## Partitioning behavior of petrodiesel/biodiesel blends in

## water — Supporting Information

Mohamad H. Yassine<sup>†</sup>, Shuyun  $Wu^{\dagger}$ , Makram T. Suidan \*, and Albert D. Venosa §

<sup>†</sup>Department of Environmental Engineering, School of Energy, Environmental, Biological, and Medical Engineering, University of Cincinnati, Ohio 45221

Faculty of Engineering and Architecture, American University of Beirut, P.O. Box 11-0236 Riad El-Solh, Beirut, Lebanon 1107 2020

§U.S. Environmental Protection Agency, 26 W. Martin Luther King Drive, Cincinnati, OH 45268

\*Corresponding author e-mail: msuidan@aub.uc.edu.lb

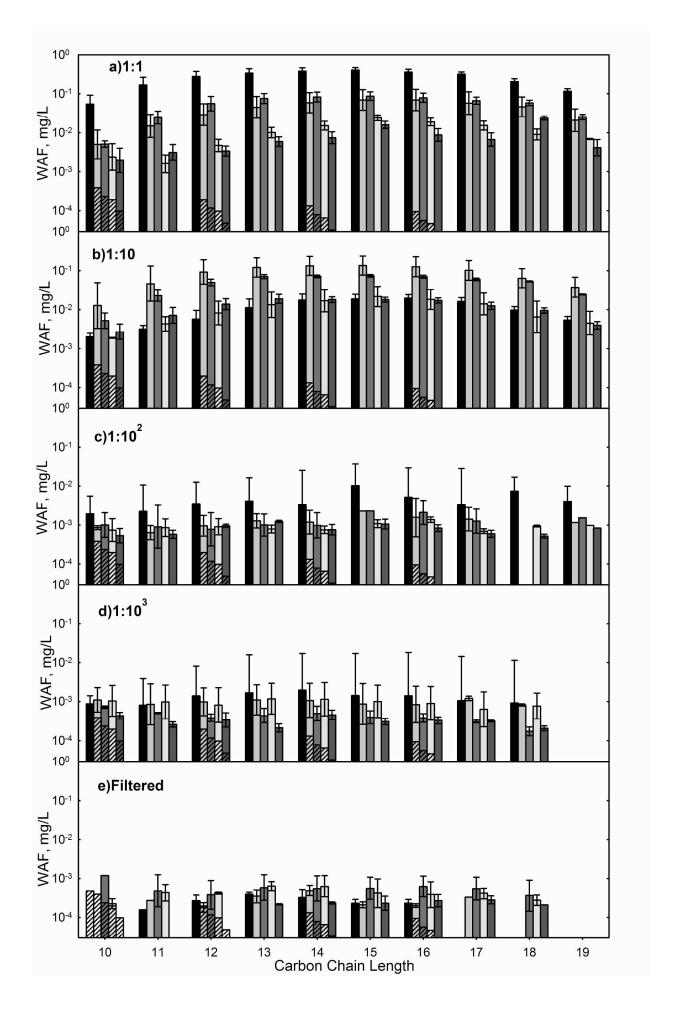


Figure S1. n-alkanes WAF concentrations in B0 ( $\blacksquare$ ), B20 ( $\blacksquare$ ), B40 ( $\blacksquare$ ), B60 ( $\blacksquare$ ), B80 ( $\blacksquare$ ), and  $S_{aq,WAF}^{sat}$  is n-alkanes concentrations calculated by Equation 1 using solubilities reported by Franks  $^1$  ( $\blacksquare$ 222) at dilution levels: (a) 1:10 $^0$ , (b) 1:10 $^1$ , (c) 1:10 $^2$ , (d) 1:10 $^3$ , (e) and 1:10 $^3$  filtered. Means and standard deviations are calculated geometrically (n = 3). Error bars are  $\pm$  1 geometric standard deviation. Not shown data were below detection limits.

