:1) δ : 173.5, 173.3, 172.8, 158.2, 80.3, 55.9, 55.0, 46.7, 44.8, 44.6, 44.1, 43.1, 40.3, 35.3, 33.9, 28.8, 28.7, 26.0, 23.6, 22.0, 20.0, 19.5, 18.3, 14.6. HRMS calcd for C₂₄H₄₇N₄O₅ (MH⁺) 471.3546; found 471.3534.

Boc-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-*N*-ferrocenyl methanamide (BocB₃Fc)



Boc- β^3 hVal- β^3 hAla- β^3 hLeu-OH (**BocB₃OH**) (0.33 g, 0.75 mmol) and ferrocene methylamine (0.24 g, 1.1 mmol) were coupled in dimethylformamide (30 mL) with HATU (0.29 g, 0.76 mmol) and diisopropylethylamine (0.53 mL, 3.0 mmol), according to general method A. The product precipitated from the ethyl acetate and was filtered to give a yellow solid (0.24 g, 49%). mp = 210 °C (dec). ¹H NMR (TFE-d₃) δ : 7.19 (t, 1H, NHCH₂, *J*=4.6 Hz), 7.04 (d, 1H, NH, *J*=8.6 Hz), 6.99 (d, 1H, NH, *J*=8.2 Hz), 5.41 (d, 1H, NHCHCH, *J*=9.6 Hz), 4.35-4.19 (m, 11H, Fc, FcCH₂), 4.12 (m, 1H, NHCHCH₂), 4.06 (m, 1H, NHCHCH₃), 3.67 (m, 1H, NHCHCH), 2.44-2.31 (m, 4H, CH(CH₂CH(CH₃)₂)CH₂, CH(CH(CH₃)₂)CHH, CH(CH₃)CHH), 2.25-2.19 (m, 2H, CH(CH(CH₃)₂)CHH, CH(CH₃)CHH), 1.73 (m, 1H, CHCH(CH₃)₂), 1.54 (m, 1H, NHCHCH₂CH), 1.44 (m, 1H, NHCHCHH), 1.42 (s, 9H, C(CH₃)₃), 1.28 (m, 1H, NHCHCHH), 1.15 (d, 3H, CHCH₃ *J*=6.4 Hz), 0.92-0.86 (m, 12H, 2 x CH(CH₃)₂). ¹³C NMR (TFE-d₃) δ : 174.8, 174.3, 174.0, 159.4, 82.5, 72.3-71.6 (br, Fc), 55.6, 47.6, 45.5, 44.9, 44.3, 43.5, 41.1, 40.9, 34.4, 29.0, 26.5, 23.5, 22.2, 20.4, 19.7, 18.3. HRMS calcd for C₃₃H₅₃FeN₄O₅ (MH⁺) 640.3287; found 640.3315.

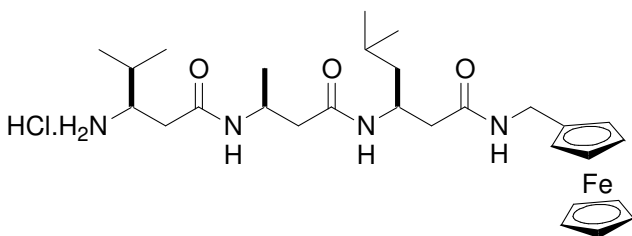
5-[1,2]Dithiolan-(3*S*/*R*)-3-yl-pentanoyl- β^3 hVal- β^3 hAla- β^3 hLeu-OH (SSB₃OH)



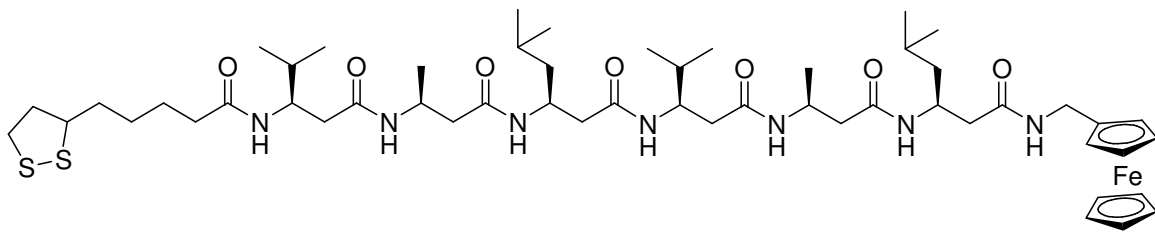
TFA- β^3 hVal- β^3 hAla- β^3 hLeu-OH (**HB₃OH**) (140 mg, 0.31 mmol) and *N*-hydroxysuccinimide-activated lipoic acid (112 mg, 0.37 mmol) were dissolved in a mixture of tetrahydrofuran (30 mL), water (20 mL) and triethylamine (205 μ L, 1.5 mmol) at rt. Triethylamine (~2 equiv) was added keeping the pH below 8.5 (universal indicator paper) and the reaction mixture was stirred for 16 h. The pH was adjusted to 3 (universal indicator paper) with 1 M aqueous citric acid and extracted twice with ethyl acetate (2 x 50 mL). The organic layers were combined, washed with water, dried (MgSO₄) and the solvent removed *in vacuo*. The product was obtained as a white solid (122 mg, 75%) that was not purified further. ¹H NMR (CD₃OD/CDCl₃ 2:1) δ : 7.76 (d, 1H, **NH**, *J*=9.4 Hz), 7.64 (d, 1H, **NH**, *J*=8.0 Hz), 7.55 (d, 1H, **NH**, *J*=9.0 Hz), 4.27 (m, 1H, **NHCH**CH₂), 4.12 (m, 1H, **NHCH**CH₃), 4.03 (m, 1H, **NHCH**CH), 3.54 (m, 1H, **SCH**), 3.18-3.06 (m, 2H, SCH₂**CH**₂), 2.48-2.36 (m, 5H, SCH₂**CHH**, CH₂CH**CH**₂CO, CHCH**CHH**CO, CH₃CH**CHH**CO), 2.24-2.16 (m, 4H, CH₂**CH**₂CO, CHCH**CHH**CO, CH₃CH**CHH**CO), 1.89 (m, 1H, SCH₂**CHH**), 1.80-1.57 (m, 6H, **NHCH**CH, **NHCH**₂**CH**, CH**CH**₂CH₂**CH**₂), 1.44 (m, 3H, CHCH₂**CH**₂CH₂, **CHH**CH(CH₃)₂), 1.30 (m, 1H, **CHH**CH(CH₃)₂), 1.13 (d, 3H, CH**CH**₃, *J*=6.7 Hz), 0.91-0.89 (m, 12H, 2 x C(**CH**₃)₂).

^{13}C NMR ($\text{CD}_3\text{OD}/\text{CDCl}_3$ 2:1) δ : 174.5, 174.2, 172.4, 172.0, 57.2, 52.9, 45.5, 44.1, 43.1, 41.0, 40.6, 39.5, 39.1, 36.9, 36.8, 35.4, 35.4, 32.9, 29.7, 29.6, 26.5, 26.0, 25.7, 23.5, 22.2, 19.9, 19.6, 18.6. HRMS calcd for $\text{C}_{25}\text{H}_{46}\text{N}_3\text{O}_5\text{S}_2$ (MH^+) 532.2879; found 532.2870.

5-[1,2]Dithiolan-(3*S*/*R*)-3-ylpentanoyl-(*R*)- $\beta^3\text{hVal}$ -(*S*)- $\beta^3\text{hAla}$ -(*S*)- $\beta^3\text{hLeu}$ -(*R*)- $\beta^3\text{hVal}$ -(*S*)- $\beta^3\text{hAla}$ -(*S*)- $\beta^3\text{hLeu}$ -1-ferrocenylmethanamide (SSB₆Fc)

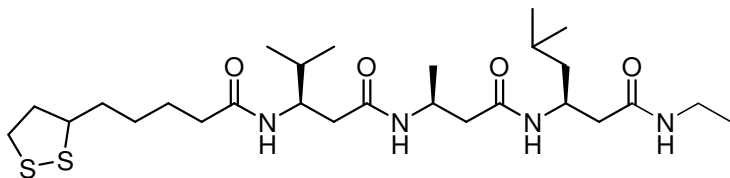


Boc- $\beta^3\text{hVal}$ - $\beta^3\text{hAla}$ - $\beta^3\text{hLeu}$ -NHCH₂Fc (**BocB₃Fc**)(92 mg, 0.14 mmol) was first dissolved in trifluoroethanol (5 mL), then HCl (4 M in dioxane, 1 mL) was added. The reaction mixture was stirred for 5 min at rt, then the solvent was removed *in vacuo*. The product (HCl· $\beta^3\text{hVal}$ - $\beta^3\text{hAla}$ - $\beta^3\text{hLeu}$ -NHCH₂Fc) was obtained as a glassy yellow solid and used without purification.



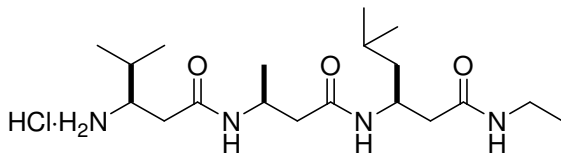
HCl· β^3 hVal- β^3 hAla- β^3 hLeu-NHCH₂Fc (**HB₃Fc**) (83 mg, 0.14 mmol) and lipoamide- β^3 hVal- β^3 hAla- β^3 hLeu-OH (**SSB₃OH**) (76 mg, 0.14 mmol) were coupled in dimethylformamide (5 mL) using HATU (54 mg, 0.14 mmol) and diisopropylethylamine (105 μ L, 0.57 mmol) according to general method B. Chromatography (lipophilic sephadex; 1:0→1:9→1:1→0:1 methanol, trifluoroethanol) gave the product as a glassy yellow solid (107 mg, 71%). ¹H NMR (TFE-*d*₃) δ : 7.30- 7.01 (m, 6H, **NH**), 6.91 (d, 1H, **NH**, *J*=8.3 Hz), 4.06-4.38 (m, 17H, 6 x **NHCH**, **FcCH₂**), 3.59 (m, 1H, **SCH**), 3.18 (m, 1H, **SCHH**), 3.12 (m, 1H, **SCHH**), 2.52-2.21 (m, 15H, 7 x **CH₂CO**, **SCH₂CHH**), 1.93 (m, 1H, **SCH₂CHH**), 1.82-1.52 (m, 8H, 2 x **NHCHCH(CH₃)₂**, 2 x **NHCHCH₂CH**, **CH₂CH₂CH₂CH₂CO**), 1.45 (m, 4H, **SCHCH₂CH₂(CH₂)₂**, 2 x **CHCHHCH(CH₃)₂**), 1.29 (m, 2H, 2 x **CHCHHCH(CH₃)₂**), 1.15 (m, 6H, 2 x **CHCH₃**), 0.98-0.81 (m, 24H, 4 x **CH(CH₃)₂**). ¹³C NMR (TFE-*d*₃) δ : 177.7, 174.5, 174.3, 174.1, 174.0, 173.9, 58.5, 54.4, 47.5, 47.4, 45.4, 44.0, 43.2, 42.0, 39.6, 38.0, 36.1, 34.0, 30.5, 30.5, 27.5, 26.7, 26.6, 23.7, 23.6, 22.7, 22.5, 20.7, 20.6, 19.8, 19.1, 19.0. HRMS calcd for C₅₃H₈₈N₇O₇S₂Fe (MH⁺) 1054.5536; found 1054.5511.

5-[1,2]Dithiolan-(3*S*/*R*)-3-yl-pentanoyl-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-N-ethanamide (SSB₃NHEt)



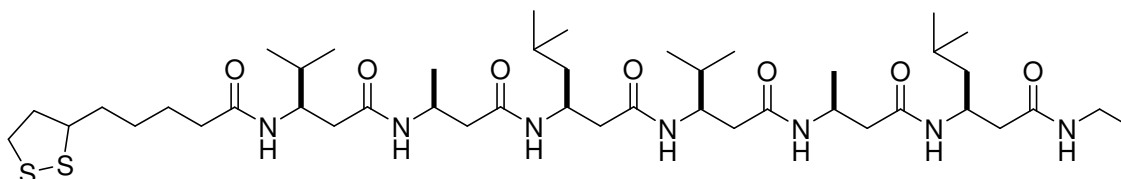
Lipoamide-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-OH (**SSB₃OH**) (50 mg, 94 μ mol) and ethylamine hydrochloride (10 mg, 0.12 mmol) were coupled in dimethylformamide (10 mL) using HATU (36 mg, 95 μ mol) and diisopropylethylamine (65 μ L, 0.38 mmol). The reaction mixture was stirred for 16 h, at which time the solvent was removed *in vacuo*. The residue was sonicated in ethanol (5 mL) and filtered to give an off-white solid (36 mg, 68%). ¹H NMR (CD₃OD/TFE-*d*₃ 2:1) δ : 4.25 (m, 1H, **CHCH₂CH(CH₃)₂**), 4.13 (m, 1H, **CHCH₃**), 4.07 (m, 1H, **CHCH(CH₃)₂**), 3.53 (m, 1H, **SCH**), 3.17-3.06 (m, 4H, **SCH₂**, **CH₂CH₃**), 2.46-2.17 (m, 9H, 3 x **NHCHCH₂**, **SCHCHH**, **(CH₂)₃CH₂CO**), 1.86 (m, 1H, **SCHCHH**), 1.78-1.44 (m, 9H, **SCHCH₂CH₂CH₂**, **CHCH(CH₃)₂**, **CHHCH(CH₃)₂**), 1.28 (m, 1H, **CHHCH(CH₃)₂**), 1.26 (m, 1H, **CHHCH(CH₃)₂**), 1.13-1.07 (m, 6H, **CHCH₃**, **CH₂CH₃**), 0.91-0.87 (m, 12H, 2 x **CH(CH₃)₂**). ¹³C NMR (CD₃OD/TFE-*d*₃ 2:1) δ : 176.9, 174.2, 174.1, 173.7, 58.4, 54.4, 47.6, 45.6, 45.4, 44.4, 43.9, 42.2, 40.8, 40.1, 38.0, 36.5, 36.3, 34.3, 30.8, 27.7, 26.9, 24.3, 22.9, 20.9, 20.3, 19.3, 15.4. HRMS calc for C₂₇H₅₁N₄O₄S₂ (MH⁺) 559.3352; found 559.3332.

