

Supporting Information

Self-assembled Inverted Micelles Stabilize Ionic Liquid Domains in Supercritical CO₂

Aneesh Chandran, Karthigeyan Prakash, and Sanjib Senapati

Table of Contents:

1. Table S1: Force field parameters for the ionic liquids, surfactants, and CO₂.
2. Table S2: Comparison of simulated crystal structure of N-EtFOSA with experimental data from the Cambridge structural database.
3. Supporting figures - Figs. S1, S2, S3.

Table S1: Force field parameters used. For atom notations, please refer to Fig. S1.

Bond Parameters

Bond	K _r (kJ mol ⁻¹ Å ⁻²)	r ₀ (Å)	Bond	K _r (kJ mol ⁻¹ Å ⁻²)	r ₀ (Å)
[N-EtFOSA]^[24]					
CT-HC ^a	1422.56	1.090	N-S	1815.86	1.670
CT-CT ^a	1297.04	1.526	S-O	2928.80	1.441
CT-H1 ^a	1422.56	1.090	S-CF ^c	985.00	1.818
CT-N ^a	1410.00	1.480	CF-F	1535.53	1.350
N-H ^a	1716.28	1.018	CF-CF	1121.30	1.550
[TMG]⁺ [23]					
CT-H1	1464.40	1.085	CA-N	1464.40	1.332
N1-CT	1255.20	1.467	N-H	2050.16	0.995
CA-N1	1464.40	1.330			
[BMIM]⁺ [22]					
CT-H1	2845.00	1.080	CW-H4	2845.00	1.080
CT-NA	2820.00	1.467	CR-H5	2845.00	1.080
CW-NA	3574.00	1.381	CT-CT	2242.00	1.528
CR-NA	3992.00	1.315	CT-HC	2845.00	1.090
CW-CW	4352.00	1.340			

[Ac]⁻ [23], [PF₆]⁻ [22], [NO₃]⁻ [22], [BF₄]⁻ [22] and CO₂ [26]

C-O2	1553.69	1.264	N-O	5307.00	1.227
C-CT	727.56	1.535	B-F	1213.36	1.392
C-HC	882.99	1.094	C _c -O _c	-	1.149
P-F	3100.00	1.605			

Angle Parameters

Angle	K _θ (kJ mol ⁻¹ rad ⁻²)	θ ₀ (deg)	Angle	K _θ (kJ mol ⁻¹ rad ⁻²)	θ ₀ (deg)
-------	--	----------------------	-------	--	----------------------

[NEtFOSA]⁺ [24]

H-CT-H ^a	146.44	109.5	N-S-CF ^c	408.00	100.2
CT-CT-H ^a	209.20	109.5	O-S-O	435.14	119.0
CT-CT-N ^b	334.72	109.7	O-S-CF ^c	435.00	102.6
H1-CT-N ^b	209.20	109.5	S-CF-CF ^b	261.08	111.9
CT-N-H ^b	209.20	118.0	S-CF-F ^c	347.00	111.8
CT-N-S	209.20	120.0	CF-CF-F	209.20	109.5
H-N-S	230.12	111.0	F-CF-F	322.17	109.1
N-S-O	502.08	107.0	CF-CF-CF	244.14	112.7

[TMG]⁺ [23]

H1-CT-H1	154.81	108.9	N1-CA-N1	292.88	121.9
N1-CT-H1	217.57	110.0	N-CA-N1	309.62	119.0
CT-N1-CT	292.88	115.0	CA-N-H	209.20	121.5
CA-N1-CT	209.20	121.9	H-N-H	108.78	117.0

[BMIM]⁺ [22]

H1-CT-H1	276.10	107.8	CR-NA-CW	585.80	108.5
H1-CT-NA	313.80	110.7	CW-CW-H4	292.90	130.9
CT-NA-CR	585.80	126.4	CT-CT-NA	488.30	112.7
CT-NA-CW	585.80	125.6	CT-CT-H1	313.80	110.7
NA-CW-H4	292.90	122.0	CT-CT-CT	488.30	112.7
NA-CW-CW	585.80	107.1	CT-CT-HC	313.80	110.7
NA-CR-H5	292.90	125.1	HC-CT-HC	276.10	107.8
NA-CR-NA	585.80	109.8			

