#### Supplementary Information for:

# Overcoming Strain-Induced Rearrangement Reactions: A Mild Dehydrative Aromatization Protocol for the Synthesis of Highly Distorted *para-*Phenylenes

Nirmal K. Mitra, Rolande Meudom, Hector H. Corzo, John D. Gorden, and Bradley L. Merner \* Department of Chemistry and Biochemistry, Auburn University, Auburn, AL, 36849

#### TABLE OF CONTENTS

- 1. Figure SI-1: Structures/compounds not numbered in the manuscript that appear in the SI
- 2. Scheme SI-1: Proposed intermediate and possible mechanistic pathways for *p*-terphenyl to *m*-terphenyl rearrangement
- 3. General experimental conditions, procedures, and characterization data
- 4. <sup>1</sup>H and <sup>13</sup>C NMR spectra
- 5. X-ray crystal structure and relevant data for compounds 28 and 34
- 6. DFT cartesian coordinates for 31 and 34

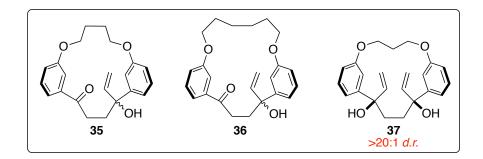
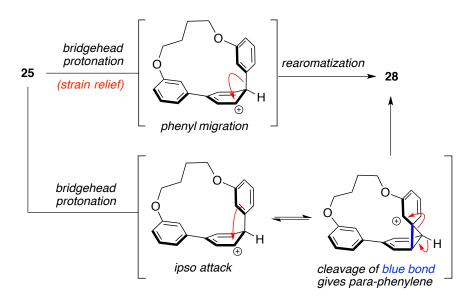


FIGURE SI-1: Structures/compounds not numbered in the manuscript that appear in the SI



SCHEME SI-1: Proposed intermediate and possible mechanistic pathways for *p*-terphenyl to *m*-terphenyl rearrangement

#### General Experimental Conditions

All reactions were run in flame or oven-dried (120 °C) glassware and cooled under a positive pressure of ultra high pure nitrogen or argon gas. All chemicals were used as received from commercial sources, unless otherwise stated. Anhydrous reaction solvents were purified and dried by passing HPLC grade solvents through activated columns of alumina (Glass Contour SDS). All solvents used for chromatographic separations were HPLC grade (hexanes, ethyl acetate, dichloromethane, chloroform, methanol, and acetone). Chromatographic separations were preformed using flash chromatography, as originally reported by Still and co-workers, on silica gel 60 (particle size 43-60 µm), and all chromatography conditions have been reported as height × diameter in centimeters. Reaction progress was monitored by thin layer chromatography (TLC), on glass-backed silica gel plates (pH = 7.0). TLC plates were visualized using a handheld UV lamp (254 nm) and stained using an aqueous ceric ammonium molybdate (CAM) solution. Plates were dipped, wiped clean, and heated from the back of the plate. <sup>1</sup>H and <sup>13</sup>C nuclear magnetic resonance (NMR) spectra were recorded at 400 or 600 MHz, calibrated using residual undeuterated solvent as an internal reference (CHCl<sub>3</sub>, δ 7.27 and 77.2 ppm), reported in parts per million relative to trimethylsilane (TMS, δ 0.00 ppm), and presented as follows: chemical shift ( $\delta$ , ppm), multiplicity (s = singlet, d = doublet, dd = doublet of doublets, ddd = doublet of doublets, dt = doublet of triplets, t = triplet, m = multiplet, p = pentet), coupling constants (J, Hz). High-resolution mass spectrometric (HRMS) data were obtained using a quadrupole time-of-flight (Q-TOF) spectrometer and electrospray ionization (ESI).

#### Experimental procedures and compound characterization data are presented in numerical order

H H C

Dialdehyde 8: 1,4-Dibromobutane (5) (3.98 g, 18.4 mmol) was added to a stirred solution of 3-hydroxybenzaldehyde (5.01 g, 40.9 mmol), K<sub>2</sub>CO<sub>3</sub> (5.66 g, 41.0 mmol) and TBAI (0.76 g, 2.1 mmol) in DMF (40 mL). The reaction was heated at 70 °C for 48 h, at which point water (100 mL) and 1 M HCl (50 mL) were added sequentially. The resulting mixture was extracted with ethyl acetate (3 × 50 mL). The organic extracts

were combined and washed with saturated solution of NaHCO<sub>3</sub> (100 mL) and brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography (18 × 5.0 cm; chloroform, 2% to 5% acetone/chloroform) to afford 8 as white solid (4.75 g, 87%):  $R_f = 0.25$  (chloroform); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  9.97 (s, 2H), 7.48-7.42 (m, 4H), 7.42-7.37 (m,

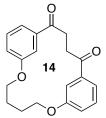
2H), 7.21-7.15 (m, 2H), 4.16-4.04 (m, 4H), 2.07-1.99 (m, 4H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 192.31, 159.67, 137.97, 130.24, 123.72, 122.09, 112.84, 67.87, 26.04; HRMS (ESI) calculated for  $C_{18}H_{19}O_4([M+H]^{\dagger})$  m/z =299.1283, found 299.1290.

10

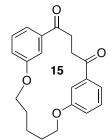
Dialdehyde 10: 1,6-Dibromohexane (7) (1.34 g, 5.49 mmol) was added to a stirred solution of 3-hydroxybenzaldehyde (2.03 g, 16.6 mmol) and K2CO3 (2.78 g, 20.1 mmol) in DMF (30 mL). The reaction was heated at 80 °C for 4 h, at which point water (75 mL) and 1 M HCl (30 mL) were added sequentially. The resulting mixture was extracted with ethyl acetate (3 × 30 mL). The organic extracts were combined and

Streamlined synthesis of macrocyclic diketone 14: Vinylmagnesium chloride (1.6 M in

washed with a saturated solution of NaHCO3 (40 mL) and brine (40 mL), dried over MgSO4, filtered and concentrated under reduced pressure. The residue was purified via flash chromatography (18 cm × 3.8 cm; 9:1 dichloromethane/hexanes, dichloromethane, and 1:9 acetone/dichloromethane) to afford **10** as white solid (1.47 g, 82%):  $R_f = 0.35$  (dichloromethane); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  9.98 (s, 2H), 7.46-7.43 (m, 4H), 7.39 (s, 2H), 7.19-7.18 (m, 2H), 4.05 (t, *J* = 6.4 Hz, 4H), 1.88-1.82 (m, 4H), 1.60-1.55 (m, 4H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 192.2, 159.6, 137.8, 130.0, 123.5, 122.0, 112.7, 68.1, 29.1, 25.8; HRMS (ESI) calculated for  $C_{20}H_{23}O_4$  ([M+H]+) m/z = 327.1596, found 327.1595.

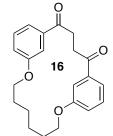


THF, 4.6 mL, 7.4 mmol) was added to a stirred solution of the dialdehyde 8 (1.02 g, 3.42 mmol) in THF (28 mL). After 10 min., the reaction was poured into water (50 mL) and further diluted with 1 M HCl (40 mL). The resulting mixture was extracted with dichloromethane (3 × 20 mL). The combined organic extracts were washed with a saturated solution NaHCO3 (30 mL) and water (30 mL), dried over Na2SO4, filtered and concentrated under reduced pressure. The pale yellow residue was dissolved in dichloromethane (224 mL), heated to 40 °C, followed by the addition of Hoveyda-Grubbs second-generation catalyst (0.052 g, 0.083 mmol). After 1 h, the reaction mixture was concentrated under reduced pressure. The dark brown residue was dissolved in 1:9 methanol/dichloromethane (34 mL), and sodium borohydride (0.380 g, 10.0 mmol) was added. After 3 h, the reaction was poured into water (50 mL) and further diluted with 1 M HCl (20 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 20 mL). The combined organic extracts were washed with water (20 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. The dark brown residue was dissolved in dichloromethane (34 mL), followed by the sequential addition of NaHCO<sub>3</sub> (0.846 g, 10.1 mmol) and Dess-Martin periodinane (2.91 g, 6.86 mmol). After 30 min., the reaction was poured into water (50 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 25 mL). The combined organic extracts were washed with water (30 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography (15 × 2.5 cm, 3:7 EtOAc/hexanes) to afford 1,4-diketone **14** as a white solid (0.551 g, 51% from **8**):  $R_f$  = 0.38 (3:7 EtOAc/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.51 (dd, J = 7.8, 1.3 Hz, 2H), 7.42-7.35 (m, 2H), 7.25-7.21 (m, 2H), 7.11 (ddd, J = 8.2, 2.5, 1.0 Hz, 2H), 4.22-4.17 (m, 4H), 3.09 (s, 4H), 2.00-1.93 (m, 4H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  199.77, 158.62, 137.57, 130.43, 120.89, 120.05, 115.89, 68.44, 36.22, 25.88; HRMS (ESI) calculated for C<sub>20</sub>H<sub>21</sub>O<sub>4</sub> ([M+H]<sup>+</sup>) m/z = 325.1440, found 325.1436.



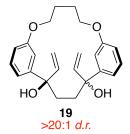
Streamlined synthesis of 1,4-diketones **15:** Vinylmagnesium chloride (1.6 M in THF, 5.5 mL, 8.8 mmol) was added to a stirred solution of **9** (1.24 g, 3.97 mmol) in THF (20 mL). After 10 min., the reaction was pored into water (100 mL) and further diluted with 1 M HCl (50 mL). The resulting mixture was extracted with dichloromethane (3 × 20 mL). The combined organic extracts were washed with a saturated solution

NaHCO3 (30 mL) and water (30 mL), dried over MgSO4, filtered and concentrated under reduced pressure. The pale yellow residue was dissolved in dichloromethane (220 mL), heated to 40 °C, followed by the addition of Hoveyda-Grubbs second-generation catalyst (0.062 g, 0.099 mmol). After 1 h, the reaction mixture was concentrated under reduced pressure. The dark brown residue was dissolved in 1:9 methanol/dichloromethane (30 mL) and sodium borohydride (0.619 g, 15.9 mmol) was added. After 3 h, the reaction was poured into water (50 mL) and further diluted with 1 M HCl (20 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 20 mL). The combined organic extracts were washed with water (20 mL), dried over MgSO<sub>4</sub>, filtered and concentrated under reduced pressure. The dark brown residue was dissolved in dichloromethane (30 mL), followed by the sequential addition of NaHCO<sub>3</sub> (0.733 g, 8.73 mmol) and Dess-Martin periodinane (3.37 g, 7.89 mmol). After 30 min., the reaction was poured into water (50 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 25 mL). The combined organic extracts were washed with water (30 mL), dried over MgSO<sub>4</sub>, filtered and concentrated under The residue was purified by flash chromatography (12 cm × 2.5 cm; 3:7 reduced pressure. EtOAc/hexanes) to afford 1,4-diketone 15 as a white solid (0.885 g, 66% from 9):  $R_f = 0.27$  (1:4 EtOAc/hexane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.49 (ddd, J = 7.7, 1.7, 1.0 Hz, 2H), 7.39-7.34 (m, 2H), 7.30 (dd, J = 2.5, 1.6 Hz, 2H), 7.07 (ddd, J = 8.2, 2.5, 1.0 Hz, 2H), 4.11 (t, J = 6.2 Hz, 4H), 3.21 (s, 4H), 1.84 (p, J = 6.3 Hz, 4H), 1.75-1.66 (m, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  200.10, 159.05, 137.72, 130.32, 120.70, 119.58, 115.59, 68.01, 36.07, 27.93, 21.98; HRMS (ESI) calculated for C<sub>21</sub>H<sub>23</sub>O<sub>4</sub> ([M+H]<sup>+</sup>) m/z = 339.1596, found 339.1598.



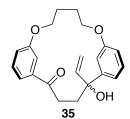
Streamlined synthesis of 1,4-diketones **16:** Vinylmagnesium chloride (1.6 M in THF, 4.8 mL, 7.7 mmol) was added to a stirred solution of **10** (1.19 g, 3.65 mmol) in THF (20 mL). After 10 min., the reaction was poured into water (100 mL) and further diluted with 1 M HCl (50 mL). The resulting mixture was extracted with dichloromethane (3 × 20 mL). The combined organic extracts were washed with a saturated solution of

NaHCO3 (40 mL) and water (40 mL), dried over MgSO4, filtered and concentrated under reduced pressure. The pale yellow residue was dissolved in dichloromethane (240 mL), heated to 40 °C, followed by the addition of Hoveyda-Grubbs second-generation catalyst (0.067 g, 0.107 mmol). After 1 h, the reaction mixture was concentrated under reduced pressure. The dark brown residue was dissolved in 1:9 methanol/dichloromethane (36 mL), and sodium borohydride (0.652 g, 17.2 mmol) was added. After 3 h, the reaction was poured into water (50 mL) and further diluted with 1 M HCl (30 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 20 mL). The combined organic extracts were washed with water (20 mL), dried over MgSO<sub>4</sub>, filtered and concentrated under reduced pressure. The dark brown residue was dissolved in dichlormethane (30 mL), followed by the sequential addition of NaHCO<sub>3</sub> (0.613 g, 7.30 mmol) and Dess-Martin periodinane (3.09 g, 7.30 mmol). After 30 min., the reaction was poured into water (50 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 25 mL). The combined organic extracts were washed with water (30 mL), dried over MgSO4, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography (12 cm × 2.5 cm; 3:7 EtOAc/hexanes) to afford 1,4-diketone 16 as a white solid (0.681 g, 53% from 10):  $R_f = 0.42$  (2:3 EtOAc/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.55 (ddd, *J* = 7.7, 1.7, 1.0 Hz, 2H), 7.43-7.35 (m, 4H), 7.09 (ddd, J = 8.2, 2.5, 1.0 Hz, 2H), 4.08 (t, J = 5.8 Hz, 4H), 3.20 (s, 4H), 1.87-1.77 (m, 4H), 1.67-1.59 (m, 4H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 199.67, 159.33, 137.50, 130.36, 120.52, 118.54, 116.23, 67.91, 35.80, 28.24, 25.29; HRMS (ESI) calculated for  $C_{22}H_{25}O_4$  ([M+H]<sup>+</sup>) m/z = 353.1753, found 353.1753.



