Supporting Information:

Rare Earth Doped Bifunctional Alkaline Earth Metal Fluoride

Nanocrystals via a Facile Microwave-Assisted Process

Min Pang, Dapeng Liu,* Yongqian Lei, Shuyan Song, Jing Feng, Weiqiang Fan and Hongjie

Zhang*

State Key Laboratory of Rare Earth Resource Utilization, Changchun Institute of Applied Chemistry,

Chinese Academy of Sciences, Changchun 130022, Jilin, China, and Graduate School of the Chinese

Academy of Sciences, Beijing 100039, P. R. China

*To whom correspondence should be addressed. E-mail: liudp@ciac.jl.cn and hongjie@ciac.jl.cn.

Tel: +86-431-85262127; Fax: +86-431-85698041

Experimental Supplement:

All the synthesis processes were performed on a programmed microwave synthesis reactor (START

SYNTH, Milestone) equipped with inner symmetrical quartz tubes. The tubes were located on a rotated

plate, which could make all the reactions in the same condition. The temperature was monitored by an

inner IR detector. All the reaction parameters were programmed with optimized heating time, target

temperature and standing time. For a typical experiment, the set parameters are as follows: microwave

irradiation power of 600 W, increasing time 10 min, target temperature 170 °C, standing time 10 min,

standing temperature 170 °C. To remove residual ethylene glycol (EG) and unreacted raw reagents, the

particles were redispersed in ethanol and centrifuged for two more times. The final nanoparticles were

redispersed and stored in ethanol for characterization.

1

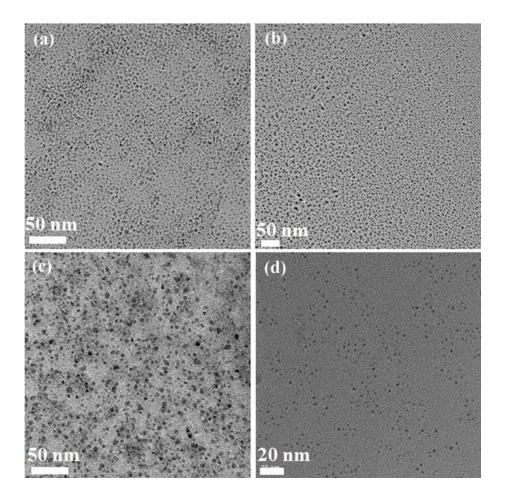


Figure S1. TEM images of (a) $CaF_2:(5\%-5\%)$, (b) $SrF_2:(5\%-5\%)$, (c) $BaF_2:(5\%-5\%)$ and (d) $SrF_2:(5\%-5\%-20\%)$ NCs.

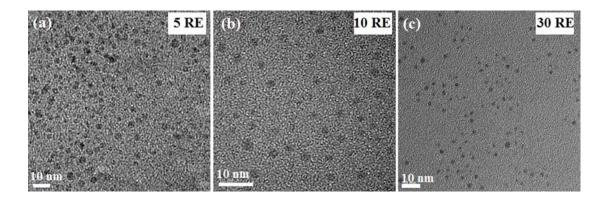


Figure S2. TEM images of SrF_2 NCs doped with various contents of trivalent rare earth ions. (a) SrF:(5%), (b) $SrF_2:(5\%-5\%)$, (c) $SrF_2:(5\%-5\%-20\%)$ [5%, 5%-5%, 5%-5%-20% (totally 5%, 10% and 30% RE; abbreviated as 5 RE, 10 RE and 30 RE, respectively)].

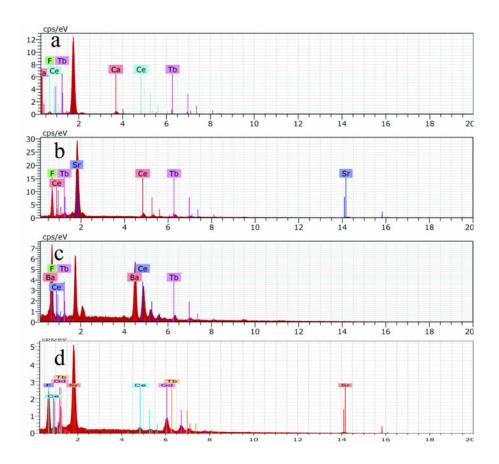


Figure S3. EDX spectra of (a) $CaF_2:(5\%-5\%)$; (b) $SrF_2:(5\%-5\%)$; (c) $BaF_2:(5\%-5\%)$ and (d) $SrF_2:(5\%-5\%-20\%)$ NCs.

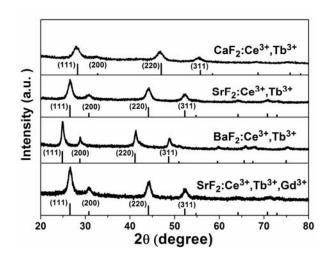


Figure S4. XRD patterns of CaF_2 :(5%-5%), SrF_2 :(5%-5%), BaF_2 :(5%-5%) and SrF_2 :(5%-5%-20%) NCs.

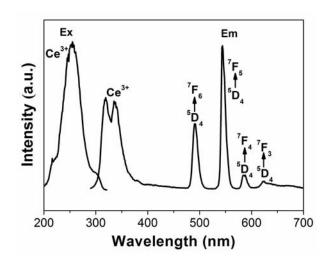


Figure S5. Excitation (Ex) and emission (Em) spectra of CaF₂:(5%-5%) NCs in ethanol solution.

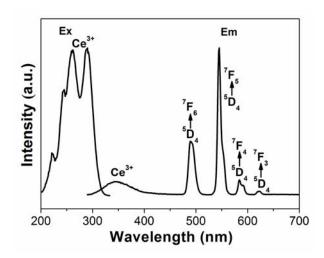


Figure S6. Excitation (Ex) and emission (Em) spectra of BaF₂:(5%-5%) NCs in ethanol solution.

Table S1. Inductively coupled plasma optical emission spectrometer (ICP) results for molar ratios of MF_2 :(5%-5%) (M = Ca, Sr, Ba) NCs.

Ca/Ce/Tb	1	0.059	0.053
Sr/Ce/Tb	1	0.050	0.049
Ba/Ce/Tb	1	0.057	0.054