# SUPPORTING INFORMATION 

## for

## Asymmetric Hydrogenation of Bicyclic Ketones Catalyzed by BINAP/IPHAN-Ru(II) Complex

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## (A) Preparation and Physical Data of trans- $\mathrm{RuCl}_{2}$ (binap)(1,4-diamine) complexes (3a-e)

The preparative method for these complexes was previously described. ${ }^{1}$ Their physical properties are as follows:
trans- $\mathbf{R u C l}_{\mathbf{2}}[(\boldsymbol{R})$-binap $][(\boldsymbol{S})$-ipban] [(R,S)-3a]. IR (ATR) 3326, 3053, 1433, 1087, 1075, 1021, 741, $697 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.27$ (s, 6H), 2.43-2.47 (br $\mathrm{m}, 2 \mathrm{H}$ ), 2.57-2.75 (m, 4H), 2.93 (br s, 2H), 3.67-3.70(m, 2H), 6.27 (d, 2H, J = 8.2 Hz ), 6.47 (br m, 6H), 6.67-6.72 (t-like m, 2H), 7.12-7.17 (t-like m, 2H), 7.39 (br m, 6H), 7.52 (d, 2H, $J=8.0 \mathrm{~Hz}$ ), 7.62 (br m, 4H), 7.73 (d, 2H, $J=8.5 \mathrm{~Hz}$ ), 7.91 (br, m, 4H), 8.25-8.30 (m, 2H). ${ }^{31} \mathrm{P}$ NMR ( $121 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 46.5$ (s). HRMS (ESI ${ }^{+} \mathrm{m} / \mathrm{z}$ 954.1606 ( $\mathrm{M}^{+}$), calcd for $\mathrm{C}_{51} \mathrm{H}_{48}{ }^{35} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2}{ }^{102} \mathrm{Ru}$ : 954.1612.
trans- $\mathbf{R u C l}_{2}[(\boldsymbol{S})$-binap][(S)-ipban] [(S,S)-3a]. IR (ATR) 3323, 3054, 1434, 1087, 1023, 741, $697 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.29(\mathrm{~s}, 6 \mathrm{H}), 2.07(\mathrm{br} \mathrm{m}, 2 \mathrm{H})$, 2.67-2.70 (br m, 2H), 2.85 (br m, 2H), 3.06-3.08 (br m, 2H), 3.72-3.74 (m, 2H), 6.28 (d, 2H, $J=8.7 \mathrm{~Hz}$ ), 6.48 (br m, 6H), 6.68-6.73 (t-like m, 2H), 7.13-7.17 (t-like m, 2H), 7.39 (br m, 6H), 7.52 (d, 2H, $J=8.0 \mathrm{~Hz}$ ), 7.61 (br m, 4H), 7.71 (d, 2H, $J=8.7 \mathrm{~Hz}$ ), 7.89 (br, m, 4H), 8.23-8.27 (m, 2H). ${ }^{31} \mathrm{P}$ NMR ( $121 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 46.5$ (s). HRMS ( $\mathrm{ESI}^{+}$) $m / z$ 954.1617 $\left(\mathrm{M}^{+}\right)$, calcd for $\mathrm{C}_{51} \mathrm{H}_{48}{ }^{35} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2}{ }^{102} \mathrm{Ru}$ : 954.1612.
trans-RuCl $\mathbf{2}_{\mathbf{2}}[\boldsymbol{S})$-binap][(R)-iphan] [(S,R)-3b]. IR (KBr-disk) 3311, 3055, 1569, 1482, 1433, 1370, 1057, 1035, 741, $697 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.54(\mathrm{~d}$, $6 \mathrm{H}, J=6.4 \mathrm{~Hz}$ ), $1.24(\mathrm{~s}, 6 \mathrm{H}), 2.66-2.75(\mathrm{br} \mathrm{m}, 4 \mathrm{H}), 3.04(\mathrm{br} \mathrm{m}, 2 \mathrm{H}), 3.47-3.51(\mathrm{~m}, 2 \mathrm{H})$, $6.25(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 6.45-6.47(\mathrm{br} \mathrm{m}, 6 \mathrm{H}), 6.64-6.70(\mathrm{t}-\mathrm{like} \mathrm{m}, 2 \mathrm{H}), 7.10-7.16$ (tlike m, 2H), 7.39-7.41 (br m, 6H), $7.51(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}$ ), 7.63 (br m, 4H), 7.72 (d, $2 \mathrm{H}, J=8.5 \mathrm{~Hz}), 7.96(\mathrm{br}, \mathrm{m}, 4 \mathrm{H}), 8.28-8.34(\mathrm{~m}, 2 \mathrm{H}) .{ }^{31} \mathrm{P}$ NMR ( $\left.121 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 45.7 (s). HRMS (ESI $) \mathrm{m} / \mathrm{z} 982.1918\left(\mathrm{M}^{+}\right)$, calcd for $\mathrm{C}_{53} \mathrm{H}_{52}{ }^{35} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2}{ }^{102} \mathrm{Ru}$ : 982.1925. Found: C, $64.62 \%, \mathrm{H}, 5.55 \%$; N, $2.70 \%$. Calcd for $\mathrm{C}_{53} \mathrm{H}_{52} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2} \mathrm{Ru}: \mathrm{C}$, $64.76 \%$; H, $5.33 \%$; N, $2.85 \%$.
