

Supporting information

Predicting Sediment Sorption Coefficients for Linear Alkylbenzene Sulfonate Congeners from Polyacrylate-Water Partition Coefficients at Different Salinities

5 *Ángeles Rico-Rico,^{1,*} Steven T.J. Droge,^{1,2} Joop L.M. Hermens¹*

¹Institute for Risk Assessment Sciences, Utrecht University, Yalelaan 2, 3508 TD, Utrecht, The Netherlands

²Helmholtz Centre for Environmental Research - UFZ, Permoserstr. 15, 04318, Leipzig, Germany

10 *Corresponding author, Phone: +31 30 253 5018; Fax: +31 30 253 5077; E-mail: a.ricorico@gmail.com

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Table S1. Ion Composition of the Artificial GP2 Seawater (SW) and Dutch Standard Water (DSW).

	SW	DSW
	mmol L ⁻¹	mmol L ⁻¹
Na ⁺ *	418	11
Mg ²⁺	46.73	0.30
Ca ²⁺	8.98	1.36
K ⁺	8.92	0.20
Sr ²⁺	0.08	
Cl ⁻	480	2.70
SO ₄ ²⁻	24.80	0.30
HCO ₃ ⁻	2.02	1.40
Br ⁻	0.74	
B ₄ O ₇ ²⁻	0.09	

*10 mM NaN₃ is considered for the calculation of Na⁺ content in SW and DSW solutions.

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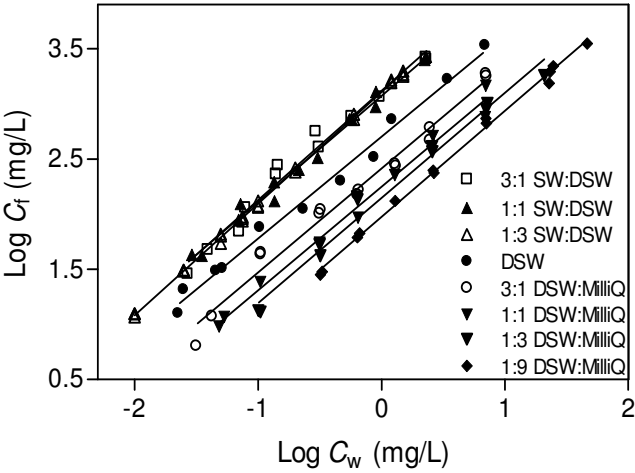
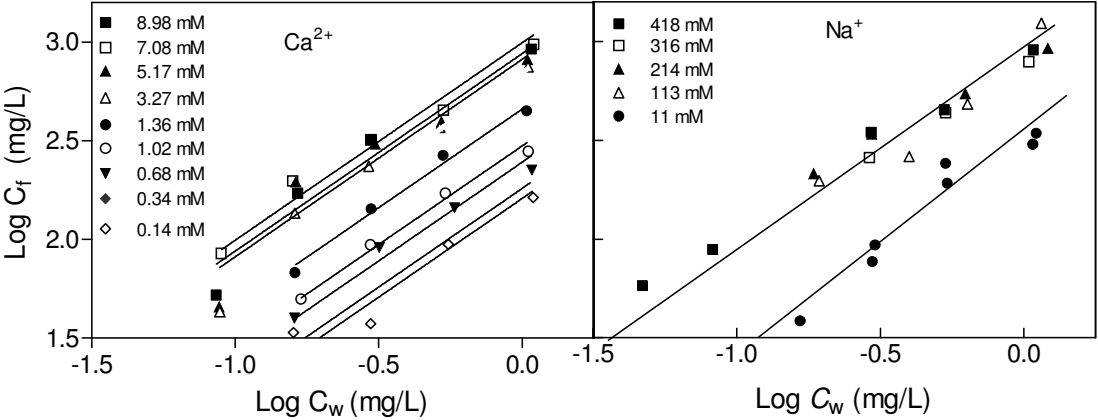


Figure S1. Fiber-water isotherms for C_{12} -2-LAS at different seawater and freshwater dilutions.

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Figure S2. Fiber-water isotherms for C_{12} -2-LAS at different Na^+ and Ca^{2+} concentrations.

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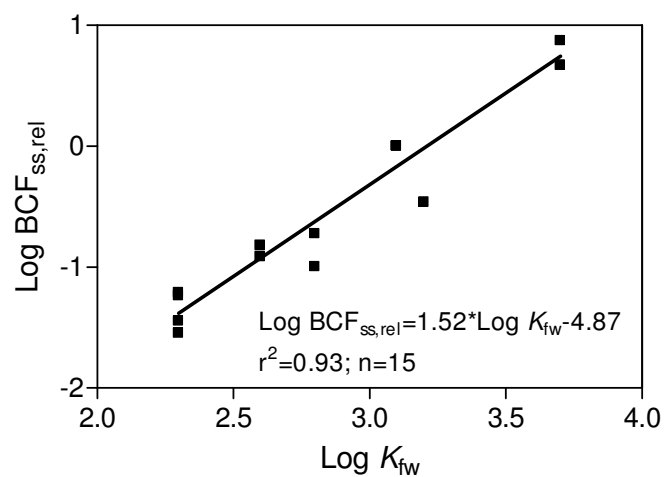
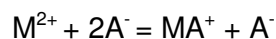


Figure S3. Relative Steady-state bioconcentration values (Reference compound: C₁₂-2-LAS) according to ref. (1) vs estimates of log K_{fw} from ref.(2).

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Section 1. Discussion on sorption of LAS congeners to polyacrylate SPME fibers.

According to Jafvert *et al.* (3) the partitioning of an anionic organic compound to the octanol phase in a single electrolyte solution that contains divalent cations can be mainly described by the following reaction:



Where M^{2+} represents the divalent cation and A^{-} is the organic anion. In this reaction, the anionic organic compound is acting as a counterion to maintain electroneutrality. When fitting the K_{iw} values and the $\log [Ca^{2+}]$ concentration from the simple electrolyte solution experiment to a sigmoidal curve with a fixed slope of 1, an apparent formation constant $\log (K_{Ca(LAS)}^{+})$ of 2.5 is obtained. This value might be considered as an indication of the affinity of the partitioning of LAS at different Ca^{2+} concentration, but we cannot state if this is reflecting the real ion pair formation, since, to our knowledge, no formation constant of this complex is yet reported.

References

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- (3) Jafvert, C. T.; Westall, J. C.; Grieder, E.; Schwarzenbach, R. P. Distribution of hydrophobic ionogenic organic compounds between octanol and water: Organic acids. *Environ. Sci. Technol.* **1990**, *24*, 1795-1803.