

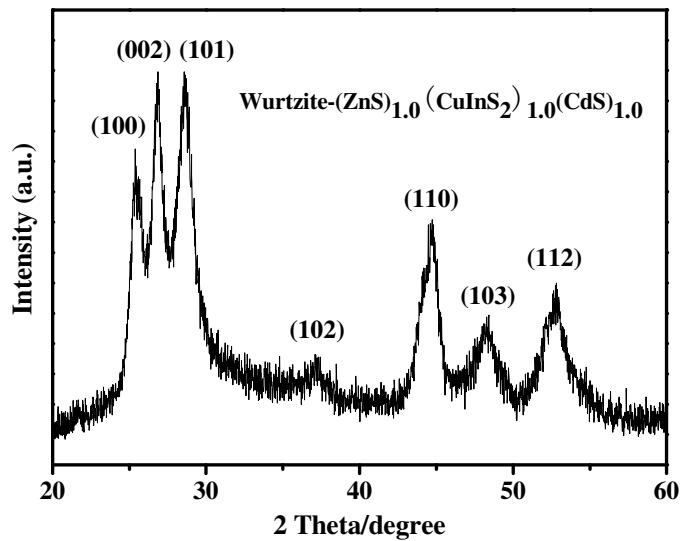
## SUPPORTING INFORMATION

### A General Synthesis of Cu-In-S Based Multicomponent Solid-Solution Nanocrystals with Tunable Band Gap, Size and Structure

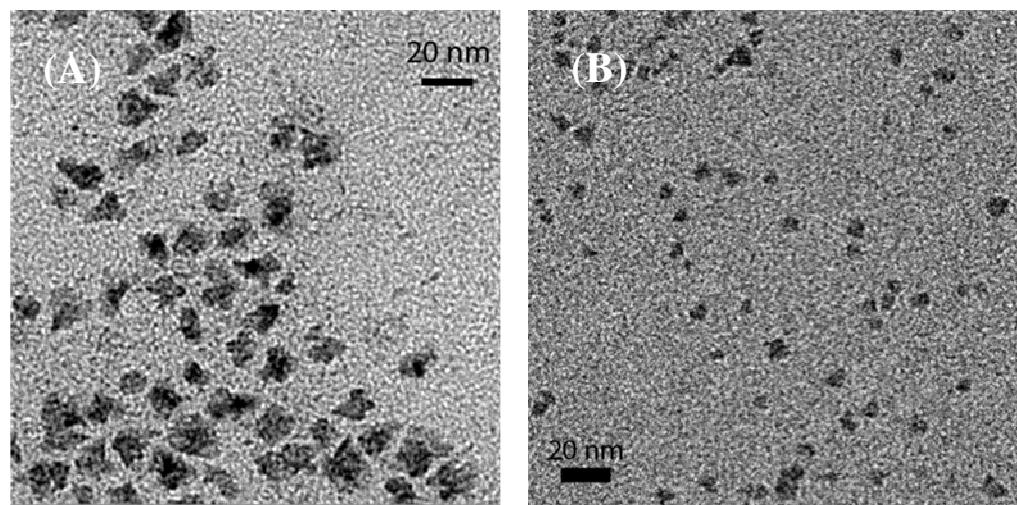
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**Table S1.** Chemical composition of the NCs of ZnS, CuInS<sub>2</sub>, CdS NCs, Zn<sub>2x</sub>(CuIn)<sub>1-x</sub>S<sub>2</sub>, (CuIn)<sub>1-x</sub>Cd<sub>2x</sub>S<sub>2</sub> and (ZnS)<sub>x</sub>(CuInS<sub>2</sub>)<sub>y</sub>(CdS)<sub>z</sub>. Formulas<sup>a</sup> were calculated by [Cu precursor]/[In precursor] ratio and Formula<sup>b</sup> by EDS results.

