

Supporting Information

Insertion of Silylenes into Si–H and Si–Cl Bonds. Comparison of Mechanism and Substituent Effects

Zheng Xu, Juan Jin, Haixia Zhang, Zhifang Li, Jianxiong Jiang, Guoqiao Lai,* and
Mitsuo Kira*

Key Laboratory of Organosilicon Chemistry and Material Technology of Ministry of
Education, Hangzhou Normal University, Hangzhou, China
E-mail: mkira@m.tohoku.ac.jp (M. K.); gqlai@hznu.edu.cn (G. L.)

Contents

1. A Full List of Authors for ref. 8
2. **Table S1.** Relative Energies of the Precursor Complexes and Activation Parameters for Possible TSs of Insertion of **2'** into Si–H and Si–Cl Bonds of H_2SiCl_2 and Me_2SiHCl Calculated at the B3LYP/6-31++G(d,p)+ZPE Level
3. **Figure S1.** A plot of $\Delta G^\ddagger$ values for reaction 1 calculated at the B3LYP/6-311++(3df,3pd) level [$\Delta G_2^\ddagger$] vs. those at the B3LYP/6-31++G(d,p) level [$\Delta G_1^\ddagger$]
4. **Table S2.** Activation Parameters for Insertion of Silylene **1** into Si–Cl Bond of $\text{YH}_2\text{Si–Cl}$ (Reaction 1) Calculated at B3LYP/6-311++G(3df,3pd) + ZPE Level
5. **Figure S2.** A plot of $\Delta G^\ddagger(\text{Y}, \mathbf{1})$ vs. $\Delta G^\ddagger(\text{Y}, \mathbf{2}')$ for the reactions with $\text{YH}_2\text{Si–H}$ (Y=H, Me, NH_2 , SiH_3 , PH_2 , SH, Cl)
6. **Figure S3.** Plots of $\Delta G^\ddagger$ values for insertion reactions of silylene **1** into an Si–X bond of $\text{YH}_2\text{Si–X}$ versus Charton σ_I constants; X = H (○), X = Cl (●). For the plot for X = H, three points for Y = F, OH, and Cl whose $\Delta G^\ddagger$ values were estimated using the plot shown in **Figure S2** are included.
7. **Table S3.** Charton σ_I Constants and Taft's Steric Substituent Constants (E_s) for Related Substituents

1. A Full List of Authors for Ref. 8

Gaussian 03 (Revision D.01), M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, and J. A. Pople, Gaussian, Inc., Wallingford CT, 2004.

2. Table S1

Table S1. Relative Energies of the Precursor Complexes and Activation Parameters for Possible TSs of Insertion of **2'** into Si–H and Si–Cl Bonds of H₂SiCl₂ and Me₂SiHCl Calculated at the B3LYP/6-31++G(d,p)+ZPE Level^a

		E _{cplx}	E _a	ΔG [‡]	ΔH [‡]	ΔS [‡]
H ₂ SiCl ₂	TS^H-18-1	-0.9	9.7	22.8	9.4	-44.9
	TS^H-18-2		12.9	25.6	12.7	-43.3
	TS^{Cl}-18-1	-1.0	6.2	19.0	5.6	-44.9
	TS^{Cl}-18-2		11.6	24.7	11.0	-46.0
Me ₂ SiHCl	TS^H-19-1	-0.6	9.0	23.0	8.6	-48.3
	TS^H-19-2		13.1	26.7	12.8	-46.6
	TS^{Cl}-19-1	-0.4	14.5	28.8	14.1	-49.3
	TS^{Cl}-19-2		15.0	28.7	14.7	-47.0

a. E_{cplx}, E_a, ΔG[‡], and ΔH[‡] are all based on the starting reagent systems (**2'** + R₂SiHCl) and given in kcal mol⁻¹. ΔS[‡] values are in cal mol⁻¹ K⁻¹.

3. Figure S1

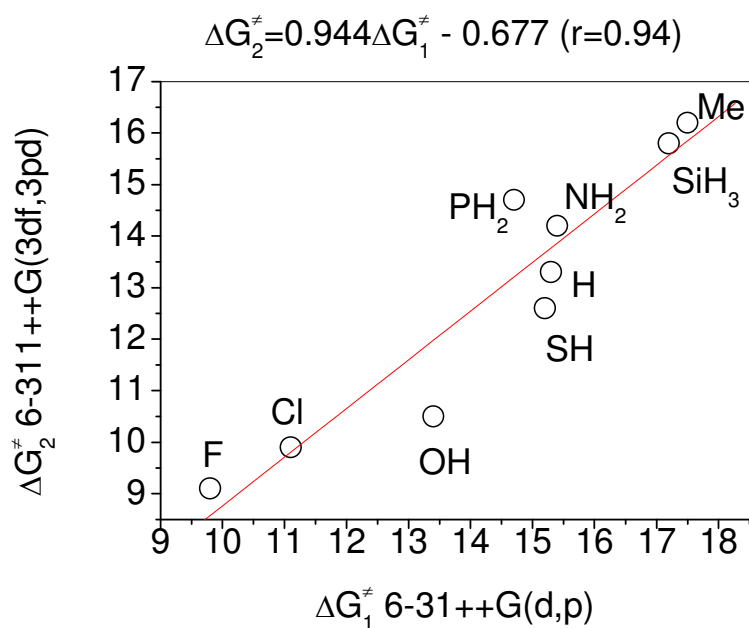


Figure S1. A plot of $\Delta G^\ddagger$ values for reaction 1 calculated at the B3LYP/6-311++(3df,3pd) level [$\Delta G_2^\ddagger$] vs. those at the B3LYP/6-31++G(d,p) level [$\Delta G_1^\ddagger$].