5-[1,2]Dithiolan-(3*S*/*R*)-3-yl-pentanoyl-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-ethanamide (SSB₆NHET)



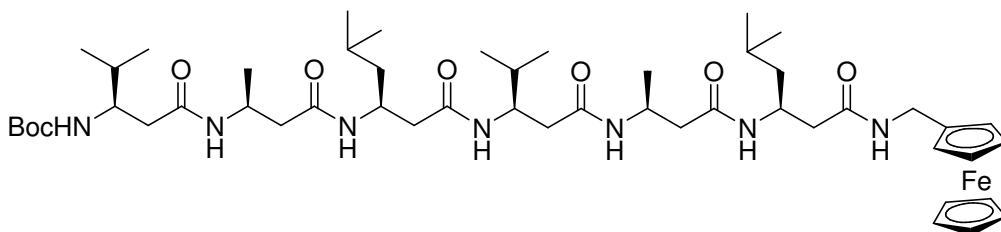
Boc- β^3 hVal- β^3 hAla- β^3 hLeu-NHET (**BocB₃NHET**) (49 mg, 0.10 mmol) was dissolved in trifluoroacetic acid (2.5 mL) and dichloromethane (2.5 mL), then stirred for 2 h at rt. The solvent was removed *in vacuo*, twice co-evaporated *in vacuo* with

dichloromethane, and once with methanol to give the deprotected amine as a trifluoroacetate salt as a light yellow oil (50 mg, quant).



HCl· β 3hVal- β 3hAla- β 3hLeu-NHEt (**HB₃NHEt**) (34 mg, 70 μ mol) and lipoamide- β 3hVal- β 3hAla- β 3hLeu-OH (**SSB₃OH**) (37 mg, 70 μ mol) were coupled in dimethylformamide (5 mL) using HATU (27 mg, 70 μ mol) and diisopropylethylamine (48 μ L, 0.28 mmol) according to general method B. The product was obtained as an off-white solid (25 mg, 41%). mp = 292°C (dec). ^1H NMR (TFE- d_3) δ : 7.24-6.93 (m, 7H, 7 x NH), 4.31-4.22 (m, 4H, 4 x NHCH), 4.08 (m, 2H, 2 x NHCH), 3.59-3.56 (m, 1H, SCH), 3.24-3.08 (m, 4H, CH₂S, NHCH₂), 2.47-2.30 (m, 15H, 6 x NHCHCH₂CO, (CH₂)₃CH₂CO, SCH₂CHH), 1.94-1.90 (m, 1H, SCH₂CHH), 1.81-1.53 (m, 8H, 2 x NHCHCH(CH₃)₂, 2 x NHCHCH₂CH, CH₂CH₂CH₂CH₂CO), 1.46 (m, 4H, SCHCH₂CH₂(CH₂)₂, 2 x CHCHHCH(CH₃)₂), 1.31 (m, 2H, 2 x CHCHHCH(CH₃)₂), 1.17-1.10 (m, 9H, 2 x CHCH₃, CH₂CH₃), 0.91- 0.90 (m, 24H, CH(CH₃)₂). ^{13}C NMR (TFE- d_3) δ : 178.9, 176.0, 175.6, 175.4, 175.1, 175.0, 175.0, 59.6, 55.5, 48.7, 48.6, 46.5, 46.4, 45.1, 44.6, 44.4, 43.2, 41.4, 40.7, 39.2, 37.6, 37.2, 35.2, 35.1, 31.7, 31.6, 28.7, 27.8, 27.7, 24.9, 24.8, 23.8, 23.6, 21.9, 21.7, 21.0, 20.9, 20.2, 20.2, 15.9. HRMS calcd for C₄₄H₈₂N₇O₇S₂ (MH⁺) 884.5717; found 884.5753.

Boc-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-(*R*)- β^3 hVal-(*S*)- β^3 hAla-(*S*)- β^3 hLeu-*N*-ferrocenylmethanamide (BocB₆Fc)

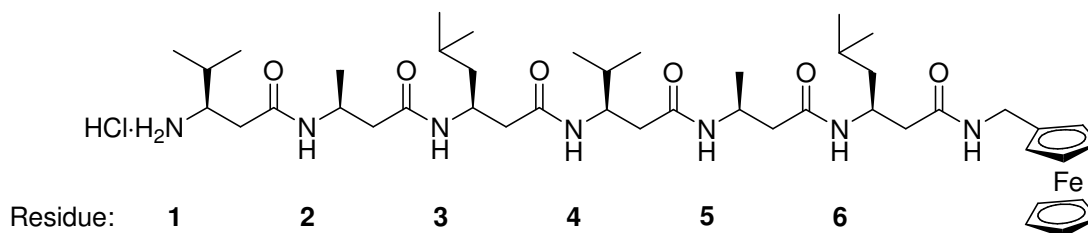


Boc- β^3 hVal- β^3 hAla- β^3 hLeu-OH (**BocB₃OH**) (38 mg, 87 μ mol) and HCl· β^3 hVal- β^3 hAla- β^3 hLeu-NHCH₂Fc (**HB₃Fc**) (50 mg, 87 μ mol) were coupled in DMF (50 mL) using HATU (33 mg, 87 μ mol) and diisopropylethylamine (60 μ L, 0.35 mmol) according to general method B to give a yellow powder (71 mg, 85%). mp = >250 °C (dec). ¹H NMR (TFE-*d*₃) δ : 7.17-7.03 (m, 5H, 5 x NH), 6.98 (d, 1H, NH, *J*=8.7 Hz), 5.46 (d, 1H, NH, *J*=9.7 Hz), 4.54-4.12 (m, 15H, CH₂Fc, 2 x NHCHCH₃, 2 x NHCHCH₂), 4.04 (m, 1H, CH₂CONHCHCH), 3.69 (m, 1H, BocNHCH), 2.45-2.21 (m, 12H, 6 x CH₂CO), 1.80-1.69 (m, 2H, 2 x CHCH(CH₃)₂), 1.56 (m, 2H, 2 x NHCHCH₂CH), 1.47-1.42 (m, 11H, C(CH₃)₃, 2 x CHCHHCH(CH₃)₂), 1.29 (m, 2H, CHCHHCH(CH₃)₂), 1.18-1.13 (m, 6H, 2 x NHCHCH₃), 0.91-0.86 (m, 24H, 2 x CH(CH₃)₂).

¹³C NMR (TFE-*d*₃) δ : 174.8, 174.2, 174.2, 174.1, 174.0, 173.9, 82.5, 74.4 (10C, Fc), 55.7, 54.4, 47.5, 47.5, 47.5, 47.5, 45.4, 45.4, 45.0, 44.9, 44.3, 44.0, 43.3, 43.3, 43.3, 41.1, 40.2, 40.2, 34.4, 33.6, 29.0, 26.6, 26.5, 23.7, 23.6, 22.4, 22.3, 20.5, 20.5, 19.7, 18.9, 18.4.

HRMS calcd for C₅₀H₈₄N₇O₈Fe (MH⁺) 966.5731; found 966.5733.

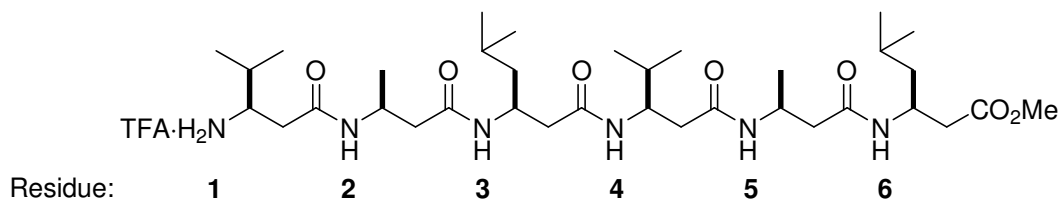
HCl·(R)- β^3 hVal-(S)- β^3 hAla-(S)- β^3 hLeu-(R)- β^3 hVal-(S)- β^3 hAla-(S)- β^3 hLeu-N-ferrocenylmethanamide (HB₆Fc)



Boc-(β^3 hVal- β^3 hAla- β^3 hLeu)₂-NHCH₂Fc (**BocB₆Fc**) (25 mg, 26 μ mol) was stirred in 4M HCl/dioxane (0.5 mL) for 10 min at rt. The solvent was removed *in vacuo* and the residue passed through an alumina plug, eluting with methanol. The solvent was removed *in vacuo* and the resulting residue was purified by column chromatography (C₁₈-coated silica; 1:1→0:1 water, methanol) to give the product as a yellow solid (22 mg, 95%). ¹H NMR (CD₃OD) δ : 8.41 (d, 1H, 2-NH, *J*=9.5 Hz), 8.32 (d, 1H, 3-NH, *J*=9.1 Hz), 8.27 (d, 1H, NH, *J*=9.9 Hz), 8.07 (d, 1H, 4-NH, *J*=9.3 Hz), 7.98 (d, 1H, NHCH₂Fc, *J*=8.4 Hz), 7.70 (d, 1H, 6-NH, *J*=10.0 Hz), 7.55 (d, 1H, 5-NH, *J*=8.4 Hz) 4.55 (m, 1H, 2-NHCH), 4.46 (m, 1H, 5-NHCH), 4.44 (m, 1H, 6-NHCH), 4.37 (m, 1H, 3-NHCH), 4.26 (m, 2H, CH₂Fc), 4.20 (m, 1H, 4-NHCH), 3.52 (m, 1H, 1-NHCH), 2.79 (dd, 1H, 1-CHHCO, *J*=15.2, 11.6 Hz), 2.72 (dd, 1H, 2-CHHCO, *J*=14.9, 11.6 Hz), 2.62 (m, 1H, 1-CHHCO), 2.60 (m, 1H, 3-CHHCO), 2.53 (m, 1H, 4-CHHCO), 2.49 (m, 1H, 2-CHHCO), 2.47 (m, 1H, 5-CHHCO), 2.41 (m, 1H, 3-CHHCO), 2.30 (m, 1H, 5-CHHCO), 2.29 (m, 2H, 6-CH₂CO), 2.20 (m, 1H, 4-CHHCO), 1.76 (m, 1H, 4-NHCHCH), 1.57 (m, 2H, 2 x NHCHCH₂CH), 1.42 (m, 1H, 3-NHCHCHH), 1.36 (m, 1H, 6-NHCHCHH), 1.25 (m, 2H, 3,6-NHCHCHH), 1.18 (d, 3H, 2-CHCH₃, *J*=6.7 Hz), 1.14 (d, 3H, 5-CHCH₃, *J*=6.7 Hz),

1H, CHCH(CH₃)₂) 3.58 (m, 1H, SCH), 3.18-3.06 (m, 2H, SCH₂), 2.48-2.19 (m, 9H, 3 x CH₂CO, SCHCHH, (CH₂)₃CH₂CO), 1.93 (m, 1H, SCHCHH), 1.80-1.40 (m, 9H, SCHCH₂CH₂CH₂, CHCH(CH₃)₂, CHHCH(CH₃)₂), 1.28 (m, 1H, CHHCH(CH₃)₂), 1.13 (d, 3H, CHCH₃, *J*=6.6 Hz), 0.93-0.86 (m, 12H, 2 x CH(CH₃)₂). ¹³C NMR (TFE-*d*₃) δ: 177.5, 174.4, 174.2, 174.0, 71.2, 70.4, 58.4, 54.2, 47.5, 45.4, 44.9, 44.0, 43.4, 41.9, 41.0, 40.4, 39.5, 37.9, 35.9, 33.8, 30.4, 27.3, 26.5, 23.5, 22.3, 20.4, 19.8, 18.8. HRMS calc for C₃₆H₅₇N₄O₄S₂Fe (MH⁺) 729.3171; found 726.3161.

