

^1H and ^2H NMR spectroscopic characterization of heterobinuclear ion pairs formed upon the activation of bis(imino)pyridine vanadium(III) pre-catalysts with $\text{AlMe}_3/[\text{Ph}_3\text{C}]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ and MAO

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Supporting information

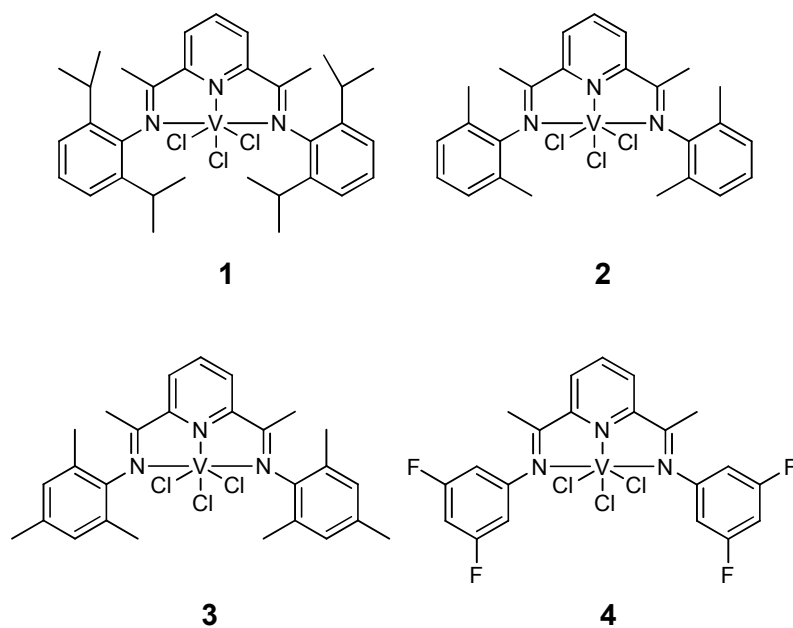


Chart S1.

^1H NMR spectra of the complexes 1-4 in CDCl_3 at 25 °C.

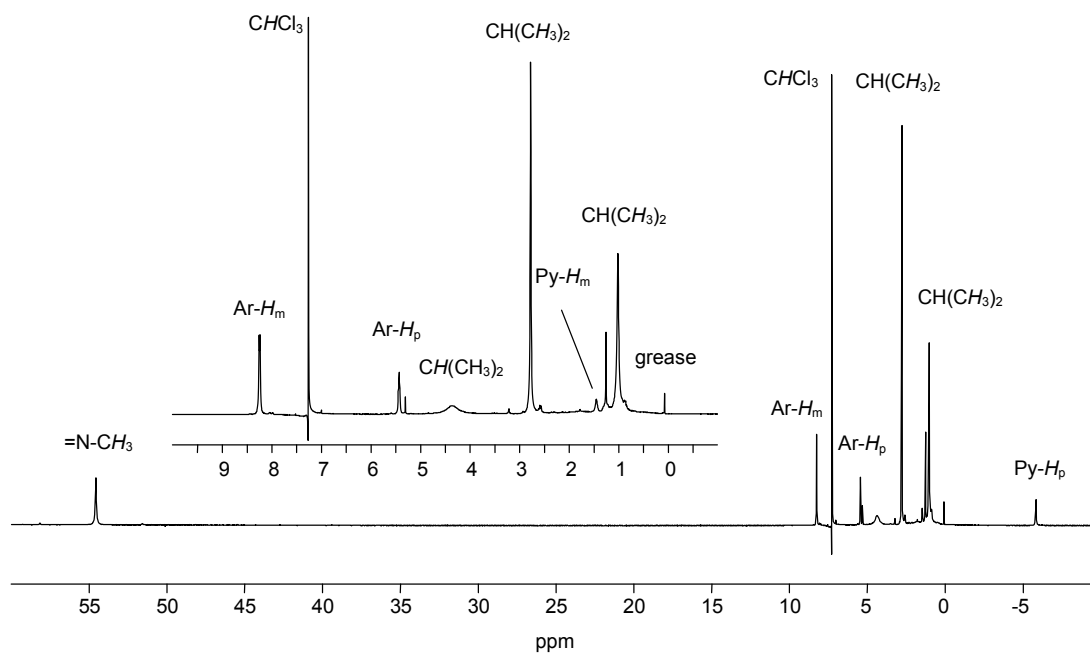


Figure S1. ^1H NMR spectrum of **1** in CDCl_3 at 25 °C.

