

# SUPPORTING INFORMATION

to

## **Timed-Release Polymer Nanoparticles**

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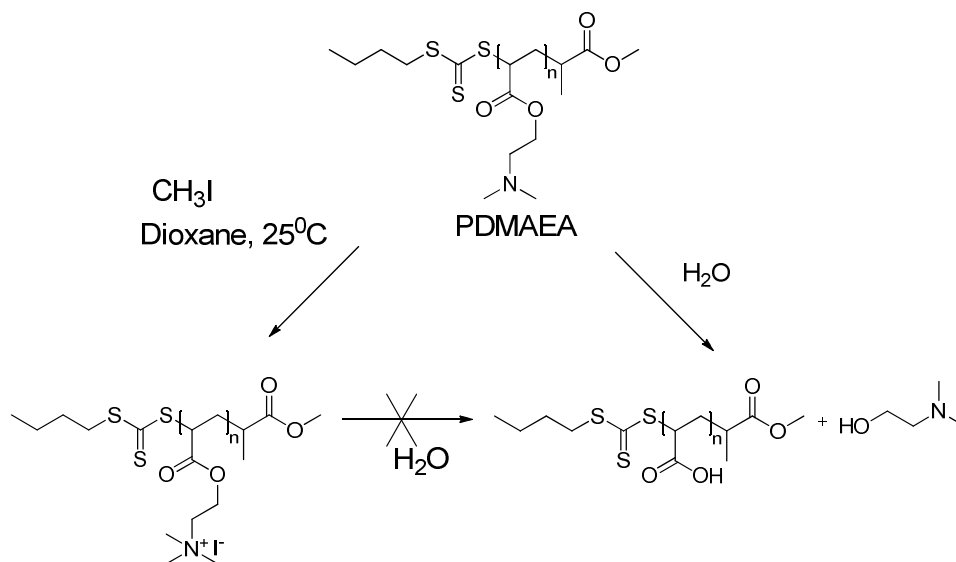
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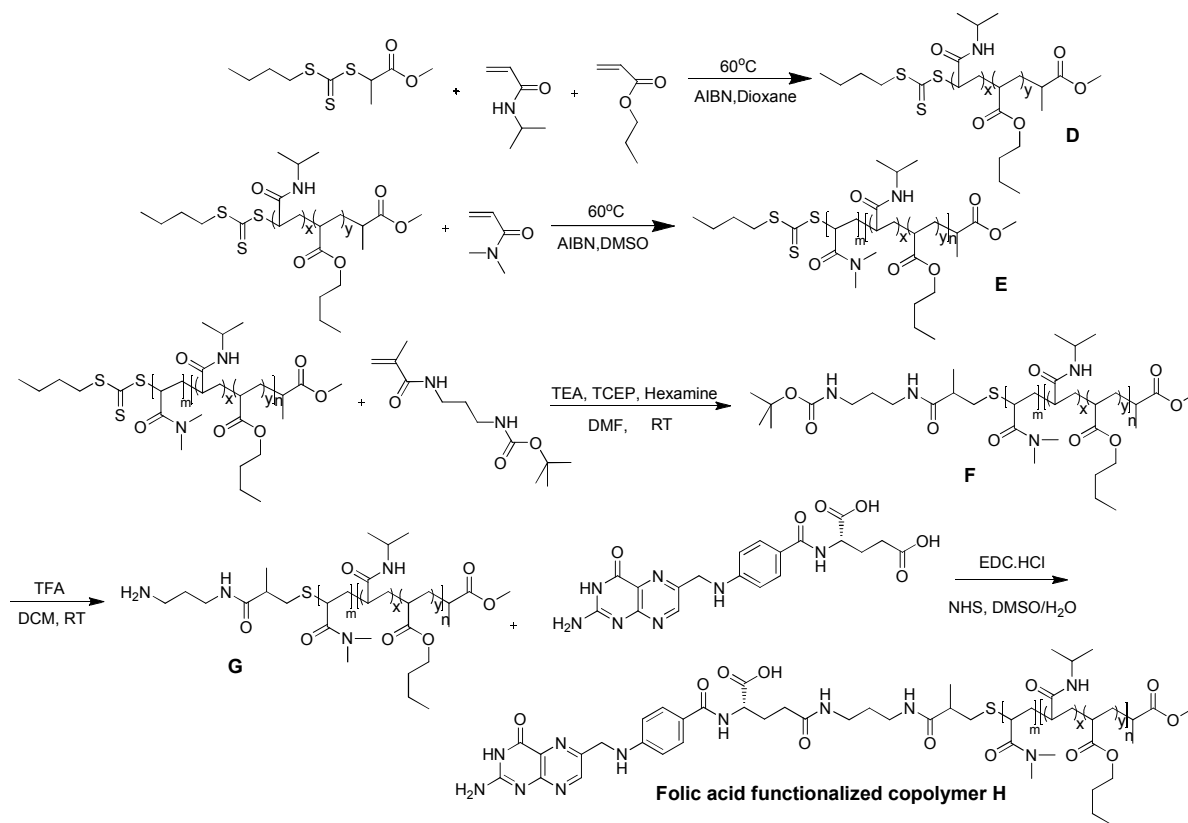
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**Scheme S1.** Self-catalyzed hydrolysis of Poly(2-dimethylaminoethyl acrylate).



**Scheme S2.** Synthesis of folic acid functional copolymer H



**Table S1. Molecular weights, polydispersities (PDI), and  $^1\text{H}$  NMR data of RAFT polymerization of thermo responsive block copolymers at 60 °C in Dioxane.**

Polymer code <sup>a</sup>		SEC <sup>b</sup>				$^1\text{H}$ NMR					
		RI		Triple detection <sup>c</sup>		Repeating units				$M_n^i$	Percent age of BA or STY (%) <sup>k</sup>
		$M_n$	PDI	$M_n$	$M_w$	NIPAM <sup>d</sup>	DMAEA	BA <sup>e</sup>	STY <sup>f</sup>		
<b>A</b>	P(DMA <sub>96</sub> -b-(NIPAM <sub>87</sub> -co-DMAEA <sub>25</sub> ))	34500	1.23	23800	24000	87	25 <sup>g</sup>	-	-	23175	-
<b>B1</b>	P(DMA <sub>96</sub> -b-(NIPAM <sub>88</sub> -co-DMAEA <sub>25</sub> -co-BA <sub>6</sub> ))	36200	1.24	28500	28900	88	25 <sup>h</sup>	6	-	24057	5.04
<b>B2</b>	P(DMA <sub>96</sub> -b-(NIPAM <sub>91</sub> -co-DMAEA <sub>25</sub> -co-BA <sub>12</sub> ))	36700	1.26	24000	24300	91	25 <sup>h</sup>	12	-	25165	9.38
<b>C1</b>	P(DMA <sub>96</sub> -b-(NIPAM <sub>84</sub> -co-DMAEA <sub>22</sub> -co-STY <sub>5</sub> ))	34300	1.32	24800	25700	84	22 <sup>g</sup>	-	5	22927	4.50
<b>C2</b>	P(DMA <sub>96</sub> -b-(NIPAM <sub>40</sub> -co-DMAEA <sub>15</sub> -co-STY <sub>13</sub> ))	24700	1.27	24100	24900	40	15 <sup>g</sup>	-	13	17786	19.11

<sup>a</sup> PDMA (Macro-CTA) with repeating unit = 96;  $M_n$  = 9769 calculated from  $^1\text{H}$  NMR. <sup>b</sup> SEC data measured in DMAc solution with 0.03 wt% of LiCl and using PSTY standards for calibration. <sup>c</sup> Triple detection data measured based on the  $dn/dc$  calculated from the polymers' concentration. <sup>d</sup> Repeating units of NIPAM ( $N_{\text{NIPAM}}$ ) determined by  $^1\text{H}$  NMR were calculated based on 96 repeating unit of Macro-CTA by the integral area of a peak at 1.08 ppm ( $I_{1.08}$ ) and a peak in the range 2.85-3.07 ppm ( $I_{2.85-3.07}$ ) using the following equation:  $N_{\text{NIPAM}} = 96 \times I_{1.08} / I_{2.85-3.07}$ . <sup>e</sup> Repeating units of BA ( $N_{\text{BA}}$ ) determined by  $^1\text{H}$  NMR were calculated based on 96 repeating unit of Macro-CTA by the integral area of a peak at 0.88 ppm ( $I_{0.88}$ ) and a peak in the range 2.85-3.07 ppm ( $I_{2.85-3.07}$ ) using the following equation:  $N_{\text{BA}} = (96 \times 2 \times I_{0.88} / I_{2.85-3.07}) - 1$ . <sup>f</sup> Repeating units of STY ( $N_{\text{STY}}$ ) determined by  $^1\text{H}$  NMR were calculated based on 96 repeating unit of Macro-CTA and the  $N_{\text{NIPAM}}$  by the integral area of the peak in the range 6.20-7.18 ppm ( $I_{6.20-7.18}$ ) and the peak in the range 2.85-3.07 ppm ( $I_{2.85-3.07}$ ) using the following equation:  $N_{\text{STY}} = [(96 \times 6 \times I_{6.20-7.18}) / I_{2.85-3.07} - N_{\text{NIPAM}}] / 5$ . <sup>g</sup> Repeating units of DMAEA ( $N_{\text{DMAEA}}$ ) determined by  $^1\text{H}$  NMR were calculated based on 96 repeating unit of Macro-CTA and the  $N_{\text{NIPAM}}$  by the integral area of the peak in the range 2.85-3.07 ppm ( $I_{2.85-3.07}$ ) and the peak in the range 3.95-4.30 ppm ( $I_{3.95-4.30}$ ) using the following equation:  $N_{\text{DMAEA}} = [(96 \times 6 \times I_{3.95-4.30} / I_{2.85-3.07}) - N_{\text{NIPAM}}] / 2$ . <sup>h</sup> Repeating units of DMAEA ( $N_{\text{DMAEA}}$ ) determined by  $^1\text{H}$  NMR were calculated based on 96 repeating unit of Macro-CTA, and the  $N_{\text{NIPAM}}$  and  $N_{\text{BA}}$  by the integral area of the peak in the range 2.85-3.07 ppm ( $I_{2.85-3.07}$ ) and the peak in the range 3.95 - 4.30 ppm ( $I_{3.95-4.30}$ ) using the following equation:  $N_{\text{DMAEA}} = [(96 \times 6 \times I_{3.95-4.30} / I_{2.85-3.07}) - (N_{\text{NIPAM}} + 2 N_{\text{BA}})] / 2$ . <sup>i</sup> Molecular weight determined by  $^1\text{H}$  NMR were calculated based on the repeating units of  $N_{\text{NIPAM}}$ ,  $N_{\text{BA}}$  or  $N_{\text{STY}}$ ,  $N_{\text{DMAEA}}$ , and the molecular weight of macro-CTA:  $M_n = (N_{\text{NIPAM}} \times 113) + (N_{\text{BA}} \times 128.2)$  or  $(N_{\text{STY}} \times 104) + (N_{\text{DMAEA}} \times 143) + 9769$ . <sup>k</sup> Percentages of BA or STY were calculated based on  $N_{\text{DMAEA}}$ ,  $N_{\text{NIPAM}}$ ,  $N_{\text{BA}}$  or  $N_{\text{STY}}$ : Percentage of BA =  $N_{\text{BA}} / (N_{\text{DMAEA}} + N_{\text{NIPAM}} + N_{\text{BA}})$  and percentage of STY =  $N_{\text{STY}} / (N_{\text{DMAEA}} + N_{\text{NIPAM}} + N_{\text{STY}})$ .

