

A Wharton-Fragmentation-Based Approach to the Carbocyclic Core of the Phomoidrides

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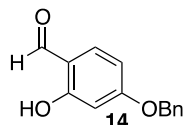
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1. General Experimental Methods

The NMR data are reported as follows: chemical shift in ppm on the δ scale, multiplicity (app = apparent, br = broad, s = singlet, d = doublet, t = triplet, q = quartet, quin = quintet, m = multiplet, coupling constants (Hz), and integration. ^1H NMR spectra were either recorded at ambient temperature at 300 MHz or 400 MHz. ^{13}C NMR spectra were recorded at ambient temperature at 100 MHz. For ^1H NMR spectra acquired in CDCl_3 , chemical shifts are reported as δ values in ppm and are calibrated according to internal CHCl_3 (7.26 ppm). For ^1H NMR spectra acquired in $\text{DMSO}-d_6$, chemical shifts are reported as δ values in ppm and are calibrated according to internal DMSO (2.50 ppm). For ^{13}C NMR spectra, chemical shifts are reported as δ values in ppm relative to chloroform and DMSO. Infrared spectra (IR) were obtained on an FTIR spectrophotometer and are reported in wavenumbers (cm^{-1}). High-resolution mass spectra were acquired on electrospray ionization (ESI) spectrometer and were obtained by peak matching. Single-crystal X-ray analyses were performed by Susie Miller or Brian Newell of Colorado State University.

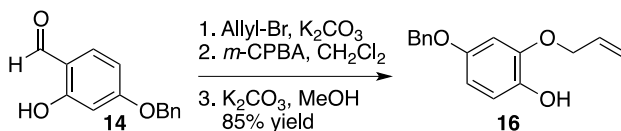
Analytical thin-layer chromatography (TLC) was performed using 0.25 mm precoated silica gel plates. Liquid chromatography was performed using forced flow (flash chromatography) of the indicated solvent system on silica gel (SiO_2) 60 Å (200-400 mesh). Unless otherwise noted all reactions were carried out using flame-dried or oven-dried glassware and inert atmosphere operations were conducted under N_2 (g) passed through a Drierite drying tube. CEM Discover Microwave was utilized for reactions performed under microwave irradiation. Anhydrous tetrahydrofuran (THF), methylene chloride (CH_2Cl_2), toluene, and diethyl ether (Et_2O) were filtered through two columns of activated basic alumina and transferred under Ar (g) according to the method described by Grubbs.¹ Anhydrous toluene was filtered through one column of activated basic alumina and one column of Q5 reactant, copper (II) oxide oxygen scavenger. Anhydrous acetonitrile (CH_3CN), dimethylsulfoxide (DMSO), and 1,2-dichloroethane were purchased from Aldrich and used without further purification. Triethylamine (Et_3N) was dried by distillation from CaH_2 under nitrogen. Zinc (II) chloride (ZnCl_2) was purchased as a 0.5 M solution in THF. All other commercial reagents were used as received, unless noted otherwise. Single-crystal X-ray analyses were performed by Susie Miller or Brian Newell of Colorado State University.

2. Experimental Details



4-Benzyloxy-2-(hydroxy)benzaldehyde (14) To a stirred solution of 2,4-dihydroxy benzaldehyde (20.0 g, 145 mmol, 1 equiv) in acetone (290 mL, 0.5 M) at room temperature was added K_2CO_3 (18.0 g, 130 mmol, 0.9 equiv) and benzyl bromide (10.3 mL, 86.9 mmol, 0.6 equiv) and the resulting mixture heated at reflux until TLC analysis (30% EtOAc / hexanes) indicated the consumption of benzyl bromide. The reaction mixture was cooled to room temperature and the solids were removed by filtration through celite. The filtrate was concentrated under reduced pressure and the resulting oil purified by gravity column chromatography (5, 10 then 20% EtOAc / hexanes until the product is recovered) to afford **14** (16.5 g, 83% based on benzyl bromide) as a purple solid.

1: R_f 0.59 (20% EtOAc / hexanes); IR (CH_2Cl_2 cast) 2833, 1643, 1626, 1576, 1371, 1338, 1228, 1215, 1121 cm^{-1} ; 1H NMR (400 MHz, $CDCl_3$) δ 11.48 (s, 1H), 9.72 (s, 1H), 7.46 – 7.32 (m, 6H), 6.62 (dd, J = 8.8, 2.2 Hz, 1H), 6.52 (d, J = 2.2 Hz, 1H), 5.11 (s, 2H); ^{13}C NMR (100 MHz, $CDCl_3$) δ 194.6, 166.0, 164.6, 135.8, 135.4, 128.9, 128.6, 127.7, 115.5, 109.1, 101.8, 70.6; HRMS (ESI) calcd for $C_{14}H_{13}O_3$ $[M+H]^+$ 229.0859; found 229.0857.



2-Allyloxy-4-(benzyloxy)benzaldehyde To a stirred solution of **14** (16.5 g, 72.3 mmol, 1 equiv) in acetone (145 mL, 0.5 M) at room temperature was added K_2CO_3 (20.0 g, 145 mmol, 2.0 equiv) and allyl bromide (8.0 mL, 94.0 mmol, 1.3 equiv) and the resulting mixture heated at reflux for 6 hours, at which time NMR analysis of a crude reaction aliquot indicated the consumption of **14**. The reaction mixture was cooled to room temperature and the solids were removed by filtration through celite. The celite was rinsed with EtOAc (2 x 50 mL) and the filtrate was concentrated under reduced pressure to afford the allyl ether as a crude oil. This was carried forward in the next step without additional purification.

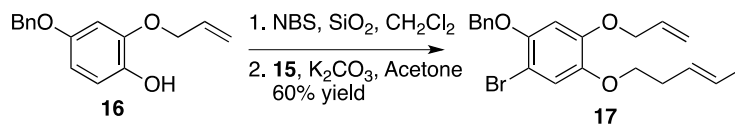
2-Allyloxy-4-(benzyloxy)phenol (16) Crude allyl ether was suspended in CH_2Cl_2 (580 mL, 0.125 M) and to this was added *m*-CPBA (25.0 g, 101 mmol, 1.4 equiv, 70%

pure). The resulting reaction mixture was heated at reflux until for 6 hours, at which time NMR analysis of a crude reaction aliquot indicated the consumption of starting material. The reaction mixture was cooled to room temperature and concentrated under reduced pressure. This was re-suspended in EtOAc (500 mL), added to a 2L separatory funnel and washed successively with saturated aqueous NaHCO₃ solution (6x150 mL), eventually resulting in the aqueous layer being clear and colorless. A sample of the organic layer was concentrated and analyzed by NMR analysis, which showed complete removal of the *m*-CBA by-product from the desired formate ester.

Into the separatory funnel containing the organic layer was then added MeOH (75 mL) and saturated aqueous K₂CO₃ solution (50 mL), and this was shook for ca. 10 minutes. A sample of the resulting dark colored organic layer was concentrated and analyzed by NMR analysis, which indicated complete cleavage of the formate ester (as indicated by the loss of the formate ester peak). To the separatory funnel was then added water (600 mL) and the resulting bi-layer separated. The organic layer was then washed with brine, dried over magnesium sulfate, filtered and concentrated. Phenol **16** was recovered as a brown solid (15.8 g, 85% over three steps), and was sufficiently pure for use in the subsequent reaction without additional purification.

Allyl Ether: *R_f* 0.59 (20% EtOAc / hexanes); IR (CH₂Cl₂ cast) 2849, 1677, 1600, 1499, 1454, 1259, 1182 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 10.36 (s, 1H), 7.81 (d, *J* = 8.8 Hz, 1H), 7.46 – 7.30 (m, 5H), 6.61 (dd, *J* = 8.8, 2.1 Hz, 1H), 6.51 (d, *J* = 2.1 Hz, 1H), 6.12 – 5.96 (m, 1H), 5.43 (d, *J* = 17.2 Hz, 1H), 5.31 (d, *J* = 10.2 Hz, 1H), 5.09 (s, 2H), 4.58 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 188.4, 165.4, 162.8, 136.2, 132.5, 130.7, 129.0, 128.6, 127.8, 119.5, 118.3, 107.1, 100.1, 70.6, 69.3; HRMS (ESI) calcd for C₁₇H₁₇O₃ [M+H⁺] 269.1172; found 269.1170.

16: *R_f* 0.59 (20% EtOAc / hexanes); IR (thin film/NaCl) 3327 (bs), 3088 (w), 3035 (w), 2911 (w), 2861 (w), 1836 (w), 1621 (m), 1513 (s), 1453 (s), 1423 (m), 1404 (m), 1378 (m), 1360 (m), 1266 (s), 1235 (m), 1219 (m), 1176 (s), 1125 (s), 1027 (s), 999 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.43 – 7.31 (m, 5H), 6.84 (d, *J* = 8.7 Hz, 1H), 6.57 (d, *J* = 2.7 Hz, 1H), 6.48 (dd, *J* = 8.7, 2.7 Hz, 1H), 6.04 (ddd, *J* = 22.6, 10.6, 5.5 Hz, 1H), 5.39 (dd, *J* = 17.3, 1.3 Hz, 1H), 5.31 (dd, *J* = 10.5, 1.2 Hz, 2H), 4.99 (s, 2H), 4.56 (d, *J* = 5.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 152.5, 145.9, 140.1, 137.1, 132.6, 128.5, 127.9, 127.5, 118.4, 114.2, 106.0, 101.6, 70.8, 69.8; HRMS (FAB) calcd for C₁₆H₁₆O₃ [M⁺] 256.1099; found 256.1099.



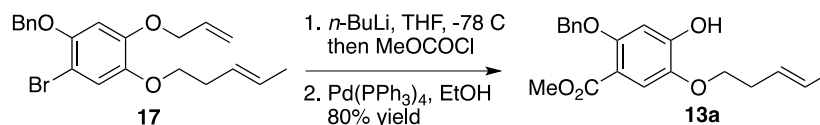
2-Allyloxy-4-benzyloxy-5-bromophenol To a stirred solution of **16** (12.0 g, 46.8 mmol, 1.0 equiv) in CH₂Cl₂ (470 mL, 0.1 M) at room temperature was added SiO₂ (11.3 g, 187 mmol, 4.0 equiv) and NBS (10.0 mL, 56.2 mmol, 1.2 equiv) and the resulting mixture stirred overnight at room temperature at which time NMR analysis of a crude reaction aliquot indicated the consumption of **16**. The reaction was then filtered through celite and washed with EtOAc (2 x 100 mL), and the filtrate concentrated under reduced

pressure. The crude oil was purified by column chromatography (30% EtOAc / hexanes until the product was recovered) to afford the bromophenol (13.3 g, 85%) as a red oil.

To a stirred solution of bromophenol (32.5 g, 97.0 mmol, 1.0 equiv) in acetone (250 mL, 0.4 M) at room temperature was added K_2CO_3 (33.5 g, 242 mmol, 2.5 equiv) and alkyl iodide **15** (47.5 g, 242.0 mmol, 2.5 equiv) and the resulting mixture heated at reflux for 48 hours, at which time TLC analysis (20% EtOAc / hexanes) indicated the near consumption of bromophenol. The reaction mixture was cooled to room temperature, and to it was added EtOAc (400 mL) and water (400 mL). The layers were separated and the aqueous layer was extracted with EtOAc (2 x 200 mL). The combined organic extracts were washed with water, brine, dried over magnesium sulfate, filtered and concentrated under reduced pressure. The resulting crude product was purified by column chromatography (5, 10 then 20% EtOAc / hexanes until the product was recovered) to afford **17** (27.3 g, 70%, 76% BRSM) and recovered bromophenol (2.7g).

Bromophenol: R_f 0.44 (30% EtOAc / hexanes); IR (CH_2Cl_2 cast) 3032, 2834, 1627, 1504, 1454, 1335, 1289, 1216, 1117 cm^{-1} ; 1H NMR (400 MHz, $CDCl_3$) δ 7.49 – 7.38 (m, 5H), 7.13 (s, 1H), 6.53 (s, 1H), 5.98 (ddt, J = 16.1, 10.8, 5.4 Hz, 1H), 5.34 (d, J = 17.6 Hz, 1H), 5.30 (d, J = 10.8 Hz, 1H), 5.05 (s, 2H), 4.50 (d, J = 5.4 Hz, 2H); ^{13}C NMR (100 MHz, $CDCl_3$) δ 148.6, 145.1, 141.3, 136.9, 132.4, 128.7, 128.1, 127.5, 118.94, 118.87, 104.2, 102.5, 72.8, 70.4; HRMS (ESI) calcd for $C_{16}H_{14}BrO_3$ $[M-H]^-$ 333.0132; found 333.0128.

17: R_f 0.61 (20% EtOAc / hexanes); IR (CH_2Cl_2 cast) 3029, 2932, 1503, 1454, 1397, 1211, 1020 cm^{-1} ; 1H NMR (500 MHz, $CDCl_3$) δ 7.41 – 7.20 (m, 5H), 7.00 (s, 1H), 6.50 (s, 1H), 5.92 (ddt, J = 16.6, 10.4, 5.1 Hz, 1H), 5.58 – 5.34 (m, 2H), 5.27 (d, J = 17.2 Hz, 1H), 5.16 (d, J = 10.4 Hz, 1H), 4.99 (s, 2H), 4.42 (d, J = 4.0 Hz, 2H), 3.86 (t, J = 7.0 Hz, 2H), 2.40 (app q, J = 6.4 Hz, 2H), 1.60 (d, J = 5.7 Hz, 3H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 149.5, 148.6, 144.4, 136.8, 133.3, 128.7, 128.1, 127.9, 127.4, 126.7, 119.4, 117.9, 104.4, 103.3, 72.3, 70.7, 70.1, 32.8, 18.2; HRMS (ESI) calcd for $C_{21}H_{24}BrO_3$ $[M+H]^+$ 403.09033; found 403.09016.



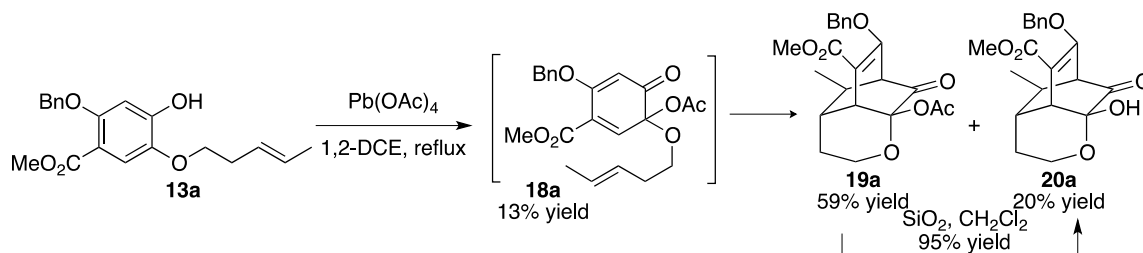
Methyl Ester Aryl bromide **17** was dried by azeotropic distillation with toluene until a constant weight (10.8 g, 26.8 mmol, 1.0 equiv) was achieved. To the flask was then added THF (270 mL, 0.1 M) and the resulting solution cooled to $-78^\circ C$. To this was added $n-BuLi$ (22.0 mL, 34.8 mmol, 1.3 equiv, 1.6 M in hexanes) and the reaction mixture was stirred at $-78^\circ C$ for 8 minutes, after which methyl chloroformate (12.4 mL, 161 mmol, 6.0 equiv) was rapidly added. The reaction was stirred for an additional 30 minutes at $-78^\circ C$ at which time TLC analysis (30% EtOAc / hexanes) indicated the consumption of **5**. The reaction mixture was warmed to room temperature, and to it was added EtOAc (200 mL) and water (200 mL). The layers were separated and the aqueous layer was extracted with EtOAc (2 x 200 mL). The combined organic extracts were

washed with water, brine, dried over magnesium sulfate, filtered and concentrated under reduced pressure. The resulting crude product was purified by column chromatography (5, 15 then 30% EtOAc / hexanes until the product was recovered) to afford the methyl ester (8.2 g, 80%) as an oil.

Phenol 13a A stirred solution of methyl ester (60 mg, 0.16 mmol, 1.0 equiv) in absolute EtOH (1.6 mL) was heated to reflux. Pd(PPh₃)₄ (16 mg, 0.016 mmol, 0.1 equiv) was then added, and the reaction was refluxed for 3 h. EtOH solvent was evaporated under reduced pressure and the residue chromatographed on silica gel (gradient elution, 4% to 14% EtOAc in hexanes) to afford **13a** as a pale yellow amorphous solid (54 mg, 100% yield).

Methyl Ester: R_f 0.48 (30% EtOAc / hexanes); IR (thin film/NaCl) 3027 (w), 2924 (m), 2857 (w), 1723 (s), 1697 (m), 1606 (m), 1585 (w), 1513 (s), 1453 (m), 1438 (s), 1412 (m), 1384 (s), 1247 (s), 1214 (s), 1079 (m), 1023 (s) cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.50 – 7.29 (m, 6H), 6.54 (s, 1H), 6 (ddd, *J* = 22.4, 10.5, 5.1 Hz, 1H), 5.61 – 5.48 (m, 2H), 5.38 (dd, *J* = 17.2, 1.4 Hz, 1H), 5.27 (dd, *J* = 10.6, 1.3 Hz, 1H), 5.13 – 5.11 (m, 2H), 4.56 (d, *J* = 5.2 Hz, 2H), 3.99 (q, *J* = 7 Hz, 2H), 3.87 (s, 3H), 2.49 (dd, *J* = 13.6, 6.8, Hz, 2H), 1.68 (d, *J* = 6.1 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 166.5, 154.8, 153.4, 143.2, 137.4, 133.0, 129.2, 128.9, 128.2, 127.5, 127.0, 118.4, 117.7, 112.7, 103.0, 72.7, 70.2, 70.1, 52.2, 33.1, 18.5; HRMS (EI) calcd for C₂₃H₂₆O₅ [M⁺] 382.1780; found 382.1767.

13a: R_f 0.46 (30% EtOAc / hexanes); IR (thin film/NaCl) 3390 (bs), 3028 (w), 2947 (m), 2879 (w), 1722 (m), 1691 (m), 1587 (m), 1513 (s), 1445 (s), 1383 (m), 1249 (s), 1216 (s), 1173 (s), 1078 (m), 1025 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.51-7.3 (m, 6H), 6.64 (s, 1H), 6.04 (s, 1H), 5.59-5.45 (m, 2H), 5.11 (s, 2H), 4.05 (t, *J* = 6.5 Hz, 2H), 3.87 (s, 3H), 2.46 (dd, *J* = 6.5, 13.1 Hz, 2H), 1.69 (dd, *J* = 1.2, 6.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.7, 155.4, 151.4, 140, 137.3, 128.9, 128.3, 128.1, 127.8, 127.6, 127.3, 126.8, 116.2, 111.5, 102.3, 71.8, 69.9, 52.3, 32.9, 18.5; HRMS (EI) calcd for C₂₀H₂₂O₅ [M+H⁺] 343.1545; found 343.1549.



Phenolic Oxidation / Diels-Alder Sequence To a stirred solution of **7** (200 mg, 0.58 mmol, 1.0 equiv) in 1,2-dichloroethane (21 mL) heated to reflux was added lead tetraacetate (340 mg, 0.77 mmol, 1.3 equiv), and the mixture was allowed to reflux for 48 h. At this point, solids were removed by filtration over a celite plug rinsing with CH₂Cl₂, and solvent was removed *in vacuo*. The residue was purified by chromatography on silica gel to afford **18a** as a pale red oil (31 mg, 13% yield), **19a** as a pale yellow oil (138 mg, 59% yield), and **20a** as a pale yellow oil (47 mg, 20% yield).

Small-Scale:

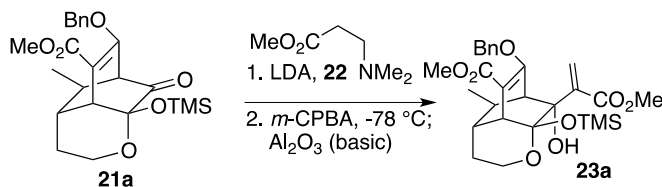
TMS Ketal 21a Hemi-ketal **20a** (23 mg, 0.066 mmol, 1.0 equiv) was dissolved in TMSCN (100 μ L, 0.73 mmol, 11 equiv) and the reaction was stirred for 5 min. Solvent (CAUTION: TMSCN liberates HCN gas upon reaction with water) was evaporated under reduced pressure to afford pure **21a** (28 mg, 99% yield).

Large-Scale:

TMS Ketal 21a To a stirred solution of **20a** (2.9 g, 8.1 mmol, 1 equiv) in CH_2Cl_2 (81 mL, 0.1 M) at -78°C was added Et_3N (1.5 mL, 10.5 mmol, 1.3 equiv) followed by TMSOTf (1.6 mL, 8.9 mmol, 1.1 equiv) and the resulting mixture stirred at -78°C for 5 minutes, at which time TLC analysis (50% EtOAc / hexanes) indicated the consumption of **9**. The reaction quenched by the addition of saturated NaHCO_3 solution (40 mL), EtOAc (100 mL) and water (100 mL), and allowed to warm to room temperature. The layers were separated and the aqueous layer was extracted with EtOAc (2 x 50 mL). The combined organic extracts were washed with water, brine, dried over magnesium sulfate, filtered and concentrated under reduced pressure. The resulting crude product was purified by column chromatography (50% EtOAc / hexanes until the product was recovered) to afford **21a** (1.9 g, 55% yield, 99% BRSM) as a yellow oil and recovered **20a** (1.3 g).

Note: A common side product observed when employing TMSCN in large-scale reactions was the cyanohydrin product. This has been partially characterized: ^1H NMR shows two TMS signals; HRMS (TOF LCMS) calcd for $\text{C}_{27}\text{H}_{39}\text{NO}_6\text{Si}_2$ $[\text{M}+\text{Na}^+]$ 552.23159, found 562.23032.

21a: R_f 0.21 (50% EtOAc / hexanes); IR (thin film/ NaCl) 2955 (s), 2873 (w), 1748 (s), 1694 (s), 1622 (s), 1498 (w), 1454 (m), 1437 (m), 1376 (m), 1333 (w), 1283 (w), 1248 (s), 1195 (s), 1152 (s), 1038 (s), 980 (m), 945 (m), 846 (s) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.39 – 7.28 (m, 5H), 5.06 (s, 2H), 3.87 (dd, J = 12.6, 5.6 Hz, 1H), 3.76 (s, 3H), 3.53 (td, J = 12.7, 2.9 Hz, 1H), 3.34 (d, J = 2.8 Hz, 1H), 3.18 (d, J = 3.4 Hz, 1H), 2.08 (qdd, J = 10.2, 6.8, 3.3 Hz, 1H), 1.98 – 1.90 (m, 1H), 1.84 (t, J = 3.3 Hz, 1H), 1.54 (dd, J = 13.6, 1.4 Hz, 1H), 1.02 (d, J = 6.9 Hz, 3H), 0.12 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 206.0, 164.9, 160.5, 136.1, 128.6, 128.2, 127.3, 109.0, 92.3, 72.2, 61.3, 58.0, 51.3, 45.7, 37.8, 35.3, 29.2, 20.5, 1.5; HRMS (EI) calcd for $\text{C}_{23}\text{H}_{31}\text{O}_6\text{Si}$ $[\text{M}+\text{H}^+]$ 431.1900; found 431.1889



Enolate of 22 To a solution of diisopropylamine (210 μ L, 1.47 mmol, 1.5 equiv) in THF (920 μ L) at -20°C was added *n*-BuLi (730 μ L, 1.17 mmol, 1.2 equiv, 1.6 M

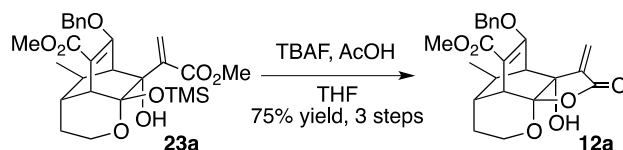
hexanes solution) dropwise over five minutes. The resultant mixture was stirred at -20°C for five minutes, and then cooled to -78°C for thirty minutes. To this mixture was added methyl-3-(dimethylamino)propionate **22** (140 μL , 0.98 mmol) dropwise over five minutes. The reaction mixture was stirred at -78°C for thirty minutes; warmed to 0°C and stirred for 15 min; warmed to room temperature and stirred for 15 min. This resulted in an orange / yellow slurry that was employed in the subsequent transformation. Any other consistency in the enolate mixture was found to be ineffective in the reaction.

Tertiary Amine A A solution of TMS ether ketone **21a** (120 mg, 0.28 mmol) in THF (2.8 mL + 1 mL) was added to enolate of **22** (0.95 mL, 0.98 mmol, 3.5 equiv, 0.5M solution) dropwise over 1 min at -78°C . The solution was slowly warmed from -78°C to 0°C over three hours. At which point the reaction mixture was cooled to -78°C and treated with 1M AcOH in THF (5mL) and allowed to warm to room temperature. At which point the reaction mixture was treated with water (5mL) and ethyl acetate (5mL). The aqueous layer was extracted with EtOAc (2 x 10mL), and the combined organic layers were washed with brine, dried over Na_2SO_4 , and concentrated *in vacuo*. Purification via silica gel chromatography (10% methanol / CH_2Cl_2) furnished **tertiary amine A** (140 mg, 89%) as an orange oily residue.

Unsaturated Ester 23a Tertiary amine (102 mg, 0.18 mmol, 1.0 equiv) was dissolved in CH_2Cl_2 (2 mL) and cooled to -78°C at which point mCPBA (61 mg, 0.27 mmol, 1.5 equiv) was added. The reaction was stirred at -78°C for 20 min at which time basic Alumina (200 mg) was added and the cold reaction mixture was quickly transferred to a short basic alumina column and filtered eluting with 10% Methanol / CH_2Cl_2 . The collected fractions contain both the desired unsaturated ester **23a** and the intermediate *N*-oxide. Upon slow rotary evaporation with warming, the *N*-oxide is converted to the desired compound. Purification via silica gel chromatography (40% EtOAc / Hexanes) furnished **23a** (88 mg, 95% yield) as a clear oily residue.

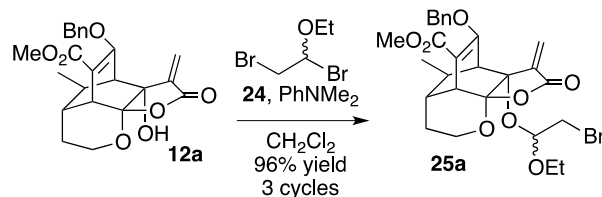
Tertiary Amine: FTIR (NaCl/thin film) 2951 (m), 1701 (s), 1624 (m), 1454 (m), 1435 (m), 1390 (m), 1375 (m), 1351 (m), 1331 (m), 1315 (w), 1295 (m), 1258 (m), 1246 (s), 1194 (m), 1182 (m), 1159 (m), 1125 (w), 1107 (w), 1089 (w), 1058 (w), 1043 (w), 1031 (w), 1013 (w), 991 (w), 956 (w), 930 (w), 911 (w), 900 (w), 864 (m), 842 (m), 791 (w), 783 (w), 757 (w), 735 (w), 697 (w), 668 (w), 649 (w), 612 (w), 594 (w), 572 (w), 557 (w), 538 (w), 524 (w), 512 (w), 504 (w) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.45 – 7.27 (m, 5H), 5.10 (d, J = 11.8 Hz, 1H), 4.92 (d, J = 11.8 Hz, 1H), 4.56 (td, J = 12.3, 4.0 Hz, 1H), 3.92 (dd, J = 11.7, 6.4 Hz, 1H), 3.75 (s, 3H), 3.52 (s, 3H), 3.07 (t, J = 10.5 Hz, 1H), 2.95 (dd, J = 10.2, 3.5 Hz, 1H), 2.92 (d, J = 2.9 Hz, 1H), 2.81 (br, 1H), 2.66 (d, J = 2.3 Hz, 1H), 2.42 (ddd, J = 7.0, 4.7, 2.3 Hz, 1H), 2.20 (s, 6H), 1.89 (tdd, J = 13.1, 6.8, 3.9 Hz, 1H), 1.62 (dt, J = 3.4, 3.2 Hz, 1H), 1.47 (t, J = 3.7 Hz, 1H), 0.91 (d, J = 7.1 Hz, 3H), 0.17 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 177.6, 165.7, 165.3, 137.0, 128.6, 128.0, 127.3, 106.0, 99.7, 80.6, 72.1, 63.2, 59.8, 54.6, 51.6, 51.3, 49.8, 47.9, 46.2, 38.9, 29.3, 28.9, 21.9, 2.2; HRMS (TOF LCMS) calcd for $\text{C}_{29}\text{H}_{44}\text{NO}_8\text{Si}$ [$\text{M}+\text{H}^+$] 562.28362; found 562.28324.

23a: FTIR (NaCl/thin film) 3445 (br), 2954 (m), 1703 (s), 1622 (m), 1453 (m), 1437 (m), 1394 (m), 1352 (m), 1324 (m), 1293 (m), 1280 (m), 1258 (m), 1246 (m), 1194 (m), 1162 (m), 1124 (m), 1116 (m), 1088 (m), 1058 (m), 1037 (m), 1022 (m), 990 (m), 957 (m), 913 (m), 864 (m), 842 (m), 814 (m), 784 (m), 735 (m), 696 (m), 661 (m), 636 (m), 608 (m), 559 (m), 538 (m), 529 (m), 512 (m), 505 (m) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.44 – 7.30 (m, 5H), 5.85 (s, 1H), 5.80 (s, 1H), 5.48 (s, 1H), 5.40 (d, J = 11.9 Hz, 1H), 4.92 (d, J = 11.9 Hz, 1H), 4.63 (td, J = 12.1, 4.3 Hz, 1H), 3.92 (dd, J = 11.8, 6.2 Hz, 1H), 3.75 (s, 3H), 3.72 (s, 3H), 2.92 (d, J = 2.8 Hz, 1H), 2.89 (d, J = 2.2 Hz, 1H), 2.52 (ddd, J = 7.0, 4.8, 2.2 Hz, 1H), 1.91 (tdd, J = 12.8, 6.9, 4.0 Hz, 1H), 1.67 (dt, J = 13.3, 3.0 Hz, 1H), 1.54 (t, J = 3.6 Hz, 1H), 1.00 (d, J = 7.1 Hz, 3H), 0.07 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 168.9, 164.6, 163.7, 141.8, 135.4, 127.0, 126.6, 126.1, 118.9, 104.8, 97.9, 79.8, 72.2, 61.4, 51.9, 50.3, 49.7, 46.0, 38.4, 27.6, 27.5, 20.2, 0.0; HRMS (TOF LCMS) calcd for $\text{C}_{27}\text{H}_{37}\text{O}_8\text{Si}$ [$\text{M}+\text{H}^+$] 517.22577; found 517.22379.



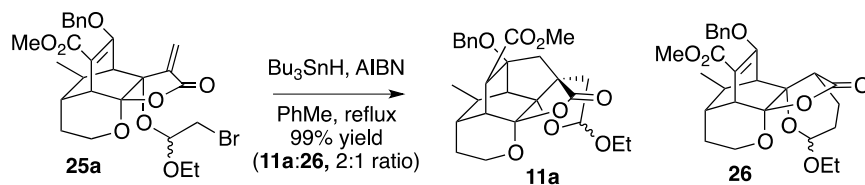
Exo-Methylene Lactone 12a To a solution of TMS-ether **23a** (43mg, 0.083 mmol) in THF (1 mL) was added glacial AcOH (24 μL , 0.416 mmol, 5 equiv) and TBAF (420 μL , 0.416 mmol, 5 equiv, 1.0 M THF solution) at 0 $^{\circ}\text{C}$. The mixture was stirred at 0 $^{\circ}\text{C}$ for 30 min, allowed to warm to room temperature for 2.5 h. At this time, additional equivalences of TBAF were added at hourly intervals until reaction was complete by TLC. The reaction was quenched by the addition of water (2 mL) and diluted with EtOAc (5 mL). The aqueous layer was extracted with EtOAc (2 x 5mL), and the combined organic layers were washed with brine, dried over sodium sulfate, and concentrated *in vacuo*. Purification via silica gel chromatography (50% EtOAc / Hexanes) furnished **12a** (30 mg, 88%) as a white foam.

12a: R_f 0.26 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 3409 (br), 2953 (w), 2924 (w), 1772 (s), 1686 (m), 1619 (m), 1454 (m), 1438 (m), 1401 (m), 1376 (m), 1317 (m), 1286 (m), 1260 (m), 1222 (m), 1194 (m), 1151 (m), 1125 (m), 1101 (m), 1083 (m), 1041 (m), 1006 (m), 986 (m), 961 (m), 911 (m), 733 (m), 697 (w) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.30-7.45 (m, 5H), 6.28 (s, 1H), 5.76 (s, 1H), 5.06 (d, J = 12.0 Hz, 1H), 4.99 (d, J = 12.0 Hz, 1H), 4.39 (ddd, J = 13.1, 9.7, 4.9 Hz, 1H), 4.11 (ddd, J = 11.4, 6.4, 4.7 Hz, 1H), 3.70 (s, 3H), 3.41 (br, 1H), 3.35 (d, J = 3.0, 1H), 3.08 (d, J = 2.3 Hz, 1H), 2.30 (ddd, J = 7.0, 4.6, 2.4 Hz, 1H), 1.98-2.10 (m, 1H), 1.66-1.78 (m, 1H), 0.97 (d, J = 7.1 Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 166.3, 165.4, 164.7, 141.1, 136.3, 128.8, 128.5, 127.5, 126.6, 104.6, 103.6, 76.2, 72.6, 64.1, 52.5, 51.8, 42.5, 39.1, 31.7, 29.0, 21.3; HRMS (TOF LCMS) calcd for $\text{C}_{23}\text{H}_{25}\text{O}_7$ [$\text{M}+\text{H}^+$] 413.16003; found 413.15974.



Bromoacetal 25a Alcohol **12a** (1.32 g, 3.25 mmol) was dissolved in CH_2Cl_2 (32.5 mL, 0.1 M) and treated with PhNMe_2 (8.3 mL, 65.0 mmol, 20 equiv) and **24** (4.35 mL, 32.5 mmol, 10 equiv) at room temperature. The solution stirred at room temperature for 16 h. The reaction was diluted with EtOAc (50 mL) and washed with 1N aqueous HCl (3 x 30 mL), followed by water and brine. The organic layer was dried over MgSO_4 , filtered, and concentrated under reduced pressure. The residue was loaded onto silica and purified by column chromatography (30 then 50% EtOAc / hexanes until the products were recovered) to yield **25a** (950 mg wet) as an orange oil, and recovered **12a** (835 mg). The recovered starting material was re-subjected to the reaction conditions to afford, after a total of 3 cycles, **25a** (1.48 g, 96% BRSM) and **12a** (0.18 g).

12a: R_f 0.51 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2972 (w), 2953 (w), 2926 (w), 1774 (s), 1703 (m), 1691 (m), 1621 (m), 1453 (m), 1438 (m), 1398 (m), 1376 (m), 1316 (m), 1286 (m), 1257 (m), 1221 (m), 1194 (m), 1150 (m), 1126 (m), 1084 (m), 1072 (m), 1055 (m), 1038 (m), 1014 (m), 913 (w), 735 (m), 697 (w) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) (1.3:1 mixture of diastereomers) δ 7.43 – 7.30 (m, 11.5H), 6.53 (s, 1.3H), 6.44 (s, 1.0H), 5.87 (s, 1.3H), 5.62 (s, 1.0H), 5.11 – 4.97 (m, 4.6H), 4.77 – 4.70 (m, 2.3H), 4.58 (td, J = 12.4, 3.9 Hz, 1.0H), 4.44 (td, J = 11.8, 4.1 Hz, 1.3H), 4.10 – 4.00 (m, 2.3H), 3.71 (s, 3.0H), 3.71 (s, 3.9H), 3.55 – 3.35 (m, 9.2H), 3.21 (m, 2.6H), 3.15 (d, J = 2.9 Hz, 1.0H), 2.97 (d, J = 2.2 Hz, 1.0H), 2.48 – 2.38 (m, 2.3H), 2.11 – 1.95 (m, 2.3H), 1.82 – 1.65 (m, 4.6H), 1.19 (t, J = 7.0 Hz, 3.9H), 1.10 (t, J = 7.0 Hz, 3H), 1.03 (d, J = 7.0 Hz, 3H), 1.00 (t, J = 7.0 Hz, 3.9H); ^{13}C NMR (100 MHz, CDCl_3) δ 166.1, 166.0, 164.9, 164.8, 164.7, 164.6, 136.5, 136.3, 136.2, 135.8, 129.0, 128.7, 128.5, 128.4, 127.7, 127.3, 126.4, 105.5, 105.0, 103.8, 97.9, 97.0, 82.0, 81.4, 73.7, 72.9, 63.6, 63.4, 62.6, 61.4, 53.8, 53.3, 51.8, 45.2, 44.3, 40.4, 39.8, 32.5, 32.3, 29.8, 29.2, 28.8, 28.7, 21.5, 21.4, 15.2, 15.0; HRMS (TOF LCMS) calcd for $\text{C}_{27}\text{H}_{32}\text{BrO}_8$ $[\text{M}+\text{H}^+]$ 563.12806, found 563.12756.

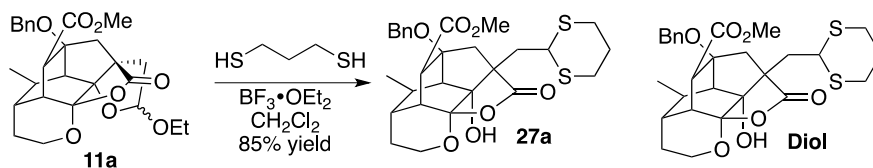


Ethyl Acetal 11a Bromide **25a** (177 mg, 0.315 mmol) was dissolved in toluene (10 mL) and to this was added Bu_3SnH (210 μL , 0.787 mmol, 2.5 equiv) followed by AIBN (155 mg, 0.945 mmol, 3.0 equiv). The mixture was placed in a 115 $^\circ\text{C}$ oil bath and stirred at reflux for one hour, at which time TLC analysis (50% EtOAc / hexanes) indicated the consumption of starting material. The reaction was cooled to room temperature and the solvent was removed under reduced pressure. The residue was purified by column chromatography (10, 25 then 40% EtOAc / Hexanes until the product

was recovered) to yield **11a** (101 mg, 66%) as an oil and cyclization byproduct **26** (52 mg, 33%).

11a: R_f 0.48 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2953 (m), 2923 (m), 2912 (m), 2359 (m), 2342 (m), 2331 (m), 1786 (s), 1731 (m), 1454 (m), 1435 (m), 1360 (m), 1327 (m), 1299 (m), 1275 (m), 1255 (m), 1245 (m), 1206 (m), 1173 (m), 1142 (m), 1133 (m), 1109 (m), 1071 (m), 1051 (m), 1030 (m), 1003 (m), 986 (m), 961 (m), 934 (m), 914 (m), 886 (w), 734 (w), 697 (w), 677 (w), 668 (w) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) (~1:1 mixture of diastereomers) δ 7.26–7.37 (m, 10H), 5.48 (dd, J = 5.7, 2.3 Hz, 1H), 5.32 (d, J = 4.1 Hz, 1H), 4.49–4.59 (m, 4H), 4.19–4.30 (m, 2H), 3.90–4.05 (m, 2H), 3.75–3.84 (m, 2H), 3.71 (s, 3H), 3.70 (s, 3H), 3.50–3.57 (m, 3H), 3.31–3.40 (m, 1H), 2.83 (d, J = 8.8 Hz, 2H), 2.77 (dd, J = 14.2, 5.8 Hz, 1H), 2.65–2.72 (m, 2H), 2.56 (d, J = 1.8 Hz, 1H), 2.53 (d, J = 1.8 Hz, 1H), 2.02–2.28 (m, 6H), 1.90 (dd, J = 12.9, 4.2 Hz, 1H), 1.55–1.75 (m, 6H), 1.31 (d, J = 7.1 Hz, 3H), 1.29 (d, J = 7.1 Hz, 3H), 1.23 (t, J = 7.1 Hz, 3H), 1.13 (t, J = 7.0 Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) 176.8, 176.1, 171.8, 138.5, 128.5, 127.5, 127.0, 126.9, 109.8, 107.5, 106.9, 106.2, 95.2, 94.6, 86.1, 84.7, 66.4, 66.2, 64.1, 63.0, 62.6, 62.3, 56.7, 56.1, 54.0, 53.2, 52.4, 51.2, 44.2, 43.9, 43.1, 42.8, 41.2, 41.1, 40.9, 39.9, 30.8, 30.7, 27.7, 27.5, 20.9, 20.8, 15.4, 14.6; HRMS (TOF LCMS) calcd for $\text{C}_{27}\text{H}_{33}\text{O}_8$ $[\text{M}+\text{H}^+]$ 485.21754, found 485.21763.

26: R_f 0.40 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2971, 1781, 1701, 1623, 1439, 1379, 1128, 1087, 1012 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.40–7.20 (m, 5H), 5.22 (d, J = 12.4 Hz, 1H), 4.96 (d, J = 12.4 Hz, 1H), 4.62 (dd, J = 4.6, 4.3 Hz, 1H), 4.41 (ddd, J = 12.3, 12.3, 3.8 Hz, 1H), 3.95 (dd, J = 12.0, 6.6 Hz, 1H), 3.87 (dt, J = 8.9, 7.2 Hz, 1H), 3.67 (s, 3H), 3.45 (dt, J = 9.0, 7.0 Hz, 1H), 3.12 (d, J = 2.7 Hz, 1H), 3.01 (d, J = 1.9 Hz, 1H), 2.32 (m, 1H), 2.25 (m, 1H), 2.02 – 1.88 (m, 2H), 1.70 (m, 1H), 1.63 (m, 1H), 1.52 – 1.37 (m, 3H), 1.17 (t, J = 7.0 Hz, 3H), 1.02 (d, J = 7.0 Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 174.6, 165.3, 164.5, 136.3, 128.9, 128.6, 127.5, 106.0, 104.7, 97.2, 81.4, 72.6, 64.3, 63.2, 51.5, 50.4, 43.9, 43.0, 40.3, 29.7, 28.5, 26.8, 21.6, 18.5, 15.1; HRMS (ESI) calcd for $\text{C}_{27}\text{H}_{33}\text{O}_8$ $[\text{M}+\text{H}^+]$ 485.2170; found 485.2173.



Dithiane 27a Acetal **11a** (35 mg, 0.072 mmol) was dissolved in CH_2Cl_2 (750 μL , 0.1 M), cooled to 0 $^\circ\text{C}$ and treated with $\text{BF}_3\cdot\text{OEt}_2$ (27 μL , 0.18 mmol, 2.5 equiv) and propanedithiol (18 μL , 0.18 mmol, 2.5 equiv). The reaction mixture was stirred at 0 $^\circ\text{C}$ for 45 min at which time TLC analysis (50% EtOAc / hexanes) indicated the consumption of starting material. The reaction was quenched with aqueous saturated NaHCO_3 (2 mL) and diluted with EtOAc (10 mL). The aqueous layer was separated and washed with EtOAc (2 x 5 mL). The combined organic layers were washed with brine, dried over Na_2SO_4 , filtered, and concentrated under reduced pressure. The residue was purified by column chromatography (20, 40 then 50% EtOAc / Hexanes until the product

was recovered) to yield **27a** (33 mg, 85%) as a white film and a trace of **diol** as a white foam.

27a: R_f 0.44 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 3451 (br), 2925 (s), 1779 (s), 1735 (s), 1322 (m), 1139 (m), 1081 (m), 983 (m), 669 (s) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.36 – 7.22 (m, 5H), 4.59 (d, J = 11.8 Hz, 1H), 4.53 (d, J = 11.8 Hz, 1H), 4.48 (t, J = 6.5 Hz, 1H), 4.35 (td, J = 11.5, 4.5 Hz, 1H), 4.07 (ddd, J = 11.7, 6.4, 2.6 Hz, 1H), 3.71 (s, 3H), 3.31 (d, J = 4.8 Hz, 1H), 3.21 (d, J = 13.6 Hz, 1H), 2.94 – 2.80 (m, 5H), 2.55 (d, J = 2.0 Hz, 1H), 2.44 – 2.32 (m, 3H), 2.15 (dd, J = 15.3, 6.4 Hz, 1H), 2.11 – 1.97 (m, 3H), 1.96 – 1.82 (m, 2H), 1.65 – 1.60 (m, 1H), 1.29 (d, J = 7.4 Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.0, 171.6, 138.7, 128.5, 127.5, 127.0, 105.5, 83.9, 80.7, 66.1, 63.9, 53.2, 52.5, 52.1, 51.2, 42.1, 41.3, 41.0, 40.4, 39.0, 30.4, 30.2, 27.9, 25.4, 20.8; HRMS (TOF LCMS) calcd for $\text{C}_{28}\text{H}_{35}\text{O}_7\text{S}_2$ [$\text{M}+\text{H}^+$] 547.18242; found 547.18182.

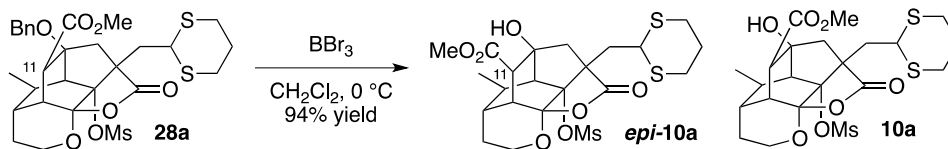
Diol: R_f 0.12 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 3461 (br), 2953 (s), 2910, 1771 (s), 1454, 1435, 1267, 1243, 1204, 1072, 983 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3) δ 4.46 (t, J = 6.3 Hz, 1H), 4.34 (ddd, J = 11.7, 11.7, 3.4 Hz, 1H), 4.07 (m, 1H), 3.75 (s, 3H), 3.38 (br. s, 1H), 2.92 – 2.78 (m, 5H), 2.48 (s, 1H), 2.37 (t_{AB} , J = 16.0 Hz, 2H), 2.30 (dd, J = 15.7, 6.3 Hz, 1H), 2.19 (s, 1H), 2.11 (dd, J = 15.2, 6.3 Hz, 1H), 2.07 – 1.92 (m, 4H), 1.92 – 1.79 (m, 3H), 1.55 (m, 1H), 1.27 (d, J = 7.3 Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 176.1, 171.4, 105.6, 84.8, 76.6, 63.8, 54.9, 54.0, 52.6, 51.3, 47.5, 42.1, 40.8, 38.8, 38.7, 30.3, 30.2, 28.4, 25.4, 20.9; HRMS (ESI) calcd for $\text{C}_{21}\text{H}_{29}\text{O}_7\text{S}_2$ [$\text{M}+\text{H}^+$] 457.1349; found 457.1352.



Mesylate 28a Alcohol **27a** (705 mg, 1.29 mmol) was dissolved in CH_2Cl_2 (13 mL, 0.1 M), cooled to 0 °C and treated with Et_3N (1.8 mL, 12.9 mmol, 10 equiv), DMAP (315 mg, 2.6 mmol, 2.0 equiv) and then MsCl (752 μL , 9.68 mmol, 7.5 equiv). The solution was allowed to stir and warm to room temperature overnight at which time TLC analysis (50% EtOAc / hexanes) indicated the consumption of starting material. The reaction was quenched with aqueous saturated NaHCO_3 (15 mL) and diluted with EtOAc (20 mL). The aqueous layer was extracted with EtOAc (2 x 15 mL). The combined organic layer was washed with water, brine, dried over MgSO_4 and concentrated. The residue was purified by column chromatography (40% EtOAc / hexanes until the product was recovered) to provide mesylate **28a** (710 mg, 88%) as a white foam.

28a: R_f 0.53 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2924 (ms), 1781 (m), 1735 (m), 1323 (m), 1185 (m), 1134 (m), 1081 (m), 981 (m) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.24–7.36 (m, 5H), 4.85 (dd, J = 8.5, 3.4 Hz, 1H), 4.59 (d, J = 11.7 Hz, 1H), 4.55 (d, J = 11.7 Hz, 1H), 4.38 (td, J = 12.3, 3.8 Hz, 1H), 4.03 (dd, J = 12.0, 6.0 Hz, 1H), 3.72 (s, 3H), 3.35 (d, J = 2.3 Hz, 1H), 3.35 (d, J = 13.5 Hz, 1H), 3.24 (s, 3H), 2.94 (td, J = 27.1, 2.3 Hz, 1H), 2.94 (dt, J = 28.3, 2.4 Hz, 1H), 2.85 (s, 1H), 2.74–2.84 (m, 2H), 2.62 (dd, J =

13.5, 1.4 Hz, 1H), 2.49-2.58 (m, 3H), 2.25 (dd, $J = 14.6, 3.4$ Hz, 1H), 1.90-2.15 (m, 4H), 1.60-1.66 (m, 1H), 1.31 (d, $J = 7.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 174.1, 171.2, 138.2, 128.5, 127.6, 127.1, 105.1, 98.0, 81.0, 66.2, 62.7, 52.6, 51.5, 51.4, 50.8, 42.1, 41.6, 40.9, 40.8, 40.5, 40.2, 31.3, 30.6, 30.1, 27.3, 25.5, 20.7; HRMS (TOF LCMS) calcd for $\text{C}_{29}\text{H}_{37}\text{O}_9\text{S}_3$ $[\text{M}+\text{H}^+]$ 625.15997; found 625.15963.

