

Hierarchical DL-Glutamic Acid Microspheres from Polymer-Induced Liquid Precursors

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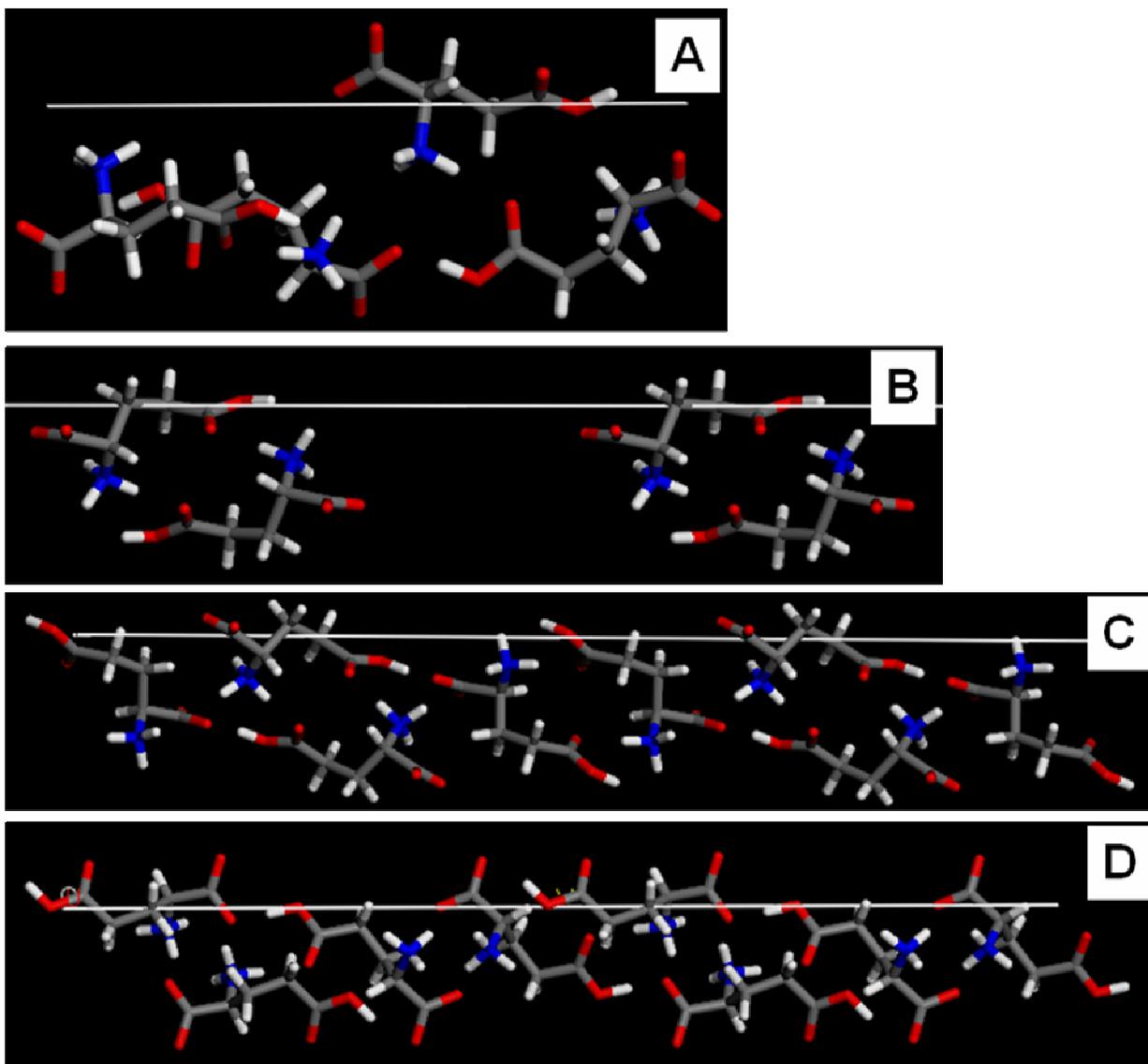


Figure 1. Images A-D are side views of the surfaces of (1 0 1), (0 0 2), (0 1 1), and (1 1 0) of β -L-Glu (white lines) Gray = carbon, red = oxygen, blue = nitrogen (from Materials studio 5.5., Accelrys).