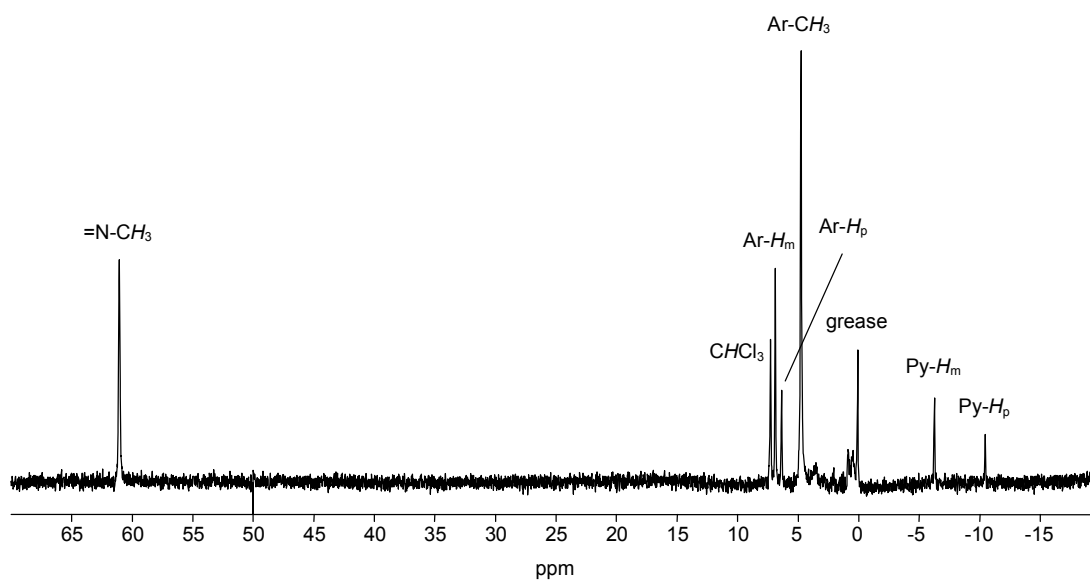


Figure S2. ^1H NMR spectrum of **2** in CDCl_3 at 25 °C.