**Table S2:** Hydrodynamic diameter ( $D_h$ ), Polydispersities (PDIs) of thermoresponsive block copolymers and their Oligo DNA 9-27 complexes in Milli-Q water at N/P Ratio 0.5, 1, 2, 5, and 10 with 1  $\mu$ g DNA. Measurements were carried out on DLS machine at 37 °C.

N/P ratio	Polymer B1/oligo DNA complexes		Polymer B2/oligo DNA complexes		Polymer C1/oligo DNA complexes		Polymer C2/oligo DNA complexes	
	$D_h$ (nm)	PDI	$D_h$ (nm)	PDI	$D_h$ (nm)	PDI	$D_h$ (nm)	PDI
0.5	$30.33 \pm 6.58$	0.413	$27.16 \pm 2.14$	0.356	$11.10 \pm 1.01$	0.453	$17.20 \pm 4.29$	0.389
1	$29.49 \pm 3.44$	0.337	$27.35 \pm 1.23$	0.218	$24.03 \pm 1.07$	0.400	$24.53 \pm 0.86$	0.243
2	$14.75 \pm 4.24$	0.450	$24.21 \pm 0.24$	0.240	$9.021 \pm 2.40$	0.430	$24.20 \pm 1.33$	0.361
5	$17.72 \pm 8.74$	0.488	$25.56 \pm 0.70$	0.168	$8.872 \pm 2.74$	0.560	$23.75 \pm 0.58$	0.214
10	$27.21 \pm 0.20$	0.048	$29.12 \pm 1.02$	0.071	$29.27 \pm 0.69$	0.080	$20.76 \pm 0.54$	0.110
Polymer only <sup>a</sup>	$27.75 \pm 1.67$	0.063	$29.87 \pm 0.53$	0.047	$28.02 \pm 1.82$	0.087	$21.22 \pm 0.78$	0.100

<sup>a</sup> For polymer only, the polymer concentration was 0.5 mg/mL. B1; P(DMA<sub>96</sub>-b-(NIPAM<sub>88</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>6</sub>)), B2; P(DMA<sub>96</sub>-b-(NIPAM<sub>91</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>12</sub>)), C1; P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>)), C2; P(DMA<sub>96</sub>-b-(NIPAM<sub>40</sub>-co-DMAEA<sub>15</sub>-co-STY<sub>13</sub>)). Data were reported as soluble polymers in Milli-Q water were incubated with DNA at N/P Ratio 0.5, 1, 2, 5, and 10 at below their LCST for 30 min prior to be measured at 37 °C on DLS machines. Data were reported as an average of five measurements. The mean standard of deviation of polymer particle sizes was calculated from five measurements.

**Table S3.** Molecular weights, polydispersities (PDIs), and  $^1\text{H}$  NMR data of RAFT polymerization for synthesis of copolymer D and E at 60 °C.

Polymer code		SEC <sup>a</sup>				$^1\text{H}$ NMR			
		RI		Triple detection <sup>b</sup>		Repeating units			$M_n$
		$M_n$	PDI	$M_n$	$M_w$	NIPAM <sup>c</sup>	BA <sup>d</sup>	DMA <sup>e</sup>	
<b>D</b>	P(NIPAM <sub>97</sub> -co-BA <sub>13</sub> )	28700	1.16	15900	16100	97	13	-	12880 <sup>f</sup>
<b>E</b>	P(DMA <sub>99</sub> -b-(NIPAM <sub>97</sub> -co-BA <sub>13</sub> ))	38500	1.19	26900	27200	-	-	99	22694 <sup>g</sup>

<sup>a</sup> SEC data measured in DMAc solution with 0.03 wt% of LiCl and using PSTY standards for calibration.

<sup>b</sup> Triple detection data measured based on the  $dn/dc$  calculated from the polymers' concentration.

<sup>c</sup> Repeating units of NIPAM ( $N_{\text{NIPAM}}$ ) determined by  $^1\text{H}$  NMR were calculated by the integral area of a peak at 1.12 ppm ( $I_{1.12}$ ) and a peak at 3.64 ppm ( $I_{3.64}$ ) using the following equation:  $N_{\text{NIPAM}} = I_{1.12} / (2 \times I_{3.64})$

<sup>d</sup> Repeating units of BA ( $N_{\text{BA}}$ ) determined by  $^1\text{H}$  NMR were calculated by the integral area of a peak at 0.89 ppm ( $I_{0.89}$ ) and a peak at 3.64 ppm ( $I_{3.64}$ ) using the following equation:  $N_{\text{BA}} = (I_{0.89} / I_{3.64}) - 1$

<sup>e</sup> Repeating units of DMA ( $N_{\text{DMA}}$ ) determined by  $^1\text{H}$  NMR were calculated based on  $N_{\text{BA}}$  by the integral area of the peak in the range 2.86-3.07 ppm ( $I_{2.86-3.07}$ ) and the peak at 0.89 ppm ( $I_{0.89}$ ) using the following equation:  $N_{\text{DMA}} = [(N_{\text{BA}} + 3) \times I_{2.86-3.07}] / (I_{0.89} \times 6)$

<sup>f</sup> Molecular weight determined by  $^1\text{H}$  NMR were calculated based on the repeating units of  $N_{\text{NIPAM}}$ ,  $N_{\text{BA}}$  and the molecular weight of MCEBTTTC:  $M_n = (N_{\text{NIPAM}} \times 113) + (N_{\text{BA}} \times 128.2) + 252.42$

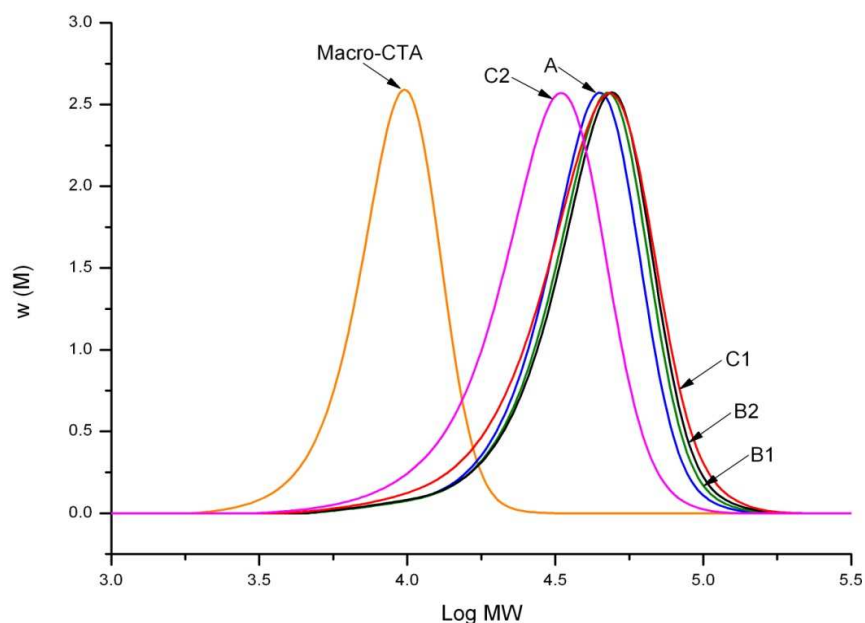
<sup>g</sup> Molecular weight determined by  $^1\text{H}$  NMR were calculated based on the repeating units of  $N_{\text{DMA}}$  and the molecular weight of macro-CTA:  $M_n = (N_{\text{DMA}} \times 99.131) + 12880$

**Table S4.** Lower critical solution temperature (LCST), hydrodynamic diameter ( $D_h$ ), Polydispersities (PDI) for thermoresponsive block copolymers G and folic acid functionalized copolymer (H) determined by dynamic light scattering (DLS).

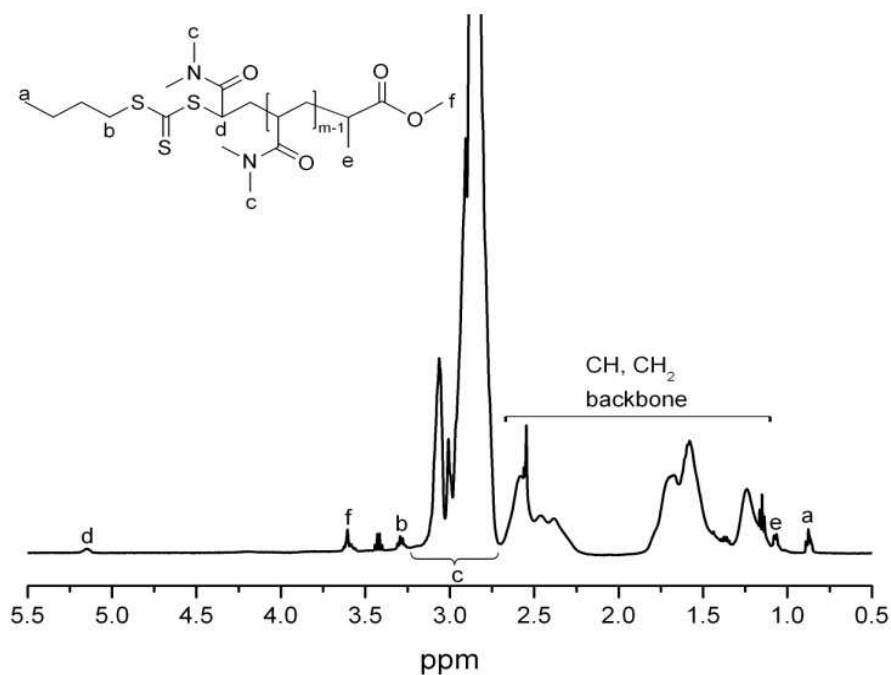
Polymer code	DLS		
	LCST (°C) <sup>a</sup>	$D_h$ (nm) <sup>b</sup>	PDI <sup>b</sup>
G	13 – 15	$33.83 \pm 0.65$	0.017
H	15 – 19	$31.66 \pm 0.98$	0.089

<sup>a</sup>LCST determined by DLS (10 mg/mL)

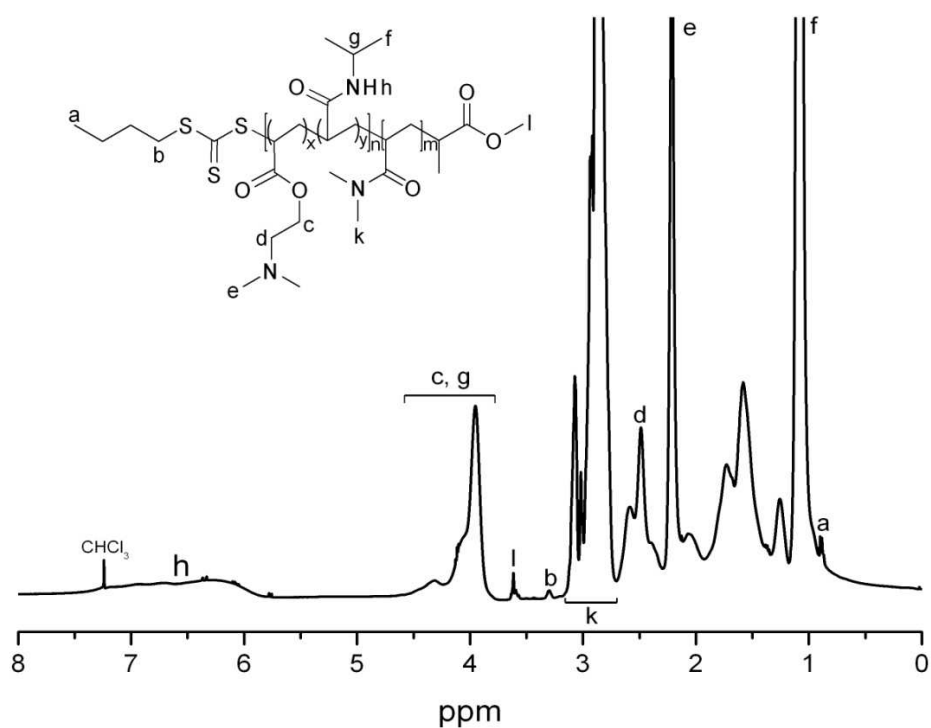
<sup>b</sup>Hydrodynamic diameter ( $D_h$ ) and Polydispersity (PDI) determined by DLS (10 mg/mL) at 37 °C.



