

Supplementary Materials

Syntheses, Structures and Magnetic Properties of a Novel *mer*- [(bbp)Fe^{III}(CN)₃]²⁻ Building Block (bbp: bis(2- benzimidazolyl)pyridine dianion) and Its Related Heterobimetallic Fe(III)-Ni(II) Complexes

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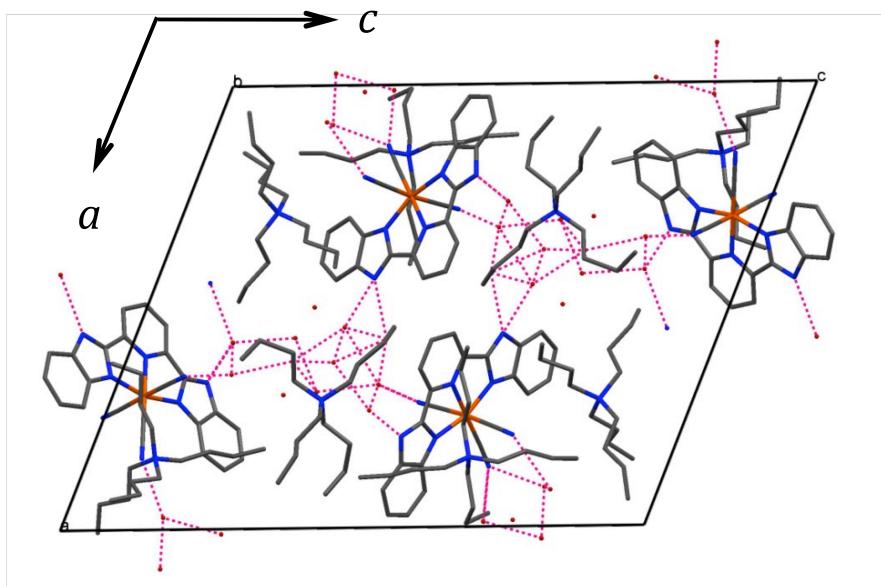


Figure S1. Asymmetric unit for complex **1** viewed perpendicular to the *b* axis.