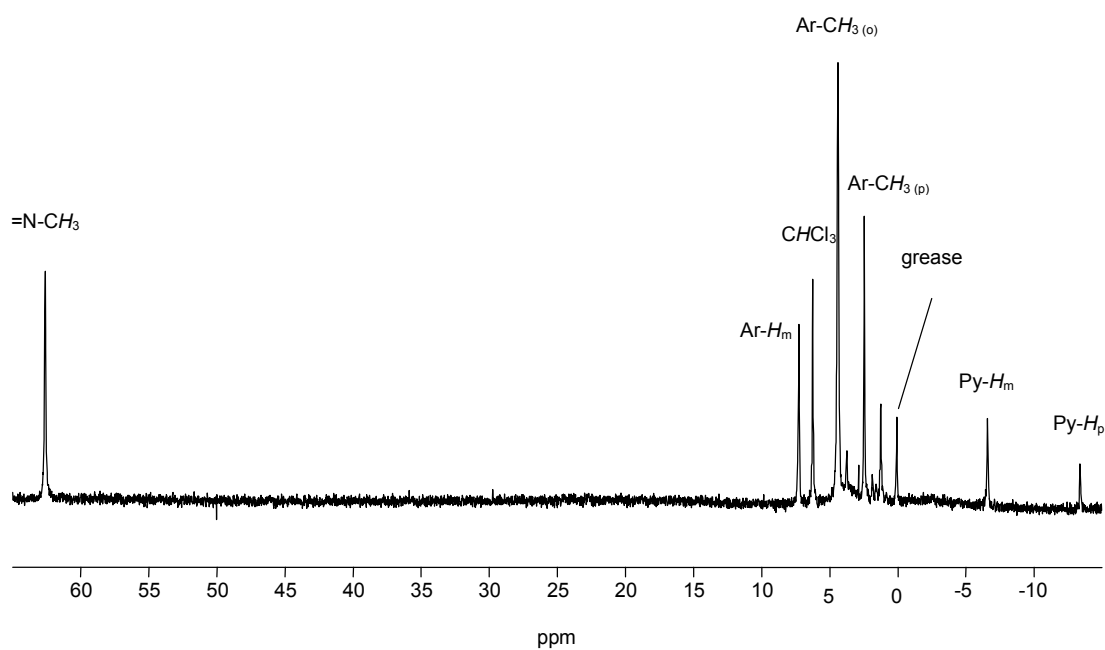


Figure S3. ^1H NMR spectrum of **3** in CDCl_3 at 25°C .

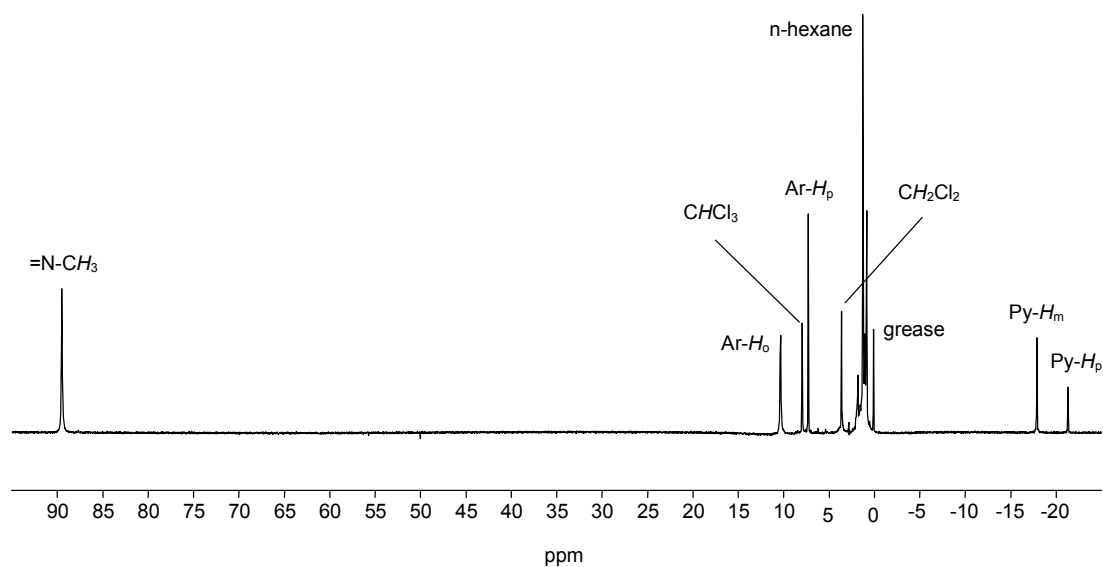


Figure S4. ^1H NMR spectrum of **4** in CDCl_3 at 25°C .

^1H NMR spectra of the samples **1-4**/ AlMe_3 / $[\text{Ph}_3\text{C}]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ in $\text{toluene}-d_8$ at -20°C .

