

Supporting information

Synthesis and Characterization of a Series of Model Complexes of the Active Site of [Fe]-Hydrogenase (Hmd)

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A. Crystallographic Details for [Fe(CO)₂(PPh₃)I(hp)] (2)

A total of 15689 reflections ($-17 \leq h \leq 17$, $-10 \leq k \leq 10$, $-24 \leq l \leq 24$) were collected at $T = 140(2)$ K in the range of 2.55 to 26.37° of which 4755 were unique ($R_{\text{int}} = 0.0220$); MoK α radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 0.352 and -0.233 eÅ $^{-3}$, respectively. The absorption coefficient was 2.056 mm $^{-1}$. The least squares refinement converged normally with residuals of $R(F) = 0.0202$, $wR(F^2) = 0.0432$ and a GOF = 1.014 ($I > 2\sigma(I)$). C₂₅H₁₉FeINO₃P, Mw = 595.13, space group $P2_1/c$, Monoclinic, $a = 14.0180(4)$, $b = 8.8000(2)$, $c = 19.9447(5)$ Å, $\beta = 107.747(3)^\circ$, $V = 2343.26(10)$ Å 3 , $Z = 4$, $\rho_{\text{calcd}} = 1.687$ Mg/m 3 .

B. Crystallographic Details for [Fe(CO)₂(PPh₃)I(hpp)] (3)

A total of 20330 reflections ($-14 \leq h \leq 14$, $-18 \leq k \leq 19$, $-17 \leq l \leq 18$) were collected at $T = 140(2)$ K in the range of 2.91 to 26.02° of which 5313 were unique ($R_{\text{int}} = 0.0399$); MoK α radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 0.845 and -0.339 eÅ $^{-3}$, respectively. The absorption coefficient was 1.787 mm $^{-1}$. The least squares refinement converged normally with residuals of $R(F) = 0.0304$, $wR(F^2) = 0.0657$ and a GOF = 0.991 ($I > 2\sigma(I)$). C₃₁H₂₃FeINO₃P, Mw = 671.22, space group $P2_1/n$, Monoclinic, $a = 12.1692(7)$, $b = 15.9364(8)$, $c = 14.6654(9)$ Å, $\beta = 107.473(6)^\circ$, $V = 2712.9(3)$ Å 3 , $Z = 4$, $\rho_{\text{calcd}} = 1.643$ Mg/m 3 .

C. Crystallographic Details for [Fe(CO)₂(PEt₃)I(hmp)] (4)

A total of 14550 reflections ($-9 \leq h \leq 9$, $-17 \leq k \leq 21$, $-18 \leq l \leq 17$) were collected at $T = 140(2)$ K in the range of 2.87 to 27.88° of which 4094 were unique ($R_{\text{int}} = 0.0334$); MoK α radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 0.464 and -0.407 eÅ $^{-3}$, respectively. The absorption coefficient was 2.764 mm $^{-1}$. The least squares refinement converged normally with residuals of $R(F) = 0.0246$, $wR(F^2) = 0.0531$ and a GOF = 1.035 ($I > 2\sigma(I)$). C₁₄H₂₁FeINO₃P, Mw = 465.04, space group $P2_1/n$, Monoclinic, $a = 7.49550(10)$, $b = 16.4875(4)$, $c = 14.0187(3)$ Å, $\beta = 95.2490(17)^\circ$, $V = 1725.19(6)$ Å 3 , $Z = 4$, $\rho_{\text{calcd}} = 1.790$ Mg/m 3 .