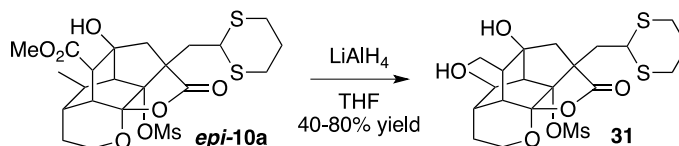


Tertiary alcohols *epi-10a* and *10a* Benzyl ether **28a** (389 mg, 0.62 mmol) was dissolved in CH_2Cl_2 (16 mL, 0.04M) and cooled to 0 °C. To this was added successive portions of BBr_3 (59 μL , 0.62 mmol, 1.0 equiv), and the reaction monitored by TLC (20% EtOAc / benzene) after each portion of BBr_3 . Upon consumption of starting material the reaction was quenched by addition of aqueous saturated NaHCO_3 (10 mL) and diluted with EtOAc (15 mL). The aqueous layer was separated and washed with EtOAc (2 x 15mL). The combined organic extracts were washed with water, brine, dried over Na_2SO_4 and concentrated *in vacuo*. The residue was purified by column chromatography (20, 40 then 60% EtOAc/hexanes until the products were recovered) to provide alcohol **epi-10a** (315 mg, 94%) as a white foam as well as **10a** (12 mg, 4%) as a pale oil.

epi-10a: R_f 0.33 (20% EtOAc/benzene); FTIR (NaCl/thin film) 3848 (w), 2932 (w), 1777 (s), 1770 (s), 1746 (m), 1731 (m), 1722 (m), 1712 (m), 1698 (m), 1680 (m), 1650 (m), 1469 (m), 1462 (m), 1453 (m), 1441 (m), 1434 (m), 1427 (m), 1416 (m), 1337 (s), 1221 (m), 1185 (m), 1151 (s), 1075 (m), 1024 (m), 1000 (m), 969 (m), 958 (m), 916 (m), 883 (m), 845 (m), 814 (m), 778 (m), 735 (m), 701 (m), 631 (w), 579 (w), 527 (w), 505 (w) cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 5.13 (s, 1H), 4.75 (dd, $J = 8.6, 4.2$ Hz, 1H), 4.38 (td, $J = 12.2, 4.0$ Hz, 1H), 4.04 (dd, $J = 12.4, 5.7$ Hz, 1H), 3.77 (s, 3H), 3.26 (s, 3H), 3.05 (d, $J = 2.3$ Hz, 1H), 2.95 (td, $J = 25.3, 2.6$ Hz, 1H), 2.95 (dt, $J = 25.0, 2.7$ Hz, 1H), 2.78-2.87 (m, 2H), 2.73 (d, $J = 3.2$ Hz, 1H), 2.67 (d, $J = 13.6$ Hz, 1H), 2.48-2.56 (m, 2H), 2.32 (dd, $J = 14.7, 4.2$ Hz, 1H), 2.21-2.28 (m, 2H), 2.04-2.14 (m, 1H), 1.79-1.98 (m, 3H), 1.56-1.65 (m, 1H), 1.29 (d, $J = 7.5$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 175.0, 173.4, 104.7, 98.4, 73.9, 62.8, 53.0, 52.4, 51.8, 51.4, 46.3, 42.9, 42.4, 40.9, 39.8, 31.8, 31.1, 30.5, 29.6, 26.5, 25.5, 19.7; HRMS (TOF LCMS) calcd for $\text{C}_{22}\text{H}_{31}\text{O}_9\text{S}_3$ $[\text{M}+\text{H}^+]$ 535.11302; found 535.11563.

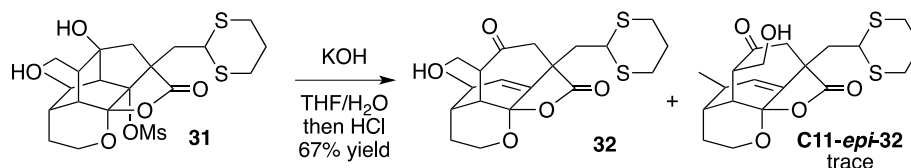
10a: R_f 0.15 (20% EtOAc/benzene); FTIR (NaCl/thin film) 3494, 2928, 1778, 1735, 1337, 1184, 1144, 1107, 1081 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 4.72 (dd, $J = 8.5, 3.6$ Hz, 1H), 4.29 (ddd, $J = 12.4, 12.4, 3.8$ Hz, 1H), 3.96 (dd, $J = 12.1, 6.2$ Hz, 1H), 3.69 (s, 3H), 3.18 (s, 3H), 3.12 (d, $J = 2.3$ Hz, 1H), 2.96 – 2.78 (m, 2H), 2.77 – 2.66 (m, 3H), 2.56 – 2.39 (m, 4H), 2.37 (d, $J = 2.7$ Hz, 1H), 2.15 (dd, $J = 14.7, 3.6$ Hz, 1H), 2.08 – 1.75 (m, 4H), 1.54 (m, 1H), 1.25 (d, $J = 7.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 174.1, 170.8, 105.0, 98.2, 76.8, 62.5, 53.8, 52.6, 51.8, 51.4, 46.7, 42.0, 40.9, 40.6, 39.9, 39.5, 31.1,

30.4, 30.0, 27.6, 25.4, 20.7; HRMS (ESI) calcd for C₂₂H₃₀O₉S₃Na [M+Na⁺] 557.0944; found 557.0943.



Diol 31 To a stirred solution of **epi-10a** (19 mg, 0.035 mmol, 1 eq) in THF (1.8 mL, 0.02 M) was added LiAlH₄ (1.2 mg, 0.032 mmol, 0.9 eq) and the mixture was vigorously stirred at RT for up to 15 minutes while monitoring by TLC (80% EtOAc / hexanes). Upon consumption of starting material the reaction was quenched according to the Fieser & Fieser protocol, filtered through celite and concentrated by rotary evaporation. The product was purified by column chromatography (30, 50, then 80% EtOAc / hexanes until the product was recovered) to afford **31** (14 mg, 78%) as a white foam. The yield of this transformation is variable, and typically falls within 40-80%.

31: R_f 0.45 (80% EtOAc / hexanes); IR (CH₂Cl₂ cast) 3448 br., 2926, 1773, 1336, 1079, 1024 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 4.77 (dd, *J* = 8.6, 3.6 Hz, 1H), 4.34 (ddd, *J* = 12.2, 12.2, 3.8 Hz, 1H), 4.13 (ddd, *J* = 10.4, 10.4, 3.8 Hz, 1H), 3.99 (dd, *J* = 12.1, 6.0 Hz, 1H), 3.68 – 3.60 (m, 1H), 3.22 (s, 3H), 3.13 (s, 1H), 3.01 (s, 1H), 3.00 – 2.86 (m, 2H), 2.80 – 2.75 (m, 2H), 2.56 – 2.44 (m, 2H), 2.31 (d, *J* = 12.6 Hz, 1H), 2.25 (d, *J* = 14.7, 3.6 Hz, 1H), 2.14 (dd, *J* = 5.1, 5.1 Hz, 1H), 2.12 – 2.03 (m, 1H), 1.94 – 1.75 (m, 4H), 1.55 (m, 1H), 1.30 (d, *J* = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 175.4, 105.3, 98.5, 75.2, 63.1, 62.6, 54.2, 52.3, 51.5, 43.7, 42.5, 42.1, 40.9, 40.1, 31.3, 31.3, 30.6, 29.9, 26.8, 25.6, 20.1; HRMS (TOF LCMS) calcd for C₂₁H₃₀O₈S₃Na [M+Na⁺] 529.0995; found 529.0994.

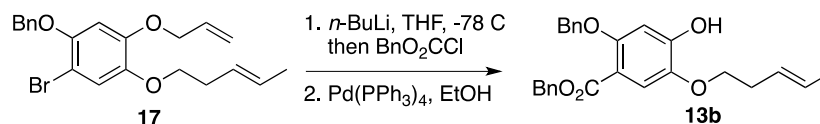


Fragmentation Products 32 and epi-32 To a stirred solution of **31** (13 mg, 0.026 mmol, 1 eq) in THF (1.3 mL, 0.02M) and water (6 μL, 0.5μL/mg) at RT was added solid KOH (5.0 mg, 0.090 mmol, 3.5 eq) and the mixture was vigorously stirred overnight while monitoring by TLC (10% MeOH / CH₂Cl₂). Upon consumption of starting material the reaction was quenched with 200μL of 1N HCl and stirred for 30 minutes. This was then diluted with EtOAc (10 mL) and water (10 mL) and the layers separated. The aqueous layer was then extracted EtOAc (2 x 5 mL) and the combined organic extracts washed with water, dried and dried over magnesium sulfate. This was filtered and concentrated by rotary evaporation. The product was purified by column chromatography (1, 2, 4 then 6% MeOH / CH₂Cl₂ until the product was recovered) to afford **32** (7.0 mg, 67 %) as an oil, as well as a trace of **epi-32**.

32: R_f 0.51 (10% MeOH / CH₂Cl₂); IR (CH₂Cl₂ cast) 3371 br., 2926, 1771, 1697, 1424, 1337, 1226, 1117 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.54 (d, J = 3.6 Hz, 1H), 4.30 (br. s, 1H), 4.15 (dd, J = 8.3, 8.3 Hz, 1H), 3.93 – 3.65 (m, 4H), 3.00 (d, J = 3.4 Hz, 1H), 2.93 (m, 1H), 2.88 – 2.67 (m, 5H), 2.83 (d, J = 15.1 Hz, 1H), 2.50 (d, J = 15.2 Hz, 1H), 2.36 (ddd, J = 7.4, 7.4, 3.4 Hz, 1H), 2.13 (dd, J = 14.2, 9.4 Hz, 1H), 2.08 – 1.99 (m, 1H), 1.96 – 1.87 (m, 1H), 1.83 – 1.74 (m, 2H), 1.48 (ddd, J = 8.7, 8.5, 4.1 Hz, 1H), 1.08 (d, J = 7.0 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 208.1, 177.4, 141.8, 133.9, 107.5, 61.6, 60.3, 59.5, 52.9, 49.8, 47.9, 43.4, 38.7, 38.2, 37.8, 34.7, 29.1, 28.8, 25.5, 19.4; HRMS (ESI) calcd for C₂₀H₂₇O₅S₂ [M+H⁺] 411.1294; found 411.1294.

epi-32: R_f 0.70 (10% MeOH / CH₂Cl₂); IR (CH₂Cl₂ cast) 3371 br., 2926, 1771, 1697, 1424, 1337, 1226, 1117 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.87 (s, 1H), 4.01 (dd, J = 10.0, 6.6 Hz, 1H), 3.97 – 3.81 (m, 2H), 3.78 – 3.68 (m, 2H), 2.99 – 2.85 (m, 3H), 2.69 (m, 1H), 2.63 (d, J = 11.5 Hz, 1H), 2.62 – 2.49 (m, 2H), 2.45 (d, J = 11.3 Hz, 1H), 2.37 (d, J = 5.8 Hz, 1H), 2.23 (dd, J = 14.7, 6.6 Hz, 1H), 2.18 (m, 1H), 2.11 (s, 1H), 2.00 – 1.89 (m, 3H), 1.75 – 1.70 (m, 1H), 1.27 (d, J = 6.9 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 208.5, 174.6, 138.4, 132.1, 105.0, 61.5, 60.9, 60.5, 55.6, 50.5, 43.1, 40.6, 39.5, 37.0, 35.5, 32.8, 27.6, 27.0, 25.4, 23.6; HRMS (ESI) calcd for C₂₀H₂₇O₅S₂ [M+H⁺] 411.1294; found 411.1297.

Benzyl Ester Series:



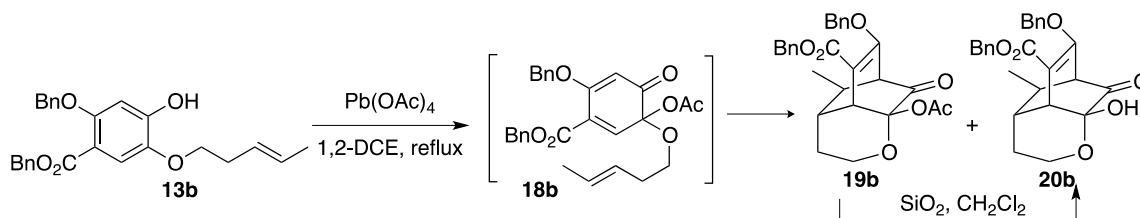
Benzyl Ester Following the procedure for the synthesis of **13a**, compound **17** (1.85g, 5.0 mmol, 1eq) was converted to **benzyl ester** using benzyl chloroformate (2.6 mL, 17.8 mmol, 3.0 eq). The resulting crude product was purified by column chromatography (5, 10 then 20% EtOAc / hexanes until the product was recovered) to afford **benzyl ester** (2.44 g, 89%) as a yellow oil.

Phenol 13b Following the procedure for the synthesis of **13a**, compound **benzyl ester** (2.44 g, 5.31 mmol, 1eq) was converted to **13b**. The resulting crude product was purified by column chromatography (10, 20 then 30% EtOAc / hexanes until the product was recovered) to afford **13b** (2.15 g, 97%) as a clear, colorless oil.

Benzyl Ester: R_f 0.68 (30% EtOAc / hexanes); IR (thin film/NaCl) 3031, 2935, 1720, 1694, 1608, 1514, 1454, 1421, 1383, 1240, 1210, 1024 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.52-7.27 (m, 11H), 6.54 (s, 1H), 5.98 (ddd, J = 17.2, 10.4, 5.2 Hz, 1H), 5.63 – 5.44 (m, 2H), 5.38 (dd, J_{AB} = 17.0 Hz, 1H), 5.34 (s, 2H), 5.27 (d, J = 11.1 Hz, 1H), 5.09 (s, 2H), 4.57 (d, J = 5.1 Hz, 2H), 3.98 (t, J = 7.1 Hz, 2H), 2.48 (q, J = 7.0 Hz, 2H), 1.67 (d, J = 5.7 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 165.9, 154.6, 153.2, 142.8, 136.9, 136.5, 132.8, 128.7, 128.6, 128.3, 128.1, 127.9, 127.4, 127.2, 126.7, 118.1, 117.8, 112.3,

102.4, 72.3, 69.95, 69.88, 66.6, 32.7, 18.2; HRMS (ESI) calcd for C₂₉H₃₀O₅ [M+Na⁺] 481.1986; found 481.1986.

13b: R_f 0.61 (30% EtOAc / hexanes); IR (thin film/NaCl) 3353 (bs), 2921 (m), 1686 (m), 1586 (m), 1512 (s), 1439 (s), 1382 (m), 1217 (s), 1173 (s), 1025 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.51 – 7.28 (m, 10H), 6.65 (s, 1H), 6.05 (s, 1H), 5.66 – 5.41 (m, 2H), 5.33 (s, 2H), 5.08 (s, 2H), 4.04 (t, *J* = 6.6 Hz, 2H), 2.45 (q, *J* = 6.4 Hz, 2H), 1.69 (d, *J* = 6.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.1, 155.2, 151.2, 139.7, 136.8, 136.6, 128.7, 128.3, 128.1, 127.9, 127.4, 126.6, 116.3, 111.3, 101.8, 71.6, 69.7, 66.7, 32.7, 18.2; HRMS (ESI) calcd for C₂₆H₂₆O₅Na [M+Na⁺] 441.16725; found 441.16735.



Phenolic Oxidation / Diels-Alder Sequence Following the procedure for the synthesis of **19a** from **13a**, compound **13b** (1.36 g, 3.25 mmol, 1eq) was converted to compounds **19b** and **18b**. The crude product was purified by column chromatography (20, 30 then 40% EtOAc / hexanes until the product was recovered) to afford **19b** (1.09 g, 71%) and **18b** (177 mg, 11%).

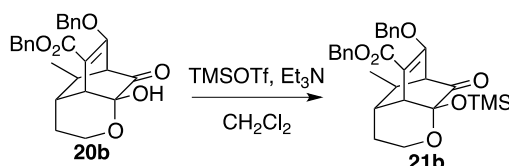
Hemiketal 20b Following the procedure for the synthesis of **20a** from **19a**, compound **19b** (5.45 g, 11.4 mmol, 1eq) was converted to compound **20b**. The crude product was purified by column chromatography (10, 20, 30 then 40% EtOAc / hexanes until the product was recovered) to afford **20b** (3.8 g, 76%) and recovered **19b** (1.4 g, 23%).

18b: R_f 0.41 (30% EtOAc / hexanes); IR (thin film/NaCl) 3033, 2949, 1741, 1673, 1580, 1499, 1456, 1366, 1236, 1126, 1070, 1012 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.36 – 7.21 (m, 10H), 6.84 (s, 1H), 5.61 (s, 1H), 5.56 – 5.41 (m, 1H), 5.37 – 5.27 (m, 1H), 5.18 (q, *J*_{AB} = 12.3 Hz, 2H), 4.97 (q, *J*_{AB} = 12.1 Hz, 2H), 3.81 – 3.72 (m, 1H), 3.70 – 3.61 (m, 1H), 2.21 (q, *J* = 7.1 Hz, 2H), 2.09 (s, 3H), 1.61 (dd, *J* = 6.2, 1.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 189.8, 169.7, 165.1, 163.6, 141.2, 135.2, 134.4, 130.3, 128.9, 128.8, 128.7, 128.6, 127.8, 126.5, 125.5, 100.9, 92.8, 71.7, 67.8, 64.5, 64.1, 33.2, 20.9, 18.2; HRMS (ESI) calcd for C₂₈H₂₈O₇Na [M+Na⁺] 499.1727; found 499.1722.

19b: R_f 0.21 (30% EtOAc / hexanes); IR (thin film/NaCl) 3032, 2957, 1747, 1682, 1621, 1454, 1406, 1368, 1240, 1189, 1089 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.39 – 7.27 (m, 10H), 5.21 (q, *J*_{AB} = 13.1 Hz, 2H), 5.17 (d, *J*_{AB} = 11.5 Hz, 1H), 5.05 (d, *J*_{AB} = 11.5 Hz, 1H), 4.07 (ddd, *J* = 12.2, 6.2, 2.6 Hz, 1H), 3.81 (d, *J* = 3.4 Hz, 1H), 3.67 (ddd, *J* = 12.0, 12.0, 3.6 Hz, 1H), 3.57 (d, *J* = 2.4 Hz, 1H), 2.11 – 1.98 (m, 2H), 1.91 (m, 1H), 1.84 (s, 3H), 1.58 (dddd, *J* = 13.8, 6.4, 3.2, 3.2 Hz, 1H), 0.95 (d, *J* = 6.9 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 200.7, 168.8, 164.4, 163.4, 136.4, 136.1, 128.8, 128.7, 128.5, 128.3,

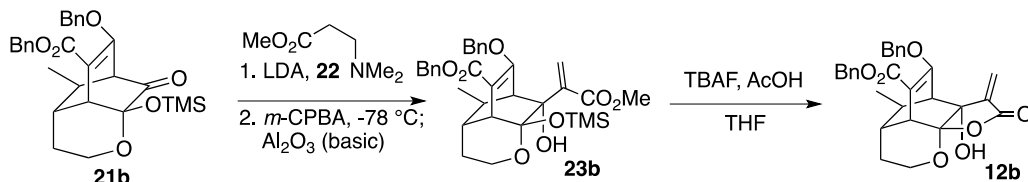
128.2, 127.8, 105.1, 94.9, 72.5, 62.3, 62.1, 57.7, 42.6, 37.9, 36.4, 28.9, 20.9, 20.8; HRMS (ESI) calcd for $C_{28}H_{28}O_7Na$ $[M+Na^+]$ 499.1727; found 499.1732.

20b: R_f 0.08 (30% EtOAc / hexanes); IR (thin film/NaCl) 3390, 2957, 2925, 2871, 1741, 1689, 1617, 1455, 1403, 1363, 1323, 1250, 1176, 1092 cm^{-1} ; 1H NMR (400 MHz, $CDCl_3$) δ 7.38 – 7.21 (m, 10H), 5.21 (s, 2H), 5.03 (s, 2H), 3.88 (dd, J = 12.3, 4.9 Hz, 1H), 3.55 (ddd, J = 12.2, 12.2, 3.4 Hz, 1H), 3.47 (br. s, 1H), 3.44 (d, J = 2.8 Hz, 1H), 3.33 (d, J = 3.4 Hz, 1H), 2.11 (m, 1H), 1.97 (m, 1H), 1.89 (m, 1H), 1.58 (m, 1H), 1.03 (d, J = 6.9 Hz, 3H); ^{13}C NMR (100 MHz, $CDCl_3$) δ 207.2, 164.5, 160.6, 136.3, 135.8, 128.8, 128.7, 128.5, 128.2, 128.1, 127.7, 108.9, 90.7, 72.5, 66.6, 61.6, 57.4, 43.3, 37.9, 35.2, 29.4, 20.6; HRMS (ESI) calcd for $C_{26}H_{26}O_6Na$ $[M+Na^+]$ 457.16216; found 457.16237.



TMS Ketal 21b Following the procedure for the synthesis of **21a** from **20a**, compound **20b** (70 mg, 0.16 mmol, 1eq) was converted to compound **21b**. The crude product was purified by column chromatography (10, 20 then 30% EtOAc / hexanes until the product was recovered) to afford **21b** (66 mg, 81%) as an oil.

21b: R_f 0.45 (30% EtOAc / hexanes); IR (thin film/NaCl) 2957, 1747, 1693, 1621, 1455, 1402, 1248, 1194, 1093, 1037 cm^{-1} ; 1H NMR (500 MHz, $CDCl_3$) δ 7.39 – 7.16 (m, 10H), 5.20 (d, J_{AB} = 12.8 Hz, 1H), 5.16 (d, J_{AB} = 12.8 Hz, 1H), 4.99 (q, J_{AB} = 11.7 Hz, 2H), 3.81 (dd, J = 12.7, 5.4 Hz, 1H), 3.48 (ddd, J = 12.7, 12.7, 2.4 Hz, 1H), 3.34 (d, J = 2.9 Hz, 1H), 3.17 (d, J = 3.4 Hz, 1H), 2.03 (m, 1H), 1.88 (m, 1H), 1.80 (m, 1H), 1.47 (m, 1H), 0.98 (d, J = 6.9 Hz, 3H), 0.04 (s, 9H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 206.1, 164.5, 160.8, 136.5, 135.9, 128.7, 128.6, 128.3, 128.1, 127.9, 127.5, 108.8, 92.4, 72.1, 66.1, 61.4, 57.9, 45.8, 37.8, 35.4, 29.3, 20.6, 1.6; HRMS (ESI) calcd for $C_{29}H_{34}O_6SiNa$ $[M+Na^+]$ 529.20224; found 529.20206.

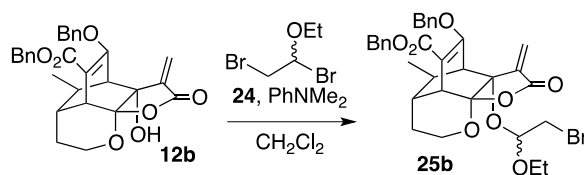


Tertiary Amine B, Unsaturated Ester 23b and Exo-Methylene Lactone 12b

Following the procedure for the synthesis of **12a**, compound **21b** (2.6 g, 5.1 mmol, 1eq) was carried through the exact three-step synthesis, affording **12b** (1.8 g, 73%) as a white foam.

23b: FTIR (NaCl/thin film) 3441, 2955, 1701, 1625, 1439, 1400, 1317, 1247, 1190, 1088, 1057, 1035, 864 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.41 – 7.28 (m, 10H), 5.85 (s, 1H), 5.82 (s, 1H), 5.48 (s, 1H), 5.30 (d, J_{AB} = 12.0 Hz, 1H), 5.22 (d, J_{AB} = 12.4 Hz, 1H), 5.18 (d, J_{AB} = 12.4 Hz, 1H), 4.93 (d, J_{AB} = 11.8 Hz, 1H), 4.59 (ddd, J = 11.9, 11.9, 4.5 Hz, 1H), 3.92 (dd, J = 11.2, 6.0 Hz, 1H), 3.76 (s, 3H), 2.95 (m, 2H), 2.52 (ddd, J = 7.2, 5.1, 2.3 Hz, 1H), 1.89 (dddd, J = 12.0, 12.0, 6.8, 4.2 Hz, 1H), 1.68 (m, 1H), 1.56 (m, 1H), 1.00 (d, J = 7.1 Hz, 3H), 0.02 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 170.6, 166.1, 165.0, 143.5, 136.9, 136.5, 128.7, 128.3, 128.2, 127.7, 120.7, 106.4, 99.6, 81.5, 73.5, 66.2, 63.0, 53.3, 52.0, 47.6, 40.0, 29.4, 29.3, 21.9, 1.6; HRMS (ESI) calcd for $\text{C}_{33}\text{H}_{40}\text{O}_8\text{SiNa}$ $[\text{M}+\text{Na}^+]$ 615.23847; found 615.23902.

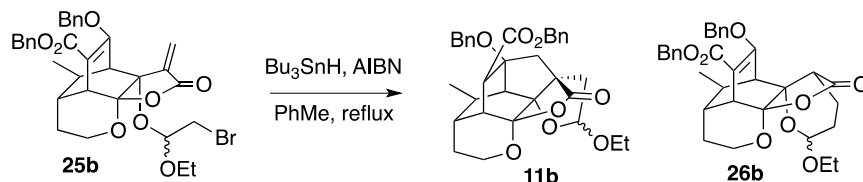
12b: R_f 0.40 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 3423, 2956, 2925, 1773, 1686, 1619, 1455, 1408, 1286, 1260, 1193, 1125, 1083, 1007 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.32 – 7.15 (m, 10H), 6.24 (s, 1H), 5.72 (s, 1H), 5.14 (d, J_{AB} = 12.8 Hz, 1H), 5.06 (d, J_{AB} = 12.8 Hz, 1H), 4.95 (d, J_{AB} = 11.7 Hz, 1H), 4.89 (d, J_{AB} = 11.7 Hz, 1H), 4.33 (ddd, J = 11.7, 11.7, 4.9 Hz, 1H), 4.04 (m, 1H), 3.64 (s, 1H), 3.34 (d, J = 2.8, 1H), 3.07 (d, J = 2.1 Hz, 1H), 2.27 (ddd, J = 6.8, 4.4, 2.4 Hz, 1H), 1.97 (m, 2H), 1.68 (m, 1H), 0.92 (d, J = 7.0 Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 166.3, 165.6, 164.3, 141.3, 136.3, 136.1, 128.8, 128.7, 128.5, 128.1, 127.6, 127.5, 126.6, 104.6, 103.5, 76.2, 72.5, 64.2, 52.3, 42.6, 39.1, 31.7, 29.0, 21.3; HRMS (ESI) calcd for $\text{C}_{29}\text{H}_{28}\text{O}_7\text{Na}$ $[\text{M}+\text{Na}^+]$ 511.17327; found 511.17332.



Bromoacetal 25b Following the procedure for the synthesis of **25a** from **12a**, compound **12b** (1.92 g, 3.93 mmol, 1eq) was converted to compound **25b**. The crude product was purified by column chromatography (20, 30, 40 then 50% EtOAc / hexanes until the product was recovered) to afford **25b** (1.59 g, 63%) and recovered **12b** (0.81 g) after the first cycle.

25b: R_f 0.58 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2924, 2854, 1773, 1682, 1621, 1454, 1406, 1285, 1192, 1084 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) (1.2:1 mixture of diastereomers) δ 7.40 – 7.20 (m, 10H), 6.50 (s, 0.5H), 6.47 (s, 0.5H), 5.89 (s, 0.5H), 5.64 (s, 0.5H), 5.22 (d, J_{AB} = 12.4 Hz, 1H), 5.14 (d, J_{AB} = 12.4 Hz, 1H), 5.06 – 4.95 (m, 2H), 4.75 (dd, J = 5.1, 5.1 Hz, 0.5H), 4.73 (dd, J = 3.6, 3.6, 0.5H), 4.59 (ddd, J = 11.8, 11.8, 4.3 Hz, 0.5H), 4.44 (ddd, J = 11.6, 11.6, 4.1 Hz, 0.5H), 4.12 – 4.01 (m, 1H), 3.56 – 3.37 (m, 4H), 3.28 (d, J = 2.8 Hz, 0.5H), 3.27 (d, J = 2.2 Hz, 0.5H), 3.25 (d, J = 2.8 Hz, 0.5H), 3.01 (d, J = 2.2 Hz, 0.5H), 2.46 (m, 1H), 2.06 (m, 1H), 1.84 – 1.66 (m, 2H), 1.21 (t, J = 7.0 Hz, 1.5H), 1.13 (t, J = 7.0 Hz, 1.5H), 1.04 (d, J = 7.0 Hz, 1.5H), 1.02 (d, J = 7.0 Hz, 1.5H); ^{13}C NMR (100 MHz, CDCl_3) δ 166.1, 166.0, 165.2, 165.0, 164.2, 164.1, 136.4, 136.3, 136.2, 136.0, 129.0, 128.8, 128.7, 128.5, 128.4, 128.1, 127.8, 127.5, 126.5, 105.6, 105.1, 103.7, 97.9, 97.1, 82.1, 81.5, 73.7, 72.9, 63.7, 63.5, 62.7, 61.5, 53.8, 53.2, 45.2,

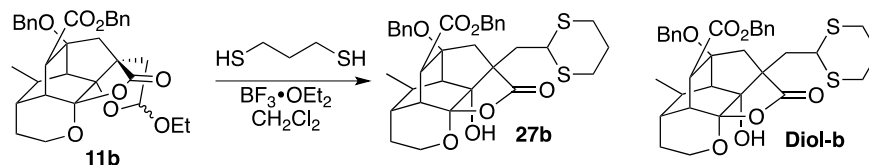
44.4, 40.4, 39.8, 32.6, 32.4, 29.9, 29.3, 28.8, 28.7, 21.4, 15.3, 15.1; HRMS (ESI) calcd for $C_{33}H_{35}BrO_8Na$ $[M+Na^+]$ 661.14075; found 661.14033.



Ethyl Acetal 11b and byproduct 26b Following the procedure for the synthesis of **11a** from **25a**, compound **25b** (2.40 g, 3.75 mmol, 1eq) was converted to compounds **11b** and **26b**. The crude product was purified by column chromatography (10, 20, 30, 40 then 50% EtOAc / hexanes until the product was recovered) to afford **11b** (1.09 g, 52%) and cyclization byproduct **26b** (882 mg, 39%).

11b: R_f 0.60 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2972, 2925, 1785, 1734, 1454, 1275, 1197, 1174, 1072 cm^{-1} ; 1H NMR (400 MHz, $CDCl_3$) (~1:1 mixture of diastereomers) δ 7.33 – 7.12 (m, 10H), 5.50 (dd, $J = 5.8, 2.1$ Hz, 0.5H), 5.34 (d, $J = 4.0$ Hz, 0.5H), 5.20 (d, $J_{AB} = 12.0$ Hz, 1H), 5.09 (d, $J_{AB} = 12.0$ Hz, 0.5H), 5.08 (d, $J_{AB} = 12.0$ Hz, 0.5H), 4.44 (d, $J_{AB} = 11.6$ Hz, 1H), 4.39 (d, $J_{AB} = 11.6$ Hz, 1H), 4.18 (m, 1H), 3.96 – 3.83 (m, 1H), 3.80 – 3.67 (m, 1H), 3.52 – 3.41 (m, 1H), 3.35 – 3.25 (m, 1H), 2.79 (d, $J = 9.4$ Hz, 1H), 2.72 (dd, $J = 14.2, 6.0$ Hz, 0.5H), 2.66 – 2.59 (m, 0.5H), 2.57 (m, 0.5H), 2.53 (m, 0.5H), 2.21 – 2.13 (m, 1H), 2.09 – 1.80 (m, 4H), 1.52 – 1.44 (m, 1H), 1.22 (app t, $J = 7.0$ Hz, 3H), 1.16 (t, $J = 7.0$ Hz, 1.5H), 1.08 (t, $J = 7.0$ Hz, 1.5H); ^{13}C NMR (100 MHz, $CDCl_3$) 176.8, 176.1, 171.2, 138.5, 135.7, 135.6, 128.9, 128.8, 128.6, 128.5, 127.6, 127.0, 126.9, 109.8, 107.6, 106.9, 106.3, 95.3, 94.7, 86.2, 84.8, 67.3, 66.4, 66.2, 64.1, 63.0, 62.7, 62.3, 56.7, 56.1, 54.0, 53.1, 51.5, 51.4, 44.3, 43.9, 43.1, 42.7, 41.3, 41.2, 40.9, 39.9, 30.7, 30.7, 27.8, 27.6, 20.9, 20.8, 15.4, 14.7; HRMS (ESI) calcd for $C_{33}H_{36}O_8Na$ $[M+Na^+]$ 583.23024; found 583.22974.

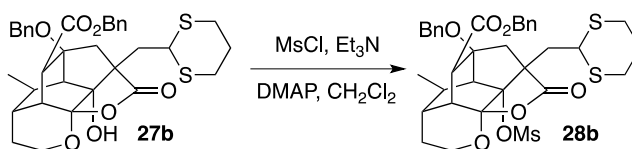
26b: R_f 0.48 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 2968, 2912, 1780, 1690, 1454, 1377, 1255, 1127, 1088, 1043 cm^{-1} ; 1H NMR (400 MHz, $CDCl_3$) δ 7.40 – 7.27 (m, 10H), 5.26 – 5.20 (m, 2H), 5.16 (d, $J_{AB} = 12.7$ Hz, 1H), 4.95 (d, $J_{AB} = 12.2$ Hz, 1H), 4.66 (dd, $J = 4.6, 4.6$ Hz, 1H), 4.46 (ddd, $J = 12.4, 12.4, 3.8$ Hz, 1H), 4.01 (dd, $J = 11.7, 6.6$ Hz, 1H), 3.92 (dq, $J = 9.1, 7.0$ Hz, 1H), 3.50 (dq, $J = 9.0, 7.0$ Hz, 1H), 3.24 (d, $J = 2.8$ Hz, 1H), 3.05 (d, $J = 2.0$ Hz, 1H), 2.39 (dd, $J = 7.4, 4.6$ Hz, 1H), 2.31 (m, 1H), 2.08 – 1.96 (m, 2H), 1.77 (m, 1H), 1.66 (m, 1H), 1.58 – 1.41 (m, 3H), 1.23 (t, $J = 7.0$ Hz, 3H), 1.07 (d, $J = 7.0$ Hz, 3H); ^{13}C NMR (100 MHz, $CDCl_3$) δ 174.6, 165.6, 164.0, 136.3, 136.2, 128.9, 128.7, 128.6, 128.1, 127.7, 106.1, 104.8, 97.4, 81.5, 72.9, 66.3, 64.4, 63.3, 50.6, 44.1, 43.1, 40.4, 29.8, 28.6, 26.9, 21.8, 18.6, 15.2; HRMS (ESI) calcd for $C_{33}H_{36}O_8Na$ $[M+Na^+]$ 561.24829; found 561.24695.



Dithiane 27b Following the procedure for the synthesis of **27a** from **11a**, compound **11b** (1.06 g, 1.89 mmol, 1eq) was converted to compounds **27b**. The crude product was purified by column chromatography (15, 30 then 45% EtOAc / hexanes until the product was recovered) to afford **27b** (920 mg, 78%) as a white foam as well as a trace amount of **Diol-b**.

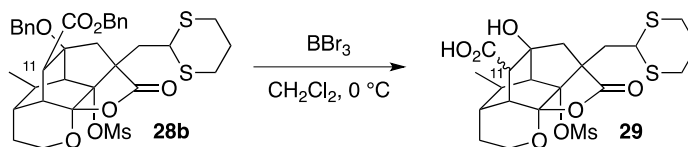
27b: R_f 0.59 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 3445, 3031, 2908, 1775, 1733, 1454, 1368, 1258, 1203, 1165, 1072, 983 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.33 – 7.13 (m, 10H), 5.13 (d, $J_{AB} = 12.2$ Hz, 1H), 5.03 (d, $J_{AB} = 12.2$ Hz, 1H), 4.49 – 4.38 (m, 3H), 4.28 (ddd, $J = 11.5, 11.5, 4.5$ Hz, 1H), 3.99 (ddd, $J = 11.3, 6.1, 2.8$ Hz, 1H), 3.26 (s, 1H), 3.11 (d, $J = 13.6$ Hz, 1H), 2.87 – 2.73 (m, 5H), 2.47 (d, $J = 2.1$ Hz, 1H), 2.33 (d, $J = 13.6$ Hz, 1H), 2.30 (d, $J = 2.6$ Hz, 1H), 2.29 (dd, $J = 15.2, 6.6$ Hz, 1H), 2.08 (dd, $J = 15.1, 6.4$ Hz, 1H), 2.04 – 1.88 (m, 3H), 1.87 – 1.75 (m, 2H), 1.54 – 1.44 (m, 1H), 1.21 (d, $J = 7.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 175.9, 171.0, 138.6, 135.5, 128.8, 128.7, 128.5, 128.4, 127.4, 127.0, 105.5, 83.9, 80.8, 67.3, 66.0, 63.8, 53.0, 52.3, 51.1, 42.1, 41.2, 40.8, 40.4, 39.0, 30.3, 30.2, 27.9, 25.4, 20.7; HRMS (ESI) calcd for $\text{C}_{34}\text{H}_{38}\text{O}_7\text{S}_2\text{Na}$ $[\text{M}+\text{Na}^+]$ 645.19512; found 645.19493.

Diol-b: R_f 0.31 (50% EtOAc / hexanes); FTIR (NaCl/thin film) 3466, 2909, 1773, 1454, 1242, 1168, 1072, 1026, 909 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.50 – 7.29 (m, 5H), 5.20 (d, $J_{AB} = 12.1$ Hz, 1H), 5.09 (d, $J_{AB} = 12.1$ Hz, 1H), 4.45 (t, $J = 6.4$ Hz, 1H), 4.32 (ddd, $J = 11.5, 11.5, 4.2$ Hz, 1H), 4.02 (ddd, $J = 11.2, 6.2, 2.2$ Hz, 1H), 3.45 (br. s, 1H), 2.94 – 2.78 (m, 5H), 2.48 (d, $J = 2.6$ Hz, 1H), 2.41 (d, $J_{AB} = 14.3$ Hz, 1H), 2.35 (d, $J_{AB} = 14.3$ Hz, 1H), 2.31 (dd, $J = 15.1, 6.2$ Hz, 1H), 2.20 (d, $J = 2.6$ Hz, 1H), 2.11 (dd, $J = 15.3, 6.6$ Hz, 1H), 2.07 – 1.79 (m, 5H), 1.53 (m, 1H), 1.26 (d, $J = 7.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.1, 170.9, 135.3, 129.0, 128.9, 128.7, 105.6, 84.9, 76.6, 67.6, 63.7, 54.8, 54.0, 51.1, 47.2, 42.1, 40.7, 38.8, 38.7, 30.3, 30.2, 28.1, 25.4, 20.9; HRMS (ESI) calcd for $\text{C}_{27}\text{H}_{32}\text{O}_7\text{S}_2\text{Na}$ $[\text{M}+\text{Na}^+]$ 555.14817; found 555.14797.



Mesylate 27b Following the procedure for the synthesis of **17a** from **27a**, compound **27b** (385 mg, 0.618 mmol, 1eq) was converted to compound **28b**. The crude product was purified by column chromatography (5, 10 then 20% EtOAc / benzene until the product was recovered) to afford **28b** (270 mg, 62%) as a white foam, and recovered **27b** (97 mg, 25%).

28b: R_f 0.53 (20% EtOAc / benzene); FTIR (NaCl/thin film) 2926, 1781, 1732, 1455, 139, 1184, 1150, 1081, 1027 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.39 – 7.17 (m, 10H), 5.20 (d, $J_{AB} = 12.1$ Hz, 1H), 5.09 (d, $J_{AB} = 12.1$ Hz, 1H), 4.83 (dd, $J = 8.6, 3.0$ Hz, 1H), 4.54 – 4.42 (m, 2H), 4.36 (ddd, $J = 12.6, 12.6, 3.8$ Hz, 1H), 4.00 (ddd, $J = 12.1, 12.1, 6.4$ Hz, 1H), 3.35 – 3.24 (m, 2H), 3.23 (s, 3H), 3.05 – 2.71 (m, 5H), 2.63 – 2.45 (m, 3H), 2.23 (dd, $J = 14.3, 3.0$ Hz, 1H), 2.12 – 1.76 (m, 5H), 1.66 (m, 1H), 1.28 (d, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 174.2, 170.7, 138.2, 135.4, 129.0, 128.8, 128.7, 128.5, 127.6, 127.1, 105.2, 98.1, 81.2, 67.6, 66.2, 62.7, 51.8, 51.5, 50.7, 42.2, 41.4, 41.0, 40.7, 40.5, 40.3, 31.4, 30.7, 30.1, 27.4, 25.6, 20.7; HRMS (ESI) calcd for $\text{C}_{35}\text{H}_{40}\text{O}_9\text{S}_3\text{Na}$ $[\text{M}+\text{Na}^+]$ 723.17267; found 723.17141.



Acids 29 Benzyl ether **28b** (40 mg, 0.057 mmol) was dissolved in CH_2Cl_2 (1 mL, 0.06M) and cooled to 0 °C. To this was added successive portions of BBr_3 (5.0 μL , 0.57 mmol, 1.0 equiv), and the reaction monitored by TLC (20% EtOAc / benzene) after each portion of BBr_3 . Upon consumption of starting material the reaction was quenched by addition of aqueous saturated NaHCO_3 (10 mL) and diluted with EtOAc (10 mL). The aqueous layer was separated and washed with EtOAc (2 x 15mL). The combined organic extracts were washed with water, brine and concentrated *in vacuo*. The residue was purified by column chromatography (0 then 10% MeOH/ CH_2Cl_2 until the products were recovered) to provide a epimeric mixture of acids **29** (23 mg, 77%) as a white foam.

Acid: R_f 0.51 (10% MeOH/ CH_2Cl_2); HRMS (ESI) calcd for $\text{C}_{21}\text{H}_{27}\text{O}_9\text{S}_3\text{Na}$ $[\text{M}+\text{Na}^+]$ 543.0793; found 543.0828.

In order to assure that the products were the acids, they were subjected to ethereal CH_2N_2 and the newly formed products isolated. Both were found to be identical to compounds **10a** and *epi-10a* by NMR spectroscopy.

3. X-ray Data

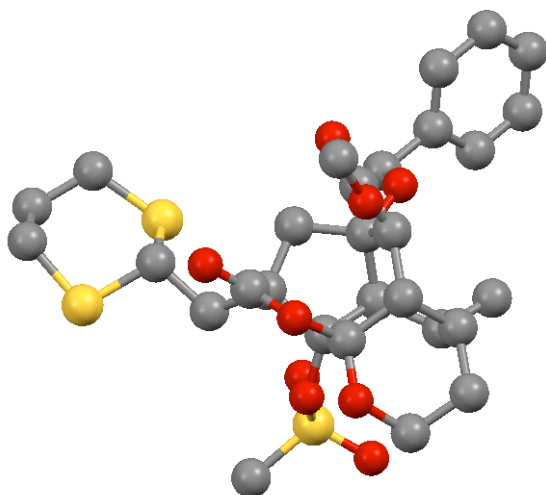
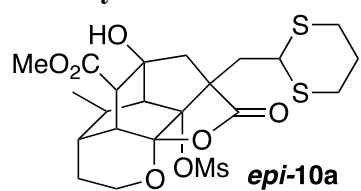


Table 1. Crystal data and structure refinement for **wood02 (C11-*epi*-10a)**.

Identification code	wood02	
Empirical formula	C ₂₂ H ₃₀ O ₉ S ₃	
Formula weight	534.64	
Temperature	120(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P -1	
Unit cell dimensions	a = 9.0924(3) Å	α = 107.430(2)°.
	b = 9.1537(4) Å	β = 91.471(2)°.
	c = 14.5615(6) Å	γ = 96.663(2)°.
Volume	1146.15(8) Å ³	
Z	2	
Density (calculated)	1.549 Mg/m ³	
Absorption coefficient	0.377 mm ⁻¹	
F(000)	564	
Crystal size	0.30 x 0.25 x 0.15 mm ³	
Theta range for data collection	2.35 to 33.23°.	
Index ranges	-12 ≤ h ≤ 13, -14 ≤ k ≤ 14, -22 ≤ l ≤ 22	
Reflections collected	32393	
Independent reflections	8602 [R(int) = 0.0344]	
Completeness to theta = 33.23°	97.9 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9463 and 0.8953	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	8602 / 0 / 311	
Goodness-of-fit on F ²	1.064	
Final R indices [I > 2σ(I)]	R ₁ = 0.0428, wR ₂ = 0.1118	
R indices (all data)	R ₁ = 0.0569, wR ₂ = 0.1210	
Largest diff. peak and hole	0.987 and -0.428 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\approx 2 \times 10^3$) for wood02. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	$U(\text{eq})$
C(1)	8564(2)	-505(2)	1244(1)	19(1)
C(2)	8447(2)	-1887(2)	1637(1)	19(1)
C(3)	8340(2)	-1436(2)	2719(1)	16(1)
C(4)	6818(2)	1088(2)	2649(1)	13(1)
C(5)	5383(2)	1778(2)	2960(1)	13(1)
C(6)	5070(1)	3277(2)	2759(1)	11(1)
C(7)	6210(2)	4636(2)	3301(1)	13(1)
C(8)	3950(2)	5596(2)	3639(1)	12(1)
C(9)	3556(1)	3801(2)	3128(1)	10(1)
C(10)	4916(2)	3157(2)	1675(1)	12(1)
C(11)	3616(2)	4045(2)	1574(1)	11(1)
C(12)	2492(1)	3547(2)	2238(1)	11(1)
C(13)	4097(2)	5845(2)	1994(1)	13(1)
C(14)	3427(2)	6470(2)	2968(1)	13(1)
C(15)	1163(2)	4464(2)	2459(1)	12(1)
C(16)	3863(2)	6661(2)	1248(1)	15(1)
C(17)	2606(2)	8415(2)	728(1)	21(1)
C(18)	93(2)	4162(2)	1571(1)	17(1)
C(19)	1718(2)	6210(2)	2918(1)	13(1)
C(20)	1218(2)	6874(2)	3934(1)	15(1)
C(21)	1899(2)	6152(2)	4640(1)	15(1)
C(22)	2326(2)	1338(2)	4812(1)	25(1)
O(1)	7528(1)	4720(1)	3320(1)	17(1)
O(2)	5539(1)	5874(1)	3782(1)	15(1)
O(3)	3497(1)	6141(1)	4567(1)	14(1)
O(4)	2987(1)	3581(1)	620(1)	15(1)
O(5)	4588(2)	6496(1)	554(1)	25(1)
O(6)	2830(1)	7601(1)	1434(1)	19(1)
O(7)	3052(1)	3141(1)	3871(1)	12(1)
O(8)	2008(1)	463(1)	2919(1)	19(1)
O(9)	380(1)	2233(1)	3838(1)	22(1)

S(1)	6910(1)	441(1)	1349(1)	16(1)
S(2)	6712(1)	-529(1)	3147(1)	16(1)
S(3)	1814(1)	1717(1)	3753(1)	12(1)

Table 3. Bond lengths [$\approx$] and angles [∞] for wood02.

C(1)-C(2)	1.531(2)
C(1)-S(1)	1.8086(15)
C(2)-C(3)	1.514(2)
C(3)-S(2)	1.8146(15)
C(4)-C(5)	1.5361(19)
C(4)-S(1)	1.8130(15)
C(4)-S(2)	1.8275(14)
C(5)-C(6)	1.5409(19)
C(6)-C(7)	1.5218(19)
C(6)-C(10)	1.5510(19)
C(6)-C(9)	1.5659(19)
C(7)-O(1)	1.1908(17)
C(7)-O(2)	1.3618(18)
C(8)-O(3)	1.3846(17)
C(8)-O(2)	1.4366(16)
C(8)-C(14)	1.534(2)
C(8)-C(9)	1.5803(19)
C(9)-O(7)	1.4486(17)
C(9)-C(12)	1.5408(18)
C(10)-C(11)	1.5387(19)
C(11)-O(4)	1.4095(16)
C(11)-C(12)	1.5477(19)
C(11)-C(13)	1.580(2)
C(12)-C(15)	1.5365(19)
C(13)-C(16)	1.516(2)
C(13)-C(14)	1.532(2)
C(14)-C(19)	1.5410(19)
C(15)-C(18)	1.5344(19)
C(15)-C(19)	1.553(2)
C(16)-O(5)	1.2003(19)
C(16)-O(6)	1.3253(18)
C(17)-O(6)	1.4626(18)
C(19)-C(20)	1.525(2)
C(20)-C(21)	1.533(2)

C(21)-O(3)	1.4610(17)
C(22)-S(3)	1.7431(16)
O(7)-S(3)	1.5839(10)
O(8)-S(3)	1.4296(12)
O(9)-S(3)	1.4333(12)

C(2)-C(1)-S(1)	113.96(11)
C(3)-C(2)-C(1)	113.22(13)
C(2)-C(3)-S(2)	114.52(10)
C(5)-C(4)-S(1)	112.43(9)
C(5)-C(4)-S(2)	102.50(9)
S(1)-C(4)-S(2)	111.49(8)
C(4)-C(5)-C(6)	120.34(12)
C(7)-C(6)-C(5)	111.37(11)
C(7)-C(6)-C(10)	110.73(11)
C(5)-C(6)-C(10)	114.66(11)
C(7)-C(6)-C(9)	103.72(11)
C(5)-C(6)-C(9)	112.96(11)
C(10)-C(6)-C(9)	102.57(10)
O(1)-C(7)-O(2)	120.52(13)
O(1)-C(7)-C(6)	128.35(13)
O(2)-C(7)-C(6)	111.11(11)
O(3)-C(8)-O(2)	102.75(10)
O(3)-C(8)-C(14)	113.01(12)
O(2)-C(8)-C(14)	109.74(11)
O(3)-C(8)-C(9)	115.86(11)
O(2)-C(8)-C(9)	105.73(11)
C(14)-C(8)-C(9)	109.20(11)
O(7)-C(9)-C(12)	117.95(11)
O(7)-C(9)-C(6)	111.72(10)
C(12)-C(9)-C(6)	107.64(11)
O(7)-C(9)-C(8)	106.22(10)
C(12)-C(9)-C(8)	108.62(10)
C(6)-C(9)-C(8)	103.74(10)
C(11)-C(10)-C(6)	105.51(11)
O(4)-C(11)-C(10)	111.82(11)

O(4)-C(11)-C(12)	109.69(11)
C(10)-C(11)-C(12)	102.08(11)
O(4)-C(11)-C(13)	113.35(11)
C(10)-C(11)-C(13)	110.75(11)
C(12)-C(11)-C(13)	108.51(11)
C(15)-C(12)-C(9)	113.24(11)
C(15)-C(12)-C(11)	115.88(11)
C(9)-C(12)-C(11)	98.64(10)
C(16)-C(13)-C(14)	117.12(12)
C(16)-C(13)-C(11)	111.81(11)
C(14)-C(13)-C(11)	109.82(11)
C(13)-C(14)-C(8)	107.84(11)
C(13)-C(14)-C(19)	114.03(12)
C(8)-C(14)-C(19)	106.17(11)
C(18)-C(15)-C(12)	112.53(12)
C(18)-C(15)-C(19)	111.97(12)
C(12)-C(15)-C(19)	110.01(11)
O(5)-C(16)-O(6)	122.83(14)
O(5)-C(16)-C(13)	122.31(14)
O(6)-C(16)-C(13)	114.83(13)
C(20)-C(19)-C(14)	108.05(11)
C(20)-C(19)-C(15)	113.33(12)
C(14)-C(19)-C(15)	109.56(11)
C(19)-C(20)-C(21)	111.45(11)
O(3)-C(21)-C(20)	112.37(12)
C(7)-O(2)-C(8)	113.07(11)
C(8)-O(3)-C(21)	115.30(10)
C(16)-O(6)-C(17)	115.51(12)
C(9)-O(7)-S(3)	127.24(8)
C(1)-S(1)-C(4)	100.44(7)
C(3)-S(2)-C(4)	103.16(7)
O(8)-S(3)-O(9)	117.05(7)
O(8)-S(3)-O(7)	111.17(6)
O(9)-S(3)-O(7)	109.70(6)
O(8)-S(3)-C(22)	111.46(8)
O(9)-S(3)-C(22)	108.93(9)

O(7)-S(3)-C(22)	96.62(7)
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Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\approx 2 \times 10^3$) for wood02. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	18(1)	24(1)	20(1)	9(1)	6(1)	11(1)
C(2)	21(1)	18(1)	19(1)	6(1)	3(1)	9(1)
C(3)	15(1)	18(1)	21(1)	10(1)	2(1)	8(1)
C(4)	12(1)	14(1)	13(1)	6(1)	2(1)	3(1)
C(5)	12(1)	14(1)	15(1)	7(1)	2(1)	3(1)
C(6)	9(1)	12(1)	12(1)	4(1)	1(1)	2(1)
C(7)	15(1)	12(1)	12(1)	5(1)	-1(1)	1(1)
C(8)	9(1)	13(1)	12(1)	3(1)	1(1)	1(1)
C(9)	10(1)	11(1)	10(1)	4(1)	1(1)	1(1)
C(10)	12(1)	15(1)	11(1)	4(1)	2(1)	4(1)
C(11)	12(1)	13(1)	10(1)	4(1)	1(1)	4(1)
C(12)	10(1)	12(1)	10(1)	3(1)	0(1)	2(1)
C(13)	14(1)	14(1)	15(1)	6(1)	4(1)	4(1)
C(14)	14(1)	11(1)	14(1)	4(1)	3(1)	3(1)
C(15)	10(1)	15(1)	13(1)	5(1)	1(1)	2(1)
C(16)	17(1)	14(1)	17(1)	6(1)	4(1)	3(1)
C(17)	26(1)	26(1)	17(1)	13(1)	4(1)	9(1)
C(18)	14(1)	22(1)	16(1)	6(1)	-2(1)	5(1)
C(19)	12(1)	14(1)	15(1)	6(1)	4(1)	5(1)
C(20)	14(1)	16(1)	15(1)	2(1)	4(1)	7(1)
C(21)	14(1)	18(1)	13(1)	4(1)	4(1)	6(1)
C(22)	34(1)	22(1)	20(1)	12(1)	-5(1)	-7(1)
O(1)	12(1)	19(1)	22(1)	7(1)	2(1)	2(1)
O(2)	11(1)	13(1)	18(1)	2(1)	-1(1)	0(1)
O(3)	14(1)	16(1)	9(1)	1(1)	0(1)	2(1)
O(4)	15(1)	22(1)	10(1)	6(1)	1(1)	4(1)
O(5)	33(1)	24(1)	23(1)	11(1)	14(1)	13(1)
O(6)	23(1)	23(1)	17(1)	12(1)	6(1)	12(1)
O(7)	12(1)	12(1)	11(1)	5(1)	1(1)	0(1)
O(8)	19(1)	16(1)	18(1)	0(1)	5(1)	-2(1)
O(9)	13(1)	20(1)	32(1)	8(1)	7(1)	2(1)

S(1)	17(1)	20(1)	14(1)	7(1)	3(1)	8(1)
S(2)	16(1)	18(1)	19(1)	11(1)	4(1)	6(1)
S(3)	12(1)	13(1)	12(1)	4(1)	2(1)	0(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\approx 2 \times 10^{-3}$) for wood02.

	x	y	z	U(eq)
H(1A)	8783	-860	556	23
H(1B)	9409	258	1595	23
H(2A)	7559	-2615	1321	22
H(2B)	9328	-2431	1466	22
H(3A)	9236	-719	3032	20
H(3B)	8338	-2374	2925	20
H(4)	7701	1859	2952	15
H(5A)	5354	1963	3665	16
H(5B)	4543	975	2651	16
H(10A)	5840	3626	1474	15
H(10B)	4703	2064	1272	15
H(12)	2126	2422	1959	13
H(13)	5194	5981	2140	16
H(14)	3784	7597	3250	15
H(15)	601	4109	2951	15
H(17A)	3556	8963	637	32
H(17B)	1903	9158	961	32
H(17C)	2209	7669	112	32
H(18A)	599	4538	1084	26
H(18B)	-772	4705	1759	26
H(18C)	-231	3051	1304	26
H(19)	1333	6787	2499	16
H(20A)	122	6679	3922	18
H(20B)	1515	8005	4157	18
H(21A)	1704	6739	5307	18
H(21B)	1412	5078	4510	18
H(22A)	1581	557	4925	37
H(22B)	2391	2290	5355	37
H(22C)	3292	955	4753	37
H(4A)	3652	3666	242	23

Table 6. Torsion angles [$^\circ$] for wood02.

