

Charge Saturation and Neutral Substitutions in Halomethanes and their Group 14 Analogues

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Supporting Information

Complete Citation for the Gaussian 03 Suite of Programs

Gaussian 03, Revision E.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, and J. A. Pople, Gaussian, Inc., Wallingford CT, 2004.

Table S.1a: Optimized *ab initio* (MP2(full)) and experimental bond distances, a (Å), and bond angles, Θ ($^\circ$), for the $\text{MH}_{4-n}\text{X}_n$ molecules ($\text{M} = \text{C}, \text{Si}, \text{Ge}$; $\text{X} = \text{F}, \text{Cl}, \text{Br}, \text{I}$)[†]

	this work	Expt.		this work	Expt.		this work	Expt.
$\text{CH}_{4-n}\text{F}_n$			$\text{SiH}_{4-n}\text{F}_n$			$\text{GeH}_{4-n}\text{F}_n$		
CH_3F			SiH_3F			GeH_3F		
$a_{\text{C-H}}$	1.090	1.086(2)	$a_{\text{Si-H}}$	1.475	1.47608(19) ^a	$a_{\text{Ge-H}}$	1.530	1.5220(30) ^a
$a_{\text{C-F}}$	1.390	1.383(1)	$a_{\text{Si-F}}$	1.626	1.59450(13) ^a	$a_{\text{Ge-F}}$	1.769	1.7350(10) ^a
$\Theta_{\text{H-C-H}}$	110.4	110.2(3)	$\Theta_{\text{H-Si-H}}$	110.9	110.63	$\Theta_{\text{H-Ge-H}}$	112.8	113.0*
$\Theta_{\text{F-C-H}}$	108.6	108.8(3)	$\Theta_{\text{F-Si-H}}$	108.0	108.269(21)	$\Theta_{\text{F-Ge-H}}$	105.9	105.92(30)
CH_2F_2			SiH_2F_2			GeH_2F_2		
$a_{\text{C-H}}$	1.089	1.084(3)	$a_{\text{Si-H}}$	1.466	1.4652(30)	$a_{\text{Ge-H}}$	1.523	
$a_{\text{C-F}}$	1.360	1.3508(5)	$a_{\text{Si-F}}$	1.608	1.5760(20)	$a_{\text{Ge-F}}$	1.747	
$\Theta_{\text{H-C-H}}$	113.7	112.8(3)	$\Theta_{\text{H-Si-H}}$	115.3	114.2(3)	$\Theta_{\text{H-Ge-H}}$	120.4	
$\Theta_{\text{F-C-H}}$	108.6		$\Theta_{\text{F-Si-H}}$	108.4		$\Theta_{\text{F-Ge-H}}$	107.6	
$\Theta_{\text{F-C-F}}$	108.6	108.49(6)	$\Theta_{\text{F-Si-F}}$	107.9	107.7(2)	$\Theta_{\text{F-Ge-F}}$	105.0	
CHF_3			SiHF_3			GeHF_3		
$a_{\text{C-H}}$	1.088	1.091(14)	$a_{\text{Si-H}}$	1.454	1.4468(5)	$a_{\text{Ge-H}}$	1.514	
$a_{\text{C-F}}$	1.337	1.3284(31)	$a_{\text{Si-F}}$	1.593	1.5624(1)	$a_{\text{Ge-F}}$	1.725	
$\Theta_{\text{F-C-H}}$	110.4		$\Theta_{\text{F-Si-H}}$	110.8	110.64(3)	$\Theta_{\text{F-Ge-H}}$	112.5	
$\Theta_{\text{F-C-F}}$	108.5	108.58(34)	$\Theta_{\text{F-Si-F}}$	108.1		$\Theta_{\text{F-Ge-F}}$	106.3	
CF_4			SiF_4			GeF_4		
$a_{\text{C-F}}$	1.322	1.319253(3) ^a	$a_{\text{Si-F}}$	1.580	1.554(3) ^b	$a_{\text{Ge-F}}$	1.707	
$\text{CH}_{4-n}\text{Cl}_n$			$\text{SiH}_{4-n}\text{Cl}_n$			$\text{GeH}_{4-n}\text{Cl}_n$		
CH_3Cl			SiH_3Cl			GeH_3Cl		
$a_{\text{C-H}}$	1.087	1.0854(5)	$a_{\text{Si-H}}$	1.474	1.47496(11)	$a_{\text{Ge-H}}$	1.529	1.5155(11)
$a_{\text{C-Cl}}$	1.778	1.7760(3)	$a_{\text{Si-Cl}}$	2.059	2.05057(6) ^a	$a_{\text{Ge-Cl}}$	2.163	2.14470(17)
$\Theta_{\text{H-C-H}}$	110.0	110.35(5)	$\Theta_{\text{H-Si-H}}$	110.7		$\Theta_{\text{H-Ge-H}}$	112.1	111.0*
$\Theta_{\text{Cl-C-H}}$	108.9		$\Theta_{\text{Cl-Si-H}}$	108.2	108.295(12)	$\Theta_{\text{Cl-Ge-H}}$	106.7	107.10(14)
CH_2Cl_2			SiH_2Cl_2			GeH_2Cl_2		
$a_{\text{C-H}}$	1.086	1.080(3)	$a_{\text{Si-H}}$	1.468	1.4671(50)	$a_{\text{Ge-H}}$	1.525	1.56(2) ^b
$a_{\text{C-Cl}}$	1.767	1.766(2)	$a_{\text{Si-Cl}}$	2.041	2.0316(30)	$a_{\text{Ge-Cl}}$	2.143	2.130(3) ^b
$\Theta_{\text{H-C-H}}$	110.9	112.10(20)	$\Theta_{\text{H-Si-H}}$	113.3	112.45(50)	$\Theta_{\text{H-Ge-H}}$	116.6	
$\Theta_{\text{Cl-C-H}}$	108.2		$\Theta_{\text{Cl-Si-H}}$	108.4		$\Theta_{\text{Cl-Ge-H}}$	108.0	106.4(15)
$\Theta_{\text{Cl-C-Cl}}$	113.0	111.96(10)	$\Theta_{\text{Cl-Si-Cl}}$	110.1	109.67(30)	$\Theta_{\text{Cl-Ge-Cl}}$	108.0	107.2(5)
CHCl_3			SiHCl_3			GeHCl_3		
$a_{\text{C-H}}$	1.085	1.100(5) ^c	$a_{\text{Si-H}}$	1.462	1.464(10)	$a_{\text{Ge-H}}$	1.522	1.55(2) ^a
$a_{\text{C-Cl}}$	1.765	1.758(2) ^c	$a_{\text{Si-Cl}}$	2.028	2.020(3) ^a	$a_{\text{Ge-Cl}}$	2.126	2.1139(50) ^a
$\Theta_{\text{Cl-C-H}}$	107.7		$\Theta_{\text{Cl-Si-H}}$	109.5	109.5(7)	$\Theta_{\text{Cl-Ge-H}}$	110.5	
$\Theta_{\text{Cl-C-Cl}}$	111.2	111.3(2)	$\Theta_{\text{Cl-Si-Cl}}$	109.5	109.4(3)	$\Theta_{\text{Cl-Ge-Cl}}$	108.4	108.3(10)
CCl_4			SiCl_4			GeCl_4		
$a_{\text{C-Cl}}$	1.771	1.7667(30) ^d	$a_{\text{Si-Cl}}$	2.018	2.0193(34) ^d	$a_{\text{Ge-Cl}}$	2.114	2.113(3) ^d
$\text{CH}_{4-n}\text{Br}_n$			$\text{SiH}_{4-n}\text{Br}_n$			$\text{GeH}_{4-n}\text{Br}_n$		
CH_3Br			SiH_3Br			GeH_3Br		
$a_{\text{C-H}}$	1.086	1.0823(5)	$a_{\text{Si-H}}$	1.474	1.47425(18) ^a	$a_{\text{Ge-H}}$	1.529	1.527(3) ^c
$a_{\text{C-Br}}$	1.936	1.9340(3)	$a_{\text{Si-Br}}$	2.228	2.21227(9) ^a	$a_{\text{Ge-Br}}$	2.317	2.297(2) ^c
$\Theta_{\text{H-C-H}}$	110.6	111.16(5)	$\Theta_{\text{H-Si-H}}$	110.7		$\Theta_{\text{H-Ge-H}}$	111.9	
$\Theta_{\text{Br-C-H}}$	108.3		$\Theta_{\text{Br-Si-H}}$	108.2	108.161(20)	$\Theta_{\text{Br-Ge-H}}$	106.9	106.3(2)

CH₂Br₂			SiH₂Br₂			GeH₂Br₂		
<i>a</i> _{C-H}	1.085	1.097(5) ^c	<i>a</i> _{Si-H}	1.469		<i>a</i> _{Ge-H}	1.526	1.52(4)
<i>a</i> _{C-Br}	1.929	1.925(2) ^c	<i>a</i> _{Si-Br}	2.211		<i>a</i> _{Ge-Br}	2.299	2.277(3)
Θ _{H-C-H}	111.5	110.9(8)	Θ _{H-Si-H}	112.8		Θ _{H-Ge-H}	115.6	
Θ _{Br-C-H}	107.9		Θ _{Br-Si-H}	108.3		Θ _{Br-Ge-H}	108.0	109.0(21)
Θ _{Br-C-Br}	113.8	112.9(2)	Θ _{Br-Si-Br}	110.9		Θ _{Br-Ge-Br}	109.2	108.4(4)
CHBr₃			SiHBr₃			GeHBr₃		
<i>a</i> _{C-H}	1.085	1.11(5) ^d	<i>a</i> _{Si-H}	1.465	1.494(10) ^a	<i>a</i> _{Ge-H}	1.525	
<i>a</i> _{C-Br}	1.931	1.924(5) ^d	<i>a</i> _{Si-Br}	2.201	2.170(2) ^a	<i>a</i> _{Ge-Br}	2.286	
Θ _{Br-C-H}	107.1	107.2(4)	Θ _{Br-Si-H}	109.0		Θ _{Br-Ge-H}	109.8	
Θ _{Br-C-Br}	111.7	111.7(4)	Θ _{Br-Si-Br}	110.0	111.36(50)	Θ _{Br-Ge-Br}	109.1	
CBr₄			SiBr₄			GeBr₄		
<i>a</i> _{C-Br}	1.943	1.935*	<i>a</i> _{Si-Br}	2.196	2.183(4) ^d	<i>a</i> _{Ge-Br}	2.279	2.272 ^d
CH_{4-n}I_n			SiH_{4-n}I_n			GeH_{4-n}I_n		
CH₃I			SiH₃I			GeH₃I		
<i>a</i> _{C-H}	1.086	1.084*	<i>a</i> _{Si-H}	1.475	1.4741(14) ^a	<i>a</i> _{Ge-H}	1.530	1.5192(30) ^a
<i>a</i> _{C-I}	2.133	2.132*	<i>a</i> _{Si-I}	2.448	2.43835(59) ^a	<i>a</i> _{Ge-I}	2.521	2.5095(10) ^a
Θ _{H-C-H}	111.1	111.2*	Θ _{H-Si-H}	110.6		Θ _{H-Ge-H}	111.5	
Θ _{I-C-H}	107.8		Θ _{I-Si-H}	108.3	108.16(17) ^a	Θ _{I-Ge-H}	107.4	107.00(30) ^a
CH₂I₂			SiH₂I₂			GeH₂I₂		
<i>a</i> _{C-H}	1.086		<i>a</i> _{Si-H}	1.473		<i>a</i> _{Ge-H}	1.529	
<i>a</i> _{C-I}	2.129		<i>a</i> _{Si-I}	2.435		<i>a</i> _{Ge-I}	2.507	
Θ _{H-C-H}	111.6		Θ _{H-Si-H}	112.0		Θ _{H-Ge-H}	114.1	
Θ _{I-C-H}	107.7		Θ _{I-Si-H}	108.5		Θ _{I-Ge-H}	108.3	
Θ _{I-C-I}	114.3		Θ _{I-Si-I}	110.8		Θ _{I-Ge-I}	109.8	
CHI₃			SiHI₃			GeHI₃		
<i>a</i> _{C-H}	1.088		<i>a</i> _{Si-H}	1.472		<i>a</i> _{Ge-H}	1.531	
<i>a</i> _{C-I}	2.137		<i>a</i> _{Si-I}	2.428		<i>a</i> _{Ge-I}	2.499	
Θ _{I-C-H}	106.6		Θ _{I-Si-H}	108.7		Θ _{I-Ge-H}	109.3	
Θ _{I-C-I}	112.2		Θ _{I-Si-I}	110.2		Θ _{I-Ge-I}	109.7	
CI₄			SiI₄			GeI₄		
<i>a</i> _{C-I}	2.157	2.15*	<i>a</i> _{Si-I}	2.428		<i>a</i> _{Ge-I}	2.499	2.499

Experimental data are from ref 29. The tetrahalides are of T_d symmetry (Θ_{X-M-X} = 109.5).

* Θ_{H-Ge-H} for GeH₃Cl and *a*_{C-Br} has been taken from D. R. Lide (Ed) CRC Handbook of Chemistry and Physics 88th ed., CRC Press Boca Raton, FL, 2007 (Section 9).

† Unless otherwise stated the bond lengths refer to the distance between the nuclear positions at the potential-energy minimum, *a_e*. Alternative distance parameters are defined as follows:

^a*a₀*; the distance between effective nuclear positions derived from rotational constants in the vibrational ground state.

^b*a_a*; the constant argument in the molecular intensity function of electron diffraction, ED, studies.

^c*a_s*; the distance between effective nuclear positions derived from isotopic differences in rotational constants.

^d*a_g*; the thermal average of the internuclear distance. It is derived from ED and is specific to the temperature of the ED experiment.

Table S.1b: Optimized *ab initio* (MP2(full)) and experimental bond distances, a (Å), and bond angles, Θ (°), for the $MH_{4-n}X_n$ molecules ($M = \text{Sn, Pb}$; $X = \text{F, Cl, Br, I}$).

	MP2	Expt.		MP2	Expt.
SnH_{4-n}F_n			PbH_{4-n}F_n		
SnH ₃ F			PbH ₃ F		
$a_{\text{Sn-H}}$	1.701		$a_{\text{Pb-H}}$	1.741	
$a_{\text{Sn-F}}$	1.948		$a_{\text{Pb-F}}$	2.056	
$\Theta_{\text{H-Sn-H}}$	113.8		$\Theta_{\text{H-Pb-H}}$	116.1	
$\Theta_{\text{F-Sn-H}}$	104.7		$\Theta_{\text{F-Pb-H}}$	101.5	
SnH ₂ F ₂			PbH ₂ F ₂		
$a_{\text{Sn-H}}$	1.693		$a_{\text{Pb-H}}$	1.723	
$a_{\text{Sn-F}}$	1.929		$a_{\text{Pb-F}}$	2.034	
$\Theta_{\text{H-Sn-H}}$	123.2		$\Theta_{\text{H-Pb-H}}$	136.5	
$\Theta_{\text{F-Sn-H}}$	107.1		$\Theta_{\text{F-Pb-H}}$	103.6	
$\Theta_{\text{F-Sn-F}}$	103.5		$\Theta_{\text{F-Pb-F}}$	101.5	
SnHF ₃			PbHF ₃		
$a_{\text{Sn-H}}$	1.683		$a_{\text{Pb-H}}$	1.718	
$a_{\text{Sn-F}}$	1.909		$a_{\text{Pb-F}}$	2.007	
$\Theta_{\text{F-Sn-H}}$	113.6		$\Theta_{\text{F-Pb-H}}$	116.0	
$\Theta_{\text{F-Sn-F}}$	105.1		$\Theta_{\text{F-Pb-F}}$	102.2	
SnF ₄			PbF ₄		
$a_{\text{Sn-F}}$	1.894		$a_{\text{Pb-F}}$	1.993	
SnH_{4-n}Cl_n			PbH_{4-n}Cl_n		
SnH ₃ Cl			PbH ₃ Cl		
$a_{\text{Sn-H}}$	1.703	1.696 (5) ^a	$a_{\text{Pb-H}}$	1.742	
$a_{\text{Sn-Cl}}$	2.323	2.328 (5) ^a	$a_{\text{Pb-Cl}}$	2.415	
$\Theta_{\text{H-Sn-H}}$	112.5		$\Theta_{\text{H-Pb-H}}$	114.6	
$\Theta_{\text{Cl-Sn-H}}$	106.2	105.5 (5) ^a	$\Theta_{\text{Cl-Pb-H}}$	103.7	
SnH ₂ Cl ₂			PbH ₂ Cl ₂		
$a_{\text{Sn-H}}$	1.698		$a_{\text{Pb-H}}$	1.736	
$a_{\text{Sn-Cl}}$	2.305		$a_{\text{Pb-Cl}}$	2.395	
$\Theta_{\text{H-Sn-H}}$	117.6		$\Theta_{\text{H-Pb-H}}$	125.4	
$\Theta_{\text{Cl-Sn-H}}$	108.1		$\Theta_{\text{Cl-Pb-H}}$	106.4	
$\Theta_{\text{Cl-Sn-Cl}}$	106.1		$\Theta_{\text{Cl-Pb-Cl}}$	104.1	
SnHCl ₃			PbHCl ₃		
$a_{\text{Sn-H}}$	1.694		$a_{\text{Pb-H}}$	1.738	
$a_{\text{Sn-Cl}}$	2.289		$a_{\text{Pb-Cl}}$	2.373	
$\Theta_{\text{Cl-Sn-H}}$	111.6		$\Theta_{\text{Cl-Pb-H}}$	113.0	
$\Theta_{\text{Cl-Sn-Cl}}$	107.2		$\Theta_{\text{Cl-Pb-Cl}}$	105.7	
SnCl ₄			PbCl ₄		
$a_{\text{Sn-Cl}}$	2.276	2.2808 (37) ^b	$a_{\text{Pb-Cl}}$	2.360	2.369(2) ^c
SnH_{4-n}Br_n			PbH_{4-n}Br_n		
SnH ₃ Br			PbH ₃ Br		
$a_{\text{Sn-H}}$	1.702		$a_{\text{Pb-H}}$	1.743	
$a_{\text{Sn-Br}}$	2.474	2.4691 (30) ^a	$a_{\text{Pb-Br}}$	2.560	
$\Theta_{\text{H-Sn-H}}$	112.3		$\Theta_{\text{H-Pb-H}}$	114.2	
$\Theta_{\text{Br-Sn-H}}$	106.4	105.9 (20) ^a	$\Theta_{\text{Br-Pb-H}}$	104.2	
SnH ₂ Br ₂			PbH ₂ Br ₂		
$a_{\text{Sn-H}}$	1.698		$a_{\text{Pb-H}}$	1.739	
$a_{\text{Sn-Br}}$	2.457		$a_{\text{Pb-Br}}$	2.540	
$\Theta_{\text{H-Sn-H}}$	116.8		$\Theta_{\text{H-Pb-H}}$	123.3	
$\Theta_{\text{Br-Sn-H}}$	108.1		$\Theta_{\text{Br-Pb-H}}$	106.7	
$\Theta_{\text{Br-Sn-Br}}$	107.2		$\Theta_{\text{Br-Pb-Br}}$	105.5	

