## **Supporting Information**

# Preparation of 4-Allylisoindoline via a Kumada Coupling with Allylmagnesium Chloride

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#### **General Methods**

Reagents and solvents were obtained from commercial sources and were used as received. Chromatography was performed using silica gel (70-230 mesh), using reagent grade solvents which were used as received.  $^{1}$ H NMR spectra were recorded at 400 MHz using Bruker Avance 400 NMR spectrometers unless otherwise noted, using the  $d_6$ -dmso resonance as an internal standard measured at 2.50 ppm.  $^{13}$ C NMR spectra were recorded on 100 or 125 MHz using Bruker Avance NMR spectrometers unless otherwise noted, using the  $d_6$ -dmso resonance as an internal standard measured at 39.5 ppm. All manipulations were carried out under an inert atmosphere of nitrogen using standard Schlenk techniques unless otherwise noted. Allyl-MgCl was purchased from Acros and was used as received. Pd(OAc)<sub>2</sub> and (neopentyl)(t-Bu<sub>2</sub>)P•HBF<sub>4</sub> were purchased from Strem Chemicals and used as received. All other reagents were purchased from Aldrich Chemical Co. unless otherwise noted.

**4-Allylisoindoline•HCl.** A 3 L flask equipped with overhead stirring, an addition funnel and a thermocouple was purged with N<sub>2</sub>, then charged under positive N<sub>2</sub> pressure with **2•HCl** (100 0.426 mol),  $Pd(OAc)_2$  (0.48) g, 2.13 (neopentyl)tBu<sub>2</sub>P•HBF<sub>4</sub> (1.3 g, 4.3 mmol). The flask was then charged with toluene (1.8 L, which had been previously de-oxygenated via N<sub>2</sub> subsurface sparging). Allylmagnesium chloride (800 mL, 1.36 mol, 1.7 M in THF) was charged to the addition funnel and then added to the slurry, with external cooling, over 1 h such that  $T_i \le 25$  °C. The resulting solution was then heated to and maintained at  $T_i = 45-50$  °C for 16 h. After cooling to ambient temperature, the reaction was inverse-quenched by slow transfer, with external cooling and at a rate such that T<sub>i</sub> < 35 °C, into a 3 L jacketed flask (equipped with a drop valve and overhead stirring) that had been charged with a solution of 15 wt% citric acid (1.1 L). The organic phase, which contained 1.2 g of 1 by assay, was rejected. The aqueous phase was assayed for 61.3 g of 1 (90% yield in organic phase, corresponding to a 92% end of reaction assay yield, along with 2.6% AY of 3). The aqueous phase was transferred to an extractor, to which was charged fresh toluene (650 mL) and the resulting biphasic mixture was stirred. Ammonium hydroxide (28-30% in water, 650 mL) was added at a rate such that  $T_i \le 30$  °C, resulting in a pH 10 aqueous phase. The phases were separated, and the aqueous phase (which contained 1.8 g of 1 by assay, 1.0 g/L) was extracted with 200 mL of toluene (0.3 g of 1 remained in the aqueous phase after extraction, or 0.18 g/L). The combined organic phases were then washed with 160 mL of 15 wt% aqueous NaCl. The toluene solution was dried via azeotropic distillation with toluene under "constant volume" conditions, then concentrated to a 650

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<sup>&</sup>lt;sup>1</sup> Evolution of propene occurs during the exothermic addition of allylmagnesium chloride at a rate proportional to the rate of addition. A mild  $N_2$  sweep efficiently removed any residual propene that was observed to evolve upon heating to  $T_i = 45$  °C.

mL total volume solution (KF = 500 ppm  $H_2O$ ). The solution was assayed by wt/wt% to contain 59.4 g of 1 (87.5% AY after workup).

The toluene solution of 1 was transferred to a 1 L flask equipped with overhead stirring, an addition funnel and a thermocouple. A solution of HCl in iPrOH (5.1 M) was transferred to the addition funnel, and then added slowly over 1 hour to the solution of 1 in toluene, at a rate such that  $T_i \le 40$  °C. The resulting slurry was aged for 6 hours, with gradual cooling to ambient temperature. The slurry was then slowly cooled to  $T_i = 0$  °C over 1 hour, then maintained at this temperature for an additional hour. The slurry was filtered, with loss to the supernatant assayed at 7.6 g/L of **1•HCl** (4.12 g total, along with 1.6 g of **3•HCl**). The cake was washed with 175 mL of 9:1 toluene:IPA (pre-cooled to T<sub>i</sub> = 0 °C), then dried with vacuum/N<sub>2</sub> sweep. **1•HCl** was isolated as an off white solid (69.5 g) that was assayed at 96 wt% (97.6 area % by HPLC), corresponding to an isolated yield of 80%. <sup>1</sup>H NMR (400 MHz,  $d_6$ -dmso)  $\delta$  9.95 (br s, 2H), 7.32 (t, J = 7.4 Hz, 1H), 7.25 (d, J = 7.4 Hz, 1H), 7.16 (d, J = 7.4 Hz, 1H), 5.96-5.84 (m, 1H), 5.10-5.04 (om, 2H), 4.48 (s, 2H), 4.45 (s, 2H), 3.37 (d, J = 6.5 Hz, 2H); <sup>13</sup>C NMR (100 MHz,  $d_6$ -dmso)  $\delta$ 135.6, 135.1, 134.6, 133.8, 128.7, 128.3, 120.8, 116.4, 49.8, 48.7, 36.9; HRMS [M+H]<sup>+</sup> for C<sub>11</sub>H<sub>13</sub>N calc'd 160.1126, found 160.1128.

### Compound 3:

The identity of 3 was initially determined using LCMS. In an effort to independently verify the structure, an authentic sample was prepared from pure 1.

A flask was charged with **1•HCl** salt (0.5 g, 3.14 mmol) and RhCl<sub>3</sub> hydrate (6 mg, 0.03 mmol), then nPrOH (15 mL). The flask was sealed under an N<sub>2</sub> atmosphere and heated to T<sub>i</sub> = 95 °C for 10 h. The solution was gradually cooled to ambient temperature, whereupon a slurry formed. The product was filtered, and the cake was washed with nPrOH. Drying afforded 300 mg of **3•HCl**. <sup>1</sup>H NMR (400 MHz,  $d_6$ -dmso)  $\delta$  9.56 (br s, 2H), 7.45 (d, J = 7.7 Hz, 1H), 7.32 (t, J = 7.7 Hz, 1H), 7.24 (d, J = 7.7 Hz, 1H), 6.44 (dd, J = 15.8, 1.5 Hz, 1H), 6.29 (dq, J = 15.8, 6.6 Hz, 1H), 4.53 (s, 2H), 4.47 (s, 2H), 1.89 (dd, J = 6.6, 1.5 Hz, 3H); <sup>13</sup>C NMR (125 MHz,  $d_6$ -dmso)  $\delta$  135.5, 132.6, 131.8, 128.8, 128.6, 127.6, 124.6, 121.1, 49.8, 49.3, 18.5; HRMS [M+H]<sup>+</sup> for C<sub>11</sub>H<sub>13</sub>N calc'd 160.1126, found 160.1130 .