[Ac]⁻ [23], [PF₆]⁻ [22], [NO₃]⁻ [22] and [BF₄]⁻ [22]

O2-C-O2	561.91	124.9	F-P-F	1165.0	90.0
O2-C-CT	592.04	117.6	O-N-O	1011.0	120.0
C-CT-HC	450.23	109.5	F-B-F	669.5	109.5
HC-CT-HC	139.33	106.8			

Dihedral Parameters

Dihedral	V_0 (kJ mol ⁻¹)	V_1 (kJ mol ⁻¹)	V_2 (kJ mol ⁻¹)	V_3 (kJ mol ⁻¹)	V_4 (kJ mol ⁻¹)
[NEtFOSA]^[24]					
HC-CT-CT-H1 ^a	0.000	0.000	0.000	1.255	0.000
HC-CT-CT-N ^b	0.000	0.000	0.000	1.941	0.000
CT-CT-N-H	0.000	0.000	0.000	0.000	0.000
CT-CT-N-S	-0.368	12.255	-10.598	2.079	0.000
H1-CT-N-H	0.000	0.000	0.000	0.000	0.000
H1-CT-N-S	0.502	5.669	-6.096	0.623	0.000
CT-N-S-O	0.000	0.000	0.000	0.000	0.000
CT-N-S-CF ^d	0.000	-19.682	-32.267	8.819	16.954
H-N-S-O	0.000	0.000	0.000	0.000	0.000
H-N-S-CF ^a	0.000	0.000	9.205	0.000	0.000
N-S-CF-F ^c	0.000	0.000	0.000	1.322	0.000
N-S-CF-CF ^d	0.000	-12.016	0.000	5.113	3.539
O-S-CF-F ^c	0.000	0.000	0.000	1.451	0.000
O-S-CF-CF	0.000	0.000	0.000	0.000	0.000
S-CF-CF-F ^b	0.000	0.000	0.000	1.464	0.000
S-CF-CF-CF ^d	0.000	-27.367	-4.527	8.192	0.000
F -CF -CF -F	0.000	-10.460	0.000	1.046	0.000
CF-CF-CF-F	0.000	1.255	0.000	1.674	0.000
CF-CF-CF-CF	0.000	27.706	3.966	-5.807	-8.862
[TMG]⁺ [23]					
CT-N1-CT-H1	0.000	0.000	0.000	1.372	0.000
CA-N1-CT-H1	0.000	0.000	0.000	0.000	0.000
N1-CA-N1-CT	0.000	0.000	23.430	0.000	0.000
N-CA-N1-CT	0.000	0.000	23.430	0.000	0.000
N1-CA-N-H	0.000	0.000	14.364	0.000	0.000
[BMIM]⁺ [22]					
H1-CT-NA-CR	0.000	0.000	0.000	0.000	0.000
H1-CT-NA-CW	0.000	0.000	0.000	0.519	0.000
X-CR-NA-X	0.000	0.000	19.460	0.000	0.000
X-CW-NA-X	0.000	0.000	12.550	0.000	0.000
X-CW-CW-X	0.000	0.000	44.980	0.000	0.000
CW-NA-CT-CT	0.000	-7.154	6.106	0.759	0.000
CR-NA-CT-CT	0.000	-5.269	0.000	0.000	0.000
NA-CT-CT-CT	0.000	-7.480	3.164	-1.203	0.000
NA-CT-CT-HC	0.000	0.000	0.000	0.367	0.000
CT-CT-CT-CT	0.000	7.280	-0.657	1.167	0.000
CT-CT-CT-HC	0.000	0.000	0.000	1.531	0.000
HC-CT-CT-HC	0.000	0.000	0.000	1.331	0.000

