Supporting Information

Multinuclear Cu-Catalysts Based on SPINOL-PHOS in

Asymmetric Conjugate Addition of Organozinc Reagents

Kohei Endo*,^{†,‡}, Daisuke Hamada[§], Sayuri Yakeishi[§], Mika Ogawa[§], Takanori Shibata*,[§]

[†]Division of Material Sciences, Graduate School of Natural Science and Technology, Kanazawa University,

Kakuma, Kanazawa 920-1192, Japan

[#] PRESTO, Japan Science and Technology Agency (JST), 4-1-8 Honcho Kawaguchi, Saitama, 332-0012, Japan

[§] Department of Chemistry and Biochemistry, School of Advanced Science and Engineering, Waseda University,

Shinjuku, Tokyo, 169-8555, Japan

kendo@se.kanazawa-u.ac.jp, tshibata@waseda.jp

Contents:

1. General Instruments and Chemicals		
2. Screening of Reaction Conditions		
· 2-1. Conditions	S 3	
· 2-2. ESI-MS of Dinuclear Zn-Complexes	S4	
3. Synthetic Procedures and Physical Property of New Compounds	5	
· 3-1. General Experimental Procedures	S5	
· 3-2. Ligands	S5	
· 3-3. Products	S7	
4. NMR Spectra	S15	

1. General Instruments and Chemicals

General: All the reactions dealing with air or moisture sensitive compounds were carried out in a dry reaction vessel under positive pressure of argon. Air- and moisture-sensitive liquids and solutions were transferred via a syringe or a stainless steel cannula. Analytical thin-layer chromatography was performed on a glass plates coated with 0.25 mm 230-400 mesh silica gel containing a fluorescent indicator (Merck, #1.05715.0009). Thin layer chromatography plates were visualized by exposure to ultraviolet light (254 nm) and/or by immersion in an acidic staining solution of *p*-anisaldehyde followed by heating. Organic solutions were concentrated by rotary evaporation at c.a. 30 mmHg. Flash column chromatography was performed on Kanto Silica gel 60 (spherical, neutral, 140–325 mesh) as described by Still *et al.*¹

Instrumentation: IR spectra were recorded with a Horiba FT730 spectrophotometer. NMR spectra were measured with JEOL AL-400 spectrometer for ¹H and ¹³C NMR, Lambda-500 for ¹⁹F NMR and ³¹P NMR, using tetramethylsilane as an internal reference, phosphoric acid and trifluoroacetic acid as external references and CDCl₃ as a solvent. Chemical shift values for protons are reported in parts per million (ppm, δ scale) downfield from tetramethylsilane and are referenced to residual proton of CDCl₃ (δ 7.26). Carbon nuclear magnetic resonance spectra (¹³C NMR) were recorded at 100 MHz: chemical shifts for carbons are reported in parts per million (ppm, δ scale) downfield from tetramethylsilane and are referenced to the carbon resonance of CDCl₃ (δ 77.0). Fluorine nuclear magnetic resonance spectra (¹⁹F NMR) were recorded at 470 MHz: chemical shifts for fluorine are reported in parts per million (ppm, δ scale) referenced to the fluorine resonance of trifluoroacetic acid (δ -76.5). Phosphorous nuclear magnetic resonance spectra (³¹P NMR) were recorded at 202 MHz: chemical shifts for phosphorous are reported in parts per million (ppm, δ scale) referenced to the phosphorous resonance of phosphoric acid (δ 0). Data are presented as following space: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, sext = sextet, sept = septet, m = multiplet and/or multiplet resonances, br = broad), coupling constant in hertz (Hz), and signal area integration in natural numbers, assignment (*italic*). Mass spectra were measured with a LOC JMS-SX102A. ESI-MS spectra were measured with LCQ-DECA.

Chemicals: All reagents were purchased as commercially available source unless otherwise noted. (*R*)-BINOL (>99.0% *ee*) was purchased from Fuji Molecular Planning Co., Ltd. Enones were purchased as commercially available source or prepared according to the previously reported protocols.

