The Direct Catalytic Enantioselective Vinylogous Aldol Reaction of α -Branched Enals with Isatins

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Supporting Information

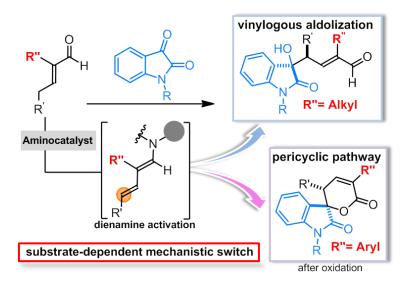


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A. General Information

The ¹H and ¹³C NMR spectra were recorded at 400 MHz and 500 MHz for ¹H or at 100 MHz and 125 MHz for ¹³C, respectively. The chemical shifts (δ) for ¹H and ¹³C are given in ppm relative to residual signals of the solvents (CHCl₃ @ 7.26 ppm ¹H NMR, 77.16 ppm ¹³C NMR). Coupling constants are given in Hz. When necessary, ¹H and ¹³C signals were assigned by means of g-COSY, g-HSQC and g-HMBC 2D-NMR sequences. The following abbreviations are used to indicate the multiplicity: s, singlet; d, doublet; t, triplet; q, quartet; qn, quintet; m, multiplet; bs, broad signal.

High-resolution mass spectra (HRMS) were obtained from the ICIQ High Resolution Mass Spectrometry Unit on Waters GCT gas chromatograph coupled time-of-flight mass spectrometer (GC/MS-TOF) with electron ionization (EI). X-ray data were obtained from the ICIQ X-Ray Unit using a Bruker-Nonius diffractometer equipped with an APPEX 2 4K CCD area detector. Optical rotations are reported as follows: $[\alpha]_{D}^{rt}$ (*c* in g per 100 mL, solvent).

The authors are indebted to the team of the Research Support Area at ICIQ, in particular to Dr. Marta Martínez and Eduardo Escudero-Adan (X-ray Unit). We thank Grace Fox for proofreading the manuscript. Mattia Silvi is acknowledged for initial spectroscopic experiments.

General Procedures. All the reactions were set up under air and using freshly distilled solvents, without any precautions to exclude moisture, unless otherwise noted - open air chemistry on the benchtop.

Chromatographic purification of products was accomplished using force-flow chromatography (FC) on silica gel (35-70 mesh). For thin layer chromatography (TLC) analysis throughout this work, Merck precoated TLC plates (silica gel 60 GF₂₅₄, 0.25 mm) were used, using UV light as the visualizing agent and an acidic mixture of ceric ammonium molybdate or basic aqueous potassium permangante (KMnO₄), and heat as developing agents. Organic solutions were concentrated under reduced pressure on a Büchi rotary evaporator.

Determination of Diastereomeric Ratios

The diastereomeric ratio was determined by ¹H NMR analysis of the crude reaction mixture.

Determination of Enantiomeric Purity. HPLC analysis on a chiral stationary phase column was performed on an Agilent 1200-series instrumentation. Daicel Chiralpak AD-H, IA, IB or IC columns and Daicel Chiralcel OD-H with *i*-PrOH/hexane as the eluent were used, as specified in the individual experiment. HPLC traces were compared to racemic samples prepared by using a racemic mixture of the commercial

available chiral amine C.

Determination of Yield and Conversion in the Optimization Studies. The conversion of the starting materials and the yield of product in the optimization studies related to the model reaction depicted in Table 1 of the main manuscript were determined by ¹H NMR spectroscopy adding an internal standard in the crude reaction: 2,5-dimethylfuran: δ 2.26 ppm (s, 6H), 5.84 (s, 2H). Since in all instances the conversion of isatin **1a** was equal to the yield of product **3a**, in some cases the yield was determined by integration of the signals of the unreacted isatin **1a** in the ¹H NMR spectra (N-benzyl isatin **1a** NMR signal @ δ 4.92 ppm (s) and product **3a** signal @ 9.34 (s) and 9.21 (s); double checked with the product signals @ 6.42 (d) and 6.16 (d)).

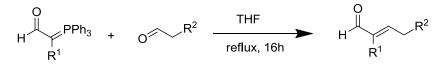
Materials. Commercial grade reagents and solvents were purchased from Sigma Aldrich, Fluka, and Alfa Aesar and used as received, without further purification; otherwise, where necessary, they were purified as

recommended.¹ The cinchona-based primary amine catalysts, 9-amino(9-deoxy)*epi*-quinine \mathbf{A}^2 and 6'hydroxy-9-amino-9-deoxyepiquinine \mathbf{B} ,³ were both prepared from commercially available quinine following the literature procedure. Chiral secondary amine catalyst \mathbf{C} is commercially available (Aldrich or Alfa Aeser); it was purified by flash column chromatography prior to use and stored at 4°C under argon to avoid undesired desilylation that would affect the catalytic potential of the amine. Catalysts \mathbf{D}^4 and \mathbf{E}^5 have been synthesized following the procedure reported in the literature.

The N-benzyl protected isatins **1** were easily synthesized from the corresponding, commercially available unprotected isatins, according to the following procedure: a solution of N-H isatin (5 mmol in 40 ml of dry DMF) was slowly added to a suspension of sodium-hydride (1.04 g, 60% dispersion in paraffin liquid, 1.3 equiv) in dry DMF (100ml) at 0 °C over a period of 10 minutes. The mixture was stirred at the same temperature for further 30 minutes. Then benzylbromide (6 mmol, 1.2 equiv) was added dropwise at the same temperature. The mixture was slowly warmed up at room temperature and stirring continued until the reaction was over (complete consumption of the starting N-H isatin, as judge by analytical TLC). The reaction was cooled at 0 °C and quenched with water (750 ml). The suspension was then filtered and the filtrated recrystallized from EtOAc and hexane to give the final product **1**.

Most of the α -branched unsaturated aldehydes 2 are commercially available and were purchased from Aldrich or Alfa Aeser and used as received. Otherwise, they were synthesized according to the following procedure.

Preparation of α-Branched Enals⁶



 R^1 = Me, Et R^2 = Et, Bn, Ar, CH₂S-Me, CH₂N-Cbz

A mixture of linear aliphatic aldehyde (1 equiv) and the appropriate triphenylphosphorane (1.5 equiv) was dissolved in THF and refluxed for 16h. The solution was then allowed to reach room temperature; the solvent was removed under reduced pressure and the crude mixture was purified by flash column chromatography (silica gel) to yield the desired product α -branched enals **2** (yield 40-50 %).

Note: For all the α -branched enals **2**, a *E*/*Z* ratio >95:5 was determined by ¹H NMR analysis. No double bond scrambling was observed neither during the catalytic reaction (checked by analysis of the crude reaction mixture) nor mixing the enal with a catalytic amount of catalyst **C**.

¹ W. L. F. Armarengo, D. D. Perrin, In *Purification of Laboratory Chemicals*, 4th ed.; Butterworth Heinemann: Oxford, 1996.

² S. H. McCooey, S.J. Connon, *Org. Lett.* **2007**, *9*, 599–602.

³ W. Chen, W. Du, Y. Duan, Y. Wu, S.-Y. Yang, Y.-C. Chen, *Angew. Chem. Int. Ed.* **2007**, *46*, 7667–7670.

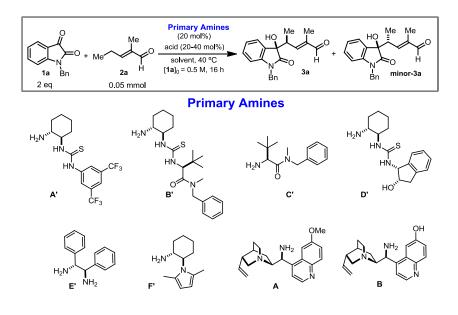
⁴ Grošelj, U.; Seebach, D.; Badine, D. M.; Schweizer, W. B.; Beck, A. K.; Krossing, I.; Klose, P.; Hayashi, Y.; Uchimaru, T. Helv. Chim. Acta **2009**, *92*, 1225-1259.

⁵ Ł. Albrecht, G. Dickmeiss, F. Cruz Acosta, C. Rodríguez-Escrich, R. L. Davis, K. A. Jørgensen, J. Am. Chem. Soc. 2012, 134, 2543– 2546.

⁶ Adapted from: Gagosz, F. Org. Lett. **2005**, 7, 4129-4132

B. Optimization Studies

Table S1. Catalyst Screening - Primary Amines



catalyst	additive	solvent	conv. (%)⁵	dr ٥	ее за (%) ^с	eeminor (%) ^c
A'	BA	toluene	28	1.3:1	6	0
B'	BA	toluene	66	1:1.9	20	20
C'	TFA	toluene	55	1:1.2	-	-
D'	BA	toluene	55	1:1	15	23
E'	TFA	toluene	47	1:1	0	0
E'	BA	toluene	n.r.	-	-	-
F'	TFA	toluene	23	1.3:1	55	43
Α	TFA	CH₃CI	42	6:1	<5	<5
Α	p-TSA	CH₃CI	40	4.5:1	<5	<5
В	TFA	CH₃CI	67	2.8:1	42	65
В	TFA	toluene	74 ^d	2.5:1	40	51
В	TFA	THF	43	3.1:1	46	58

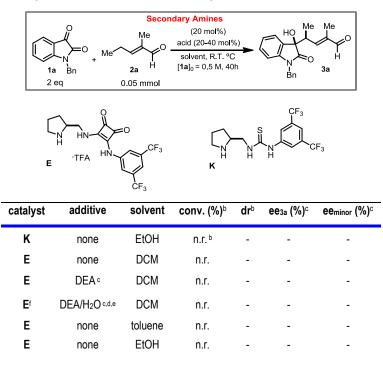
BA: benzoic acid; TFA: trifluoroacetic acid; p-TSA: p-toluensulfonic acid. The absolute configuration of the minor isomer was not univocally determined. ^a Reactions performed at 40 °C on a 0.05 mmol scale using 2 equivalents of (*E*)-2-methylpent-2-enal **2a** with [**1a**]₀ = 0.5 M, reaction time 16h. ^b Both conversion and diastereomeric ratios (dr) were determined by ¹H NMR analysis of the crude reaction mixture. ^c Determined by HPLC analysis on a chiral stationary phase. ^d Reaction time 40 h.

Table S2. Catalyst screening - Secondary Amines

	1a 2 ec		Secon Me 2a 0.05 mmol	dary Amines (20 mol%) acid (20-40 mol% solvent, 40 °C [1a] ₀ = 0.5 M, 16		HO Me Me N O Jan Bn Ja	н Н
$\bigcap_{\mathbf{C}}^{\mathbf{N}}$	Ph Ph OTMS		n 🚺	Ph Ph OSiPh ₂ Me	_	F ₃ C	
G	< ^Y NH NH NH NH NH NH NH NH NH NH NH NH NH	H Ph H	<			K OTMS CF	J H H
	catalyst	additive	solvent	conv. (%) ^b	dr ^b	ee 3a (%) ^c	eeminor (%) ^c
	С	BA	toluene	45	1/1	40	20
	В	BA	toluene	n.r.	-	-	-
	К	BA	EtOH	n.r.	-	-	-
	F	BA	toluene	n.r.	-	-	-
	G	TFA	toluene	30	1.9:1	5	5
	Н	HCI	EtOH	n.r.	-	-	-
	I	TFA	EtOH	n.r.	-	-	-
	J	AcOH _(1eq)	EtOH	n.r.	-	-	-
	С	BA	EtOH	>95	1.2:1	75	65
	D	BA	EtOH	>95	1.2:1	83	76
	Dd	BA	EtOH	>95	1.4:1	89	84

n.r.: no reaction. BA: benzoic acid; TFA: trifluoroacetic acid; p-TSA: p-toluensulfonic acid. a Reactions performed at 40 c on a 0.05 mmol scale using 2 equivalents of (*E*)-2-methylpent-2-enal **2a** with [**1a**]a = 0.5 M, reaction time 16h. b Both conversion and diastereomeric ratios (dr) were determined by ¹H NMR analysis of the crude reaction mixture. c Determined by HPLC analysis on a chiral stationary phase. d Reaction carried out at 25 c.

Table S3. Catalyst screening – Bifunctional Secondary Amines



n.r.: no reaction. DCM dichloromethane; DEA N,N-diethylacetamide. ^a Reactions performed at 25 °C on a 0.05 mmol scale using 2 equivalents of (*E*)-2methylpent-2-enal **2a** with [**1a**]₀ = 0.5 M, reaction time 16h. ^b Reactions performed at 40 °C. ^c 1 eq (0.05 mmol) of DEA was used. ^d 2.8 eq of water were used. ^a Reactions performed with [**1a**]₀ = 0.25 M. ^f These reaction conditions (selected for entry 4, Table 1 of the main manuscript) reflect the optimized system as reported in the original papers describing the preparation and the synthesis of catalyst **E**.^{5,7}

	0 benzo	D (20 mol%) ic acid (20 mo lvent, 25 °C = 0.5 M, 16 h	→ _()	Me Me \sim_{O} H 3a
solvent	conv. (%) ^b	drb	ее за (%) с	eeminor (%) ^c
EtOH	>95	1.2	83	76
DCE	10	2.5:1	92	67
dioxane	n.r.	-	-	-
CHCl₃	n.r.	-	-	-
Et ₂ O	n.r.	-	-	-
THF	n.r.	-	-	-
EtOAc	n.r.	-	-	-
CH₃CN	30	3:1	92	77
2,2,2-trifluoroethanol	n.r.	-	-	-
EtOH/CH ₃ CN (1/1)	75	2.2:1	91	79
EtOH/CH3CN (1/9)	49	2.7:1	92	76

 Table S4. Solvent screening

n.r.: no reaction. ^a Reactions performed at 25 °C on a 0.05 mmol scale using 2 equiv. of **2a** with [**1a**]₀ = 0.5 M, reaction time 16 h. 20 mol% of amine **D** and benzoic acid was were used. ^b Determined by ¹H NMR analysis of the crude reaction mixture. ^c Determined by HPLC analysis.

⁷ Ł. Albrecht, F. Cruz Acosta, A. Fraile, A. Albrecht, J. Christensen, K. A. Jørgensen, Angew. Chem. Int. Ed. 2012, 51, 9088–9092. See also Ref 5 in the S.I.

Table S5. Acidic additive screening

la la		Me H H H H H H H H H H H H H H H H H H H	mol%) I (9/1), 25 ℃ N C		OSiPh ₂ Me Ph Ph D
		Acids OH OH Binol	s tested Ph BocHN CO ₂ H N-Boc-PhGly		
Add	litive	Conv. (%) ^b	drb	ee 3a (%) ^c	eeminor (%) ^c
Ac	ОН	45	2.8:1	92	81
CH ₂ CI	-CO ₂ H	44	2.9:1	88	80
2-F-C ₆ ł	H4CO2H	45	2.8:1	88	78
(<i>L</i>)-N-Bo	oc-PhGly	52	3.2:1	83	75
(<i>D</i>)-N-Во	oc-PhGly	52	3.2:1	83	70
(S)-I	Binol	38	2:1	91	83
(<i>R</i>)-	Binol	47	2.2:1	91	82
Salicy	lic Acid	50	2.5:1	89	81
2-NO2-C	6H4CO2H	60	2.6:1	84	74
p-NO ₂ -	Phenol	65	1.6:1	80	60
2,6-F-Ce	H3CO2H	57	3.1:1	87	77
2,4,6-Me-	C6H2CO2H	50	2.7:1	88	73
4-NO ₂ -C	₆ H ₄ CO ₂ H	43	3:1	89	73
2,6-CF ₃ -C	C ₆ H ₃ CO ₂ H	59	3.2:1	91	77
3,5- <i>t</i> -butyl-	C6H3CO2H	45	2.8:1	90	73
2-Ph-C ₆	H4CO2H	44	2.8:1	90	73
2,6-MeO-(C6H3CO2H	60	2.7:1	91	75

^a Reactions performed at 25 °C on a 0.05 mmol scale using 20 mol% of amine **D** and 2 equiv. of (*E*)-2-methylpent-2-enal **2a** with [**1a**]₀ = 0.5 M, reaction time 16h. ^b Both conversion and diastereomeric ratios (dr) were determined by ¹H NMR analysis of the crude reaction mixture.

Table S6. Amine D/2,6-CF₃-C₆H₃CO₂H ratio

Ner Ner				OSiPh ₂ Me Ph D
x	Conv. (%) ^b	Dr _{3a:4a} b	ee₃a (%)°	eeminor (%) ^c
40	35	3.2/1	90	77
20	59	3.2/1	91	77
10	48	3.2/1	91	78
0	20	3.3/1	91	76

^a Reactions performed on a 0.05 mmol scale using 20 mol% of amine **D** in combination with different amount of 2,6-bis (trifluoromethyl) benzoic acid. ^b Determined by ¹H NMR analysis of the crude reaction mixture. ^c Determined by HPLC analysis.

Table S7. Temperature effect

1a Bn	Me 2a H 2a H <u>2a</u> H C20 H 2.6-CF ₃ BA ElOH/CH ₃ C [1a] ₀ = 0.	(20 mol%) N (9/1), T °C		OSiPh ₂ Me Ph D
T (°C)	Conv. (%) ^b	dr ^b	ee₃a (%)º	ee _{minor} (%) ^c
40	100	2/1	83	64
30	74	2.7/1	89	70
r.t.	59	3.2/1	91	77
10	<10	3.5/1	n.d.	n.d.

^a Reactions performed on a 0.05 mmol scale using 20 mol% of amine **D** in combination with 20 mol% of 2,6-bis (trifluoromethyl) benzoic acid. ^b Determined by ¹H NMR analysis of the crude reaction mixture. ^c Determined by HPLC analysis

Table S8. Concentration effect.

$ \begin{array}{c} & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & $							
[1a]	Conv. (%) ^b	Dr _{3a:4a} b	ee 3a (%) ^c	eeminor (%) ^c			
2	>95	2.8/1	89	70			
2	95 (87) ^d	3.2/1	90	76			
1	>95	2.9/1	89	72			
0.5	55	3.2/1	91	77			
0.25	30	3.3/1	91	79			

^a Reactions performed on a 0.05 mmol scale using 2 equivalents of 2-methyl pentenal 1, reaction time 16h. ^b Yield and d.r. determined by ¹H NMR analysis of the crude reaction mixture. ^c Determined by HPLC analysis on a chiral stationary phase. ^d 10 mol % of **D** and 10 mol % of acid, reaction time 36 h. Value between brackets refers to the yield of the isolated compound **3a** after chromatography.

Table S9. The importance of different N-protecting groups on the isatin derivatives 1

$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array} \\ \begin{array}{c} \end{array}\\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\$						
PG	R ¹	Conv. (%)	dr	ee 3 (%)		
Bn	CH ₃	49	3.2:1	92		
Ме	CH ₃	50	2.9:1	91		
Boc	CH ₃	<10 ^b	-	-		
Cbz	CH ₃	<10 ^b	-	-		
Н	$CH_2C_6H_5$	43	2:1	84		

^a Reactions performed on a 0.05 mmol scale using 2 equivalents of **2** with $[1]_0 = 0.5$ M, reaction time 16h. A combination of catalyst **D** and benzoic acid was used. ^b the formation of the hemiacetal deriving from the attack of ethanol on the isatin derivatives was oserved.

Table S10. Optimizing the reaction conditions for the vinylogous aldolization with unprotected N-H isatin.

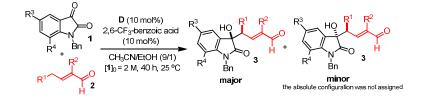
additive	solvent	conv. (%) ^b	dr ^b	ee 3 (%) ^c	eeminor (%) ^c
BA	DCM	n.r.	-	-	-
BA	CH ₃ CN	41	2.5:1	85	81
BA	EtOH	72	1.1:1	85	80
BA	CH ₃ CN/EtOH (9/1)	43	2:1	84	81
BA	CH ₃ CN/EtOH (1/1)	53	1.6:1	79	60
AcOH	CH ₃ CN	38	2:1	85	81
(L)-N-Boc-PhGly	CH ₃ CN	30	2.8:1	88	80
(D)-N-Boc-PhGly	CH ₃ CN	25	3:1	87	80
$2,6\text{-}\text{F-}\text{C}_6\text{H}_3\text{CO}_2\text{H}$	CH ₃ CN	35	3.3:1	88	-
2,4,6-Me-C ₆ H ₂ CO ₂ H	CH ₃ CN	46	2.2:1	81	-
$4-NO_2-C_6H_4CO_2H$	CH₃CN	14	3.3:1	90	82
2,6-OH-C6H3CO2H	CH ₃ CN	n.r.	-	-	-

 $\begin{array}{c|c}
 & Me \\
 & H \\
 & H$

n.r.: no reaction. BA: benzoic acid; N-Boc-PhGly: N-Boc phenylglycine, see Table S5 for the structure.

^a Reactions performed at 25 °C on a 0.05 mmol scale using 2 equivalents of enal **2a** with [**1a**]₀ = 0.5 M, reaction time 16h. ^b Both conversion and diastereomeric ratios (dr) were determined by ¹H NMR analysis of the crude reaction mixture. ^c Determined by HPLC analysis on a chiral stationary phase.

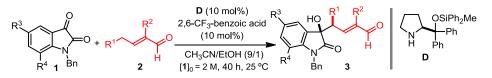
Table S11. Scope of the Direct Vinylogous Aldolization - ee of the minor diastereoisomer



entry	R ¹	R ²	R ³	R ⁴	3	yield (%) ^b	dr∘	e.e. % ^d major/minor
1	Me	Me	Н	Н	а	87	3.2:1	90 / 76
2	Bn	Me	Н	Н	b	68	2.5:1	95 / 77
3	CH ₂ -SMe	Me	Н	Н	С	89	1.6:1	94 / 70
4	CH ₂ NHCbz	Me	Н	Н	d	63	3:1	94 / 74
5 ^e	Bn	Bn	Н	Н	е	65	1.5:1	90 / <mark>78</mark>
6 ^e	Et	Et	Н	Н	f	28 ^f	1.5:1	94 / -
7	Me	Me	CI	Н	g	92	1.7:1	86 / 73
8	Bn	Me	CI	Н	ĥ	69	1.6:1	92 / <mark>78</mark>
9	Me	Me	Br	Н	i	68	1.9:1	85 / <mark>78</mark>
10	Me	Me	Me	Н	j	76	3.8:1	92 / <mark>81</mark>
11	Me	Me	NO ₂	Н	k	87	1.5:1	87 / <mark>75</mark>
12	Me	Me	CF₃O	Н	1	71	2.9:1	89 / <mark>63</mark>
13	Me	Me	Me	Me	m	65	3.9:1	91 / 77
14	Me	Me	Н	Br	n	88	2.4:1	92 / <mark>71</mark>

^a Reactions performed on a 0.2 mmol scale using 2 equiv of **2**. *E/Z* ratio of **2**: >95:5. Only the (*E*)-isomer of the aldol products **3** has been detected. ^b Yield of the isolated product **3** after chromatographic purification on silica gel. ^c Determined by ¹H NMR analysis of the crude mixture. ^d Ee values determined by HPLC analysis. ^fYield of the isolated major diastereomer of **3f**.

C. General Procedure for the Vinylogous Aldol Reaction



All the reactions were carried out in a 9/1 mixture of acetonitrile and ethanol without any precaution for excluding air and moisture (open air chemistry on the benchtop). An ordinary vial equipped with a Teflon-coated stir bar and a plastic screw cap was charged with (*S*)-(–)- α , α -diphenyl-2-pyrrolidinemethanol methyldiphenylsilyl ether **D** (9.00 mg, 0.02 mmol, 10 mol%) and 2,6-bis(trifluromethyl)benzoic acid (5.2 mg, 0.02 mmol, 10 mol%). Then the solvent mixture (100µL) and the α -branched enal **2** (0.4 mmol) were sequentially added and the resulting solution stirred at ambient temperature for 5 minutes. The reaction was started by the addition of the *N*-benzyl protected isatin derivative **1** (0.2 mmol). The vial was sealed and immerged in a water bath (thermostated at 25 °C) and stirring continued over 40 hours. Then the crude mixture was flushed through a short plug of silica, using dichloromethane/diethyl ether 1:1 as the eluent (5 ml). Solvent was removed under reduced pressure and the crude mixture was isolated by flash column chromatography using the specified eluent.

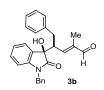
(S,E)-4-((R)-1-benzyl-3-hydroxy-2-oxoindolin-3-yl)-2-methylpent-2-enal (3a)

The reaction was carried out according to the general procedure to furnish the crude product as a 3.2:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.42 ppm (d), δ_{minor} 6.21 ppm (d).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.24$ hexane/ethyl acetate 9/1) in 87% yield (white solid). The enantiomeric excess was determined to be 90% for the major diasteroisomer (76% EE for the minor) by HPLC analysis on a Daicel Chiralpak IB column: 95:5 hexane/i-PrOH, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 26.7$ min, $\tau_{minor} = 62.5$ min. [α]_D²⁸ = +61.0 (c = 0.79, CHCl₃, d.r. 3.2/1, major 90% ee, minor 76% ee). HRMS calc. for ($C_{21}H_{21}NO_3+Na$): 358.1419, found 358.1419.

¹H NMR (400 MHz, CDCl₃): δ 9.33 (s, 1H), 7.39 (d, 1H, J_1 = 7.3 Hz, J_2 = 1.1 Hz), 7.35-7.19 (m, 7H, overlap with the signal from the minor diastereomer), 7.08 (dt, 1H, J_d = 7.6 Hz, J_t = 0.9 Hz), 6.78 (d, 1H, J = 7.8 Hz, overlap with the signal from the minor diastereomer), 6.42 (dq, 1H, J_d = 10.5 Hz, J_q = 1.3 Hz), 5.05 (d, 1H, J = 15.6 Hz, overlap with the signal from the minor diastereomer), 4.71 (d, 1H, J = 15.6 Hz), 3.49-3.39 (m, 1H, overlap with the signal from the minor diastereomer), 1.74 (d, 3H, J = 1.2 Hz), 1.00 (d, 3H, J = 6.8 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 195.2, 177.3, 151.7, 142.9, 141.3, 135.5, 130.2, 129.0, 128.3, 128.1, 127.6, 127.5, 124.7, 123.3, 109.7, 78.4, 44.2, 41.5, 14.1, 9.8 ppm.

(S,E)-4-((R)-1-benzyl-3-hydroxy-2-oxoindolin-3-yl)-2-methyl-5-phenylpent-2-enal (3b)



The reaction was carried out following the general procedure to furnish the crude products as a 2.5:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 9.18 ppm (s), δ_{minor} 9.11 ppm (s).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.3$ hexane/ethyl acetate 8/2) in 68% yield (white solid). The enantiomeric excess was determined to be 95%

for the major diasteroisomer (77% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 90/10 hexane/i-PrOH, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 22.9$ min, $\tau_{minor} = 41.0$ min. $[\alpha]_D^{26} = +149.0$ (c = 1.0, CHCl₃, d.r. 2.5/1, major 95% ee, minor 77% ee). HRMS calc. for (C₂₇H₂₅NO₃+Na): 434.1732, found 434.1740.

