

Organocatalytic Asymmetric Strecker Reaction of Di- and Trifluoromethyl Ketoimines. Remarkable Fluorine Effect

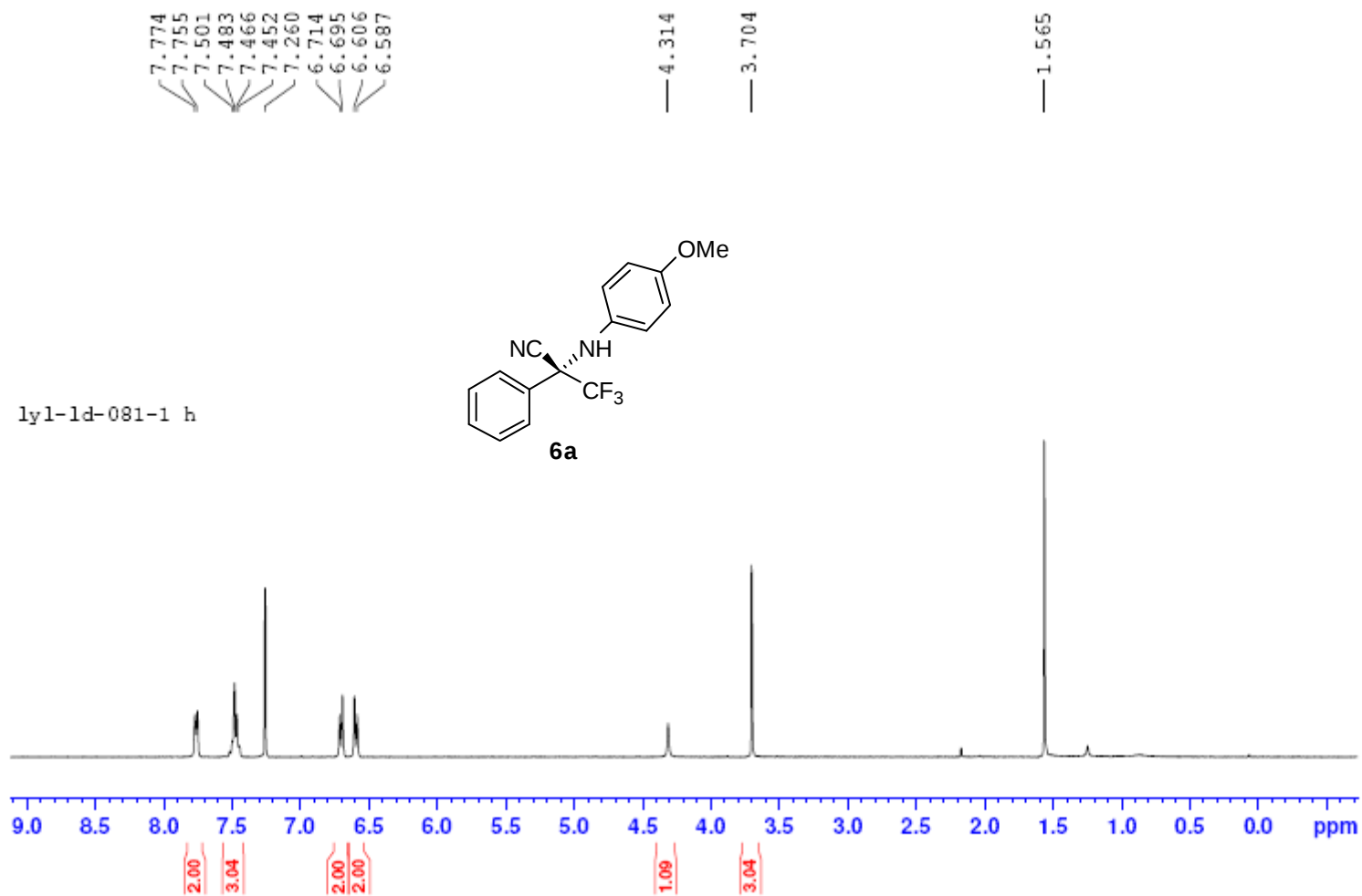
Yun-Lin Liu,^a Tao-Da Shi,^a Feng Zhou,^a Xiao-Li Zhao,^a Xin Wang^{*b} and Jian Zhou^{*a}

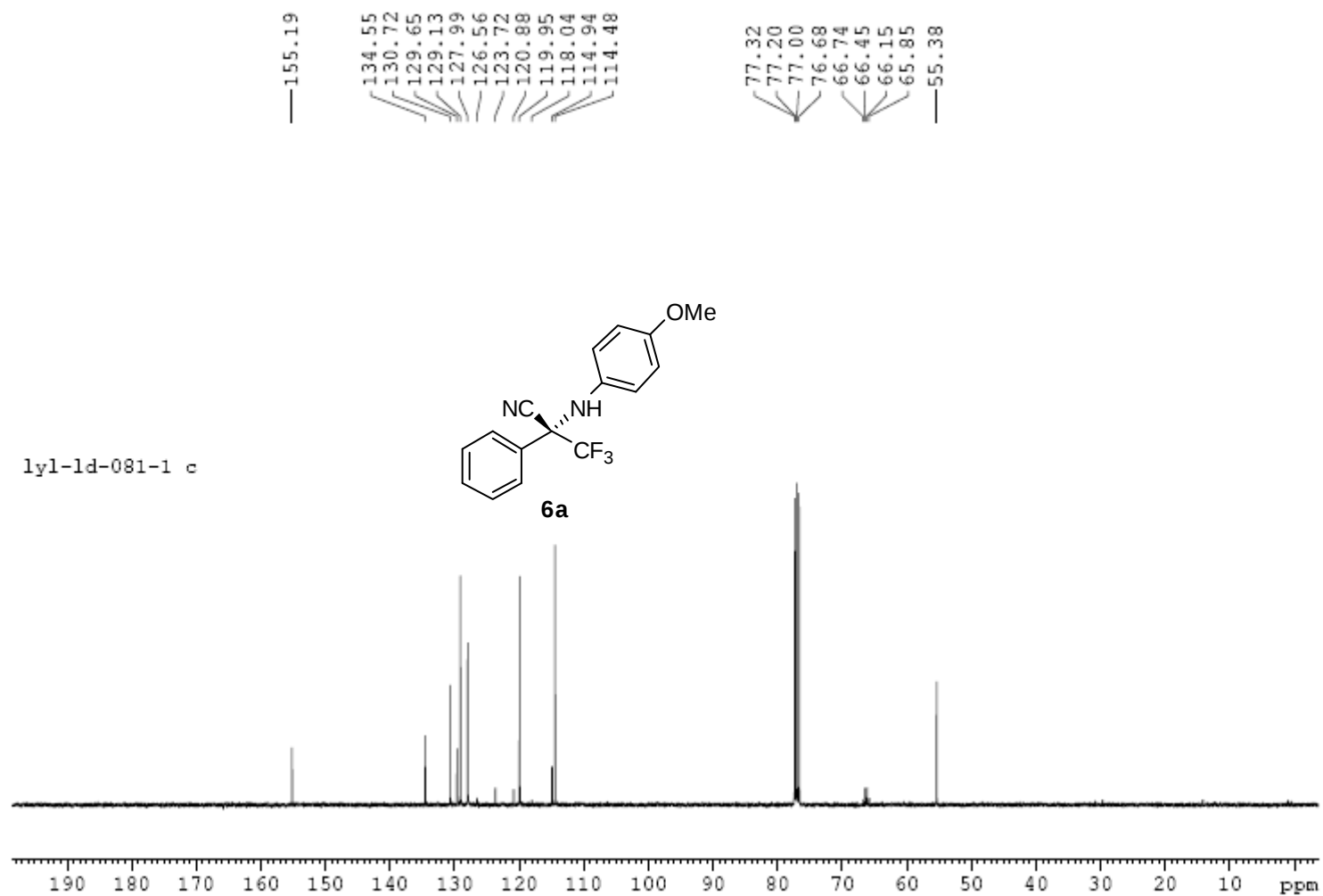
^aShanghai Key Laboratory of Green Chemistry and Chemical Processes, Department of Chemistry, East China Normal University, 3663N Zhongshan Road, Shanghai 200062, China, and College of Chemistry, Sichuan University, Chengdu 610064, China, *E-mail:* wangxin@scu.edu.cn, jzhou@chem.ecnu.edu.cn

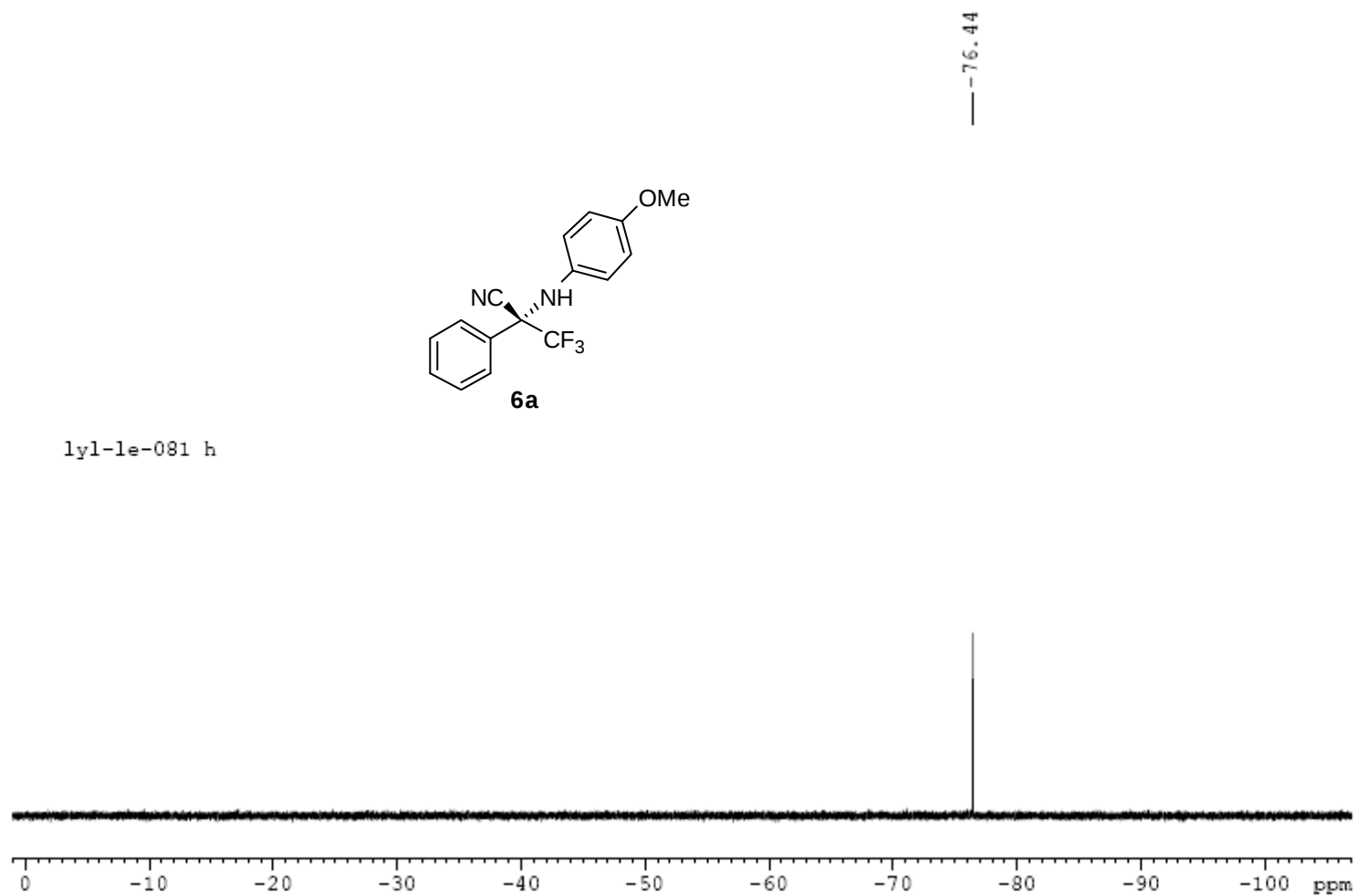
Supporting Information (Part III)

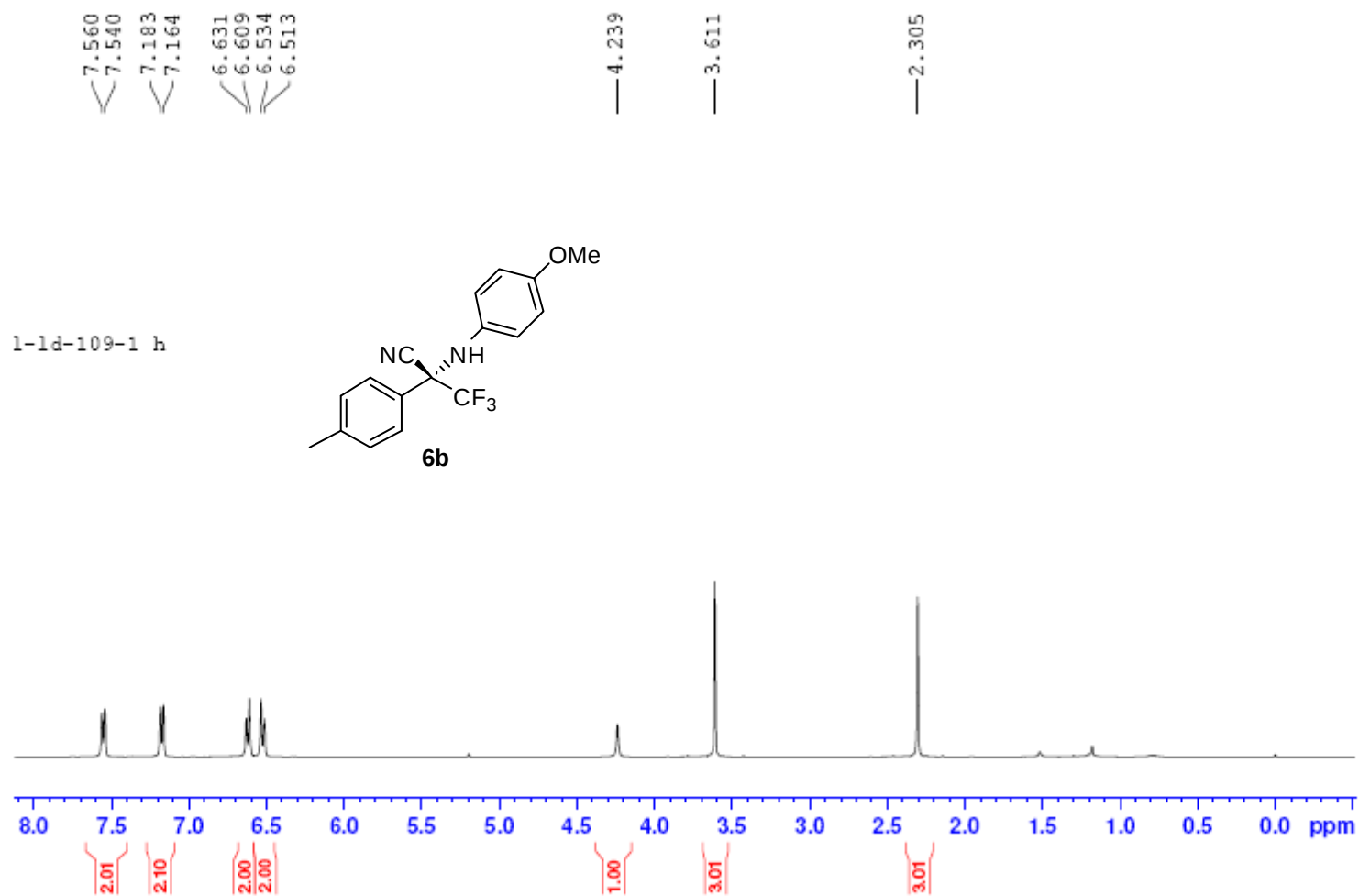
Content	Page
¹ H, ¹³ C NMR and ¹⁹ F NMR spectra of α -CF ₃ α -tetrasubstituted α -aminonitriles 6a-r	S2-S55
¹ H, ¹³ C NMR and ¹⁹ F NMR spectra of product 19-21	S56-S64

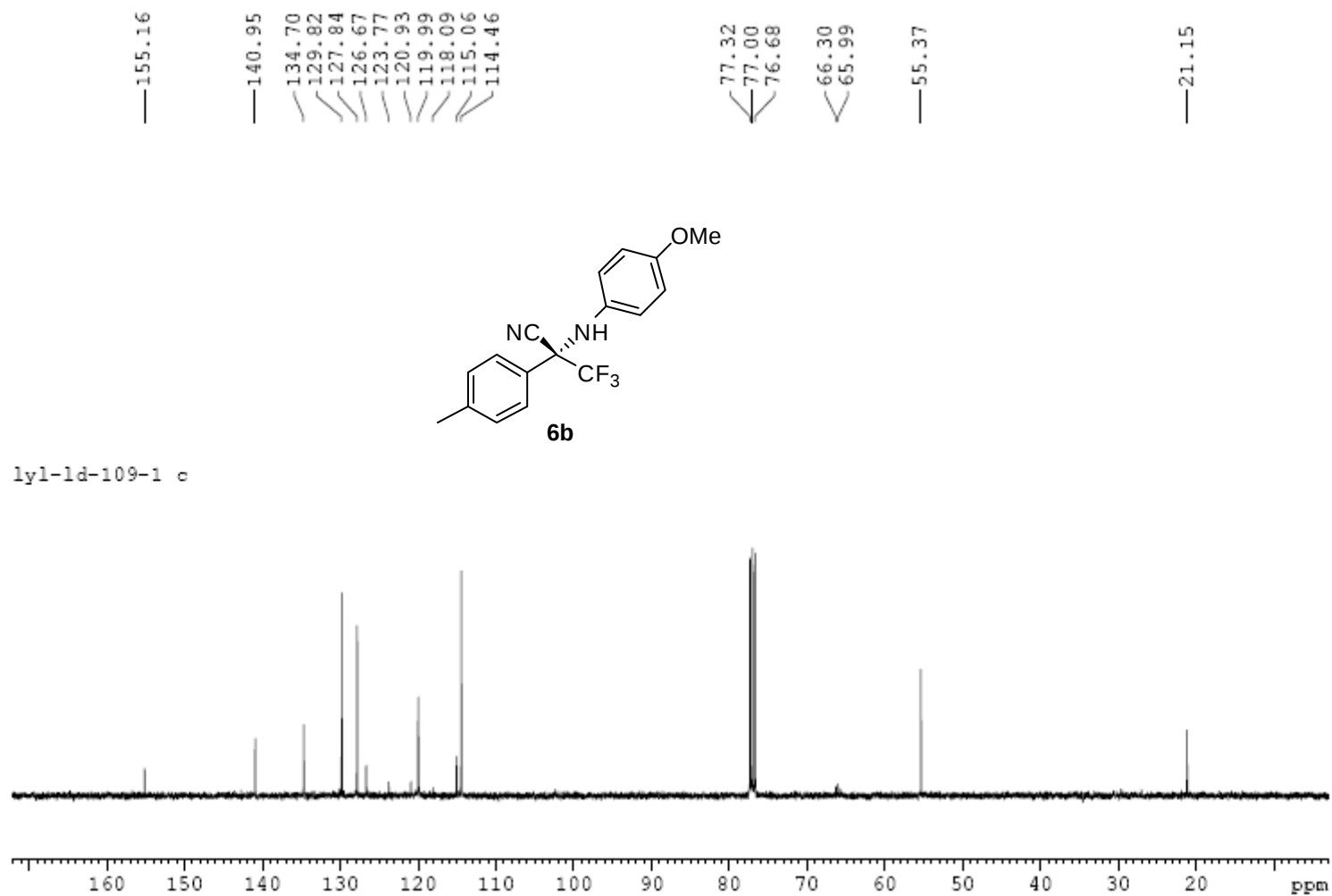
¹H, ¹³C NMR and ¹⁹F NMR spectra were recorded on 400 MHz. Chemical shifts are reported in ppm from CDCl₃, (CD₃)₂SO or D₂O with the solvent resonance as the internal standard. The following abbreviations were used to designate chemical shift multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, h = heptet, m = multiplet, br = broad. Coupling constants, *J*, are reported in Hertz.

