## Supporting Information

# Total Synthesis of (+)-Chinensiolide B via Tandem Allylboration/Lactonization 

Tim G. Elford and Dennis G. Hall*

## Table of Contents

1) GENERAL INFORMATION ..... 3
2) EXPERIMENTAL PROCEDURES ..... 3
(Z)-Methyl 6-((tert-butyldiphenylsilyl)oxy)-2-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl) methyl)hex-2-enoate (Z-5) ..... 3
( $2 S, 3 R, 5 S$ )-3-((tert-Butyldimethylsilyl)oxy)-2-chloro-2-methyl-5-(prop-1-en-2-yl) cyclohexanone (SI- 12). ..... 4
(1R,2S,3R,5R)-Methyl 3-((tert-butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2-yl) cyclopentanecarboxylate (7) ..... 4
((1R,2S,3S,5R)-3-((tert-Butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2-yl) cyclopentyl) methanol (SI- 13) ..... 5
(1R,2S,3R,5R)-3-((tert-Butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2-yl) cyclopentanecarbaldehyde (4) ..... 6
(4S,5S)-5-((1'R,2'S,3'R,5'R)-3'-((tert-Butyldimethylsilyl)oxy)-2'-methyl-5'-(prop-1-en-2-yl) cyclopentyl)-4-(3-((tert-butyldiphenylsilyl)oxy)propyl)-3-methylenedihydrofuran-2(3H)-one (3) ..... 6
(4S,5S)-5-((1'R,2'S,3'R,5'R)-3'-((tert-Butyldimethylsilyl)oxy)-2'-methyl-5'-(prop-1-en-2-yl) cyclopentyl)-4-(3-hydroxypropyl)-3-methylenedihydrofuran-2(3H)-one (SI-14) ..... 7
(4S,5S)-4-allyl-5-(( $1 R, 2 S, 3 S, 5 R)$-3-((tert-Butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2- yl)cyclopentyl)-3-methylenedihydrofuran-2(3H)-one (9) ..... 8
(3S,4S,5S)-5-((1'R,2'S,3'S,5’R)-3’-(t-Butyl(dimethyl)siloxy)-2’-methyl-5'-(prop-1-en-2-yl)cyclopentyl)- 5-cyanomethyl-4-(prop-2-en-1-yl)tetrahydrofuran-2-one (8) ..... 8
(3aS,6aR, $8 R, 9 S, 9 \mathrm{a} R, 9 \mathrm{~b} R)-8$-((tert-Butyldimethylsilyl)oxy)-6,9-dimethyl-3-methylene- 3a,4,6a,7,8,9,9a,9b-octahydroazuleno[4,5-b]furan-2(3H)-one (10) ..... 9
(1S,2R,3aR,3bS,4aR,5aS,8aR,8bS)-2-((tert-Butyldimethylsilyl)oxy)-1,3b-dimethyl-6- methylenedecahydrooxireno[2',3':7,8]azuleno[4,5-b]furan-7(2H)-one (2) ..... 10
(3aS, $6 R, 6 \mathrm{a} R, 8 R, 9 S, 9 \mathrm{a} S, 9 \mathrm{~b} R)-8$-((tert-Butyldimethylsilyl)oxy)-6-hydroxy-6,9-dimethyl-3- methylenedecahydroazuleno[4,5-b]furan-2(9bH)-one (11) ..... 11
(3aS,6R,6aR,9S,9aR,9bR)-6-Hydroxy-6,9-dimethyl-3-methyleneoctahydroazuleno[4,5-b]furan- 2,8(3H,9bH)-dione (1) ..... 11
3) COMPARISON OF ${ }^{13}$ C-NMR DATA FOR 1 ..... 13
4) CRYSTALLOGRAPHIC DATA FOR 8 ..... 14
5) COPIES OF NMR SPECTRA ..... 22
6) REFERENCES ..... 36

## 1) General information

Unless otherwise noted, all reactions were performed under an argon atmosphere. Toluene, HMPA, $\mathrm{CH}_{3} \mathrm{CN}$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ were distilled over $\mathrm{CaH}_{2}$. THF was distilled over sodium/ benzophenone ketyl. $\mathrm{NH}_{4} \mathrm{Cl}(\mathrm{aq})$ and $\mathrm{NaHCO}_{3}(\mathrm{aq})$ refer to saturated aqueous solutions. All other chemicals were used as received from commercial sources. Thin layer chromatography (TLC) was performed on Macherey-Nagel Polygram Sil G/UV 254 plates and was visualized with UV light, potassium permanganate stain or Seebach's stain. NMR spectra were recorded on 300, 400, 500 or 600 MHz instruments as indicated in procedures. The residual solvent protons $\left({ }^{1} \mathrm{H}\right)$ or the solvent carbons $\left({ }^{13} \mathrm{C}\right)$ were used as internal standards. ${ }^{1} \mathrm{H}$ NMR data are presented as follows: chemical shift in ppm downfield from tetramethylsilane (multiplicity, integration, coupling constant). The following abbreviations are used in reporting NMR data: br, broad; app, apparent; s , singlet; d, doublet; t, triplet; q, quartet; dd, doublet of doublets; ddd, doublet of doublet of doublets; dddd, doublet of doublet of doublet of doublets; $p$, pentet; qt, quartet of triplets; $m$, multiplet. High-resolution mass spectra were recorded by the University of Alberta Mass Spectrometry Services Laboratory using either electron impact (EI) or electrospray (ES) ionization techniques. Infrared spectra and optical rotations were recorded by the University of Alberta Spectral Services Laboratory.

## Full authorship for reference 3:

Zhang, S.; Zhao, M.; Bai, L.; Hasegawa, T.; Wang, J.; Wang, L.; Xue, H.; Deng, Q.; Xing, F.; Bai, Y.; Sakai, J.; Bai, J.; Koyanagi, R.; Tsukumo, Y.; Kataoka, T.; Nagai, K.; Hirose, K.; Ando, M. J. Nat. Prod. 2006, 69, 1425-1428.

## 2) Experimental procedures


(Z)-Methyl 6-((tert-butyldiphenylsilyl)oxy)-2-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl) methyl)hex-2-enoate (Z-5)

To a solution of HMPA ( 10.0 ml , 6.0 equiv) in toluene ( 100 ml ) under argon at $0{ }^{\circ} \mathrm{C}$ was added DIBALH ( $10.0 \mathrm{ml}, 1.5 \mathrm{M}$ solution in toluene, 1.5 equiv). This mixture was left to stir for 1 h , at which point, known propiolic ester derivative ${ }^{1}(3.81 \mathrm{~g}, 1.0$ equiv) was added. The mixture was stirred for 5 h at $0{ }^{\circ} \mathrm{C}$, then freshly distilled chloromethylpinacolboronate ${ }^{2}$ was added. The reaction mixture was allowed to warm to room temperature and stirred overnight. The reaction was quenched with 1 M HCl and extracted with ether ( x 4 ). The combined organics were washed with $1 \mathrm{M} \mathrm{HCl}(x 3), \mathrm{NaHCO}_{3}(\mathrm{aq})(\mathrm{x} 2)$, water and brine. The organics were then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuo. The crude product was then purified by flash chromatography ( $3 \%$ ethyl acetate in dichloromethane) to provide allylboronate 5 as a mixture of alkene isomers ( $3.92 \mathrm{~g}, 68 \%$ yield). The major isomer was the desired one and was often obtained in a ratio $>3.5: 1$. The mixture of isomers was used for the next step as separation on
large scale was difficult and the minor isomer does not react in the next step of the synthesis. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.70-7.63(\mathrm{~m}, 4 \mathrm{H}), 7.45-7.33(\mathrm{~m}, 6 \mathrm{H}), 5.92(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.69$ $(\mathrm{s}, 3 \mathrm{H}), 3.68(\mathrm{t}, 2 \mathrm{H}, J=6.5 \mathrm{~Hz}), 2.60(\mathrm{q}, 2 \mathrm{H}, J=7.8 \mathrm{~Hz}), 1.82(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.68(\operatorname{app} \mathrm{p}, 2 \mathrm{H}, J=$ $7.3 \mathrm{~Hz}), 1.23(\mathrm{~s}, 12 \mathrm{H}), 1.05(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 168.3,142.9,135.6,134.1$, $129.5,128.1,127.6,83.3,63.6,51.2,32.5,26.9,26.4,24.8,19.3 .{ }^{11} \mathrm{~B}-\mathrm{NMR}\left(160 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta$ 33.1. IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right): 3071,2932,1722$. HRMS (EI, m/z) Calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{5} \mathrm{Si}^{35} \mathrm{~B}^{11}\left[\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9}\right]^{+}: 465.22687$. Found: 465.22809.