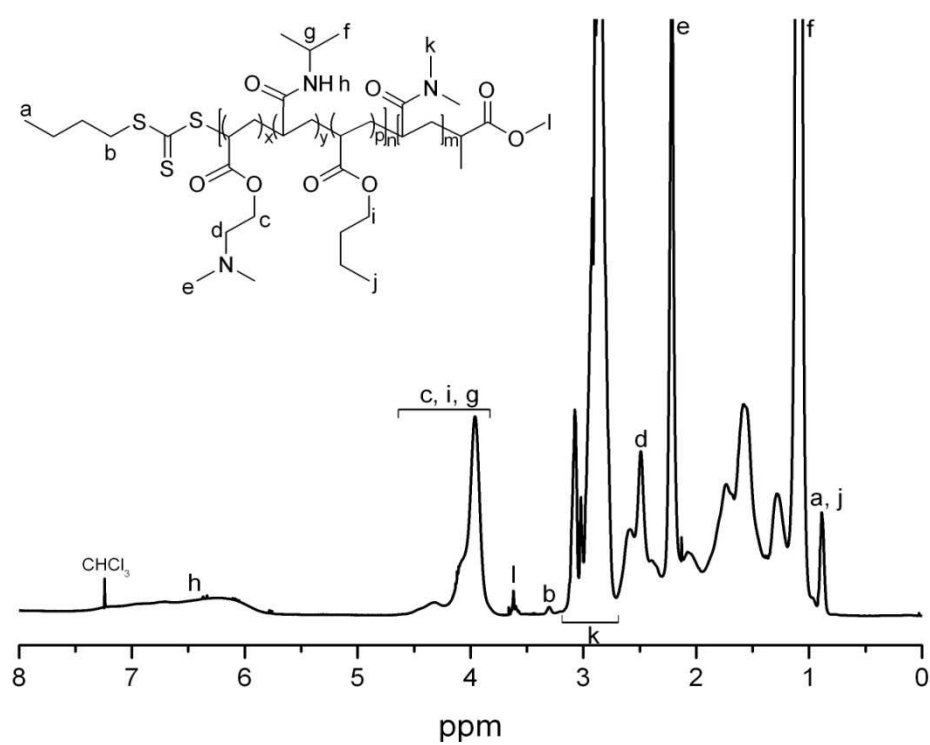
**Figure S1.** Size Exclusion Chromatography (SEC) traces of macro CTA; PDMA<sub>96</sub>, A; P(DMA<sub>96</sub>-b-(NIPAM<sub>87</sub>-co-DMAEA<sub>25</sub>)), B1; P(DMA<sub>96</sub>-b-(NIPAM<sub>88</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>6</sub>)), B2; P(DMA<sub>96</sub>-b-(NIPAM<sub>91</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>12</sub>)), C1; P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>)), C2; P(DMA<sub>96</sub>-b-(NIPAM<sub>40</sub>-co-DMAEA<sub>15</sub>-co-STY<sub>13</sub>)). The data were measured by DMAc SEC. The intensity for different distribution curves was normalized.



**Figure S2** <sup>1</sup>H NMR spectrum of Macro-CTA: PDMA<sub>96</sub> in CDCl<sub>3</sub>.

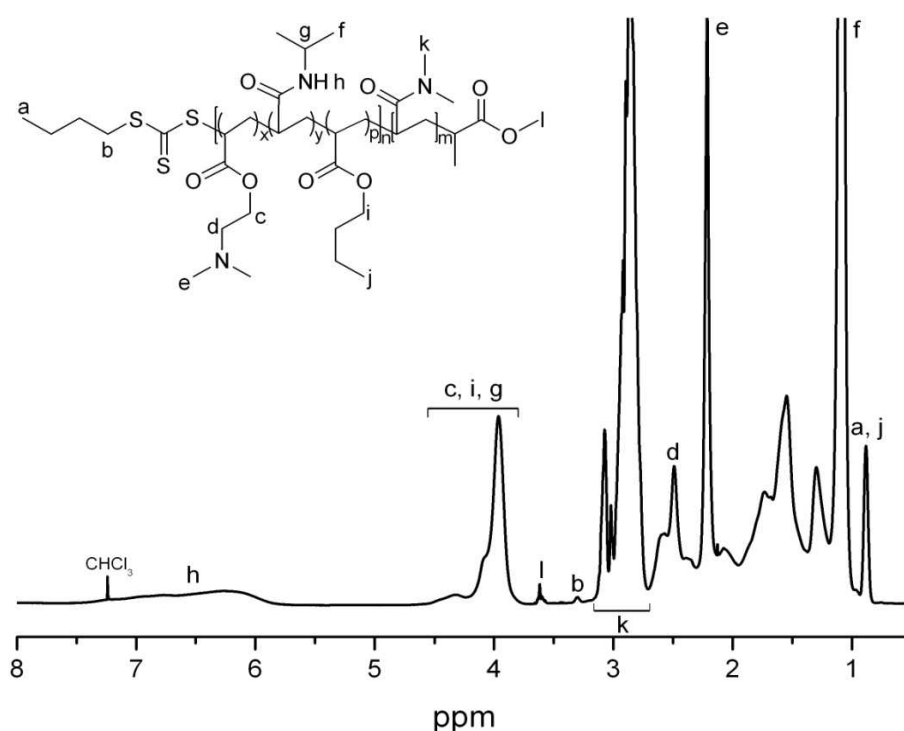


**Figure S3.**  $^1\text{H}$  NMR spectrum of A: P(DMA<sub>96</sub>-b-(NIPAM<sub>87</sub>-co-DMAEA<sub>25</sub>)) in  $\text{CDCl}_3$ .

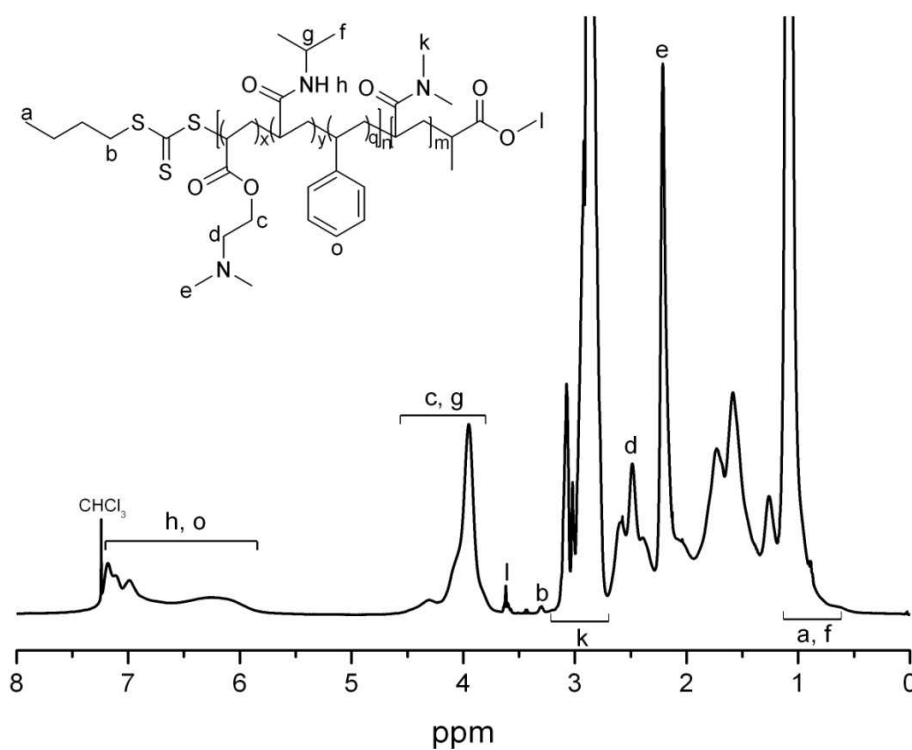


**Figure S4.**  $^1\text{H}$  NMR spectrum of B1: P(DMA<sub>96</sub>-b-(NIPAM<sub>88</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>6</sub>)) in  $\text{CDCl}_3$ .

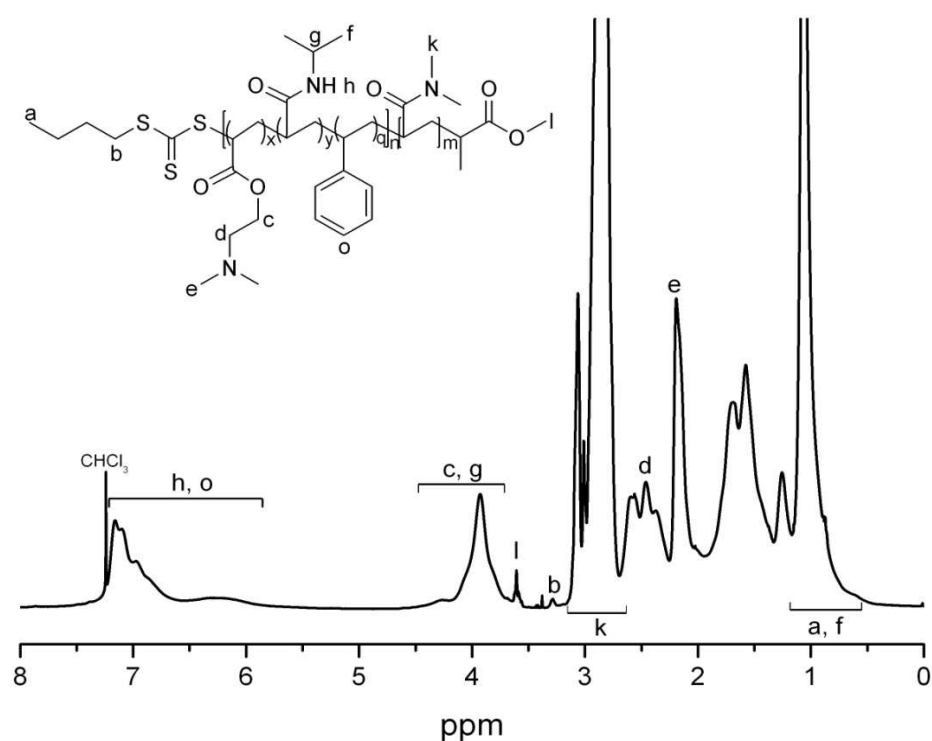




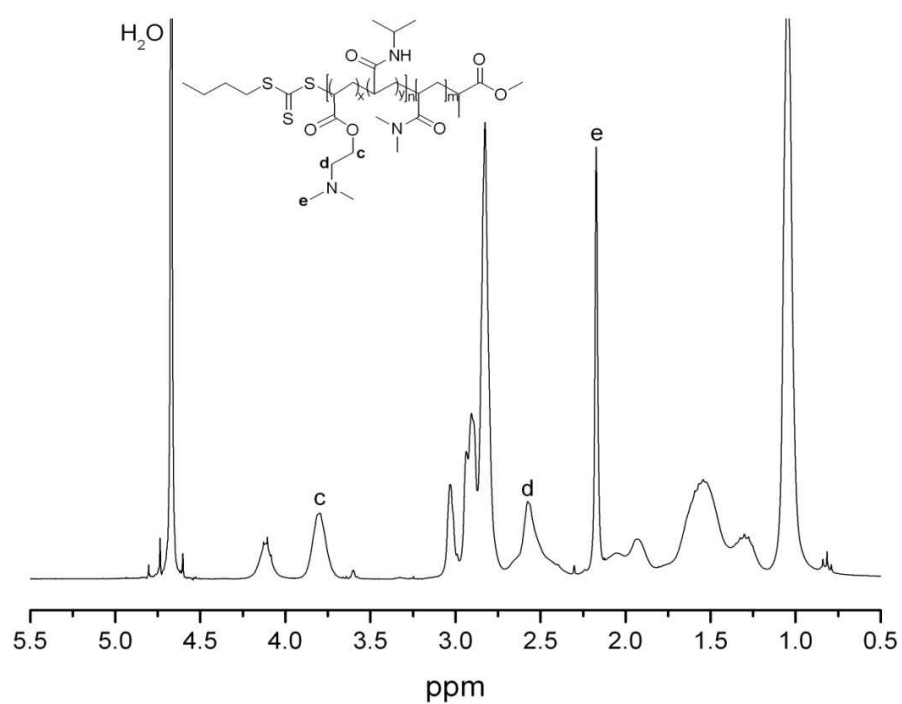
**Figure S5.**  $^1\text{H}$  NMR spectrum of B2: P(DMA<sub>96</sub>-b-(NIPAM<sub>91</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>12</sub>)) in CDCl<sub>3</sub>.