SnHBr₃			PbHBr₃		
$a_{\text{Sn-H}}$	1.698		$a_{\text{Pb-H}}$	1.744	
$a_{\text{Sn-Br}}$	2.444		$a_{\text{Pb-Br}}$	2.523	
$\Theta_{\text{Br-Sn-H}}$	110.9		$\Theta_{\text{Br-Pb-H}}$	112.1	
$\Theta_{\text{Br-Sn-Br}}$	108.0		$\Theta_{\text{Br-Pb-Br}}$	106.7	
SnBr₄			PbBr₄		
$a_{\text{Sn-Br}}$	2.434		$a_{\text{Pb-Br}}$	2.515	
SnH_{4-n}I_n			PbH_{4-n}I_n		
SnH₃I			PbH₃I		
$a_{\text{Sn-H}}$	1.702	1.701 (5) ^a	$a_{\text{Pb-H}}$	1.745	
$a_{\text{Sn-I}}$	2.676	2.6746 (50) ^a	$a_{\text{Pb-I}}$	2.752	
$\Theta_{\text{H-Sn-H}}$	112.0		$\Theta_{\text{H-Pb-H}}$	113.6	
$\Theta_{\text{I-Sn-H}}$	106.8	107.1 (5) ^a	$\Theta_{\text{I-Pb-H}}$	104.9	
SnH₂I₂			PbH₂I₂		
$a_{\text{Sn-H}}$	1.701		$a_{\text{Pb-H}}$	1.744	
$a_{\text{Sn-I}}$	2.663		$a_{\text{Pb-I}}$	2.735	
$\Theta_{\text{H-Sn-H}}$	115.3		$\Theta_{\text{H-Pb-H}}$	120.0	
$\Theta_{\text{I-Sn-H}}$	108.2		$\Theta_{\text{I-Pb-H}}$	107.3	
$\Theta_{\text{I-Sn-I}}$	108.4		$\Theta_{\text{I-Pb-I}}$	106.8	
SnHI₃			PbHI₃		
$a_{\text{Sn-H}}$	1.703		$a_{\text{Pb-H}}$	1.753	
$a_{\text{Sn-I}}$	2.654		$a_{\text{Pb-I}}$	2.725	
$\Theta_{\text{I-Sn-H}}$	110.2		$\Theta_{\text{I-Pb-H}}$	111.0	
$\Theta_{\text{I-Sn-I}}$	108.8		$\Theta_{\text{I-Pb-I}}$	107.9	
SnI₄			PbI₄		
$a_{\text{Sn-I}}$	2.651		$a_{\text{Pb-I}}$	2.725	

The tetrahalides are of T_d symmetry ($\Theta_{\text{X-M-X}} = 109.5^\circ$).

^a a_o (and θ°): parameters obtained from the nuclear positions derived from rotational constants in the vibrational ground state. ^b a_g : thermal averages of the internuclear distance, derived from electron diffraction (ED), they are specific to the temperature of the ED experiment (18 °C for SnCl₄). ^c a_a : the bond distance obtained directly from ED experiments (For PbCl₄, T = 20 °C). Note: The Sn—H distances are estimated based on an empirical relationship between $a_o(\text{Sn—H})$ and the vibrational wavenumber $\nu(\text{Sn—H})$ for various stannane derivatives. See ref 1 in the footnote below.

¹ Graner, G.; Hirota, E.; Iijima, T.; Kuchitsu, K.; Ramsay, D. A.; Vogt, J.; Vogt N. In Landolt-Börnstein - Numerical Data and Functional Relationships in Science and Technology; Group II: Molecules and Radicals Volume 25. Structure Data of Free Polyatomic Molecules, Sub-volume A: Inorganic Molecules. Martienssen, W., Kuchitsu, K., Eds. Springer-Verlag: Berlin, Heidelberg; 2006.

Table S.2a: CCSD NBO charges for atoms in the $MH_{4-n}X_n$ molecules. The basis sets indicated were used for all elements except Sn, Pb, and I for which we used the pseudopotentials and basis sets used are described in the methods section.

6-311+G*					6-311++G**				cc-pVTZ				aug-cc-pVTZ			
M = C																
CH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
C	-0.044	-0.518	-0.602	-0.704	-0.022	-0.495	-0.579	-0.681	-0.055	-0.492	-0.570	-0.682	-0.051	-0.507	-0.589	-0.703
H	0.146	0.193	0.200	0.206	0.138	0.184	0.191	0.196	0.141	0.186	0.193	0.197	0.146	0.191	0.197	0.200
X	-0.394	-0.061	0.001	0.087	-0.392	-0.057	0.005	0.092	-0.369	-0.068	-0.009	0.089	-0.386	-0.067	-0.002	0.104
CH ₂ X ₂																
C	0.504	-0.387	-0.556	-0.758	0.522	-0.365	-0.534	-0.735	0.482	-0.330	-0.489	-0.718	0.495	-0.360	-0.529	-0.760
H	0.121	0.201	0.211	0.217	0.110	0.187	0.196	0.202	0.110	0.191	0.198	0.200	0.115	0.195	0.201	0.201
X	-0.373	-0.008	0.066	0.162	-0.371	-0.004	0.070	0.165	-0.351	-0.026	0.047	0.159	-0.363	-0.015	0.064	0.179
CHX ₃																
C	0.941	-0.326	-0.583	-0.877	0.952	-0.311	-0.567	-0.862	0.915	-0.235	-0.486	-0.837	0.922	-0.298	-0.562	-0.911
H	0.115	0.208	0.215	0.217	0.099	0.186	0.193	0.196	0.097	0.193	0.190	0.185	0.099	0.195	0.192	0.184
X	-0.352	0.039	0.122	0.220	-0.350	0.042	0.125	0.222	-0.337	0.014	0.099	0.217	-0.340	0.034	0.123	0.243
CX ₄																
C	1.326	-0.320	-0.669	-1.048	1.326	-0.320	-0.669	-1.048	1.268	-0.201	-0.572	-1.049	1.277	-0.292	-0.669	-1.142
X	-0.331	0.080	0.167	0.262	-0.331	0.080	0.167	0.262	-0.317	0.050	0.143	0.262	-0.319	0.073	0.167	0.285
M=Si																
SiH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Si	1.282	0.867	0.775	0.660	1.357	0.952	0.863	0.753	1.367	0.996	0.895	0.762	1.385	0.997	0.894	0.751
H	-0.215	-0.166	-0.156	-0.149	-0.240	-0.195	-0.187	-0.181	-0.245	-0.201	-0.192	-0.182	-0.245	-0.200	-0.191	-0.184
X	-0.637	-0.369	-0.306	-0.214	-0.637	-0.367	-0.303	-0.210	-0.632	-0.393	-0.321	-0.215	-0.651	-0.396	-0.320	-0.197
SiH ₂ X ₂																
Si	1.778	1.022	0.844	0.617	1.819	1.073	0.901	0.679	1.834	1.172	0.977	0.708	1.867	1.163	0.959	0.669
H	-0.252	-0.168	-0.152	-0.141	-0.272	-0.195	-0.182	-0.175	-0.283	-0.209	-0.194	-0.182	-0.281	-0.205	-0.191	-0.183
X	-0.637	-0.343	-0.270	-0.168	-0.637	-0.342	-0.269	-0.165	-0.633	-0.377	-0.294	-0.172	-0.652	-0.377	-0.288	-0.152
SiHX ₃																
.Si	2.171	1.118	0.859	0.515	2.189	1.142	0.887	0.547	2.193	1.292	1.002	0.586	2.248	1.202	0.952	0.510
H	-0.252	-0.165	-0.147	-0.138	-0.289	-0.191	-0.178	-0.175	-0.304	-0.212	-0.199	-0.190	-0.301	-0.170	-0.191	-0.191
X	-0.631	-0.318	-0.237	-0.126	-0.633	-0.317	-0.236	-0.124	-0.630	-0.360	-0.268	-0.132	-0.649	-0.344	-0.254	-0.107
SiX ₄																
Si	2.509	1.178	0.835	0.364	2.509	1.178	0.835	0.364	2.490	1.361	0.975	0.393	2.567	1.291	0.902	0.294
X	-0.627	-0.294	-0.209	-0.091	-0.627	-0.294	-0.209	-0.091	-0.623	-0.340	-0.244	-0.098	-0.642	-0.323	-0.226	-0.073
M=Ge																
GeH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Ge	1.155	0.783	0.691	0.535	1.228	0.864	0.775	0.628	1.197	0.848	0.751	0.623	1.216	0.863	0.765	0.624
H	-0.180	-0.132	-0.123	-0.106	-0.204	-0.159	-0.152	-0.139	-0.189	-0.149	-0.141	-0.133	-0.191	-0.152	-0.144	-0.139
X	-0.616	-0.388	-0.321	-0.218	-0.616	-0.386	-0.318	-0.211	-0.631	-0.400	-0.328	-0.223	-0.642	-0.408	-0.332	-0.207
GeH ₂ X ₂																
Ge	1.665	0.973	0.794	0.528	1.702	1.027	0.852	0.589	1.695	1.058	0.870	0.616	1.735	1.066	0.871	0.591
H	-0.210	-0.128	-0.115	-0.096	-0.228	-0.157	-0.147	-0.131	-0.215	-0.150	-0.137	-0.128	-0.216	-0.149	-0.138	-0.133
X	-0.623	-0.358	-0.282	-0.168	-0.623	-0.357	-0.279	-0.164	-0.633	-0.380	-0.298	-0.180	-0.652	-0.384	-0.298	-0.162

GeHX₃																
Ge	2.089	1.102	0.838	0.461	2.104	1.127	0.865	0.492	2.105	1.212	0.936	0.552	2.164	1.214	0.921	0.483
H	-0.220	-0.121	-0.107	-0.090	-0.236	-0.148	-0.138	-0.128	-0.225	-0.146	-0.134	-0.130	-0.225	-0.143	-0.133	-0.132
X	-0.623	-0.327	-0.244	-0.124	-0.622	-0.326	-0.242	-0.122	-0.627	-0.356	-0.267	-0.141	-0.646	-0.357	-0.263	-0.117
GeX₄																
Ge	2.467	1.191	0.840	0.351	2.467	1.191	0.840	0.351	2.460	1.329	0.964	0.434	2.542	1.309	0.913	0.334
X	-0.617	-0.298	-0.210	-0.088	-0.617	-0.298	-0.210	-0.088	-0.615	-0.332	-0.241	-0.108	-0.635	-0.327	-0.228	-0.084
M=Sn																
SnH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Sn	1.329	0.992	0.908	0.798	1.401	1.070	0.988	0.881	1.409	1.097	1.004	0.882	1.445	1.130	1.039	0.909
H	-0.213	-0.178	-0.170	-0.164	-0.238	-0.205	-0.198	-0.194	-0.243	-0.209	-0.201	-0.194	-0.249	-0.216	-0.209	-0.204
X	-0.689	-0.458	-0.397	-0.306	-0.689	-0.455	-0.393	-0.300	-0.680	-0.472	-0.401	-0.301	-0.697	-0.482	-0.413	-0.297
SnH₂X₂																
Sn	1.847	1.221	1.060	0.846	1.885	1.266	1.108	0.899	1.890	1.319	1.138	0.899	1.932	1.346	1.168	0.915
H	-0.233	-0.174	-0.162	-0.155	-0.253	-0.199	-0.189	-0.185	-0.262	-0.206	-0.194	-0.185	-0.267	-0.210	-0.199	-0.195
X	-0.690	-0.436	-0.368	-0.268	-0.689	-0.434	-0.365	-0.265	-0.683	-0.453	-0.375	-0.265	-0.699	-0.463	-0.385	-0.262
SnHX₃																
Sn	2.291	1.391	1.161	0.845	2.306	1.410	1.182	0.873	2.301	1.489	1.223	0.870	2.355	1.513	1.250	0.877
H	-0.240	-0.166	-0.152	-0.148	-0.257	-0.189	-0.178	-0.184	-0.270	-0.199	-0.186	-0.180	-0.272	-0.201	-0.189	-0.190
X	-0.684	-0.408	-0.336	-0.232	-0.683	-0.407	-0.335	-0.230	-0.677	-0.430	-0.346	-0.230	-0.694	-0.437	-0.354	-0.229
SnX₄																
Sn	2.689	1.508	1.217	0.803	2.689	1.508	1.217	0.803	2.663	1.598	1.257	0.803	2.777	1.619	1.279	0.803
X	-0.672	-0.377	-0.304	-0.201	-0.672	-0.377	-0.304	-0.201	-0.666	-0.400	-0.314	-0.201	-0.694	-0.405	-0.320	-0.201
M=Pb																
PbH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Pb	1.150	0.848	0.771	0.666	1.216	0.920	0.845	0.743	1.214	0.939	0.854	0.737	1.259	0.976	0.894	0.768
H	-0.153	-0.123	-0.118	-0.114	-0.175	-0.149	-0.144	-0.142	-0.179	-0.151	-0.145	-0.140	-0.186	-0.159	-0.154	-0.152
X	-0.692	-0.478	-0.418	-0.324	-0.690	-0.473	-0.412	-0.316	-0.678	-0.487	-0.420	-0.317	-0.700	-0.499	-0.432	-0.312
PbH₂X₂																
Pb	1.661	1.108	0.962	0.757	1.692	1.147	1.005	0.804	1.688	1.191	1.029	0.800	1.738	1.222	1.062	0.817
H	-0.141	-0.104	-0.098	-0.100	-0.158	-0.127	-0.124	-0.129	-0.166	-0.132	-0.127	-0.127	-0.170	-0.137	-0.133	-0.139
X	-0.690	-0.450	-0.383	-0.279	-0.688	-0.447	-0.378	-0.273	-0.678	-0.463	-0.387	-0.273	-0.699	-0.474	-0.399	-0.270
PbHX₃																
Pb	2.110	1.308	1.093	0.788	2.121	1.323	1.111	0.811	2.098	1.388	1.148	0.808	2.161	1.401	1.176	0.814
H	-0.133	-0.090	-0.086	-0.095	-0.147	-0.111	-0.112	-0.127	-0.152	-0.119	-0.117	-0.126	-0.151	-0.118	-0.119	-0.137
X	-0.659	-0.406	-0.335	-0.231	-0.658	-0.404	-0.333	-0.228	-0.649	-0.423	-0.344	-0.227	-0.670	-0.428	-0.352	-0.226
PbX₄																
Pb	2.494	1.448	1.184	0.776	2.494	1.448	1.184	0.776	2.452	1.525	1.218	0.776	2.547	1.552	1.240	0.776
X	-0.623	-0.362	-0.296	-0.194	-0.623	-0.362	-0.296	-0.194	-0.613	-0.381	-0.305	-0.194	-0.637	-0.388	-0.310	-0.194

Table S.2b: MP2 NBO charges for atoms in the $MH_{4-n}X_n$ molecules. The basis sets indicated were used for all elements except Sn, Pb, and I for which we used the pseudopotentials and basis sets used are described in the methods section.

6-311+G*					6-311++G**				cc-pVTZ				aug-cc-pVTZ			
M = C																
CH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
C	-0.060	-0.536	-0.616	-0.718	-0.042	-0.516	-0.597	-0.700	-0.074	-0.518	-0.591	-0.701	-0.071	-0.527	-0.610	-0.722
H	0.151	0.199	0.207	0.213	0.144	0.192	0.199	0.205	0.146	0.196	0.200	0.205	0.151	0.199	0.204	0.208
X	-0.392	-0.062	-0.004	0.080	-0.390	-0.058	0.001	0.085	-0.365	-0.071	-0.010	0.085	-0.383	-0.069	-0.003	0.099
CH ₂ X ₂																
C	0.495	-0.394	-0.555	-0.753	0.510	-0.375	-0.536	-0.734	0.470	-0.337	-0.497	-0.719	0.482	-0.368	-0.537	-0.760
H	0.123	0.207	0.217	0.224	0.113	0.193	0.203	0.210	0.113	0.198	0.204	0.207	0.118	0.202	0.207	0.208
X	-0.371	-0.009	0.061	0.153	-0.369	-0.006	0.065	0.157	-0.348	-0.029	0.044	0.152	-0.359	-0.017	0.061	0.172
CHX ₃																
C	0.933	-0.325	-0.567	-0.851	0.944	-0.311	-0.553	-0.836	0.887	-0.232	-0.481	-0.818	0.911	-0.296	-0.558	-0.891
H	0.117	0.212	0.220	0.223	0.100	0.191	0.198	0.202	0.098	0.198	0.195	0.190	0.101	0.200	0.196	0.188
X	-0.350	0.038	0.116	0.209	-0.348	0.040	0.118	0.211	-0.328	0.011	0.095	0.209	-0.337	0.032	0.121	0.234
CX ₄																
C	1.319	-0.312	-0.639	-0.994	1.319	-0.312	-0.639	-0.994	1.257	-0.189	-0.556	-1.012	1.265	-0.283	-0.656	-1.102
X	-0.330	0.078	0.160	0.248	-0.330	0.078	0.160	0.248	-0.314	0.047	0.139	0.253	-0.316	0.071	0.164	0.275
M=Si																
SiH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Si	1.295	0.886	0.796	0.680	1.344	0.953	0.867	0.754	1.364	0.995	0.895	0.764	1.383	0.997	0.894	0.753
H	-0.221	-0.171	-0.161	-0.153	-0.237	-0.194	-0.186	-0.180	-0.246	-0.200	-0.191	-0.181	-0.246	-0.200	-0.191	-0.183
X	-0.632	-0.372	-0.313	-0.221	-0.633	-0.369	-0.309	-0.216	-0.626	-0.394	-0.323	-0.221	-0.646	-0.398	-0.322	-0.203
SiH ₂ X ₂																
Si	1.778	1.033	0.862	0.631	1.803	1.075	0.908	0.683	1.824	1.171	0.976	0.711	1.858	1.161	0.957	0.672
H	-0.257	-0.172	-0.155	-0.144	-0.269	-0.194	-0.181	-0.173	-0.284	-0.208	-0.193	-0.181	-0.282	-0.204	-0.190	-0.181
X	-0.632	-0.345	-0.276	-0.172	-0.632	-0.343	-0.274	-0.168	-0.627	-0.378	-0.295	-0.175	-0.647	-0.377	-0.288	-0.155
SiHX ₃																
.Si	2.161	1.124	0.873	0.524	2.172	1.144	0.897	0.552	2.176	1.288	0.998	0.587	2.233	1.267	0.946	0.511
H	-0.274	-0.168	-0.150	-0.140	-0.286	-0.191	-0.177	-0.173	-0.305	-0.211	-0.197	-0.188	-0.301	-0.205	-0.190	-0.189
X	-0.629	-0.318	-0.241	-0.128	-0.629	-0.318	-0.240	-0.126	-0.624	-0.359	-0.267	-0.133	-0.644	-0.354	-0.252	-0.107
SiX ₄																
Si	2.490	1.178	0.846	0.369	2.490	1.178	0.846	0.369	2.467	1.353	0.967	0.391	2.546	1.317	0.891	0.290
X	-0.622	-0.294	-0.212	-0.092	-0.622	-0.294	-0.212	-0.092	-0.617	-0.338	-0.242	-0.098	-0.636	-0.329	-0.223	-0.072
M=Ge																
GeH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Ge	1.129	0.749	0.658	0.540	1.189	0.815	0.727	0.611	1.183	0.836	0.738	0.612	1.203	0.852	0.753	0.614
H	-0.166	-0.121	-0.113	-0.106	-0.186	-0.144	-0.137	-0.132	-0.186	-0.145	-0.136	-0.128	-0.189	-0.148	-0.140	-0.134
X	-0.632	-0.384	-0.320	-0.223	-0.633	-0.382	-0.316	-0.216	-0.624	-0.401	-0.329	-0.227	-0.636	-0.409	-0.334	-0.212
GeH ₂ X ₂																
Ge	1.646	0.940	0.765	0.531	1.678	0.984	0.813	0.584	1.677	1.049	0.860	0.611	1.719	1.057	0.862	0.585
H	-0.191	-0.117	-0.103	-0.094	-0.207	-0.141	-0.130	-0.125	-0.213	-0.145	-0.132	-0.122	-0.214	-0.145	-0.133	-0.128
X	-0.632	-0.353	-0.279	-0.172	-0.632	-0.351	-0.276	-0.167	-0.625	-0.379	-0.298	-0.183	-0.645	-0.384	-0.298	-0.164

