

## Supporting Information

### Structure, stability and property modulations of stoichiometric graphene oxide

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In the supporting Information, we provide some additional results for better understanding our studied system, including: structural parameters of all the studied structures, and the geometric structure, stability and electronic properties of a-GMO configuration, and single layer deposition of metal atoms on z-GMO.

## 1. Structural parameters

We list the structural parameters in Table S1 for all the studied systems.

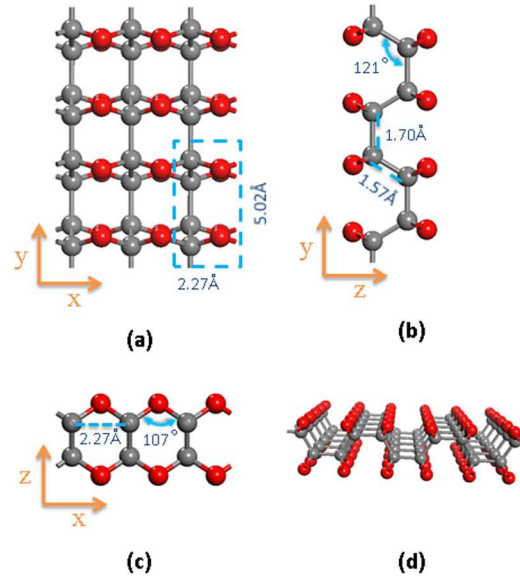
**Table S1.** Geometrical parameters of GMO and GDO (the vacuum layer is set as 15Å, the lattice parameters are given in order of a, b, c,  $\alpha$ ,  $\beta$ ,  $\gamma$ ).

Structure	Space group	Lattice parameters (Å)			Atom	Wyckoff positions		
<b>z-GMO</b>	<i>Pmma</i> (51)	2.25	2.68	15.0	C(2e)	0.7500	0.0000	0.4703
		90°	90°	90°	O(2f)	0.2500	0.5000	0.5854
<b>a-GMO</b>	<i>Pmma</i> (51)	5.02	2.27	15.0	C(4i)	0.4192	0.0000	0.4551
		90°	90°	90°	O(4j)	0.0100	0.5000	0.4045
<b>ep-GMO</b> <sup>1,2</sup>	<i>Cmmm</i> (65)	2.64	5.68	15.0	C(4h)	0.3259	0.0000	0.5000
		90°	90°	90°	O(4l)	0.5000	0.0000	0.5697
<b>mix-GMO</b> <sup>2</sup>	<i>Cm</i> (8)	9.42	5.27	15.0	C(4b)	0.0484	0.1415	0.4401
		90°	90°	90°	C(4b)	0.9144	0.3110	0.4319
					C(4b)	0.2856	0.6432	0.4081
					O(4b)	0.1599	0.2344	0.3906
					O(2a)	0.8976	0.5000	0.5006
					O(2a)	0.9300	0.5000	0.3644
					O(2a)	0.3230	0.5000	0.3313
					O(2a)	0.5703	0.5000	0.5203
<b>GDO</b>	<i>P4<sub>m</sub>2</i> (115)	2.34	2.34	15.0	C(1c)	0.5000	0.5000	0.5000
		90°	90°	90°	O(2g)	0.5000	0.0000	0.4587

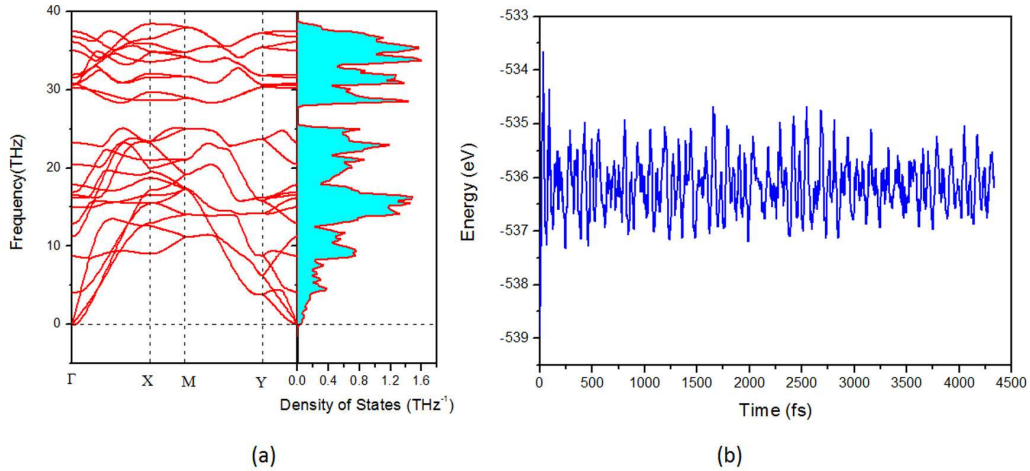
## 2. The stability and properties of a-GMO configuration