		$[\text{Ac}]^{- [23]}$			
HC-CT-C-O2	0.000	6.694	0.000	0.669	0.000

Nonbonded Parameters

atom ID	ϵ_{ii} (kJ mol ⁻¹)	σ_{ii} (Å)	atom ID	ϵ_{ii} (kJ mol ⁻¹)	σ_{ii} (Å)
[NEtFOSA]^[24]					
CT	0.4577	3.399	S	1.0460	3.550
HC	0.0657	2.650	O	0.7113	2.960
H1	0.0657	2.471	CF	0.2761	3.500
N	0.7113	3.250	F	0.2218	2.950
H	0.0660	1.069			
[TMG]^{+ [23]}					
CA	0.3598	3.399	CT	0.4577	3.399
N	0.7113	3.250	H1	0.0657	2.471
N1	0.7113	3.250	H	0.0657	1.069
[BMIM]^{+ [22]}					
HC	0.1255	2.500	CT	0.2761	3.500
H1	0.1255	2.500	CR	0.2929	3.550
H4	0.1255	2.420	CW	0.2929	3.550
H5	0.1255	2.420	NA	0.7113	3.250
$[\text{Ac}]^{- [23]}$, $[\text{PF}_6]^{- [22]}$, $[\text{NO}_3]^{- [22]}$, $[\text{BF}_4]^{- [22]}$, $[\text{Cl}]^{- [22]}$ and $\text{CO}_2^{[26]}$					
O2	0.8786	2.960	O _N	0.6100	2.770
C	0.3598	3.399	B	0.3975	3.581
CT	0.4577	3.399	F _B	0.2552	3.118
HC	0.0657	2.650	Cl	0.8300	3.650
P	0.8368	3.740	C _c	0.2340	2.757
F _P	0.2552	3.118	O _c	0.6690	3.033
N	0.3380	3.060			

F_P: Fluorine in $[\text{PF}_6]^{-}$; O_N: Oxygen in $[\text{NO}_3]^{-}$; F_B: Fluorine in $[\text{BF}_4]^{-}$; and c stands for CO₂. Atomic charge on chloride is set to -1.

^aFrom AMBER parameter set: a) W. D. Cornell, P. Cieplak, C. I. Bayly, I. R. Gould, K. M. Merz, D. M. Ferguson, D. C. Spellmeyer, T. Fox, J. W. Caldwell, P. A. Kollman, *J. Am. Chem. Soc.* **1995**, *117*, 5179-5197. b) J. Wang, R. M. Wolf, J. W. Caldwell, P. A. Kollman, D. A. Case, *J. Comput. Chem.* **2004**, *25*, 1157-1174.

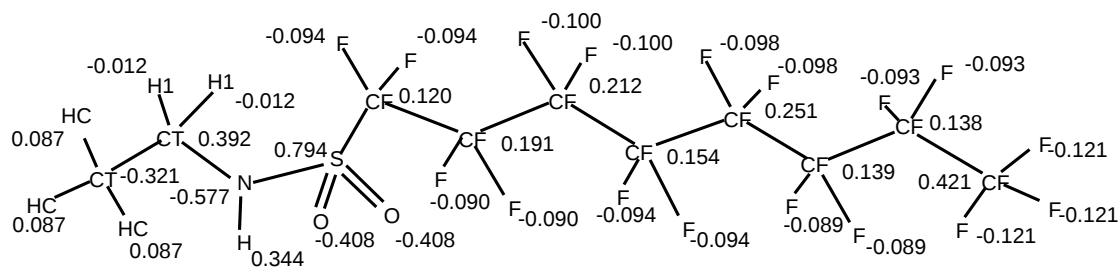
^bFrom OPLS parameter set: W. L. Jorgensen, D. S. Maxwell, J. Tirado-Rives, *J. Am. Chem. Soc.* **1996**, *118*, 11225-11236.

^cFrom Ref. 22.

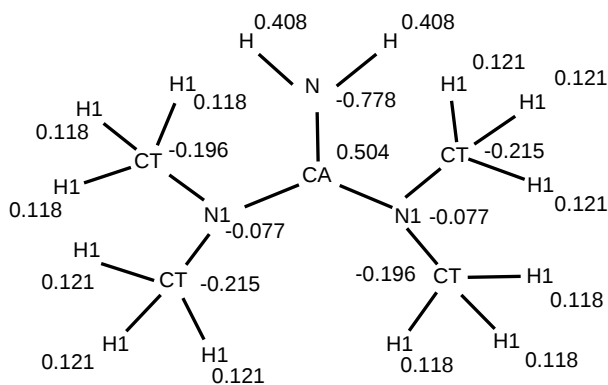
^dDeveloped in this work.

