

Supporting information

Attempted Isolation and Characterization of Diazirinone (N₂CO)

Christopher J. Shaffer,[†] Brian J. Esselman,[†]

Robert J. McMahon,^{*,†} John F. Stanton,^{*,‡} R. Claude Woods^{*,†}

*Department of Chemistry, University of Wisconsin-Madison,
1101 University Avenue, Madison, Wisconsin 53706-1322*

*Institute for Theoretical Chemistry, Department of Chemistry and Biochemistry,
University of Texas, 1 University Station A5300, Austin, Texas 78712-0165*

General Experimental Methods	2
Cartesian Coordinates for Diazirinone (1) (CCSD(T)/ANO2).....	2
¹ H NMR Spectrum of 3-chloro-3-(<i>p</i> -nitrophenoxy)diazirine (5)	3
¹ H NMR Spectrum of reaction mixture	4
¹⁹ F NMR Spectrum of reaction mixture	5
¹ H NMR Spectrum of 3-fluoro-3-(<i>p</i> -nitrophenoxy)diazirine (9)	6
¹⁹ F NMR Spectrum of 3-fluoro-3-(<i>p</i> -nitrophenoxy)diazirine (9)	7
¹ H NMR Spectrum of <i>p</i> -Nitrophenol	8

[†] University of Wisconsin

[‡] University of Texas

General Information.

Acetonitrile was distilled from CaH₂ and CDCl₃ from CaCl₂ shortly before use. 3-Chloro-3-(*p*-nitrophenoxy)diazirine (**5**) was prepared according to Moss et al. (*J. Am. Chem. Soc.* **2005**, 127, 2408-2409). Of note is the sensitivity of diazirine **5** to acid. When stored at -25 °C in the presence of a small amount of chloroform, decomposition was observed within one month. These problems were minimized by not reusing whatever material had been dedicated to spectroscopy.

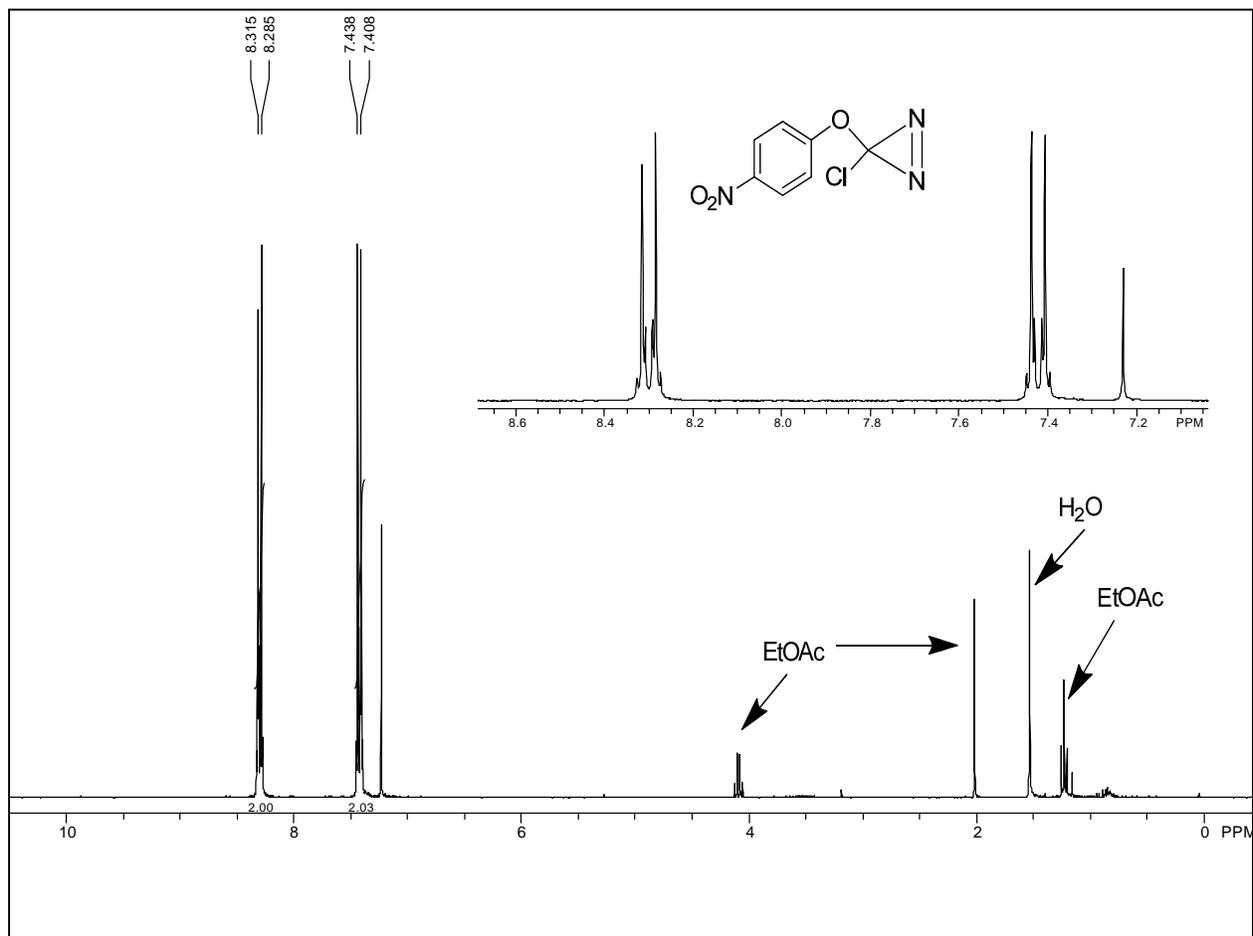
NMR spectra (¹H, 500 MHz or 300 MHz; ¹⁹F, 282 MHz) were obtained in CDCl₃ or CD₂Cl₂ and referenced to residual solvent (NSF CHE-8813550, CHE-9208463, CHE-9629688; NIH 1 S10 RR04981-10, 1 S10 RR08389-01). Mass spectra were obtained using either electrospray ionization with time-of-flight analyzer (NSF CHE-9974739) or electron impact ionization (NSF CHE-9304546).

