

Supporting Information

Infrared laser spectroscopy of the CH₃OO radical formed from the reaction of CH₃ and O₂ within a helium nanodroplet

Alexander M. Morrison,^b Jay Agarwal,^a H. F. Schaefer III^a and Gary E. Douberly^{b,*}

^a Center for Computational Chemistry, University of Georgia, Athens, GA 30602

^b Department of Chemistry, University of Georgia, Athens, GA 30602

- Corresponding Author: douberly@uga.edu

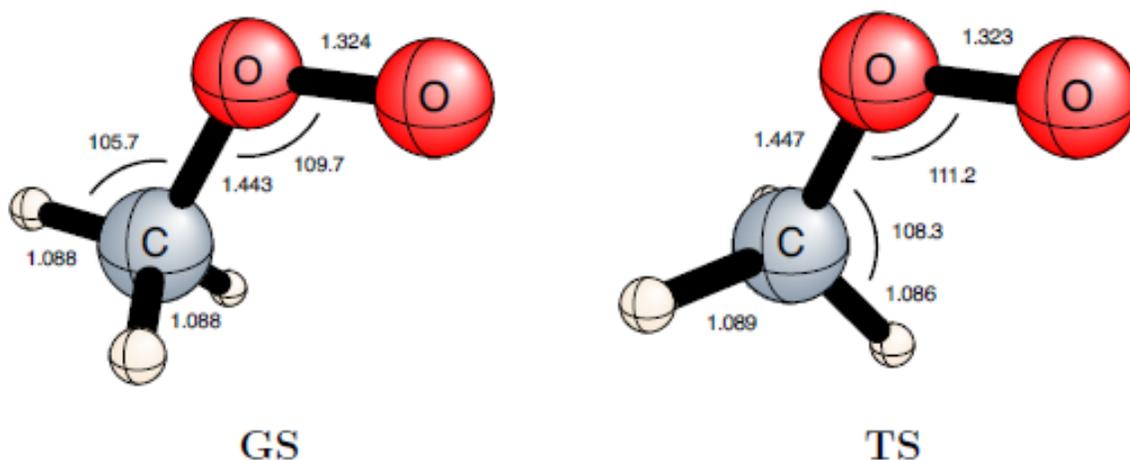


Figure S1: Optimized geometry for the ground-state and transition-state at the CCSD(T)/cc-pVTZ level of theory. Bond lengths are shown in Angstroms and bond angles in degrees.

Supporting Information

Table S1: Focal Point analysis of the CH₃OO torsional barrier in the (X)²A'' electronic state.^a

| Basis set | ΔE_e HF | + δ MP2 | + δ CCSD | + δ CCSD(T) | NET |
|-----------|-----------------|----------------|-----------------|--------------------|----------------|
| cc-pVTZ | +1.36 | -0.66 | +0.33 | -0.08 | [+0.95] |
| cc-pVQZ | +1.38 | -0.69 | +0.33 | -0.08 | [+0.94] |
| cc-pV5Z | +1.38 | -0.69 | [+0.33] | [-0.08] | [+0.94] |
| CBS LIMIT | [+1.38] | [-0.70] | [+0.33] | [-0.08] | [+0.93] |

$\Delta E_t(\text{final}) = \Delta E_t[\text{CBS CCSD(T)}] + \Delta_{\text{ZPVE}}[\text{CCSD(T)/cc-pVTZ}]$
 $\Delta E_t(\text{final}) = 0.93 - 0.21 = \mathbf{0.72 \text{ kcal mol}^{-1}}$

^aEnergies shown in kcal mol⁻¹. Delta (δ) denotes the change in relative energy (ΔE_e) with respect to the preceding level of theory.

Note: The barrier for torsional rotation (0.72 kcal mol⁻¹), has a suggested accuracy of ± 0.1 kcal mol⁻¹. This can be deduced from looking at the focal-point table, where we see oscillatory convergence to within 0.08 kcal mol⁻¹.

Ground-State: (Cartesian Coordinates)

| | | | |
|---|-----------------|-----------------|-----------------|
| H | -1.97621996614 | 0.487796731042 | 0.0 |
| C | -1.1661393889 | -0.238632305755 | 0.0 |
| O | 0.0526172527545 | 0.534467925974 | 0.0 |
| O | 1.09753254814 | -0.278193009495 | 0.0 |
| H | -1.19623505374 | -0.856855813713 | 0.895207315689 |
| H | -1.19623505374 | -0.856855813713 | -0.895207320981 |

Ground-State Frequencies [CCSD(T)/cc-pVTZ]:

0.0000i
 0.0000i
 0.0000i
 0.0000i
 0.0000
 0.0000
 149.7274
 493.5895
 949.3199
 1144.0846
 1160.2828
 1212.0275
 1449.0832
 1484.1431
 1497.6945
 3061.9382
 3159.3911
 3168.8276
 ZPVE: 27.0619 kcal/mol

