SUPPORTING INFORMATION

"Reported, but still unknown." A closer look into 3,4-*bis*- and 3,4,5-*tris*(trifluoromethyl)pyrazoles.

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1. Copies of NMR spectra



Date: 26-Jan-2011

Solvent: dmso

SW: 32680 Hz



TE: 681 K

AQ: 1.57 sec, RD: 0.00 sec

Parameter file, XWIN-NMR⊓⊓Version 3.5⊓

DY2234-F19.fid



| МЧЧ | 0 | -10 | -20 | -30 | -40 | -50 | -60 | -70 | -80 | -90 | -100 | -110 | -120 | -130 | -140 | -150 | -160 | -170 | -180 | -190 | -200 |
|-----|-------------------|----------|----------|-----------|-----|-----|------------|---------|-----|------------------|----------|------|------|----------|----------------------------|------|------|------------|------|------|------|
| | | File nam | e: DY223 | 4-F19.fid | | | Opera | itor: | | SF: 376.2981 MHz | | | | NSC: 0 | | P | | SI: 131072 | | | |
| | Date: 27-Jan-2011 | | | | | | Solvent: d | lmso-d6 | | S | N: 82816 | Hz | Т | E: 293 K | AQ: 0.77 sec, RD: 0.00 sec | | | | | | П |







| <u>a</u> | 130 | 100 | 170 | 100 | 150 | 140 | 150 | 120 | 110 | 100 | 30 | 00 | 10 | 00 | 50 | 40 | 50 | 4 | 0 | 10 | |
|----------|--------------|-----------|-----|----------|---------|-------|------------|------|----------|-----|-------------|-------------|-------|----|---------|-------------|-----------|-------|-----------|----|--|
| | File name: R | ₹523525-C | 13 | Operator | ∵ root⊓ | SF: 1 | 25.7422 N | /IHz | NSC: 284 | | PW: 0.00 u | sec, RG: 5 | 1200 | | | S | 81: 65536 | | | | |
| | Date: 21- | Jan-2011 | | Solvent: | dmso | SM | /: 32680 H | lz | TE: 0 K | / | AQ: 1.57 se | ec, RD: 0.0 | 0 sec | | Paramet | er file, XV | VIN-NMR | ⊓⊓Ver | rsion 3.5 | iΠ | |



| Mdd | 0 | -10 | -20 | -30 | -40 | -50 | -60 | -70 | -80 | -90 | -100 | -110 | -120 | -130 | -140 | -150 | -160 | -170 | -180 | -190 | -200 |
|-----|----------------------------|-----|-----|-----|-----|-----|----------|---------|-----|------------------|-----------|------|------|-----------|------|------|------------|------------|-------|------|------|
| | File name: R523525-F19.fid | | | | | | Ope | rator: | | SF: 376.2981 MHz | | | | NSC: 0 | | F | | SI: 131072 | | | |
| | Date: 21-Jan-2011 | | | | | | Solvent: | dmso-d6 | | S | GW: 82816 | ∂ Hz | | TE: 293 K | | AQ | : 0.77 sec | , RD: 0.0 |) sec | | Π |













Wdd

-60.17 -63.01

n7-F19



| | | | | | | | | L | | | | | | | | | | | | | |
|-----|-------------------|----------|--------|-----|-----|-------------|------------------|-----|-----------|----------|------|------|-------|----------------------------|------|----------|------------|-------|------|------|------------|
| МЧЧ | 0 | -10 | -20 | -30 | -40 | -50 | -60 | -70 | -80 | -90 | -100 | -110 | -120 | -130 | -140 | -150 | -160 | -170 | -180 | -190 | -200 |
| | Fi | le name: | n7-F19 | | C | Operator: | | | SF: 376.2 | 2973 MHz | | N | SC: 0 | | | PW: 3.00 |) usec, RG | G: 24 | | 5 | SI: 131072 |
| | Date: 06-Jun-2011 | | | | So | lvent: cdcl | : cdcl3 SW: 8281 | | | | | TE: | 293 K | AQ: 0.77 sec, RD: 0.00 sec | | | | | | П | |