TFA-(*R*)-β³hVal-(*S*)-β³hAla-(*S*)-β³hLeu-(*R*)-β³hVal-(*S*)-β³hAla-(*S*)-β³hLeu methyl ester (HB₆OMe)



Boc-(β³hVal-β³hAla-β³hLeu)₂ methyl ester (**BocB₆OMe**) (21 mg, 27 μmol) was dissolved in trifluoroacetic acid (2 mL) and the solution stirred for 2 h at rt. The solvent was removed *in vacuo* and the residue twice resuspended in dichloromethane and concentrated *in vacuo*. The residue was redissolved in methanol and concentrated *in vacuo* to give a light brown solid (21 mg, quant.) that gave spectral and physical properties in agreement with those in the literature.^{S1} mp= 214-216 °C (220-222 °C).^{S1} Where necessary, the chemical shift data is appended with the residue number indicated in the chemical structure above. ¹H NMR (CD₃OD) δ: 8.74 (d, 1H, NH, *J*=4.8 Hz), 8.43

(d, 1H, **NH**, $J=7.7$ Hz), 8.39 (d, 1H, **NH**, $J=9.3$ Hz), 8.26 (d, 1H, **NH**, $J=8.9$ Hz), 7.80 (d, 1H, **NH**, $J=7.0$ Hz), 7.48 (d, 1H, **NH**, $J=9.4$ Hz), 4.54 (m, 1H, 2-NH**CHCH**₃), 4.45 (m, 2H, 5-NH**CHCH**₃, 6-NH**CHCH**₂), 4.38 (m, 1H, 3-NH**CHCH**₂), 4.22 (m, 1H, 4-NH**CHCH**), 3.70 (s, 3H, **OCH**₃), 3.52 (m, 1-NH₂**CHCH**), 2.76 (dd, 1H, 1-NH₂**CHCHH**, $J=15.4, 11.5$ Hz), 2.69 (m, 1H, 2-NH**CHCHH**), 2.64 (m, 1H, 6-NH**CHCHHCO**), 2.59 (m, 1H, 1-NH**CHCHH**), 2.52 (m, 1H, 6-NH**CHCHHCO**), 2.50 (m, 2H, 3-NH**CHCHHCO**, 4-NH**CHCHH**), 2.47 (m, 1H, 2-NH**CHCHH**), 2.45 (m, 1H, 5-NH**CHCHH**), 2.39 (m, 1H, 3-NH**CHCHHCO**), 2.33 (dd, 1H, 5-NH**CHCHH**, $J=15.1, 10.8$ Hz), 2.22 (dd, 1H, 4-NH**CHCHH**, $J=14.7, 11.8$ Hz), 2.07 (m, 1H, 1-NH₂**CHCH**), 1.70 (m, 1H, 4-NH**CHCH**), 1.56 (m, 2H, 3,6-NH**CHCH**₂**CH**), 1.42 (m, 2H, 3,6-NH**CHCHH**), 1.24 (m, 2H, 3,6-NH**CHCHH**), 1.20 (d, 3H, 2-NH**CHCH**₃, $J=6.6$ Hz), 1.14 (d, 3H, 5-NH**CHCH**₃, $J=6.7$ Hz), 1.08 (d, 6H, 1-**CHCH(CH**₃)₂, $J=6.9$ Hz), 0.92 (m, 12H, 4-**CHCH(CH**₃)₂), 0.88 (m, 12H, 3,6-**CH**₂**CH(CH**₃)₂).

S1. Seebach, D.; Overhand, M.; Kuehnle, F. N. M.; Martinoni, B., *Helvetica Chimica Acta* **1996**, 79, (4), 913-941.

S2. Beer, P. D.; Smith, D. K., *Journal of the Chemical Society, Dalton Transactions: Inorganic Chemistry* **1998**, (3), 417-423.

S3. Sun, H. W., Qingmin; Huang, Runqiu; Li, Heng; Li, Yonghong., *Journal of Organometallic Chemistry* **2002**, 655, (1-2), 182-185. 12.

S4. Gruzman, A.; Hidmi, A.; Katzhendler, J.; Haj-Yehie, A.; Sasson, S., *Bioorganic & Medicinal Chemistry* **2004**, 12, (5), 1183-1190.

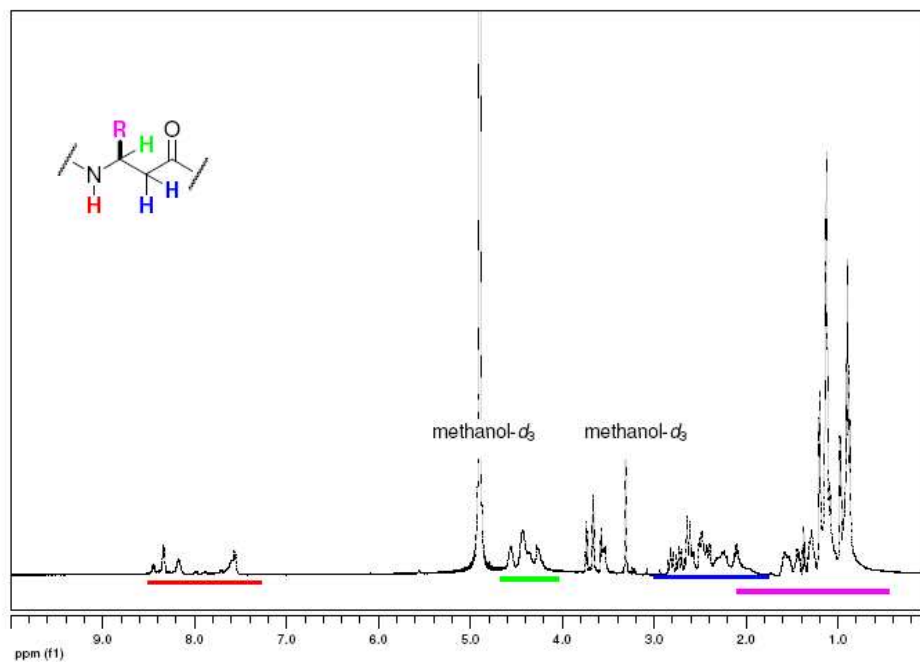


Figure S1 NMR spectrum of **B₆Fc** in *d*₃-methanol. The colored bands indicate the different proton environments of the peptide.

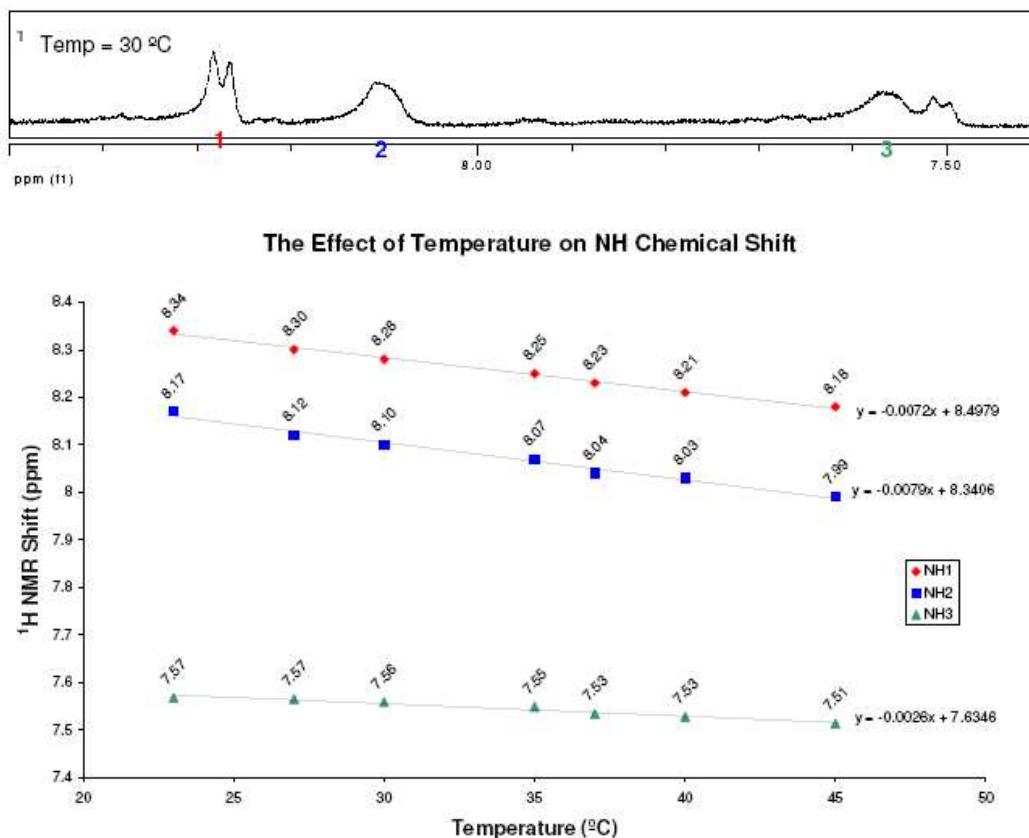


Figure S2 (top) NMR spectrum of **B₆Fc** in *d*₃-methanol at 30°C. (bottom) Chemical shift (ppm) of three NH protons of **B₆Fc** at different temperatures. The chemical shift values are shown next to the data points.

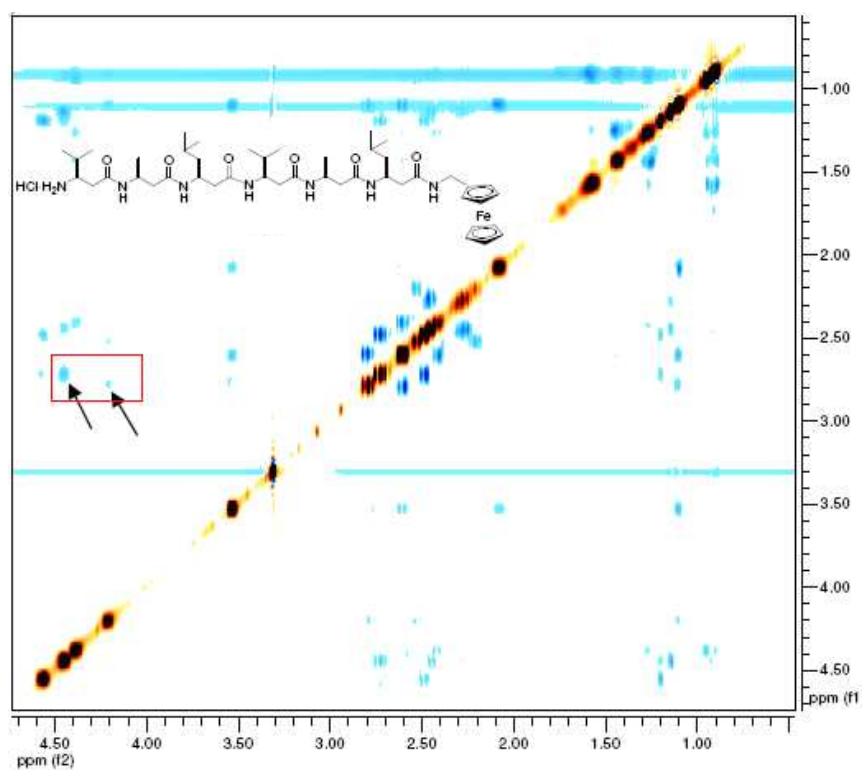
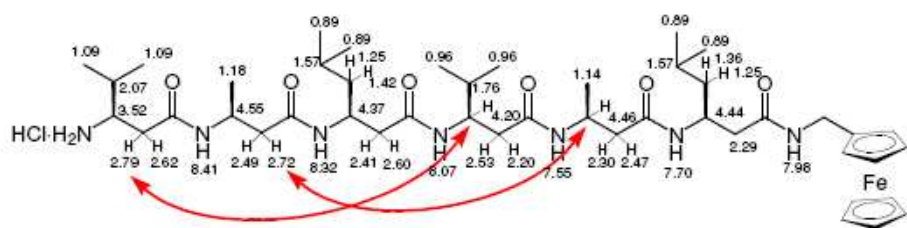


Figure S3 The partially assigned structure of **B₆Fc** showing: (top) NMR shifts, the red arrows indicate proton-proton correlations observed in the (bottom) 2D ROESY spectrum. The black arrows show the important structural correlations between backbone protons as illustrated above.