GeHX₃																
Ge	2.077	1.070	0.811	0.463	2.090	1.089	0.835	0.490	2.080	1.203	0.928	0.550	2.143	1.205	0.912	0.480
H	-0.198	-0.108	-0.093	-0.087	-0.211	-0.132	-0.121	-0.122	-0.223	-0.140	-0.128	-0.124	-0.223	-0.138	-0.127	-0.127
X	-0.627	-0.321	-0.239	-0.125	-0.626	-0.319	-0.238	-0.123	-0.619	-0.354	-0.267	-0.142	-0.640	-0.356	-0.262	-0.118
GeX₄																
Ge	2.461	1.160	0.818	0.351	2.461	1.160	0.818	0.351	2.429	1.320	0.956	0.433	2.516	1.299	0.902	0.330
X	-0.615	-0.290	-0.204	-0.088	-0.615	-0.290	-0.204	-0.088	-0.607	-0.330	-0.239	-0.108	-0.629	-0.325	-0.225	-0.083
M=Sn																
SnH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Sn	1.337	1.002	0.916	0.805	1.395	1.065	0.982	0.874	1.391	1.082	0.988	0.868	1.429	1.115	1.024	0.896
H	-0.218	-0.181	-0.173	-0.166	-0.238	-0.203	-0.196	-0.191	-0.240	-0.204	-0.196	-0.188	-0.247	-0.211	-0.204	-0.199
X	-0.683	-0.459	-0.398	-0.308	-0.683	-0.455	-0.393	-0.302	-0.670	-0.470	-0.400	-0.303	-0.689	-0.481	-0.412	-0.299
SnH₂X₂																
Sn	1.844	1.225	1.062	0.848	1.875	1.262	1.102	0.893	1.867	1.303	1.124	0.889	1.911	1.331	1.155	0.906
H	-0.238	-0.177	-0.164	-0.155	-0.254	-0.198	-0.187	-0.181	-0.261	-0.201	-0.189	-0.179	-0.265	-0.206	-0.194	-0.190
X	-0.684	-0.436	-0.368	-0.269	-0.683	-0.434	-0.364	-0.265	-0.672	-0.450	-0.373	-0.265	-0.691	-0.459	-0.383	-0.263
SnHX₃																
Sn	2.278	1.393	1.155	0.842	2.290	1.408	1.173	0.867	2.270	1.472	1.210	0.863	2.328	1.495	1.236	0.869
H	-0.245	-0.168	-0.153	-0.148	-0.259	-0.187	-0.176	-0.179	-0.270	-0.194	-0.181	-0.175	-0.271	-0.196	-0.183	-0.184
X	-0.678	-0.408	-0.334	-0.232	-0.677	-0.407	-0.333	-0.230	-0.667	-0.426	-0.343	-0.229	-0.686	-0.433	-0.351	-0.228
SnX₄																
Sn	2.665	1.516	1.210	0.796	2.665	1.516	1.210	0.796	2.621	1.584	1.238	0.796	2.706	1.604	1.260	0.796
X	-0.666	-0.379	-0.302	-0.199	-0.666	-0.379	-0.302	-0.199	-0.655	-0.396	-0.310	-0.199	-0.677	-0.401	-0.315	-0.199
M=Pb																
PbH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Pb	1.134	0.833	0.754	0.647	1.189	0.894	0.817	0.712	1.160	0.903	0.817	0.701	1.221	0.941	0.858	0.732
H	-0.150	-0.119	-0.113	-0.108	-0.169	-0.141	-0.136	-0.133	-0.165	-0.140	-0.134	-0.129	-0.178	-0.149	-0.143	-0.141
X	-0.683	-0.476	-0.415	-0.322	-0.681	-0.471	-0.409	-0.313	-0.664	-0.482	-0.415	-0.315	-0.688	-0.494	-0.428	-0.309
PbH₂X₂																
Pb	1.640	1.094	0.944	0.738	1.666	1.128	0.981	0.779	1.637	1.162	0.998	0.772	1.690	1.192	1.032	0.789
H	-0.140	-0.100	-0.093	-0.094	-0.154	-0.120	-0.116	-0.120	-0.155	-0.123	-0.117	-0.116	-0.158	-0.127	-0.122	-0.128
X	-0.680	-0.447	-0.379	-0.275	-0.679	-0.444	-0.375	-0.269	-0.664	-0.458	-0.382	-0.270	-0.687	-0.469	-0.393	-0.266
PbHX₃																
Pb	2.082	1.292	1.076	0.771	2.091	1.306	1.092	0.791	2.052	1.359	1.123	0.787	2.121	1.389	1.150	0.794
H	-0.132	-0.085	-0.081	-0.089	-0.143	-0.104	-0.104	-0.118	-0.149	-0.109	-0.107	-0.115	-0.146	-0.109	-0.109	-0.126
X	-0.650	-0.402	-0.332	-0.227	-0.649	-0.401	-0.329	-0.224	-0.634	-0.417	-0.339	-0.224	-0.659	-0.426	-0.347	-0.223
PbX₄																
Pb	2.458	1.447	1.167	0.759	2.458	1.447	1.167	0.759	2.392	1.502	1.196	0.759	2.500	1.524	1.217	0.759
X	-0.615	-0.362	-0.292	-0.190	-0.615	-0.362	-0.292	-0.190	-0.598	-0.376	-0.299	-0.190	-0.625	-0.381	-0.304	-0.190

Table S.2b: B3PW91 NBO charges for atoms in the $\text{MH}_{4-n}\text{X}_n$ molecules. The basis sets indicated were used for all elements except Sn, Pb, and I for which we used the pseudopotentials and basis sets used are described in the methods section.

6-311+G*					6-311++G**				cc-pVTZ				aug-cc-pVTZ			
M = C																
CH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
C	-0.085	-0.554	-0.636	-0.740	-0.090	-0.556	-0.638	-0.742	-0.103	-0.544	-0.616	-0.719	-0.108	-0.551	-0.632	-0.737
H	0.158	0.206	0.213	0.217	0.159	0.207	0.214	0.217	0.156	0.205	0.211	0.216	0.161	0.208	0.213	0.216
X	-0.387	-0.064	-0.003	0.091	-0.386	-0.064	-0.003	0.090	-0.365	-0.072	-0.016	0.072	-0.375	-0.073	-0.007	0.088
CH ₂ X ₂																
C	0.473	-0.413	-0.575	-0.785	0.475	-0.408	-0.570	-0.780	0.449	-0.360	-0.512	-0.723	0.449	-0.387	-0.548	-0.769
H	0.130	0.215	0.224	0.228	0.128	0.213	0.221	0.225	0.123	0.212	0.218	0.217	0.127	0.213	0.219	0.219
X	-0.367	-0.009	0.063	0.165	-0.366	-0.008	0.064	0.165	-0.348	-0.032	0.038	0.144	-0.351	-0.019	0.055	0.166
CHX ₃																
C	0.915	-0.338	-0.587	-0.898	0.920	-0.328	-0.577	-0.889	0.880	-0.243	-0.486	-0.819	0.881	-0.295	-0.548	-0.887
H	0.122	0.224	0.230	0.229	0.114	0.213	0.219	0.219	0.105	0.214	0.212	0.207	0.105	0.213	0.212	0.206
X	-0.346	0.038	0.119	0.223	-0.345	0.039	0.119	0.223	-0.328	0.010	0.091	0.204	-0.329	0.027	0.112	0.227
CX ₄																
C	1.310	-0.320	-0.663	-1.057	1.310	-0.320	-0.663	-1.057	1.232	-0.191	-0.553	-1.013	1.238	-0.268	-0.630	-1.080
X	-0.327	0.080	0.166	0.264	-0.327	0.080	0.166	0.264	-0.308	0.048	0.138	0.253	-0.310	0.067	0.157	0.270
M=Si																
SiH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Si	1.210	0.791	0.693	0.572	1.233	0.815	0.720	0.601	1.254	0.858	0.758	0.641	1.261	0.871	0.767	0.632
H	-0.194	-0.144	-0.133	-0.128	-0.200	-0.152	-0.143	-0.139	-0.207	-0.160	-0.151	-0.144	-0.208	-0.164	-0.154	-0.147
X	-0.629	-0.360	-0.293	-0.189	-0.632	-0.360	-0.290	-0.183	-0.634	-0.377	-0.306	-0.209	-0.638	-0.379	-0.304	-0.190
SiH ₂ X ₂																
Si	1.731	0.954	0.775	0.528	1.738	0.969	0.793	0.550	1.766	1.064	0.883	0.620	1.770	1.053	0.864	0.581
H	-0.233	-0.143	-0.129	-0.122	-0.235	-0.150	-0.138	-0.134	-0.247	-0.171	-0.157	-0.143	-0.247	-0.169	-0.155	-0.145
X	-0.633	-0.334	-0.258	-0.142	-0.634	-0.334	-0.258	-0.141	-0.636	-0.362	-0.285	-0.167	-0.638	-0.358	-0.277	-0.146
SiHX ₃																
.Si	2.144	1.063	0.795	0.431	2.146	1.069	0.807	0.448	2.165	1.203	0.906	0.537	2.171	1.202	0.882	0.468
H	-0.252	-0.139	-0.122	-0.118	-0.254	-0.145	-0.135	-0.136	-0.270	-0.173	-0.157	-0.148	-0.270	-0.170	-0.151	-0.147
X	-0.631	-0.308	-0.224	-0.104	-0.631	-0.308	-0.224	-0.104	-0.632	-0.343	-0.250	-0.130	-0.634	-0.344	-0.244	-0.107
SiX ₄																
Si	2.498	1.134	0.782	0.284	2.498	1.134	0.782	0.284	2.494	1.336	0.942	0.386	2.505	1.291	0.876	0.298
X	-0.624	-0.284	-0.195	-0.071	-0.624	-0.284	-0.195	-0.071	-0.633	-0.334	-0.236	-0.097	-0.626	-0.323	-0.219	-0.075
M=Ge																
GeH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Ge	1.060	0.681	0.589	0.466	1.088	0.715	0.625	0.504	1.074	0.719	0.624	0.505	1.100	0.742	0.648	0.521
H	-0.144	-0.100	-0.093	-0.088	-0.153	-0.112	-0.106	-0.103	-0.150	-0.110	-0.102	-0.095	-0.155	-0.115	-0.108	-0.106
X	-0.627	-0.380	-0.310	-0.201	-0.629	-0.379	-0.308	-0.195	-0.624	-0.389	-0.319	-0.221	-0.633	-0.396	-0.324	-0.205
GeH ₂ X ₂																
Ge	1.590	0.883	0.704	0.458	1.604	0.903	0.724	0.483	1.605	0.953	0.772	0.545	1.628	0.964	0.786	0.506
H	-0.167	-0.093	-0.082	-0.078	-0.173	-0.103	-0.094	-0.094	-0.177	-0.109	-0.096	-0.092	-0.181	-0.111	-0.104	-0.097
X	-0.628	-0.349	-0.270	-0.151	-0.629	-0.349	-0.268	-0.147	-0.625	-0.367	-0.290	-0.180	-0.633	-0.371	-0.289	-0.156

GeHX₃																
Ge	2.038	1.028	0.759	0.395	2.042	1.035	0.769	0.408	2.043	1.118	0.858	0.499	2.069	1.117	0.839	0.446
H	-0.173	-0.081	-0.070	-0.071	-0.176	-0.089	-0.081	-0.088	-0.186	-0.101	-0.092	-0.088	-0.190	-0.104	-0.092	-0.094
X	-0.622	-0.315	-0.229	-0.108	-0.622	-0.315	-0.229	-0.107	-0.619	-0.339	-0.255	-0.137	-0.626	-0.338	-0.249	-0.117
GeX₄																
Ge	2.436	1.133	0.773	0.294	2.436	1.133	0.773	0.294	2.421	1.254	0.899	0.417	2.454	1.234	0.855	0.328
X	-0.609	-0.283	-0.193	-0.074	-0.609	-0.283	-0.193	-0.074	-0.605	-0.313	-0.225	-0.104	-0.613	-0.308	-0.214	-0.082
M=Sn																
SnH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Sn	1.276	0.949	0.861	0.744	1.307	0.984	0.898	0.784	1.265	0.946	0.853	0.740	1.314	0.990	0.902	0.782
H	-0.198	-0.162	-0.155	-0.151	-0.209	-0.175	-0.169	-0.166	-0.199	-0.164	-0.156	-0.149	-0.210	-0.176	-0.169	-0.165
X	-0.681	-0.463	-0.395	-0.292	-0.680	-0.460	-0.391	-0.286	-0.668	-0.454	-0.384	-0.292	-0.683	-0.464	-0.396	-0.288
SnH₂X₂																
Sn	1.793	1.183	1.015	0.790	1.806	1.202	1.037	0.816	1.772	1.186	1.009	0.789	1.820	1.218	1.047	0.820
H	-0.215	-0.154	-0.143	-0.139	-0.222	-0.165	-0.157	-0.156	-0.217	-0.159	-0.148	-0.138	-0.225	-0.167	-0.157	-0.157
X	-0.682	-0.438	-0.364	-0.256	-0.681	-0.436	-0.362	-0.252	-0.669	-0.434	-0.357	-0.256	-0.685	-0.442	-0.367	-0.254
SnHX₃																
Sn	2.238	1.362	1.119	0.792	2.240	1.368	1.127	0.803	2.204	1.373	1.116	0.796	2.261	1.396	1.148	0.806
H	-0.218	-0.141	-0.130	-0.131	-0.221	-0.149	-0.141	-0.147	-0.222	-0.149	-0.137	-0.135	-0.227	-0.153	-0.147	-0.148
X	-0.673	-0.407	-0.329	-0.220	-0.673	-0.406	-0.329	-0.219	-0.661	-0.408	-0.326	-0.220	-0.678	-0.414	-0.334	-0.219
SnX₄																
Sn	2.633	1.496	1.181	0.746	2.633	1.496	1.181	0.746	2.580	1.518	1.185	0.746	2.660	1.535	1.204	0.746
X	-0.658	-0.374	-0.295	-0.186	-0.658	-0.374	-0.295	-0.186	-0.645	-0.379	-0.296	-0.186	-0.665	-0.384	-0.301	-0.186
M=Pb																
PbH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Pb	1.121	0.836	0.756	0.645	1.148	0.867	0.789	0.679	1.100	0.825	0.742	0.635	1.158	0.876	0.797	0.682
H	-0.147	-0.117	-0.112	-0.110	-0.156	-0.128	-0.125	-0.125	-0.146	-0.117	-0.112	-0.108	-0.159	-0.131	-0.126	-0.125
X	-0.681	-0.485	-0.420	-0.313	-0.681	-0.482	-0.414	-0.305	-0.663	-0.473	-0.406	-0.312	-0.682	-0.484	-0.419	-0.308
PbH₂X₂																
Pb	1.621	1.097	0.945	0.731	1.629	1.112	0.963	0.752	1.587	1.091	0.933	0.725	1.643	1.128	0.975	0.752
H	-0.131	-0.094	-0.090	-0.097	-0.136	-0.103	-0.102	-0.112	-0.132	-0.098	-0.094	-0.094	-0.141	-0.107	-0.104	-0.110
X	-0.679	-0.455	-0.382	-0.269	-0.679	-0.453	-0.379	-0.264	-0.662	-0.448	-0.373	-0.269	-0.681	-0.457	-0.384	-0.266
PbHX₃																
Pb	2.064	1.300	1.078	0.763	2.064	1.305	1.084	0.773	2.014	1.302	1.070	0.761	2.079	1.327	1.100	0.773
H	-0.124	-0.077	-0.077	-0.094	-0.123	-0.082	-0.087	-0.110	-0.126	-0.083	-0.083	-0.091	-0.130	-0.088	-0.089	-0.107
X	-0.647	-0.408	-0.333	-0.223	-0.647	-0.407	-0.332	-0.221	-0.630	-0.406	-0.329	-0.223	-0.650	-0.413	-0.337	-0.222
PbX₄																
Pb	2.436	1.460	1.171	0.753	2.436	1.460	1.171	0.753	2.361	1.470	1.170	0.753	2.454	1.488	1.191	0.753
X	-0.609	-0.365	-0.293	-0.188	-0.609	-0.365	-0.293	-0.188	-0.590	-0.367	-0.293	-0.188	-0.614	-0.372	-0.298	-0.188

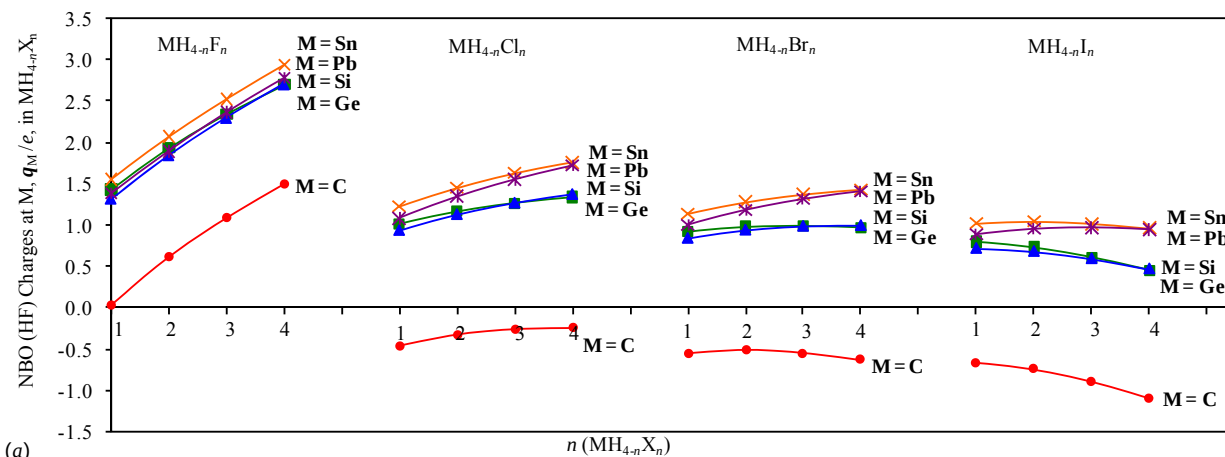
Table S.2b: HF NBO charges for atoms in the $MH_{4-n}X_n$ molecules. The basis sets indicated were used for all elements except Sn, Pb, and I for which we used the pseudopotentials and basis sets used are described in the methods section.