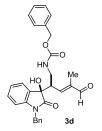
¹H NMR (400 MHz, CDCl₃): δ 9.18 (s, 1H), 7.46 (d, 1H, $J_1 = 7.4$ Hz, $J_2 = 0.9$ Hz), 7.37-6.69 (m, 12H, overlap with the signal from the minor diastereomer), 6.81 (d, 1H, J = 7.7 Hz), 6.31 (dq, 1H, $J_d = 10.8$ Hz, $J_q = 1.2$ Hz), 5.05 (d, 1H, J = 15.8 Hz, overlap with the signal from the minor diastereomer), 4.73 (d, 1H, J = 15.8 Hz), 3.66 (td, 1H, $J_t = 11.0$ Hz, $J_d = 3.3$ Hz, overlap with the signal from the minor diastereomer), 3.08 (dd, 1H, $J_1 = 13.4$ Hz, $J_2 = 11.2$ Hz), 1.26 (d, 3H, J = 1.3 Hz, overlap with the signal from the minor diastereomer) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 194.8, 177.2, 149.5, 143.2, 142.8, 138.4, 135.4, 130.4, 129.1, 129.0, 128.6, 128.5, 128.0, 127.6, 127.5, 126.6, 124.5, 123.5, 109.9, 78.1, 49.5, 44.2, 35.1, 9.5 ppm

(S,E)-4-((R)-1-benzyl-3-hydroxy-2-oxoindolin-3-yl)-2-methyl-5-(methylthio)pent-2-enal (3c)



The reaction was carried out following the general procedure **A** to furnish the crude products as a 1.6:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.14 ppm (d), δ_{minor} 5.89 ppm (d).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.28$ hexane/ethyl acetate 8/2) in 89% yield (colourless solid). The enantiomeric excess was determined to be 95% for the major diasteroisomer (77% ee for the minor) by HPLC analysis on a Daicel Chiralpak IB column: 95:5 hexane/i-PrOH, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 32.0$ min, $\tau_{minor} = 97.4$ min. [α]_D²⁶= +103.5 (c = 0.68, CHCl₃, d.r. 1.6/1, major 95% ee, minor 77% ee). HRMS calc. for (C₂₂H₂₃NO₃S+Na): 404.1296, found 404.1286. ¹H NMR (400 MHz, CDCl₃): δ 9.17 (s, 1H), 7.38 (d, 1H, J = 7.7 Hz), 7.34-7.15 (m, 5H, overlap with the signal from the minor diastereomer), 7.06 (dt, 1H, $J_d = 7.6$ Hz, $J_t = 0.8$ Hz), 6.78 (d, 1H, J = 7.8 Hz), 6.14 (dq, 1H, $J_d = 10.9$ Hz, $J_q = 1.4$ Hz), 5.02 (d, 1H, J = 15.5 Hz, overlap with the signal from the minor diastereomer), 4.70 (d, 1H, J = 15.5 Hz), 3.68 (bs, 1H), 3.65-3.55 (m, 1H, overlap with the signal from the minor diastereomer), 2.94 (dd, 1H, $J_I = 13.1$ Hz, $J_2 = 4.6$ Hz), 2.54 (dd, 1H, $J_I = 13.0$ Hz, $J_2 = 10.0$ Hz), 2.07 (s, 3H), 1.66 (d, 3H, J = 1.3 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 194.8, 176.7, 148.8, 143.3, 143.3, 142.5, 135.4, 130.4, 129.0, 128.2, 127.4, 124.3, 123.5, 109.7, 77.9, 46.5, 44.2, 33.3, 16.1, 10.1 ppm.



benzyl ((**R**,**E**)-2-((**R**)-1-**benzyl-3-hydroxy-2-oxoindolin-3-yl**)-4-methyl-5-oxopent-3-en-1-yl)carbamate (3d). The reaction was carried out following the general procedure to furnish the crude products as a 3.0:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.10 ppm (bd), δ_{minor} 5.91 ppm (bd).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.25$ hexane/ethyl acetate 7/3) in 63% yield (white solid). The enantiomeric excess was determined to be 94% for the major diasteroisomer by HPLC analysis on a Daicel Chiralpak IB column: 85/15

hexane/i-PrOH, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 27.4$ min, $\tau_{minor} = 45.1$ min. $[\alpha]_D^{26} = +55.5$ (c = 0.90, CHCl₃, d.r. 3.0/1, 94% ee_{major}, 74% ee_{minor}). HRMS calc. for (C₂₉H₂₈N₂O₅+Na): 507.1896, found 507.1911. ¹H NMR (400 MHz, CDCl₃): δ 9.04 (bs, 1H), 7.50-7.13 (m, 14H, overlap with the signal from the minor diastereomer), 7.05 (t, 1H, $J_t = 7.4$ Hz), 6.73 (d, 1H, J = 7.8 Hz), 6.10 (bd, 1H, J = 10.0 Hz), 5.34-5.25 (m, 1H), 5.07-5.03 (m, 1H, overlap with the signal from the minor diastereomer), 4.99 (d, 1H, J = 15.6 Hz, overlap with the signal from the minor diastereomer), 4.66 (d, 1H, J = 15.6 Hz), 3.82-3.73 (m, 1H, overlap

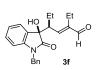
with the signal from the minor diastereomer), 3.64-3.47 (m, 2H, overlap with the signal from the minor diastereomer), 1.50 (bs, 3H) ppm.¹³C NMR (100 MHz, CDCl₃): δ194.7, 176.6, 153.7, 147.2, 143.3, 142.2, 136.4, 135.4, 130.3, 129.4, 129.0, 128.6, 128.2, 127.5, 124.1, 123.6, 109.6, 77.2, 66.9, 47.2, 44.0, 40.0, 9.7 ppm.

(S,E)-2-benzyl-4-((R)-1-benzyl-3-hydroxy-2-oxoindolin-3-yl)-5-phenylpent-2-enal (3e). The reaction was carried out following the general procedure (using 20% mol of catalyst loading) to furnish the crude products as a 1.5:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{maior} 9.32 ppm (s), δ_{minor} 9.24 ppm (s).

The title compound was isolated as a mixture of diastereoisomers (hexane/ethyl acetate 10/1) in 65% yield (white solid). The enantiomeric excess was determined to be 90% for the major diasteroisomer by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 9.3$ min, $\tau_{minor} = 11.7$ min. $[\alpha]_D^{26} = +92.5$ (c = 0.75, CHCl₃, d.r. 1.5/1, major 90% ee, minor 76% ee). HRMS calc. for (C₃₃H₂₉NO₃+Na): 510.2045, found 510.2021.

¹H NMR (400 MHz, CDCl₃): δ 9.32 (bs, 1H), 7.36-6.93 (m, 20H, overlap with the signal from the minor diastereomer), 6.79 (bd, 1H, *J* = 7.8 Hz, overlap with the signal from the minor diastereomer), 6.76-6.71 (m, 2H), 6.48 (d, 1H, *J* = 11.0Hz), 5.00 (d, 1H, *J* = 15.9 Hz, overlap with the signal from the minor diastereomer), 4.72 (d, 1H, *J* = 15.9 Hz), 3.72 (td, 1H, *J_t* = 11.0 Hz, *J_d* = 3.4 Hz), 3.41 (d, 1H, *J* = 15.1 Hz), 3.02-2.95 (m, 1H, overlap with the signal from the minor diastereomer), 2.87 (bs, 1H), 2.39 (dd, 1H, *J_t* = 13.3 Hz, *J₂* = 10.7 Hz) ppm.¹³C NMR (100 MHz, CDCl₃): δ 194.1, 177.0, 150.9, 146.0, 142.4, 138.3, 138.1, 135.3, 130.2, 129.1, 129.0, 128.7, 128.6, 128.5, 128.4, 128.3, 128.2, 128.0, 127.5, 126.6, 126.0, 124.7, 123.4, 109.8, 77.9, 49.3, 44.1, 35.4, 29.7 ppm.

(S,E)-4-((R)-1-benzyl-3-hydroxy-2-oxoindolin-3-yl)-2-ethylhex-2-enal (3f)



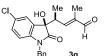
The reaction was carried out following the general procedure (using 20% mol of the catalyst **D**/acid combination) to furnish the crude products as a 1.5:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.22 ppm (d), δ_{minor} 5.87 ppm (d).

The title compound was isolated as a single diastereoisomer (hexane/ethyl acetate 10:1) in 28% yield (white solid). The enantiomeric excess was determined to be 94% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 9.1$ min, $\tau_{minor} = 12.9$ min. [α]_D²⁶= +84.5 (c = 1.45, CHCl₃, 94% ee). HRMS calc. for (C₂₃H₂₅NO₃+Na): 386.1732, found 386.1739.

¹H NMR (400 MHz, CDCl₃): δ 9.35 (s, 1H), 7.37 (dd, 1H, J_1 = 7.4 Hz, J_2 = 1.0 Hz), 7.34-7.22 (m, 6H), 7.07 (d, 1H, J_t = 7.5 Hz, J_d = 1.0 Hz), 6.77 (d, 1H, J = 7.8 Hz), 6.22 (d, 1H, J = 11.0Hz), 5.01 (d, 1H, J = 15.4 Hz), 4.74 (d, 1H, J = 15.4 Hz), 3.26 (td, 1H, J_t = 11.1 Hz, J_d = 3.0 Hz), 2.96 (bs, 1H), 2.36-2.23 (m, 2H), 1.67-1.58 (m, 1H), 1.20-1.10 (m, 1H), 0.95 (t, 3H, J = 7.5 Hz), 0.76 (t, 3H, J = 7.5 Hz) ppm.

¹³C NMR (100 MHz, CDCl₃): *δ* 194.9, 177.4, 150.4, 149.3, 142.9, 135.5, 1302, 129.0, 128.0, 127.6, 124.7, 123.3, 109.8, 78.2, 48.6, 44.2, 21.7, 18.1, 13.4, 12.1 ppm.

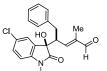
(S,E)-4-((R)-1-benzyl-5-chloro-3-hydroxy-2-oxoindolin-3-yl)-2-methylpent-2-enal (3g)



The reaction was carried out following the general procedure to furnish the crude products as a 1.7:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: $\delta_{major} 6.39 \text{ ppm}$ (d), $\delta_{minor} 6.17 \text{ ppm}$ (d).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 8/2) in 92% yield (white solid). The enantiomeric excess was determined to be 86% for the major diasteroisomer (73% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 90:10 hexane/i-PrOH, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 15.3$ min, $\tau_{minor} = 22.2$ min. [α]_D²⁶= +31.0 (c = 1.15, CHCl₃, d.r. 1.7/1, _{major} 86% ee, _{minor} 73% ee). HRMS calc. for (C₂₁H₂₀NO₃Cl+Na): 392.1029, found 392.1038.

¹H NMR (400 MHz, CDCl₃): δ 9.36 (s, 1H), 7.39 (d, 1H, J = 2.1 Hz), 7.37-7.18 (m, 7H, overlap with the signal from the minor diastereomer), 6.72 (d, 1H, J = 8.5 Hz, , overlap with the signal from the minor diastereomer), 6.39 (dq, 1H, $J_d = 10.2$ Hz, $J_q = 1.3$ Hz), 5.05 (d, 1H, J = 15.7 Hz, overlap with the signal from the minor diastereomer), 4.72 (d, 1H, J = 15.7 Hz), 3.49-3.39 (m, 1H, overlap with the signal from the minor diastereomer), 1.77 (d, 3H, J = 1.3 Hz), 1.05 (d, 3H, J = 6.8 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 195.0, 176.9, 150.8, 141.6, 141.4, 135.0, 130.1, 129.1, 128.9, 128.3, 127.5, 127.4, 125.2, 110.7, 78.4, 44.3, 41.5, 14.0, 9.9 ppm.



(S,E)-4-((R)-1-benzyl-5-chloro-3-hydroxy-2-oxoindolin-3-yl)-2-methyl-5-phenylpent-2-enal (3h)

The reaction was carried out following the general procedure to furnish the crude products as a 1.6:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.42 ppm (d), δ_{minor} 5.97 ppm (d).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 8/2) in 69% yield (white solid). The enantiomeric excess was determined to be 92% for the major diasteroisomer (78% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 52.5/2.5/50 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 7.9$ min, $\tau_{minor} = 9.2$ min. [α]_D²⁶= +80.0 (c = 0.77, CHCl₃, d.r. 1.6/1, _{major} 92% ee, _{minor} 78% ee). HRMS calc. for (C₂₇H₂₄NO₃Cl+Na): 468.1342, found 468.1358.

¹H NMR (400 MHz, CDCl₃): δ 9.19 (s, 1H), 7.42 (d, 1H, J = 2.2 Hz), 7.36-7.12 (m, 9H, overlap with the signal from the minor diastereomer), 7.05-7.01 (m, 2H overlap with the signal from the minor diastereomer), 6.71 (d, 1H, J = 8.4 Hz, overlap with the signal from the minor diastereomer), 6.42 (dq, 1H, $J_d = 10.9$ Hz, $J_q = 1.3$ Hz), 5.02 (d, 1H, J = 15.4 Hz, overlap with the signal from the minor diastereomer), 4.73 (d, 1H, J = 15.4 Hz, 3.62 (td, 1H, $J_t = 10.9$ Hz, $J_d = 3.6$ Hz, overlap with the signal from the minor diastereomer), 3.34 (bs, 1H), 3.07 (dd, 1H, $J_1 = 13.6$ Hz, $J_2 = 3.1$ Hz), 2.49 (dd, 1H, $J_1 = 13.6$ Hz, $J_2 = 10.9$ Hz), 1.27 (s, 3H, signal overlapped with minor isomer) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 194.6, 176.8, 148.7, 143.5, 141.2, 138.1, 134.9, 130.4, 130.3, 129.2, 129.0, 128.6, 128.3, 127.6, 126.7, 125.0, 110.9, 78.0, 49.5, 44.3, 34.9, 9.5 ppm.

(S,E)-4-((R)-1-benzyl-5-bromo-3-hydroxy-2-oxoindolin-3-yl)-2-methylpent-2-enal

(3i). The reaction was carried out following the general procedure to furnish the crude products as a 1.9:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.36 ppm (d), δ_{minor} 6.19 ppm (d). The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 8/2) in 68% yield (white solid). The enantiomeric excess

was determined to be 85% for the major diasteroisomer (78% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 90:10 hexane/i-PrOH, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 8.3$ min, $\tau_{minor} = 11.5$ min. $[\alpha]_D^{26} = +51.0$ (c = 1.33, CHCl₃, d.r. 1.9/1, major 85% ee, minor 78% ee). HRMS calc. for (C₂₁H₂₀NO₃Br+Na): 436.0524, found 436.0533.

¹H NMR (400 MHz, CDCl₃): δ 9.32 (s, 1H), 7.50 (d, 1H, J = 2.0 Hz), 7.39-7.23 (m, 7H, overlap with the signal from the minor diastereomer), 6.64 (d, 1H, J = 8.3 Hz, overlap with the signal from the minor diastereomer), 6.36 (dq, 1H, $J_d = 10.4$ Hz, $J_q = 1.4$ Hz), 5.01 (d, 1H, J = 15.6 Hz, overlap with the signal from the minor diastereomer), 4.67 (d, 1 H, J = 15.6 Hz), 3.51 (bs, 1H), 3.47-3.36 (m, 1 H, overlap with the signal from the minor diastereomer), 1.74 (d, 3H, J = 1.4 Hz), 1.05 (d, 3H, J = 6.9 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 195.1, 176.9, 150.9, 141.9, 141.6, 134.9, 132.9, 130.6, 129.1, 128.2, 127.9, 127.5, 116.1, 111.2, 78.4, 44.2, 41.5, 14.1, 9.8 ppm.

HO Me Me (S,E)-4-((R)-1-benzyl-3-hydroxy-5-methyl-2-oxoindolin-3-yl)-2-methylpent-2-enal <math>(3j). The reaction was carried out following the general procedure to furnish the crude products as a 3.8:1 mixture of diastereoisomers; d.r. determined by integration of ¹H

NMR signal: δ_{major} 6.46 ppm (d), δ_{minor} 6.22 ppm (d). The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 8/2) in 76% yield (white solid). The enantiomeric excess was determined to be 95% for the major diasteroisomer (81% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 49:2:49 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 7.4 \text{ min}$, $\tau_{minor} = 10.3 \text{ min}$. [α]_D²⁶= +66.7 (c = 1.345, CHCl₃, d.r. 3.8/1, major 95% ee, minor 81% ee). HRMS calc. for ($C_{22}H_{23}NO_3+Na$): 372.1576, found 372.1586.

¹H NMR (400 MHz, CDCl₃): δ 9.36 (s, 1H), 7.35-7.19 (m, 6H, overlap with the signal from the minor diastereomer), 7.06 (bd, 1H, J = 7.9 Hz), 6.68 (d, 1H, J = 7.9 Hz, overlap with the signal from the minor diastereomer), 6.46 (dq, 1H, $J_d = 10.5$ Hz, $J_q = 1.3$ Hz), 5.02 (d, 1H, J = 15.6 Hz, overlap with the signal from the minor diastereomer), 4.69 (d, 1H, J = 15.6 Hz), 3.49-3.41 (m, 2H, overlap with the signal from the minor diastereomer), 2.34 (s, 3H), 1.76 (d, 3H, J = 1.3 Hz), 1.02 (d, 3H, J = 6.8 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 195.2, 177.4, 152.1, 141.2, 140.5, 135.6, 132.9, 130.4, 128.9, 128.5, 127.9, 127.6, 127.4, 125.4, 109.5, 78.6, 44.1, 41.5, 21.2, 14.2, 9.8 ppm.

(S,E)-4-((R)-1-benzyl-3-hydroxy-5-nitro-2-oxoindolin-3-yl)-2-methylpent-2-enal (3k)

^{Me} _H The reaction was carried out following the general procedure to furnish the crude products as a 1.5:1 mixture of diastereoisomers; d.r. determined by integration of ¹H ^{3k} NMR signal: δ_{major} 6.33 ppm (d), δ_{minor} 6.15 ppm (d).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 7/3) in 87% yield (white solid). The enantiomeric excess was determined to be 87% for the major diasteroisomer (75% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 49/2/49 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 10.7$ min, $\tau_{minor} = 14.0$ min. [α]_D²⁶= +80.3 (c = 0.80, CHCl₃, d.r. 1.5/1, major 87% ee, minor 75% ee). HRMS calc. for (C₂₁H₂₀N₂O₅+Na): 403.1270, found 403.1270.

¹H NMR (400 MHz, CDCl₃): δ 9.32 (s, 1H), 8.35-8.25 (m, 2H, overlap with the signal from the minor diastereomer), 7.39-7.23 (m, 5H), 6.88 (d, 1H, J = 6.7 Hz, overlap with the signal from the minor diastereomer), 6.33 (dq, 1H, $J_d = 10.4$ Hz, $J_q = 1.4$ Hz), 5.09 (d, 1H, J = 15.4 Hz, signal overlapped with minor isomer), 4.78 (d, 1H, J = 15.4 Hz), 3.55-3.39 (m, 1 H, signal overlap with the signal from the minor diastereomer), 3.26 (bs, 1H, overlap with the signal from the minor diastereomer), 1.71 (d, 3H, J = 1.4 Hz),

1.08 (d, 3H, J = 6.7 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 195.1, 178.3, 151.2, 141.3, 140.3, 136.9, 136.1, 132.0, 128.7, 127.5, 126.8, 124.6, 123.8, 103.1, 77.4, 44.8, 41.8, 14.0, 9.8 ppm.

(S,E)-4-((R)-1-benzyl-3-hydroxy-2-oxo-5-(trifluoromethoxy)indolin-3-yl)-2-methylpent-2-enal (31)

The reaction was carried out following the general procedure to furnish the crude products as a 2.9:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.28 ppm (d), δ_{minor} 6.16ppm (d).

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 8/2) in 71% yield (white solid). The enantiomeric excess was determined to be 89% for the major diasteroisomer (63% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 49/2/49 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 8.2$ min, $\tau_{minor} = 10.5$ min. HRMS calc. for ($C_{22}H_{20}NO_4F_3+Na$): 442.1242, found 442.1259.

¹H NMR (400 MHz, CDCl₃): δ 9.28 (s, 1H), 7.36-7.26 (m, 6H, overlap with the signal from the minor diastereomer), 7.14-7.10 (m, 1H), 6.76 (d, 1H, J = 8.4 Hz, overlap with the signal from the minor diastereomer), 6.28 (dq, 1H, $J_d = 10.3$ Hz, $J_q = 1.2$ Hz), 5.06 (d, 1H, J = 15.6 Hz, signal overlap with the signal from the minor diastereomer), 4.70 (d, 1H, J = 15.6 Hz), 3.45-3.37 (m, 1H, signal overlapped with minor isomer), 3.02 (bs, 1H, overlap with the signal from the minor diastereomer), 1.70 (d, 3H, J = 1.2 Hz), 1.07 (d, 3H, J = 6.6 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 194.9, 177.1, 150.5, 145.1, 141.6, 1414, 134.9, 129.2, 128.4, 127.6, 123.3, 118.6, 110.3, 78.3, 44.4, 41.7, 13.9, 9.8 ppm.

(S,E)-4-((R)-1-benzyl-3-hydroxy-5,7-dimethyl-2-oxoindolin-3-yl)-2-methylpent-2-enal (3m)

The reaction was carried out following the general procedure to furnish the crude products as a 3.9:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.47 ppm (d), δ_{minor} 6.39 ppm (d) in deutereted toluene.

The title compound was isolated as a mixture of diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 10/1) in 65% yield (pale-pink solid). The enantiomeric excess was determined to be 91% for the major diasteroisomer (77% for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 48.5:3:48.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 6.4$ min, $\tau_{minor} = 7.7$ min. [α]_D²⁶= +64.5 (c = 1.40, CHCl₃, d.r. 3.9/1, major 90% ee, minor 77% ee). HRMS calc. for (C₂₃H₂₅NO₃+Na): 386.1732, found 386.1751.

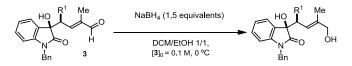
¹H NMR (400 MHz, CDCl₃): δ 9.40 (s, 1H), 7.32-7.20 (m, 3H, overlap with the signal from the minor diastereomer), 7.17-7.12 (m, 2H, overlap with the signal from the minor diastereomer), 7.08 (bs, 1H, signal overlap with the signal from the minor diastereomer), 6.81 (bs, 1H), 6.47 (dq, 1H, $J_d = 10.4$ Hz, $J_q = 1.4$ Hz), 5.15 (s, 1H, overlap with the signal from the minor diastereomer), 5.10 (s, 1H, overlap with the signal from the minor diastereomer), 5.10 (s, 1H, overlap with the signal from the minor diastereomer), 2.28 (s, 3H), 2.21 (s, 3H, overlap with the signal from the minor diastereomer), 1.73 (d, 3H, J = 1.4 Hz), 1.06 (d, 3H, J = 6.9 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 195.3, 178.5, 152.4, 141.1, 138.5, 137.3, 134.6, 132.9, 129.0, 127.5, 125.9, 123.3, 120.2, 77.6, 45.3, 41.6, 20.9, 18.8, 14.3, 9.8 ppm.

(S,E)-4-((R)-1-benzyl-7-bromo-3-hydroxy-2-oxoindolin-3-yl)-2-methylpent-2-enal (3n)

The reaction was carried out following the general procedure to furnish the crude products as a 2.4:1 mixture of diastereoisomers; d.r. determined by integration of ¹H NMR signal: δ_{major} 5.46 ppm (d), δ_{minor} 5.40 ppm (d). The title compound was isolated as a mixture of 3n diastereoisomers ($R_f = 0.30$ hexane/ethyl acetate 8/2) in 88% yield (white solid). The enantiomeric excess was determined to be 92% for the major diasteroisomer (71% ee for the minor) by HPLC analysis on a Daicel Chiralpak IA column: 49:2:49 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} =$ 8.2 min, $\tau_{minor} = 10.3$ min. $[\alpha]_D^{-26} = +38.6$ (c = 1.15, CHCl₃, d.r. 2.4/1, major 92% ee, minor 71% ee). HRMS calc. for (C₂₁H₂₀NO₃Br+Na): 436.0524, found 436.0508.

¹H NMR (400 MHz, CDCl₃): δ 9.30 (s, 1 H, overlap with the signal from the minor diastereomer), 7.43 (dd, 1H, $J_1 = 8.1$ Hz, $J_2 = 1.2$ Hz, overlap with the signal from the minor diastereomer), 7.36 (dd, 1H, $J_1 = 7.4$ Hz, J_2 = 1.2 Hz), 7.33-7.21 (m, overlap with the signal from the minor diastereomer), 6.97 (dd, 1 H, $J_1 = 8.2$ Hz, $J_2 = 7.3$ Hz, signal overlapped with minor isomer), 6.35 (dq, 1H, $J_d = 10.6$ Hz, $J_a = 1.5$ Hz, overlap with the signal from the minor diastereomer), 5.46 (d, 1H, J = 16.1 Hz), 5.29 (d, 1H, J = 16.1 Hz, overlap with the signal from the minor diastereomer), 3.45-3.33 (m, 2H, overlap with the signal from the minor diastereomer), 1.69 (d, 3H, J = 1.3 Hz), 1.03 (d, 3H, J = 6.7 Hz) ppm. ¹³C NMR (100 MHz, $CDCl_3$): δ 195.1, 178.3, 151.1, 141.4, 140.3, 136.9, 136.2, 132.0, 127.6, 126.8, 126.5, 124.6, 123.7, 103.1, 77.4, 44.8, 41.8, 14.0, 9.8 ppm.

D. General Procedure for the Reduction of Products 3



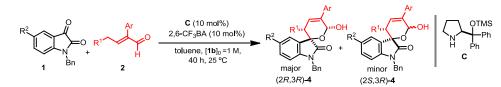
0.1 mmol of adducts 3a, 3h, and 3j were transferred in a vial and dissolved in 1 mL of a DCM/ethanol mixture (1 to 1 v/v) and cooled to 0 $^{\circ}$ C (ice bath). After the addition of NaBH₄ (1.5 equivalents, added in small portions) the mixture was stirred until the reaction was considered complete by TLC analysis (typically 1-2h). The reaction was then quenched with water and the compound extracted with diethyl ether. The aqueous phase was washed three times with diethyl ether and the combined organic phases dried over sodium sulphate. Solvent was removed under reduced pressure and the crude mixture was purified by chromatography column. Separation of the two diastereoisomers was straightforward, securing access to diastereomerically pure alcohol adducts.

Compound **3a** was reduced to the corresponding alcohol following the general procedure. The crude was purified by flash column chromatography (gradient from hexane/ethyl acetate 7/3 to 1/1, $R_f = 0.15$ in hexane/ethyl acetate 7/3) to afford compound **3a-red** as a single diastereoisomer in 71% yield (colorless solid, 47% overall yield).