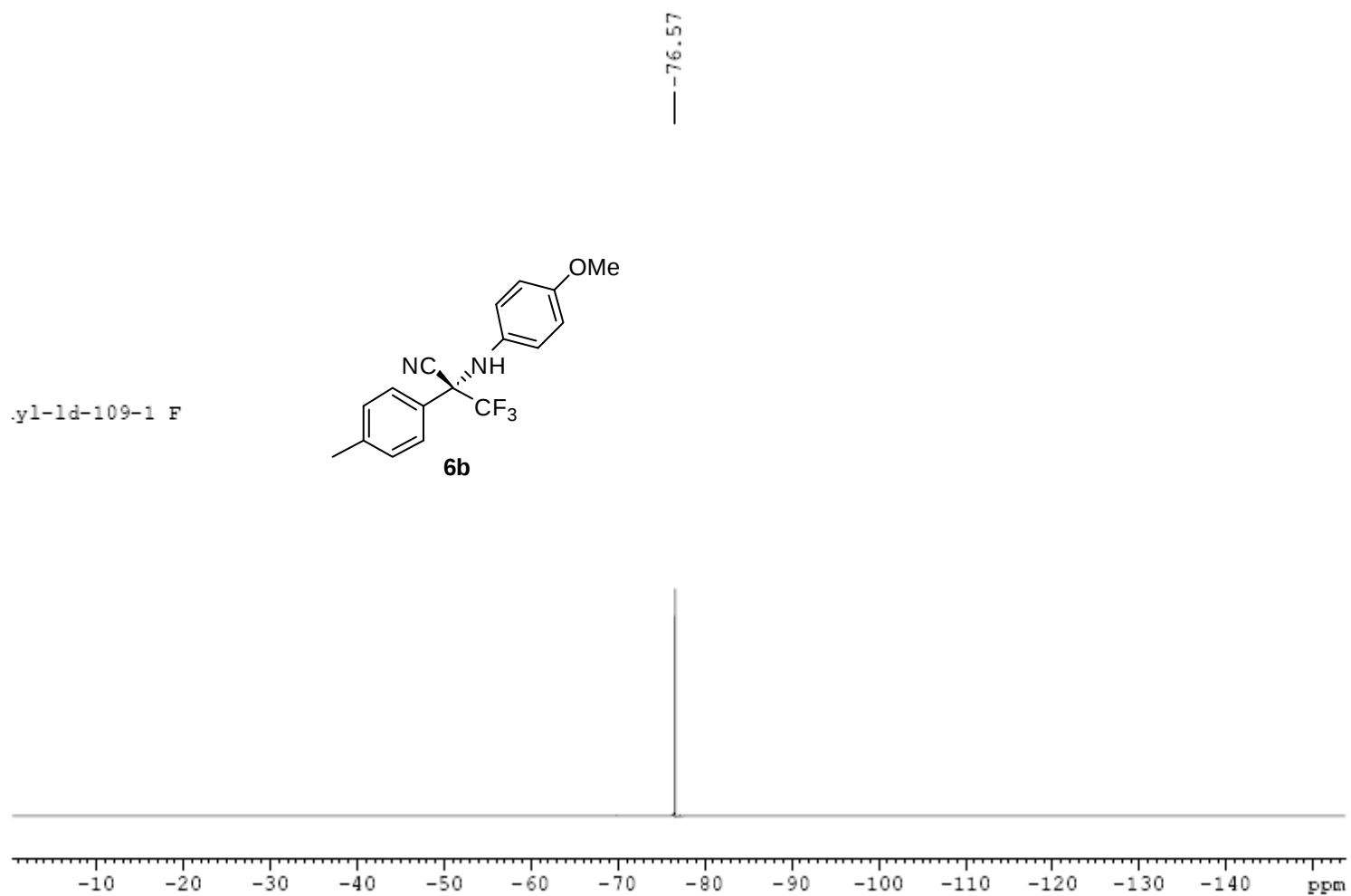


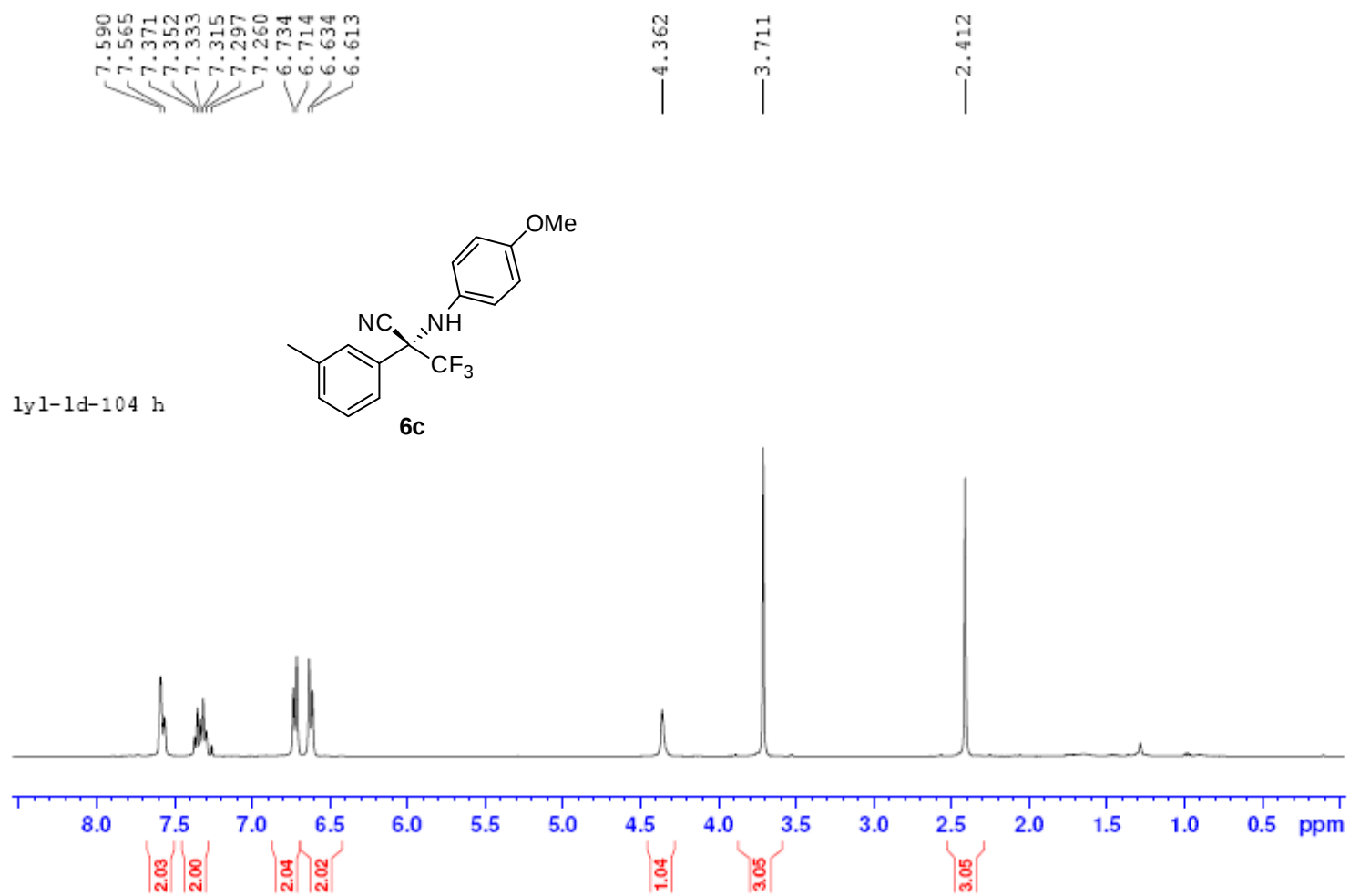


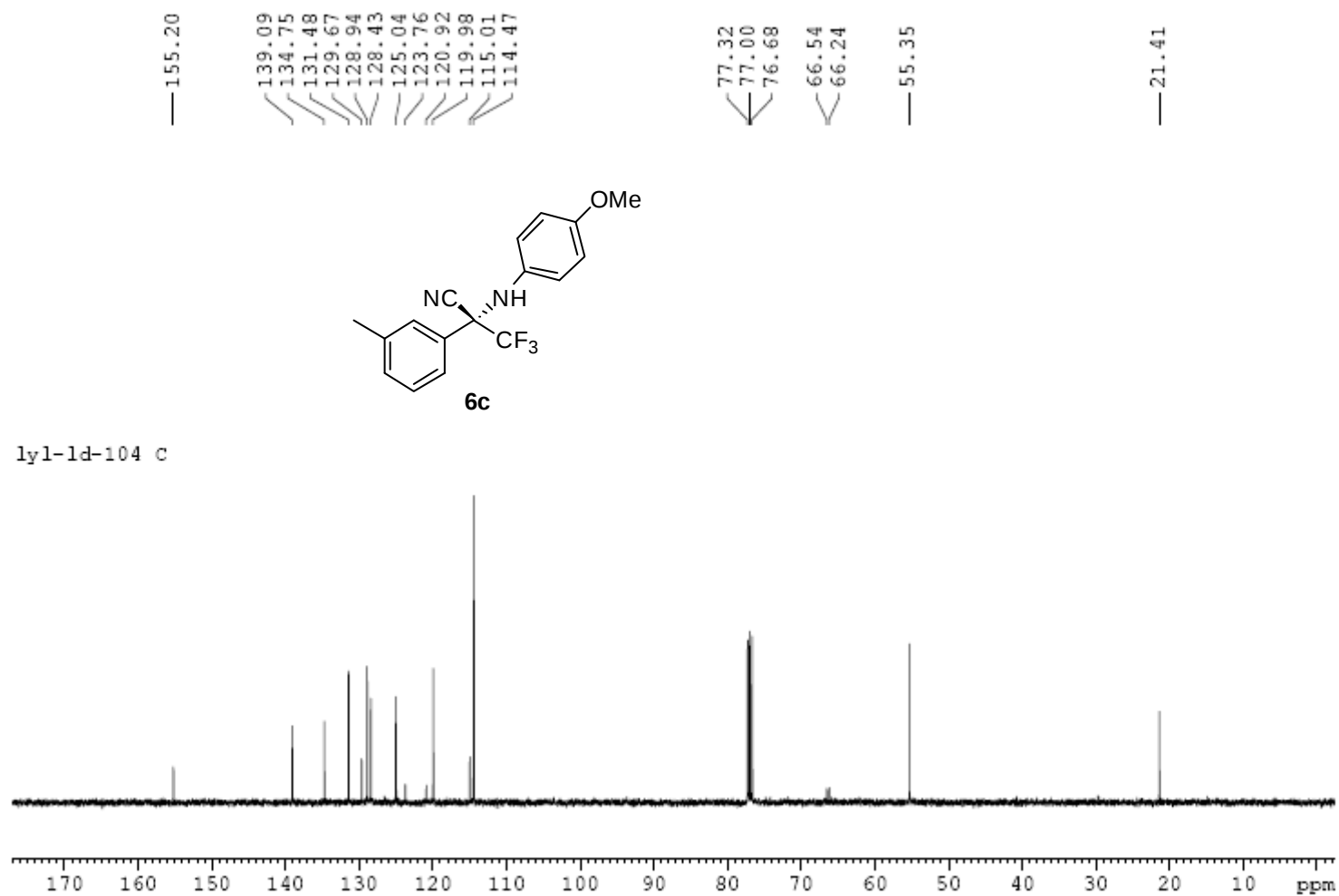


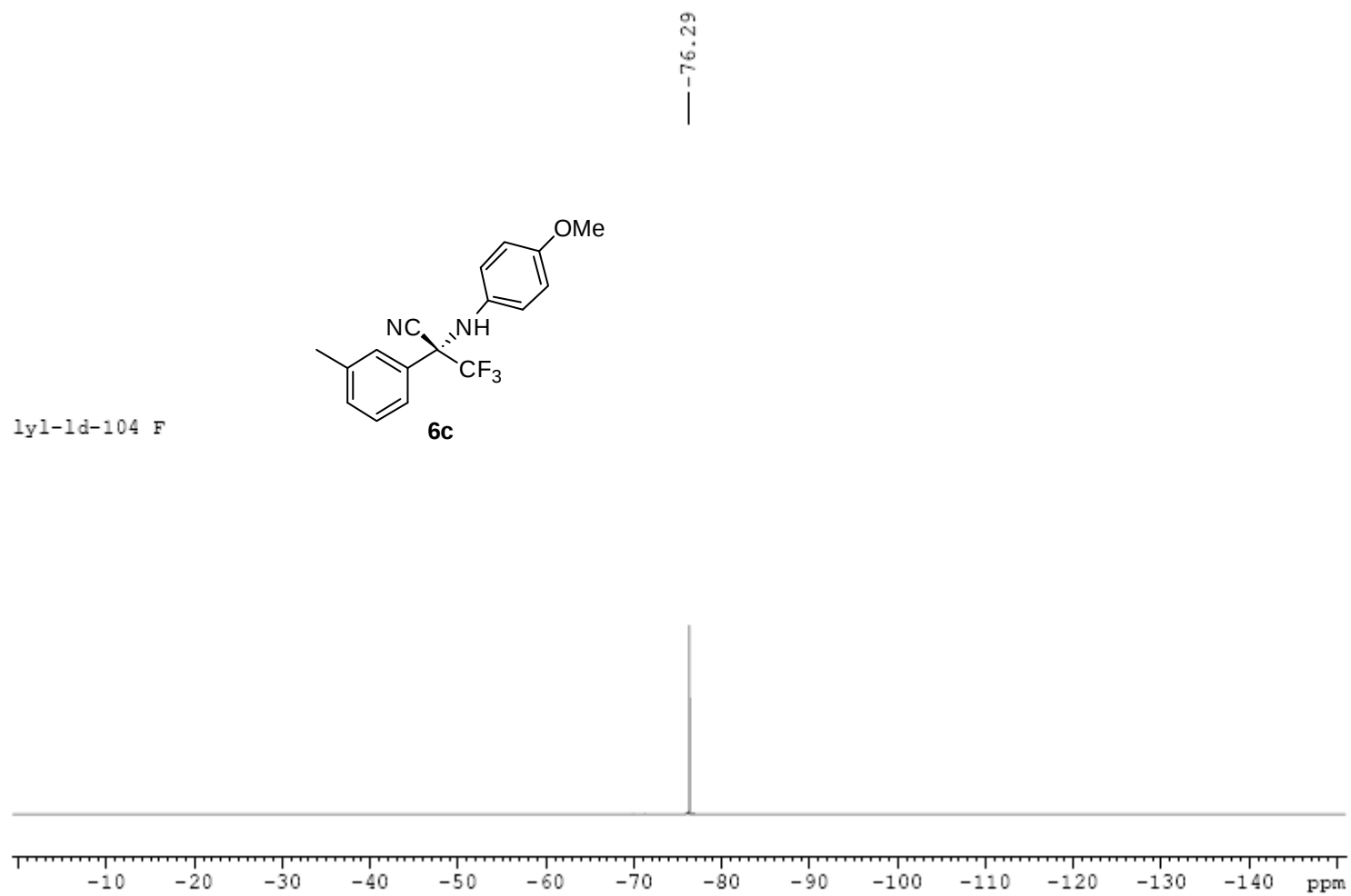




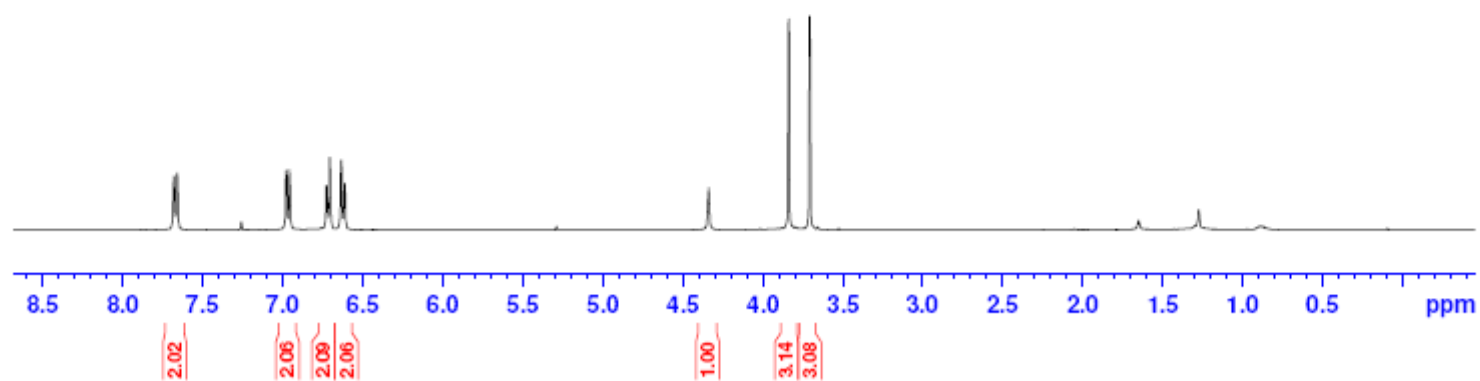
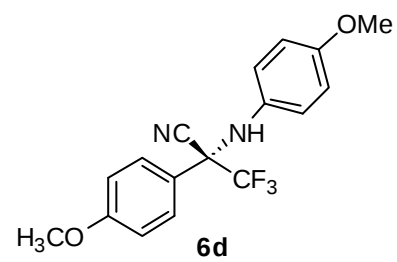






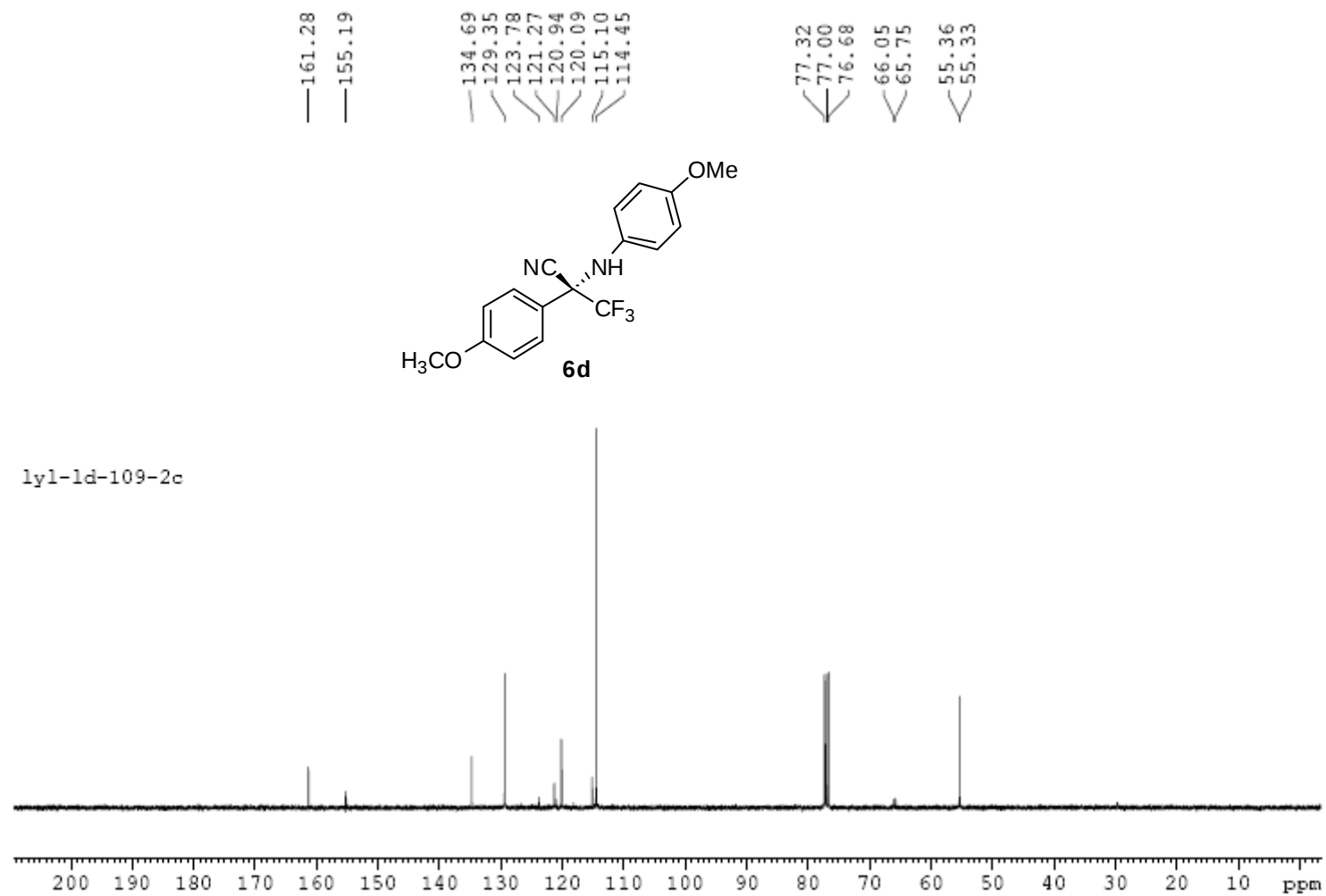


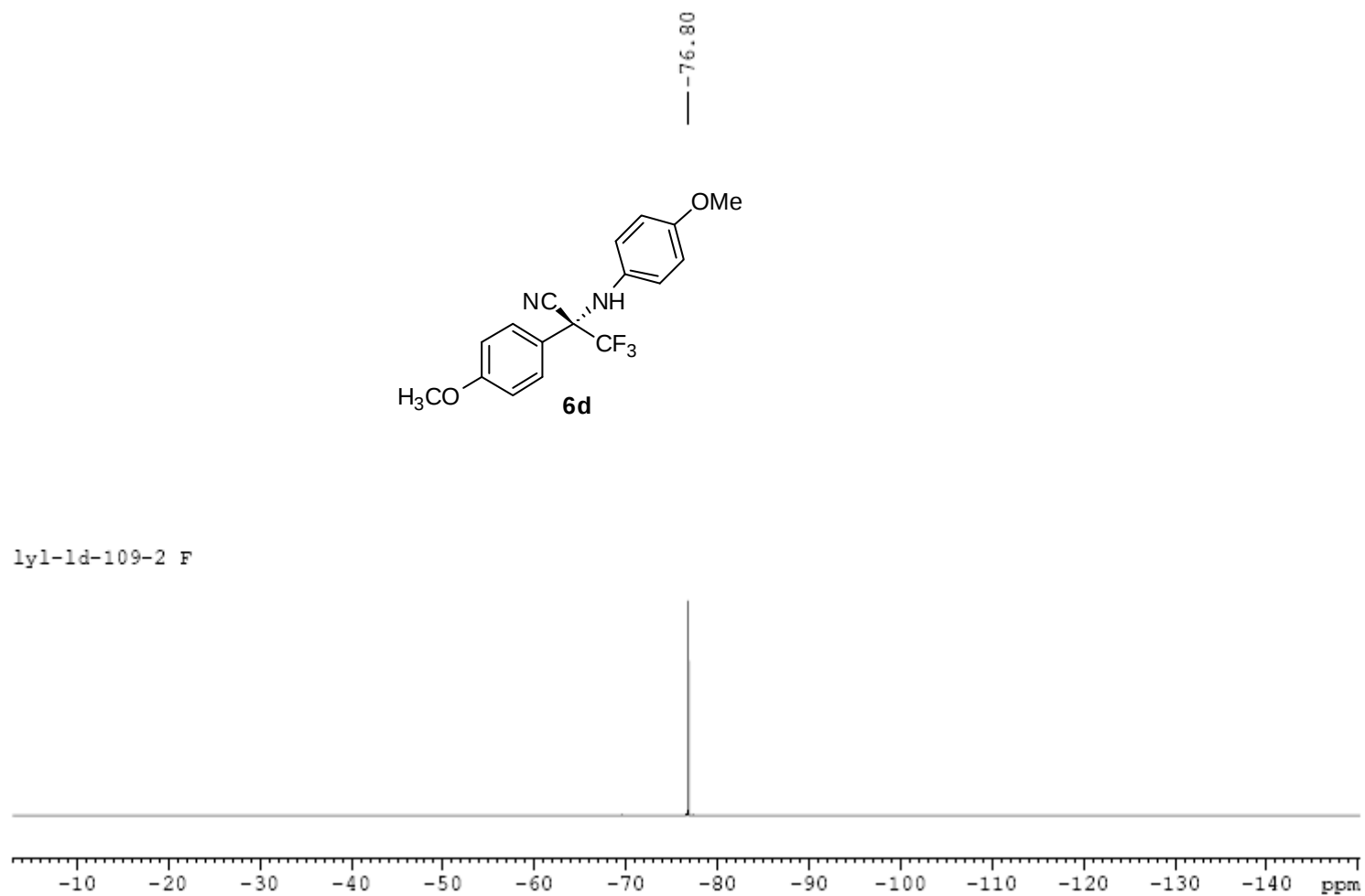
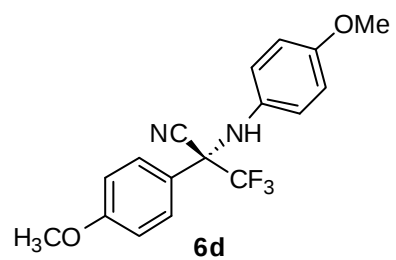
1y1-1d-109-2 h

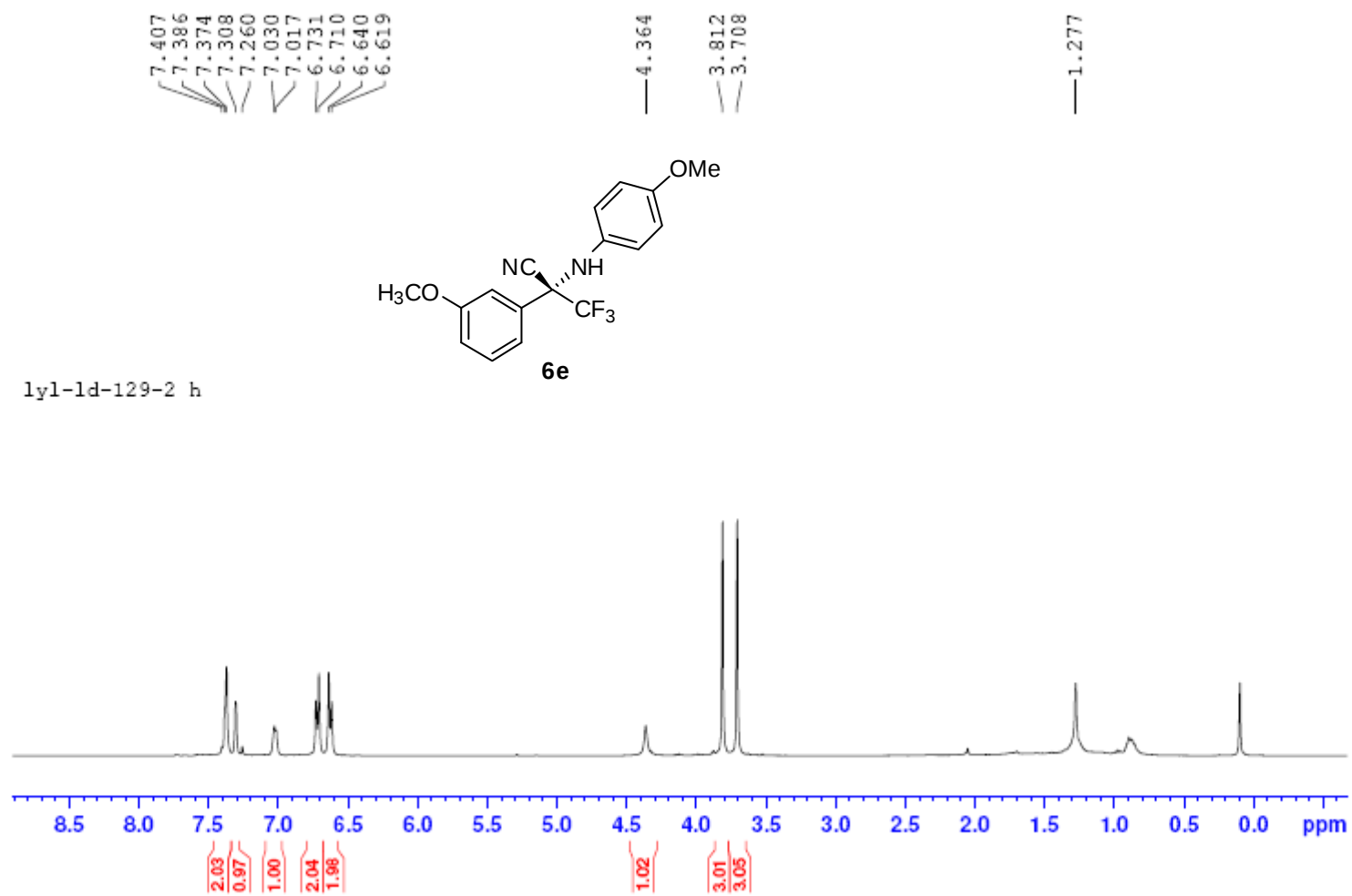


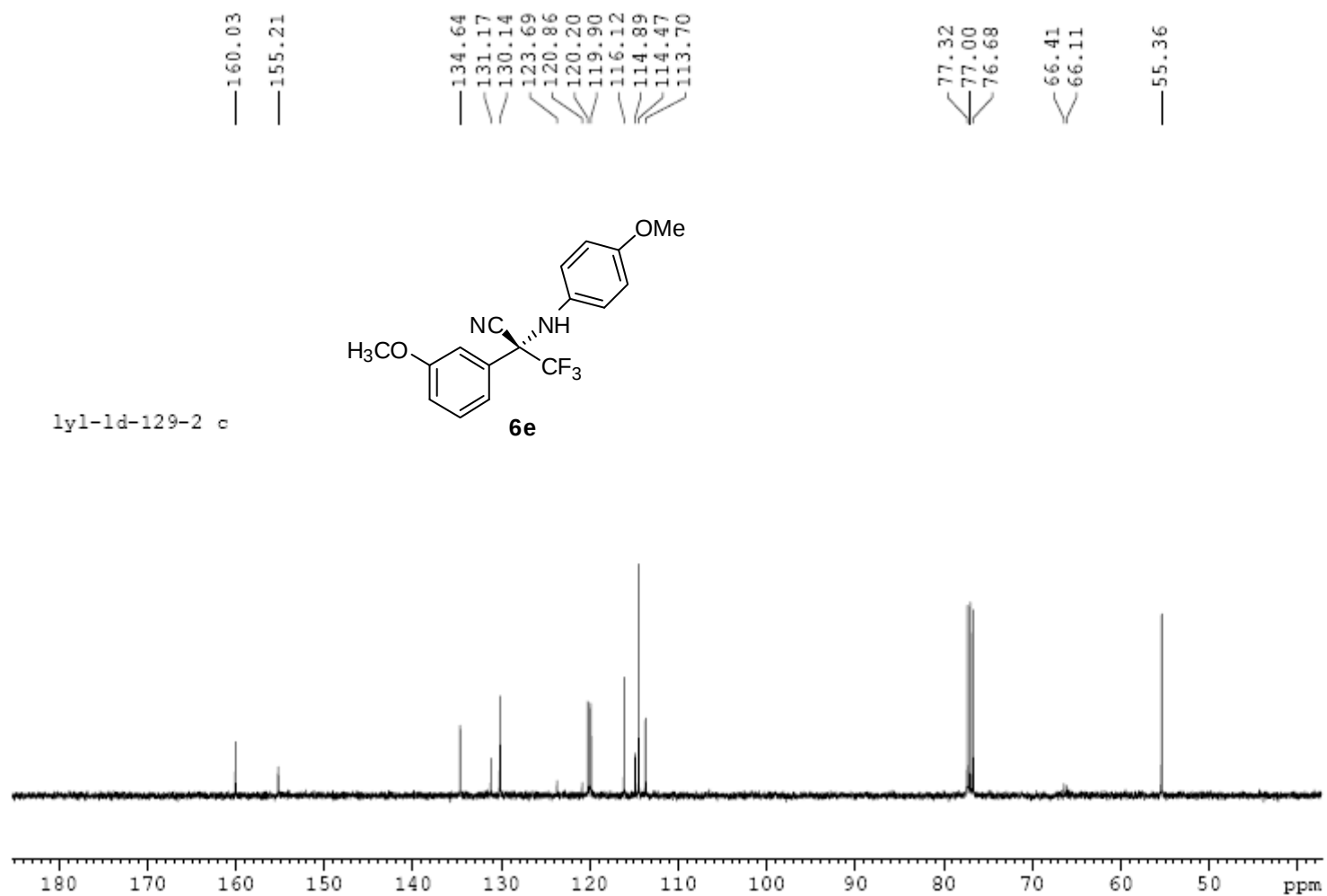
7.683
7.662
7.261
6.979
6.959
6.728
6.708
6.636
6.615

4.339
3.839
3.708

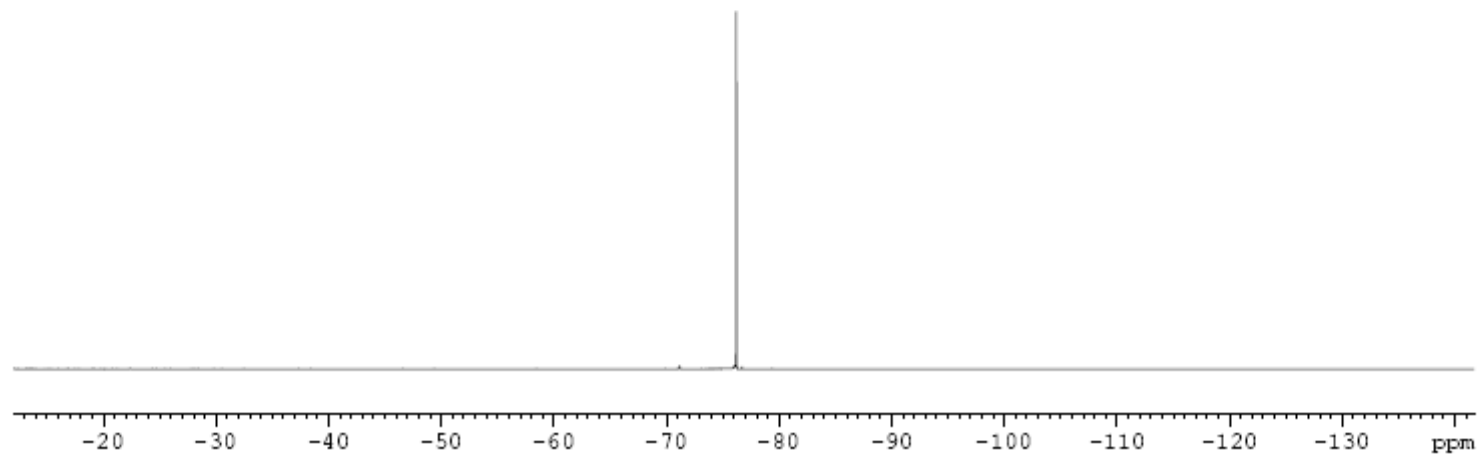
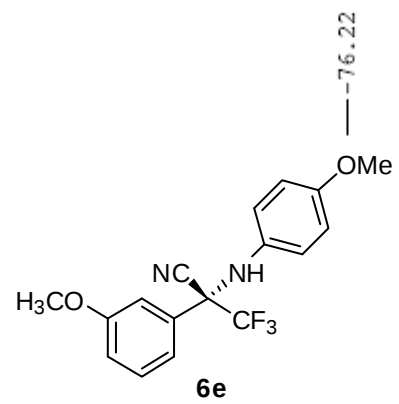