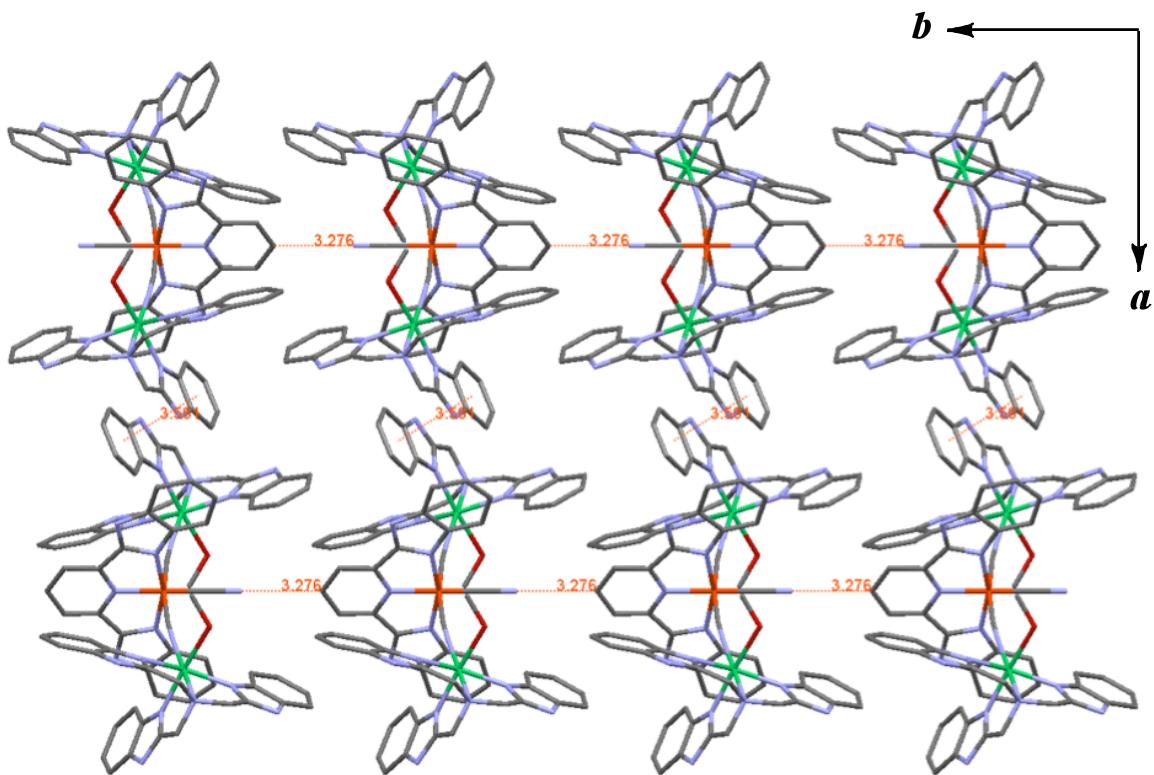


Figure S2. Molecular packing of complex **2** in the *ab* plane showing hydrogen bonding and π - π stacking.

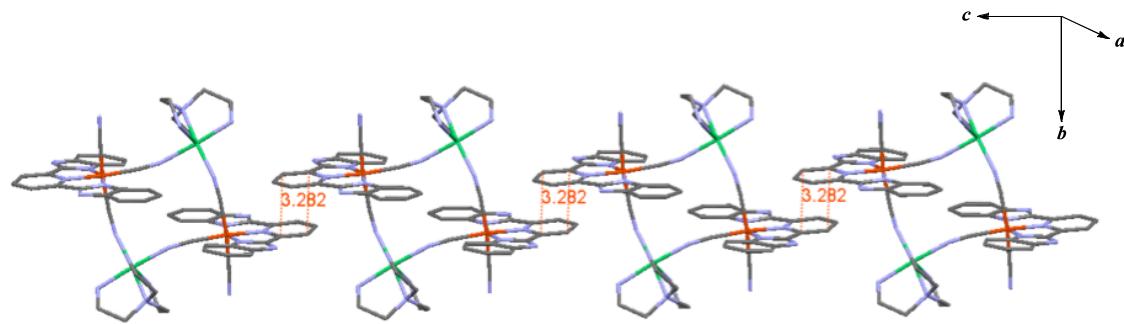


Figure S3. Supramolecular chain of complex 3 running along the *c* axis.

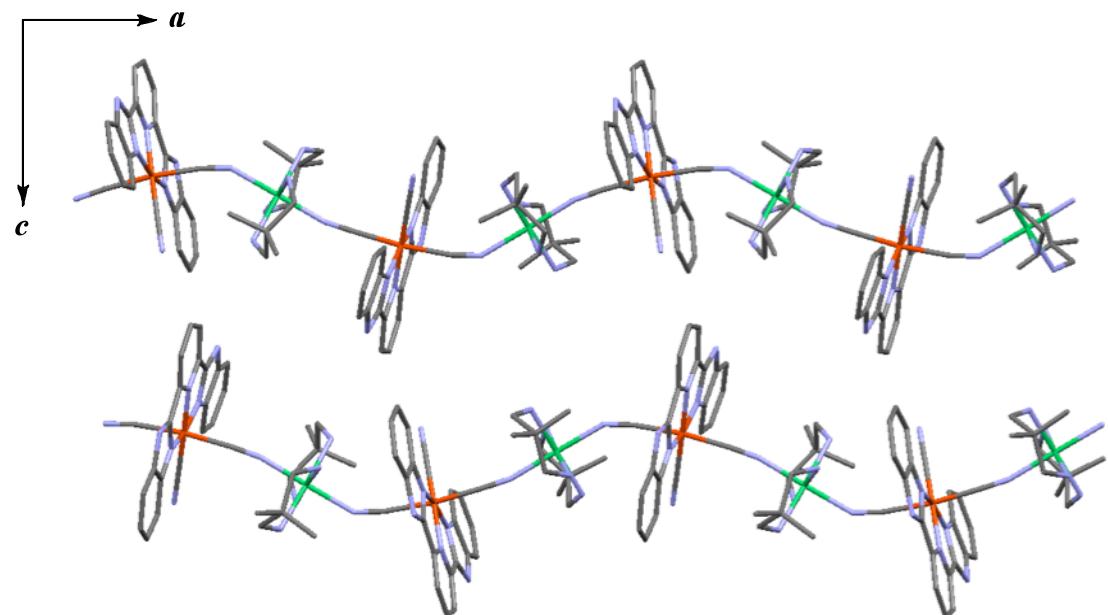


Figure S4. One dimensional coordination polymer observed in 4 running parallel to the *a* axis.