D. Crystallographic Details for [Fe(CO)₂(PPh₃)(hmp){S(2-*i*Pr-C₆H₄)}] (6)

A total of 12233 reflections ($-13 \leq h \leq 13$, $-14 \leq k \leq 13$, $-18 \leq l \leq 18$) were collected at $T = 140(2)$ K in the range of 2.77 to 26.02° of which 6506 were unique ($R_{\text{int}} = 0.1119$); MoK α radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 2.170 and -0.974 eÅ $^{-3}$, respectively. The absorption coefficient was 0.764 mm $^{-1}$. The least squares refinement converged normally with residuals of $R(F) = 0.1128$, $wR(F^2) = 0.2409$ and a GOF = 0.930 ($I > 2\sigma(I)$). C₃₆H₃₄Cl₂FeNO₃PS, Mw = 718.42, space group $P-1$, Triclinic, $a = 11.0759(19)$, $b = 11.5177(14)$, $c = 15.082(2)$ Å, $\alpha = 71.085(12)^\circ$, $\beta = 78.739(13)^\circ$, $\gamma = 66.159(14)^\circ$, $V = 1660.2(4)$ Å 3 , $Z = 2$, $\rho_{\text{calcd}} = 1.437$ Mg/m 3 .

E. Crystallographic Details for $[\text{Fe}(\text{CO})_2(\text{PPh}_3)(\text{hmp})\{\text{S}(2,4,6\text{-Me}_3\text{C}_6\text{H}_2)\}]$ (7)

A total of 7201 reflections ($-20 \leq h \leq 19$, $-18 \leq k \leq 18$, $-18 \leq l \leq 19$) were collected at $T = 100(2)$ K in the range of 3.38 to 27.54° of which 7201 were unique ($R_{\text{int}} = 0.0000$); $\text{MoK}\alpha$ radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 1.150 and -0.464 $\text{e}\text{\AA}^{-3}$, respectively. The absorption coefficient was 0.634 mm^{-1} . The least squares refinement converged normally with residuals of $R(F) = 0.0653$, $wR(F^2) = 0.1390$ and a GOF = 1.191 ($I > 2\sigma(I)$). $\text{C}_{35}\text{H}_{32}\text{FeNO}_3\text{PS}$, Mw = 633.50 , space group $P2_1/c$, Monoclinic, $a = 15.649(3)$, $b = 13.890(2)$, $c = 14.8057(13)$ Å, $\beta = 102.833(10)^\circ$, $V = 3137.9(8)$ Å³, $Z = 4$, $\rho_{\text{calcd}} = 1.341$ Mg/m^3 .

F. Crystallographic Details for *cis*-(*I*, PPh_3)- $\text{Fe}(\text{CO})_2(\text{PPh}_3)\text{I}(\text{OMe-PyS})$ (10a)

A total of 43518 reflections ($-11 \leq h \leq 11$, $-25 \leq k \leq 25$, $-22 \leq l \leq 22$) were collected at $T = 100(2)$ K in the range of 3.09 to 27.50° of which 6737 were unique ($R_{\text{int}} = 0.0947$); $\text{MoK}\alpha$ radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 2.401 and -2.651 $\text{e}\text{\AA}^{-3}$, respectively. The absorption coefficient was 1.885 mm^{-1} . The least squares refinement converged normally with residuals of $R(F) = 0.0619$, $wR(F^2) = 0.1595$ and a GOF = 1.302 ($I > 2\sigma(I)$). $\text{C}_{26}\text{H}_{21}\text{FeINO}_3\text{PS}\cdot\text{CH}_2\text{Cl}_2$, Mw = 726.14 , space group $P2_1/c$, Monoclinic, $a = 8.7884(8)$, $b = 19.780(2)$, $c = 17.558(3)$ Å, $\beta = 103.934(8)^\circ$, $V = 2962.3(7)$ Å³, $Z = 4$, $\rho_{\text{calcd}} = 1.628$ Mg/m^3 .

G. Crystallographic Details for *trans*-(*I*, PPh_3)- $\text{Fe}(\text{CO})_2(\text{PPh}_3)\text{I}(\text{OMe-PyS})$ (10b)