trans- $\mathbf{R u C l}_{\mathbf{2}}[(\boldsymbol{S})$-tolbinap][(R)-iphan] [(S,R)-3c]. IR (ATR) 3309, 2989, 1500, 1223, 1194, 1053, 1038, 808, 756, $744 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.53(\mathrm{~d}, 6 \mathrm{H}, J=$ $6.4 \mathrm{~Hz}), 1.24(\mathrm{~s}, 6 \mathrm{H}), 1.77(\mathrm{~s}, 6 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}), 2.63-2.73(\mathrm{~m}, 4 \mathrm{H}), 3.01(\mathrm{br} \mathrm{m}, 2 \mathrm{H})$, 3.46-3.51 (m, 2H), 6.20-6.24 (m, 6H), 6.65-6.69 (t-like m, 2H), 7.12-7.16 (t-like m, $2 \mathrm{H}), 7.21$ (d, 4H, $J=8.0 \mathrm{~Hz}$ ), 7.46 (br m, 4H), 7.50 (d, 2H, $J=7.9 \mathrm{~Hz}$ ), 7.72 (d, 2H, $J=$ 8.7 Hz ), 7.87 (br, m, 4H), 8.28-8.33 (m, 2H). ${ }^{31} \mathrm{P}$ NMR ( $161 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 44.5$ (s). HRMS (ESI $)^{+} m / z$ 1038.2580 ( $\mathrm{M}^{+}$), calcd for $\mathrm{C}_{57} \mathrm{H}_{60}{ }^{35} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2}{ }^{102} \mathrm{Ru}$ : 1038.2551. Found: C, $61.34 \%, \mathrm{H}, 5.34 \%$; N, $2.40 \%$. Calcd for $\mathrm{C}_{57} \mathrm{H}_{60} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2} \mathrm{Ru} \bullet 0.67 \mathrm{CHCl}_{3}$ : C, $61.92 \%$; H, $5.47 \%$; N, $2.50 \%$.
trans- $\mathbf{R u C l}_{\mathbf{2}}[(\boldsymbol{S})$-binap $][(\boldsymbol{R}, \boldsymbol{R})$-2,5-hexanediamine $][(\boldsymbol{S}, \boldsymbol{R})-\mathbf{3 d}] . \quad$ IR (ATR) 3320, 3054, $2960,1574,1482,1432,1260,1085,1027,806,738,696 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 0.44(\mathrm{~d}, 6 \mathrm{H}, J=6.4 \mathrm{~Hz}), 1.25-1.31(\mathrm{br} \mathrm{m}, 2 \mathrm{H}), 1.52(\mathrm{br} \mathrm{m}, 2 \mathrm{H}), 2.73-2.74$ (br m, 4H), 2.99 (br m, 2H), $6.25(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 6.45-6.47(\mathrm{br} \mathrm{m}, 6 \mathrm{H}), 6.65-6.70(\mathrm{t}-$ like m, 2H), 7.10-7.15 (t-like m, 2H), 7.38 (br m, 6H), 7.52 (d, 2H, $J=7.8 \mathrm{~Hz}$ ), 7.64 (br $\mathrm{m}, 4 \mathrm{H}), 7.72(\mathrm{~d}, 2 \mathrm{H}, J=8.5 \mathrm{~Hz}), 7.98(\mathrm{br}, \mathrm{m}, 4 \mathrm{H}), 8.26-8.32(\mathrm{~m}, 2 \mathrm{H}) .{ }^{31} \mathrm{P}$ NMR (121 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \quad \delta 45.7$ (s). HRMS (ESI $) ~ m / z 910.1706\left(\mathrm{M}^{+}\right)$, calcd for $\mathrm{C}_{50} \mathrm{H}_{48}{ }^{35} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2}{ }^{102} \mathrm{Ru}: 910.1713$. Found: C, $65.93 \%, \mathrm{H}, 5.31 \%$; N, 3.08\%. Calcd for $\mathrm{C}_{50} \mathrm{H}_{48} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2} \mathrm{Ru} \bullet 0.25 \mathrm{CHCl}_{3}: \mathrm{C}, 64.16 \% ; \mathrm{H}, 5.17 \% ; \mathrm{N}, 2.98 \%$.
trans- $\mathrm{RuCl}_{2}[(\boldsymbol{R})$-tolbinap $][(\boldsymbol{S}, \boldsymbol{S})$-2,5-hexanediamine] [(R,S)-3e]. IR (ATR) 3316, 3053, 2919, 1499, 1191, 1089, 1037, $804 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.43(\mathrm{~d}$, $6 \mathrm{H}, J=6.4 \mathrm{~Hz}), 1.21-1.30(\mathrm{~m}, 2 \mathrm{H}), 1.51(\mathrm{br} \mathrm{m}, 2 \mathrm{H}), 1.78(\mathrm{~s}, 6 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}), 2.71$ (br m, 4H), 2.96 (br m, 2H), 6.20-6.24 (m, 6H), 6.63-6.69 (t-like m, 2H), 7.09-7.19 (m, $6 \mathrm{H}), 7.48-7.52(\mathrm{~m}, 6 \mathrm{H}), 7.72(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.5 \mathrm{~Hz}), 7.90(\mathrm{br} \mathrm{m}, 4 \mathrm{H}), 8.26-8.32(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{31} \mathrm{P}$ NMR (121 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 43.9(\mathrm{~s})$. HRMS (ESI ${ }^{+}$) $m / z 966.2343\left(\mathrm{M}^{+}\right)$, calcd for $\mathrm{C}_{54} \mathrm{H}_{56}{ }^{35} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}_{2}{ }^{102} \mathrm{Ru}: 966.2339$.
(B) Asymmetric Hydrogenation of 3-Quinuclidinone (1a) ( $\mathbf{S} / \mathbf{C}=\mathbf{5 0 , 0 0 0}$ )