6-311+G*					6-311++G**					cc-pVTZ				aug-cc-pVTZ			
M = C																	
CH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	
C	0.033	-0.457	-0.548	-0.664	0.031	-0.457	-0.548	-0.663	0.020	-0.448	-0.533	-0.646	0.009	-0.459	-0.550	-0.664	
H	0.134	0.183	0.191	0.195	0.134	0.183	0.191	0.195	0.131	0.181	0.189	0.194	0.137	0.185	0.191	0.195	
X	-0.434	-0.091	-0.024	0.080	-0.433	-0.091	-0.024	0.079	-0.413	-0.096	-0.035	0.063	-0.420	-0.097	-0.024	0.080	
CH ₂ X ₂																	
C	0.613	-0.323	-0.507	-0.743	0.617	-0.317	-0.500	-0.736	0.594	-0.266	-0.441	-0.691	0.594	-0.295	-0.478	-0.729	
H	0.110	0.195	0.206	0.208	0.107	0.191	0.202	0.205	0.100	0.189	0.198	0.201	0.104	0.191	0.200	0.200	
X	-0.416	-0.033	0.048	0.163	-0.415	-0.033	0.048	0.163	-0.397	-0.056	0.022	0.145	-0.401	-0.044	0.040	0.165	
CHX ₃																	
C	1.081	-0.258	-0.544	-0.891	1.087	-0.250	-0.535	-0.879	1.043	-0.157	-0.432	-0.810	1.048	-0.207	-0.495	-0.872	
H	0.105	0.204	0.214	0.211	0.097	0.194	0.203	0.199	0.087	0.194	0.195	0.190	0.088	0.193	0.196	0.191	
X	-0.395	0.018	0.110	0.227	-0.395	0.019	0.110	0.227	-0.377	-0.013	0.079	0.207	-0.379	0.005	0.100	0.227	
CX ₄																	
C	1.491	-0.244	-0.630	-1.090	1.415	-0.244	-0.630	-1.090	1.415	-0.109	-0.522	-1.033	1.418	-0.199	-0.603	-1.100	
X	-0.373	0.061	0.158	0.272	-0.354	0.061	0.158	0.272	-0.354	0.027	0.131	0.258	-0.355	0.050	0.151	0.275	
M=Si																	
SiH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	
Si	1.430	1.013	0.916	0.795	1.450	1.040	0.946	0.827	1.475	1.104	1.003	0.877	1.494	1.099	1.002	0.869	
H	-0.249	-0.201	-0.192	-0.187	-0.255	-0.210	-0.201	-0.198	-0.263	-0.226	-0.215	-0.207	-0.269	-0.224	-0.216	-0.210	
X	-0.683	-0.409	-0.341	-0.234	-0.684	-0.411	-0.342	-0.233	-0.686	-0.427	-0.356	-0.256	-0.688	-0.427	-0.354	-0.240	
SiH ₂ X ₂																	
Si	1.933	1.165	0.977	0.727	1.941	1.188	1.004	0.759	1.969	1.282	1.087	0.828	1.971	1.285	1.066	0.793	
H	-0.282	-0.198	-0.183	-0.178	-0.285	-0.209	-0.197	-0.194	-0.298	-0.230	-0.215	-0.204	-0.297	-0.226	-0.213	-0.204	
X	-0.684	-0.384	-0.305	-0.186	-0.685	-0.385	-0.306	-0.185	-0.687	-0.411	-0.328	-0.210	-0.688	-0.417	-0.320	-0.192	
SiHX ₃																	
.Si	2.344	1.268	0.988	0.608	2.347	1.279	1.003	0.627	2.363	1.438	1.141	0.732	2.367	1.413	1.099	0.667	
H	-0.300	-0.194	-0.176	-0.174	-0.301	-0.204	-0.190	-0.193	-0.318	-0.232	-0.217	-0.209	-0.317	-0.227	-0.212	-0.205	
X	-0.682	-0.358	-0.270	-0.145	-0.682	-0.359	-0.271	-0.144	-0.682	-0.402	-0.308	-0.174	-0.683	-0.395	-0.296	-0.154	
SiX ₄																	
Si	2.700	1.337	0.970	0.455	2.700	1.337	0.970	0.455	2.692	1.541	1.132	0.551	2.698	1.501	1.069	0.470	
X	-0.675	-0.334	-0.243	-0.114	-0.675	-0.334	-0.243	-0.114	-0.673	-0.385	-0.283	-0.138	-0.675	-0.375	-0.267	-0.118	
M=Ge																	
GeH ₃ X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	
Ge	1.316	0.935	0.839	0.717	1.350	0.975	0.877	0.754	1.342	0.986	0.887	0.780	1.365	1.022	0.922	0.773	
H	-0.208	-0.166	-0.157	-0.154	-0.218	-0.179	-0.171	-0.169	-0.217	-0.179	-0.170	-0.169	-0.222	-0.189	-0.181	-0.173	
X	-0.693	-0.438	-0.367	-0.254	-0.695	-0.438	-0.364	-0.246	-0.693	-0.449	-0.376	-0.274	-0.700	-0.453	-0.378	-0.255	
GeH ₂ X ₂																	
Ge	1.845	1.129	0.939	0.680	1.862	1.150	0.964	0.708	1.867	1.205	1.015	0.762	1.887	1.212	1.015	0.737	
H	-0.229	-0.158	-0.146	-0.142	-0.236	-0.168	-0.158	-0.158	-0.240	-0.179	-0.166	-0.157	-0.244	-0.181	-0.169	-0.163	
X	-0.694	-0.406	-0.324	-0.198	-0.695	-0.407	-0.324	-0.196	-0.694	-0.424	-0.341	-0.224	-0.699	-0.425	-0.339	-0.205	

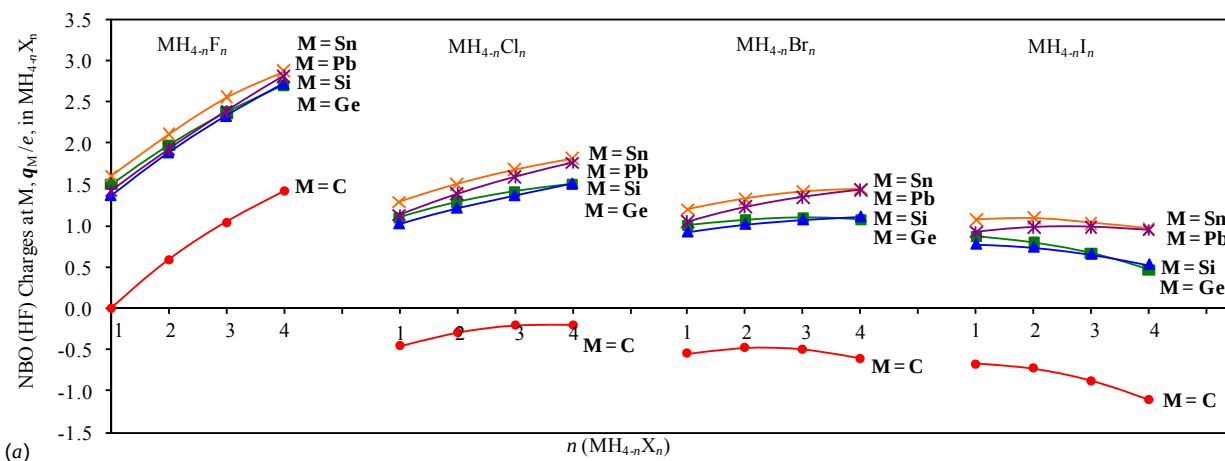
GeHX₃																
Ge	2.299	1.268	0.986	0.591	2.305	1.280	1.001	0.610	2.310	1.367	1.088	0.700	2.330	1.364	1.069	0.650
H	-0.235	-0.147	-0.134	-0.136	-0.240	-0.159	-0.150	-0.157	-0.249	-0.173	-0.161	-0.157	-0.253	-0.174	-0.162	-0.161
X	-0.688	-0.374	-0.284	-0.152	-0.688	-0.374	-0.284	-0.151	-0.687	-0.398	-0.309	-0.181	-0.692	-0.397	-0.302	-0.163
GeX₄																
Ge	2.706	1.373	0.995	0.468	2.706	1.373	0.995	0.468	2.700	1.492	1.123	0.607	2.721	1.509	1.107	0.525
X	-0.677	-0.343	-0.249	-0.117	-0.677	-0.343	-0.249	-0.117	-0.675	-0.373	-0.281	-0.152	-0.680	-0.377	-0.277	-0.131
M=Sn																
SnH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Sn	1.558	1.229	1.139	1.021	1.591	1.265	1.178	1.062	1.561	1.234	1.139	1.041	1.602	1.287	1.196	1.070
H	-0.270	-0.235	-0.228	-0.223	-0.281	-0.248	-0.242	-0.238	-0.273	-0.238	-0.231	-0.230	-0.283	-0.254	-0.247	-0.241
X	-0.748	-0.523	-0.456	-0.351	-0.748	-0.521	-0.453	-0.347	-0.742	-0.519	-0.478	-0.351	-0.752	-0.525	-0.456	-0.347
SnH₂X₂																
Sn	2.071	1.452	1.279	1.046	2.086	1.473	1.303	1.074	2.070	1.477	1.294	1.064	2.106	1.504	1.327	1.087
H	-0.286	-0.227	-0.215	-0.210	-0.294	-0.238	-0.228	-0.227	-0.291	-0.239	-0.227	-0.219	-0.298	-0.247	-0.236	-0.232
X	-0.750	-0.499	-0.425	-0.313	-0.749	-0.498	-0.423	-0.310	-0.744	-0.499	-0.420	-0.313	-0.755	-0.505	-0.428	-0.311
SnHX₃																
Sn	2.524	1.629	1.372	1.019	2.529	1.637	1.381	1.031	2.516	1.655	1.371	1.026	2.553	1.676	1.412	1.033
H	-0.293	-0.216	-0.204	-0.204	-0.297	-0.225	-0.214	-0.219	-0.299	-0.231	-0.215	-0.210	-0.304	-0.237	-0.224	-0.220
X	-0.744	-0.471	-0.389	-0.272	-0.744	-0.471	-0.389	-0.271	-0.739	-0.475	-0.385	-0.272	-0.750	-0.480	-0.396	-0.271
SnX₄																
Sn	2.937	1.762	1.431	0.963	2.937	1.762	1.431	0.963	2.916	1.792	1.422	0.963	2.869	1.809	1.440	0.963
X	-0.734	-0.440	-0.358	-0.241	-0.734	-0.440	-0.358	-0.241	-0.729	-0.448	-0.355	-0.241	-0.742	-0.452	-0.360	-0.241
M=Pb																
PbH₃X	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I	X=F	X=Cl	X=Br	X=I
Pb	1.384	1.086	1.003	0.888	1.412	1.118	1.037	0.923	1.375	1.084	0.997	0.881	1.422	1.129	1.046	0.924
H	-0.210	-0.180	-0.174	-0.172	-0.219	-0.191	-0.187	-0.186	-0.210	-0.182	-0.175	-0.170	-0.221	-0.194	-0.188	-0.185
X	-0.756	-0.547	-0.481	-0.371	-0.755	-0.544	-0.477	-0.365	-0.746	-0.540	-0.471	-0.371	-0.759	-0.548	-0.481	-0.367
PbH₂X₂																
Pb	1.896	1.345	1.184	0.955	1.906	1.362	1.205	0.979	1.883	1.350	1.182	0.961	1.923	1.382	1.226	0.986
H	-0.193	-0.155	-0.149	-0.155	-0.199	-0.164	-0.161	-0.171	-0.196	-0.160	-0.154	-0.158	-0.203	-0.168	-0.167	-0.172
X	-0.754	-0.518	-0.443	-0.323	-0.754	-0.517	-0.441	-0.319	-0.746	-0.515	-0.437	-0.323	-0.759	-0.523	-0.446	-0.321
PbHX₃																
Pb	2.364	1.552	1.314	0.971	2.365	1.557	1.323	0.983	2.343	1.564	1.319	0.975	2.386	1.587	1.346	0.986
H	-0.181	-0.136	-0.135	-0.151	-0.181	-0.143	-0.145	-0.168	-0.185	-0.143	-0.145	-0.155	-0.186	-0.148	-0.152	-0.168
X	-0.728	-0.472	-0.393	-0.273	-0.728	-0.472	-0.393	-0.272	-0.719	-0.474	-0.391	-0.274	-0.733	-0.480	-0.398	-0.273
PbX₄																
Pb	2.777	1.719	1.409	0.950	2.777	1.719	1.409	0.950	2.740	1.740	1.412	0.950	2.808	1.758	1.433	0.950
X	-0.694	-0.430	-0.352	-0.237	-0.694	-0.430	-0.352	-0.237	-0.685	-0.435	-0.353	-0.237	-0.702	-0.439	-0.358	-0.237

Figures

Charges plotted for the Central atoms (q_M) for two sample cases: 6-311+G* and aug-cc-pVTZ for all four methods.

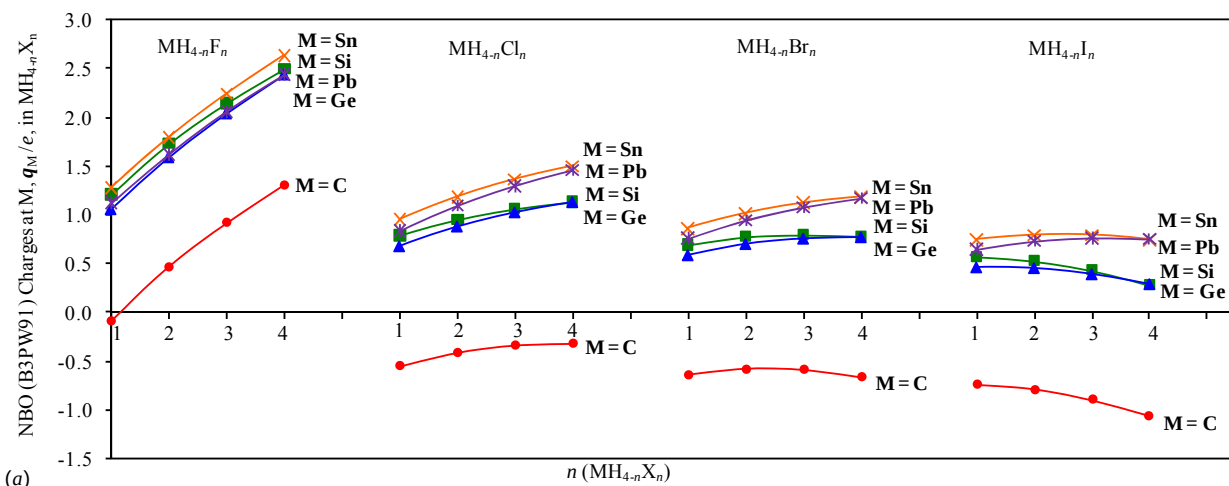


Basis Set: 6-311+G*

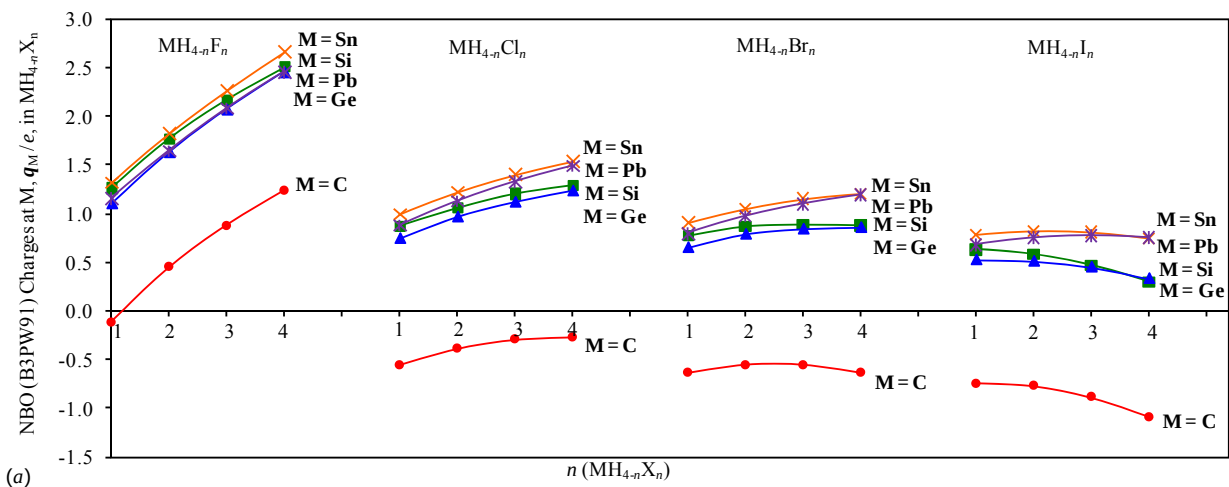


Basis Set: aug-cc-pVTZ

Figure S.1a: NBO point charges obtained at the HF level of theory for the central atoms, q_M . The q_X data plotted below and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

