

STRUCTURAL AND ELECTRONIC PROPERTIES OF LITHIATED SnO₂. A PERIODIC DFT STUDY

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Table S1. Intercalation sites (fractional coordinates) considered for lithium intercalation and calculated cohesive energies (in eV) for all Li_xSnO_2 systems

X	Model	Intercalation sites for Li_xSnO_2 periodic models								E_{coh}
0										283.2
1/16	1/16A	$(0 \frac{1}{4} \frac{3}{4})$								286.4
	1/8A	$(0 \frac{1}{4} \frac{1}{4}) (0 \frac{1}{4} \frac{3}{4})$								288.7
	1/8B	$(0 \frac{1}{4} \frac{3}{4}) (0 \frac{3}{4} \frac{3}{4})$								288.8
	1/8C	$(0 \frac{3}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4})$								288.5
1/8	1/8D	$(0 \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4})$								288.1
	1/8E	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4})$								287.7
	1/8F	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{1}{4})$								287.6
	1/8G	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{4} \frac{1}{2} \frac{3}{4})$								288.4
	1/8H	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{4} \frac{1}{2} \frac{1}{4})$								288.6
	1/4A	$(0 \frac{1}{4} \frac{1}{4}) (0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4})$								293.6
	1/4B	$(0 \frac{3}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4})$								292.9
	1/4C	$(0 \frac{1}{4} \frac{3}{4}) (0 \frac{3}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4})$								292.8
1/4	1/4D	$(0 \frac{1}{4} \frac{3}{4}) (0 \frac{3}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{3}{4} \frac{1}{4})$								293.8
	1/4E	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4})$								294.0
	1/4F	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{1}{4})$								294.3
	1/4G	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{4} \frac{1}{2} \frac{3}{4}) (\frac{3}{4} \frac{1}{2} \frac{1}{4})$								294.0
	1/4H	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{4} \frac{1}{2} \frac{3}{4}) (\frac{3}{4} \frac{1}{2} \frac{3}{4})$								293.4
1/2	1/2A	$(0 \frac{1}{4} \frac{1}{4}) (0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{1}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4})$								303.4
	1/2B	$(0 \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{1}{4}) (\frac{1}{4} 0 \frac{1}{4}) (\frac{1}{4} \frac{1}{2} \frac{3}{4}) (\frac{3}{4} 0 \frac{3}{4}) (\frac{3}{4} \frac{1}{2} \frac{1}{4})$								305.0
	1/2C	$(0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4}) (\frac{1}{4} 0 \frac{3}{4}) (\frac{1}{4} \frac{1}{2} \frac{3}{4}) (\frac{3}{4} 0 \frac{3}{4}) (\frac{3}{4} \frac{1}{2} \frac{3}{4})$								301.2
	1	$(0 \frac{1}{4} \frac{1}{4}) (0 \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{1}{4} \frac{1}{4}) (\frac{1}{2} \frac{1}{4} \frac{3}{4}) (\frac{1}{2} \frac{3}{4} \frac{1}{4}) (\frac{1}{2} \frac{3}{4} \frac{3}{4})$								325.1
		$(\frac{1}{4} 0 0) (\frac{1}{4} 0 \frac{1}{2}) (\frac{1}{4} \frac{1}{2} 0) (\frac{1}{4} \frac{1}{2} \frac{3}{4}) (\frac{3}{4} 0 0) (\frac{3}{4} 0 \frac{1}{2}) (\frac{3}{4} \frac{1}{2} 0) (\frac{3}{4} \frac{1}{2} \frac{1}{2})$								