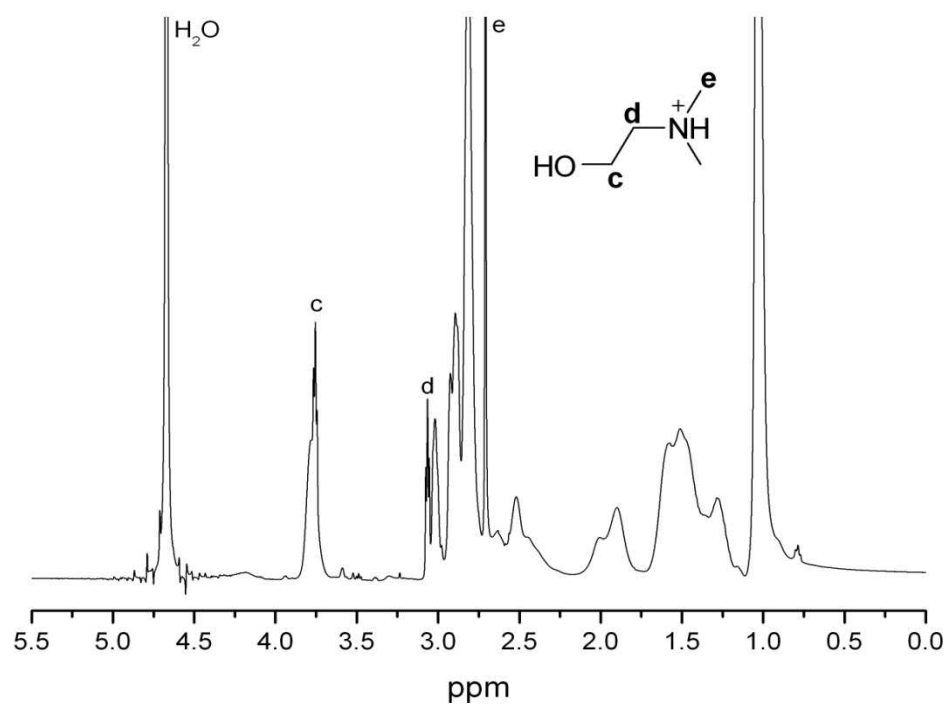
**Figure S6.**  $^1\text{H}$  NMR spectrum of C1: P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>)) in CDCl<sub>3</sub>.



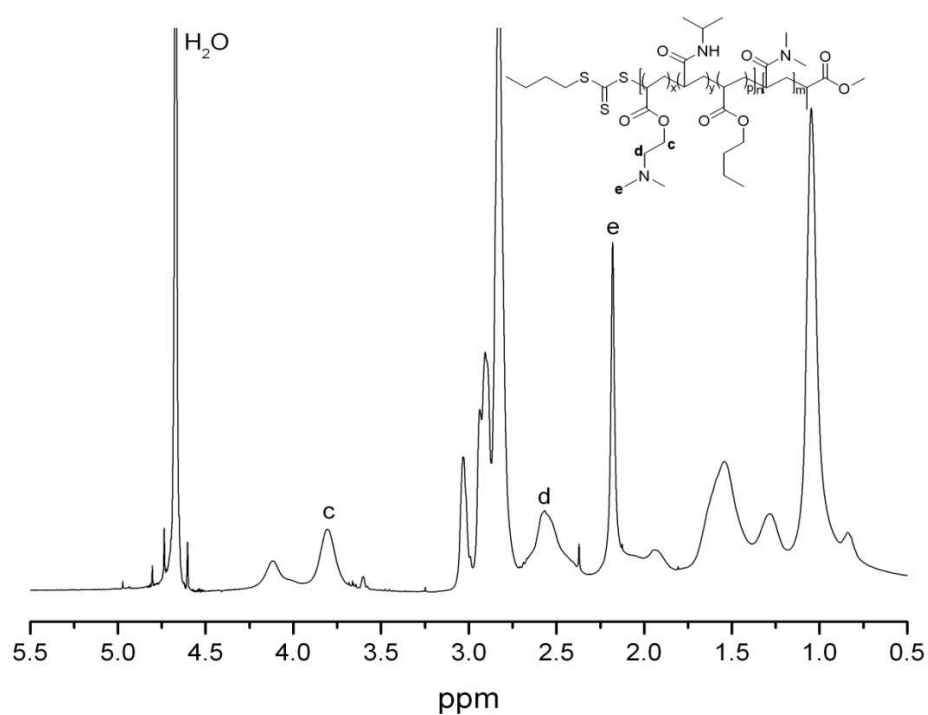
**Figure S7.**  $^1\text{H}$  NMR spectrum of C2: P(DMA<sub>96</sub>-b-(NIPAM<sub>40</sub>-co-DMAEA<sub>15</sub>-co-STY<sub>13</sub>)) in CDCl<sub>3</sub>.



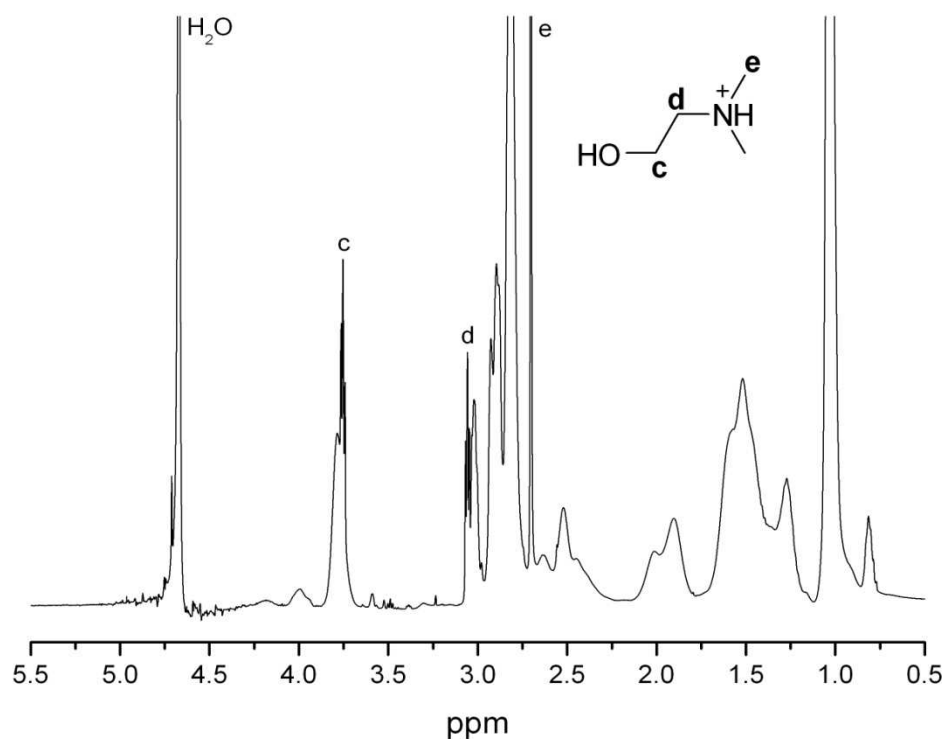
**Figure S8.**  $^1\text{H}$  NMR spectrum of A: P(DMA<sub>96</sub>-b-(NIPAM<sub>87</sub>-co-DMAEA<sub>25</sub>)) after being dissolved in D<sub>2</sub>O and immediately measured at 25 °C.



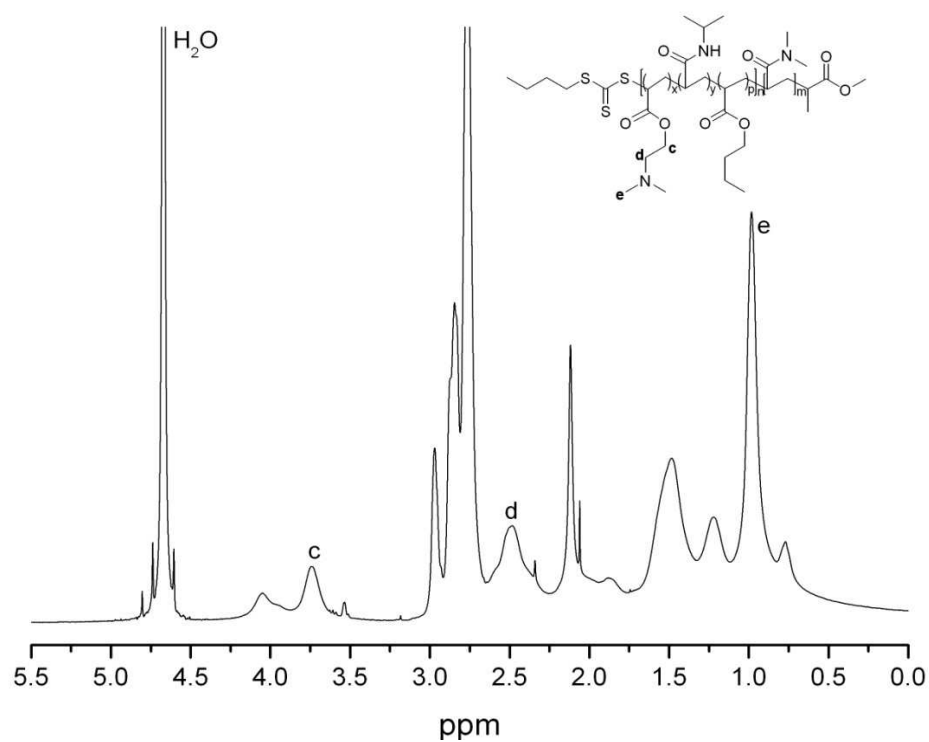
**Figure S9.**  $^1\text{H}$  NMR spectrum of A;  $\text{P}(\text{DMA}_{96}\text{-b-(NIPAM}_{87}\text{-co-DMAEA}_{25}))$  after being dissolved and kept in  $\text{D}_2\text{O}$  for 80 h. The spectrum was measured at 25  $^\circ\text{C}$ .