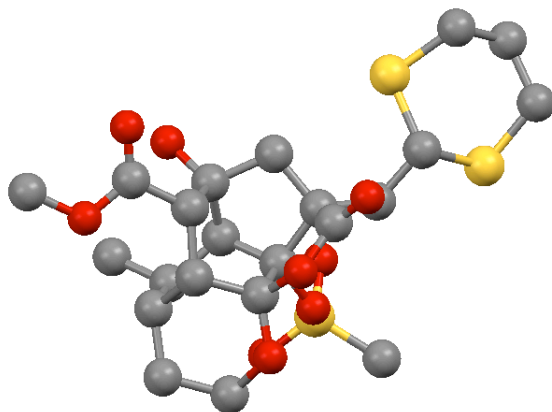
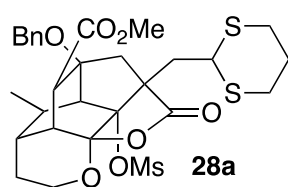


Table 1. Crystal data and structure refinement for **wood13 (28a)**.

Identification code	wood13	
Empirical formula	C ₃₁ H ₄₀ Cl ₄ O ₉ S ₃	
Formula weight	794.61	
Temperature	120(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	a = 9.6413(4) Å	α = 93.135(2)°.
	b = 12.5011(5) Å	β = 100.872(2)°.
	c = 16.1465(7) Å	γ = 111.766(2)°.
Volume	1758.25(13) Å ³	
Z	2	
Density (calculated)	1.501 Mg/m ³	
Absorption coefficient	0.567 mm ⁻¹	
F(000)	828	
Crystal size	0.62 x 0.18 x 0.06 mm ³	
Theta range for data collection	2.05 to 28.45°.	
Index ranges	-12 ≤ h ≤ 12, -16 ≤ k ≤ 16, -21 ≤ l ≤ 21	
Reflections collected	30817	
Independent reflections	8600 [R(int) = 0.0288]	
Completeness to theta = 28.45°	97.1 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9668 and 0.7202	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	8600 / 0 / 424	
Goodness-of-fit on F ²	1.050	
Final R indices [I > 2σ(I)]	R ₁ = 0.0377, wR ₂ = 0.0905	
R indices (all data)	R ₁ = 0.0550, wR ₂ = 0.0990	
Largest diff. peak and hole	0.566 and -0.568 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\approx 2 \times 10^3$) for wood13. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	$U(\text{eq})$
C(1)	2065(2)	8750(2)	-173(1)	14(1)
C(2)	3924(2)	8329(2)	-1178(1)	20(1)
C(3)	3508(3)	9087(2)	-1803(1)	24(1)
C(4)	3250(2)	10092(2)	-1383(1)	24(1)
C(5)	702(2)	8242(2)	259(1)	14(1)
C(6)	5311(2)	10575(2)	4084(2)	25(1)
C(7)	3770(2)	8752(2)	3274(1)	14(1)
C(8)	2370(2)	5336(2)	1687(1)	17(1)
C(9)	4343(2)	4696(2)	2490(1)	20(1)
C(10)	1769(2)	3293(2)	2034(1)	20(1)
C(11)	4780(3)	3866(2)	2870(2)	28(1)
C(12)	2210(3)	2462(2)	2402(2)	28(1)
C(13)	3715(3)	2751(2)	2824(2)	29(1)
C(14)	2833(2)	4414(2)	2070(1)	14(1)
C(15)	1795(2)	6967(2)	2214(1)	12(1)
C(16)	2292(2)	7620(2)	1469(1)	13(1)
C(17)	2221(2)	7734(2)	3085(1)	12(1)
C(18)	3258(3)	5359(2)	4348(2)	29(1)
C(19)	313(3)	9047(2)	6160(2)	38(1)
C(20)	1821(2)	9462(2)	1679(1)	15(1)
C(21)	1163(2)	8226(2)	1221(1)	12(1)
C(22)	131(2)	8428(2)	2474(1)	12(1)
C(23)	-2340(2)	7728(2)	2847(1)	16(1)
C(24)	-1730(2)	7167(2)	3561(1)	17(1)
C(25)	-408(2)	6868(2)	3375(1)	14(1)
C(26)	844(2)	7997(2)	3240(1)	12(1)
C(27)	-3687(2)	7027(2)	-222(1)	20(1)
C(28)	12(2)	6460(2)	1910(1)	11(1)
C(29)	-182(2)	7569(2)	1642(1)	11(1)
C(30)	-903(2)	5945(2)	2577(1)	13(1)
C(31)	-804(2)	4794(2)	2792(1)	17(1)

Cl(1)	5006(1)	6588(1)	4613(1)	36(1)
Cl(2)	1984(1)	5456(1)	4978(1)	38(1)
Cl(3)	516(1)	7749(1)	5873(1)	50(1)
Cl(4)	-1450(1)	9058(1)	5612(1)	38(1)
O(1)	4853(2)	8778(1)	2990(1)	21(1)
O(2)	3838(2)	9603(1)	3838(1)	20(1)
O(3)	2434(2)	6122(1)	2394(1)	14(1)
O(4)	2788(2)	10296(1)	1513(1)	24(1)
O(5)	1198(2)	9536(1)	2355(1)	14(1)
O(6)	-1117(2)	8674(1)	2612(1)	14(1)
O(7)	-1648(1)	7458(1)	1128(1)	12(1)
O(8)	-2075(2)	5747(1)	109(1)	19(1)
O(9)	-3949(2)	5678(1)	960(1)	23(1)
S(1)	2413(1)	7550(1)	-656(1)	17(1)
S(2)	1570(1)	9640(1)	-934(1)	20(1)
S(3)	-2854(1)	6327(1)	503(1)	13(1)

Table 3. Bond lengths [$\approx$] and angles [∞] for wood13.

C(1)-C(5)	1.546(3)
C(1)-S(2)	1.8160(19)
C(1)-S(1)	1.8209(19)
C(2)-C(3)	1.520(3)
C(2)-S(1)	1.812(2)
C(3)-C(4)	1.518(3)
C(4)-S(2)	1.810(2)
C(5)-C(21)	1.537(3)
C(6)-O(2)	1.450(2)
C(7)-O(1)	1.207(2)
C(7)-O(2)	1.336(2)
C(7)-C(17)	1.523(3)
C(8)-O(3)	1.441(2)
C(8)-C(14)	1.505(2)
C(9)-C(14)	1.388(3)
C(9)-C(11)	1.388(3)
C(10)-C(12)	1.384(3)
C(10)-C(14)	1.386(3)
C(11)-C(13)	1.379(3)
C(12)-C(13)	1.383(3)
C(15)-O(3)	1.424(2)
C(15)-C(16)	1.543(3)
C(15)-C(17)	1.552(3)
C(15)-C(28)	1.562(2)
C(16)-C(21)	1.549(2)
C(17)-C(26)	1.539(2)
C(18)-Cl(1)	1.766(3)
C(18)-Cl(2)	1.769(2)
C(19)-Cl(3)	1.754(3)
C(19)-Cl(4)	1.767(3)
C(20)-O(4)	1.197(2)
C(20)-O(5)	1.355(2)
C(20)-C(21)	1.519(2)
C(21)-C(29)	1.555(3)

C(22)-O(6)	1.400(2)
C(22)-O(5)	1.435(2)
C(22)-C(26)	1.523(3)
C(22)-C(29)	1.580(3)
C(23)-O(6)	1.455(2)
C(23)-C(24)	1.518(3)
C(24)-C(25)	1.530(3)
C(25)-C(26)	1.538(3)
C(25)-C(30)	1.555(3)
C(27)-S(3)	1.750(2)
C(28)-C(30)	1.537(3)
C(28)-C(29)	1.541(2)
C(29)-O(7)	1.450(2)
C(30)-C(31)	1.531(2)
O(7)-S(3)	1.5967(13)
O(8)-S(3)	1.4248(14)
O(9)-S(3)	1.4286(15)

C(5)-C(1)-S(2)	107.28(12)
C(5)-C(1)-S(1)	108.45(12)
S(2)-C(1)-S(1)	113.55(10)
C(3)-C(2)-S(1)	114.30(14)
C(4)-C(3)-C(2)	113.44(18)
C(3)-C(4)-S(2)	113.59(15)
C(21)-C(5)-C(1)	114.61(15)
O(1)-C(7)-O(2)	122.90(18)
O(1)-C(7)-C(17)	124.88(17)
O(2)-C(7)-C(17)	112.10(16)
O(3)-C(8)-C(14)	106.09(15)
C(14)-C(9)-C(11)	120.34(19)
C(12)-C(10)-C(14)	120.49(19)
C(13)-C(11)-C(9)	120.1(2)
C(13)-C(12)-C(10)	120.1(2)
C(11)-C(13)-C(12)	119.9(2)
C(10)-C(14)-C(9)	119.11(18)
C(10)-C(14)-C(8)	121.29(18)

C(9)-C(14)-C(8)	119.57(18)
O(3)-C(15)-C(16)	113.72(15)
O(3)-C(15)-C(17)	103.92(14)
C(16)-C(15)-C(17)	115.79(15)
O(3)-C(15)-C(28)	114.64(14)
C(16)-C(15)-C(28)	101.56(14)
C(17)-C(15)-C(28)	107.45(14)
C(15)-C(16)-C(21)	105.25(14)
C(7)-C(17)-C(26)	116.93(15)
C(7)-C(17)-C(15)	115.07(15)
C(26)-C(17)-C(15)	110.55(15)
Cl(1)-C(18)-Cl(2)	111.34(13)
Cl(3)-C(19)-Cl(4)	112.05(14)
O(4)-C(20)-O(5)	121.12(17)
O(4)-C(20)-C(21)	127.56(18)
O(5)-C(20)-C(21)	111.29(15)
C(20)-C(21)-C(5)	109.24(15)
C(20)-C(21)-C(16)	110.71(15)
C(5)-C(21)-C(16)	113.93(15)
C(20)-C(21)-C(29)	104.14(14)
C(5)-C(21)-C(29)	115.13(15)
C(16)-C(21)-C(29)	103.17(14)
O(6)-C(22)-O(5)	103.07(13)
O(6)-C(22)-C(26)	111.98(15)
O(5)-C(22)-C(26)	109.75(14)
O(6)-C(22)-C(29)	115.91(15)
O(5)-C(22)-C(29)	106.24(14)
C(26)-C(22)-C(29)	109.44(14)
O(6)-C(23)-C(24)	112.06(15)
C(23)-C(24)-C(25)	111.36(16)
C(24)-C(25)-C(26)	107.75(15)
C(24)-C(25)-C(30)	112.94(15)
C(26)-C(25)-C(30)	109.45(15)
C(22)-C(26)-C(25)	106.71(15)
C(22)-C(26)-C(17)	111.67(15)
C(25)-C(26)-C(17)	109.07(14)

C(30)-C(28)-C(29)	112.75(14)
C(30)-C(28)-C(15)	116.37(15)
C(29)-C(28)-C(15)	98.61(13)
O(7)-C(29)-C(28)	118.40(14)
O(7)-C(29)-C(21)	111.66(14)
C(28)-C(29)-C(21)	107.78(14)
O(7)-C(29)-C(22)	106.45(13)
C(28)-C(29)-C(22)	108.29(14)
C(21)-C(29)-C(22)	103.13(14)
C(31)-C(30)-C(28)	112.56(15)
C(31)-C(30)-C(25)	112.95(15)
C(28)-C(30)-C(25)	109.80(15)
C(7)-O(2)-C(6)	115.19(16)
C(15)-O(3)-C(8)	117.74(14)
C(20)-O(5)-C(22)	112.40(14)
C(22)-O(6)-C(23)	115.92(13)
C(29)-O(7)-S(3)	125.76(11)
C(2)-S(1)-C(1)	100.74(9)
C(4)-S(2)-C(1)	99.98(9)
O(8)-S(3)-O(9)	118.18(9)
O(8)-S(3)-O(7)	110.30(8)
O(9)-S(3)-O(7)	107.91(8)
O(8)-S(3)-C(27)	111.11(10)
O(9)-S(3)-C(27)	109.51(10)
O(7)-S(3)-C(27)	97.88(8)

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\approx 2 \times 10^3$) for wood13. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	15(1)	15(1)	15(1)	5(1)	5(1)	8(1)
C(2)	18(1)	29(1)	19(1)	7(1)	8(1)	12(1)
C(3)	21(1)	40(1)	18(1)	14(1)	9(1)	14(1)
C(4)	20(1)	28(1)	26(1)	16(1)	9(1)	9(1)
C(5)	13(1)	15(1)	15(1)	4(1)	4(1)	6(1)
C(6)	20(1)	19(1)	26(1)	-5(1)	2(1)	0(1)
C(7)	16(1)	15(1)	13(1)	2(1)	1(1)	8(1)
C(8)	26(1)	16(1)	15(1)	1(1)	5(1)	14(1)
C(9)	16(1)	19(1)	24(1)	3(1)	5(1)	6(1)
C(10)	16(1)	16(1)	26(1)	-1(1)	2(1)	5(1)
C(11)	21(1)	35(1)	30(1)	4(1)	-1(1)	18(1)
C(12)	34(1)	13(1)	34(1)	3(1)	6(1)	8(1)
C(13)	44(1)	24(1)	29(1)	7(1)	5(1)	24(1)
C(14)	18(1)	14(1)	13(1)	1(1)	5(1)	10(1)
C(15)	14(1)	12(1)	13(1)	1(1)	4(1)	8(1)
C(16)	14(1)	14(1)	14(1)	3(1)	4(1)	8(1)
C(17)	13(1)	13(1)	13(1)	2(1)	3(1)	7(1)
C(18)	27(1)	37(1)	28(1)	5(1)	7(1)	18(1)
C(19)	41(2)	28(1)	42(2)	-9(1)	3(1)	13(1)
C(20)	14(1)	14(1)	19(1)	1(1)	4(1)	7(1)
C(21)	12(1)	11(1)	15(1)	3(1)	5(1)	6(1)
C(22)	12(1)	10(1)	14(1)	1(1)	4(1)	5(1)
C(23)	13(1)	18(1)	19(1)	1(1)	6(1)	8(1)
C(24)	17(1)	20(1)	15(1)	2(1)	6(1)	9(1)
C(25)	14(1)	15(1)	13(1)	2(1)	5(1)	7(1)
C(26)	12(1)	14(1)	12(1)	0(1)	2(1)	7(1)
C(27)	22(1)	21(1)	17(1)	2(1)	-4(1)	11(1)
C(28)	13(1)	10(1)	12(1)	0(1)	3(1)	6(1)
C(29)	10(1)	10(1)	11(1)	1(1)	1(1)	5(1)
C(30)	12(1)	12(1)	15(1)	2(1)	4(1)	5(1)
C(31)	21(1)	13(1)	17(1)	4(1)	6(1)	7(1)

Cl(1)	29(1)	40(1)	40(1)	6(1)	10(1)	13(1)
Cl(2)	31(1)	62(1)	26(1)	8(1)	10(1)	20(1)
Cl(3)	64(1)	37(1)	47(1)	-12(1)	-10(1)	32(1)
Cl(4)	43(1)	35(1)	40(1)	-5(1)	10(1)	20(1)
O(1)	14(1)	23(1)	26(1)	-2(1)	6(1)	6(1)
O(2)	15(1)	18(1)	22(1)	-6(1)	2(1)	3(1)
O(3)	20(1)	14(1)	12(1)	0(1)	3(1)	12(1)
O(4)	22(1)	14(1)	34(1)	0(1)	14(1)	1(1)
O(5)	15(1)	8(1)	17(1)	0(1)	4(1)	4(1)
O(6)	14(1)	15(1)	19(1)	0(1)	5(1)	9(1)
O(7)	11(1)	11(1)	14(1)	-1(1)	0(1)	5(1)
O(8)	21(1)	17(1)	18(1)	-4(1)	1(1)	10(1)
O(9)	19(1)	23(1)	18(1)	1(1)	4(1)	-3(1)
S(1)	18(1)	17(1)	19(1)	4(1)	8(1)	8(1)
S(2)	20(1)	21(1)	24(1)	12(1)	9(1)	11(1)
S(3)	13(1)	12(1)	13(1)	1(1)	1(1)	3(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\approx 2 \times 10^{-3}$) for wood13.

	x	y	z	U(eq)
H(1A)	2998	9251	268	17
H(2A)	4841	8826	-738	24
H(2B)	4201	7758	-1486	24
H(3A)	4341	9400	-2109	29
H(3B)	2564	8598	-2229	29
H(4A)	3147	10613	-1810	28
H(4B)	4163	10544	-925	28
H(5A)	59	8704	161	17
H(5B)	67	7437	-19	17
H(6A)	5254	11151	4500	37
H(6B)	5574	10933	3581	37
H(6C)	6098	10297	4340	37
H(8A)	3083	5753	1336	21
H(8B)	1320	4985	1325	21
H(9A)	5082	5461	2518	24
H(10A)	729	3094	1754	24
H(11A)	5813	4066	3161	33
H(12A)	1478	1692	2366	33
H(13A)	4016	2182	3081	35
H(16A)	2224	7071	983	16
H(16B)	3360	8201	1648	16
H(17A)	2348	7211	3514	15
H(18A)	3463	4649	4434	35
H(18B)	2775	5295	3739	35
H(19A)	1165	9710	6034	46
H(19B)	379	9150	6781	46
H(23A)	-2910	7133	2343	19
H(23B)	-3064	8032	3030	19
H(24A)	-2568	6449	3633	20
H(24B)	-1371	7705	4101	20

H(25A)	18	6565	3880	16
H(26A)	1176	8590	3755	15
H(27A)	-4485	6446	-670	31
H(27B)	-2899	7557	-478	31
H(27C)	-4143	7470	74	31
H(28A)	-312	5865	1397	13
H(30A)	-2004	5775	2321	15
H(31A)	-1121	4249	2267	25
H(31B)	-1482	4463	3175	25
H(31C)	253	4932	3071	25

Table 6. Torsion angles [$^\circ$] for wood13.

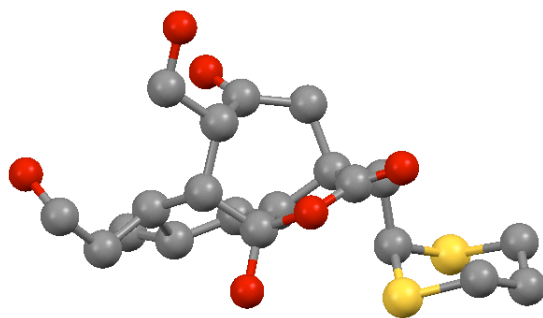
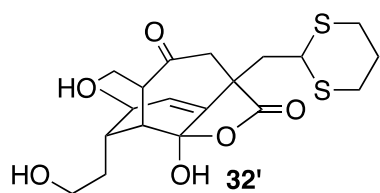


Table 1. Crystal data and structure refinement for **wood16 (32)**.

Identification code	wood16	
Empirical formula	C ₂₀ H ₂₉ O ₆ S ₂	
Formula weight	429.55	
Temperature	120(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P 2(1)/c	
Unit cell dimensions	a = 12.5635(17) Å	α = 90°.
	b = 6.5271(10) Å	β = 90.859(7)°.
	c = 23.877(3) Å	γ = 90°.
Volume	1957.7(5) Å ³	
Z	4	
Density (calculated)	1.457 Mg/m ³	
Absorption coefficient	0.308 mm ⁻¹	
F(000)	916	
Crystal size	0.24 x 0.08 x 0.07 mm ³	
Theta range for data collection	2.34 to 20.86°.	
Index ranges	-12 ≤ h ≤ 12, -6 ≤ k ≤ 6, -23 ≤ l ≤ 22	
Reflections collected	18741	
Independent reflections	2043 [R(int) = 0.1376]	
Completeness to theta = 20.86°	99.2 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9797 and 0.9303	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2043 / 168 / 131	
Goodness-of-fit on F ²	1.035	
Final R indices [I > 2σ(I)]	R1 = 0.0683, wR2 = 0.1703	
R indices (all data)	R1 = 0.1038, wR2 = 0.1937	
Largest diff. peak and hole	1.151 and -0.653 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\times 10^3$) for wood16. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	$U(\text{eq})$
C(1)	1475(5)	957(10)	8698(3)	22(1)
C(2)	855(5)	-550(9)	9057(3)	22(1)
C(3)	1028(5)	-2764(10)	8892(3)	22(1)
C(4)	1218(5)	-1637(10)	7767(3)	22(1)
C(5)	2386(5)	-2287(10)	7680(3)	22(1)
C(6)	2984(5)	-828(10)	7304(3)	22(1)
C(7)	3375(5)	1078(10)	7606(3)	22(1)
C(8)	2407(5)	62(10)	6805(3)	22(1)
C(9)	2735(5)	2262(10)	6742(3)	22(1)
C(10)	3987(5)	-1908(10)	7058(3)	22(1)
C(11)	4314(5)	-1148(10)	6480(3)	22(1)
C(12)	4531(5)	1126(9)	6383(3)	22(1)
C(13)	5343(5)	1553(10)	5939(3)	22(1)
C(14)	3478(5)	2370(10)	6253(3)	22(1)
C(15)	2852(5)	1644(10)	5710(3)	22(1)
C(16)	1957(5)	-842(10)	6360(3)	22(1)
C(17)	1813(5)	404(10)	5835(3)	22(1)
C(18)	2583(5)	3465(9)	5331(3)	22(1)
C(19)	3486(5)	4284(10)	4998(3)	22(1)
C(20)	1491(5)	-932(10)	5328(3)	22(1)
O(1)	3774(3)	1176(6)	8066(2)	22(1)
O(2)	3295(3)	2760(6)	7271(2)	22(1)
O(3)	1930(3)	3741(6)	6669(2)	22(1)
O(4)	4411(3)	-2371(6)	6099(2)	22(1)
O(5)	6346(4)	750(8)	6064(2)	47(2)
O(6)	3824(4)	2744(7)	4623(2)	35(1)
S(1)	1033(1)	1021(2)	7972(1)	22(1)
S(2)	495(1)	-3417(3)	8199(1)	23(1)

Table 3. Bond lengths [$\approx$] and angles [∞] for wood16.

C(1)-C(2)	1.526(9)
C(1)-S(1)	1.813(6)
C(2)-C(3)	1.515(9)
C(3)-S(2)	1.825(6)
C(4)-C(5)	1.545(8)
C(4)-S(2)	1.808(7)
C(4)-S(1)	1.818(6)
C(5)-C(6)	1.516(9)
C(6)-C(8)	1.501(9)
C(6)-C(7)	1.517(9)
C(6)-C(10)	1.566(9)
C(7)-O(1)	1.202(7)
C(7)-O(2)	1.362(7)
C(8)-C(16)	1.335(8)
C(8)-C(9)	1.502(9)
C(9)-O(3)	1.408(7)
C(9)-O(2)	1.472(7)
C(9)-C(14)	1.507(9)
C(10)-C(11)	1.528(9)
C(11)-O(4)	1.218(7)
C(11)-C(12)	1.527(9)
C(12)-C(13)	1.509(9)
C(12)-C(14)	1.579(9)
C(13)-O(5)	1.393(8)
C(14)-C(15)	1.580(9)
C(15)-C(18)	1.529(9)
C(15)-C(17)	1.568(9)
C(16)-C(17)	1.503(9)
C(17)-C(20)	1.540(9)
C(18)-C(19)	1.495(9)
C(19)-O(6)	1.416(8)
C(2)-C(1)-S(1)	113.7(4)
C(3)-C(2)-C(1)	113.1(5)

C(2)-C(3)-S(2)	114.0(4)
C(5)-C(4)-S(2)	112.8(4)
C(5)-C(4)-S(1)	115.1(4)
S(2)-C(4)-S(1)	113.1(3)
C(6)-C(5)-C(4)	112.8(5)
C(8)-C(6)-C(5)	118.4(5)
C(8)-C(6)-C(7)	102.1(5)
C(5)-C(6)-C(7)	113.1(6)
C(8)-C(6)-C(10)	105.0(5)
C(5)-C(6)-C(10)	110.4(5)
C(7)-C(6)-C(10)	106.9(5)
O(1)-C(7)-O(2)	121.4(6)
O(1)-C(7)-C(6)	127.4(6)
O(2)-C(7)-C(6)	111.1(5)
C(16)-C(8)-C(6)	130.9(6)
C(16)-C(8)-C(9)	117.1(6)
C(6)-C(8)-C(9)	108.6(5)
O(3)-C(9)-O(2)	106.8(5)
O(3)-C(9)-C(8)	118.1(5)
O(2)-C(9)-C(8)	104.7(5)
O(3)-C(9)-C(14)	108.9(5)
O(2)-C(9)-C(14)	111.1(5)
C(8)-C(9)-C(14)	107.2(5)
C(11)-C(10)-C(6)	114.9(5)
O(4)-C(11)-C(12)	120.2(6)
O(4)-C(11)-C(10)	119.5(6)
C(12)-C(11)-C(10)	120.2(6)
C(13)-C(12)-C(11)	114.2(5)
C(13)-C(12)-C(14)	109.9(5)
C(11)-C(12)-C(14)	112.2(5)
O(5)-C(13)-C(12)	113.5(5)
C(9)-C(14)-C(12)	110.5(5)
C(9)-C(14)-C(15)	108.3(5)
C(12)-C(14)-C(15)	114.4(5)
C(18)-C(15)-C(17)	109.7(5)
C(18)-C(15)-C(14)	110.8(5)

C(17)-C(15)-C(14)	113.9(5)
C(8)-C(16)-C(17)	118.1(6)
C(16)-C(17)-C(20)	112.1(5)
C(16)-C(17)-C(15)	110.3(5)
C(20)-C(17)-C(15)	110.7(5)
C(19)-C(18)-C(15)	115.5(5)
O(6)-C(19)-C(18)	108.7(5)
C(7)-O(2)-C(9)	110.9(5)
C(1)-S(1)-C(4)	101.3(3)
C(4)-S(2)-C(3)	100.7(3)

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\approx 2 \times 10^3$) for wood16. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

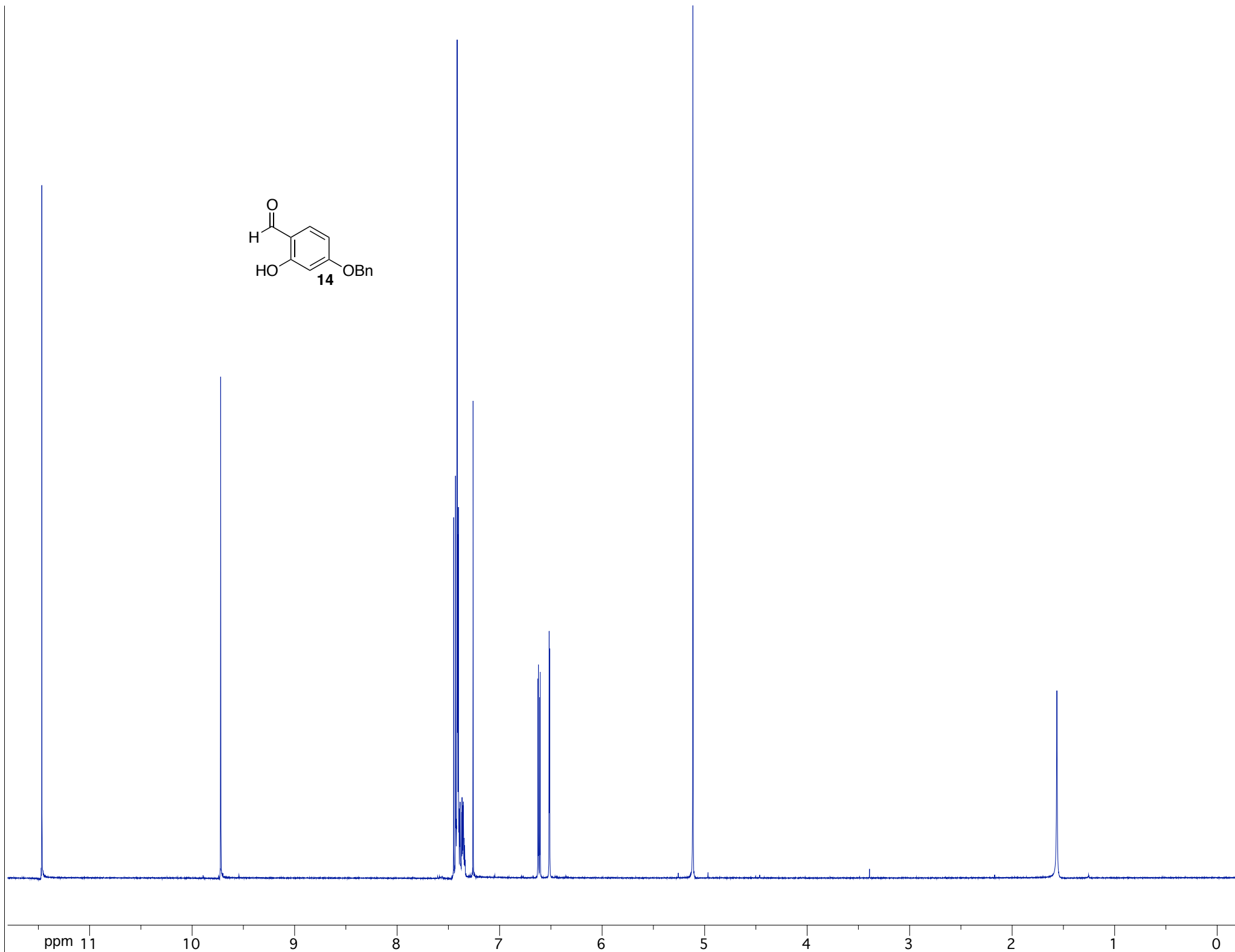
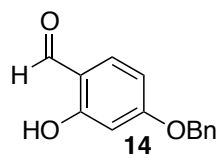
	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(2)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(3)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(4)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(5)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(6)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(7)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(8)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(9)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(10)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(11)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(12)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(13)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(14)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(15)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(16)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(17)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(18)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(19)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
C(20)	9(1)	31(1)	24(1)	-1(1)	-10(1)	2(1)
O(1)	10(2)	32(2)	25(2)	-2(1)	-12(1)	-2(1)
O(2)	10(2)	32(2)	25(2)	-2(1)	-12(1)	-2(1)
O(3)	5(2)	29(3)	32(3)	1(2)	-10(2)	2(2)
O(4)	10(2)	32(2)	25(2)	-2(1)	-12(1)	-2(1)
O(5)	23(3)	54(4)	63(4)	-9(3)	-15(3)	0(3)
O(6)	24(3)	43(3)	38(3)	-5(3)	3(2)	-6(2)
S(1)	10(1)	28(1)	28(1)	0(1)	-14(1)	2(1)
S(2)	10(1)	33(1)	25(1)	1(1)	-11(1)	-3(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\approx 2 \times 10^{-3}$) for wood16.

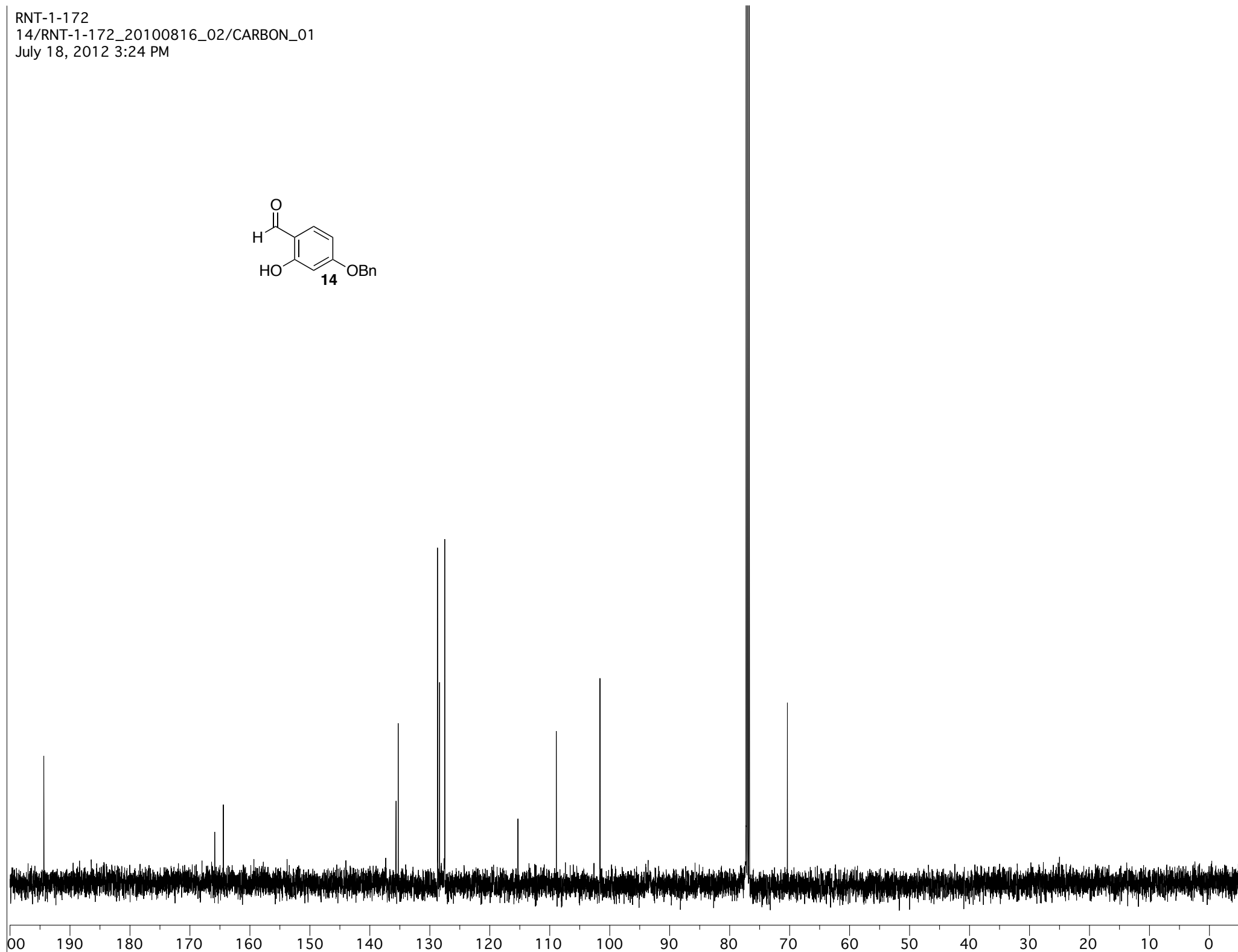
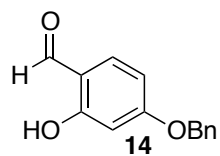
	x	y	z	U(eq)
H(1A)	2238	584	8713	26
H(1B)	1404	2347	8859	26
H(2A)	1074	-369	9454	26
H(2B)	87	-230	9025	26
H(3A)	691	-3659	9173	26
H(3B)	1802	-3053	8900	26
H(4)	874	-1747	7388	26
H(5A)	2755	-2358	8049	26
H(5B)	2398	-3676	7513	26
H(8)	1728	350	7003	26
H(10A)	4593	-1710	7322	26
H(10B)	3845	-3398	7034	26
H(12)	4831	1679	6743	26
H(13A)	5085	968	5579	26
H(13B)	5406	3054	5889	26
H(14)	3680	3838	6199	26
H(15)	3335	720	5498	26
H(16)	1734	-2231	6374	26
H(17)	1227	1409	5899	26
H(18A)	2005	3050	5069	26
H(18B)	2304	4587	5567	26
H(19A)	3254	5513	4786	26
H(19B)	4082	4681	5251	26
H(20A)	2043	-1964	5265	32
H(20B)	1413	-66	4995	32
H(20C)	814	-1616	5403	32
H(3)	1403	3424	6863	33
H(5)	6405	-410	5915	70
H(6)	4289	3226	4411	52

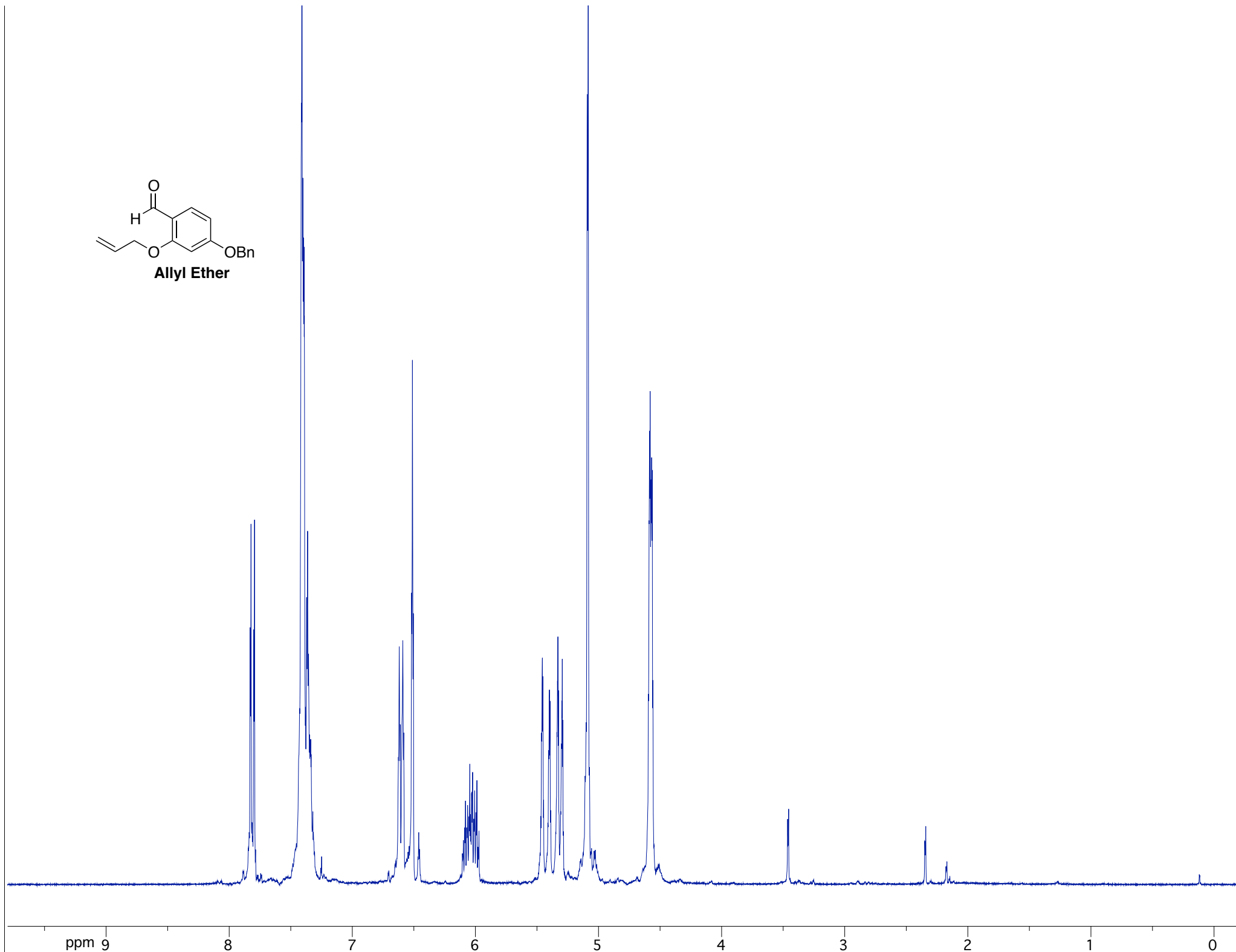
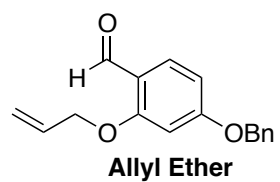
Table 6. Torsion angles [∞] for wood16.