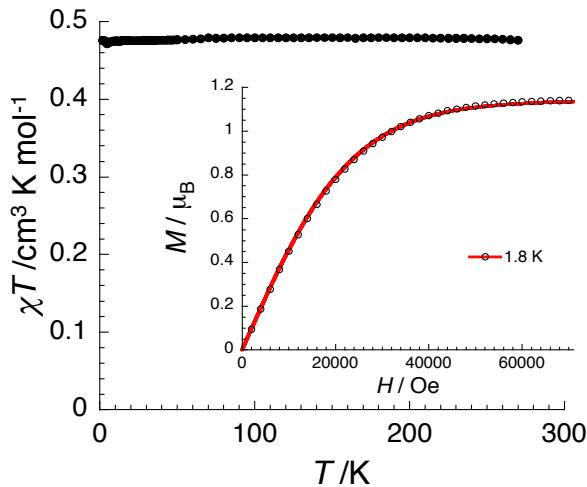


Figure S5. Temperature dependence of the χT product for compound **1** (where $\chi = M/H$ per complex) measured at 0.1 T. Inset: field dependence of the magnetization versus H at 1.8 K. The solid red line is the best fit of the experimental data to the $S = 1/2$ Brillouin function $g_{\text{Fe}} = 2.29$.

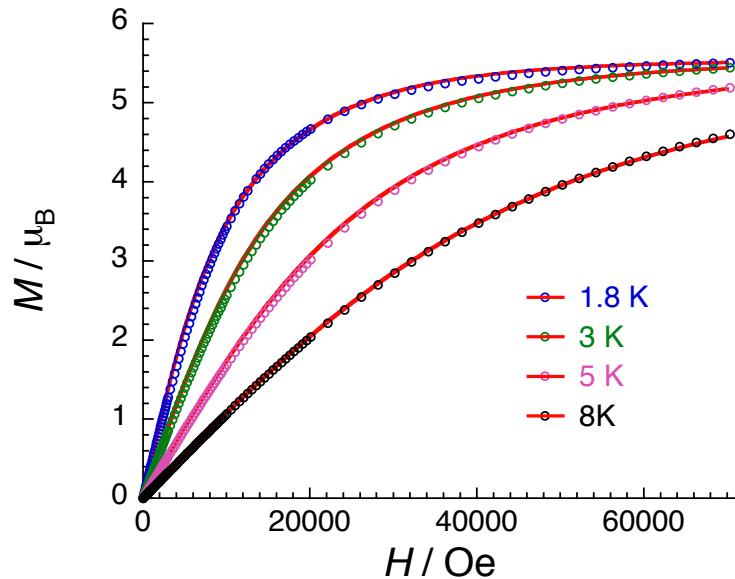


Figure S6. Field dependence of the magnetization at 1.8, 3, 5 and 8 K for compound **2**. The solid red lines are the best simulation obtained with the MAGPACK program¹ using the Hamiltonian $\hat{\mathbf{H}} = -2JS_{\text{Fe(III)}} \cdot (S_{\text{Ni(II)}} + S_{\text{Ni(II)}}) + D_{\text{Ni}} (S_{z,\text{Ni1}}^2 + S_{z,\text{Ni2}}^2)$ with the parameters $g_{\text{Ni}} = 2.21(1)$, $g_{\text{Fe}} = 2.29(1)$, $J/k_B = 12.4(2)$ K, and $D_{\text{Ni}}/k_B = -4(1)$ K.