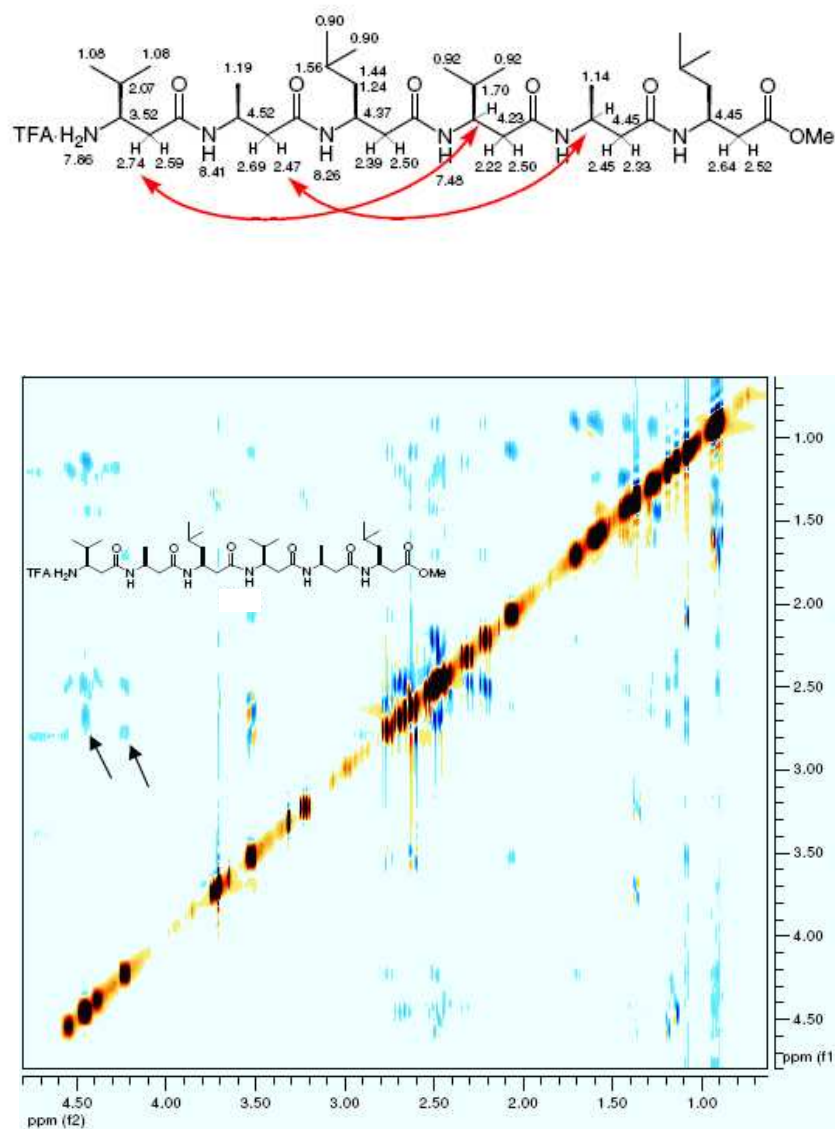
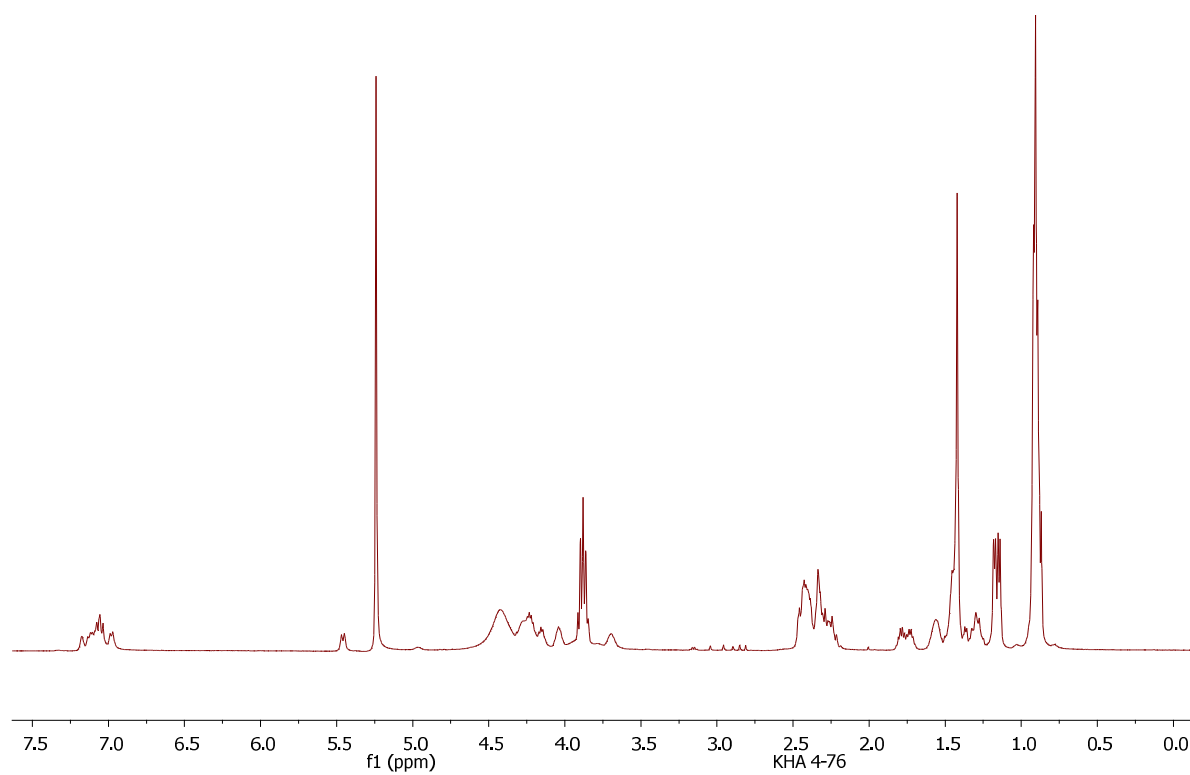
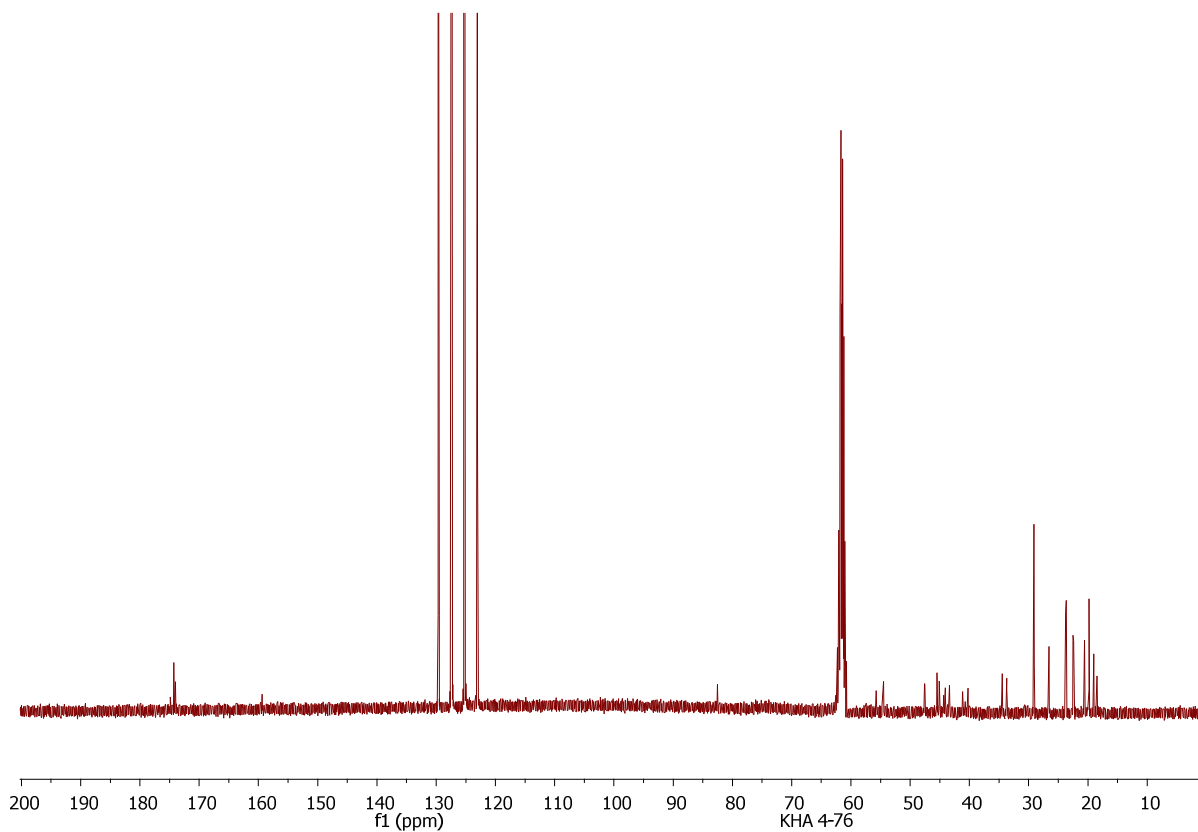


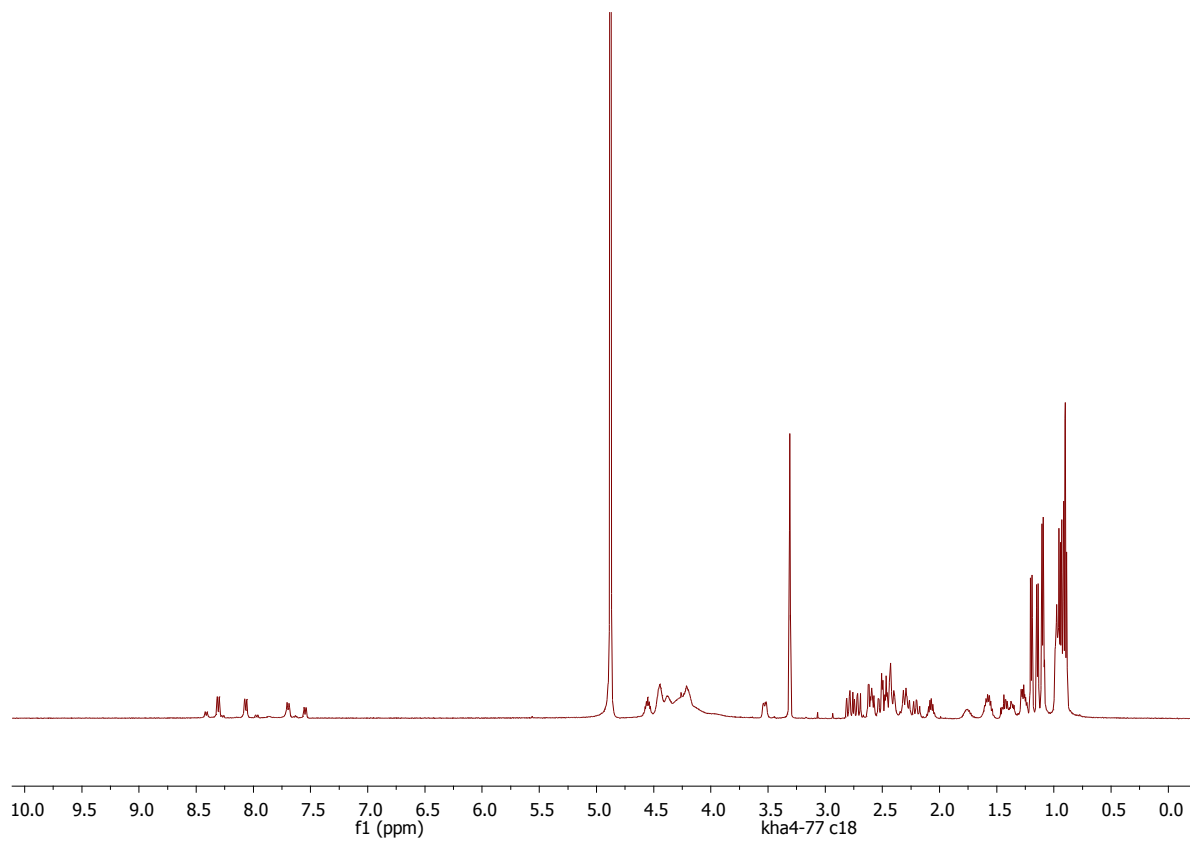
Figure S4 The partially assigned structure of **pB₆** showing: (top) NMR shifts, the red arrows indicate proton-proton correlations observed in the (bottom) 2D ROESY spectrum. The black arrows show the important structural correlations between backbone protons as illustrated above.