Allylic alcohol **19:** 1,4-diketone **12** (0.298 g, 0.925 mmol), as a solution in THF (7.5 mL) was added to a stirred 65 °C solution of vinylmagnesium chloride (1.6 M in THF, 1.8 mL, 2.8 mmol). After 1 min., the reaction mixture was poured into water (20 mL) and further diluted with 1 M HCl (20 mL). The resulting mixture was extracted with dichloromethane (3  $\times$  10 mL). The organic extracts were combined

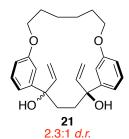
and washed with a saturated solution of NaHCO<sub>3</sub> (20 mL) and brine (20 mL), then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The solid residue was purified by flash chromatography (15 × 2.5 cm, 1:4 EtOAc/hexanes) to give hydroxyketone **35** (0.048 g, 15%) and allylic alcohol **19** (0.220 g, 63%; 77% based on recovered **35**) predominately as the *syn*-diastereomer (> 20:1 d.r.).



Hydroxy ketone **35**:  $R_f$  = 0.35 (1:4 EtOAc/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.53 (dt, J = 7.7, 1.3 Hz, 1H), 7.38-7.30 (m, 2H), 7.12 (ddd, J = 7.7, 1.7, 0.9 Hz, 1H), 7.04 (ddd, J = 8.2, 2.5, 1.0 Hz, 1H), 6.96-6.93 (m, 1H), 6.92-6.90 (m, 1H), 6.88 (dd, J = 2.5, 1.0 Hz, 1H), 6.86 (dd, J = 2.5, 1.0 Hz, 1H), 5.31 (dd, J = 17.3, 0.9 Hz, 1H), 5.15 (dd, J =

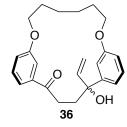
10.7, 0.8 Hz, 1H), 4.27-4.17 (m, 2H), 4.18-4.10 (m, 1H), 4.03-3.93 (m, 1H), 2.79-2.56 (m, 2H), 2.44-2.24 (m, 2H), 2.13-1.88 (m, 3H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 201.84, 158.68, 158.59, 145.39, 144.29, 137.62, 129.98, 129.74, 120.38, 119.80, 118.30, 115.98, 113.38, 113.34, 113.05, 77.07, 69.16, 67.06, 39.19, 33.75, 26.45, 25.80; HRMS (ESI) calculated for C<sub>22</sub>H<sub>23</sub>O<sub>3</sub> ([M-(H<sub>2</sub>O)+H]<sup>+</sup>) *m/z* = 335.1647, found 335.1647.

Allylic alcohol **19**:  $R_f$  = 0.22 (1:4 EtOAc/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.25-7.17 (m, 2H), 6.98-6.90 (m, 2H), 6.80 (ddd, J = 8.2, 2.5, 0.9 Hz, 2H), 6.72-6.64 (m, 2H), 6.19 (dd, J = 17.2, 10.7 Hz, 2H), 5.32 (dd, J = 17.3, 1.3 Hz, 2H), 5.16 (dd, J = 10.7, 1.3 Hz, 2H), 4.15-4.00 (m, 4H), 3.08 (s, 2H), 2.04-1.90 (m, 4H), 1.84-1.65 (m, 4H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  158.71, 146.45, 143.18, 143.16, 129.16, 129.14, 118.47, 113.54, 113.31, 113.29, 112.81, 76.86, 67.50, 36.77, 26.03; HRMS (ESI) calculated for C<sub>24</sub>H<sub>25</sub>O<sub>2</sub> ([M-(2H<sub>2</sub>O)+H]+) m/z = 345.1855, found 345.1868.



Allylic alcohol 21: 1,4-diketone 12 (0.560 g, 1.59 mmol), as a solution in THF (10 mL), was added to a stirred 65 °C solution vinylmagnesium chloride (1.6 M in

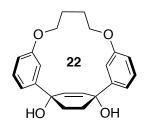
THF, 5.2 mL, 8.3 mmol). After 1 h, the reaction mixture was poured into water (100 mL) and further diluted with 1 M HCl (30 mL). The resulting mixture was extracted with dichloromethane (3 × 30 mL). The organic extracts were combined and washed with a saturated solution of NaHCO<sub>3</sub> (30 mL) and brine (30 mL), dried over anhydrous MgSO<sub>4</sub> and concentrated under reduced pressure. The solid residue was purified by flash chromatography (15 × 2.5 cm, 1:4 EtOAc/hexanes) to give hydroxyketone **36** (0.250 g, 41%) and allylic alcohol **21** (0.310 g, 47%; 86% based on recovery of **36**) as an inseparable mixture of diastereomers (2.3:1 d.r.):



*Hydroxy ketone* **36**:  $R_f$  = 0.33 (1:4 EtOAc/hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.58-7.52 (m, 1H), 7.37-7.32 (m, 1H), 7.29-7.24 (m, 1H), 7.21 (s, 1H), 7.12 (s, 1H), 7.06 (dd, J = 8.1, 2.4 Hz, 1H), 6.93 (dd, J = 7.7, 1.6 Hz, 1H), 6.79 (dd, J = 8.1, 2.6 Hz, 1H), 6.26 (dd, J = 17.3, 10.7 Hz, 1H), 5.36 (d, J = 17.4 Hz, 1H), 5.17 (d, J = 10.6 Hz, 1H), 4.12-3.96 (m, 4H), 2.95 (dt, J = 15.5, 7.6 Hz, 1H), 2.57 (dt, J = 15.1, 6.6 Hz, 1H), 2.45

(dt, J = 14.8, 7.6 Hz, 1H), 2.33-2.19 (m, 1H), 2.09 (s, 1H), 1.97-1.74 (m, 4H), 1.69-1.56 (m, 4H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  201.35, 159.19, 158.89, 145.49, 144.45, 138.18, 129.74, 129.45, 120.02, 118.67, 118.00, 116.16, 112.99, 112.76, 111.58, 76.61, 67.99, 66.45, 37.18, 33.02, 28.09, 27.90, 24.66, 24.33; HRMS (ESI) calculated for C<sub>24</sub>H<sub>29</sub>O<sub>4</sub> ([M+H]<sup>+</sup>) m/z = 381.2066, found 381.2075.

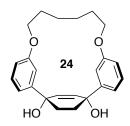
Allylic alcohol **21** (major isomer):  $R_f$  = 0.14 (1:4 EtOAc/hexanes), 0.59 (1:1 EtOAc/hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.31 (d, J = 8.0 Hz, 2H), 7.19 (dd, J = 7.7, 1.7 Hz, 2H), 6.80 (dd, J = 8.2, 2.4 Hz, 2H), 6.59-6.54 (m, 2H), 6.08 (dd, J = 17.3, 10.6 Hz, 2H), 5.18 (d, J = 17.3 Hz, 2H), 5.01 (d, J = 10.6 Hz, 2H), 4.11-4.03 (m, 2H), 3.98-3.94 (m, 2H), 2.51 (s, 2H), 2.02-1.94 (m, 2H), 1.93-1.86 (m, 2H), 1.83-1.76 (m, 4H), 1.73-1.56 (m, 4H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  159.2, 146.2, 145.7, 129.4, 117.4, 113.2, 112.3, 111.3, 76.8, 66.8, 35.6, 28.4, 24.7; HRMS (ESI) calculated for C<sub>26</sub>H<sub>33</sub>O<sub>4</sub> ([M+H]<sup>+</sup>) m/z = 409.2379, found 409.2380.



*Cyclohex-2-ene-1,4-diol* **22:** Grubbs' second-generation catalyst (0.023 g, 0.026 mmol) was added to a stirred solution of **19** (>20:1 *d.r.*; 0.201 g, 0.526 mmol) in dichloromethane (35 mL) and the reaction was heated to 40 °C. After 2 h, the solvent was removed under reduced pressure and residue was purified by flash

chromatography (15 × 2.5 cm, 1:1 EtOAc/hexanes) to give compound 22 as an off-white solid (0.159 g,

86%);  $R_f = 0.27$  (1:1 EtOAc/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.36-7.27 (m, 4H), 7.05-6.99 (m, 2H), 6.80 (ddd, J = 7.7, 2.5, 1.4 Hz, 2H), 6.08 (s, 2H), 4.26-4.11 (m, 2H), 4.06-3.93 (m, 2H), 2.17 (s, 2H), 2.14-1.98 (m, 4H), 1.97-1.78 (m, 4H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  158.64, 147.77, 134.96, 130.23, 117.59, 114.70, 113.81, 73.25, 69.81, 37.00, 26.98. HRMS (ESI) calculated for C<sub>22</sub>H<sub>23</sub>O<sub>3</sub> ([M-(H<sub>2</sub>O)+H]<sup>+</sup>) m/z = 335.1647, found 335.1641.

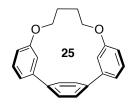


Cyclohex-2-ene-1,4-diol 22: Grubbs' second-generation catalyst (0.020 g, 0.023 mmol) was added to a stirred solution of 21 (2.3:1 *d.r.*; 0.380 g, 0.930 mmol) in dichloromethane (23 mL) and the reaction was heated to 40 °C. After 2 h, the solvent was removed under reduced pressure and residue was purified by flash

chromatography (15 x 2.5 cm, 3:7 EtOAc/hexanes) to give compound **24** as an off-white solid (0.225 g, 59%, 85% based on recovered *anti-***21**) and (uncyclized) *anti-***21** (0.106 g, 92% recovery).

anti-21:  $R_f$  = 0.33 (1:4 EtOAc/hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.30-7.23 (m, 2H), 7.15-7.10 (m, 2H), 6.79-6.73 (m, 2H), 6.58-6.52 (m, 2H), 6.06 (dd, J = 17.2, 10.6 Hz, 2H), 5.15 (dd, J = 17.2, 1.0 Hz, 2H), 5.00 (dd, J = 10.6, 1.0 Hz, 2H), 4.03 (dt, J = 9.3, 4.8 Hz, 2H), 3.94 (td, J = 9.2, 3.9 Hz, 2H), 2.00 (s, 2H), 1.95-1.88 (m, 2H), 1.88-1.80 (m, 2H), 1.79-1.72 (m, 1H), 1.69-1.50 (m, 6H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  159.15, 146.18, 145.66, 129.43, 117.42, 113.21, 112.31, 111.30, 76.80, 66.75, 35.64, 28.38, 24.71; HRMS (ESI) calculated for C<sub>26</sub>H<sub>33</sub>O<sub>4</sub> ([M+H]<sup>+</sup>) m/z = 409.2379, found 409.2372.

Cyclohex-2-ene-1,4-diol **22:**  $R_f$ = 0.29 (1:1 EtOAc/hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.32-7.26 (m, 4H), 6.93-6.88 (m, 2H), 6.80-6.74 (m, 2H), 5.96 (s, 2H), 4.02-3.94 (m, 4H), 2.54 (s, 2H), 2.08-2.03 (m, 2H), 1.86-1.74 (m, 6H), 1.63-1.55 (m, 4H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  158.83, 147.84, 134.66, 130.13, 117.49, 113.76, 112.62, 72.98, 67.62, 36.33, 27.78, 24.63; HRMS (ESI) calculated for C<sub>24</sub>H<sub>27</sub>O<sub>3</sub> ([(M-H<sub>2</sub>O)+H]<sup>+</sup>) m/z = 363.1960, found 363.1968.



1,6-dioxa[6](3,3')-p-Terphenylenophane (25): p-Toluene sulfonic acid monohydrate (0.130 g, 0.684 mmol) was added to a stirred solution of 22 (0.040 g, 0.11 mmol) in toluene (6 mL). The reaction was heated at 50 °C for 10 h and then to 60 °C for 5 h. After 15 h, a saturated solution of NaHCO<sub>3</sub> (20 mL) was added. The layers were

were combined and washed with brine (20 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (15 × 1.3 cm, 5% EtOAc/hexanes) to afford **25** as a white solid (0.015 g, 42%):  $R_f$  = 0.43 (5% EtOAc/hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.37 (s, 4H), 7.36-7.32 (m, 2H), 7.26-7.24 (m, 2H), 6.78 (dd, J = 8.4, 2.7 Hz, 2H), 5.31 (d, J = 2.9 Hz, 2H), 3.95-3.89 (s, 4H), 1.46-1.40 (m, 4H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  156.29, 144.82, 144.55, 130.30, 117.70, 115.81, 115.69, 67.29, 22.77; HRMS (EI) calculated for C<sub>22</sub>H<sub>21</sub>O<sub>2</sub> (M<sup>+</sup>) m/z = 316.1463, found 316.1437.