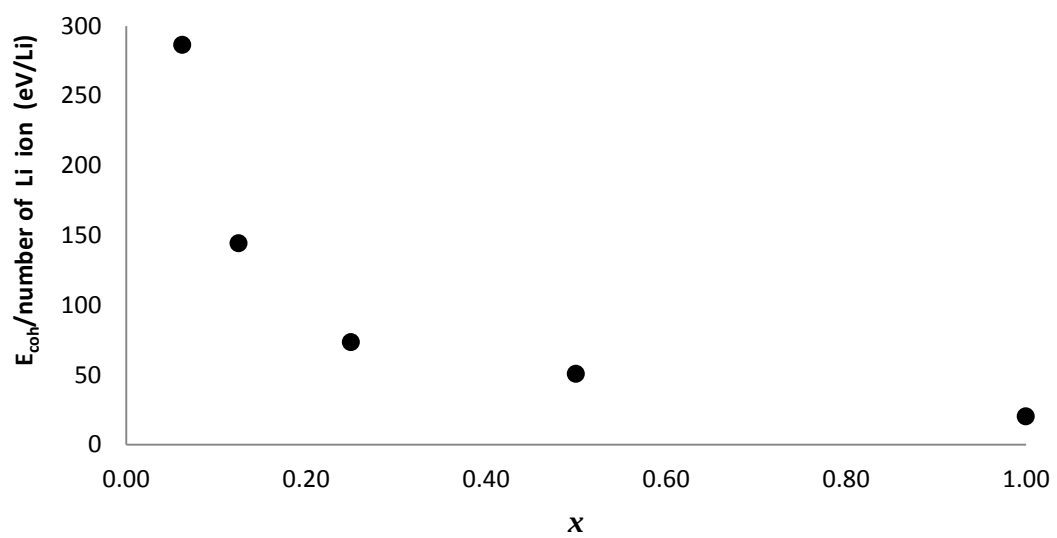


Figure S1. Cohesive energies per number of intercalated Li ions in the $2\times 2\times 2$ supercell.

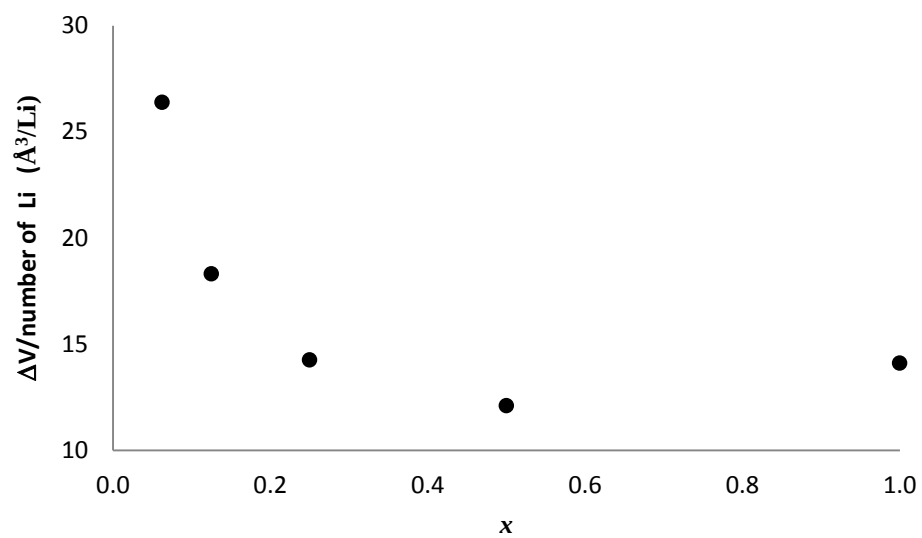


Figure S2. Variation in volume per lithium atom in the $2\times 2\times 2$ supercell.