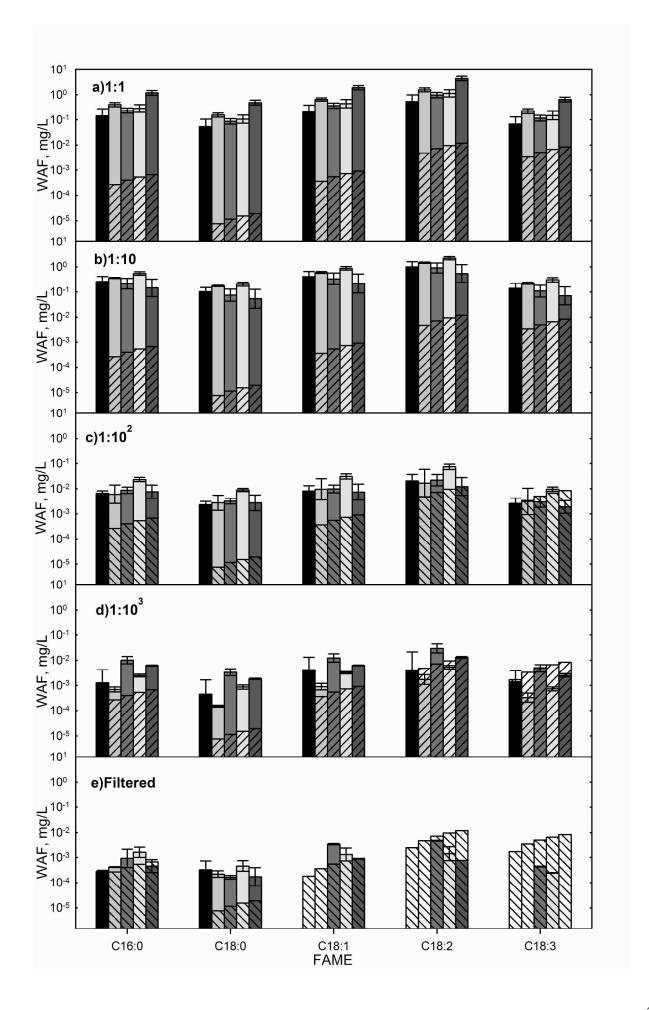


Figure S2. FAMEs WAF concentrations in B20 ( $\blacksquare$ ), B40 ( $\blacksquare$ ), B60 ( $\blacksquare$ ), B80 ( $\blacksquare$ ), B100 ( $\blacksquare$ ), and  $S_{aq,WAF}^{sat}$  is FAMEs concentrations calculated by Equation 1 using solubilities reported by Krop et al.<sup>2</sup> ( $\blacksquare$ ) at dilution levels: (a) 1:10<sup>0</sup>, (b) 1:10<sup>1</sup>, (c) 1:10<sup>2</sup>, (d) 1:10<sup>3</sup>, (e) and 1:10<sup>3</sup> filtered. Means and standard deviations are calculated geometrically (n=3). Error bars are  $\pm$  1 geometric standard deviation. Not shown data were below detection limits.

