

Influence of Urea on Shifting Hydrophilic to Hydrophobic Interactions of $\text{Pr}(\text{NO}_3)_3$, $\text{Sm}(\text{NO}_3)_3$, and $\text{Gd}(\text{NO}_3)_3$ with BSA in Aqueous Citric Acid: A Volumetric, Viscometric, and Surface Tension Study

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Table S1. Comparison of Density ρ , viscosity η , and Surface Tension γ , of Liquids with Literature values at $T = 298.15$ K and $p = 0.1$ MPa

	Literature	Experimental	Percentage deviation
1.27 mol·kg ⁻¹ aqueous sucrose			
$\rho/10^3 \text{ kg}\cdot\text{m}^{-3}$	1.127374 ¹	1.127287	0.01
$\eta/10^{-3} \text{ kg}\cdot\text{m}^{-1}\cdot\text{s}^{-1}$	2.7164 ¹	2.7268	0.38
DMSO			
$\gamma/\text{mN}\cdot\text{m}^{-1}$	42.70 ²	42.33	0.86

Table S2. Transfer Apparent Molar Volumes ($\Delta_{tr}V_\phi^0/10^{-6} \text{ m}^3\cdot\text{mol}^{-1}$) of $\text{Ln}(\text{NO}_3)_3$ from WC to WCB, WCU, and WCUB

	WCB-WC	WCU-WC	WCUB-WCU	WCU-WC
$\text{Pr}(\text{NO}_3)_3$	-101.43	-38.58	-76.48	-115.06
$\text{Sm}(\text{NO}_3)_3$	-112.26	-69.32	-67.43	-136.75
$\text{Gd}(\text{NO}_3)_3$	-105.58	-85.53	-62.90	-148.43

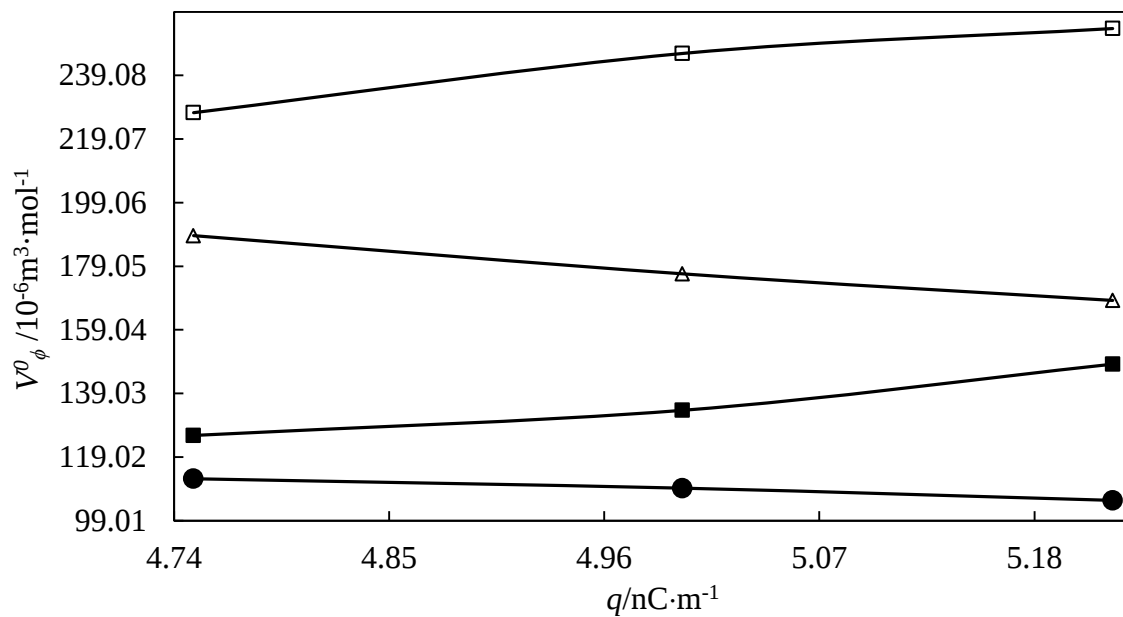


Figure S1. Effect of charge density of Ln^{3+} on limiting apparent molar volume, V_{ϕ}^0 , with WC ($\square$), WCU (Δ), WCB ($\blacksquare$), and WCUB ($\bullet$) at $T = 298.15$ K