Cartesian Coordinates.

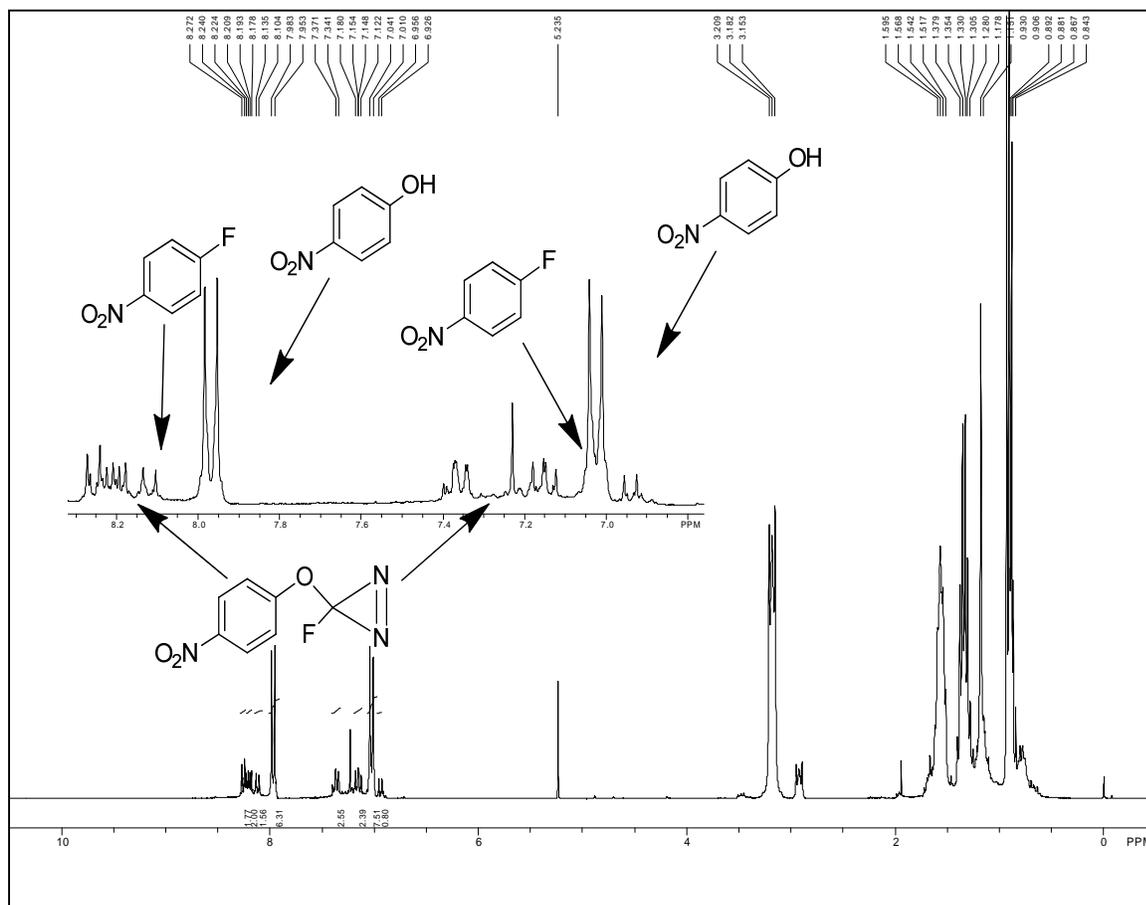
Diazirinone (**1**); CCSD(T)/ANO2

Z-matrix	Atomic	Coordinates (in bohr)		
Symbol	Number	X	Y	Z
O	8	0.00000000	0.00000000	-2.75110880