¹ W. C. Still, M. Kahn, A. Mitra, J. Org. Chem. **1978**, 43, 2923.

2. Other Dialkylzinc Reagents

Table S1. Screening of Organozinc Reagents with SP2

Ph 1	0 + R Ph + (3 e	2Zn 2quiv) CuCl ₂ :2H ₂ O SP2 (5 r THF, 0 °C	$(5 \text{ mol }\%)$ $\xrightarrow{\text{nol }\%)} \qquad \xrightarrow{\text{R}} O$ $\xrightarrow{\overline{\vdots}} \qquad \xrightarrow{\text{Ph}} Ph$ $(R) \qquad \qquad Ph$
entry	R	time (h)	yield, ee $(\%)^a$
1	Et	2	93, 95
2	Me	17	20, 93
3	<i>n</i> -Bu	17	19, 86
^a Isola	ted vields Fe	was determined by	v chiral HPLC analysis

3. ESI-MS of Dinuclear Zn-Complexes



3. Synthetic Procedures and Physical Property of New Compounds

3-1. Representative Procedures

A mixture of $CuCl_2 2H_2O$ (0.005 mmol) and **SP1** (0.005 mmol) in THF (1 mL) was stirred at room temperature for 30 minutes and then cooled to 0 °C. A solution of Et_2Zn in hexane (0.3 mmol, 0.3 mL, 1 M) was added to the mixture. To the clear yellow solution was added enone (0.1 mmol) at once. The reaction was carried out at 0 °C and monitored by TLC analysis. After the completion of the reaction, a minimal amount of Sat. NH₄Cl aq. was added. After stirring at 0 °C for 30 minutes, the mixture was passed through a pad of silica gel with ether (50 mL). Concentration and purification by silica gel columun chromatography gave the desired product.

A mixture of $CuCl_2 \cdot 2H_2O$ (0.025 mmol) and **SP2** (0.025 mmol) in THF (5 mL) was stirred at room temperature for 30 minutes and then cooled to 0 °C. A solution of Et₂Zn in hexane (1.5 mmol, 1.5 mL, 1 M) was added to the mixture. To the clear yellow solution was added enone (0.5 mmol) at once. The reaction was carried out at 0 °C and monitored by TLC analysis. After the completion of the reaction, a minimal amount of Sat. NH₄Cl aq. was added. After stirring at 0 °C for 30 minutes, the mixture was passed through a pad of silica gel with ether (50 mL). Concentration and purification by silica gel columun chromatography gave the desired product.

3-2. Ligands

Synthesis of BP2



To a solution of Br₂Cl₂BINOL² (ca. 5 mmol including inseparable impurities) in THF (50 mL) was added NaH (ca. 12 mmol, 55% oil dispersion) at rt. After 1 h stirring, MOMCl (15 mmol) was added dropwise at rt. The reaction mixture was stirred at rt for 1 h. The excess NaH was killed with a minimal amount of Sat. NH.Cl ag. The mixture was filtrated through a pad of silica gel with EtOAc (200 mL). Concentration gave a solid containing Br₂Cl₂-MOM-BINOLate, which was used without purification. To a solution of Br₂Cl₂-MOM-BINOLate (ca. 5 mmol including some impurities) in THF 50 mL was added n-BuLi (12 mmol, 1.65 M in hexane) dropwise at -78 °C. After stirring at the temperature for 3 h, Ph₂PCl (12 mmol) was added dropwise at -78 °C. The mixture was stirred at -78 °C for 1 h and then room temperature for 1 h. The reaction was terminated with a minimal amount of Sat. NaHCO₃ aq. The mixture was passed through a pad of silica gel with ether to give a crude mixture, which was concentrated under vacuo to give a residue. The deprotection was carried out in refluxing MeOH 50 mL with a few drops of conc. HCl aq. for 16 h. The resulting solution was treated with a minimal amount of Sat. NaHCO₃ aq. and concentrated. The residue was purified by silica gel column chromatography (10% EtOAc/hexane) to give BP2 in 832 mg, 23% yield as a white powder; mp 145 °C (decompose); ¹H NMR (400 MHz, CDCl₃) δ 4.98 (s, 2H), 6.94 (d, J = 4.4 Hz, 2H), 7.06 (d, J = 8.8 Hz, 2H), 7.24 (d, J = 8.8 Hz, 2H), 7.37–7.40 (m, 20H), 8.43–8.45 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 111.0, 124.6, 125.8, 126.4, 128.5, 128.9, 129.0, 129.4, 130.6, 131.7, 131.8, 132.0, 132.2, 134.1, 134.2, 134.3, 134.4, 134.9, 135.0 (2C), 139.1, 139.2, 152.1; ³¹P NMR (202 MHz, CDCl₃) δ –13.5 (s); IR (cm⁻¹, CH₂Cl₂) 3525, 1571, 1494, 1434, 1371, 1319, 1182, 1139, 1093; $[\alpha]^{27.1}_{D} = 84.2$ (c = 1.02, CHCl₃); HRMS (FAB, positive) m/z calcd. for $C_{44}H_{30}Cl_2O_2P_2$ [M]⁺, 722.1098; found 722.1108.