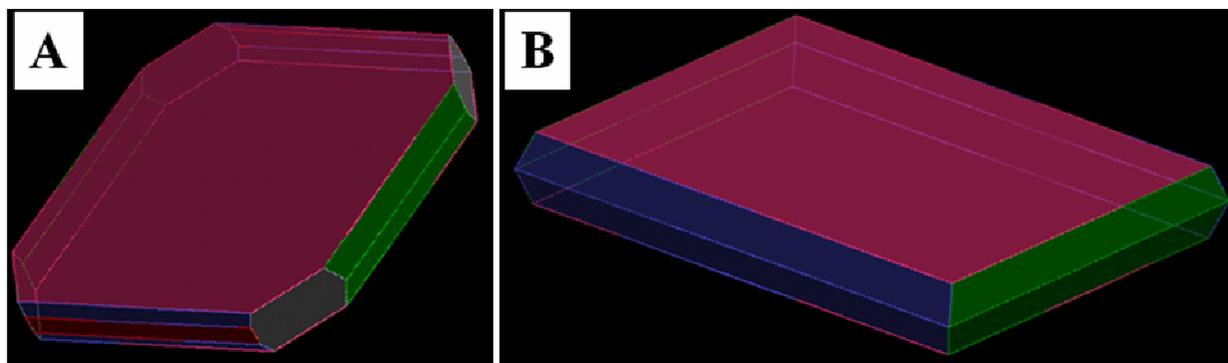


Figure 2. Image **A** shows the morphology of a representative nanoplatelet. Image **B** represents a single crystalline sheet, which models the precipitates collected by mixing DL-Glu aqueous solution with EtOH in the absence of PEI additive. Pink = (0 2 0), blue = (0 1 1), green = (1 1 0), red = (0 0 2), and grey = (1 0 1). Both figures were drawn by using software Cerius² (Accelrys).

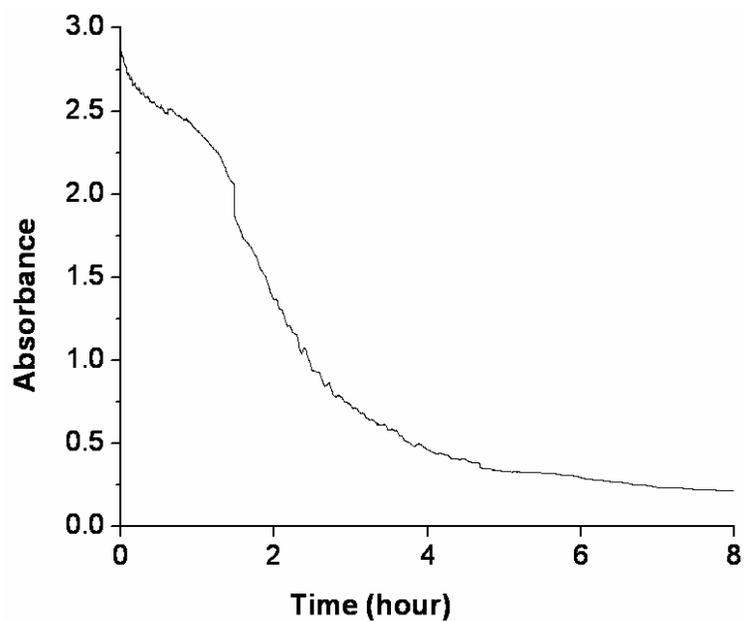


Figure 3. Reaction kinetics from a time-resolved turbidity measurement for the crystallization of microspheres from PILP droplets. The wavelength used was 500 nm. $[\text{DL-Glu}] = 0.3 \text{ mol}\cdot\text{L}^{-1}$, $[\text{EI}] = 0.05 \text{ mol}\cdot\text{L}^{-1}$, $V_{\text{Et}} / V_{\text{Aq}} = 9$, $\text{pH} = 0.5$, 60°C . The measurement was performed at R.T.

Table 1 Compositions for the preparation of microspheres ($V_{Et} / V_{Aq} = 9$, pH = 0.5, 60°C)

	[DL-Glu] (mol·L ⁻¹)	[EI] (mol·L ⁻¹)
1	0.2	0.04
2	0.7	0.15
3	0.7	0.25
4 ^a	1	0.05
5	1	0.15
6	1	0.25
^a some insoluble crystals remained in the aqueous solution		

Table 2 Compositions for the preparation of microspheres ($V_{Et} / V_{Aq} = 9$, pH = 0.5, R.T.)

	[DL-Glu] (mol·L ⁻¹)	[EI] (mol·L ⁻¹)
7 ^a	1	0.05
8 ^a	1	0.15
9 ^a	1	0.25
^a some insoluble crystals remained in the aqueous solution		