(2S,3R,5S)-3-((tert-Butyldimethylsilyl)oxy)-2-chloro-2-methyl-5-(prop-1-en-2-yl) cyclohexanone (SI-12)

Known alcohol $\mathbf{6}^{3}$ ( $127.8 \mathrm{mmol}, 1.00$ equiv) was dissolved in 150 ml acetonitrile and cooled to 0 ${ }^{\circ} \mathrm{C}$ under argon. To this stirred solution was added pyridine ( $12.4 \mathrm{ml}, 1.20$ equiv) followed by tert-butyldimethylsilyltrifluoromethane sulfonate ( $32.3 \mathrm{ml}, 1.10$ equiv). The reaction solution was then warmed to room temperature and allowed to stir overnight. The reaction was then quenched by the addition of $\mathrm{NH}_{4} \mathrm{Cl}(\mathrm{aq})$ and the mixture was extracted with ethyl acetate ( x 3 ). The organic layers were combined, washed with water and brine, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. Crude product SI-12 was obtained as a colorless oil and was pure enough by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ to be used without further purification $\left(39.5 \mathrm{~g}, 97 \%\right.$ yield). $[\alpha]_{\mathrm{D}}=+60.5(c 0.34$, $\mathrm{CHCl}_{3}$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 4.80(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.76(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.19(\mathrm{dd}, 1 \mathrm{H}, J=3.5,1.9$ $\mathrm{Hz}), 3.04(\mathrm{dd}, 1 \mathrm{H}, J=14.0,14.0 \mathrm{~Hz}), 2.83(\mathrm{app} \mathrm{tt}, 1 \mathrm{H}, J=12.5,3.4 \mathrm{~Hz}), 2.42(\mathrm{ddd}, 1 \mathrm{H}, J=14.3$, $12.6,2.0 \mathrm{~Hz}$ ), $2.35(\mathrm{ddd}, 1 \mathrm{H}, 13.5,3.7,2.2 \mathrm{~Hz}$ ), 1.84-1.73 (m, 1H), $1.75(\mathrm{~s}, 3 \mathrm{H}), 1.60(\mathrm{~s}, 3 \mathrm{H})$, $0.87(\mathrm{~s}, 9 \mathrm{H}), 0.09(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 204.5,146.7,110.4,77.9,68.2,41.1$, 39.2, 33.7, 25.7, 23.0, 20.4, 18.0, -4.5, -4.9. IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right): 2957,2936,1729$. HRMS (EI, m/z) Calcd for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{2} \mathrm{Si}^{35} \mathrm{Cl}\left[\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9}\right]^{+}: 259.09210$. Found: 259.09199.

(1R,2S,3R,5R)-Methyl 3-((tert-butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2-yl) cyclopentanecarboxylate (7)

Sodium hydride ( $3.23 \mathrm{~g}, 1.50$ equiv, $60 \mathrm{wt} \%$ in oil) was washed with pentane and then suspended in 60 ml dry methanol at $0^{\circ} \mathrm{C}$ under argon. Crude TBS-protected alcohol SI-12 (53.9 mmol) was dissolved in dry ether at $0^{\circ} \mathrm{C}$ under argon, and to this solution was added the sodium methoxide/methanol solution by cannula. The reaction was left to stir at $0^{\circ} \mathrm{C}$ for 1.5 h , then quenched with $30 \mathrm{ml} \mathrm{NH}_{4} \mathrm{Cl}(\mathrm{aq})$ and 10 ml water. The reaction was warmed to room temperature and stirred vigorously for 1 h . The mixture was then extracted with ether (x3) and the combined organics were washed with brine. The organics were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. THF ( 100 ml ) was added to help removed methanol during concentration. Crude product 7 was obtained as an orange oil and was essentially pure by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and could be used without further purification $(14.34 \mathrm{~g}, 85 \%$ yield $) .[\alpha]_{\mathrm{D}}=-20.5\left(c 1.00, \mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 4.78$ (br s, 1 H ), 4.70 (br s, 1 H ), 4.17 (dd, $1 \mathrm{H}, J=3.9,3.9 \mathrm{~Hz}$ ), 3.60 (s, 3H), 3.20 (ddd, $1 \mathrm{H}, J=10.5,10.5,7.0 \mathrm{~Hz}), 2.82(\mathrm{dd}, 1 \mathrm{H}, J=10.1,9.3 \mathrm{~Hz}), 2.48-2.34(\mathrm{~m}, 1 \mathrm{H}), 2.01-$ $1.90(\mathrm{~m}, 1 \mathrm{H}), 1.78-1.68(\mathrm{~m}, 1 \mathrm{H}), 1.73(\mathrm{~s}, 3 \mathrm{H}), 0.98(\mathrm{~d}, 3 \mathrm{H}, J=7.3 \mathrm{~Hz}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}$, $6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 175.2,145.5,111.2,75.5,53.3,51.1,46.5,43.2,40.1,25.8$, $22.5,18.1,14.3,-4.7,-5.0$. IR ( $\mathrm{CDCl}_{3}$, cast film, $\mathrm{cm}^{-1}$ ): 2956, 2931, 1738. HRMS (ESI, m/z) Calcd for $\mathrm{C}_{17} \mathrm{H}_{33} \mathrm{O}_{3} \mathrm{Si}^{35}[\mathrm{M}+\mathrm{H}]^{+}: 313.21935$. Found: 313.21904.

((1R,2S,3S,5R)-3-((tert-Butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2-yl) cyclopentyl) methanol (SI-13)

Crude ester $7(45.9 \mathrm{mmol})$ was dissolved in 150 ml THF and cooled to $0{ }^{\circ} \mathrm{C}$ under argon. $\mathrm{LiAlH}_{4}$ ( $1.82 \mathrm{~g}, 1.00$ equiv) was added slowly to the reaction mixture. The reaction was stirred at $0^{\circ} \mathrm{C}$ for 1 h , then warmed to room temperature and stirred for an additional 1 h . The reaction was quenched by slowly adding water and ethyl acetate until all the bubbling stopped. The reaction mixture was then filtered through a pad of celite and concentrated in vacuo. THF was used to help coevaporate the water in the mixture. This provided the crude product as a white solid that was pure enough by ${ }^{1} \mathrm{H}$-NMR to be used without further purification $\left(10.11 \mathrm{~g}, 77 \%\right.$ yield). $[\alpha]_{\mathrm{D}}$ $=-17.5\left(c 1.00, \mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 4.90(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.81(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.12$ (dd, $1 \mathrm{H}, J=3.6,3.6 \mathrm{~Hz}$ ), $3.56(\mathrm{dd}, 1 \mathrm{H}, J=11.6,4.8), 3.49(\mathrm{dd}, 1 \mathrm{H}, J=11.6,5.9 \mathrm{~Hz}), 3.09-2.98$ $(\mathrm{m}, 1 \mathrm{H}), 2.02-1.84(\mathrm{~m}, 2 \mathrm{H}), 1.86(\mathrm{~s}, 3 \mathrm{H}), 1.80(\mathrm{dd}, 1 \mathrm{H}, J=11.9,3.6 \mathrm{~Hz}), 1.66(\mathrm{ddd}, 1 \mathrm{H}, J=$ $12.7,6.7,1.3 \mathrm{~Hz}), 1.00(\mathrm{~d}, 3 \mathrm{H}, J=7.0 \mathrm{~Hz}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 125 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 147.9,110.4,75.0,63.9,48.3,44.8,42.1,39.7,25.8,24.0,18.2,14.6,-4.6,-5.0$ IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right): 3353,2957,2929$. HRMS (EI, m/z) Calcd for $\mathrm{C}_{16} \mathrm{H}_{32} \mathrm{O}_{2} \mathrm{Si}: 284.21716$. Found: 284.21731.

(1R,2S,3R,5R)-3-((tert-Butyldimethylsilyl)oxy)-2-methyl-5-(prop-1-en-2-yl) cyclopentanecarbaldehyde (4)
Oxalyl chloride ( 2.0 ml , 1.3 equiv) was dissolved in 30 ml DCM under argon at $-78^{\circ} \mathrm{C}$. To this stirred solution was slowly added a solution of DMSO ( $2.75 \mathrm{ml}, 2.2$ equiv) in 3 ml DCM. The reaction mixture was stirred for 15 minutes, then a solution of crude alcohol SI-13 ( 17.6 mmol ) dissolved in 10 ml DCM was added. The reaction mixture was stirred for 15 minutes, then triethylamine ( $12.0 \mathrm{ml}, 5.0$ equiv) was added. The reaction was then allowed to warm to room temperature over 1 h . The reaction was quenched with water and extracted with DCM (x3). The organic layers were combined, washed with brine, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. The crude mixture was purified via flash chromatography ( $5 \%$ ethyl acetate in hexanes) to provide 4 as a colorless oil ( $4.89 \mathrm{~g}, 99 \%$ yield). There was a trace amount of a contaminant in the final product that could not be removed despite several attempts. This contaminant was not characterized and did not affect subsequent reactions. $[\alpha]_{\mathrm{D}}=-21.6$ (c 1.00, $\left.\mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 9.53(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 4.86(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.83(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 4.19 (dd, $1 \mathrm{H}, J=3.8,3.8 \mathrm{~Hz}$ ), 3.31 (ddd, $1 \mathrm{H}, J=10.7,10.7,6.9$ ), 2.62 (ddd, $1 \mathrm{H}, J=10.7,8.6$, 3.9 Hz ), 2.32-2.25 (m, 1H), 1.87 (ddd, $1 \mathrm{H}, J=12.8,12.8,3.4 \mathrm{~Hz}$ ), 1.80 (ddd, $1 \mathrm{H}, J=12.8,6.8$, $1.0 \mathrm{~Hz}), 1.73(\mathrm{~s}, 3 \mathrm{H}), 0.99(\mathrm{~d}, 3 \mathrm{H}, J=6.9 \mathrm{~Hz}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 204.2,144.1,111.8,75.6,58.1,46.0,40.6,40.0,25.9,22.9,18.2,14.1,-4.6$. IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right): 2957,2930,1721$. HRMS (EI, m/z) Calcd for $\mathrm{C}_{16} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{Si}$ : 282.20151 . Found: 282.20131.