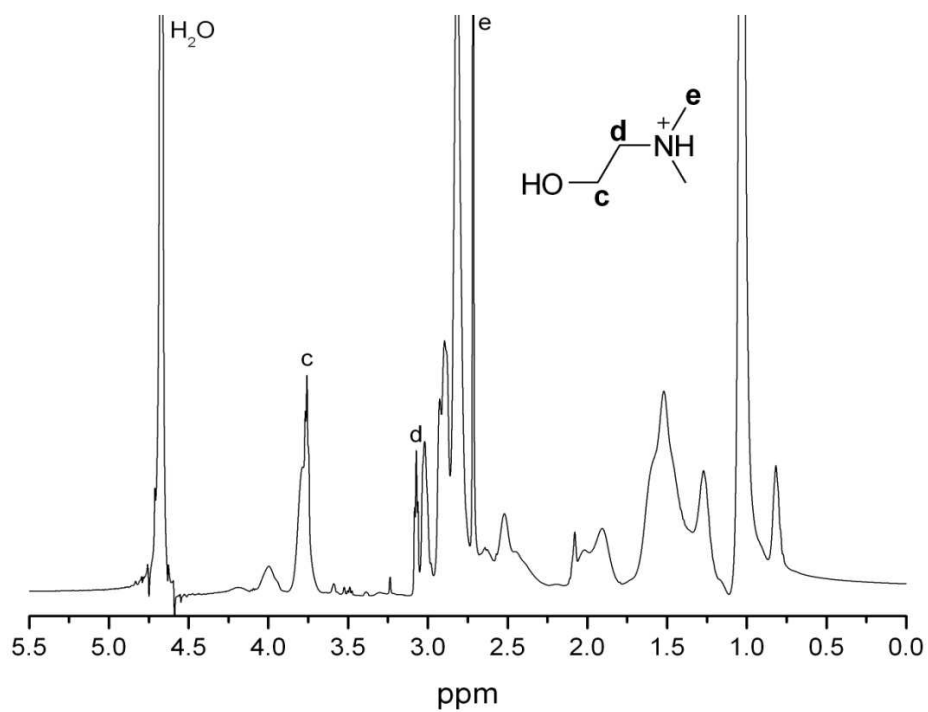
**Figure S10.**  $^1\text{H}$  NMR spectrum of B1:  $\text{P}(\text{DMA}_{96}\text{-b-(NIPAM}_{88}\text{-co-DMAEA}_{25}\text{-co-BA}_6))$  after being dissolved in  $\text{D}_2\text{O}$  and immediately measured at 25  $^\circ\text{C}$ .



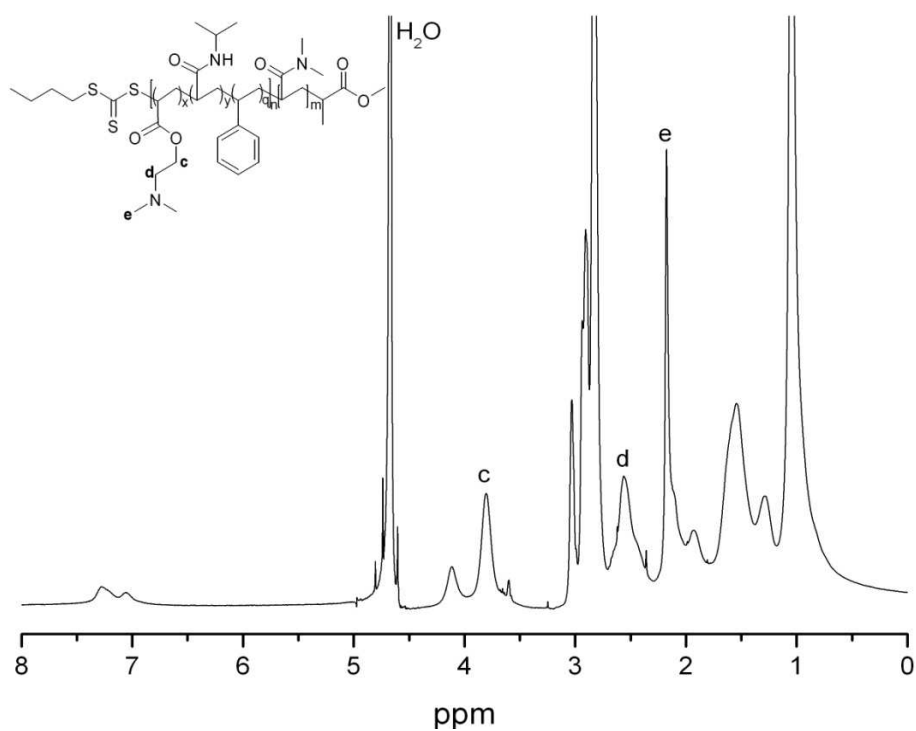
**Figure S11.**  $^1\text{H}$  NMR spectrum of B1:  $\text{P(DMA}_{96}\text{-b-(NIPAM}_{88}\text{-co-DMAEA}_{25}\text{-co-BA}_6\text{))}$  after being dissolved and kept in  $\text{D}_2\text{O}$  for 80 h. The spectrum was measured at 25  $^\circ\text{C}$ .



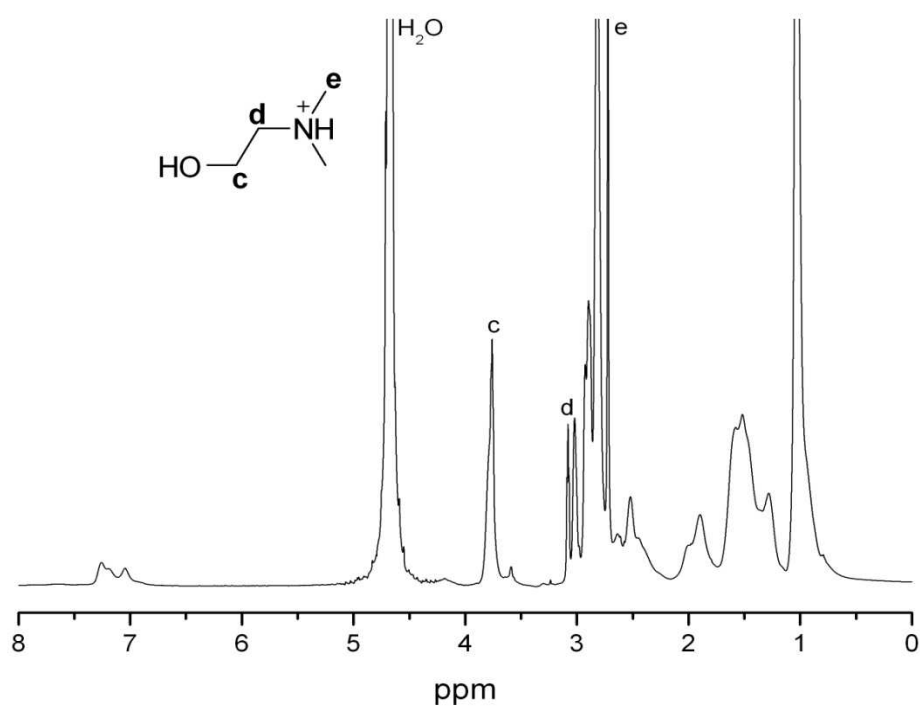
**Figure S12.**  $^1\text{H}$  NMR spectrum of B2:  $\text{P(DMA}_{96}\text{-b-(NIPAM}_{91}\text{-co-DMAEA}_{25}\text{-co-BA}_{12}\text{))}$  after being dissolved in  $\text{D}_2\text{O}$  and immediately measured at 15  $^\circ\text{C}$ .



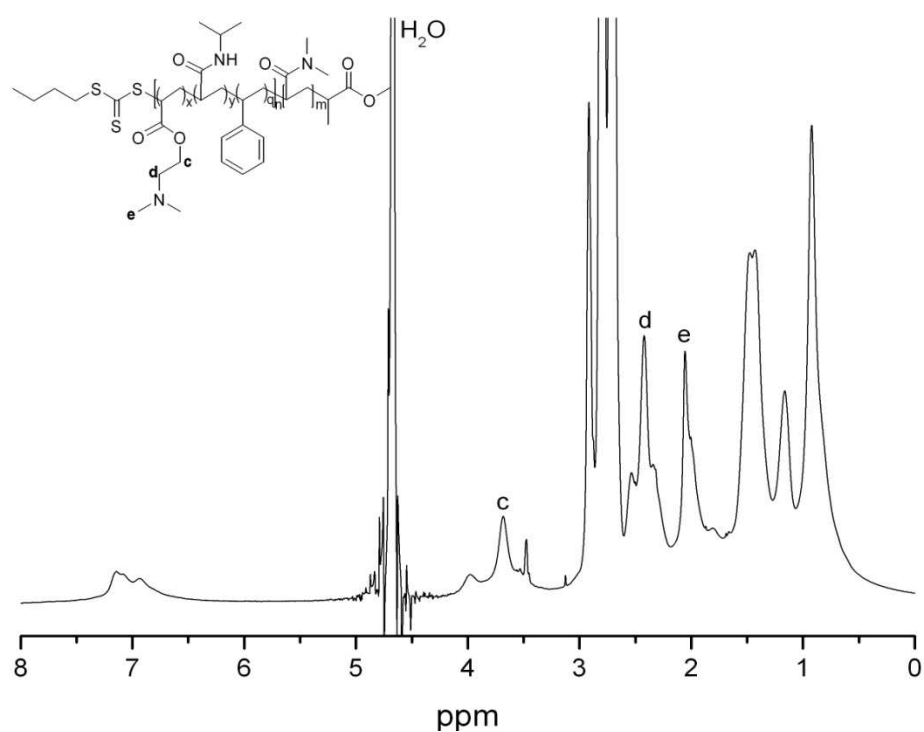
**Figure S13.**  $^1\text{H}$  NMR spectrum of B2: P(DMA<sub>96</sub>-b-(NIPAM<sub>91</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>12</sub>)) after being dissolved and kept in D<sub>2</sub>O for 80 h. The spectrum was measured at 25 °C.



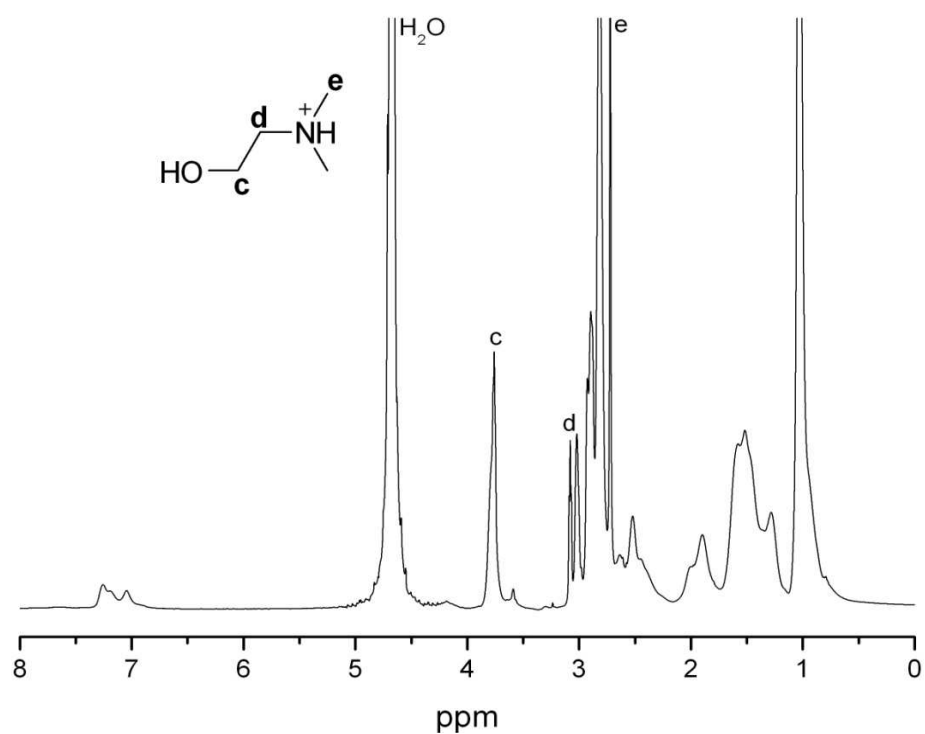
**Figure S14.**  $^1\text{H}$  NMR spectrum of C1: P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>)) after being dissolved in D<sub>2</sub>O and immediately measured at 25 °C.