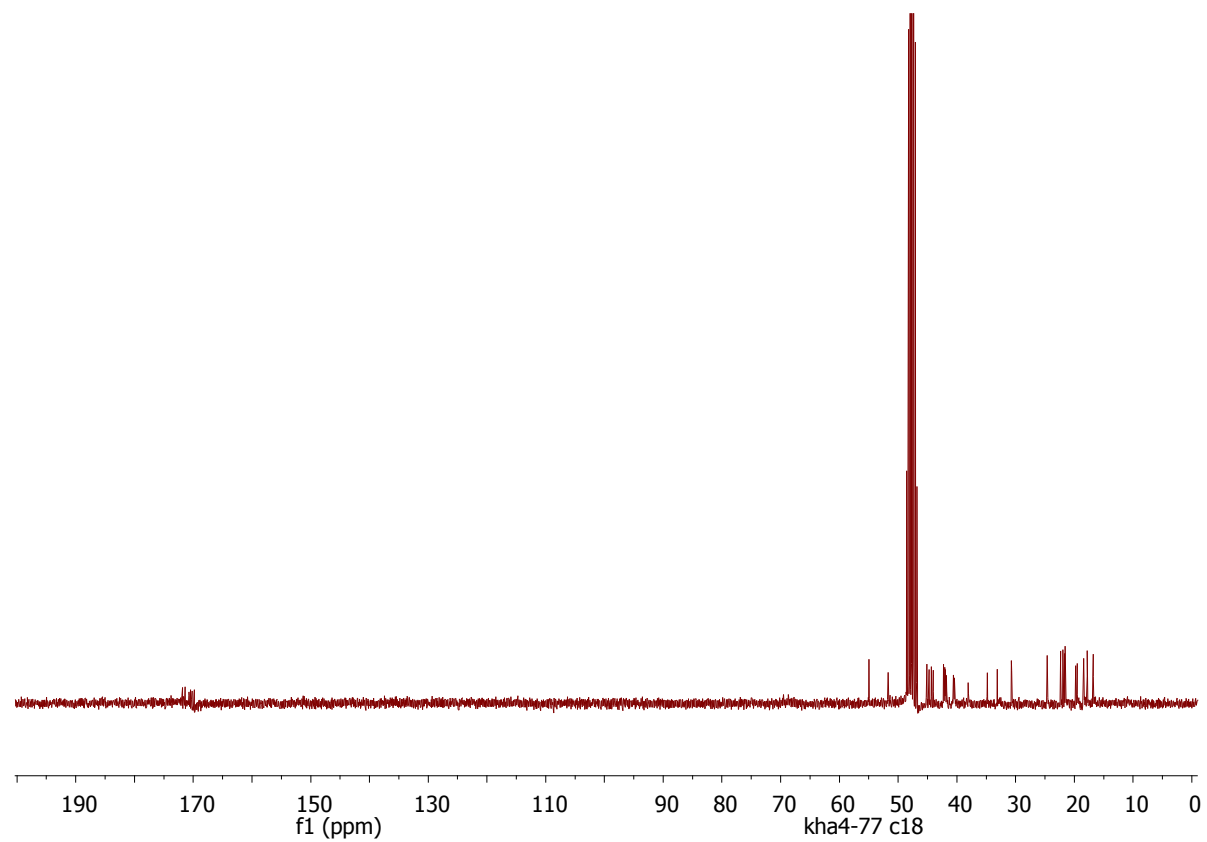
BocB₆Fc



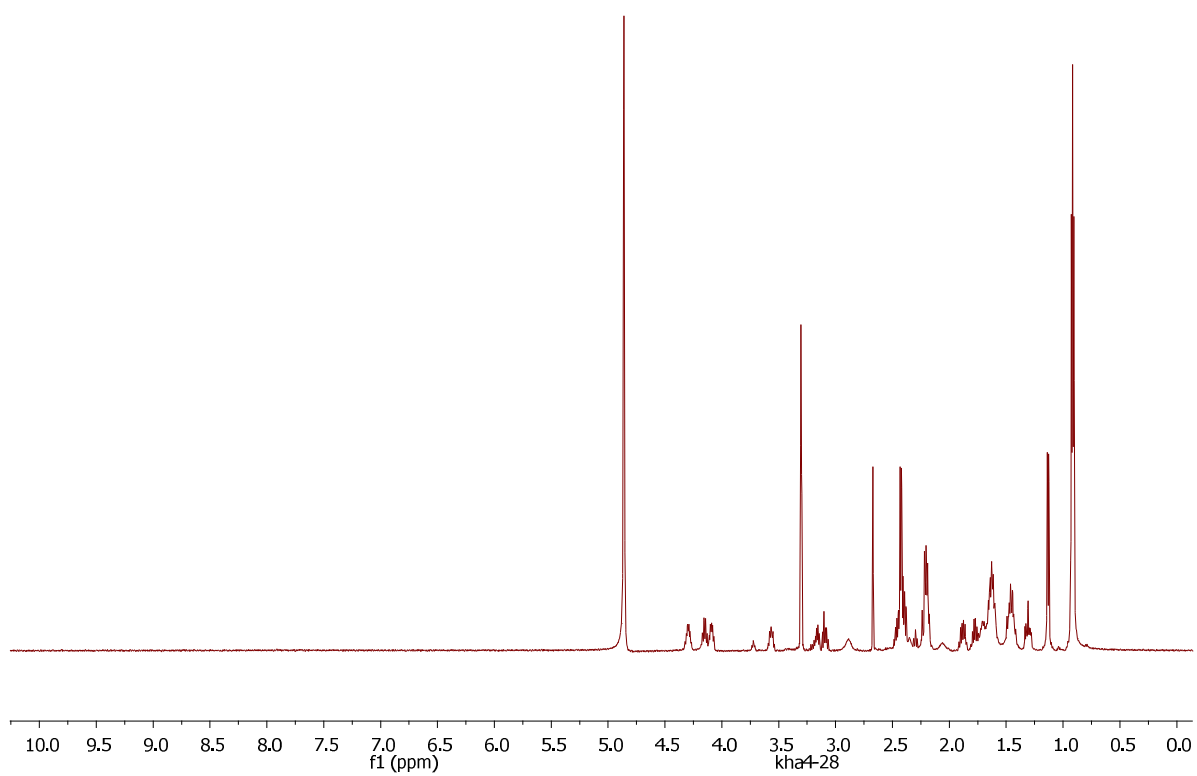


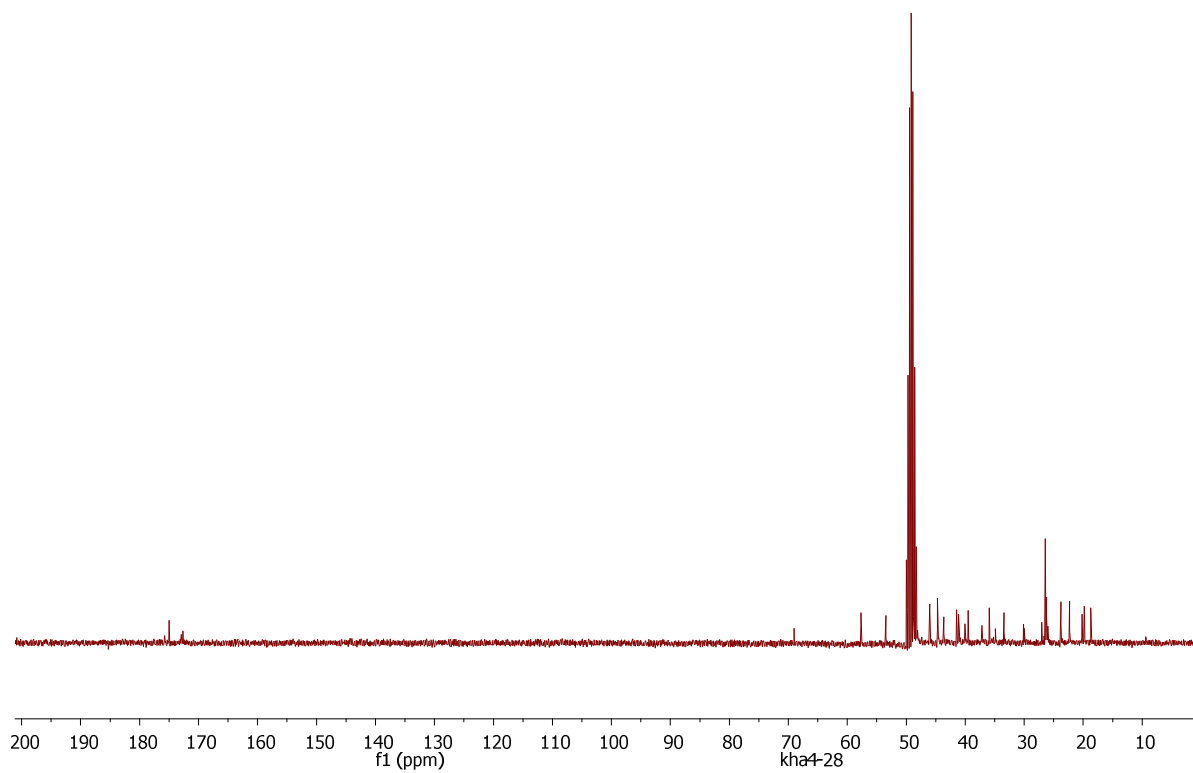
B_6Fc



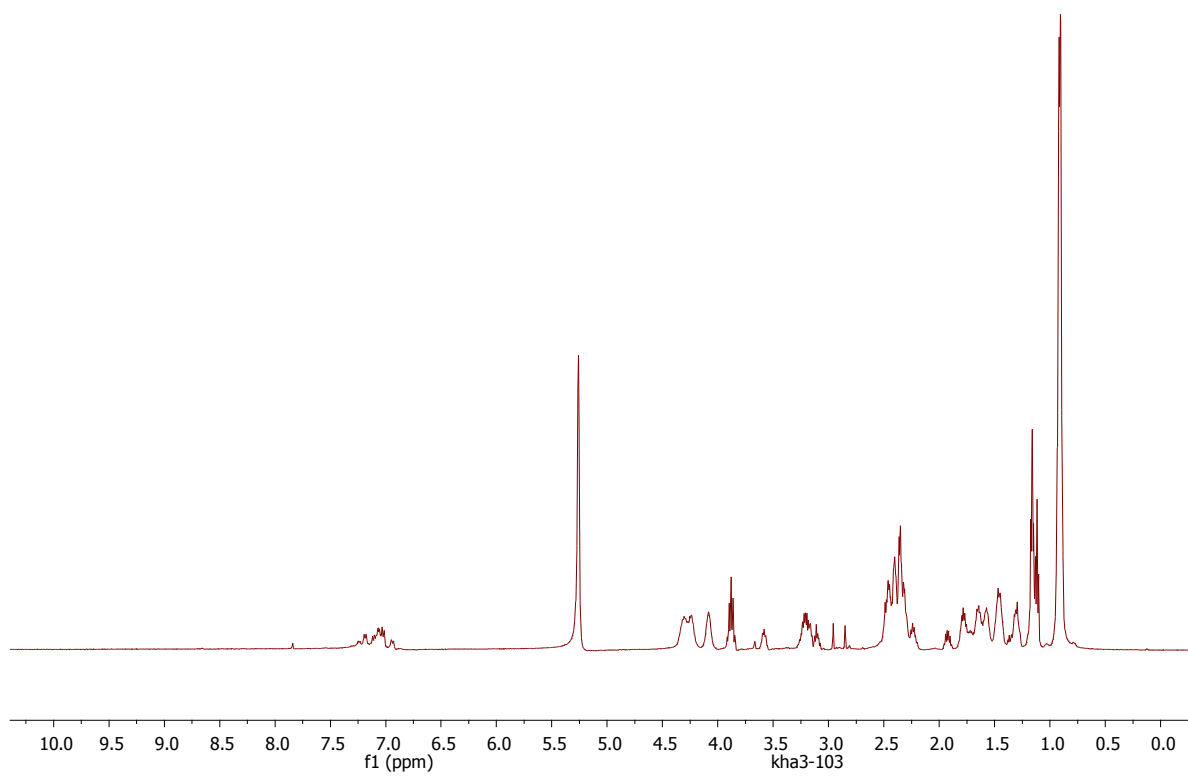


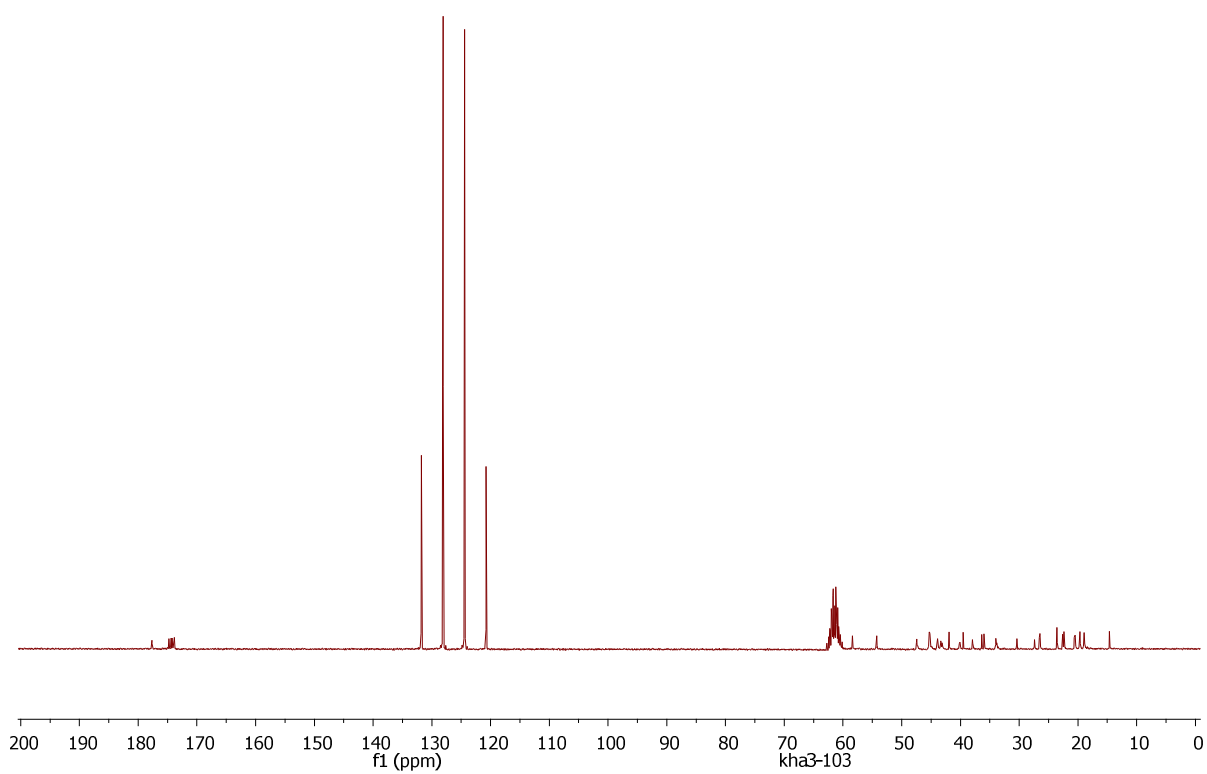
SSB_3OH