(4S,5S)-5-(( $\left.1^{\prime} R, 2^{\prime} S, 3^{\prime} R, 5^{\prime} R\right)-3^{\prime}-(($ tert-Butyldimethylsilyl)oxy)-2'-methyl-5'-(prop-1-en-2-yl) cyclopentyl)-4-(3-((tert-butyldiphenylsilyl)oxy)propyl)-3-methylenedihydrofuran-2(3H)-one (3)

Allylboronate 5 ( $6.2 \mathrm{mmol}, 1.0$ equiv) and aldehyde 4 ( $6.2 \mathrm{mmol}, 1.0$ equiv) were dissolved in toluene under argon and cooled to $0^{\circ} \mathrm{C}$. To this stirred solution was added $\mathrm{BF}_{3} \bullet \mathrm{Et}_{2} \mathrm{O}(20 \mu$, 0.025 equiv) and the reaction was stirred for 48 h . Crude ${ }^{1} \mathrm{H}-\mathrm{NMR}$ at this point showed complete consumption of $Z-5$, a significant amount of $E-5$ still remaining, and a small amount of unreacted aldehyde 4. The reaction was quenched with a solution of $9: 1 \mathrm{NH}_{4} \mathrm{Cl}(\mathrm{aq}): \mathrm{NH}_{4} \mathrm{OH}$ and extracted with ethyl acetate (x3). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuo. The crude product was purified by flash chromatography ( $5 \%$ ethyl acetate in hexanes) to
provide 3 as a colorless oil ( $3.49 \mathrm{~g}, 87 \%$ yield). $[\alpha]_{\mathrm{D}}=+1.3\left(c 1.00, \mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H}$ NMR (300 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.69-7.62(\mathrm{~m}, 4 \mathrm{H}), 7.47-7.34(\mathrm{~m}, 6 \mathrm{H}), 6.22(\mathrm{~d}, 1 \mathrm{H}, J=2.6 \mathrm{~Hz}), 5.51(\mathrm{~d}, 1 \mathrm{H}, J=$ 2.4 Hz ), $4.92(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.85(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.19-4.12(\mathrm{~m}, 2 \mathrm{H}), 3.75-3.61(\mathrm{~m}, 2 \mathrm{H}), 3.04(\mathrm{ddd}, 1 \mathrm{H}, J$ $=13.7,8.9,6.0 \mathrm{~Hz}), 2.77-2.67(\mathrm{~m}, 1 \mathrm{H}), 2.07-1.88(\mathrm{~m}, 3 \mathrm{H}), 1.77(\mathrm{~s}, 3 \mathrm{H}), 1.73-1.54(\mathrm{~m}, 5 \mathrm{H}), 1.06$ $(\mathrm{s}, 9 \mathrm{H}), 1.01(\mathrm{~d}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.04(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ $171.0,144.3,139.6,135.8,134.0,129.9,127.9,121.7,112.8,83.6,74.8,63.6,50.4,46.2,42.8$, $40.0,38.7,30.0,29.6,27.1,26.1,23.7,19.5,18.4,17.4,-4.4,-4.8$. IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right)$ : 3072, 2956, 2931, 1767. HRMS (ESI, m/z) Calcd for $\mathrm{C}_{39} \mathrm{H}_{58} \mathrm{O}_{4} \mathrm{Si}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}: 669.37658$. Found: 669.37507.

(4S,5S)-5-(( $\left.1^{\prime} R, 2^{\prime} S, 3^{\prime} R, 5^{\prime} R\right)-3^{\prime}-(($ tert-Butyldimethylsilyl)oxy)-2'-methyl-5'-(prop-1-en-2-yl) cyclopentyl)-4-(3-hydroxypropyl)-3-methylenedihydrofuran-2(3H)-one (SI-14)

A solution of tetrabutylammonium fluoride ( $2.6 \mathrm{ml}, 1.0 \mathrm{M}$ solution in THF, 1.5 equiv) and glacial acetic acid ( $146 \mu \mathrm{l}, 1.50$ equiv) was stirred in 10 ml DMF for 30 minutes under argon. This mixture was then cannulated into a solution of $\mathbf{3}$ in 10 ml DMF at room temperature. The reaction mixture was stirred and monitored by TLC ( $20 \%$ ethyl acetate in hexanes) until consumption of $\mathbf{3}$ was complete (typically 3 to 4 hours). The reaction was quenched with water and extracted with ether (x3). The organics were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. The crude mixture was purified by flash chromatography ( 20 to $30 \%$ ethyl acetate in hexanes) to provide alcohol SI-14 as a colorless oil $(487 \mathrm{mg}, 70 \%$ yield $) .[\alpha]_{\mathrm{D}}=+14.2\left(c 0.84, \mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 6.24(\mathrm{~d}, 1 \mathrm{H}$, $J=3.1 \mathrm{~Hz}), 5.58(\mathrm{~d}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 4.93(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.85(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.19(\mathrm{dd}, 1 \mathrm{H}, J=5.3,2.8$ Hz ), 4.14 (dd, $1 \mathrm{H}, J=4.8,4.8 \mathrm{~Hz}$ ), $3.70-3.65(\mathrm{~m}, 2 \mathrm{H}), 3.05(\mathrm{ddd}, 1 \mathrm{H}, J=12.7,9.0,6.1 \mathrm{~Hz}$ ), 2.80-2.73 (m, 1H), 2.05-1.91 (m, 3H), $1.79(\mathrm{~s}, 3 \mathrm{H}), 1.70-1.51(\mathrm{~m}, 6 \mathrm{H}), 1.03(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz})$, $0.88(\mathrm{~s}, 9 \mathrm{H}), 0.03(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 170.7,144.1,139.2,121.7,112.6$, $83.4,74.5,62.4,50.3,45.9,42.6,39.8,38.5,29.8,29.3,25.8,23.4,18.1,17.1,-4.7,-5.0$. IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right): 3433,2956,2930$, 1764. HRMS (ESI, m/z) Calcd for $\mathrm{C}_{23} \mathrm{H}_{44} \mathrm{O}_{4} \mathrm{SiNa}$ $[\mathrm{M}+\mathrm{Na}]^{+}: 432.25881$. Found: 431.25961.

(4S,5S)-4-allyl-5-(( $\left.1^{\prime} R, 2^{\prime} S, 3^{\prime} S, 5^{\prime} R\right)-3^{\prime}-(($ tert-Butyldimethylsilyl)oxy)-2'-methyl-5'-(prop-1-en-2-yl)cyclopentyl)-3-methylenedihydrofuran-2(3H)-one (9)

Alcohol SI-14 ( $1.05 \mathrm{~g}, 2.57 \mathrm{mmol}$ ) was dissolved in 24 ml of THF under argon at room temperature. To this was added 2-nitrophenylselenium cyanate ( $583 \mathrm{mg}, 2.57 \mathrm{mmol}$ ) and the dark red reaction mixture was cooled down to $0^{\circ} \mathrm{C}$. Once cooled, tributylphosphine ( 0.63 ml , 2.57 mmol ) was added to create a black solution, which was stirred at $0^{\circ} \mathrm{C}$ for 3 h and eventually became a orange/brown, transparent solution. The reaction was quenched with $\mathrm{NH}_{4} \mathrm{Cl}(\mathrm{aq})$ and extracted with ethyl acetate (x3). The organics were then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The crude mixture was the redissolved in 24 ml of THF, put under argon and cooled to $0{ }^{\circ} \mathrm{C}$. Hydrogen peroxide ( $2.6 \mathrm{ml}, 1.0 \mathrm{ml} / \mathrm{mmol}$ substrate) was added and the mixture was allowed to warm to room temperature and stirred for 2 h . The reaction was quenched with saturated $\mathrm{NaHSO}_{3}$ and extracted with ethyl acetate (x3). The organics were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash chromatography (3-5\% ethyl acetate:hexanes) provided 9 as a pale yellow oil ( $602 \mathrm{mg}, 60 \%$ yield). $[\alpha]_{\mathrm{D}}=-1.1$ (c 0.07, $\mathrm{CHCl}_{3}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.25(\mathrm{~d}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}$ ), 5.74 (dddd, $1 \mathrm{H}, J=16.8,10.8$, $7.3,7.3), 5.59(\mathrm{~d}, 1 \mathrm{H}, J=2.6 \mathrm{~Hz}), 5.15(\mathrm{~m}, 1 \mathrm{H}), 5.12(\mathrm{~m}, 1 \mathrm{H}), 4.92(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.85(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, $4.19(\mathrm{dd}, 1 \mathrm{H}, J=6.2,2.7 \mathrm{~Hz}), 4.15(\mathrm{dd}, 1 \mathrm{H}, J=4.6,4.6 \mathrm{~Hz}), 3.02(\mathrm{ddd}, 1 \mathrm{H}, J=13.8,9.0,5.9$ $\mathrm{Hz}), 2.85-2.79(\mathrm{~m}, 1 \mathrm{H}), 2.41-2.34(\mathrm{~m}, 1 \mathrm{H}), 2.34-2.26(\mathrm{~m}, 1 \mathrm{H}), 2.06-1.93(\mathrm{~m}, 3 \mathrm{H}), 1.76(\mathrm{~s}, 3 \mathrm{H})$, $1.59(\mathrm{dd}, 1 \mathrm{H}, J=12.7,6.0 \mathrm{~Hz}), 1.03(\mathrm{~d}, 3 \mathrm{H}, J=6.9 \mathrm{~Hz}), 0.88(\mathrm{~s}, 9 \mathrm{H}), 0.03(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 170.6,144.0,138.7,134.0,121.7,118.3,112.3,82.8,74.5,49.6,45.9$, $42.0,39.8,38.3,37.5,25.9,23.4,18.2,17.2,-4.7,-5.0$. IR $\left(\mathrm{CDCl}_{3}\right.$, cast film, $\left.\mathrm{cm}^{-1}\right): 2957,2930$, 1767. HRMS (EI, m/z) Calcd for $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{O}_{3} \mathrm{Si}\left[\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9}\right]^{+}: 333.18860$. Found: 333.18994.