¹H NMR (400 MHz, CDCl₃): δ 7.44-7.39 (m, 1H), 7.36-7.26 (m, 5H), 7.22 (td, 1H, J_t = 7.9 Hz, J_d = 1.3 Hz), 7.05 (td, 1H, $J_t = 7.9$ Hz, $J_d = 1.3$ Hz), 6.75 (d, 1H, J = 7.9 Hz), 5.48 (d, 1H, J = 10.3 Hz), 5.07 (d, 1H, J = 10 15.3 Hz), 4.70 (d, 1H, J = 15.3 Hz), 4.02 (bs, 1H), 3.25-3.17 (m, 1H), 3.03 (bs, 1H), 1.71 (s, 3H), 0.88 (d, 3H, J = 6.8 Hz) ppm. ¹³C NMR (125 MHz, CDCl₃): δ 177.9, 143.1, 139.0, 135.8, 129.6, 129.1, 128.9, 127.9, 127.6, 124.9, 123.8, 122.9, 109.4, 78.9, 68.6, 44.1, 40.2, 15.1, 14.4 ppm.

Compound **3h** was reduced to the corresponding alcohol following the general procedure. Chromatographic purification on silica gel (gradient from hexane/ethyl acetate 8:2 to 6/4, $R_f = 0.2$ in hexane/ethyl acetate 7/3) afforded the reduced adduct **3h-red** as single diasteroisomer in 55% yield (colorless solid, 35% overall yield). Since the alcohol was not solid, the major diasteroisomer was then re-oxidized to the aldehyde adduct **3h** in order to get suitable crystals for X-ray crystallographic analysis. Oxidation of the alcohol **3h-red**: compound **3h-red** was transferred in a round bottom flask, then DCM (1.5 ml, 0.05M) and activated MnO₂ (10 equivalents) were added and the stirring continued over a period of 16 hours. The mixture was filtered on celite and the solvent removed under reduced pressure. The crude product was purified by flash chromatography (hexane/ethyl acetate 7/3) to give the pure **3h** as single diastereoisomer in a 32% overall yield.

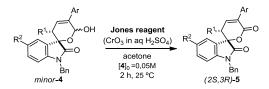
E. General Procedure for the Hetero-Diels-Alder-type Reaction of α-Aryl Substituted Enals



All the reactions were carried out in toluene (synthesis grade, >99%) without any precaution for excluding air and moisture (open air chemistry on the benchtop). An ordinary vial equipped with a Teflon-coated stir bar and a plastic screw cap was charged with (*S*)-(–)- α , α -diphenyl-2-pyrrolidinemethanol trimethylsilyl ether **C** (6.5 mg, 0.02 mmol, 10 mol%) and 2,6-bis(trifluromethyl)benzoic acid (5.2 mg, 0.02 mmol, 10 mol%). Then the solvent (200µL) and the α -branched enal **2** (0.4 mmol) were sequentially added and the resulting solution stirred at ambient temperature for 5 minutes. The reaction was started by the addition of the N-benzyl protected isatin derivative **1** (0.2 mmol). The vial was sealed and immerged in a water bath (thermostated at 25 °C) and stirring continued over 40 hours. Then the crude mixture was flushed through a short plug of silica, using dichloromethane/diethyl ether 1:1 as the eluent (5 ml). Solvent was removed under reduced pressure and the crude mixture was analyzed by ¹H NMR spectroscopy to determine the diastereomeric ratio. The two diastereoisomers for product **4** were isolated by flash column chromatography using the specified eluent.

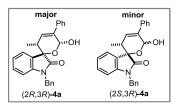
While the major isomers, the spirooxindole lactols (2R,3R)-4 are stable compounds that can be stored after isolation, the minor isomers (the (2S,3R)-4 adducts) are stable on the bench only for 2-3 days. In addition, given the difficulties of determining their enantiomeric excess by HPLC analysis, spirooxindole lactols (2S,3R)-4 were oxidized suddenly after their isolation. The procedure for the oxidation is as follows:

Procedure for the Oxidation of Lactol 4 to Lactons 5



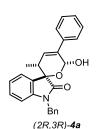
The oxidation of compounds **4** to the corresponding lactones **5** was performed following a slightly modified procedure reported in literature.⁸ 0.1 mmol of compound **4** were placed and dried in a 5 ml vial, followed by the sequential addition of acetone (0.5 mL) and the Jones reagent (dropwise, 0.2 mL). The mixture was stirred at room temperature over 2 hours, then diluted with diethyl ether and quenched with water. The aqueous phase was washed three times with diethyl ether and the combined organic phases dried over sodium sulphate. Solvent was removed under reduced pressure and the crude mixture was purified by chromatography on silica gel (typically with hexane/ethyl acetate 9/1 as the eluent) to afford the pure spirooxindole dihydropyran-2-ones (2*S*,3*R*)-**5**.

Jones reagent was prepared carefully diluting a solution of CrO_3 (5g) in 5 mL of H_2SO_4 with 25 mL of water at 0 °C.



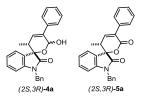
The reaction was carried out following the general procedure to furnish the crude products as a mixture of 2.2:1 diastereoisomers (2R,3R)-**4a**/(2S,3R)-**4a**; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.13 ppm (bs), δ_{minor} 6.22 ppm and 6.25 ppm (bs). (2R,3R)-**4a**/(2S,3R)-**4a** were individually isolated by chromatographic purification on silica gel (gradient from hexane/diethyl

ether 9/1 to 8/2).



The major diastereoisomer (2R,3R)-**4a** was isolated as single diastereoisomer ($R_f = 0.3$ hexane/diethyl ether 8/2) in 45% yield (white solid). The enantiomeric excess was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 4.8$ min, $\tau_{minor} = 5.7$ min. [α]_D²⁶= -59.0 (c = 1.60, CHCl₃, 99% ee). HRMS calc. for (C₂₆H₂₃NO₃+Na): 420.1576, found 420.1571.

¹H NMR (400 MHz, CDCl₃): δ 7.55-7.51 (m, 2H), 7.35-7.30 (m, 3H), 7.28-7.17 (m, 7H), 7.06 (td, 1H, J_t = 7.5 Hz, J_d = 0.9 Hz), 6.67 (d, 1H, J= 7.7 Hz), 6.06-6.04 (m, 1H), 5.85 (d, 1H, J= 12.8 Hz), 5.09 (d, 1H, J= 12.7 Hz), 5.00 (d, 1H, J= 15.5 Hz), 4.65 (d, 1H, J= 15.5 Hz), 3.03-2.96 (m, 1H), 0.87 (d, 3H, J= 7.5 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 177.9, 142.8, 137.7, 137.2, 135.4, 130.1, 129.7, 129.0, 128.6, 127.9, 127.5, 126.7, 126.6, 124.0, 123.9, 109.5, 92.1, 78.6, 44.3, 36.9, 14.8 ppm.



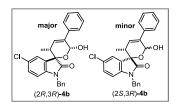
The minor (2S,3R)-**4a** isomer was isolated as a 1/1 mixture of anomers ($R_f = 0.20$ hexane/ethyl acetate 8/2) in 22% yield (white solid). The title compound was directly oxidized to the corresponding lactone (2S,3R)-**5a** using Jones reagent following the reported procedure. The corresponding lactone was obtained as a

⁸ A. Füstner, T. Nagano, J. Am. Chem. Soc., **2007**, 129, 1906-1907

single diasteroisomer and isolated after chromatography column ($R_f = 0.3$ hexane/diethyl ether 8/2) in 95% yield.

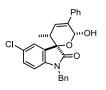
(2*S*,3*R*)-**5a.** The enantiomeric excess was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 50/50 hexane/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 5.1$ min, $\tau_{minor} = 6.1$ min. HRMS calc. for (C₂₆H₂₁NO₃+Na): 418.1419, found 418.1407.

¹H NMR (400 MHz, CDCl₃): δ 7.61-7.55 (m, 2H), 7.48-7.23 (m, 10H), 6.97 (td, 1H, $J_t = 7.7$ Hz, $J_d = 0.9$ Hz), 6.84 (d, 1H, J = 7.9 Hz), 5.02 (d, 1H, J = 15.6 Hz), 4.86 (d, 1H, J = 15.6 Hz), 3.66 (qd, 1H, $J_q = 7.2$ Hz, $J_d = 2.4$ Hz), 0.98 (d, 3H, J = 7.3 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 172.1, 162.1, 144.9, 143.4, 135.1, 134.8, 132.5, 131.2, 129.1, 128.8, 128.3, 128.3, 128.1, 127.4, 125.1, 125.0, 123.3, 110.3, 84.6, 44.3, 34.6, 14.6 ppm.



The reaction was carried out following the general procedure to furnish the crude products as a mixture of 2.3:1 diastereoisomers (2R,3R)-**4b**/(2S,3R)-**4b**; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.12 ppm (bs), δ_{minor} 6.19 ppm and 6.23 ppm (bs). (2R,3R)-**4b**/(2S,3R)-**4b** were individually isolated by chromatographic purification on silica gel (gradient from hexane/diethyl

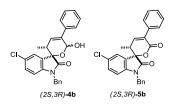
ether 9/1 to 8/2).



(2R.3R)-**4b**

(2R,3R)-4b was isolated as a single diastereoisomer ($R_f = 0.3$ hexane/diethyl ether 8/2) in 47% yield (white solid). The enantiomeric excess was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 5.5$ min, $\tau_{minor} = 6.9$ min. $[\alpha]_D^{26} = -20.7$ (c = 1.3, CHCl₃, 99% ee). HRMS calc. for ($C_{26}H_{22}NO_3Cl+Na$): 454.1186, found 454.1183.

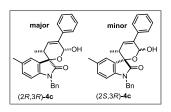
¹H NMR (400 MHz, CDCl₃): δ 7.61-7.57 (m, 2H), 7.42-7.18 (m, 11H), 6.65 (d, 1H, *J*= 8.5 Hz), 6.12-6.09 (m, 1H), 5.91 (s, 1H), 5.04 (d, 1H, *J*= 15.8 Hz), 4.69 (d, 1H, *J*= 15.8 Hz), 3.07-2.99 (m, 1H), 0.94 (d, 3H, *J*= 7.4 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 177.4, 141.2, 137.4, 137.2, 134.9, 131.4, 130.0, 129.4, 129.1, 128.6, 128.1, 127.9, 127.5, 126.7, 126.3, 124.6, 110.5, 92.1, 44.4, 36.9, 14.7 ppm.



(2S,3R)-4b was isolated as a 1/1 mixture of anomers ($R_f = 0.20$ hexane/ethyl acetate 8/2) in 17% yield (white solid). The title compound was directly oxidized to the corresponding lactone (2S,3R)-5b using the Jones reagent following the reported procedure. The corresponding lactone was obtained as a single diasteroisomer and isolated after chromatography column ($R_f = 0.3$

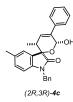
hexane/diethyl ether 8/2) in 98% yield. (2*S*,3*R*)-**5b.** The enantiomeric excess was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 40/60 hexane/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 7.3 \text{ min}$, $\tau_{minor} = 9.3 \text{ min}$. HRMS calc. for (C₂₆H₂₁NO₃+Na): 418.1419, found 418.1407.

¹H NMR (400 MHz, CDCl₃): δ 7.58-7.53 (m, 2H), 7.48-7.40 (m, 3H), 7.38-7.21 (m, 8H), 6.84 (t, 1H, J = 2.8 Hz), 6.70 (d, 1H, J = 8.4 Hz), 4.99 (d, 1H, J = 15.6 Hz), 4.85 (d, 1H, J = 15.6 Hz), 3.65 (qd, 1H, $J_q = 7.5$ Hz, $J_q = 2.9$ Hz), 1.00 (d, 3H, J = 7.3 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 171.7, 161.6, 144.6, 141.8, 134.6, 134.6, 132.6, 131.1, 129.2, 128.9, 128.7, 128.7, 128.3, 128.3, 127.4, 126.6, 125.5, 111.4, 84.3, 44.4, 34.5, 14.5 ppm.



(2R,3R)-4c and (2S,3R)-4c. The reaction was carried out following the general procedure, using 20 mol% of the catalyst, to furnish the crude products as a mixture of 3.0:1 diastereoisomers (2R,3R)-4c/(2S,3R)-4c; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.11 ppm (bs), δ_{minor} 6.19 ppm and 6.22 ppm (bs). (2R,3R)-4c/(2S,3R)-4c were individually isolated by chromatographic

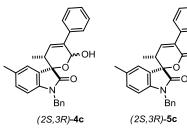
purification on silica gel (gradient from hexane/diethyl ether 9/1 to 8/2) as described below.



(2R,3R)-4c was isolated as a single diastereoisomer ($R_f = 0.3$ hexane/diethyl ether 8/2) in 63% yield (white solid). The enantiomeric excess was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215, 254$ nm: $\tau_{major} = 4.5$ min, $\tau_{minor} = 5.6$ min. $[\alpha]_D^{27} = -30.8$ (c = 1.25, CHCl₃, 99% ee). HRMS calc. for ($C_{27}H_{25}NO_3+Na$): 434.1732, found 434.1716.

¹H NMR (400 MHz, CDCl₃): δ 7.61-7.56 (m, 2H), 7.42-7.25 (m, 10H), 7.20-7.17 (m, 1H), 7.04 (d, 1H, *J*= 8.3 Hz), 6.61 (d, 1H, *J*= 8.2 Hz), 6.11-6.09 (m, 1H), 6.10 (bs, 1H), 5.89 (dq, 1H, *J*_d= 12.6 Hz, *J*_q= 1.2 Hz), 5.20 (d, 1H, *J*= 12.7 Hz), 5.04 (d, 1H, *J*= 15.6 Hz), 4.70 (d, 1H, *J*= 15.6 Hz), 4.70 (d, 1H, *J*= 15.8 Hz), 3.08-2.99 (m, 1H), 2.32 (s, 3H), 0.94 (d, 3H, *J*= 7.4 Hz) ppm.

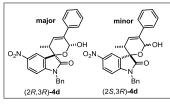
¹³C NMR (100 MHz, CDCl₃): *δ* 177.8, 140.3, 137.7, 137.2, 135.5, 133.7, 130.4, 129.7, 129.0, 128.6, 127.9, 127.8, 127.5, 126.7, 126.7, 124.7, 109.3, 92.2, 78.7, 44.3, 36.9, 21.2, 14.9 ppm.



(2S,3R)-4c was isolated as a 1/1 mixture of anomers ($R_f = 0.20$ hexane/ethyl acetate 8/2) in 17% yield (white solid). The title compound was directly oxidized to the corresponding lactone (2S,3R)-5c using the Jones reagent following the reported procedure. The corresponding lactone 5c was obtained as a single diasteroisomer and isolated after chromatography column ($R_f = 0.3$ hexane/ethyl acetate 9/1) in 94%

yield.

(2*S*,3*R*)-**5c.** The enantiomeric excess was determined to be 98% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 4.5$ min, $\tau_{minor} = 5.3$ min. [α]_D²⁶= -73.63 (c = 0.4, CHCl₃, 98% ee). ¹H NMR (400 MHz, CDCl₃): δ 7.60-7.54 (m, 2H), 7.48-7.39 (m, 3H), 7.36-7.22 (m, 5H), 7.18 (bs, 1H), 7.05 (bd, 1H, *J* = 7.8 Hz), 6.83 (d, 1H, *J* = 2.6 Hz), 6.65 (d, 1H, *J* = 8.1 Hz), 4.99 (d, 1H, *J* = 15.7 Hz), 4.83 (d, 1H, *J* = 15.7 Hz), 3.65 (qd, 1H, *J*_q = 7.4 Hz, *J*_d= 2.7 Hz), 2.63 (s, 3H), 0.98 (d, 3H, *J* = 7.4 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 172.1, 162.2, 145.1, 140.9, 135.2, 135.0, 132.9, 132.5, 131.4, 129.1, 128.8, 128.7, 128.4, 128.0, 127.4, 125.8, 125.0, 110.1, 84.8, 44.4, 34.6, 21.3, 14.6 ppm.



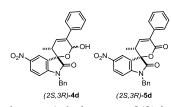
(2*R*,3*R*)-4d and (2*S*,3*R*)-4d. The reaction was carried out following the general procedure, using 20 mol % of the catalyst, to furnish the crude products as a mixture of 1.3:1 diastereoisomers (2*R*,3*R*)-4d/(2*S*,3*R*)-4d; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.14 ppm (bs), δ_{minor} 6.18 ppm and 6.22 ppm (bs). (2*R*,3*R*)-4d/(2*S*,3*R*)-4d were individually isolated by

chromatographic purification on silica gel (gradient hexane/ethyl acetate 9/1 to hexane/ethyl acetate 7/3) as described below.



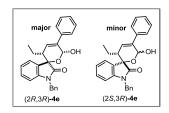
(2R,3R)-4d was isolated as mixture of diastereoisomers (18/1) (R_f = 0.3 hexane/diethyl ether 7/3) in 45% yield (white solid). The enantiomeric excess of (2R, 3R)-4d was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 6.3 \text{ min}$, $\tau_{minor} = 9.2 \text{ min}$. $[\alpha]_D^{26} = +15.4 \text{ (c} = -10.2 \text{ min})$ 0.85, CHCl₃, 99% ee). HRMS calc. for (C₂₆H₂₂N₂O₅+Na): 465.1426, found 465.1429. ¹H

NMR (400 MHz, CDCl₃): δ 8.27 (d, 1H, J= 2.4 Hz), 8.22 (dd, 1H, J₁= 8.7 Hz, J₁= 2.3 Hz), 7.60-7.55 (m, 5H), 7.43-7.26 (m, 9H), 6.82 (d, 1H, J= 8.7 Hz), 6.12-6.09 (bs, 1H), 5.93 (d, 1H, J= 12.9 Hz), 5.10 (d, 1H, J= 12.9 Hz), 5 J= 15.5 Hz), 4.77 (d, 1H, J= 15.5 Hz), 4.73 (d, 1H, J= 12.3 Hz), 3.16-3.07 (m, 1H), 0.93 (d, 3H, J= 7.4 Hz) ppm.¹³C NMR (100 MHz, CDCl₃): δ178.0, 148.3, 144.5, 137.3, 137.2, 134.2, 130.8, 129.3, 128.7, 128.5, 128.1, 127.5, 127.1, 126.7, 125.9, 120.1, 109.3, 77.9, 11.6, 36.8, 14.7 ppm.



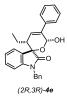
(2S,3R)-4d was isolated as a 1/1 mixture of anomers ($R_f = 0.20$ hexane/ethyl acetate 8/2) in 17% yield (white solid). The title compound was directly oxidized to the corresponding lactone (2S,3R)-5d using the Jones reagent and following the reported procedure. The corresponding lactone was obtained as a single diasteroisomer and isolated after chromatography column ($R_f = 0.3$ hexane/ethyl acetate 8/2) in 94% yield.

(2S,3R)-5d. The enantiomeric excess was determined to be 97% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 7.3$ min, $\tau_{minor} = 7.3$ minor mino 9.3 min. $[\alpha]_D^{27}$ = -122.75 (c = 0.75, CHCl₃, 97% ee). HRMS calc. for (C₂₆H₂₀N₂O₅+Na): 463.1270, found 463.1266. ¹H NMR (400 MHz, CDCl₃): δ 831 (d, 1H, J = 2.3 Hz), 8.24 (dd, 1H, J_1 = 8.6 Hz, J_2 = 2.2 Hz), 7.59-7.54 (m, 2H), 7.50-7.27 (m, 9H), 6.89 (d, 1H, J = 8.8 Hz), 6.86 (d, 1H, J = 2.9 Hz), 5.06 (d, 1H, J = 15.8 Hz), 4.93 (d, 1H, J = 15.8 Hz), 3.67 (qd, 1H, $J_q = 7.3$ Hz, $J_d = 2.0$ Hz), 1.03 (d, 3H, J = 7.5 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ172.1, 162.2, 145.1, 140.9, 135.2, 135.0, 132.9, 132.5, 131.4, 129.1, 128.8, 128.7, 128.4, 128.0, 127.4, 125.8, 125.0, 110.1, 84.8, 44.4, 34.6, 21.3, 14.6 ppm.



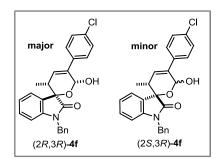
(2R,3R)-4e and (2S,3R)-4e. The reaction was carried out following the general procedure **B**, using 20 mol % of the catalyst, to furnish the crude products as a mixture of 2.2:1 diastereoisomers (2R, 3R)-4e/(2S, 3R)-4e; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.31 ppm (bs), δ_{minor} 6.38 ppm and 6.40 ppm (bs). The reaction conversion was approximately 60% after 72 hours reaction

time.

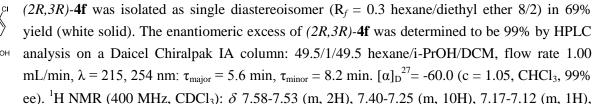


(2R,3R)-4e was isolated as a single diastereoisomer ($R_f = 0.3$ hexane/diethyl ether 8/2) in 36% yield (white solid). The enantiomeric excess of (2R, 3R)-4e was determined to be 99% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 4.8 \text{ min}$, $\tau_{minor} = 5.6 \text{ min}$. $[\alpha]_D^{26} = -51.5 \text{ (c} = 0.65, \text{ CHCl}_3, 99\%$ ee). HRMS calc. for (C₂₇H₂₅NO₃+Na): 434.1737, found 434.1716.

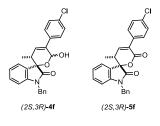
¹H NMR (400 MHz, CDCl₃): δ 7.26-7.57 (m, 2H), 7.41-7.36 (m, 3H), 7.36-7.21 (m, 7H), 7.21 (td, 1H J_t = 2.8 Hz, J_d= 0.7 Hz), 6.74 (d, 1H, J= 7.7 Hz), 6.31-6.28 (m, 1H), 5.90 (bd, 1H, J= 11.8 Hz), 5.08 (bd, 1H, J= 11.8 Hz), 5.04 (d, 1H, J= 15.6 Hz), 4.72 (d, 1H, J= 15.6 Hz), 2.80-2.75 (m, 1H), 1.35-1.24 (m, 1H), 1.16-1.06 (m, 1H), 0.92 (t, 1H, J= 7.4 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 178.1, 142.9, 137.9, 137.4, 135.3, 130.1, 129.9, 129.0, 128.6, 127.9, 127.8, 127.5, 126.7, 123.9, 123.9, 109.4, 92.1, 77.9, 44.2, 43.7, 21.5, 11.7 ppm.



(2R,3R)-4f and (2S,3R)-4f. The reaction was carried out following the general procedure, using 20 mol% of catalyst, to furnish the crude products as a mixture of 3.2:1 diastereoisomers (2R,3R)-4f/(2S,3R)-4f; d.r. determined by integration of ¹H NMR signal: δ_{major} 6.12 ppm (bs), δ_{minor} 6.22 ppm and 6.28 ppm (bs). (2R,3R)-4f/(2S,3R)-4f were individually isolated by chromatographic purification on silica gel (gradient from hexane/diethyl ether 9/1 to hexane/ethyl acetate 8/2) as described below.



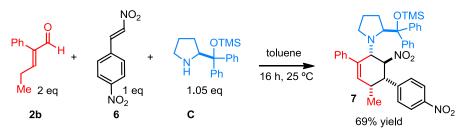
 $_{(2R,3R)-4f}$ (28). If NMR (400 MHz, CDCl₃). δ 7.58-7.55 (III, 2H), 7.40-7.25 (III, 10H), 7.17-7.12 (III, 1H), 6.77 (d, 1H, J= 7.8 Hz), 6.14-6.11 (bs, 1H), 5.86 (d, 1H, J= 12.8 Hz), 5.20 (d, 1H, J= 12.6 Hz), 5.07 (d, 1H, J= 15.5 Hz), 4.74 (d, 1H, J= 15.5 Hz), 3.10-3.39 (m, 1H), 0.94 (d, 3H, J= 7.3 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 177.9, 142.7, 136.2, 136.1, 135.3, 133.7, 130.2, 129.6, 129.0, 128.7, 128.0, 127.5, 127.0, 124.0, 123.9, 109.5, 91.9, 78.6, 44.3, 36.9, 14.8 ppm.



(2S,3R)-4f was isolated as a 1/1 mixture of anomers ($R_f = 0.20$ hexane/ethyl acetate 8/2) in 22% yield (white solid). The title compound was directly oxidized to the corresponding lactone (2S,3R)-5f using the Jones reagent and following the reported procedure. The corresponding lactone was obtained as a single diasteroisomer and isolated by chromatography column ($R_f = 0.3$ hexane/ethyl acetate 9/1) in 98% yield.

(2*S*,3*R*)-**5f.** The enantiomeric excess was determined to be 98% by HPLC analysis on a Daicel Chiralpak IA column: 49.5/1/49.5 hexane/i-PrOH/DCM, flow rate 1.00 mL/min, $\lambda = 215$, 254 nm: $\tau_{major} = 5.7$ min, $\tau_{minor} = 7.0$ min. HRMS calc. for (C₂₆H₂₀NO₃Cl+Na): 452.1017, found 452.1029. ¹H NMR (400 MHz, CDCl₃): δ 7.54-7.49 (m, 2H), 7.43-7.38 (m, 2H), 7.37-7.24 (m, 7H), 6.98 (td, 1H, $J_r = 7.7$ Hz, $J_d = 0.7$ Hz), 6.84 (d, 1H, J = 2.8 Hz), 6.78 (d, 1H, J = 7.9 Hz), 5.02 (d, 1H, J = 15.6 Hz), 4.85 (d, 1H, J = 15.6 Hz), 3.62 (td, 1H, $J_r = 7.4$ Hz, $J_d = 2.8$ Hz), 0.99(d, 3H, J = 7.4 Hz) ppm. ¹³C NMR (100 MHz, CDCl₃): δ 172.10, 161.9, 145.1, 143.4, 135.1, 135.0, 133.2, 131.5, 131.3, 129.7, 129.1, 128.9, 128.1, 127.4, 125.0, 124.9, 123.4, 110.4, 84.5, 44.4, 34.7, 14.7 ppm.

F. Synthesis of 7

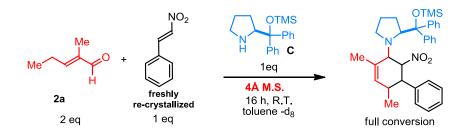


An ordinary vial equipped with a Teflon-coated stir bar and a plastic screw cap was charged with (*S*)-(–)- α , α diphenyl-2-pyrrolidinemethanol trimethylsilyl ether **C** (17.0 mg, 0.0525 mmol, 1.05 equivalents) and the α branched enal **2b** (15.9 µL, 0.1 mmol, 2 equivalents). Then toluene (100 µL), was added and the reaction stirred for 5 minutes. Finally the (*E*)-1-nitro-4-(2-nitrovinyl)benzene **6** (9.7 mg, 0.05 mmol, 1 equivalent) was added, the vial closed and stirring continued over 40 hours at 25 °C. After this time the mixture was directly charged on a preparative TLC (20x20cm) and eluted with toluene ($R_f = 0.9$ in toluene). The silica containing the compound was washed with diethyl ether and DCM and the solvent removed under vacuum to afford the pure product **7** as single diasteroisomer in 69% yield.

Compound 7 was characterized by X-ray crystallographic analysis, see page S36.

As reported in Figure 2c and discussed within the text of the main manuscript, when running the same experiment under the same reaction condition but using the enal **2a** bearing a methyl substituent (that is, mixing 2 equivalents of enal **2a**, 1.05 equivalents of amine **C** and 1 equivalent of nitrostyrene in toluene d_8), the reaction did not proceed at all.