A total of 23989 reflections ($-10 \leq h \leq 10$, $-16 \leq k \leq 16$, $-17 \leq l \leq 17$) were collected at $T = 100(2)$ K in the range of 3.06 to 25.00° of which 5759 were unique ($R_{\text{int}} = 0.0854$); $\text{MoK}\alpha$ radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 1.395 and -0.632 $\text{e}\text{\AA}^{-3}$, respectively. The absorption coefficient was 1.700 mm^{-1} . The least squares refinement converged normally with residuals of $R(F) = 0.0807$, $wR(F^2) = 0.1879$ and a GOF = 1.323 ($I > 2\sigma(I)$). $\text{C}_{26}\text{H}_{21}\text{FeINO}_3\text{PS}\cdot\text{CH}_2\text{Cl}_2$, Mw = 726.14 , space group $P-1$, Triclinic, $a = 8.7114(15)$, $b = 14.042(3)$, $c = 15.126(3)$ Å, $\alpha = 109.470(17)$, $\beta = 92.157(15)$, $\gamma = 107.654(14)^\circ$, $V = 1642.4(5)$ Å³, $Z = 2$, $\rho_{\text{calcd}} = 1.468$ Mg/m^3 .

H. Crystallographic Details for $[\text{Fe}(\text{CO})_2(\text{PPh}_3)_2\{\text{S}(6\text{-Me-C}_5\text{H}_3\text{N})\}]^+(\text{PF}_6)^-\cdot 3\text{CH}_3\text{CN}$ (11)

A total of 21507 reflections ($-14 \leq h \leq 14$, $-18 \leq k \leq 18$, $-18 \leq l \leq 18$) were collected at $T = 140(2)$ K in the range of 2.85 to 26.02° of which 9450 were unique ($R_{\text{int}} = 0.0305$); $\text{MoK}\alpha$ radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 0.400 and -0.286 $\text{e}\text{\AA}^{-3}$, respectively. The absorption coefficient was 0.522 mm^{-1} . The least squares refinement converged normally with residuals of $R(F) = 0.0441$, $wR(F^2) = 0.0983$ and a GOF = 1.037 ($I > 2\sigma(I)$). $\text{C}_{50}\text{H}_{45}\text{F}_6\text{FeN}_4\text{O}_2\text{P}_3\text{S}$, Mw = 1028.72 , space group $P-1$, Triclinic, $a = 11.6427(3)$, $b = 14.8175(5)$, $c = 14.9136(5)$ Å, $\alpha = 76.110(3)$, $\beta = 76.397(3)$, $\gamma = 81.470(3)^\circ$, $V = 2416.00(14)$ Å³, $Z = 2$, $\rho_{\text{calcd}} = 1.414$ Mg/m^3 .

I. Crystallographic Details for [Fe(CO)₂I(2,6-Me₂C₆H₃NC)(SC₅H₄N-CO)] (13)

A total of 14104 reflections ($-9 \leq h \leq 9$, $-22 \leq k \leq 22$, $-17 \leq l \leq 17$) were collected at $T = 140(2)$ K in the range of 2.72 to 26.37° of which 3751 were unique ($R_{\text{int}} = 0.0568$); MoK α radiation ($\lambda = 0.71073$ Å). The structure was solved by the direct methods. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were placed in calculated idealized positions. The residual peak and hole electron densities were 3.010 and -1.176 eÅ⁻³, respectively. The absorption coefficient was 2.612 mm⁻¹. The least squares refinement converged normally with residuals of $R(F) = 0.0567$, $wR(F^2) = 0.1521$ and a GOF = 1.020 ($I > 2\sigma(I)$). C₁₇H₁₃FeIN₂O₃S, Mw = 508.10 , space group $P2_1/n$, Monoclinic, $a = 7.5304(9)$, $b = 17.855(2)$, $c = 13.8288(15)$ Å, $\beta = 95.161(10)^\circ$, $V = 1851.8(4)$ Å³, $Z = 4$, $\rho_{\text{calcd}} = 1.823$ Mg/m³.