Solid ( $S, R$ )-3b ( $3.0 \mathrm{mg}, 0.0031 \mathrm{mmol}$ ) and 3-quinuclidinone (1a) ( $18.78 \mathrm{~g}, 150 \mathrm{mmol}$ ) were placed in a 200 mL SUS autoclave with a Teflon-coated magnetic stirring bar. Air present in the autoclave was replaced by argon. 2-Propanol ( 49 mL ) was added to the autoclave and degassed. A solution of $t-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{OK}$ in 2-propanol ( $1 \mathrm{M}, 1.0 \mathrm{~mL}, 1.0$ mmol) which had been degassed was added to the autoclave. Hydrogen was initially introduced into the autoclave at a pressure of 5 atm , before being reduced to 1 atm by carefully releasing the stop valve. After this procedure was repeated 5 times, the vessel was pressurized to 50 atm . The reaction mixture was vigorously stirred at $25^{\circ} \mathrm{C}$. After stirring for 24 h and carefully venting the hydrogen gas, the solvent was removed under reduced pressure. The residue was purified by silica-gel (Fuji silysia, Chromatorex NH) column chromatography giving ( $R$ )-3-quinuclidinol ${ }^{2,3}$ [(R)-2a] (colorless powder, $18.9 \mathrm{~g}, 99 \%$ yield, $97 \%$ ee). The enantiomeric excess of 2a was determined by HPLC analysis of acetylated derivative. Column, CHIRALCEL AD-H; eluent, hexane:ethanol:2-propanol $=90: 5: 5$; flow, $0.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}$; column temp, $25^{\circ} \mathrm{C}$; retention time $\left(t_{\mathrm{R}}\right)$ of $\mathrm{Ac}-(S) \mathbf{- 2 a}, 18.3 \mathrm{~min}(1.5 \%) ; t_{\mathrm{R}}$ of $\mathrm{Ac}-(R) \mathbf{- 2 a}, 24.6 \min (98.5 \%)$. $[\alpha]_{\mathrm{D}}{ }^{28}-42.5(c 1.06,1 \mathrm{M} \mathrm{HCl})\left(\right.$ lit. $^{2}[\alpha]_{\mathrm{D}}{ }^{25}-44.9(c 2.0,1 \mathrm{M} \mathrm{HCl}), 96 \%$ ee $\left.(R)\right)$. IR (KBr-disk) 3109 (br), 2941, 2871, 1456, 1346, 1309, 1116, 1045, 988, 817, 795, 773 $\mathrm{cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.31-1.40(\mathrm{~m}, 1 \mathrm{H}), 1.41-1.51(\mathrm{~m}, 1 \mathrm{H}), 1.64-$ $1.72(\mathrm{~m}, 1 \mathrm{H}), 1.79-1.83(\mathrm{~m}, 1 \mathrm{H}), 1.90-1.98(\mathrm{~m}, 1 \mathrm{H}), 2.58-2.69(\mathrm{~m}, 2 \mathrm{H}), 2.72-2.82(\mathrm{~m}$, 2H), 2.87-2.95 (m including a broad signal, 2H), 3.13 (ddd, $J=14.1,8.2,2.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.83-3.87(m, 1H). ${ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 18.9\left(\mathrm{CH}_{2}\right), 24.8\left(\mathrm{CH}_{2}\right), 28.3(\mathrm{CH})$, $46.3\left(\mathrm{CH}_{2}\right), 47.3\left(\mathrm{CH}_{2}\right), 57.9\left(\mathrm{CH}_{2}\right), 67.5(\mathrm{CH}) . \quad \mathrm{HRMS}\left(\mathrm{EI}^{+}\right), m / z 127.0997\left(\mathrm{M}^{+}\right)$, calcd $\mathrm{C}_{7} \mathrm{H}_{13} \mathrm{NO}: 127.0997$. Found: C, $65.81 \%, \mathrm{H}, 10.32 \%$; N, 10.94\%. Calcd for $\mathrm{C}_{7} \mathrm{H}_{13} \mathrm{NO}: \mathrm{C}, 66.10 \%$; $\mathrm{H}, 10.30 \%$; N, $11.01 \%$.