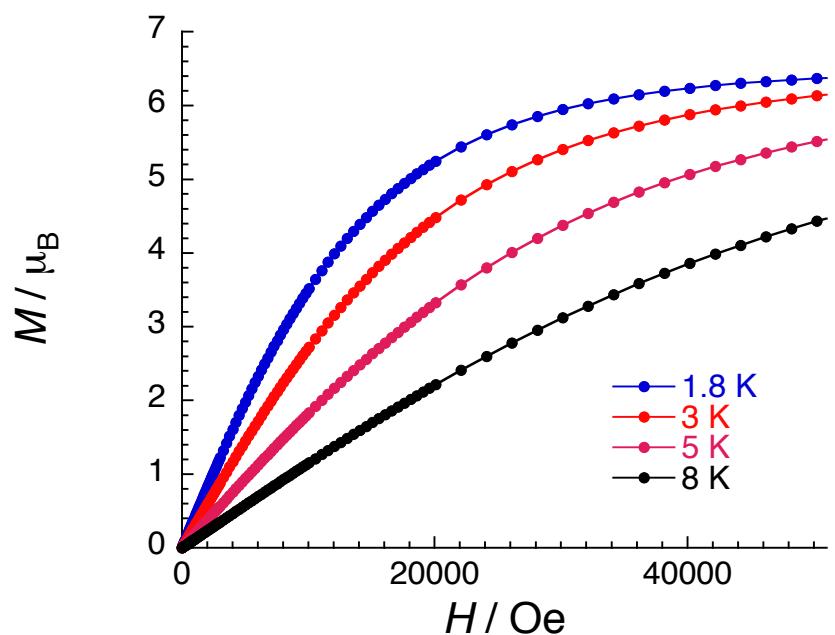


Figure S7. Field dependence of the magnetization at 1.8, 3, 5 and 8 K for **3** (the lines are eye guides).

Table S1. Selected magneto-structural data for related heterobimetallic $\text{Fe}^{\text{III}}_{\text{LS}}(\mu\text{-CN})\text{Ni}^{\text{II}}$ compounds.