As is mentioned in the main text, the total energy of the metastable a-GMO is slightly higher than z-GMO but still more energetically favorable than ep-GMO and mix-GMO. Here we present its optimized geometrical structure (Figure S1), dynamic

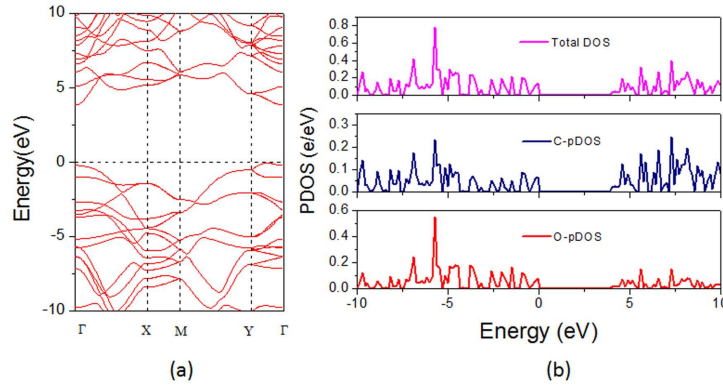
and thermodynamic stability (Figure S2), and electronic structure (Figure S3).



**Figure S1.** Different views of optimized a-GMO: (a) along z direction; (b) along x direction; (c) along y direction; (d) perspective view. Red and grey balls are oxygen and carbon atoms, respectively.



**Figure S2.** (a) Phonon dispersion band structures and the corresponding phonon density of states of a-GMO. The high symmetric Brillouin zone points are  $\Gamma$  (0, 0, 0), X (1/2, 0, 0), M (1/2, 1/2, 0), and Y (0, 1/2, 0). (b) Total energy fluctuation with respect to time during molecular dynamics simulations within a (3 $\times$ 3) supercell at the constant temperature of 300K.



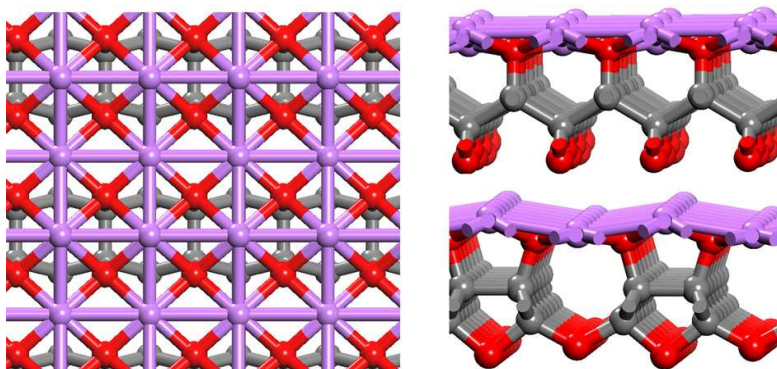
**Figure S3.** (a) Electronic band structure and (b) PDOS of a-GMO. The high symmetric Brillouin zone points in (a) are  $\Gamma$  (0, 0, 0), X (1/2, 0, 0), M (1/2, 1/2, 0) and Y (0, 1/2, 0).

The a-GMO structure (consisting of armchair carbon chains linked by O atoms) is a metastable state. It is interesting to note that the armchair carbon nanotubes are more metallic-like than the zigzag one, but a-GMO shows much larger energy band gap (3.90 eV) as compared with z-GMO (0.90 eV). Moreover, we found a-GMO displays larger anisotropy in its in-plane elastic modulus (See Table 1 in the main text). Since a-GMO can be stable up to 2000 K as the z-GMO does, and energy difference is quite small, a-GMO is also expected to be observed in experiment. The big differences between z-GMO and a-GMO in energy band gap and in-plane elastic modulus clearly indicate that the properties of graphene oxide are very sensitive to the structural details and synthesis details even in stoichiometric condition.

### 3. Deposition of metal atoms on z-GMO

Due to the special configuration of oxygen in z-GMO, the following question would be asked: is it possible to use z-GMO as a substrate for the single layer deposition of metal atoms? To explore this, we extensively studied the stability of

metal atoms on z-GMO including Li, Be, Mg, Al, K, Ca, Sc, Ti, V, Cr, Mn, Fe, Co, Ni Cu and Zn. We found that only Li can be stably absorbed on the z-GMO forming a single layer without clustering (Figure S4 in SI), where (1×2) reconstruction occurs resulting in slight bucklings. This is because the Li-O ionic bond (341 kJ/mol bond dissociation energy) is much more stable than Li-Li metal bond (106 kJ/mol bond dissociation energy). The stability of Li-z-GMO is verified through *ab initio* molecular dynamics simulations under 300 K for 5 ps.



**Figure S4.** Different views of optimized geometry for Li-z-GMO.

## REFERENCES

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- <sup>2</sup> Xiang, H. J.; Wei, S.-H.; Gong, X. G. *Phys. Rev. B* **2010**, 82, 035416.