## SUPPORTING INFORMATION

## Surfactant-free RAFT emulsion polymerization using poly( $\mathbf{N}, \mathbf{N}$-dimethyl acrylamide) trithiocarbonate macromolecular chain transfer agents.

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## A- NMR characterization of the RAFT agents TTCA-12, TTCA-4 and ATTCA




acetone

Figure SI-1. $250 \mathrm{MHz}^{1} \mathrm{H}$ NMR spectrum of TTCA-12 in DMSO-d6 and $63 \mathrm{MHz}{ }^{13} \mathrm{C}$ NMR spectrum in acetone-d6.
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Figure SI－2． $250 \mathrm{MHz}^{1} \mathrm{H}$ NMR and $63 \mathrm{MHz}{ }^{13} \mathrm{C}$ NMR spectra of TTCA－4 in $\mathrm{CDCl}_{3}$ ．


a
DMSO





Figure SI-3. $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR and $50 \mathrm{MHz}{ }^{13} \mathrm{C}$ NMR spectra of ATTCA in DMSO-d6.

## B- SEC characterization of the PDMAAm macroRAFT agents



Figure SI-4. SEC chromatograms for the PDMAAm macroRAFT agent B3 in THF and in DMF (see Table 1 in the article).

## C- Solution polymerization of $n$-butyl acrylate and styrene in the presence of the PDMAAm macroRAFT agents and characterization of the formed diblock copolymers

## Solution polymerization of $n$-butyl acrylate ( $n B A$ )

For the synthesis of PDMAAm-PnBA block copolymers, polymerizations of $n B A$ were performed in 1,4-dioxane at $70^{\circ} \mathrm{C}$ with ACPA as an initiator, in the presence of the different macromolecular RAFT agents (see Table SI- 1 for detailed polymerization conditions). In a typical experiment (Table SI-1, entry A2-S1), the polymerization of 1.5 g $n$ BA ( 4.0 mol. $\mathrm{L}^{-1}$ ) was carried out with 126 mg PDMAAm-TTC-12, A2, ( $M_{\mathrm{n}}=7300 \mathrm{~g} . \mathrm{mol}^{-1} ; 5.8 \times 10^{-3} \mathrm{~mol} . \mathrm{L}^{-1}$ ) in 1.3 mL of 1,4 -dioxane with 1.1 mg ACPA ( 0.5 mL of a stock solution at $2.2 \mathrm{~g} . \mathrm{L}^{-1}$ ). The solution was placed in a septum-sealed flask, purged for 30 min with nitrogen in an ice bath and heated to $70^{\circ} \mathrm{C}$ in a thermostated oil bath under stirring. Sampling was performed at regular time intervals, and the polymerization was stopped by immersion of the flask in iced water. The monomer conversion was determined by gravimetry

## Solution polymerization of styrene (S)

For the synthesis of PDMAAm-PS block copolymers, polymerizations of $S$ were performed in 1,4-dioxane at $80^{\circ} \mathrm{C}$ with ACPA as an initiator, in the presence of the different macromolecular RAFT agents (see Table SI-2 for detailed polymerization conditions). In a typical experiment (Table SI-2, entry A3-SS1), the polymerization of 1.51 $\mathrm{g} \mathrm{S}\left(2.0 \mathrm{~mol}^{\left.-\mathrm{L}^{-1}\right)}\right.$ was carried out with 211 mg PDMAAm-TTC-12, A3, $\left(M_{\mathrm{n}}=10700 \mathrm{~g} \cdot \mathrm{~mol}^{-1} ; 2.7 \times 10^{-3} \mathrm{~mol}^{-\mathrm{L}^{-1}}\right)$ in 5.6 mL of 1,4 -dioxane with 1.3 mg ACPA ( 0.5 mL of a stock solution at $2.6 \mathrm{~g} . \mathrm{L}^{-1}$ ). The solution was placed in a septum-sealed flask, purged for 30 min with nitrogen in an ice bath and heated to $80^{\circ} \mathrm{C}$ in a thermostated oil bath under stirring. The monomer conversion was determined by gravimetry.