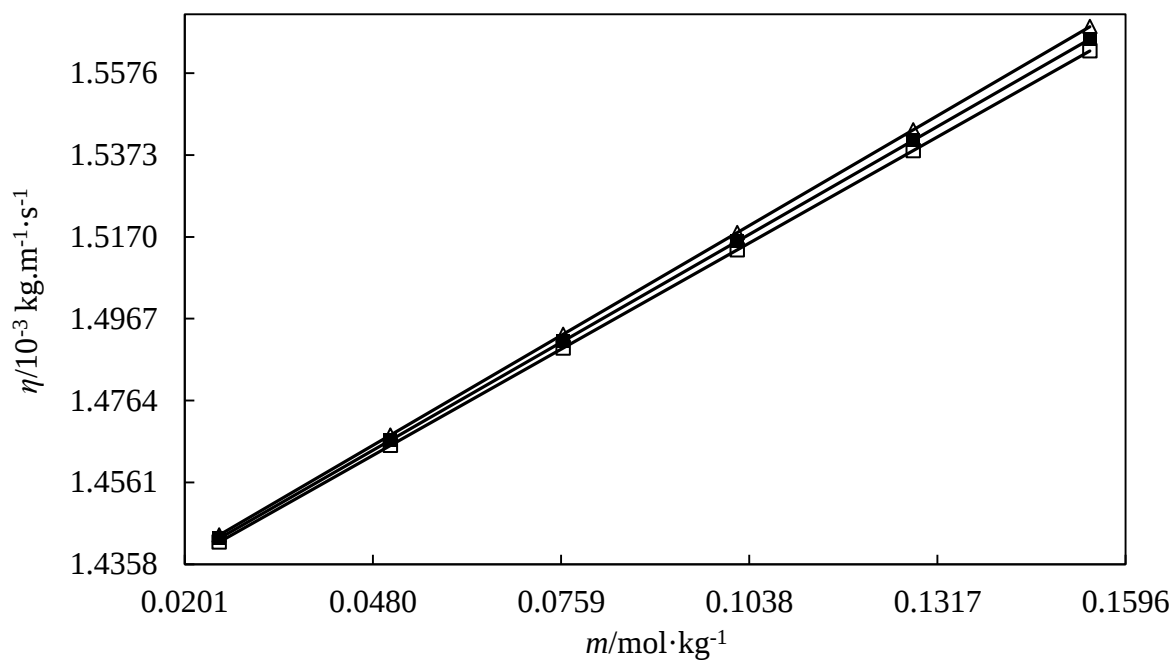


Figure S2. Effect of concentration on viscosity of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WC at 298.15 K