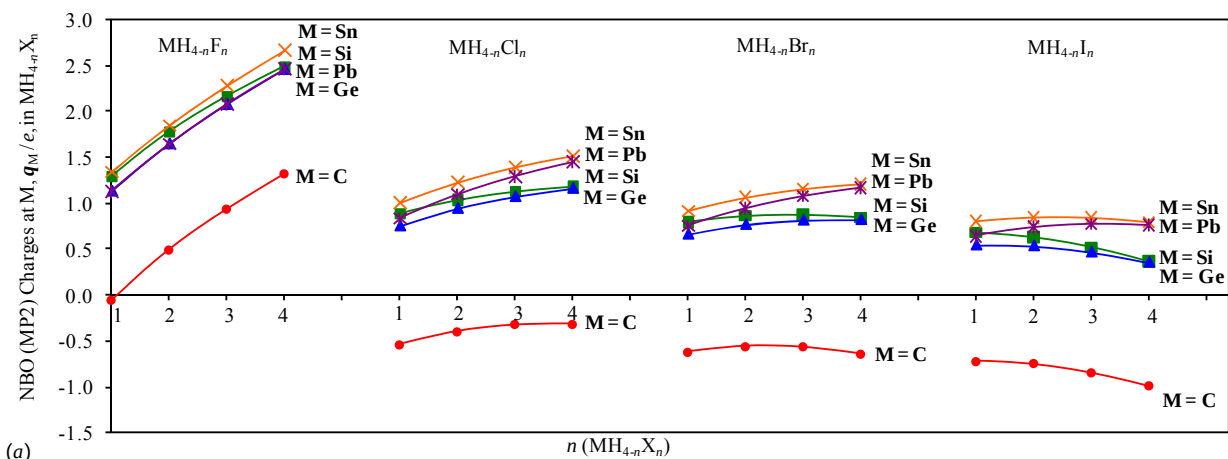


Basis Set: 6-311+G*

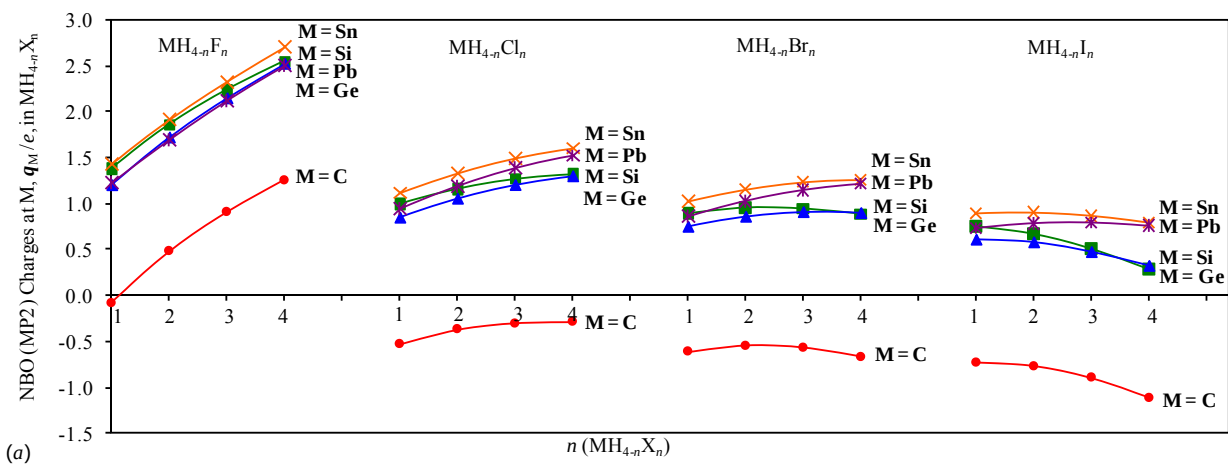


Basis Set: aug-cc-pVTZ

Figure S.1b: NBO point charges obtained at the B3PW91 level of theory for the central atoms, q_M . The q_X data plotted below and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

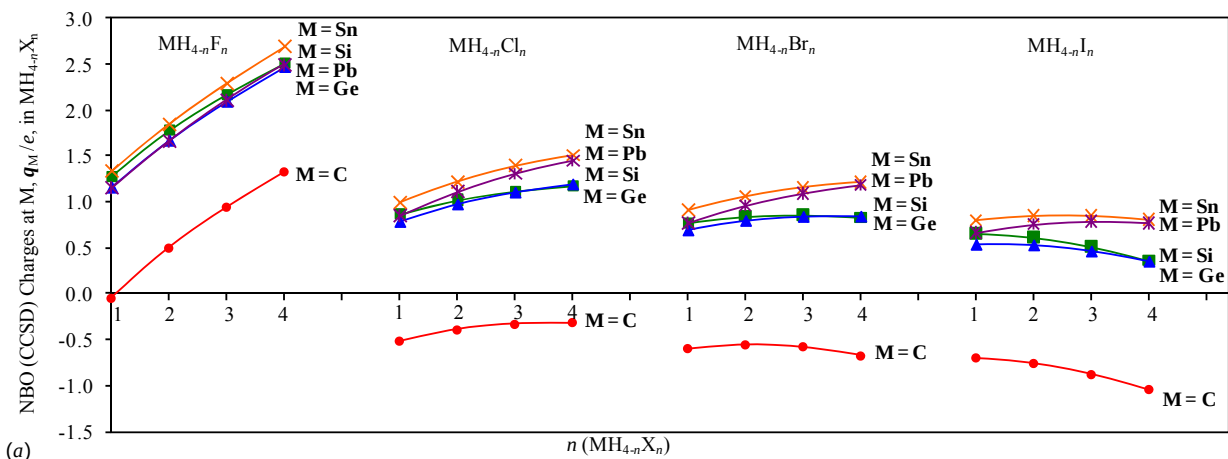


Basis Set: 6-311+G*

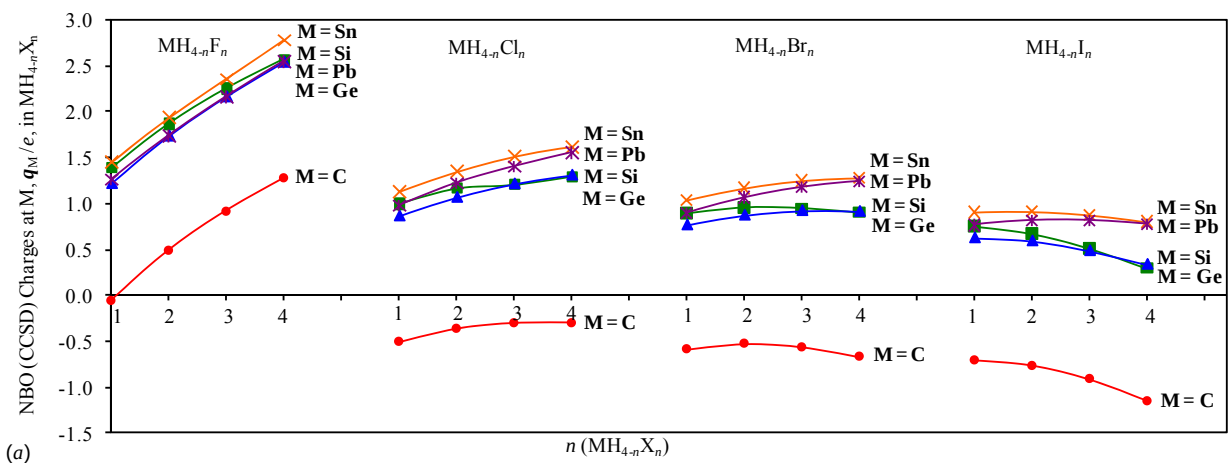


Basis Set: aug-cc-pVTZ

Figure S.1c: NBO point charges obtained at the MP2(full) level of theory for the central atoms, q_M . The q_X data plotted below and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.



Basis Set: 6-311+G*



Basis Set: aug-cc-pVTZ

Figure S.1d: NBO point charges obtained at the CCSD level of theory for the central atoms, q_M . The q_X data plotted below and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

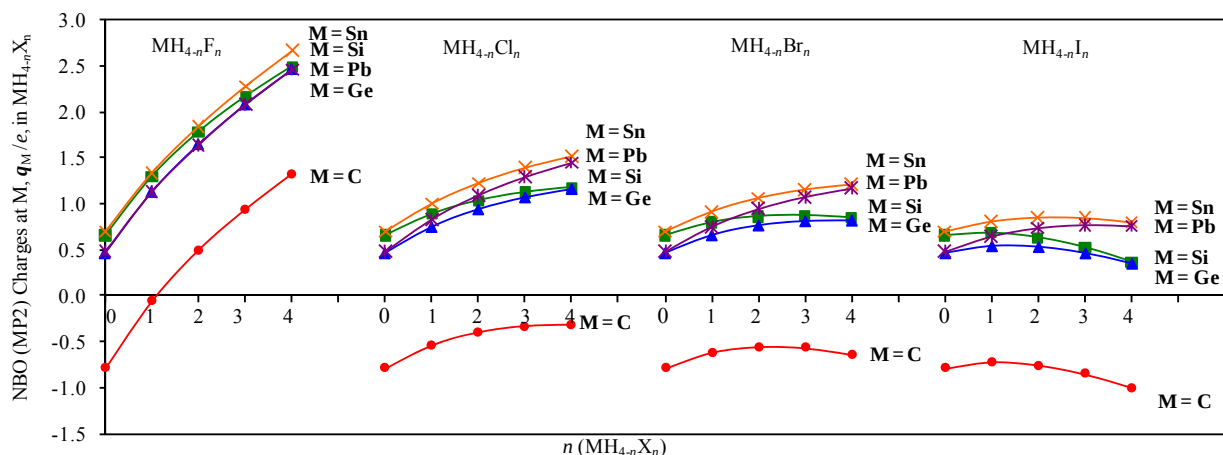


Figure S.1e: NBO point charges obtained at the MP2(full) level of theory for the central atoms, q_M , extended to include the MH_4 ($n = 0$) molecules.. For the elements preceding Sn in the periodic table, the 6-311+G* basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section. The additional tetrahydride charges are given below.

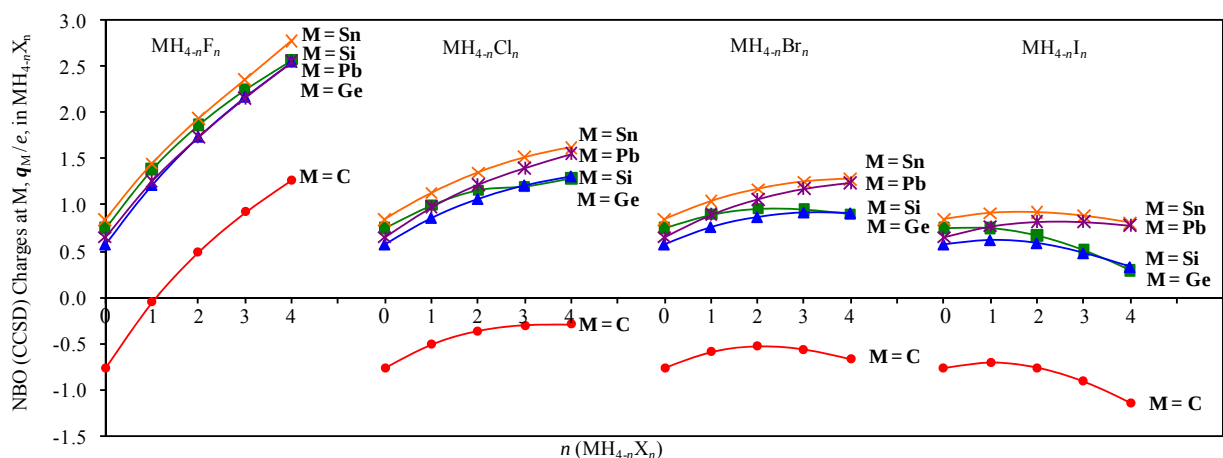
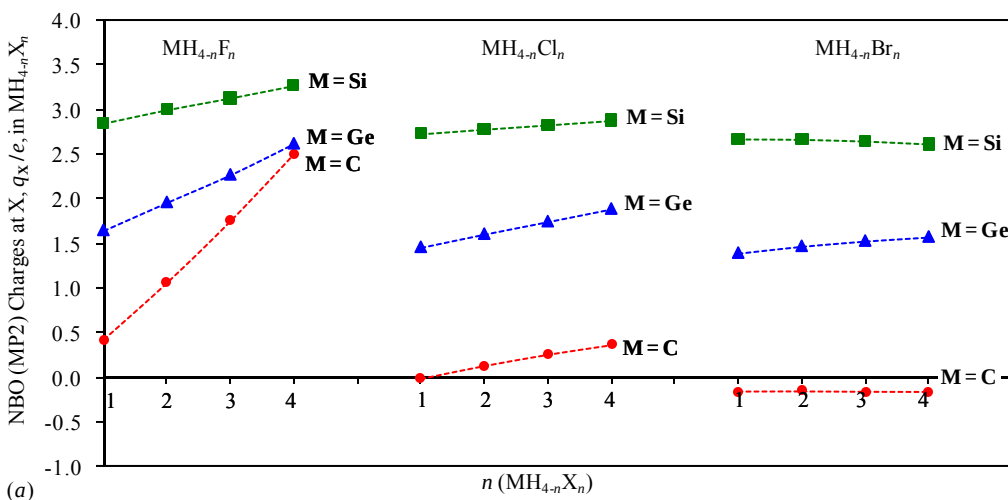


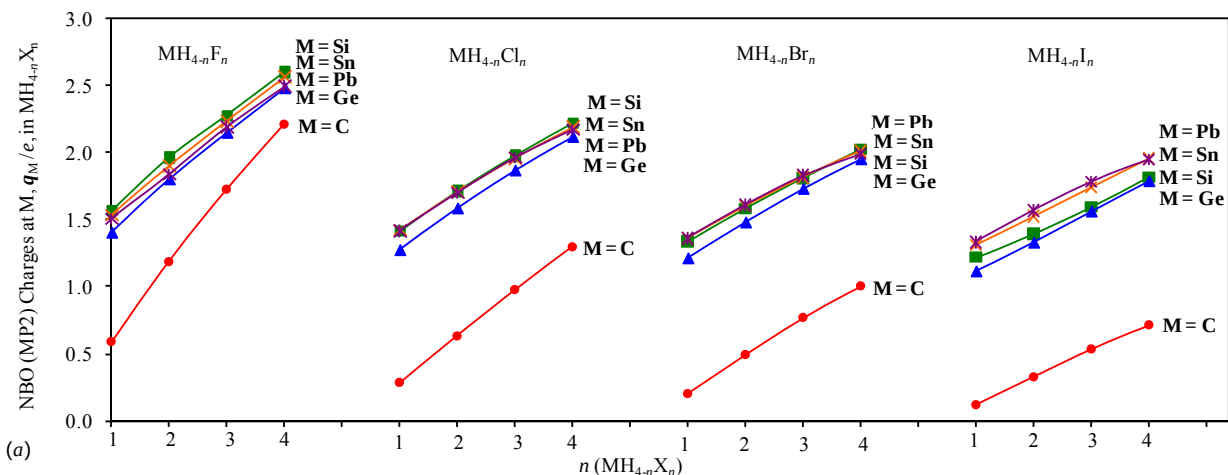
Figure S.1f: NBO point charges obtained at the CCSD level of theory for the central atoms, q_M , extended to include the MH_4 ($n = 0$) molecules. For the elements preceding Sn in the periodic table, the aug-cc-pVTZ basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section. The additional tetrahydride charges are given below.

	MP2(full)/6-311+G*	CCSD/aug-cc-pVTZ		MP2(full)/6-311+G*	CCSD/aug-cc-pVTZ
CH₄			SnH₄		
C	-0.790	-0.762	Sn	0.696	0.843
H	0.198	0.190	H	-0.174	-0.211
SiH₄			PbH₄		
Si	0.654	0.750	Pb	0.480	0.652
H	-0.163	-0.188	H	-0.120	-0.163
GeH₄					
Ge	0.464	0.577			
H	-0.116	-0.144			



Basis Set: 6-311+G*

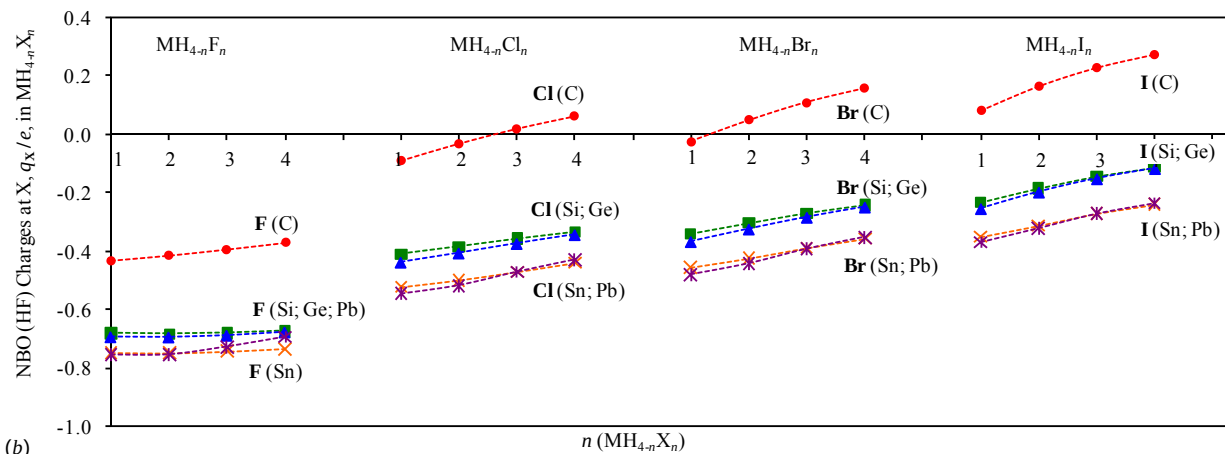
Figure S.2a: AIM point charges obtained at the MP2(full) level of theory for the central atoms, q_M . The q_X data plotted below and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.