4. Table S2

Table S2. Activation Parameters for the Insertion of Silylene **1** into Si-Cl Bond of $\text{YH}_2\text{Si-Cl}$ (Reaction 1) Calculated at B3LYP/6-311++G(3df,3pd) + ZPE Level^a

Y	TS1A(in-plane Y)		
	E_a	$\Delta H^\ddagger$	$\Delta G^\ddagger$
H	2.6	2.1	13.3
Me	4.4	4.0	16.2
NH ₂	2.4	2.0	14.2
OH	-1.0	-1.1	10.5
F	-2.0	-2.3	9.1
SiH ₃	4.2	3.9	15.8
PH ₂	2.9	2.7	14.7
SH	0.9	0.6	12.6
Cl	-1.2	-1.4	9.9

a. E_a , $\Delta G^\ddagger$, and $\Delta H^\ddagger$ are all based on the starting reagent systems (**1** + $\text{YH}_2\text{Si-Cl}$) and given in kcal mol^{-1} .

5. Figure S2

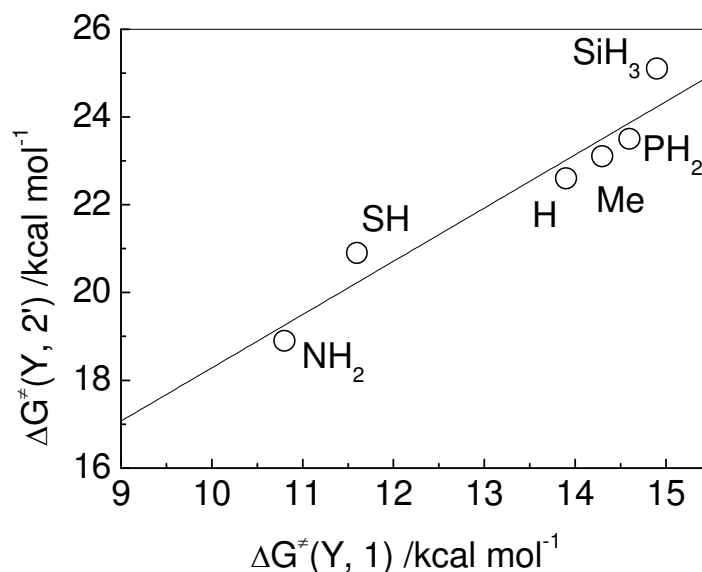


Figure S2. A plot of $\Delta G^\ddagger(\text{Y}, \mathbf{1})$ versus $\Delta G^\ddagger(\text{Y}, \mathbf{2}')$ for the reactions with $\text{YH}_2\text{Si-H}$ ($\text{Y}=\text{H}, \text{Me}, \text{NH}_2, \text{SiH}_3, \text{PH}_2, \text{SH}, \text{Cl}$). The linear equation is $\Delta G^\ddagger(\text{Y}, \mathbf{1}) = 0.826\Delta G^\ddagger(\text{Y}, \mathbf{2}') - 5.08$ and the correlation coefficient $r = 0.96$. The $\Delta G^\ddagger(\text{Y}, \mathbf{1})$ for $\text{Y}=\text{F}, \text{OH}$, and Cl are evaluated to be 9.2, 10.1 and 10.6 kcal mol^{-1} by the linear equation.

6. Figure S3

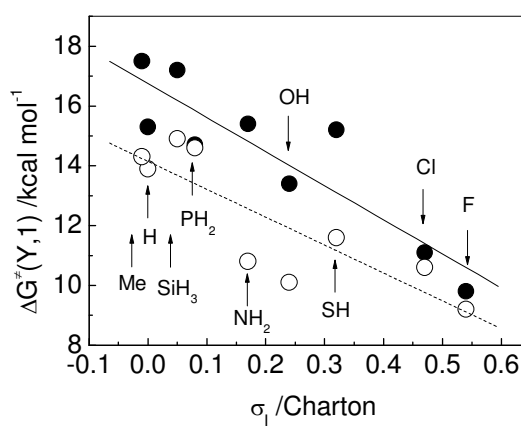


Figure S3. Plots of $\Delta G^\ddagger$ values for insertion reactions of silylene **1** into an Si-X bond of $\text{YH}_2\text{Si-X}$ versus Charton σ_1 constants; $\text{X} = \text{H}$ (○), $\text{X} = \text{Cl}$ (●). For the plot for $\text{X} = \text{H}$, three points for $\text{Y} = \text{F}, \text{OH}$, and Cl whose $\Delta G^\ddagger$ values were estimated using the plot shown in Figure S2 are included. See ref. 12 in text.

7. Table 3

Table S3. Charton σ_I Constants and Taft's Steric Substituent Constants (E_s) for Related Substituents

substituent	Charton σ_I
H	0
Me	-0.01
NH ₂	0.17
OH	0.24
F	0.54
SiH ₃	0.05
PH ₂	0.08
SH	0.32
Cl	0.47

substituent	E_s
H	1.24
Me	0
Cl	0.18
<i>i</i> -Pr	-0.47
<i>t</i> -Bu	-1.54