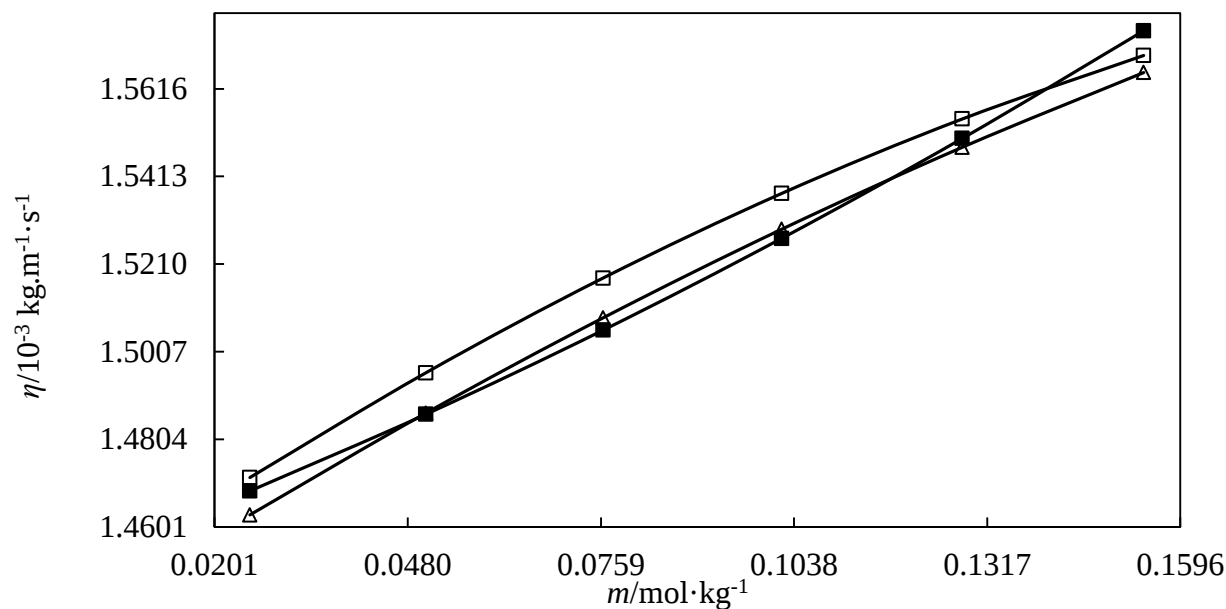


Figure S3. Effect of concentration on viscosity of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WCB at 298.15 K

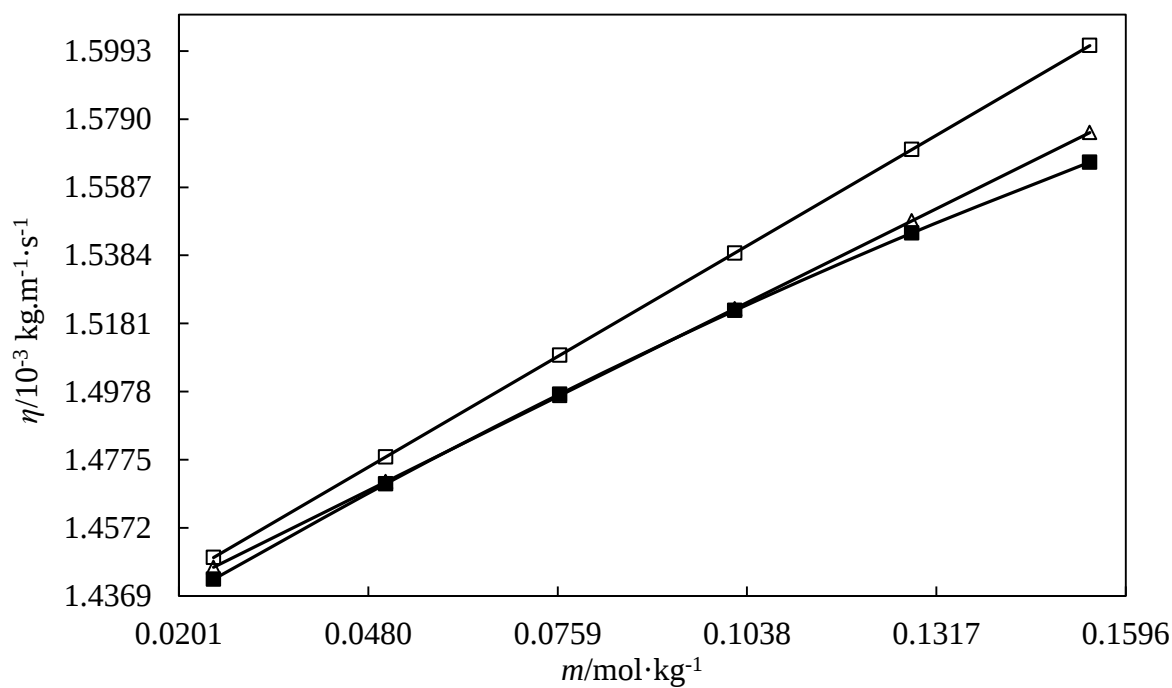


Figure S4. Effect of concentration on viscosity of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WCU at 298.15 K