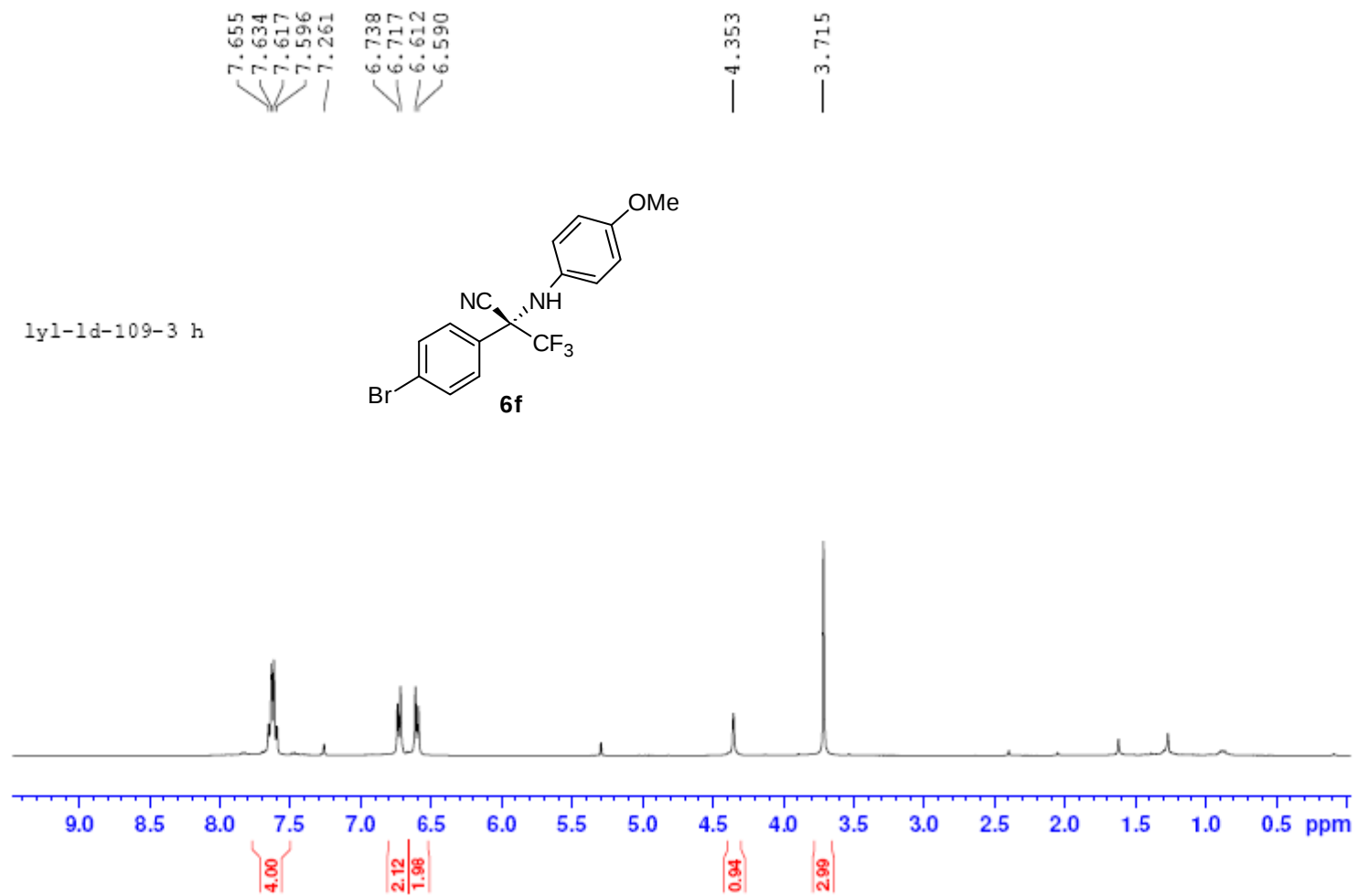
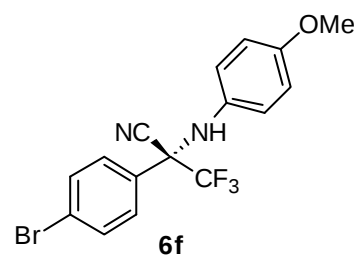