C	6	0.00000000	0.00000000	-0.50938536
N	7	0.00000000	1.24562519	1.78947761
N	7	0.00000000	-1.24562519	1.78947761

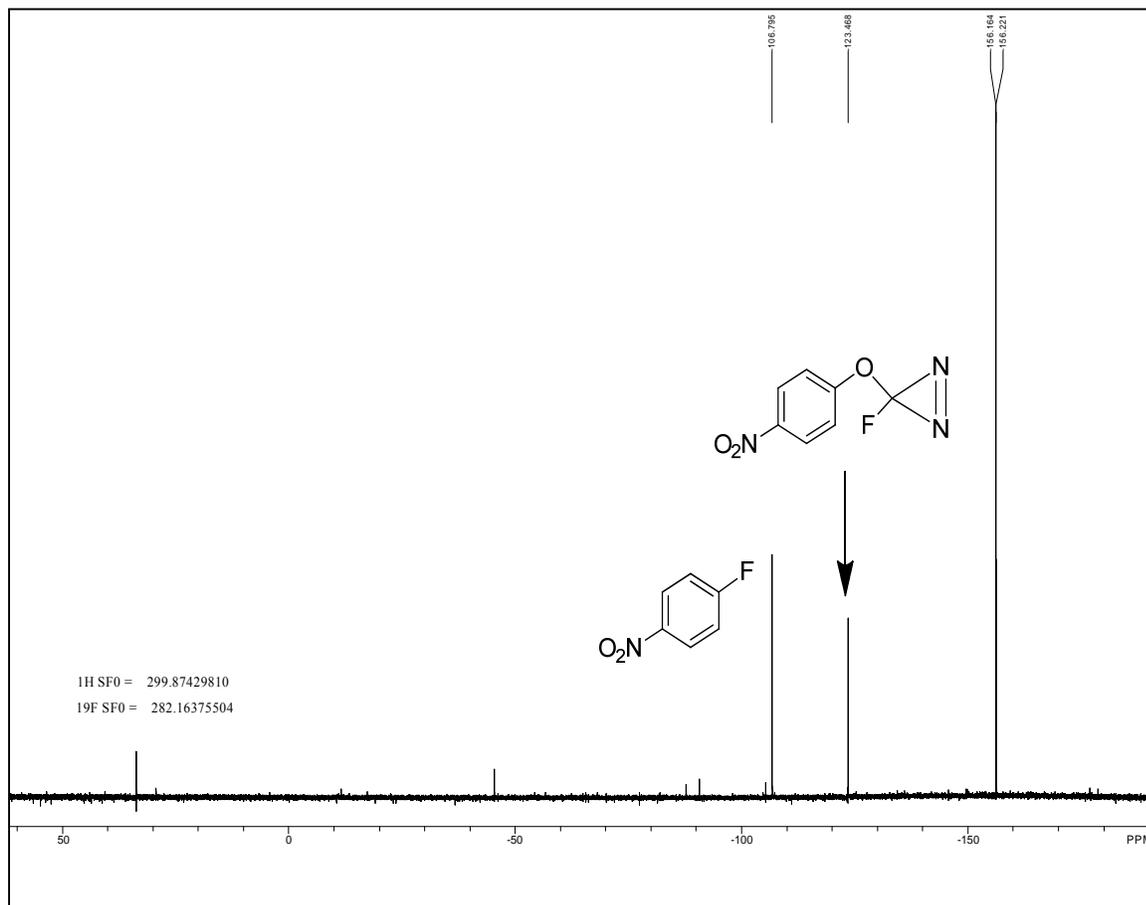
3-Chloro-3-(*p*-nitrophenoxy)diazirine (**5**)
(¹H NMR, 300 MHz)