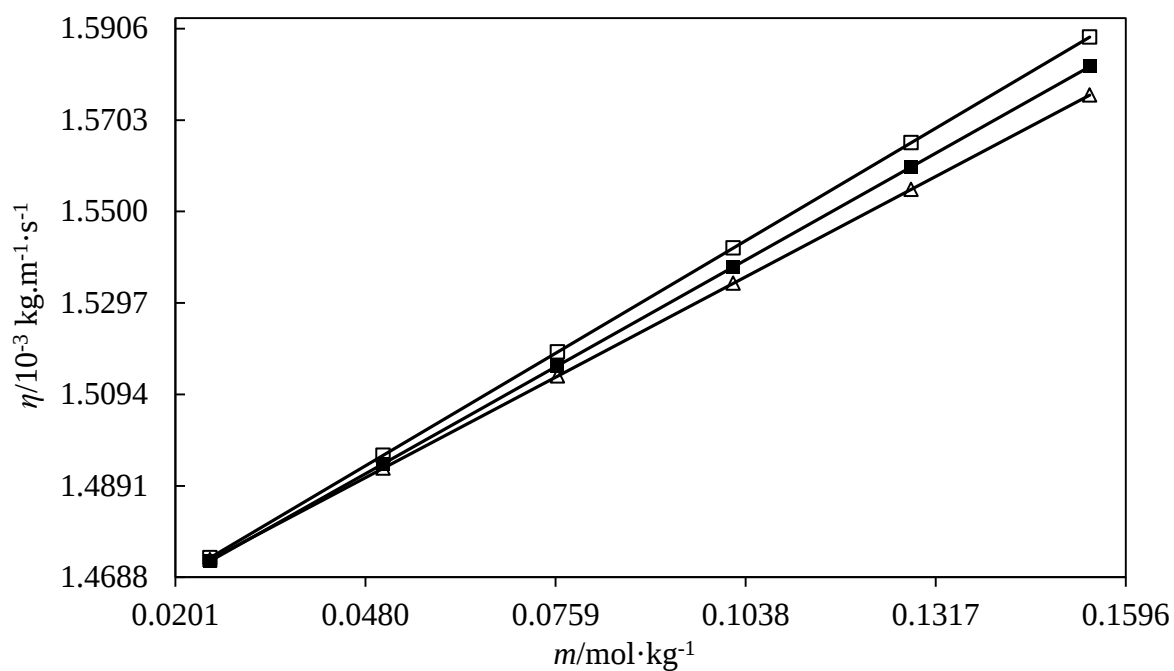


Figure S5. Effect of concentration on viscosity of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WCUB at 298.15 K

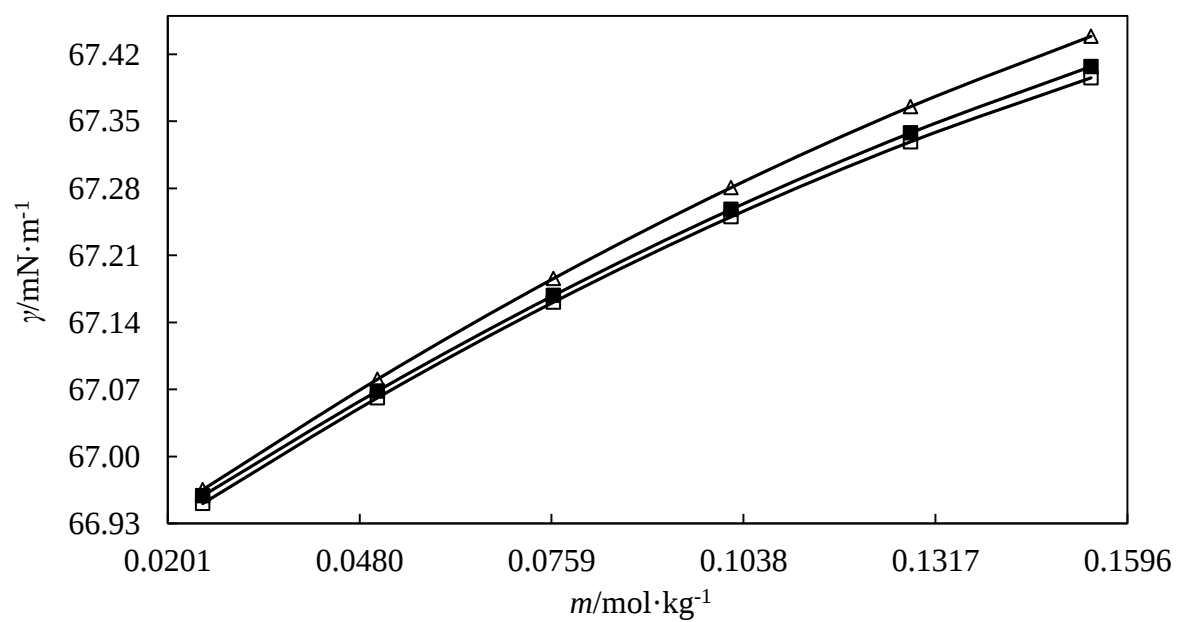


Figure S6. Effect of concentration on surface tension of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WC at 298.15 K