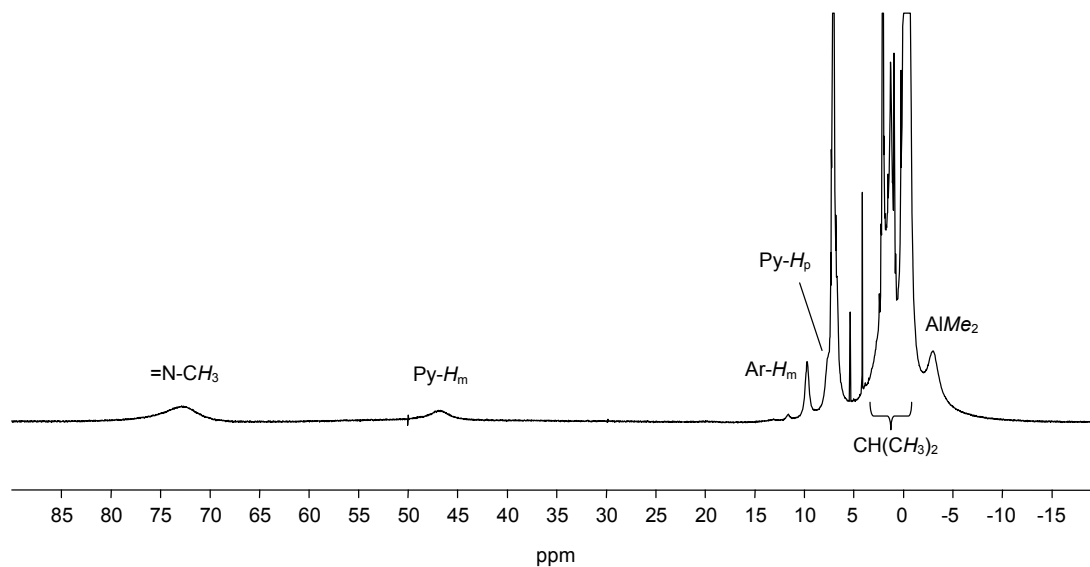


Figure S5. ^1H NMR spectrum (−20 °C, toluene- d_8) of the sample **1**/AlMe₃/[Ph₃C]⁺[B(C₆F₅)₄][−] ([V]:[Al]:[B] = 1:10:1.2).

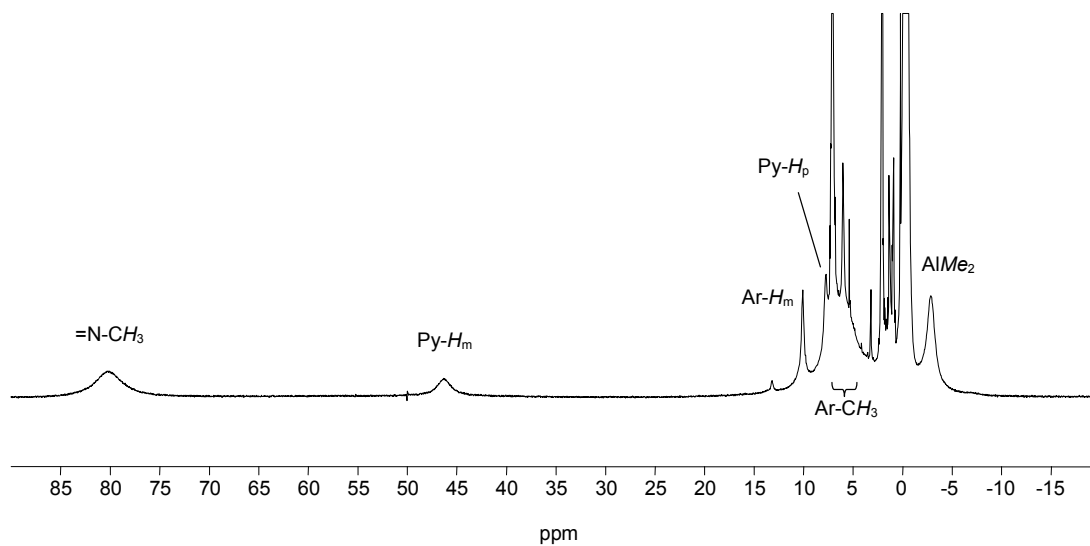


Figure S6. ^1H NMR spectrum (−20 °C, toluene- d_8) of the sample **2**/AlMe₃/[Ph₃C]⁺[B(C₆F₅)₄][−] ([V]:[Al]:[B] = 1:10:1.2).