We however found that the cyclic structure of type 7 incorporating the aminocatalyst C can form also from the α -methyl substituted enal **2a** when running the same experiment under strictly anhydrous conditions and in the presence of freshly activated molecular sieves (4Å), see Scheme S1.⁹



Scheme S1.

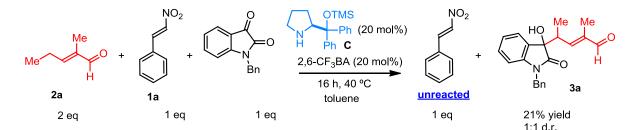
In a Schlenk tube equipped with a Teflon-coated stir bar was charged with 4Å M.S. (50 mg) and the molecular sieves activated by heating under vacuum. Than catalyst **C** (0.05mmol), nitrostyrene (0.05mmol), toluene- d_8 (100 µL) and enal **2a** (0.1mmol) were sequentially added under argon atmosphere. The mixture was stirred for 16 h at room temperature and then analyzed by ¹H NMR after filtration on 20 µm Teflon HPLC filter under argon. The crude ¹H NMR spectrum revealed complete conversion of the nitrostyrene into the product depicted in the scheme and the unreacted excess of enal **2a**.

⁹ Experiment under strictly anhydrous conditions and using a freshly re-crystallized nitrostyrene

A similar experiment has been performed adding the nitrostyrene to the preformed dienamine of 2a generated in the presence of freshly activated molecular sieves (4Å) and in toluene d_8 . In a first attempt, the addition of the nitrostyrene resulted, after 4 hours, in the complete hydrolysis of the dienamine intermediate back to the starting components, the amine C and the aldehyde 2a, without providing any trace of possible products. When repeating the experiment under strictly anhydrous conditions and using a freshly recrystallized nitrostyrene, we observed the formation of the Diels-Alder-type product of type 7 reported in Scheme S1.

The formation of cyclic adduct requires anhydrous conditions and the presence of freshly activated molecular sieves (4Å), conditions that do not reflect (are very far from) the catalytic reaction system. To provide direct evidence that the dienamine of **2a** can form in the presence of nitrostyrene but under the *actual reaction conditions*, we designed the competitive experiment described in Scheme S2. When adding two different electrophiles, such as the nitrostyrene (1 equiv) and the isatin **2a** (1 equiv), to a mixture of (*E*)-2-methylpent-2-enal **2a** (2 equiv), amine **C** (20 mol%), and the 2,6-bis (trifluoromethyl) benzoic acid (20 mol%) in toluene, only the formation of the aldol product **3a** was observed. The nitrostyrene remained totally unreacted. This indicates that, under the catalytic reaction conditions, the dienamine intermediate generated by the condensation of amine **C** and enal **2a** is formed, with the reaction exclusively channeled through the aldol pathway.

This experiment also indicates that the presence of nitrostyrene does not affect the aldol reaction. Indeed, the aldol process performed in the absence of the nitrostyrene gave very similar results (as detailed in Table 1, entry 3 of the main text).



Scheme S2. Competitive experiment

An ordinary vial equipped with a Teflon-coated stir bar and a plastic screw cap was charged with (S)-(-)- α,α -С diphenyl-2-pyrrolidinemethanol trimethylsilyl ether (0.01)mmol. 0.2 equivalents). 2.6bis(trifluromethyl)benzoic acid (0.01 mmol, 0.2 equivalents) and the α -branched enal 2a (0.1 mmol, 2 equivalents). Then toluene (200 µL) was added and the reaction stirred for 5 minutes. Finally the (E)-(2nitrovinyl)benzene (0.05 mmol, 1 equivalent) and the N-benzyl protected isatin derivative 1a (0.05 mmol, 1 equivalent) were added, the vial closed and stirring continued over 16 hours at 25 °C. Then the crude mixture was flushed through a short plug of silica gel using dichloromethane/diethyl ether 1:1 as the eluent (5 ml). Solvent was removed under reduced pressure and the crude mixture was analyzed by ¹H NMR spectroscopy to determine the diastereometric ratio. The product **3a** was isolated by flash column chromatography using the specified eluent.

G. Conformational Investigations on the Dienamine Intermediates

We focused on the conformational analysis of the covalent dienamine intermediate actively involved in the stereo-defining step. An intimate appreciation of the interactions that allow the aminocatalyst of effectively controlling the molecular topology of the dienamine intermediate may provide fundamental clues to understand and rationalize the origin of the stereoselectivity. We investigated spectroscopically the dienamine intermediate generated by direct condensation of the α -branched enals **2a** and **2b** (bearing a methyl or a phenyl α -branched substituent, respectively) with the catalyst **C**. The formation of the dienamine intermediates were achieved under anhydrous conditions (using Schlenk technique) by mixing an almost equimolar amount of the catalyst **C** (1.05 equivalent) and enals **2** (1 equivalent, 0.15 mmol) in presence of freshly activated molecular sieves (4Å) directly in deuterated solvent ([**2**]₀= 1M). After the complete disappearance of the aldehyde the reaction was filtered through a 0.2 µm PTFE filter directly into the NMR tube. After dilution (till approximately 0.2-0.3M) with the same deuterated solvent (previously anhydrified on activated molecular sieves in pellets) the sample was analyzed by NMR spectroscopy.

The dienamine adduct (II in Figure 2 of the main manuscript) has a moderate half-life (less than 1 day) depending on the presence of water in the media.

We first studied the dienamine adduct derived by aldehyde 2b and catalyst C condensation in deuterated chloroform. In this solvent two different conformers (shown in Figure S1) were found in solution with a ratio of 2.7:1.

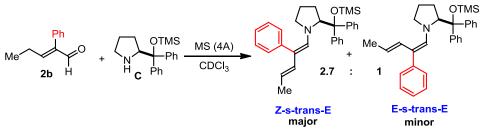


Figure S1. The dienamines from 2-phenyl-pentenal 2b and catalyst C.

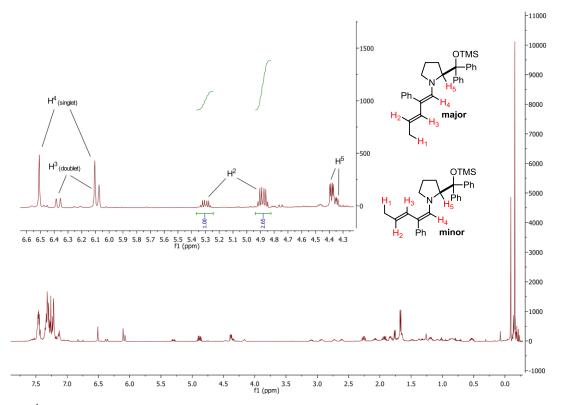


Figure S2. H¹NMR of the dienamine derived from 2b and catalyst C in CDCl₃.

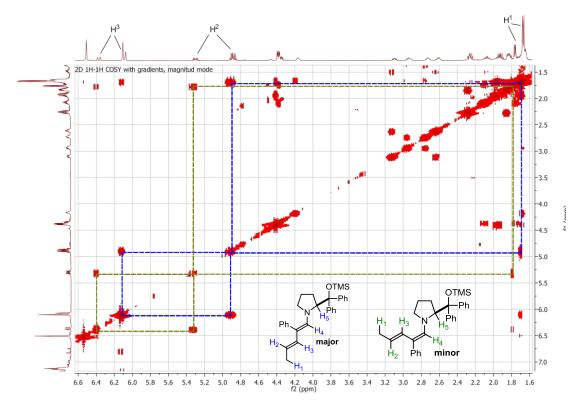


Figure S3. COSY experiment in $CDCl_3$. Comments: $H^1 H^2$ and H^3 were assigned using COSY experiment. Blue lines refer to the major conformer and green lines to the minor.

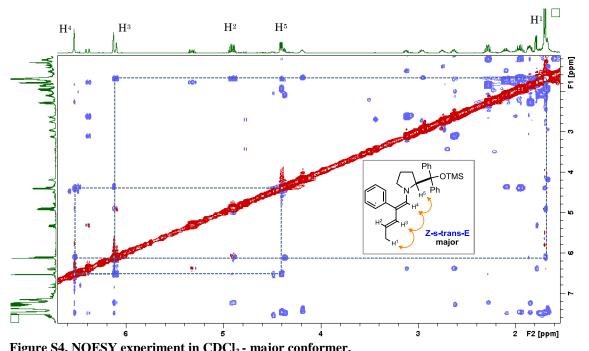


Figure S4. NOESY experiment in CDCl₃ - major conformer. Comments: the most diagnostic signals of the NOESY experiment are highlighted. Strong nuclear Overhauser effects are shown between $(H^1) - (H^3)$, $(H^3) - (H^4)$, and $(H^4) - (H^5)$, indicating a Z-s-trans-E conformation of the dienamine.

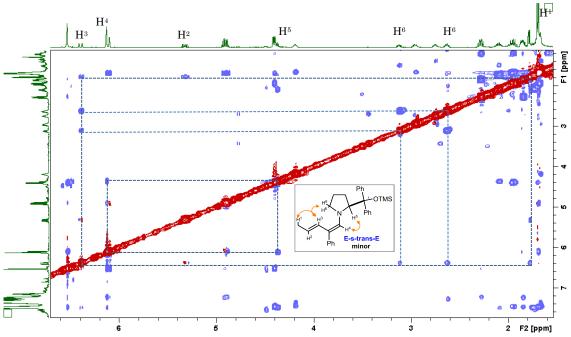


Figure 5S. NOESY experiment in CDCl₃ - minor conformer.

Comments: the most diagnostic signals of the NOESY experiment are highlighted. Strong nuclear Overhauser effects are shown between (H^1-H^3) , (H^3-H^6) , and (H^4-H^5) , indicating a E-s-trans-E conformation of the dienamine.

The NOESY experiment in deuterated chloroform revealed that the two conformers of the dienamine are both *s*-*trans*. However, different geometries of the double bond closer to the nitrogen atom can be inferred. The major isomer has a Z-configure double bond, while the minor has an *E*-configuration.

In addition, the second double bond, more distant from the nitrogen atom, is *E*-configured in both of the conformations detectable by spectroscopic analysis. In support of the results obtained in the NOESY analysis, the *J* (15.5 Hz) for protons H^3 and H^2 is identical for both conformers, clearly pointing to a relative *E* geometry.

On the basis of the spectroscopic analysis, we can conclude that the major isomer has a Z-s-trans-E conformation, while the minor isomer has an E-s-trans-E topology.

These findings are in contrast to the spectroscopic studies by Jørgensen and co-workers¹⁰ carried out on the dienamine obtained by condensation of catalyst **C** with a linear, non-substituted α , β -unsaturated aldehyde (see Figure S6). For this system, two possible conformers were detected in CDCl₃ solution, differing in the geometry of the second double bond, more distant from the nitrogen atom.

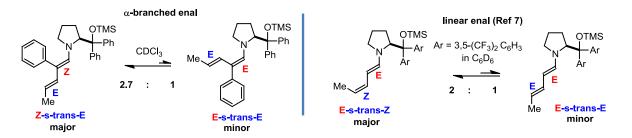


Figure S6. Comparison of the conformational behavior of the dienamines formed by condensation of α-branched and linear enals with catalyst C.

¹⁰ S. Bertelsen, M. Marigo, S. Brandes, P. Dinér, K. A. Jørgensen, J. Am. Chem. Soc. **2006**, 128, 12973–12980.

Remarkably, the detected dienamines from α -branched enal **2b** both show an exclusive *E* geometry at the remote double bond. There thus arises the interesting prospect that the α -branched enals, which are difficult substrates for enamine and iminium ion catalysis, have the structural properties (namely the α -substituent) to bias the dienamine geometry, a necessary requirement for forging a stereogenic centre at the γ position with high fidelity.

We then carried out conformational studies in toluene- d_8 , the reaction medium. In contrast to the experiments carried out in CDCl₃, the Z-*s*-trans-*E* dienamine shows a much higher stability than *E*-*s*-trans-*E* in toluene- d_8 . As shown Figure S7, almost only one conformer can be detected (ratio > 16:1).

NOESY (Figure S8), COSY experiments (Figure S9) and *J* analysis confirm that the thermodynamically most stable conformation of the dienamine derived from aldehyde 2b has a Z-s-trans-*E* geometry, the same observed for the major conformer in CDCl₃.

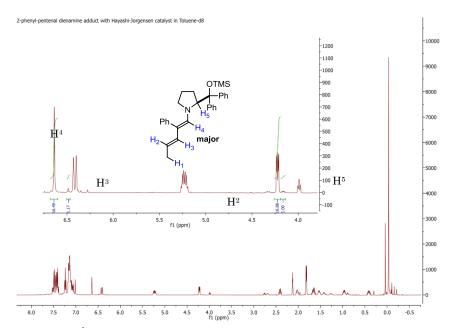


Figure S7. H¹NMR of the dienamine derived from 2b and catalyst C in toluene-d₈.

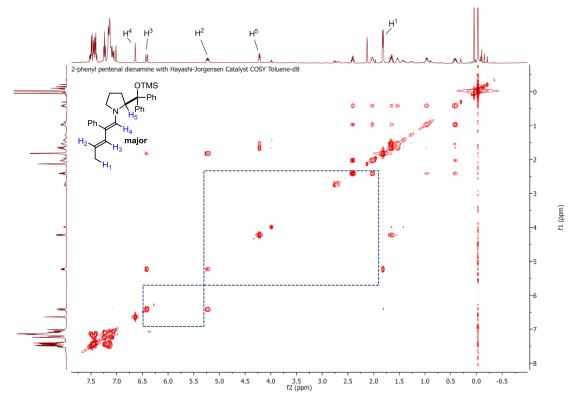


Figure S8. COSY experiment in toluene-d_{8.}

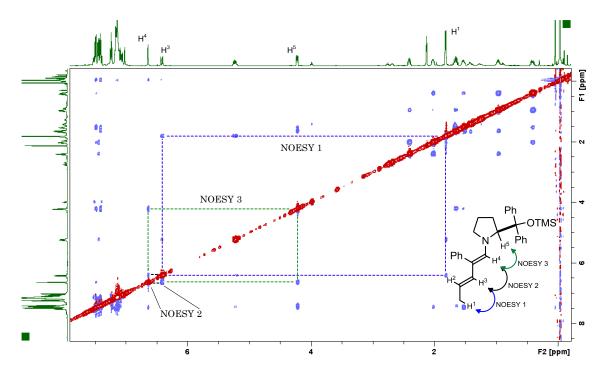


Figure S9. NOESY experiment in toluene-d_{8.}

We then studied the conformational behavior of the dienamine adduct derived from aldehyde 2a, bearing a methyl alpha-substitutent, and catalyst C condensation in deuterated toluene. The same major conformer observed in the previous case (using enal 2b) was observed, in this case in a 7:1 ratio with respect to the minor conformer (Figure S10).

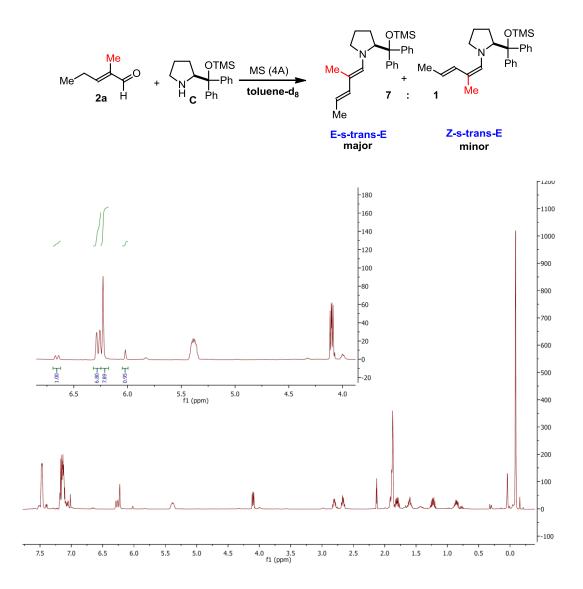


Figure S10. The dienamine from 2-methyl-pentenal 2a and catalyst C in toluene-d₈: two conformers (7:1 ratio) were detected.

The same sequence of experiments carried out for the dienamine derived from **2b** (Figures S2-9) served to establish that the dienamines derived from **2a** and **2b** ($R_{branched} = Me$ and Ph, respectively) have similar ground state thermodynamic stability: we can conclude that the nature of the α -branched substituent did not alter the conformational preference of the dienamine, being the *s*-trans dienamine with the same geometry of the two double bonds the more stable species in both the cases.¹¹

¹¹ It should be noted that different α -branched substituents (Ph vs Me) change the priority for the double bond, thereby switching the nomenclature of the first insaturation. The more stable dienamine derived from enal **2a** (R = Me) has a *E*-s-*trans-E* configuration, while when R= Ph the nomenclature change to *Z*-s-*trans-E*; still the two dienamines show the same structural topology.

H. X-ray Crystallographic Data

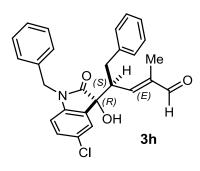
Single Crystal X-ray Diffraction Data for compound 3h

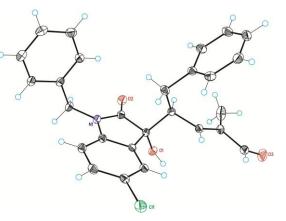
X-ray structure determinations: Crystals of compound **3h** were obtained by slow evaporation of a mixture of hexane/diethyl ether at room temperature. The measured crystals were unstable under atmosphere conditions; they were prepared under inert conditions immersed in perfluoropolyether as protecting oil for manipulation.

Data Collection. Measurements were made on a Bruker-Nonius diffractometer equipped with an APPEX 2 4K CCD area detector, a FR591 rotating anode with $Mo_{K\alpha}$ radiation, Montel mirrors and a Cryostream Plus low temperature device (T = 100K). Full-sphere data collection was used with ω and φ scans.

Programs used: Data collection Apex2 V2009.11 (Bruker-Nonius 2008), data reduction Saint + Version 7.60A (Bruker AXS 2008) and absorption correction TWINABS V. 2008-1 (2008).

Structure Solution. SIR2008 Structure Refinement. SHELXTL V6.14





Crystal data for 3h at 100 K: CCDC 885390

Empirical formula	C27 H24 Cl N O3	
Formula weight	445.92	
Temperature	100(2)K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P2(1)2(1)2(1)	
Unit cell dimensions	a = 9.784 Å	α = 90.00 °.
	b = 13.467 Å	$\beta=90.00$ °.
	c = 16.619 Å	$\gamma=~90.00$ °.
Volume	2189.7 Å ³	
Z	4	
Density (calculated)	1.353 Mg/m ³	
Absorption coefficient	0.205 mm ⁻¹	
F(000)	936	
Crystal size	$0.20 \ge 0.10 \ge 0.05 \text{ mm}^3$	
Theta range for data collection	1.95 to 37.20 °.	
Index ranges	-11 <=h<=16 ,-22 <=k<=22 ,-	28 <=l<=27
Reflections collected	29424	

Independent reflections	10964 [R(int) = 0.0447]
Completeness to theta =37.20 $^{\circ}$	0.982 %
Absorption correction	Empirical
Max. and min. transmission	0.9898 and 0.9602
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	10964 / 0 / 291
Goodness-of-fit on F ²	1.030
Final R indices [I>2sigma(I)]	R1 = 0.0416, $wR2 = 0.1008$
R indices (all data)	R1 = 0.0525, $wR2 = 0.1070$
Flack parameter	x =0.01(3)
Largest diff. peak and hole	0.409 and -0.263 e.Å ⁻³

Single Crystal X-ray Diffraction Data for compound (2R,3R)-4b

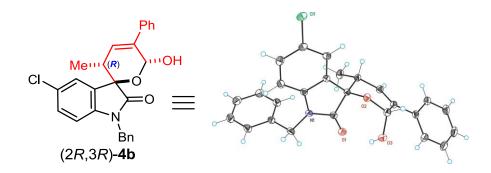
X-ray structure determinations: Crystals of compound (2R, 3R)-4b (major diastereomer) were obtained by slow evaporation of a mixture of hexane/diethyl ether at room temperature. The measured crystals were unstable under atmosphere conditions; they were prepared under inert conditions immersed in perfluoropolyether as protecting oil for manipulation.

Data Collection. Measurements were made on a Bruker-Nonius diffractometer equipped with an APPEX 2 4K CCD area detector, a FR591 rotating anode with $Mo_{K\alpha}$ radiation, Montel mirrors and a Cryostream Plus low temperature device (T = 100K). Full-sphere data collection was used with ω and φ scans.

Programs used: Data collection Apex2 V2009.11 (Bruker-Nonius 2008), data reduction Saint + Version 7.60A (Bruker AXS 2008) and absorption correction TWINABS V. 2008-1 (2008).

Structure Solution. SIR2008

Structure Refinement. SHELXTL V6.14



Crystal data for (2R,3R)-4b at 100 K: CCDC 885391

Empirical formula	C26 H22 Cl N O3	
Formula weight	431.90	
Temperature	100(2)K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	C2	
Unit cell dimensions	a = 26.891(4) Å	$\alpha = ~90.00$ °.

	b = 6.5139(9) Å c = 13.524(2) Å	$\beta = 117.976(7)$ °. $\gamma = 90.00$ °.
Volume	2092.1(5) Å ³	
Z	4	
Density (calculated)	1.371 Mg/m ³	
Absorption coefficient	0.212 mm ⁻¹	
F(000)	904	
Crystal size	$0.20 \ x \ 0.05 \ x \ 0.05 \ mm^3$	
Theta range for data collection	1.70 to 34.54 $^\circ.$	
Index ranges	-40 <=h<=32 ,-9 <=k<=9 ,-21	<=l<=21
Reflections collected	12374	
Independent reflections	6476 [R(int) = 0.0568]	
Completeness to theta =34.54 $^{\circ}$	0.869 %	
Absorption correction	Empirical	
Max. and min. transmission	0.9895 and 0.9589	
Refinement method	Full-matrix least-squares on F	72
Data / restraints / parameters	6476 / 7 / 292	
Goodness-of-fit on F ²	1.051	
Final R indices [I>2sigma(I)]	R1 = 0.0441, $wR2 = 0.1027$	
R indices (all data)	R1 = 0.0635, $wR2 = 0.1092$	
Flack parameter	x = -0.02(4)	
Largest diff. peak and hole	0.374 and -0.264 e.Å ⁻³	

Single Crystal X-ray Diffraction Data for compound (2S,3R)-5b

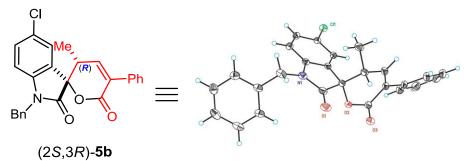
X-ray structure determinations: Crystals of compound (2S,3R)-5b were obtained by slow evaporation of a mixture of hexane/diethyl ether at room temperature. The measured crystals were unstable under atmosphere conditions; they were prepared under inert conditions immersed in perfluoropolyether as protecting oil for manipulation.

Data Collection. Measurements were made on a Bruker-Nonius diffractometer equipped with an APPEX 2 4K CCD area detector, a FR591 rotating anode with $Mo_{K\alpha}$ radiation, Montel mirrors and a Cryostream Plus low temperature device (T = 100K). Full-sphere data collection was used with ω and φ scans.

Programs used: Data collection Apex2 V2009.11 (Bruker-Nonius 2008), data reduction Saint + Version 7.60A (Bruker AXS 2008) and absorption correction TWINABS V. 2008-1 (2008).

Structure Solution. SIR2008

Structure Refinement. SHELXTL V6.14

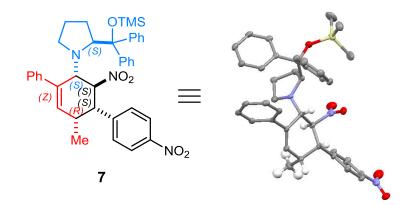


Crystal data for (2*S*,3*R*)-**5b** at 100 K: **CCDC 885392**

Empirical formula	C26 H20 Cl N O3
Formula weight	429.88
Temperature	100(2)K
Wavelength	0.71073 Å
Crystal system	Orthorhombic
Space group	P2(1)2(1)2(1)
Unit cell dimensions	$a = 6.9520(8) \text{ Å}$ $\alpha = 90.00 ^{\circ}.$
	$b = 12.8684(13) \text{ Å} \qquad \beta = 90.00 ^{\circ}.$
	$c = 23.405(3) \text{ Å}$ $\gamma = 90.00 ^{\circ}.$
Volume	2093.9(4) Å ³
Z	4
Density (calculated)	$1.364 Mg/m^3$
Absorption coefficient	0.211 mm ⁻¹
F(000)	896
Crystal size	0.20 x 0.20 x 0.20 mm ³
Theta range for data collection	1.81 to 36.54 °.
Index ranges	-9 <=h<=11 ,-20 <=k<=19 ,-29 <=l<=38
Reflections collected	16479
Independent reflections	9299 [R(int) = 0.0215]
Completeness to theta =36.54 $^{\circ}$	0.932 %
Absorption correction	Empirical
Max. and min. transmission	0.9589 and 0.9589
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	9299 / 0 / 281
Goodness-of-fit on F ²	1.026
Final R indices [I>2sigma(I)]	R1 = 0.0385, $wR2 = 0.0976$
R indices (all data)	R1 = 0.0442, $wR2 = 0.1016$
Flack parameter	x = -0.01(4)
Largest diff. peak and hole	0.477 and -0.233 e.Å ⁻³

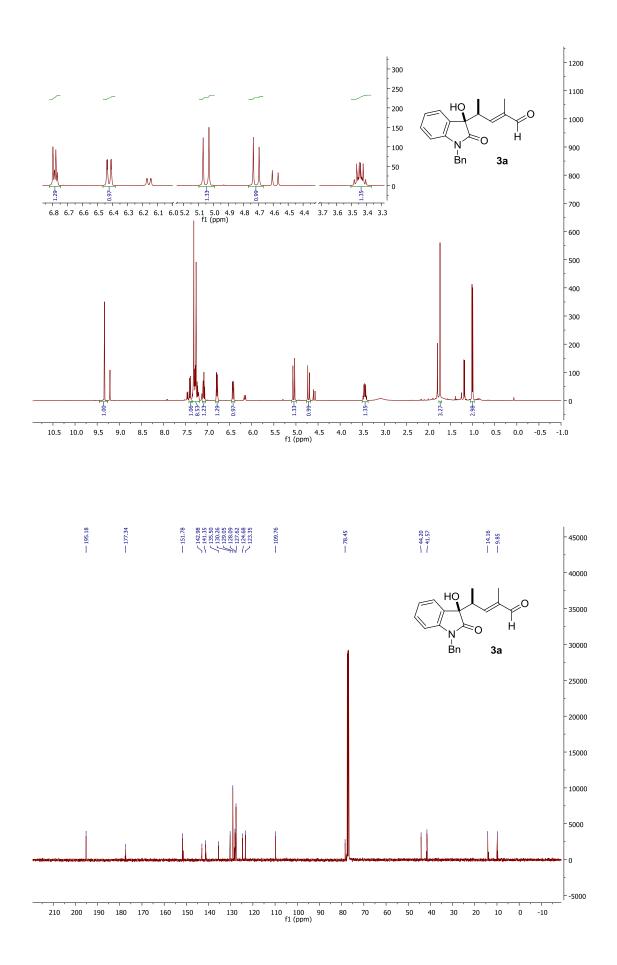
Single Crystal X-ray Diffraction Data for compound 7

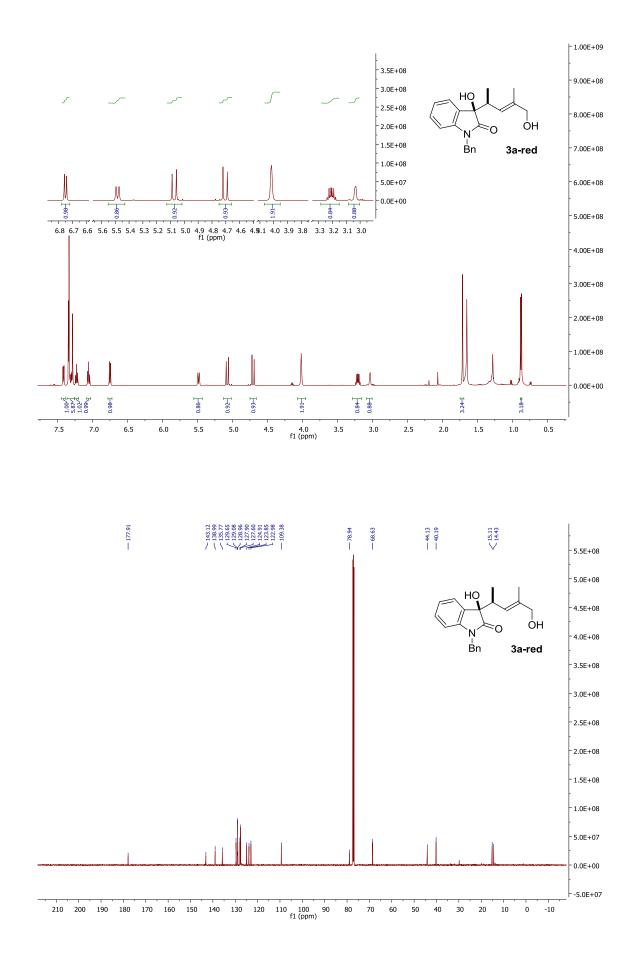
X-ray structure determinations: Crystals of compound **7** were obtained by slow evaporation of hexane at room temperature.