Supporting Information

Transition State: (Cartesian Coordinates)

| | | | |
|---|-----------------|------------------|-----------------|
| H | -0.925965420482 | 1.29146315085 | 0.0 |
| C | -1.17814575391 | 0.234897748448 | 0.0 |
| O | 0.053280041967 | -0.525593962184 | 0.0 |
| O | 1.10782046825 | 0.272944428453 | 0.0 |
| H | -1.73678128951 | -0.0393094821325 | -0.893941920592 |
| H | -1.73678128951 | -0.0393094821325 | 0.893941920592 |

Transition-State Frequencies [CCSD(T)/cc-pVTZ]:

142.6587i
0.0000i
0.0000i
0.0000i
0.0000
0.0000
0.0000
533.7710
941.3993
1126.2488
1138.5946
1229.6803
1437.7611
1486.6697
1496.9888
3059.1017
3148.8551
3182.0058
ZPVE: 26.8489 kcal/mol

Supporting Information

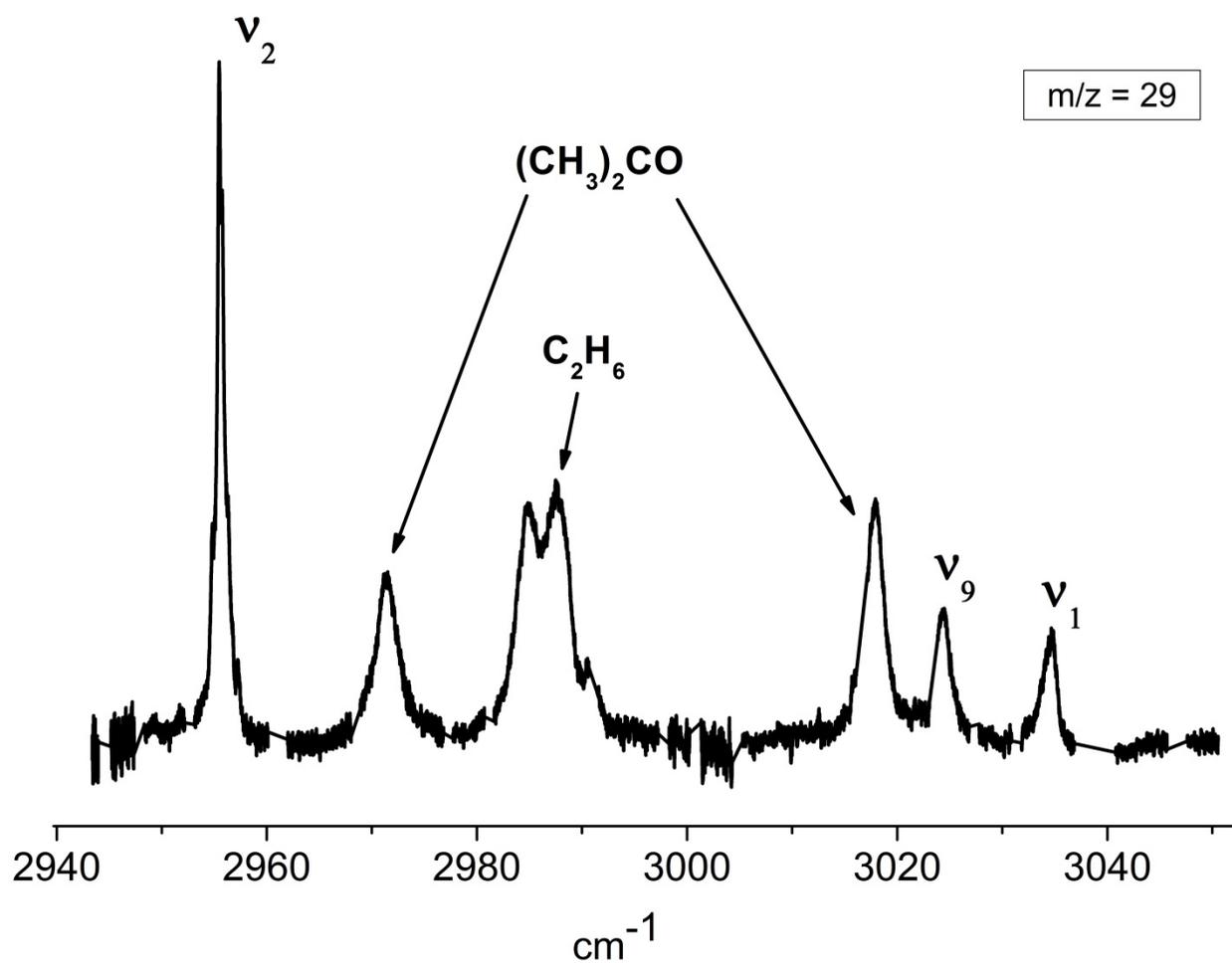


Figure S2: Survey Spectrum of the CH stretch region by measuring the depletion in mass channel $m/z = 29$. Two acetone bands as well as an ethane band are observed in addition to the three transitions reported in this manuscript for CH_3O_2 .