Alternate procedure for **25** (**Table 1, entry 4**): Sodium hydrogen sulfate monohydrate (0.008 g, 0.06 mmol) was added to a stirred 130 °C solution of **22** (0.010 g, 0.028 mmol) and *o*-chloranil (0.035 g, 0.14 mmol) in DMSO (0.75 mL) and xylenes (2 mL). After 24 h, the reaction mixture was cooled to room temperature and a saturated solution of NaHCO<sub>3</sub> (10 mL) and dichloromethane (10 mL) were added. The layers were separated and the aqueous phase was extracted with dichloromethane (2 × 10 mL). The corganic extracts were combined, filtered through a pad of Celite (2 cm), and the filtrate was concentrated under reduced pressure. The residue was purified by flash chromatography (7.5 × 0.6 cm; 5% EtOAc/hexanes) to afford the **25** as a white solid (0.0032 g, 36%). A trace amount of the [6]MTPP isomer was observed in the <sup>1</sup>H NMR spectrum of **25**.

Alternate procedure for **25** (**Table 1**, **entry 10**): Burgess reagent (0.021 g, 0.088 mmol) was added to a stirred solution of **22** (0.010 g, 0.028 mmol) in toluene (2 mL) at 80 °C. After 15 min., the reaction was cooled to room temperature, water (10 mL) was added, and the resulting mixture was stirred for 5 min. The layers were separated and the mixture was extracted with dichloromethane (3 × 5 mL). The organic extracts were combined and washed with brine (10 mL), dried over anhydrous MgSO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (12 × 1.3 cm, 1:1 dichloromethane/hexanes) to afford **25** as a white solid (0.005 g, 56%).



1,7-dioxa[7](3,3')-p-Terphenylenophane (**26**) (**Table 1, entry 11**): Burgess reagent (0.050 g, 0.21 mmol) was added to a stirred solution of **23** (0.026 g, 0.071 mmol) in toluene (3 mL) at 80 °C. After 15 min., the reaction was cooled to room temperature, water (10 mL) was added, and the resulting mixture was stirred for 5 min. The layers were separated and the mixture was extracted with

dichloromethane (3  $\times$  10 mL). The organic extracts were combined and washed with brine (10 mL),

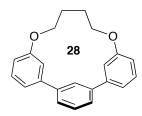
dried over anhydrous MgSO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (12 × 1.3 cm, 1:1 dichloromethane/hexanes) to afford **26** as a white solid (0.016 g, 68%):  $R_f$  = 0.32 (1:19 EtOAc/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.44 (s, 4H), 7.35 (dd, J = 8.2, 7.4 Hz, 2H), 7.30-7.24 (m, 2H), 6.78 (ddd, J = 8.3, 2.8, 1.0 Hz, 2H), 5.81 (dd, J = 2.8, 1.5 Hz, 2H), 4.10-4.05 (m, 4H), 1.51-1.42 (m, 4H), 1.21-1.12 (m, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  157.2, 144.7, 144.1, 130.6, 129.5, 118.7, 115.9, 115.4, 68.5, 26.8, 23.3; HRMS (EI) calculated for C<sub>23</sub>H<sub>22</sub>O<sub>2</sub> ([M]<sup>+</sup>) m/z = 330.1618, found, 330.1620.



1,8-dioxa[8](3,3")-p-Terphenylenophane (27): p-Toluene sulfonic acid monohydrate (0.502 g, 2.92 mmol) was added to a stirred solution of 24 (0.184 g, 0.484 mmol) in toluene (20 mL) and the reaction was heated to 60 °C. After 2 h, a saturated solution of NaHCO<sub>3</sub> (20 mL) was added to the reaction. The layers were separated

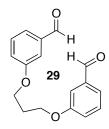
and the aqueous phase was extracted with dichloromethane (3 x 15 mL). The organic extracts were combined and washed with brine (30 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (15 x 1.3 cm, 1:1 dichloromethane/hexanes) to afford **27** as a white solid (0.120 g, 74%):  $R_f = 0.41$  (1:1 dichloromethane/hexanes); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.39 (s, 4H), 7.36 (dd, J = 8.3, 7.4 Hz, 2H), 7.26-7.20 (m, 2H), 6.84 (ddd, J = 8.3, 2.7, 0.9 Hz, 2H), 5.92 (dd, J = 2.8, 1.4 Hz, 2H), 4.08-3.99 (m, 4H), 1.62-1.50 (m, 4H), 1.12-1.04 (m, 4H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  156.86, 144.54, 143.59, 130.21, 128.76, 117.46, 116.63, 115.88, 68.42, 27.81, 27.63; HRMS (EI) calculated for C<sub>24</sub>H<sub>24</sub>O<sub>2</sub> ([M]<sup>+</sup>) m/z = 344.1931 found 344.1896.

Alternate procedure for 28 (Table 1, entry 13): Burgess reagent (0.058 g, 0.24 mmol) was added to a stirred solution of 24 (0.021 g, 0.082 mmol) in toluene (3 mL) at 80 °C. After 15 min., the reaction was cooled to room temperature, water (10 mL) was added, and the resulting mixture was stirred for 5 min. The layers were separated and the mixture was extracted with dichloromethane (3 × 10 mL). The organic extracts were combined and washed with brine (10 mL), dried over anhydrous MgSO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (12 × 1.3 cm, 1:1 dichloromethane/hexanes) to afford 25 as a white solid (0.017 g, 60%).



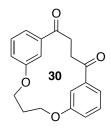
1,6-dioxa[6](3,3")m-Terphenylophane (28): para-Toluensulfonic acid monohydrate (0.033 g, 0.17 mmol) was added to a stirred 70 °C of 25 (0.011 g, 0.035 mmol) in toluene (2 mL). After 12 h, a saturated solution of NaHCO<sub>3</sub> (10 mL) was added to the reaction. The layers were separated and the aqueous phase was extracted

with dichloromethane (3 × 5 mL). The organic extracts were combined and washed with brine (10 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (7 × 0.5 cm, 1:1 dichloromethane/hexanes) to afford **28** as a white solid (0.060 g, 55%):  $R_f$  = 0.31 (5% EtOAc/hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  8.33-8.27 (m, 1H), 7.66 (dd, J = 7.6, 2.0 Hz, 2H), 7.52-7.45 (m, 1H), 7.40-7.34 (m, 4H), 7.34-7.30 (m, 2H), 6.95-6.89 (m, 2H), 4.36-4.22 (m, 4H), 2.08 (t, J = 4.2 Hz, 4H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  158.73, 141.71, 140.49, 131.38, 130.47, 129.34, 124.04, 117.44, 116.85, 114.30, 69.01, 24.25; HRMS (EI) calculated for C<sub>22</sub>H<sub>21</sub>O<sub>2</sub> (M<sup>+</sup>) m/z = 316.1463, found 316.1442.



*Dialdehyde* **29:** 1,3-Dibromopropane (2.76 g, 13.6 mmol) was added to a stirred solution of 3-hydroxybenzaldehyde (5.00 g, 40.9 mmol), K<sub>2</sub>CO<sub>3</sub> (6.50 g, 47.1 mmol) and TBAI (0.375 g, 1.01 mmol) in DMF (75 mL). The reaction was heated at 70 °C for 15 h, at which point water (100 mL) and 1 M HCl (50 mL) were added sequentially. The resulting mixture was extracted with ethyl acetate (3 × 50 mL). The organic extracts

were combined and washed with saturated solution of NaHCO<sub>3</sub> (100 mL) and brine (100 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. The residue was purified via flash chromatography (16 cm × 5.0 cm; dichloromethane, and 2% acetone/dichloromethane) to afford **2** as colorless oil. (2.80 g, 72%):  $R_f$  = 0.27 (dichloromethane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  9.80 (s, 2H), 7.30-7.28 (m, 4H), 7.25-7.23 (m, 2H), 7.06-6.98 (m, 2H), 4.07 (t, J = 6.1 Hz, 4H), 2.15 (p, J = 6.1 Hz, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  192.17, 159.44, 137.88, 130.19, 123.70, 121.97, 112.82, 64.60, 29.14; HRMS (ESI) calculated for C<sub>17</sub>H<sub>17</sub>O<sub>4</sub> ([M+H]<sup>+</sup>) m/z = 285.1127, found 285.1124.



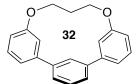
Streamlined synthesis of macrocyclic diketone **30**: Vinylmagnesium chloride (1.6 M in THF, 2.5 mL, 4.0 mmol) was added to a stirred 0 °C solution of dialdehyde **2** (0.500 g, 1.76 mmol) in THF (10 mL). After 30 min, the reaction mixture was poured into water (50 mL) and further diluted with 1 M HCl (30 mL). The resulting mixture was

extracted with dichloromethane (3 × 15 mL). The combined organic extracts were washed with a saturated solution of NaHCO3 (30 mL) and brine (30 mL), dried over anhydrous Na2SO4, and concentrated under reduced pressure. The pale yellow residue was dissolved in dichloromethane (150 mL), stirred and heated to 40 °C, followed by the addition of Hoveyda-Grubbs second-generation catalyst (0.040 g, 0.060 mmol). After 2 h, the reaction mixture was cooled to room temperature and concentrated under reduced pressure. The residue was dissolved in 1:9 methanol/dichloromethane (15 mL) and sodium borohydride (0.230 g, 5.88 mmol) was added. After 1 h, the reaction mixture was poured into water (100 mL) and the layers were separated. The aqueous phase was extracted with dichloromethane (2 × 20 mL) and the combined organic extracts were washed with water (20 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The residue was dissolved in dichlormethane (20 mL), followed by the sequential addition of NaHCO<sub>3</sub> (0.270 g, 3.21 mmol) and Dess-Martin periodinane (1.45 g, 3.20 mmol). After 2 h, a 10% aqueous solution of Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (50 mL) was added and stirring was continued for 10 min. The resulting mixture was extracted with dichloromethane (3 × 20 mL). The organic extracts were combined and washed with water (50 mL) and brine (50 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. The residue was purified via flash chromatography (15 cm × 2.5 cm; 3:7 ethyl acetate/hexane) to give 1,4diketone 30 as a beige solid (0.120 g, 22%, over 4 steps):  $R_f = 0.38$  (2:3 ethyl acetate/hexane); <sup>1</sup>H NMR  $(600 \text{ MHz}, \text{CDCl}_3) \delta 7.42 \text{ (dd, } I = 7.7, 1.3 \text{ Hz}, 2\text{H}), 7.36-7.29 \text{ (m, 2H)}, 7.24-7.20 \text{ (m, 2H)}, 7.10 \text{ (ddd, } I = 8.1, 1.3 \text{ Hz})$ 2.6, 1.0 Hz, 2H), 4.34 (t, *J* = 6.1 Hz, 4H), 3.11 (s, 4H), 2.19 (p, *J* = 6.1 Hz, 2H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 200.85, 159.14, 138.03, 130.43, 121.57, 121.39, 117.10, 66.38, 36.42, 29.09.

Cyclohex-2-ene-1,4-diol 31: Vinylmagnesium chloride (1.6 M in THF, 0.45 mL, 0.70 mmol) was added to a stirred solution of 1,4-diketone 30 (0.100 g, 0.322 mmol), in THF (4 mL) at 65 °C. After 30 min., the reaction mixture was cooled to room temperature, poured into water (20 mL), and further diluted with 1 M HCl (10 mL). The resulting mixture was extracted with dichloromethane (3 × 10 mL). The combined organic extracts were washed with a saturated solution of NaHCO<sub>3</sub> (30 mL) and brine (30 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The pale yellow residue was dissolved in dichloromethane (10 mL), the Grubbs second-generation catalyst (0.007g, 0.008 mmol) was added, and the reaction was

heated to 40 °C. After 2 h, the reaction was cooled to room temperature and the solvent was removed

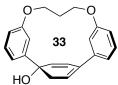
under reduced pressure. The brown residue was purified by flash chromatography (15 × 1.3 cm, 3:2 EtOAc/hexane) to give compound 37 as an off-white solid (0.065 g, 60%);  $R_f = 0.41$  (7:3 EtOAc/hexane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.36-7.30 (m, 4H), 7.03-6.99 (m, 2H), 6.92-6.84 (m, 2H), 6.14 (s, 2H), 4.46-4.30 (m, 2H), 4.29-4.23 (m, 2H), 2.29-2.08 (m, 6H), 1.84-1.71 (m, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 158.88, 146.95, 142.95, 129.36, 119.04, 115.62, 114.07, 113.62, 76.93, 65.44, 37.24, 27.78; HRMS (ESI) calculated for  $C_{21}H_{21}O_3([M-(H_2O)]+H]^+)$  m/z = 321.1491, found 321.1493.



*1,5-dioxa*[*5*](*3,3*")*m-Terphenylophane* (**32**): para-Toluensulfonic acid monohydrate (0.060 g, 0.31 mmol) was added to a stirred 60 °C of 31 (0.012 g, 0.038 mmol) in toluene (2.5 mL). After 3 h, the reaction was heated to 70 °C for an

Tin(II) chloride dihydrate (0.053 g, 0.230 mmol) was added to a stirred solution of

additional 1 h, followed by the addition of a saturated solution of NaHCO<sub>3</sub> (10 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (3 × 5 mL). The organic extracts were combined and washed with brine (10 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by flash chromatography (7 × 0.5 cm, 1:1 dichloromethane/hexanes) to afford 32 as a white solid (0.004 g, 40%):  $R_f = 0.41$  (2:3 dichloromethane /hexanes); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.62 (dd, J = 7.5, 2.0 Hz, 2H), 7.50 (d, J = 2.1 Hz, 1H), 7.46-7.40 (m, 1H), 7.35-7.29 (m, 2H), 7.28 (s, 1H), 7.07-7.02 (m, 2H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 7.07-7.02 (m, 2H), 7.35-7.29 (m, 2H), 7.28 (s, 1H), 7.07-7.02 (m, 2H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 7.07-7.02 (m, 2H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 7.07-7.02 (m, 2H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, <math>J = 8.2, 2.8 Hz, 2H), 4.36-4.19 (m, 4H), 6.92 (dd, J = 8.2, 2.8 Hz, 2H), 6.92 (dd, J = 8.2, 2.8 Hz)2.65 (s, 2H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 157.24, 144.91, 142.84, 142.64, 129.70, 127.63, 123.74, 118.28, 117.98, 116.15, 64.23, 24.77; HRMS (EI) calculated for  $C_{21}H_{18}O_2$  (M<sup>+</sup>) m/z = 302.1307, found 302.1336.