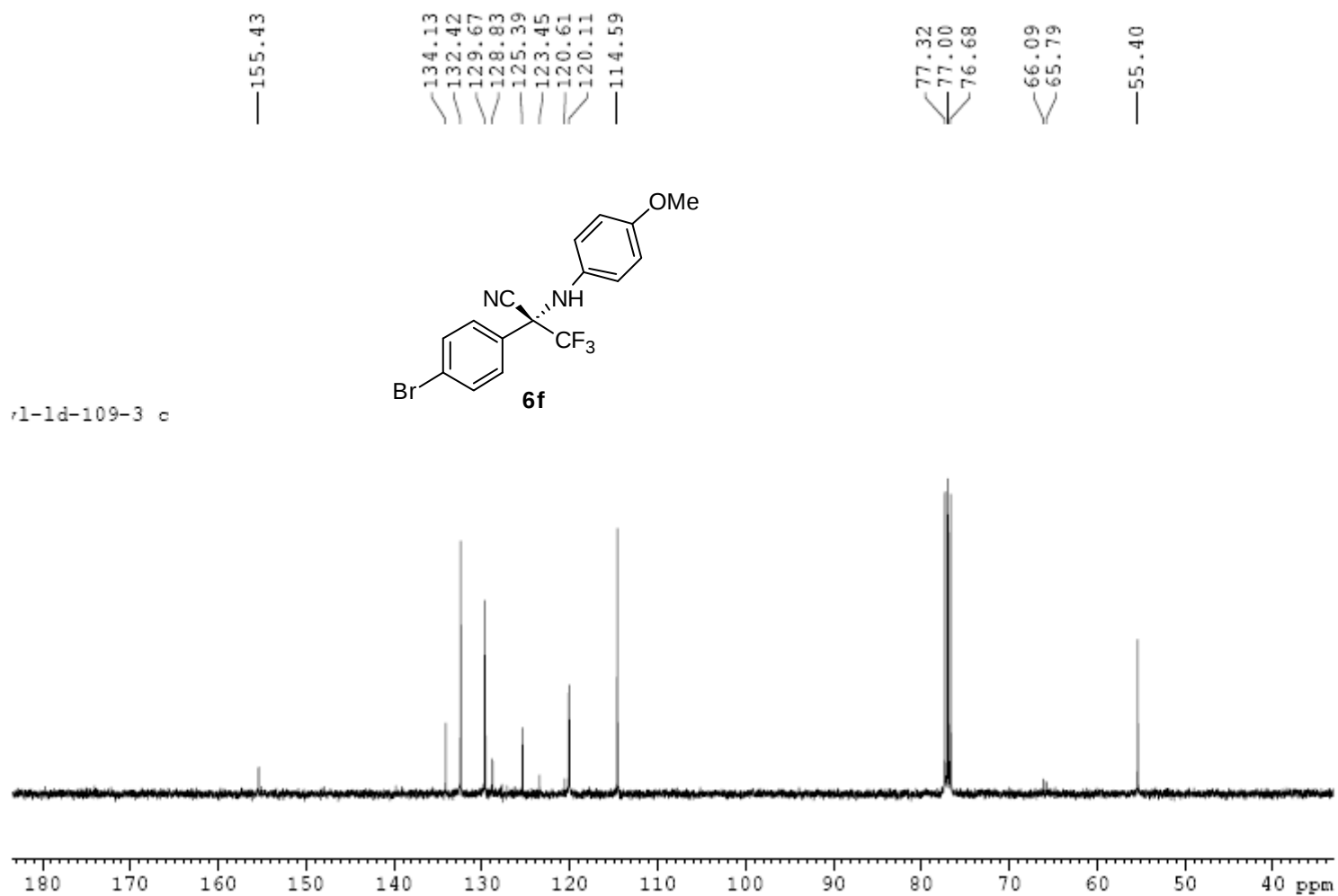


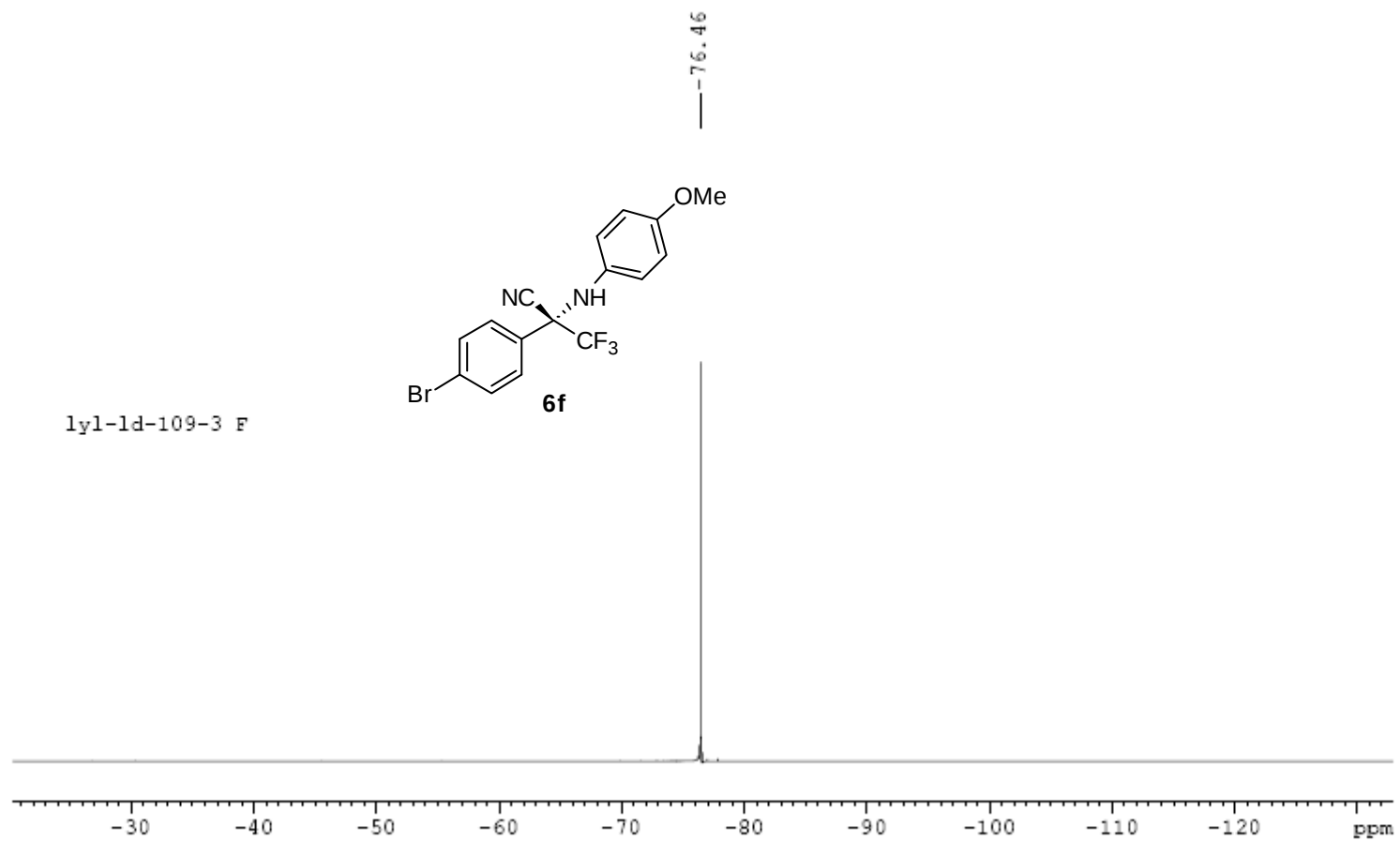
1y1-1d-129-2 f

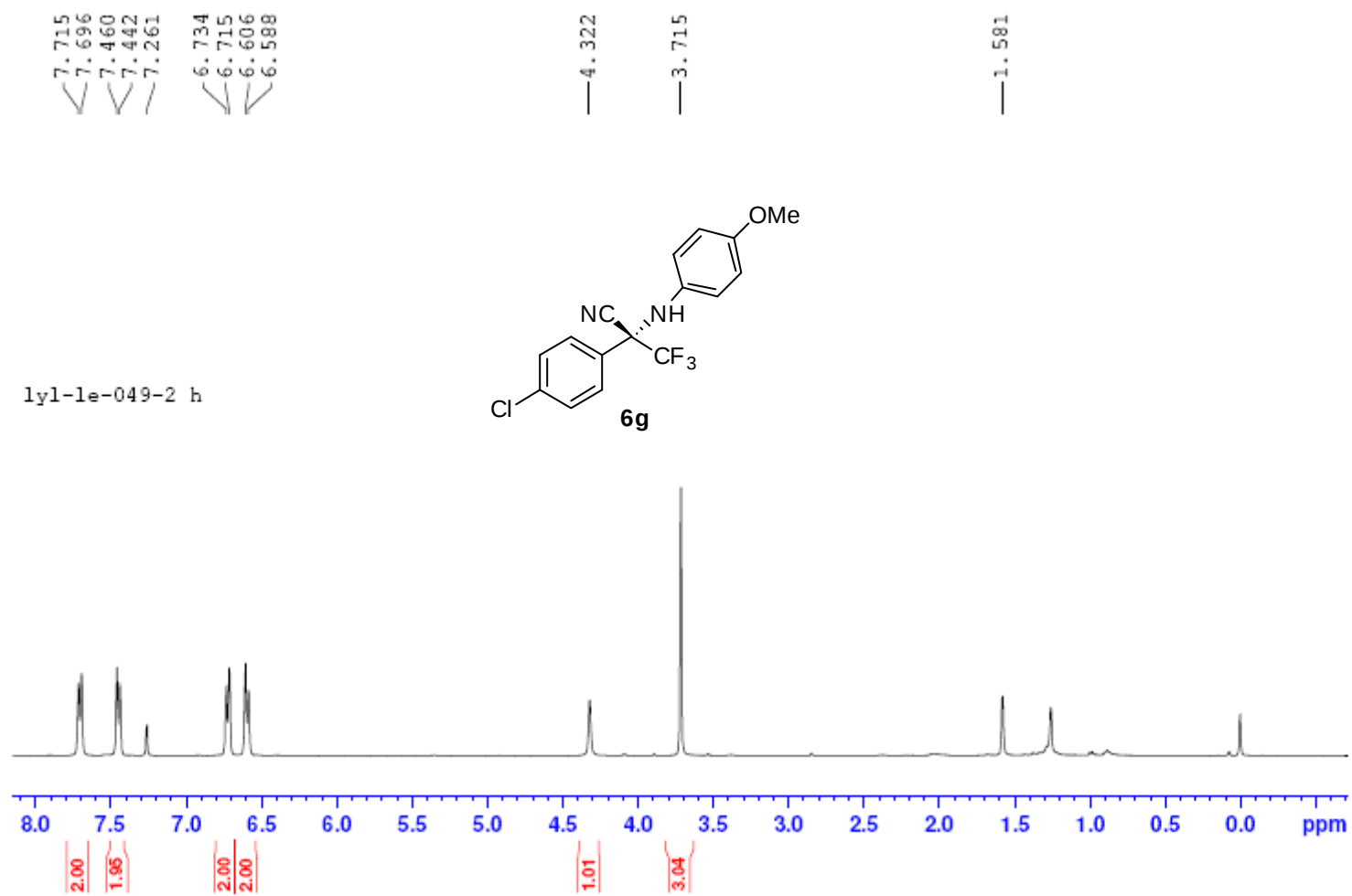


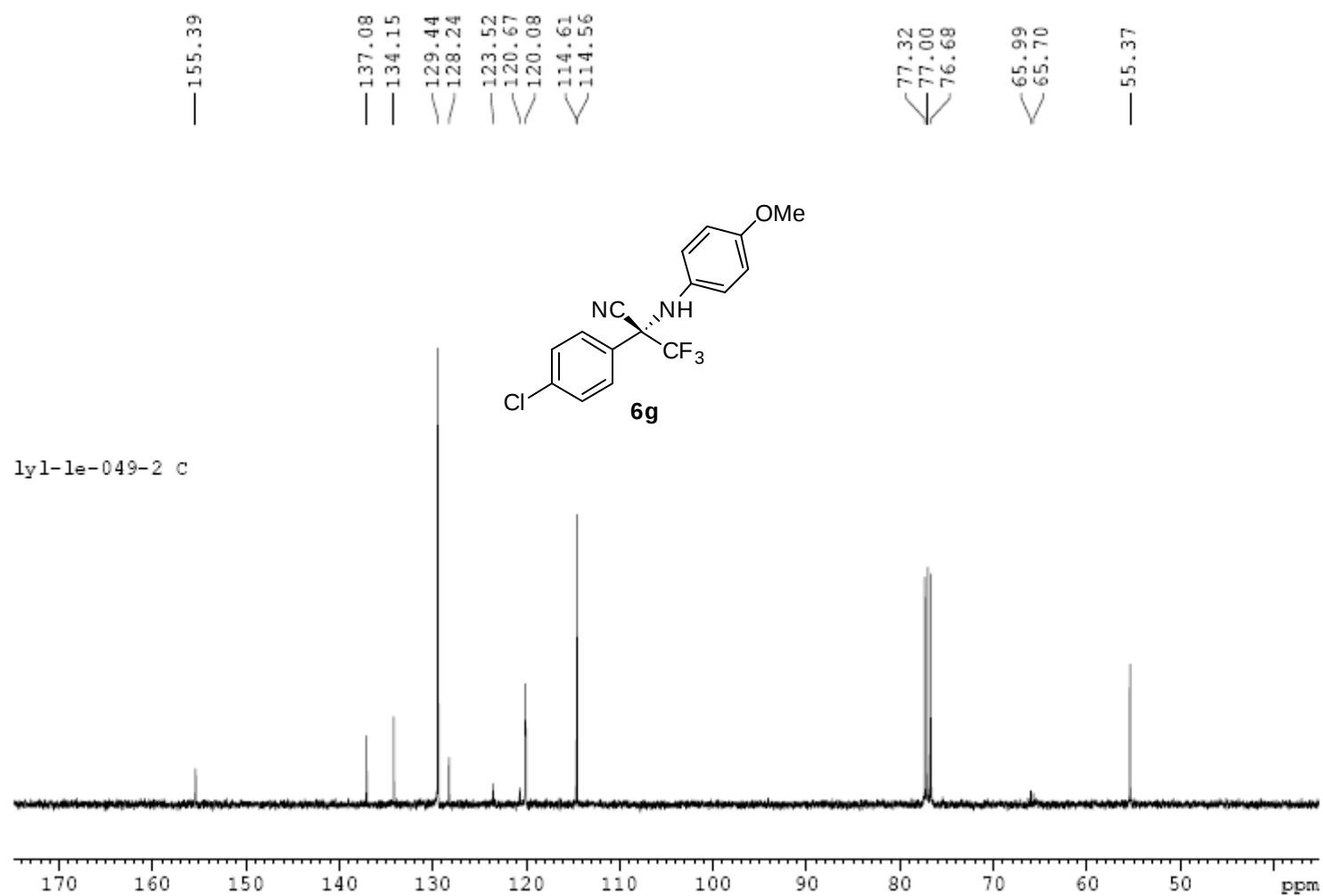
1y1-1d-109-3 h

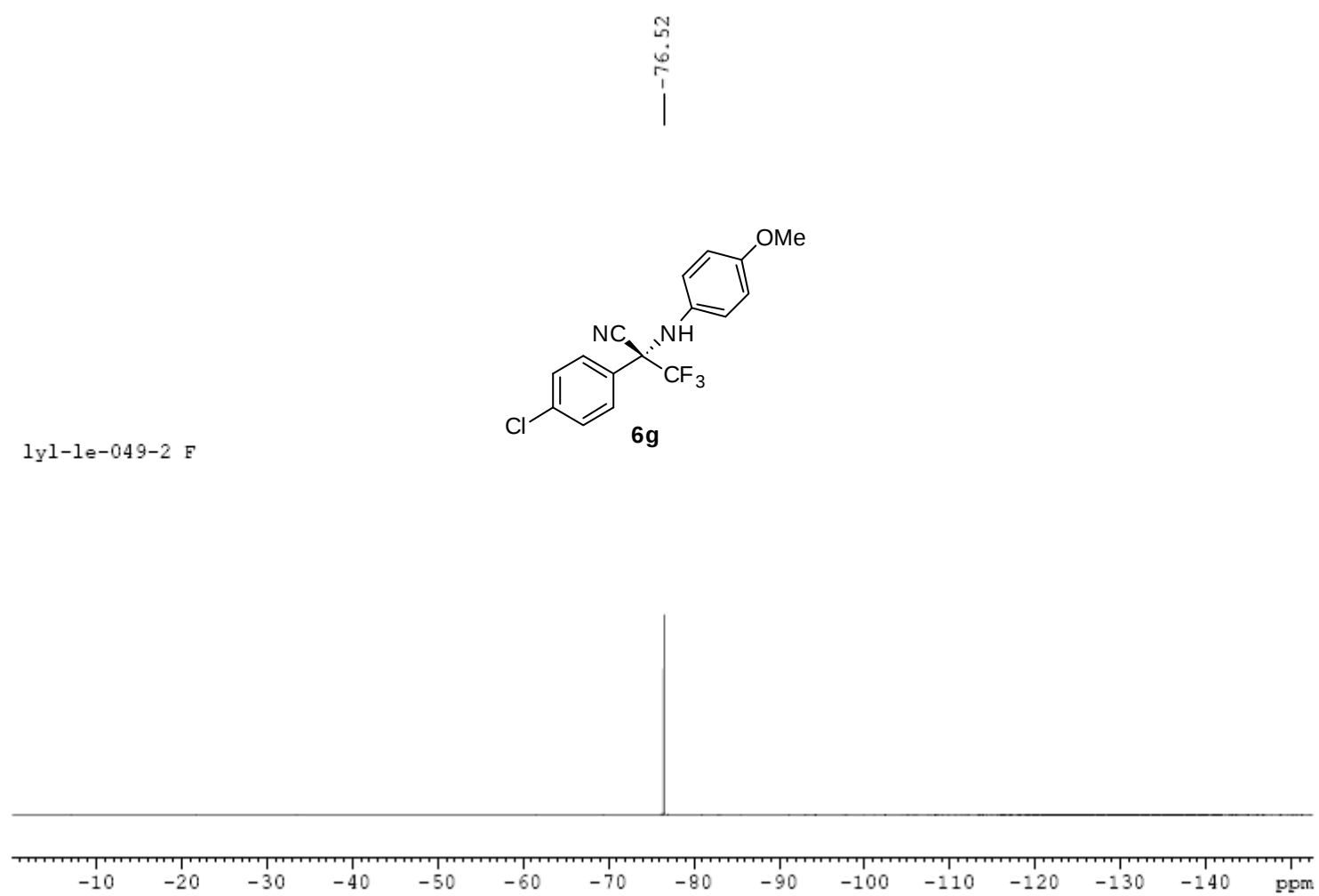


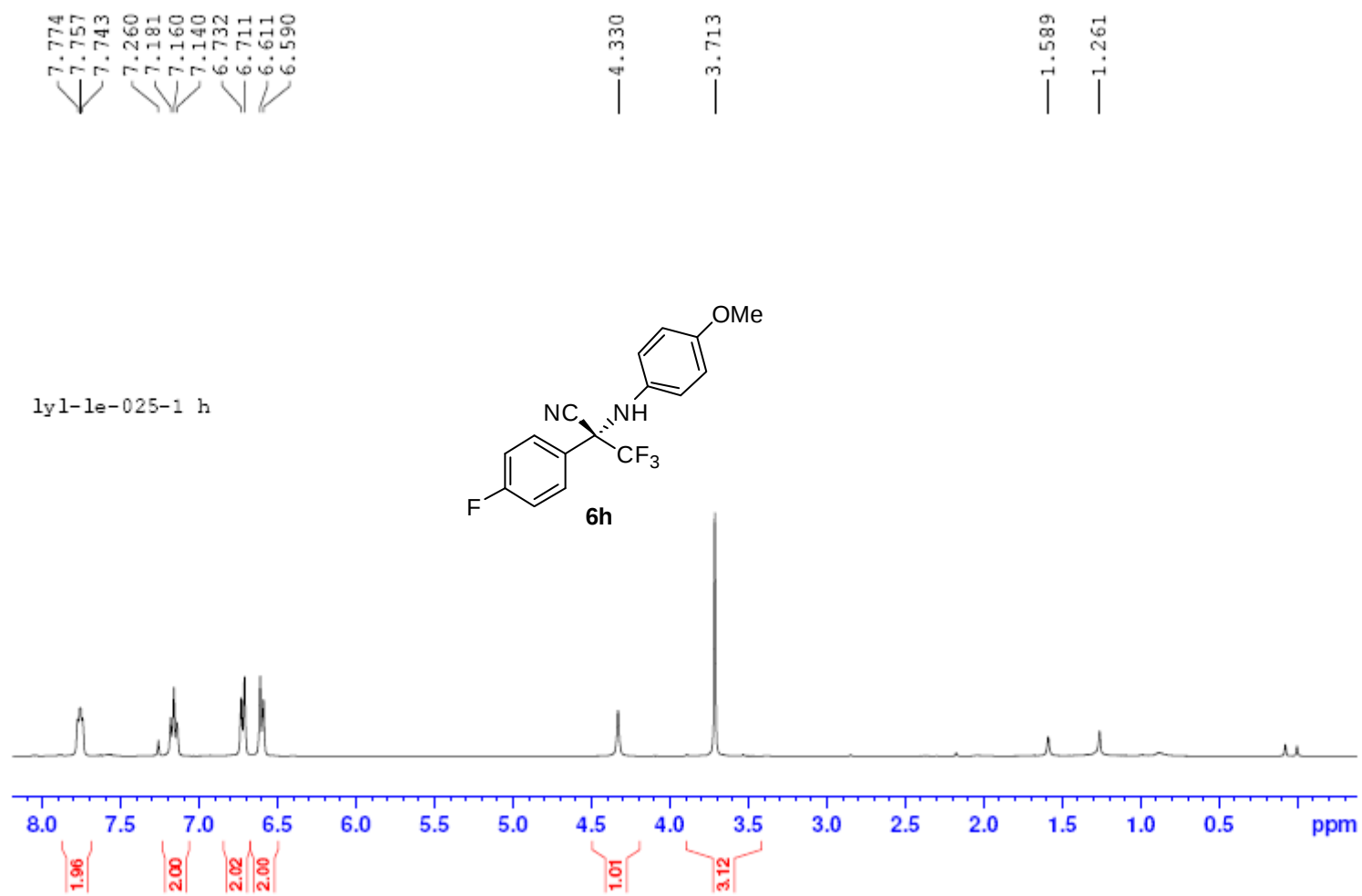


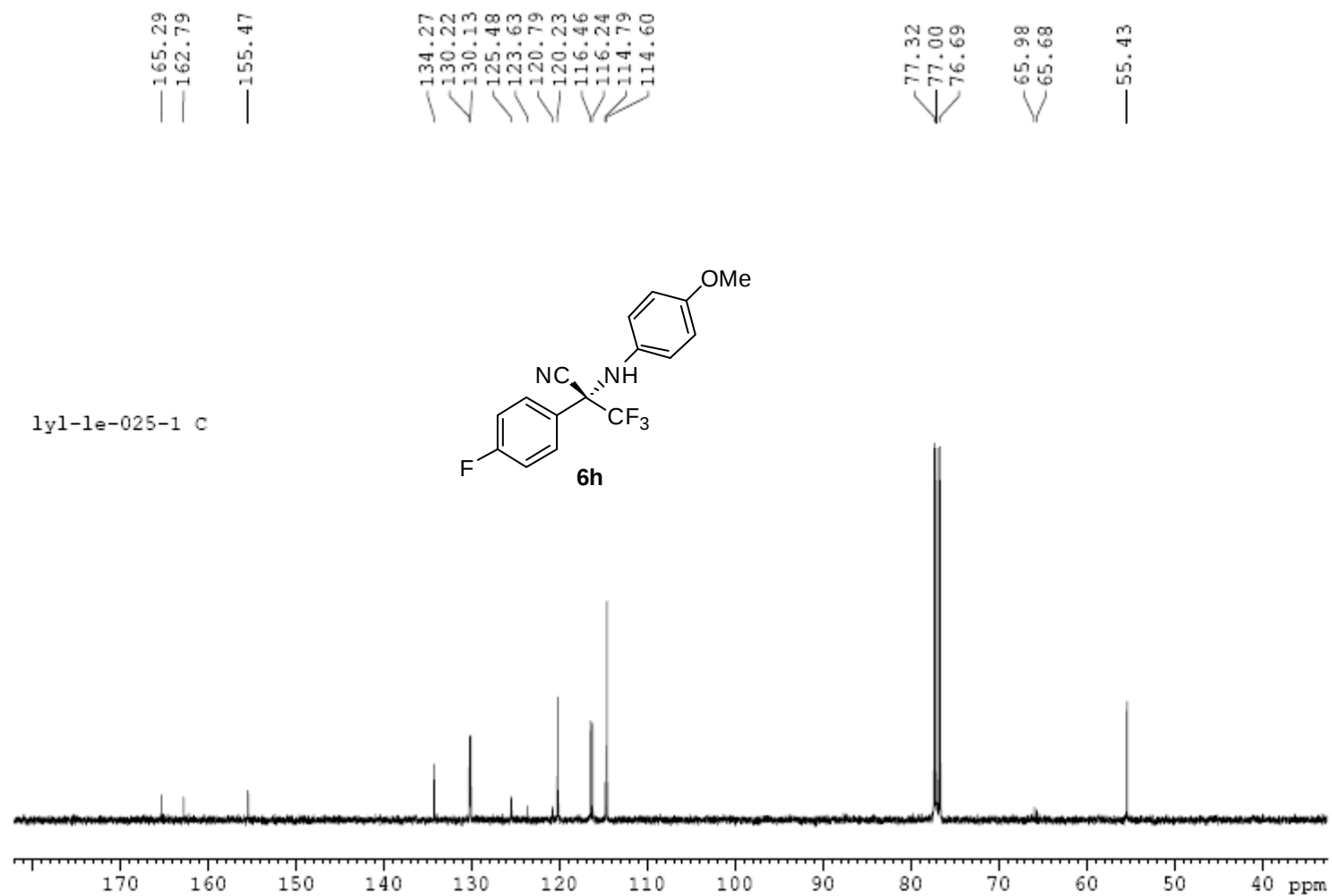


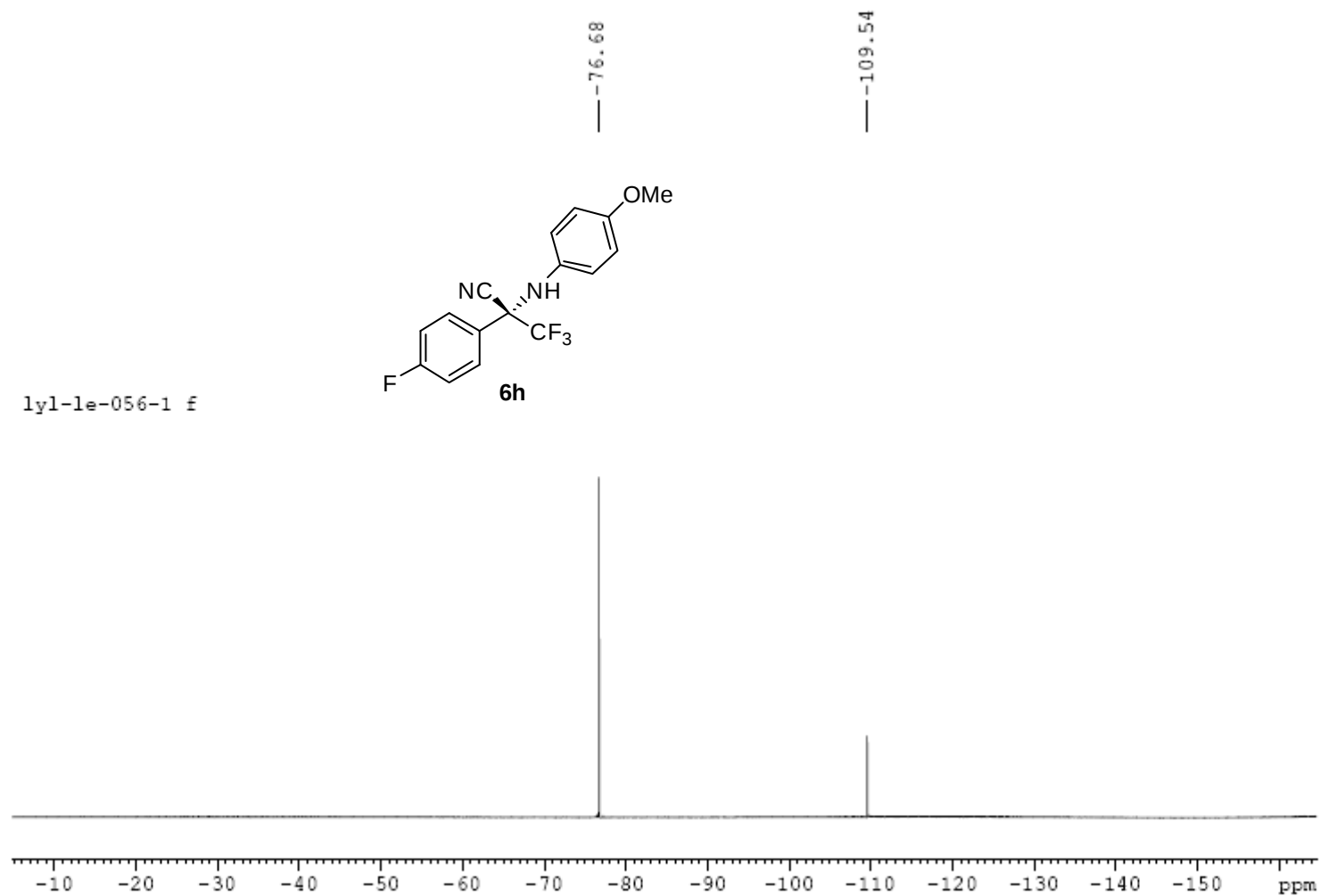


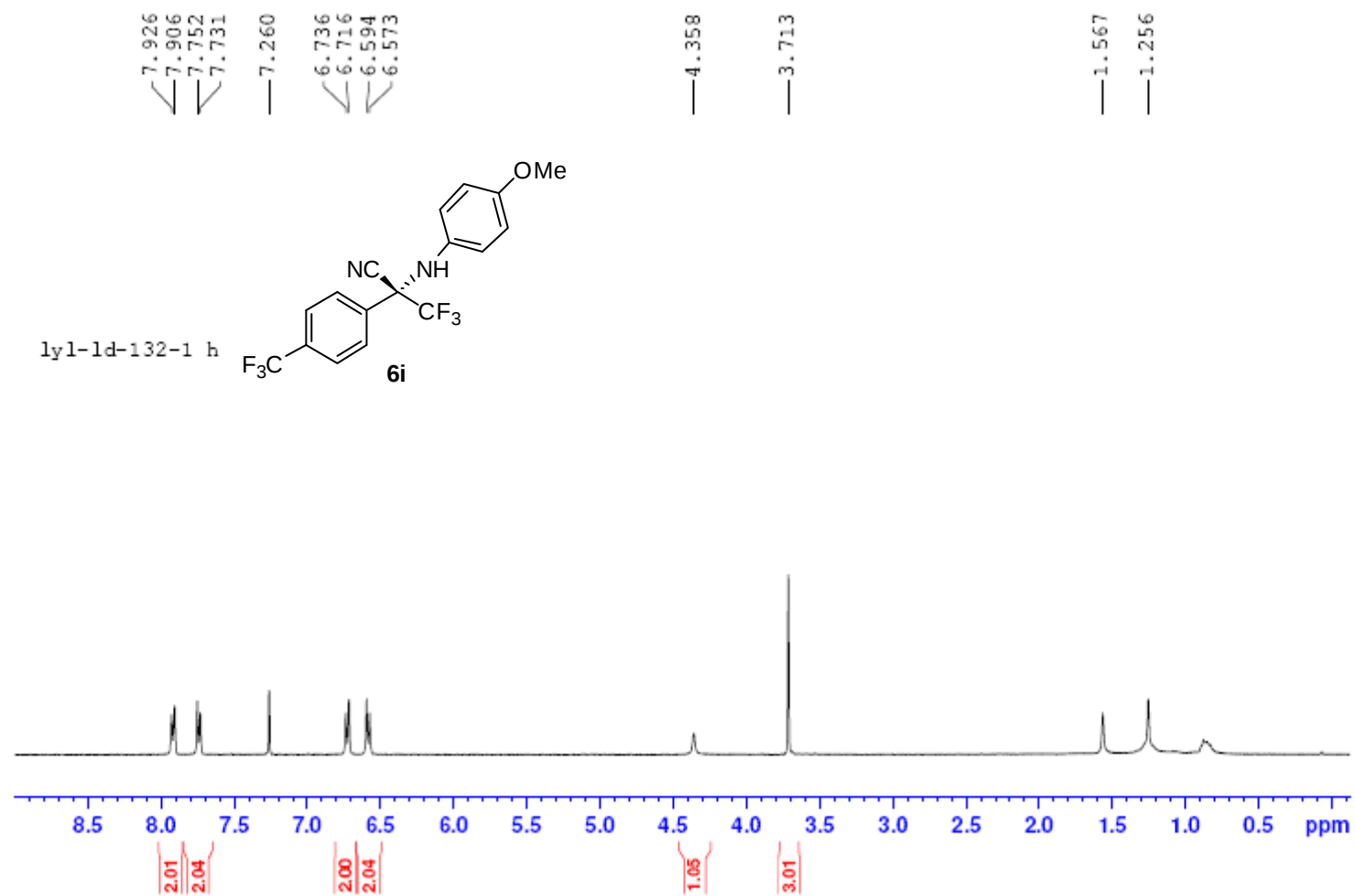




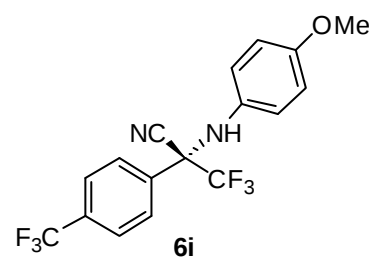




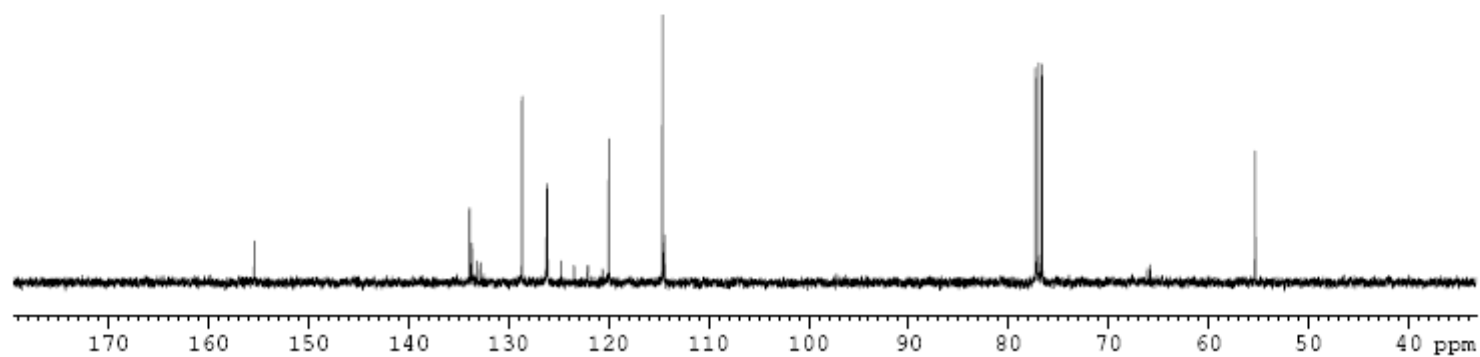




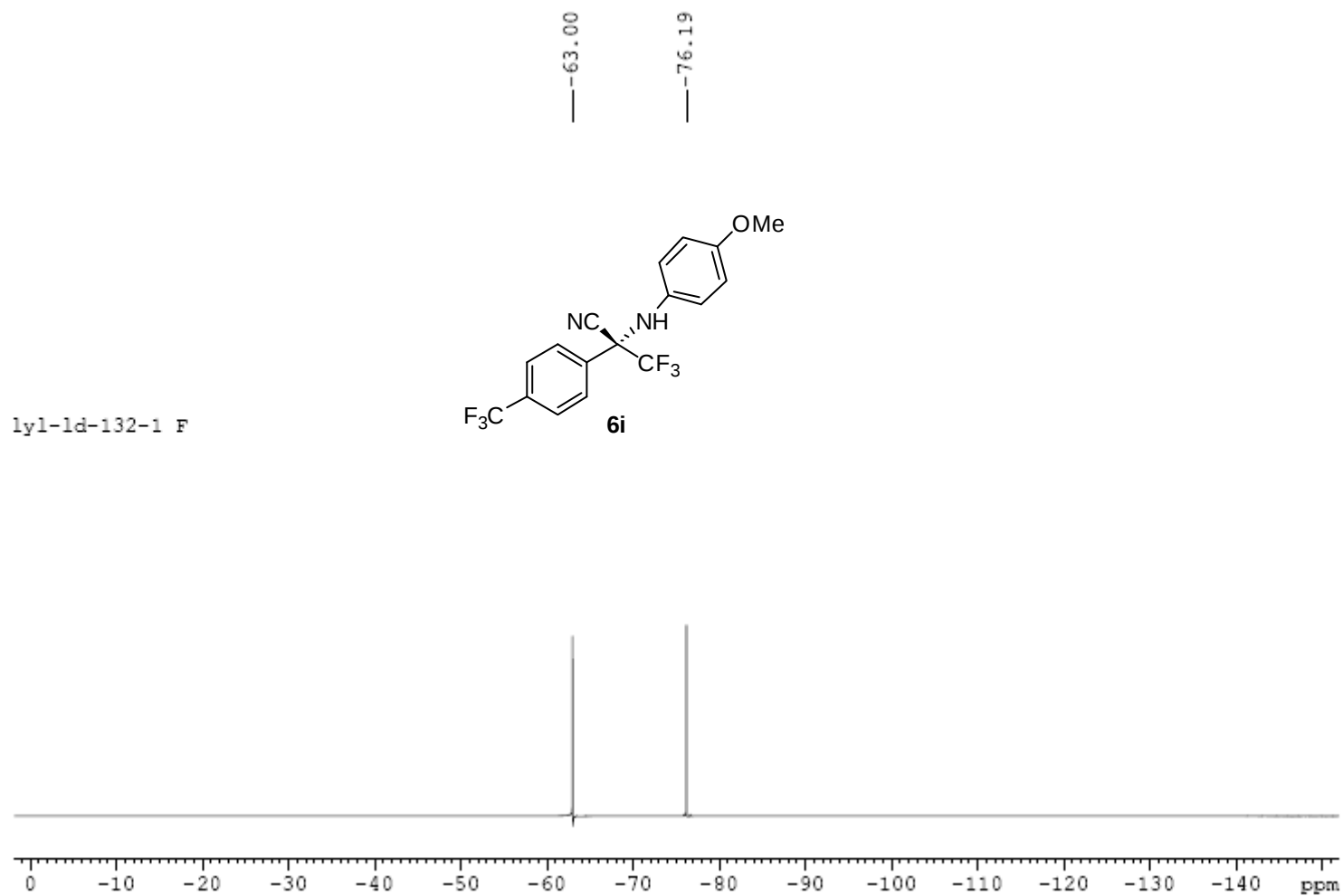
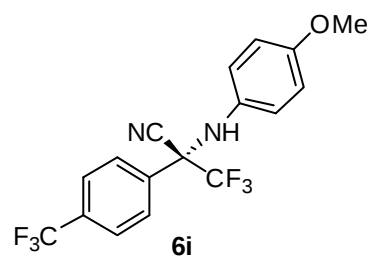
— 155.46
 133.94
 133.70
 133.15
 132.82
 128.69
 126.20
 126.17
 124.80
 123.47
 122.09
 120.62
 120.00
 114.63
 114.43
 77.32
 77.00
 76.68
 66.13
 65.83
 — 55.36