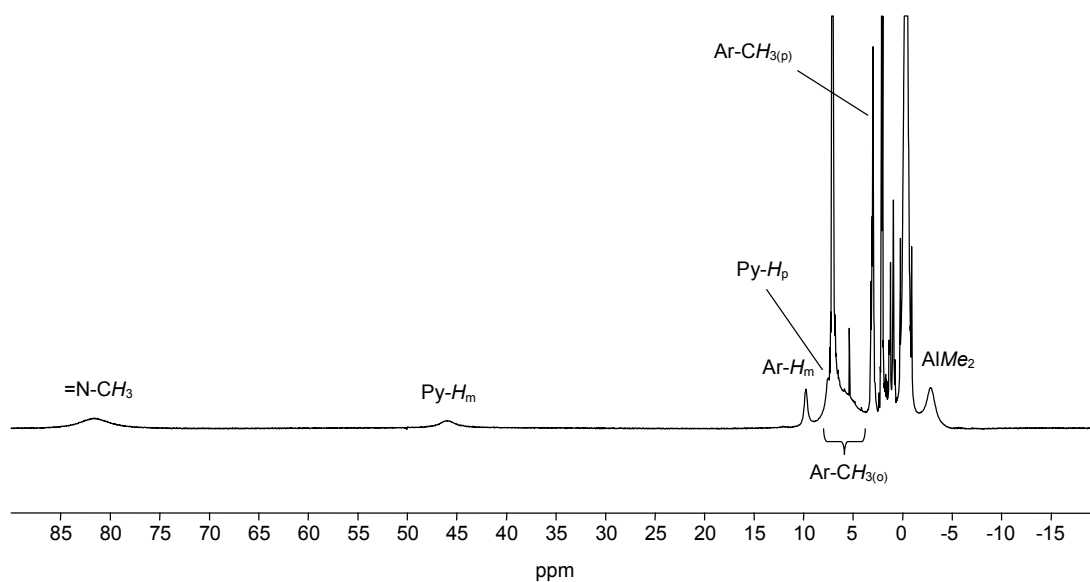


Figure S7. ^1H NMR spectrum ($-20\text{ }^\circ\text{C}$, toluene- d_8) of the sample **3**/ AlMe_3 / $[\text{Ph}_3\text{C}]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ ($[\text{V}]:[\text{Al}]:[\text{B}] = 1:10:1.2$).

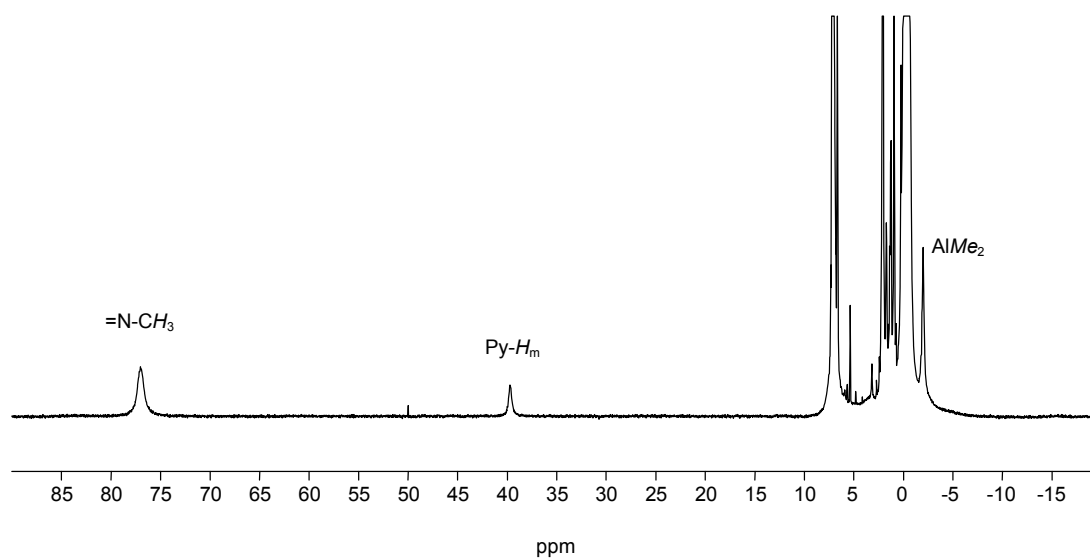


Figure S8. ^1H NMR spectrum ($-20\text{ }^\circ\text{C}$, toluene- d_8) of the sample **4**/ AlMe_3 / $[\text{Ph}_3\text{C}]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ ($[\text{V}]:[\text{Al}]:[\text{B}] = 1:10:1.2$).

¹H NMR data (δ, ppm and Δv_{1/2}, Hz) for the complexes 1b-4b observed in the systems 1-4/AlMe₃/[Ph₃C]⁺[B(C₆F₅)₄]⁻ ([V]:[Al]:[B] = 1:10:1.2) at -20 °C.

Table S1. ¹H NMR data for the complex 1b (toluene-*d*₈, -20 °C).