4. Spectra

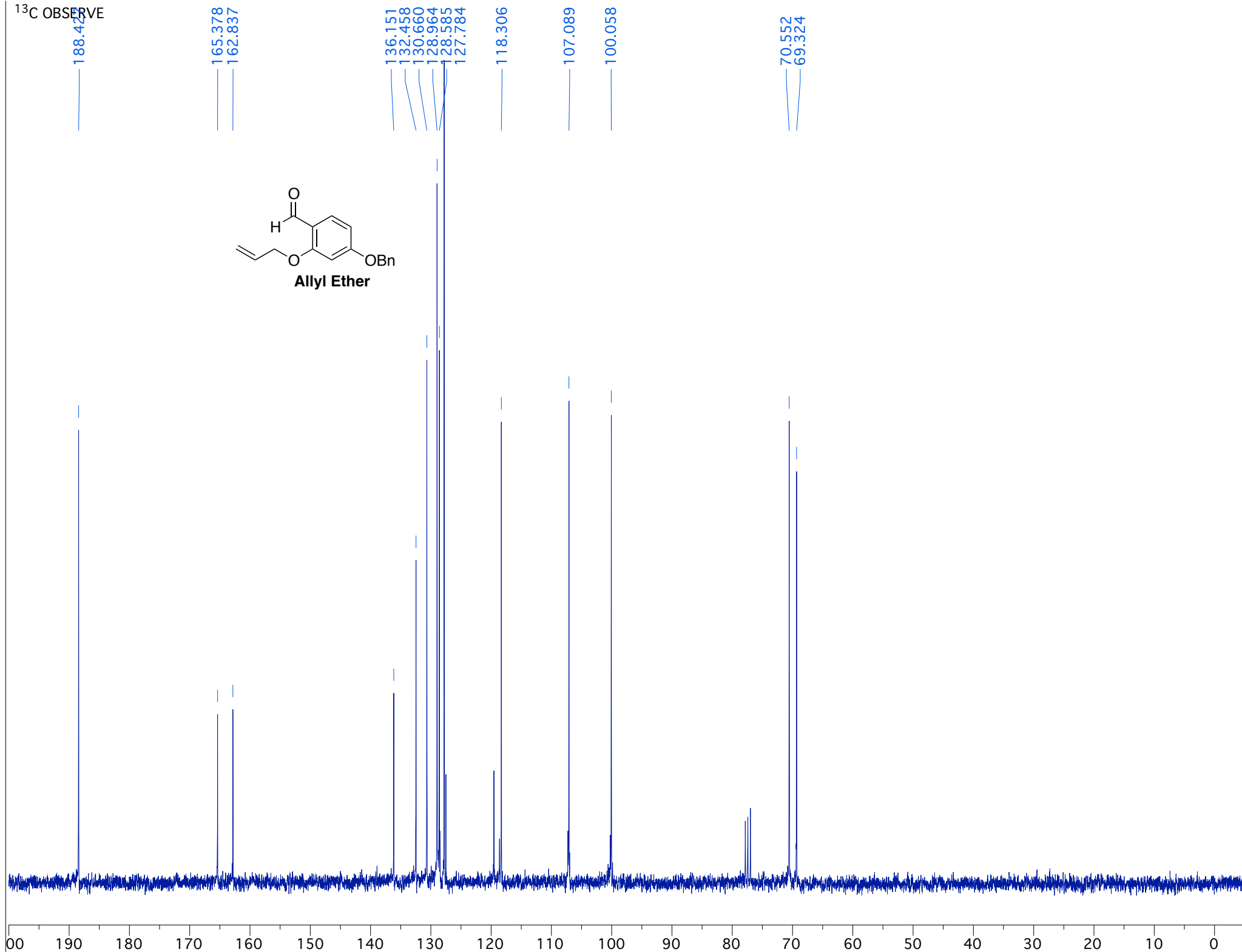
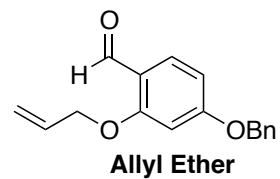


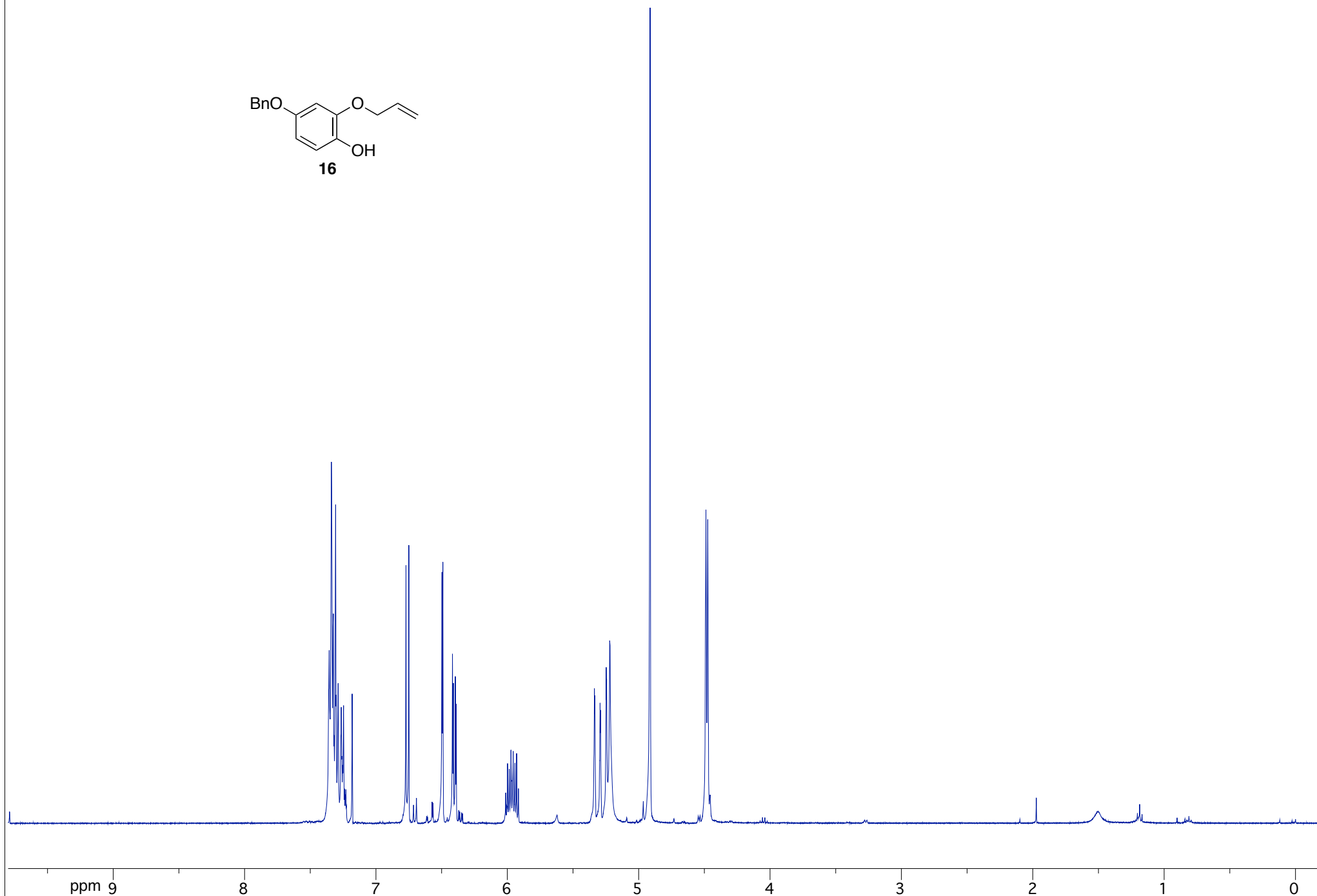
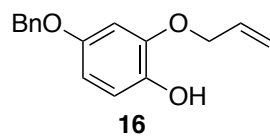
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14/RNT-1-172_20100816_02/CARBON_01
July 18, 2012 3:24 PM



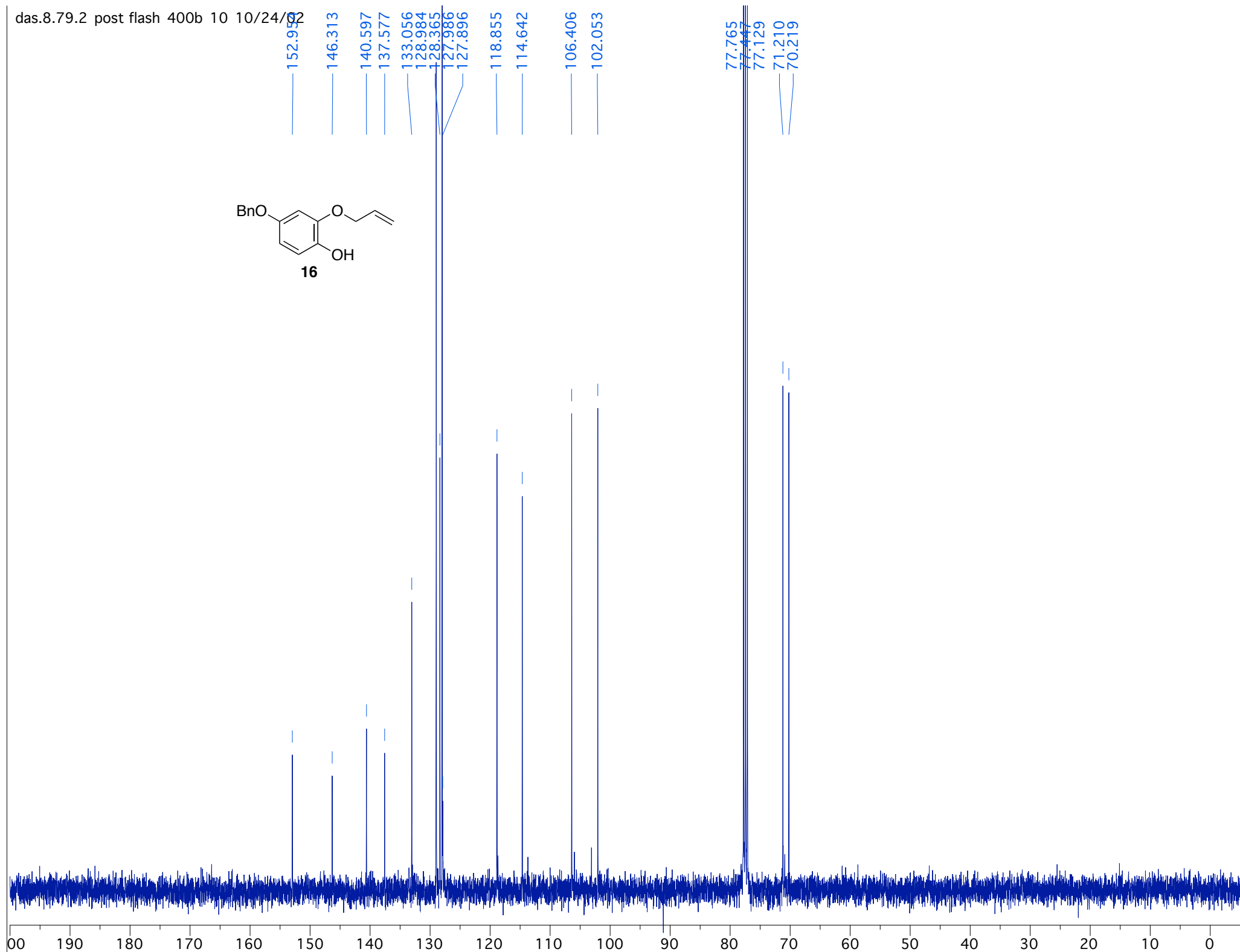
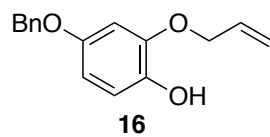


¹³C OBSERVE

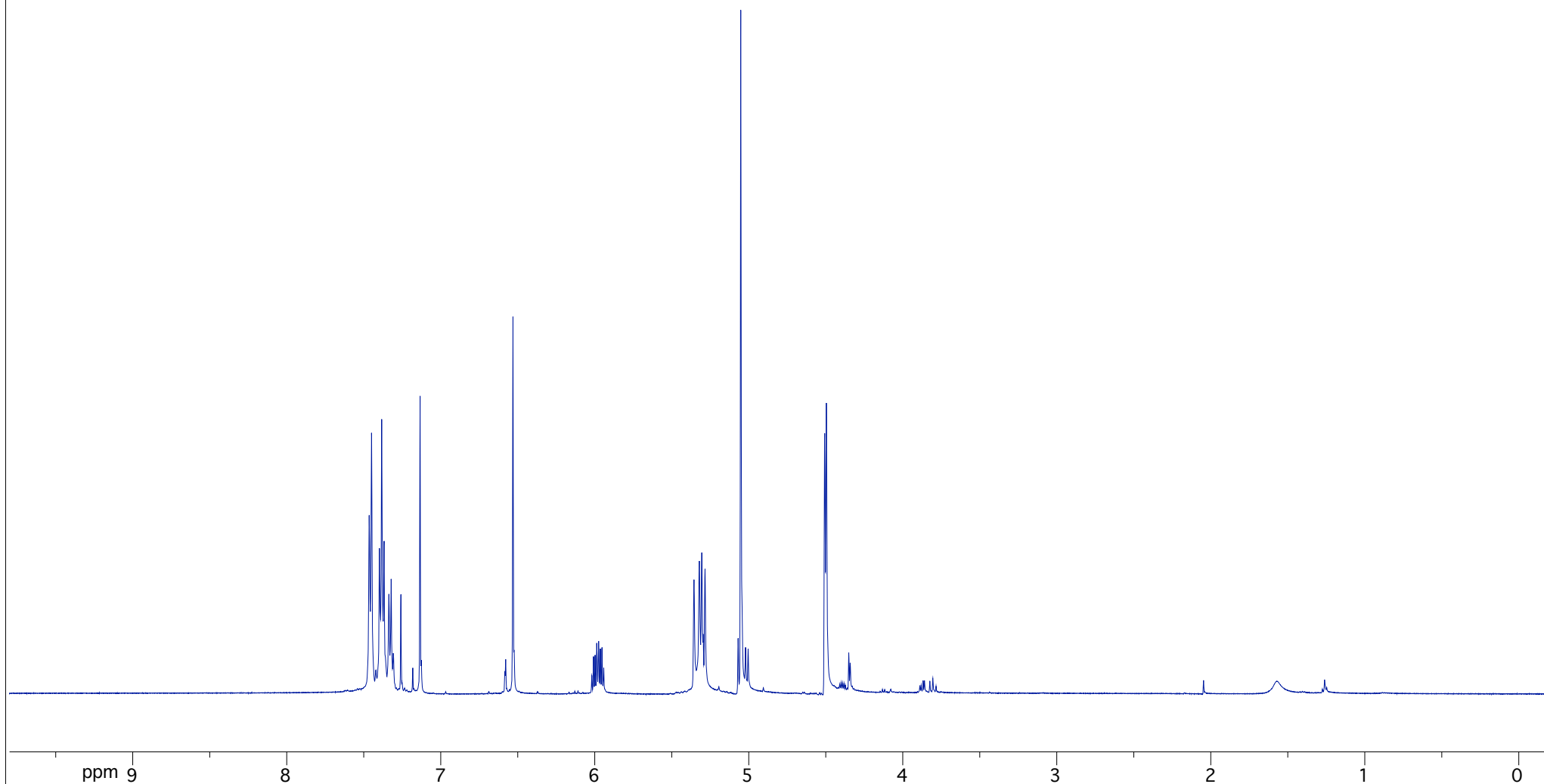
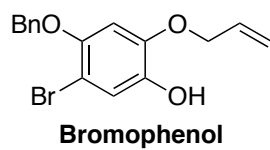




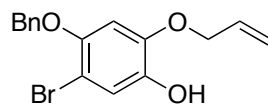
das.8.79.2 post flash 400b 10 10/24/92



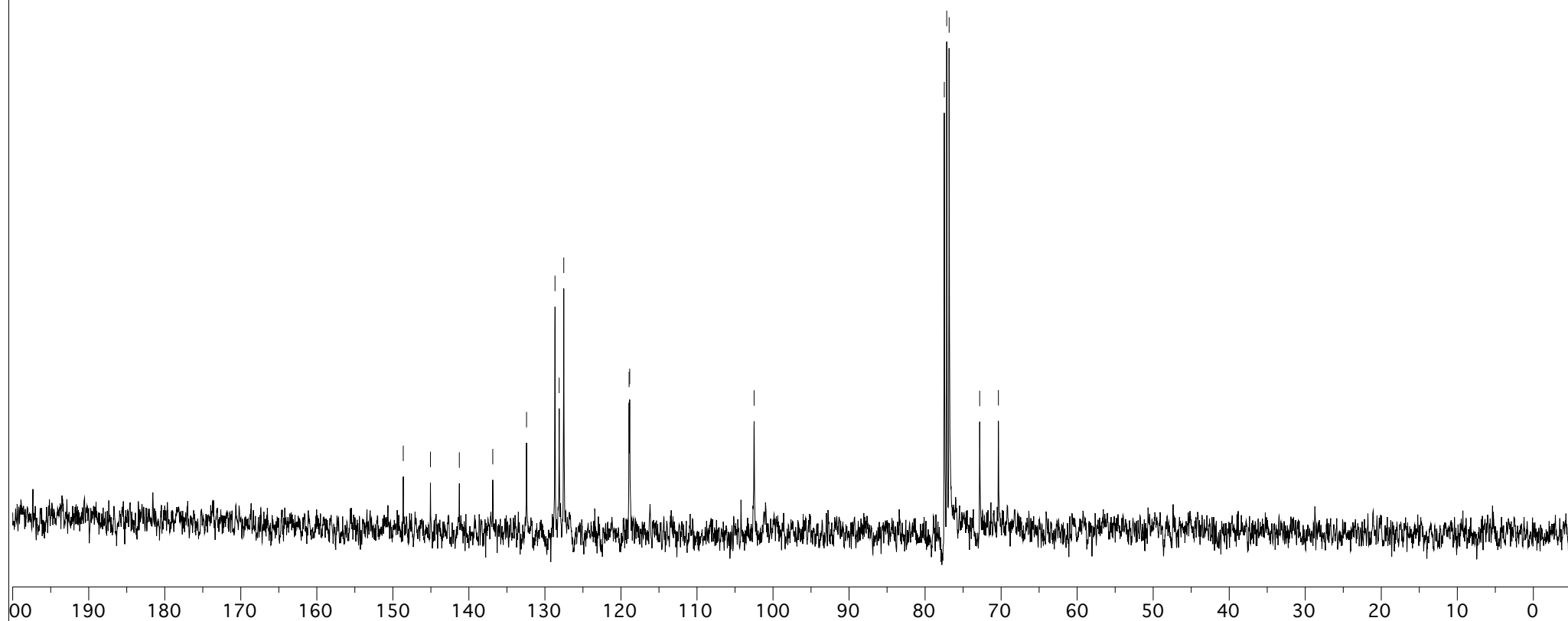
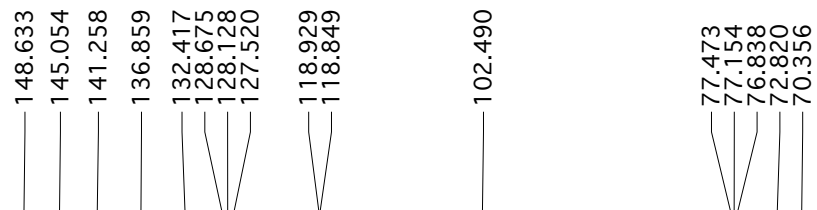
GKM-4-098p
Bromophenol/GKM-4-098p_20100813_01/PROTON_01
July 18, 2012 1:18 PM



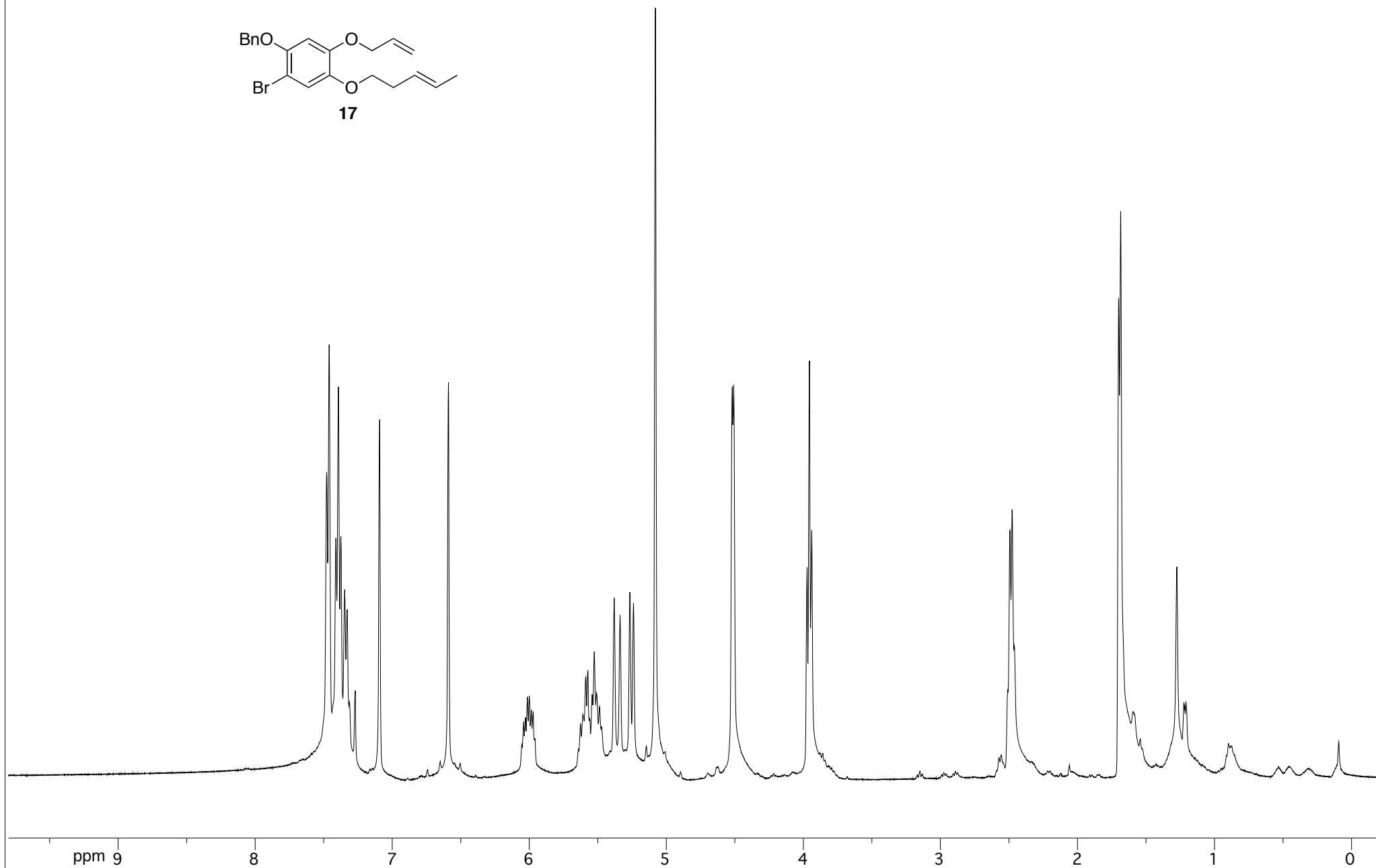
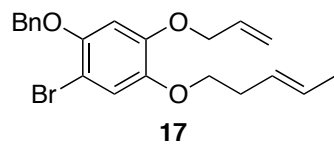
GKM-4-098 Bromination product
FIDs/Bromophenol/GKM-4-098C
July 18, 2012 1:19 PM



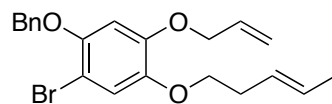
Bromophenol



STANDARD ^1H OBSERVE
FIDs/17/GKM-10-064H
July 18, 2012 1:21 PM



GKM-10-064 Side chain alkylation product
FIDs/17/GKM-10-064C
July 18, 2012 1:22 PM

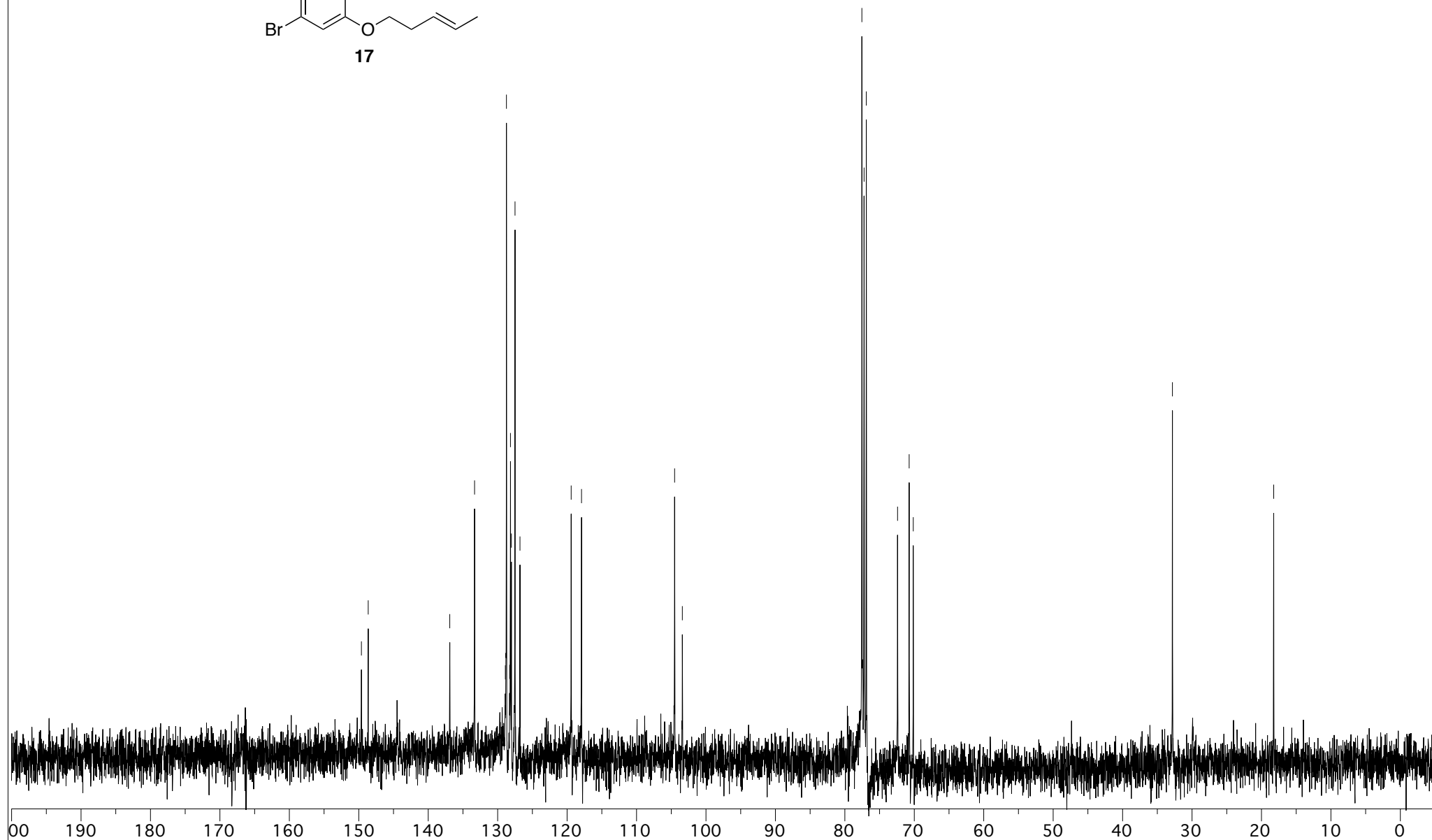


17

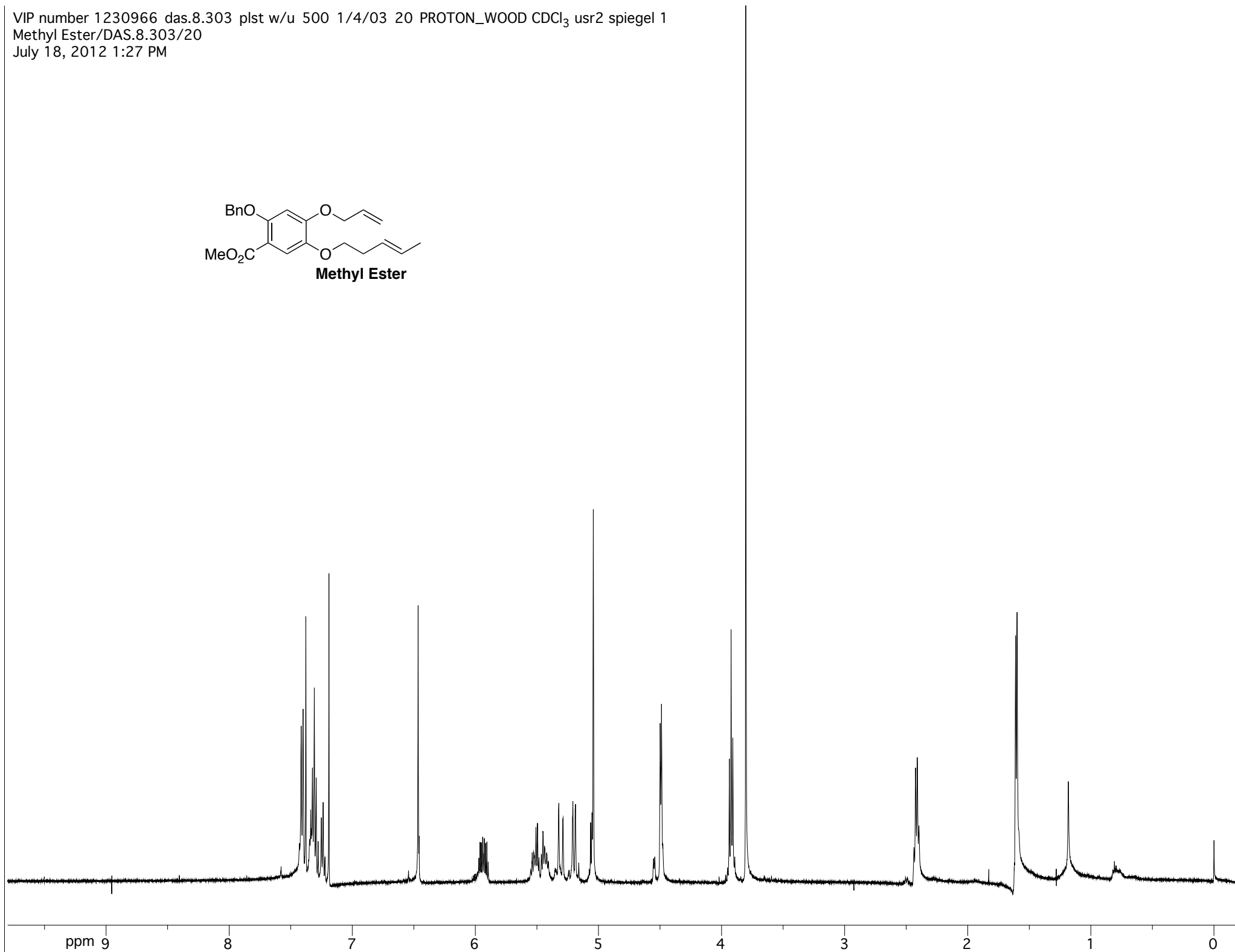
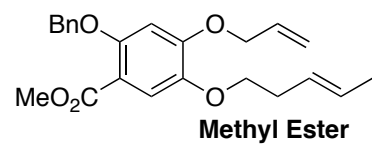
149.621
148.641
136.897
133.326
128.893
128.736
128.280
128.156
128.024
127.500
126.786
119.413
117.923
104.516
103.405
77.548
77.232
76.912
72.409
70.738
70.147

32.824

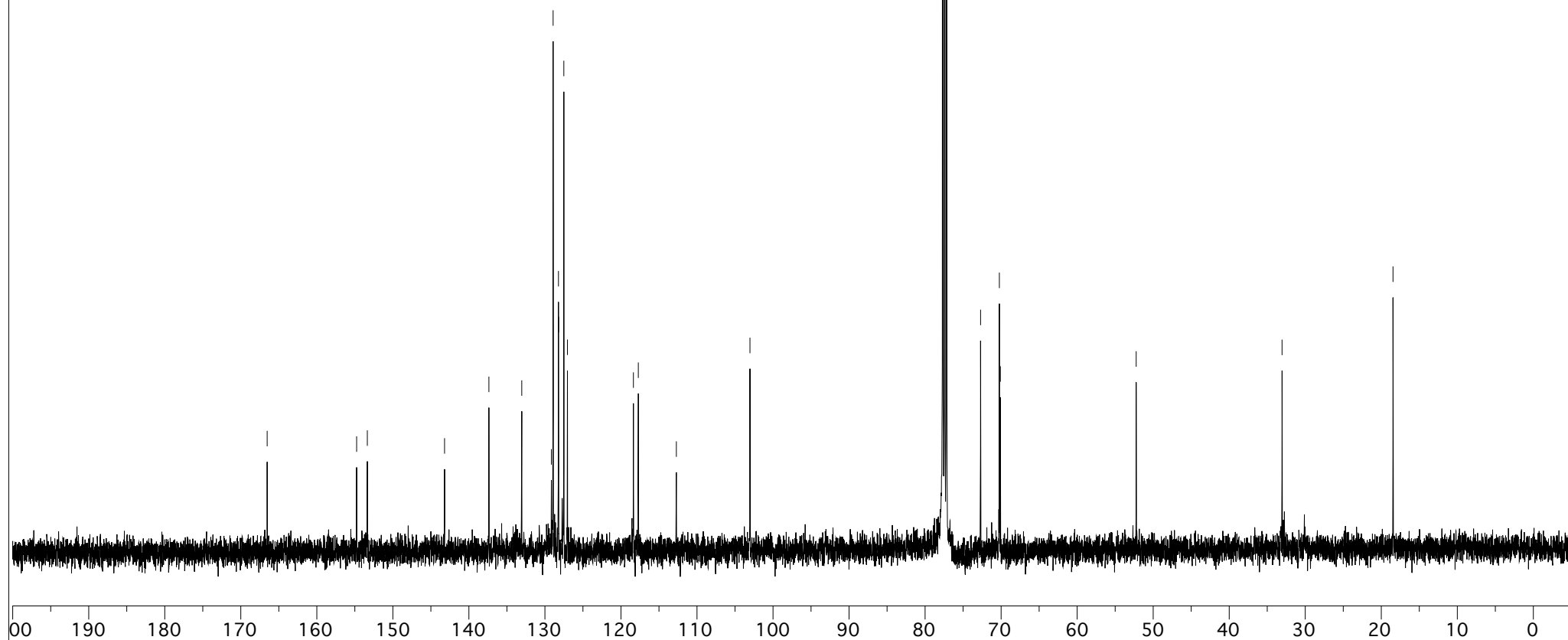
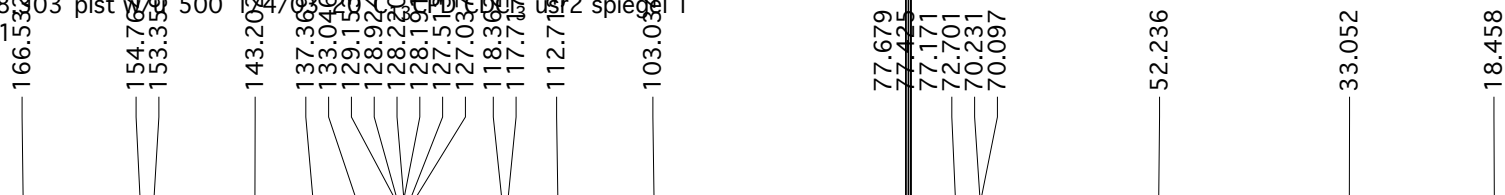
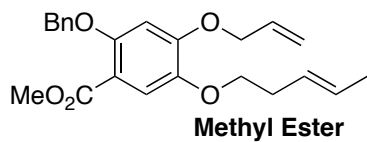
18.238



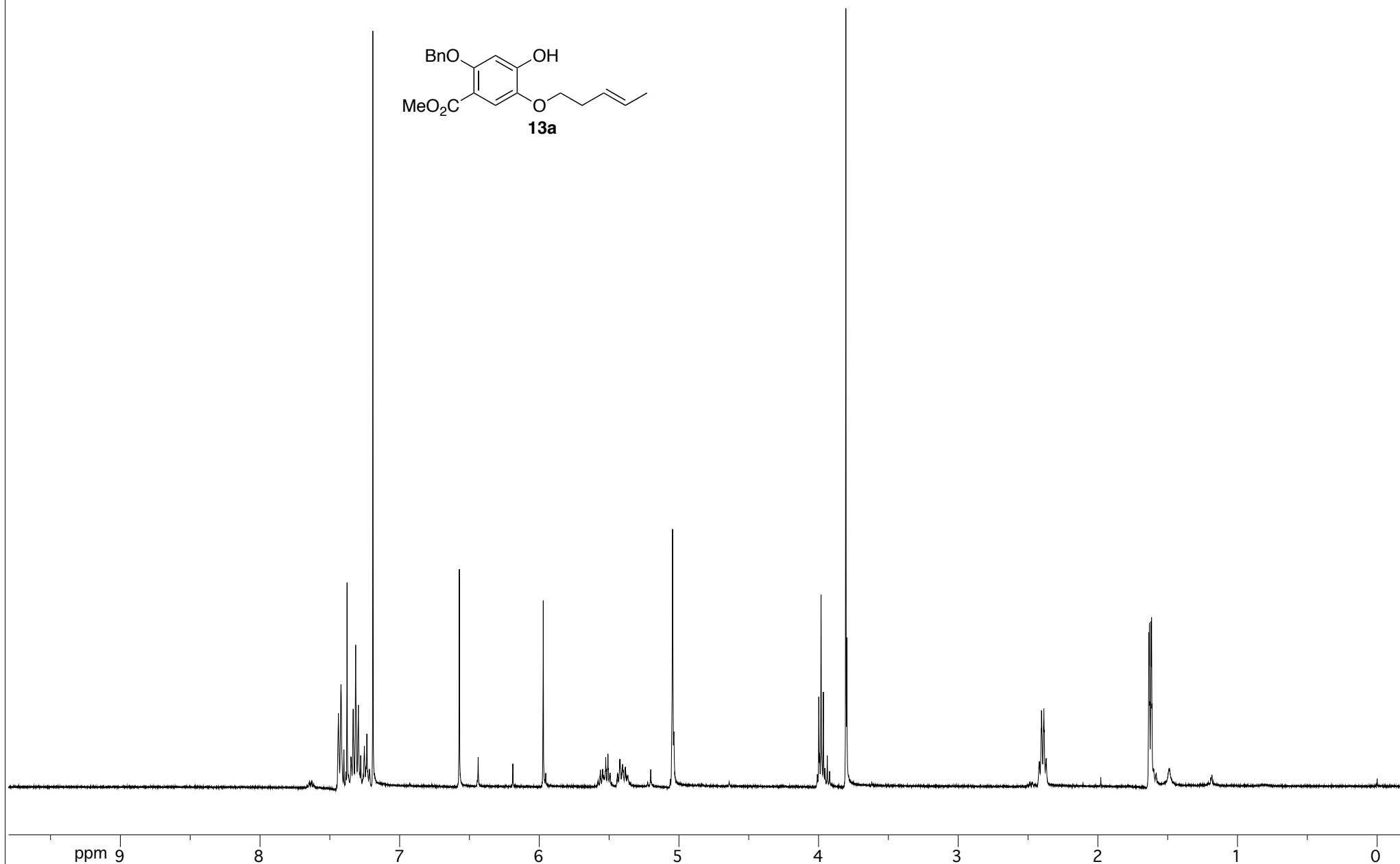
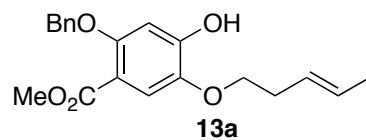
VIP number 1230966 das.8.303 plst w/u 500 1/4/03 20 PROTON_WOOD CDCl₃ usr2 spiegel 1
Methyl Ester/DAS.8.303/20
July 18, 2012 1:27 PM



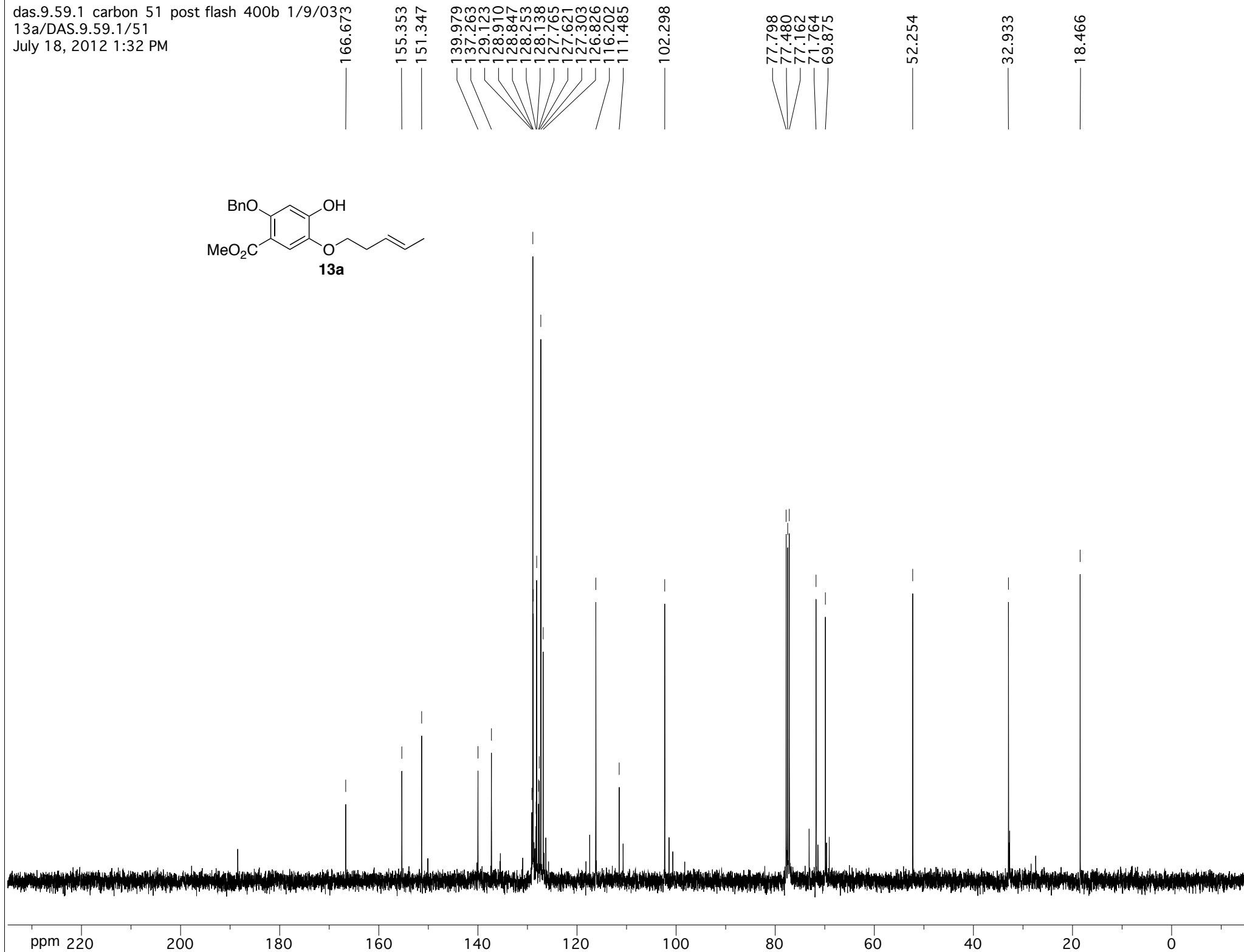
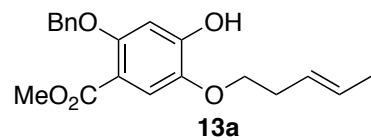
VIP number 1230966 das.8.303 plst 500 124/03/01 10:00:00 us 2 spiegl 1
Methyl Ester/DAS.8.303/21
July 18, 2012 1:25 PM



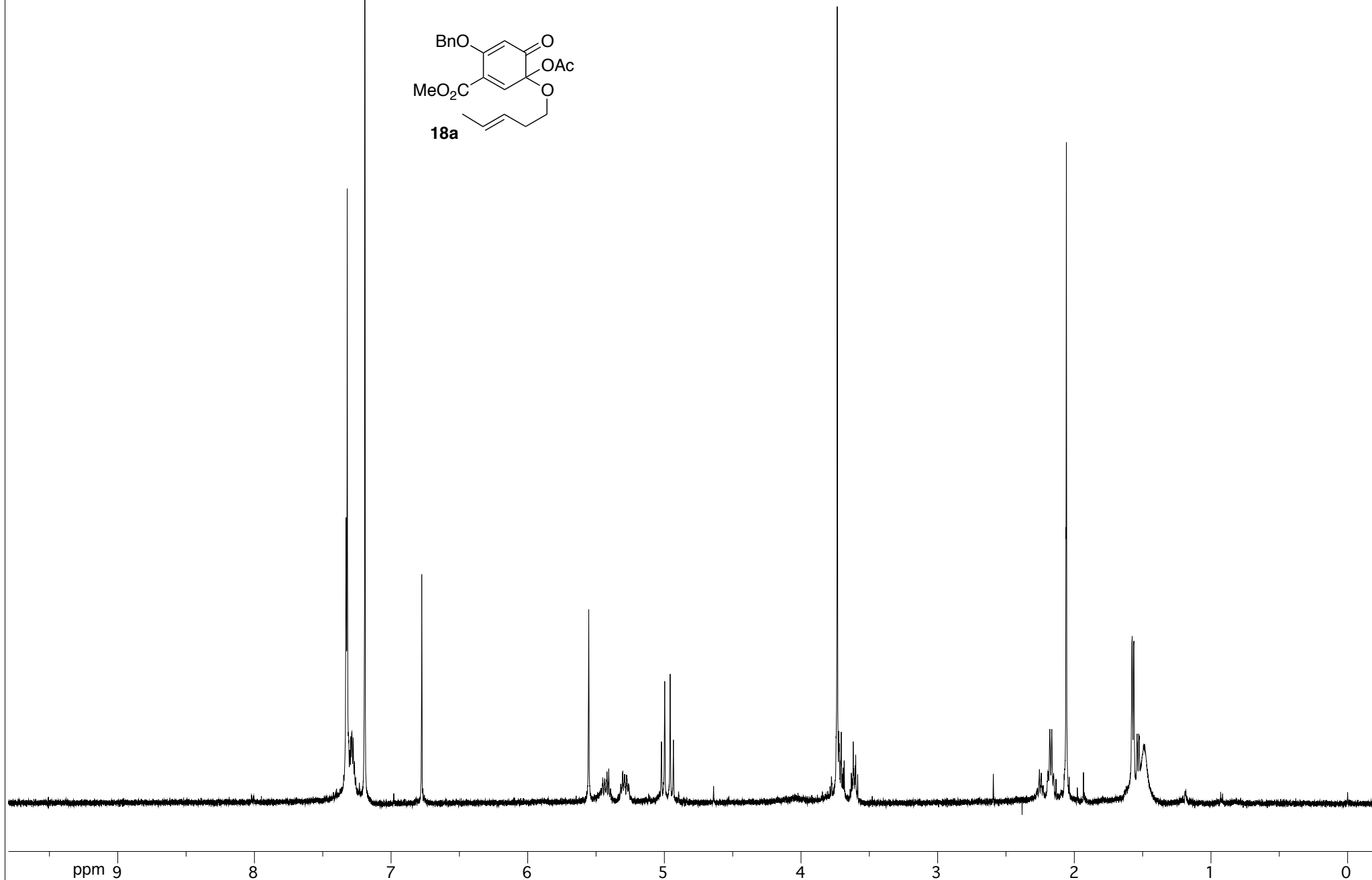
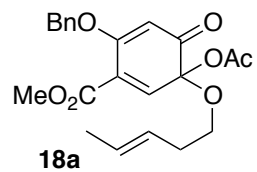
das.9.59.1 post flash 400b 1/9/03 40
13a/DAS.9.59.1/40
July 18, 2012 1:31 PM



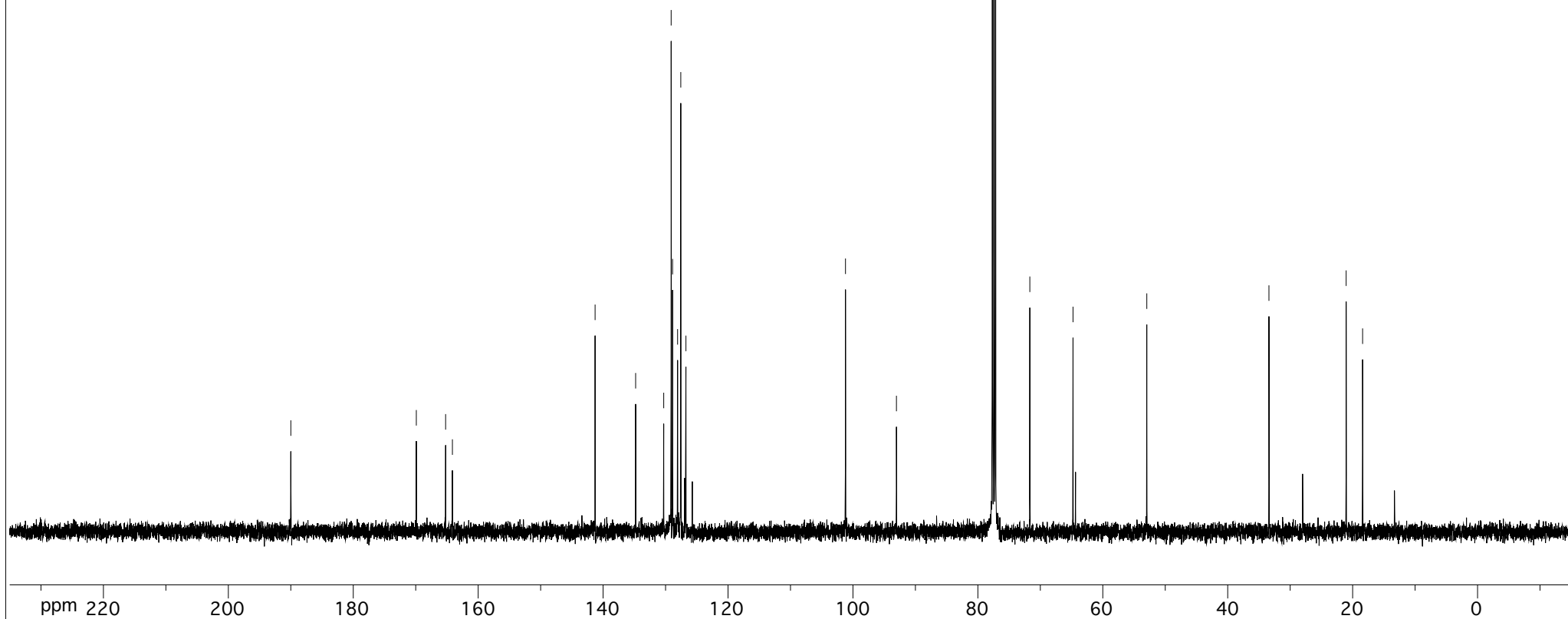
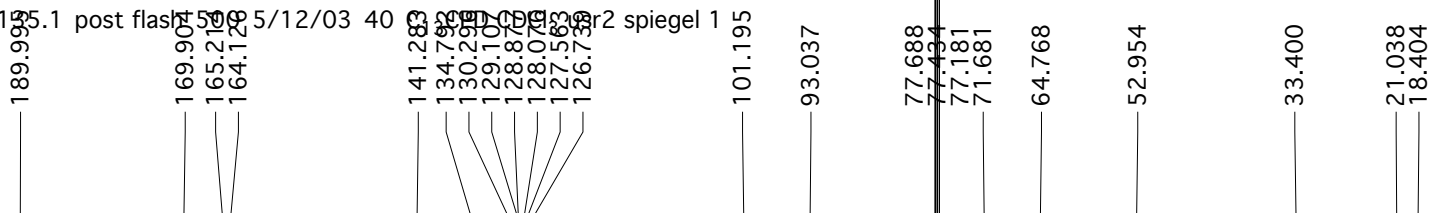
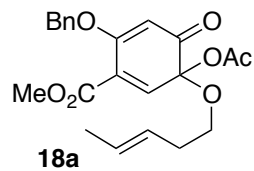
das.9.59.1 carbon 51 post flash 400b 1/9/03
13a/DAS.9.59.1/51
July 18, 2012 1:32 PM



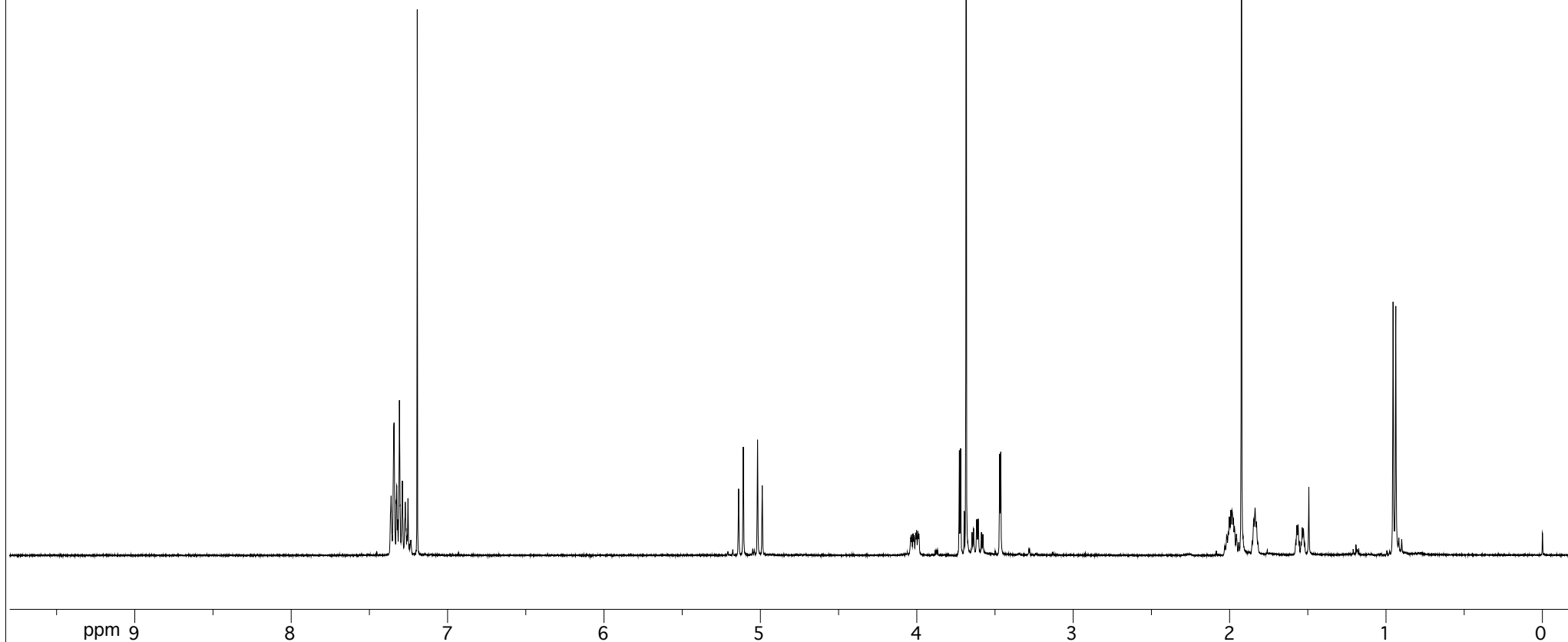
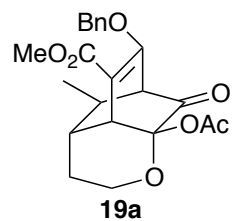
VIP number 1230966 das.10.135.1 post flash 500 5/12/03 20 PROTON_PEAK CDCl₃ usr2 spiegel 2
18a/DAS.10.135.1/20
July 18, 2012 1:34 PM



VIP number 1230966 das.10.135.1 post flash 1500 5/12/03 40 0
18a/DAS.10.135.1/40
July 18, 2012 1:34 PM