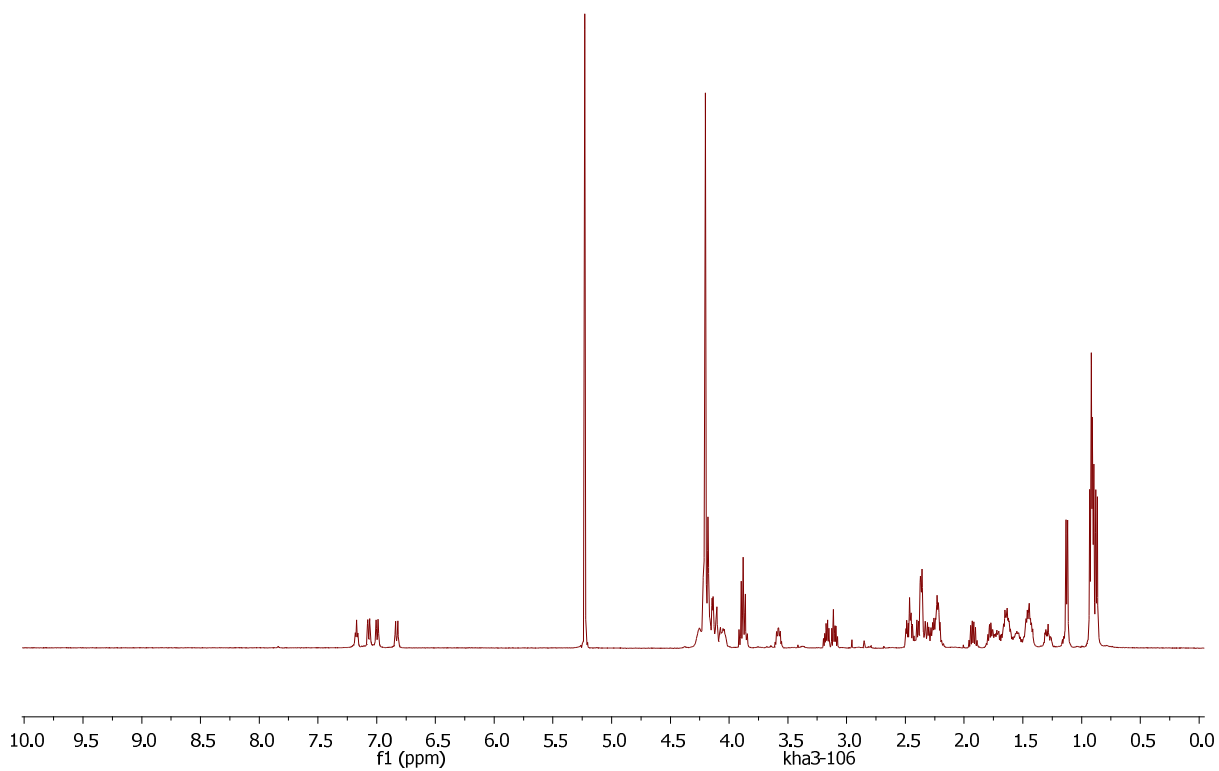


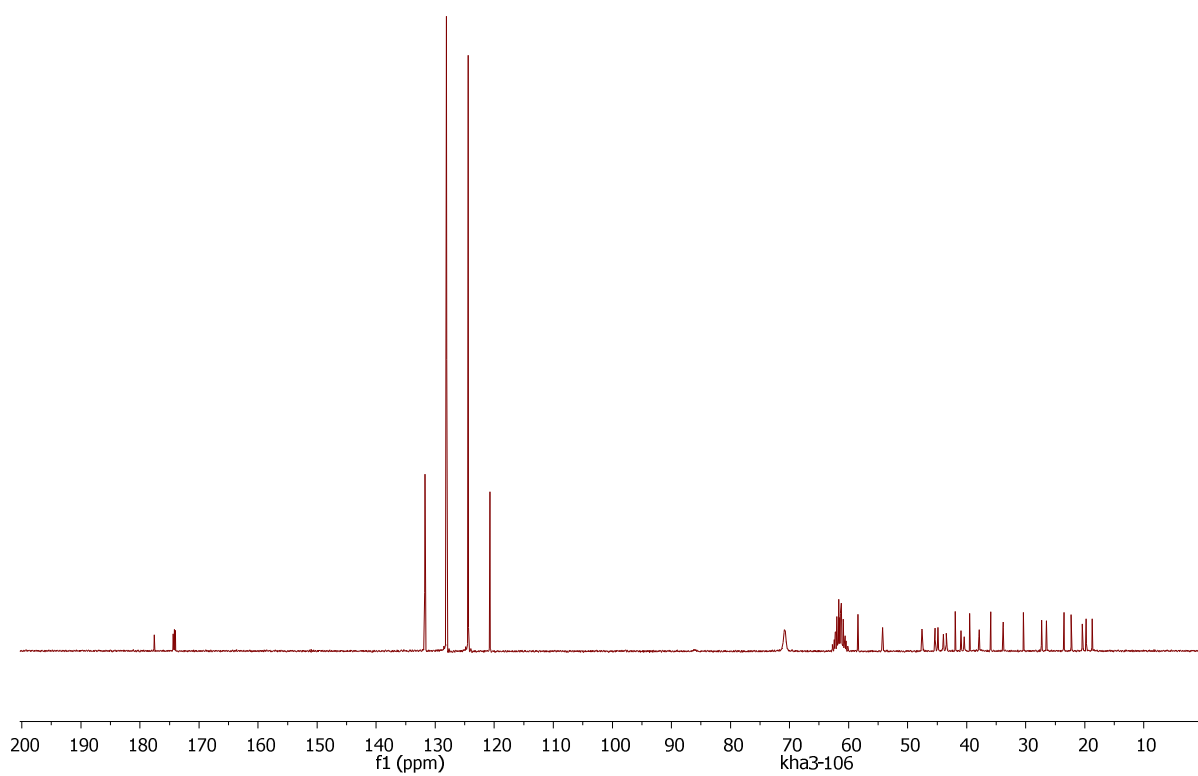
SSB₆Et



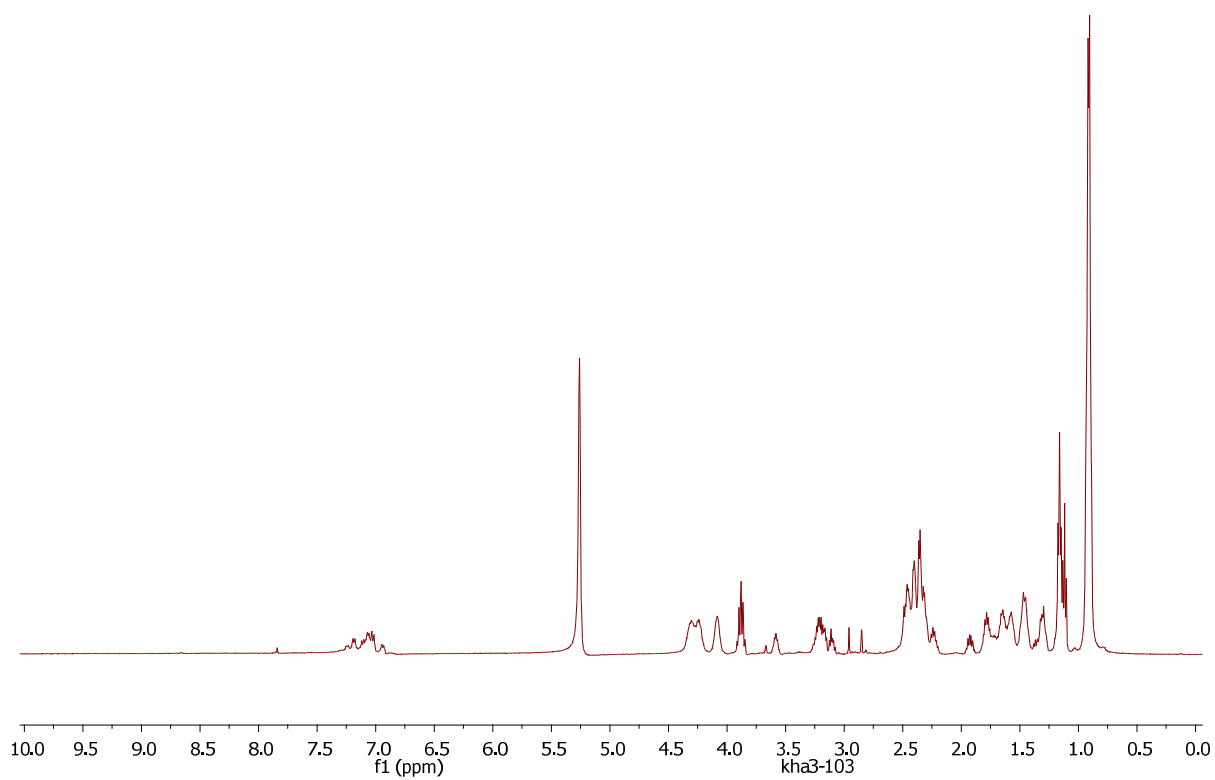


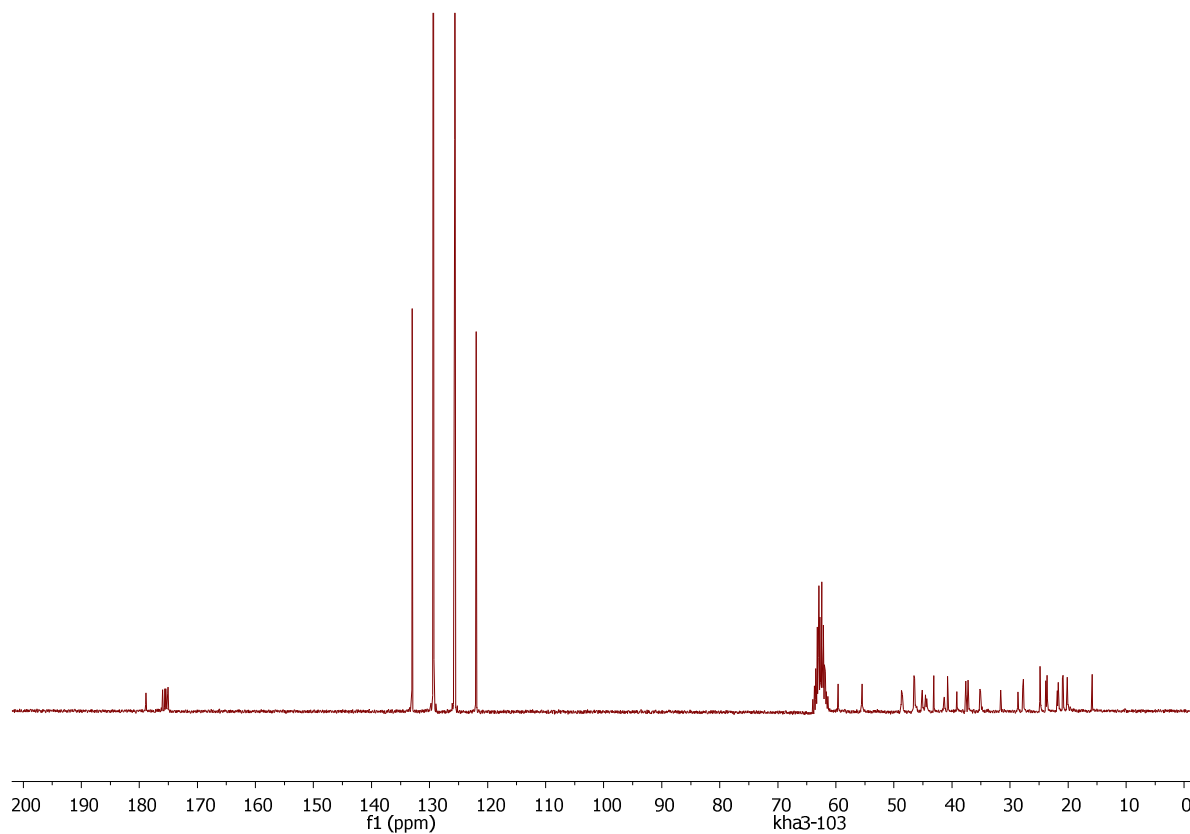
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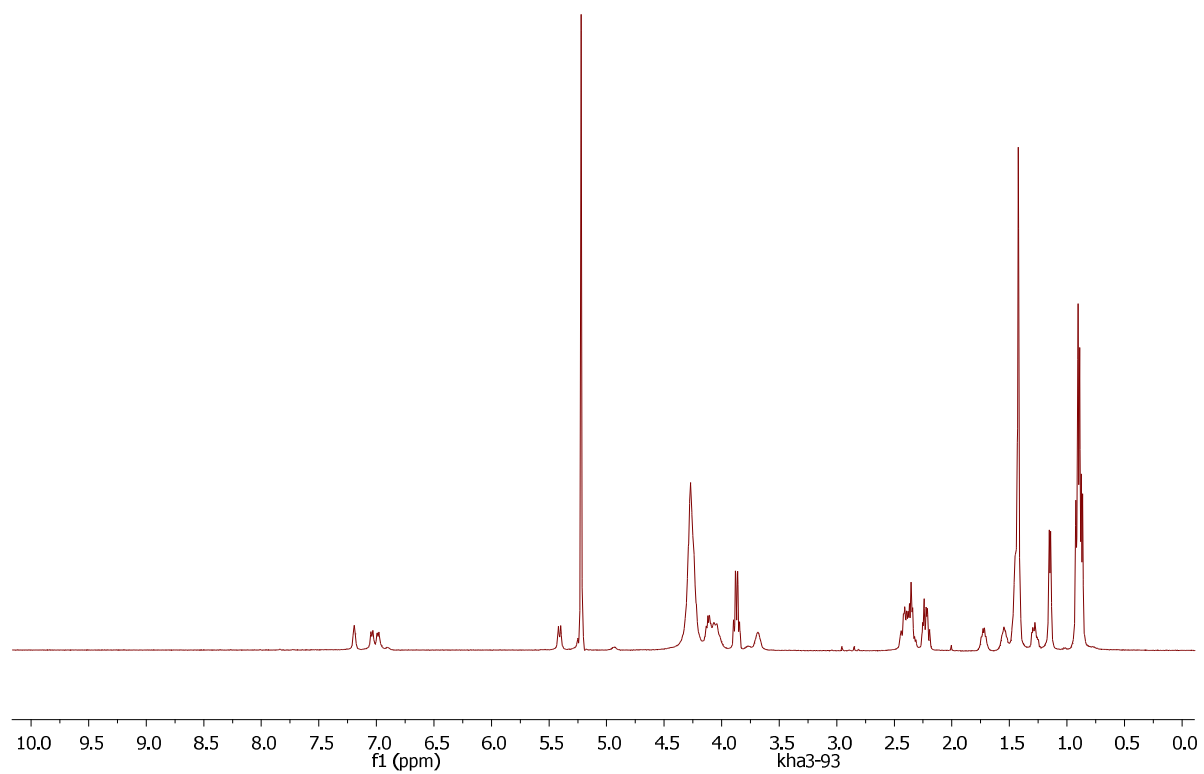