Compounds	Structure	Ni-NC (°)	Ni···Fe (Å)	<i>g</i>	<i>J/K</i>	Ref.
$[(\text{Tp})_2\text{Fe}_2(\text{CN})_6\text{Ni}(\text{cyclam})]\cdot 2\text{H}_2\text{O}$	trinuclear	171.5(3)	5.132(2)	2.37	24.6	2a
$[(\text{Tp})_2\text{Fe}_2(\text{CN})_6\text{Ni}(\text{en})_2]$	trinuclear	153.8(3)	5.0669(11)	2.25	1.7	2b
$[\text{Ni}(\text{L1})][\text{Fe}(\text{bpb})(\text{CN})_2]_2\cdot \text{H}_2\text{O}$	trinuclear	160.2(7)	5.093(2)	2.5	9.14	2c
$[\text{Ni}(\text{L3})][\text{Fe}(\text{bpb})(\text{CN})_2]_2\cdot 7\text{H}_2\text{O}$	trinuclear	163.2(2)	5.110(1)	7.8	2c	2c
$[\text{Ni}(\text{L4})][\text{Fe}(\text{bpb})(\text{CN})_2]_2\cdot 4\text{H}_2\text{O}$	trinuclear	173.4(5)	5.133(1)	11.1	8.81	2c
$[\text{Ni}(\text{L5})][\text{Fe}(\text{bpb})(\text{CN})_2]_2$	trinuclear	157.0(2)	5.068(1)	2.5	2c	2d
$[(\text{pzTp})_2\text{Fe}_2(\text{CN})_6\text{Ni}(\text{L})]\cdot 1/2\text{CH}_3\text{OH}$	trinuclear	149.2(6)-150.7(6)	4.900(2)-4.910(2)	2.31	1.3	2d
$[(\text{pzTp})_2\text{Fe}_2(\text{CN})_6\text{Ni}(\text{bipy})_2]\cdot 2\text{H}_2\text{O}$	square	169.8(4)	5.080(1)	2.30	3.8	2e
$[\text{TpFe}(\text{CN})_3]_2[\text{Ni}(\text{DMSO})_4]$	trinuclear	165.4(3)	5.020(4)	2.20	7.5	3a
$\{[\text{Tp}^*\text{Fe}(\text{CN})_3\text{Ni}(\text{DMF})_4]\}_2\cdot 2\text{DMF}$	square	176.4(3)	5.1144(9)-5.0963(8)	2.30	7.1	3b
$\{[\text{Tp}^{*\text{Bn}}\text{Fe}(\text{CN})_3\text{Ni}(\text{DMF})_4]\}\cdot 2\text{DMF}$	trinuclear	173.3(3)	5.0422(6)	2.30	15.3-13.8	3c
$[\text{TpFe}(\text{CN})_2(\mu\text{-CN})\text{Ni}(\text{dmphen})_2]_2(\text{ClO}_4)_2\cdot 2\text{CH}_3\text{OH}$	square	169.5(2)-168.6(2)	5.170(6)-5.166(8)	2.30-	2.21	3c
$[(\text{pzTp})\text{Fe}(\text{CN})_2(\mu\text{-CN})\text{Ni}(\text{dmphen})_2]_2(\text{ClO}_4)_2\cdot 2\text{CH}_3\text{OH}$	square	170.3(3)-171.6(3)	5.150(6)-5.168(5)	2.39-	12.1-8.9	3c
$[(\text{pzTp})\text{Fe}(\text{CN})_3]_4[\text{Ni}]_4[\text{OTf}]_4\cdot 10\text{DMF}\cdot \text{Et}_2\text{O}$	cube	177.4(8)-179.1(2)	5.120(3)	2.20	9.5	4a
$\{[(\text{Tp}^*\text{Me})\text{Fe}(\text{CN})_3]_4[\text{Ni}(\text{DMF})_3]_2\}\cdot 4\text{DMF}\cdot \text{H}_2\text{O}$	hexanuclear	170.2(3)-177.1(3)	5.0986(8)-5.0571(9)	2.30	9.0	4b
$\{[(\text{Tp}^*\text{Me})\text{Fe}(\text{CN})_3]_6[\text{Ni}(\text{MeOH})_3]_2\}\cdot 3\text{H}_2\text{O}\cdot 8\text{MeOH}$	nonanuclear	157.5(4)-172.8(4)1	5.0458(7)-5.0775(8)	2.50	9.0	4b
$\{(\text{MeTp})_2\text{Fe}_2(\text{CN})_6\text{Ni}[(1R,2R)\text{-chxn}]_2\}$	square	163.0(5)-167.1(5)	5.096(1)-5.081(1)	2.34	4.16	5a
$\{(\text{Tp}^*\text{Me})_2\text{Fe}(\text{CN})_3\}_4[\text{Ni}(\text{tren})]_4[\text{ClO}_4]_4\}\cdot 7\text{H}_2\text{O}\cdot 4\text{MeCN}$	octanuclear	158.7(4)-170.6(3)	5.0859(9)-5.093(1)- 5.1188(8)	2.40	9.5	5b
$[(\text{pzTp})_2\text{Fe}_2\text{Ni}_2(\text{dpa})_2(\text{CN})_6](\text{ClO}_4)_2\cdot 2\text{CH}_3\text{OH}\cdot 6\text{H}_2\text{O}$	square	171.2(6)-174.4(5)	5.100(1)	2.23	10.1	5c
$\{[\text{TpFe}(\text{CN})_3]_2[\text{Ni}(\text{dpt})]\}_{\text{n}}\cdot 3\text{nH}_2\text{O}$	chain	159.5(2)-172.0(3)	5.066-5.188	2.30	16.3	5d
$[(\text{PhTp})\text{Fe}(\text{CN})_3\text{Ni}(\text{tren})]_2(\text{ClO}_4)_2$	square	160.3(4)-173.6(4)	5.016-5.157	2.284	6.0	6
$[(\text{MeTp})\text{Fe}(\text{CN})_3\text{Ni}(\text{tren})]_2(\text{ClO}_4)_2\cdot 2\text{H}_2\text{O}$	square	161.6(8)-177.2(8)	5.047-5.153	2.305	4	6
$[(i\text{BuTp})\text{Fe}(\text{CN})_3\text{Ni}(\text{tren})]_2(\text{ClO}_4)_2\cdot 2\text{H}_2\text{O}\cdot 2\text{CH}_3\text{OH}$	square	164.7(4)-175.2(4)	5.047-5.189	2.285	7.8	6
$[(\text{pzTp})\text{Fe}(\text{CN})_3]_4[\text{Ni}]_4[\text{OTf}]_4$	cube	179.0(1)-177.0(1)	5.188(5)	2.30	9.5-8.5	7
$[\text{TpFe}(\text{CN})_3\text{Ni}(\text{tren})]_2(\text{ClO}_4)_2\cdot 2\text{H}_2\text{O}$	square	161.7(3)-173.8(3)	5.042-5.156	2.22	6.4	8a
$[\text{TpFe}(\text{CN})_3\text{Ni}(\text{bipy})_2]_2[\text{TpFe}(\text{CN})_3]_2\cdot 6\text{H}_2\text{O}$	square	169.8(2)-171.7(2)	5.090-5.100	2.67	10.5	8a
$\{[\text{Tp}^*\text{Fe}(\text{CN})_3\text{Ni}(\text{bipy})_2]_2[\text{OTf}]_2\}\cdot 2\text{H}_2\text{O}$	square	167.1(4)-171.9(3)	5.097(1)	2.29	9.4	8b
<i>catena</i> - $[\text{Ni}((\text{R})\text{-pabn})][\text{TpFe}(\text{CN})_3]\text{PF}_6\cdot 2\text{MeOH}$	chain	164.9(3)-160.0(3)	5.0654(8)-4.9853(8)	2.40	9.14	8c