Table SI-1. Experimental conditions and results for the solution polymerizations of $n B A$ in 1,4 -dioxane at $70^{\circ} \mathrm{C}$ in the presence of different PDMAAm-TTC macroRAFT agents.

| Entry | Macro-RAFT |  | $\begin{gathered} {[n B A]_{0}} \\ \left(\mathrm{~mol}^{-1}\right) \end{gathered}$ | $\begin{gathered} {[\text { Macro-RAFT] }]_{0}} \\ \left(\text { mmol.L }{ }^{-1}\right) \end{gathered}$ | $\begin{gathered} {[\mathrm{ACPA}]_{0}} \\ \left(\mathrm{mmol} . \mathrm{L}^{-1}\right) \end{gathered}$ | $\begin{gathered} {D P_{n, t h}{ }^{b}}_{\text {PnBA block }} \end{gathered}$ |  | conv.$(\%)^{c}$ | $\begin{gathered} M_{\mathrm{n}, \mathrm{th}}{ }^{b} \\ \left(\mathrm{~kg} \cdot \mathrm{~mol}^{-1}\right) \end{gathered}$ | $\begin{gathered} M_{\mathrm{n}, \exp }{ }^{d} \\ \left(\mathrm{~kg} \cdot \mathrm{~mol}^{-1}\right) \end{gathered}$ | $M_{\mathrm{w}} / M_{\mathrm{n}}^{d}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | \# | $M_{n}{ }^{\text {a }}$ |  |  |  |  |  |  |  |  |  |
| A1-S1 | A1 | 3.2 | 4.0 | 13 | 2.7 | 300 | 30 | 42 | 19.1 | 19.9 | 1.15 |
|  |  |  |  |  |  |  | 52 | 64 | 27.7 | 26.3 | 1.14 |
| A2-S1 | A2 | 7.3 | 4.0 | 5.8 | 1.4 | 690 | 25 | 25 | 35.4 | 42.5 | 1.27 |
|  |  |  |  |  |  |  | 90 | 35 | 43.3 | 48.7 | 1.22 |
| A2-S2 ${ }^{\text {e }}$ | A2 | 7.3 | bulk | 8.9 | $0.9{ }^{\text {d }}$ | 790 | 160 | 70 | 78.0 | 77.9 | 1.16 |
| B1-S1 | B1 | 4.2 | 2.0 | 2.8 | 0.7 | 710 | 100 | 43 | 43.2 | 28.4 | 1.36 |
|  |  |  |  |  |  |  | 190 | 73 | 70.3 | 38.5 | 1.46 |
|  |  |  |  |  |  |  | 270 | 87 | 83.0 | 42.0 | 1.50 |
| B2-S1 | B2 | 8.7 | 2.0 | 3.1 | 0.7 | 650 | 40 | 33 | 36.4 | 26.7 | 1.36 |
|  |  |  |  |  |  |  | 65 | 50 | 50.6 | 32.2 | 1.42 |
|  |  |  |  |  |  |  | 240 | 90 | 84.1 | 39.8 | 1.58 |
| B3-S1 | B3 | 12.4 | 4.0 | 6.6 | 1.4 | 600 | 43 | 66 | 63.4 | 50.7 | 1.28 |
| C1-S1 | C1 | 6.6 | 4.0 | 5.5 | 1.3 | 730 | 48 | 37 | 41.7 | 35.1 | 1.35 |
|  |  |  |  |  |  |  | 94 | 65 | 67.6 | 52.6 | 1.27 |
|  |  |  |  |  |  |  | 150 | 79 | 80.7 | 59.0 | 1.26 |
| C2-S1 | C2 | 9.8 | 4.0 | 6.5 | 1.3 | 620 | 31 | 19 | 24.9 | 15.7 | 1.81 |
|  |  |  |  |  |  |  | 51 | 33 | 35.9 | 23.8 | 1.56 |
| C2-S2 | C2 | 9.8 | 2.0 | 3.8 | 0.8 | 520 | 41 | 33 | 31.9 | 22.3 | 1.75 |
|  |  |  |  |  |  |  | 69 | 52 | 44.7 | 33.0 | 1.51 |
|  |  |  |  |  |  |  | 145 | 76 | 60.8 | 36.7 | 1.52 |
|  |  |  |  |  |  |  | 258 | 91 | 70.9 | 37.5 | 1.54 |