² Valente, C.; Choi, E.; Belowich, M. E.; Doonan, C. J.; Li, Q.; Gasa, T. B.; Botros, Y. Y.; Yaghi, O. M.; Stoddart, J. F. *Chem. Commun.* **2010**, *46*, 4911.

Synthesis of BP3



n-BuLi (2.4 equiv), THF, -78 °C, 3 h then Ph₂PCI (2.4 equiv), -78 °C, 1 h then rt, 1 h
 cat. HCI, MeOH, reflux, 16 h

To a solution of 6,6'-Br₂-MOM-BINOLate³ (7 mmol) in THF 50 mL was added *n*-BuLi (16.8 mmol, 1.65 M in hexane) dropwise at -78 °C. After stirring at the temperature for 3 h, Ph₂PCl (16.8 mmol) was added dropwise at -78 °C. The mixture was stirred at -78 °C for 1 h and then room temperature for 1 h. The reaction was terminated with a minimal amount of Sat. NaHCO₃ aq. The mixture was passed through a pad of silica gel with ether to give a crude mixture, which was concentrated under vacuo to give a residue. The deprotection was carried out in refluxing MeOH 50 mL with a few drops of conc. HCl aq. for 16 h. The resulting solution was treated with a minimal amount of Sat. NaHCO₃ aq. and concentrated. The residue was purified by silica gel column chromatography (20% EtOAc/hexane) to give **BP3** in 990 mg, 21% yield as a white powder; mp. 125 °C (decomposed); ¹H NMR (400 MHz, CDCl₃) δ 5.21 (broad d, 2H), 7.08–7.35 (m, 26H), 7.79–7.91 (m, 4H); ¹³C NMR (100 MHz, CDCl₃, some of the peaks overlapped) δ 117.7, 118.1, 124.1 (2C), 127.5, 128.4, 128.5 (3C), 128.7, 131.5, 131.6, 133.6 (2C), 133.8 (2C), 153.5; ³¹P NMR (202 MHz, CDCl₃) δ –4.8 (s); IR (cm⁻¹, CH₂Cl₂) 3523, 3054, 1734, 1514, 1269, 897; [α]³⁴_D = -127.6 (c = 1.02, CHCl₃); HRMS (FAB, positive) *m*/*z* calcd. for C₄₄H₃₂O₂P₂ [M]⁺, 654.1878; found 654.1870.