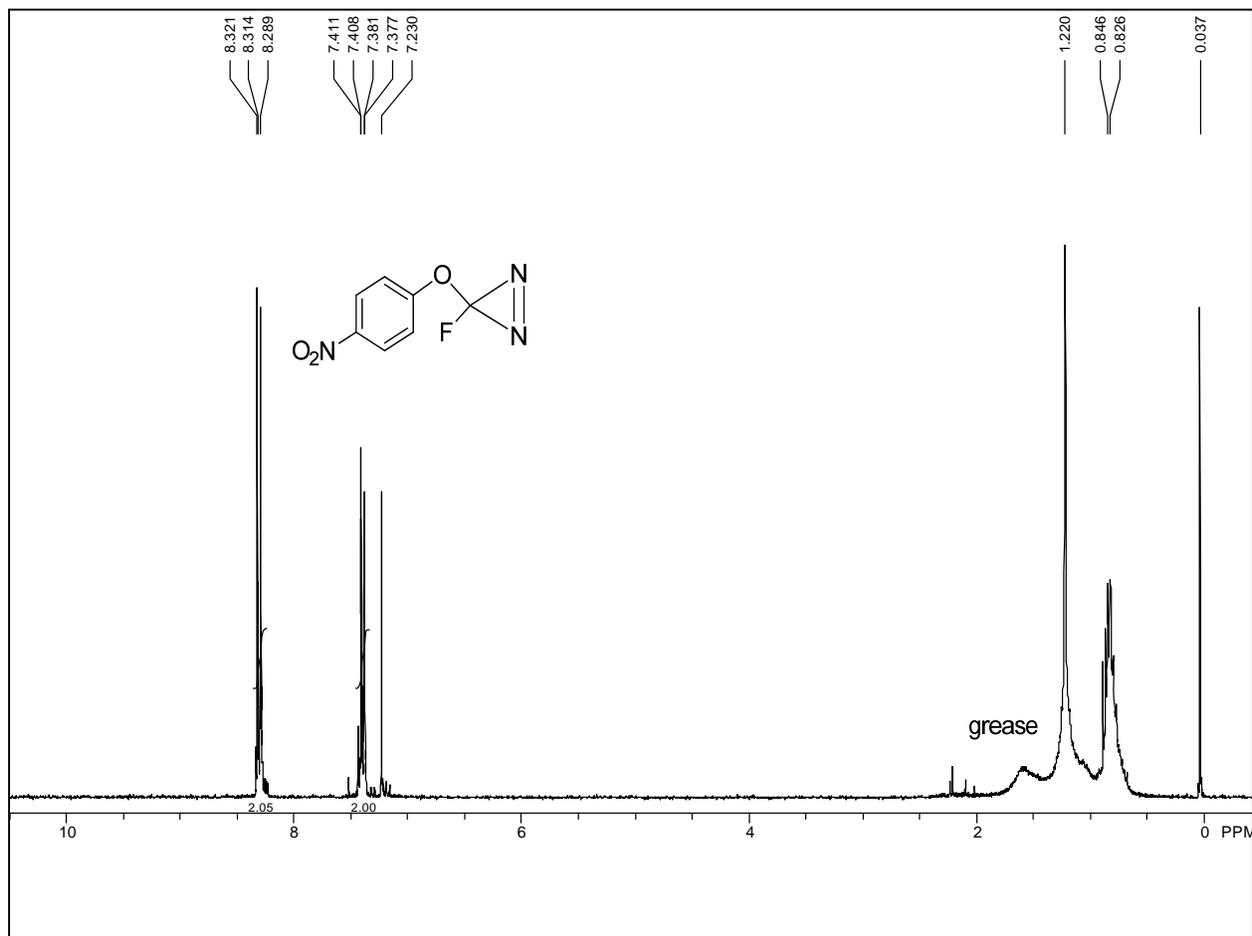
^1H NMR (300 MHz) spectrum of the residual reaction mixture from an attempted matrix isolation experiment using 3-chloro-3-(*p*-nitrophenoxy)diazirine (**5**) and TBAF followed by solvation in CH_2Cl_2 and two aqueous washes as described in the main paper. In this crude mixture can be identified 3-fluoro-3-(*p*-nitrophenoxy)diazirine (**9**), *p*-floronitrobenzene (**7**), and *p*-nitrophenol.