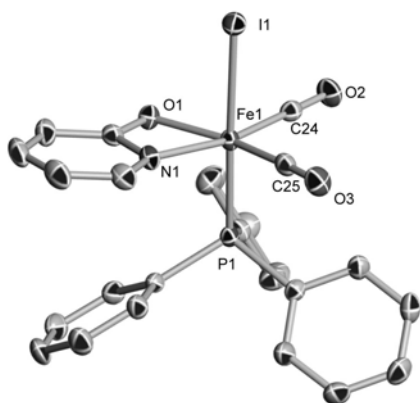


Figure S1. Solid-state structure of **2**. The thermal ellipsoids are displayed in 50% probability. Selected bond distances (Å) and angles (°): Fe1-N1, 1.9575(16); Fe1-O1, 2.0274(13); Fe1-P1, 2.2556(6); Fe1-C24, 1.794(2); Fe1-C25, 1.778(2); Fe1-I1, 2.6536(3); C24-O2, 1.136(2); C25-O3, 1.140(2); C5-O1, 1.309(2); C5-N1, 1.349(2); C1-N1, 1.342(3); C24-Fe1-C25, 94.74(9); O1-Fe1-N1, 66.42(6).

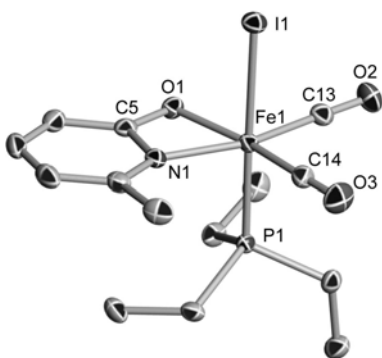


Figure S2. Solid-state molecular structure of complex **4**. The thermal ellipsoids are displayed at 50% probability. Selected bond distances (Å) and angles (°): Fe1-N1, 1.973(2); Fe1-O1, 2.0104(16); Fe1-P1, 2.2477(6); Fe1-C13, 1.787(3); Fe1-C14, 1.774(2); Fe1-I1, 2.6752(3); C13-O2, 1.135(3); C14-O3, 1.140(3); C5-O1, 1.307(3); C5-N1, 1.355(3); C13-Fe1-C14, 90.57(11); O1-Fe1-N1, 66.63(7).

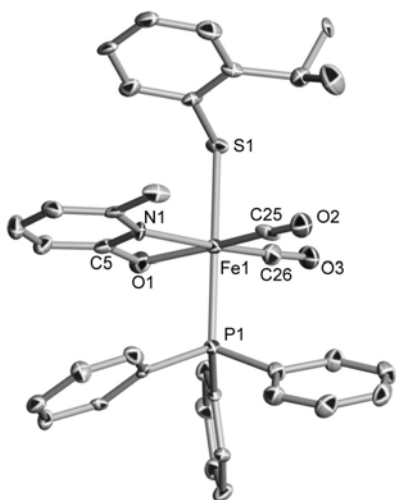


Figure S3. Solid-state molecular structure of complex **6**. The thermal ellipsoids are displayed at 50% probability. Selected bond distances (Å) and angles (°): Fe1-N1, 1.967(8); Fe1-O1, 2.001(6); Fe1-P1, 2.273(3); Fe1-C25, 1.808(11); Fe1-C26, 1.812(10); Fe1-S1, 2.365(3); C25-O2, 1.128(10); C26-O3, 1.123(10); C5-O1, 1.323(12); C5-N1, 1.373(12); C25-Fe1-C26, 93.6(4); O1-Fe1-N1, 67.3(3).

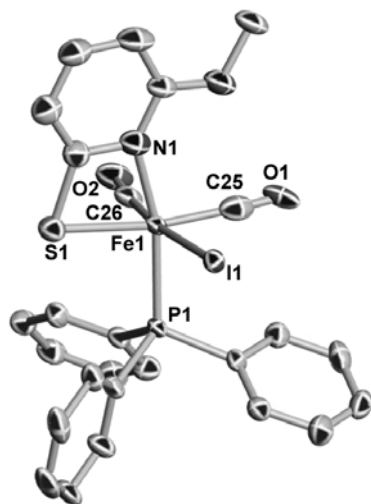


Figure S4. Solid-state molecular structure of complex **10a**. The thermal ellipsoids are displayed at 50% probability. One CH₂Cl₂ molecule was omitted. Selected bond distances (Å) and angles (°): Fe1-I1, 2.6716(9); Fe1-N1, 2.002(6); Fe1-P1, 2.2769(16); Fe1-C25, 1.823(9); Fe1-C26, 1.788(6); Fe1-S1, 2.3888(19); C25-O1, 1.021(9); C26-O2, 1.135(8); C25-Fe1-C26, 89.8(3).