		=N-CH ₃	Py-H _m	Py-H _p	Ar-H _m	Ar-H _p	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	AlMe ₂
1	δ	65.2	-8.3	-2.5	8.6	5.5	3.4	1.1	5.9
	Δv _{1/2}	(80)	(30)	(40)	(20)	(20)	(15)	(25)	(120)
1b	δ	72.9	47.0	7.7	9.7	N/O ^a	~2 ^b	N/O ^a	-3.0
	Δv _{1/2}	(1370)	(800)	(~200)	(170)	-	(~700)	-	(530)

^a N/O – not observed.

^b The signal was observed using inversion-recovery ¹H NMR experiment with 10 ms delay between 180° and 90° pulses.

Table S2. ¹H NMR data for the complex 2b (toluene-*d*₈, -20 °C).

		=N-CH ₃	Py-H _m	Py-H _p	Ar-H _m	Ar-H _p	Ar-CH ₃	AlMe ₂
2	δ	72.6	-8.3	-19.8	6.5	7.1	5.7	-
	Δv _{1/2}	(100)	(90)	(100)	(60)	(70)	(80)	-
2b	δ	80.2	47.0	7.7	10.1	6.0	~6 ^a	-2.9
	Δv _{1/2}	(1100)	(580)	(140)	(100)	(65)	(~1000)	(290)

^a The signal was observed using inversion-recovery ¹H NMR experiment with 10 ms delay between 180° and 90° pulses.

Table S3. ¹H NMR data for the complex 3b (toluene-*d*₈, -20 °C).

		=N-CH ₃	Py-H _m	Py-H _p	Ar-H _m	Ar-CH _{3(p)}	Ar-CH _{3(o)}	AlMe ₂
3	δ	74.9	-8.8	-23.9	5.8	1.6	5.1	-
	Δv _{1/2}	(90)	(40)	(60)	(20)	(20)	(40)	-
3b	δ	81.5	45.9	7.5	9.7	3.0	~5.8 ^a	-2.8
	Δv _{1/2}	(1400)	(740)	(180)	(130)	(40)	(~800)	(400)

^a The signal was observed using inversion-recovery ¹H NMR experiment with 10 ms delay between 180° and 90° pulses.

Table S4. ¹H NMR data for the complex 4b (toluene-*d*₈, -20 °C).

		=N-CH ₃	Py-H _m	Py-H _p	Ar-H _o	Ar-H _p	AlMe ₂
4	δ	109.9	-24.8	-32.0	12.5	9.7	-
	Δv _{1/2}	(140)	(50)	(70)	(60)	(20)	-
4b	δ	77.0	39.7	N/O ^a	N/O ^a	N/O ^a	-2.0
	Δv _{1/2}	(290)	(160)	-	-	-	(80)

^a N/O – not observed.

^1H NMR spectra of the complex $[\text{LAlMe}_2]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ in toluene- d_8 at 25 °C.