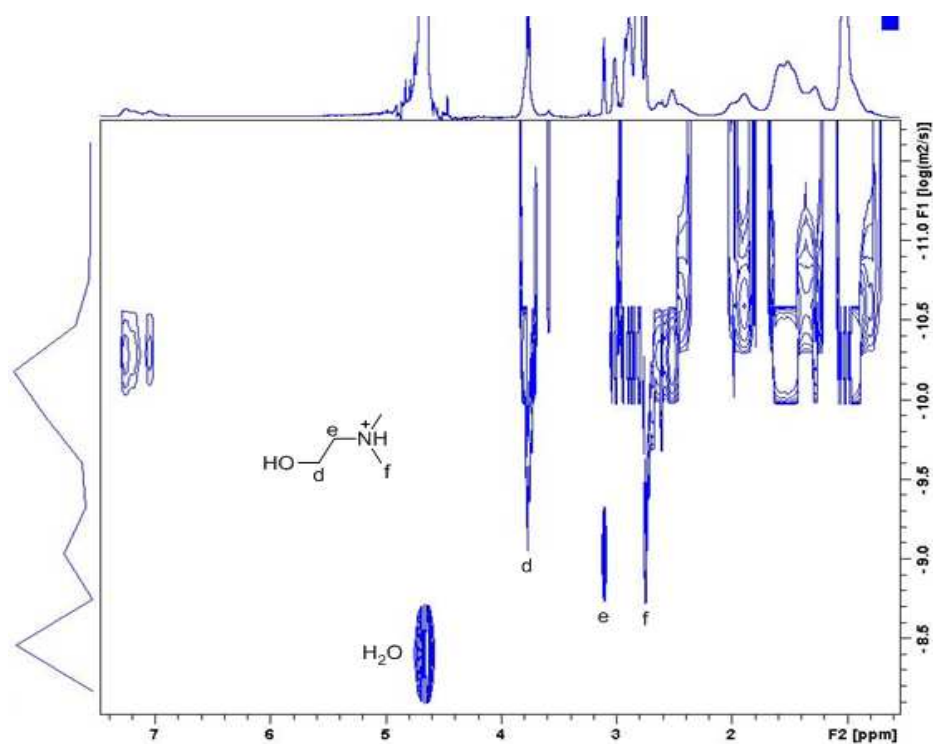
**Figure S15.**  $^1\text{H}$  NMR spectrum of C1:  $\text{P}(\text{DMA}_{96}\text{-b-(NIPAM}_{84}\text{-co-DMAEA}_{22}\text{-co-STY}_5)$  after being dissolved and kept in  $\text{D}_2\text{O}$  for 80 h. The spectrum was measured at  $25\text{ }^\circ\text{C}$ .



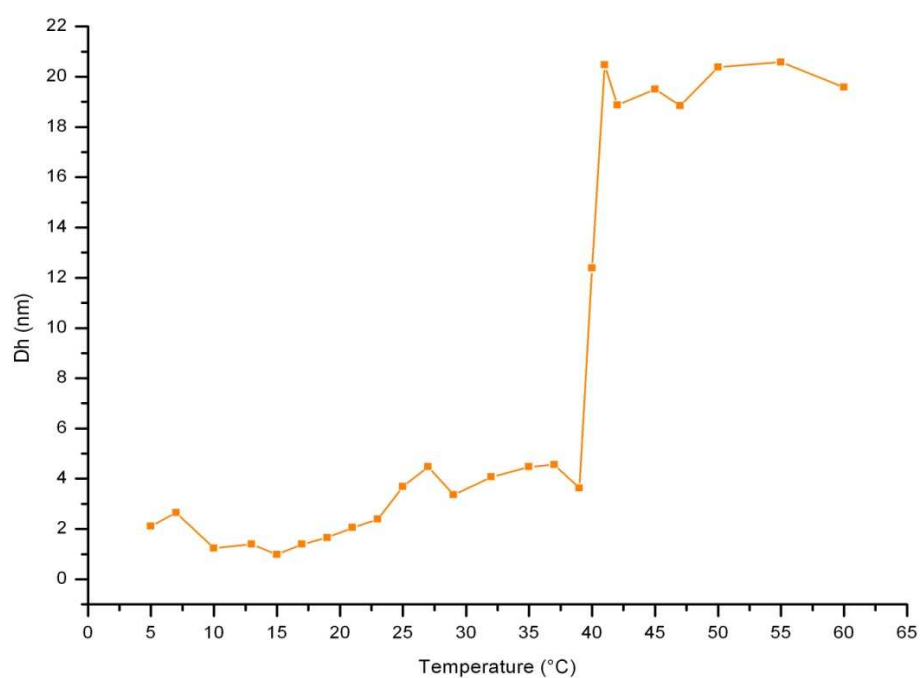
**Figure S16.**  $^1\text{H}$  NMR spectrum of C2:  $\text{P}(\text{DMA}_{96}\text{-b-(NIPAM}_{40}\text{-co-DMAEA}_{15}\text{-co-STY}_{13}))$  after being dissolved in  $\text{D}_2\text{O}$  and immediately measured at  $15\text{ }^\circ\text{C}$ .



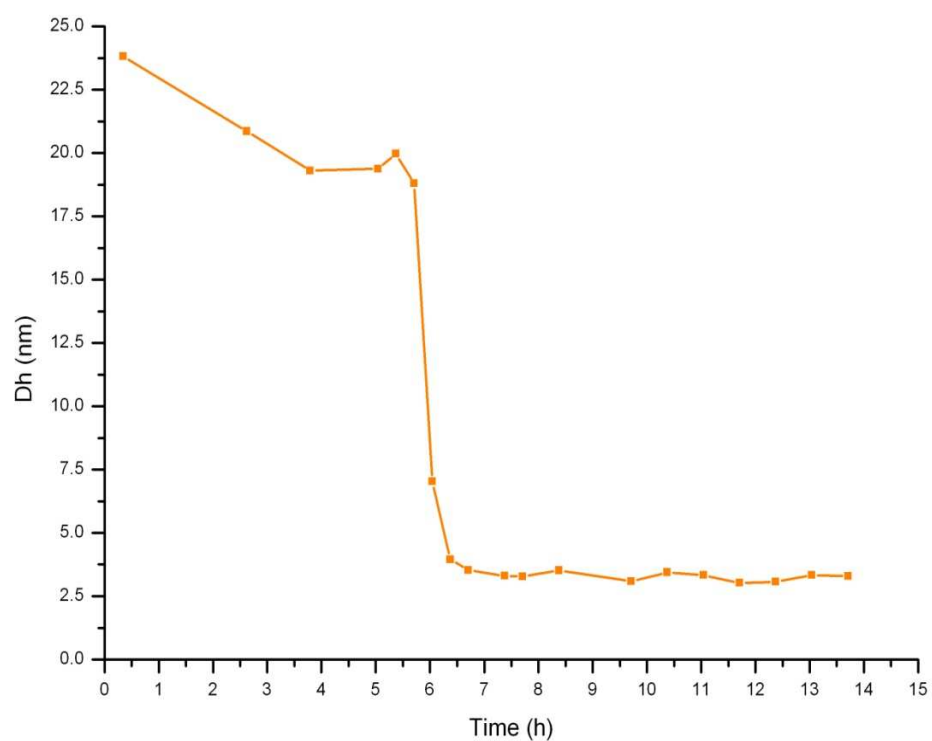
**Figure S17.**  $^1\text{H}$  NMR spectrum of C2: P(DMA<sub>96</sub>-b-(NIPAM<sub>40</sub>-co-DMAEA<sub>15</sub>-co-STY<sub>13</sub>)) after being dissolved and kept in D<sub>2</sub>O for 80 h. The spectrum was measured at 25 °C.



**Figure S18.** DOSY NMR spectrum of C1: P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>)) after being dissolved in D<sub>2</sub>O for 80h. The spectrum was measured at 25 °C.

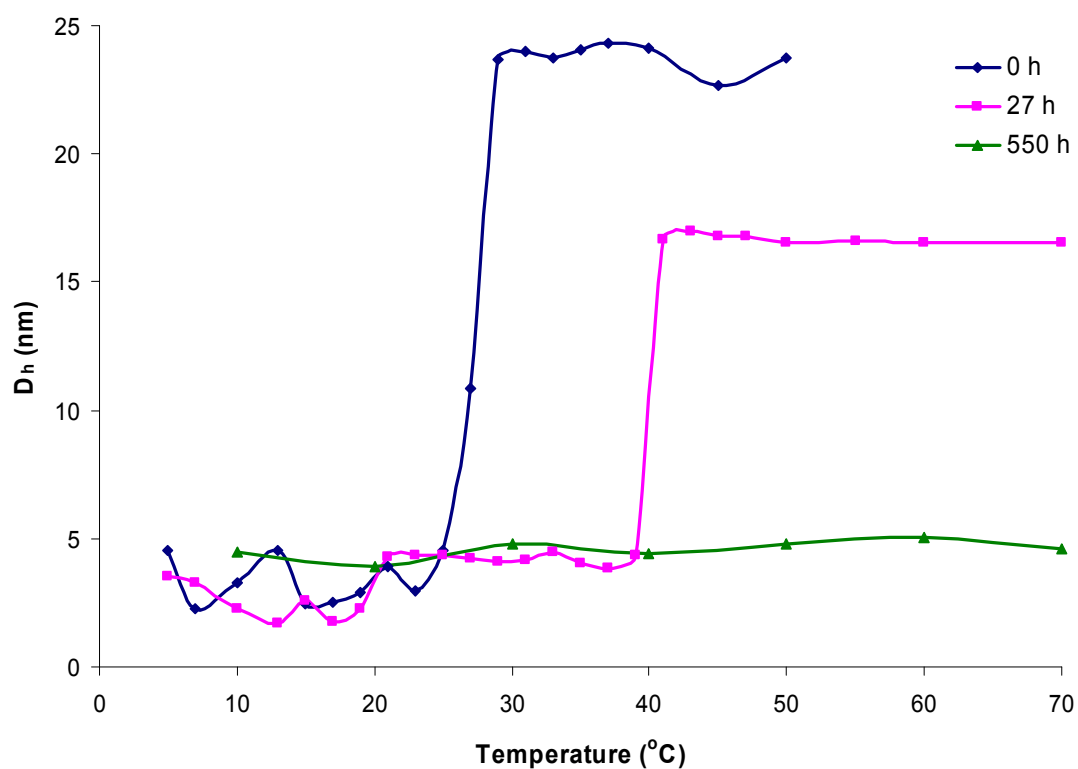


**Figure S19.** LCST of A: P(DMA<sub>96</sub>-b-(NIPAM<sub>87</sub>-co-DMAEA<sub>25</sub>)). The data were reported as average numbers from five measurements on DLS machine at concentration 10 mg/mL.

