## Formation of a Ruthenium(IV)-Oxo Complex by Electron-Transfer Oxidation of a Coordinatively Saturated Ruthenium(II) Complex and Detection of an Oxygen-Rebound Intermediates in C-H Bond Oxygenation

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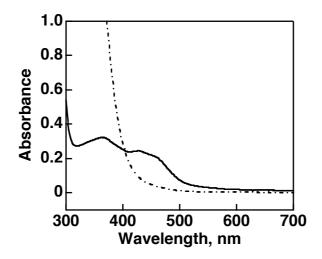
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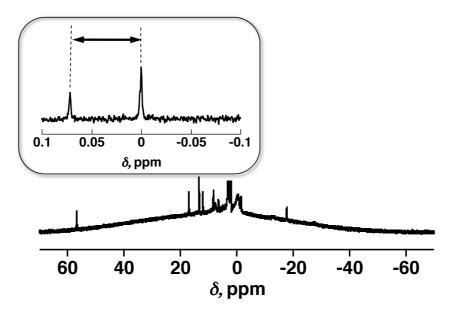
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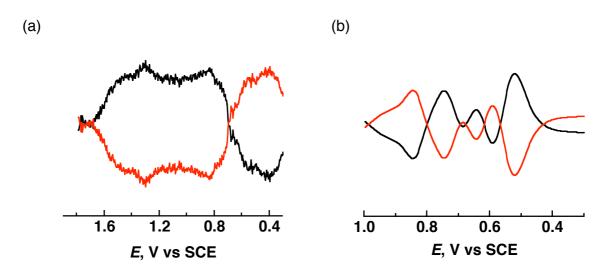
 $^{\Delta}$ ALCA, JST.



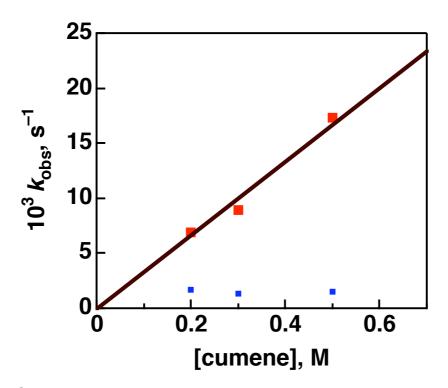
**Figure S1.** UV–vis spectral change upon addition of CAN (10  $\mu$ mol) to an aqueous solution of **1** (0.9  $\mu$ mol) in 3 mL of H<sub>2</sub>O at pH 2.0. The solid line and dotted line are spectra before and after addition of CAN, respectively.



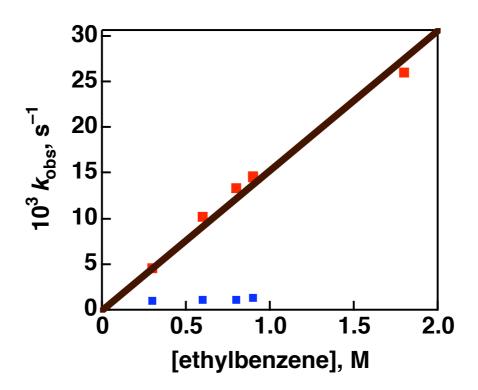
*Figure S2.* <sup>1</sup>H NMR spectrum of 2 in  $(CD_3)_2CO$ . Inset: Signals due to TMS for the Evans' method.