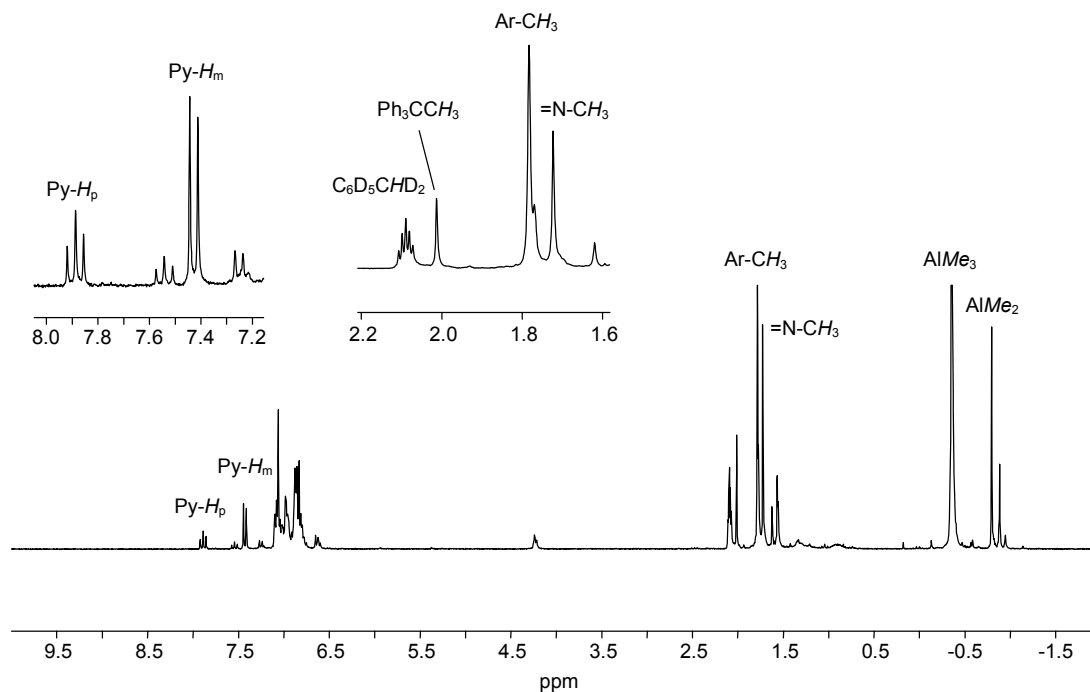


Figure S9. ^1H NMR spectrum (25 °C, toluene- d_8) of the complex $[\text{LAlMe}_2]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ { $\text{L} = [2,6-(\text{ArN}=\text{CMe})_2\text{C}_5\text{H}_3\text{N}]$, $\text{Ar} = 2,6\text{-Me}_2\text{C}_6\text{H}_3$ } independently prepared by the reaction of L with AlMe_3 and $[\text{Ph}_3\text{C}]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$.

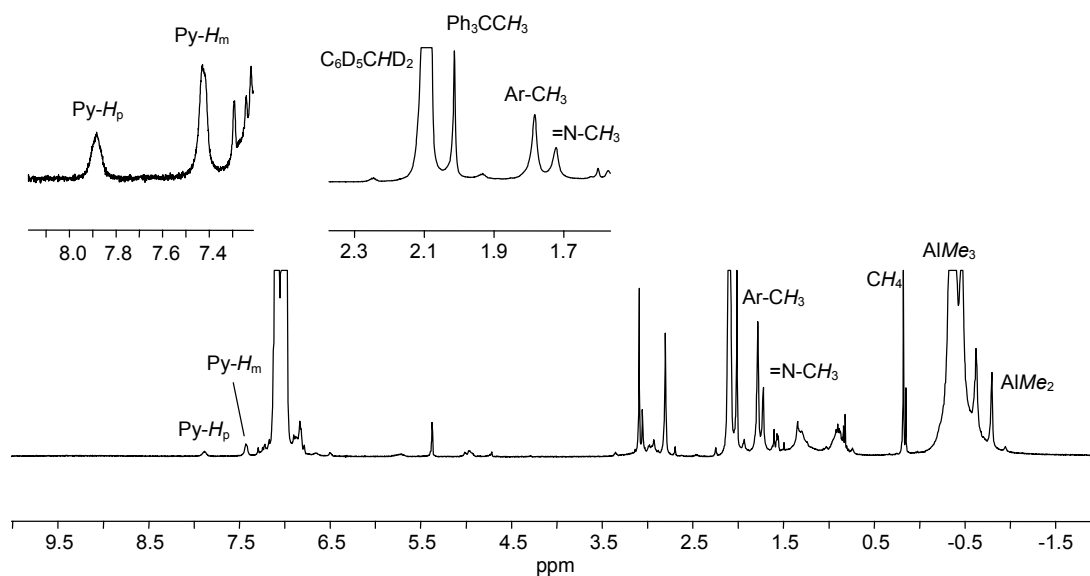


Figure S10. ^1H NMR spectrum (25 °C, toluene- d_8) of the complex $[\text{LAlMe}_2]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-$ { $\text{L} = [2,6-(\text{ArN}=\text{CMe})_2\text{C}_5\text{H}_3\text{N}]$, $\text{Ar} = 2,6\text{-Me}_2\text{C}_6\text{H}_3$ } observed in the catalyst system $2/\text{AlMe}_3/[\text{Ph}_3\text{C}]^+[\text{B}(\text{C}_6\text{F}_5)_4]^-/\text{C}_2\text{H}_4$ ($[\text{V}]:[\text{Al}]:[\text{B}] = 1:10:1.2$; $\text{N}(\text{V}):\text{N}(\text{C}_2\text{H}_4) = 1:50$).