[TpFe(CN) ₃ (Nil ²) ₂] ₂ ·2ClO ₄ ·6H ₂ O	square	175.5(3)-177.9(2)	5.135(4)	2.14	6.1	9
[(<i>i</i> -BuTp)Fe(CN) ₃ Ni(L ³) ₂] ₂ ·2ClO ₄ ·6H ₂ O	square	175.5(3)-177.9(2)	4.978(9)	2.14	6.0	9
[Tp] ₂ Fe ₂ (CN) ₆ Ni(L ⁴) ₂ ·8H ₂ O	square	159.6(4)	4.978(9)	2.34	5.9	9
{[Ni(ntb)(MeOH)] ₂ [Fe(bbp)(CN) ₃][ClO ₄] ₂ }·2MeOH 2	trinuclear	176.5(3)	5.0817(4)	2.22	12.4	This work
{[Ni(tren)] ₂ [Fe(bbp)(CN) ₃] ₂ }·7MeOH 3	square	149.4(2)-164.3(2)	4.9443(5)-5.1107(5)	2.23	4.4	This work
{[Ni(dpd) ₂] ₂ [Fe(bbp)(CN) ₃] ₂ }·9MeOH·3H ₂ O 4	chain	148.6(4)-167.3(4)	4.9239(8)-5.0839(9)	2.30	18.5	This work

Tp = hydrotris(pyrazol-1-yl)borate, Tpm^{Me} = tris(3,5-dimethyl-1-pyrazolyl)methane, cyclam = 1,4,8,11-tetraazacyclooctadecane, en = ethylenediamine, , pZTp = tetra(pyrazol-1-yl)borate, L = 1,5,8,12-tetraazadodecane, bipy = 2,2'-bipyridine, Tp* = hydrotris(3,5-dimethylpyrazol-1-yl)borate, OTf = trifluoromethanesulfonate, Tp*Bn = tris(3,5-dimethyl-4-benzyl)pyrazolylborate, L1 = 1,8-dimethyl-1,3,6,8,10,13-hexaazacyclotetradecane, L3 = 1,8-dihydroxyethyl-1,3,6,8,10,13-hexaazacyclotetradecane, L4 = 3,10-bis(2-phenylethyl)-1,3,5,8,10,12-hexaazacyclotetradecane, L5 = 3-methyl-1,3,5,8,12-pentaazacyclotetradecane, dmphen = 2,9-dimethyl-1,10-phenanthroline, MeTp = methyltrispyrazolylborate, bbp = 1,2-Bis(pyridine-2-carboxamido)benzenate, chxn = 1,2-diaminocyclohexane, Tp*^{Me} = tris(3,4,5-trimethylpyrazolyl)borate, phen = phenanthroline, PhTp = tris(pyrazolyl)phenylborate, iBuTp = 2-methylpropyltris(pyrazolyl)borate, L¹ = 1-S(acetyl)-tris(pyrazolyl)hexane, pabn = N²,N²-bis(pyridin-2-ylmethyl)-1,10-binaphthyl-2,2'-diamine, cyclen = 1,4,7,10-tetraazacyclododecane, L² = 4,5-[1',4']dithiolo[2',3'-b]quinoxaline-2-bis(2-pyridyl)methylene-1,3-dithiole, dpt = dipropyleneetriamine.

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