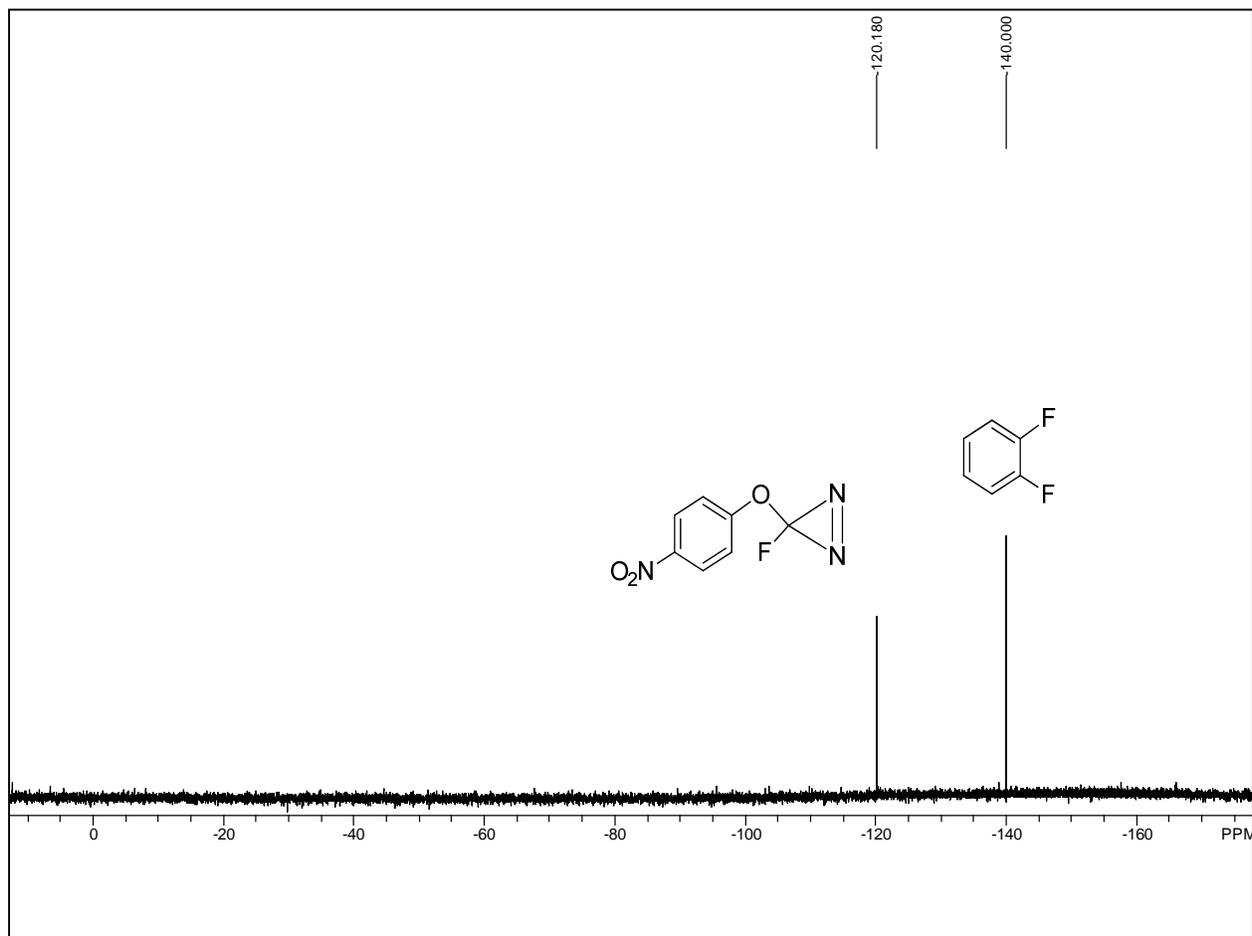
^{19}F NMR (282 MHz) spectrum of the above reaction mixture.



3-Fluoro-3-(*p*-nitrophenoxy)diazirine (**9**)
(¹H NMR, 300 MHz)



3-Fluoro-3-(*p*-nitrophenoxy)diazirine (**9**)
(¹⁹F NMR, 282 MHz)



p-Nitrophenol
(¹H NMR, 300 MHz)