Table S2: Comparison of simulated crystal structure of N-EtFOSA with experimental data from the Cambridge structural database (Lehmmler, H. J.; Rama Rao, V.V.V.N.S.; Nauduri, D.; Vargo, J. D.; Parkin, S.; *J. Fluorine Chem.* **2007**, 128, 595–607). The objective of the simulations was to test the performance of the force field in predicting both the crystal density and its structural properties, such as the dimensions and director angles of the crystalline unit cell. A similar simulation protocol, as prescribed by Padua and coworkers, was followed.²² The liquid-state properties could not be compared due to the unavailability of experimental data.

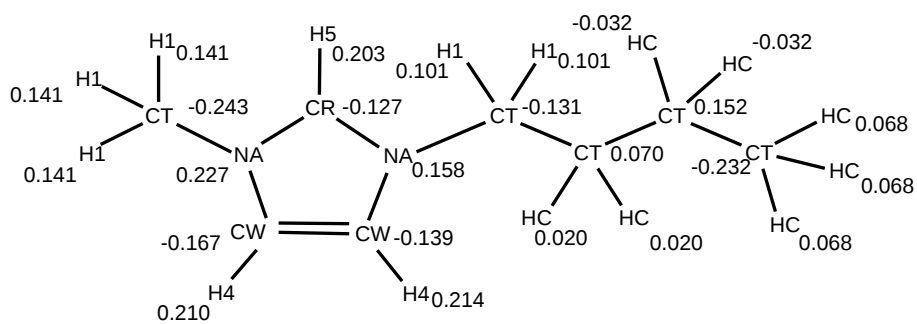
Formula	C ₂ H ₅ NHSO ₂ C ₈ F ₁₇
Molecular mass	527.22
Crystal system	Triclinic
Space group	P -1
Molecules/box	48
Stacked cells	4x3x1
T/K	90
Cut off/Å	12.0
Molecules/cell	4
a_{exp} / Å	6.341
a_{sim} / Å	6.265
b_{exp} / Å	9.567
b_{sim} / Å	9.180
c_{exp} / Å	28.230
c_{sim} / Å	28.476
α_{exp} /deg	88.579
α_{sim} /deg	90.0
β_{exp} /deg	85.758
β_{sim} /deg	90.0
γ_{exp} /deg	87.348
γ_{sim} /deg	90.0
V_{exp} / Å ³	1705.69
V_{sim} / Å ³	1636.54
ρ_{exp} /g cm ⁻³	2.053
ρ_{sim} /g cm ⁻³	2.139



(a)



(b)



(c)

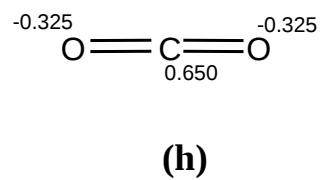
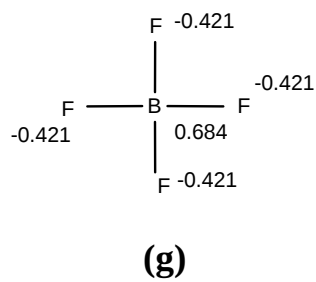
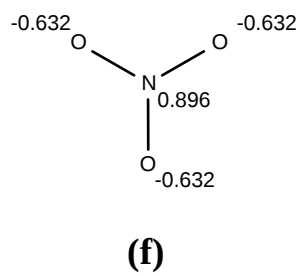
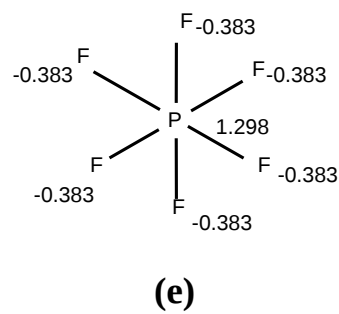
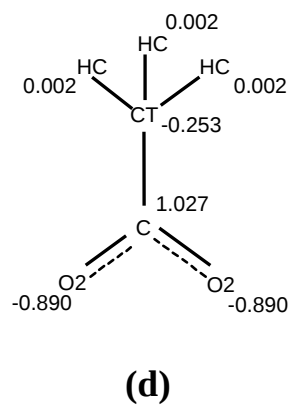


Figure S1: Atom-centered point charges on surfactant and IL ions. (a) N-EtFOSA (b) [TMG]⁺ (c) [BMIM]⁺ (d) Ac⁻ (e) PF₆⁻ (f) NO₃⁻ (g) BF₄⁻ (h) CO₂.