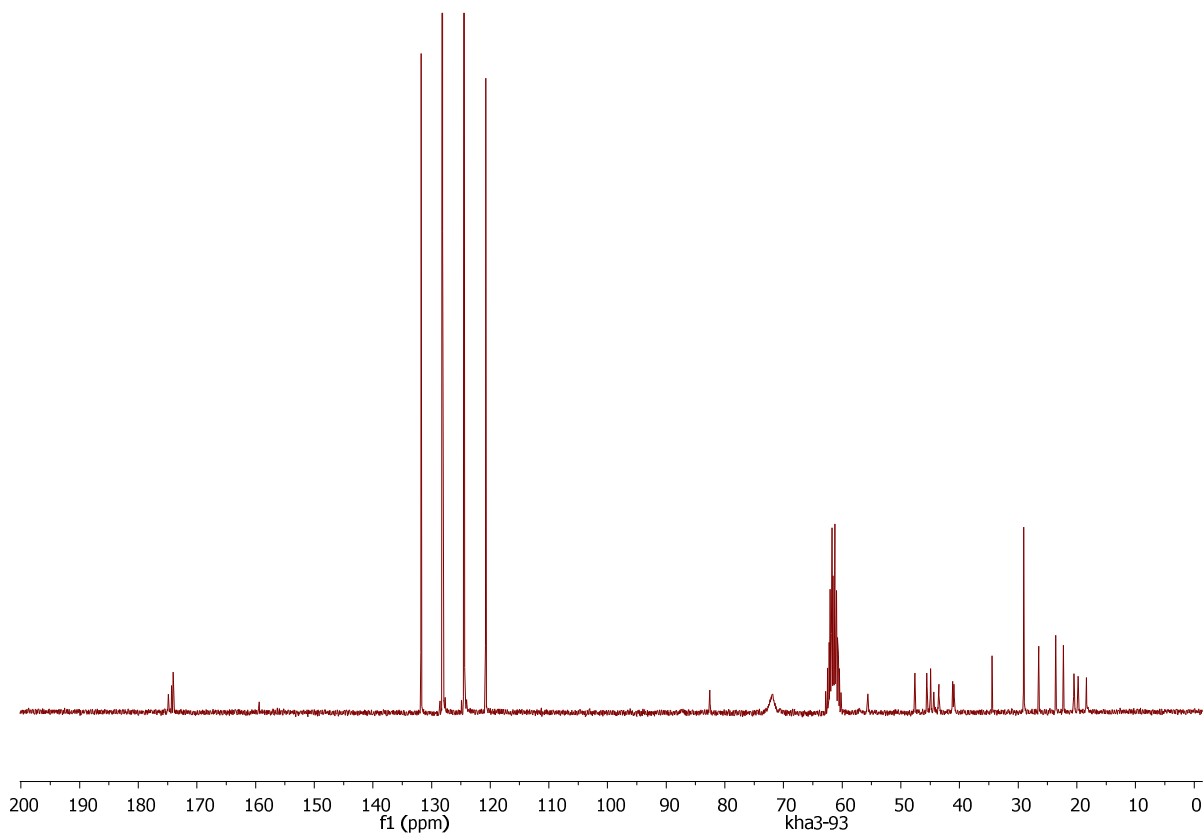
SSB_6Et



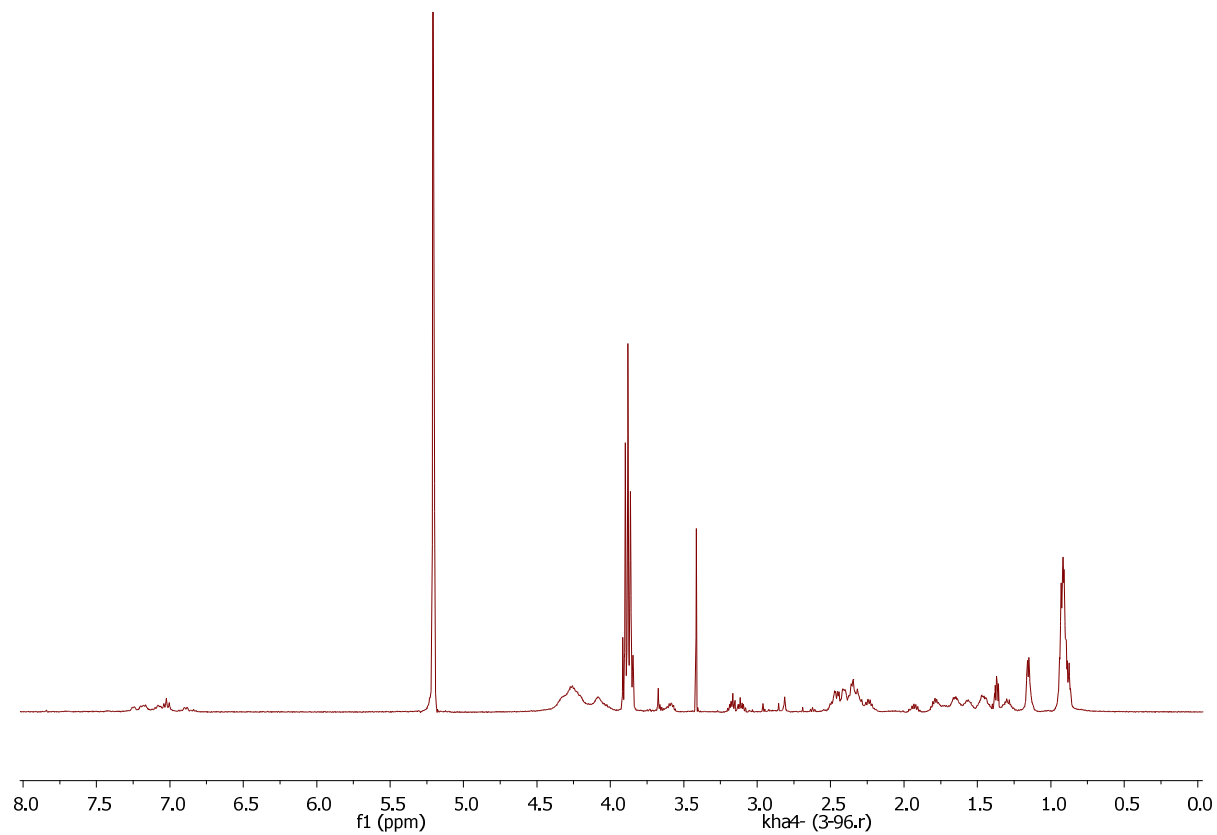


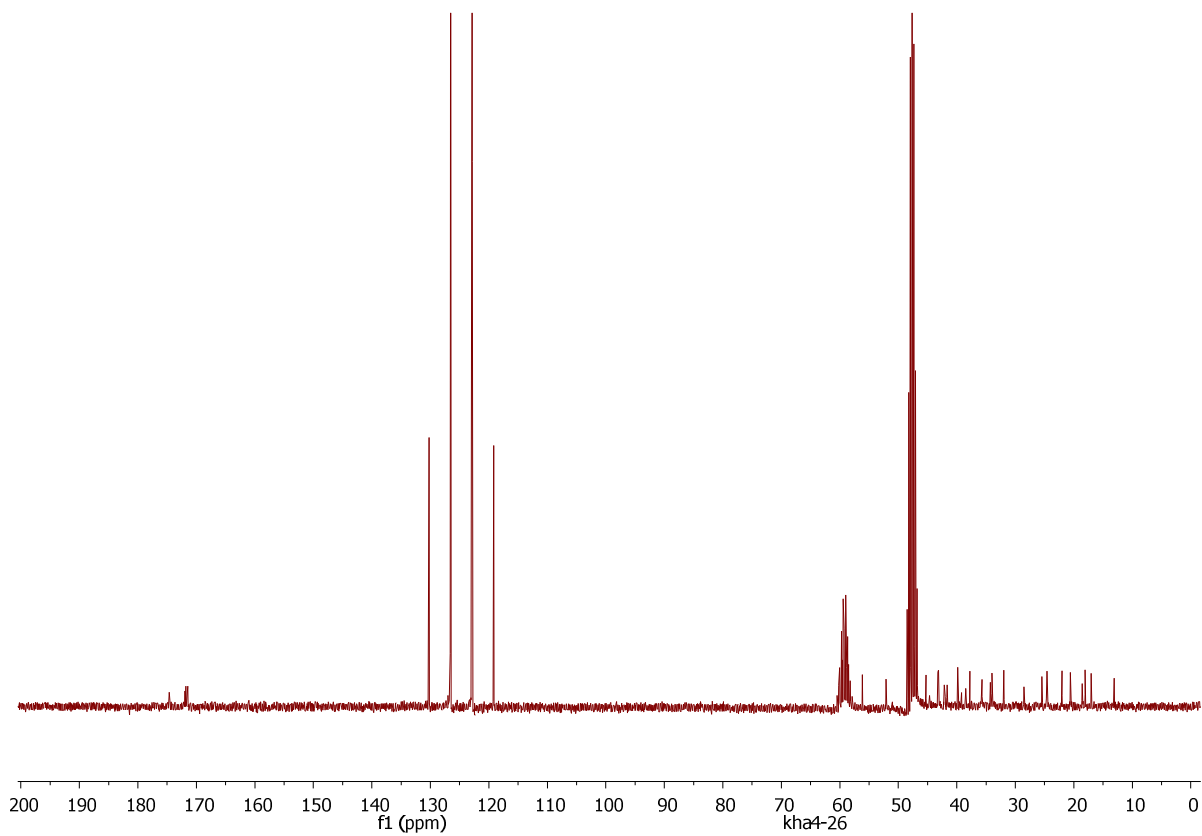
BocB₃Fc



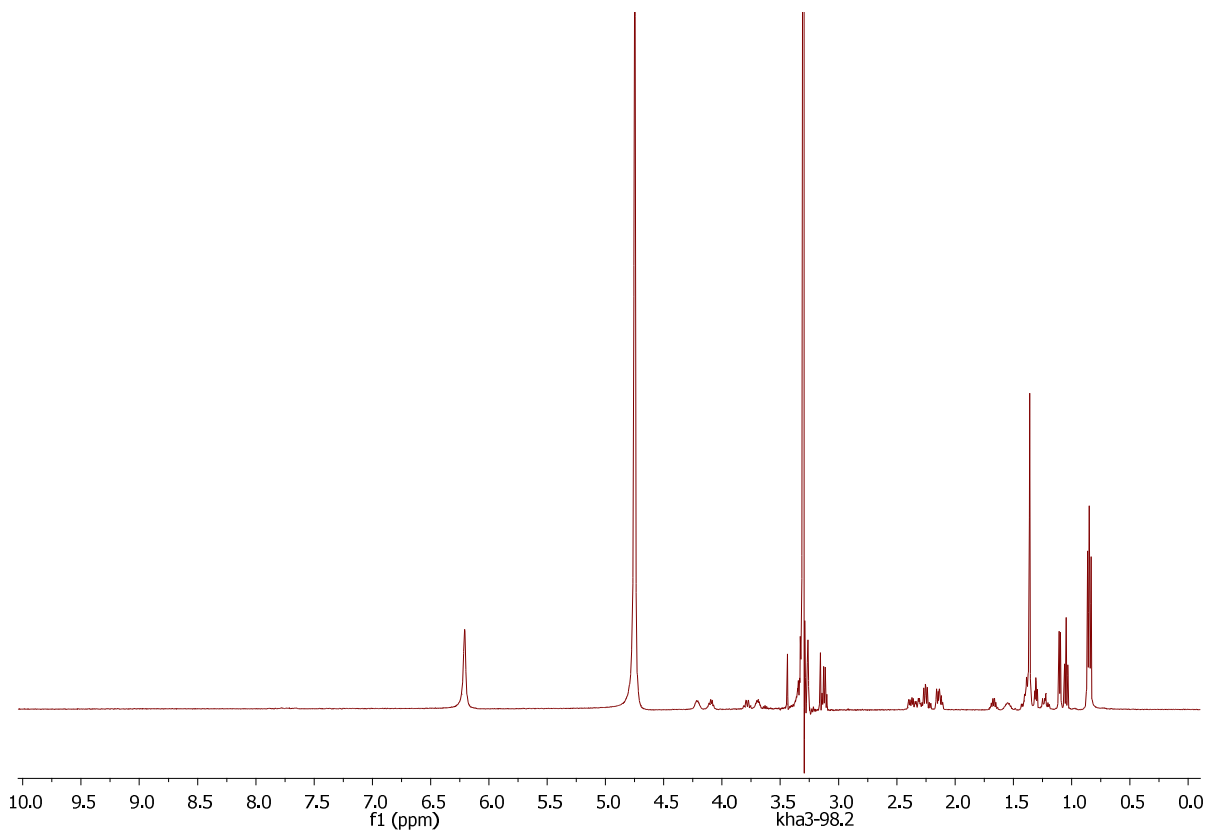


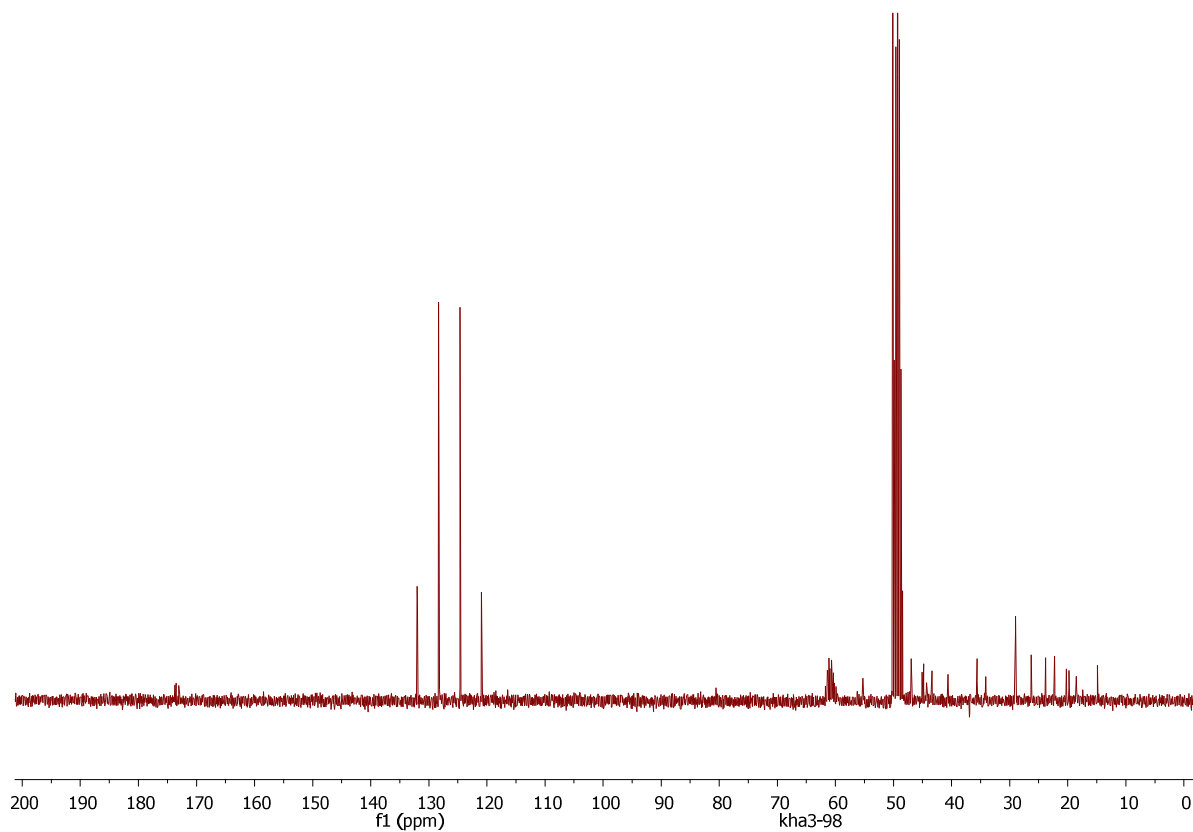
SSB₆Fc



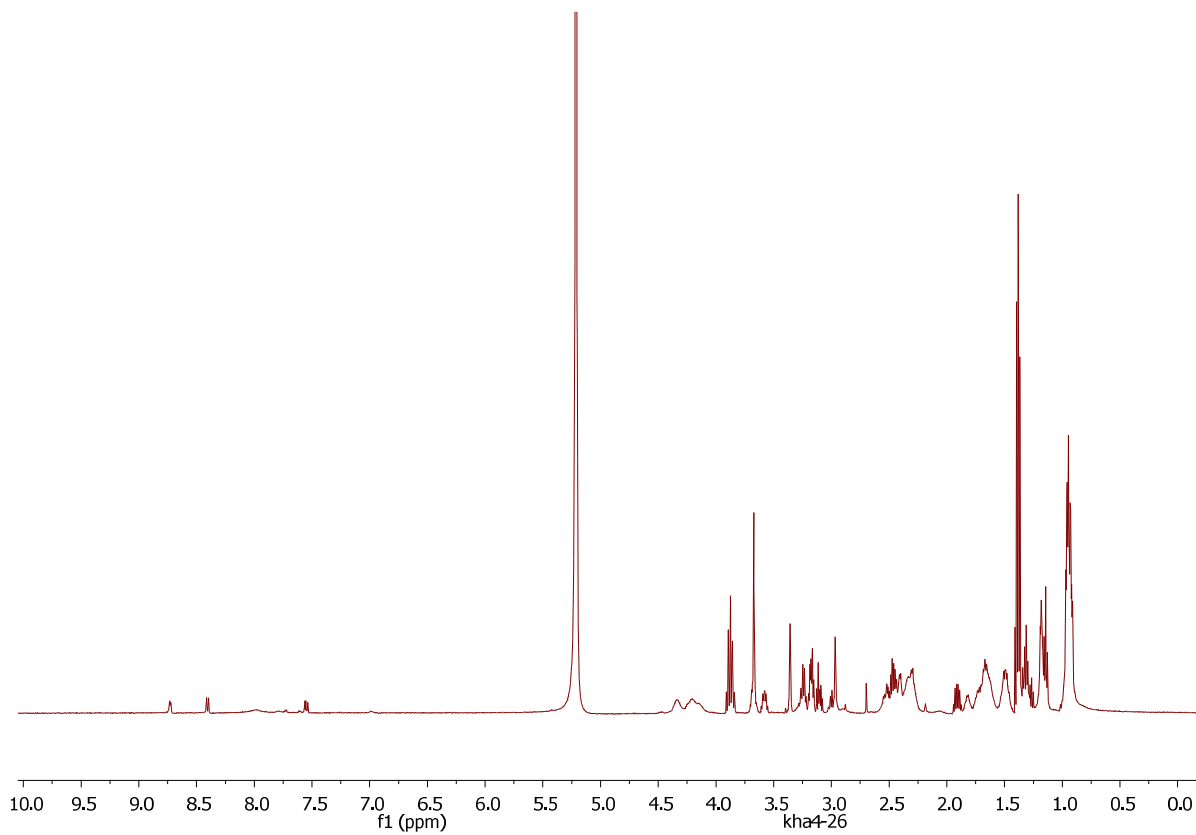


BocB₃Et





SSB₃Et



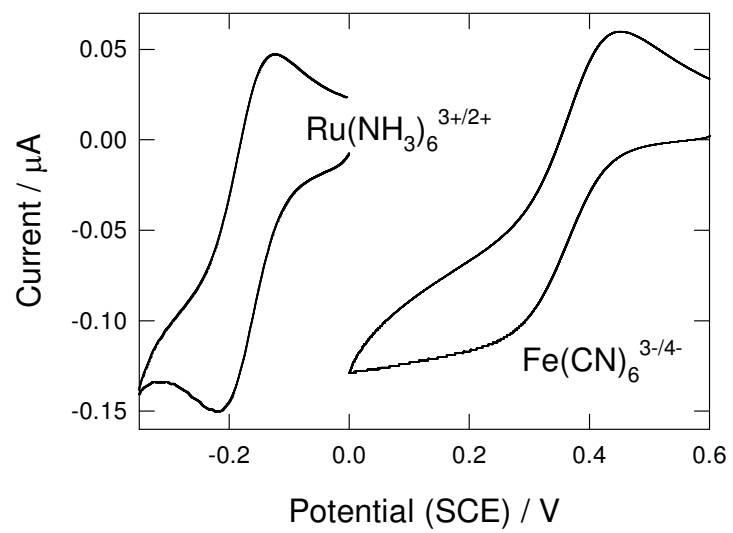
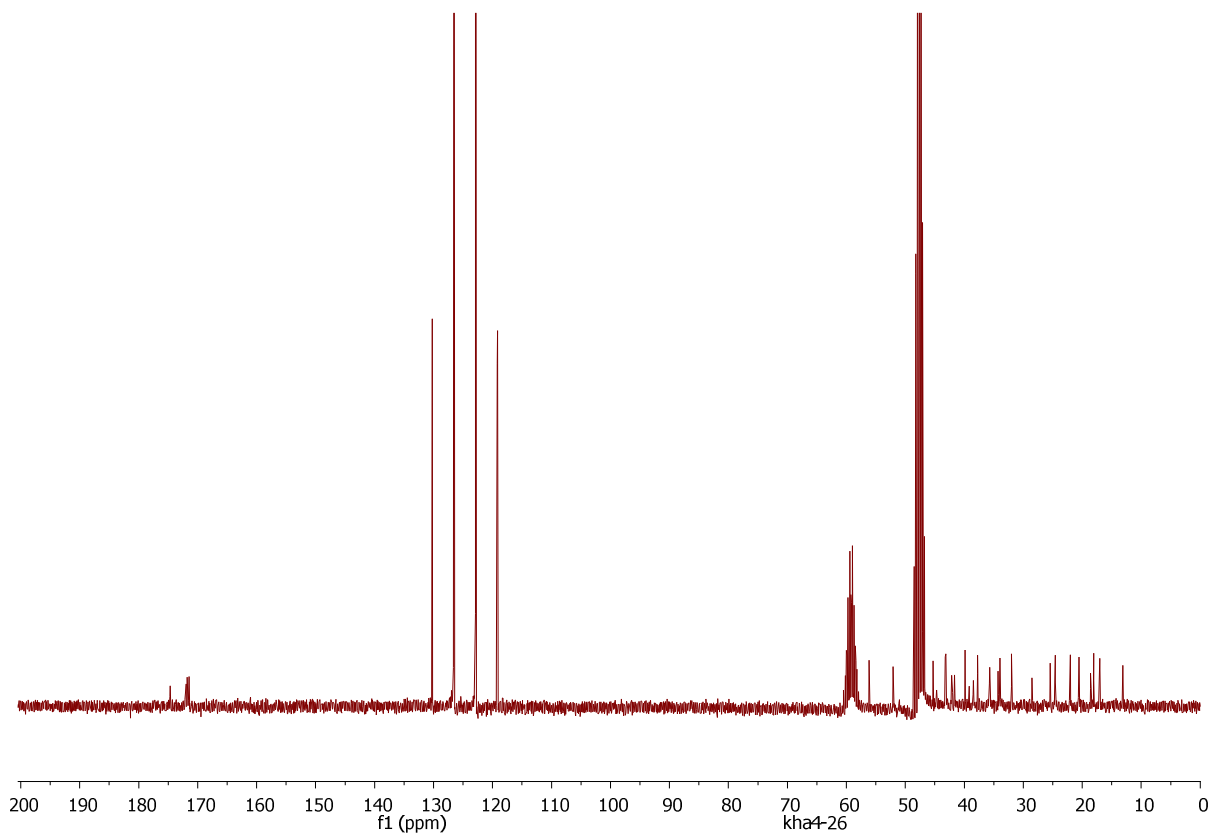


Figure S5 The cyclic voltammograms collected at 80 mV s^{-1} of aqueous $11.7 \text{ mM Fe(CN)}_6^{3-} + 0.1 \text{ M HClO}_4$ and $0.38 \text{ mM Ru(NH}_3)_6^{3+} + 0.1 \text{ M HClO}_4$ solutions at **SSB₆**. The ferricyanide current was scaled by 0.5.