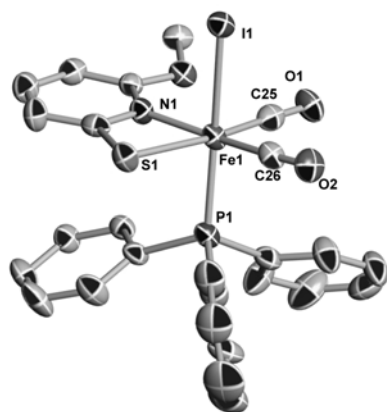


Figure S5. Solid-state molecular structure of complex **10b**. The thermal ellipsoids are displayed at 50% probability. One CH₂Cl₂ molecule was omitted. Selected bond distances (Å) and angles (°): Fe1-I1, 2.6777(16); Fe1-N1, 2.005(8); Fe1-P1, 2.257(3); Fe1-C25, 1.784(13); Fe1-C26, 1.790(11); Fe1-S1, 2.365(3); C25-O1, 1.167(13); C26-O2, 1.134(12); C25-Fe1-C26, 90.3(5).

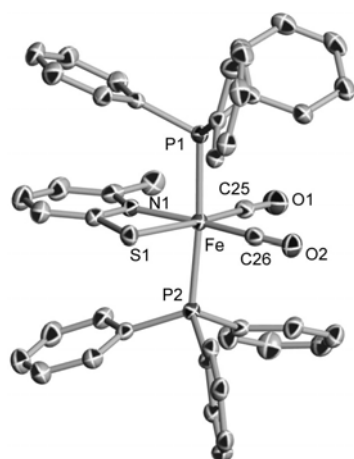


Figure S6. Solid-state molecular structure of complex **11**. The thermal ellipsoids are displayed at 50% probability. Selected bond distances (Å) and angles (°): Fe1-N1, 1.997(2); Fe1-P1, 2.3146(8); Fe1-P2, 2.3045(8); Fe1-C25, 1.807(3); Fe1-C26, 1.779(3); Fe1-S1, 2.3322(8); C25-O2, 1.131(3); C26-O3, 1.144(3); C25-Fe1-C26, 89.58(12).

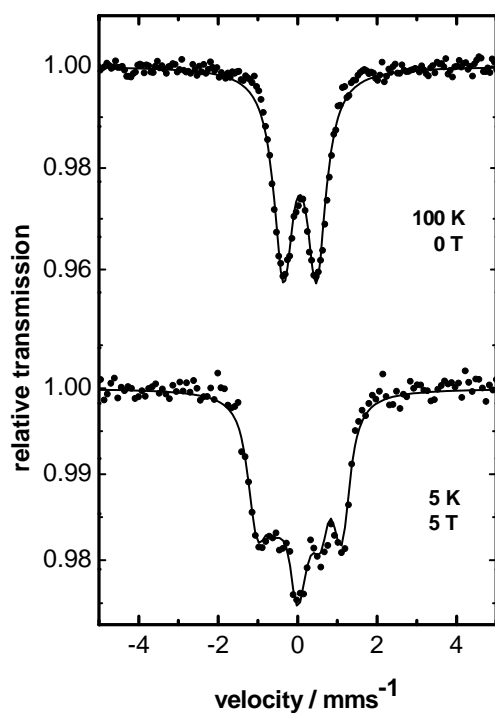


Fig. S7. (Top) Mössbauer spectrum of complex **5** taken at 100 K. The isomer shift is $\delta = 0.06 (\pm 0.02) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = (-) 0.83 (\pm 0.02) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.58 (\pm 0.02) \text{ mms}^{-1}$. (Bottom) Mössbauer spectrum of complex **5** taken at 5 K with a magnetic field of 5 T perpendicular to the γ -ray. The isomer shift is $\delta = 0.04 (\pm 0.02) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = - 0.87 (\pm 0.03) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.44 (\pm 0.02) \text{ mms}^{-1}$.