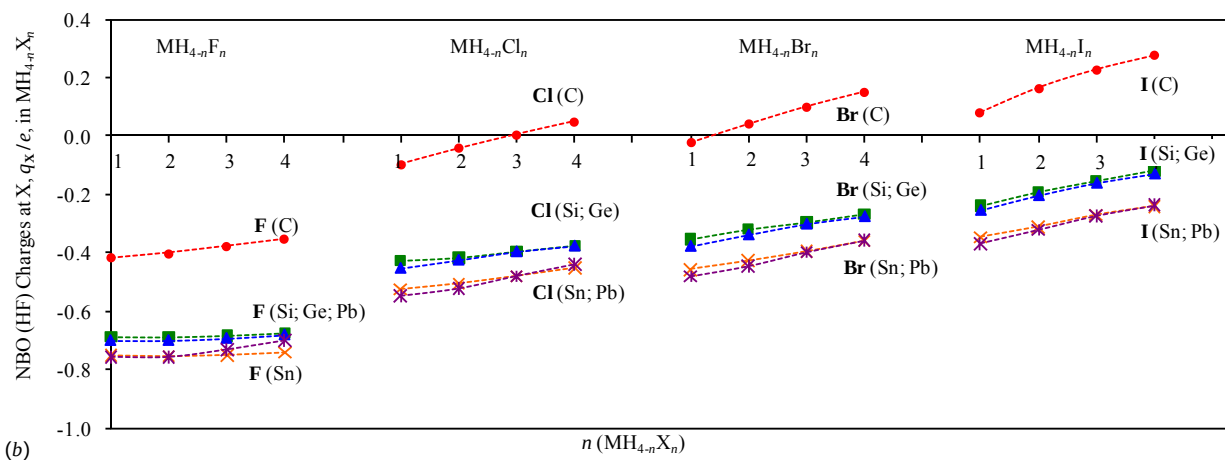
Basis Set: 6-311+G*

Figure S.2b: APT point charges obtained at the MP2(full) level of theory for the central atoms, q_M . The q_X data plotted below and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

Charges plotted for the terminal atoms (q_X) for two sample cases: 6-311+G* and aug-cc-pVTZ for all four methods.

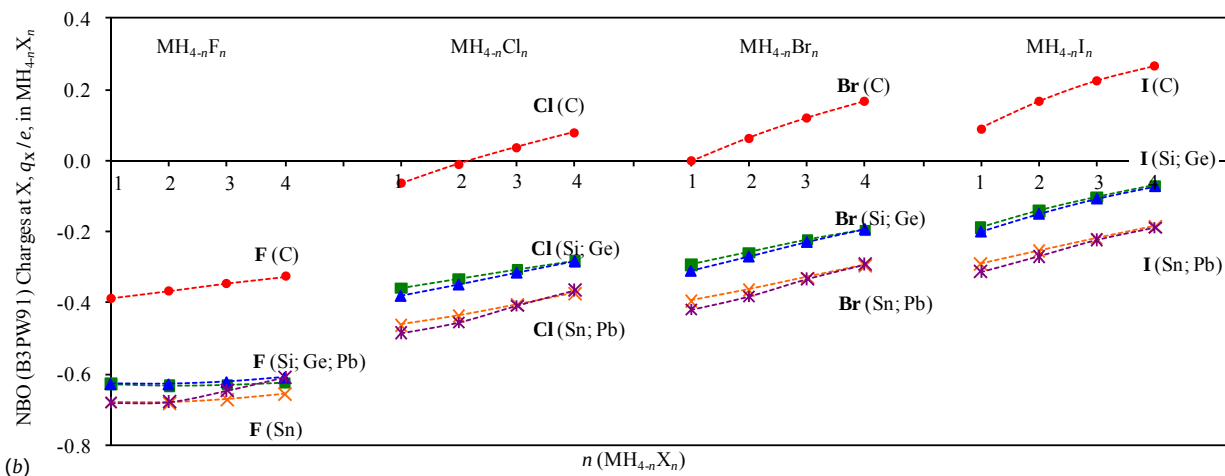


Basis Set: 6-311+G*

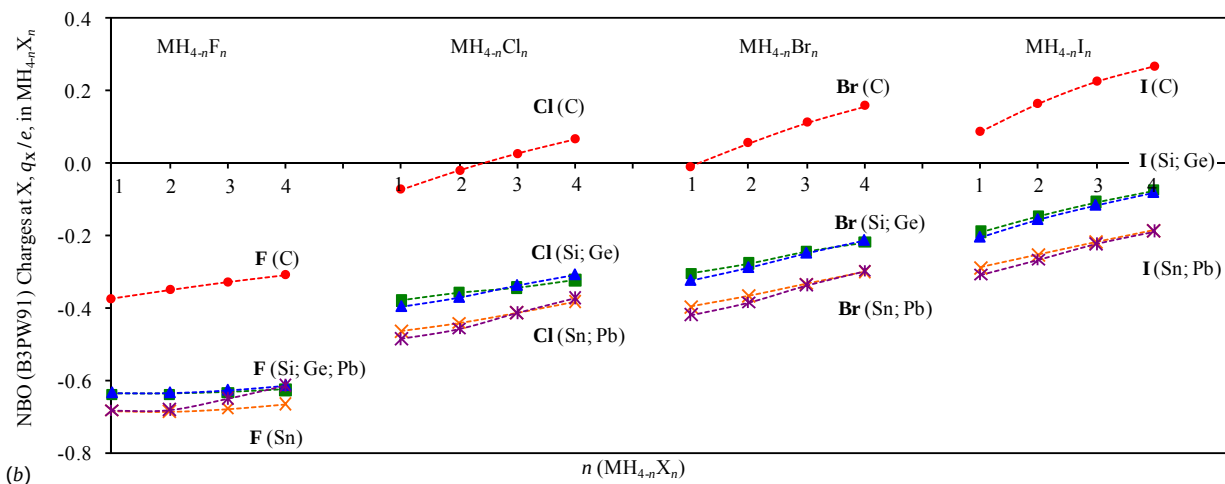


Basis Set: aug-cc-pVTZ

Figure S.3a: NBO point charges obtained at the HF level of theory for the terminal atoms, q_X . The q_M data plotted above and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

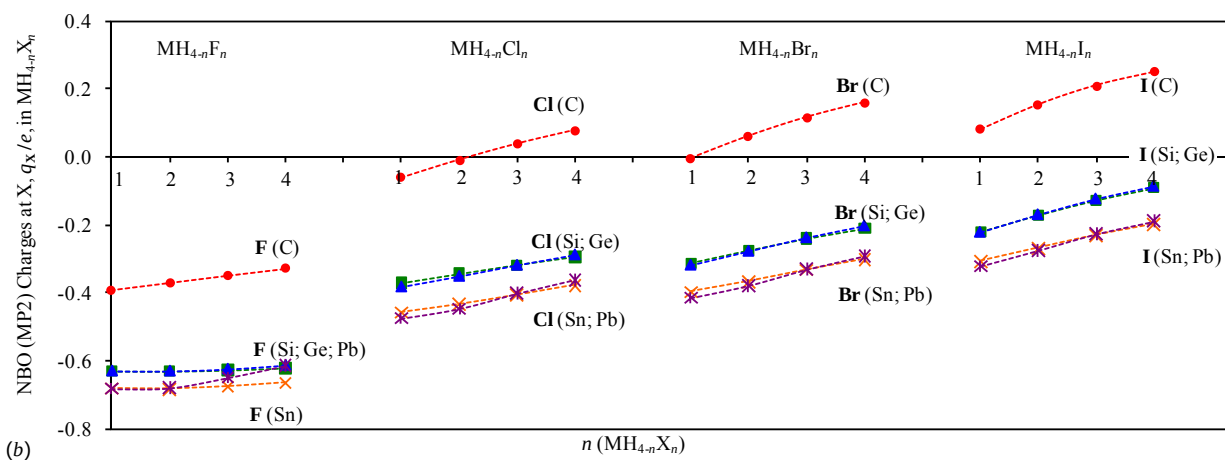


Basis Set: 6-311+G*

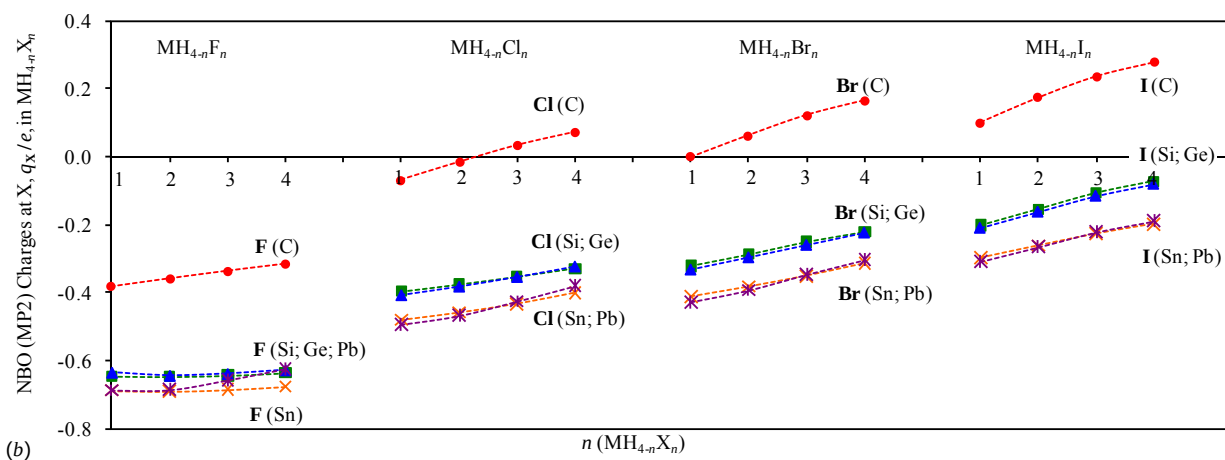


Basis Set: aug-cc-pVTZ

Figure S.3b: NBO point charges obtained at the B3PW91 level of theory for the terminal atoms, q_X . The q_M data plotted above and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

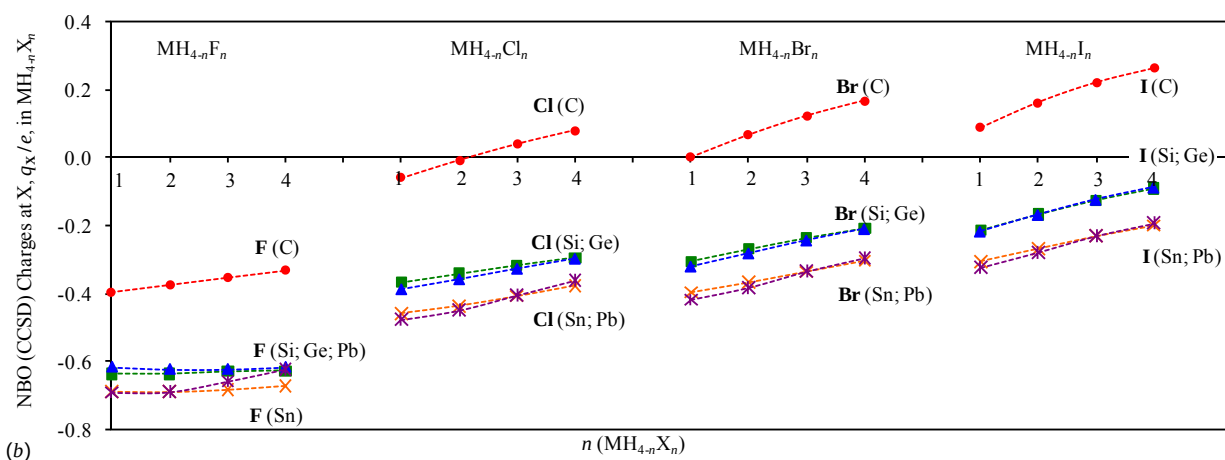


Basis Set: 6-311+G*

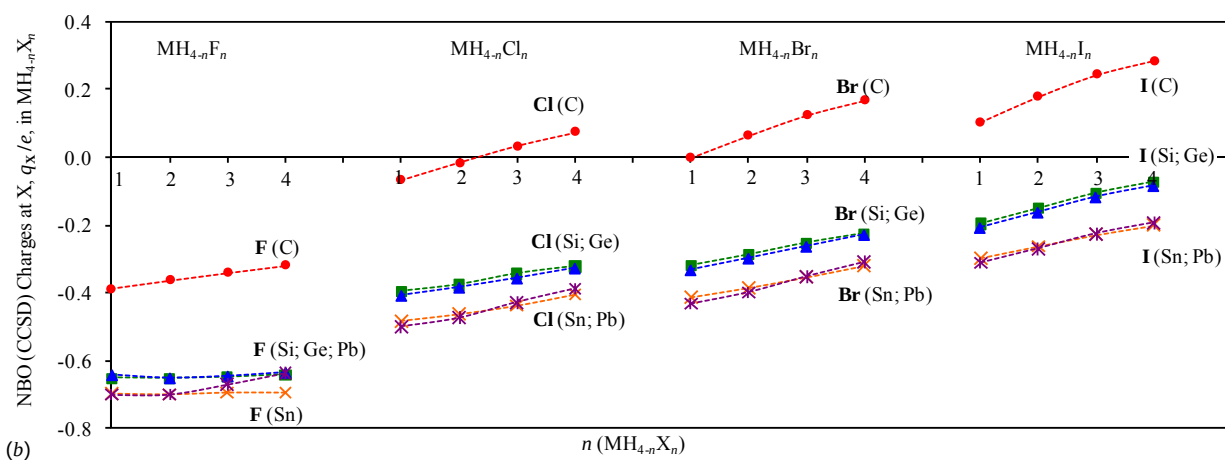


Basis Set: aug-cc-pVTZ

Figure S.3c: NBO point charges obtained at the MP2(full) level of theory for the terminal atoms, q_X . The q_M data plotted above and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

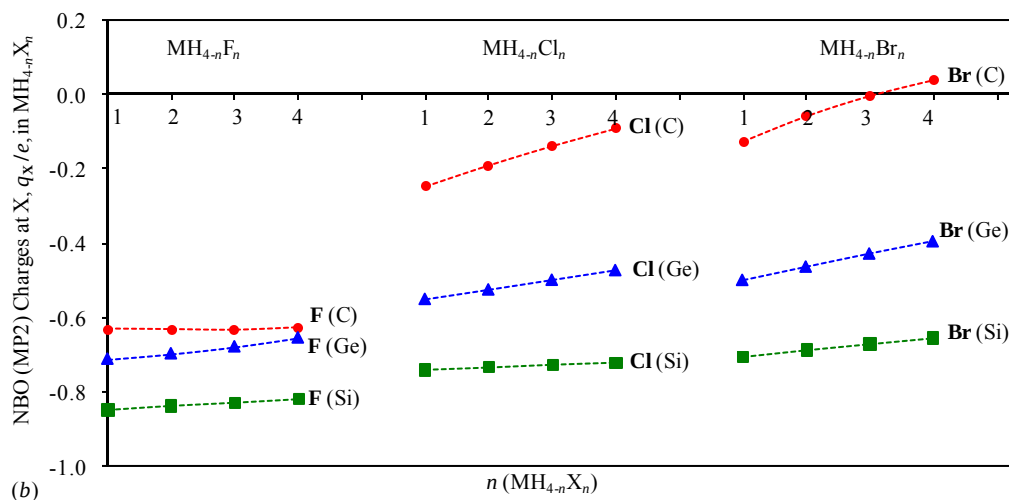


Basis Set: 6-311+G*



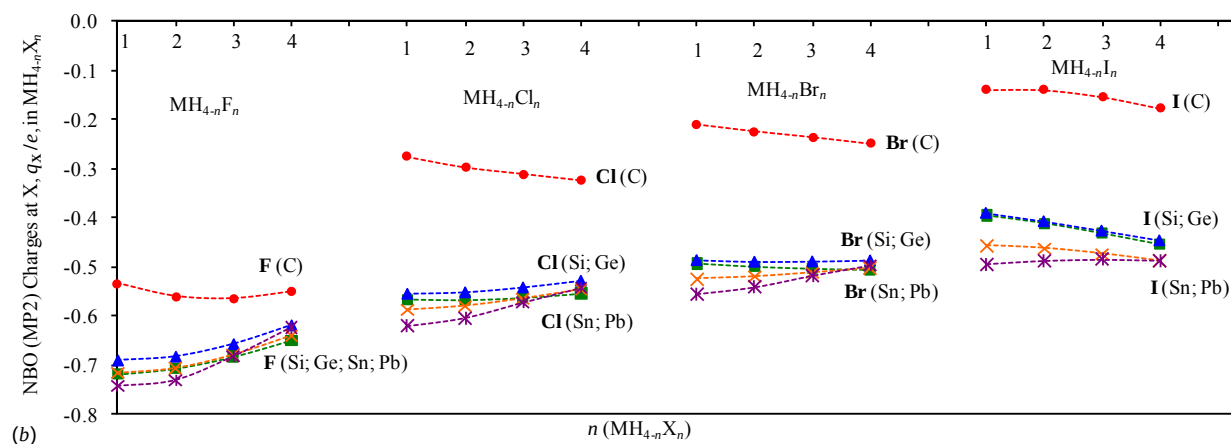
Basis Set: aug-cc-pVTZ

Figure S.3d: NBO point charges obtained at the CCSD level of theory for the terminal atoms, q_X . The q_M data plotted above and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.



Basis Set: 6-311+G*

Figure S.4a: AIM point charges obtained at the CCSD level of theory for the terminal atoms, q_X . The q_M data plotted above and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.