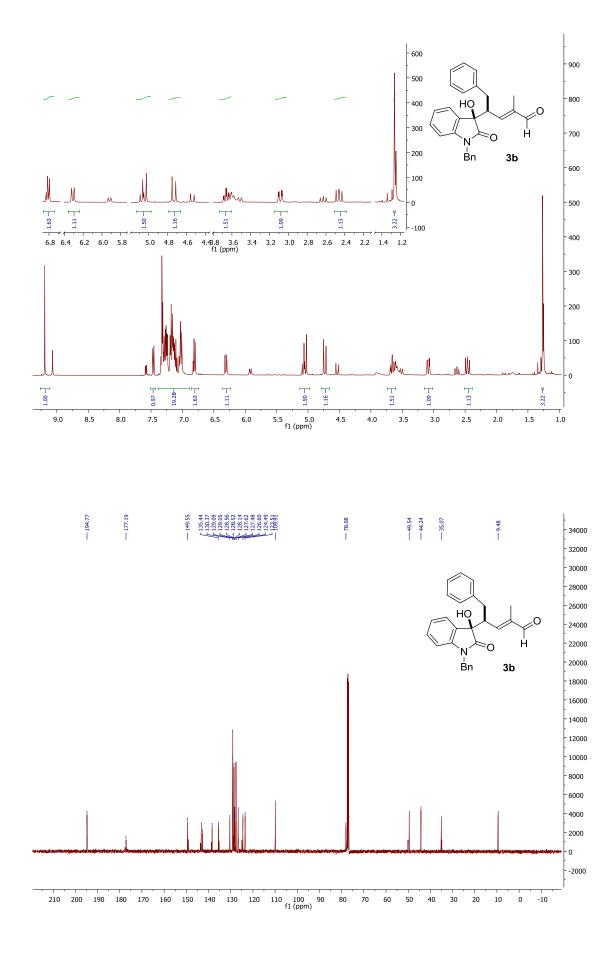


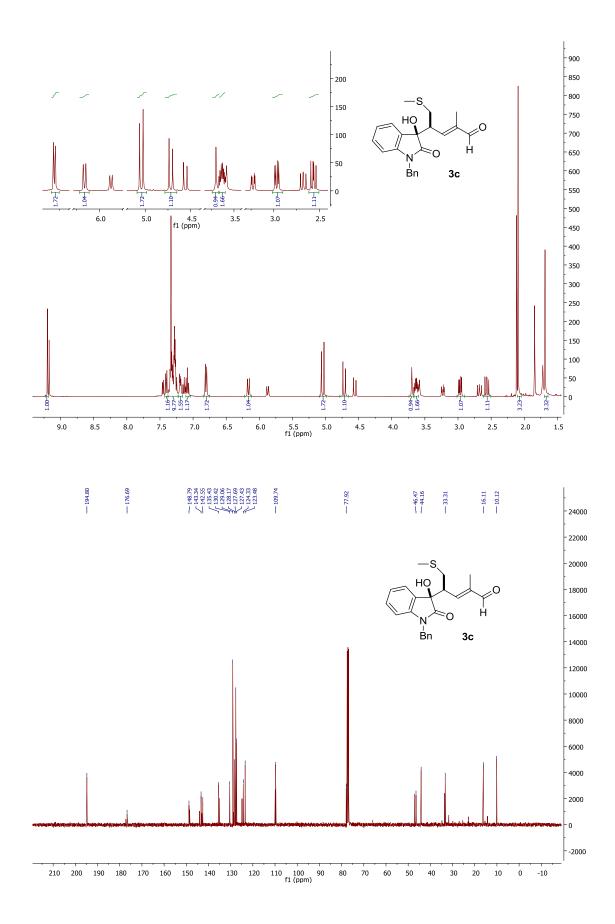
Crystal data for **7** at 100 K: CCDC 885695

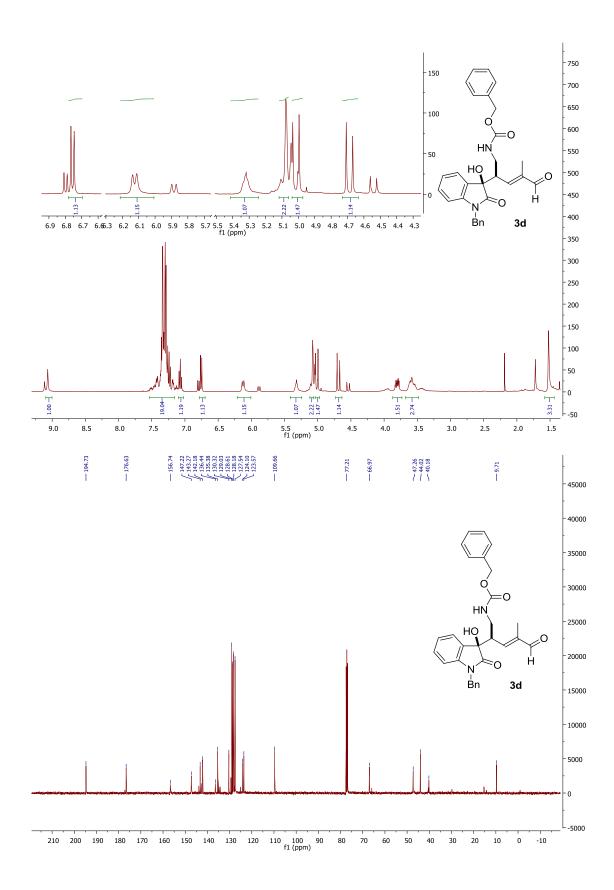
Empirical formula	C39 H43 N3 O5 Si	
Formula weight	661.85	
Temperature	100(2)K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2(1)	
Unit cell dimensions	a = 10.6855(16) Å	α = 90.00 °.
	b = 9.3798(15) Å	$\beta = 99.839(5)$ °.
	c = 18.076(3) Å	$\gamma=~90.00~^{\circ}.$
Volume	1785.1(5) Å ³	
Z	2	
Density (calculated)	1.231 Mg/m ³	
Absorption coefficient	0.113 mm ⁻¹	
F(000)	704	
Crystal size	0.30 x 0.15 x 0.01 mm ³	
Theta range for data collection	1.14 to 26.26 °.	
Index ranges	-13 <=h<=10 ,-9 <=k<=11 ,-2	1 <=l<=22
Reflections collected	15425	
Independent reflections	6511 [R(int) = 0.0493]	
Completeness to theta =26.26 $^{\circ}$	0.989 %	
Absorption correction	Empirical	
Max. and min. transmission	0.9989 and 0.9670	
Refinement method	Full-matrix least-squares on F	72
Data / restraints / parameters	6511 / 1 / 437	
Goodness-of-fit on F ²	1.024	
Final R indices [I>2sigma(I)]	R1 = 0.0539, w $R2 = 0.1244$	
R indices (all data)	R1 = 0.0737, $wR2 = 0.1394$	
Flack parameter	x =0.09(17)	
Largest diff. peak and hole	0.500 and -0.325 e.Å ⁻³	