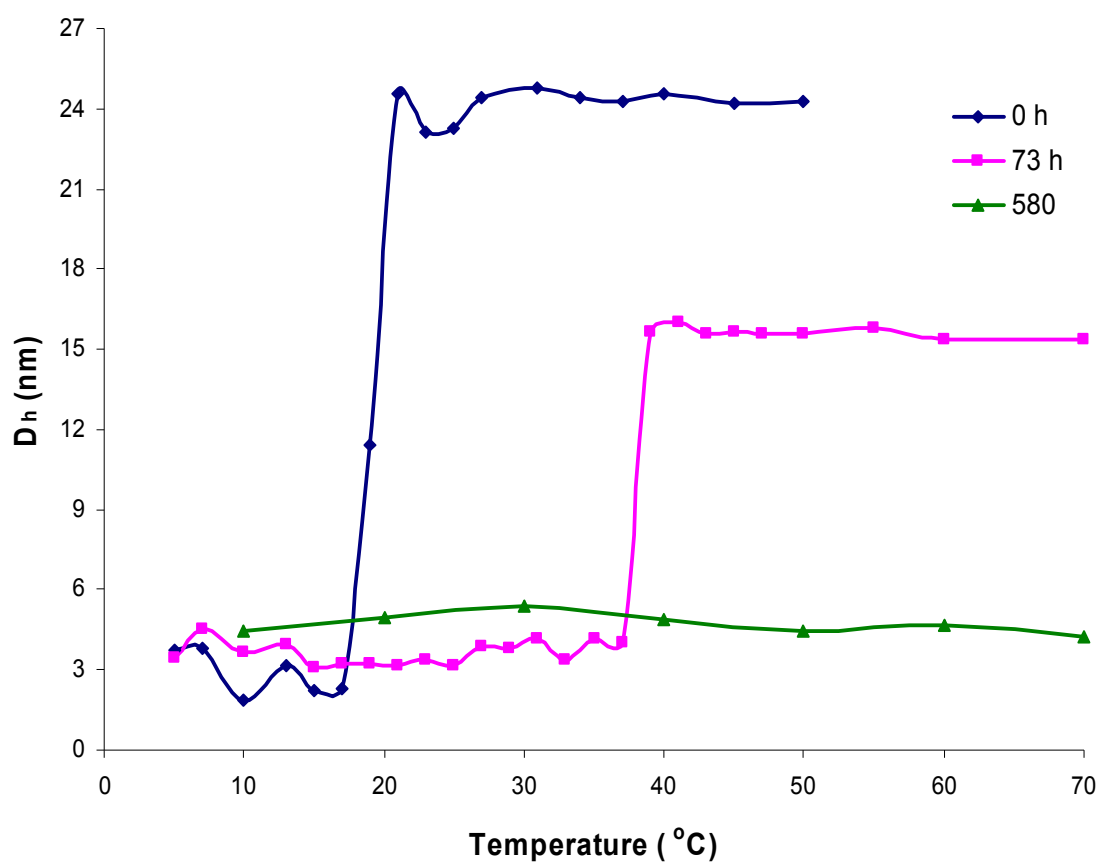


**Figure S20.** Degradation kinetic of A: P(DMA<sub>96</sub>-b-(NIPAM<sub>87</sub>-co-DMAEA<sub>25</sub>)). The data were averaged from five measurements on DLS machine at polymer solution concentration 5 mg/mL at 45 °C.

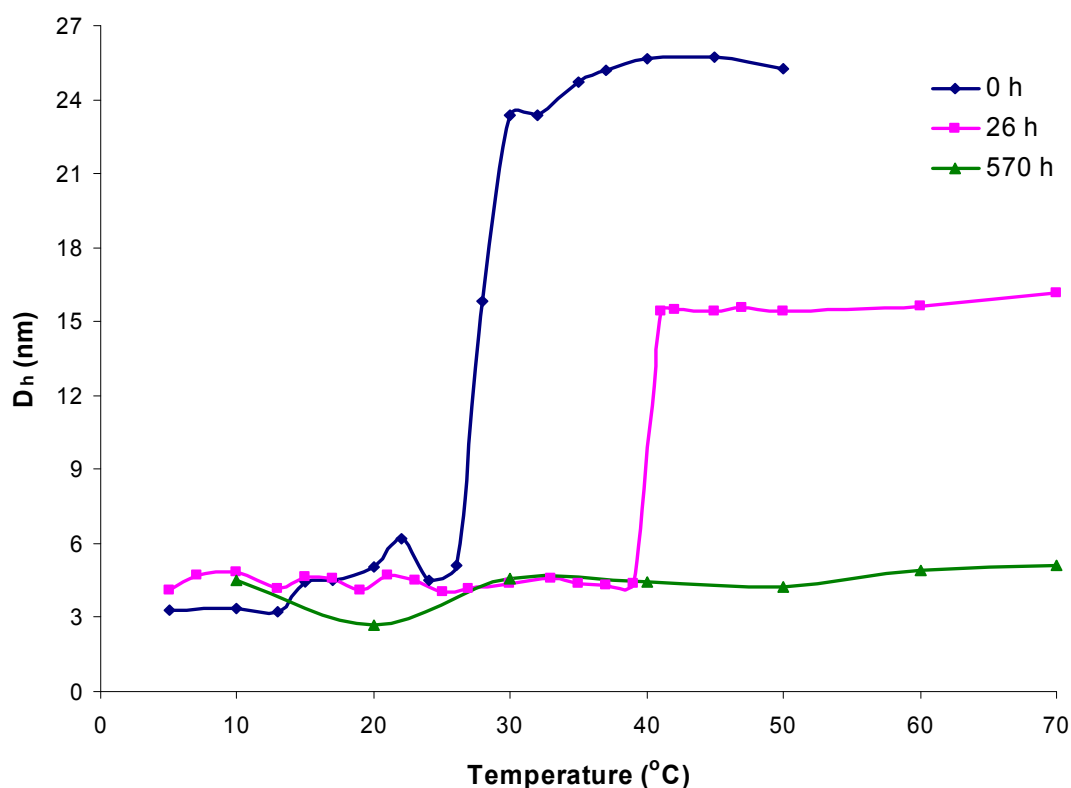




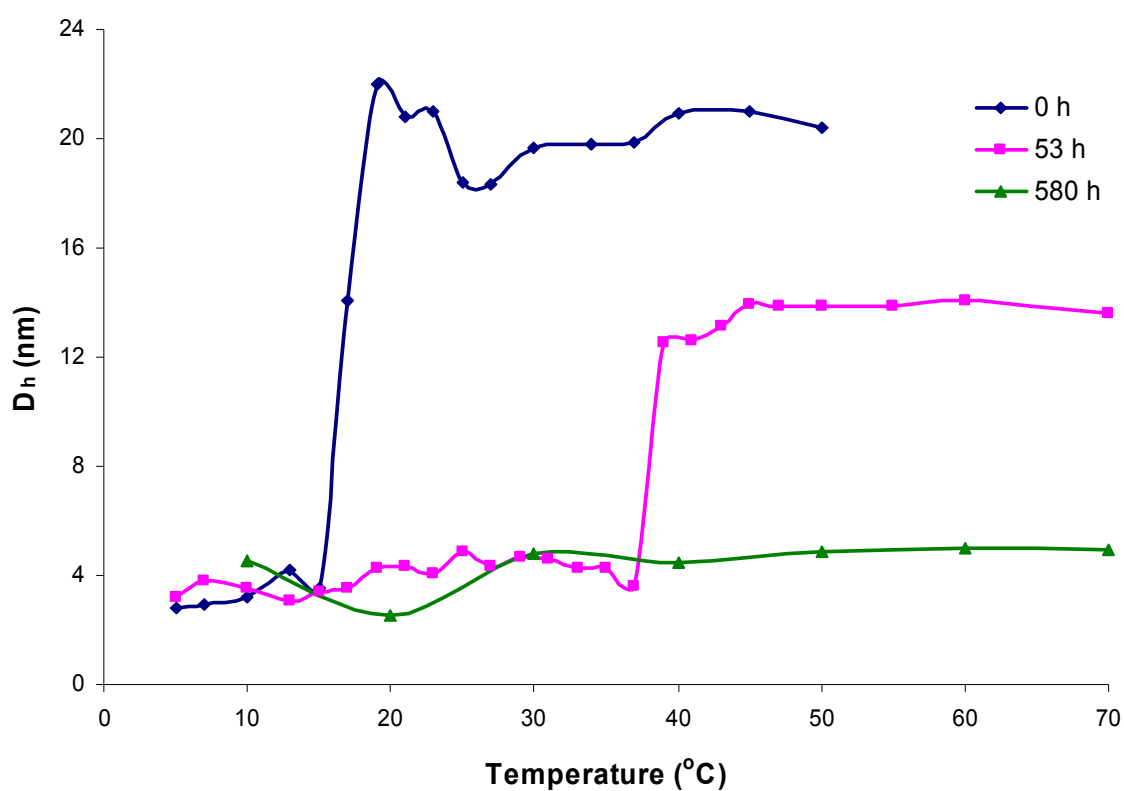
**Figure S21.** LCST of B1: P(DMA<sub>96</sub>-b-(NIPAM<sub>88</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>6</sub>)) at different times after the polymer was dissolved in water. The data were reported as average numbers from five measurements on DLS machine at concentration 10 mg/mL.



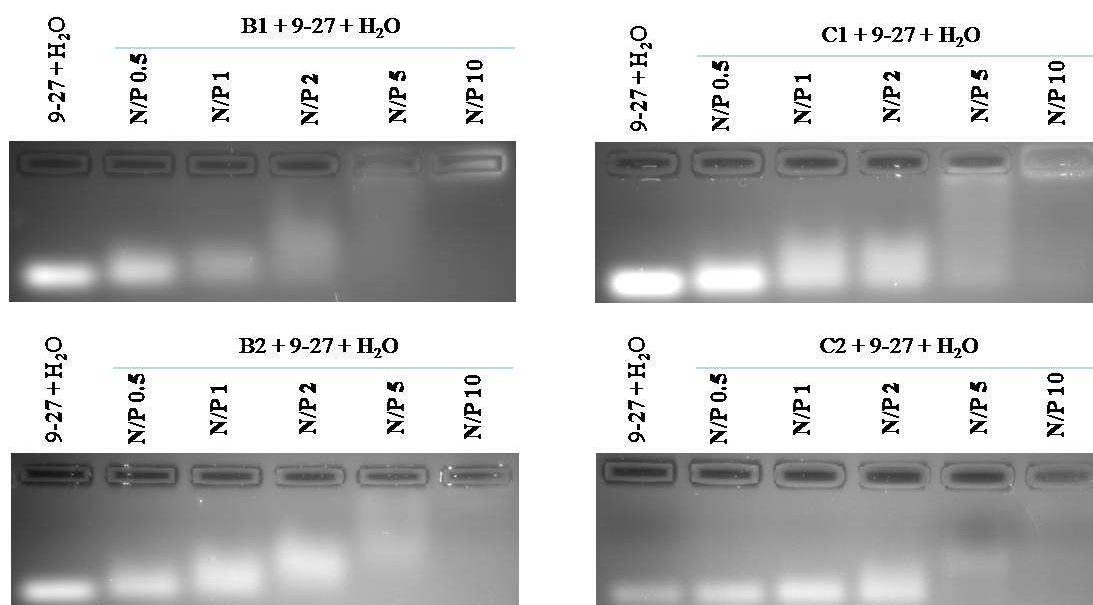
**Figure S22.** LCST of B2: P(DMA<sub>96</sub>-b-(NIPAM<sub>91</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>12</sub>)) at different times after the polymer was dissolved in water. The data were reported as average numbers from five measurements on DLS machine at concentration 10 mg/mL.