31 (0.008 g, 0.023 mmol) in 1:1 THF/PhMe (4 mL) at 80 °C. After 12 h, the reaction was cooled to room temperature and 3 M NaOH (5 mL) was added, followed by dichloromethane (10 mL). The layers were separated and the aqueous phase was extracted with dichloromethane (3 × 5 mL). The combined organic extracts was washed with brine (30 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The off-white residue was purified by flash chromatography (4.0 × 0.7 cm, dichloromethane to 2% acetone/dichloromethane) to give compound 33 as colorless solid (0.006 g, 78%);  $R_f = 0.41$  (1% acetone/dichloromethane); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.41 (dd, J = 2.5, 1.7 Hz, 1H), 7.34-7.24 (m, 2H), 7.22-7.15 (m, 1H), 6.91 (ddd, J = 7.8, 2.6, 1.2 Hz, 1H), 6.84 - 6.80 (m, 1H), 6.79 - 6.72 (m, 2H), 6.36 (dd, J = 9.6, 0.9 Hz, 1H), 6.12 (dt, J = 9.6, 1.3 Hz, 1H), 5.66 (ddd, J = 7.0, 2.8, 1.4 Hz, 1H), 4.52-4.40 (m, 1H), 4.41-4.15 (m, 3H), 2.97-2.84 (m, 1H), 2.71 (ddd,

I = 16.0, 7.0, 1.9 Hz, 1H), 2.16 (s, 1H), 2.06-1.91 (m, 1H), 1.88-1.76 (m, 1H); 13C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$ 156.98, 156.95, 145.10, 142.78, 139.09, 135.73, 130.31, 130.20, 129.11, 124.46, 120.22, 117.53, 116.56, 116.36, 116.05, 111.99, 75.50, 65.96, 64.25, 39.18, 26.69; HRMS (EI) calculated for  $C_{21}H_{20}O_3$  (M<sup>+</sup>) m/z = 320.1412, found 320.1410

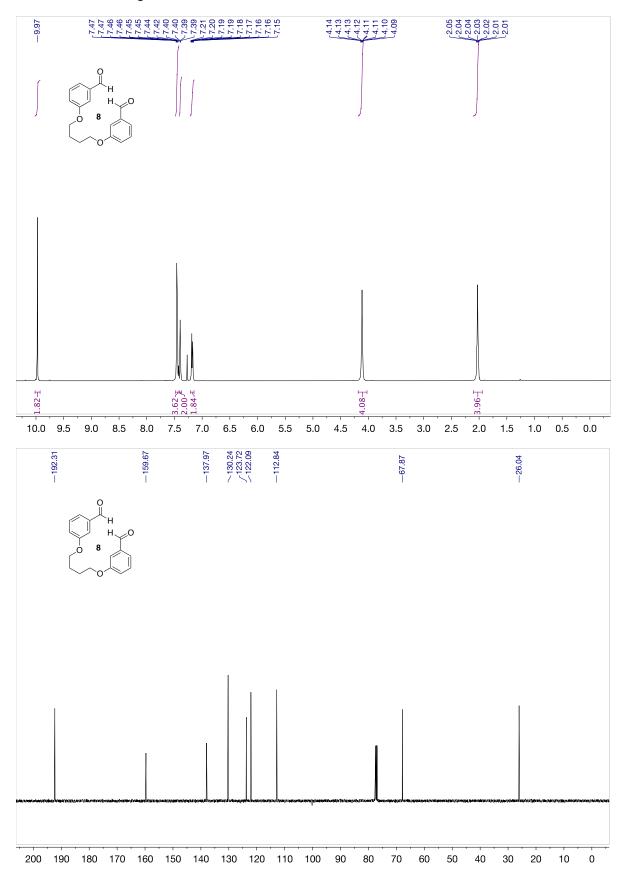


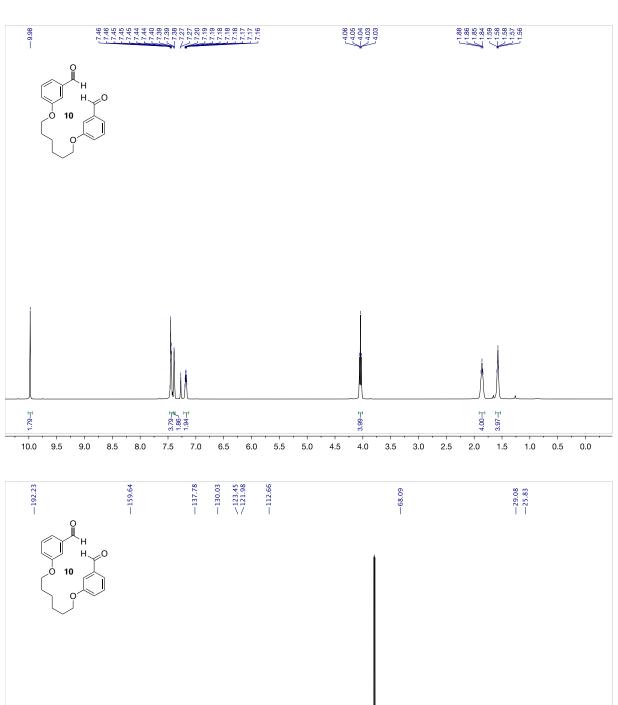
for  $C_{21}H_{18}O_2$  (M+) m/z = 302.1307, found 302.1336.

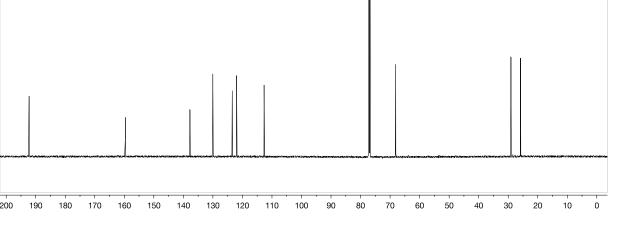
1,5-dioxa[5](3,3")p-terphenylenophane (34) (Table 1, entry 9): Burgess reagent (0.126) g, 0.530 mmol) was added to a stirred solution of 31 (0.044 g, 0.130 mmol) in toluene (2.5 mL) at 80 °C. After 15 min., the reaction was cooled to room temperature, water (20 mL) was added, and the resulting mixture was stirred for 5 min. The layers were separated and the mixture was extracted with dichloromethane (3 × 10 mL). The organic extracts were combined and washed with brine (10 mL), dried over anhydrous MgSO4 and concentrated under reduced pressure. The residue was purified by flash chromatography (12 × 1.3 cm, 1:1 dichloromethane/hexanes) to afford 34 as a white solid (0.023 g, 58%):  $R_f = 0.43$  (2:3 dichloromethane/hexane); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.42 (s, 4H), 7.33-7.26 (m, 2H), 7.25-7.21 (m, 2H), 6.72 (dd, J = 8.3, 2.8 Hz, 2H), 5.35 (s, 2H), 4.12-3.81 (m, 4H), 1.96-1.86 (m, 2H);  ${}^{13}$ C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  157.25, 145.19, 145.13, 131.75, 130.53, 118.50, 115.48, 115.42, 64.67, 25.16; HRMS (EI) calculated

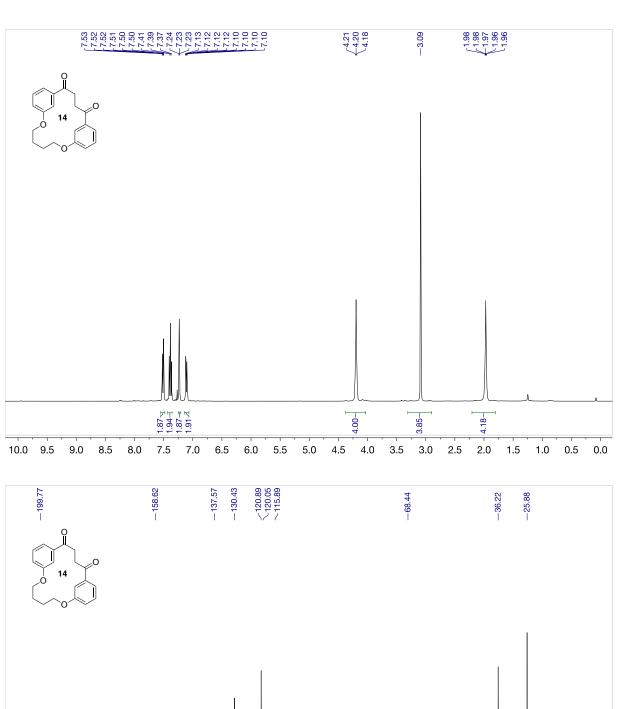
Alternate procedure for 34 (Table 1 entry 8): Trifluoromethanesulfonic anhydride 0.088 g, 0.31 mmol) and pyridine (0.5 mL) were added to a stirred solution of 33 (0.020 g, 0.062 mmol) in dichloromethane (2 mL) at 0 °C. After 30 min., the cooling bath was removed and the reaction was warmed to room temperature, poured into water (10 mL), and further diluted by 1 M HCl (5 mL). The layers were separated and the aqueous layer was extracted with dichloromethane (2 × 10 mL). The combined organic extracts were washed with saturated solution of NaHCO<sub>3</sub> (10 mL), brine (10 mL), dried over anhydrous MgSO4 and concentrated under reduced pressure. The residue was purified by flash chromatograph (12 × 0.5 cm, 2:3 dichloromethane/hexanes) to afford 34 as a white solid (0.003 g, 16%).

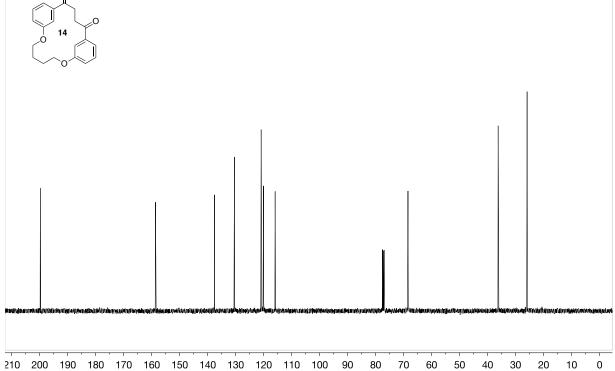
<sup>1</sup>H and <sup>13</sup>C NMR Spectra (listed in numerical order)