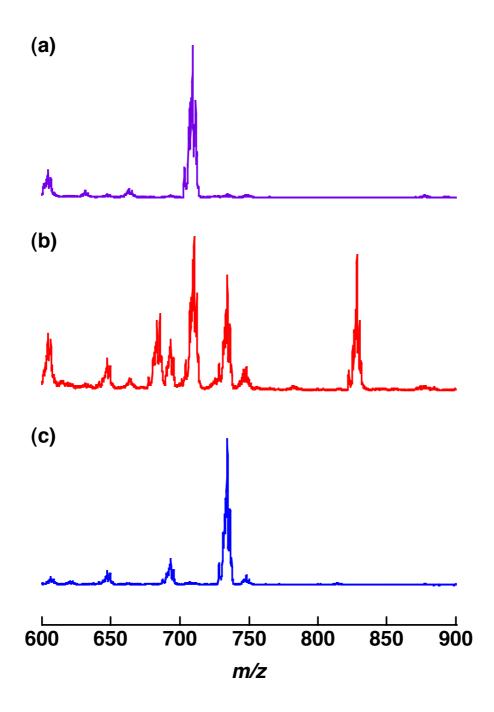
*Figure S3.* Second harmonic ac voltammograms of 2: (a) in  $CH_3CN$  (b) in aqueous  $Na_2SO_4$  (0.1 M) at pH 1 (adjusted by portions of  $H_2SO_4$ ).



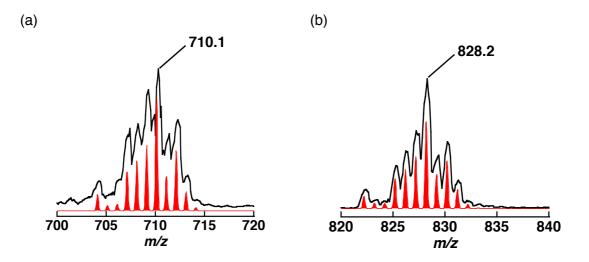
*Figure S4.* Plots of pseudo-first-order rate constants of reaction of 2 with cumene in  $CH_3CN$  at 298 K under  $O_2$  atmosphere vs concentration of cumene: red, the first step; blue, the second step.



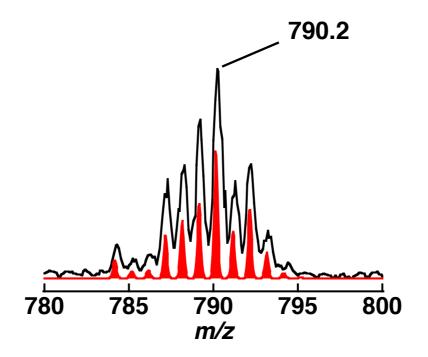
*Figure S5.* Plots of the observed first-order rate constant of the first step (red squares) and the second step (blue squares) in the oxidation of ethylbenzene in CH<sub>3</sub>CN at 298 K vs concentration of ethylbenzene.



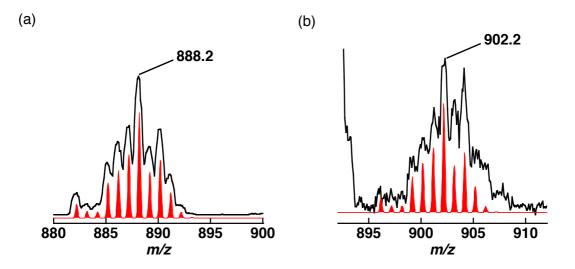
*Figure S6.* Time-course of ESI-MS spectra of the reaction mixture of 2 (0.10 mM) and cumene (200 mM) in CH<sub>3</sub>CN: (a) 0 min, (b) 15 min, (c) 300 min.



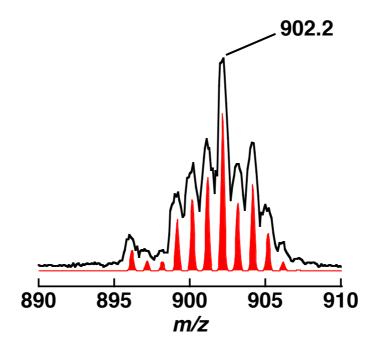
**Figure S7.** Positive-ion ESI-MS spectra of the mixture of cumene (200 mM) and 2 (0.10 mM) (see also Figure 12): (a) A peak cluster at m/z = 710.1 ascribed to  $\{3(^{16}O) - H^+ - 2(PF_6^-)\}^+$ ; (b) a peak cluster at m/z = 828.2 assigned to  $\{4(^{16}O) - H^+ - 2(PF_6^-)\}^+$ .