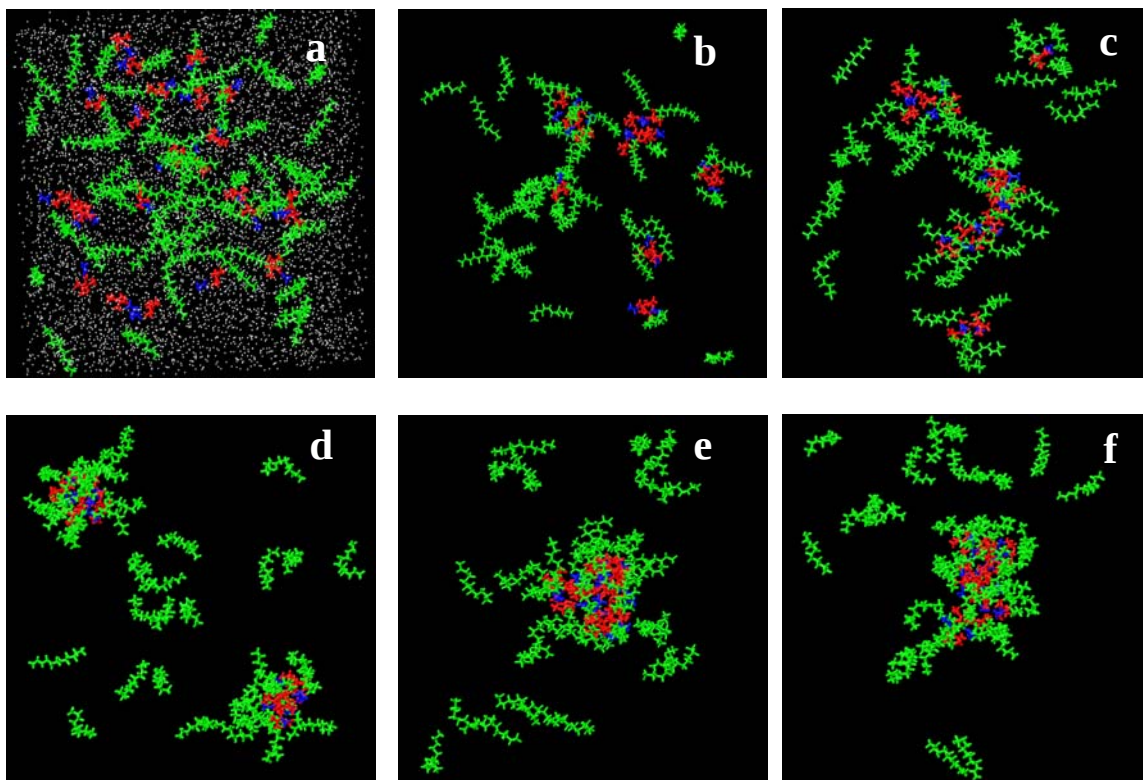


Figure S2. Time evolution of the ternary mixture of [TMG][Ac]/N-EtFOSA/CO₂ at $W_o = 0.4$. Snapshots are taken at (a) 0 ns (b) 0.5 ns (c) 30 ns (d) 70 ns (e) 120 ns and (f) 200 ns. Color scheme: red, cations; blue, anions; green, surfactants; white dots, C of CO₂. In (b) - (f), CO₂ is omitted for clarity.