## (C) Asymmetric Hydrogenation of Bicyclo[2.2.2]octan-2-one (1b) ( $\mathbf{S} / \mathbf{C}=\mathbf{1 , 0 0 0}$ )

Solid ( $S, R$ )-3b ( $1.3 \mathrm{mg}, 1.3 \mu \mathrm{~mol}$ ), $t-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{OK}(14.9 \mathrm{mg}, 0.133 \mathrm{mmol}$ ), and bicyclo[2.2.2]octan-2-one (1b) ( $167.9 \mathrm{mg}, 1.35 \mathrm{mmol}$ ) were placed in a $100-\mathrm{mL}$ glass
autoclave with a Teflon-coated magnetic stirring bar under Ar. A degassed (three freeze-thaw cycles) mixture of 2-propanol ( 4.8 mL ) and tert-butylalcohol ( 1.6 mL ) was added to the autoclave. Hydrogen was initially introduced into the autoclave at a pressure of 10 atm , before being reduced to 1 atm by carefully releasing the stop valve. After this procedure was repeated several times, the vessel was pressurized to 20 atm , and then the reaction mixture was vigorously stirred at $20^{\circ} \mathrm{C}$ for 2 h . After venting the hydrogen gas, the solvent was carefully removed under reduced pressure. The residue was purified by silica-gel column chromatography eluted with a hexane-ethyl acetate (7:1 then 4:1) giving (S)-bicyclo[2.2.2]octan-2-ol ${ }^{4}[(S)$-2b] (colorless crystals, $153.4 \mathrm{mg}, 90 \%$ yield, $98 \%$ ee). The enantiomeric excess of $\mathbf{2 b}$ was determined by GC analysis. Column, BETA DEX-120 ( $0.25 \times 30, \mathrm{DF}=0.25$ ); carrier, $\mathrm{He}(100 \mathrm{kPa})$; oven temp, $80^{\circ} \mathrm{C}, 2 \mathrm{~min}$ hold, $1^{\circ} \mathrm{C} / \mathrm{min}$ to $140^{\circ} \mathrm{C}$; $t_{\mathrm{R}}$ of $(R)-\mathbf{2 b}, 47.98 \mathrm{~min}(1 \%) ; t_{\mathrm{R}}$ of $(S)-\mathbf{2 b}$, $48.34 \mathrm{~min}(99 \%) . \quad[\alpha]_{\mathrm{D}}{ }^{28}+33.0\left(c 1.01, \mathrm{CHCl}_{3}\right) . \quad\left(\right.$ lit. $^{4}[\alpha]_{\mathrm{D}}{ }^{27}+31.3\left(c 1.4, \mathrm{CHCl}_{3}\right)$, $97.8 \%$ ee $(S)$ ). IR (KBr-disk) 3357, 2932, 2861, 1456, 1362, 1092, $1034 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.30-1.51(\mathrm{~m}, 6 \mathrm{H}), 1.53-1.65(\mathrm{~m}, 4 \mathrm{H}), 1.84-1.92(\mathrm{~m}, 1 \mathrm{H})$, $1.97-2.05(\mathrm{~m}, 1 \mathrm{H}), 3.95(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 18.5$ $\left(\mathrm{CH}_{2}\right), 23.7\left(\mathrm{CH}_{2}\right), 24.4\left(\mathrm{CH}_{2}\right), 24.7(\mathrm{CH}), 25.6\left(\mathrm{CH}_{2}\right), 31.7(\mathrm{CH}), 37.5\left(\mathrm{CH}_{2}\right), 69.6$ (CH). $\operatorname{HRMS}\left(\mathrm{EI}^{+}\right), m / z 126.1045\left(\mathrm{M}^{+}\right)$, calcd for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{O}: 126.1045$. Found: C, $76.14 \%, \mathrm{H}, 11.37 \%$. Calcd for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{O}: \mathrm{C}, 76.14 \% ; \mathrm{H}, 11.18 \%$.

