

Supporting Information
For
Synthesis, Characterization, and Reactivity of the
Heterometallic Dinuclear μ -PH₂ and μ -PPhH Complexes
FeMn(CO)₈(μ -PH₂) and FeMn(CO)₈(μ -PPhH)

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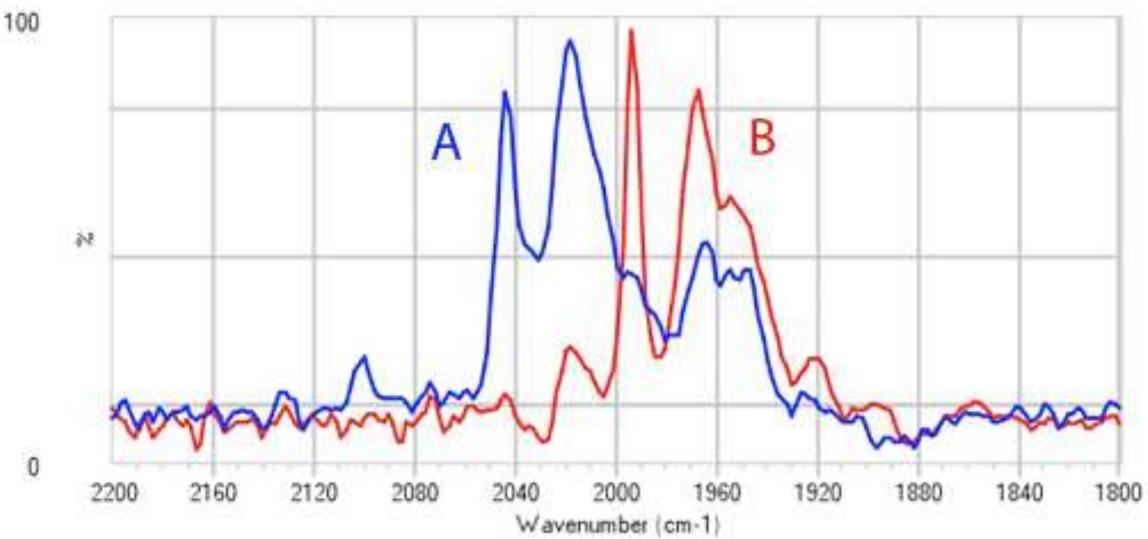


Figure S1. *In situ* FTIR spectrum of **1a** before (A) and after (B) addition of *n*-butyllithium to form [2a]⁻ in THF.

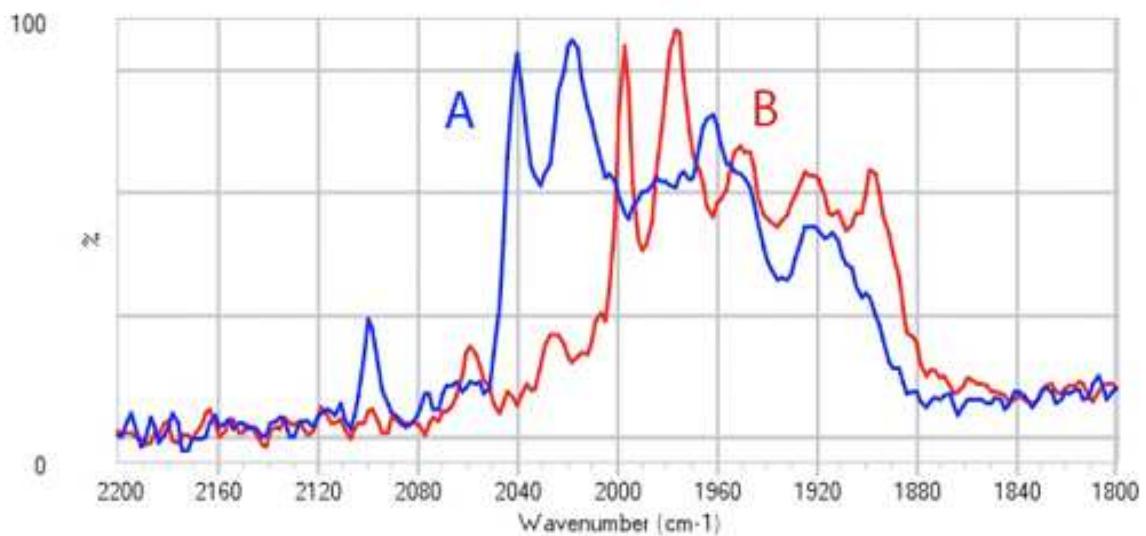


Figure S2. *In situ* FTIR spectrum of **1b** before (A) and after (B) addition of *n*-butyllithium to form [2b]⁻ in THF.