$(3 S, 4 S, 5 S)-5-\left(\left(1^{\prime} R, 2^{\prime} S, 3^{\prime} S, 5^{\prime} R\right)-3^{\prime}-(t-B u t y l(d i m e t h y l) s i l o x y)-2^{\prime}-m e t h y l-5 '-(p r o p-1-e n-2-\right.$ yl)cyclopentyl)-5-cyanomethyl-4-(prop-2-en-1-yl)tetrahydrofuran-2-one (8)

Obtained during the purification of $\mathbf{9}$ in an amount of 325 mg ( $30 \%$ yield) and was recrystallized from diethyl ether. ${ }^{1}$ H NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 5.83-5.73(\mathrm{~m}, 1 \mathrm{H}), 5.24-5.21(\mathrm{~m}, 1 \mathrm{H}), 5.20$ (br s, 1H), 4.91 (br s, 1H), 4.82 (br s, 1H), 3.10-3.03 (m, 1H), 4.17 (br dd, 1H, $J=4.9,4.9 \mathrm{~Hz}$ ), $4.14(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, J=9.1 \mathrm{~Hz}), 3.10-3.03(\mathrm{~m}, 1 \mathrm{H}), 2.81(\mathrm{dd}, 1 \mathrm{H}, J=17.2,5.4 \mathrm{~Hz}), 2.76(\mathrm{dd}, 1 \mathrm{H}, J=$
$17.1,5.0 \mathrm{~Hz}$ ), 2.53 (ddd, $1 \mathrm{H}, J=11.0,5.1,5.1 \mathrm{~Hz}), 2.46-2.29(\mathrm{~m}, 3 \mathrm{H}), 2.13-2.03(\mathrm{~m}, 2 \mathrm{H}), 1.97$ (ddd, $1 \mathrm{H}, J=13.2,13.2,4.7 \mathrm{~Hz}), 1.75(\mathrm{~s}, 3 \mathrm{H}), 1.58(\mathrm{dd}, 1 \mathrm{H}, J=12.5,5.4 \mathrm{~Hz}), 1.09(\mathrm{~d}, 3 \mathrm{H}, J=$ $7.1 \mathrm{~Hz}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 3 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 174.9,143.8$, $133.3,119.5,116.7,112.3,82.8,74.4,46.9,45.9,42.9,41.7,39.6,37.8,34.5,25.9,23.5,18.2$, 17.7, 17.3, -4.6, -4.9. IR ( $\mathrm{CDCl}_{3}$ cast film, $\left.\mathrm{cm}^{-1}\right)$ : 3083, 2957, 2930, 2857, 2251, 1776. HRMS (EI, m/z) Calcd for $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{NO}_{3} \mathrm{Si}\left[\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9}\right]^{+}: 360.19949$. Found: 360.19935 . This product was also analyzed by X-ray analysis. See Section 4 for crystallographic data.

(3aS,6aR, $8 R, 9 S, 9 \mathrm{a}, 9 \mathrm{bR})$-8-((tert-Butyldimethylsilyl)oxy)-6,9-dimethyl-3-methylene-3a,4,6a,7,8,9,9a,9b-octahydroazuleno[4,5-b]furan-2(3H)-one (10)
$\gamma$-Lactone 9 ( $300 \mathrm{mg}, 0.77 \mathrm{mmol}, 1.0$ equiv) was dissolved in 150 ml DCM under argon. Grubbs second generation catalyst ( $32 \mathrm{mg}, 0.038 \mathrm{mmol}, 0.05$ equiv) was added and the reaction was heated to $40^{\circ} \mathrm{C}$ for 12 h . The solvent was removed and the crude mixture was purified by flash chromatography ( $10 \%$ ethyl acetate:hexanes) to provide 10 as a light beige solid ( $259 \mathrm{mg}, 93 \%$ yield). $[\alpha]_{\mathrm{D}}=-42.8\left(c 0.50, \mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 6.20(\mathrm{~d}, 1 \mathrm{H}, J=3.5 \mathrm{~Hz})$, 5.47 (d, 1H, $J=3.2 \mathrm{~Hz}$ ), $5.44(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.11$ (br. t, $1 \mathrm{H}, J=4.4 \mathrm{~Hz}$ ), 3.92 (dd, $1 \mathrm{H}, J=11.3,8.9$ Hz), 3.19-3.07 (m, 1H), 2.74-2.67 (m, 1H), 2.64-2.56 (m, 1H), 2.29-2.17 (m, 1H), 2.08-2.03 (m, $1 \mathrm{H}), 2.03-1.95(\mathrm{~m}, 1 \mathrm{H}), 1.91$, (ddd, $1 \mathrm{H}, J=11.2,8.6,5.4 \mathrm{~Hz}), 1.78(\mathrm{~s}, 3 \mathrm{H}), 1.54(\mathrm{ddd}, 1 \mathrm{H}, J=$ $12.5,12.5,3.9 \mathrm{~Hz}), 1.10(\mathrm{~d}, 3 \mathrm{H}, J=7.0 \mathrm{~Hz}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 3 \mathrm{H}), 0.06(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.4,140.3,137.0,120.1,119.9,87.3,73.6,52.3,45.3,44.7,42.7,42.3$, 30.7, 28.5, 25.9, 18.2, 15.8, -4.6, -4.9. IR (cast film microscope, $\mathrm{cm}^{-1}$ ): 2957, 2930, 1771. HRMS (EI, m/z) Calcd for $\mathrm{C}_{21} \mathrm{H}_{34} \mathrm{O}_{3} \mathrm{Si}: 362.22772$. Found: 362.22799 .

(1S,2R,3aR,3bS,4aR,5aS,8aR,8bS)-2-((tert-Butyldimethylsilyl)oxy)-1,3b-dimethyl-6methylenedecahydrooxireno $\left[2^{\prime}, 3^{\prime}: 7,8\right]$ azuleno $[4,5-b]$ furan-7(2H)-one (2)
$\gamma$-Lactone 10 ( $90 \mathrm{mg}, 0.25 \mathrm{mmol}, 1.0$ equiv) was dissolved in 5 ml DCM under argon and cooled to $0^{\circ} \mathrm{C} . m$-CPBA ( $111 \mathrm{mg}, 0.50 \mathrm{mmol}, 2.0$ equiv) was added and the reaction was stirred at $0^{\circ} \mathrm{C}$ for 1 h . The reaction mixture was quenched with water and diluted with ethyl acetate. The water layer was extracted with ethyl acetate (x3), then the organics were washed with $1 \mathrm{~N} \mathrm{NaOH}(\mathrm{x} 2)$ and brine, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. The crude mixture was purified by flash chromatography ( $20 \%$ ethyl acetate:hexanes) to provide $\mathbf{2}$ as an inseparable mixture of diastereomers ( $85 \mathrm{mg}, 90 \%$ yield). The major diastereomer was the desired one and was obtained in a ratio ranging from 3.5:1 to 4.5:1, depending on the batch.

Major diastereomer: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.18(\mathrm{~d}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz}), 5.48(\mathrm{~d}, 1 \mathrm{H}, J=$ $3.2 \mathrm{~Hz}), 4.23(\mathrm{td}, 1 \mathrm{H}, J=5.8,2.4 \mathrm{~Hz}), 3.60(\mathrm{dd}, 1 \mathrm{H}, J=11.7,9.3 \mathrm{~Hz}), 3.11(\mathrm{dt}, 1 \mathrm{H}, J=13.9,7.5$ Hz ), $3.06(\mathrm{~d}, 1 \mathrm{H}, J=5.0 \mathrm{~Hz}$ ), 2.89-2.76 (m, 1H), $2.67(\mathrm{ddd}, 1 \mathrm{H}, J=15.4,5.0,3.6 \mathrm{~Hz}), 2.43(\mathrm{pd}$, $1 \mathrm{H}, J=7.2,2.5 \mathrm{~Hz}), 1.98(\mathrm{dd}, 1 \mathrm{H}, J=8.8,2.9 \mathrm{~Hz}), 1.96-1.73(\mathrm{~m}, 3 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.02(\mathrm{~d}, 3 \mathrm{H}$, $J=6.8 \mathrm{~Hz}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 3 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H})$.

Minor diastereomer: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.20(\mathrm{~d}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz}), 5.47(\mathrm{~d}, 1 \mathrm{H}, J=$ $3.3 \mathrm{~Hz}), 4.42(\mathrm{dd}, 1 \mathrm{H}, J=10.3,8.9 \mathrm{~Hz}), 4.09-4.06(\mathrm{~m}, 1 \mathrm{H}), 3.05(\mathrm{~d}, 1 \mathrm{H}, J=3.9), 3.01-2.93(\mathrm{~m}$, $1 \mathrm{H}), 2.89-2.76(\mathrm{~m}, 1 \mathrm{H}), 2.57$ (ddd, $1 \mathrm{H}, J=15.4,5.0,5.0 \mathrm{~Hz}$ ), 2.14 (dd, $1 \mathrm{H}, J=16.0,11.9 \mathrm{~Hz}$ ), 2.09-2.02 (m, 1H), 1.96-1.73 (m, 3H), $1.29(\mathrm{~s}, 3 \mathrm{H}), 1.09(\mathrm{~d}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.06$ (s, 3H), 0.05 (s, 3H).

Mixture of diastereomers: ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 169.8,139.9,139.6,120.0,119.5$, $85.4,83.5,75.2,72.1,64.9,64.7,63.6,62.3,60.4,51.8,51.0,45.7,42.9,41.9,41.5,40.5,40.2$, $40.1,39.4,29.3,29.0,27.2,25.8,25.8,24.9,18.1,15.4,14.9,14.2,11.9,-4.6,-4.8,-5.0$. IR (microscope, $\mathrm{cm}^{-1}$ ): 2957, 2929, 1769. HRMS (EI, m/z) Calcd for $\mathrm{C}_{21} \mathrm{H}_{34} \mathrm{O}_{4} \mathrm{Si}: 378.22263$.
Found: 378.22250.