## (D) Asymmetric Hydrogenation of ( $\pm$ )-Benzobicyclo[2.2.2]octen-2-one [( $\pm$ )-4]

Solid ( $S, R$ )-3b ( $1.3 \mathrm{mg}, 1.3 \mu \mathrm{~mol}$ ), $t$ - $\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{OK}(16.6 \mathrm{mg}, 0.148 \mathrm{mmol})$, and ( $\pm$ )-4 $\mathbf{4}^{5}(224.0$ $\mathrm{mg}, 1.30 \mathrm{mmol}$ ) were placed in a $100-\mathrm{mL}$ glass autoclave with a Teflon-coated magnetic stirring bar under Ar. A degassed (three freeze-thaw cycles) mixture of 2propanol ( 4.9 mL ) and tert-butylalcohol ( 1.5 mL ) was added to the autoclave. Hydrogen was initially introduced into the autoclave at a pressure of 10 atm , before being reduced to 1 atm by carefully releasing the stop valve. After this procedure was repeated several times, the vessel was pressurized to 20 atm , and then the reaction mixture was vigorously stirred at $20^{\circ} \mathrm{C}$ for 2 h . After venting the hydrogen gas, the
solvent was carefully removed under reduced pressure. The residue was purified on silica-gel preparative thin-layer chromatography developed with a toluene-ethyl acetate (10:1) giving ( $1 R, 2 S, 4 R$ )-benzobicyclo[2.2.2]octen-2-ol ${ }^{4.5}$ (5, exo-alcohol) (colorless crystals, $94.2 \mathrm{mg}, 42 \%$ yield, $99 \%$ ee), and ( $1 S, 2 S, 4 S$ )-benzobicyclo[2.2.2]octen-2-ol ${ }^{4,5}$ ( 6 , endo-alcohol) (colorless crystals, $103.5 \mathrm{mg}, 46 \%$ yield, $96 \%$ ee). The enantiomeric excess of 5 and $\mathbf{6}$ was determined by HPLC analysis. For 5: Column, CHIRALCEL OD-H; eluent, hexane:2-propanol $=95: 5$; flow, $0.5 \mathrm{~mL} \mathrm{~min}^{-1}$; column temp, $40{ }^{\circ} \mathrm{C}$; retention time $\left(t_{\mathrm{R}}\right)$ of $(1 S, 2 R, 4 S)-\mathbf{5}, 13.7 \mathrm{~min}(0.3 \%) ; t_{\mathrm{R}}$ of $(1 R, 2 S, 4 R)-\mathbf{5}, 16.0 \mathrm{~min}$ (99.7\%). For 6: Column, CHIRALCEL OJ-H; eluent, hexane:2-propanol $=90: 10$; flow, $0.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}$; column temp, $40^{\circ} \mathrm{C}$; retention time $\left(t_{\mathrm{R}}\right)$ of $(1 S, 2 S, 4 S)-6,23.8 \mathrm{~min}$ (97.9\%); $t_{\mathrm{R}}$ of $(1 R, 2 R, 4 R)-\mathbf{6}, 31.5 \mathrm{~min}(2.1 \%)$. exo-alcohol $(1 R, 2 S, 4 R)-5:[\alpha]_{\mathrm{D}}{ }^{28}+7.3$ (c 1.03, $\mathrm{CHCl}_{3}$ ), lit. ${ }^{4}[\alpha]_{\mathrm{D}}+7.7\left(\mathrm{CHCl}_{3}\right.$, (S)-config., extrapolated). IR (KBr-disk) 3319 (br), 2945, 1482, 1460, 1077, 1051, 1012, $750 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $1.23-1.31(\mathrm{~m}, 1 \mathrm{H}), 1.37-1.48(\mathrm{~m}, 2 \mathrm{H}), 1.61(\mathrm{br} \mathrm{d}, 1 \mathrm{H}), 1.86-1.94(\mathrm{~m}, 1 \mathrm{H}), 1.95-2.03$ (m, 1H), 2.26-2.34 (m, 1H), 2.96-2.99 (m, 1H), 3.01-3.03 (m, 1H), 3.91-3.94 (m, 1H), 7.11-7.14 (m, 1H), 7.17-7.21 (m, 3H). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 18.0\left(\mathrm{CH}_{2}\right)$, $26.4\left(\mathrm{CH}_{2}\right)$, $34.6(\mathrm{CH}), 36.8\left(\mathrm{CH}_{2}\right), 42.2(\mathrm{CH}), 69.5(\mathrm{CH}), 123.5(\mathrm{CH}), 124.7(\mathrm{CH})$, $126.0(\mathrm{CH}), 126.4(\mathrm{CH}), 141.4(\mathrm{C}), 143.8(\mathrm{C}) . \quad$ endo-alcohol ( $1 S, 2 S, 4 S)-6:[\alpha]_{\mathrm{D}}{ }^{28}-$ 20.0 (c 1.04, $\mathrm{CHCl}_{3}$ ), lit. ${ }^{4}[\alpha]_{\mathrm{D}}{ }^{27}-18.55\left(c 0.9, \mathrm{CHCl}_{3}\right), 85 \%$ ee $(S)$ ). IR ( KBr , disk) 3281 (br), 3211 (br), 2941, 2864, 1485, 1330, 1080, 1035, $754 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.02(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.19(\mathrm{~m}, 1 \mathrm{H}), 1.26-1.34(\mathrm{~m}, 1 \mathrm{H}), 1.42-1.50(\mathrm{~m}$, $1 \mathrm{H}), 1.57-1.72$ (m, 2H), 2.26 (ddd, $J=14.0,8.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.03-3.06 (m, 1H), 3.09$3.11(\mathrm{~m}, 1 \mathrm{H}), 4.09-4.14(\mathrm{~m}, 1 \mathrm{H}), 7.15-7.18(\mathrm{~m}, 1 \mathrm{H}), 7.21-7.24(\mathrm{~m}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 22.5\left(\mathrm{CH}_{2}\right), 24.6\left(\mathrm{CH}_{2}\right), 34.1(\mathrm{CH}), 39.5\left(\mathrm{CH}_{2}\right), 42.0\left(\mathrm{CH}_{2}\right), 69.7$ $(\mathrm{CH}), 123.5(\mathrm{CH}), 126.2(\mathrm{CH}), 126.65(\mathrm{CH}), 126.72(\mathrm{CH}), 138.4(\mathrm{C}), 144.0(\mathrm{C})$.