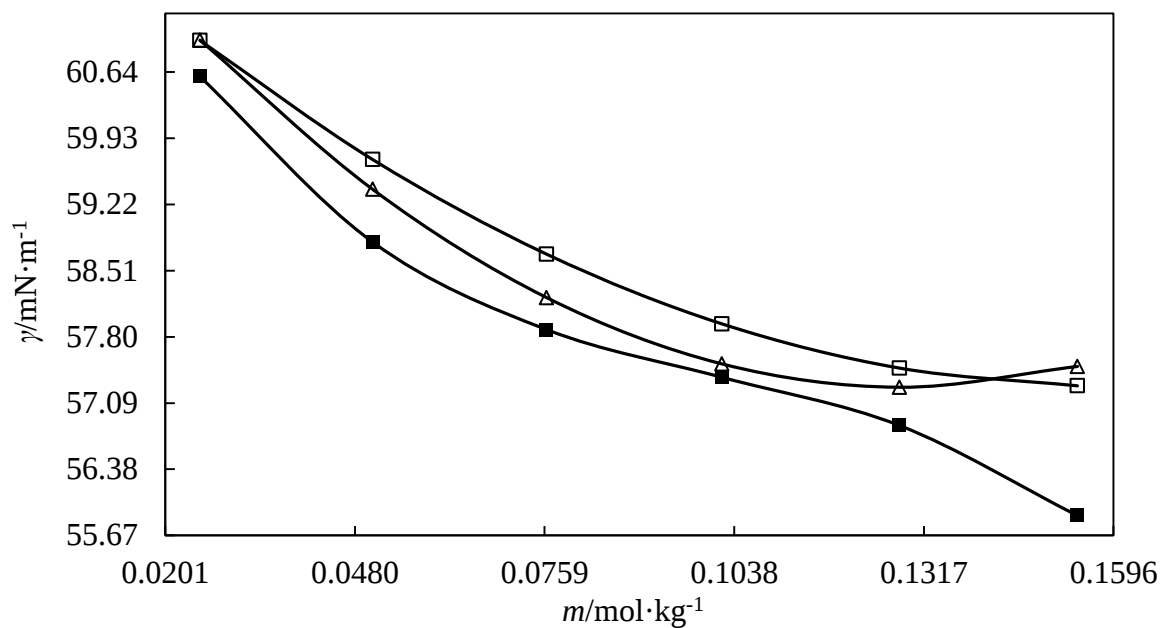


Figure S7. Effect of concentration on surface tension of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WCB at 298.15 K