(3aS,6R,6aR,8R,9S,9aS,9bR)-8-((tert-Butyldimethylsilyl)oxy)-6-hydroxy-6,9-dimethyl-3-methylenedecahydroazuleno[4,5-b]furan-2(9bH)-one (11)

Epoxide 2 ( $30 \mathrm{mg}, 0.08 \mathrm{mmol}, 1.0$ equiv) was dissolved in THF under argon and cooled to -78 ${ }^{\circ} \mathrm{C}$. To this mixture was added $\operatorname{DIBALH}(0.19 \mathrm{ml}, 0.28 \mathrm{mmol}, 3.5$ equiv) as a 1.5 M solution in toluene. The reaction mixture was stirred for 2 h , then $\mathrm{LiEt}_{3} \mathrm{BH}(87 \mu \mathrm{l}, 87 \mu \mathrm{~mol}, 1.1$ equiv) was added and the reaction was allowed to warm to rt over 30 min . The reaction was quenched with a $30 \%$ solution of NaKtartrate and extracted with ethyl acetate (x4). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The crude residue was dissolved in 15 ml DCM and to this was added powdered $90 \% \mathrm{MnO}_{2}$ ( $\sim 20$ equiv). The reaction was stirred overnight at rt , then filtered through celite, washed with DCM and the solvent was removed. Flash chromatography (20 to $50 \%$ ethyl acetate:hexanes) provided $\mathbf{1 1}$ as a colorless oil ( $15 \mathrm{mg}, 50 \%$ yield). $[\alpha]_{\mathrm{D}}=$ $-22.2\left(c 0.48, \mathrm{CHCl}_{3}\right) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.19(\mathrm{~d}, 1 \mathrm{H}, J=3.5 \mathrm{~Hz}), 5.47(\mathrm{~d}, 1 \mathrm{H}, J=$ $3.2 \mathrm{~Hz})$, 4.11-4.04 (m, 2H), 2.88-2.77 (m, 1H), 2.25-2.12 (m, 2H), 2.00-1.87 (m, 2H), 1.87-1.81 $(\mathrm{m}, 2 \mathrm{H}), 1.78-1.68(\mathrm{~m}, 1 \mathrm{H}), 1.51-1.39(\mathrm{~m}, 1 \mathrm{H}), 1.19(\mathrm{~s}, 3 \mathrm{H}), 1.11(\mathrm{~d}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 0.89(\mathrm{~s}$, 9H), $0.05(\mathrm{~s}, 3 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 170.3,140.1,119.5,85.9,75.5$, $75.0,49.8,46.4,45.6,41.6,37.8,29.8,25.9,25.3,18.2,15.0,-4.6,-4.8$. IR (microscope, $\mathrm{cm}^{-1}$ ): 3474, 2955, 2928, 1762. HRMS (EI, m/z) Calcd for $\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{O}_{4} \mathrm{Si}\left[\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9}\right]^{+}: 323.16788$. Found: 323.16800 .

(3aS,6R,6aR,9S,9aR,9bR)-6-Hydroxy-6,9-dimethyl-3-methyleneoctahydroazuleno[4,5-b]furan-2,8(3H,9bH)-dione (1)

To a suspension of PDC ( $15 \mathrm{mg}, 1.5$ equiv) in 0.5 ml DCM under argon at $0^{\circ} \mathrm{C}$ was added $\mathrm{TMSCl}(12 \mu \mathrm{l}, 3.5$ equiv). After stirring this mixture for $5 \mathrm{~min}, \gamma$-lactone 11 ( $10 \mathrm{mg}, 1.0$ equiv) was dissolved in 0.5 ml DCM and added to the reaction mixture. This mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 4 h , then quenched by the addition of moist silica gel. The resulting suspension was filtered through a short pad of silica with DCM and ethyl acetate. Purification of the reaction mixture
was done via flash chromatography ( $20 \%$ ethyl acetate:hexanes) to provide pure $\mathbf{1}$ as a crystalline solid ( $5 \mathrm{mg}, 71 \%$ yield).

The initial isolation ${ }^{4}$ of chinensiolide B indicates that the NMR spectra were obtained using $\mathrm{CDCl}_{3}$ as the solvent. However, upon consultation with the corresponding author (Professor Masayoshi Ando), it was found that the NMR spectra were actually recorded in deuterated pyridine. Upon running NMR analyses of the synthetic sample of $\mathbf{1}$ in deuterated pyridine, the spectral properties ( ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR) were identical to those provided by Professor Ando.

At this point, it was also noticed that the ${ }^{1} \mathrm{H}$ NMR data listed in the isolation paper did not match the spectrum provided by Professor Ando. The peak listed at 2.27 ppm in the ${ }^{1} \mathrm{H}$ NMR data in the paper was observed at 2.76 ppm in the actual spectrum. Also, the peak listed at 1.36 ppm was observed at 1.84 ppm in the actual spectrum. These errors must have been made in describing the chemical shifts for these two protons in the isolation paper, as the actual spectrum of natural chinensiolide B is different from that listed in the isolation paper.

After these corrections and errors are taken into account, the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of synthetic 1 was identical to that of natural $\mathbf{1}$. Other analyses (IR, HRMS and $[\alpha]^{20}{ }_{\mathrm{D}}$ ) were consistent with that reported in the original isolation paper. ${ }^{4}$
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 6.24(\mathrm{~d}, 1 \mathrm{H}, J=3.5 \mathrm{~Hz}), 5.52(\mathrm{~d}, 1 \mathrm{H}, J=3.1 \mathrm{~Hz}), 4.11(\mathrm{dd}, 1 \mathrm{H}, J$ $=9.9,9.9 \mathrm{~Hz}), 3.12-3.02(\mathrm{~m}, 1 \mathrm{H}), 2.66-2.57(\mathrm{~m}, 2 \mathrm{H}), 2.51-2.34(\mathrm{~m}, 3 \mathrm{H}), 2.32-2.22(\mathrm{~m}, 1 \mathrm{H}), 1.99$ $(\mathrm{d}, 1 \mathrm{H}, J=14.5,6.1,6.1 \mathrm{~Hz}), 1.78(\mathrm{ddd}, 1 \mathrm{H}, J=14.3,8.3,5.8 \mathrm{~Hz}), 1.57-1.47(\mathrm{~m}, 1 \mathrm{H}), 1.25(\mathrm{~d}$, $3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.22(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 219.0,169.7,139.8,120.5,85.0$, $74.2,50.5,48.0,46.0,44.3,39.8,39.8,28.0,25.3,15.9$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \operatorname{Pyr-D} 5$ ): $\delta 6.24(\mathrm{~d}, 1 \mathrm{H}, J=3.7 \mathrm{~Hz}), 6.22(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 5.44(\mathrm{~d}, 1 \mathrm{H}, J=3.2$ Hz ), $4.12(\mathrm{dd}, 1 \mathrm{H}, J=9.9 \mathrm{~Hz}), 3.09-2.96(\mathrm{~m}, 2 \mathrm{H}), 2.79(\mathrm{~m}, 1 \mathrm{H}), 2.54-2.44(\mathrm{~m}, 2 \mathrm{H}), 2.34$ (ddd, $1 \mathrm{H}, J=10.0,8.4,8.4 \mathrm{~Hz}$ ), 2.18-2.03 (m, 2H), 1.88-1.81 (m, 1H), 1.39-1.29 (m, 1H), $1.27(\mathrm{~s}$, $3 \mathrm{H}), 1.24(\mathrm{~d}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR (100 MHz, Pyr-D 5 ): $\delta 219.7,170.1,141.2,119.6,86.0$, 73.3, 50.5, 48.3, 46.4, 44.5, 40.4, 27.1, 25.8, 15.8.

IR (microscope, $\mathrm{cm}^{-1}$ ): 3459, 2968, 2932, 1761, 1738. HRMS (EI, m/z) Calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}_{4}$ : 264.13617. Found: 264.13638. $[\alpha]_{\mathrm{D}}=+4.8\left(c 0.12, \mathrm{CHCl}_{3}\right)$. Literature ${ }^{4}=+2.6\left(c 0.469, \mathrm{CHCl}_{3}\right)$.

## 3) Comparison of ${ }^{13} \mathrm{C}$-NMR data for 1

| Carbon | Natural product ${ }^{4}$ in Pyr-D <br> $(\mathrm{ppm})$ | Synthetic product in Pyr-D <br> $(\mathrm{ppm})$ | Synthetic product in $\mathrm{CDCl}_{3}$ <br> $(\mathrm{ppm})$ |
| :---: | :---: | :---: | :---: |
| 1 | 46.5 | 46.4 | 46.0 |
| 2 | 40.3 | 40.4 | 39.8 |
| 3 | 219.3 | 219.7 | 219.0 |
| 4 | 48.3 | 48.3 | 48.0 |
| 5 | 50.5 | 50.5 | 50.5 |
| 6 | 85.8 | 86.0 | 85.0 |
| 7 | 44.5 | 44.5 | 44.3 |
| 8 | 25.9 | 25.8 | 25.3 |
| 9 | 40.6 | $-a^{\text {a }}$ | 39.8 |
| 10 | 73.1 | 73.3 | 74.2 |
| 11 | 141.3 | 141.2 | 139.8 |
| 12 | 169.9 | 170.1 | 169.7 |
| 13 | 119.4 | 119.6 | 120.5 |
| 14 | 27.3 | 27.1 | 28.0 |
| 15 | 15.9 | 15.8 | 15.9 |

${ }^{\text {a }}$ Due to conformational flexibility in the seven-member ring, the carbon peaks for C9 and C14 appear as very broad peaks. The peak for C 9 in the ${ }^{13} \mathrm{C}$ NMR is sometimes not observed due to overlap with the C 2 peak. This issue was mentioned and discussed in the original isolation paper. ${ }^{4}$