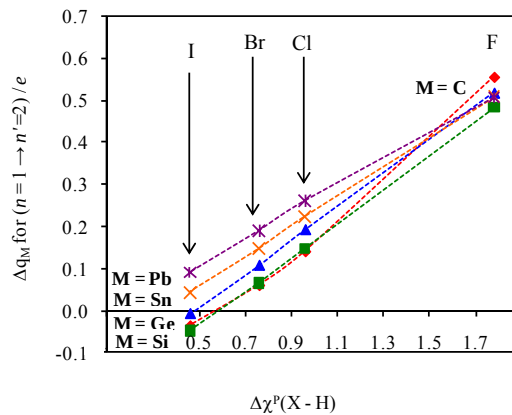
Basis Set: 6-311+G*

Figure S.4b: APT point charges obtained at the CCSD level of theory for the terminal atoms, q_X . The q_M data plotted above and the corresponding q_H values are all listed in the tables in the supporting information. For the elements preceding Sn in the periodic table, the stated basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

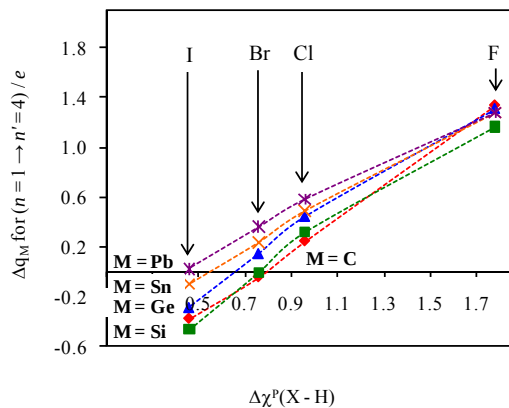
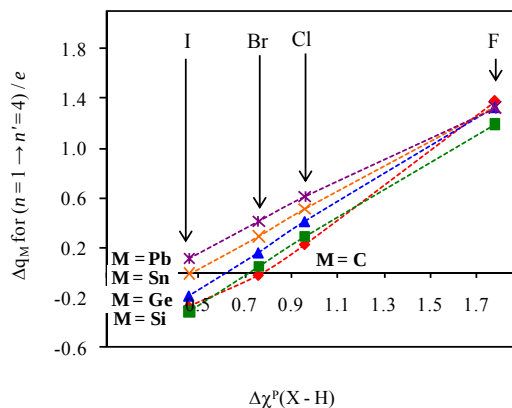
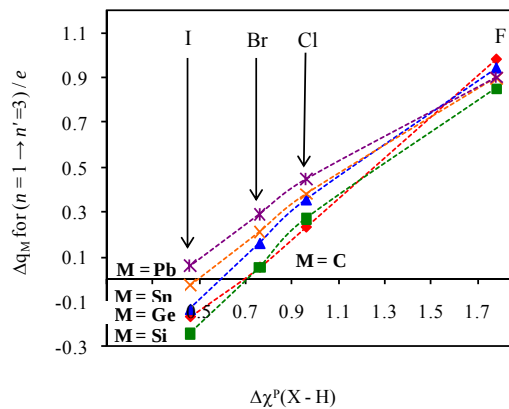
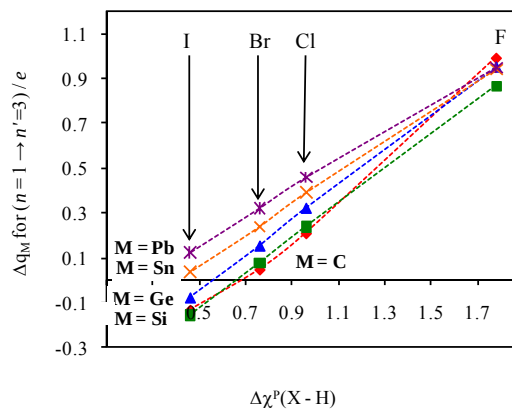
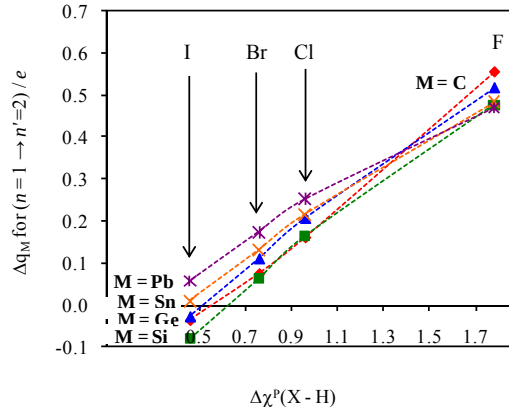
Sample plots of the equation ($[q_{M(n')} - q_{M(n)}] = a[\chi_{\Delta'} - \chi_{\Delta}] + b$) [Figure S.5]

MP2 (NBO)

Basis Set: 6-311+G*

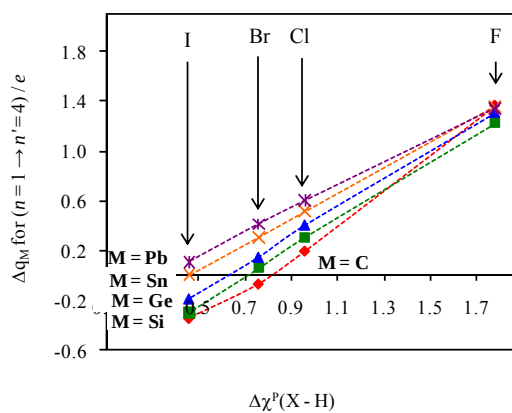
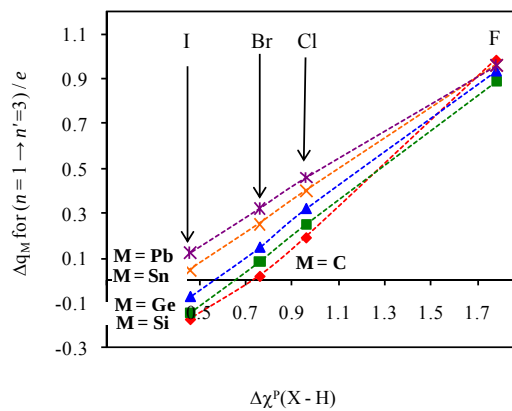
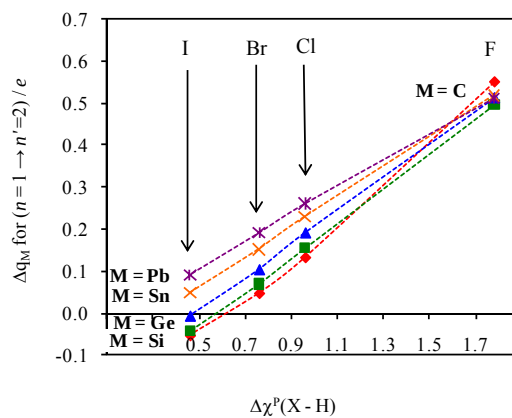


Basis Set aug-cc-pVTZ

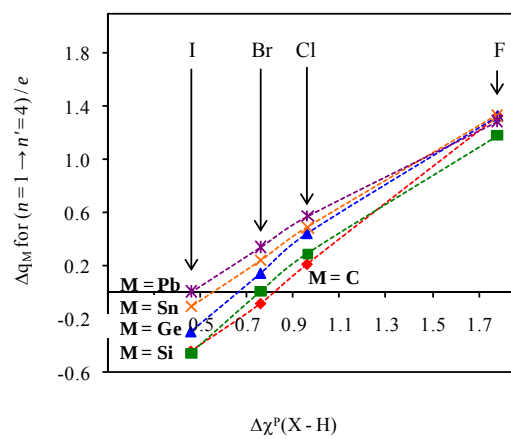
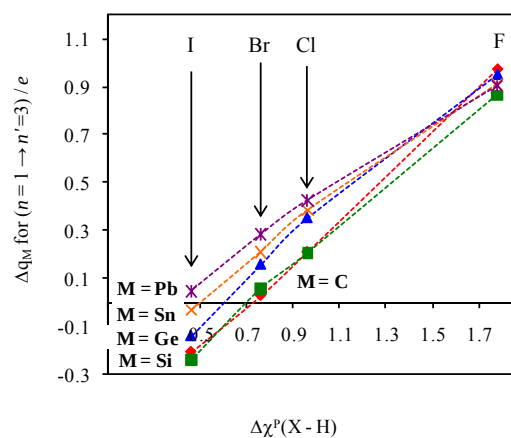
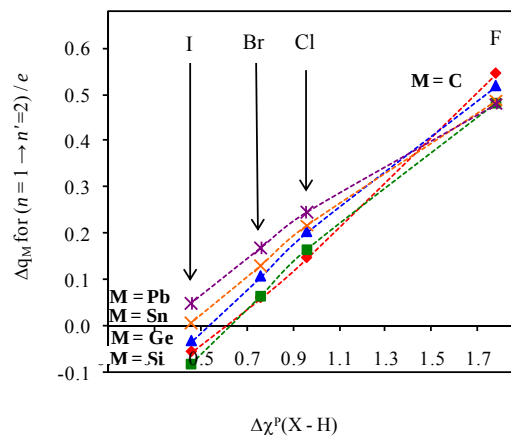


CCSD (NBO)

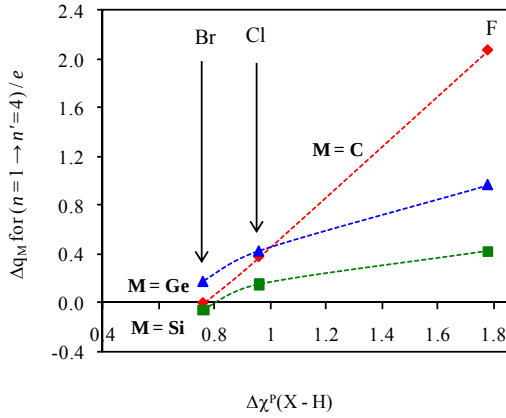
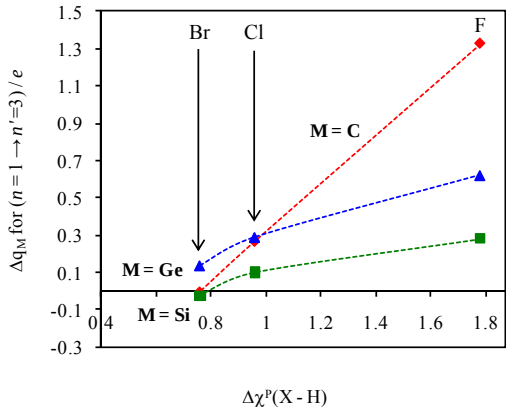
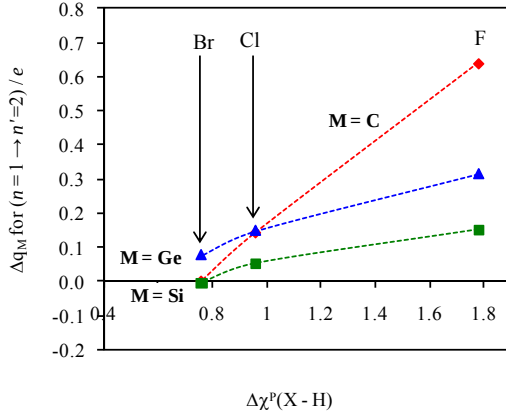
Basis Set: 6-311+G*



Basis Set aug-cc-pVTZ

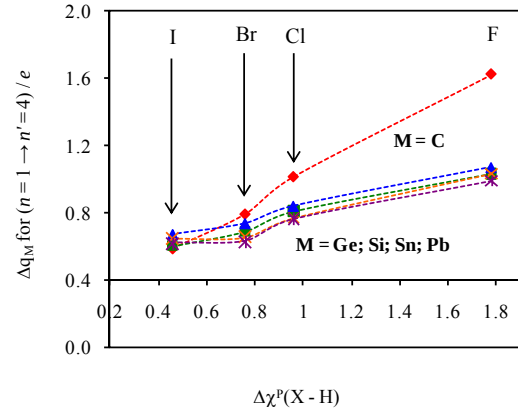
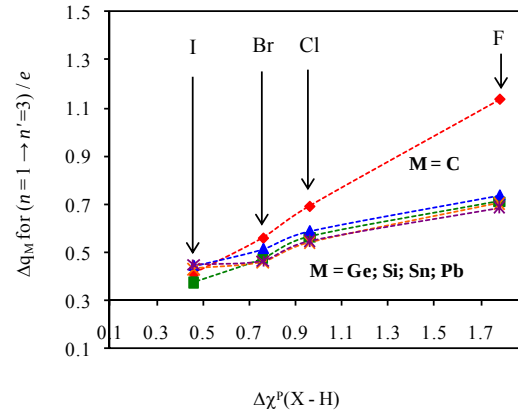
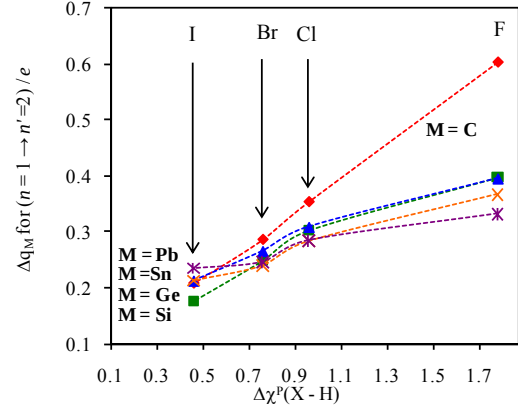


MP2 (AIM) Basis Set: 6-311+G*



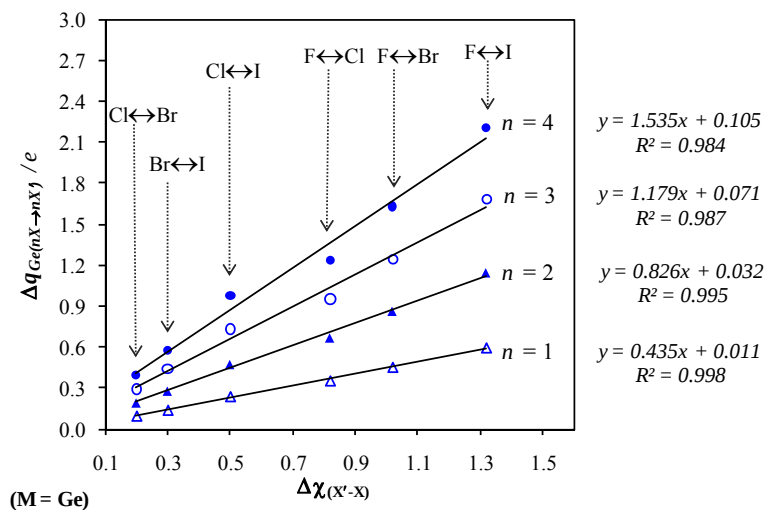
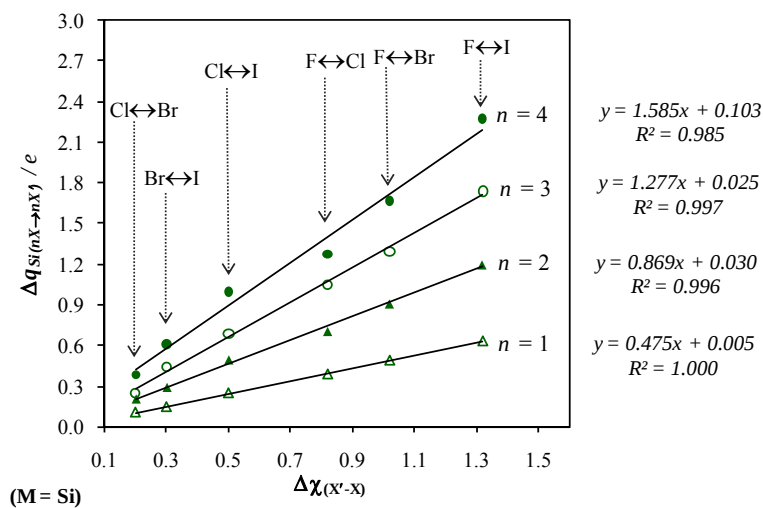
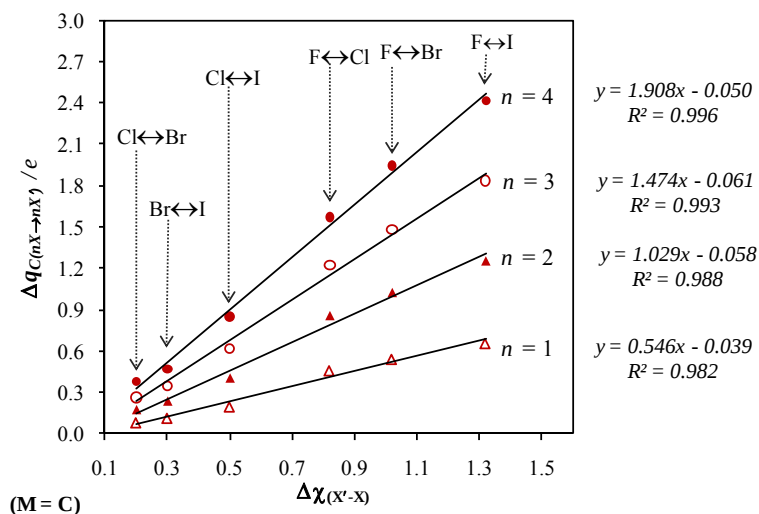
Δn	M	a	b	R^2
1	C	0.619	-0.461	0.999
2	C	1.303	-0.987	1.000
3	C	2.052	-1.577	1.000
1	Si	0.143	-0.101	0.967
2	Si	0.278	-0.205	0.949
3	Si	0.429	-0.329	0.936
1	Ge	0.224	-0.082	0.988
2	Ge	0.458	-0.185	0.984
3	Ge	0.742	-0.339	0.984

MP2 (APT) Basis Set: 6-311+G*



Δn	M	a	b	R^2	M	a	b	R^2
1	C	0.303	0.062	0.999	Sn	0.118	0.158	0.985
2	C	0.552	0.154	0.999	Sn	0.213	0.322	0.979
3	C	0.790	0.223	0.997	Sn	0.310	0.467	0.959
1	Si	0.161	0.122	0.953	Pb	0.076	0.200	0.953
2	Si	0.247	0.288	0.959	Pb	0.189	0.350	0.965
3	Si	0.329	0.456	0.983	Pb	0.300	0.451	0.957
1	Ge	0.135	0.162	0.974				
2	Ge	0.223	0.351	0.983				
3	Ge	0.310	0.520	0.990				

Sample plots of the equation ($[q_{M(n)} - q_{M(n)}] = a[\chi_{A'} - \chi_A] + b$), for A = X. See caption on next page.