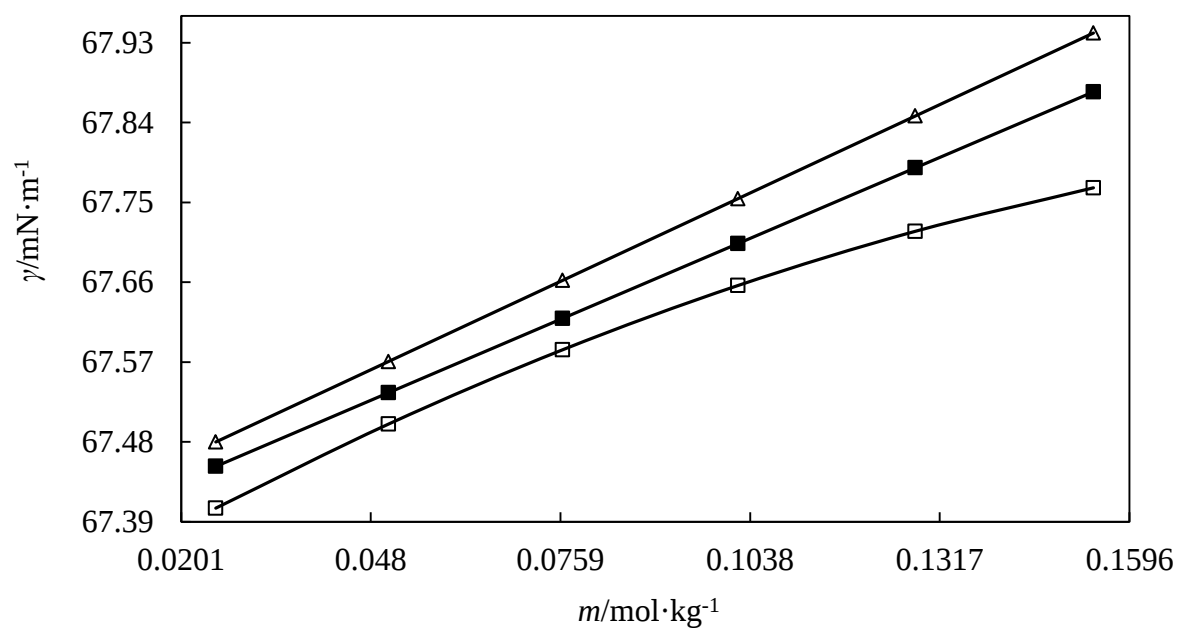


Figure S8. Effect of concentration on surface tension of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WCU at 298.15 K

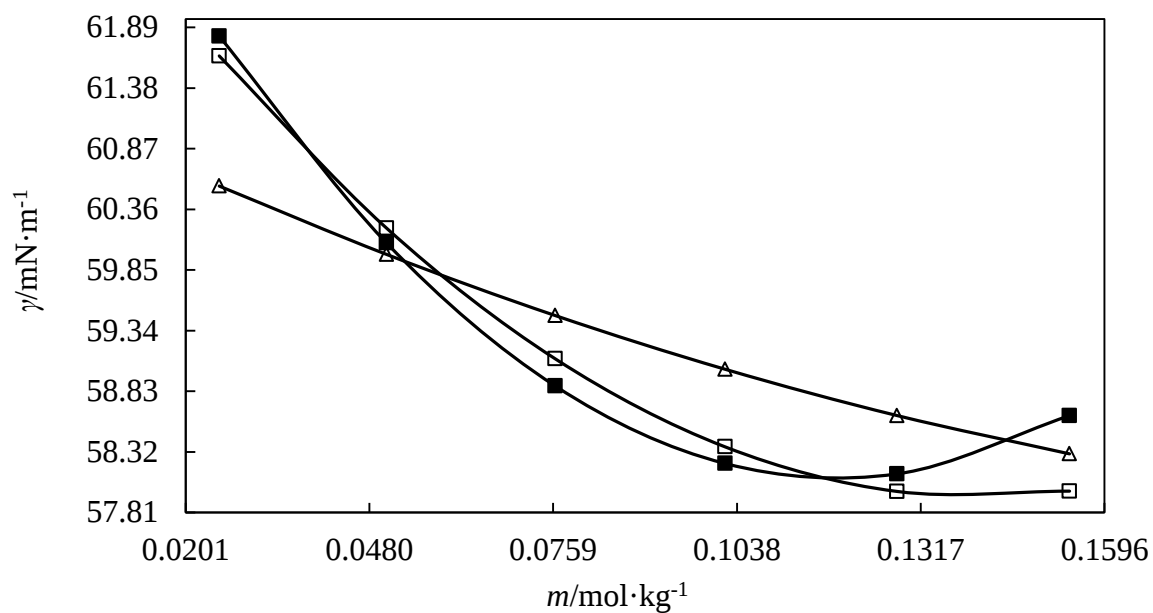


Figure S9. Effect of concentration on surface tension of $\text{Pr}(\text{NO}_3)_3$ ($\square$), $\text{Sm}(\text{NO}_3)_3$ ($\blacksquare$), and $\text{Gd}(\text{NO}_3)_3$ ($\triangle$) with WCUB at 298.15 K

■ REFERENCES

- (1) Riyazuddeen; Usmani, M. A. Density, speed of sound, and viscosities of (L-proline + aqueous glucose) and (L-proline + aqueous sucrose) solutions in the temperature range (298.15 to 323.15) K. *J. Chem. Eng. Data.* **2011**, 56, 3504-3509.
- (2) Markarian, S. M.; Terzyan, A. M. Surface tension and refractive index of dialkylsulfoxide + water mixtures at several temperatures. *J. Chem. Eng. Data.* **2007**, 52, 1704-1709.