## 4) Crystallographic data for 8

ORTEP diagram of crystal structure


Table 1. Crystallographic Experimental Details

```
A. Crystal Data
formula
formula weight
crystal dimensions (mm)
crystal system
space group
unit cell parameters }\mp@subsup{}{}{a
a(\AA)
    b(\AA)
    c(\AA)
    \beta(deg)
    V(\AA}\mp@subsup{\AA}{}{3}
    Z
\rhocalcd
\mu(mm-1)
C24H39 NO
4 1 7 . 6 5
0.47\times0.25 \times 0.12
monoclinic
P21 (No. 4)
10.9585 (4)
7.9293 (3)
15.5064 (6)
107.7359 (4)
1283.36 (8)
2
1.081
\(\mu\left(\mathrm{mm}^{-1}\right)\)
0.113
```


## B. Data Collection and Refinement Conditions

diffractometer
radiation $(\lambda[\AA])$
temperature $\left({ }^{\circ} \mathrm{C}\right)$
scan type
data collection $2 \theta$ limit (deg)
total data collected
independent reflections
number of observed reflections ( $N O$ )
structure solution method
refinement method
absorption correction method range of transmission factors data/restraints/parameters
Flack absolute structure parameter ${ }^{e}$ goodness-of-fit ( $S)^{f}$
final $R$ indices $g$

$$
R_{1}\left[F_{\mathrm{o}}^{2} \geq 2 \sigma\left(F_{\mathrm{o}}^{2}\right)\right] \quad 0.0319
$$

$w R_{2}\left[F_{0}^{2} \geq-3 \sigma\left(F_{0}^{2}\right)\right]$
largest difference peak and hole

Bruker D8/APEX II CCD ${ }^{b}$
graphite-monochromated Mo $\mathrm{K} \alpha(0.71073)$
-100
$\omega$ scans ( $0.3^{\circ}$ ) (20 s exposures)
55.00
$11315(-14 \leq h \leq 14,-10 \leq k \leq 10,-20 \leq l \leq 19)$
$5820\left(R_{\text {int }}=0.0160\right)$
$5554\left[F_{\mathrm{o}}{ }^{2} \geq 2 \sigma\left(F_{\mathrm{o}}^{2}\right)\right]$
direct methods (SHELXD ${ }^{c}$ )
full-matrix least-squares on $F^{2}$ (SHELXL-97d)
Gaussian integration (face-indexed)
0.9867-0.9482
$5820\left[F_{0}^{2} \geq-3 \sigma\left(F_{\mathrm{o}}^{2}\right)\right] / 0 / 263$
0.04(7)
$1.045\left[F_{0}^{2} \geq-3 o\left(F_{0}^{2}\right)\right]$
0.0868
0.388 and -0.159 e $\AA^{-3}$
${ }^{a}$ Obtained from least-squares refinement of 9952 reflections with $5.42^{\circ}<2 \theta<54.90^{\circ}$.
${ }^{b}$ Programs for diffractometer operation, data collection, data reduction and absorption correction were those supplied by Bruker.
${ }^{c}$ Schneider, T. R.; Sheldrick, G. M. Acta Crystallogr. 2002, D58, 1772-1779.
${ }^{d}$ Sheldrick, G. M. Acta Crystallogr. 2008, A64, 112-122.
${ }^{e}$ Flack, H. D. Acta Crystallogr. 1983, A39, 876-881; Flack, H. D.; Bernardinelli, G. Acta Crystallogr. 1999, A55, 908-915; Flack, H. D.; Bernardinelli, G. J. Appl. Cryst. 2000, 33, 1143-1148. Ideally, the Flack parameter will refine to a value near zero if the structure is in the correct configuration and will refine to a value near one for the inverted configuration. In this case, the relatively large standard uncertainty indicates that the structural data alone should not be used to confirm absolute stereochemistry. The conformation of the model presented herein is based upon the established stereochemistry of the precursor compound (specifically the configurations of the stereogenic centers of the 3 - $\{t$-butyl(dimethyl)siloxy $\}$ -2-methyl-5-(prop-1-en-2-yl)cyclopentyl group).
$f_{S}=\left[\Sigma w\left(F_{0}^{2}-F_{\mathrm{c}^{2}}^{2}\right)^{2 /(n-p)}\right]^{1 / 2}(n=$ number of data; $p=$ number of parameters varied; $w=$ $\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0570 P)^{2}+0.0794 P\right]^{-1}$ where $\left.P=\left[\operatorname{Max}\left(F_{\mathrm{o}}^{2}, 0\right)+2 F_{\mathrm{c}^{2}}{ }^{2}\right] / 3\right)$.
$g_{R_{1}}=\Sigma| | F_{\mathrm{o}}\left|-\left|F_{\mathrm{c}}\right|\right| \Sigma\left|F_{\mathrm{o}}\right| ; w R_{2}=\left[\Sigma w\left(F_{\mathrm{o}}^{2}-F_{\mathrm{c}^{2}}\right)^{2} / \Sigma w\left(F_{\mathrm{o}}{ }^{4}\right)\right]^{1 / 2}$.

Table 2. Atomic Coordinates and Equivalent Isotropic Displacement Parameters

| Atom | $x$ | $y$ | $z$ | $U_{\mathrm{eq}}, \AA^{2}$ |
| :--- | :---: | :--- | :--- | :--- |
| Si | $-0.17321(3)$ | $0.52206(4)$ | $0.04847(2)$ | $0.02365(8)^{*}$ |
| O1 | $0.34611(8)$ | $0.55436(13)$ | $0.38211(6)$ | $0.0317(2)^{*}$ |
| O2 | $0.47670(10)$ | $0.76824(15)$ | $0.44097(8)$ | $0.0418(3)^{*}$ |
| O3 | $-0.07215(8)$ | $0.46586(12)$ | $0.14665(6)$ | $0.0287(2)^{*}$ |
| N | $0.14210(17)$ | $0.9668(2)$ | $0.46920(13)$ | $0.0605(4)^{*}$ |
| C1 | $0.40153(12)$ | $0.66359(19)$ | $0.44880(9)$ | $0.0301(3)^{*}$ |
| C2 | $0.35633(13)$ | $0.63258(18)$ | $0.53070(9)$ | $0.0281(3)^{*}$ |
| C3 | $0.23970(11)$ | $0.51706(18)$ | $0.49444(8)$ | $0.0256(2)^{*}$ |
| C4 | $0.26391(12)$ | $0.43460(17)$ | $0.41129(8)$ | $0.0251(3)^{*}$ |
| C5 | $0.14563(12)$ | $0.39474(16)$ | $0.33250(8)$ | $0.0235(2)^{*}$ |
| C6 | $0.05835(12)$ | $0.55107(17)$ | $0.29753(8)$ | $0.0290(3)^{*}$ |
| C7 | $0.03885(12)$ | $0.56094(17)$ | $0.19503(8)$ | $0.0286(3)^{*}$ |
| C8 | $0.15888(13)$ | $0.47833(19)$ | $0.18460(9)$ | $0.0320(3)^{*}$ |
| C9 | $0.17337(12)$ | $0.32347(17)$ | $0.24605(8)$ | $0.0265(3)^{*}$ |
| C10 | $0.33532(16)$ | $0.80013(19)$ | $0.57389(10)$ | $0.0378(3)^{*}$ |
| C11 | $0.22748(17)$ | $0.8951(2)$ | $0.51583(12)$ | $0.0419(4)^{*}$ |
| C12 | $0.22169(15)$ | $0.38815(19)$ | $0.56324(9)$ | $0.0337(3)^{*}$ |
| C13 | $0.2031(2)$ | $0.4701(2)$ | $0.64526(11)$ | $0.0487(4)^{*}$ |
| C14 | $0.2859(3)$ | $0.4598(3)$ | $0.72698(12)$ | $0.0747(7)^{*}$ |
| C15 | $-0.06660(14)$ | $0.5473(3)$ | $0.32158(10)$ | $0.0450(4)^{*}$ |
| C16 | $-0.22779(14)$ | $0.74276(18)$ | $0.05361(12)$ | $0.0387(3)^{*}$ |
| C17 | $-0.09707(13)$ | $0.5053(2)$ | $-0.04333(9)$ | $0.0393(3)^{*}$ |
| C18 | $-0.31049(13)$ | $0.36902(17)$ | $0.02907(10)$ | $0.0295(3)^{*}$ |
| C19 | $-0.26065(16)$ | $0.18782(19)$ | $0.03268(13)$ | $0.0435(4)^{*}$ |
| C20 | $-0.40858(15)$ | $0.4003(2)$ | $-0.06451(12)$ | $0.0456(4)^{*}$ |
| C21 | $-0.37666(16)$ | $0.3914(2)$ | $0.10211(12)$ | $0.0459(4)^{*}$ |
| C22 | $0.29202(14)$ | $0.2170(2)$ | $0.25999(9)$ | $0.0361(3)^{*}$ |
| C23 | $0.39190(15)$ | $0.2661(3)$ | $0.23399(11)$ | $0.0494(4)^{*}$ |
| C24 | $0.28853(19)$ | $0.0474(2)$ | $0.30016(11)$ | $0.0520(4)^{*}$ |

Anisotropically-refined atoms are marked with an asterisk $(*)$. The form of the anisotropic displacement parameter is: $\exp \left[-2 \pi^{2}\left(h^{2} a^{* 2} U_{11}+k^{2} b^{* 2} U_{22}+l^{2} c^{* 2} U_{33}+2 k l b^{*} c^{*} U_{23}\right.\right.$ $\left.\left.+2 h l a^{*} c^{*} U_{13}+2 h k a^{*} b^{*} U_{12}\right)\right]$.