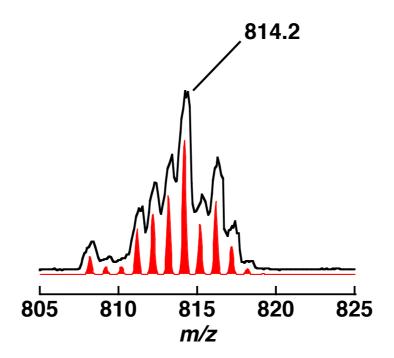
**Figure S8.** Positive-ion ESI-MS spectrum of the mixture of cyclohexene (120 mM) and **2** (0.10 mM) in CH<sub>3</sub>CN: A peak cluster at m/z = 790.2 assigned to  $\{[Ru^{III}(OC_6H_9)(TPA)(bpy)](PF_6^-)\}^+$ .



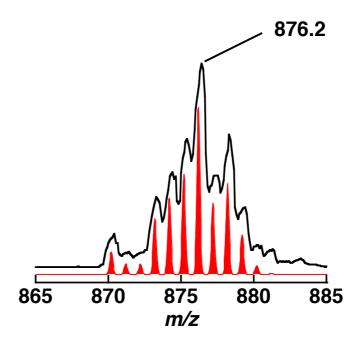
**Figure S9.** Positive-ion ESI-MS spectra of the mixture of DHA (50 mM) and **2** (5.0 mM) in CD<sub>3</sub>CN: (a) A peak cluster at m/z 888.2 assigned to  $\{[Ru^{III}(OC_{14}H_{11})(TPA)(bpy)](PF_6^-)\}^+;$  (b) a peak cluster at m/z 902.2 ascribed to  $\{[Ru^{III}(OC_{14}H_9O)(TPA)(bpy)](PF_6^-)\}^+.$ 



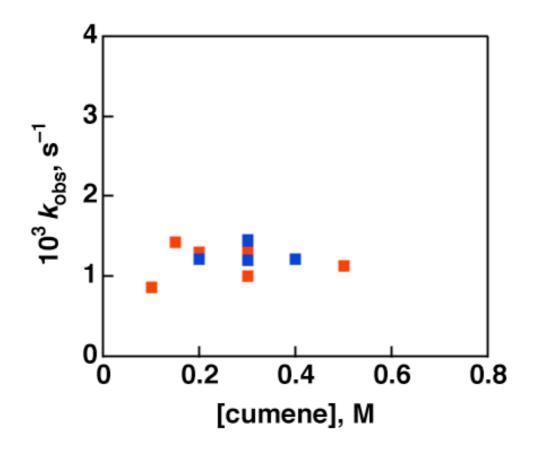
**Figure S10.** Positive-ion ESI-MS spectrum of the mixture of anthrone (50 mM) and **2** (5.0 mM) in CD<sub>3</sub>CN: A peak cluster at m/z = 902.2 assigned to  $\{[Ru^{III}(OC_{14}H_9O)(TPA)(bpy)](PF_6^-)\}^+$ .



**Figure S11.** Positive-ion ESI-MS spectrum of the mixture of ethylbenzene (100 mM) and **2** (5.0 mM) in CD<sub>3</sub>CN: A peak cluster at m/z = 814.2 assigned to  $\{[Ru^{III}(OCCH(CH_3)(C_6H_5))(TPA)(bpy)](PF_6^-)\}^+$ .



**Figure S12.** Positive-ion ESI-MS spectrum of the mixture of diphenylmethane (15 mM) and 2 (5.0 mM) in CD<sub>3</sub>CN: A peak cluster at m/z = 876.2 ascribed to  $\{[Ru^{III}(OCCH(C_6H_5)_2)(TPA)(bpy)](PF_6^-)\}^+$ .



*Figure S13.* Concentration dependence of first-order rate constants for the second step in the cumene oxygenation by 2 in CH<sub>3</sub>CN at 298 K (red, in the absence of  $D_2O$ ; blue, in the presence of excess amount of  $D_2O$  (2M)).