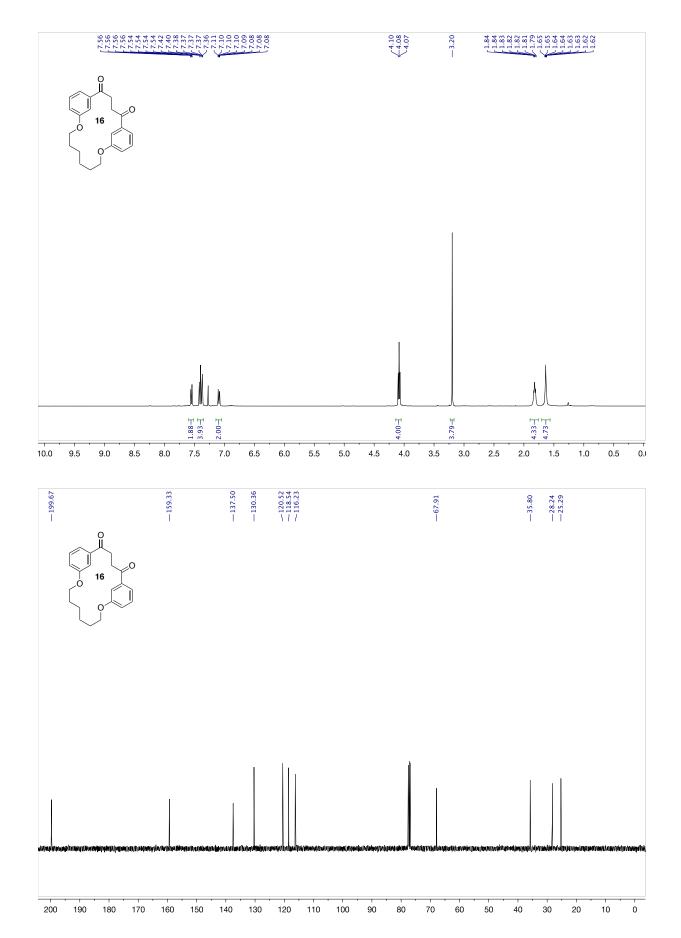


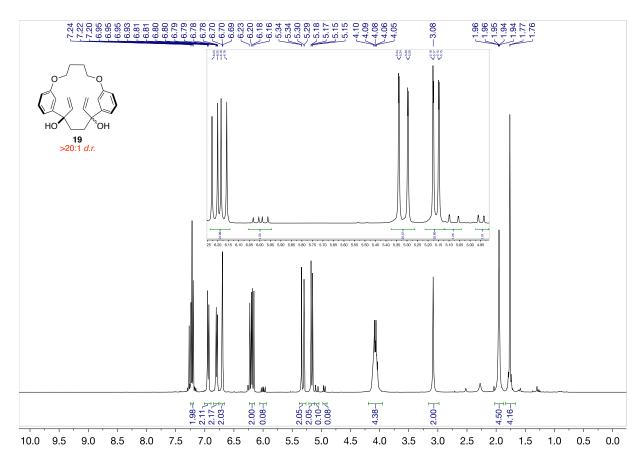


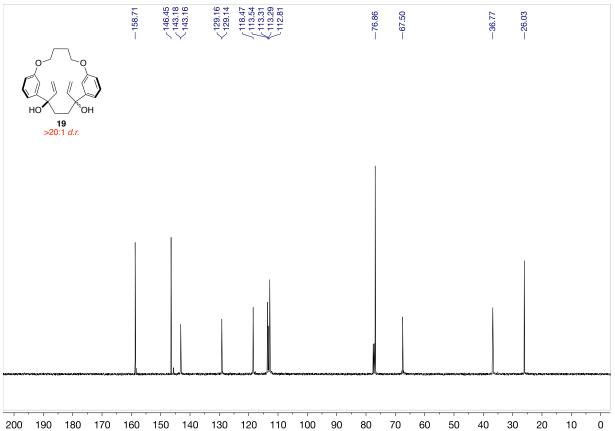


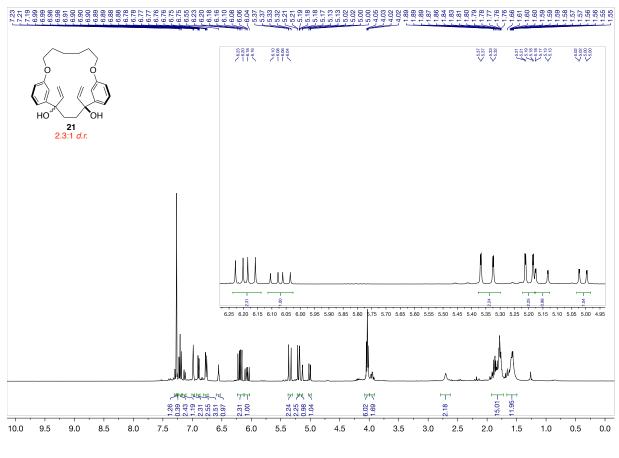


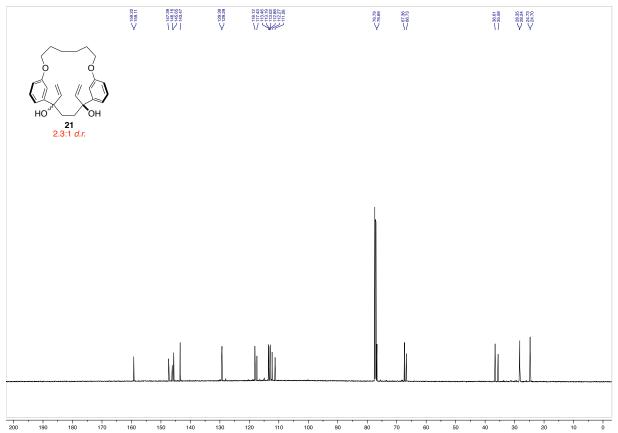


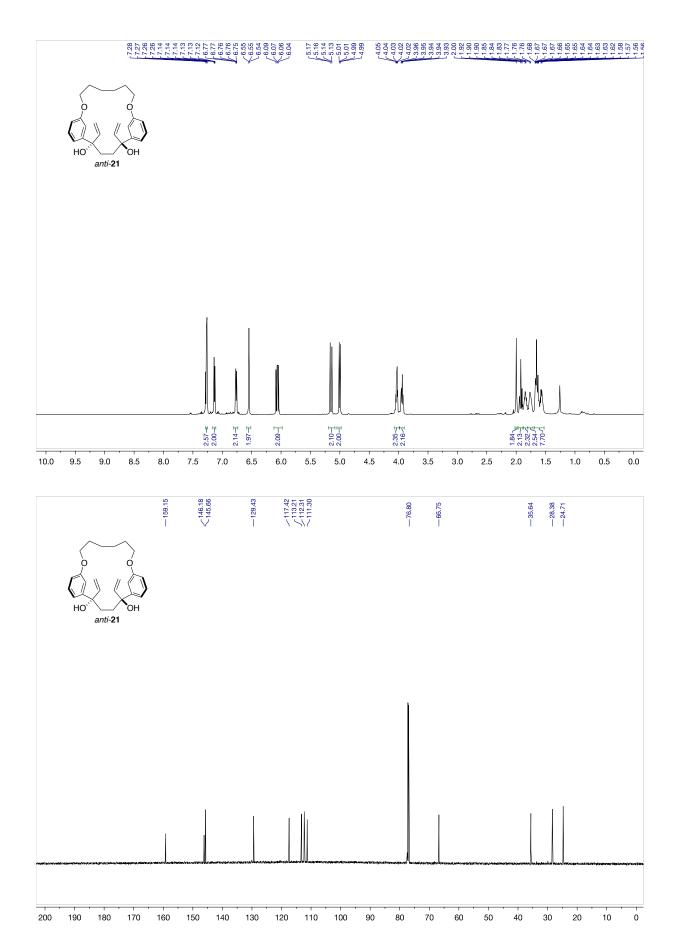


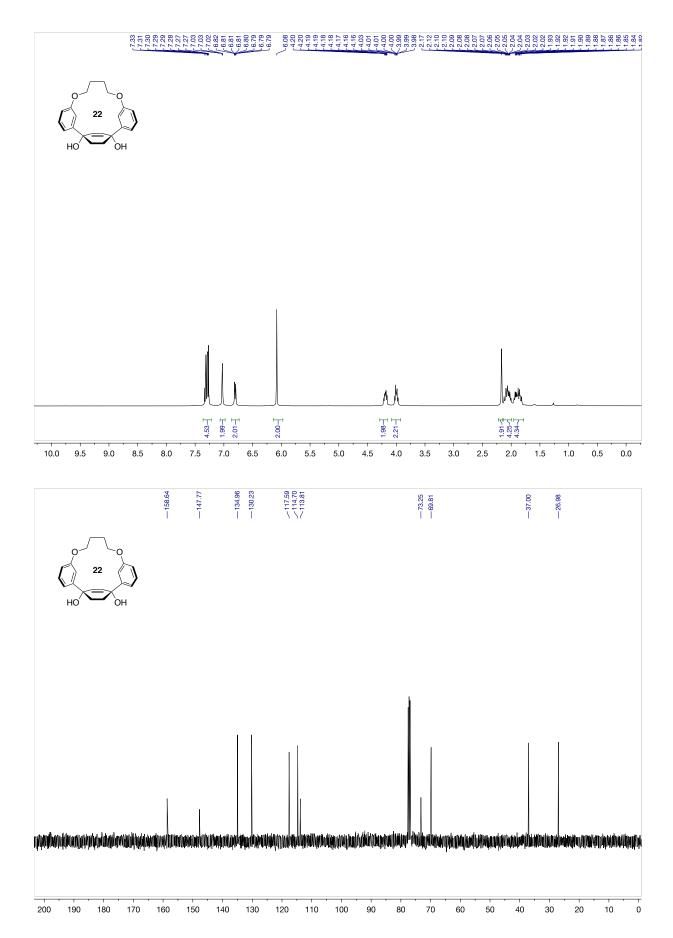


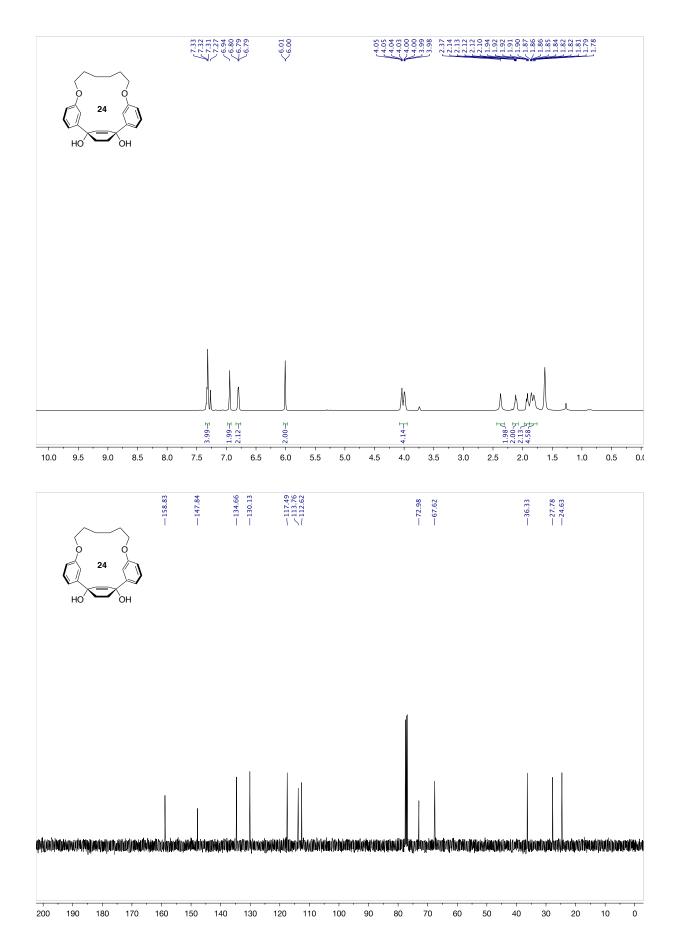


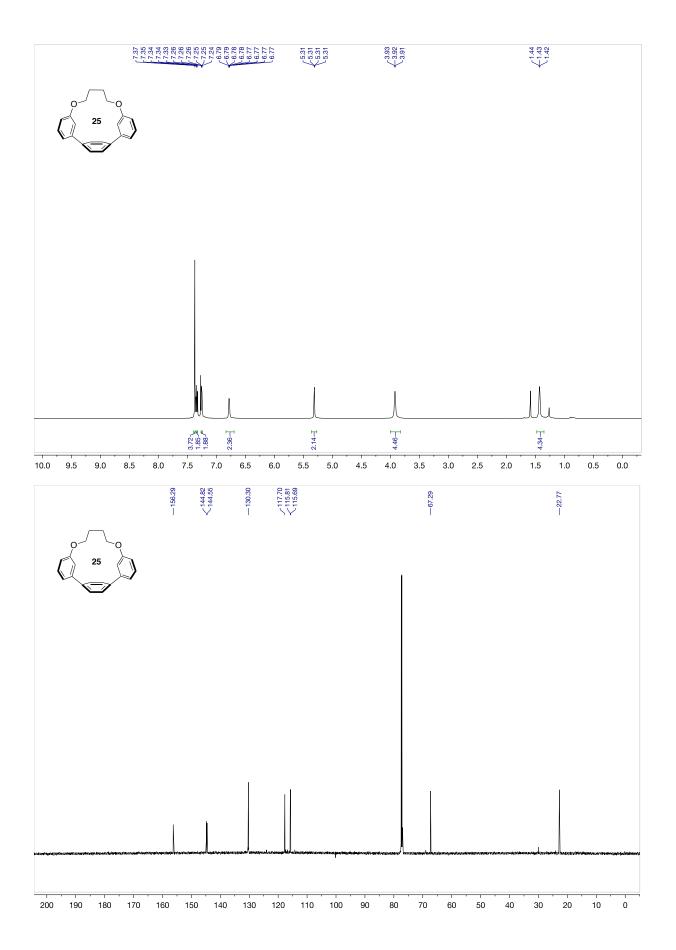


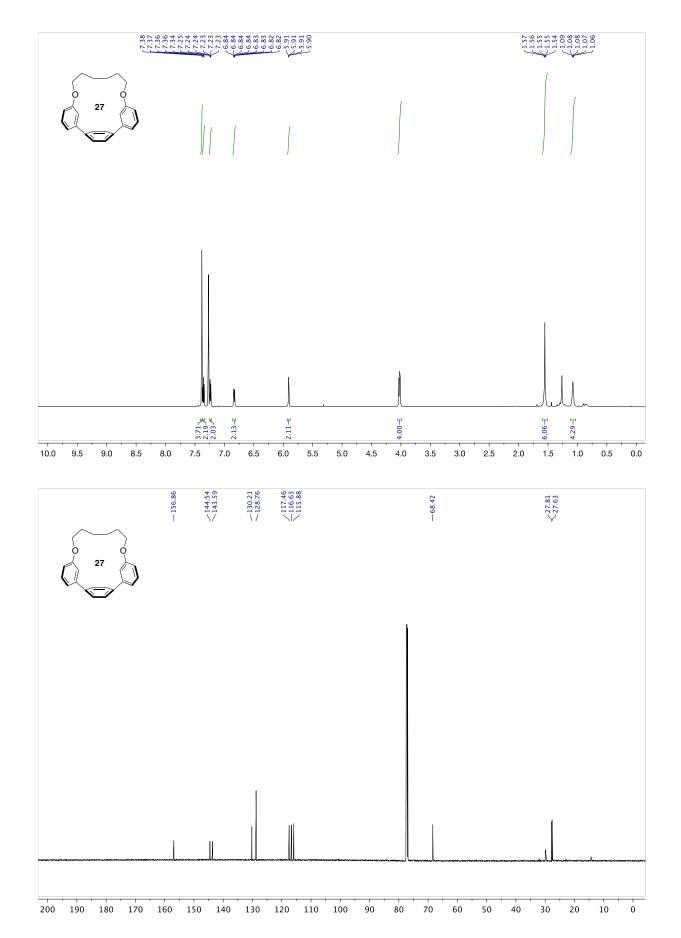


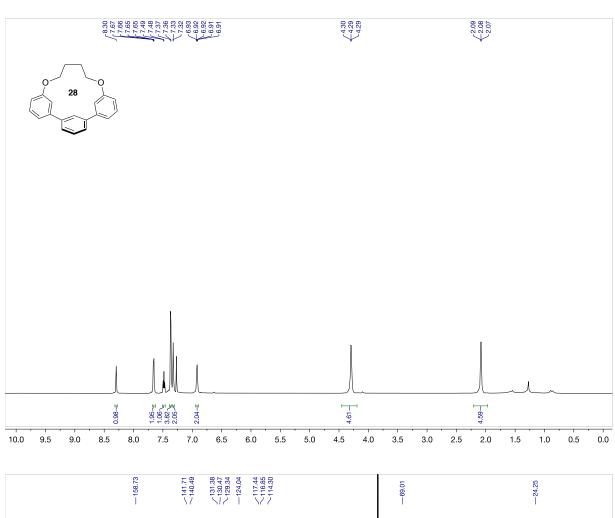


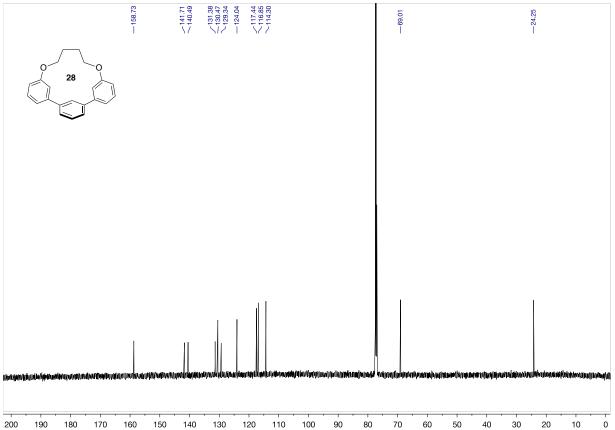


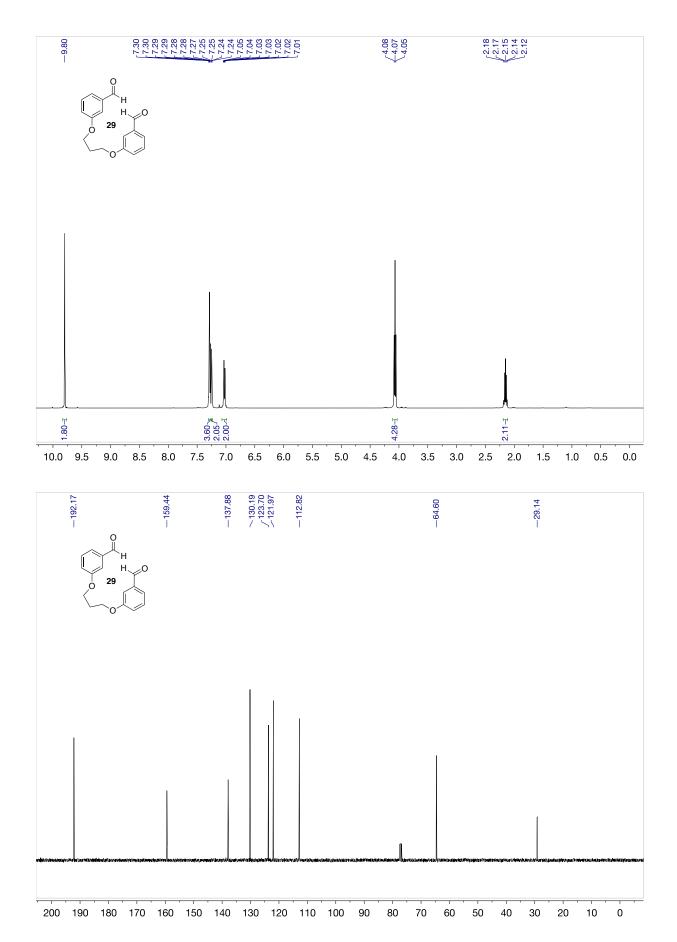


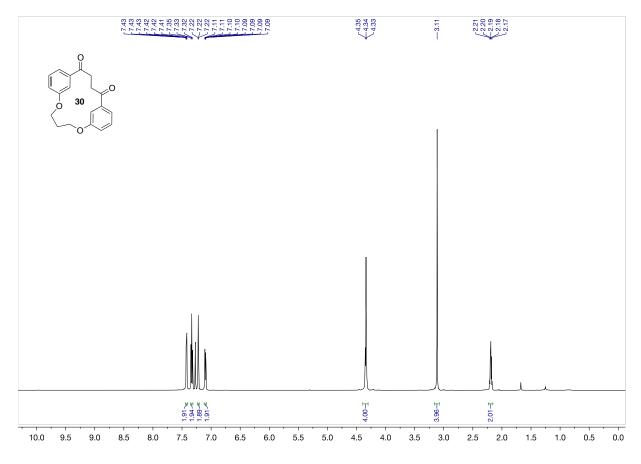


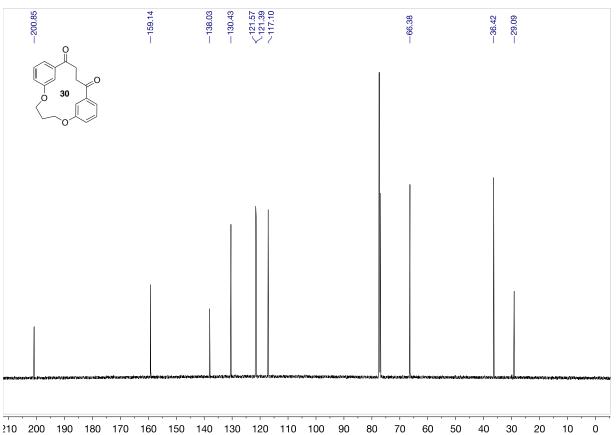


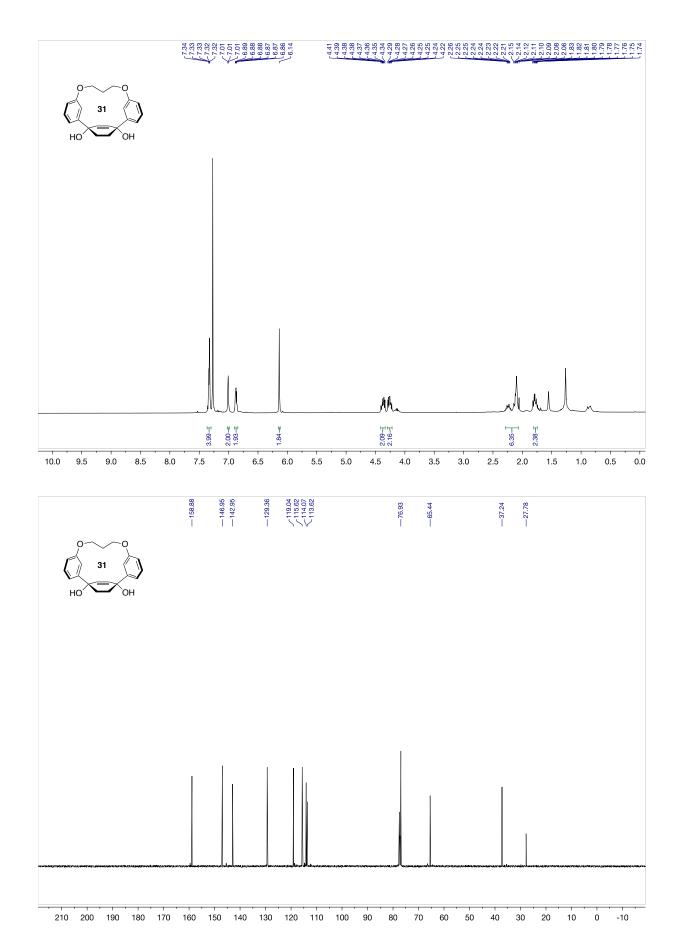


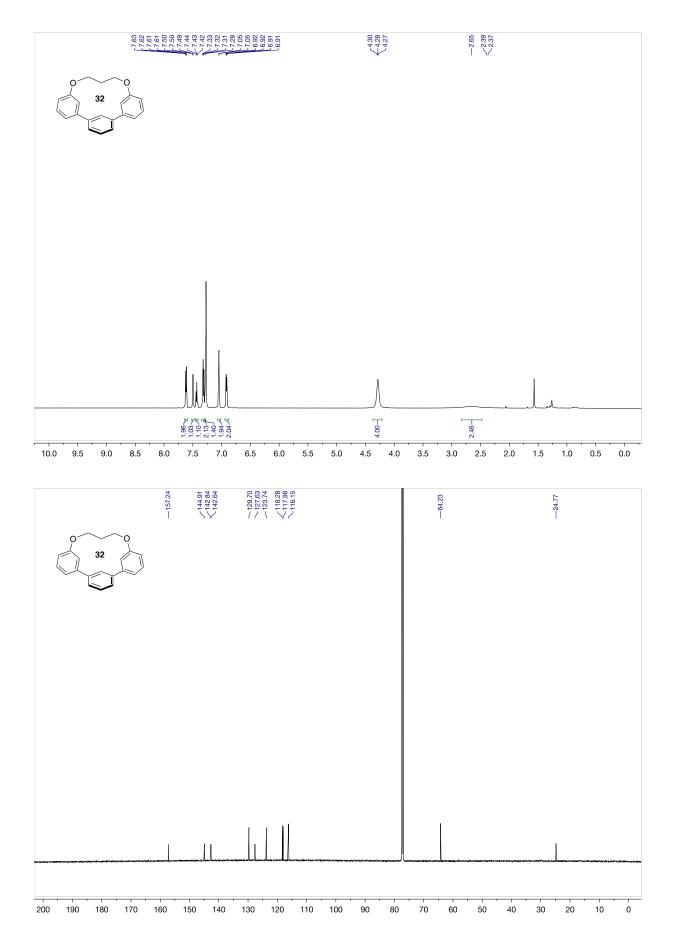


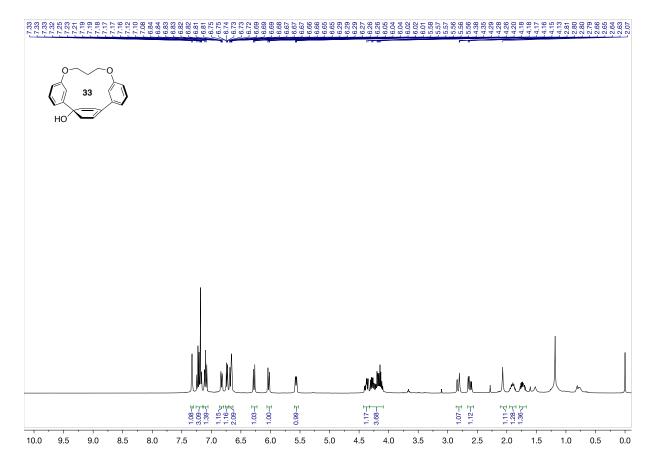


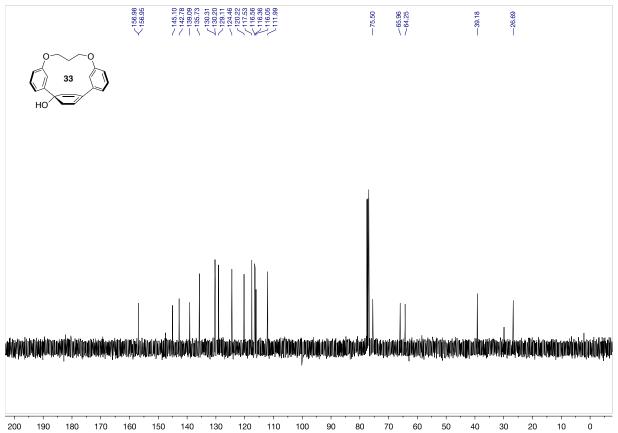


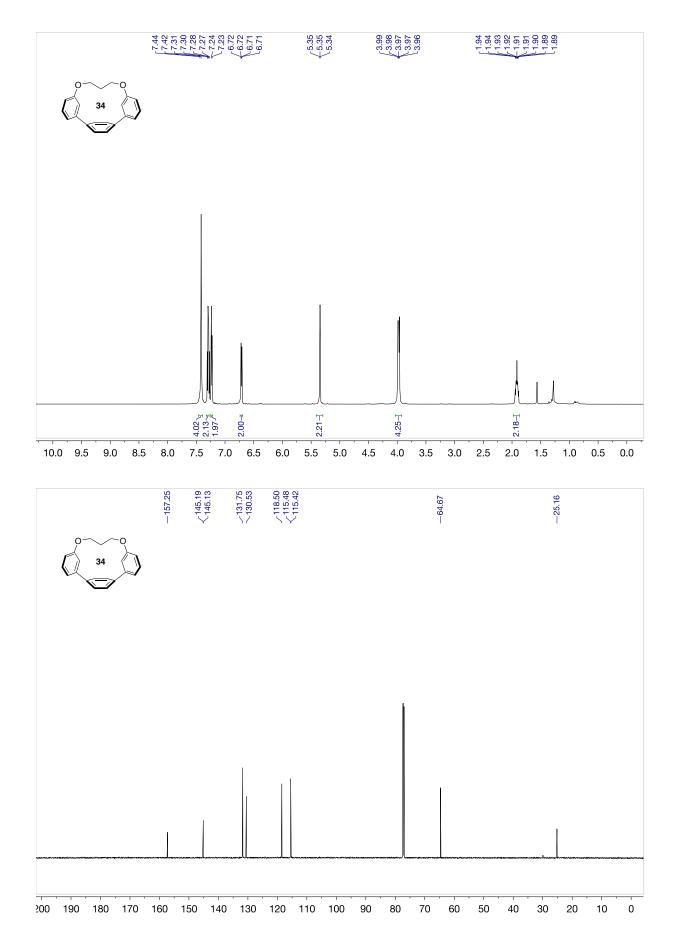


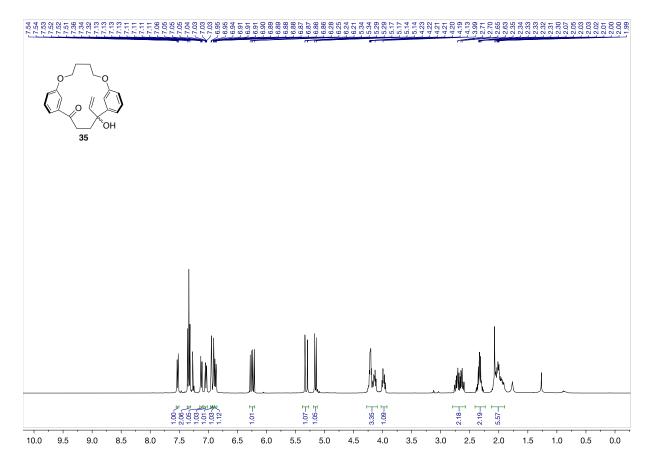


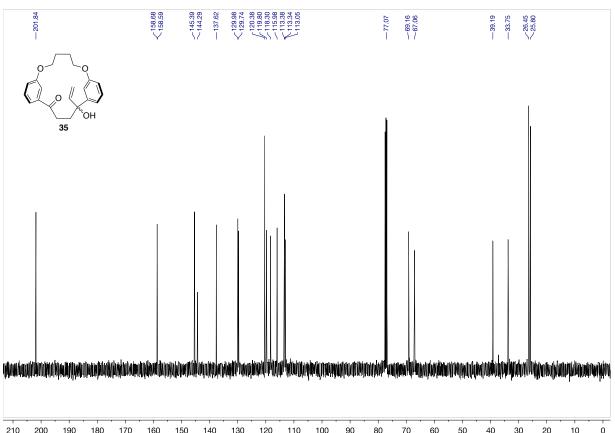


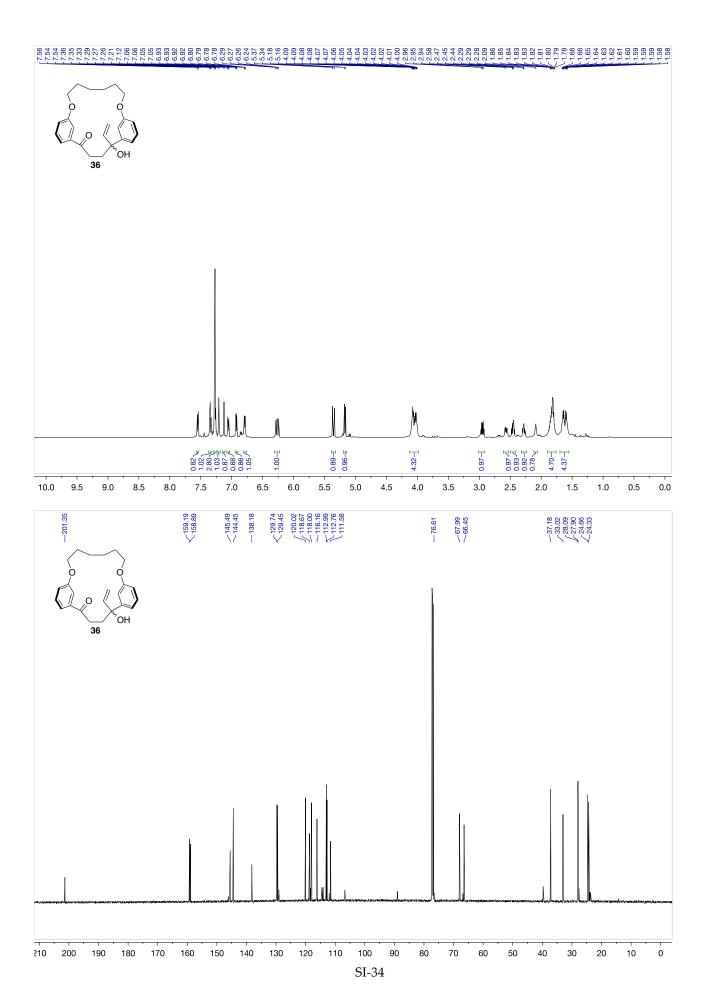


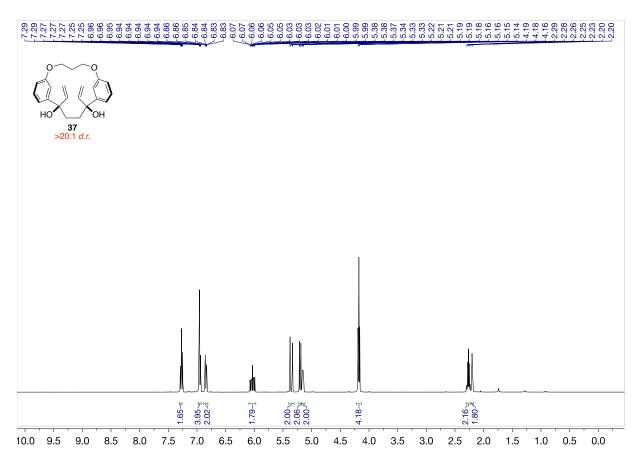


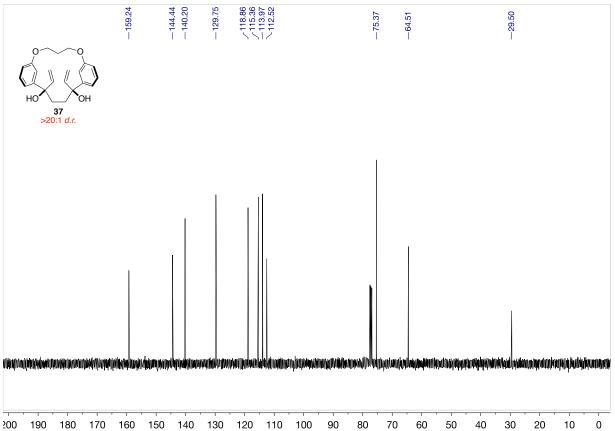












# X-ray Crystal Structure and Relevant Data for Compound 34

Table SI-1: Crystal Data and Structure Refinement for (Merner060815) Compound 34

Identification code	Merner060815	
Empirical formula	$C_{21}H_{18}O_2$	
Formula weight	304.37	
Temperature	180(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	Pbca	
Unit cell dimensions	a = 19.549(2)  Å	$\alpha$ = 90°
	b = 8.1172(8)  Å	$\beta = 90^{\circ}$
	c = 19.927(2)  Å	$\gamma = 90^{\circ}$
Volume	3162.1(6) Å <sup>3</sup>	
Z	8	
Density (calculated)	1.279 g/cm <sup>3</sup>	
Absorption coefficient	0.081 mm <sup>-1</sup>	
F(000)	1296	
Crystal size	0.17 x 0.22 x 0.25 mm	3