${ }^{a} M_{n}$ determined by SEC in DMF by light scattering (LS) ( $\left.d n / d c=0.081 \mathrm{~mL} \cdot g^{-1}\right) ;{ }^{b}$ theoretical number-average molar mass, $M_{\mathrm{n}, \mathrm{th}}$, and number-average degree of polymerization, $\mathrm{DP}_{\mathrm{n}}$, calculated using conversion ${ }^{c}$; ${ }^{c}$ Monomer conversion determined by gravimetry; ${ }^{d} M_{\mathrm{n}}$ and $M_{\mathrm{w}} / M_{\mathrm{n}}$ determined by SEC in THF with a PS calibration; ${ }^{e}$ Experiment A2-S2 was performed in bulk with 2,2'-azobis(isobutyronitrile) (AIBN) as initiator.


Figure SI-5. Evolution of the monomer conversion with time for the solution polymerization of $n \mathrm{BA}$ in the presence of different PDMAAm macroRAFT agents ( $[n B A]_{0}=2 \mathrm{~mol}^{-\mathrm{L}^{-1}}$ (B2-S1, C2-S2) or $4 \mathrm{~mol} \mathrm{~L}^{-1}$ (A1-S1, C1-
 higher initiator concentration being responsible for the faster kinetics.


Figure SI-6. Solution polymerization of $n B A$ in 1,4-dioxane in the presence of different PDMAAm macroRAFT agents (A1, B2, C1 and C2): a) an example of the evolution of the size exclusion chromatograms with monomer conversion (Experiment C1-S1); b) evolution of the number average molar mass, $M_{\mathrm{n}}$, (full symbols) and polydispersity index $\left(M_{w} / M_{\mathrm{n}}\right)$ (open symbols) with time (determined by SEC in THF and calculated with a PS calibration curve). The straight lines correspond to the theoretical $M_{\mathrm{n}} v s$. conversion.

Table SI-2. Experimental conditions and results for the solution polymerizations of styrene (S) in 1,4-dioxane at $80^{\circ} \mathrm{C}$ in the presence of PDMAAm-TTC-12, A1 and A3. ${ }^{\text {a }}$

| Entry |  | croRAFT | $\begin{gathered} \mathrm{DP}_{\mathrm{n}}{ }^{c} \\ (\mathrm{~S}) \end{gathered}$ | $\begin{gathered} {[\text { RAFT }]_{0}} \\ (\text { mmol.L } \end{gathered}$ | $\begin{gathered} {[\text { ACPA }]_{0}} \\ \left(\text { mmol. }{ }^{-1}\right) \end{gathered}$ | t <br> (h) | conv. ${ }^{\text {b }}$ (\%) ${ }^{a}$ | $\begin{gathered} M_{\mathrm{n}, \mathrm{th}}{ }^{c} \\ \left(\mathrm{~kg} \cdot \mathrm{~mol}^{-1}\right) \end{gathered}$ | $\begin{gathered} M_{\mathrm{n}, \exp }^{d} \\ \left(\mathrm{~kg}_{\mathrm{mol}}{ }^{-1}\right) \end{gathered}$ | $M_{w} / M_{n}{ }^{\text {d }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | \# | $\begin{gathered} M_{\mathrm{n}} \\ \left(\mathrm{~kg} \cdot \mathrm{~mol}^{-1}\right) \end{gathered}$ |  |  |  |  |  |  |  |  |
| A1-SS1 | A1 | 3.2 | 620 | 3.2 | 0.6 | 20 | 19 | 15.5 | 12.4 | 1.44 |
| A3-SS1 | A3 | 10.7 | 736 | 2.7 | 0.6 | 20 | 20 | 25.6 | 14.3 | 1.46 |

${ }^{a}[\mathbf{S}]_{0}=2 \mathrm{M} ;{ }^{b}$ Monomer conversion by gravimetry; ${ }^{c}$ Theoretical number-average molar mass, $M_{\mathrm{n}, \mathrm{th}}$, and numberaverage degree of polymerization, $\mathrm{DP}_{\mathrm{n}}$, calculated using conversion ${ }^{b} ;{ }^{d} M_{n}$ and $M_{w} / M_{n}$ determined by SEC in THF with a PS calibration.


Figure SI-7. SEC chromatograms in THF for sample A3-SS1 prepared by polymerization of styrene in 1,4dioxane at $80^{\circ} \mathrm{C}$ with macroRAFT agent A3.