## (E) Asymmetric Hydrogenation of ( $\mathbf{\pm})$-Norcamphor [( $\pm$ )-7] ( $\mathbf{S / C}=\mathbf{1 , 0 0 0}$ )

Solid ( $S, R$ )-3b ( $1.4 \mathrm{mg}, 1.4 \mu \mathrm{~mol}$ ), $t-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{OK}(16.0 \mathrm{mg}, 0.143 \mathrm{mmol}$ ), and ( $\pm$ )norcamphor $[( \pm)-7](158.0 \mathrm{mg}, 1.43 \mathrm{mmol})$ were placed in a $100-\mathrm{mL}$ glass autoclave with a Teflon-coated magnetic stirring bar under Ar. A degassed (three freeze-thaw cycles) mixture of 2-propanol ( 4.8 mL ) and tert-butylalcohol ( 1.6 mL ) was added to the autoclave. Hydrogen was initially introduced into the autoclave at a pressure of 10 atm, before being reduced to 1 atm by carefully releasing the stop valve. After this procedure was repeated several times, the vessel was pressurized to 10 atm , and then the reaction mixture was vigorously stirred at $20{ }^{\circ} \mathrm{C}$ for 2 h . After venting the hydrogen gas, the solvent was carefully removed under reduced pressure. The residue was purified by silica-gel thin layer chromatography developed with a 7:1 hexane-ethyl acetate to give a diastereomeric mixture of norborneol [8 (exo) and $\mathbf{9}$ (endo), 8:9 = 11:89] (colorless crystals, $115.1 \mathrm{mg}, 72 \%$ yield). This sample contained a small amount of ethyl acetate $(4 \%(\mathrm{w} / \mathrm{w}))$. The ratio of diastereomers was determined by GC analysis. Column, BETA DEX-325 ( $0.25 x 30, \mathrm{DF}=0.25$ ); carrier, $\mathrm{He}(100 \mathrm{kPa})$; oven temp, $100{ }^{\circ} \mathrm{C}$ isothernal, $t_{\mathrm{R}}$ of $(1 R, 2 R, 4 S)-\mathbf{8}, 17.14 \min (0.28 \%) ; t_{\mathrm{R}}$ of $(1 S, 2 S, 4 R)-\mathbf{8}$, $17.59 \min (11.3 \%), t_{\mathrm{R}}$ of $(1 S, 2 R, 4 R)-9,18.19 \mathrm{~min}(38.5 \%) ; t_{\mathrm{R}}$ of $(1 R, 2 S, 4 S)-9,18.65$ $\min (49.9 \%) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.84\left(\mathrm{dt}\right.$, endo- $\left.1 \mathrm{H}, J_{\mathrm{d}}=12.7, J_{\mathrm{t}}=3.4 \mathrm{~Hz}\right)$, 0.98-1.05 (m, exo-2H), 1.10-1.14 (m, exo-1H), 1.26-1.70 (m, endo-6H and exo-6H), 1.84-1.99 (m, endo-2H), 2.14-2.26 (m, endo-2H and exo-2H), 3.76 (br d, exo-1H, $J=6.8 \mathrm{~Hz}$ ), 4.23 (br m, endo-1H). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 19.8$ (endo, $\mathrm{CH}_{2}$ ), 24.3 (exo, $\mathrm{CH}_{2}$ ), 28.0 (exo, $\mathrm{CH}_{2}$ ), 29.8 (endo, $\mathrm{CH}_{2}$ ), 34.3 (exo, $\mathrm{CH}_{2}$ ), 35.3 (exo, CH ), 37.1 (endo, CH ), 37.5 (endo, $\mathrm{CH}_{2}$ ), 39.3 (endo, $\mathrm{CH}_{2}$ ), 42.1 (exo, $\mathrm{CH}_{2}$ ), 42.4 (endo, CH ), 44.1 (exo, CH), 72.8 (endo, CH), 74.7 (exo, CH).