Theta range for data collection	2.04 to 28.35°	
Index ranges	-25≤h≤26, -10≤k≤9, -26	6≤l≤25
Reflections collected	31449	
Independent reflections	3941 [R(int) = 0.0561]	
Completeness to theta = 28.35°	99.7%	
Absorption correction	Multiscan	
Max. and min. transmission	0.9860 and 0.9800	
Refinement method	Full-matrix least-squa	ares on F2
Data / restraints / parameters	3941 / 0 / 220	
Goodness-of-fit on F2	1.212	
Final R indices [I>2sigma(I)]	R1 = 0.0602, $wR2 = 0$ .	1070
R indices (all data)	R1 = 0.0725, $wR2 = 0.3$	1120
Largest diff. peak and hole	0.245 and -0.196	

Table SI-2: Bond Lengths and Angles for (Merner060815) Compound 34

O1-C26	1.378(2)
O1-C2	1.442(2)
O8-C21	1.380(2)
O8-C4	1.446(2)
C9-C10	1.385(3)
C9-C21	1.393(2)
C21-C20	1.381(2)
C20-C22	1.404(2)
C22-C11	1.381(2)

C22-C23 C23-C12 C23-C19 C19-C18 C18-C24 C24-C13 C24-C25 C25-C14 C25-C17 C14-C15 C15-C16 C16-C26 C26-C17 C2-C3 C3-C4 C13-C12 C11-C10	1.502(2) 1.399(2) 1.401(2) 1.385(2) 1.402(2) 1.397(2) 1.493(2) 1.388(2) 1.408(2) 1.389(3) 1.382(3) 1.396(2) 1.381(2) 1.532(2) 1.528(2) 1.393(2) 1.397(3)
C26-O1-C2 C21-O8-C4 C10-C9-C21 O8-C21-C9 C20-C21-C9 C20-C21-C9 C21-C20-C22 C11-C22-C23 C20-C22-C23 C12-C23-C19 C12-C23-C22 C19-C23-C22 C18-C19-C23 C19-C18-C24 C13-C24-C18 C13-C24-C25 C14-C25-C17 C14-C25-C24 C17-C25-C24 C15-C16-C15 C16-C15-C14 C15-C16-C26 O1-C26-C16 C17-C26-C16 C17-C26-C16	118.22(13) 117.43(13) 119.00(16) 122.78(15) 118.17(15) 119.04(16) 122.07(16) 118.72(16) 131.59(16) 109.46(14) 117.73(16) 117.76(15) 117.27(15) 120.23(16) 119.54(16) 117.81(17) 118.04(15) 118.95(16) 118.71(17) 128.43(17) 111.69(15) 119.29(18) 121.99(17) 119.10(17) 123.68(16) 117.00(16) 119.28(17) 114.53(15)

C4-C3-C2	108.95(14)