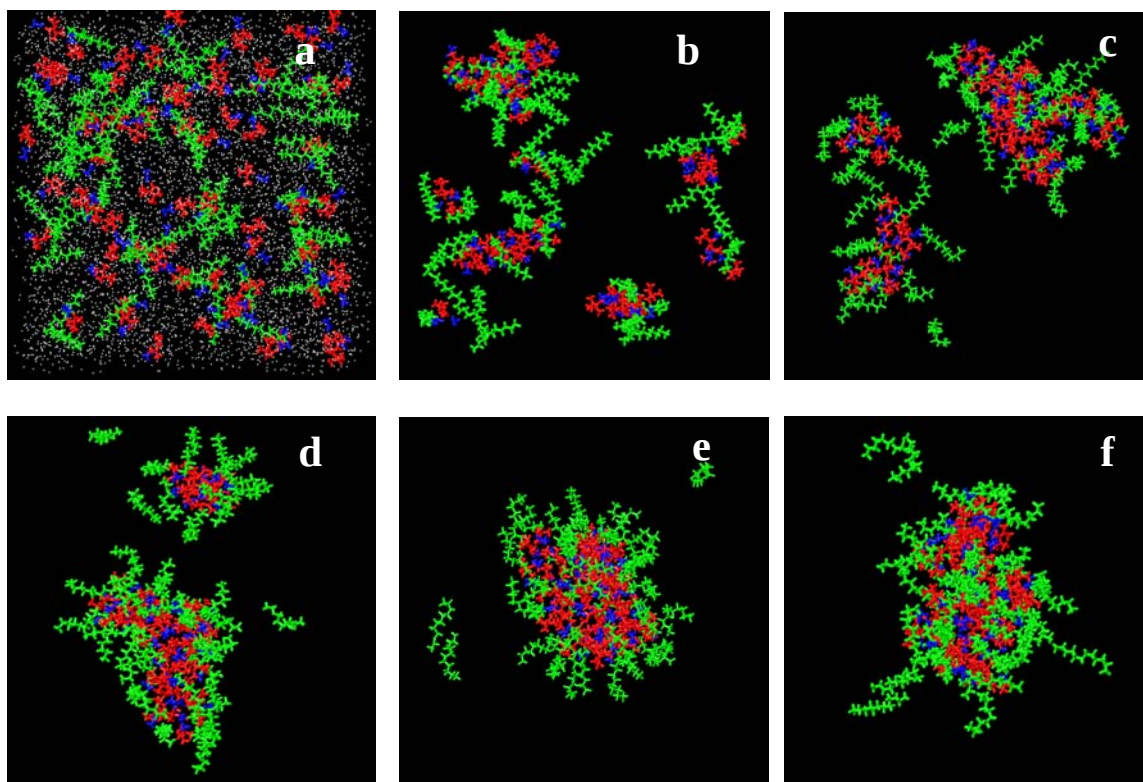


Figure S3. Time evolution of the ternary mixture of [TMG][Ac]/N-EtFOSA/CO₂ at $W_0 = 1.2$. Snapshots are taken at (a) 0 ns (b) 0.3 ns (c) 20 ns (d) 70 ns (e) 100 ns and (f) 200 ns. Color scheme: red, cations; blue, anions; green, surfactants; white dots, C of CO₂. In (b) - (f), CO₂ is omitted for clarity.