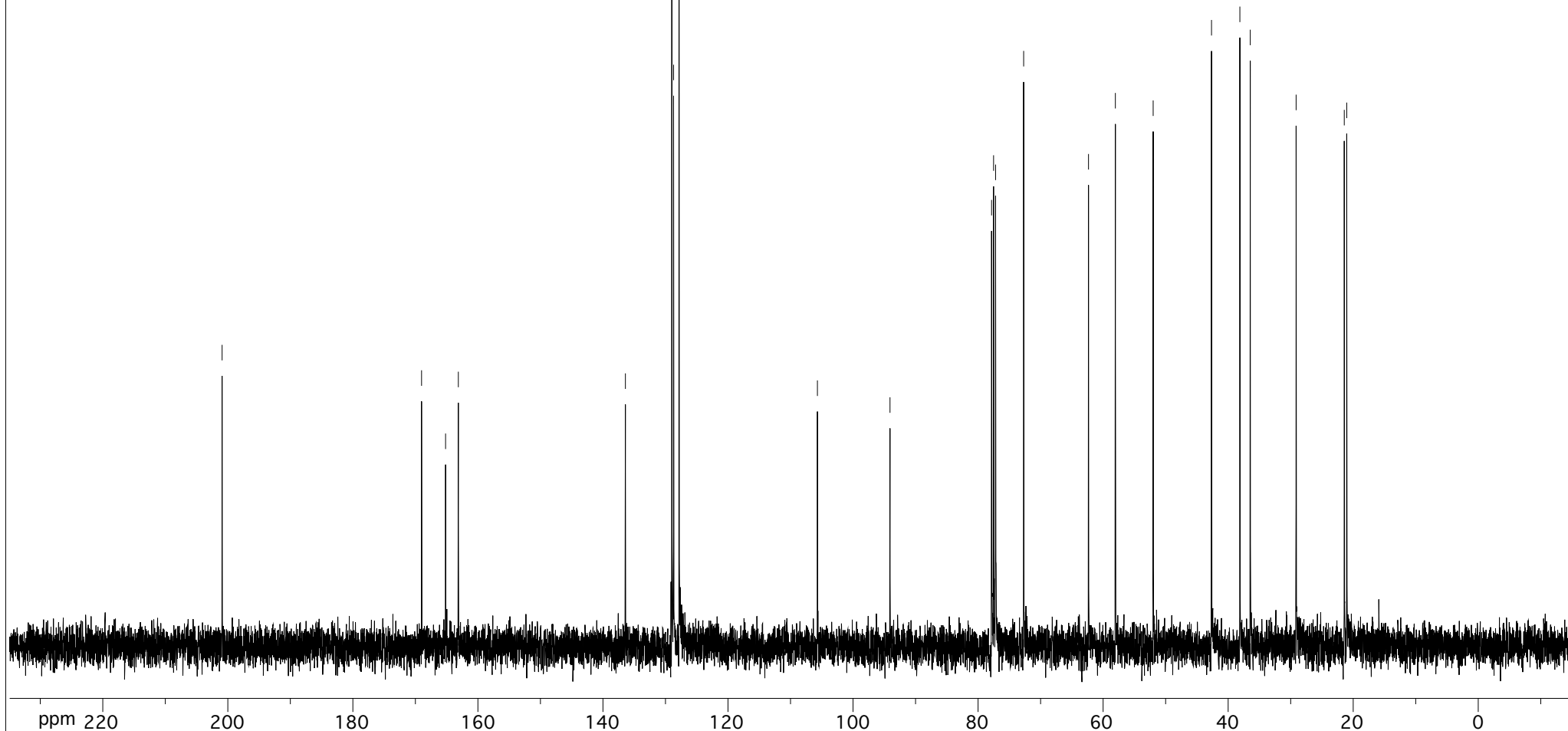
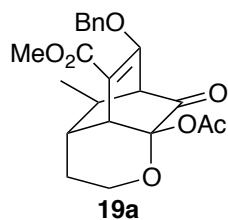


das.9.241.1 flash 400b 3/10/03 70
19a/DAS.9.241.1/70
July 18, 2012 1:35 PM

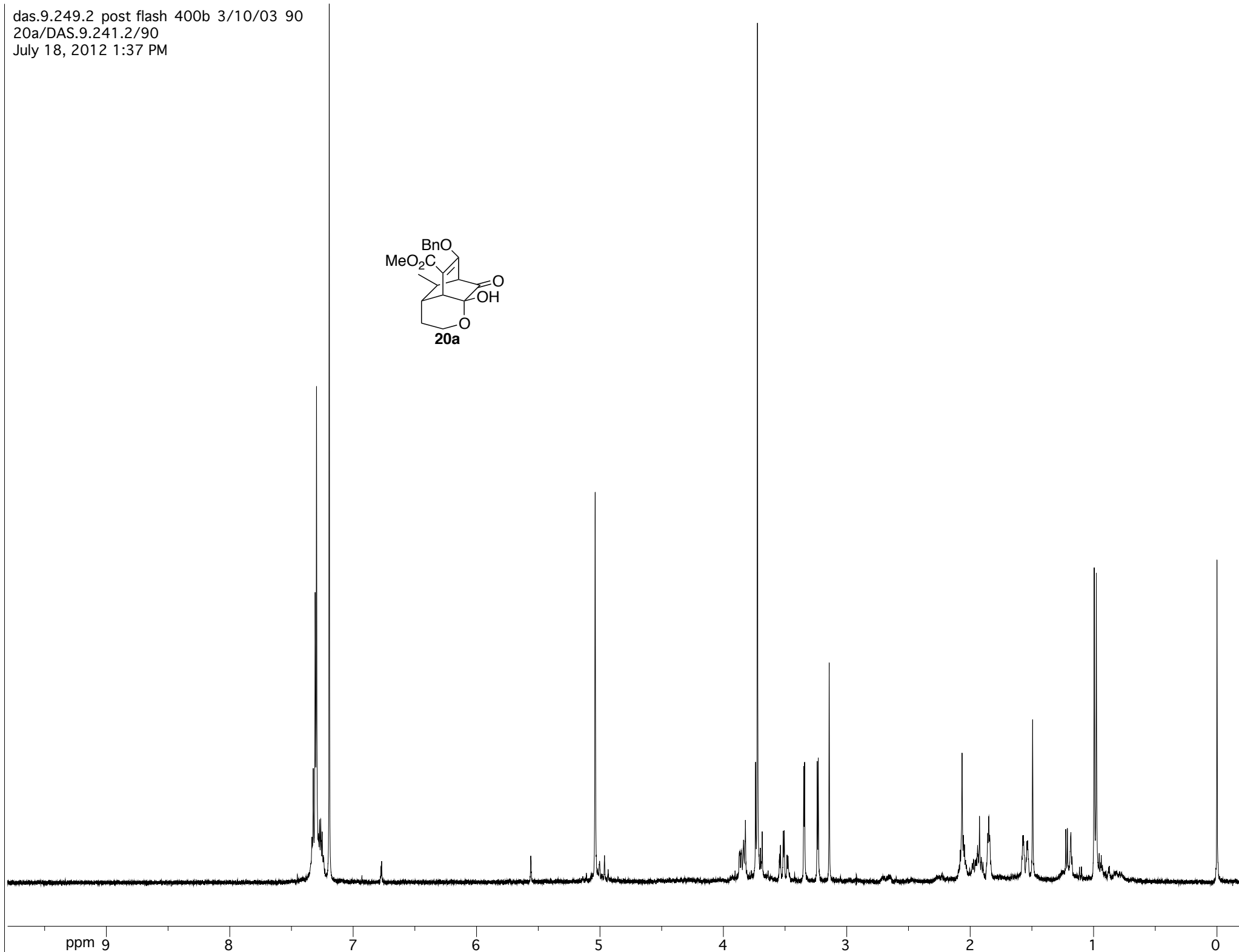
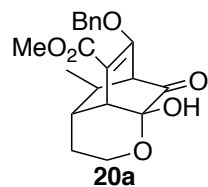


das.9.241.1carbon 400 3/10/03 80
19a/DAS.9.241.1/80
July 18, 2012 1:36 PM

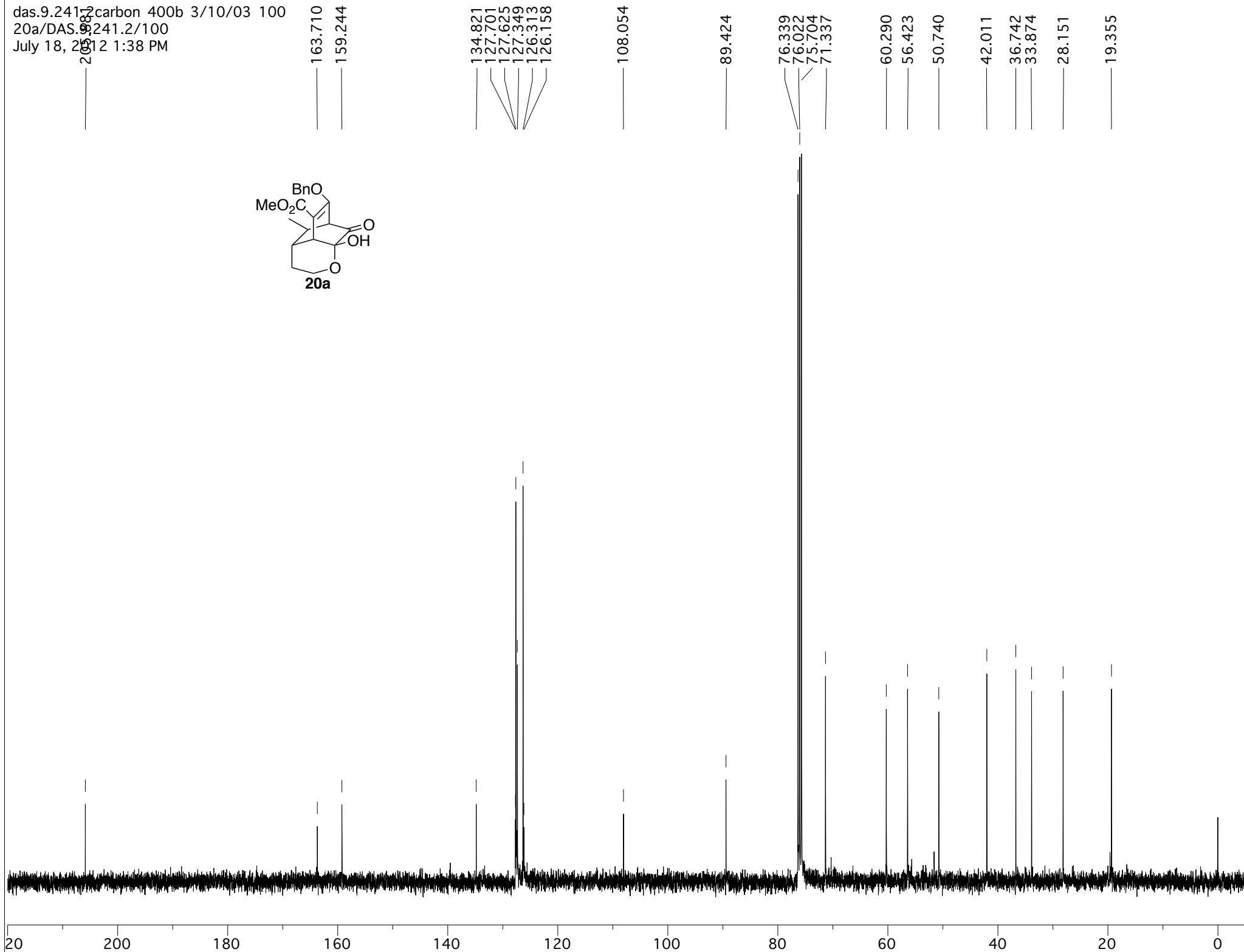
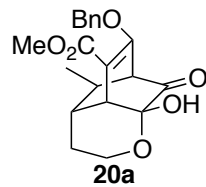
200.910 169.013 165.172 163.130 136.408 128.986 128.700 127.834 105.697 94.087 77.839 77.520 77.201 72.709 62.337 58.040 51.995 42.661 38.126 36.463 29.130 21.429 21.032



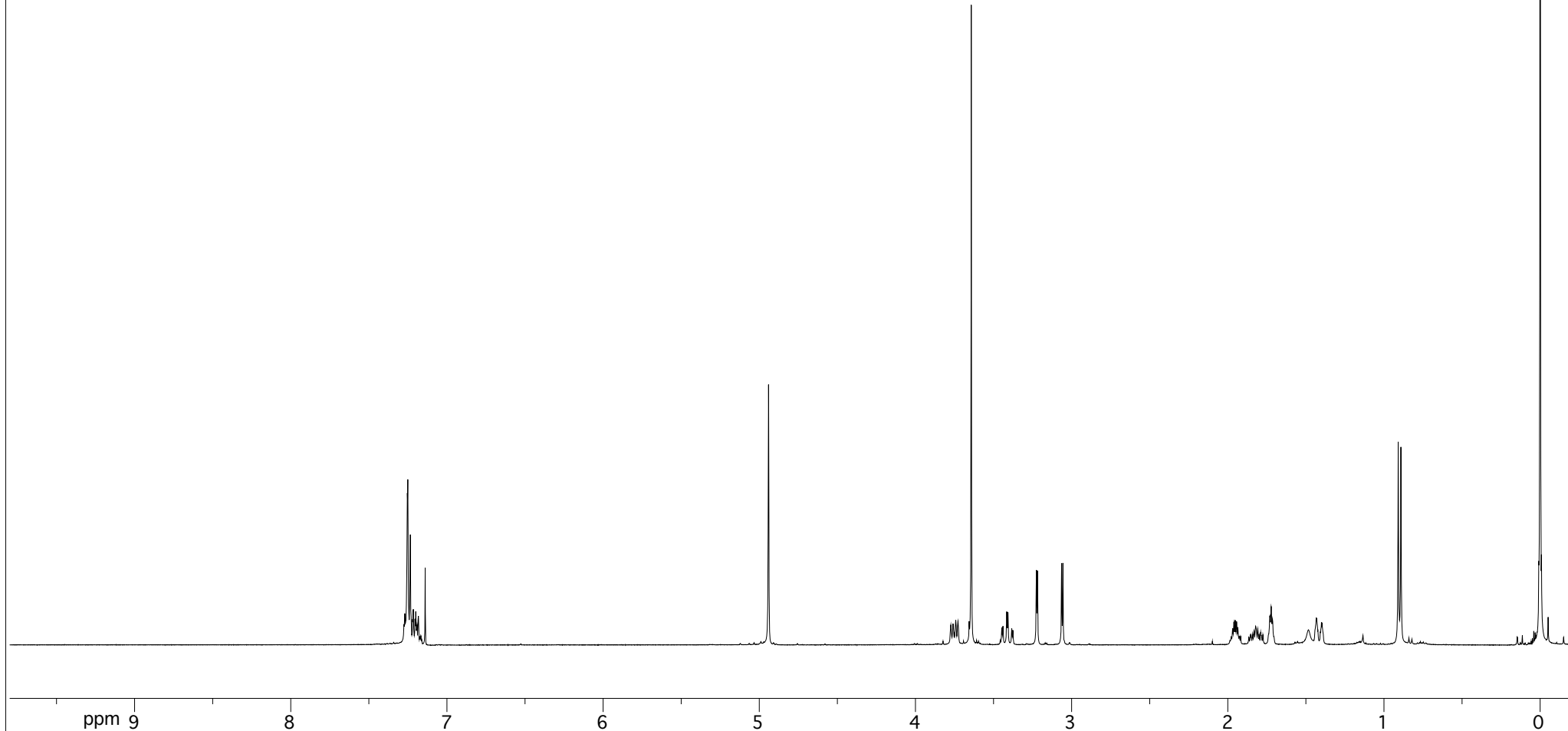
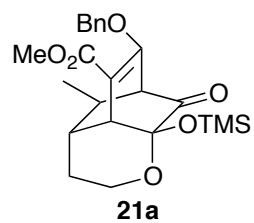
das.9.249.2 post flash 400b 3/10/03 90
20a/DAS.9.241.2/90
July 18, 2012 1:37 PM



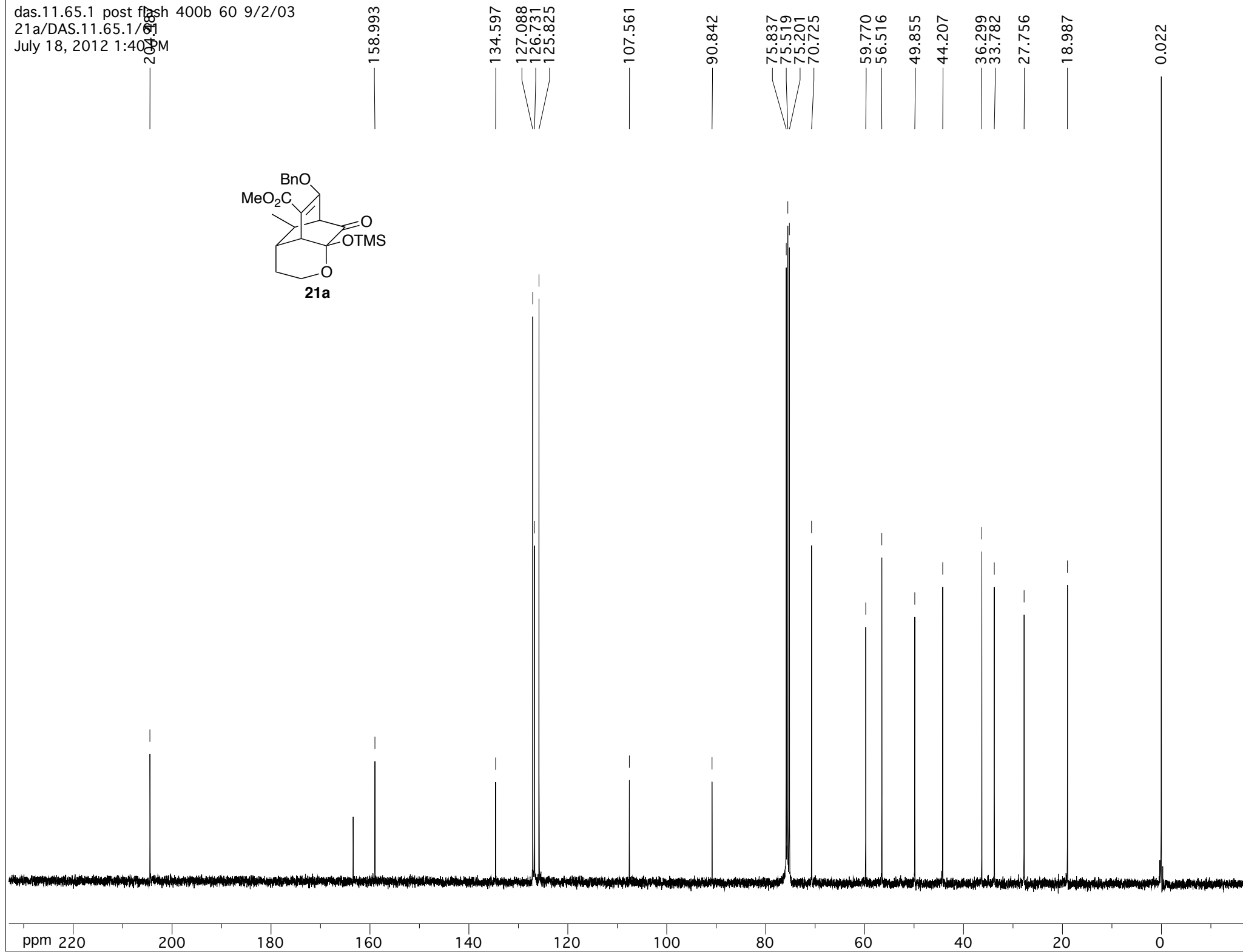
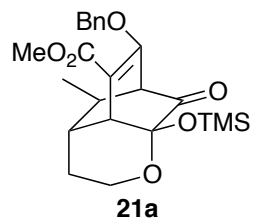
das.9.241-2 carbon 400b 3/10/03 100
20a/DAS.9.241.2/100
July 18, 2012 1:38 PM



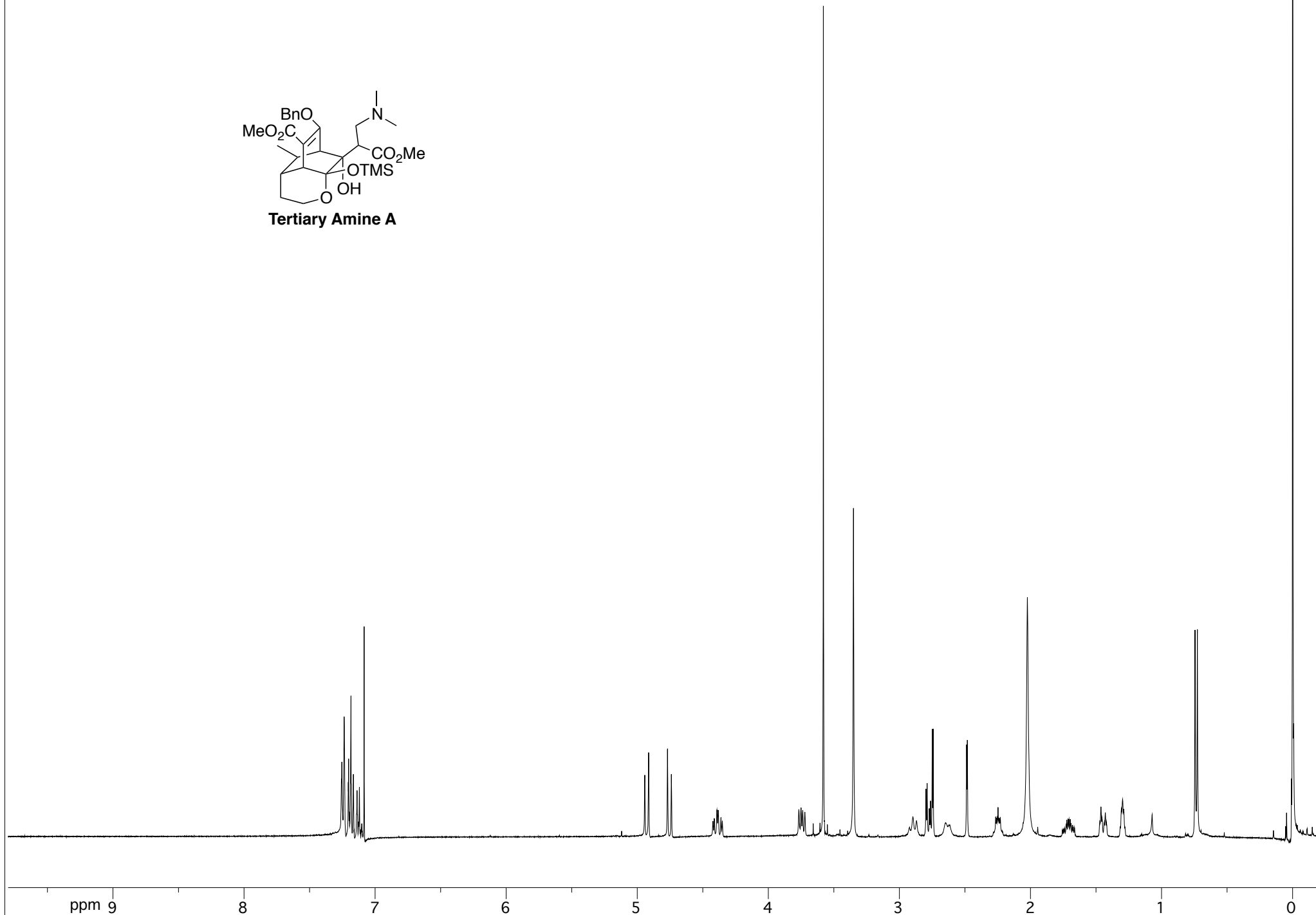
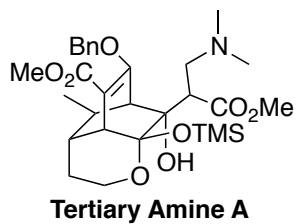
das.11.65.1 post flash 400b 60 9/2/03
21a/DAS.11.65.1/60
July 18, 2012 1:39 PM



das.11.65.1 post flash 400b 60 9/2/03
21a/DAS.11.65.1/61
July 18, 2012 1:40 PM

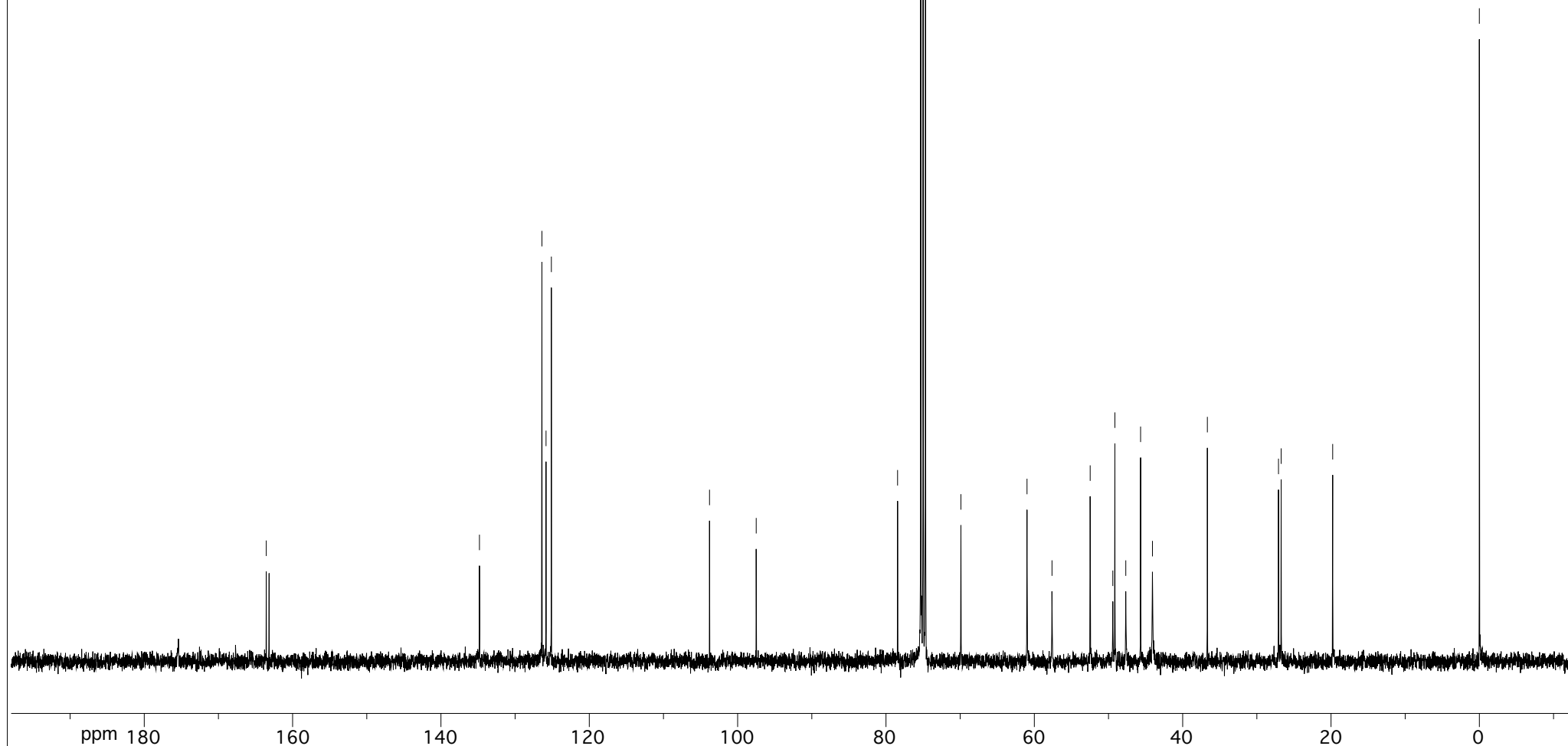
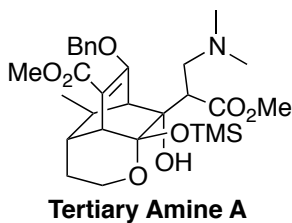


bmt6-2_30May2008
Tertiary Amine a/TAA/PROTON
July 18, 2012 1:41 PM

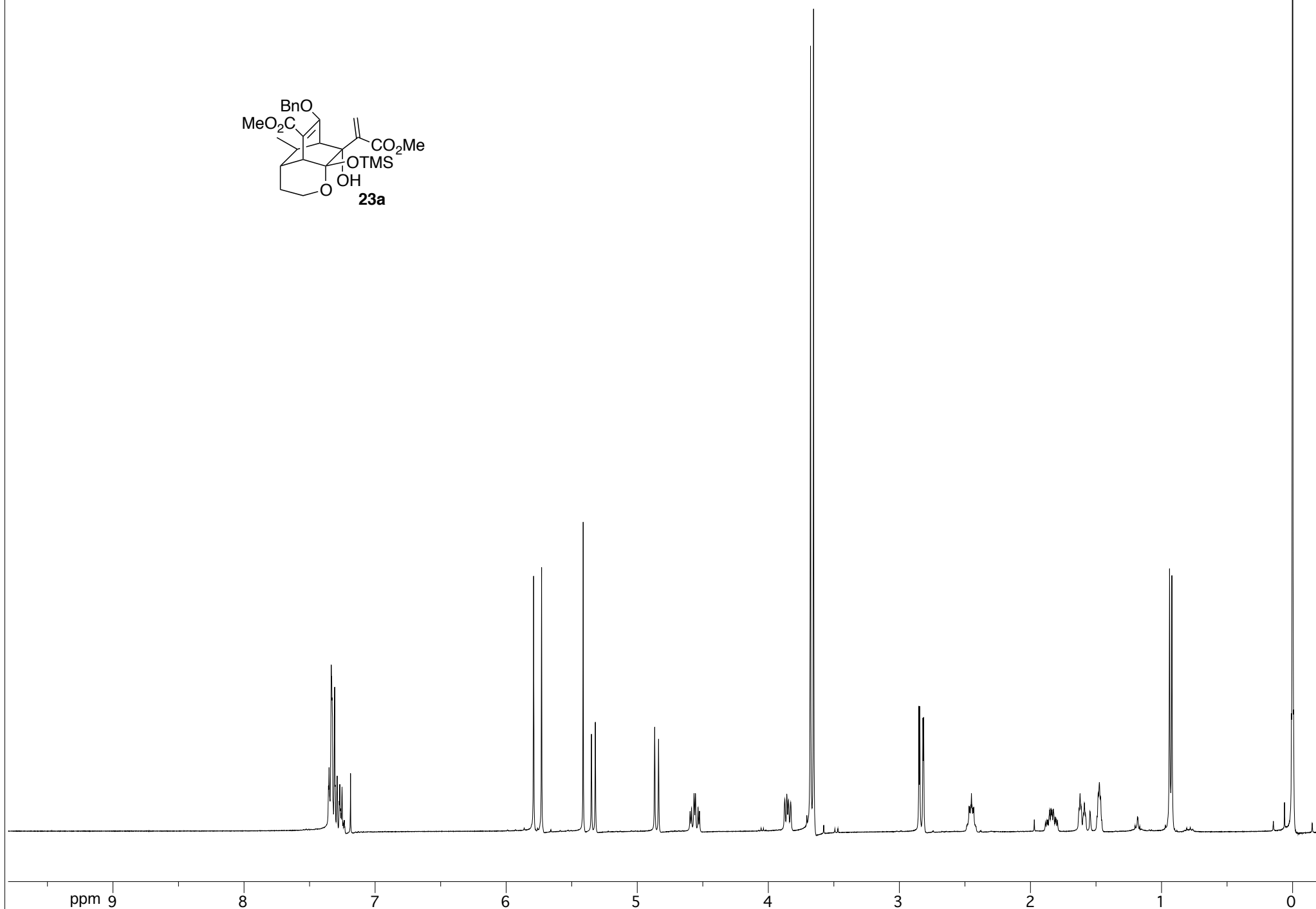
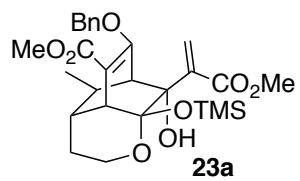


bmt6-2_30May2008
Tertiary Amine a/TAA/CARBON
July 18, 2012 1:41 PM

163.52
134.809
126.385
125.832
125.113
103.791
97.496
78.438
75.312
74.991
74.675
69.899
60.986
57.618
52.471
49.420
49.138
47.670
45.674
44.074
36.673
27.082
26.724
19.777
0.005



bmt6-37_02Jun2008
23a/23a/PROTON
July 18, 2012 1:42 PM



bmt6-37_02Jun2008
23a/23a/CARBON
July 18, 2012 1:43 PM

168.984
164.617
163.712

141.868
135.455
127.079
126.621
126.125
118.963

104.824
97.949

79.899
75.918
75.599
75.282
72.247

61.454

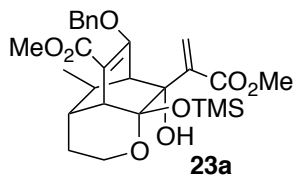
51.963
50.319
49.731
46.082

38.406

27.683
27.560

20.256

0.006



ppm 180

160

140

120

100

80

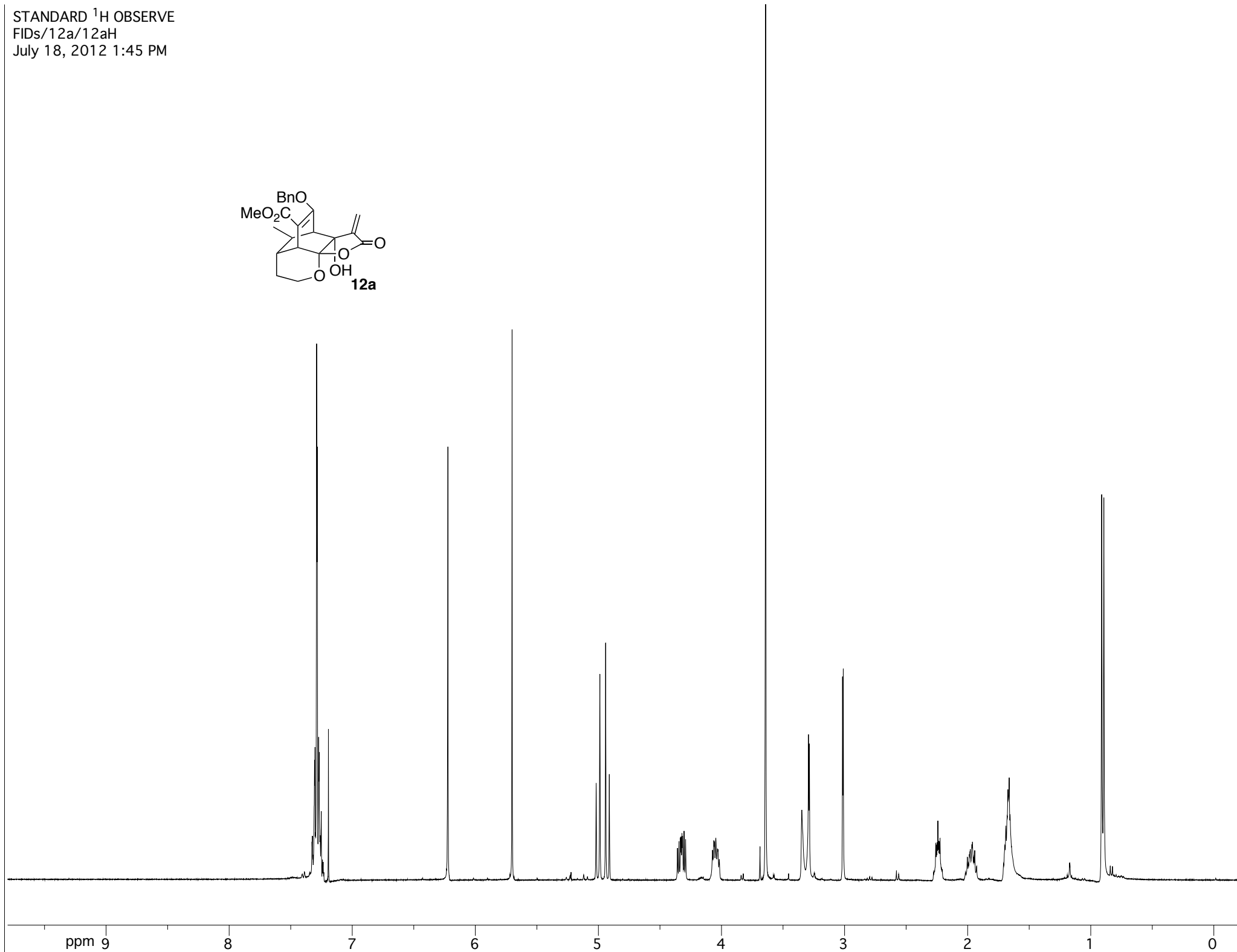
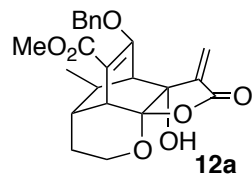
60

40

20

0

STANDARD ^1H OBSERVE
FIDs/12a/12aH
July 18, 2012 1:45 PM



¹³C OBSERVE
FIDs/12a/12a
July 18, 2012 1:46 PM

166.396
165.442
164.714

141.110

136.365

128.826

128.517

127.550

126.658

104.623

103.608

77.589

77.273

76.958

76.254

72.688

64.199

52.584

51.890

42.584

42.531

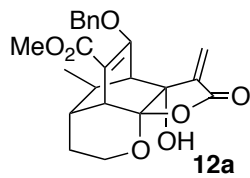
39.122

39.095

31.785

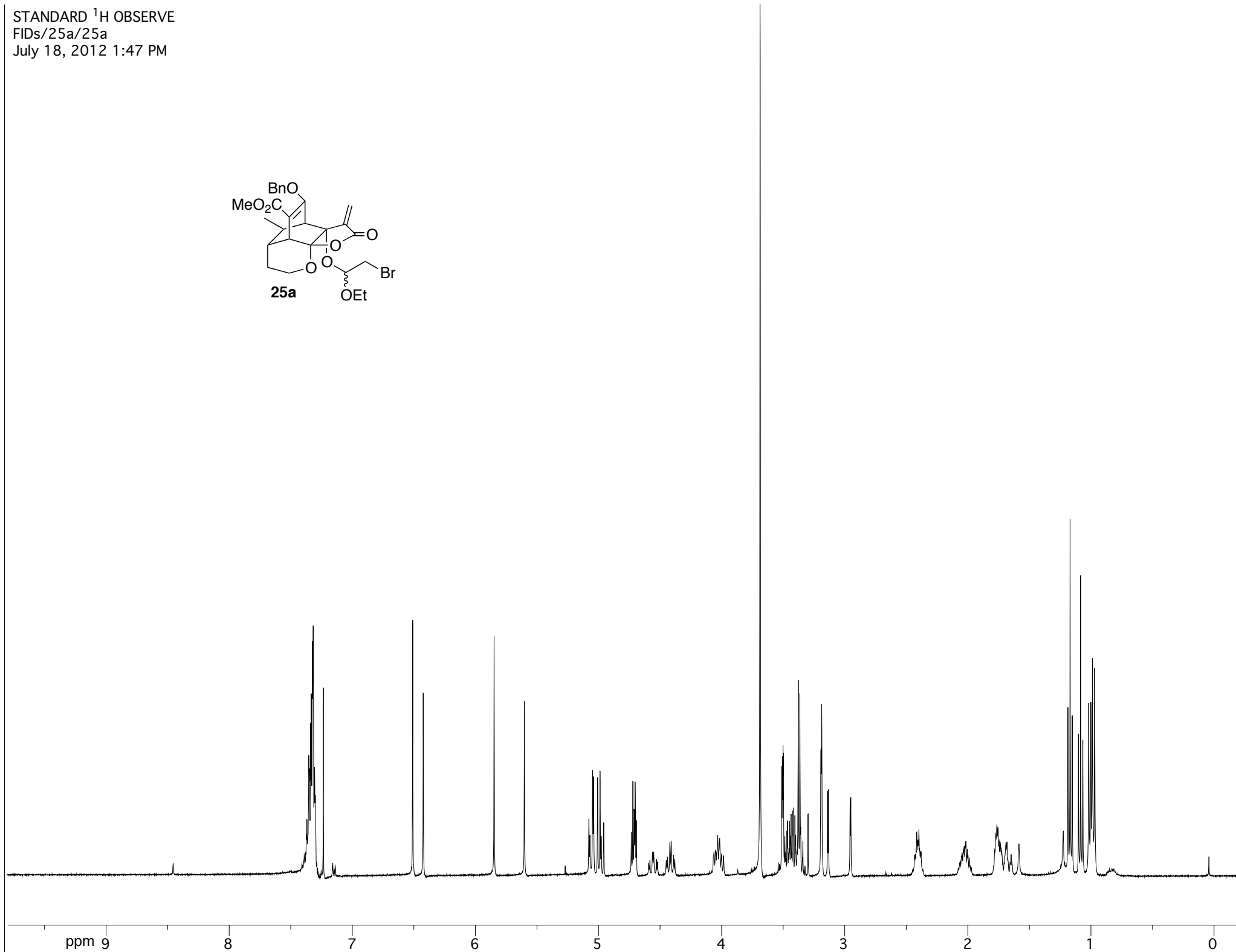
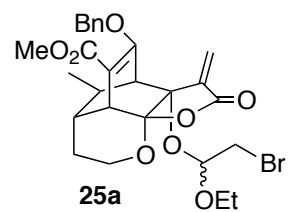
29.068

21.351

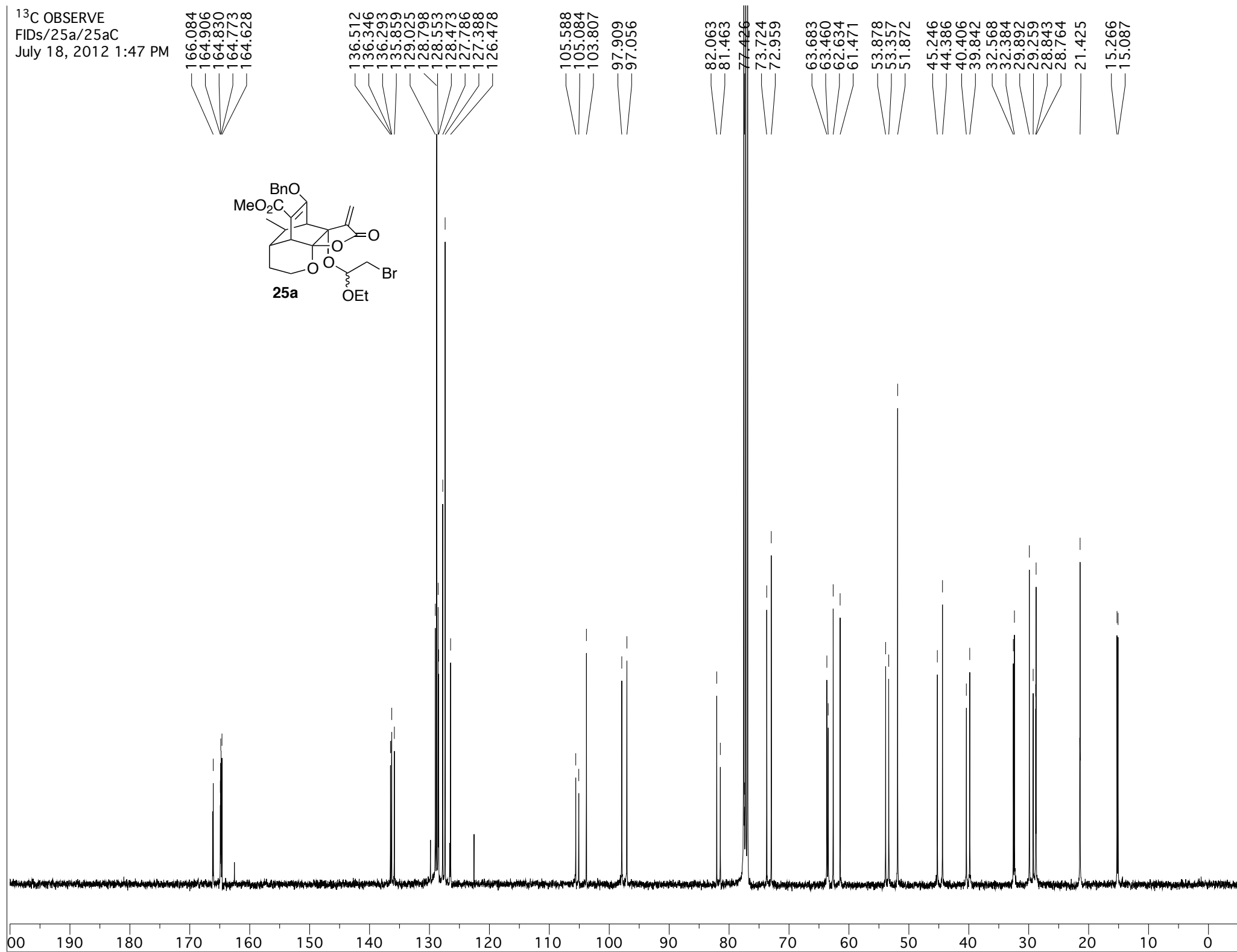
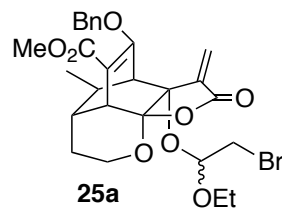


00 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0

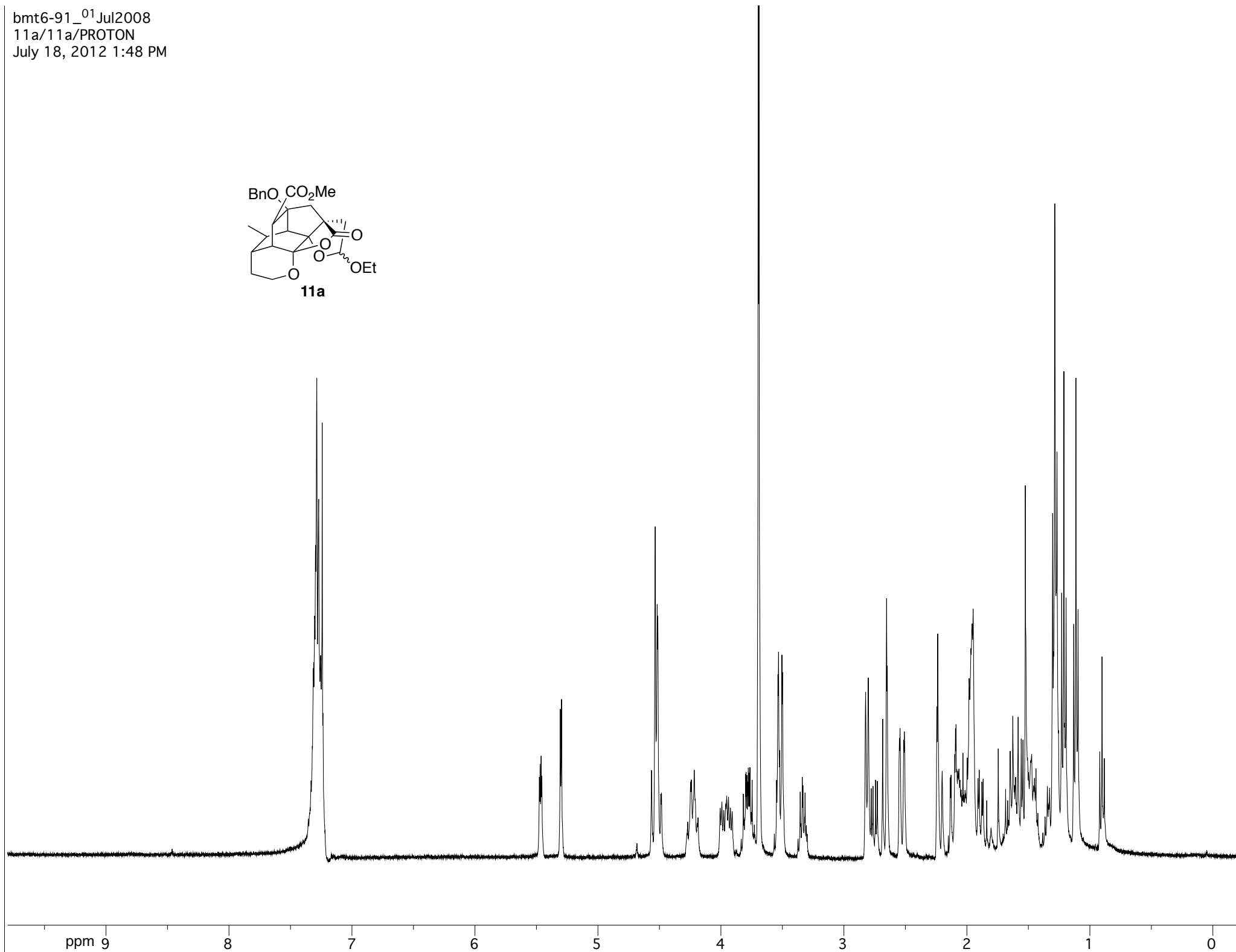
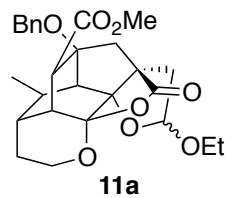
STANDARD ^1H OBSERVE
FIDs/25a/25a
July 18, 2012 1:47 PM



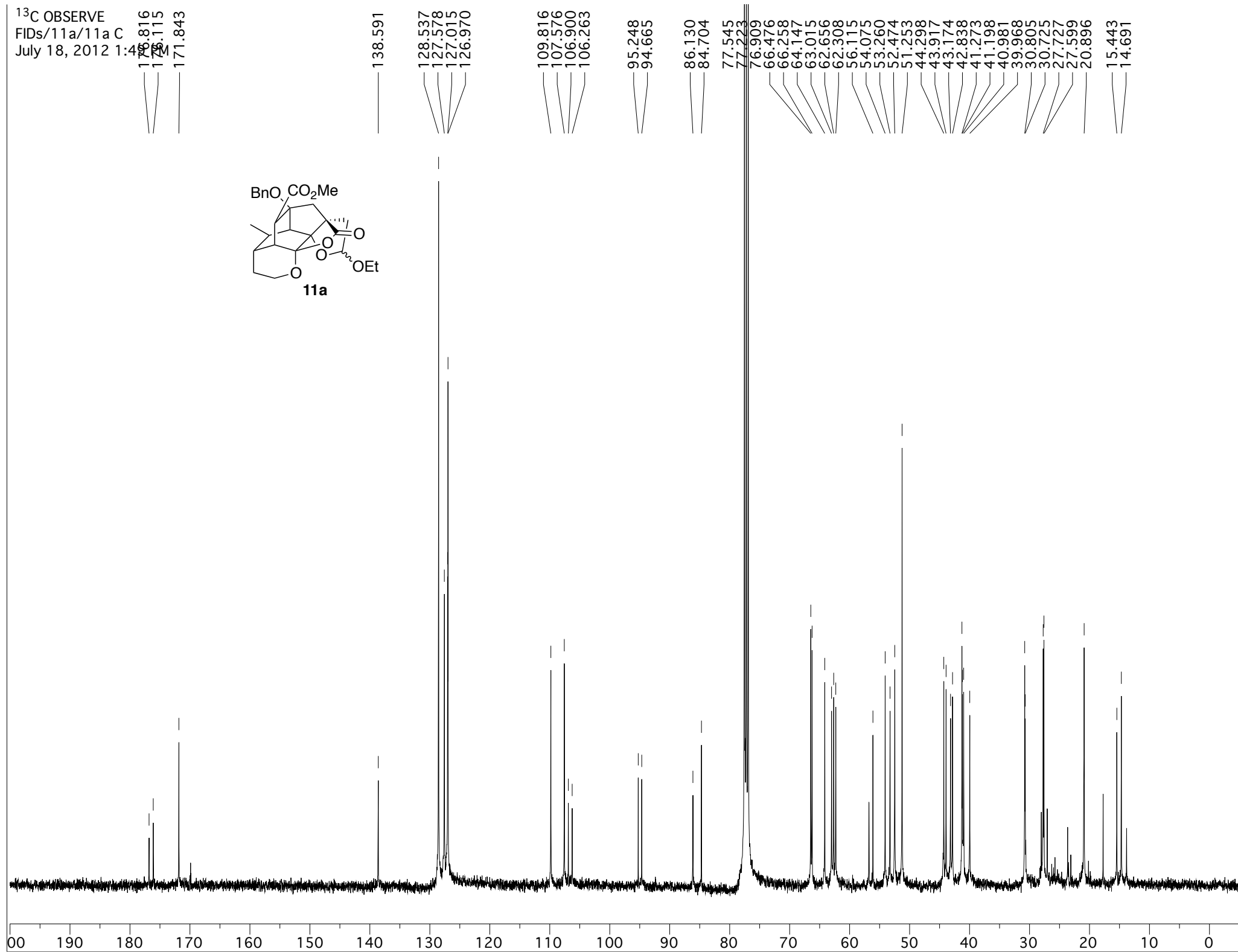
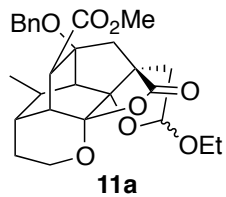
166.084
164.906
164.830
164.773
164.628