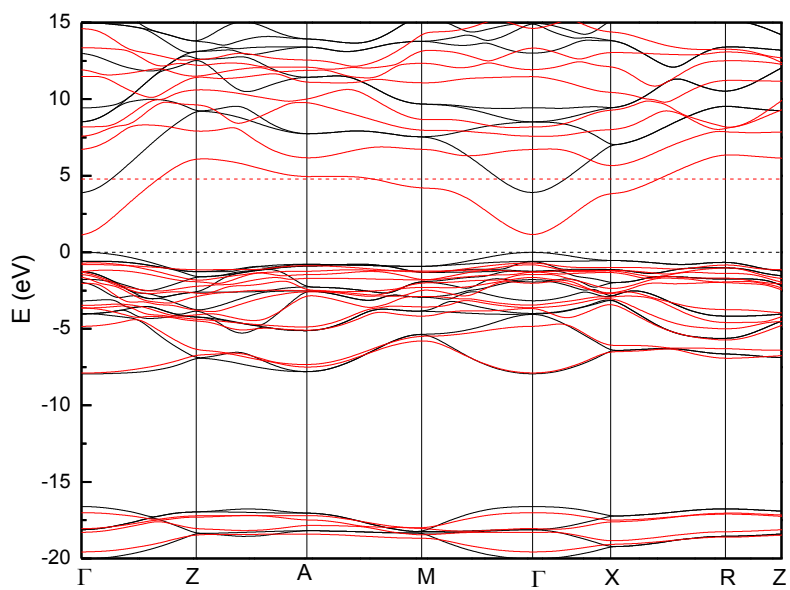


Figure S3. Calculated electronic band structure diagrams for pure SnO_2 (black line), $\text{Li}_{0.5}\text{SnO}_2$ (red line). The dashed lines refer to the corresponding Fermi level.

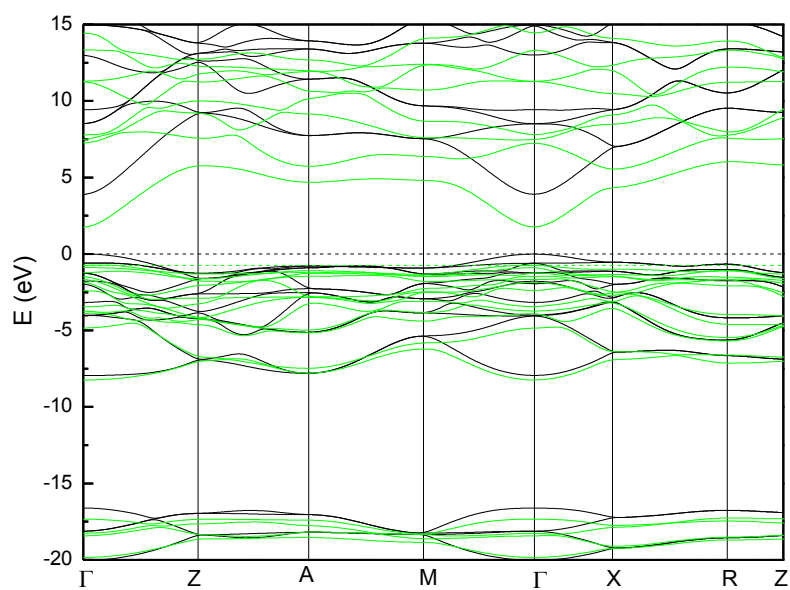


Figure S4. Calculated electronic band structure diagrams for pure SnO_2 (black line) and Li_0SnO_2 (green line). The dashed lines refer to the corresponding Fermi level.

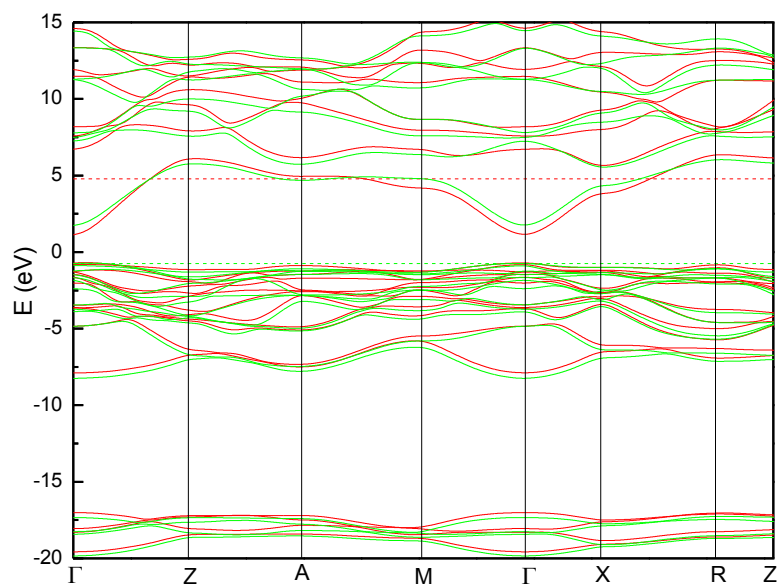


Figure S5. Calculated electronic band structure diagrams for $\text{Li}_{0.5}\text{SnO}_2$ (red line) and Li_0SnO_2 (green line). The dashed lines refer to the corresponding Fermi level.