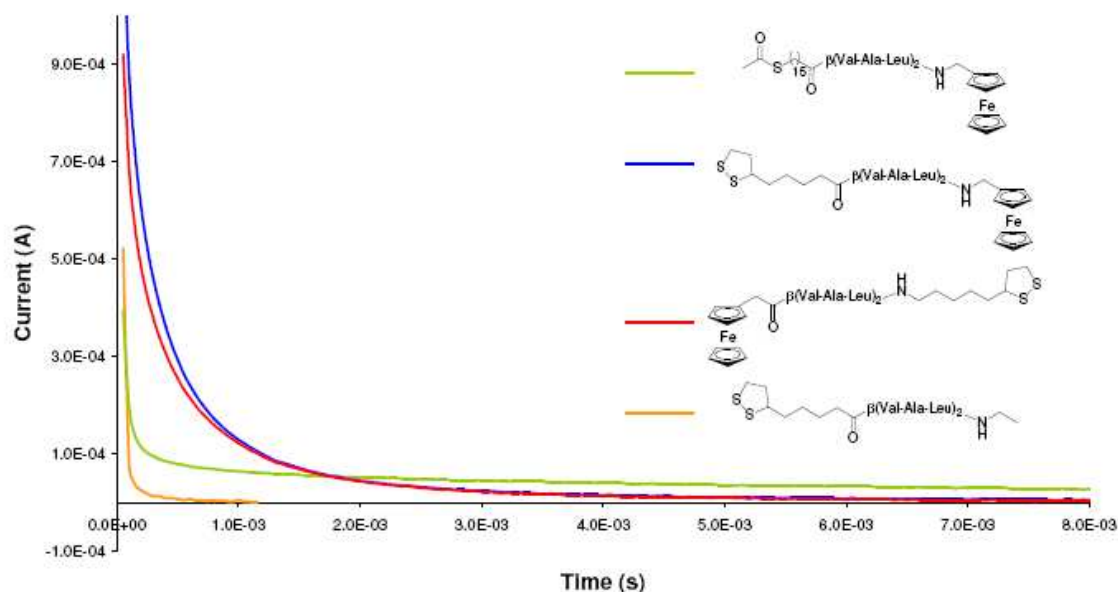


Figure S6 CA graphs of a variety of B_6 analogues, including (blue) SSB_6Fc and (yellow) SSB_6 . Also shown are a reverse dipole (red) SSB_6Fc and a long chain alkanethiol peptide (green) $SC_{15}B_6Fc$ not discussed in this article. The overpotential was 0.05 V (SCE) for each trace. The initial potential was 0.1 V for all species, and the step potential that equates to an overpotential = 0.05 V depends on the value of $E_{1/2}$ for each species.

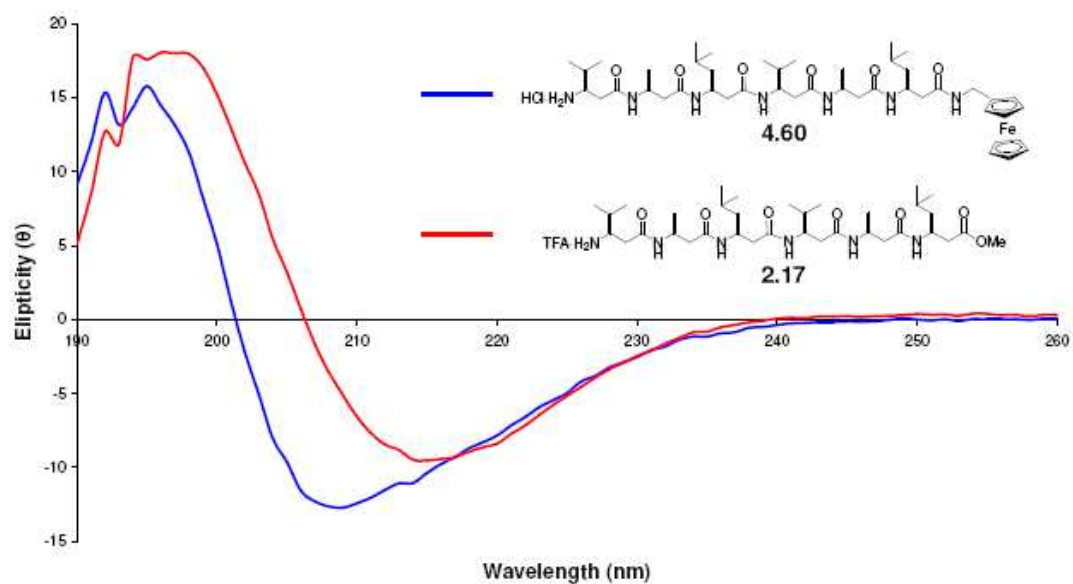


Figure S7 The CD spectra of 0.2 mM **B₆Fc** and **B₆** taken in methanol.