## Photocatalytic and Conductive MWCNT/TiO<sub>2</sub> Nanocomposite Thin Films

*Kwadwo E. Tettey*<sup>§</sup>, *Michael Q. Yee*<sup>‡</sup>, and *Daeyeon Lee*<sup>\* §</sup>

Department of Chemical and Biomolecular Engineering, University of Pennsylvania,

Philadelphia, Pennsylvania 19104 (USA)

\* daeyeon@seas.upenn.edu

<sup>§</sup> Department of Chemical and Biomolecular Engineering, University of Pennsylvania

<sup>±</sup> Department of Bioengineering, University of Pennsylvania

## SUPPORTING INFORMATION

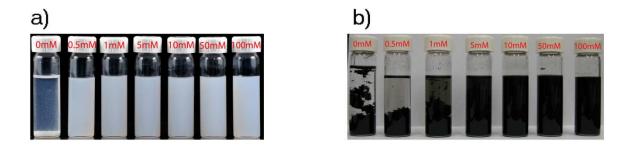
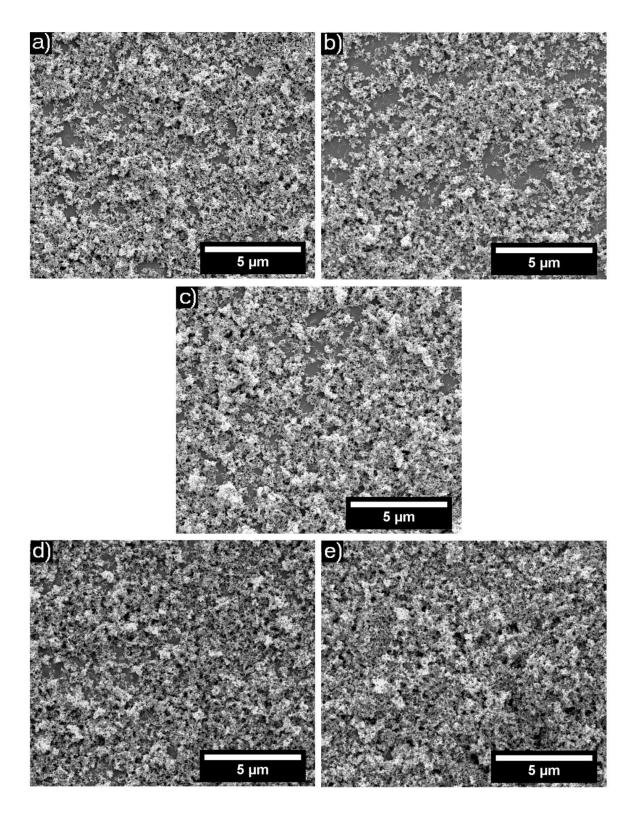
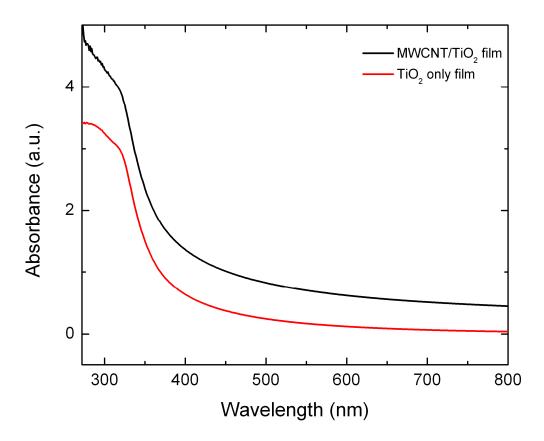


Figure S1. (a) TiO<sub>2</sub> and (b) MWCNTs dispersed in toluene containing different AOT

concentrations shown in red text.



**Figure S2.** SEM images for 20-bilayer MWCNT/TiO<sub>2</sub> films composed with (a) 5 mM, (b) 10 mM, (c) 50 mM, (d) 100 mM and (e) 200 mM AOT.



**Figure S3.** UV–Vis absorbance spectra for MWCNT/TiO<sub>2</sub> film (black line) and TiO<sub>2</sub> only film (red line).

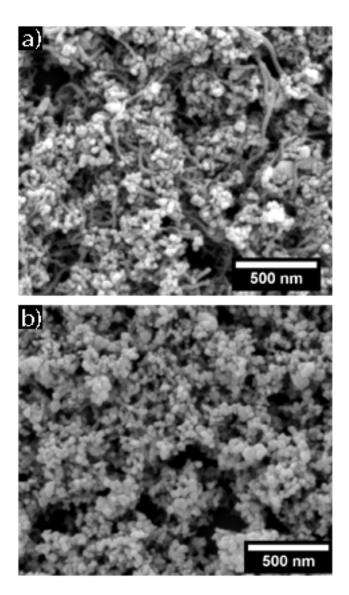
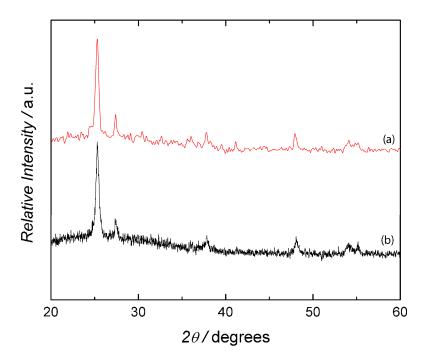


Figure S4. SEM images showing topography of (a) MWCNT/TiO<sub>2</sub> and (b) TiO<sub>2</sub> films.



**Figure S5.** XRD pattern for (a)  $TiO_2$ –P25 calcined at 600°C for 1 hour and (b) untreated  $TiO_2$ –P25

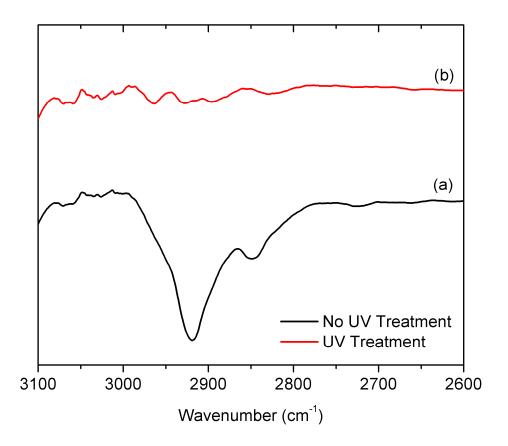


Figure S6. FTIR spectra for (a) MWCNT/TiO<sub>2</sub> nanocomposite before UV treatment (black) and (b) after UV treatment (red). Peaks between 3000 and 2800 cm<sup>-1</sup> represent alkane groups of AOT.