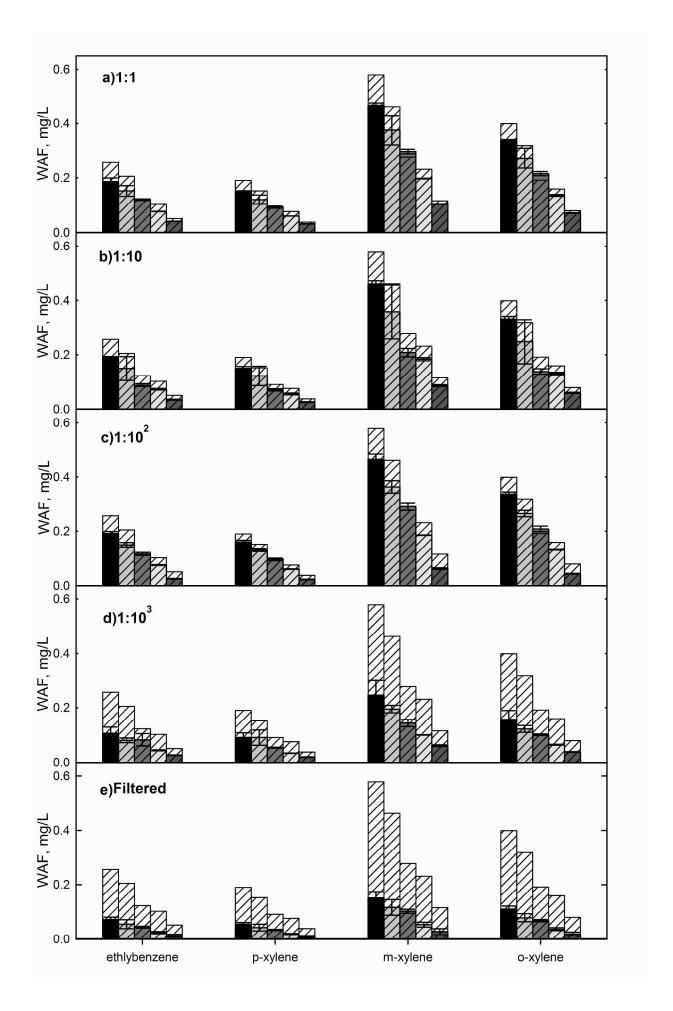


Figure S 3 Target aromatic compounds WAF concentrations in B0 ( $\blacksquare$ ), B20 ( $\blacksquare$ ), B40 ( $\blacksquare$ ), B60 ( $\blacksquare$ ), B80 ( $\blacksquare$ ), and  $S^{sat}_{aq,WAF}$  is aromatic compounds concentrations calculated by Equation 1 using solubilities reported by Sanemasa et al.<sup>3</sup> ( $\blacksquare$ ) at dilution levels: (a) 1:10<sup>0</sup>, (b) 1:10<sup>1</sup>, (c) 1:10<sup>2</sup>, (d) 1:10<sup>3</sup>, (e) and 1:10<sup>3</sup> filtered. Means and standard deviations are calculated arithmetically (n=3). Error bars are  $\pm$  1 arithmetic standard deviation.

Table S1. Reported aqueous solubility of pure selected hydrocarbons in deionized water at room temperature ( $22 \pm 1$  °C)

| Compound         | Pure solubility @ room temperature (µg/L) | Reference |
|------------------|---|-----------|
| <u>n-alkanes</u> |   |           |
| C10              | 26  | 1         |
| C16              | 6   | 1         |
| <u>FAMEs</u>     |   |           |
| C16:0            | 4   | 2         |
| C18:0            | 0.3                                       | 2         |
| C18:1            | 4.39                                      | 2         |
| C18:2            | 21  | 2         |
| C18:3            | 91.85                                     | 2         |
| mono-aromatic    |   |           |
| benzene          | 1,780,000 (or 1.78g/L)                    | 4         |
| toluene          | 563,000                                   | 5         |
| ethylbenzene     | 187,000                                   | 6         |
| p-xylene         | 153,000                                   | 6         |
| m-xylene         | 159,000                                   | 6         |
| o-xylene         | 212,000                                   | 6         |

| <u>PAHs</u>  |        |   |
|--------------|--------|---|
| Naphthalene  | 30,618 | 7 |
| Fluorene     | 1,685  | 8 |
| phenanthrene | 994    | 8 |
|              |        |   |

## References

- 1. Franks, F., Solute-Water Interactions and the Solubility Behaviour of Long-chain Paraffin Hydrocarbons. *Nature* **1966**, 210, (5031), 87-88.
- 2. Krop, H. B.; van Velzen, M. J. M.; Parsons, J. R.; Govers, H. A. J., n-Octanol-water partition coefficients, aqueous solubilities and Henry's law constants of fatty acid esters. *Chemosphere* **1997**, 34, (1), 107-119.
- 3. Sanemasa, I.; Araki, M.; Deguchi, T.; Nagai, H., Solubility measurements of benzene and the alkylbenzenes in water by making use of solute vapor. *Bulletin of the Chemical Society of Japan* **1982**, 55, (4), 1054-1062.
- 4. Goral, M., IUPAC-NIST solubility data series. 81. Hydrocarbons with water and seawater Revised and updated. Part 2. Benzene with water and heavy water. *Journal of physical and chemical reference data* **2005**, 34, (2), 477-552.
- 5. Goral, M., IUPAC-NIST solubility data series. 81. Hydrocarbons with water and seawater-revised and updated. Part 5. C7 hydrocarbons with water and heavy water. *Journal of physical and chemical reference data* **2005**, 34, (3), 1399-1487.
- 6. Goral, M., IUPAC-NIST solubility data series. 81. Hydrocarbons with water and seawater Revised and updated. Part 6. C8H8-C8H 10 hydrocarbons with water. *Journal of physical and chemical reference data* **2005**, 34, (3), 1489-1553.
- 7. Goral, M., IUPAC-NIST solubility data series. 81. Hydrocarbons with water and seawater Revised and updated. Part 9. C10 hydrocarbons with water. *Journal of physical and chemical reference data* **2006**, 35, (1), 93-151.
- 8. Goral, M., IUPAC-NIST solubility data series. 81. Hydrocarbons with water and seawater Revised and updated. Part 11. C13-C36 hydrocarbons with water. *Journal of physical and chemical reference data* **2006**, 35, (2), 687-784.