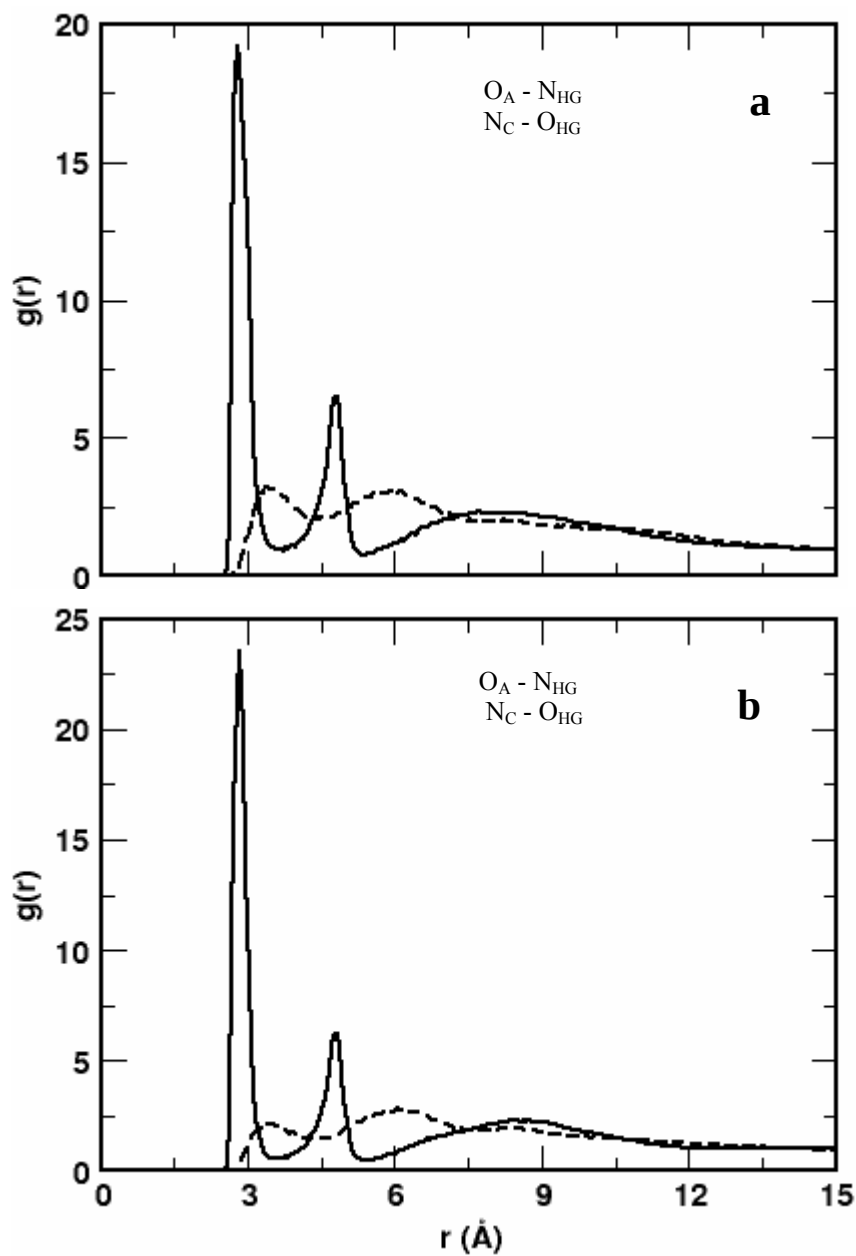


Figure S4. Radial distribution functions for the [TMG][Ac] – surfactant pairs in [TMG][Ac]/N-EtFOSA/CO₂ systems with (a) $Wo = 0.4$ and (b) $Wo = 1.2$. Solid line: anion – headgroups, dashed line: cation - headgroups. The site-site distribution functions with maximum intensity in each category are compared.

Multi-author Complete References:

(25) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Zakrzewski, V. G.; Montgomery, J. A.; Stratmann, R. E.; Burant, J. C.; Dapprich, S.; Millam, J. M.; Daniels, A. D.; Kudin, K. N.; Strain, M. C.; Farkas, O.; Tomasi, J.; Barone, V.; Cossi, M.; Cammi, R.; Mennucci, B.; Pomelli, C.; Adamo, C.; Clifford, S.; Ochterski, J.; Petersson, G. A.; Ayala, P. Y.; Cui, Q.; Morokuma, K.; Salvador, P.; Dannenberg, J. J.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Cioslowski, J.; Ortiz, J. V. B., A. G.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Gomperts, R.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B. G.; Chen, W.; Wong, M. W.; Andres, J. L.; Gonzalez, C.; Head-Gordon, M.; Replogle, E. S.; Pople, J. A.; Gaussian 03, Revision C.02; Gaussian, Inc. Wallingford CT, **2004**.

(26) Case, D.A.; Darden, T.A.; Cheatham, III, T.E.; Simmerling, C.L.; Wang, J.; Duke, R.E.; Luo, R.; Merz, K.M.; Pearlman, D.A.; Crowley, M.; Walker, R.C.; Zhang, W.; Wang, B.; Hayik, S.; Roitberg, A.; Seabra, G.; Wong, K.F.; Paesani, F.; Wu, X.; Brozell, S.; Tsui, V.; Gohlke, H.; Yang, L.; Tan, C.; Mongan, J.; Hornak, V.; Cui, G.; Beroza, P.; Mathews, D.H.; Schafmeister, C.; Ross, W.S.; Kollman, P.A. *Amber 9.0*; : University of California, San Francisco, CA., **2006**.