Synthesis of SP1



To a suspension mixture of MOM-protected (*S*)-SPINOL⁴ (275mg, 0.81 mmol) in ether (15 mL) was added *n*-BuLi (1.60 M, 2.0 mmol, 2.4 equiv) at room temperature. After 3 h stirring at ambient conditions, THF (10 mL) was added. The orange solution with some precipitates was stirred further for 1 h. Then Ph₂PCl (2.0 mmol, 2.4 equiv) was added at 0 °C. After stirring at room temperature for 30 min, Sat. NH₄Cl aq. (1 mL) was added. Concentration under vacuum gave a residue, which was filtered through a pad of silica gel with AcOEt. Concentration gave a white solid, which was treated with a few drops of conc. HCl aq. in refluxing MeOH for 3 h. Concentration gave the residue, which was passed through a pad of silica gel with CH₂Cl₂. Concentration and purification by silica gel column chromatography gave **SP1** in 124 mg, 0.2 mmol, 25% yield; white powder; mp. 83 °C; ¹H NMR (400 MHz, CDCl₃) δ 2.22–2.36 (m, 4H), 2.93–3.07 (m, 4H), 5.70 (broad s, 2H), 6.73–6.79 (m, 4H), 7.24–7.35 (m, 20H); ¹³C NMR (100 MHz, CDCl₃) δ 31.1, 37.6, 58.2, 117.7 (2C), 119.7, 119.8, 128.3, 128.4, 128.5 (2C), 128.6, 128.8, 132.3, 133.0, 133.2, 133.5, 133.7, 134.3, 135.8, 135.9 (3C), 147.9, 155.3, 155.5; ³¹P NMR (202 MHz, CDCl₃) δ –24.7 (s); IR (cm⁻¹, CH₂Cl₂) 3515, 2950, 1434, 1419, 1261, 809; [α]^{28.2}_D = –5.0 (c = 1.09, CHCl₃); HRMS (FAB, positive) *m/z* calcd. for C₄₁H₃₅O₂P₂ [M + H]⁺, 621.2107; found 621.2118.

Synthesis of SP2



SP2, 58%

To a solution of (S)-Br₂-SPINOLate⁵ (8.7 mmol) in THF (200 mL) was added *n*-BuLi (21 mmol, 1.59 M in hexane) dropwise at -78 °C. After stirring at the temperature for 2 h, Ph₂PCl (21 mmol) was added at -78 °C. The mixture was stirred at -78 °C

³ Yang, L.; Yang, F.; Lan, J.; Gao, G.; You, J.; Su, X. Org. Biomol. Chem. 2011, 9, 2618.

⁴ Yang, Y.; Zhu, S.-F.; Duan, H.-F.; Zhou, C.-Y.; Wang, L.-X.; Zhou, Q.-L. J. Am. Chem. Soc. 2007, 129, 2248.

⁵ Birman, V. B.; Rheingold, A. L.; Lam, K.-C.; *Tetrahedron: Asymm.* 1999, 10, 125.

for 1 h and then room temperature for 0.5 h. The mixture was treated with a minimal amount of H₂O and concentrated. The residue was passed through a pad of silica gel with 5% EtOAc/hexane. The volatiles were removed under vacuo to give a crude product, which was treated with LiAlH₄ (44 mmol) in THF at -78 °C. After stirring at room temperature for 1 h, the mixture was treated with H₂O carefully at 0 °C. The resulting mixture was passed through a pad of silica gel. The volatiles were removed under vacuo to give a residue. A plenty of hexane was added and placed at -78 °C to form white powder. The filtration gave a white powder. The mother liquor was concentrated and placed at -78 °C to form second white powder. The combined yield of **SP2** was 58%; white powder; mp. 111 °C; ¹H NMR (400 MHz, CDCl₃) δ 2.14–2.25 (m, 4H), 2.73–2.82 (m, 2H), 3.09–3.14 (m, 2H), 4.83 (broad s, 2H), 6.61–6.72 (m, 4H), 7.26–6.36 (m, 20H); ¹³C NMR (100 MHz, CDCl₃) δ 31.1, 31.2, 57.6, 115.1 (2C), 124.7, 124.8, 128.5 (2C), 128.6 (2C), 128.7, 130.7 (2C), 133.6 (2C), 133.8 (2C), 134.3, 136.4, 136.5, 136.6, 136.7, 150.3, 150.6, 153.7; ³¹P NMR (202 MHz, CDCl₃) δ –14.4 (s); IR (cm⁻¹, CH₂Cl₂) 3532, 1581, 1471, 1205, 823; [α]^{28.3}_D = 47.6 (c = 1.06, CHCl₃); HRMS (FAB, positive) *m*/*z* calcd. for C₄₁H₃₄O₂P₂, 620.2034; found 620.2034.