## (F) Asymmetric Hydrogenation of ( $\pm$ )-2-Diphenylmethyl-3-quinuclidinone [ $\pm$ )10]

Solid ( $S, R$ )-3b ( $2.6 \mathrm{mg}, 0.0027 \mathrm{mmol}$ ) and ( $\pm$ )-2-diphenylmethyl-3-quinuclidinone ${ }^{6,7}$ $[( \pm)-10](7.71 \mathrm{~g}, 26.5 \mathrm{mmol})$ were placed in a 200 mL SUS autoclave with a Teflon-
coated magnetic stirring bar. Air present in the autoclave was replaced by argon. 2Propanol ( 108 mL ) and dimethylacetamide ( 18 mL ) were added to the autoclave. Argon was introduced into the autoclave at a pressure of 5 atm , then released to 1 atm . This procedure was repeated 5 times. A solution of $t-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{OK}$ in 2-propanol (1 M, 2.6 $\mathrm{mL}, 2.6 \mathrm{mmol}$ ) which had been degassed was added to the autoclave. Hydrogen was initially introduced into the autoclave at a pressure of 5 atm , before being reduced to 1 atm by carefully releasing the stop valve. After this procedure was repeated 5 times, the vessel was pressurized to 50 atm . The reaction mixture was vigorously stirred at $25^{\circ} \mathrm{C}$. After stirring for 24 h and carefully venting the hydrogen gas, the solvent was removed under reduced pressure. The residue was purified by silica-gel (Fuji silysia, Chromatorex NH) column chromatography giving (2S,3S)-2-diphenylmethyl-3quinuclidinol ${ }^{8}[(S, S)-\mathbf{1 1}]$ (colorless powder, $7.73 \mathrm{~g}, 99 \%$ yield, $>99 \%$ ee) as a single diastereomer. The enantiomeric excess of $\mathbf{1 1}$ was determined by HPLC analysis. Column, CHIRALCEL OD-RH; eluent, $\mathrm{CH}_{3} \mathrm{CN}: 0.1 \mathrm{M}$ aq. $\mathrm{KPF}_{6}=70: 30$; flow, 0.5 $\mathrm{mL} \mathrm{min}{ }^{-1}$; column temp, $25^{\circ} \mathrm{C}$; retention time $\left(t_{\mathrm{R}}\right)$ of $(R, R)$-11, $33.7 \mathrm{~min}\left(t_{\mathrm{R}}\right.$ of racemic 11) $\left(<0.5 \%\right.$ not detected); $t_{\mathrm{R}}$ of $(S, S)-\mathbf{1 1}, 41.1 \mathrm{~min}(>99.5 \%) . \quad[\alpha]_{\mathrm{D}}{ }^{26}+11.6(c \quad 1.0$, $\left.\mathrm{CHCl}_{3}\right)\left(\right.$ lit. ${ }^{8}[\alpha]_{\mathrm{D}}+11.6\left(c \quad 1, \mathrm{CDCl}_{3}\right), 99.5 \%$ ee $\left.(S, S)\right) . \quad$ IR (KBr-disk) 3430 (br), 3024, 2935, 2869, 1597, 1495, 1450, 1066, 1042, 755, 741, $702 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 1.12-1.32(\mathrm{~m}, 1 \mathrm{H}), 1.35(\mathrm{~d}, 1 \mathrm{H}, J=3.3 \mathrm{~Hz}), 1.47-1.69(\mathrm{~m}, 2 \mathrm{H}), 1.88-1.96$ $(\mathrm{m}, 2 \mathrm{H}), 1.61-1.69(\mathrm{~m}, 1 \mathrm{H}), 2.77-2.83(\mathrm{~m}, 2 \mathrm{H}), 3.14-3.25(\mathrm{~m}, 1 \mathrm{H}), 3.61-3.67(\mathrm{~m}, 1 \mathrm{H})$, 3.94-3.99 (m, 1H), $4.48(\mathrm{~d}, 1 \mathrm{H}, J=12.0 \mathrm{~Hz}), 7.07-7.34(\mathrm{~m}, 8 \mathrm{H}), 7.42-7.46(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 19.2\left(\mathrm{CH}_{2}\right), 24.8\left(\mathrm{CH}_{2}\right), 28.4(\mathrm{CH}), 41.5\left(\mathrm{CH}_{2}\right), 49.0$ $\left(\mathrm{CH}_{2}\right), 49.2(\mathrm{CH}), 63.3(\mathrm{CH}), 68.9(\mathrm{CH}), 126.1(\mathrm{CH}), 126.7(\mathrm{CH}), 127.6(\mathrm{CH}), 127.7$ $(\mathrm{CH}), 128.4(\mathrm{CH}), 129.3(\mathrm{CH}), 143.6(\mathrm{C}), 144.3(\mathrm{C}) . \quad$ HRMS ( $\left.\mathrm{EI}^{+}\right) \mathrm{m} / z 293.1778\left(\mathrm{M}^{+}\right)$, calcd for $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{NO}: 293.1780$. Found: C, $81.36 \%, \mathrm{H}, 8.01 \%$; N, 4.67\%. Calcd for $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{NO}: \mathrm{C}, 81.87 \% ; \mathrm{H}, 7.90 \% ; \mathrm{N}, 4.77 \%$.