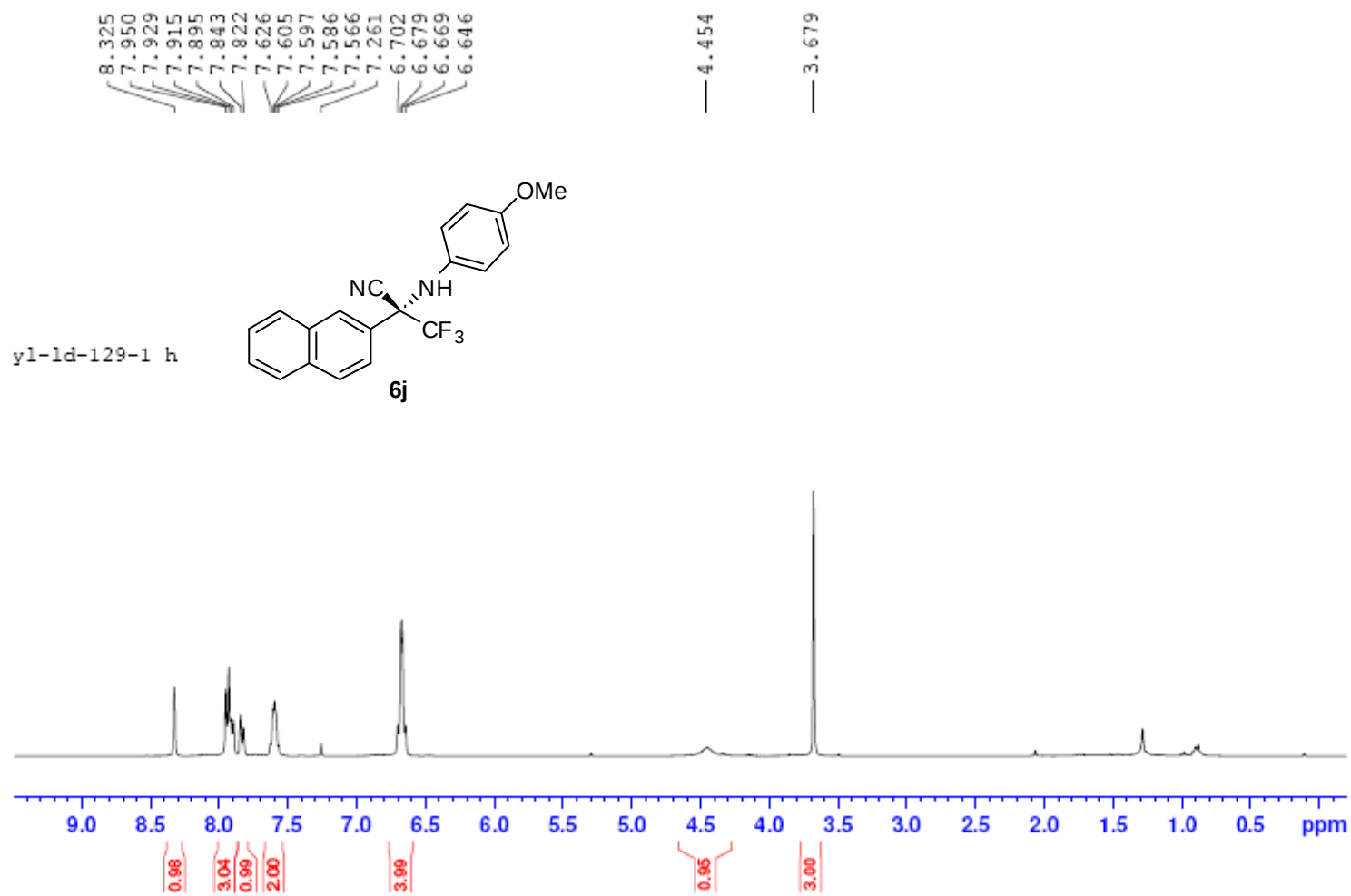


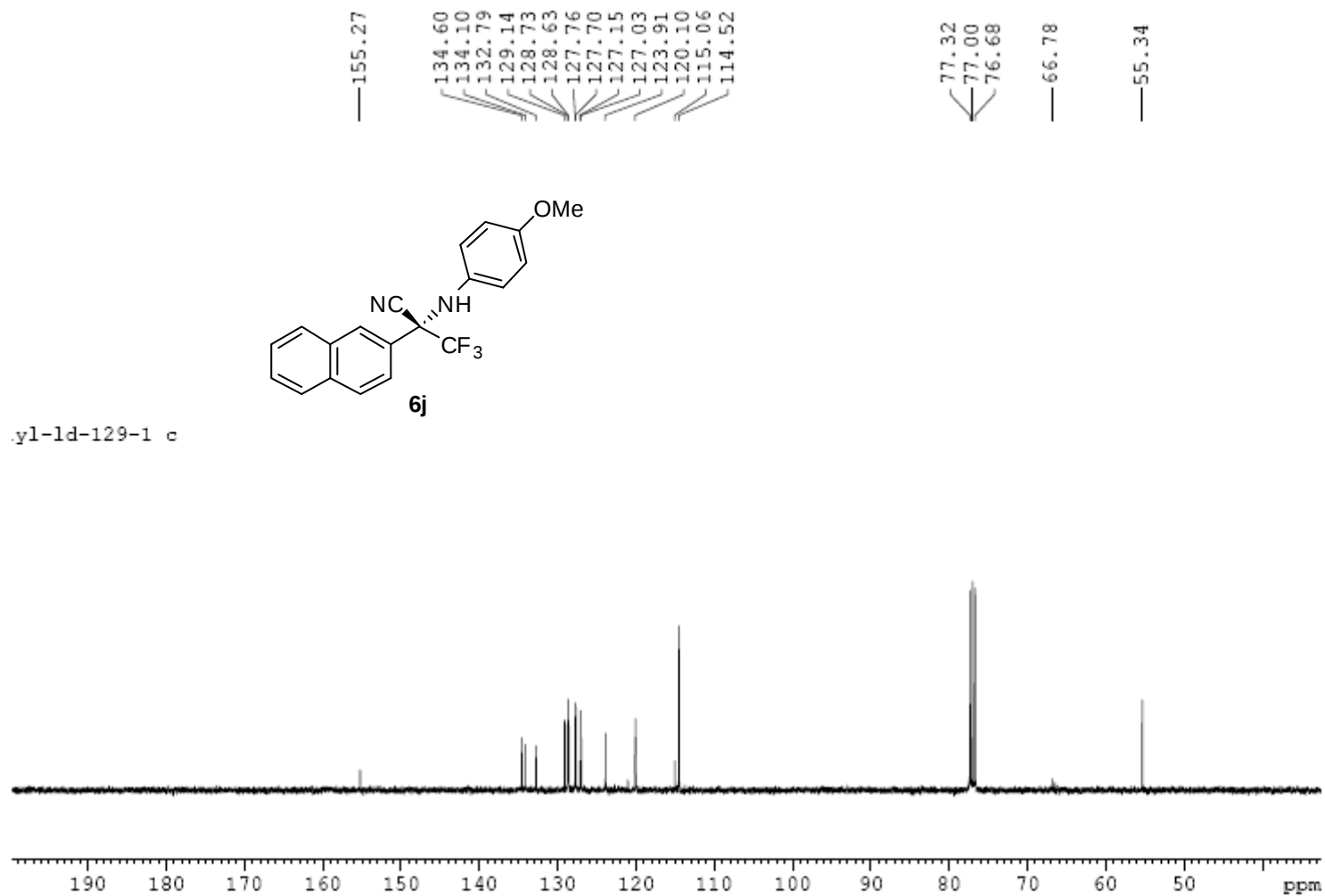
1y1-1e-056-2 c

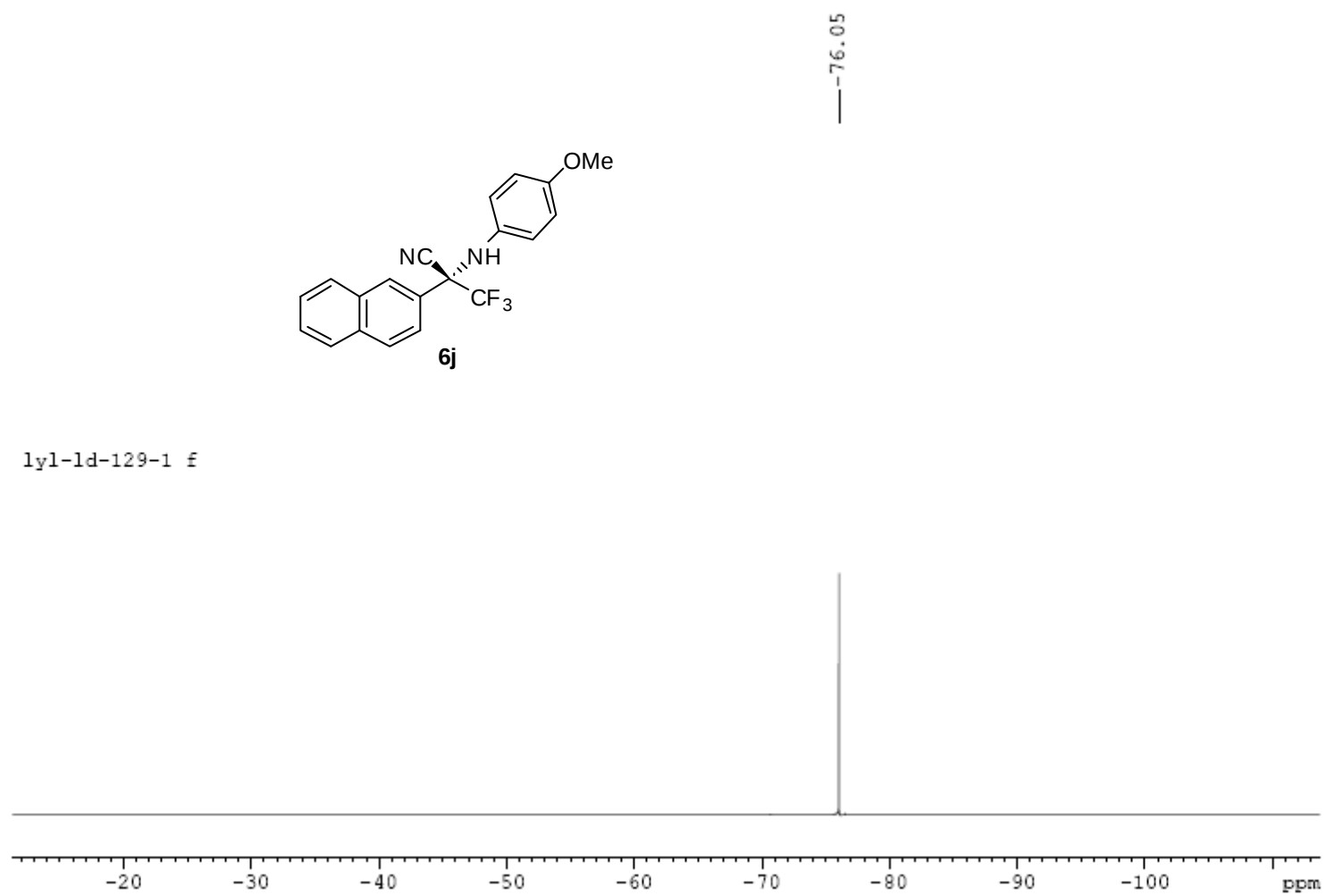


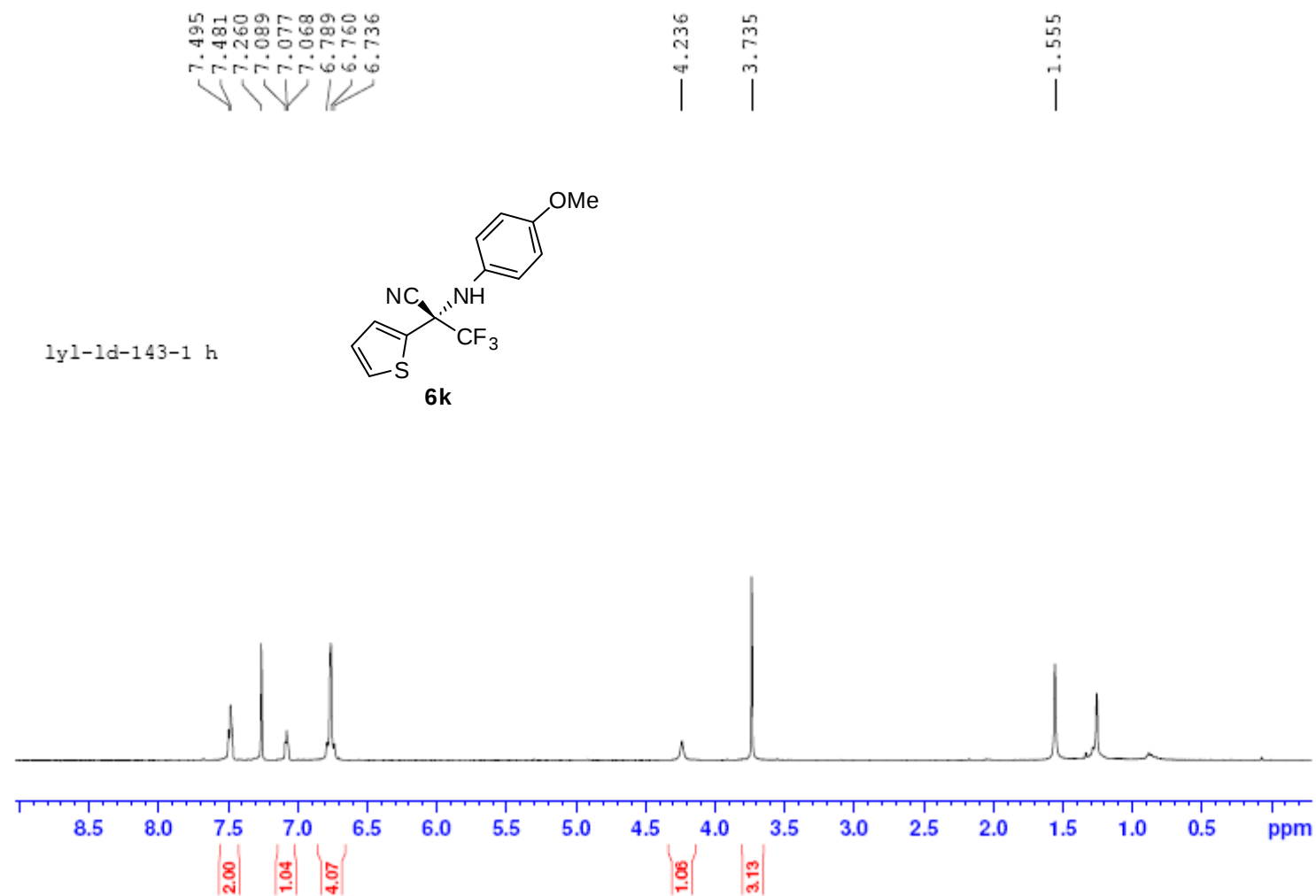
1y1-1d-132-1 F

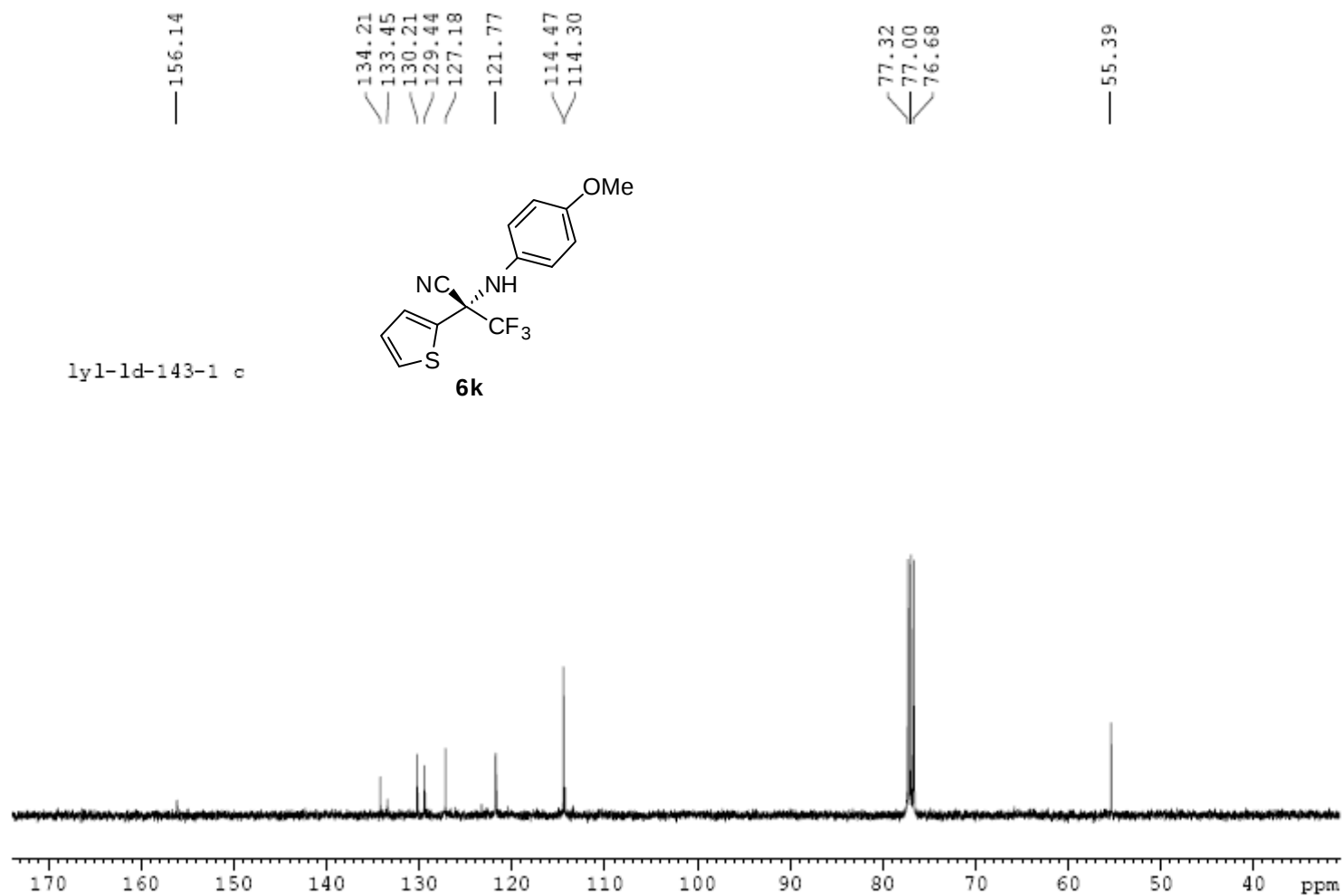


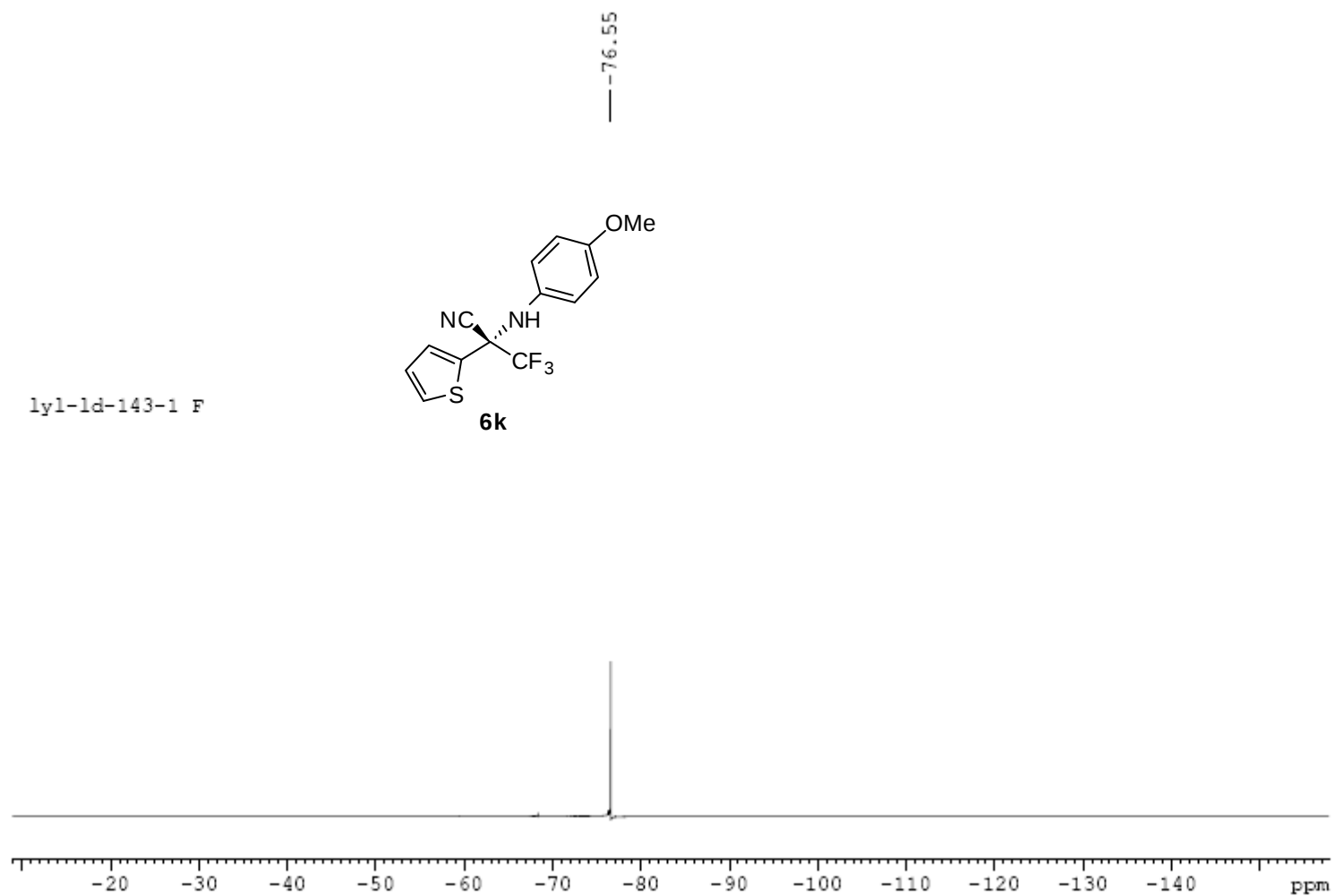




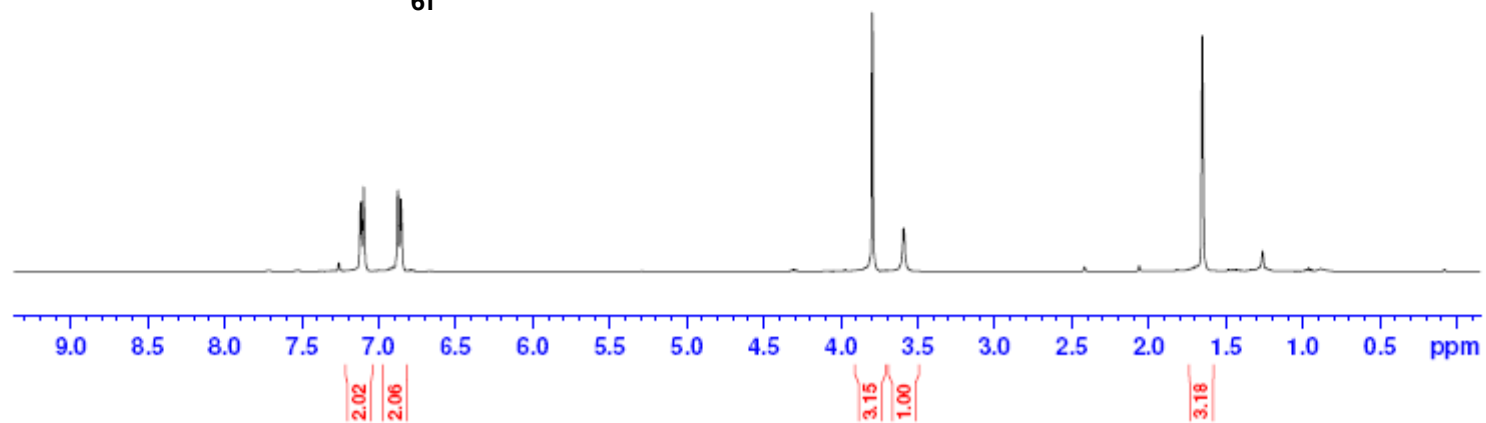
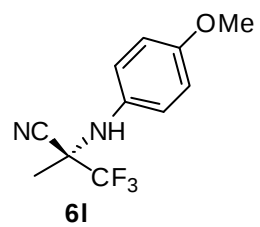