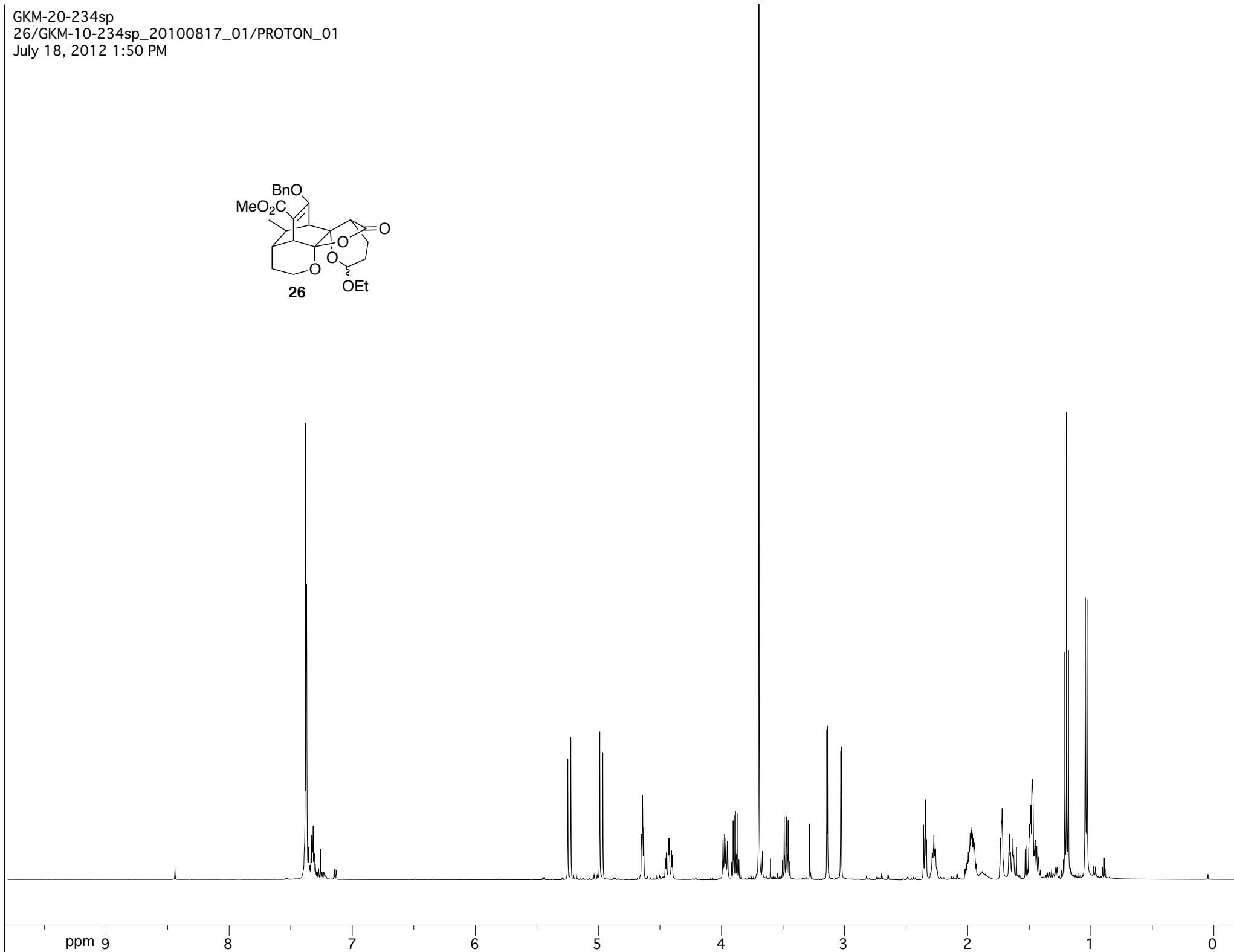
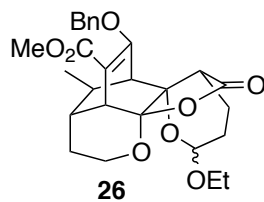
bmt6-91_01Jul2008
11a/11a/PROTON
July 18, 2012 1:48 PM



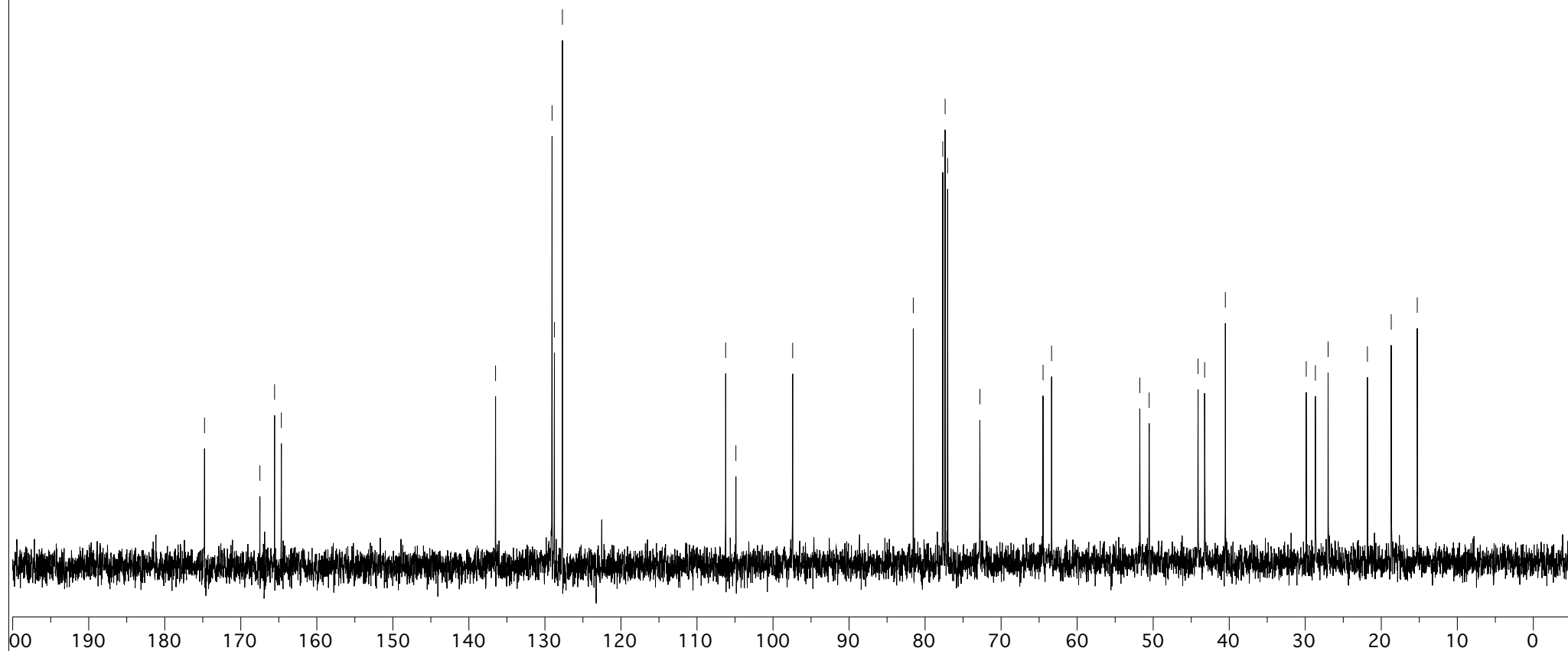
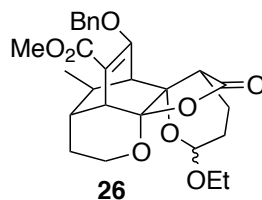
¹³C OBSERVE 8.16 11.15 71.843
FIDs/11a/11a C
July 18, 2012 1:49 PM



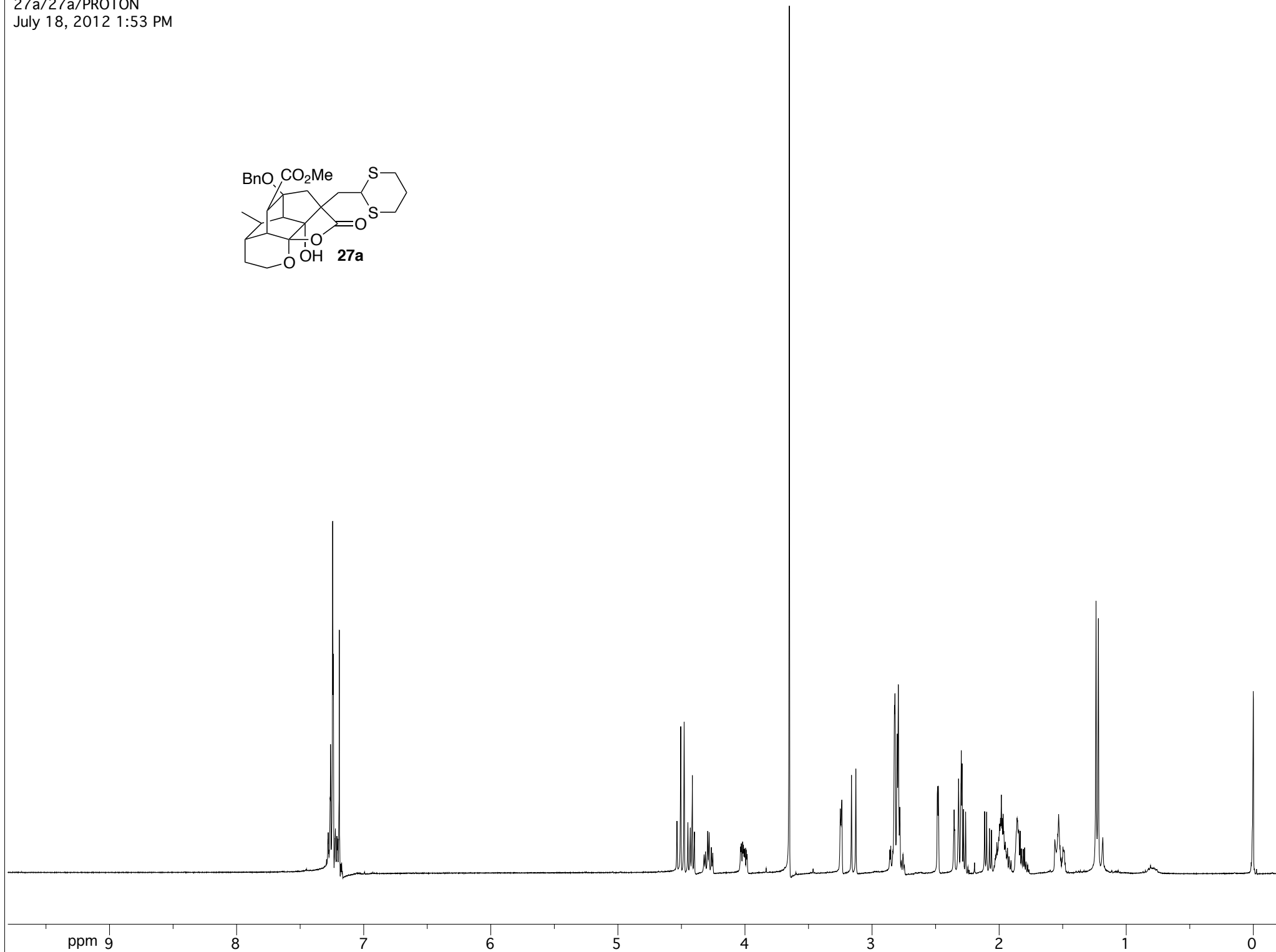
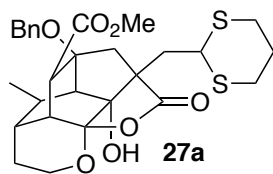
GKM-20-234sp
26/GKM-10-234sp_20100817_01/PROTON_01
July 18, 2012 1:50 PM



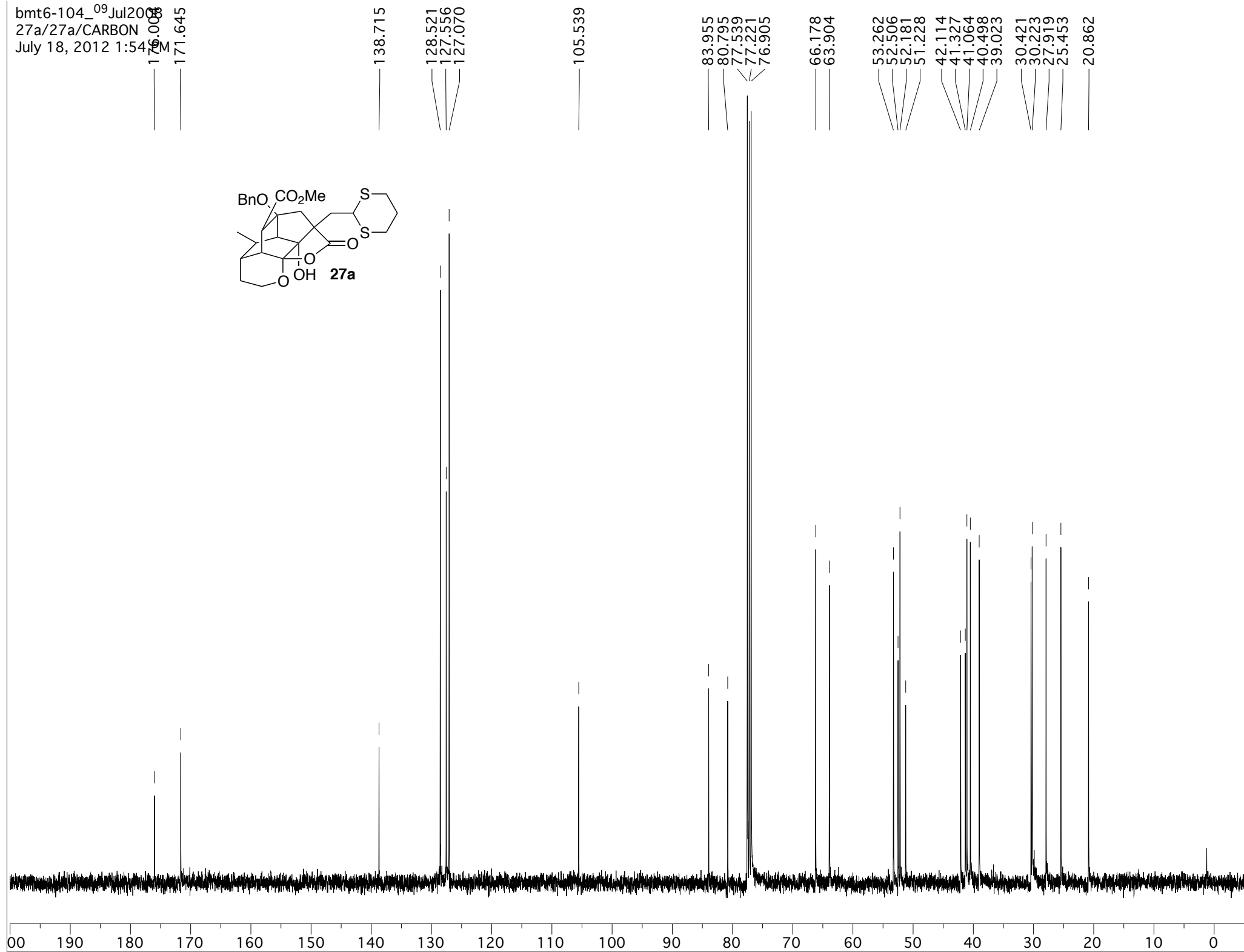
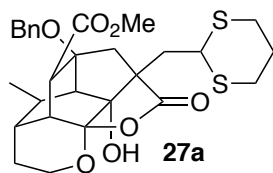
GKM-10-234 Side product 80-12 radical reaction
FIDs/26/GKM-10-234
July 18, 2012 1:52 PM



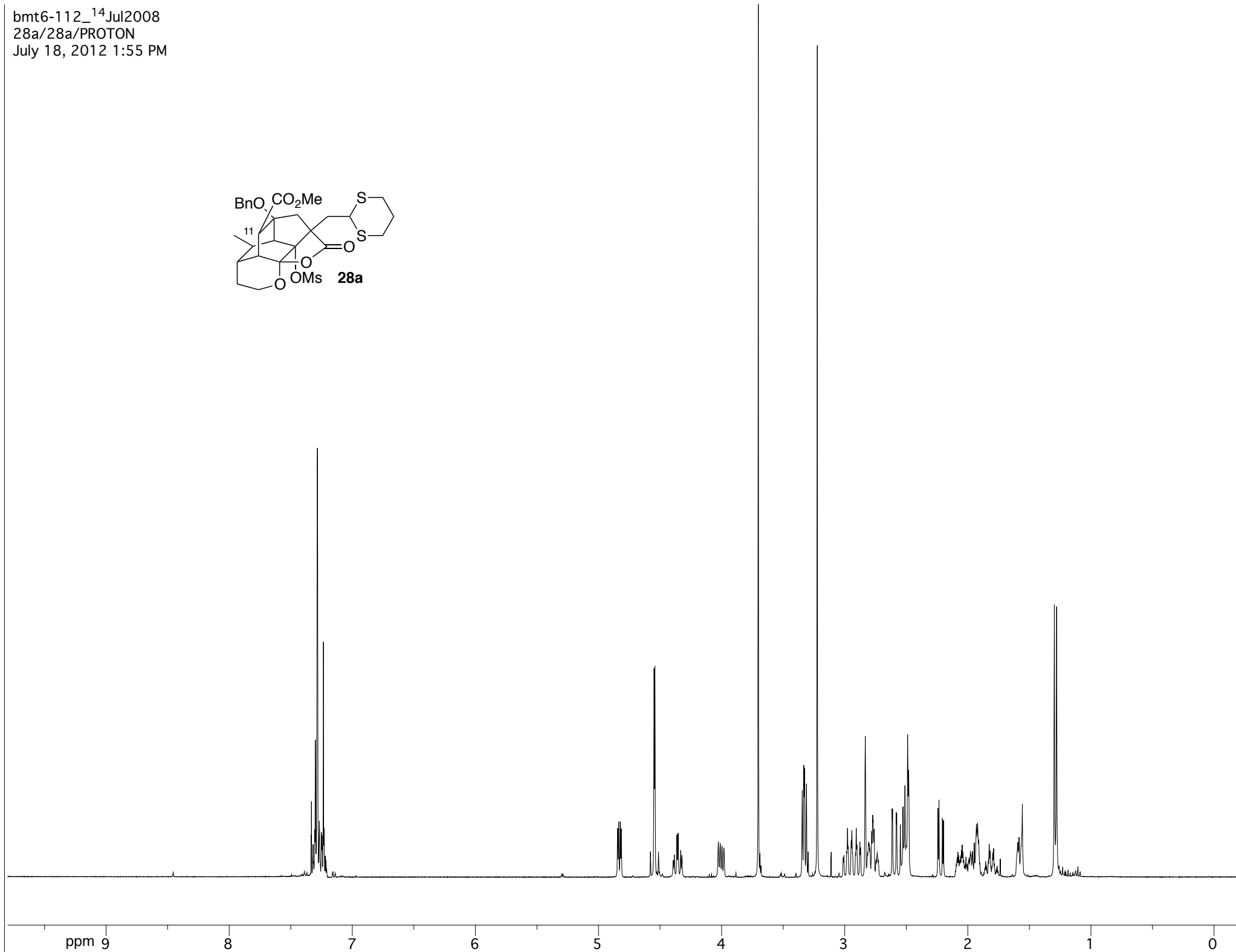
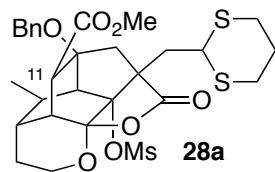
bmt6-104_09Jul2008
27a/27a/PROTON
July 18, 2012 1:53 PM



bmt6-104_09Jul2008
27a/27a/CARBON 0.008
July 18, 2012 1:54 PM 71.645



bmt6-112_14Jul2008
28a/28a/PROTON
July 18, 2012 1:55 PM



— ~~174.148~~⁹
 — 171.251

— ~~174.148~~⁹
 — 171.251

— ~~174.148~~⁹
 — 171.251

138.292

128.518

128.518	127.608
---------	---------

127.156

105.188

98.066

81.095

77.549	—
77.230	—

77.230	
76.911	

76.911

66.237

62.741

52.667

51.528	51.490
--------	--------

[illegible]

42.154
41.662

41.962
40.975

40.837
40.522

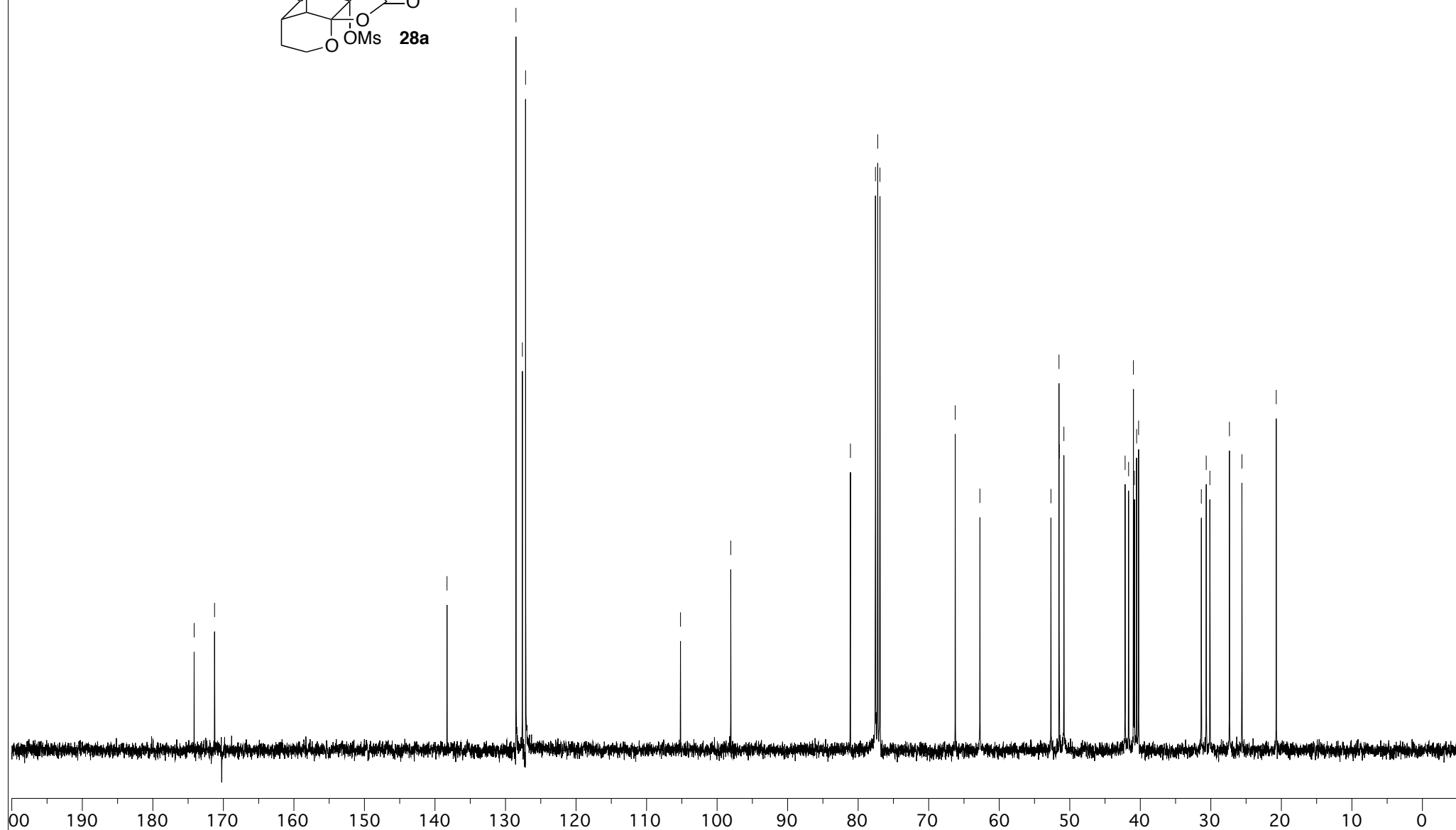
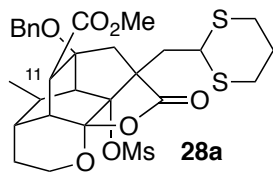
40.522
40.242

31.351	—
30.655	—

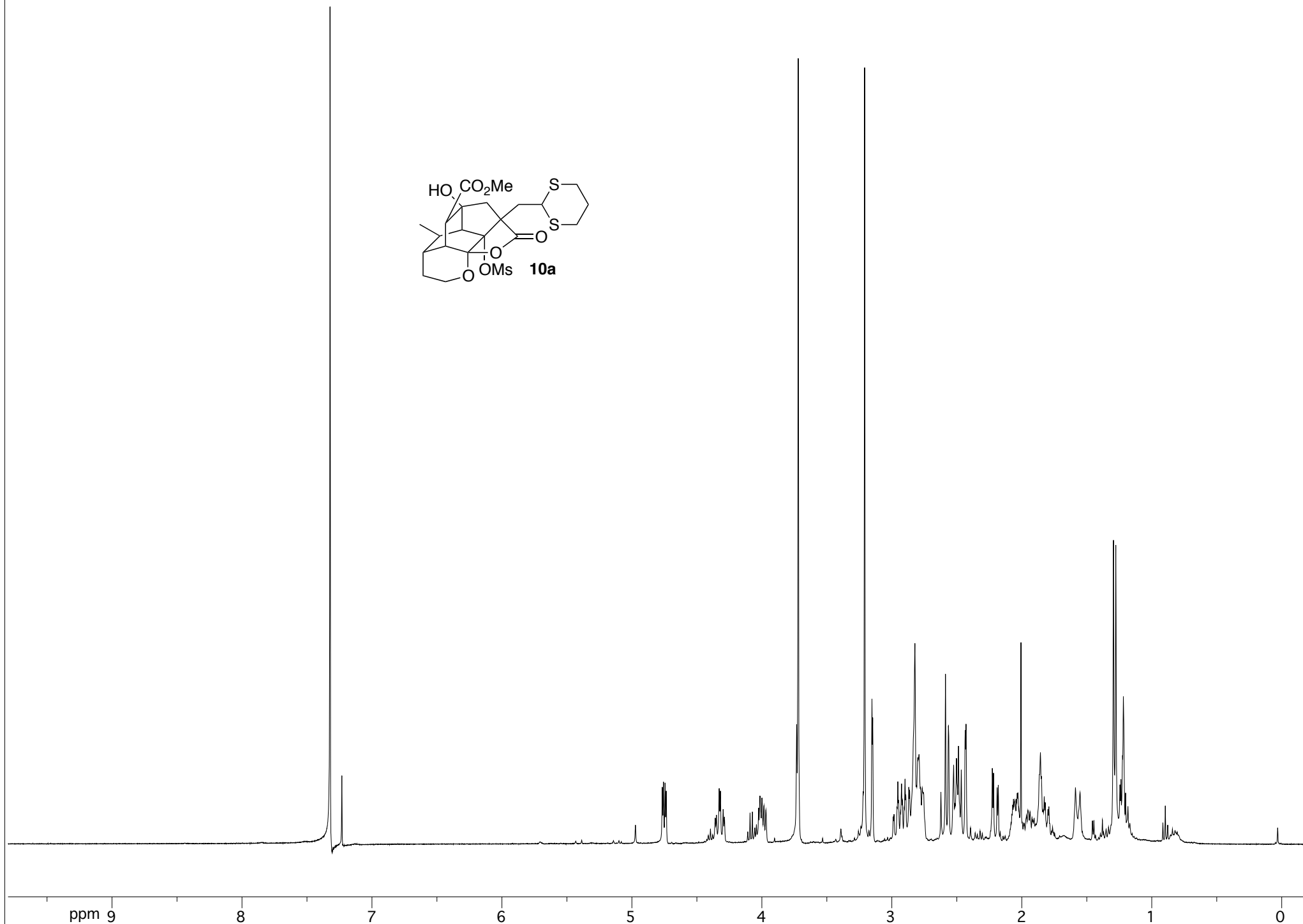
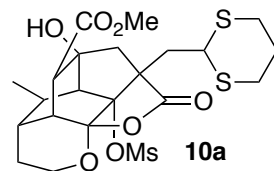
30.635	
30.135	

27.363
25.590

25.590
20.727

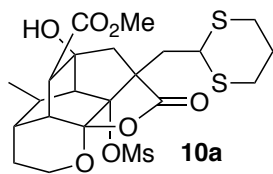


GKM-9-022 Side product, low Rf, from BBr₃ reaction
FIDs/10a/GKM-9-022H_sp
July 18, 2012 2:00 PM



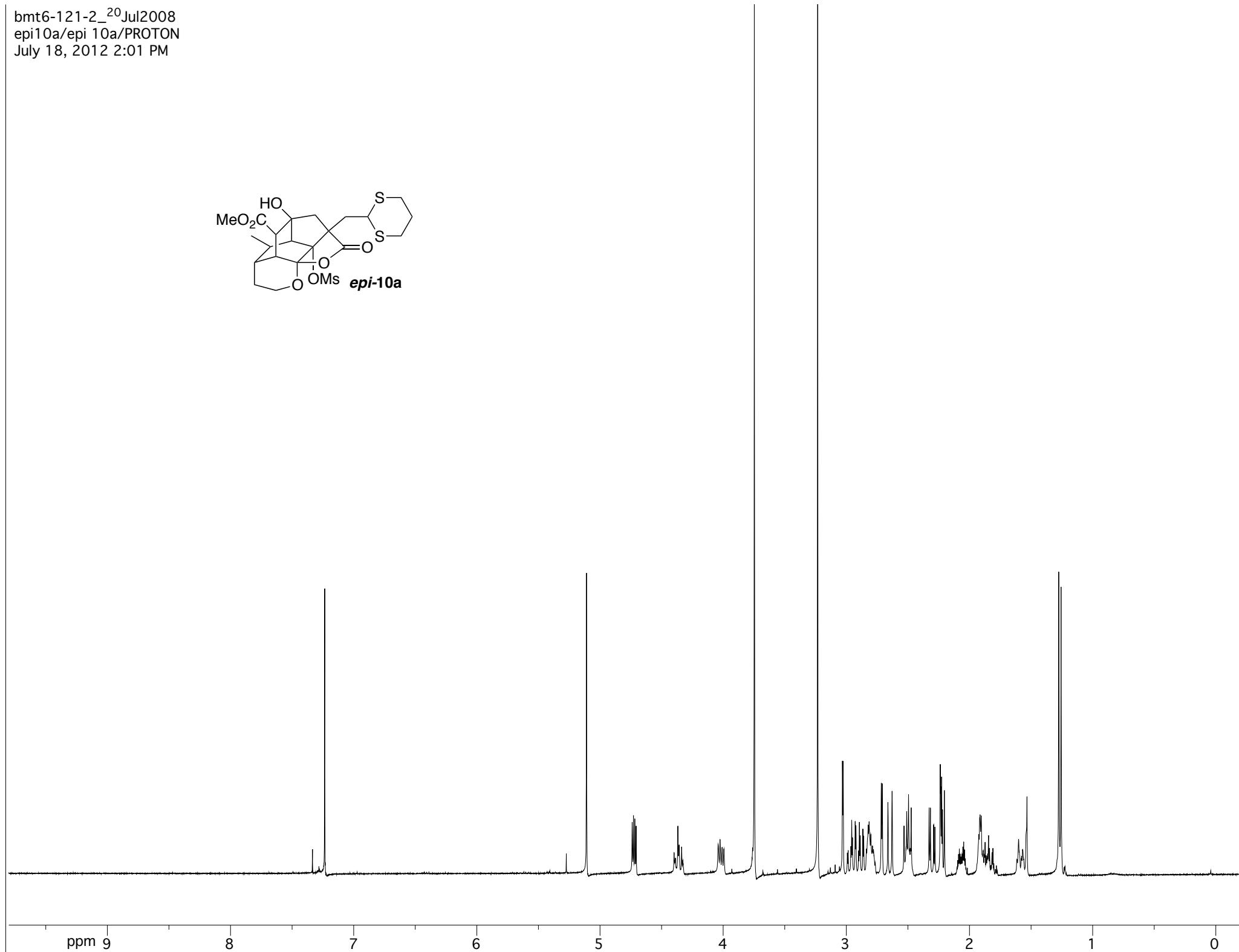
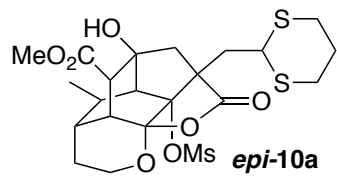
GKM-10-240 Side product from BBr₃ reaction
FIDs/10a/GKM-10-240 Comp
July 18, 2012 1:59 PM

170.84
170.84
105.070
98.168
77.475
77.155
76.841
62.553
53.821
52.646
51.840
51.423
46.709
41.991
40.895
40.644
39.859
39.514
31.110
30.478
29.959
27.660
25.476
20.724

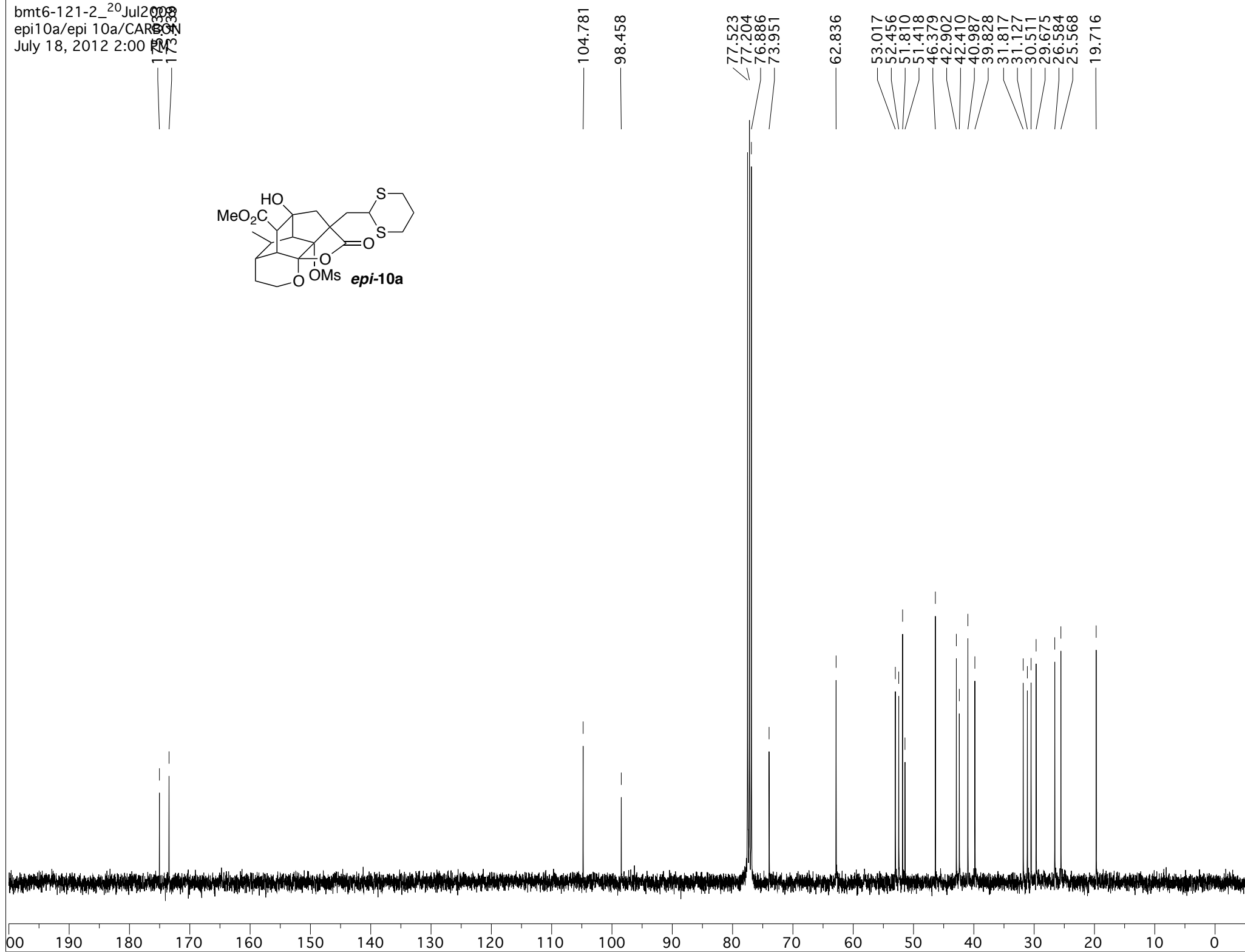
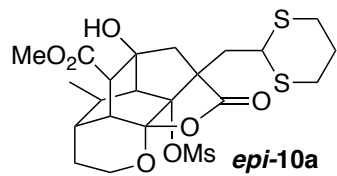


00 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0

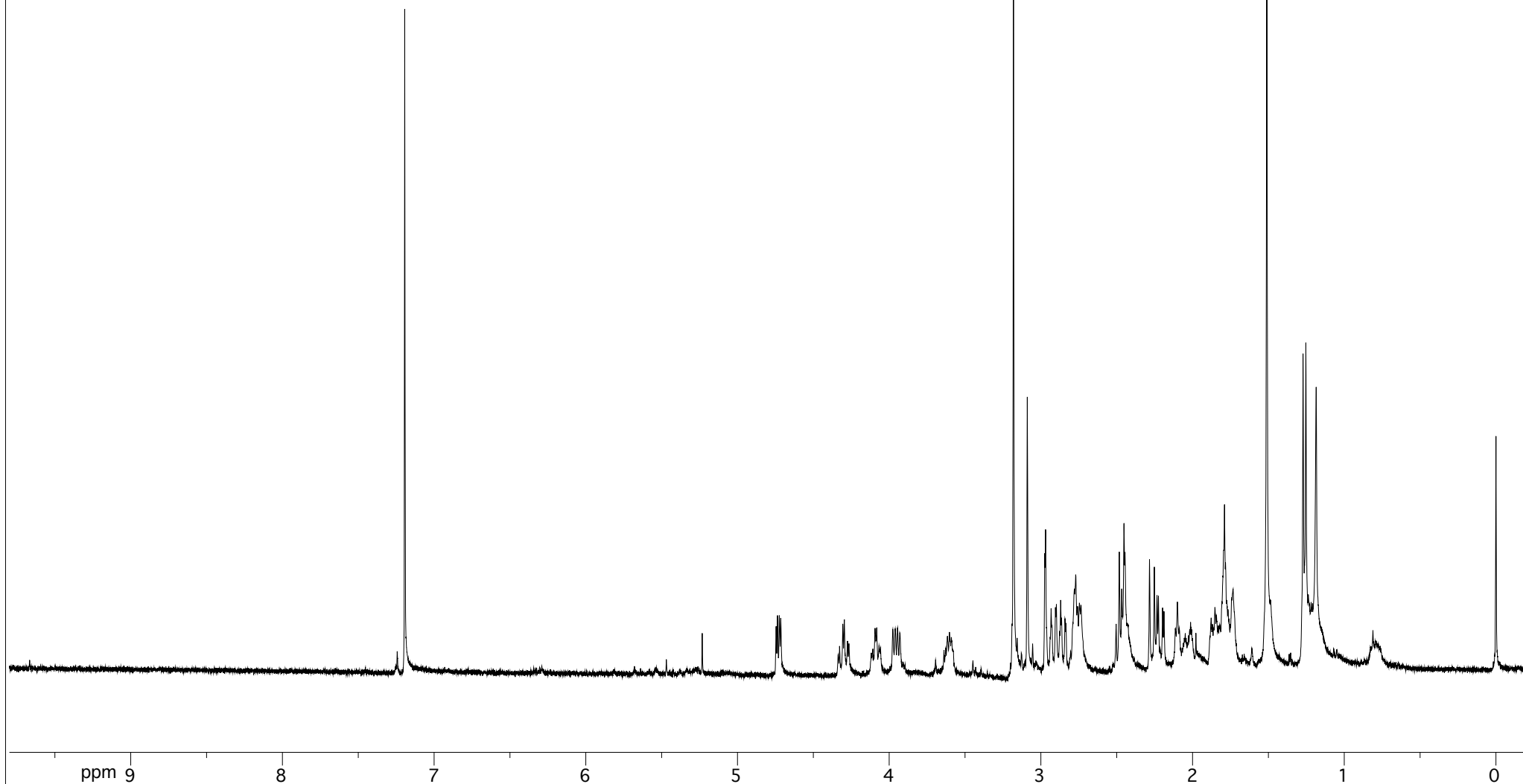
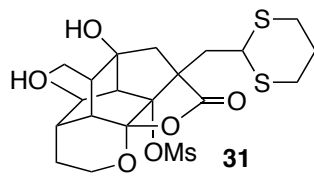
bmt6-121-2_20Jul2008
epi10a/epi 10a/PROTON
July 18, 2012 2:01 PM



bmt6-121-2_20 Jul 2008
epi10a/epi 10a/CARSON
July 18, 2012 2:00 PM

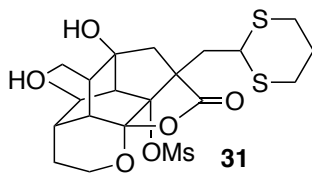


LiAlH₄ reduction product
31/GKM-7-232_14Aug2009/PROTON
July 18, 2012 2:03 PM



GKM-7-232 LiAlH₄ reduction product
FIDs/31/GKM-7-232
July 18, 2012 2:03 PM

1.00



105.259

98.454

77.405

75.161

63.085

62.552

54.150

52.193

51.449

43.624

42.432

42.018

40.883

40.036

31.274

31.223

30.567

29.887

29.643

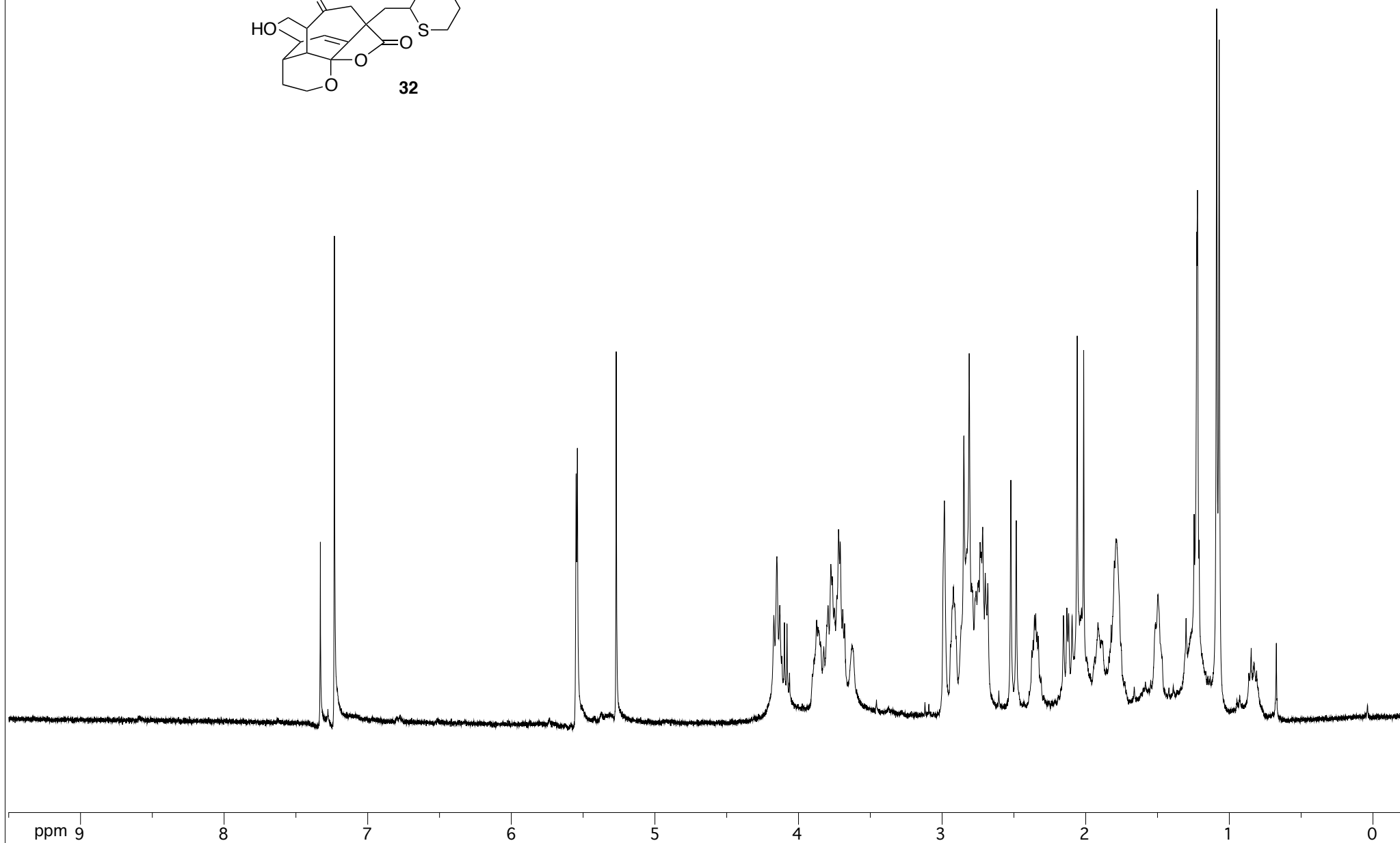
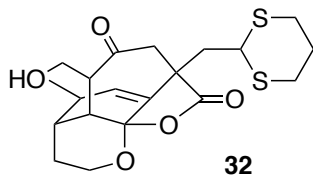
26.737

25.574

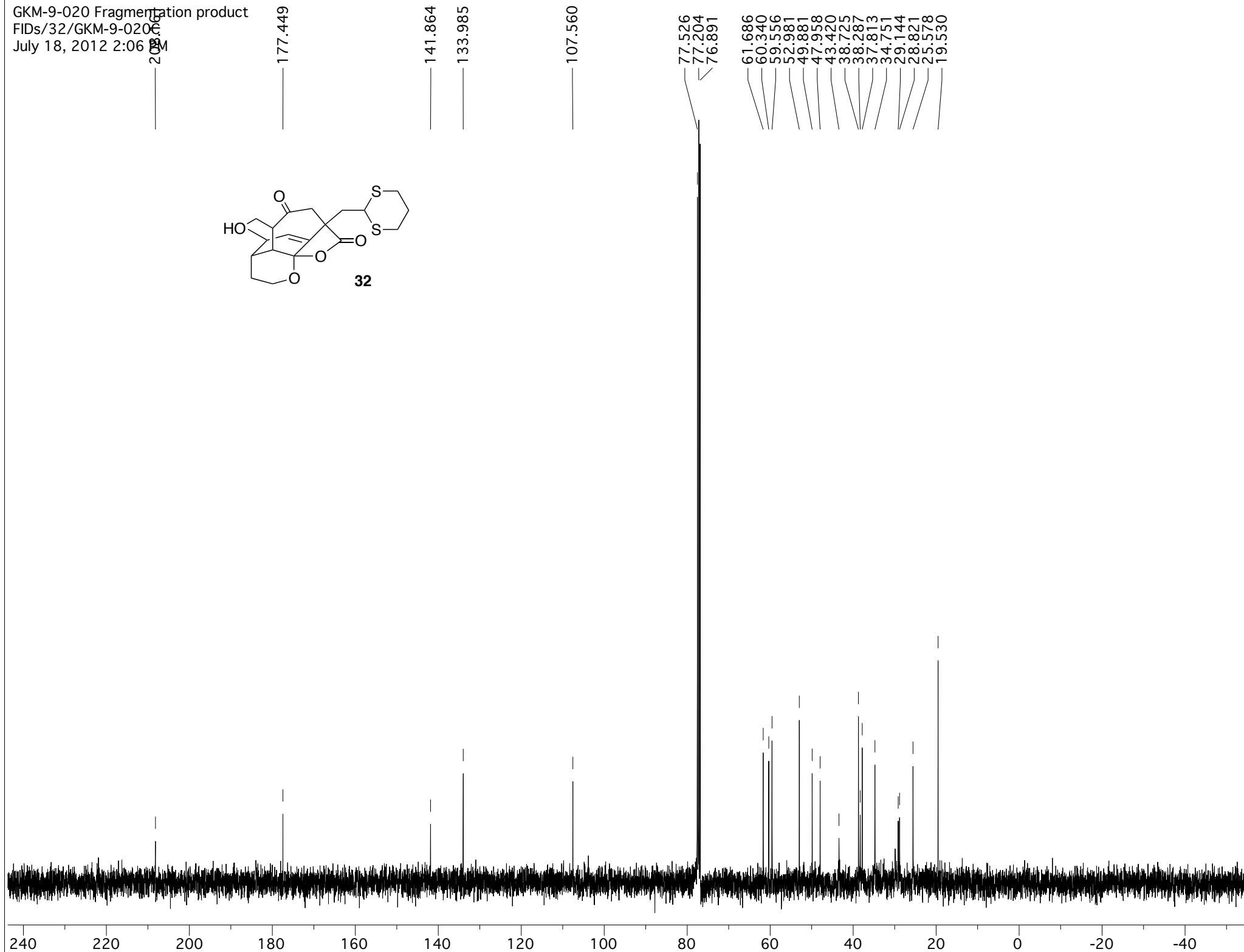
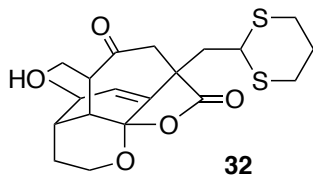
19.999