## (G) Comment on the Catalytic Cycle

A plausible catalytic cycle for the hydrogenation of ketones with BINAP/1,4-diamineRu complexes 3 in 2-propanol is shown in Scheme S1 by analogy with the mechanism of BINAP/DPEN-Ru catalyzed hydrogenation. ${ }^{9}$ The excellent activity of this catalyst is rationalized by a non-classical "metal-ligand cooperative mechanism" using the NH functionality. The precatalyst dichloride complex $\mathbf{A}$ is converted to the cationic species $\mathbf{B}$ by releasing two HCl molecules with the assistance of a base, followed by hydride donation from 2-propanol or $\mathrm{H}_{2}$. Then, $\boldsymbol{B}$ reacts with an $\mathrm{H}_{2}$ molecule to form a cationic intermediate $\mathbf{C}$, which undergoes deprotonation with a solvent molecule to afford the active $\mathrm{RuH}_{2}$ species $\mathbf{D}$. This process is promoted by a base. Ketone is promptly reduced by $\mathbf{D}$, resulting in the alcoholic product and the 16 -electron $\mathrm{Ru}-$ amide complex E. This species is easily protonated in an alcoholic solvent to regenerate the cationic amino complex $\mathbf{B}$, while it partially returns to $\mathbf{D}$ by reaction with $\mathrm{H}_{2}$. The active species $\mathbf{D}$ has a $f a c$-structure for the hydride and two nitrogen atoms of the diamine, so that this species and a ketonic substrate react smoothly through the sixmembered pericyclic transition state $\mathbf{F}$. Ketone is hydrogenated in the outer coordination sphere of $\mathbf{D}$, where neither ketone $/ \mathrm{Ru}$ nor alkoxy $/ \mathrm{Ru}$ interaction is involved.


SCHEME S1. Plausible Catalytic Cycle for BINAP/1,4-diamine-Ru-catalyzed Hydrogenation of Ketones. $\mathrm{P}-\mathrm{P}=(S)$-BINAP. $\mathrm{NH}_{2}-\mathrm{NH}_{2}=1,4$-diamine.

## (H) References

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## (I) Spectral Charts


$\square$
(s)






C: \WINNMR98\COMMON \_DEFAULT.ALS
(S,R)-3d
$\qquad$

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ss_433 13C
(R)-2a