| ۹. | 14.0 | 10.0 | 12.0 | 11.0 | 10.0 | 0.0 | 0.0 | 1.0 | 0.0 | 0.0 | 4.0 | 0.0 | 2.0 | 1.0 | 0.0 | |
|-------------------|-----------------|------|---------------|------|-------------|-------|---------|-------|----------------|----------|-----|----------|------------------|----------|----------|--|
| F | File name: n9-H | 11 | Operator: roc | ot⊓ | SF: 500.130 | 0 MHz | NSC: 1 | PW | : 0.00 usec, R | RG: 40 | | | SI: 327 | 68 | | |
| Date: 17-Oct-2009 | | 09 | Solvent: CDC | CI3 | SW: 8013 | 8 Hz | TE: 0 K | AQ: 2 | .04 sec, RD: (| 0.00 sec | | Paramete | er file, XWIN-NI | MR⊓⊓Vers | ion 3.5⊓ | |





| File name: n9-C13 | Operator: root⊓ | SF: 125.7578 MHz | NSC: 627 | PW: 0.00 usec, RG: 51200 | SI: 65536 |
|-------------------|-----------------|------------------|-----------|----------------------------|--|
| Date: 10-Sep-2009 | Solvent: CDCl3 | SW: 28986 Hz | TE: 682 K | AQ: 1.77 sec, RD: 0.00 sec | Parameter file, XWIN-NMR⊓⊓Version 3.5⊓ |



°CF₃

| | 1 | | | | | | | | | | | | | | | | | | | | |
|-----|--------------|-------|---------------------|------------|-----|----------|-----------------------|-----|--------|-----|----------------------------|------------|-----------|------|------|--|------|----------|------|------|------|
| Mdd | -10 | -20 | -30 | -40 | -50 | -60 | -70 | -80 |) -9(| 0 - | 100 | -110 | -120 | -130 | -140 | -150 | -160 | -170 | -180 | -190 | -200 |
| | File name: n | 9-F19 | Oper | ator: psm⊓ | | SF: 470. | 5926 MHz | | NSC: 8 | 3 | F | PW: 0.00 เ | usec, RG: | 512 | | | | SI: 6553 | 6 | | |
| | Date: 14-Aug | 2009 | Solvent: C6D6 SW: ! | | | | /: 99010 Hz TE: 679 K | | | | AQ: 0.66 sec, RD: 0.00 sec | | | | | Parameter file, XWIN-NMR⊓⊓Version 3.5⊓ | | | | | |









2. Determinations of pK_a values of pyrazoles 4 and 5.

pK_a values of pyrazoles **4** and **5** were determined by potentiometric titration using an automatic pHmeter with glass electrode. To measure ionization constants a certain amount of the sample was dissolved in a water-acetonitrile solution containing 0.15 M KCl to adjust the ionic strength. Titrations of the compounds were performed in aqueous-acetonitrile mixture with the percentage ratio of the solvents ranging from 80:20 to 35:65, respectively. The solutions were titrated with 0.07 M KOH at a constant temperature of 23 °C. On the basis of the obtained potentiometric curves (Figure 1) the ionization constants in the aqueous-acetonitrile solutions (ps K_a values) were calculated. Aqueous p K_a values were obtained by extrapolation to 100% water using a linear relationship between ps K_a value and acetonitrile concentration (Figure 2).^{1,2}

The obtained pK_a values are summarized in Table 1.



Table 1. Ionization constants of pyrazoles 4 and 5 in aqueous solution.



Figure 1. Potentiometric titration curves for pyrazole **5** (a) and pyrazole **4** (b). $C_{BH} = 0.01$ M, $C_{NaOH} = 0.07$ M, C_{H_2O} : C_{CH_2O} (%) = 80:20 (2). $V_0 = 20$ mL.

Figure 2. Determination of ionization constants of pyrazole 5 (1) and pyrazole 4 (2) in aqueous solution.

- 1. Albert, A.; Sergeant, E. P. Ionization constants of acids and bases, Wiley, Inc., New York, 1962.
- 2. Denesh, I. Titration in non-aqueous media [Russian translation], Mir, Moscow, 1971.

3. Fluorescence measurements.

Fluorescent properties of compounds **4** and **5** were investigated in DMSO solution. The organic solvent was of spectroscopic grade. Fluorescence emission spectra were measured at LS55 spectrofluorometer The spectra were recorded within the range 300-600 nm. The data collection frequency was 0.5 nm. The excitation and emission slits were 15 and 20 nm, respectively. The excitation wavelength was 277 nm. The spectra were obtained after subtracting the background of the solvent. The fluorescence maxima agreed within 0.5-1.0 nm in independent parallel measurements (Figure 1).

Figure 1. Fluorescence emission spectra of pyrazole 5 (1) and pyrazole 4 (2) in DMSO solution. $C_{sub}=5\cdot10^{-4}$ M. $\lambda^{exi}=277$ nm. L = 1cm, slits 15; 20.