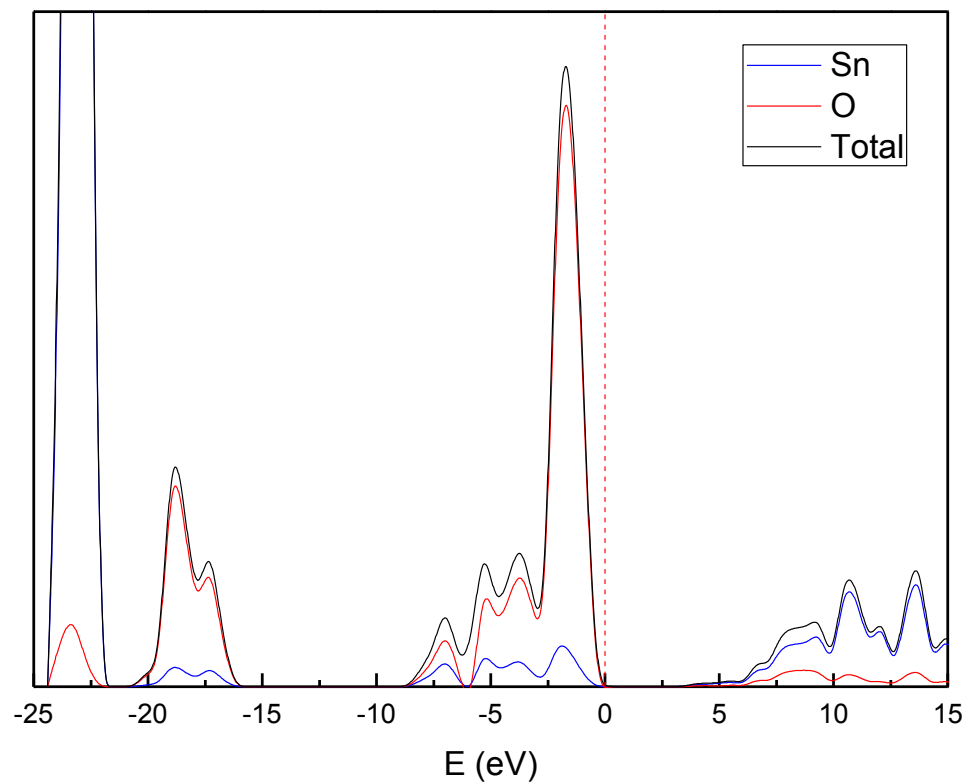


Figure S6. Calculated DOS and PDOS of pure SnO_2 . The dashed lines refer to the Fermi level. The bands centered about -23 eV and -19 eV correspond to tin 4*d*-like (VB3) and O 2*s*-like (VB2) states, respectively. See text for a detailed description of VB1 and CB.