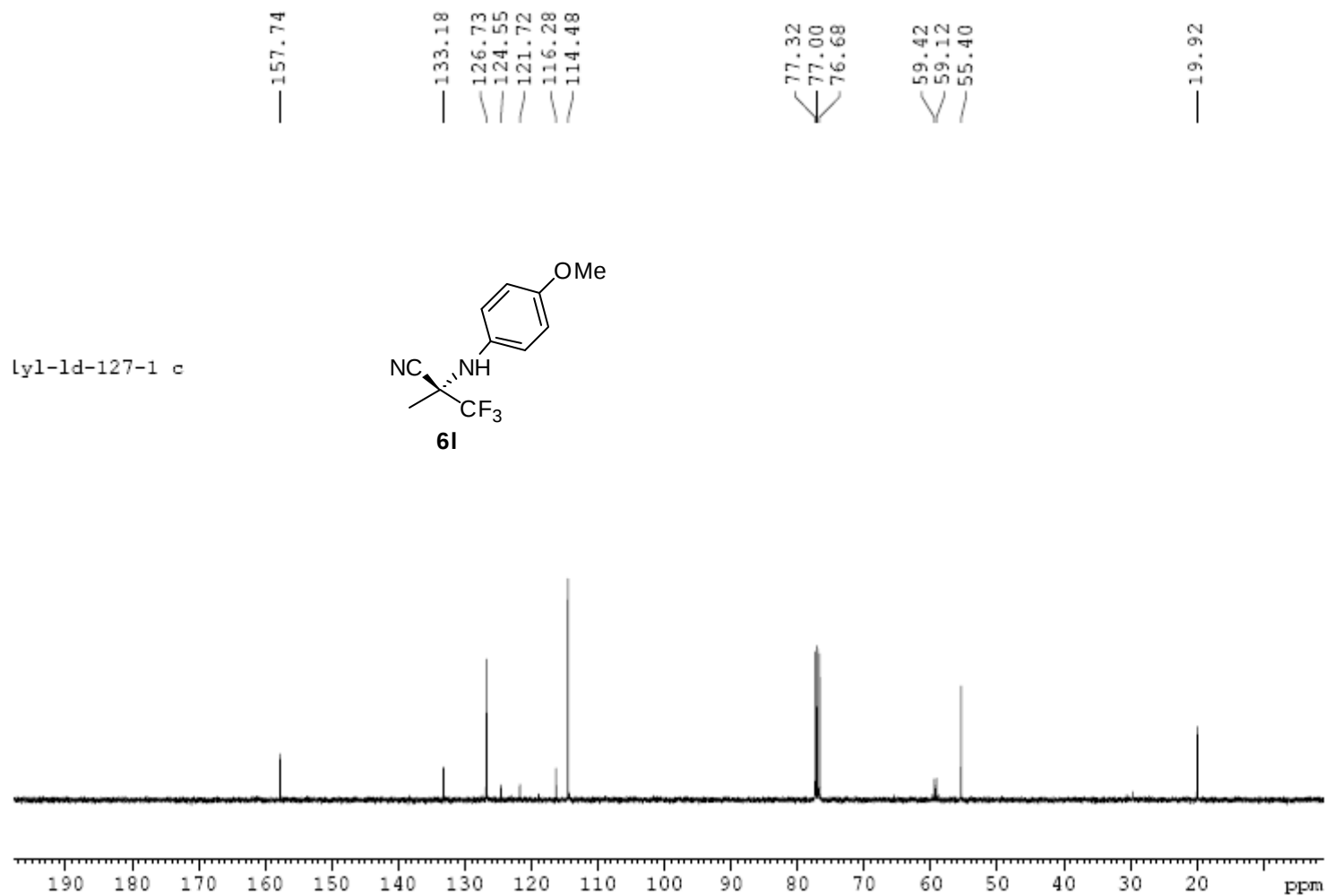
.yl-ld-127-1 h

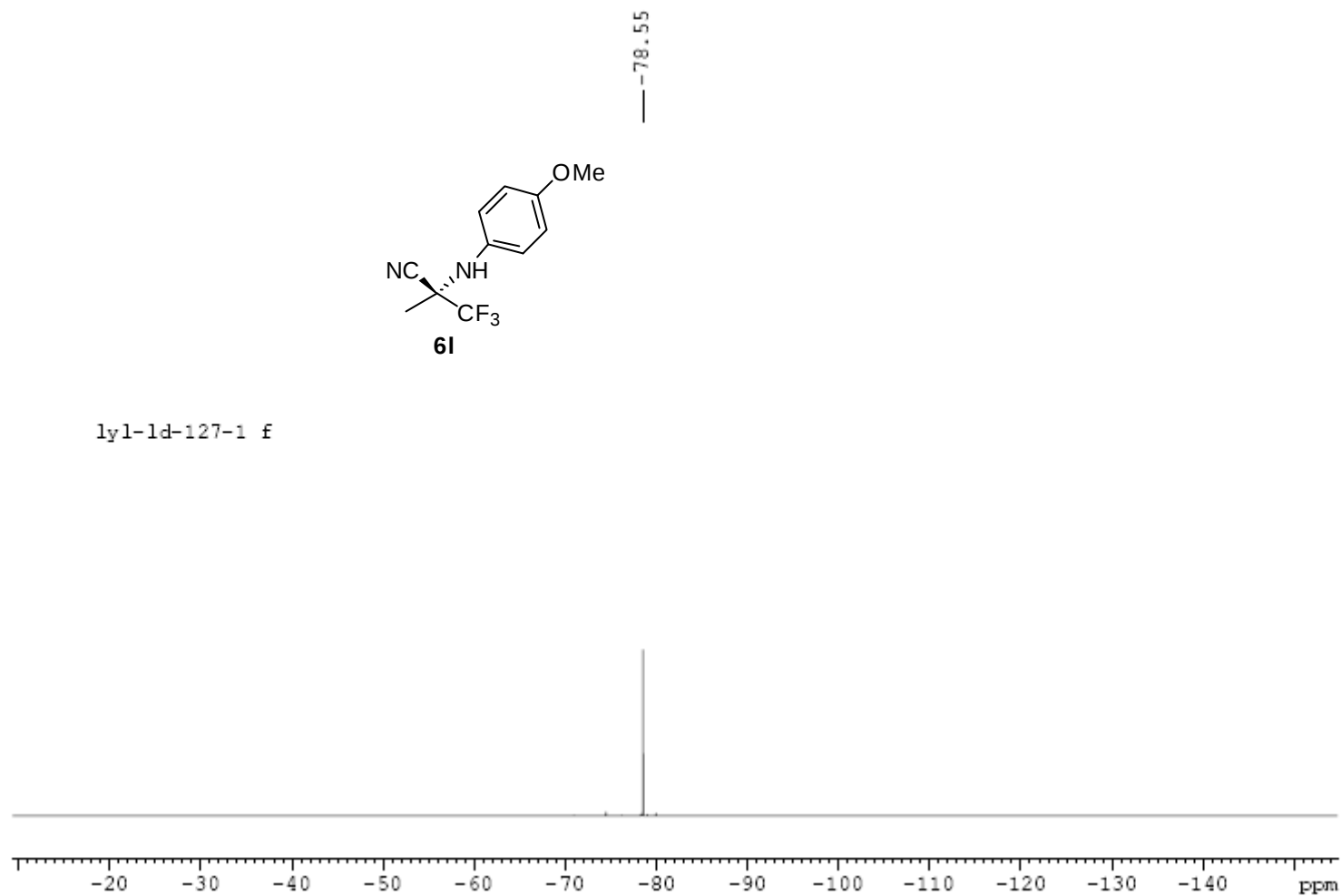


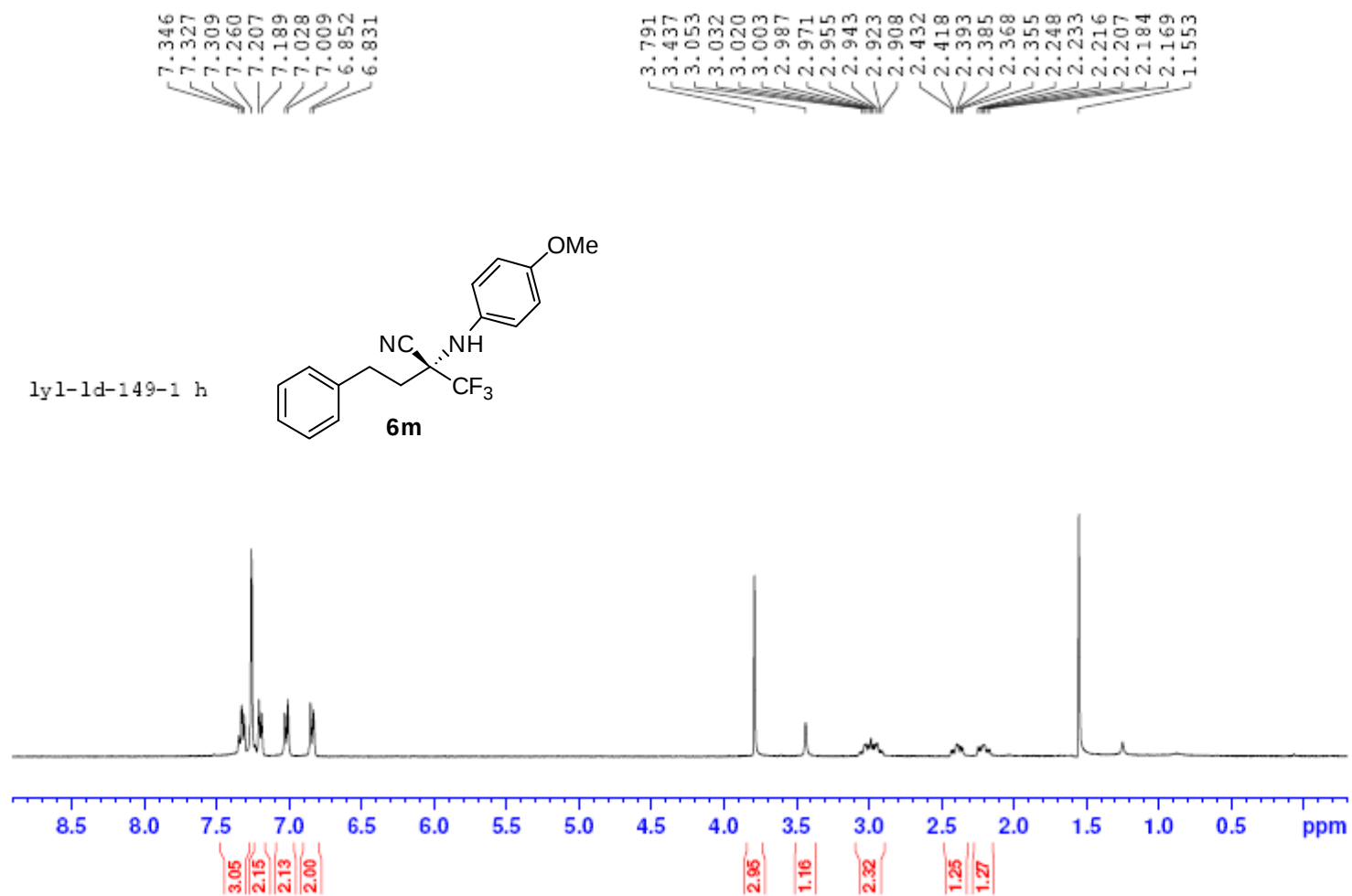
7.261
 7.118
 7.099
 6.876
 6.857

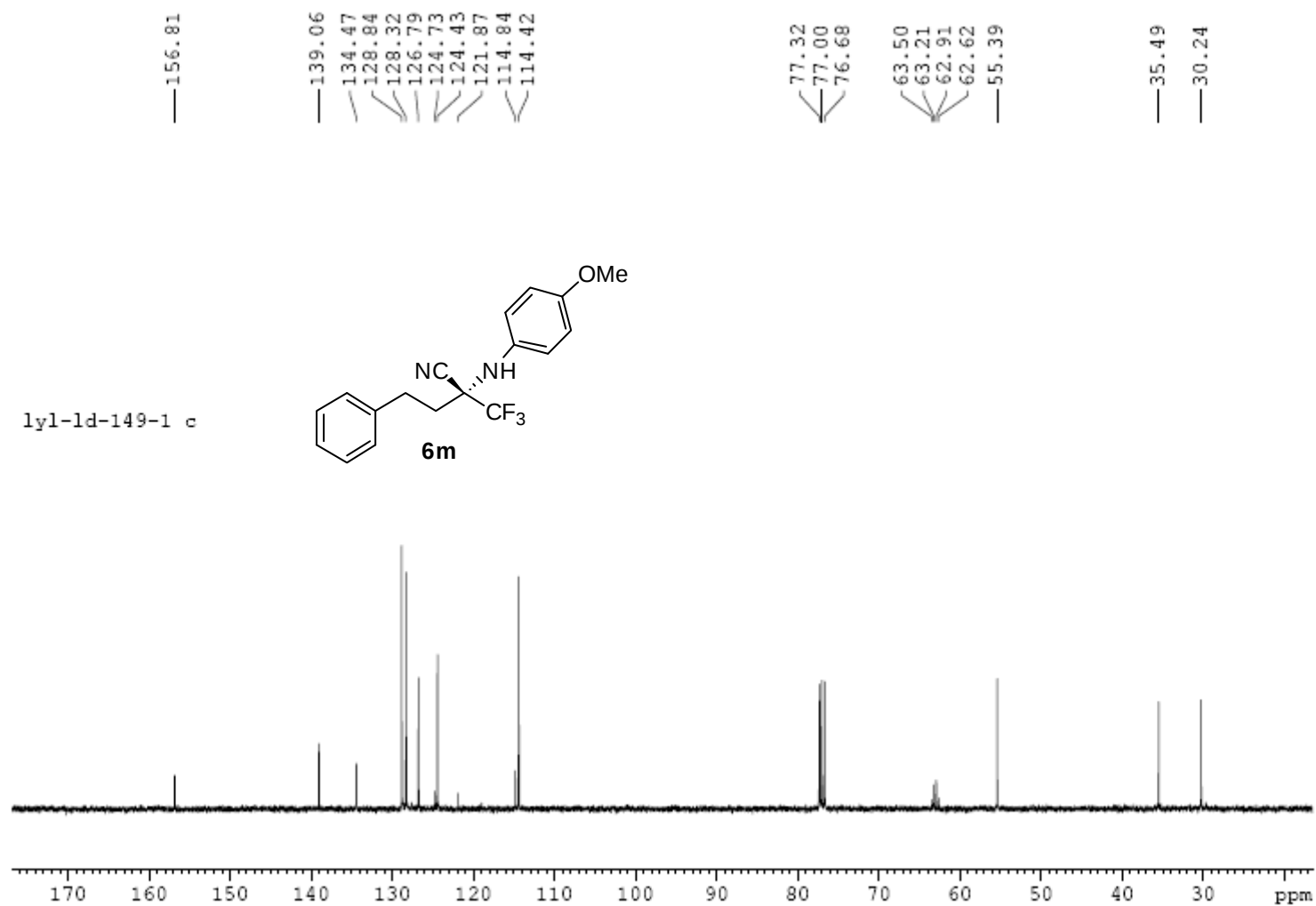
3.798
 3.593

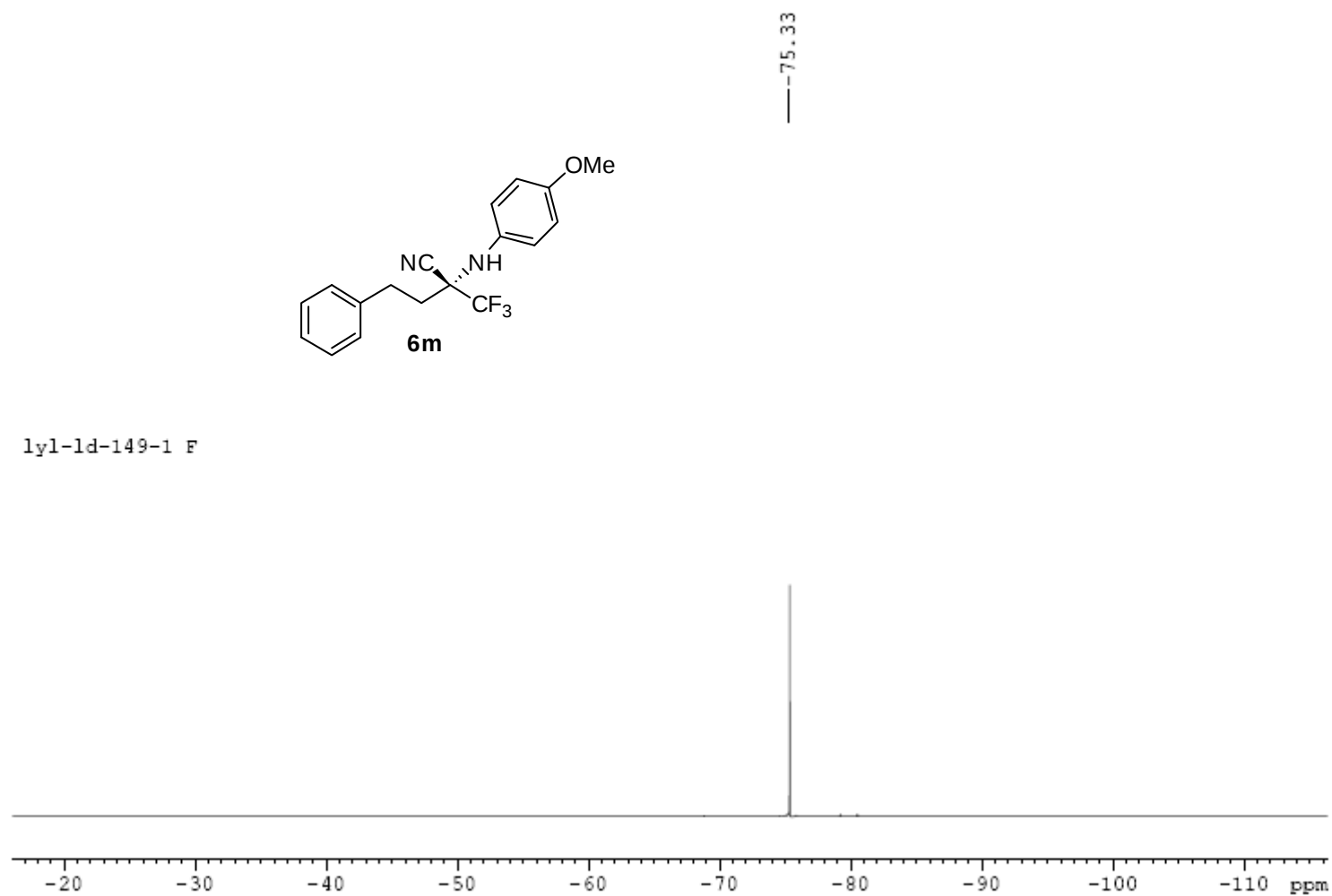
1.654
 1.263

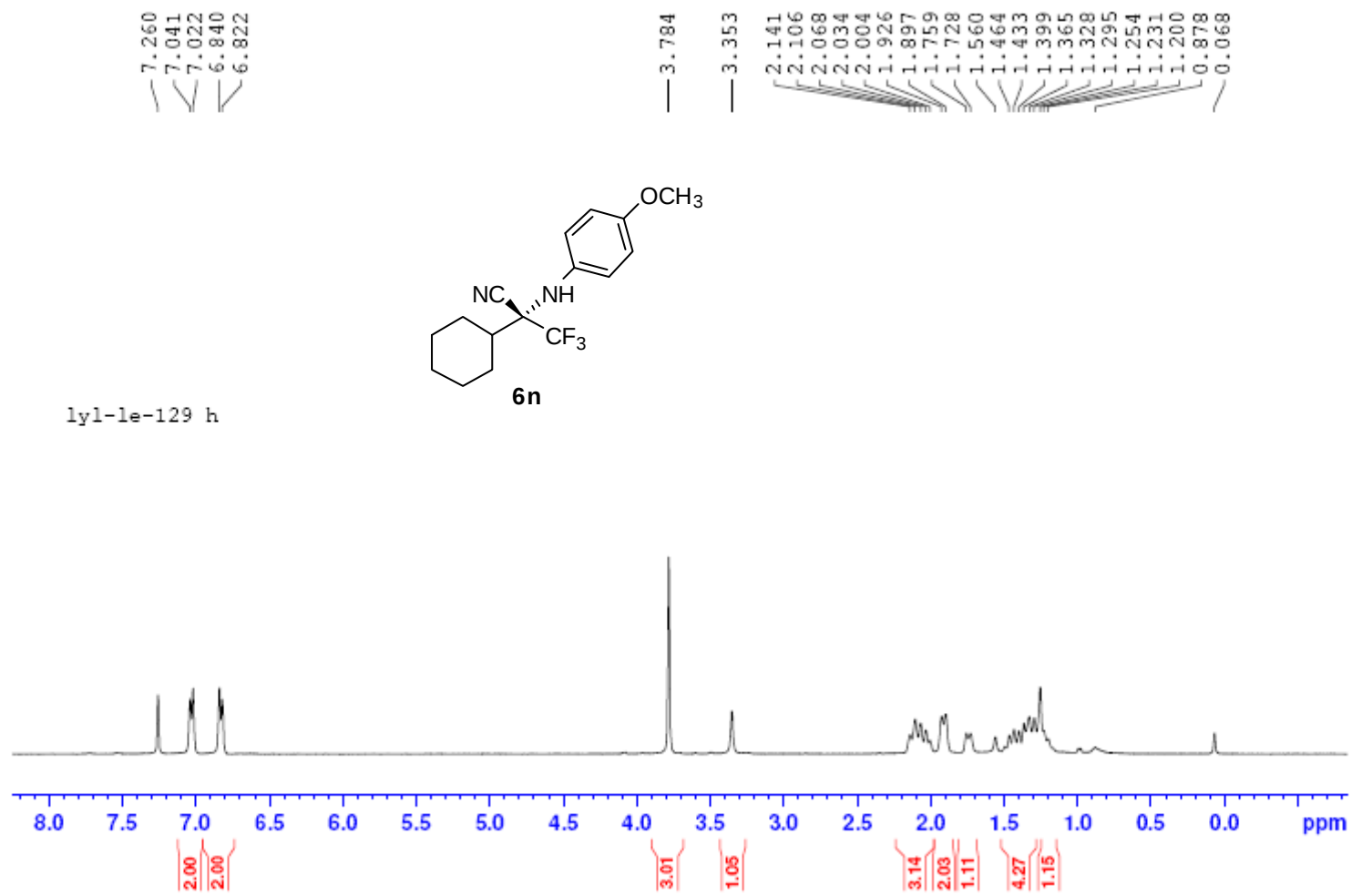


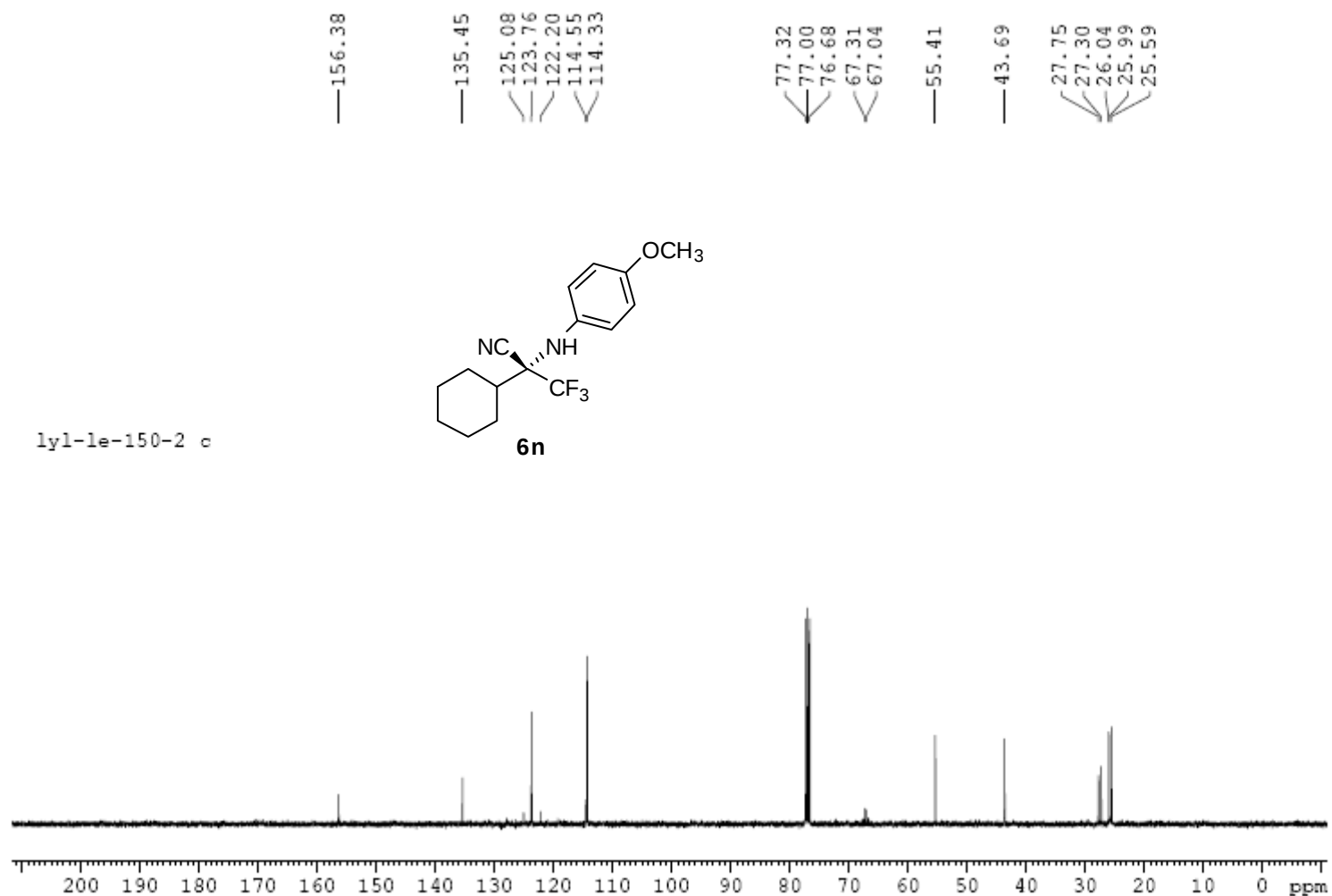


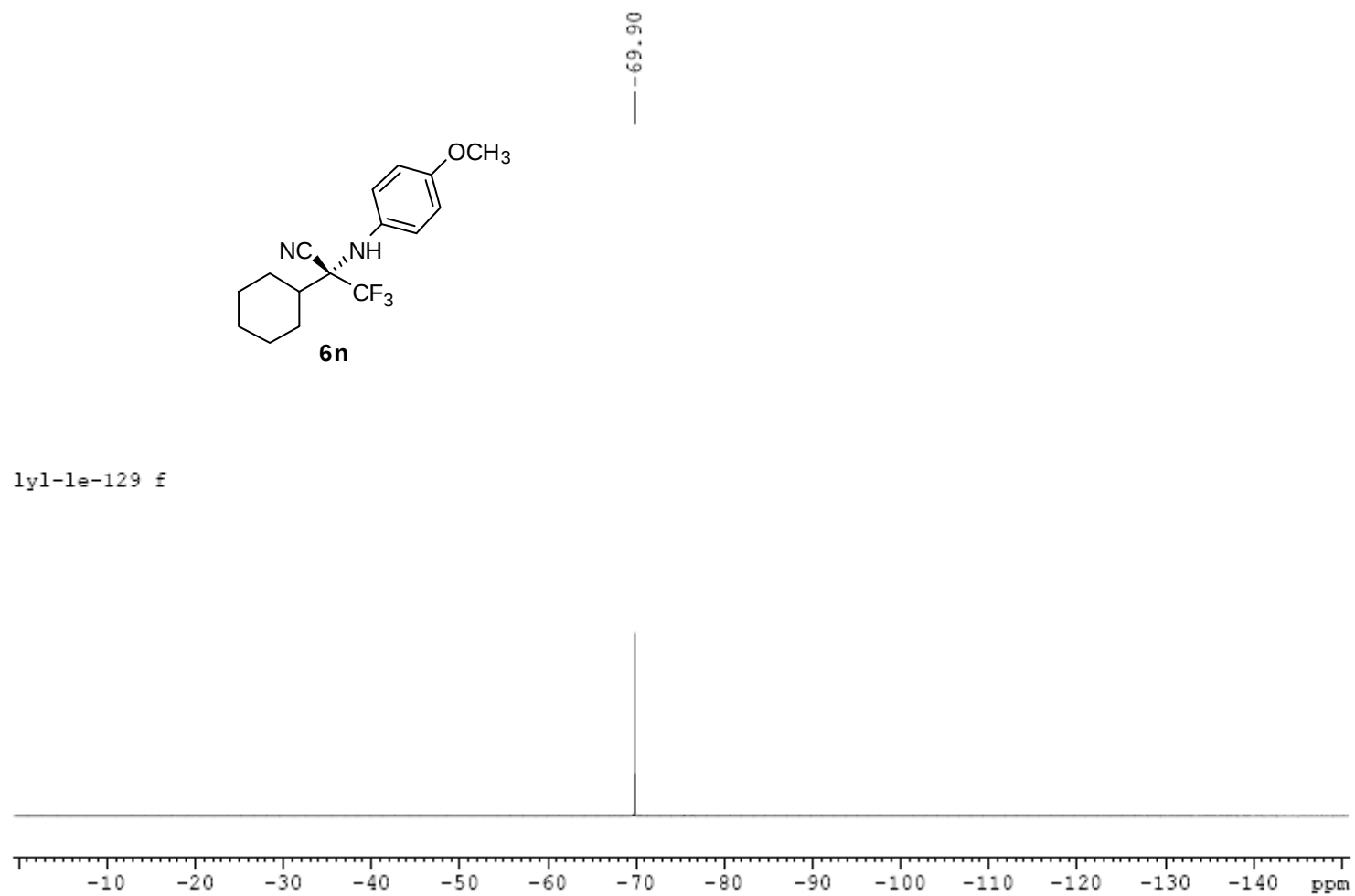
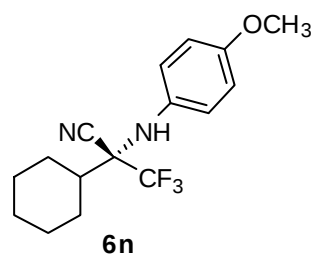


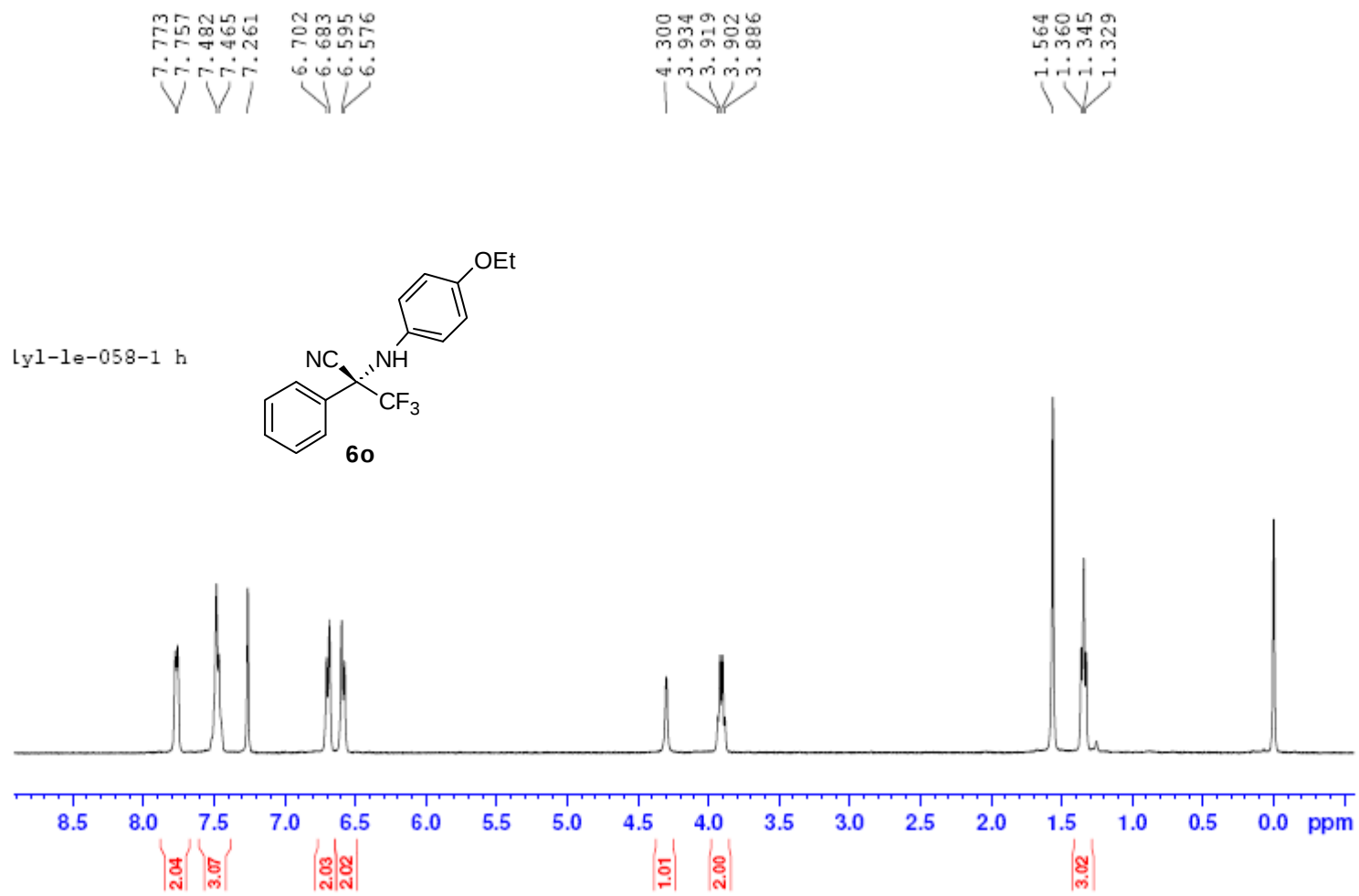


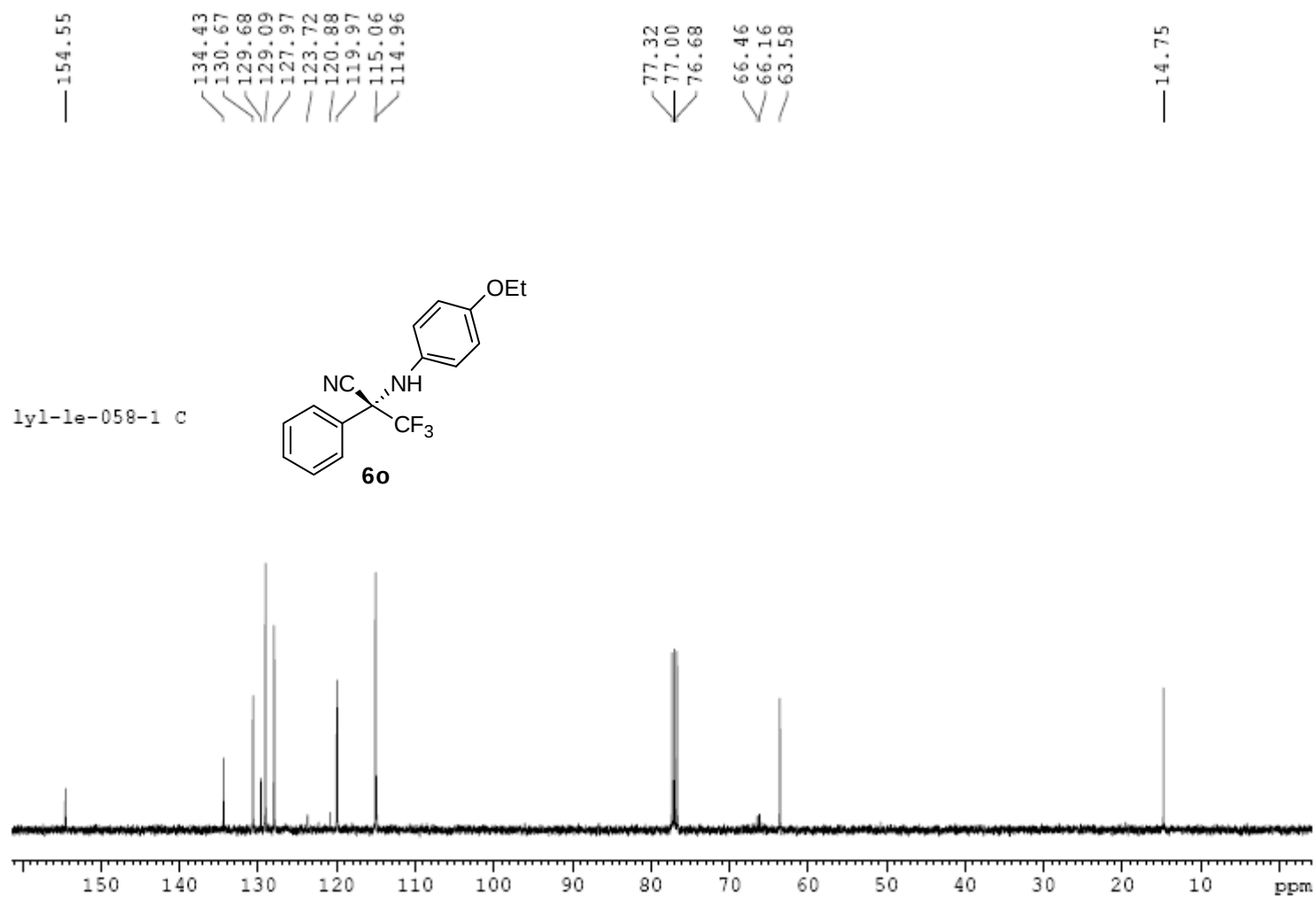


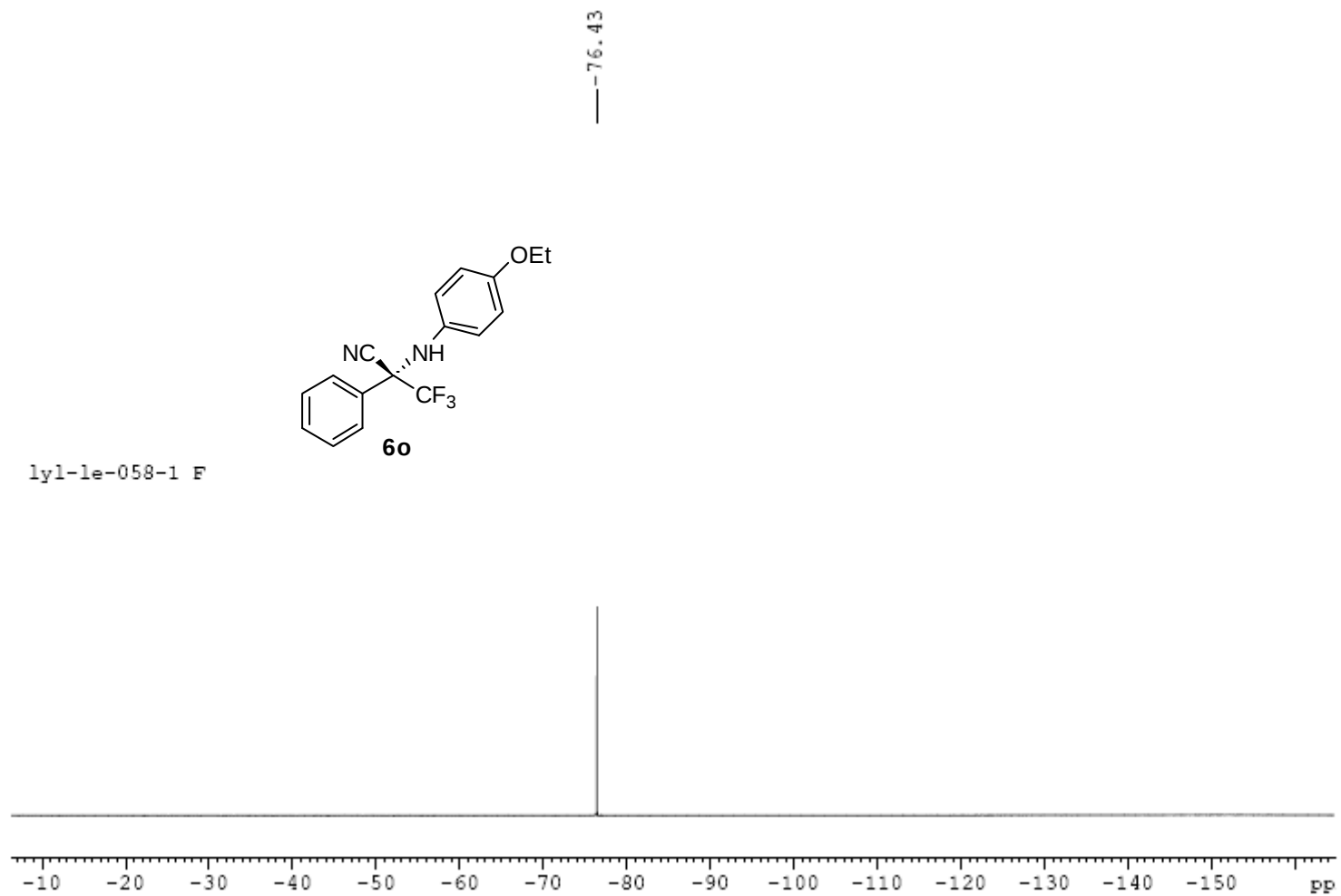


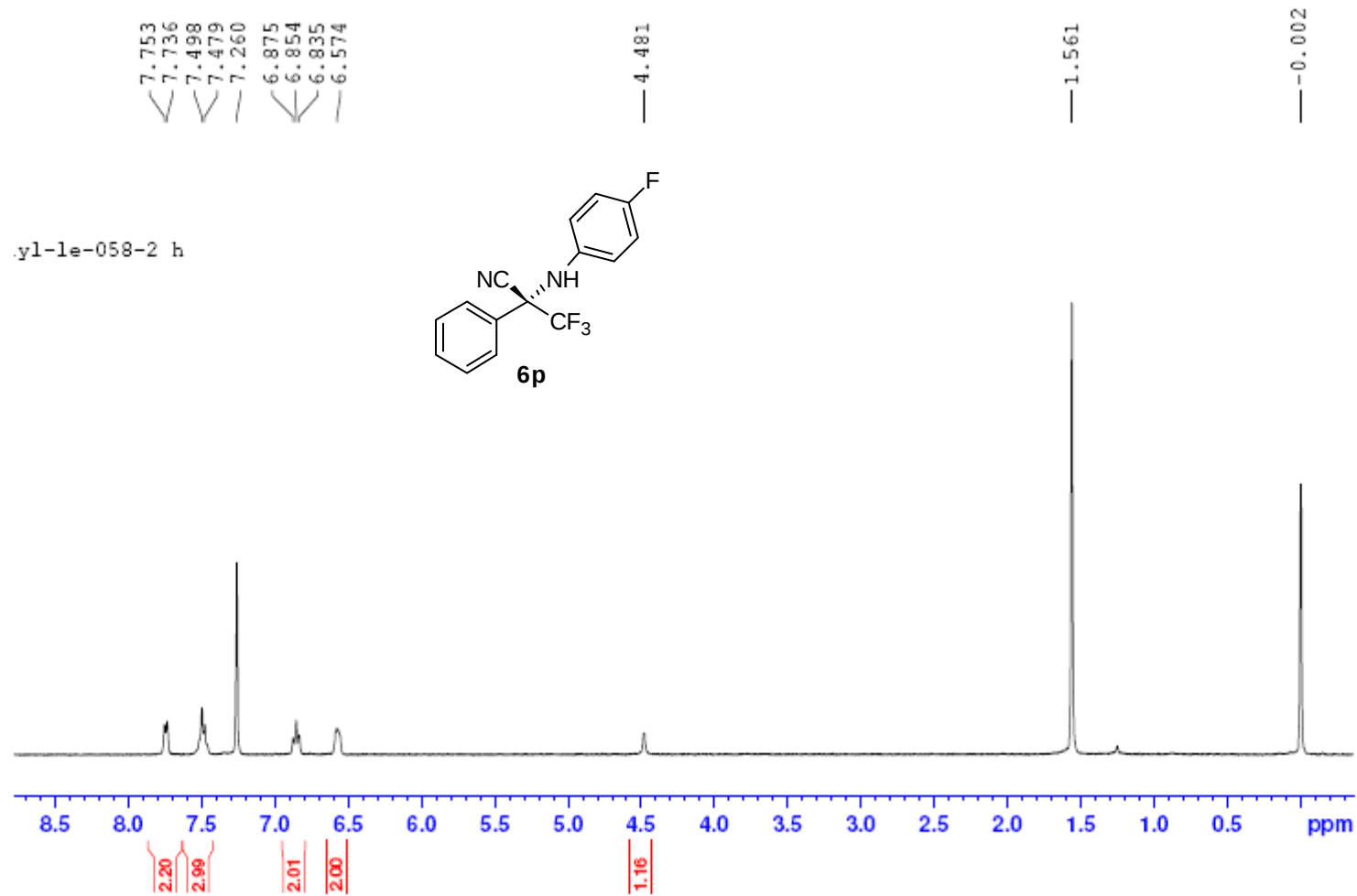










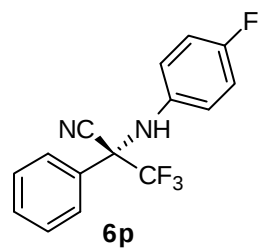


— 159.40
— 157.00

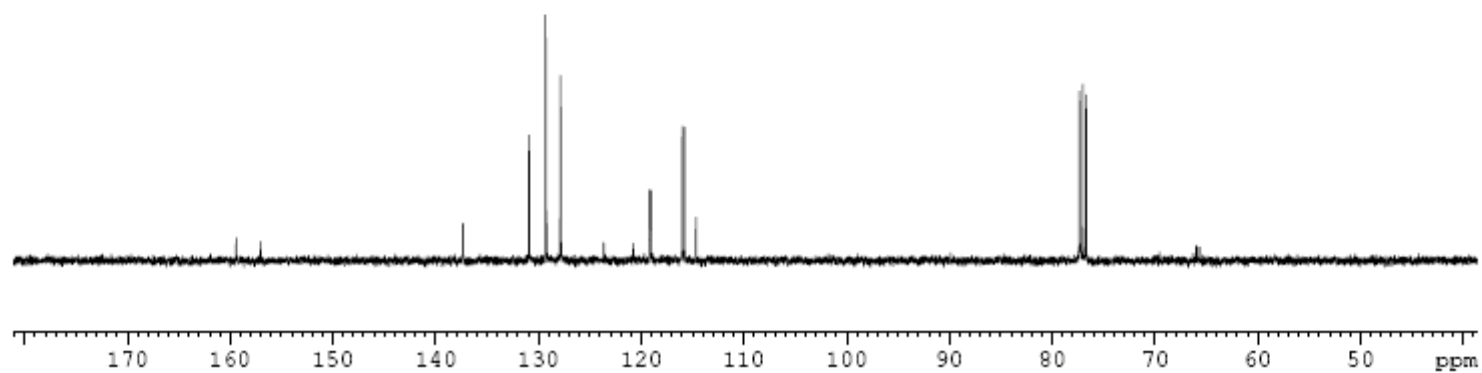
— 137.32
130.89
129.28
129.21
127.84
123.62
120.78
119.14
119.07
115.98
115.76
114.65