O8-C4-C3	115.00(14)
C26-C17-C25	121.48(16)
C12-C13-C24	120.11(16)
C13-C12-C23	119.46(16)
C22-C11-C10	119.08(17)
C9-C10-C11	122.05(17)

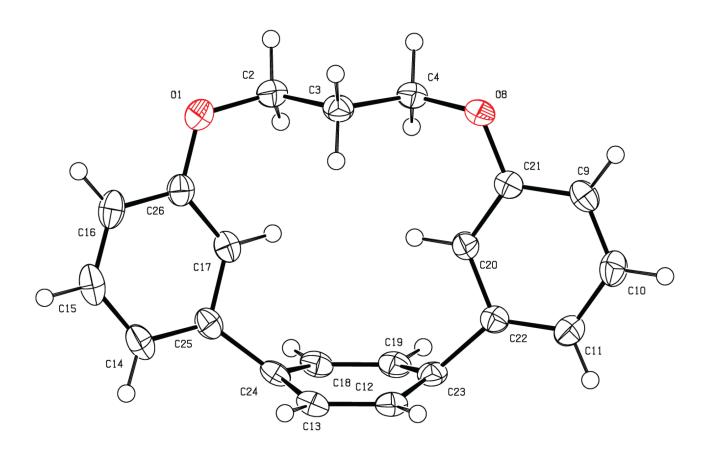


FIGURE SI-2: X-ray crystal structure of compound 34

Table SI-1: Crystal Data and Structure Refinement for (Merner121114) Compound 28

 $\begin{array}{cccc} & & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & \\ & & \\$ 

Unit cell dimensions a = 12.5333(3) Å  $\alpha = 90^{\circ}$ 

	b = 7.8729(2)  Å	β = 91°
	c = 16.2274(4)  Å	$\gamma = 90^{\circ}$
Volume	1600.48(7) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.3130 g/cm <sup>3</sup>	
Absorption coefficient	0.083 mm <sup>-1</sup>	
F(000)	672.3	
Crystal size	$0.5 \times 0.4 \times 0.2 \text{ mm}^3$	
Theta range for data collection	4.04 to 74.34°	
Index ranges	$-21 \le h \le 20$ , $-13 \le k \le 13$ , $-27$	≤1≤27
Reflections collected	47977	
Independent reflections	8058 [R(int) = 0.0586, R(sigm)]	a) = $0.0452$ ]
Data / restraints / parameters	8058 / 0 / 217	
Goodness-of-fit on F2	1.099	
Final R indices [I>2sigma(I)]	R1 = 0.0682, $wR2 = 0.1560$	
R indices (all data)	R1 = 0.0984, $wR2 = 0.1826$	
Largest diff. peak and hole	0.70 and -0.47	

Table SI-3: Bond Lengths and Angles for (Merner121114) Compound 28

O1-C0aa	1.371(2)
O1-C2	1.449(2)
O8-C5	1.446(2)
O8-C7	1.374(2)
C15-C0aa	1.402(2)
C15-C14	1.389(2)
C16-C0aa	1.389(2)
C2-C3	1.536(2)
C3-C4	1.536(2)
C4-C5	1.532(2)
C7-C9	1.400(2)
C7-C20	1.391(2)
C9-C10	1.388(2)
C14-C13	1.397(2)
C13-C3aa	1.400(2)
C16-C3aa	1.406(2)
C3aa-C4aa	1.488(2)
C20-C1aa	1.405(2)
C1aa-C5aa	1.490(2)
C1aa-C11	1.395(2)
C5aa-C19	1.401(2)
C12-C4aa	1.399(2)
C4aa-C17	1.402(2)
C17-C18	1.398(2)

C18-C19	1.394(2)
C11-C10	1.399(2)