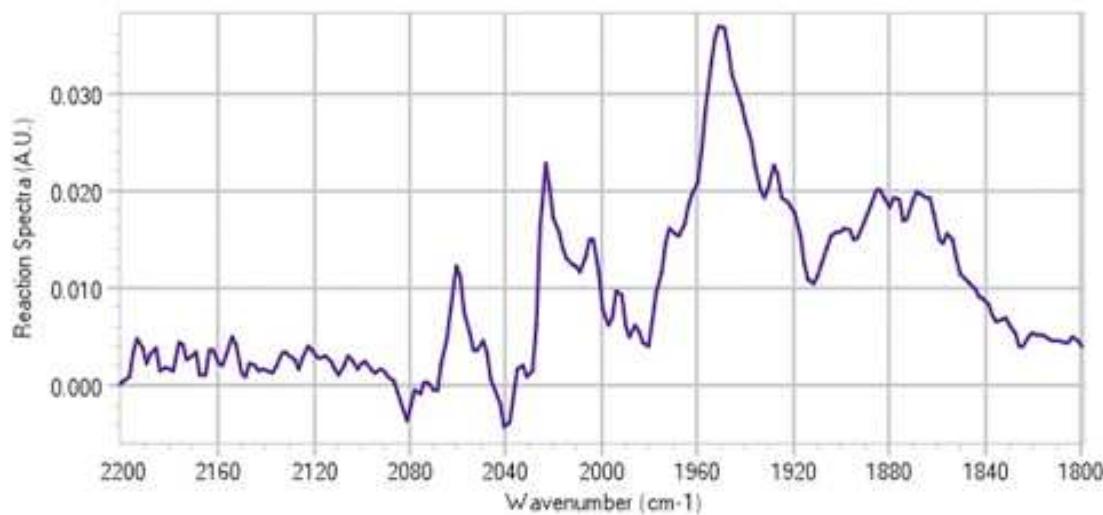


Figure S3. *In situ* FTIR spectrum obtained after stirring **[2a]** at -40 °C for 30 minutes in THF.

Exposure of this product to atmosphere immediately gave rise to **[4]**.