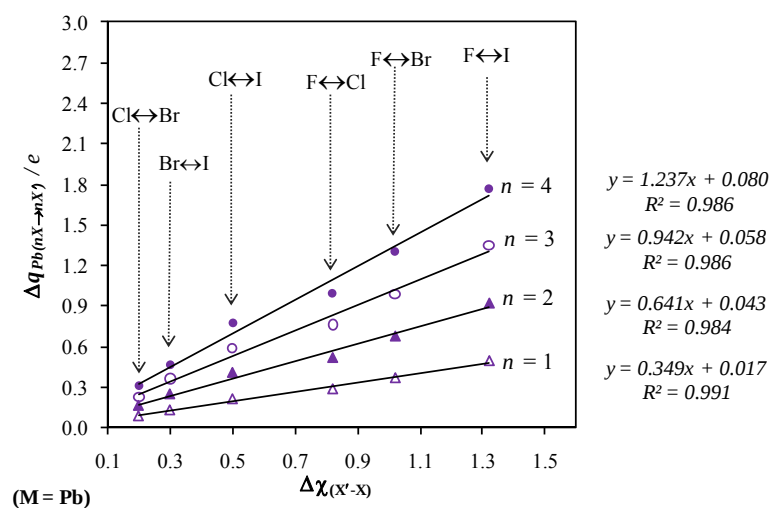
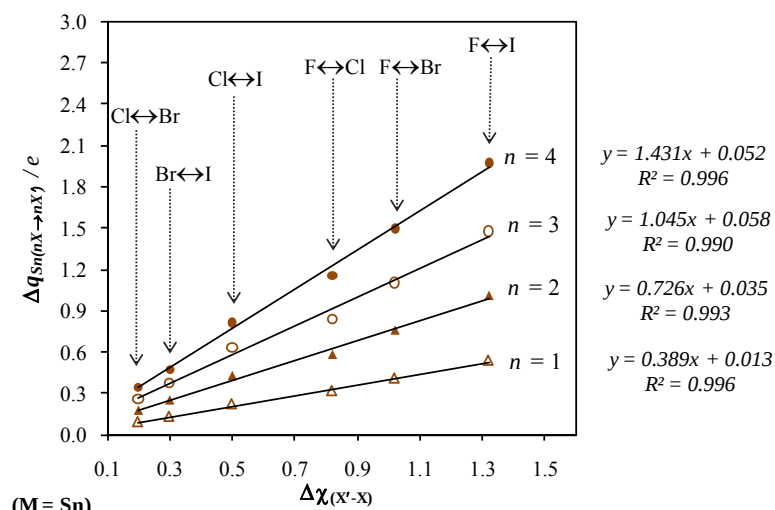
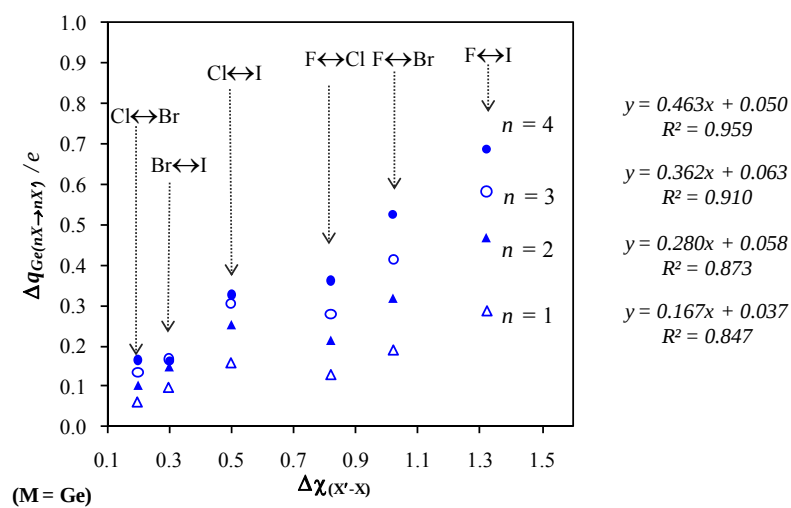
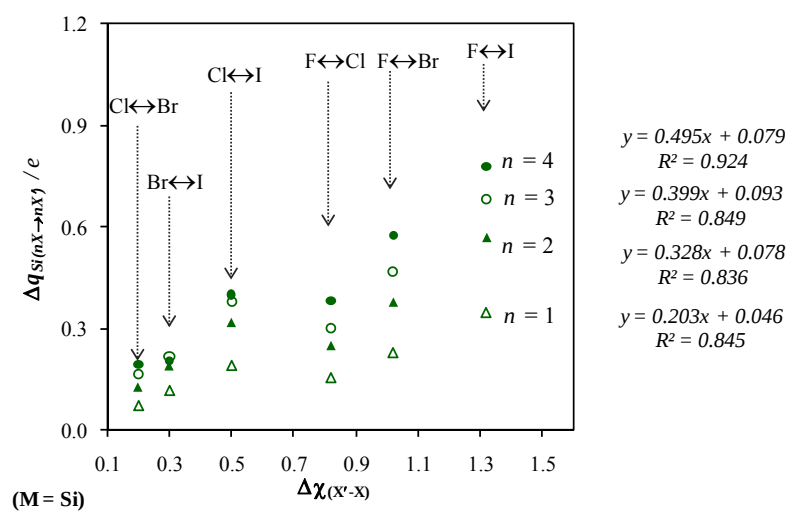
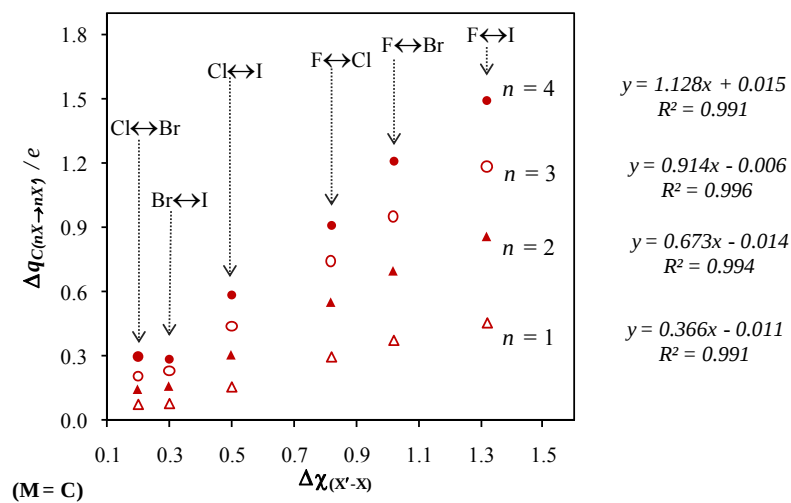


Figure S.6: $\Delta q_{M(nX \rightarrow nX')}$ vs. $\Delta \chi_{(X'X)}$ going from $MH_{4-n}X_n$ to $MH_{4-n}X'_n$ for $M = C, Ge, \text{ and } Pb$ for the CCSD/aug-cc-pVTZ (NBO) q_M data ($M = Si, \text{ and } Sn$ are included in the supporting information).



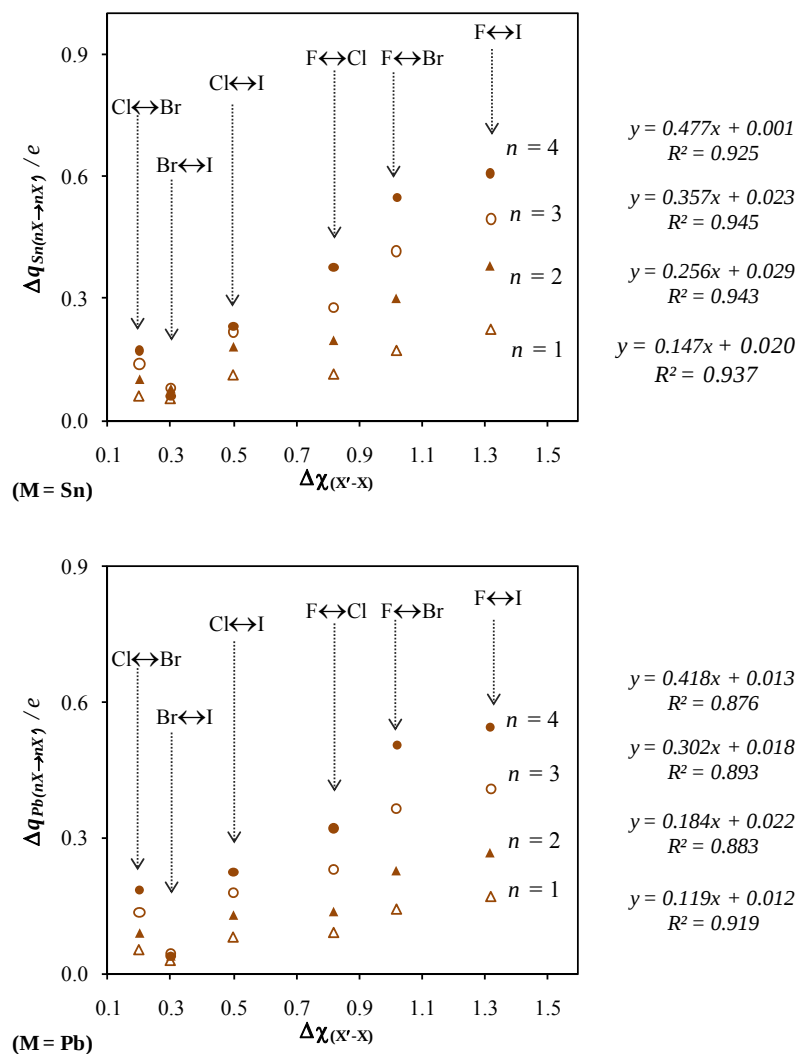


Figure S.7: $\Delta q_{M(nX \rightarrow nX')}$ vs. $\Delta \chi_{(X'X)}$ going from $MH_{4-n}X_n$ to $MH_{4-n}X'_n$ for $M = \text{C}, \text{Ge},$ and Pb for the MP2(full)/6-311+G* (APT) q_M data ($M = \text{Si},$ and Sn are included in the supporting information).

NOTE: The linear trends are observed for the NBO data, but they are unreliable for the APT values. A linear dependence is evident for $M = \text{C}$ above, but the situation deteriorates thereafter: (we have excluded the best fit lines to allow for careful examination of the more scattered APT values. The $\text{Br} \rightarrow \text{I}$ data points are particularly low for $M = \text{Sn}$ and Pb (relative to the rest of the values) due to the similarity of APT q_M values for $X = \text{Br}$ and I for each n . We do not consider the AIM data here since we have been able to include the iodides in our AIM calculations.

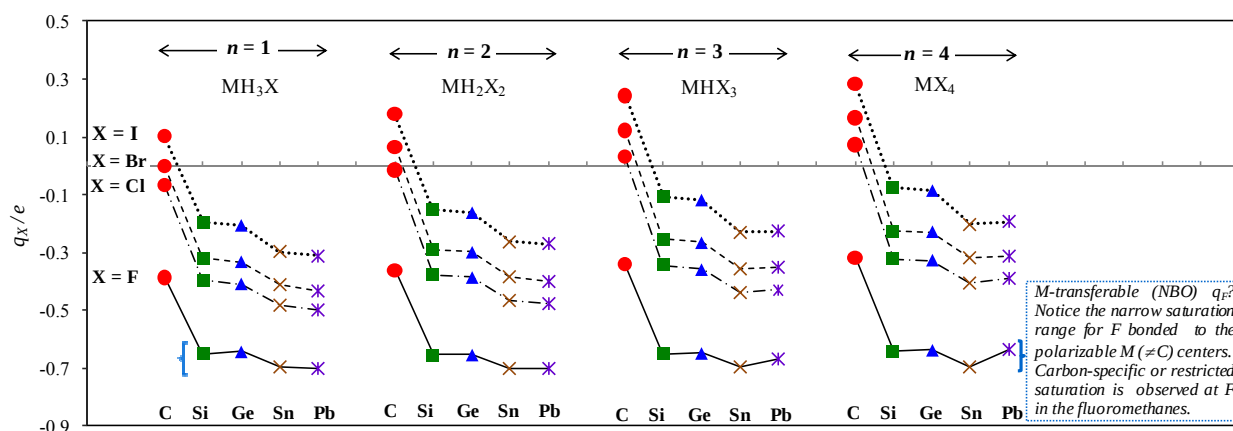


Figure S.8a: NBO point charges obtained at the CCSD level of theory for the halides, q_X . For the elements preceding Sn in the periodic table, the aug-cc-pVTZ basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

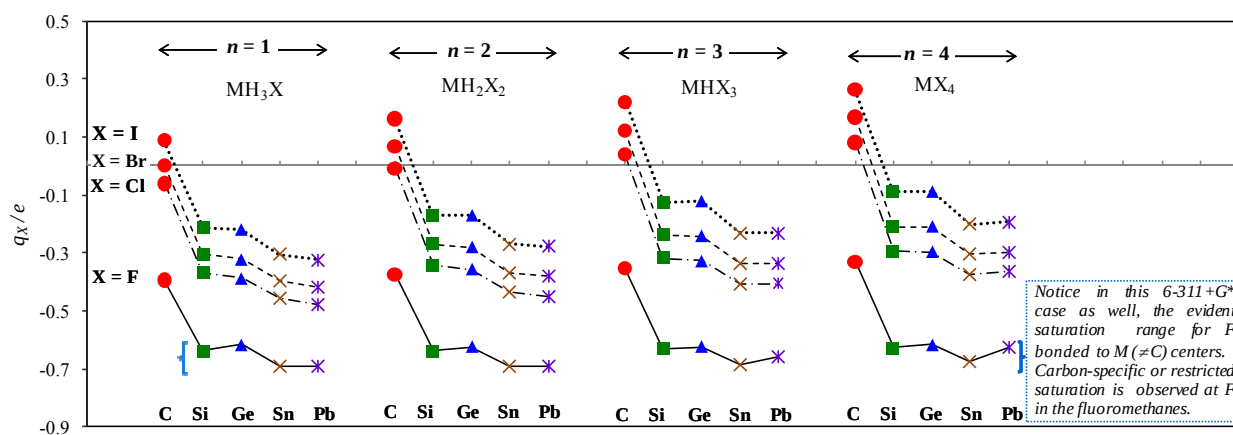


Figure S.9b: NBO point charges obtained at the CCSD level of theory for the halides, q_X . For the elements preceding Sn in the periodic table, the 6-311+G* basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section. The differences in the data for the two basis sets (cf. Figure S.6a above) are hardly perceptible.

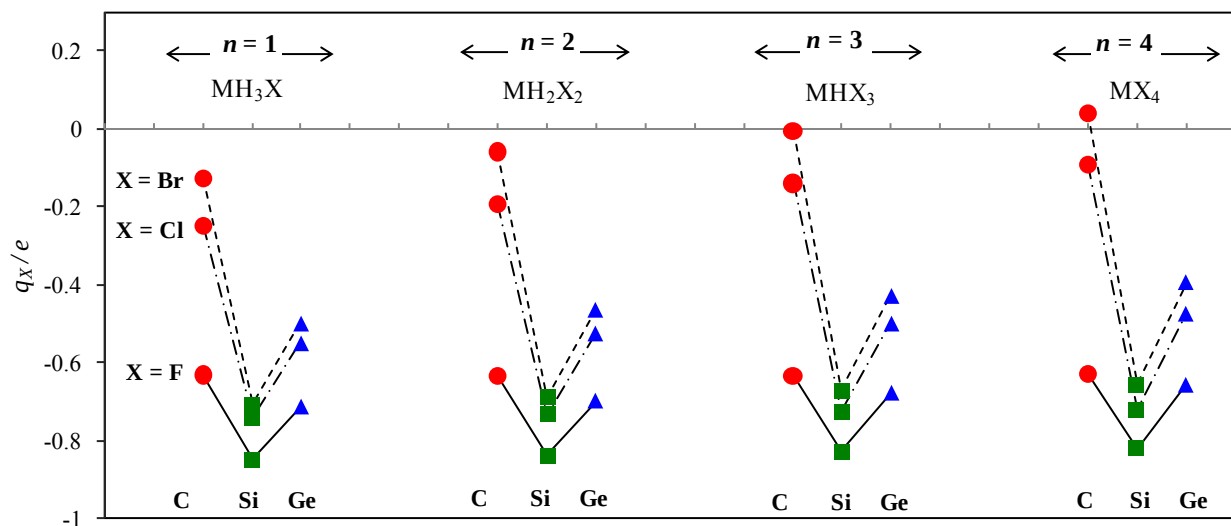


Figure S.10: AIM point charges obtained at the MP2(full) level of theory for the halides, q_X . For the elements preceding Sn in the periodic table, the 6-311+G* basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.

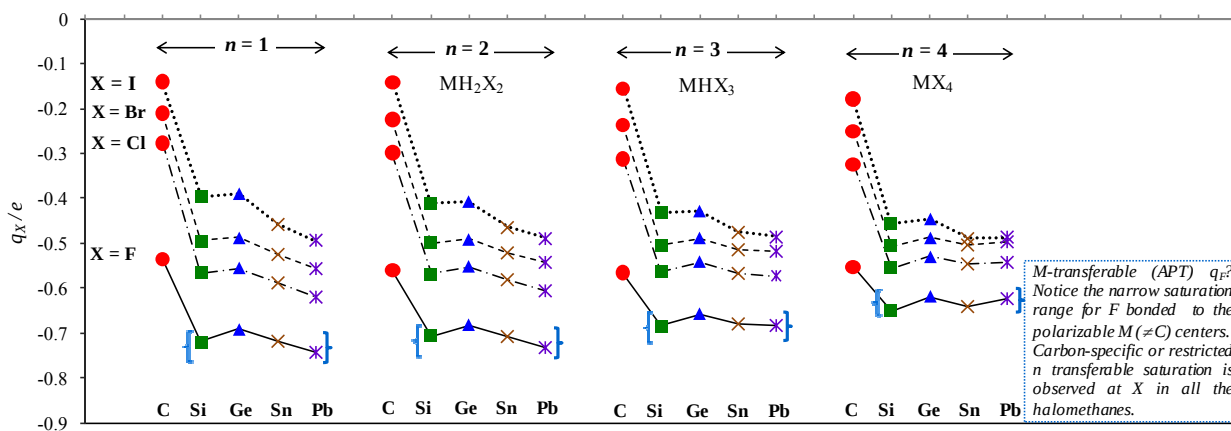


Figure S.11: APT point charges obtained at the MP2(full) level of theory for the halides, q_X . For the elements preceding Sn in the periodic table, the 6-311+G* basis sets were used. For the heavier elements, the basis sets and ECPs are given in the methods section.