3-3. Products⁶



(*R*)-1,3-diphenylpentan-1-one (2a); $[\alpha]_{D}^{24.3} = -3.18$ (c = 1.38, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1.0 mL/min, 254 nm), t_r (major) = 6.0 min, t_r (minor) = 7.2 min.

Chiral HPLC: (S)-2a (90% ee) with SP1



(-)-3-(4-fluorophenyl)-1-phenylpentan-1-one (2b)



 $[\alpha]^{26.3}_{D} = -1.95$ (c = 1.31, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1.0 mL/min, 254 nm), t_r = 6.3 min, t_r = 8.0 min.

Chiral HPLC: (+)-2b (75% ee) with SP1



⁶ We already reported the physical properties of products: Endo, K. Ogawa, M.; Shibata, T. Angew. Chem. Int. Ed. 2010, 49, 2410.

(R)-3-(4-chlorophenyl)-1-phenylpentan-1-one (2c)



 $[\alpha]_{D}^{264} = +2.1$ (c = 1.78, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 0.5 mL/min, 254 nm), t_r = 6.3 min, t_r = 8.6 min.

Chiral HPLC: (S)-2c (85% ee) with SP1



(R)-1-phenyl-3-(4-(trifluoromethyl)phenyl)pentan-1-one (2d)



 $[\alpha]_{D}^{18.0}$ = +7.76 (c = 1.39, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 0.5 mL/min, 254 nm), t_r = 5.2 min, t_r = 6.7 min.

Chiral HPLC: (S)-2d (84% ee) with SP1



(-)-1-phenyl-3-(p-tolyl)pentan-1-one (2e)



 $[\alpha]_{D}^{24.6} = -6.31 \text{ (c} = 1.33, \text{CHCl}_3); \text{HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1.0 mL/min, 254 nm), t_r = 5.6 min, t_r = 7.5 min.$

Chiral HPLC: (+)-2e (83% ee) with SP1



(+)-3-([1,1'-biphenyl]-4-yl)-1-phenylpentan-1-one (2f)



 $[\alpha]_{D}^{25.3} = -3.96$ (c = 1.23, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1.0 mL/min, 254 nm), t_r = 7.6 min, t_r = 10.9 min.

Chiral HPLC: (-)-2f (59% ee) with SP1



(*R*)-3-(4-methoxyphenyl)-1-phenylpentan-1-one (2g)



 $[\alpha]_{D}^{25.1} = -5.78 \text{ (c} = 2.1, \text{CHCl}_3); \text{HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1.0 mL/min, 254 nm), t_r (major) = 7.8 min, t_r (minor) = 11.4 min.$

Chiral HPLC: (S)-2g (73% ee) with SP1



(+)-3-(naphthalen-2-yl)-1-phenylpentan-1-one (2h)



 $[\alpha]_{D}^{26.2}$ = +5.4 (c = 2.1, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1.0 mL/min, 254 nm), t_r = 7.6 min, t_r = 9.5 min.

Chiral HPLC: (-)-2h (76% ee) with SP1



Chiral HPLC: (+)-2h (94% ee) with SP2

		7. SN 1	- 1. Sa.,					
**	CALCU.	ATION REPO	17 **					
CH	PKNO	TIME	AGLA.	HE IGHT	MK	10\0	CONC	
1	-1	7.883	22166	1755	1		5. 2034	
	6	9.867	669777	13055			96.7966	
		TOTAL.	691943	45410			100	

(+)-3-(furan-2-yl)-1-phenylpentan-1-one (2i)



 $[\alpha]_{D}^{25.6}$ (90% *ee*) = +0.28 (c = 1.47, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 0.5 mL/min, 254 nm), t_r = 11 min, t_r = 12.7 min.