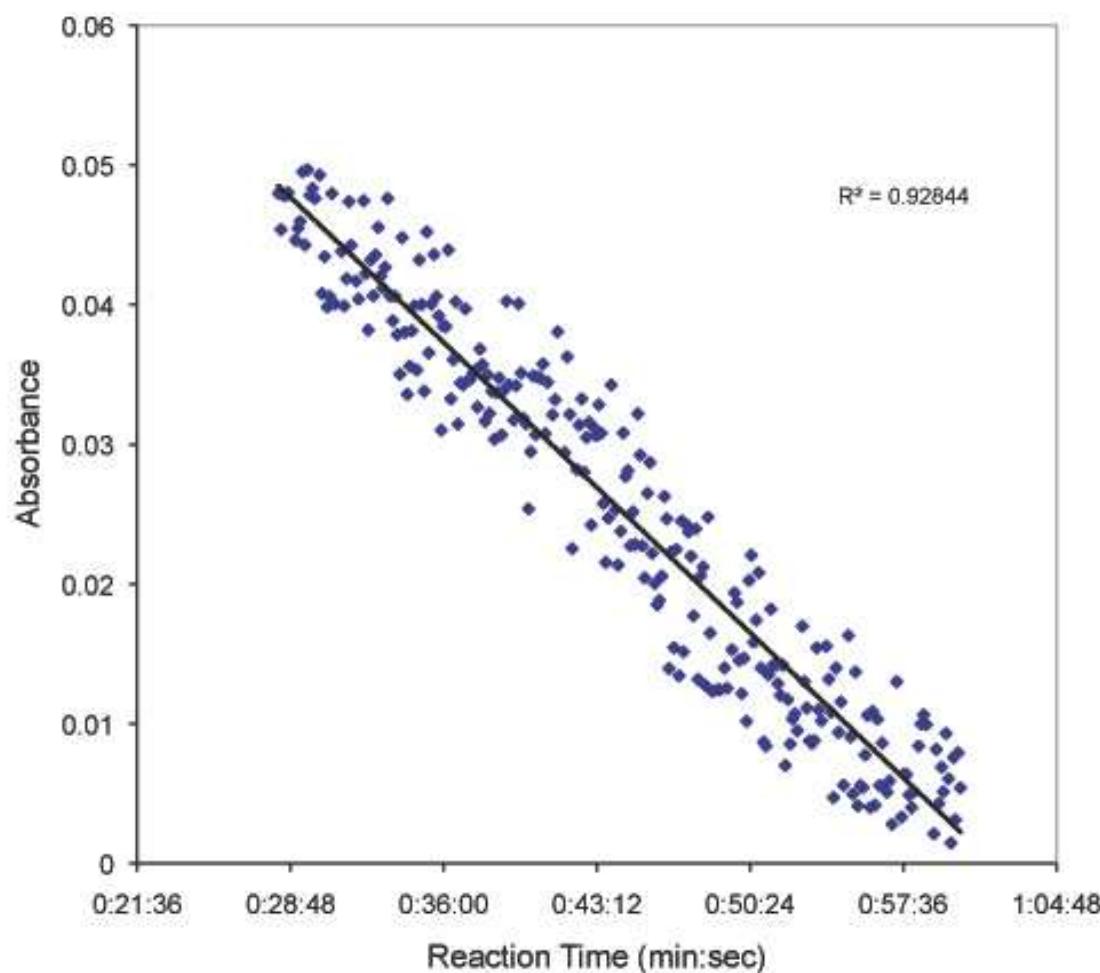


Figure S4. Change in IR absorbance of representative peak of **[2a]** (1993 cm⁻¹) over time.

	1a	1b	3a	3b
Formula	C ₈ H ₂ FeMnO ₈ P	C ₁₄ H ₆ FeMnO ₈ P	C ₁₉ H ₅ FeMn ₂ O ₁₃ P	C ₃₂ H ₂₀ AuFeMnO ₈ P ₂
Formula weight	367.86	443.95	637.93	902.18
crystal system	triclinic	monoclinic	triclinic	orthorhombic
space group	P-1	P 21/c	P-1	Pbcn
a (Å)	7.8647(8)	10.242(4)	12.325(2)	18.4784(14)
b (Å)	9.2226(10)	14.193(6)	14.204(3)	12.6296(9)
c (Å)	9.3681(10)	11.935(5)	14.926(3)	28.196(2)
α (deg)	90.966(2)	90.00	107.600(3)	90.00
β (deg)	91.141(2)	97.927(8)	106.384(3)	90.00
γ (deg)	110.032(2)	90.00	95.561(4)	90.00
V (Å ³)	638.06(12)	1718.4(13)	2341.8(8)	6580.3(8)
Z	2	4	4	8
D _{calc} (g·cm ⁻³)	1.915	1.716	1.809	1.821
λ (Mo Kα), (Å)	0.71073	0.71073	0.71073	0.71073
T (K)	223(2)	293(2)	213(2)	213(2)
θ _{max} (deg)	28.33	28.370	28.420	28.29
abs. coeff (mm ⁻¹)	2.280	1.710	1.808	5.407
rflns(coll)	5030	20831	23216	76896
rflns (unique)	2066	4226	9627	8102
Params	181	230	649	407
F(000)	360	880	1256	3488
R ₁ [I > 2σ(I)]	0.0276	0.0775	0.0310	0.0249
wR ₂ [I > 2σ(I)]	0.0591	.1328	0.0532	0.0551
GOF	0.795	1.122	0.453	1.111

Table S1: X-ray collection and refinement parameters for complexes **1a**, **1b**, **3a**, and **3b**.