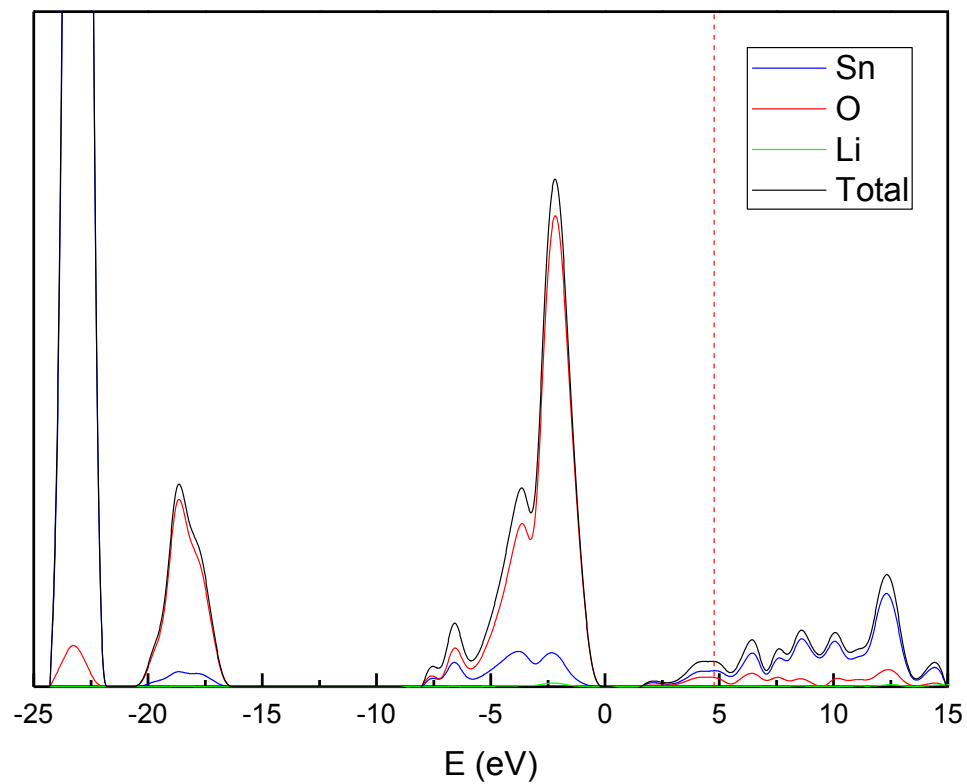


Figure S7. Calculated DOS and PDOS of $\text{Li}_{0.5}\text{SnO}_2$. The dashed lines refer to the Fermi level.

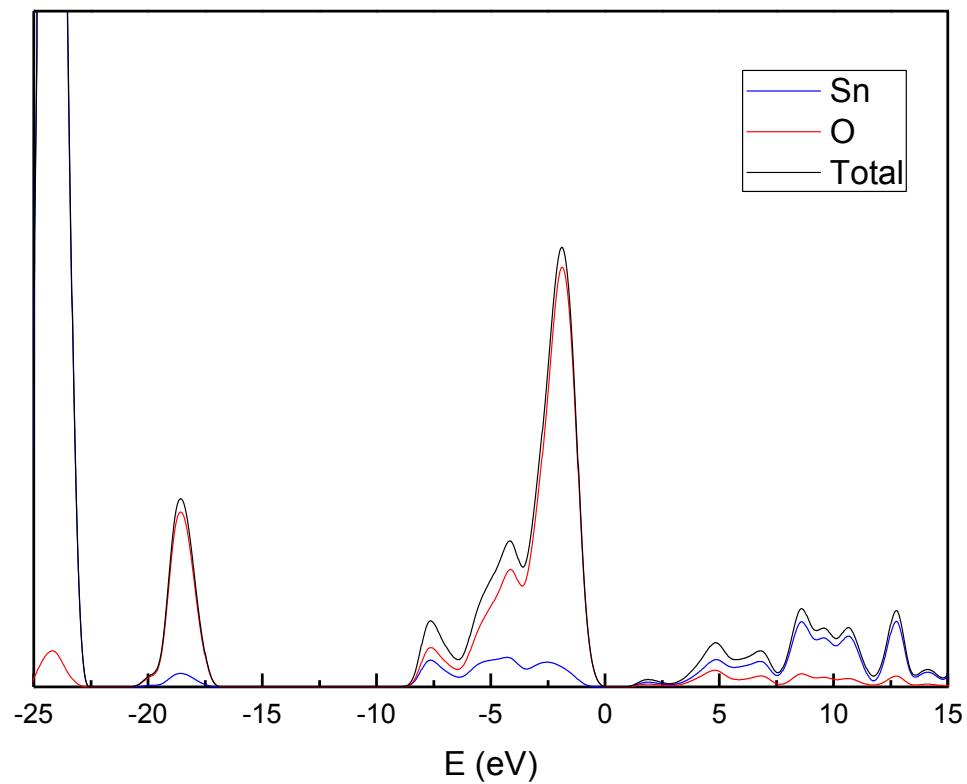


Figure S8. Calculated DOS and PDOS of delithiated Li_0SnO_2 (Li_0SnO_2 which is similar to $\text{Li}_{0.5}\text{SnO}_2$ where Li ions have been removed without relaxing the lattice).