Chiral HPLC: (-)-2i (79% ee) with SP1



Chiral HPLC: (+)-2i (90% ee) with SP2



(-)-1-phenyl-3-(thiophen-2-yl)pentan-1-one (2j)



 $[\alpha]_{D}^{24.2}$ = -7.47 (c = 1.06, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 95/5, 1 mL/min, 254 nm), t_r = 6 min, t_r = 7 min.

Chiral HPLC: (+)-2j (78% ee) with SP1



5, 3496	1732	17189	0.341	2
94.6503	26639	501114	7.386	3
1				
100	28371	321302	TOTAL	

(R)-3-cyclohexyl-1-phenylpentan-1-one (2k)



 $[\alpha]^{24.2}_{D}$ = +3.53 (c = 1.79, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 98/2, 0.7 mL/min, 254 nm), t_r (major) = 7.8 min, t_r (minor) = 8.6 min.

Chiral HPLC: (S)-2k (91% ee) with SP1



Chiral HPLC: (R)-2k (99% ee) with SP2



(-)-3-ethyl-1-phenyloctan-1-one (2l)



 $[\alpha]^{23.4}_{D} = -2.43$ (c = 1.76, CHCl₃); HPLC (CHIRALPAK AD-H x 2, Daicel, 4.6 x 250 mm, hexane/IPA = 99.9/0.1, 1.0 mL/min, 254 nm), t_r = 23.4 min, t_r = 24.2 min.

Chiral HPLC: (+)-2l (75% ee) with SP1



Chiral HPLC: (-)-2l (96% ee) with SP2

** CH

	23.522	24.764				
CALCUI PKNO 8	LATION REPORT TIME 23.522	** AREA 18792	HEIGHT 777	MK	IDNO	

2.2837	777	18792	23.522	8	1
97.7163	23180	804068	24.764	9	
will fight this little rate and the last off have also not used by t					
100	23956	822860	TOTAL		

CONC

(R)-4-phenylhexan-2-one (2m)

Et O . Me

 $[\alpha]^{26.7}_{D} = -33.3$ (c = 0.95, CHCl₃); chiral HPLC (CHIRALPAK OJ-H, Daicel, 4.6 x 250 mm, hexane/IPA = 98/2, 1.0 mL/min, 254 nm), t_r = 10.4 min, t_r = 12.2 min.

Chiral HPLC: (S)-2m (64% ee) with SP1





(S)-4-ethyloctan-2-one (2n)



 $[\alpha]_{D}^{21.5} = -1.80 (c = 1.41, CHCl_3);$ chiral GC (RT-bDEXsm, 0.25 mm x 30 m, 90 °C, 30 mL/min), $t_r = 11.2 min, t_r = 11.6 min.$

Chiral GC: (*R*)-2n (78% ee) with SP1



Chiral GC: (S)-2n (88% ee) with SP2

-	8.0						
-	10.0						
-	12.0	11.657	11.293				
**	CALCU	LATION REPO	ORT **				
CH	PKNO	TIME	AREA	HEIGHT	MK	IDNO	CONC
1	1	11.293	338771	40657			93, 9777
	2	11.657	21709	2293	V		6.0223
		TOTAL	360480	42949		_	100

(*R*)-3-ethylcyclohexanone (20)



 $[\alpha]^{22.5}_{D}$ = +2.8 (c = 0.51, CHCl₃); HPLC (CHIRALPAK AD-H, Daicel, 4.6 x 250 mm, hexane/IPA = 99.5/0.5, 1.0 mL/min, 254 nm), t_r = 4.7 min, t_r = 5.9 min.

Chiral GC: (S)-20 (92% ee) with SP1



Chiral GC: (R)-20 (33% ee) with SP2



4. NMR Spectra of New Compounds

¹H NMR (400 MHz) BP2



¹³C NMR (100 MHz) BP2





¹³C NMR (100 MHz) BP3



¹H NMR (400 MHz) SP1



¹³C NMR (100 MHz) SP1







¹³C NMR (100 MHz) SP2