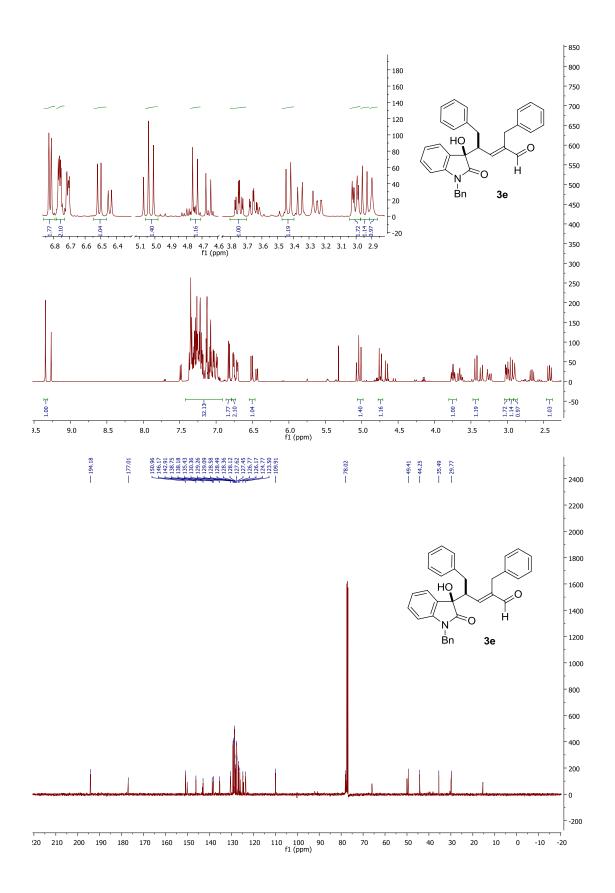


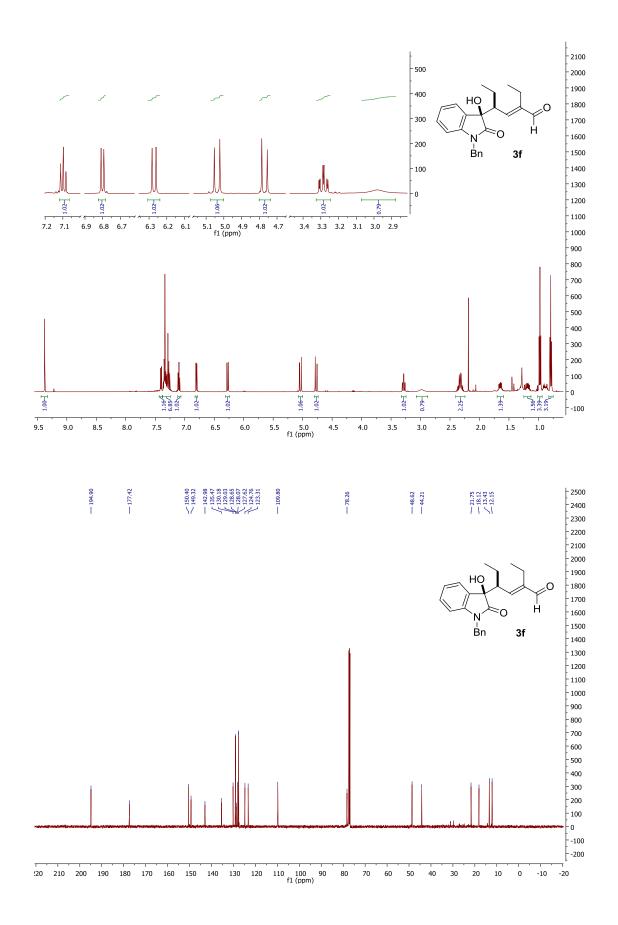


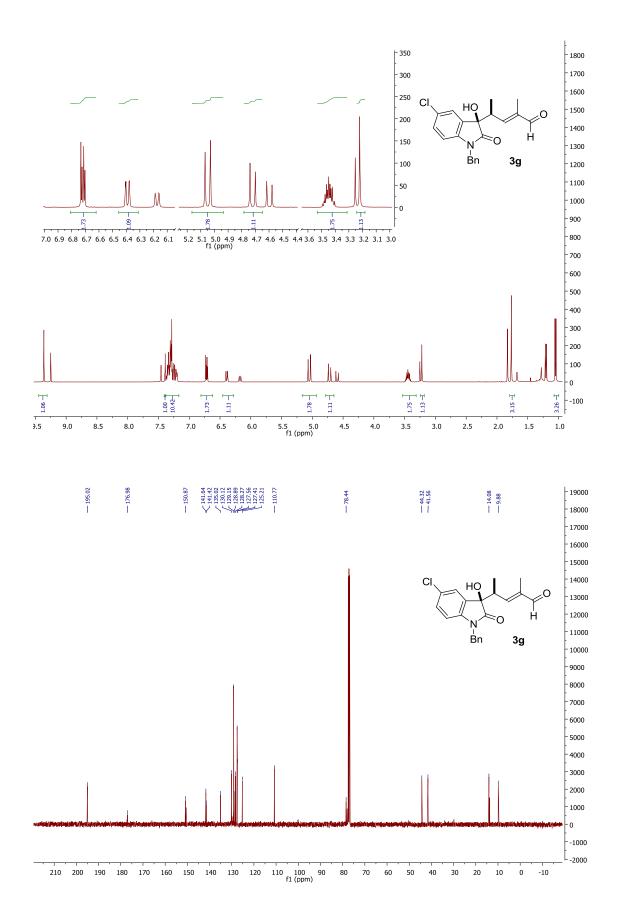


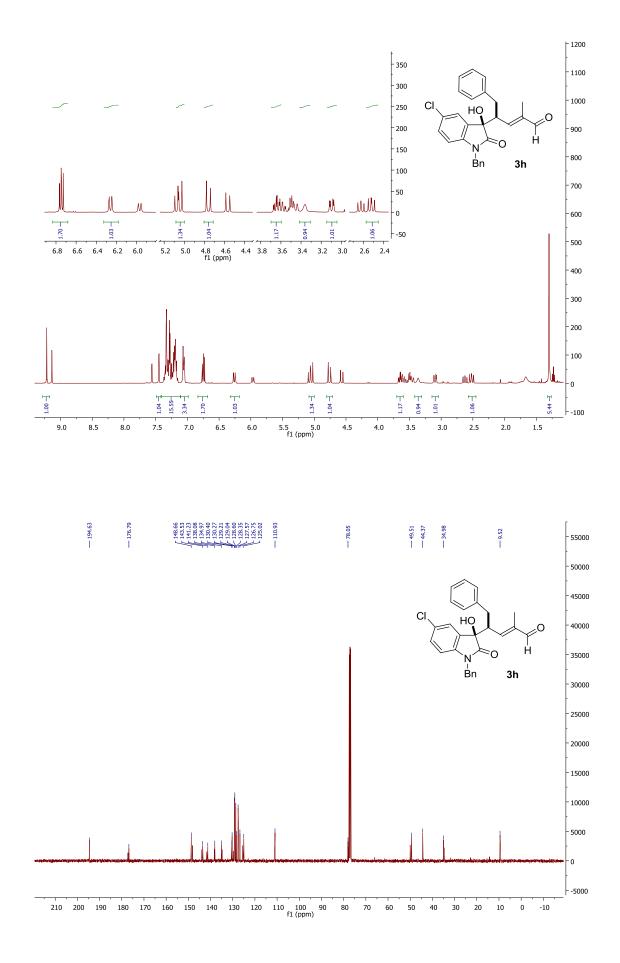


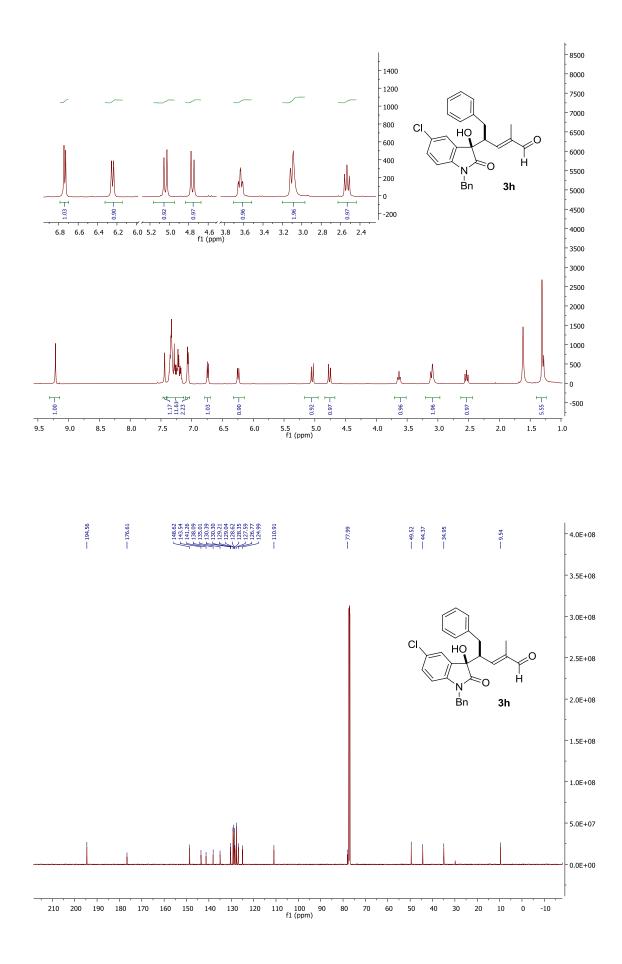




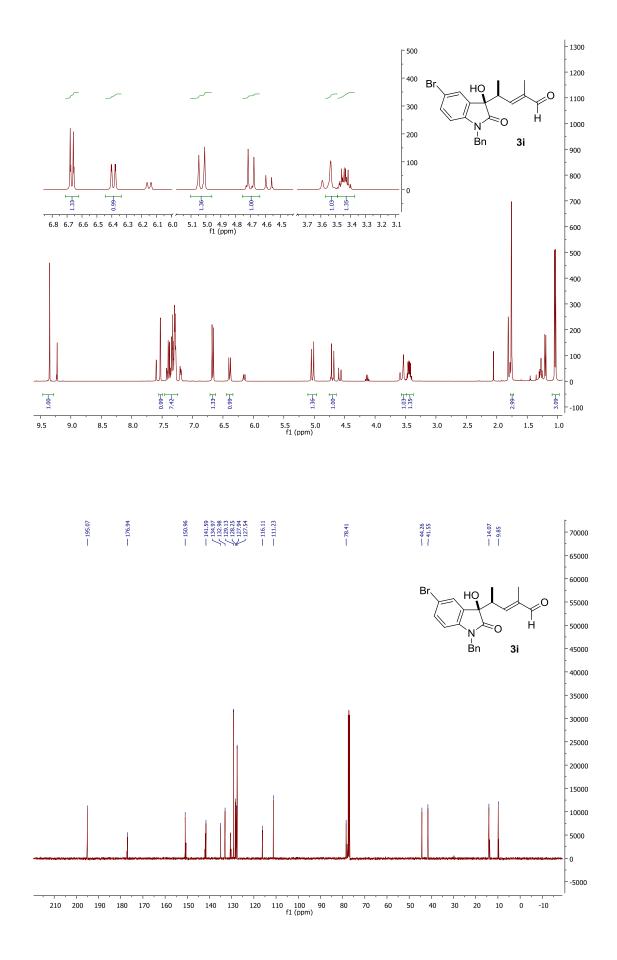


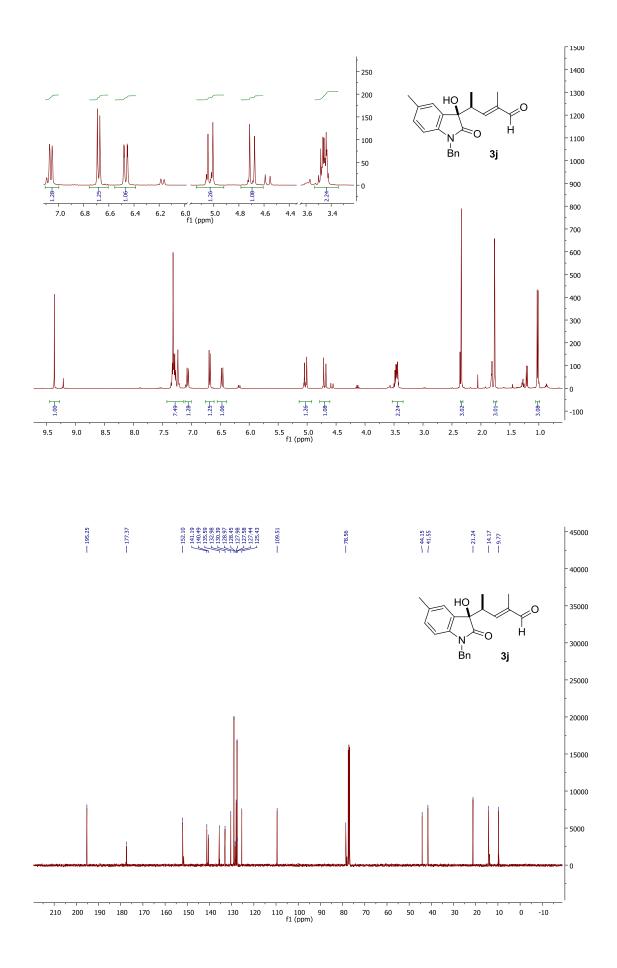


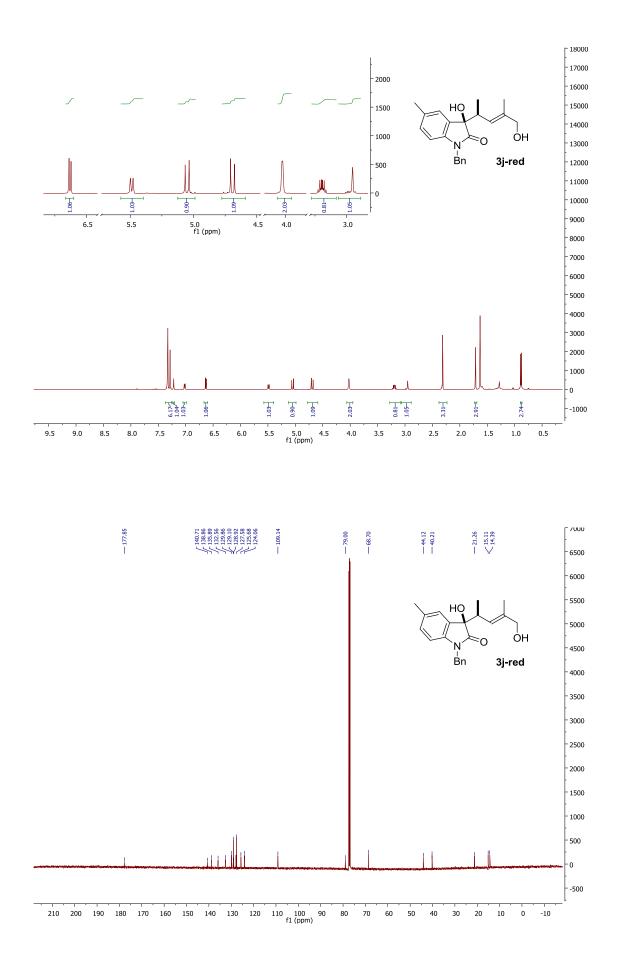


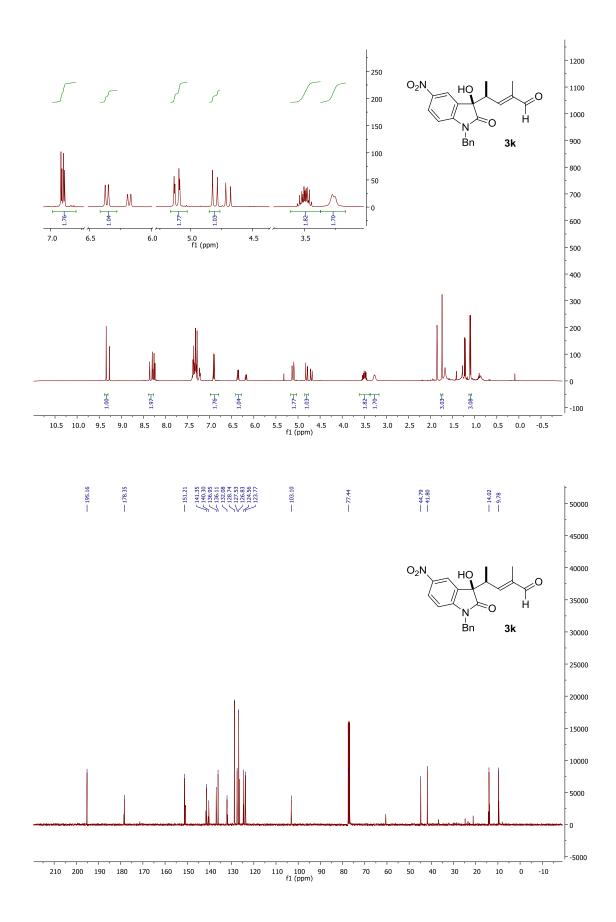


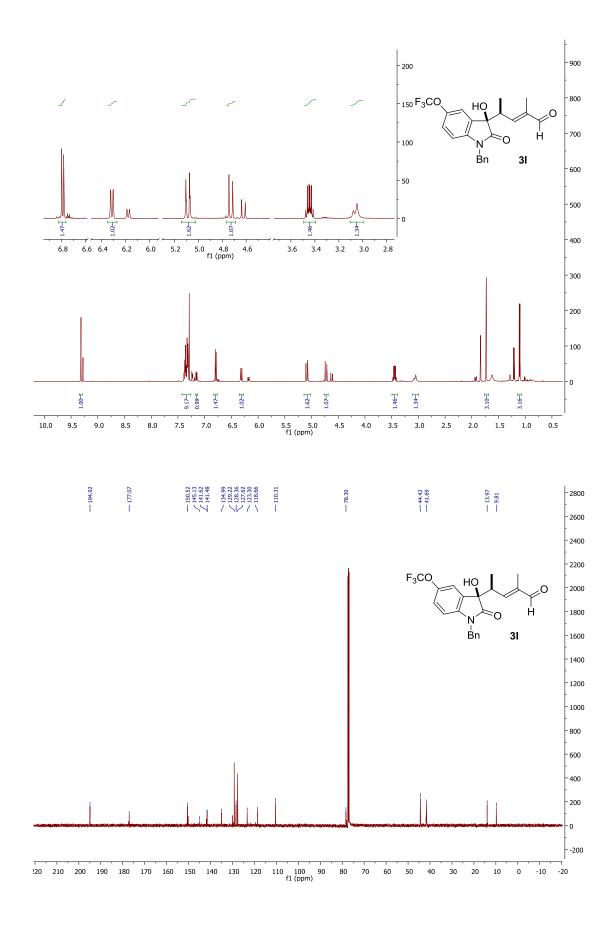
S46

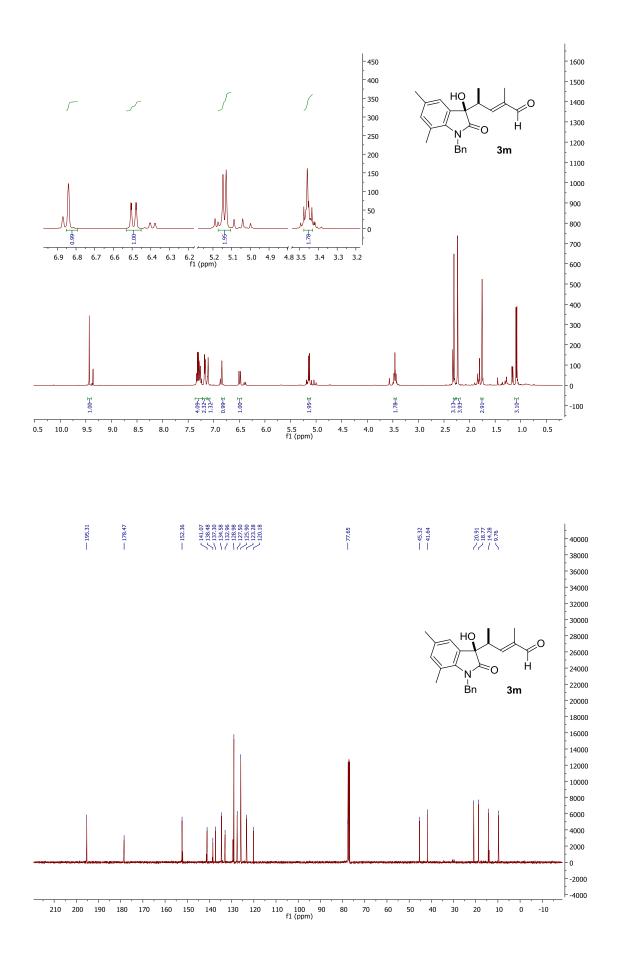


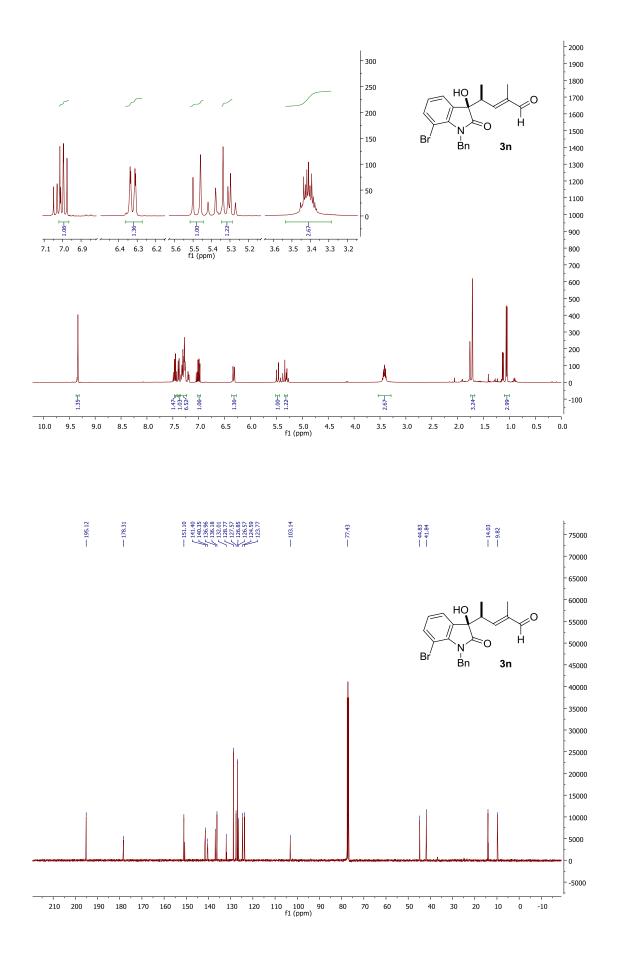


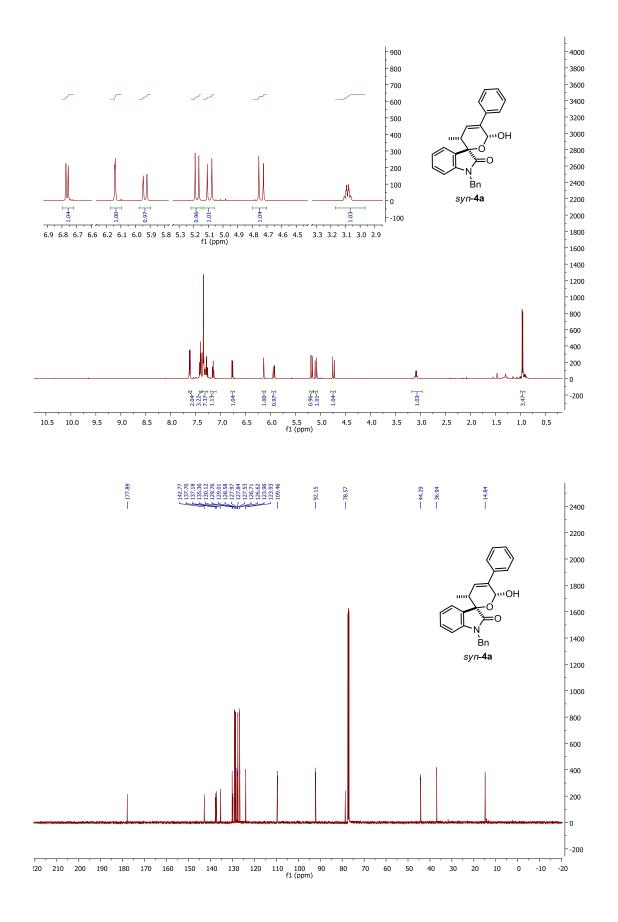


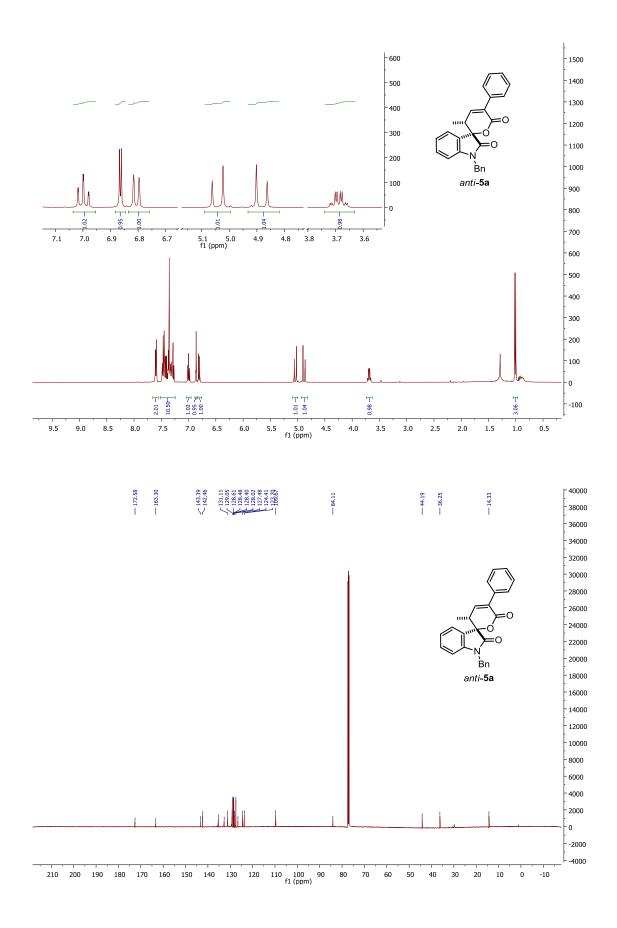


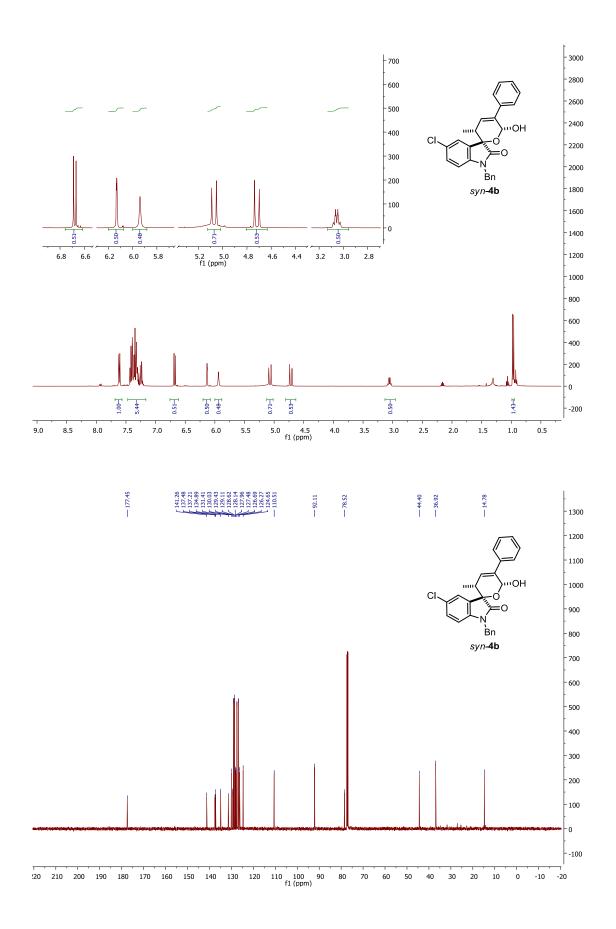


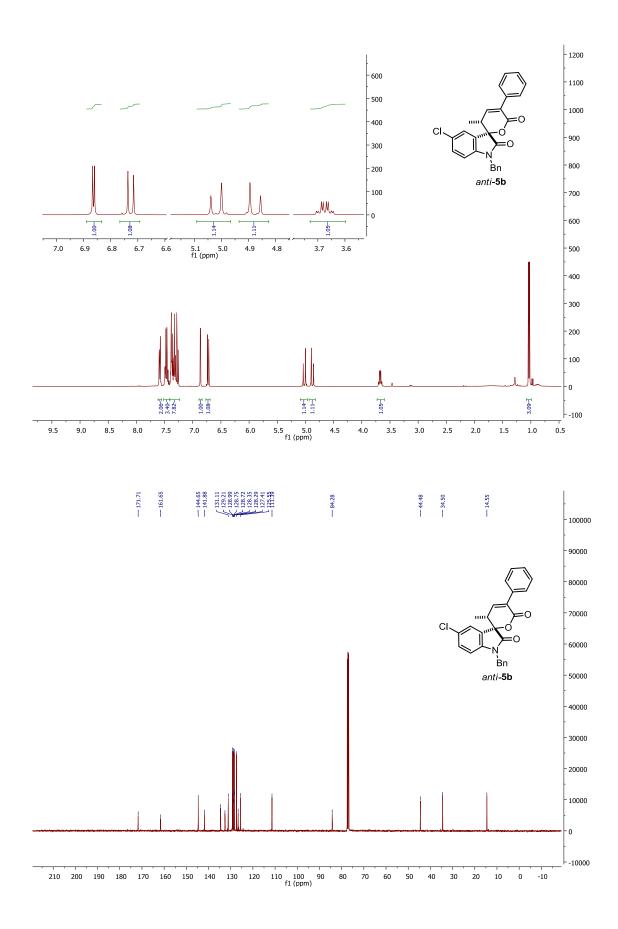


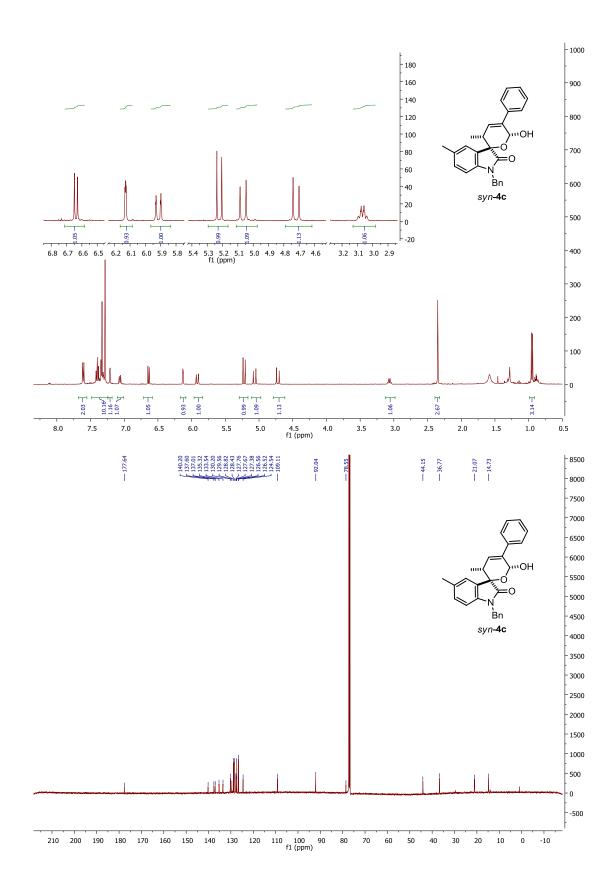


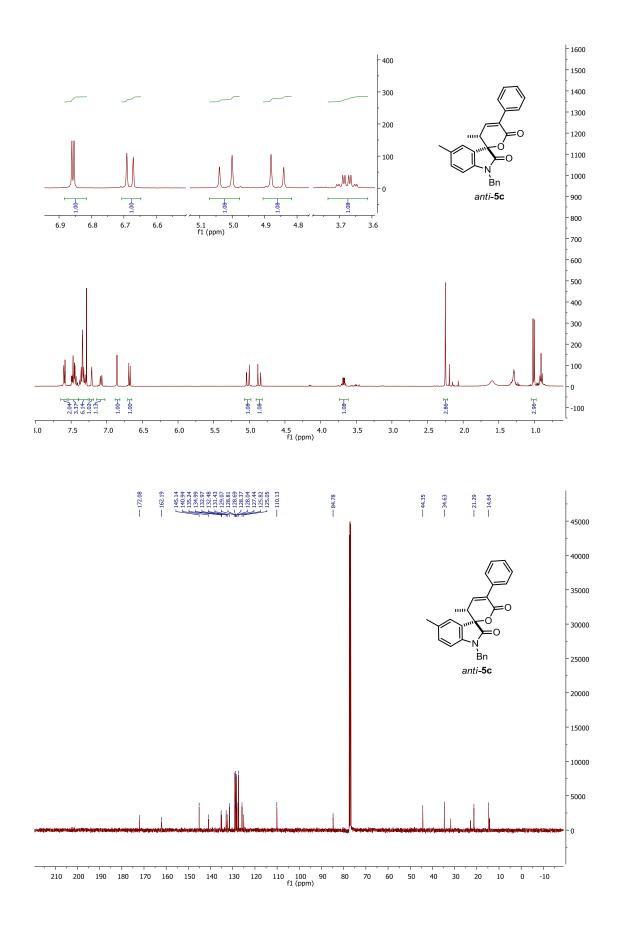


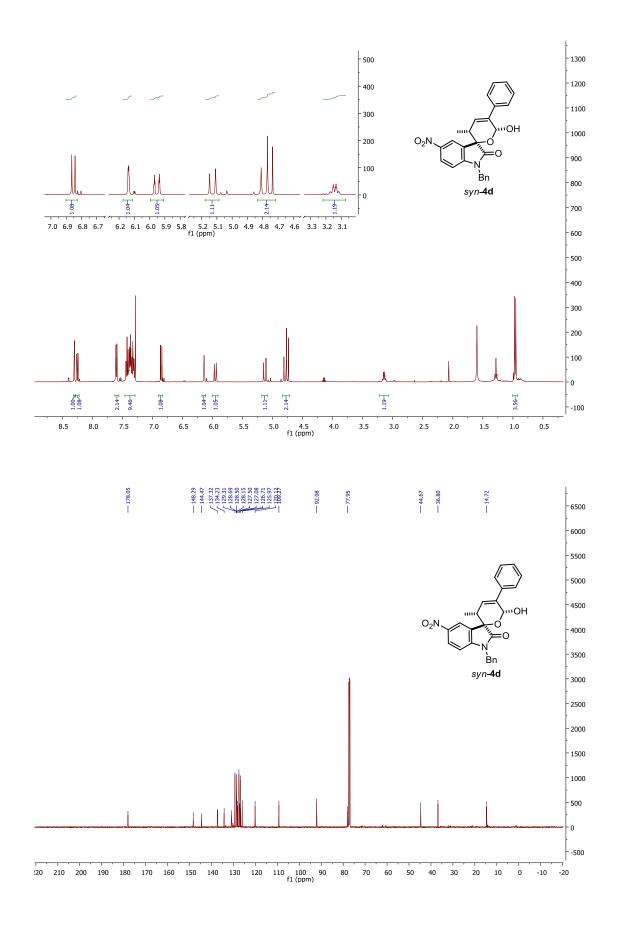


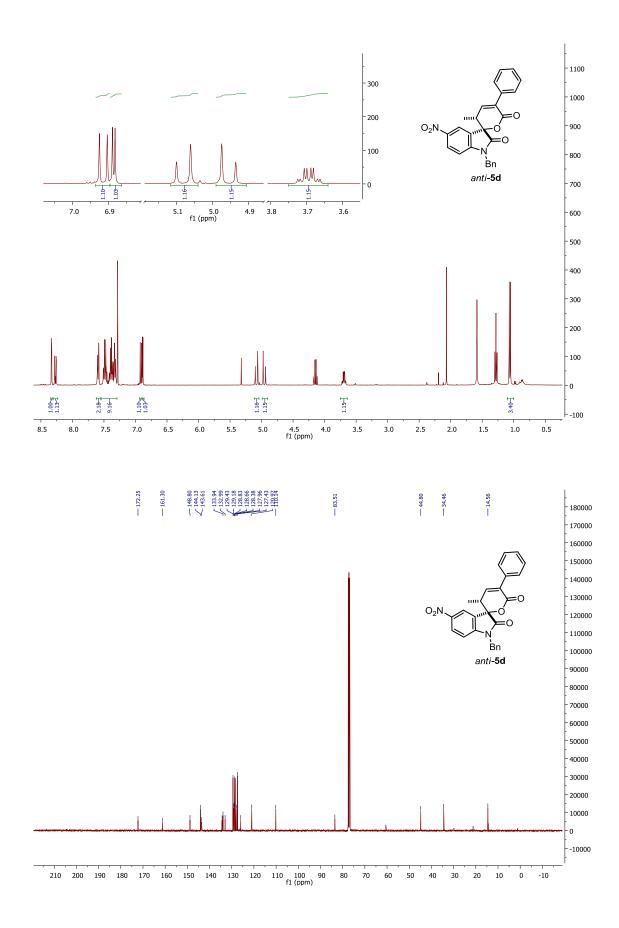


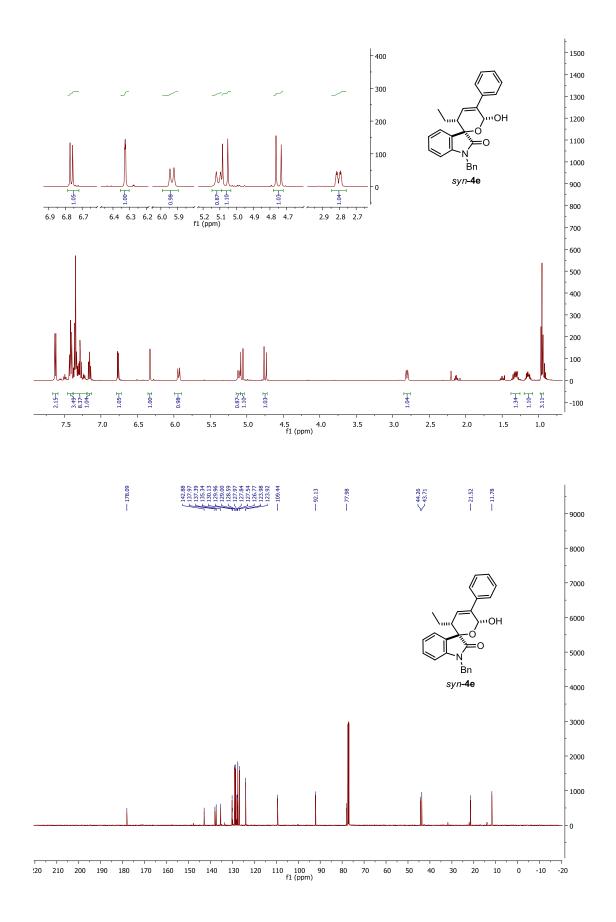


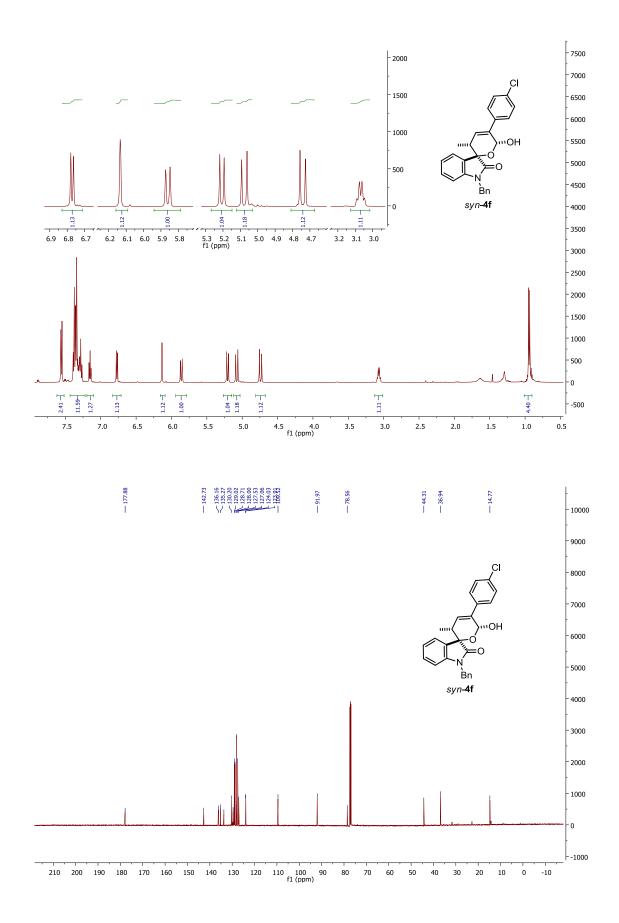


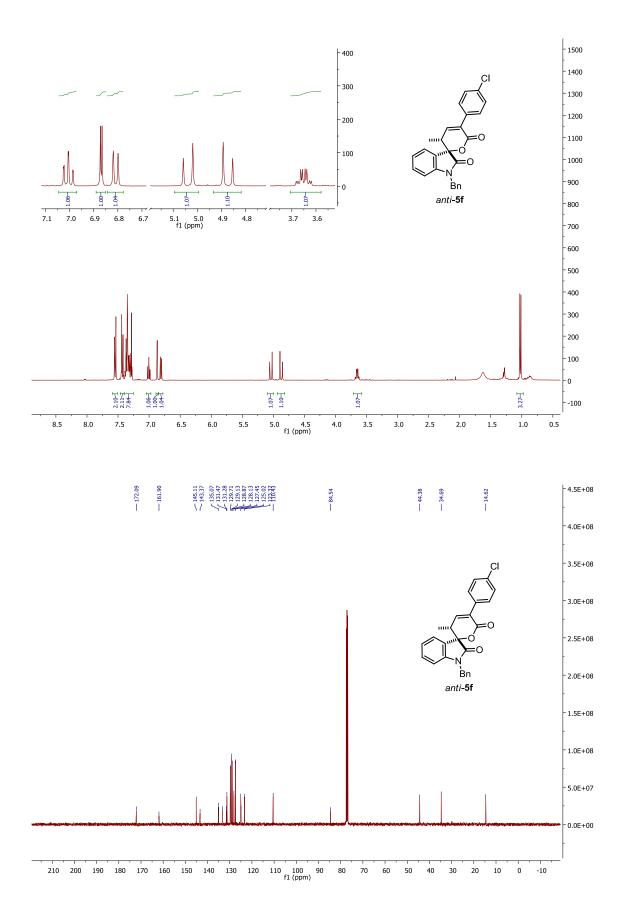


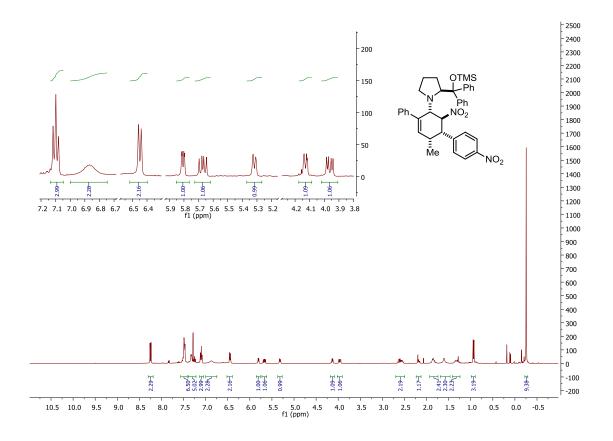


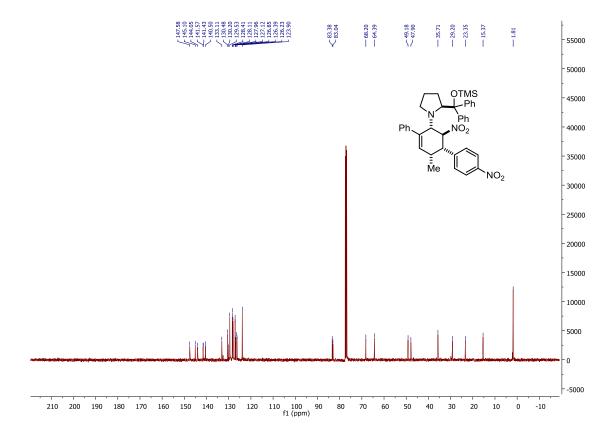


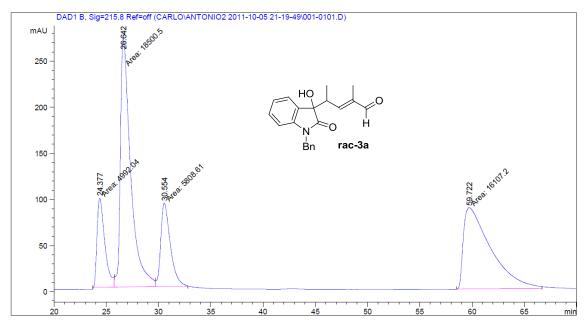




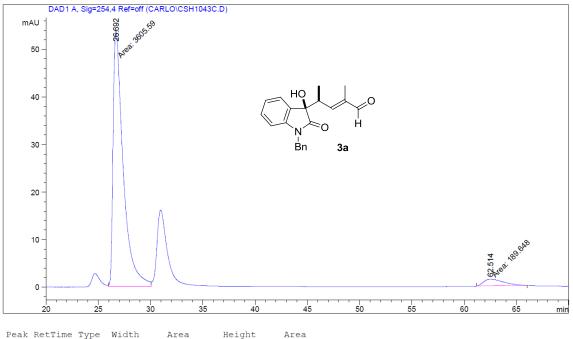




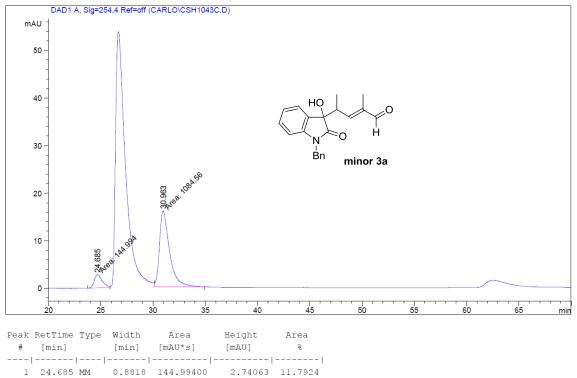




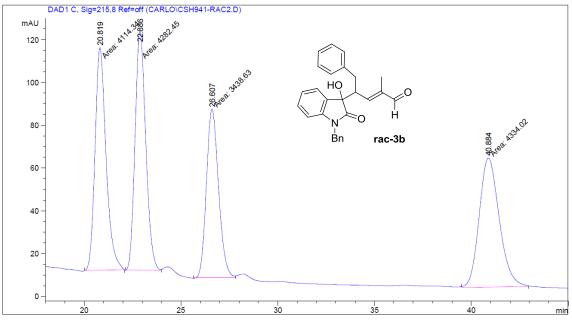
Peak RetTime # [min]	Type Widt [min		Height [mAU]	Area %
			-	
1 24.377	MF 0.80	624 4992.03955	5 96.47308	10.9937
2 26.642	FM 1.14	435 1.85005e4	269.65146	40.7425
3 30.554	FM 1.00	690 5808.60791	90.56314	12.7919
4 59.722	MM 3.04	434 1.61072e4	88.20726	35.4719