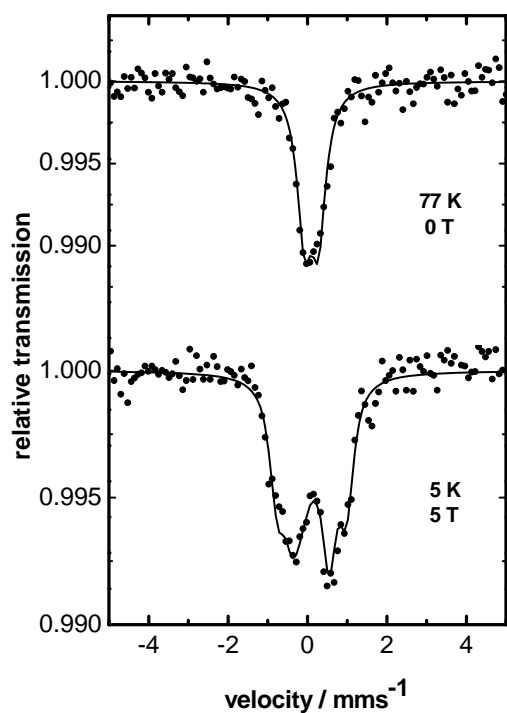


Fig. S8. (Top) Mössbauer spectrum of complex **9** taken at 77 K. The isomer shift is $\delta = 0.10 (\pm 0.01) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = -0.35 (\pm 0.03) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.45 (\pm 0.05) \text{ mms}^{-1}$. (Bottom) Mössbauer spectrum of complex **9** taken at 5 K with a magnetic field of 5 T perpendicular to the γ -ray. The isomer shift is $\delta = 0.11 (\pm 0.02) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = -0.36 (\pm 0.03) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.40 (\pm 0.02) \text{ mms}^{-1}$.

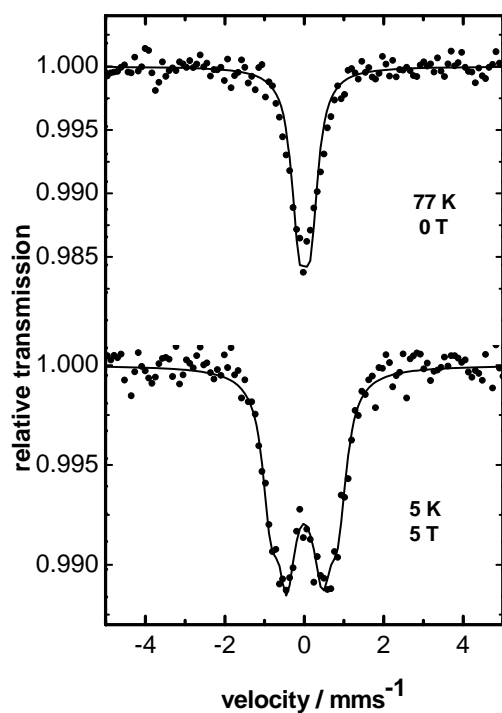


Fig. S9. (Top) Mössbauer spectrum of complex **13** taken at 77 K. The isomer shift is $\delta = 0.01$ (± 0.02) mms^{-1} , the quadrupole splitting is $\Delta E_Q = (+) 0.29$ (± 0.02) mms^{-1} and the linewidth is $\Gamma = 0.45$ (± 0.05) mms^{-1} . (Bottom) Mössbauer spectrum of complex **13** taken at 5 K with a magnetic field of 5 T perpendicular to the γ -ray. The isomer shift is $\delta = 0.01$ (± 0.02) mms^{-1} , the quadrupole splitting is $\Delta E_Q = + 0.29$ (± 0.04) mms^{-1} and the linewidth is $\Gamma = 0.47$ (± 0.03) mms^{-1} .