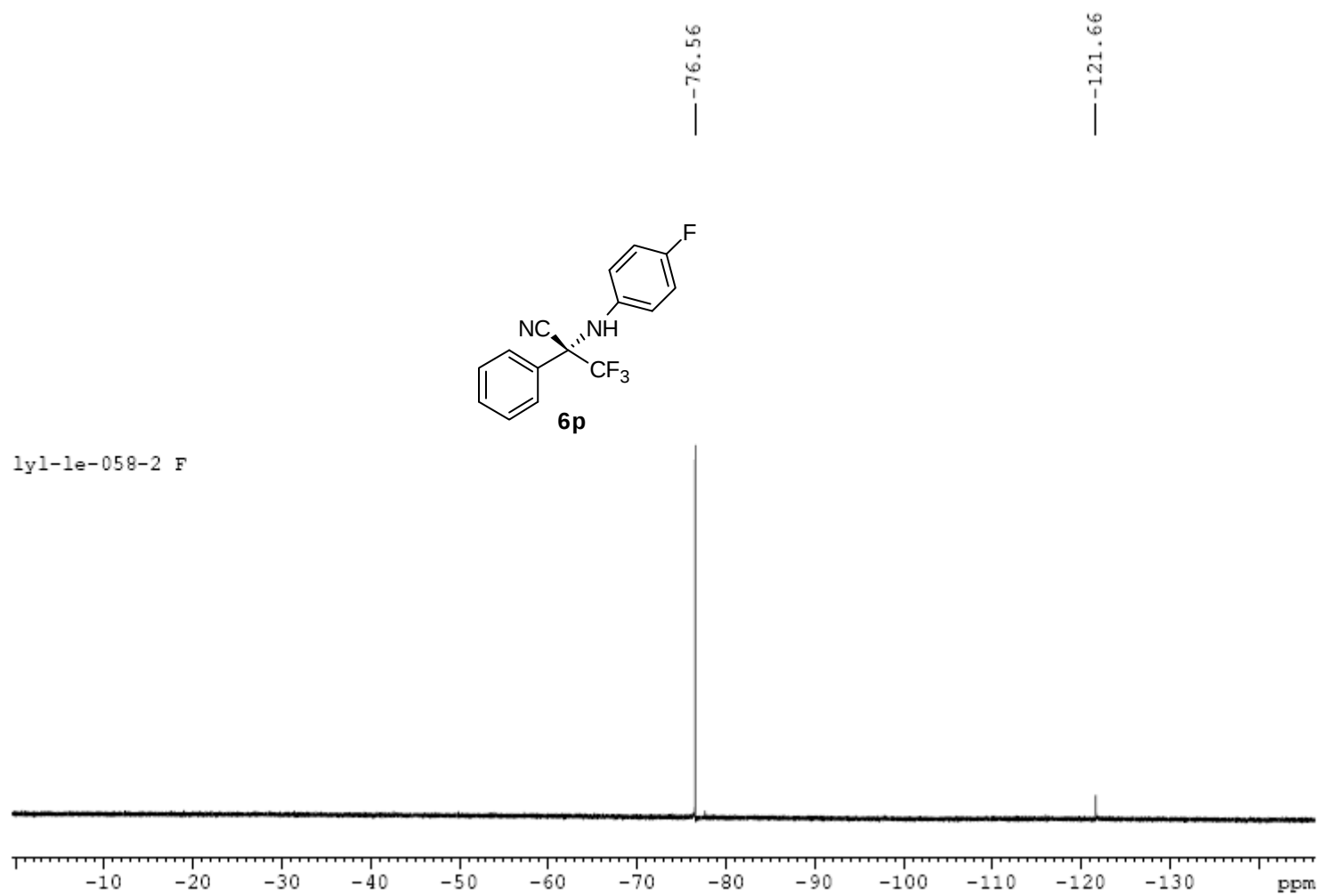
77.32
77.00
76.68

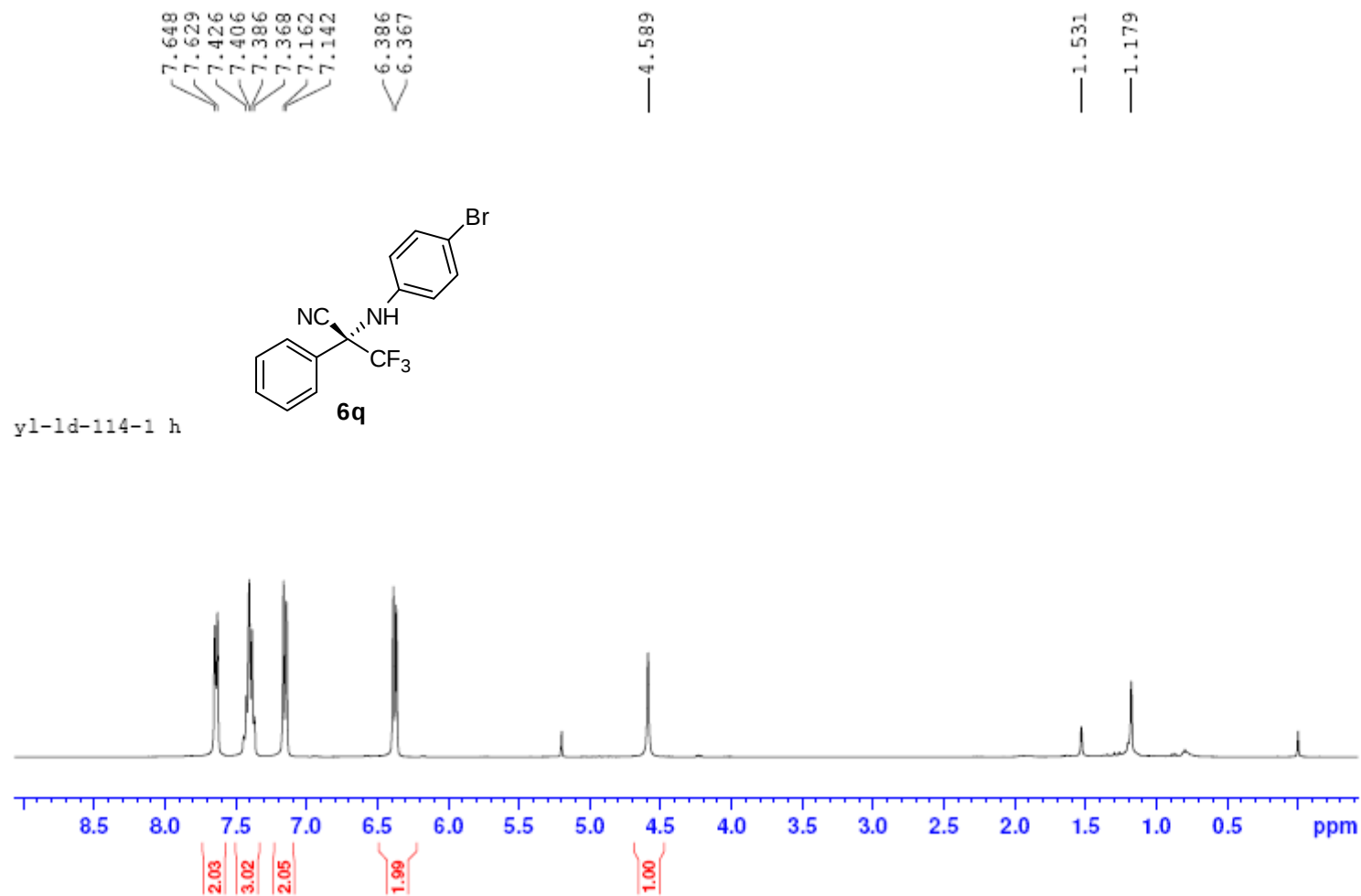
65.95
65.66

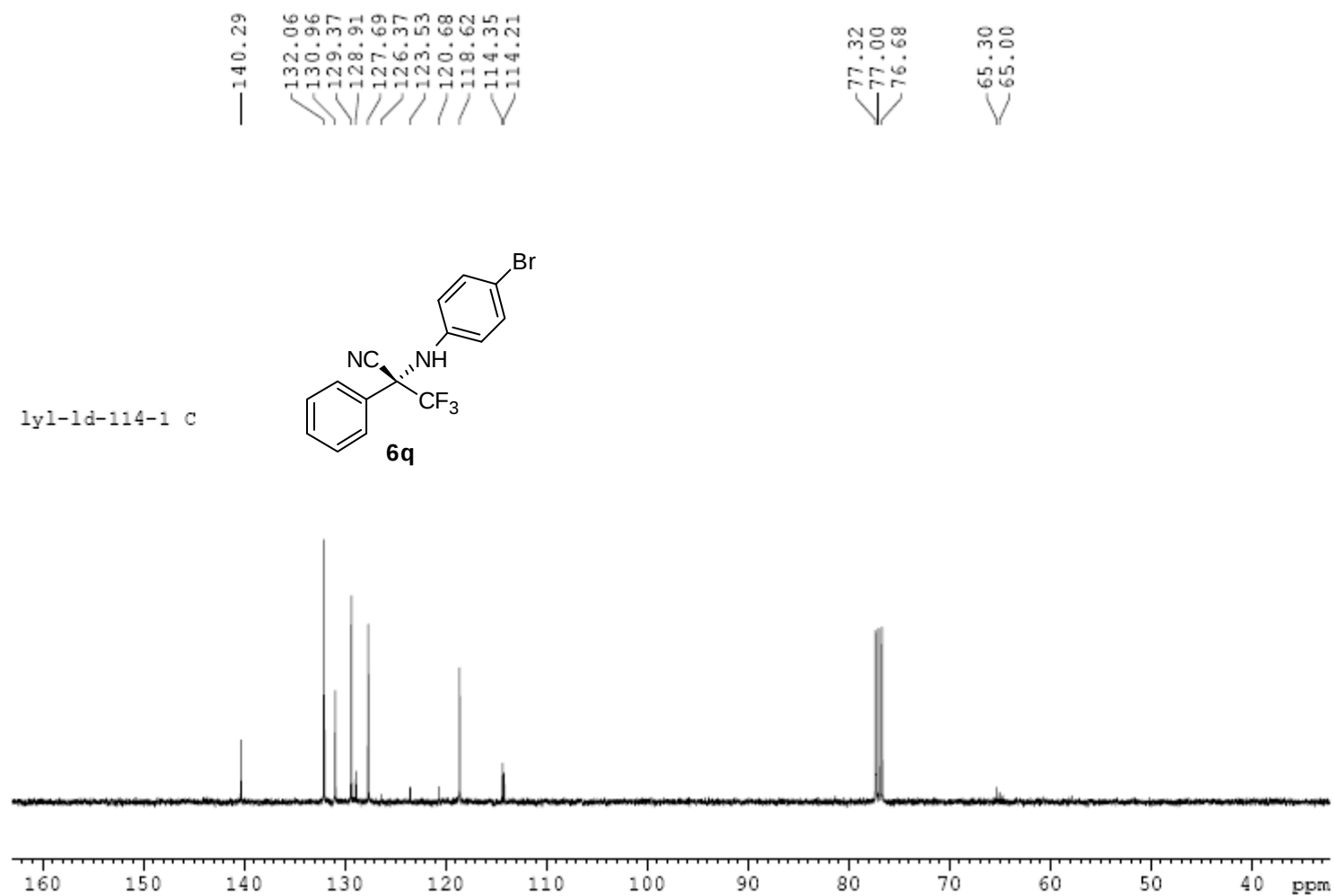


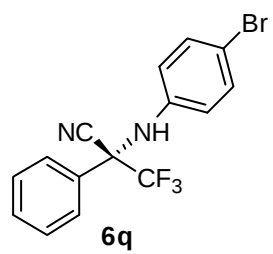
1y1-le-058-1c



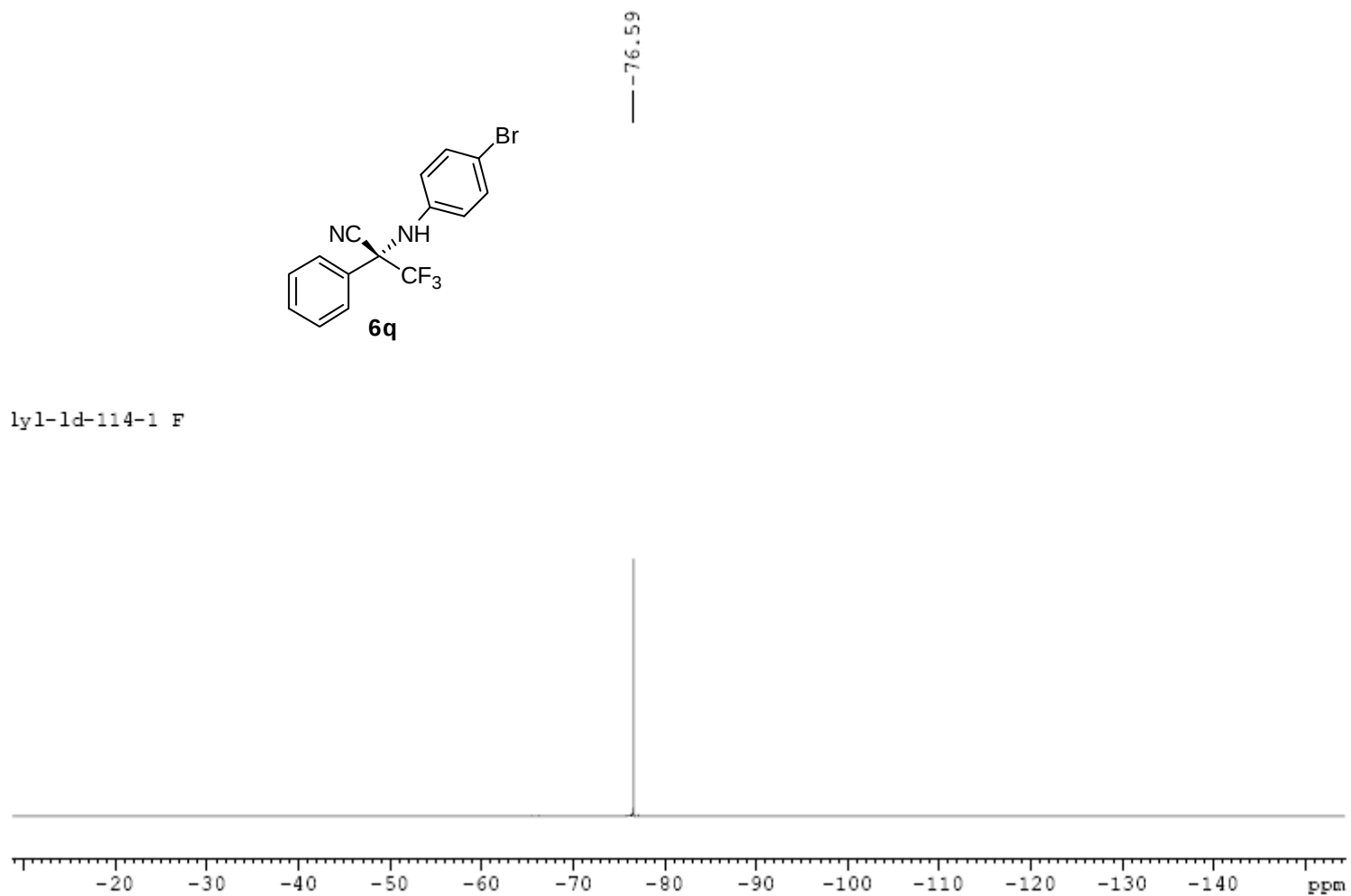


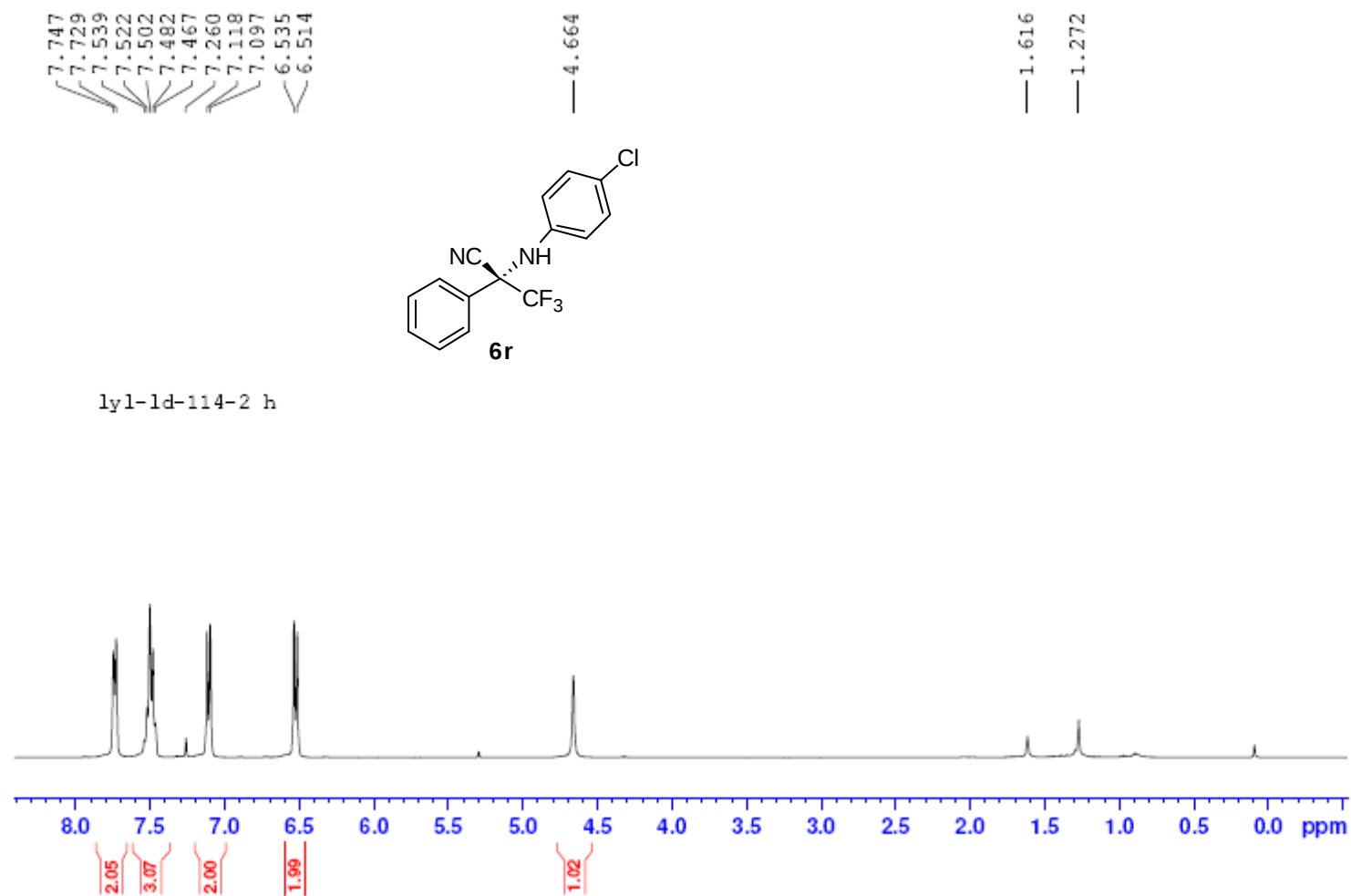






1y1-1d-114-1 F

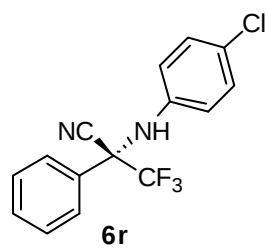




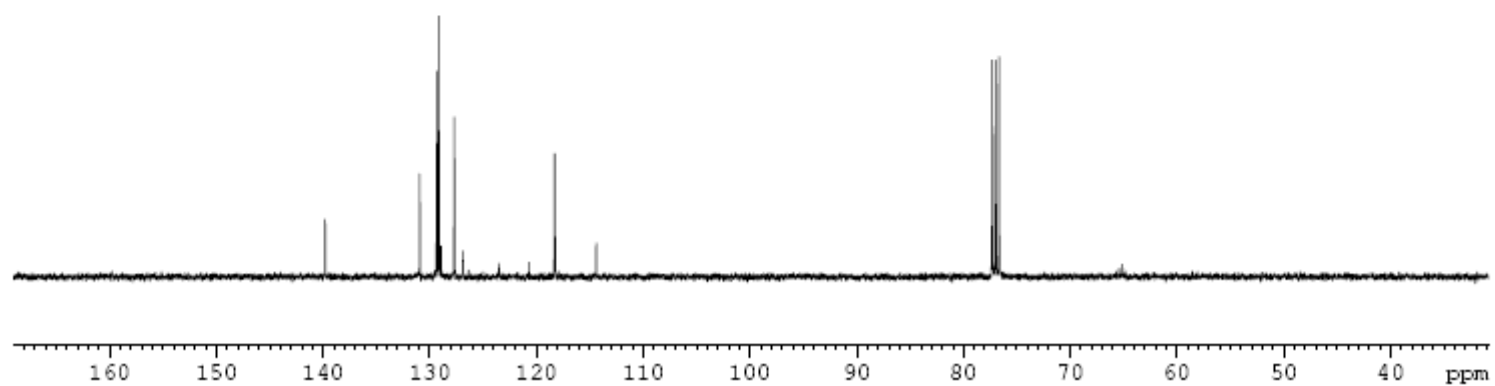
— 139.81
 130.96
 129.36
 129.17
 128.98
 127.72
 126.93
 123.55
 120.71
 118.31
 — 114.42