00 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0

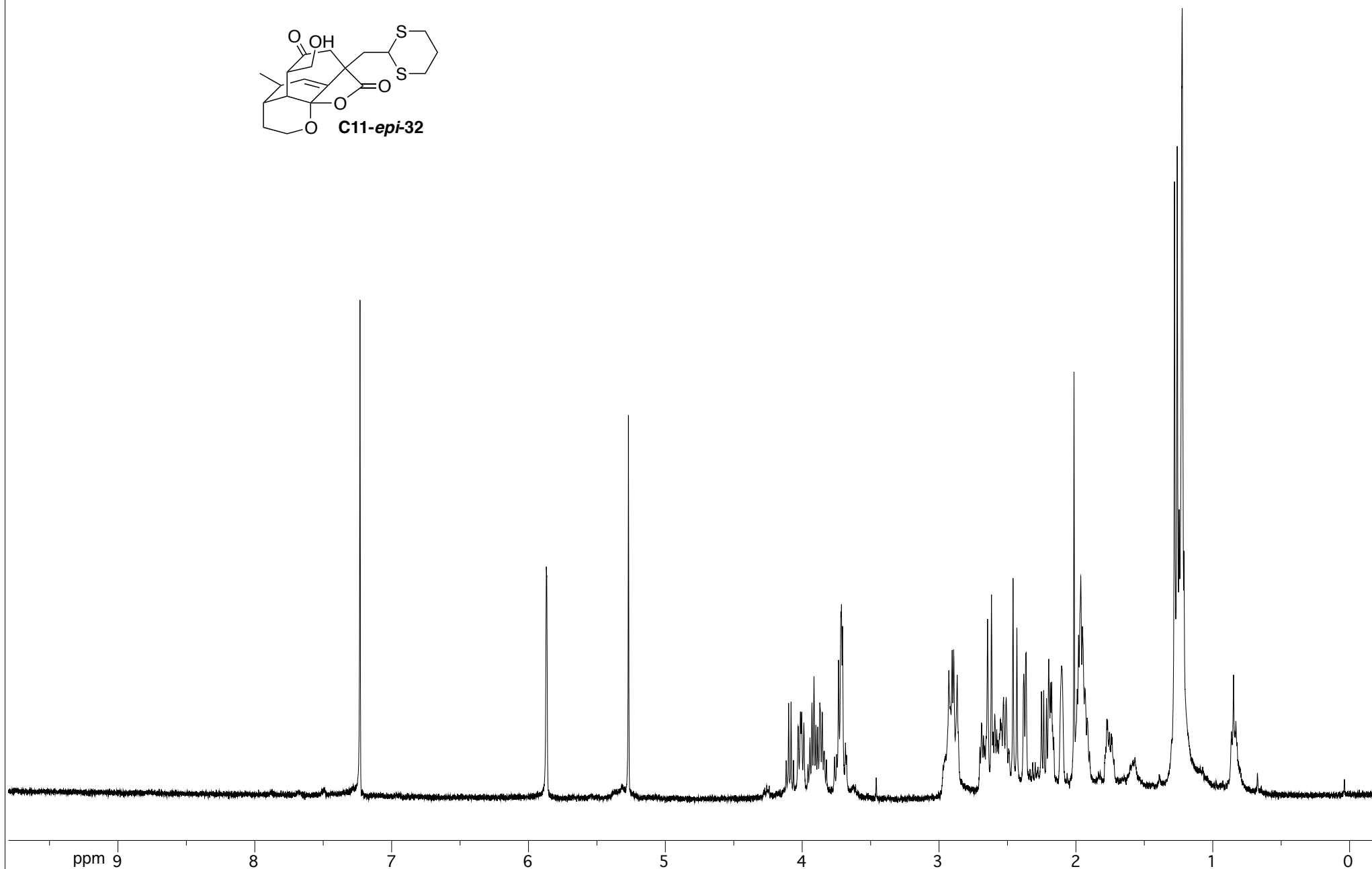
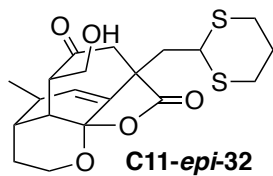
GKM-9-020_13Jan2010
32/GKM-9-020_13Jan2010/PROTON
July 18, 2012 2:06 PM



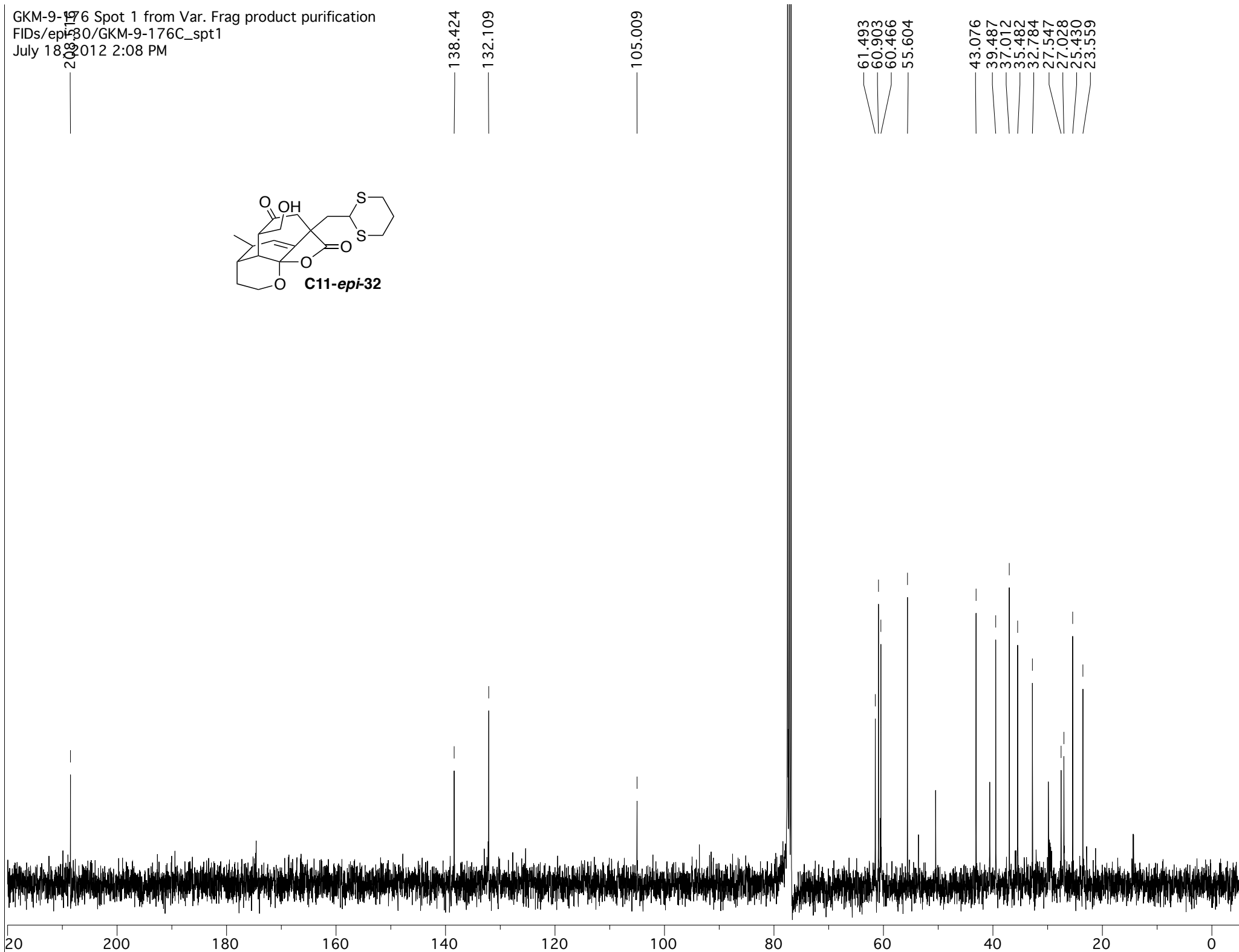
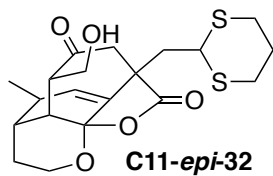
GKM-9-020 Fragmentation product
FIDs/32/GKM-9-020
July 18, 2012 2:06 PM



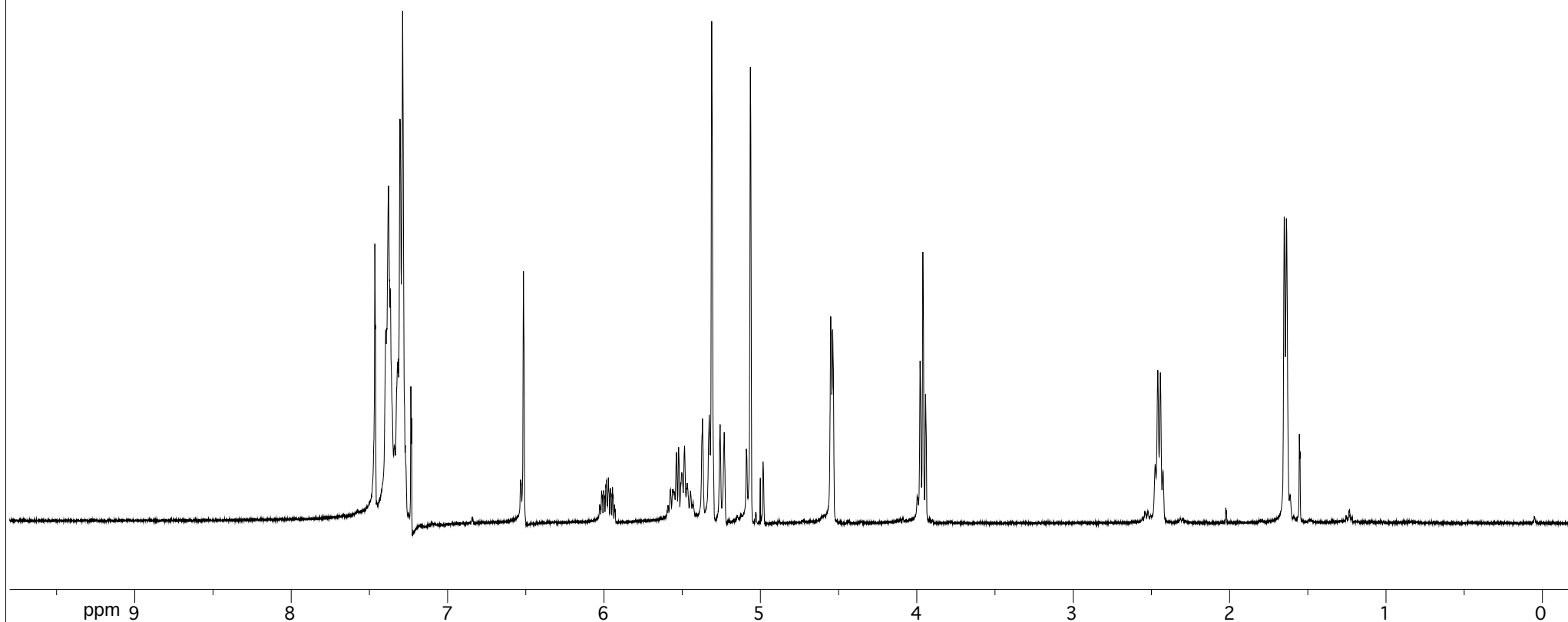
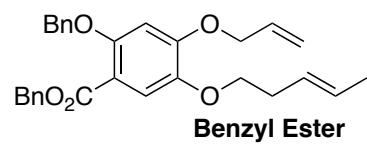
GKM-9-176_spt1_26Feb2010
epi-30/GKM-9-176_spt1_26Feb2010/PROTON
July 18, 2012 2:08 PM



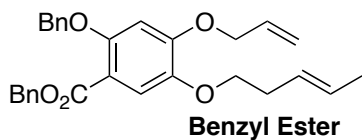
GKM-9-176 Spot 1 from Var. Frag product purification
FIDs/epi-30/GKM-9-176C_spt1
July 18 2012 2:08 PM



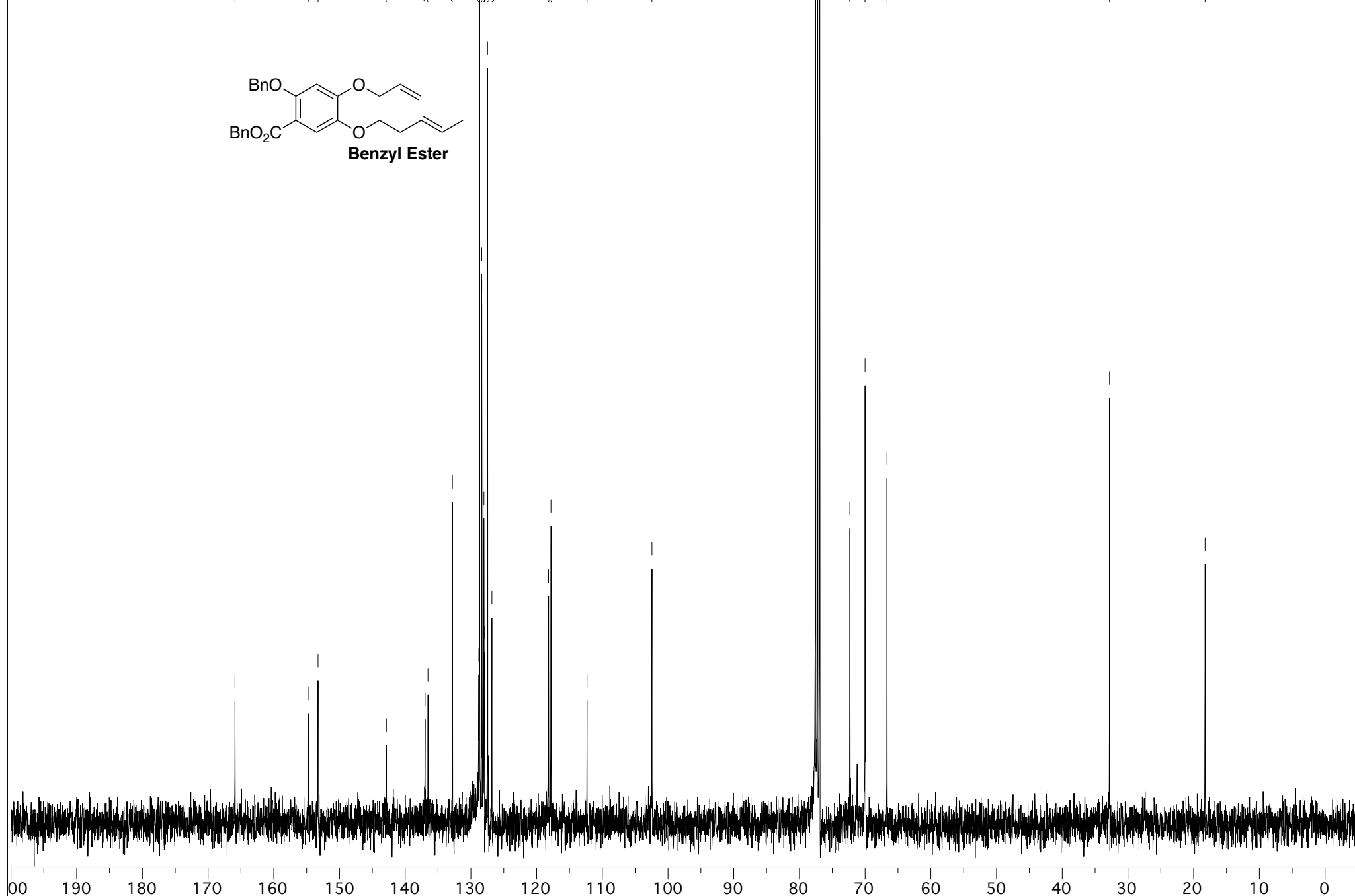
GKM-3-276 Bn Ester formation
FIDs/Benzyl Ester/GKM-3-276H_p
July 18, 2012 2:10 PM



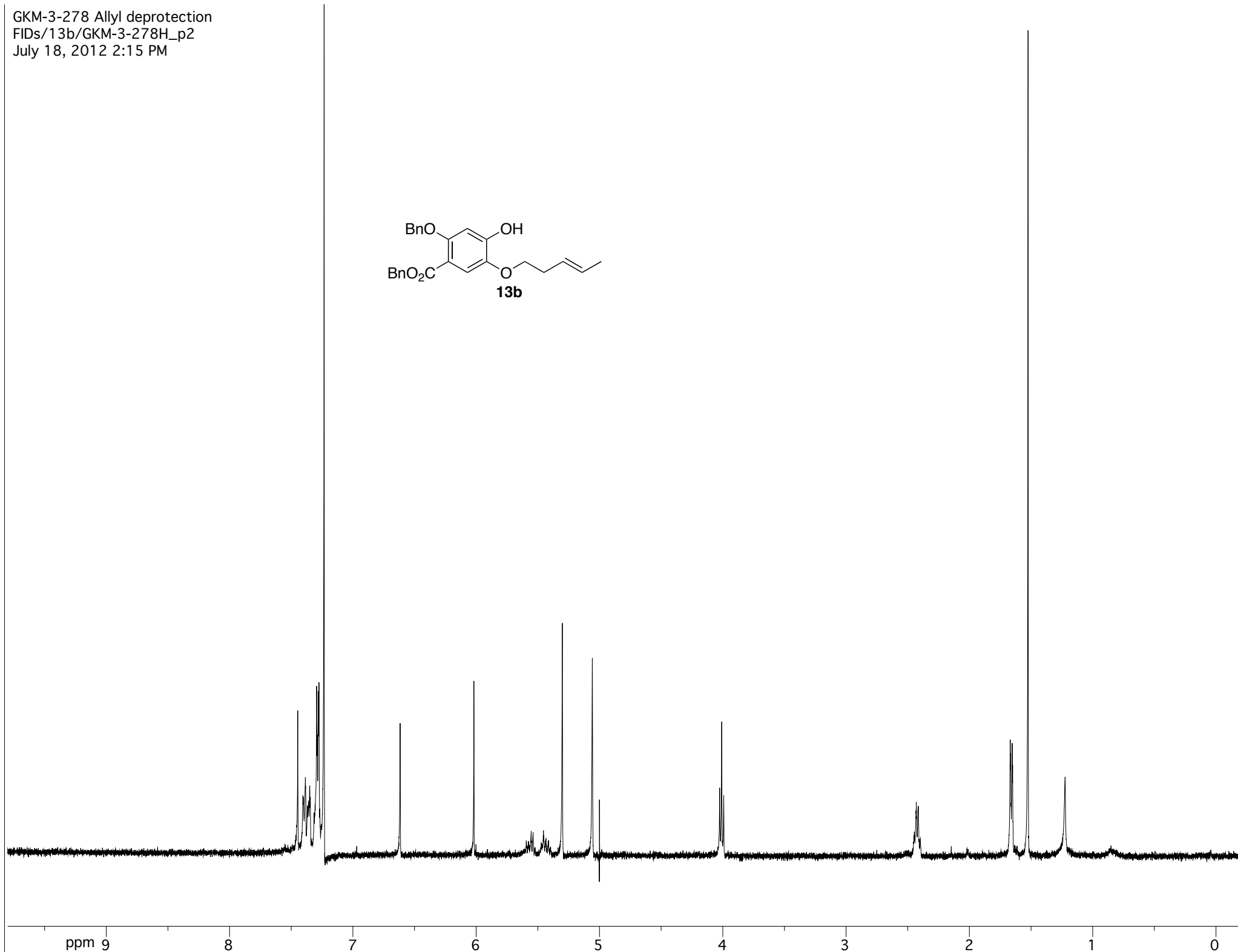
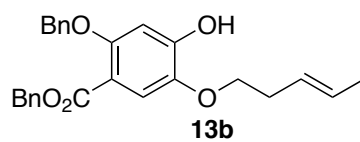
GKM-3-276 Bn Ester formation
FIDs/Benzyl Ester/GKM-3-276
July 18, 2012 2:10 PM



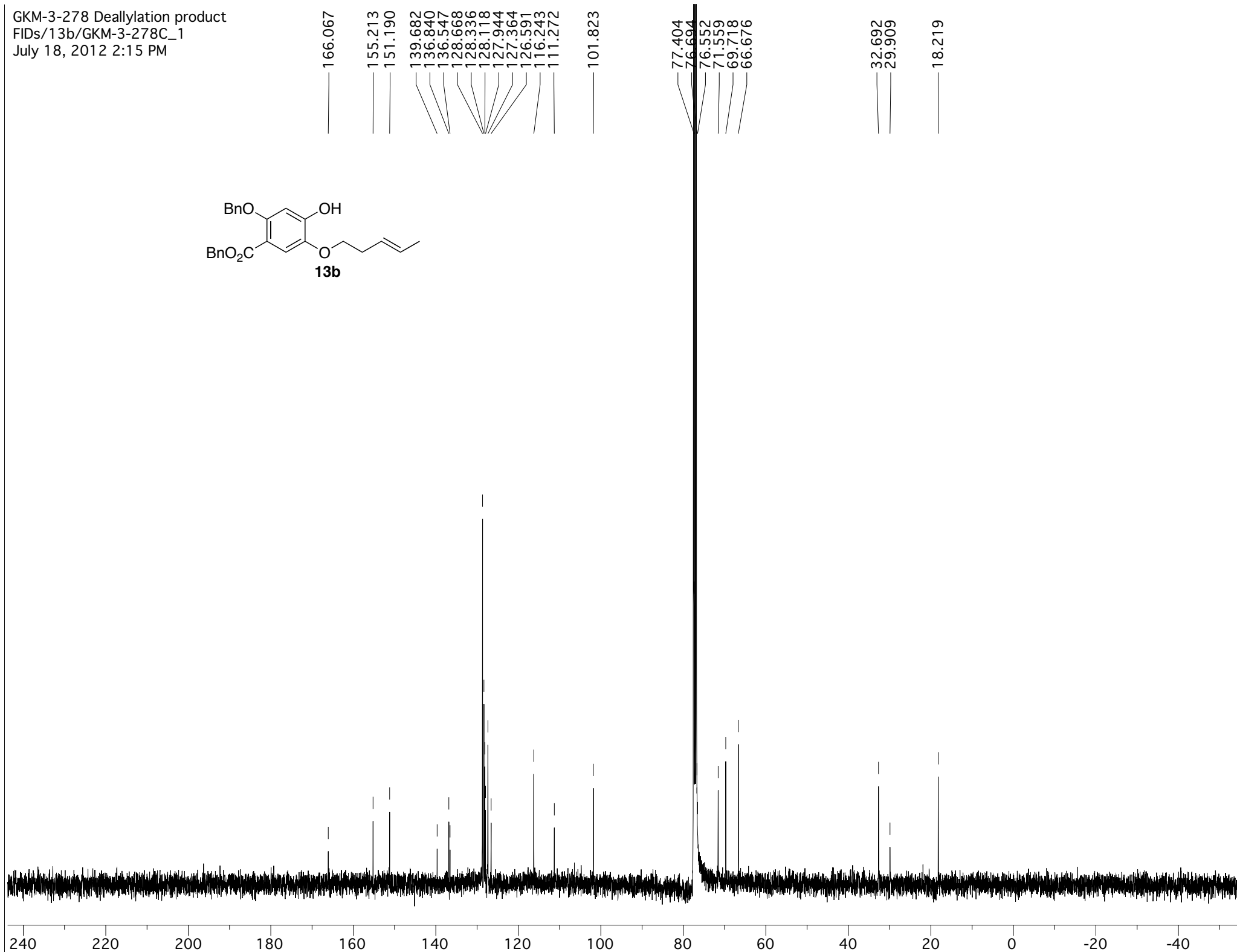
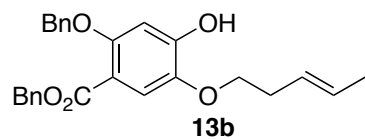
165.054
154.667
153.265
142.863
136.970
136.531
132.814
128.791
128.696
128.353
128.154
128.020
127.976
127.462
126.802
118.181
117.812
112.329
102.444
77.547
77.225
76.911
72.325
70.009
69.924
66.675
32.807
18.260



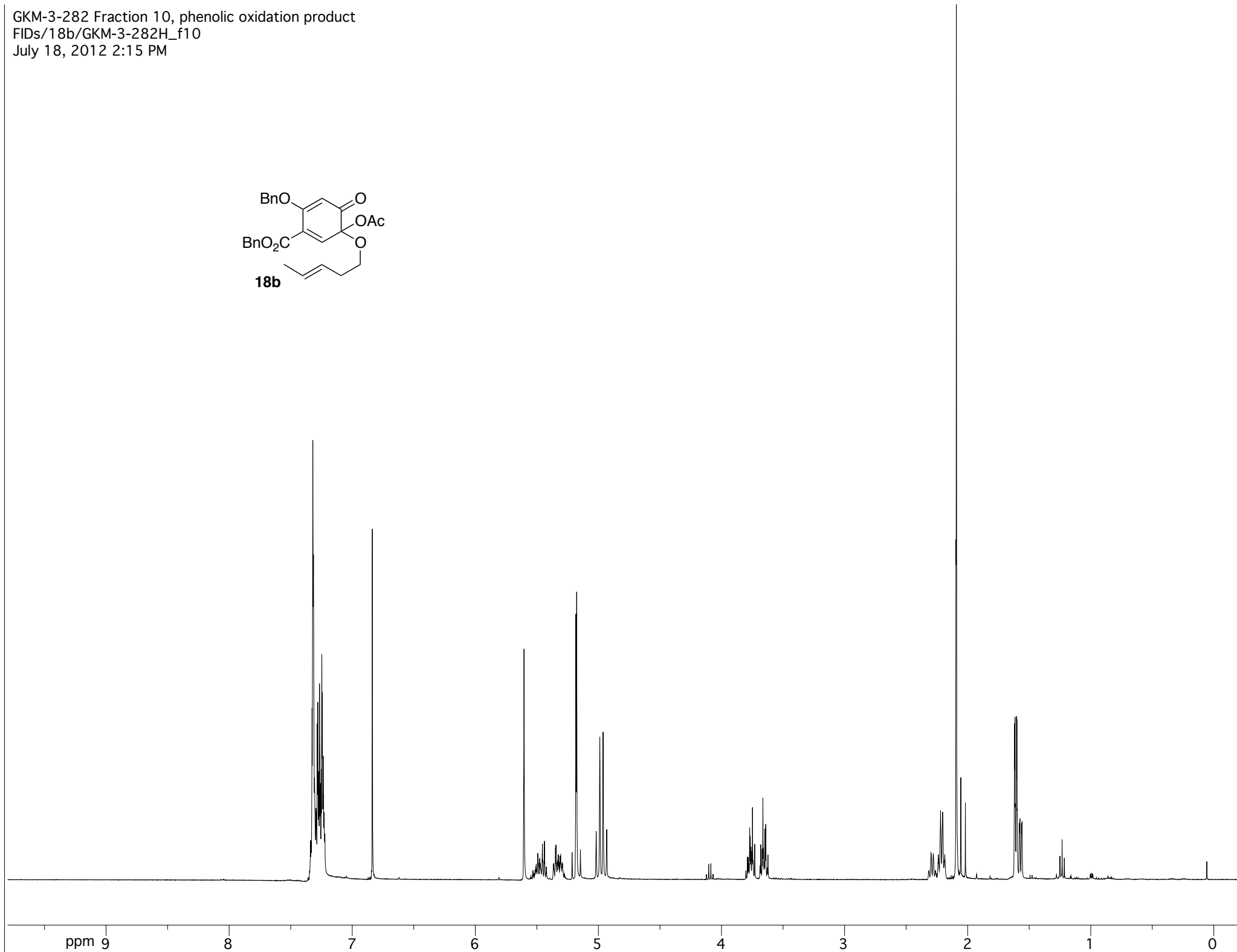
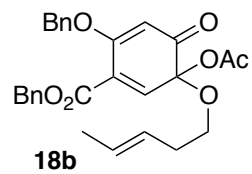
GKM-3-278 Allyl deprotection
FIDs/13b/GKM-3-278H_p2
July 18, 2012 2:15 PM



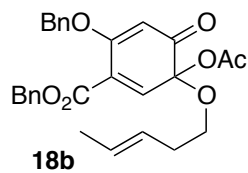
GKM-3-278 Deallylation product
FIDs/13b/GKM-3-278C_1
July 18, 2012 2:15 PM



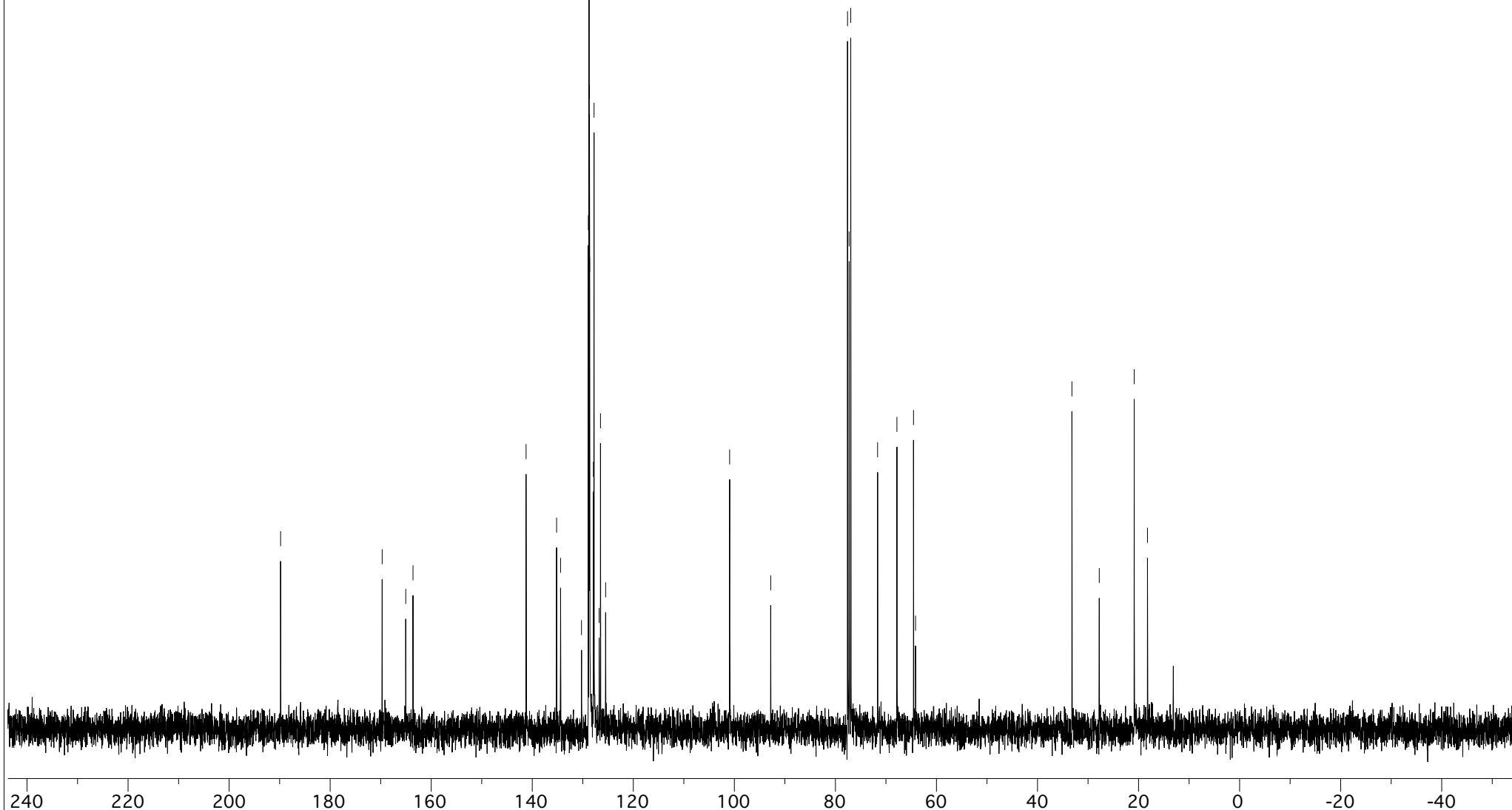
GKM-3-282 Fraction 10, phenolic oxidation product
FIDs/18b/GKM-3-282H_f10
July 18, 2012 2:15 PM



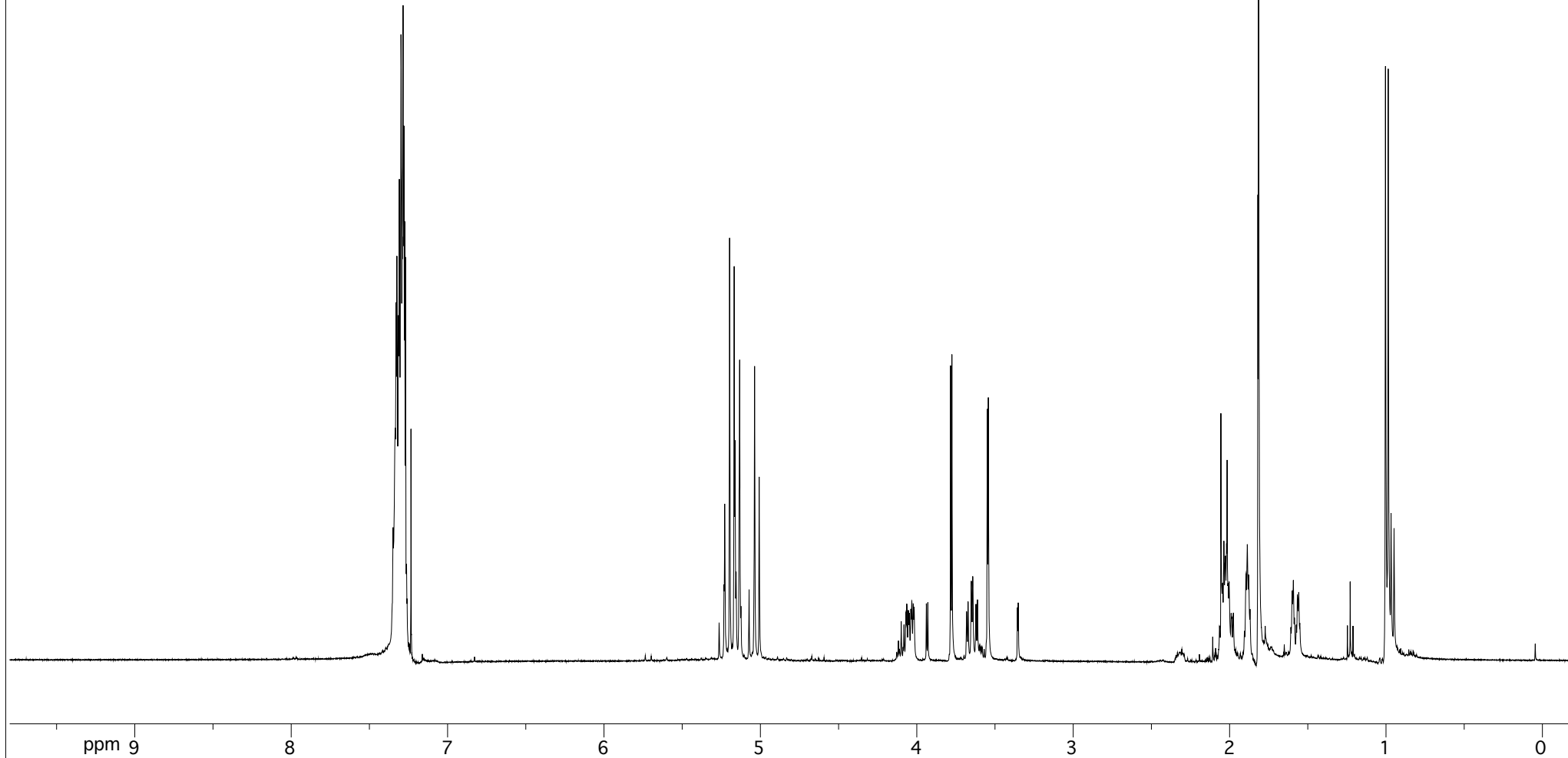
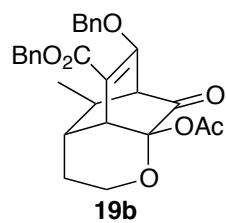
GKM-3-282 Fraction 10, phenolic oxidation product
FIDs/18b/GKM-3-282C_f10_1
July 18, 2012 2:15 PM



189.801
169.761
165.030
163.601
141.223
135.182
134.400
130.255
128.901
128.752
128.682
128.626
127.897
127.768
126.766
126.495
125.475
100.938
92.805
77.601
77.283
76.964
71.652
67.827
64.544
64.141
33.188
27.779
20.859
18.235



GKM-3-282 Fraction 16, Oxidation/DA product, acetate
FIDs/19b/GKM-3-282H_f16
July 18, 2012 2:14 PM



GKM-3-282 Fraction 16, Oxidation/DA product, ac 0.081
 FIDs/19b/GKM-3-282C_F16
 July 18, 2012 2:14 PM

168.116
 164.356
 163.400
 136.435
 136.038
 128.859
 128.773
 128.681
 128.569
 128.497
 128.371
 128.298
 128.196
 127.770
 127.553

105.116

93.846

77.579
 77.257
 76.943
 72.485

66.259

62.104

57.709

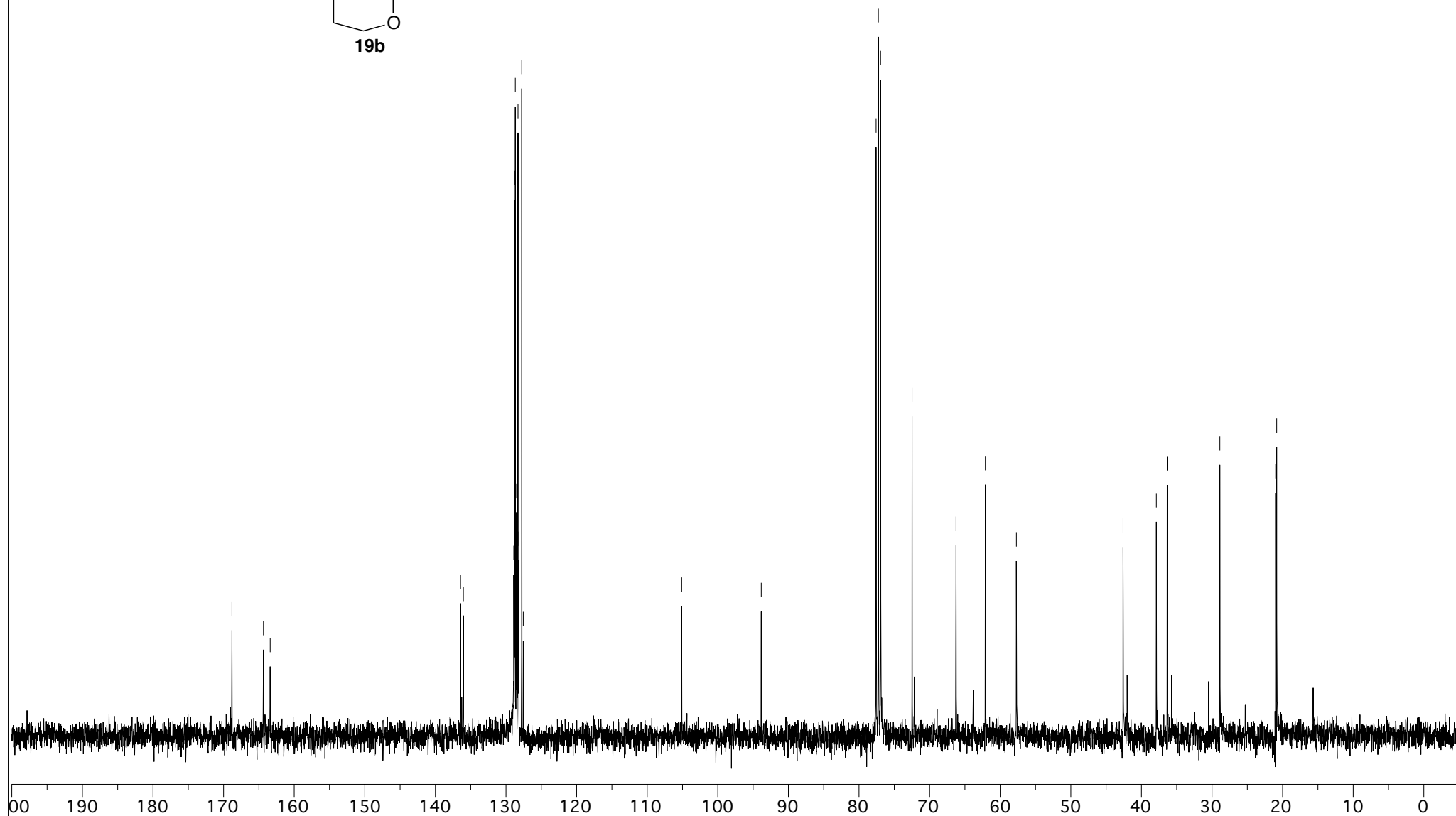
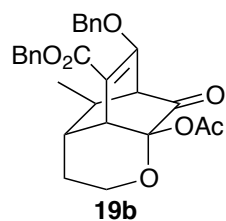
42.598

37.887

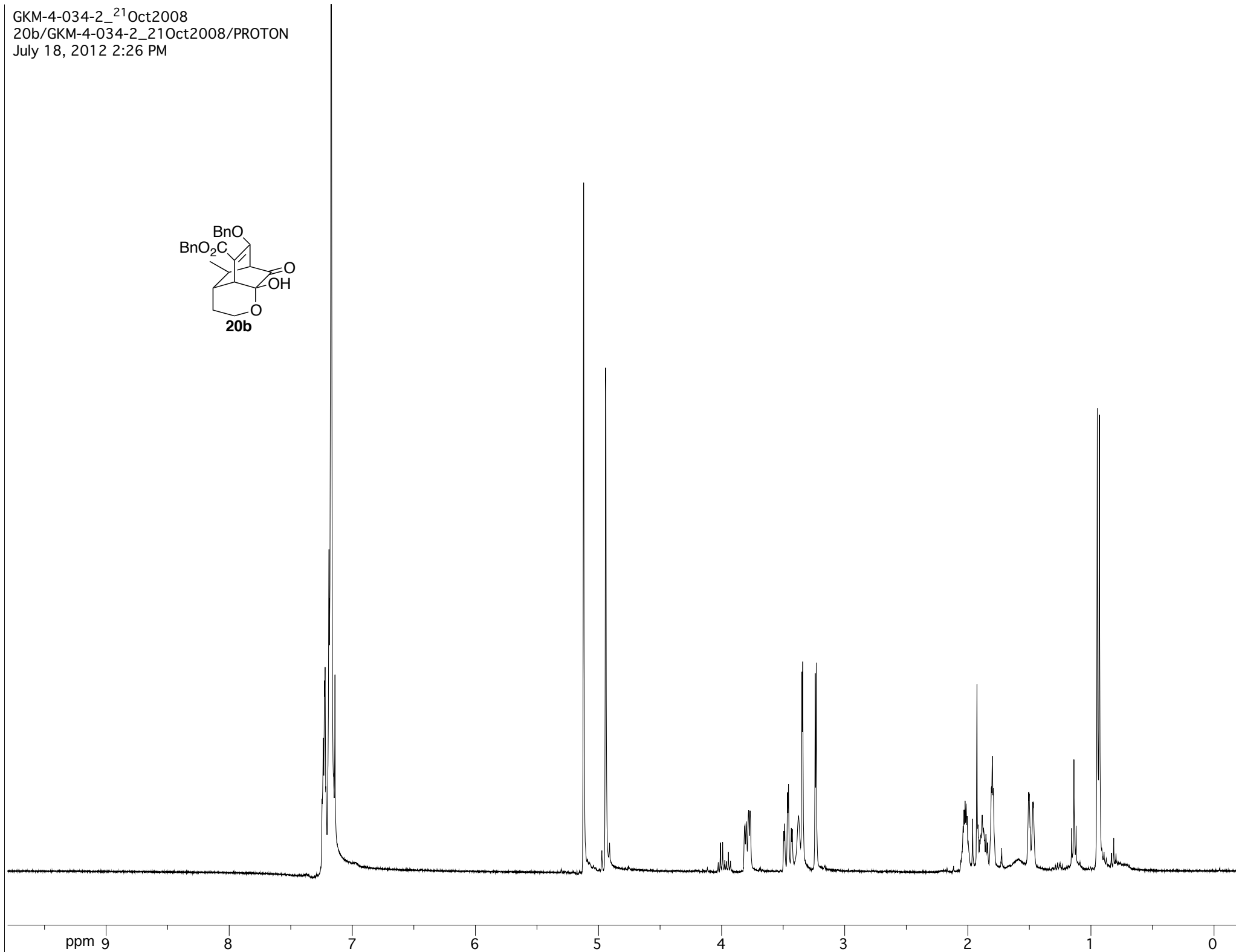
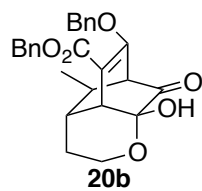
36.356

28.890

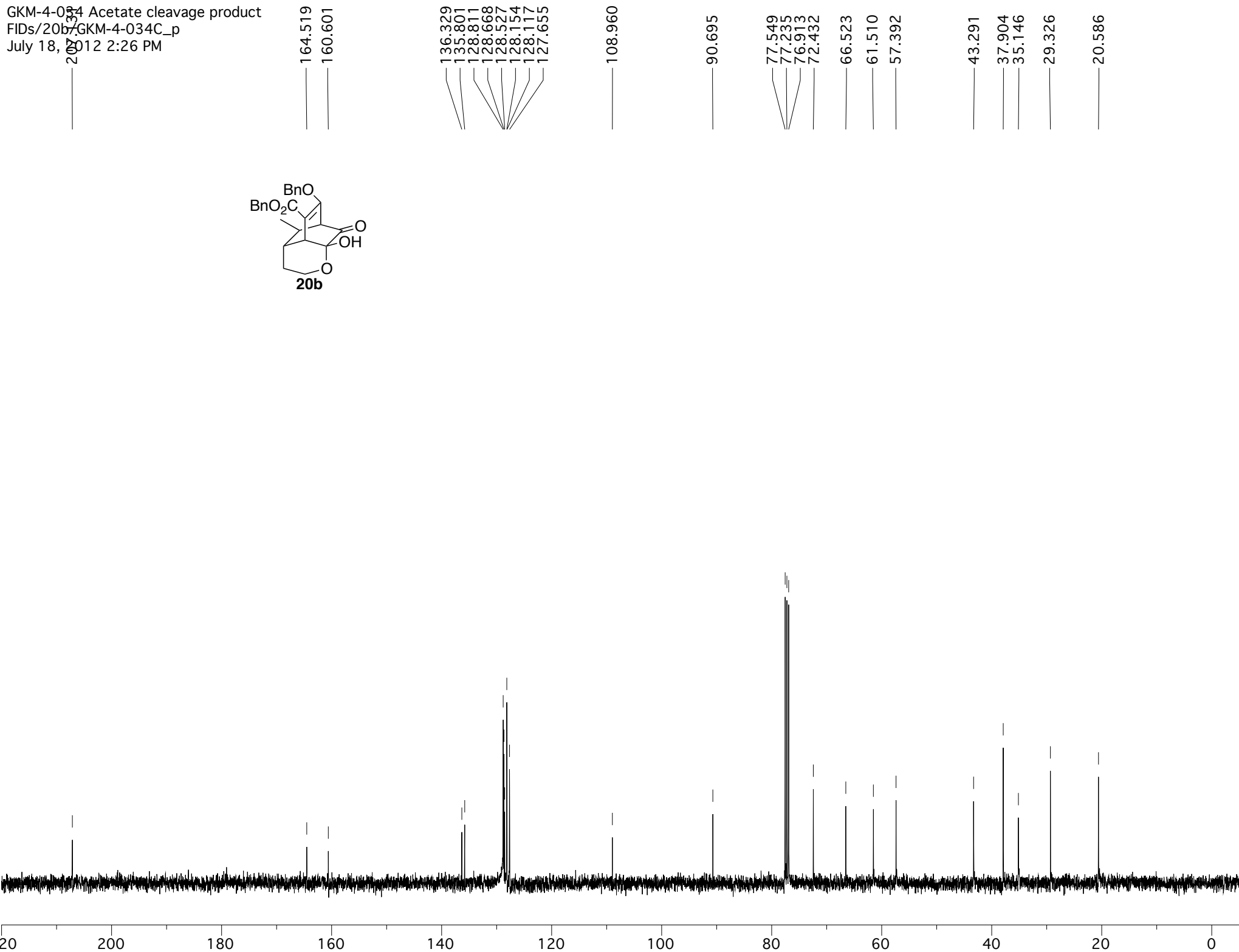
20.986
 20.841



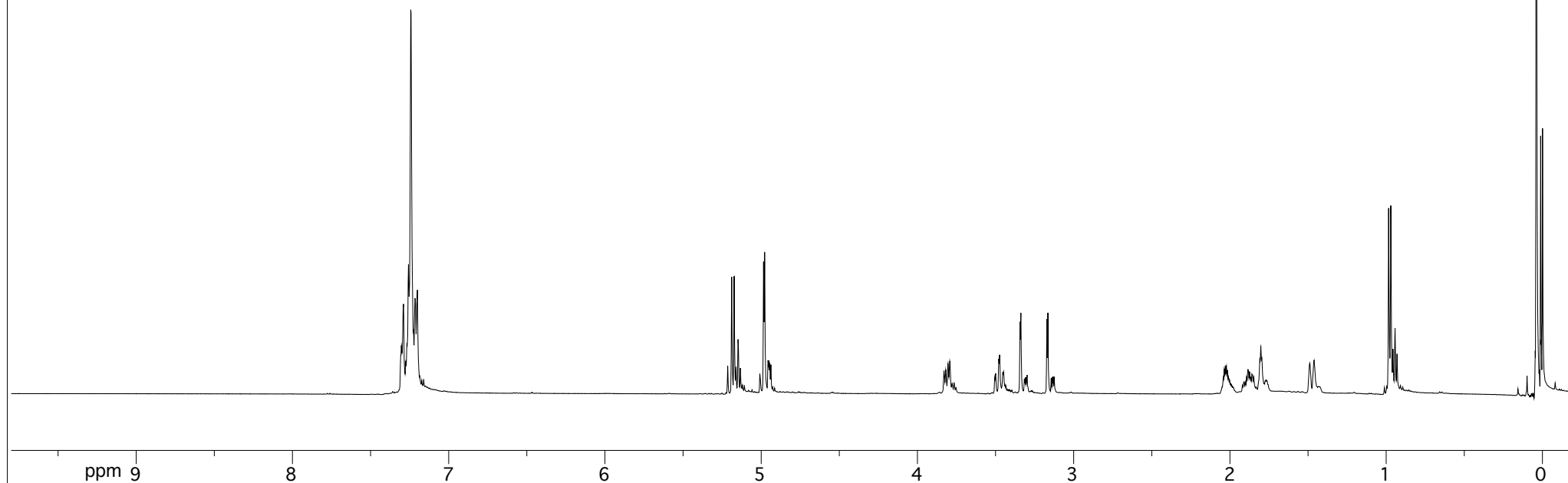
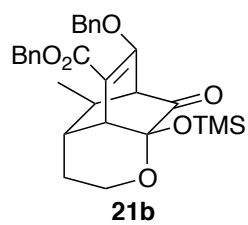
GKM-4-034-2_21Oct2008
20b/GKM-4-034-2_21Oct2008/PROTON
July 18, 2012 2:26 PM