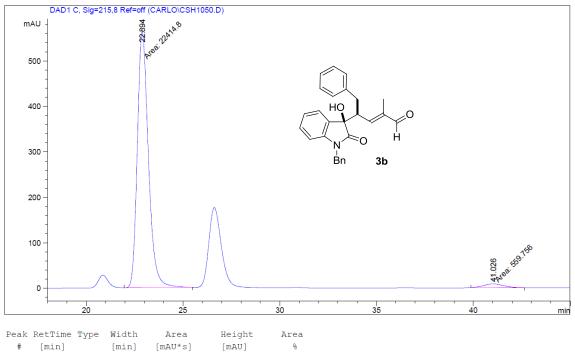
Реак	RetTime	туре	Width	Area	Height	Area	
#	[min]		[min]	[mAU*s]	[mAU]	8	
1	26.692	MM	1.1182	3605.58667	53.74288	95.0030	
2	62.514	MM	2.2273	189.64813	1.41914	4.9970	



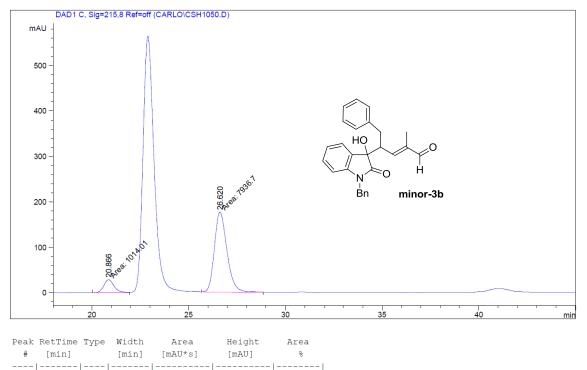
1	24.685	MM	0.8818	144.99400	2.74063	11.7924
2	30.963	MM	1.1325	1084.55872	15.96165	88.2076



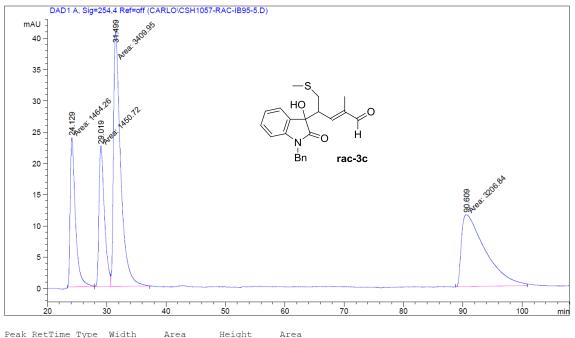
				Area [mAU*s]	~	Area %
		-				
1	20.819	MM	0.6586	4114.34473	104.11303	25.4452
2	22.886	MM	0.6390	4282.44678	111.70278	26.4848
3	26.607	MM	0.7274	3438.63062	78.78328	21.2662
4	40.884	MM	1.1975	4334.02393	60.32115	26.8038



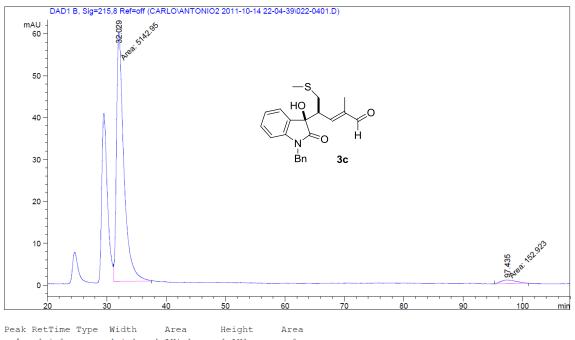
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1	22.894	MM	0.6634	2.24148e4	563.15527	97.5636	
2	41.026	MM	1.1516	559.75647	8.10107	2.4364	



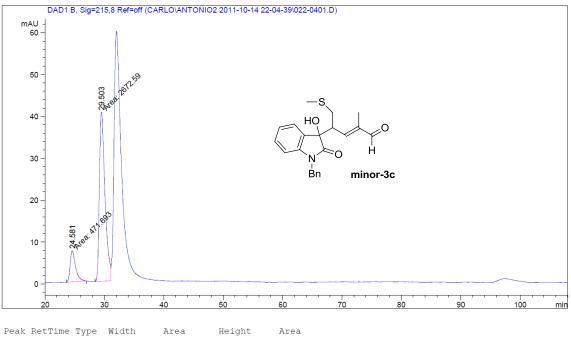
1	20.866	MM	0.5989	1014.01294	28.21827	11.3288
2	26.620	MM	0.7513	7936.70313	176.06934	88.6712



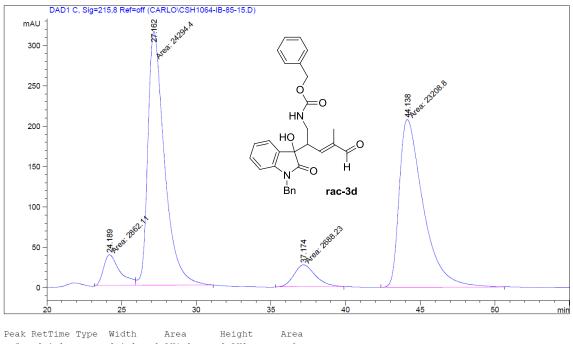
	10011110	-100		112.000	11019110	112 0 00	
#	[min]		[min]	[mAU*s]	[mAU]	8	
	·						
1	24.129	MF	1.0219	1464.25732	23.88078	15.3619	
2	29.019	MF	1.0747	1450.72424	22.49841	15.2199	
З	31.499	FM	1.3968	3409.94678	40.68789	35.7745	
4	90.609	MM	4.6529	3206.84326	11.48697	33.6437	



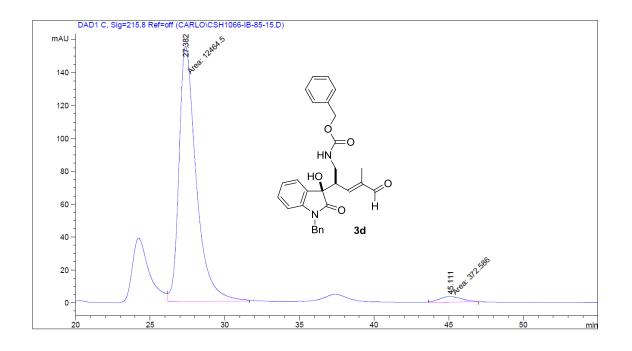
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1	32.029	MM	1.4429	5142.94971	59.40366	97.1124
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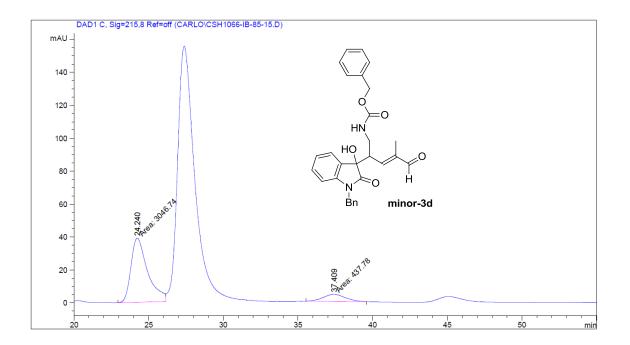
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1	24.581	MM	1.0589	471.69299	7.42394	15.0016
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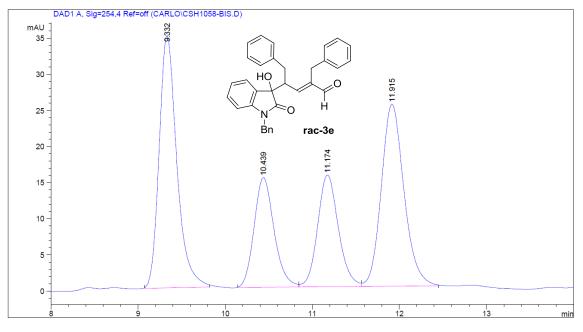
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		-				
1	24.189	MF	1.2616	2862.11279	37.80940	5.3948
2	27.162	FM	1.2902	2.42944e4	313.82260	45.7922
3	37.174	MM	1.6591	2688.23047	27.00568	5.0670
4	44.138	MM	1.8643	2.32088e4	207.48540	43.7460



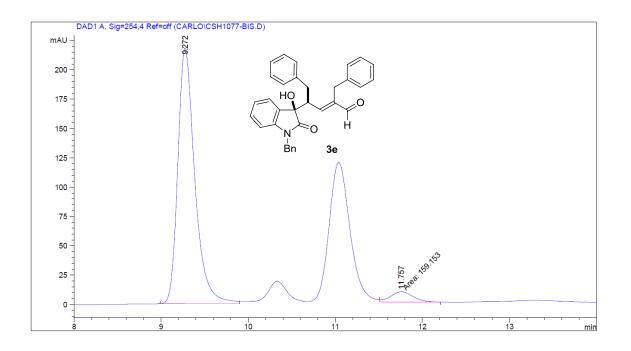
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1 27.382	MM 1.3397	1.24645e4	155.06746	97.0976
2 45.111	MM 1.7626	372.58612	3.52305	2.9024



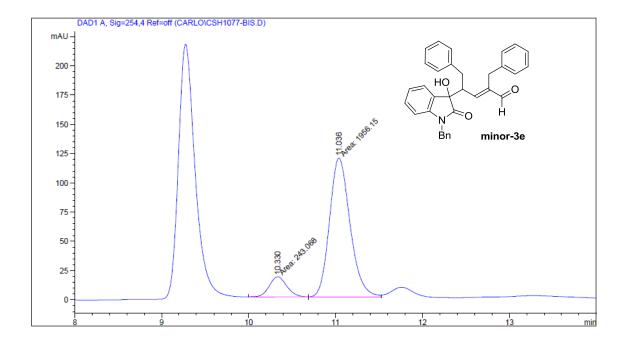
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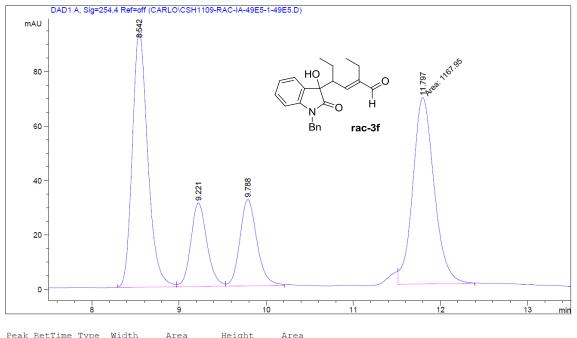
			Width		Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
		-	-			
1	9.332	BB	0.2143	492.16507	34.93966	34.0602
2	10.439	BV	0.2437	241.45648	15.16767	16.7100
3	11.174	VV	0.2548	255.67877	15.46775	17.6942
4	11.915	VB	0.2771	455.68457	25.18669	31.5356



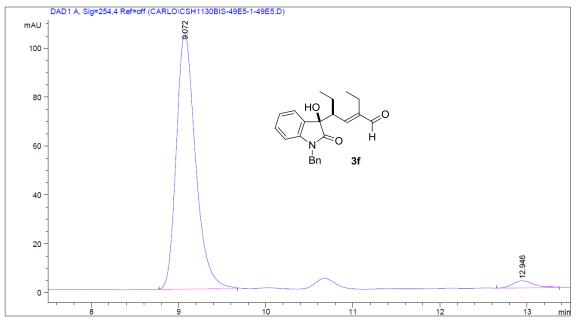
Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	9.272	BB	0.2041	2952.27417	217.92471	94.8849
2	11.757	MM	0.2996	159.15332	8.85478	5.1151



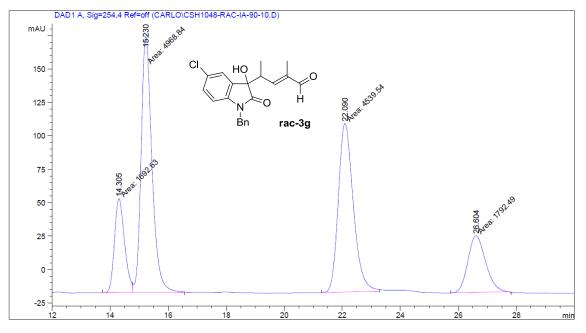
Peak RetTime # [min]			Area [mAU*s]	Height [mAU]	Area %
			-	-	
1 10.330	MF C	.2341	243.06848	17.30475	11.0525
2 11.036	FM C	.2745	1956.15308	118.75937	88.9475



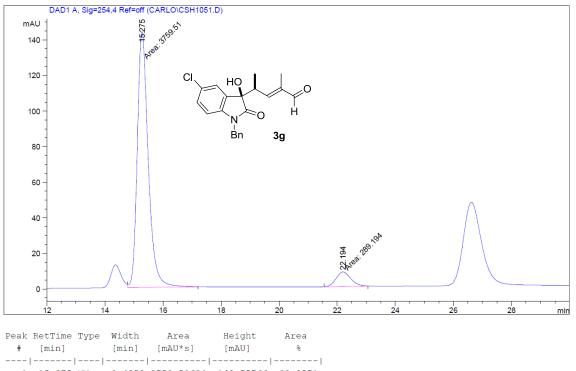
Реак	RetTime	туре	Wiath	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	8.542	BV	0.1879	1159.97192	94.01902	36.7376
2	9.221	VV	0.1960	395.64096	30.76184	12.5304
3	9.788	VB	0.2092	433.88327	31.79554	13.7416
4	11.797	MM	0.2840	1167.95422	68.54762	36.9904

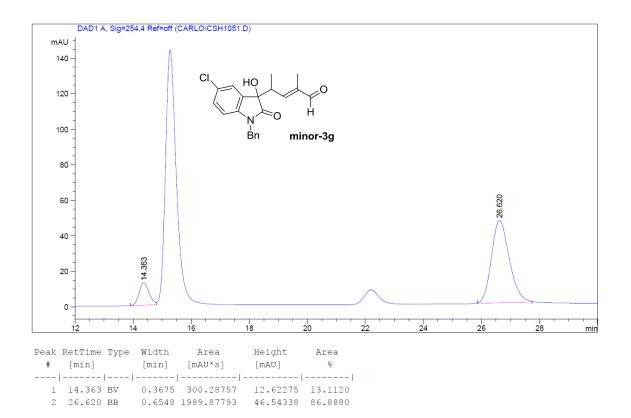


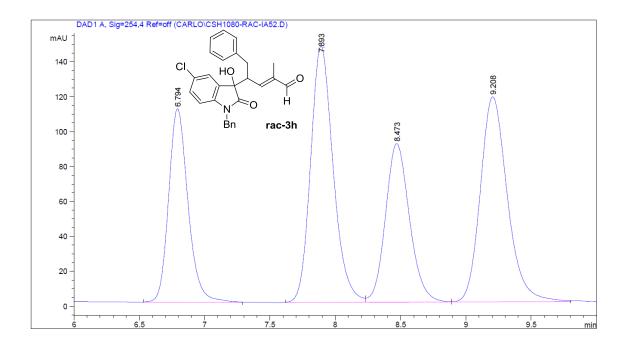
Peak	RetTime	Type	Width	Area	Height	Area	
#	[min]		[min]	[mAU*s]	[mAU]	8	
1	9.072	BB	0.2342	1600.85596	104.79333	97.0881	
2	12.946	BB	0.2519	48.01389	2.88847	2.9119	



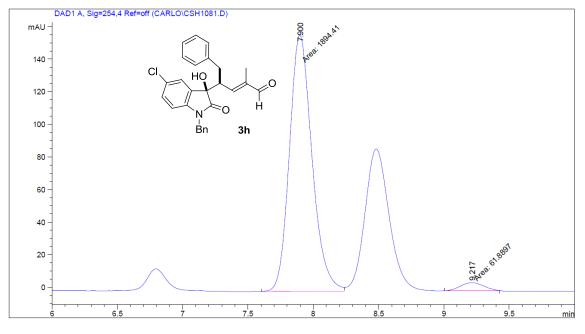
	RetTime	Туре		Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	14.305	MF	0.4011	1692.63147	70.33507	13.0268
2	15.230	FM	0.4326	4968.83691	191.43402	38.2410
3	22.090	MM	0.5995	4539.54004	126.19344	34.9370
4	26.604	MM	0.7050	1792.48804	42.37347	13.7953



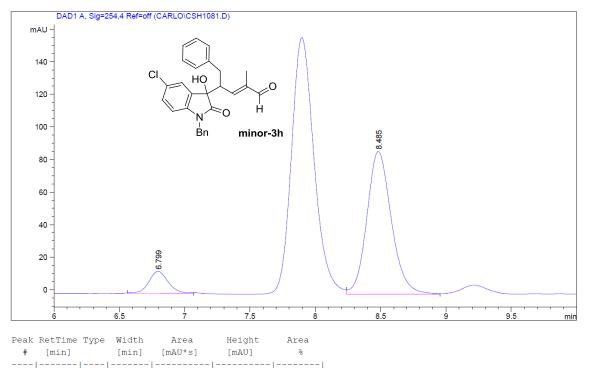




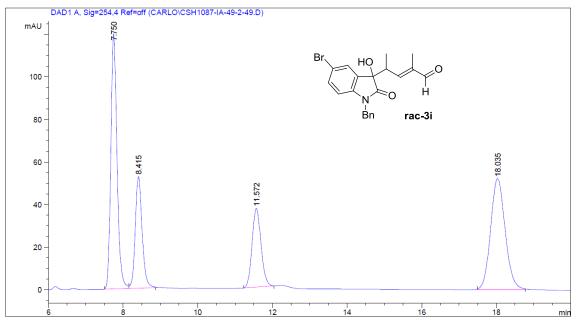
	RetTime [min]			Area [mAU*s]	Height [mAU]	Area %
		-				
1	6.794	BB	0.1583	1143.34375	110.89497	19.8464
2	7.893	BV	0.1841	1762.59595	146.79938	30.5955
3	8.473	VV	0.1973	1161.95654	90.78934	20.1695
4	9.208	VB	0.2201	1693.06262	117.48557	29.3886



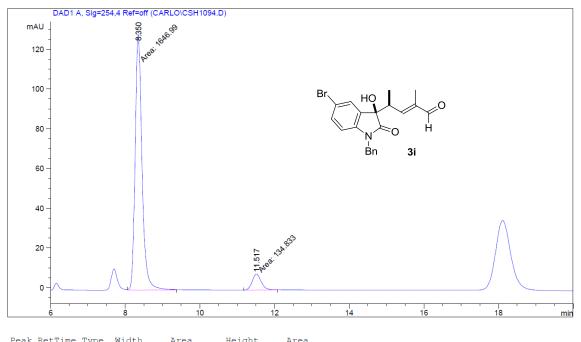
Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	7.900	MM	0.2004	1894.41309	157.54710	96.8364
2	9.217	MM	0.2072	61.88970	4.97807	3.1636



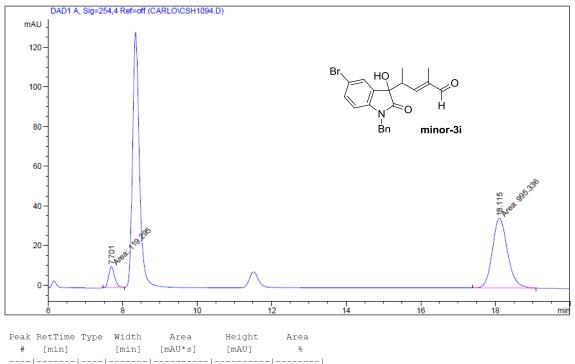
1	6.799	BB	0.1583	142.68256	13.61268	11.1329	
2	8.485	VB	0.1998	1138,95020	87.54805	88.8671	



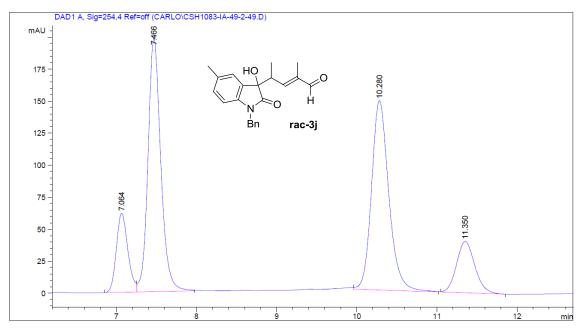
Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	%
1	7.750	BV	0.1761	1376.19531	119.70969	33.7870
2	8.415	VB	0.1950	669.62732	52.44112	16.4400
3	11.572	BB	0.2583	615.44397	36.94635	15.1098
4	18.035	BB	0.4178	1411.88379	52.16154	34.6632



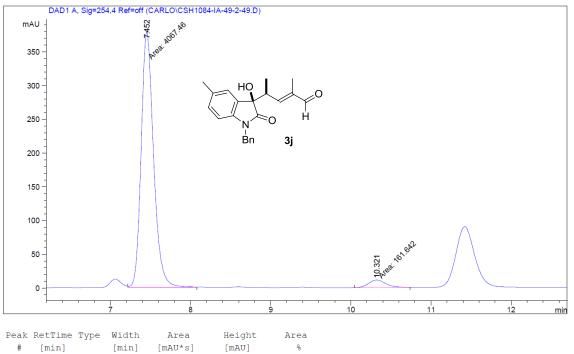
Pe	eak	RetTime	Туре	Width	Area	Height	Area	
	#	[min]		[min]	[mAU*s]	[mAU]	%	
	1	8.350	MM	0.2134	1646.98682	128.65370	92.4328	
	2	11.517	MM	0.2809	134.83328	7.99874	7.5672	



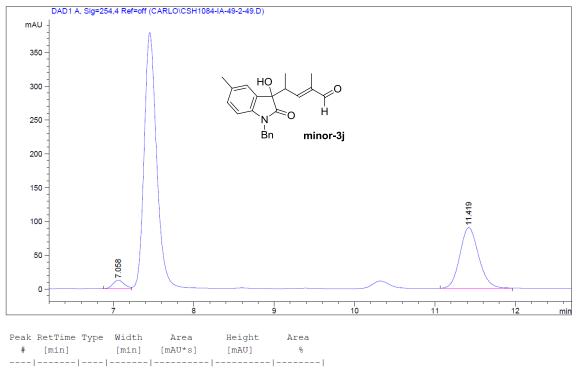
1	7.701	MM	0.1880	119.29466	10.57328	10.7026
2	18.115	MM	0.4711	995.33588	35.21309	89.2974



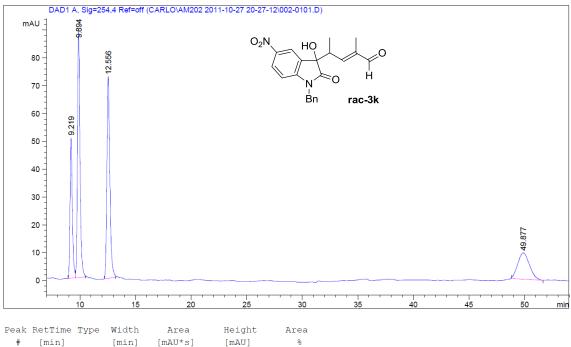
				Area [mAU*s]	Height [mAU]	Area %
		-				
1	7.064	BV	0.1545	615.89227	61.68438	10.9023
2	7.466	VB	0.1694	2193.22412	197.80807	38.8237
3	10.280	BB	0.2259	2202.62793	147.80676	38.9902
4	11.350	BB	0.2454	637.43884	40.11445	11.2837



	L		C			-
1	7.452	MM	0.1790	4067.45654	378.66370	96.1779
2	10.321	MM	0.2444	161.64211	11.02412	3.8221

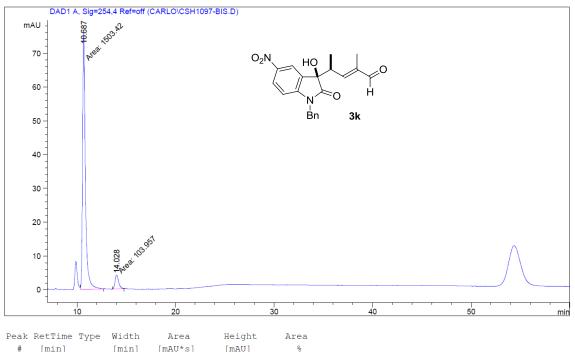


1	7.058 BV	0.1514	121.21336	12.46582	7.6296
2	11.419 BB	0.2522	1467.51990	90.00616	92.3704

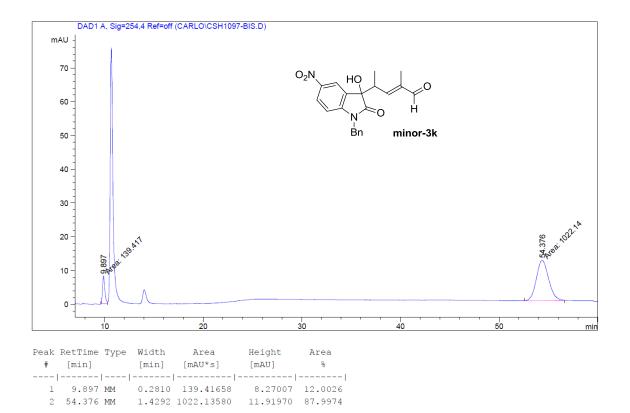


	#	[min]		[min]	[mAU*s]	[mAU]	*
-							
	1	9.219	BV	0.2168	720.84735	50.43407	17.3572
	2	9.894	VB	0.2400	1397.14832	88.57747	33.6419
	3	12.556	BB	0.2781	1344.59045	72.57484	32.3763
	4	49.877	BB	1.0440	690.42035	9.45374	16.6246