77.32
 77.00
 76.68

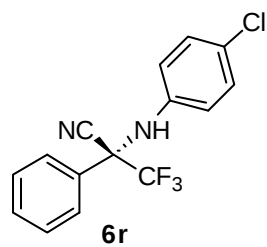
65.42
 65.12



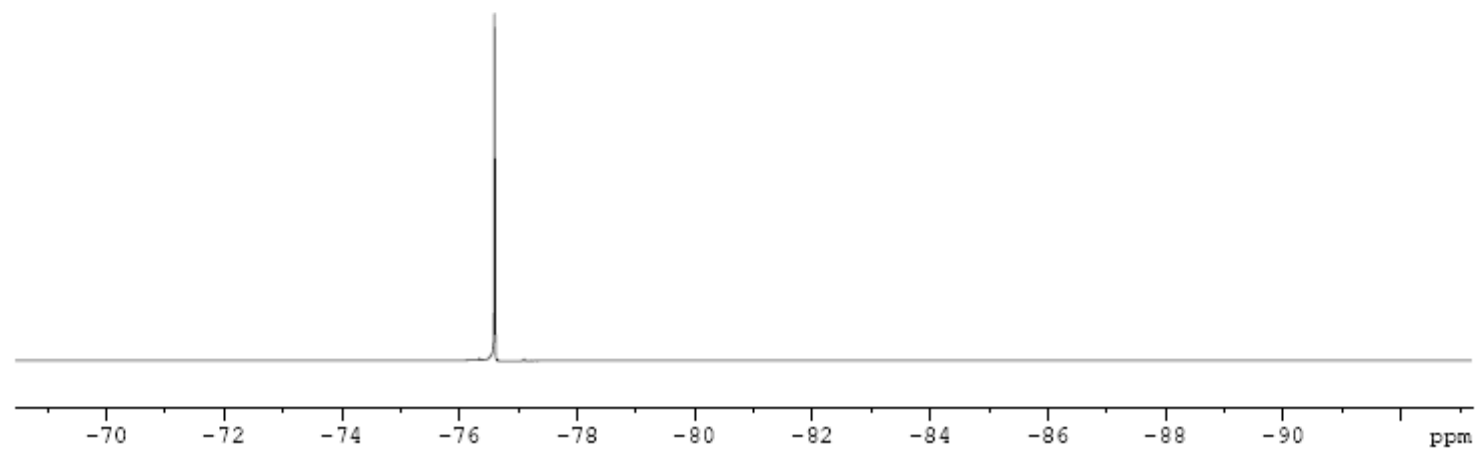
1y1-1d-114-2 c

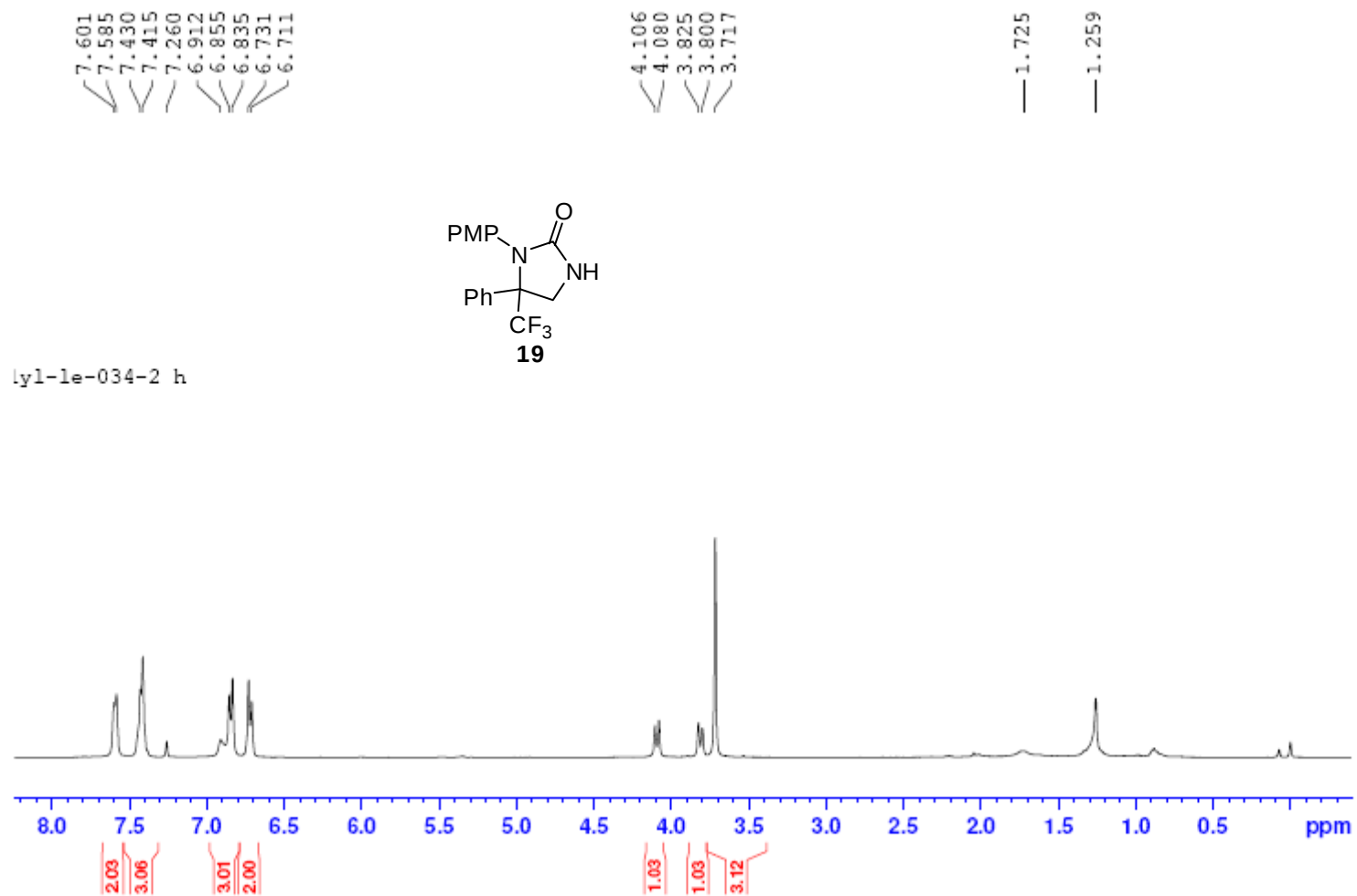


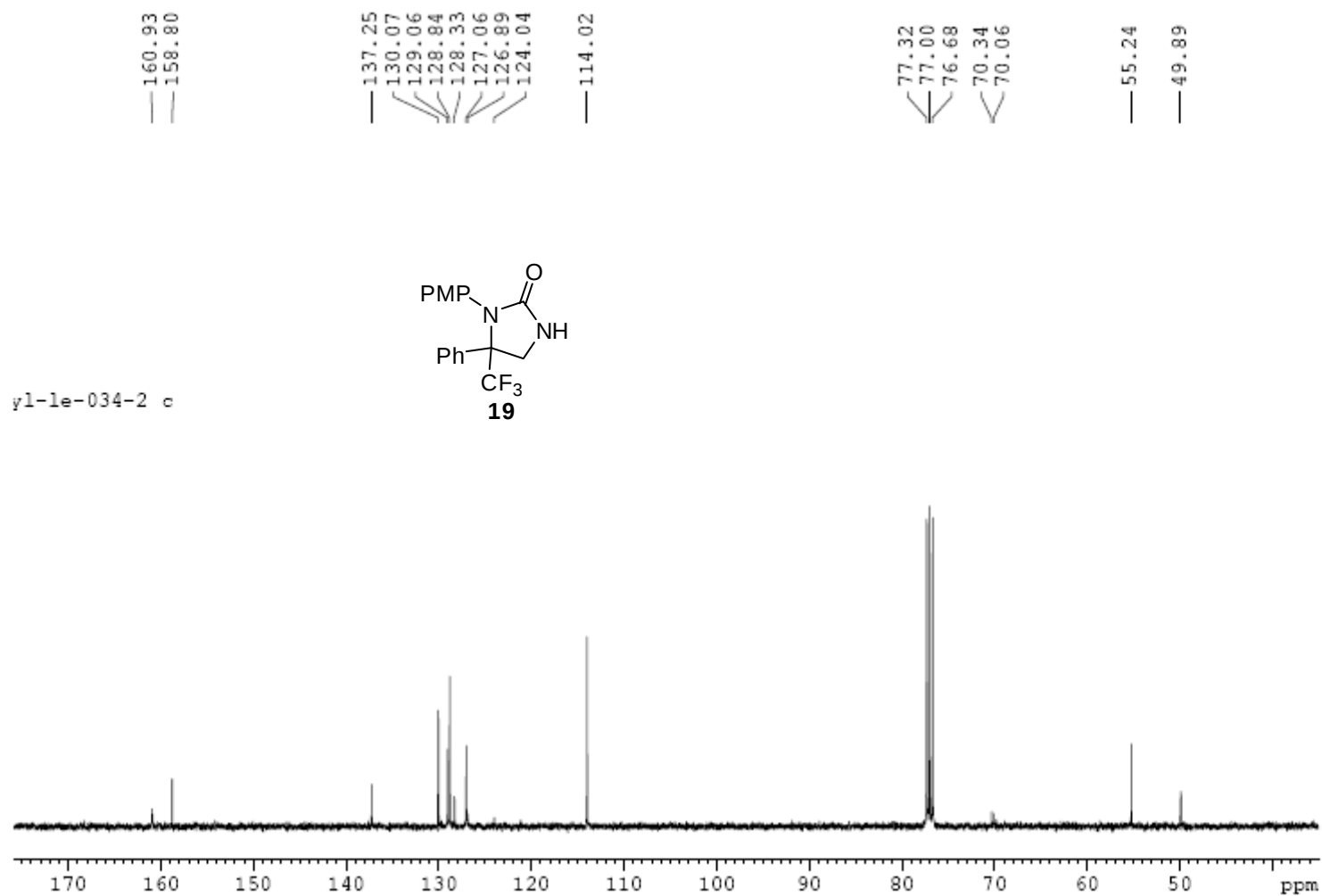
— -76.60

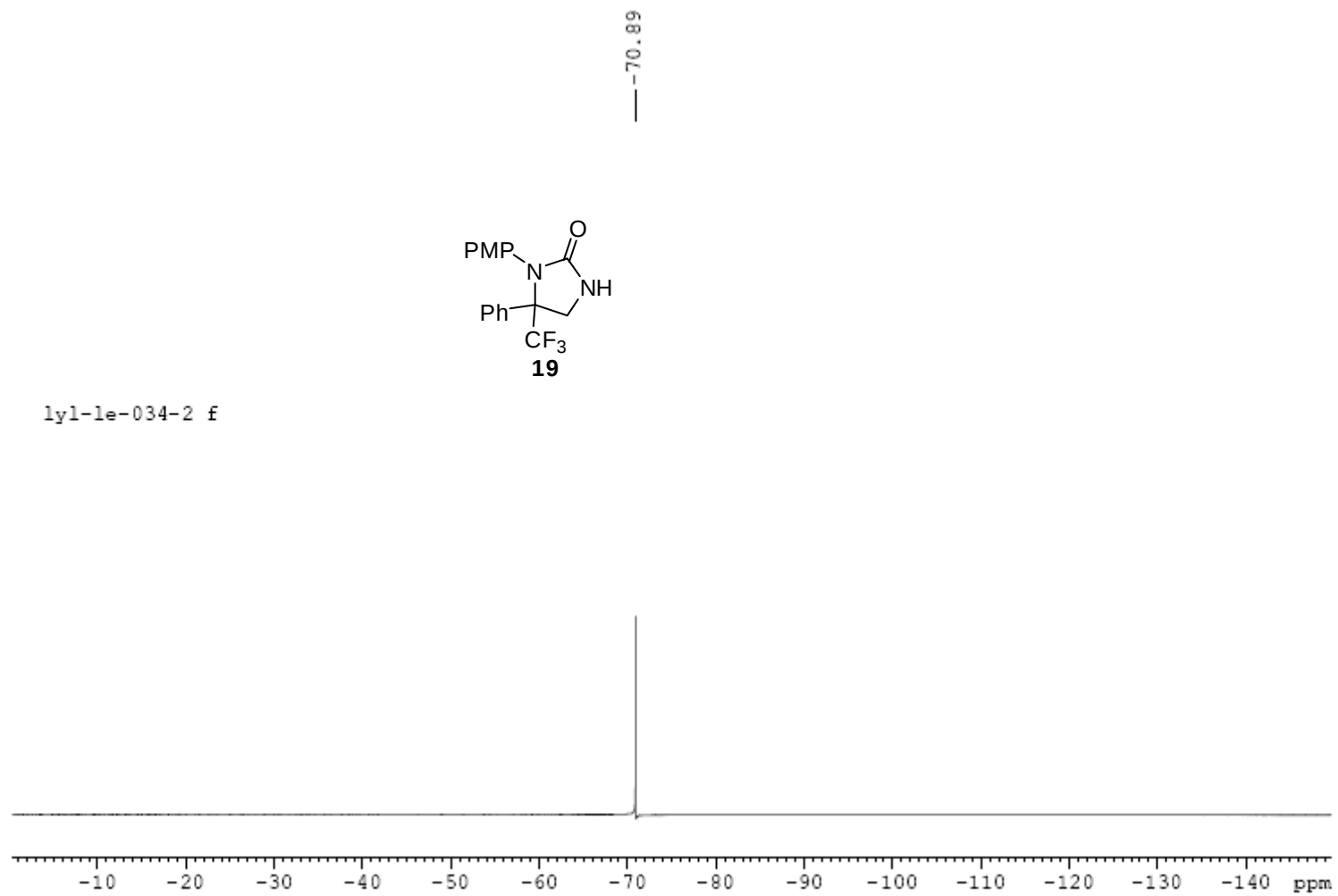


.yl-1d-114-2 F

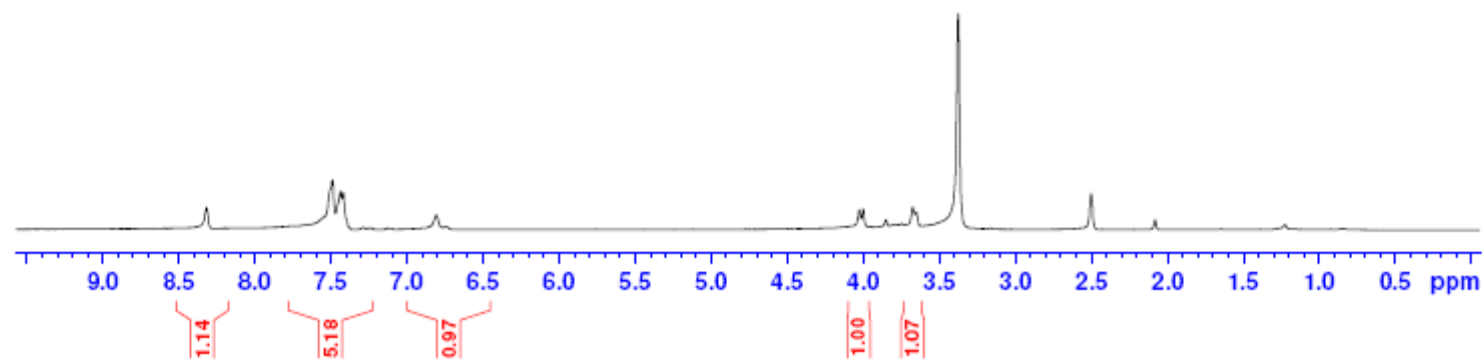
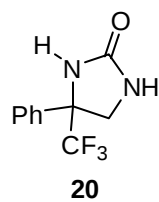








lyl-le-040-3 h



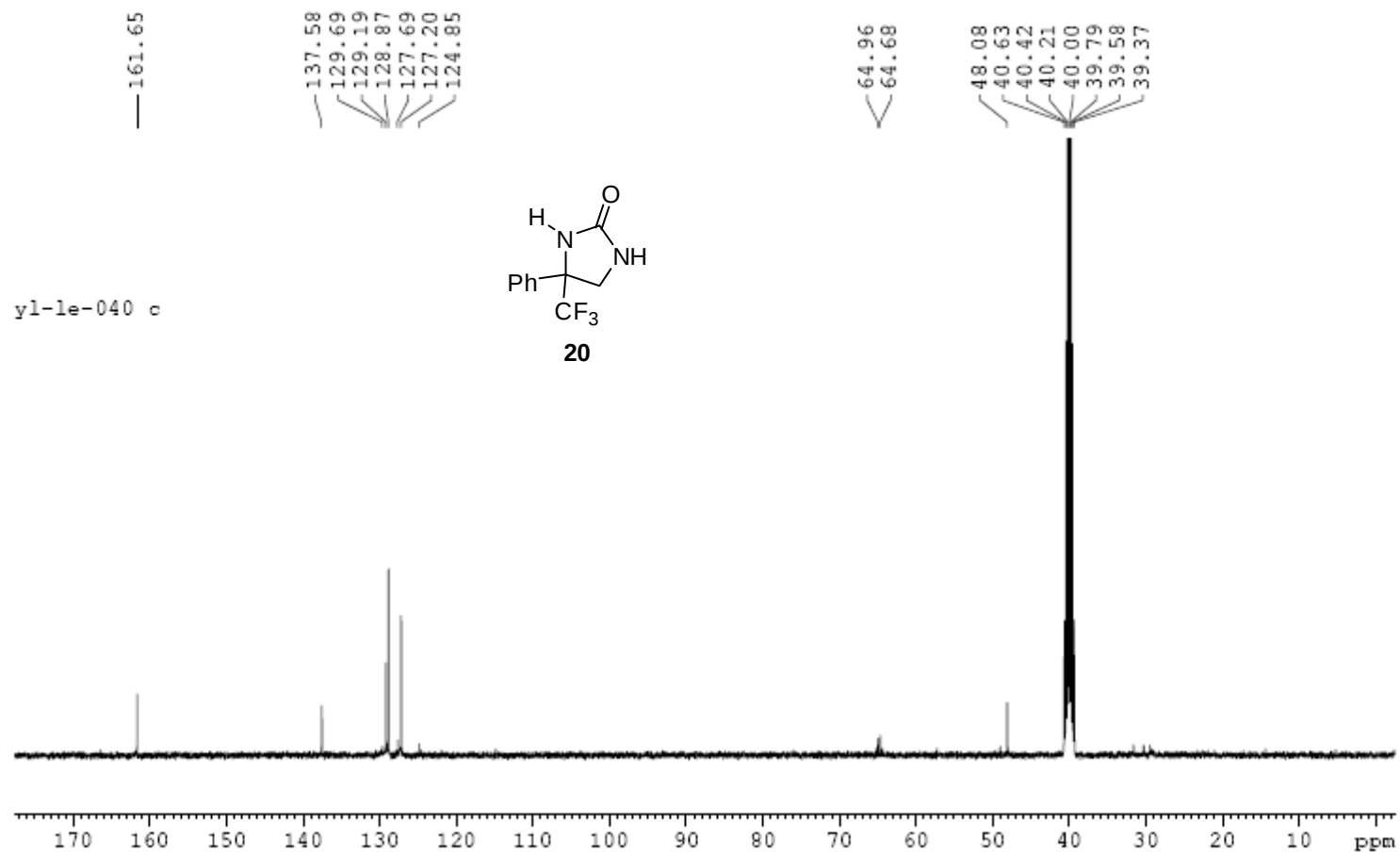
— 8.318

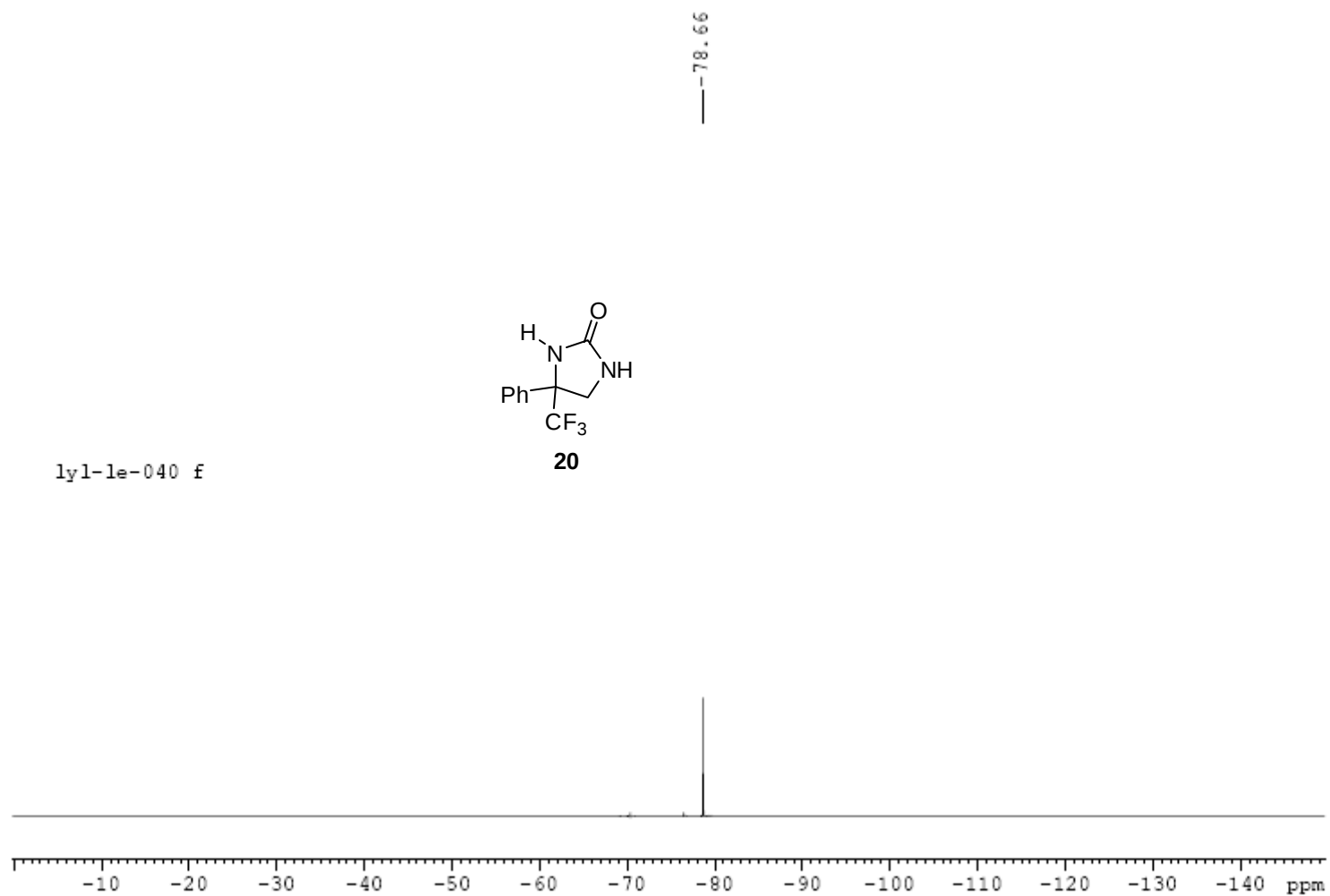
7.490
7.438
7.419

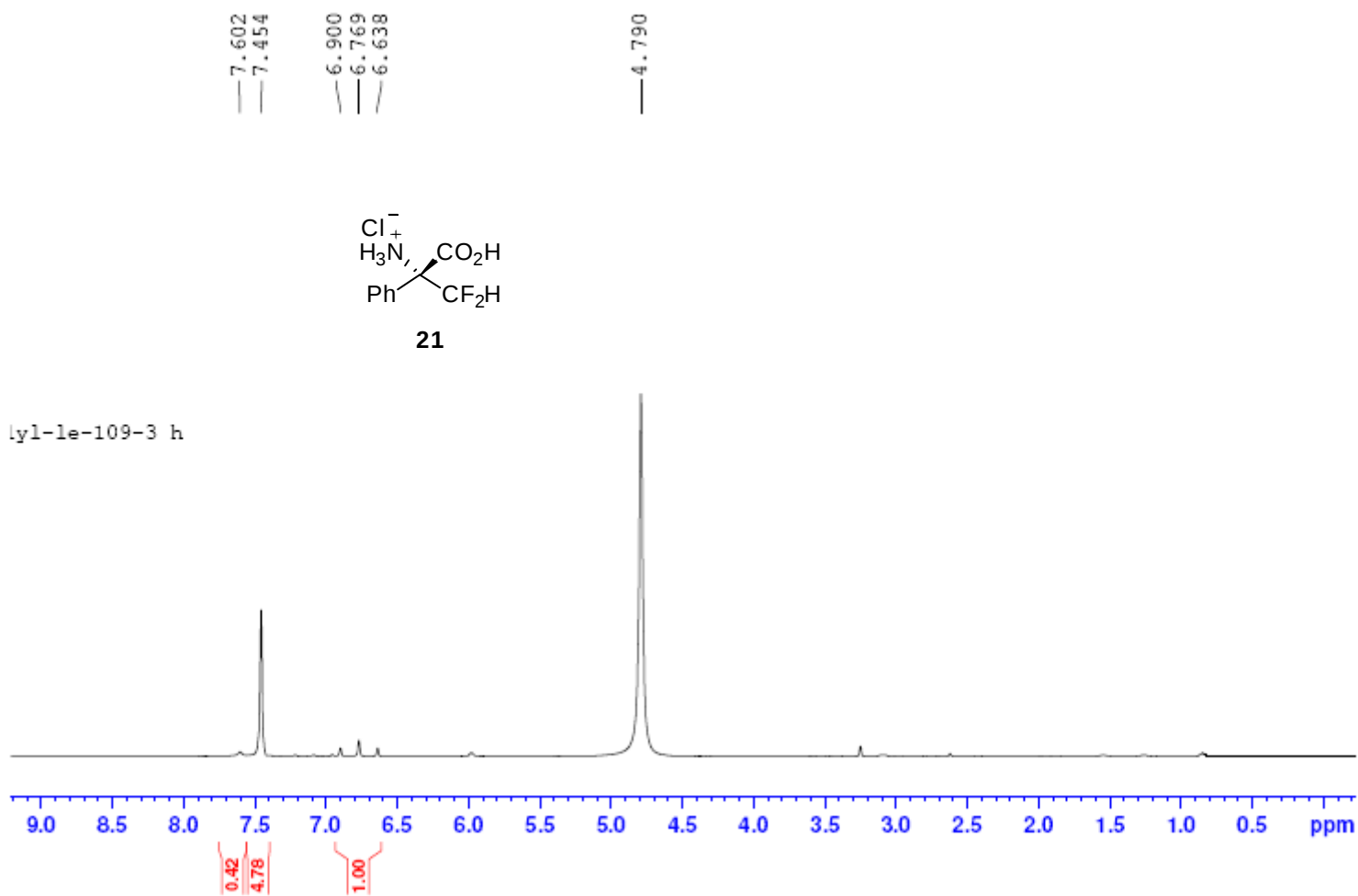
— 6.807

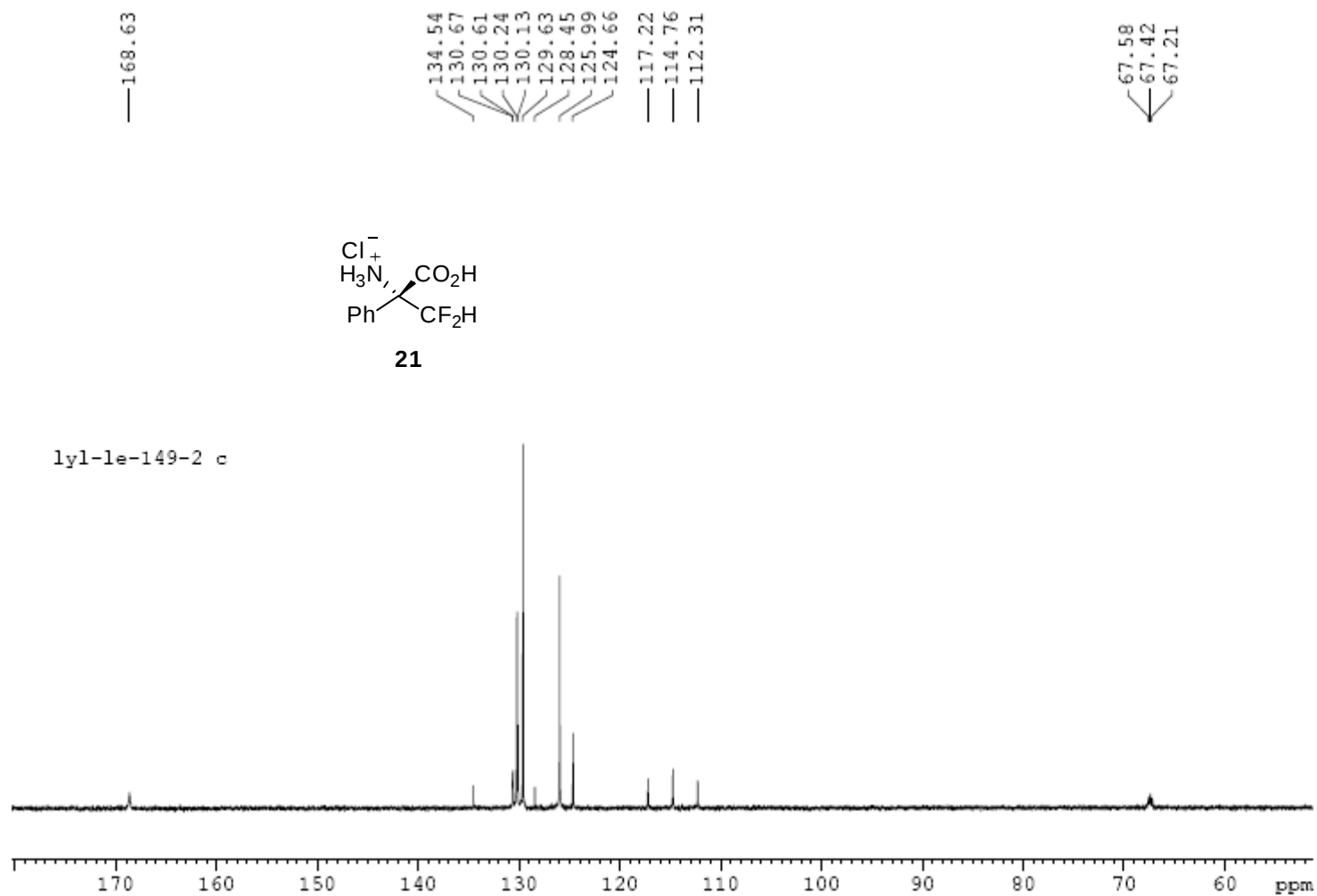
4.025
3.999
3.674
3.649
3.375

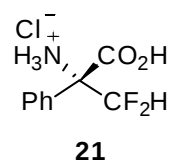
— 2.501











δ 126.16
 δ 126.90
 δ 129.53
 δ 130.28

lyl-le-109-3 f