C2-O1-C0aa	116.75(9)
C7-O8-C5	117.99(9)
C14-C15-C0aa	118.66(11)
C15-C0aa-O1	117.83(10)
C16-C0aa-O1	122.70(11)
C16-C0aa-C15	119.45(12)
C2-C3-O1	114.34(10)
C4-C3-C2	111.52(10)
C5-C4-C3	112.70(11)
C4-C5-O8	112.90(11)
C9-C7-O8	117.14(11)
C20-C7-O8	123.94(11)
C20-C7-C9	118.91(12)
C10-C9-C7	119.77(11)
C13-C14-C15	121.99(11)
C13-C24-C18	119.73(10)
C3aa-C13-C14	117.93(10)
C4aa-C3aa-C13	124.74(11)
C4aa-C3aa-C16	117.09(9)
C3aa-C16-C0aa	122.18(11)
C1aa-C20-C7	121.30(11)
C5aa-C1aa-C20	115.27(18)
C11-C1aa-C20	119.47(11)
C11-C1aa-C5aa	125.09(11)
C12-C5aa-C1aa	115.29(10)
C19-C5aa-C12	118.34(10)
C4aa-C12-C5aa	123.12(10)
C12-C4aa-C3aa	118.00(10)
C17-C4aa-C3aa	124.22(10)
C17-C4aa-C12	117.21(10)
C18-C17-C4aa	120.09(10)
C19-C18-C17	121.52(11)
C18-C19-C5aa	119.18(12)
C10-C11-C1aa	118.89(12)
C11-C10-C9	121.55(12)

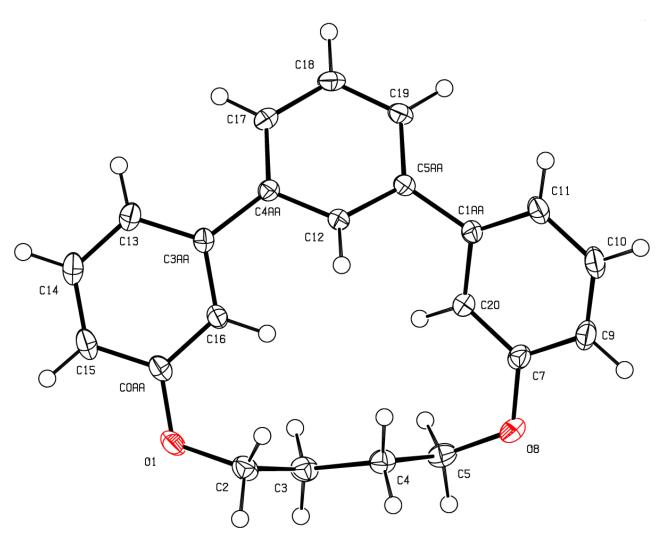


FIGURE SI-3: X-ray crystal structure of compound 28

### DFT Computed Strain Energies of 31 and 34

Full geometry optimization using the geometry from the CIF of **34** as a starting guess was performed using the B3LYP functional in conjunction with a 6-31G(d) basis set. The harmonic vibrational analysis was done at the same level to verify the nature of the stationary point. All of the electronic calculations were performed with Gaussian 09¹ package of programs.

Cartesian Coordinates for <b>31</b> (in Å):			
C	0.3212660	2.7493050	-0.0453390
Н	0.5628480	2.2260540	-0.9830630
Н	0.3868140	2.0230740	0.7656570
C	1.3645940	3.8134910	0.2380560
Н	1.1964530	4.2491770	1.2294520
Н	1.3286080	4.6288670	-0.4951910
C	2.7511280	3.1445680	0.1691690
Н	3.0038250	2.9129420	-0.8720530
Н	3.5222540	3.8097520	0.5701870
C	3.7852380	0.4135700	-0.6593360
Н	4.5455890	1.1438780	-0.9195320
C	3.7792020	-0.8501160	-1.2566500
Н	4.5414470	-1.0945860	-1.9918820
C	2.8174130	-1.7988090	-0.9178490
Н	2.8345970	-2.7801400	-1.3814740
C	-0.1690700	-2.1574780	1.4915120
Н	0.2253650	-2.2545950	2.5032360
C	-1.4251280	-1.7360260	1.3102530
Н	-2.0536680	-1.5014170	2.1674850
C	-3.9485900	0.2517810	-0.1915970
Н	-4.7181960	-0.5108470	-0.1931570
C	-4.2848820	1.6134600	-0.2351330
Н	-5.3325700	1.9006140	-0.2777970
C	-3.3129450	2.6108900	-0.2156330
Н	-3.5787150	3.6632990	-0.2369020
C	-1.6270970	0.8893910	-0.1399920
Н	-0.5952630	0.5799690	-0.1027110
C	-1.1311970	-1.8995930	-1.2039300
Н	-1.7444830	-2.2248080	-2.0509350
Н	-0.6049580	-0.9975070	-1.5289930
C	-0.0924760	-2.9678500	-0.8463340
Н	0.5507580	-3.1710620	-1.7085080
Н	-0.5764160	-3.9168520	-0.5826090
C	1.8593210	-0.2450900	0.6552210
Н	1.1293830	0.0083000	1.4160000

C	2.8157150	0.7193440	0.2987620
C	1.8291710	-1.4997650	0.0352870
C	0.7682730	-2.5695350	0.3712640
C	-2.1041710	-1.5566990	-0.0393240
C	-2.6038140	-0.1112260	-0.1455760
C	-1.9610920	2.2440040	-0.1595640
Н	2.1206790	-3.5544940	1.4020730
Н	-2.9602900	-3.3030080	0.1494380
O	-1.0143240	3.2407660	-0.1153490
O	2.7833510	1.9332760	0.9542970
Ο	1.4453800	-3.7878210	0.7441140
Ο	-3.2605580	-2.4090100	-0.0778510

Cartesian Coordinates for *p*-terphenyl precursor of **31** (in Å):

C	4.0938360	2.3203580	-0.2895740	
Н	4.7114240	3.2018340	-0.4396960	
C	4.5106410	1.0780030	-0.7740900	
Н	5.4558050	0.9888220	-1.3035800	
C	3.7188600	-0.0524580	-0.5795430	
Н	4.0467160	-1.0191730	-0.9499840	
C	0.6683050	-1.1281390	1.4258760	
Н	1.1189150	-1.1549740	2.4186910	
C	-0.6604510	-1.0390050	1.3071900	
Н	-1.2902580	-1.0232530	2.1946540	
C	-3.8375350	-0.3927280	-0.3698930	
Н	-4.0494110	-1.4293470	-0.6026870	
C	-4.8585410	0.5601300	-0.3966920	
Н	-5.8686490	0.2547650	-0.6580880	
C	-4.5901970	1.8936820	-0.0886440	
Н	-5.3868130	2.6327770	-0.1082950	
C	-2.2693220	1.3203550	0.2726540	
Н	-1.2600750	1.6259920	0.5374600	
C	-0.4042270	-0.7750700	-1.1812630	
Н	-0.9096890	-1.0207090	-2.1210590	
Н	-0.1622310	0.2935320	-1.2142040	
C	0.8848070	-1.5867140	-1.0292740	
Н	1.5430460	-1.4391260	-1.8913950	
Н	0.6590090	-2.6600150	-0.9868440	
C	2.0886720	1.2851140	0.5864590	
Н	1.1535990	1.3710860	1.1316950	
C	2.8811930	2.4192950	0.3914310	
C	2.4937190	0.0361220	0.1005730	
C	1.6435830	-1.2336100	0.2670970	
C	-1.3942800	-1.0447570	-0.0228810	

C	-2.5301470	-0.0221220	-0.0364780
C	-3.2885870	2.2712480	0.2476700
Н	3.1189280	-2.1203240	1.2147190
Н	-1.3170260	-2.9992820	0.0087150
Ο	2.5089670	-2.3643380	0.4987390
Ο	-2.0025680	-2.3414970	-0.1843790
Н	-3.0657640	3.3065400	0.4930230
Н	2.5493010	3.3793090	0.7782900

# Cartesian Coordinates for alkoxy bridging group of **31** (in Å):

C	1.0016090	-0.3573120	0.2886430
Н	0.9827770	-0.3976060	1.3921430
Н	0.6823180	-1.3458830	-0.0834540
C	0.0375080	0.7147830	-0.2039330
Н	0.0637950	0.7379960	-1.2995760
Н	0.3934530	1.6905240	0.1524140
C	-1.4046250	0.5057520	0.2635150
Н	-1.4508640	0.4962440	1.3656130
Н	-2.0223160	1.3428360	-0.0760810
Ο	2.3002940	-0.0319070	-0.1910680
Ο	-2.0185440	-0.6605370	-0.2811930
Н	2.9093110	-0.7292060	0.0952710
Н	-1.6194180	-1.4346830	0.1424080

### Cartesian Coordinates for **34** (in Å):

Cart	coluit Coordina	(111 11)	•
Ο	2.5008600	2.4769540	-0.0663690
O	-2.5014700	2.4767900	-0.0661030
C	-4.1810510	0.9095340	-0.6688080
Н	-4.7964100	1.7394290	-1.0019400
C	-2.9197620	1.1707590	-0.1102330
C	-2.1481990	0.1000130	0.3274770
Н	-1.1628060	0.2712500	0.7349810
C	-2.5507840	-1.2384260	0.1382700
C	-1.4005580	-2.1747890	0.3530230
C	-0.6964980	-2.1262590	1.5686650
C	0.6970860	-2.1262570	1.5685580
C	1.4009580	-2.1748010	0.3528020
C	2.5510250	-1.2382610	0.1379540
C	3.8009490	-1.4923690	-0.4263360
Н	4.1368800	-2.5121720	-0.5901390
C	4.6098300	-0.4092790	-0.8028310
Н	5.5890240	-0.6012880	-1.2340140

C	4.1810410	0.9101030	-0.6686090
Н	4.7963420	1.7401210	-1.0015280
C	2.9194940	1.1710300	-0.1104680
C	1.2702900	2.8011090	0.5947750
Н	1.3708210	3.8625900	0.8413240
Н	1.2084350	2.2507010	1.5436210
C	-0.0003560	2.5756450	-0.2525400
Н	-0.0003810	3.2626500	-1.1058070
Н	-0.0004650	1.5632290	-0.6645240
C	-1.2709150	2.8012130	0.5949060
Н	-1.2089050	2.2510100	1.5438660
Н	-1.3715240	3.8627310	0.8412410
C	2.1480260	0.1000840	0.3268960
Н	1.1624350	0.2710810	0.7340060
C	0.6983360	-2.6030540	-0.7856340
C	-0.6981190	-2.6030650	-0.7855200
C	-3.8004470	-1.4928280	-0.4265040
Н	-4.1360110	-2.5127210	-0.5904740
C	-4.6094880	-0.4099420	-0.8032070
Н	-5.5884960	-0.6021560	-1.2347140
Н	-1.2284750	-1.8857580	2.4860930
Н	-1.2239940	-2.7029460	-1.7318650
Н	1.2240680	-2.7029390	-1.7320590
Н	1.2292030	-1.8857230	2.48588

Cartesian Coordinates for alkoxy bridging group of 34 (in Å):				
Ο	2.4612860	-0.2841330	-0.1052280	
O	-2.4612860	-0.2841330	-0.1052270	
C	1.2781130	0.4911610	0.0459260	
Н	1.3095390	1.2271150	-0.7649240	
Н	1.2914070	1.0549930	0.9941160	
C	0.0000000	-0.3454170	-0.0538550	
Н	0.0000000	-0.8947320	-1.0022630	
Н	0.0000000	-1.0978860	0.7523190	
C	-1.2781130	0.4911610	0.0459260	
Н	-1.2914070	1.0549940	0.9941150	
Н	-1.3095390	1.2271150	-0.7649250	
Н	-2.4862720	-0.9234540	0.6236060	
Н	2.4862760	-0.9234500	0.6236090	

## Cartesian Coordinates for *p*-terphenyl system of **34** (in Å):

C	-5.7300110	0.0001460	-0.0001880
Н	-6.8166550	0.0001850	-0.0001910

C	-5.0259130	1.1822850	-0.2365940
C	-3.6319500	1.1815670	-0.2373110
Н	-3.0949770	2.1018100	-0.4500820
C	-2.9087790	-0.0000600	0.0000040
C	-1.4246130	-0.0001210	0.0001530
C	-0.6956500	1.0835930	0.5181180
C	0.6956500	1.0835920	0.5181180
C	1.4246130	-0.0001210	0.0001530
C	2.9087790	-0.0000600	0.0000040
C	3.6321460	-1.1815610	0.2373210
Н	3.0953000	-2.1018500	0.4502030
C	5.0260970	-1.1820670	0.2363700
Н	5.5633960	-2.1068890	0.4299920
C	5.7300110	0.0001460	-0.0001880
Н	6.8166540	0.0001850	-0.0001900
C	5.0259120	1.1822860	-0.2365910
C	3.6319490	1.1815670	-0.2373080
Н	3.0949760	2.1018110	-0.4500760
C	0.6956480	-1.0838220	-0.5178670
C	-0.6956470	-1.0838220	-0.5178660
C	-3.6321460	-1.1815600	0.2373240
Н	-3.0953000	-2.1018490	0.4502080
C	-5.0260970	-1.1820670	0.2363730
Н	-5.5633950	-2.1068890	0.4299970
Н	-1.2267650	1.9225390	0.9592850
Н	-1.2267790	-1.9227620	-0.9590250
Н	1.2267790	-1.9227620	-0.9590250
Н	1.2267660	1.9225380	0.9592850
Н	-5.5630360	2.1071990	-0.4302730
Н	5.5630350	2.1072000	-0.4302670

[1.] Gaussian 09, Revision D.01: M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.