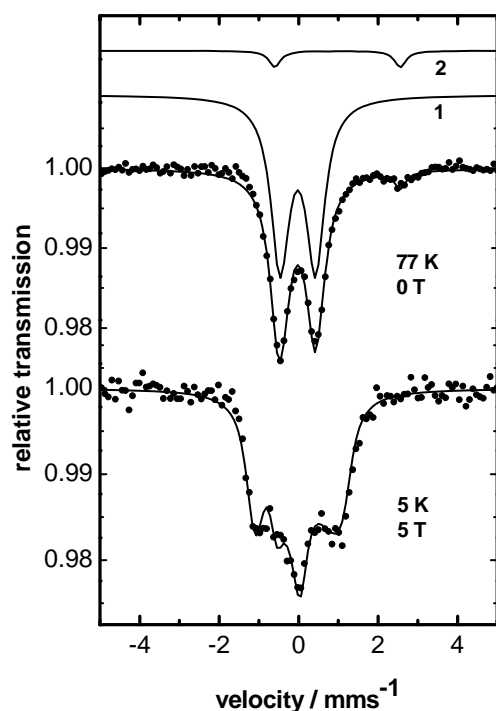


Fig. S10. (Top) Mössbauer spectrum of complex **17** taken at 77 K. There are two different components. The isomer shift for the first component is $\delta = -0.02 (\pm 0.02) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = (+) 0.89 (\pm 0.02) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.57 (\pm 0.02) \text{ mms}^{-1}$. The relative contribution of this component is 94 %. The isomer shift for the second component is $\delta = 0.98 (\pm 0.02) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = 3.16 (\pm 0.02) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.35 (\pm 0.02) \text{ mms}^{-1}$. The iron is therefore in a Fe(II) $S = 2$ state. The relative contribution of this component is 6 %. (Bottom) Mössbauer spectrum of complex **17** taken at 5 K with a magnetic field of 5 T perpendicular to the γ -ray. The isomer shift is $\delta = -0.01 (\pm 0.02) \text{ mms}^{-1}$, the quadrupole splitting is $\Delta E_Q = +0.91 (\pm 0.03) \text{ mms}^{-1}$ and the linewidth is $\Gamma = 0.47 (\pm 0.02) \text{ mms}^{-1}$. The minor component splits into multiple lines at high field and is therefore not detectable under high field conditions.

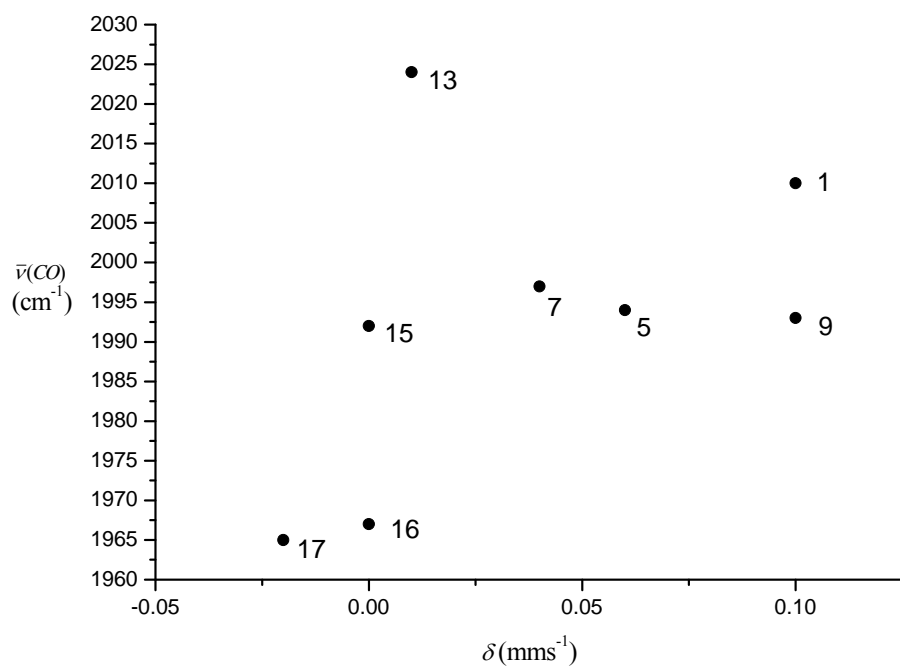


Fig. S11. Correlation of Mössbauer isomeric shifts with the averaged $\nu(\text{CO})$ frequencies of 8 model complexes.