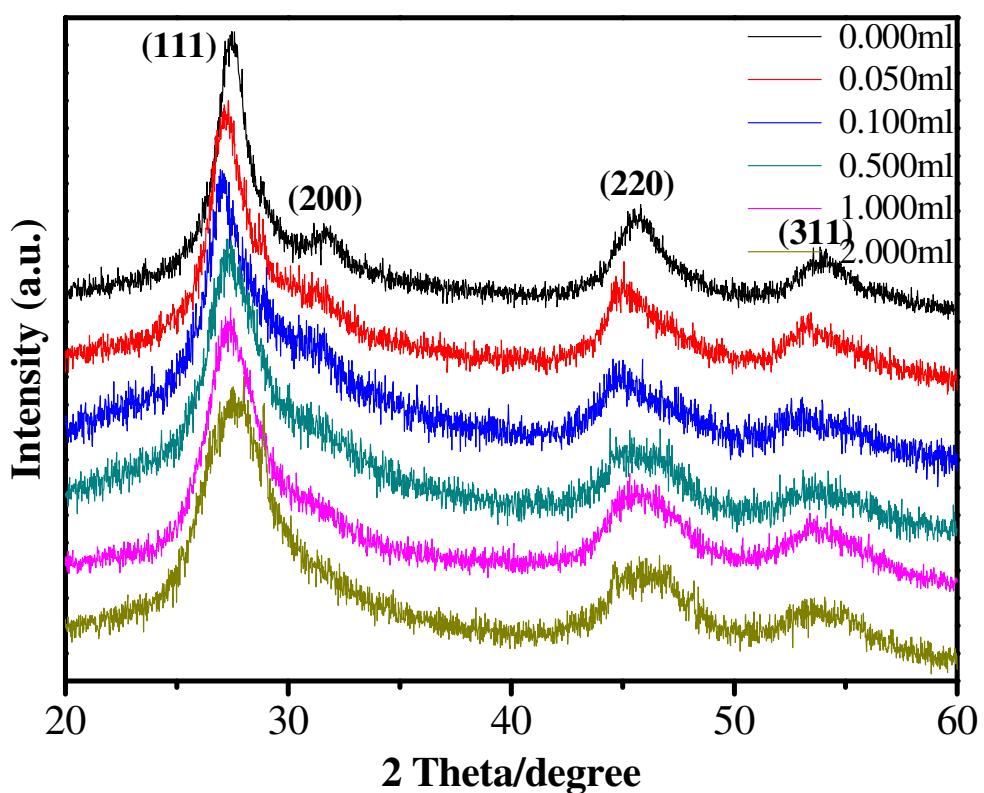
Formula <sup>a</sup>	Zn%	Cu%	In%	Cd%	Formula <sup>b</sup>
Zn <sub>1.0</sub> S <sub>1.0</sub>	44.17	----	----	----	ZnS
Zn <sub>3.0</sub> (CuIn) <sub>0.5</sub> S <sub>4.0</sub>	33.94	5.78	5.09	----	Zn <sub>3.2</sub> Cu <sub>0.6</sub> In <sub>0.5</sub>
Zn <sub>2.0</sub> (CuIn) <sub>1.0</sub> S <sub>4.0</sub>	22.39	10.62	9.97	----	Zn <sub>2.2</sub> Cu <sub>1.1</sub> In <sub>1.0</sub>
Zn <sub>1.0</sub> (CuIn) <sub>1.5</sub> S <sub>4.0</sub>	11.15	15.78	15.02	----	Zn <sub>1.1</sub> Cu <sub>1.6</sub> In <sub>1.5</sub>
Cu <sub>1.0</sub> In <sub>1.0</sub> S <sub>2.0</sub>	----	21.37	20.15	----	Cu <sub>1.1</sub> In <sub>1.0</sub>
(CuIn) <sub>1.5</sub> Cd <sub>1.0</sub> S <sub>4.0</sub>	----	15.10	14.79	10.92	Cu <sub>1.4</sub> In <sub>1.4</sub> Cd <sub>1.0</sub>
(CuIn) <sub>1.0</sub> Cd <sub>2.0</sub> S <sub>4.0</sub>	----	10.81	9.96	20.87	Cu <sub>1.0</sub> In <sub>1.0</sub> Cd <sub>2.0</sub>
(CuIn) <sub>0.5</sub> Cd <sub>3.0</sub> S <sub>4.0</sub>	----	6.08	5.75	33.53	Cu <sub>0.6</sub> In <sub>0.5</sub> Cd <sub>3.0</sub>
Cd <sub>1.0</sub> S <sub>1.0</sub>	----	----	----	46.31	CdS
(ZnS) <sub>1.0</sub> (CuInS <sub>2</sub> ) <sub>0.5</sub> (CdS) <sub>2.0</sub>	10.62	5.17	4.98	22.43	Zn <sub>0.9</sub> Cu <sub>0.5</sub> In <sub>0.4</sub> Cd <sub>2.0</sub>
(ZnS) <sub>1.0</sub> (CuInS <sub>2</sub> ) <sub>1.0</sub> (CdS) <sub>1.0</sub>	10.08	11.14	9.98	10.62	Zn <sub>0.9</sub> Cu <sub>1.0</sub> In <sub>0.9</sub> Cd <sub>1.0</sub>
(ZnS) <sub>2.0</sub> (CuInS <sub>2</sub> ) <sub>0.5</sub> (CdS) <sub>1.0</sub>	21.51	6.07	5.86	10.55	Zn <sub>2.0</sub> Cu <sub>0.6</sub> In <sub>0.5</sub> Cd <sub>1.0</sub>