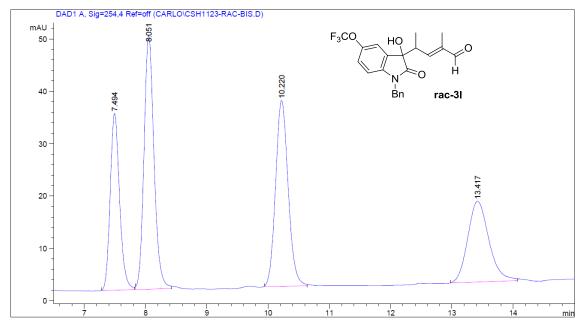
_



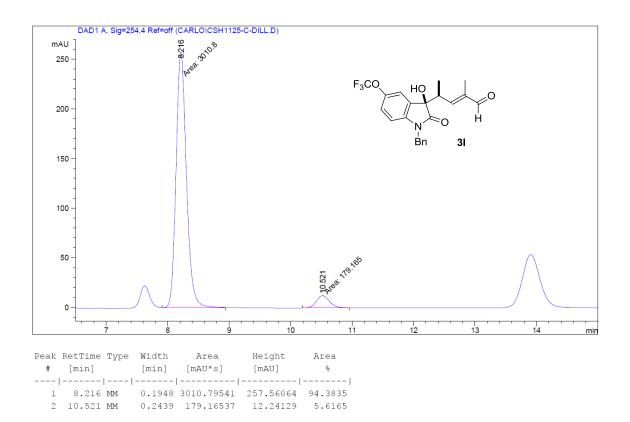
Реак	RetTime	туре	Wiath	Area	Height	Area	
#	[min]		[min]	[mAU*s]	[mAU]	8	
1	10.687	MM	0.3306	1503.41895	75.78211	93.5325	
2	14.028	MM	0.4252	103.95670	4.07523	6.4675	

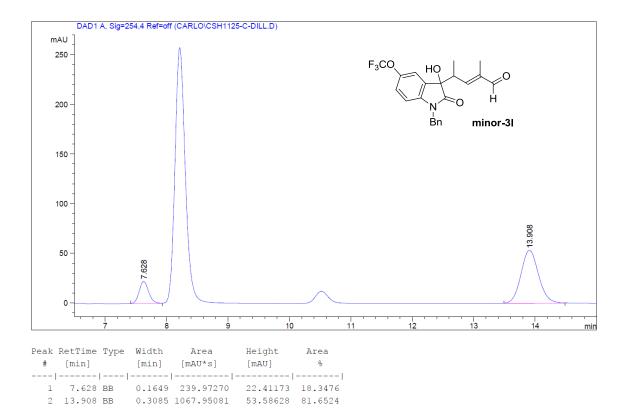


S	8	6
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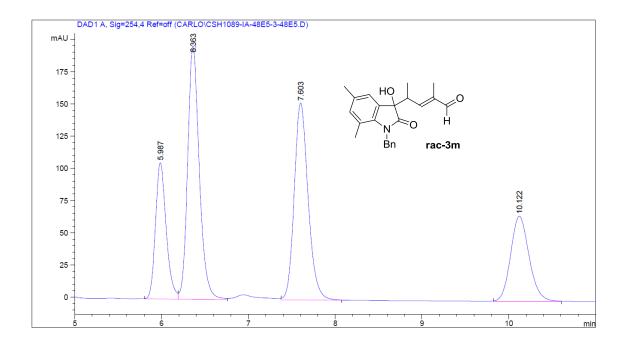


Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	7.494	BB	0.1623	359.62213	33.74640	20.1593
2	8.051	BB	0.1739	547.87671	48.44818	30.7123
3	10.220	BB	0.2203	505.76489	35.49067	28.3517
4	13.417	BB	0.3681	370.63464	15.43935	20.7767

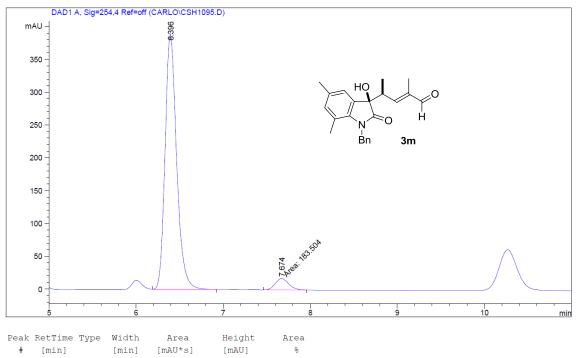


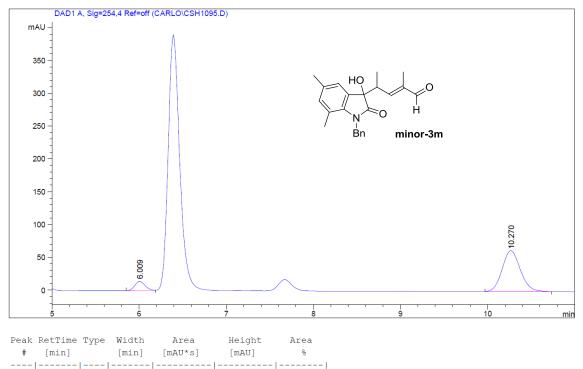


S88

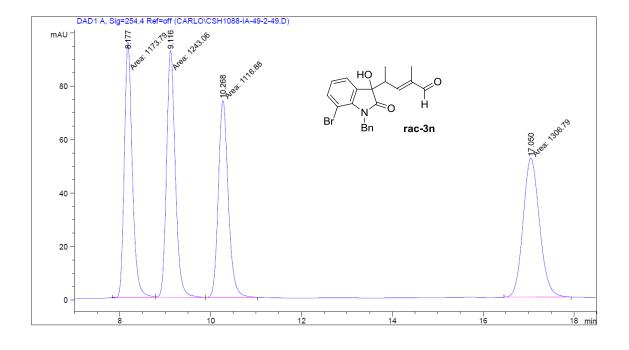


				Area [mAU*s]	-	Area %
		-				
1	5.987	BV	0.1298	904.70215	105.74374	16.7248
2	6.363	VB	0.1426	1830.04907	196.55920	33.8313
3	7.603	BB	0.1697	1698.77612	152.80586	31.4045
4	10.122	BB	0.2264	975.81372	66.02898	18.0394

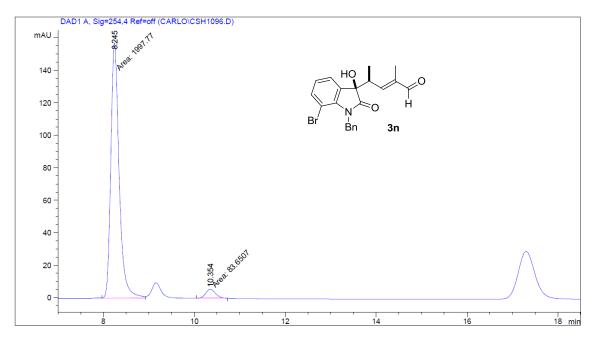




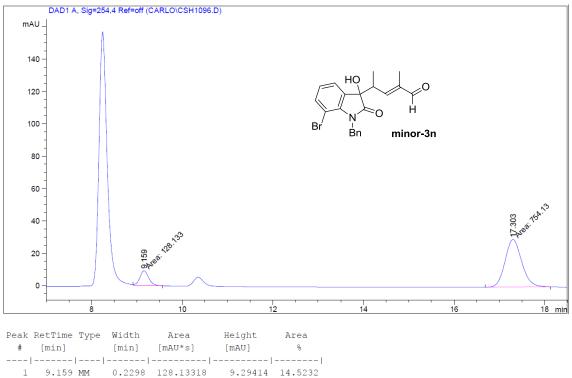
1	6.009	BV	0.1292	119.53513	14.34897	11.6028
2	10.270	BB	0.2243	910.69336	62.39051	88.3972



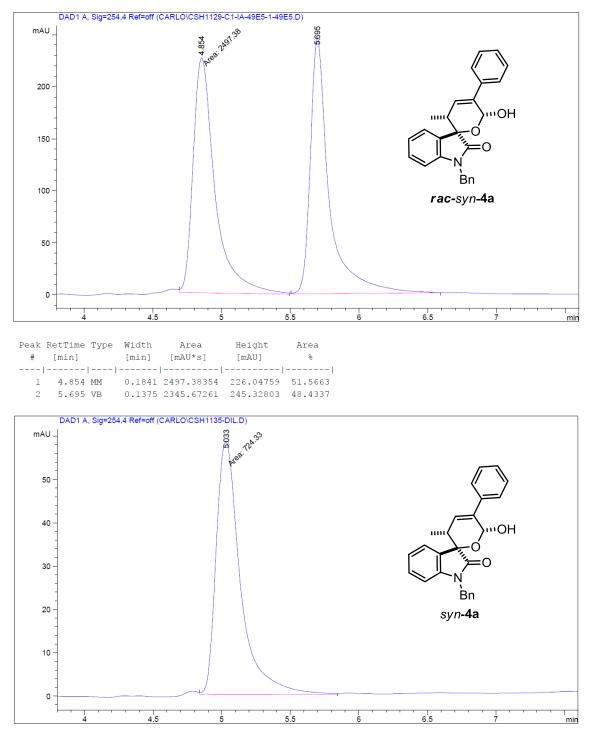
#	[min]		[min]	Area [mAU*s]	Height [mAU]	%
		-				
1	8.177	MF	0.2050	1173.79138	95.43773	24.2493
2	9.116	FM	0.2241	1243.06018	92.42852	25.6803
3	10.268	MM	0.2528	1116.88257	73.62360	23.0736
4	17.050	MM	0.4193	1306.79102	51.93951	26.9969

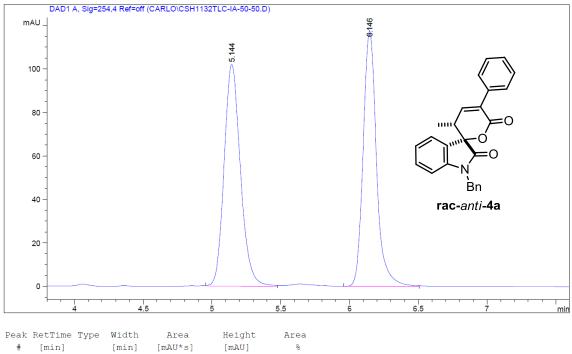


				Area [mAU*s]	Height [mAU]	Area %
1	8.245	MM	0.2117	1997.76758	157.30171	95.9811
2	10.354	MM	0.2499	83.65068	5.57990	4.0189

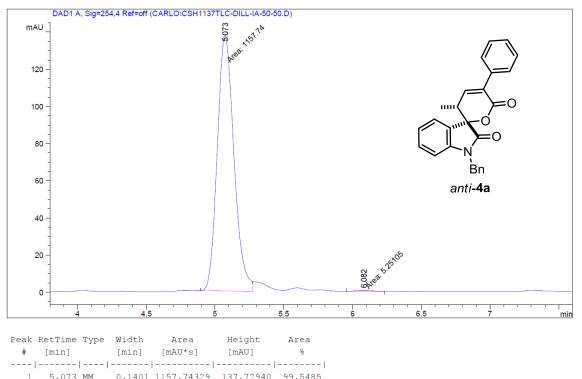


T	9.159	MIM	0.2298	128.13318	9.29414	14.5232
2	17.303	MM	0.4270	754.13013	29.43535	85.4768



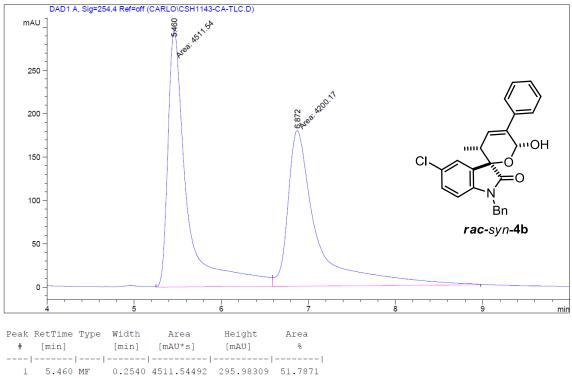


1	5.144	BB	0.1284	842.88971	102.04626	51.4832	
2	6.146	VB	0.1048	794.32196	117.66401	48.5168	

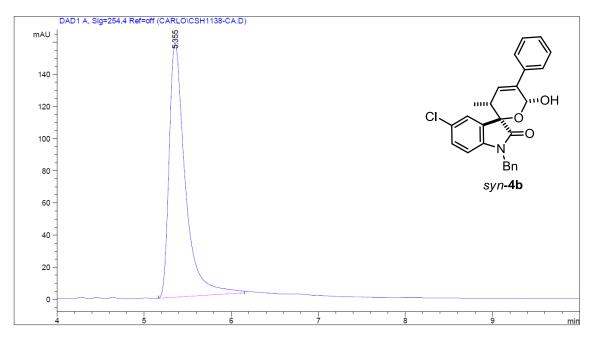


 1
 5.073 MM
 0.1401 1157.74329
 137.72940
 99.5485

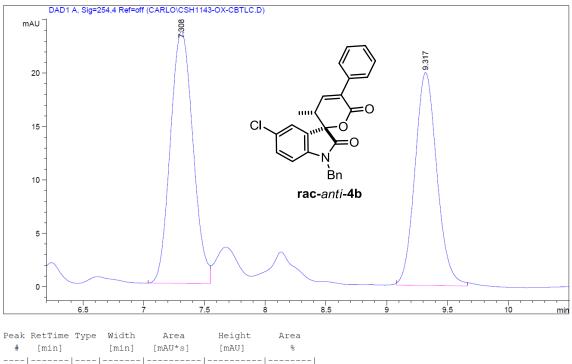
 2
 6.082 MM
 0.1365
 5.25105
 6.41304e-1
 0.4515



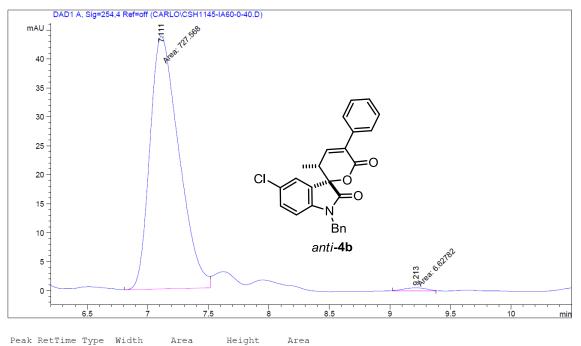
2	6.872 FM	0.3899	4200.17285	179.52324	48.2129



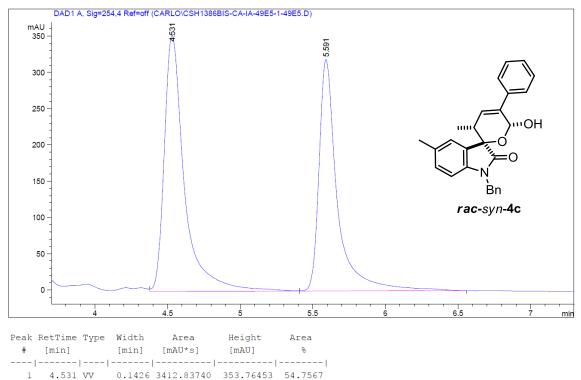




1	7.308	BV	0.2051	305.39990	23.58515	54.8761
2	9.317	BB	0.1948	251.12614	19.95225	45.1239

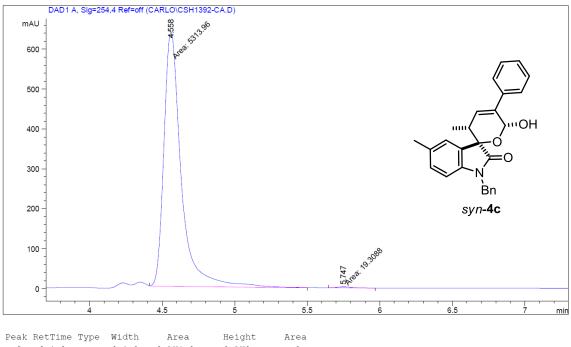


Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	7.111	MM	0.2775	727.56799	43.69289	99.0973
2	9.213	MM	0.2075	6.62782	5.32330e-1	0.9027

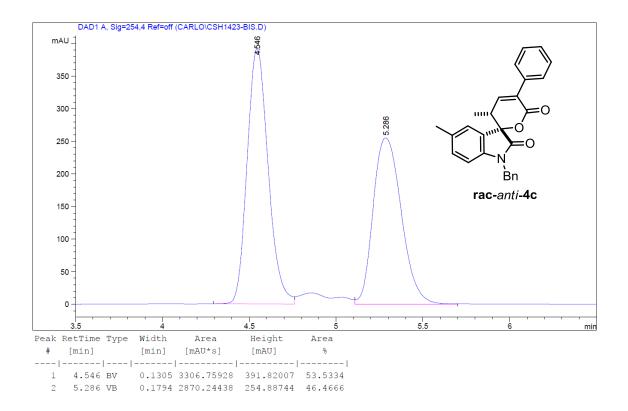


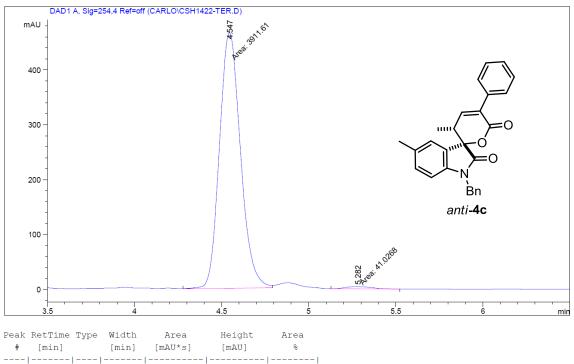
 1
 4.531
 VV
 0.1426
 3412.83740
 353.76453
 54.7567

 2
 5.591
 VB
 0.1269
 2819.88940
 319.57239
 45.2433

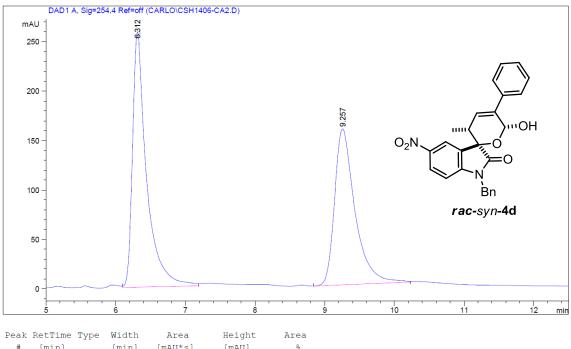


#	[min]		[min]	[mAU*s]	[mAU]	8
1	4.558	MM	0.1386	5313.95508	639.13660	99.6380
2	5.747	MM	0.1285	19.30883	2.50364	0.3620

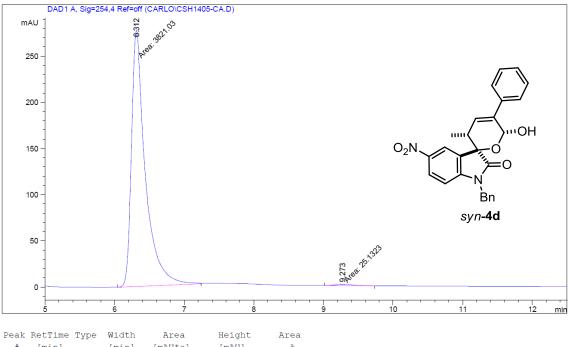




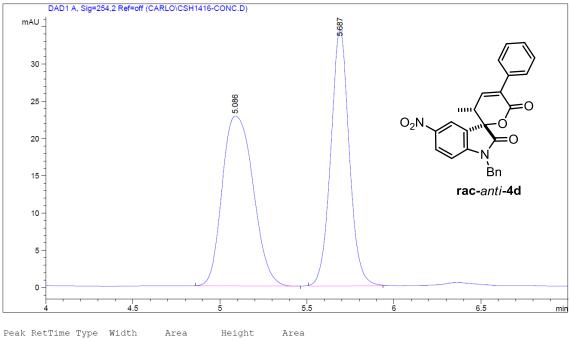
1	4.547 1	MM	0.1398	3911.60571	466.	.39664	98.9620
2	5.282 1	MM	0.1711	41.02678	3.	99631	1.0380



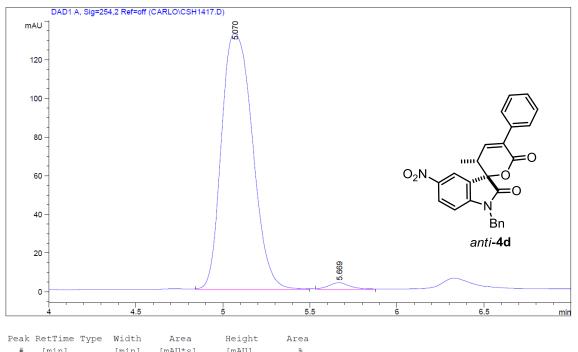
#	[min]		[min]	[mAU*s]	[mAU]	8
1	6.312	VB	0.1994	3618.26196	258.28412	52.9052
2	9.257	VB	0.3039	3220.88379	157.97133	47.0948



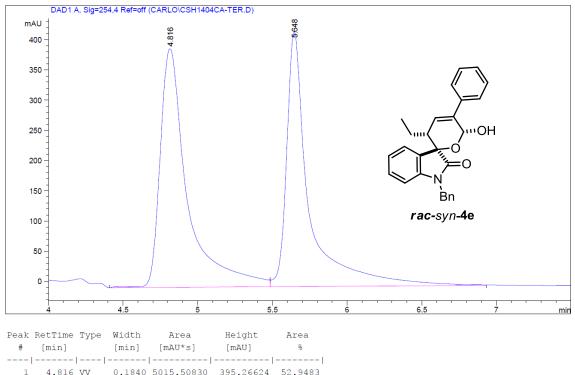
#	[min]		[min]	[mAU*s]	[mAU]	8	
1	6.312	MM	0.2299	3821.03198	276.96030	99.3466	
2	9.273	MM	0.2912	25.13235	1.43834	0.6534	



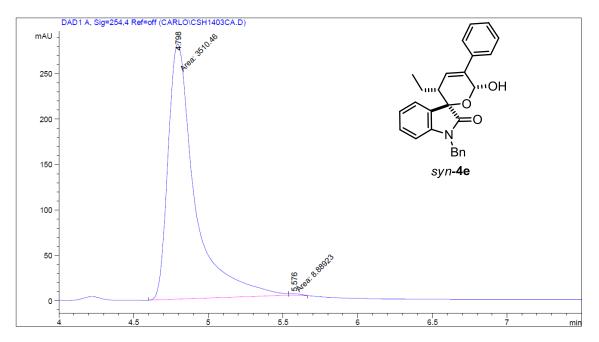
#	[min]		[min]	[mAU*s]	[mAU]	8
1	5.086	BB	0.2008	282.90732	22.79844	52.7355
2	5.687	BB	0.1139	253.55757	34.44197	47.2645



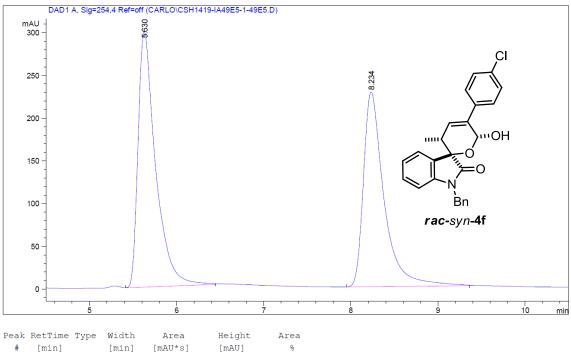
π	[111]		[10111]	[IIIAO S]	[IIIAO]	-0	
1	5.070	BB	0.1985	1636.89514	132.17401	98.3714	
2	5.669	BB	0.1237	27.10050	3.30509	1.6286	



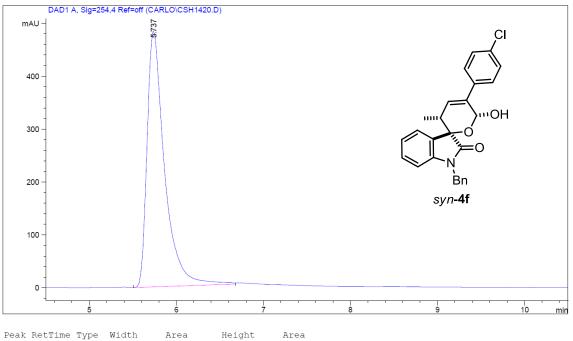
-	4.010 00	0.1040	5015.50050	555.20024	52.5405
2	5.648 VB	0.1491	4456.95508	422.41815	47.0517

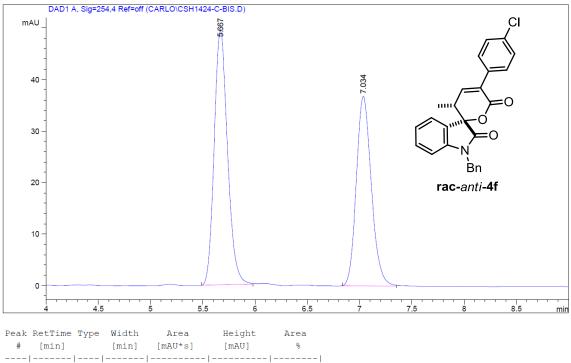


Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	4.798	MM	0.2081	3510.45752	281.11945	99.7474
2	5.576	MM	0.0919	8.88923	1.60976	0.2526

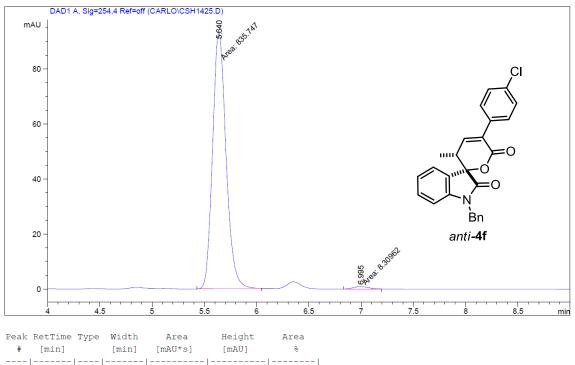


#	[min]	[min]	[mAU*s]	[mAU]	8
-					
1	5.630 VB	0.2051	4182.32764	299.09503	52.7277
2	8.234 BB	0.2439	3749.61230	227.98657	47.2723





1	5.667	BB	0.1377	438.94666	49.38423	54.4000
2	7.034	BB	0.1543	367,94055	36.89619	45,6000



-						
1	5.640	MM	0.1494	835.74683	93.21142	99.0155
2	6.995	MM	0.1611	8.30962	8.59917e-1	0.9845