Table 3. Selected Interatomic Distances ( $\AA$ )

| Atom1 | Atom2 | Distance | Atom1 | Atom2 | Distance |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Si | O 3 | $1.6479(9)$ | C 5 | C 6 | $1.5587(17)$ |
| Si | C 16 | $1.8589(14)$ | C 5 | C 9 | $1.5675(17)$ |
| Si | C 17 | $1.8610(14)$ | C 6 | C 7 | $1.5397(18)$ |
| Si | C 18 | $1.8843(14)$ | C 6 | C 15 | $1.5252(19)$ |
| O 1 | C 1 | $1.3438(17)$ | C 7 | C 8 | $1.5215(19)$ |
| O 1 | C 4 | $1.4718(16)$ | C 8 | C 9 | $1.5326(19)$ |
| O 2 | C 1 | $1.2012(18)$ | C 9 | C 22 | $1.5095(19)$ |
| O 3 | C 7 | $1.4335(15)$ | C 10 | C 11 | $1.458(2)$ |
| N | C 11 | $1.144(2)$ | C 12 | C 13 | $1.497(2)$ |
| C 1 | C 2 | $1.517(2)$ | C 13 | C 14 | $1.317(3)$ |
| C 2 | C 3 | $1.5332(18)$ | C 18 | C 19 | $1.532(2)$ |
| C 2 | C 10 | $1.536(2)$ | C 18 | C 20 | $1.540(2)$ |
| C 3 | C 4 | $1.5391(16)$ | C 18 | C 21 | $1.530(2)$ |
| C 3 | C 12 | $1.5330(19)$ | C 22 | C 23 | $1.335(2)$ |
| C 4 | C 5 | $1.5191(16)$ | C 22 | C 24 | $1.487(3)$ |

Table 4. Selected Interatomic Angles (deg)

| Atom1 | Atom2 | Atom3 | Angle | Atom1 | Atom2 | Atom3 | Angle |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| O3 | Si | C16 | 110.45(6) | C5 | C6 | C7 | 106.09(10) |
| O3 | Si | C17 | 110.83(6) | C5 | C6 | C15 | 113.62(12) |
| O3 | Si | C18 | 104.62(5) | C7 | C6 | C15 | 113.63(11) |
| C16 | Si | C17 | 108.98(8) | O3 | C7 | C6 | 110.02(10) |
| C16 | Si | C18 | 111.18(7) | O3 | C7 | C8 | 109.95(11) |
| C17 | Si | C18 | 110.73(7) | C6 | C7 | C8 | 103.60(10) |
| C1 | O1 | C4 | 110.78(10) | C7 | C8 | C9 | 102.01(10) |
| Si | O3 | C7 | 124.44(8) | C5 | C9 | C8 | 103.44(11) |
| O1 | C1 | O2 | 122.04(13) | C5 | C9 | C22 | 117.52(10) |
| O1 | C1 | C2 | 110.70(11) | C8 | C9 | C22 | 117.30(12) |
| O2 | C1 | C2 | 127.26(13) | C2 | C10 | C11 | 112.24(12) |
| C1 | C2 | C3 | 103.78(10) | N | C11 | C10 | 178.59(19) |
| C1 | C2 | C10 | 110.81(12) | C3 | C12 | C13 | 112.43(13) |
| C3 | C2 | C10 | 117.23(12) | C12 | C13 | C14 | 123.9(2) |
| C2 | C3 | C4 | 102.94(10) | Si | C18 | C19 | 109.86(10) |
| C2 | C3 | C12 | 114.38(10) | Si | C18 | C20 | 109.92(10) |
| C4 | C3 | C12 | 113.03(12) | Si | C18 | C21 | 110.03(10) |
| O1 | C4 | C3 | 105.14(10) | C19 | C18 | C20 | 108.92(14) |
| O1 | C4 | C5 | 109.82(10) | C19 | C18 | C21 | 109.01(14) |
| C3 | C4 | C5 | 116.08(10) | C20 | C18 | C21 | 109.06(13) |
| C4 | C5 | C6 | 113.40(10) | C9 | C22 | C23 | 122.92(16) |
| C4 | C5 | C9 | 115.00(10) | C9 | C22 | C24 | 115.76(14) |
| C6 | C5 | C9 | 104.33(10) | C23 | C22 | C24 | 121.23(16) |

Table 5. Torsional Angles (deg)

| Atom1 | Atom2 | Atom3 | Atom4 | Angle | Atom1 | Atom2 | Atom3 | Atom4 | Angle |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C16 | Si | O3 | C7 | -51.38(12) | C12 | C3 | C4 | O1 | -149.14(11) |
| C17 | Si | O3 | C7 | 69.50(12) | C12 | C3 | C4 | C5 | 89.29(14) |
| C18 | Si | O3 | C7 | -171.11(10) | C2 | C3 | C12 | C13 | 60.50(17) |
| O3 | Si | C18 | C19 | -56.30(13) | C4 | C3 | C12 | C13 | 177.88(13) |
| O3 | Si | C18 | C20 | -176.15(10) | O1 | C4 | C5 | C6 | -63.40(13) |
| O3 | Si | C18 | C21 | 63.72(11) | O1 | C4 | C5 | C9 | 56.56(14) |
| C16 | Si | C18 | C19 | -175.54(12) | C3 | C4 | C5 | C6 | 55.65(15) |
| C16 | Si | C18 | C20 | 64.62(13) | C3 | C4 | C5 | C9 | 175.61(11) |
| C16 | Si | C18 | C21 | -55.51(13) | C4 | C5 | C6 | C7 | 126.60(11) |
| C17 | Si | C18 | C19 | 63.15(13) | C4 | C5 | C6 | C15 | -107.86(13) |
| C17 | Si | C18 | C20 | -56.69(12) | C9 | C5 | C6 | C7 | 0.74(13) |
| C17 | Si | C18 | C21 | -176.82(11) | C9 | C5 | C6 | C15 | 126.28(12) |
| C4 | O1 | C1 | O2 | 176.92(13) | C4 | C5 | C9 | C8 | -98.91(12) |
| C4 | O1 | C1 | C2 | -2.61(15) | C4 | C5 | C9 | C22 | 32.11(17) |
| C1 | O1 | C4 | C3 | 17.99(13) | C6 | C5 | C9 | C8 | 25.94(12) |
| C1 | O1 | C4 | C5 | 143.56(11) | C6 | C5 | C9 | C22 | 156.96(12) |
| Si | O3 | C7 | C6 | 142.74(9) | C5 | C6 | C7 | O3 | 90.12(12) |
| Si | O3 | C7 | C8 | -103.78(11) | C5 | C6 | C7 | C8 | -27.38(13) |
| O1 | C1 | C2 | C3 | -13.82(15) | C15 | C6 | C7 | O3 | -35.42(17) |
| O1 | C1 | C2 | C10 | -140.49(12) | C15 | C6 | C7 | C8 | -152.92(13) |
| O2 | C1 | C2 | C3 | 166.68(14) | O3 | C7 | C8 | C9 | -73.81(12) |
| O2 | C1 | C2 | C10 | 40.01(19) | C6 | C7 | C8 | C9 | 43.74(13) |
| C1 | C2 | C3 | C4 | 23.33(13) | C7 | C8 | C9 | C5 | -43.19(12) |
| C1 | C2 | C3 | C12 | 146.34(11) | C7 | C8 | C9 | C22 | -174.33(11) |
| C10 | C2 | C3 | C4 | 145.84(12) | C5 | C9 | C22 | C23 | -113.14(16) |
| C10 | C2 | C3 | C12 | -91.14(15) | C5 | C9 | C22 | C24 | 70.28(16) |
| C1 | C2 | C10 | C11 | 67.93(16) | C8 | C9 | C22 | C23 | 11.2(2) |
| C3 | C2 | C10 | C11 | -50.89(17) | C8 | C9 | C22 | C24 | -165.39(12) |
| C2 | C3 | C4 | O1 | -25.22(12) | C2 | C10 | C11 | N | 13(8) |
| C2 | C3 | C4 | C5 | -146.80(11) | C3 | C12 | C13 | C14 | -113.09(19) |

Table 6. Anisotropic Displacement Parameters $\left(U_{\mathrm{ij}}, \AA^{2}\right)$