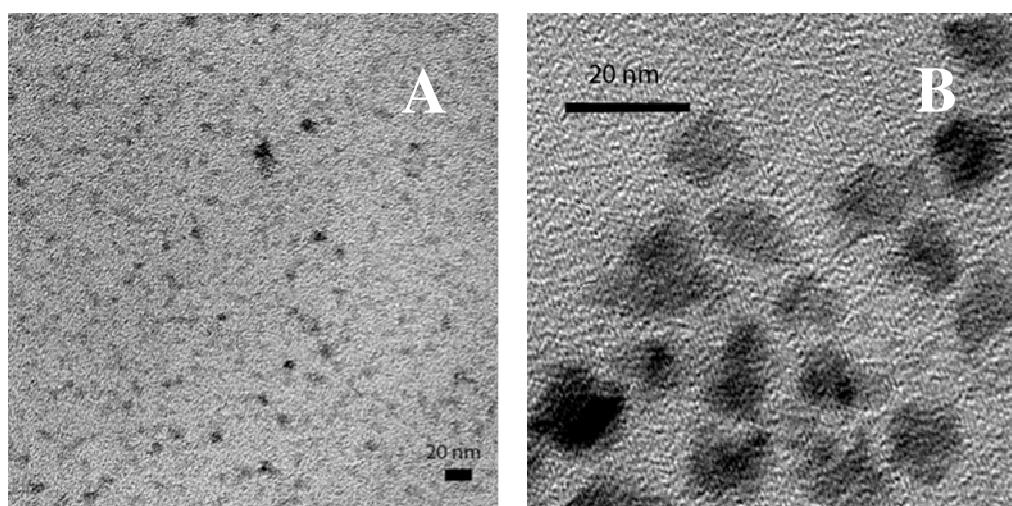
**Figure S1.** XRD patterns of  $(\text{ZnS})_{1.0}(\text{CuInS}_2)_{1.0}(\text{CdS})_{1.0}$  NCs prepared using 1-dodecanethiol was used as the capping agent showing a wurtzite structure



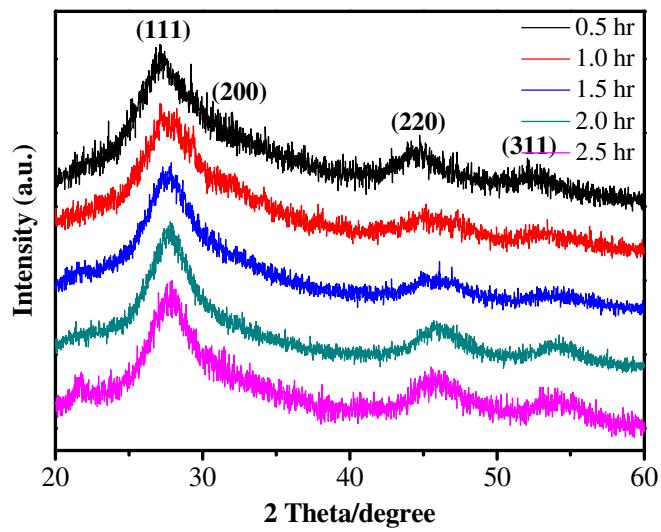
**Figure S2.** TEM images of  $(\text{ZnS})_{1.0}(\text{CuInS}_2)_{1.0}(\text{CdS})_{1.0}$  NCs with (A) 0.05 mL and (B) 0.5 mL oleylamine as the activation agent



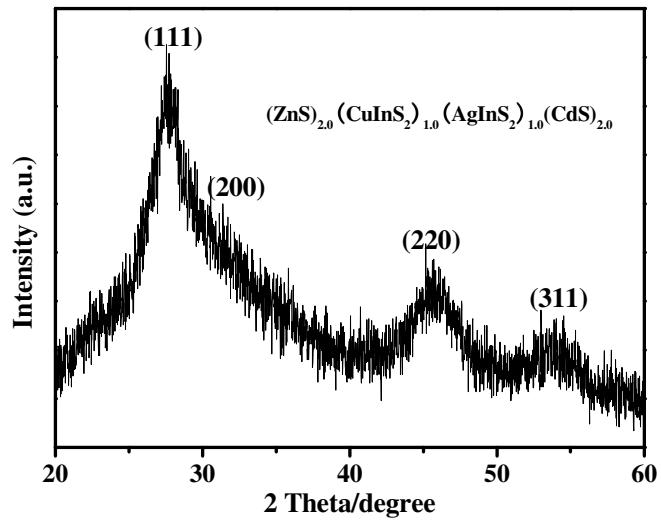
**Figure S3.** XRD patterns of (ZnS)<sub>1.0</sub>(CuInS<sub>2</sub>)<sub>1.0</sub>(CdS)<sub>1.0</sub> NCs synthesized with various oleic acid amounts (0.0 to 2.0 mL) and a fixed volume of oleylamine (1.0 mL)



**Figure S4.** TEM images of (ZnS)<sub>1.0</sub>(CuInS<sub>2</sub>)<sub>1.0</sub>(CdS)<sub>1.0</sub> NCs synthesized at (A) 100°C and (B) 200°C with a fixed volume of oleylamine and oleic acid at 1.0 mL



**Figure S5.** XRD patterns of  $(\text{ZnS})_{1.0}(\text{CuInS}_2)_{1.0}(\text{CdS})_{1.0}$  NCs synthesized at various reaction time (0.5 to 2.5 hr) with a fixed amount of oleylamine and oleic acid at 1.0 mL



**Figure S6.** XRD patterns of  $(\text{ZnS})_{2.0}(\text{CuInS}_2)_{1.0}(\text{AgInS}_2)_{1.0}(\text{CdS})_{1.0}$  NCs

