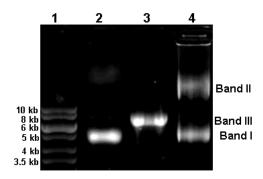
## **Supporting Information**

## DNA Cleavage System of Nanosized Graphene Oxide Sheets and Copper Ions

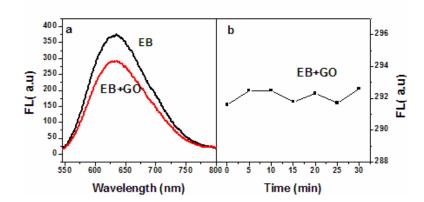
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## **Physical methods:**

Electronic absorption spectra of GO and DNA were recorded on a Cary 50 spectrophotometer (Varian, USA). IR spectra of GO was recorded on a Perkin-Nicolet FT-IR 200 in the range of 4000-400 cm<sup>-1</sup>. Samples were run as KBr pellets. Atomic Force Microscopic images of graphene oxide were taken on a Nanoscope MultiMode V scanning probe microscopy (SPM) system (Veeco, USA). The scanning rate was set usually at 0.7-1 Hz. The samples for AFM were prepared by dropping aqueous suspension (~0.02 mg mL<sup>-1</sup>) of the graphene oxide on freshly cleaved mica surface and dried under vacuum at 80 °C. A JASCO J815 spectropolarimeter (Jasco International Co. Ltd., Japan) equipped with a Jasco temperature controller (model PTC-423S) and controlled by a PC was used for all circular dichrosim measurements at 22 °C. A 1 mL quartz cell of 1 cm path length was used. Each spectrum was averaged from five successive accumulations at a scan rate of 50 nm/min. Fluorescence measurements were carried out with Hitachi f4600. Agarose gel electrophoresis was carried out with DYY-6C electrophoresis apparatus (Liuyi Instrumental Co., China). The agarose gels were visualized and digitized with FR-200A Gel Image Analysis System and analyzed by SmartView software.



**Figure S1**. Agarose gel electrophoresis of the DNA ladder (lane 1), the DNA (0.75  $\mu$ g/ml, lane 2), the DNA linearized by EcoRI (lane 3), and the DNA cleaved with GO/Cu<sup>2+</sup> (GO, 75 $\mu$ g/ml and Cu<sup>2+</sup>, 10 mM).



**Figure S2.** a) Effect of the GO on the FL intensity of EB alone (GO 10  $\mu$ g/ml, EB 12  $\mu$ g/ml), b) FL intensity variation of the EB with GO as function of time.

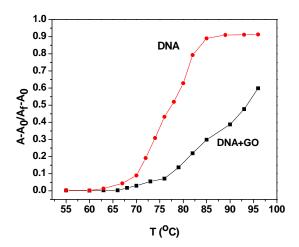
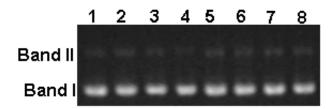
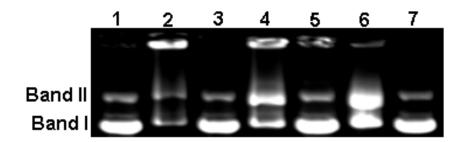


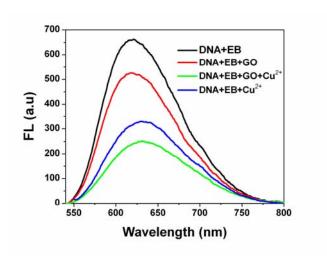
Figure S3. Thermal denaturation plots of DNA (10  $\mu$ g/mL) at 260 nm in the absence and presence of the GO 3  $\mu$ g/mL. Where  $A_0$ ,  $A_f$ , and A are the absorption intensities at 50 °C, 97 °C, and at a given temperature between 50-97 °C, respectively.  $A_f$  for the sample of the DNA with GO was calculated as  $A_f$ (DNA) + ( $A_0$ (GO)-  $A_0$ (DNA)).



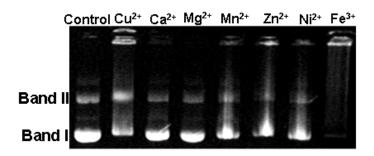
**Figure S4.** Agarose gel electrophoresis of the DNA (0.75  $\mu$ g) treated with different amount of Cu<sup>2+</sup>. Incubation time 2 hours (37 °C). **Lane 1** is control, Cu<sup>2+</sup> concentrations were **2**: 2.5, **3**: 7.5, **4**: 10, **5**: 15, **6**: 20, **7**: 25, and **8**: 27.5 mM.



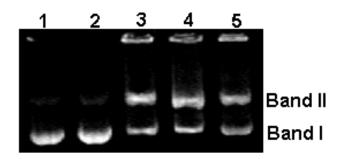
**Figure S5**. Comparison of the DNA cleavage activity of the  $GO/Cu^{2+}$  and GO alone in the buffer with the same ionic strength. DNA was 1.25 µg, incubation time 2 h (37 °C). **Lane 1** control, **Lanes 2, 4**, and **6** were DNA with GO in the presence of 2.5, 10, and 22.5 mM  $Cu^{2+}$ , respectively. **Lanes 3, 5,** and **7** are DNA with GO in the presence of 5, 20, and 45 mM  $Na^+$ , respectively.



**Figure S6.** Effects of the GO,  $Cu^{2+}$ , and  $GO/Cu^{2+}$  on the FL spectra of DNA with EB (GO 10  $\mu$ g/ml, EB 12  $\mu$ g/ml, and  $Cu^{2+}$  1 mM).



**Figure S7**. DNA cleavage activity of the GO with different metal ions incubated at 37  $^{\circ}$ C for 2 h. DNA was 1.2 µg, GO was 50 µg/mL, metal ions were all 10 mM except for Fe<sup>3+</sup> (1 mM).



**Figure S8**. Agarose gel electrophoresis of the DNA (0.75 μg) treated with 125 μg/mL GO and 10 mM  $Cu^{2+}$  in the presence of DMSO (5%) and BuOH (5%). Incubation time was 2 hours (37  $^{\circ}$ C ). **Lane 1** is control, **Lane 2**: DNA+GO, **Lane 3**: DNA+GO+Cu<sup>2+</sup>, **Lane 4**: DNA+GO+Cu<sup>2+</sup>+DMSO, and **Lane 5**: DNA+GO+Cu<sup>2+</sup>+BuOH.

**Table S1**. Cleavage of the DNA (1.25  $\mu$ g) under constant Cu<sup>2+</sup> concentration.

reagent			
$(Cu^{2+}(10 \text{ mM})/GO \mu g/mL)$	band I	band II	well
0	98	0	2
12.5	93	5	2
25	67	17	16
50	50	33	17
75	40	44	16
100	41	46	13
125	30	50	20