GKM-4-034 Acetate cleavage product
FIDs/20b7/GKM-4-034C_p
July 18, 2012 2:26 PM



OTMS Stuff
21b/GKM-4-182_20100813_01/PROTON_01
July 18, 2012 2:26 PM



OTMS Stuff
21b/GKM-4-182_201100813_01/CARBON_01
July 18, 2012 3:12 PM

204.932

162.856

159.162

134.844

134.340

127.045

126.924

126.675

126.480

126.361

125.864

107.125

90.816

75.784

75.530

75.276

70.547

64.444

59.725

56.238

44.131

36.204

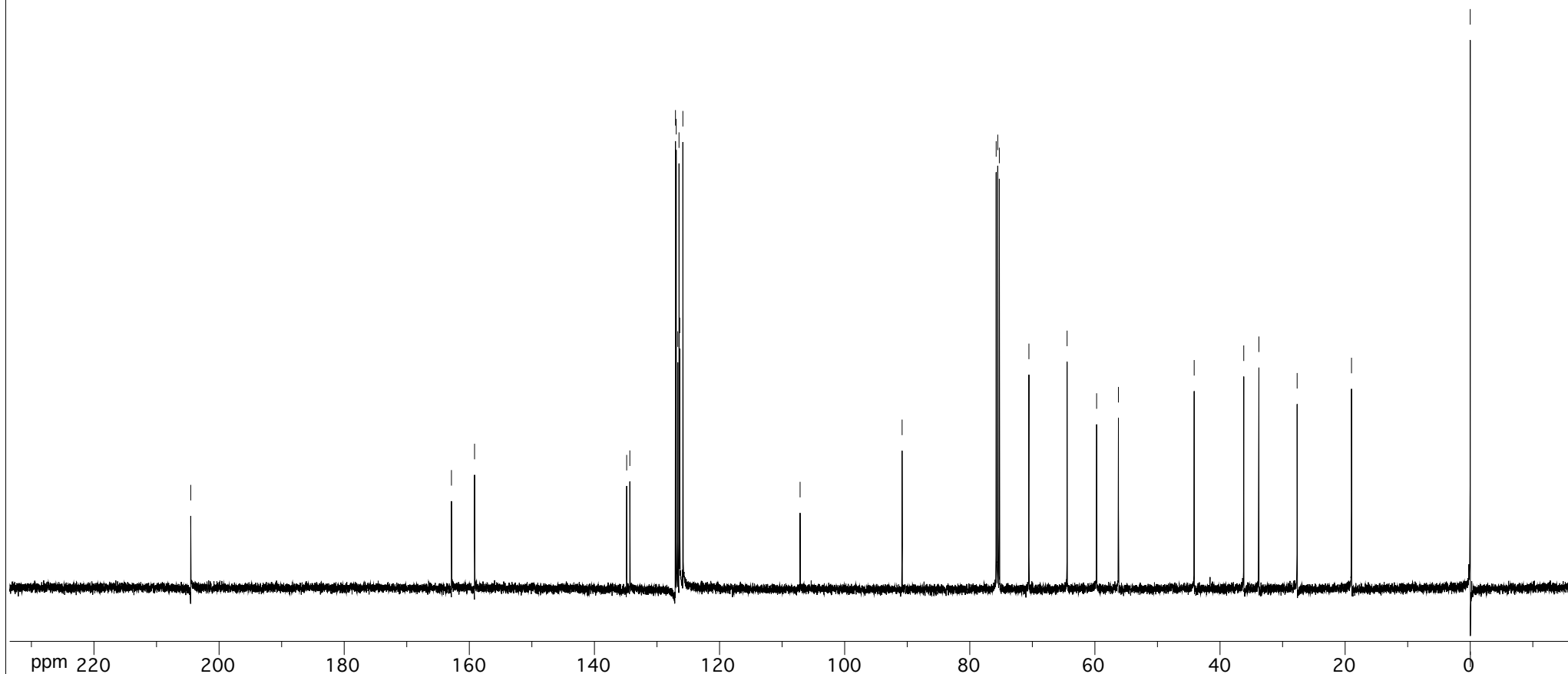
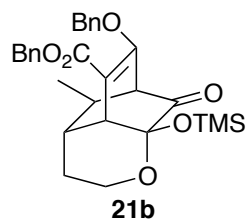
33.783

27.673

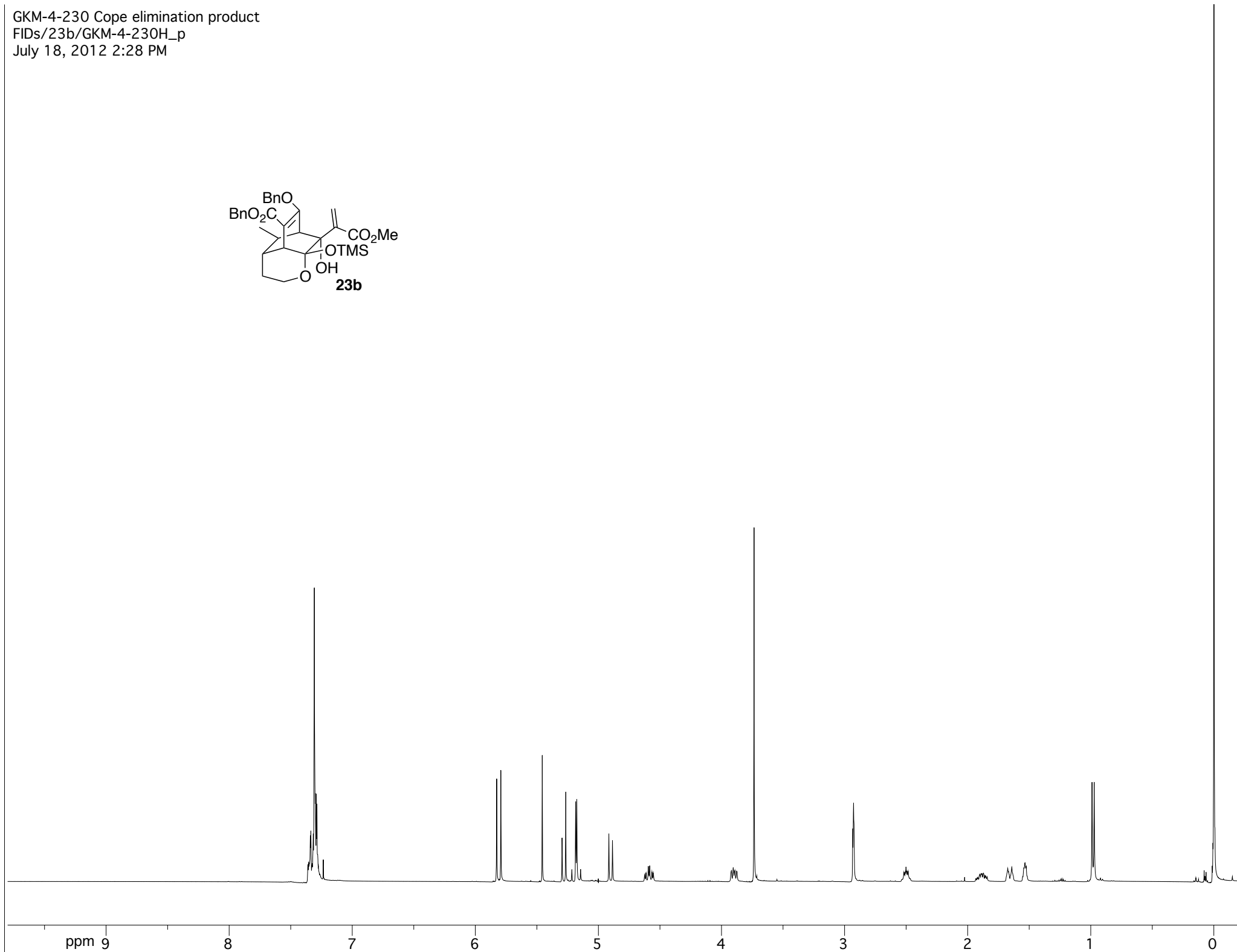
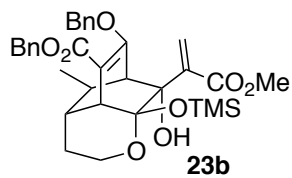
18.980

0.004

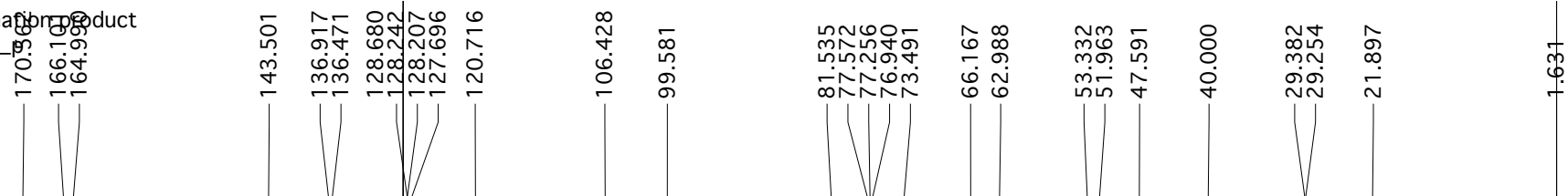
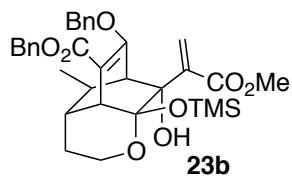
-0.018



GKM-4-230 Cope elimination product
FIDs/23b/GKM-4-230H_p
July 18, 2012 2:28 PM

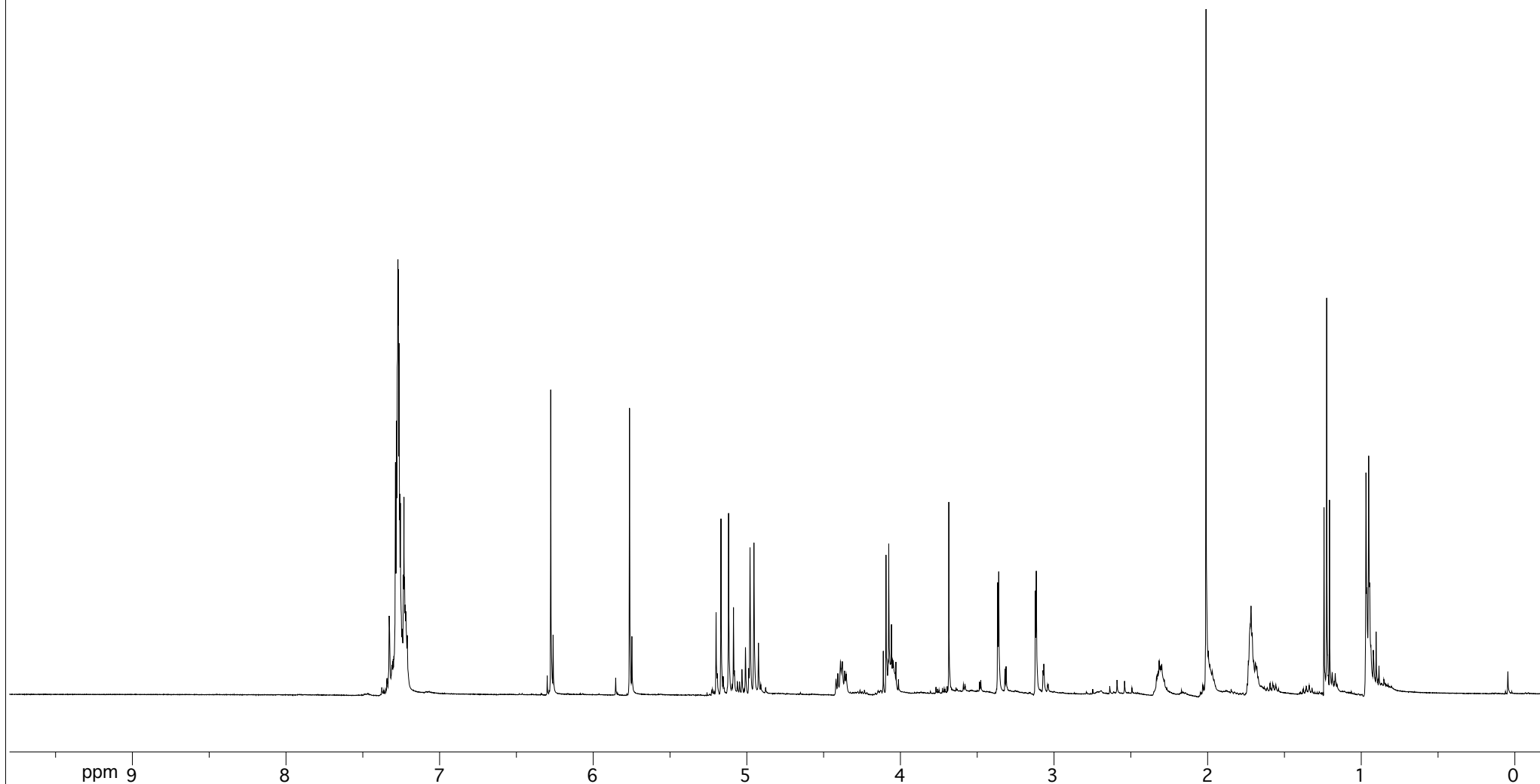
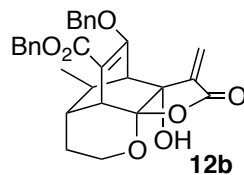


GKM-4-230 Cope elimination product
FIDs/23b/GKM-4-230C_1
July 18, 2012 2:28 PM

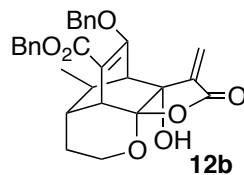


00 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0

GKM-4-038 Lactone closure product
FIDs/12b/GKM-4-038H
July 18, 2012 2:38 PM



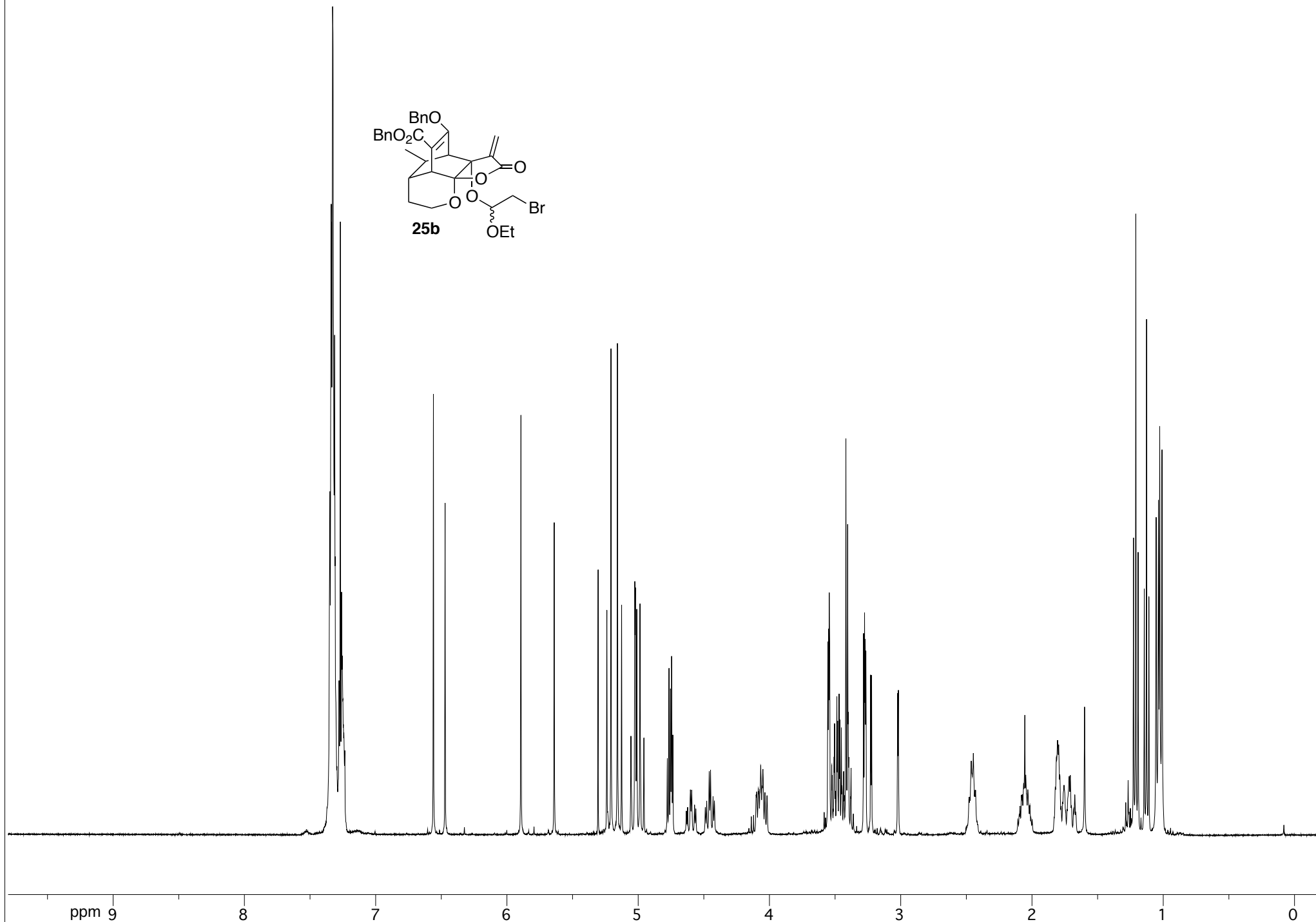
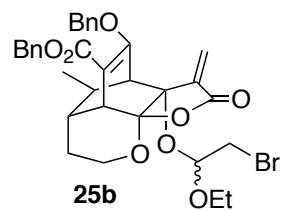
GKM-4-038 Lactone closure product
FIDs/12b/GKM-4-038C_p
July 18, 2012 2:38 PM



166.344
165.518
164.244
141.245
136.363
136.110
128.773
128.662
128.463
127.697
127.607
127.516
126.586
104.609
103.443
77.564
77.242
76.927
76.244
72.521
66.385
64.166
52.253
42.598
39.052
31.694
28.964
21.296

00 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0

GKM-4-214 Bromoacetal product
FIDs/25b/GKM-4-214H_p
July 18, 2012 2:38 PM



GKM-4-214 Bromoacetal product
FIDs/25b/GKM-4-214C_p
July 18, 2012 2:38 PM

166.100
165.135
164.975

136.351
136.307
136.147
135.981
128.954
128.777
128.716
128.682
128.517
128.444
128.099
127.789
127.460
126.495

105.572
103.731

97.958
97.085

82.083
81.514

77.429

73.667
72.887

66.438
63.691

63.473
62.705

61.520

53.775
53.164

45.242
44.371

40.361
39.789

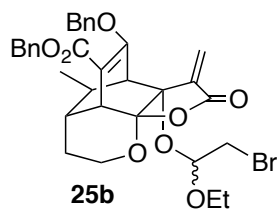
32.594
32.404

29.964
29.327

28.817
28.739

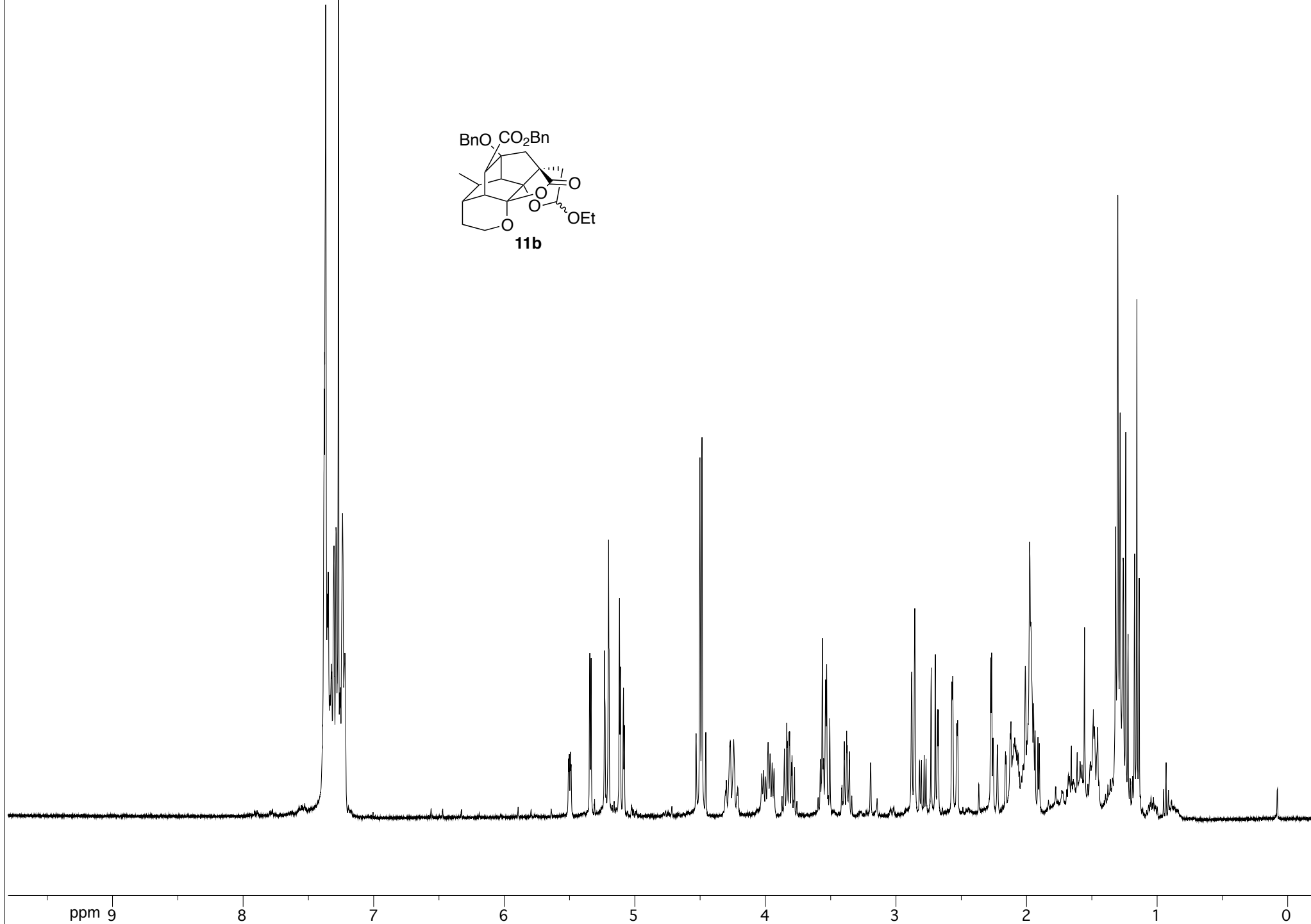
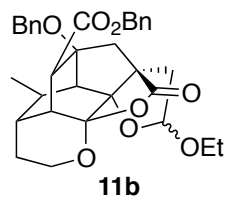
21.424

15.269
15.102

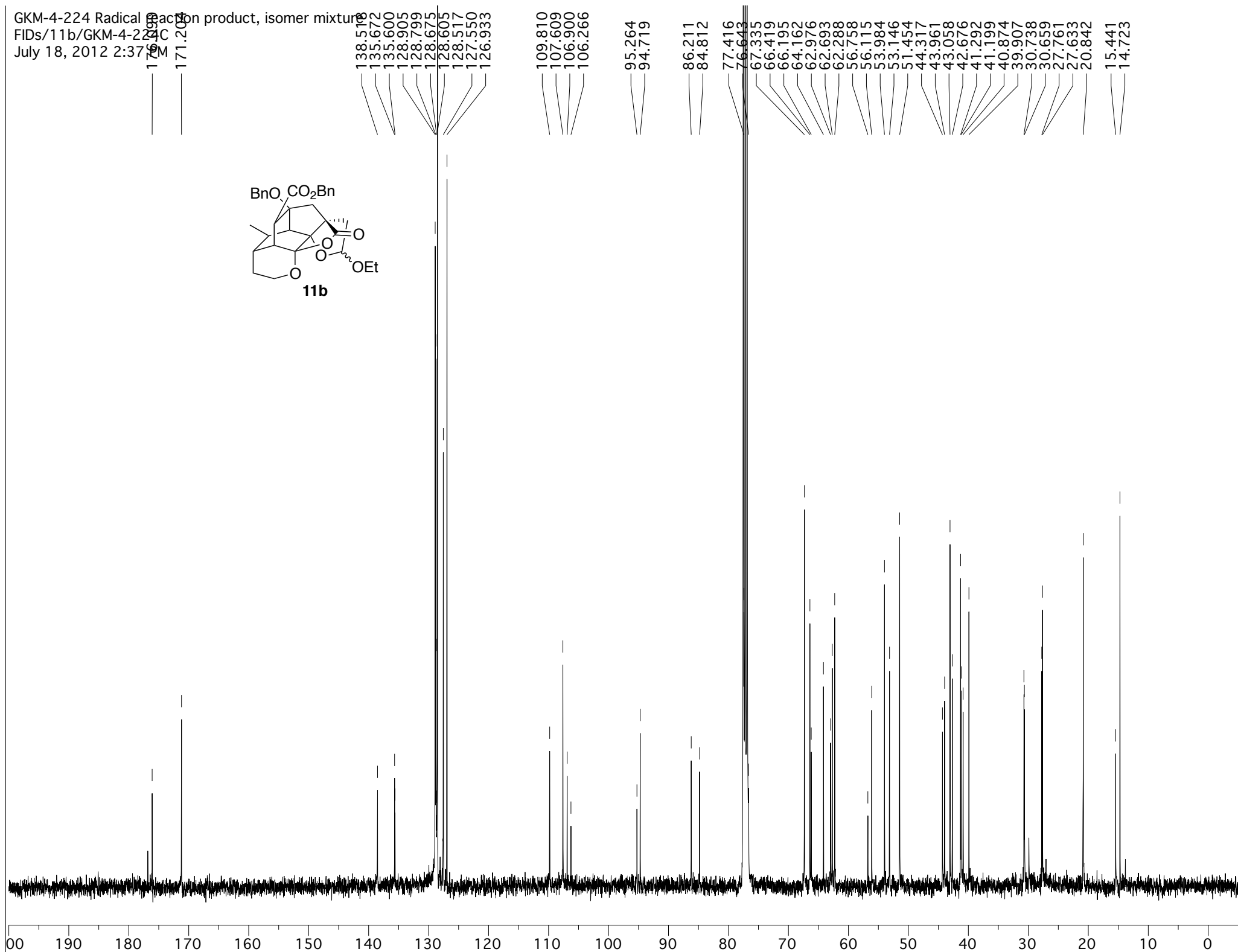
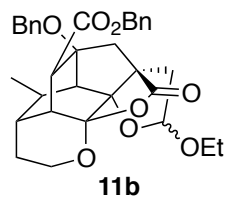


00 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0

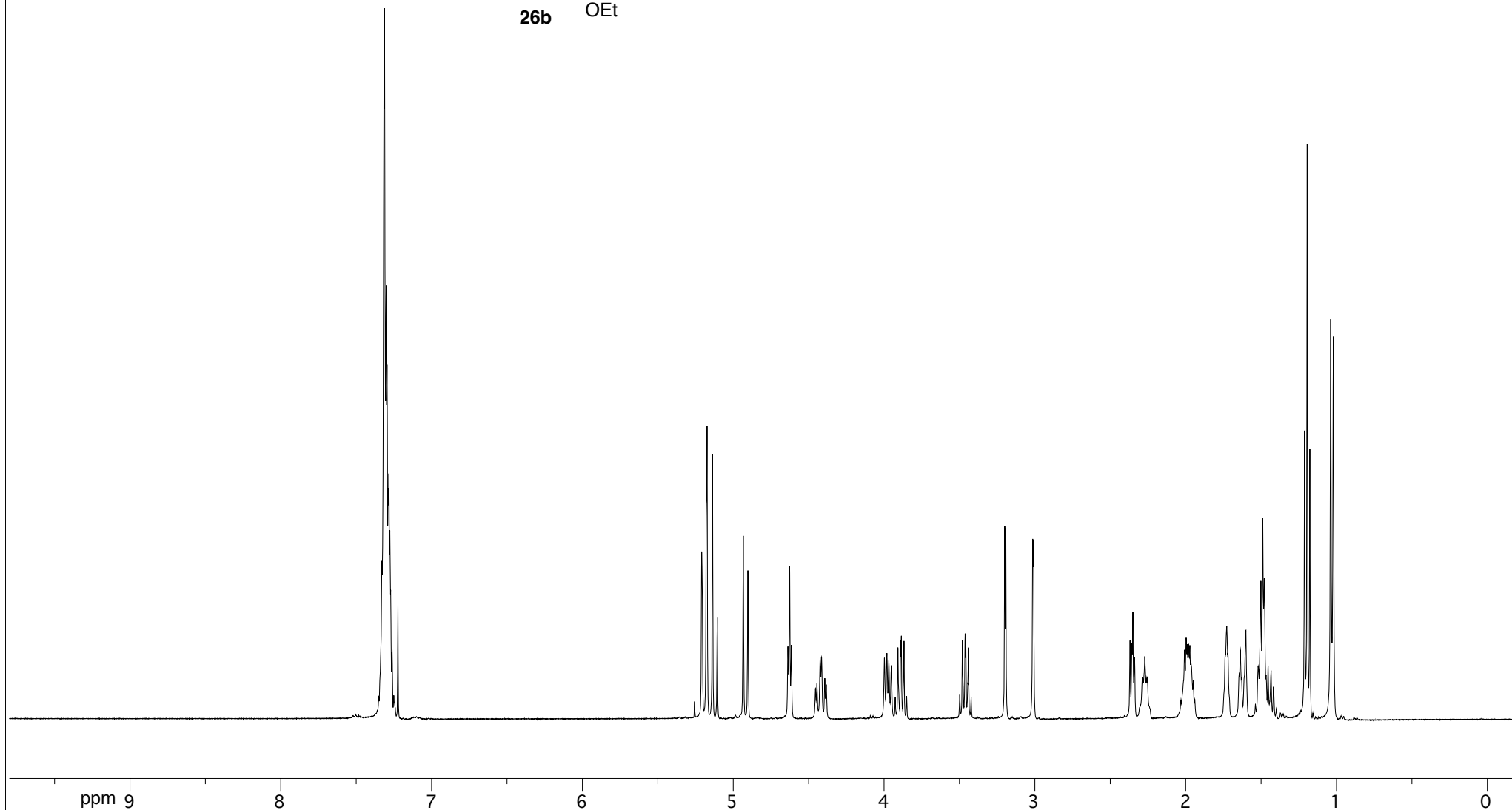
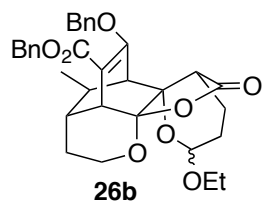
GKM-4-224 Radical reaction product, isomer mixture
FIDs/11b/GKM-4-224H_p
July 18, 2012 2:37 PM



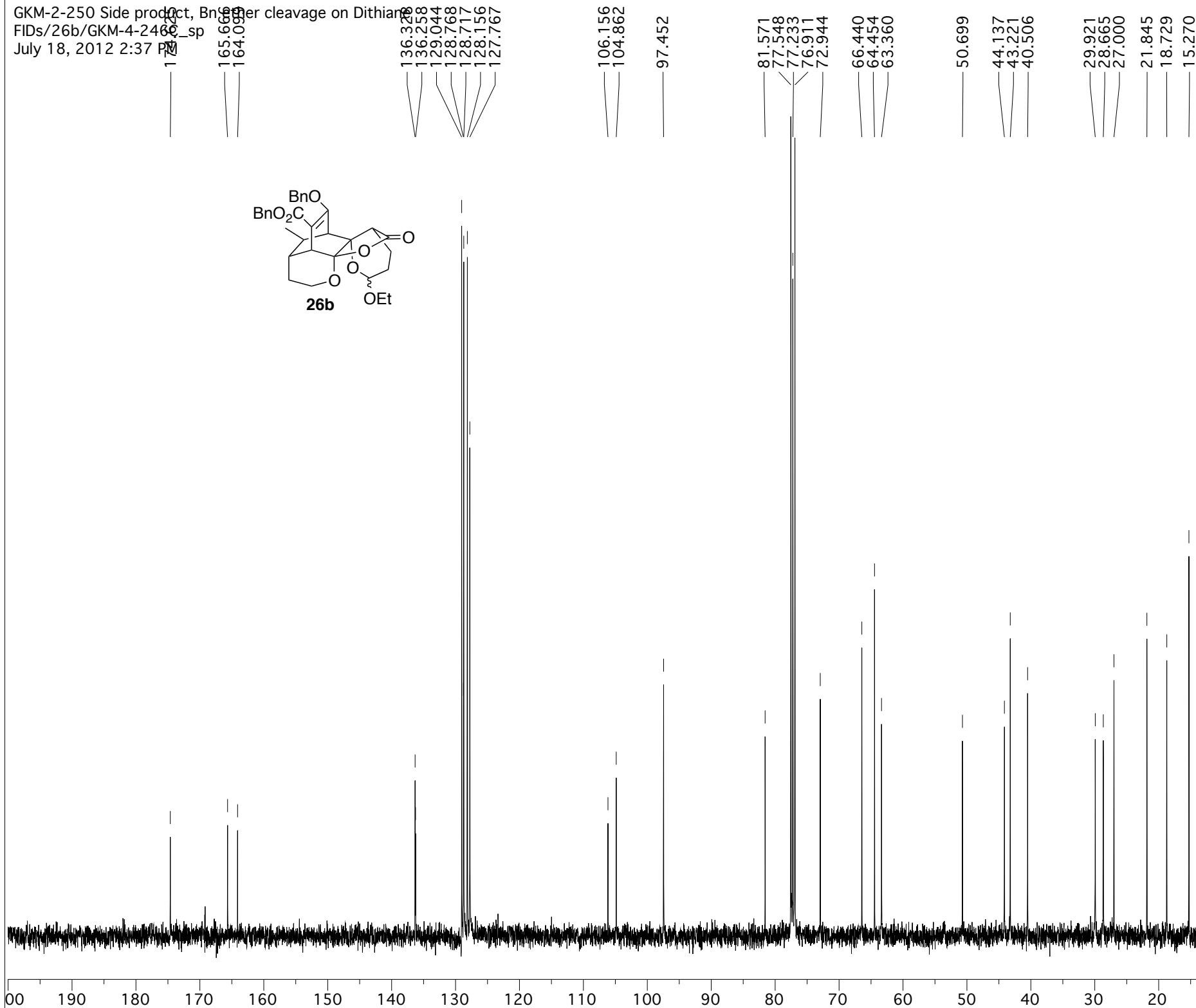
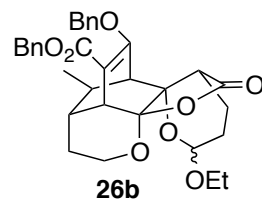
GKM-4-224 Radical reaction product, isomer mixture
FIDs/11b/GKM-4-224C
July 18, 2012 2:37:59



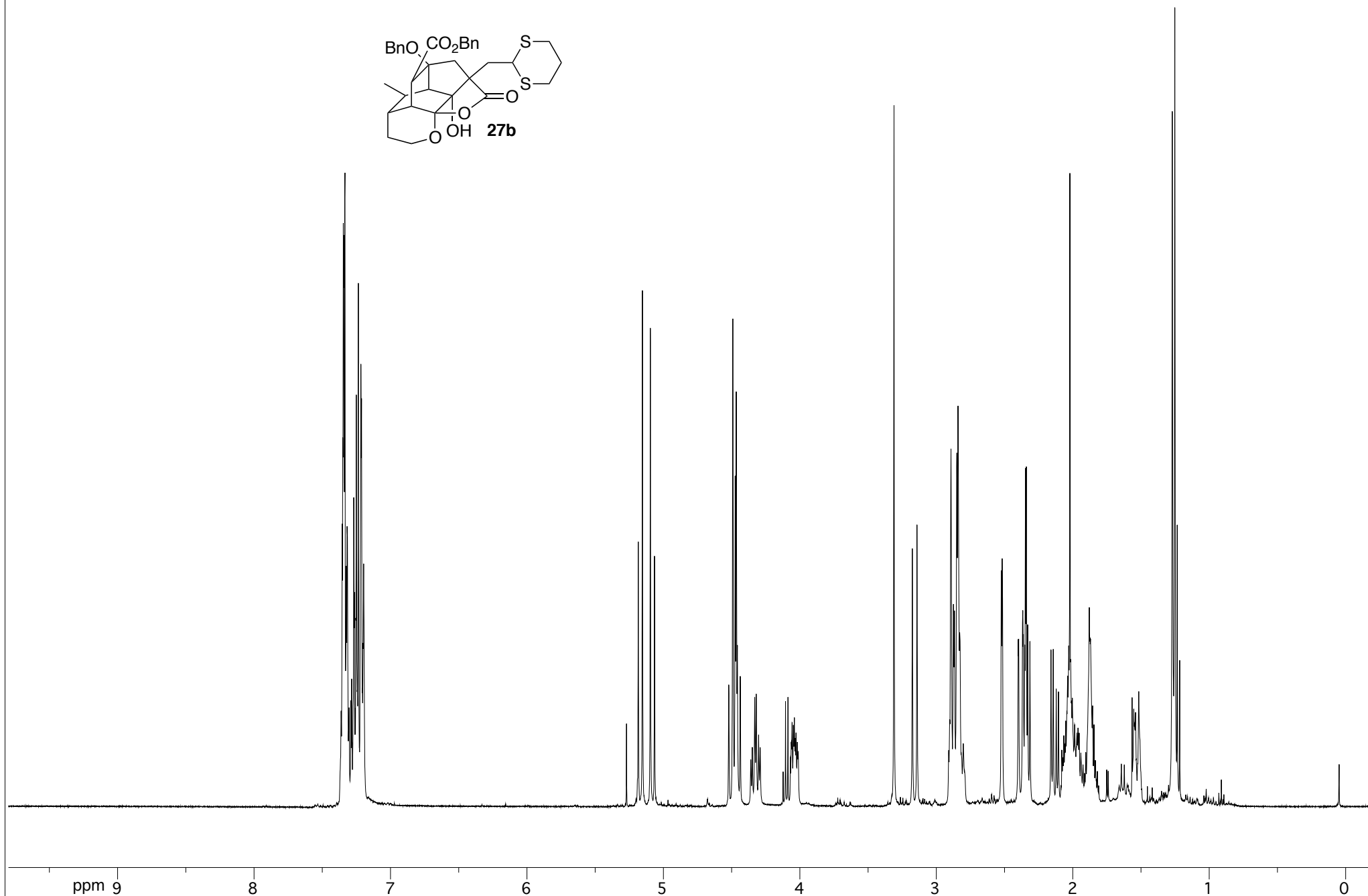
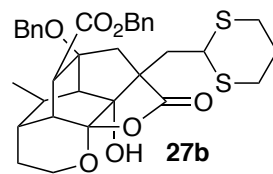
GKM-2-250 Side product, Bn ether cleavage on Dithiane
FIDs/26b/GKM-4-246H_sp
July 18, 2012 2:37 PM



GKM-2-250 Side product, BnO-ether cleavage on Dithiane
FIDs/26b/GKM-4-2466.sp
July 18, 2012 2:37 PM



GKM-4-226 Dithiane product
FIDs/27b/GKM-4-226H_p
July 18, 2012 2:45 PM



GKM-4-226 dithiane product
FIDs/27b/GKM-4-226C_1
July 18, 2012 2:45 PM

171.93
171.93

138.642
135.605
128.918
128.808
128.589
128.499
127.516
127.059

105.554

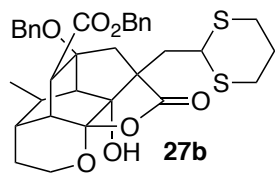
83.994
80.898
77.564
77.241
76.928

67.379
66.108
63.873

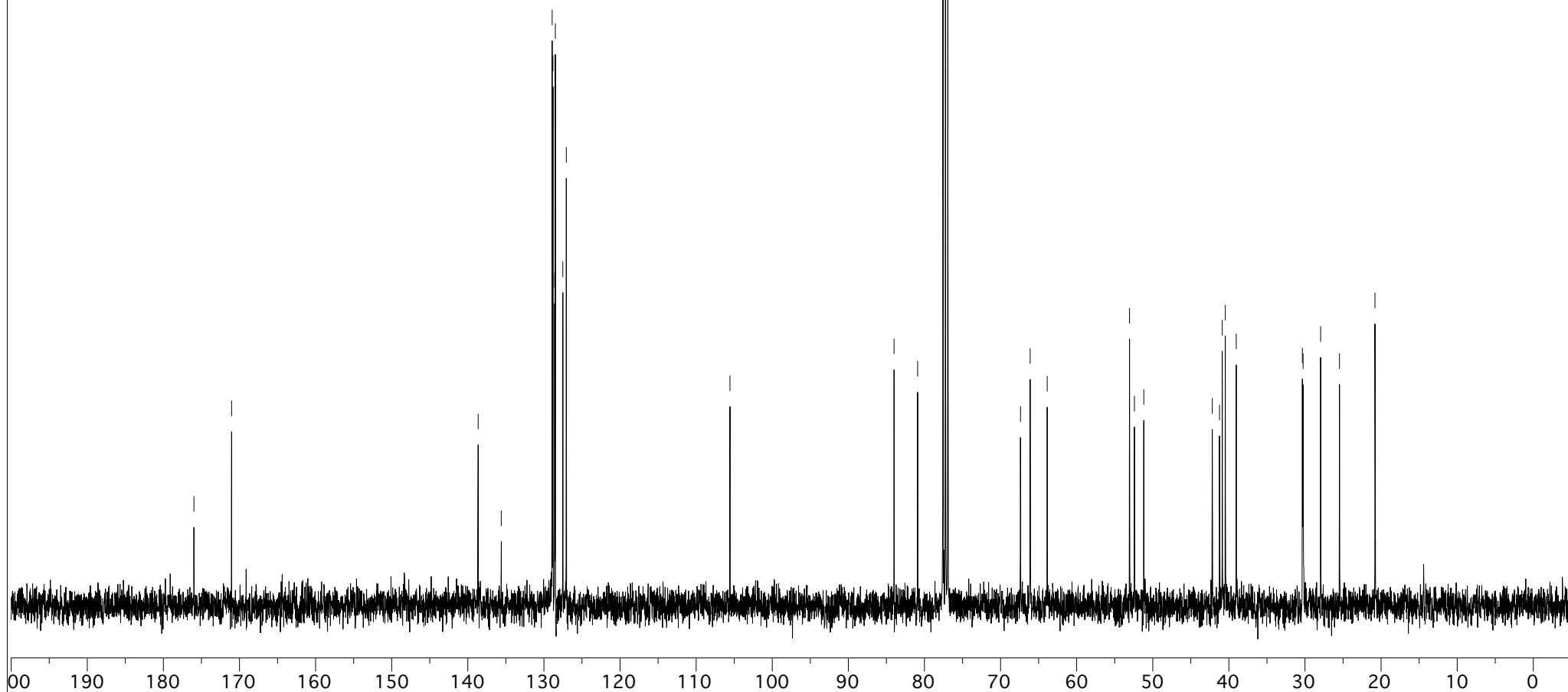
53.054
52.409
51.179

42.179
41.232
40.872
40.471
39.034

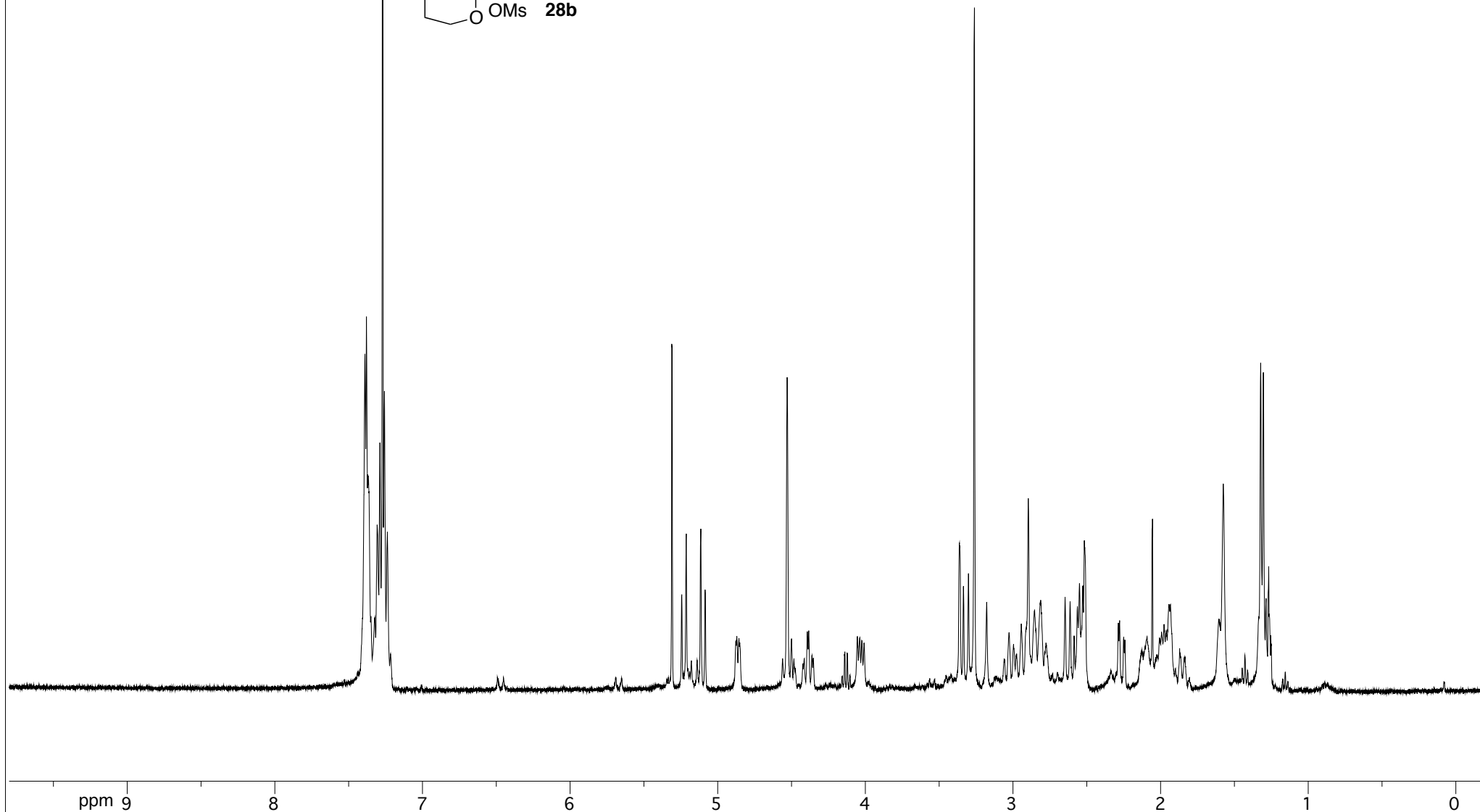
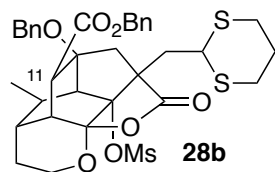
30.346
30.249
27.943
25.468
20.804



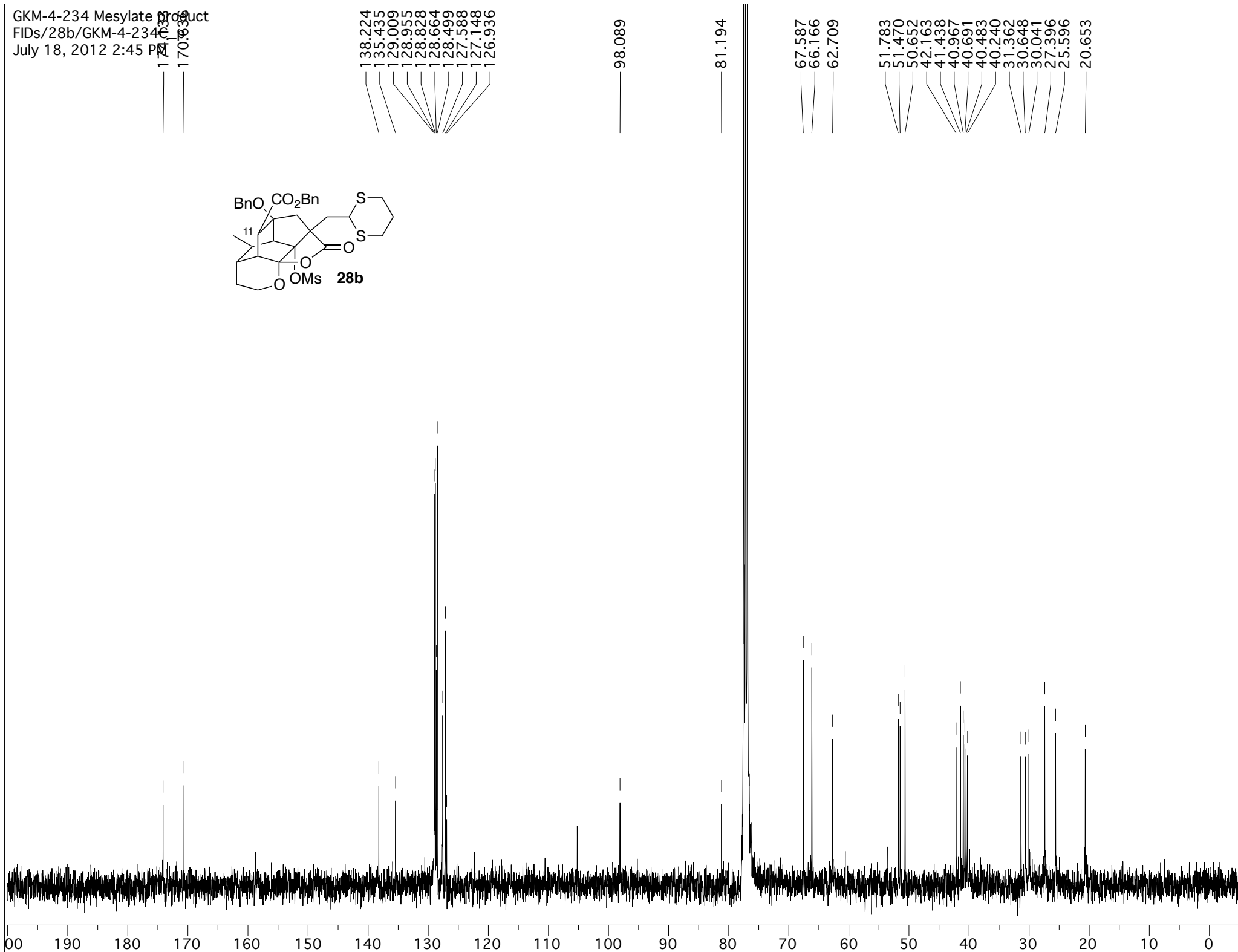
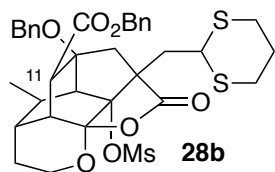
27b



GKM-4-234 Mesylate product
FIDs/28b/GKM-4-234H_p
July 18, 2012 2:44 PM



GKM-4-234 Mesylate product
FIDs/28b/GKM-4-234C
July 18, 2012 2:45 PM



July 18, 2012 2:44 PM