| Atom | $U_{11}$ | $U_{22}$ | $U_{33}$ | $U_{23}$ | $U_{13}$ | $U_{12}$ |
| :--- | :--- | :--- | :--- | ---: | :--- | ---: |
| Si | $0.02151(15)$ | $0.02406(16)$ | $0.02326(16)$ | $0.00321(13)$ | $0.00365(11)$ | $0.00183(13)$ |
| O 1 | $0.0272(4)$ | $0.0439(6)$ | $0.0235(4)$ | $-0.0025(4)$ | $0.0067(3)$ | $-0.0050(4)$ |
| O 2 | $0.0328(5)$ | $0.0503(7)$ | $0.0400(6)$ | $0.0016(5)$ | $0.0076(4)$ | $-0.0128(5)$ |
| O 3 | $0.0270(4)$ | $0.0306(5)$ | $0.0237(4)$ | $0.0037(4)$ | $0.0004(3)$ | $-0.0017(4)$ |
| N | $0.0679(10)$ | $0.0501(9)$ | $0.0678(10)$ | $0.0031(8)$ | $0.0269(9)$ | $0.0196(8)$ |
| C 1 | $0.0225(6)$ | $0.0372(7)$ | $0.0272(7)$ | $0.0007(5)$ | $0.0024(5)$ | $-0.0003(5)$ |
| C 2 | $0.0285(6)$ | $0.0301(6)$ | $0.0222(6)$ | $-0.0024(5)$ | $0.0025(5)$ | $-0.0020(5)$ |
| C 3 | $0.0271(5)$ | $0.0281(5)$ | $0.0201(5)$ | $-0.0028(5)$ | $0.0052(4)$ | $-0.0005(5)$ |
| C4 | $0.0251(6)$ | $0.0288(6)$ | $0.0203(6)$ | $-0.0007(5)$ | $0.0053(5)$ | $0.0005(5)$ |
| C5 | $0.0249(6)$ | $0.0256(6)$ | $0.0189(5)$ | $0.0000(5)$ | $0.0051(5)$ | $-0.0001(5)$ |
| C6 | $0.0289(6)$ | $0.0306(7)$ | $0.0235(6)$ | $-0.0009(5)$ | $0.0021(5)$ | $0.0042(5)$ |
| C7 | $0.0280(6)$ | $0.0295(7)$ | $0.0227(6)$ | $0.0041(5)$ | $-0.0004(5)$ | $-0.0022(5)$ |
| C8 | $0.0294(6)$ | $0.0434(8)$ | $0.0216(6)$ | $0.0050(5)$ | $0.0053(5)$ | $-0.0019(5)$ |
| C9 | $0.0273(6)$ | $0.0321(7)$ | $0.0182(5)$ | $-0.0020(5)$ | $0.0040(5)$ | $0.0020(5)$ |
| C10 | $0.0469(8)$ | $0.0329(7)$ | $0.0331(7)$ | $-0.0086(6)$ | $0.0112(6)$ | $-0.0077(6)$ |
| C11 | $0.0536(9)$ | $0.0323(7)$ | $0.0458(9)$ | $-0.0058(7)$ | $0.0237(8)$ | $0.0010(7)$ |
| C12 | $0.0419(8)$ | $0.0350(7)$ | $0.0243(6)$ | $-0.0019(6)$ | $0.0104(6)$ | $-0.0077(6)$ |
| C13 | $0.0693(11)$ | $0.0501(10)$ | $0.0345(8)$ | $-0.0068(7)$ | $0.0274(8)$ | $-0.0183(8)$ |
| C14 | $0.1220(19)$ | $0.0741(14)$ | $0.0273(8)$ | $-0.0092(9)$ | $0.0215(10)$ | $-0.0426(14)$ |
| C15 | $0.0335(7)$ | $0.0681(12)$ | $0.0329(7)$ | $-0.0016(8)$ | $0.0093(6)$ | $0.0156(8)$ |
| C16 | $0.0342(7)$ | $0.0240(6)$ | $0.0530(9)$ | $0.0041(6)$ | $0.0062(7)$ | $0.0045(6)$ |
| C17 | $0.0334(7)$ | $0.0568(10)$ | $0.0279(6)$ | $0.0073(7)$ | $0.0096(5)$ | $0.0015(7)$ |
| C18 | $0.0246(6)$ | $0.0272(6)$ | $0.0345(7)$ | $0.0021(5)$ | $0.0058(5)$ | $0.0006(5)$ |
| C19 | $0.0406(8)$ | $0.0257(7)$ | $0.0617(11)$ | $-0.0015(7)$ | $0.0116(8)$ | $-0.0010(6)$ |
| C20 | $0.0307(7)$ | $0.0481(9)$ | $0.0467(9)$ | $0.0045(8)$ | $-0.0050(6)$ | $-0.0068(7)$ |
| C21 | $0.0375(8)$ | $0.0478(9)$ | $0.0578(10)$ | $-0.0014(8)$ | $0.0227(8)$ | $-0.0090(7)$ |
| C22 | $0.0353(7)$ | $0.0490(9)$ | $0.0198(6)$ | $-0.0072(6)$ | $0.0024(5)$ | $0.0124(6)$ |
| C23 | $0.0346(8)$ | $0.0721(12)$ | $0.0398(8)$ | $-0.0126(8)$ | $0.0087(7)$ | $0.0110(8)$ |
| C24 | $0.0664(10)$ | $0.0505(10)$ | $0.0348(8)$ | $0.0023(7)$ | $0.0090(7)$ | $0.0247(9)$ |

The form of the anisotropic displacement parameter is:
$\exp \left[-2 \pi^{2}\left(h^{2} a^{* 2} U_{11}+k^{2} b^{* 2} U_{22}+l^{2} c^{* 2} U_{33}+2 k l b^{*} c^{*} U_{23}+2 h l a^{*} c^{*} U_{13}+2 h k a^{*} b^{*} U_{12}\right)\right]$

Table 7. Derived Atomic Coordinates and Displacement Parameters for Hydrogen Atoms

| Atom | $x$ | $y$ | $z$ | $U_{\text {eq }}, \AA^{2}$ |
| :--- | :---: | :---: | :---: | :---: |
| H2 | 0.4245 | 0.5680 | 0.5764 | 0.034 |
| H3 | 0.1609 | 0.5885 | 0.4736 | 0.031 |
| H4 | 0.3130 | 0.3278 | 0.4310 | 0.030 |
| H5 | 0.0942 | 0.3088 | 0.3535 | 0.028 |
| H6 | 0.1071 | 0.6537 | 0.3262 | 0.035 |
| H7 | 0.0317 | 0.6807 | 0.1739 | 0.034 |
| H8A | 0.2340 | 0.5540 | 0.2054 | 0.038 |
| H8B | 0.1470 | 0.4453 | 0.1210 | 0.038 |
| H9 | 0.0994 | 0.2482 | 0.2160 | 0.032 |
| H10A | 0.3197 | 0.7768 | 0.6324 | 0.045 |
| H10B | 0.4140 | 0.8694 | 0.5864 | 0.045 |
| H12A | 0.1463 | 0.3169 | 0.5339 | 0.040 |
| H12B | 0.2978 | 0.3138 | 0.5821 | 0.040 |
| H13 | 0.1270 | 0.5334 | 0.6380 | 0.058 |
| H14A | 0.3629 | 0.3975 | 0.7364 | 0.090 |
| H14B | 0.2686 | 0.5148 | 0.7764 | 0.090 |
| H15A | -0.1166 | 0.6488 | 0.2980 | 0.054 |
| H15B | -0.0475 | 0.5434 | 0.3875 | 0.054 |
| H15C | -0.1160 | 0.4471 | 0.2947 | 0.054 |
| H16B | -0.2652 | 0.7536 | 0.1033 | 0.046 |
| H16A | -0.2924 | 0.7716 | -0.0037 | 0.046 |
| H16C | -0.1546 | 0.8195 | 0.0640 | 0.046 |
| H17A | -0.0263 | 0.5860 | -0.0322 | 0.047 |
| H17B | -0.1607 | 0.5303 | -0.1017 | 0.047 |
| H17C | -0.0642 | 0.3906 | -0.0445 | 0.047 |
| H19B | -0.1993 | 0.1661 | 0.0924 | 0.052 |
| H19C | -0.2182 | 0.1725 | -0.0140 | 0.052 |
| H19A | -0.3326 | 0.1089 | 0.0219 | 0.052 |
| H20A | -0.4793 | 0.3198 | -0.0743 | 0.055 |
| H20B | -0.3669 | 0.3857 | -0.1117 | 0.055 |
| H20C | -0.4420 | 0.5155 | -0.0671 | 0.055 |
| H21A | -0.4477 | 0.3111 | 0.0916 | 0.055 |
| H21B | -0.4098 | 0.5066 | 0.0997 | 0.055 |
| H21C | -0.3149 | 0.3708 | 0.1618 | 0.055 |
| H23A | 0.4625 | 0.1923 | 0.2410 | 0.059 |
| H23B | 0.3924 | 0.3749 | 0.2085 | 0.059 |
| H24A | 0.3640 | -0.0172 | 0.2986 | 0.062 |
| H24B | 0.2108 | -0.0121 | 0.2653 | 0.062 |
| H24C | 0.2885 | 0.0599 | 0.3630 | 0.062 |
| H3 |  |  |  |  |

## 5) Copies of NMR spectra

Compound $\boldsymbol{Z}-5$ in $\mathrm{CDCl}_{3}$

 $\begin{array}{lllllllllllllllllll}240 & 220 & 200 & 180 & 160 & 140 & 120 & 100 & 80 & 60 & 40 & 20 & 0 & -20 & \mathrm{ppm}\end{array}$

Compound SI-12 in $\mathrm{CDCl}_{3}$




## Compound 7 in $\mathrm{CDCl}_{3}$




Compound SI-13 in $\mathrm{CDCl}_{3}$




## Compound $\mathbf{4}$ in $\mathrm{CDCl}_{3}$





## Compound $\mathbf{3}$ in $\mathrm{CDCl}_{3}$





Compound SI-14 in $\mathrm{CDCl}_{3}$




## Compound 9 in $\mathrm{CDCl}_{3}$




## Compound 10 in $\mathrm{CDCl}_{3}$




Mixture of diastereomers of compound $\mathbf{2}$ in $\mathrm{CDCl}_{3}$




## Compound 11 in $\mathrm{CDCl}_{3}$





NMR spectra for synthetic $\mathbf{1}$ in $\mathrm{CDCl}_{3}$ :



NMR spectra for synthetic $\mathbf{1}$ in pyridine- $D_{5}$ :


NMR spectra provided from corresponding author for natural $\mathbf{1}^{3}$ (in pyridine- $D_{5}$ ):



## 6) References

${ }^{1}$ Hall, D.G.; Deslongchamps, P. J. Org. Chem. 1995, 60, 7796-7814.
${ }^{2}$ Whiting, A. Tetrahedron Lett. 1991, 32, 1503-1506.
${ }^{3}$ Andrews, S.P.; Ball, M.; Wierschem, F.; Cleator, E.; Oliver, S.; Hogenauer, K.; Simic, O.;
Antonello, A.; Hunger, U.; Smith, M.D.; Ley, S.V. Chem. Eur. J. 2007, 13, 5688-5712.
${ }^{4}$ Zhang, S.; Wang, J.; Xue, H.; Deng, Q.; Xing, F.; Ando, M. J. Nat. Prod. 2002, 65, 1927-1929.

## Full authorship for reference 3:

Zhang, S.; Zhao, M.; Bai, L.; Hasegawa, T.; Wang, J.; Wang, L.; Xue, H.; Deng, Q.; Xing, F.; Bai, Y.; Sakai, J.; Bai, J.; Koyanagi, R.; Tsukumo, Y.; Kataoka, T.; Nagai, K.; Hirose, K.; Ando, M. J. Nat. Prod. 2006, 69, 1425-1428.