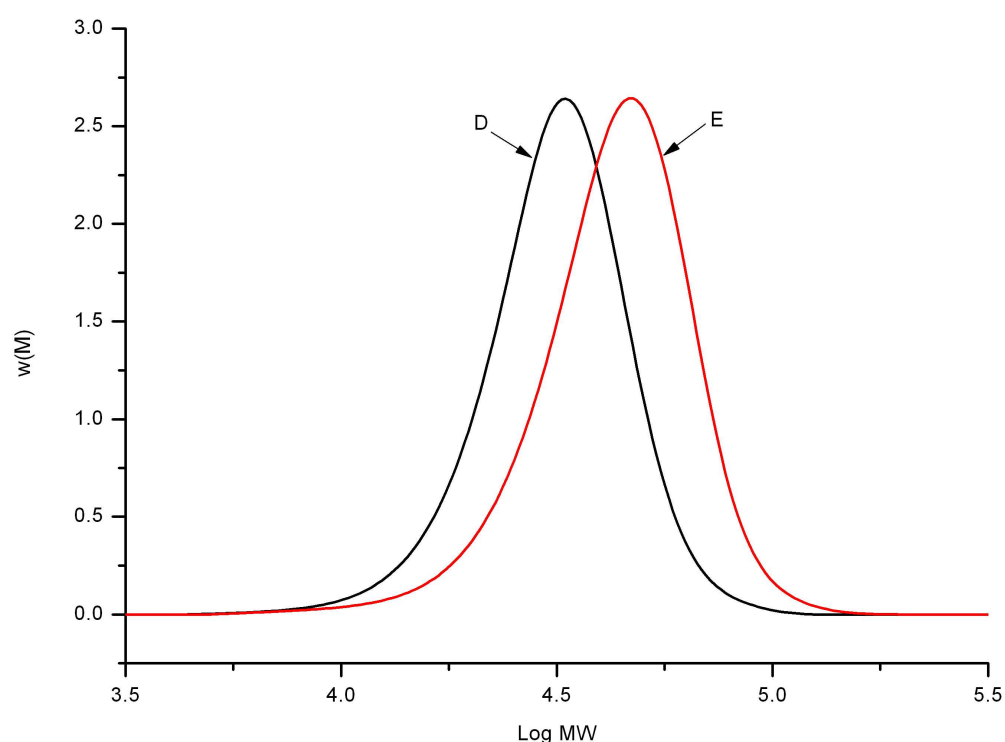
**Figure S23.** LCST of C1: P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>) at different times after the polymer was dissolved in water. The data were reported as average numbers from five measurements on DLS machine at concentration 10 mg/mL



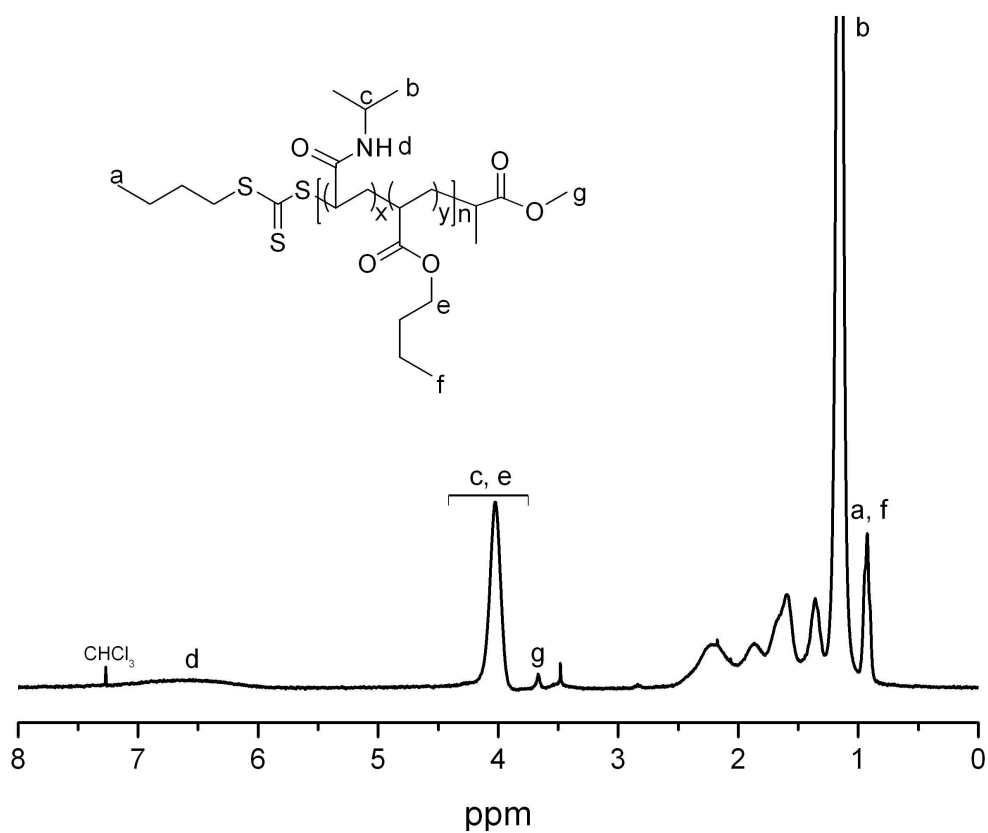
**Figure S24.** LCST of C2: P(DMA<sub>96</sub>-b-(NIPAM<sub>40</sub>-co-DMAEA<sub>15</sub>-co-STY<sub>13</sub>)) at different times after the polymer was dissolved in water. The data were reported as average numbers from five measurements on DLS machine at concentration 10 mg/mL.



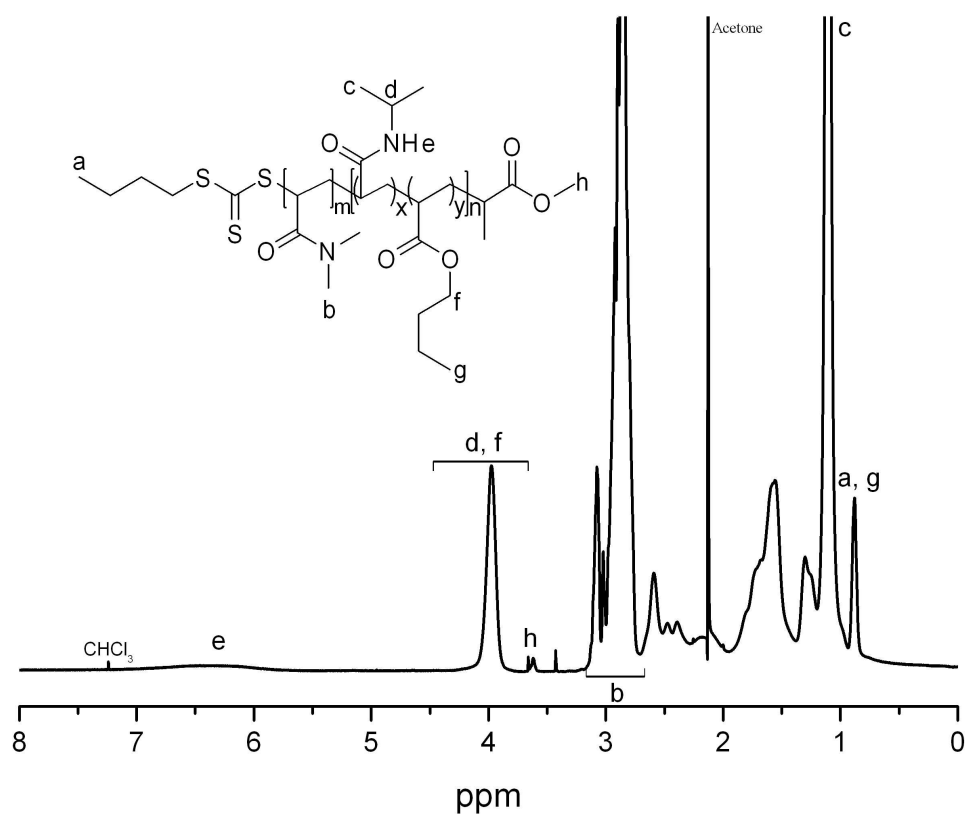
**Figure S25.** Agarose gel assay of Oigo DNA 9-27 / thermoresponsive block copolymers complexes in Milli-Q water at N/P Ratio 0.5, 1, 2, 5, and 10. B1: P(DMA<sub>96</sub>-b-(NIPAM<sub>88</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>6</sub>)), B2: P(DMA<sub>96</sub>-b-(NIPAM<sub>91</sub>-co-DMAEA<sub>25</sub>-co-BA<sub>12</sub>)), C1: P(DMA<sub>96</sub>-b-(NIPAM<sub>84</sub>-co-DMAEA<sub>22</sub>-co-STY<sub>5</sub>)), C2: P(DMA<sub>96</sub>-b-(NIPAM<sub>40</sub>-co-DMAEA<sub>15</sub>-co-STY<sub>13</sub>)). Soluble copolymers in Milli-Q water were incubated with DNA at different N/P ratios at below their LCST for 30 min and then being heated up to 37 °C for 15 min before doing Agarose gel retardation assay.



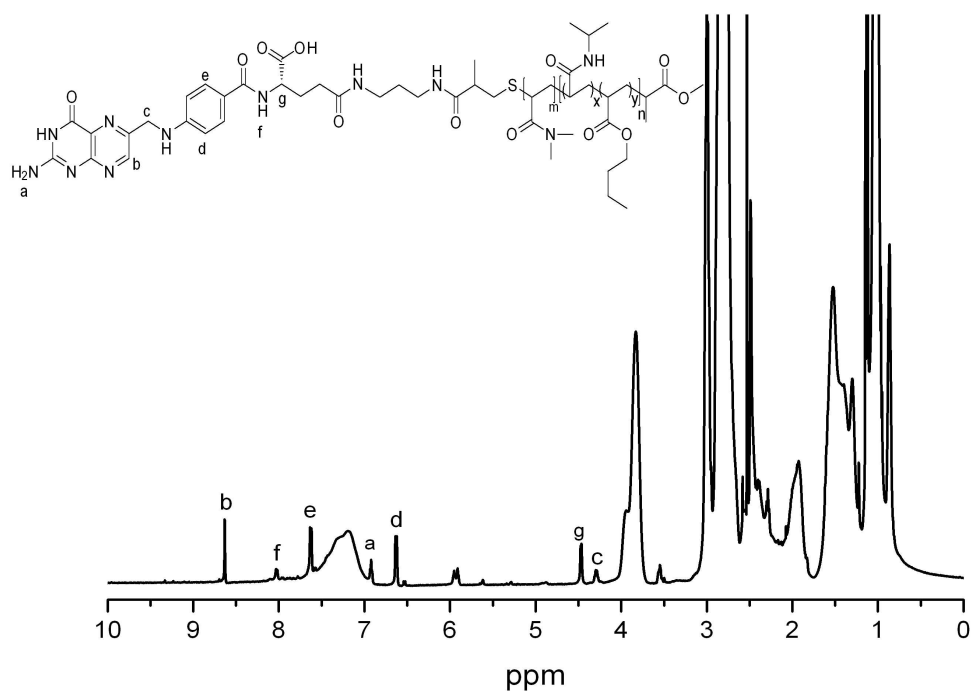
**Figure S26.** Size Exclusion Chromatography (SEC) traces of D: P(NIPAM<sub>97</sub>-co-BA<sub>13</sub>)) and E: P(DMA<sub>99</sub>-b-(NIPAM<sub>97</sub>-co-BA<sub>13</sub>)). The samples were measured by DMAc SEC. The intensity for different distribution curves was normalized.



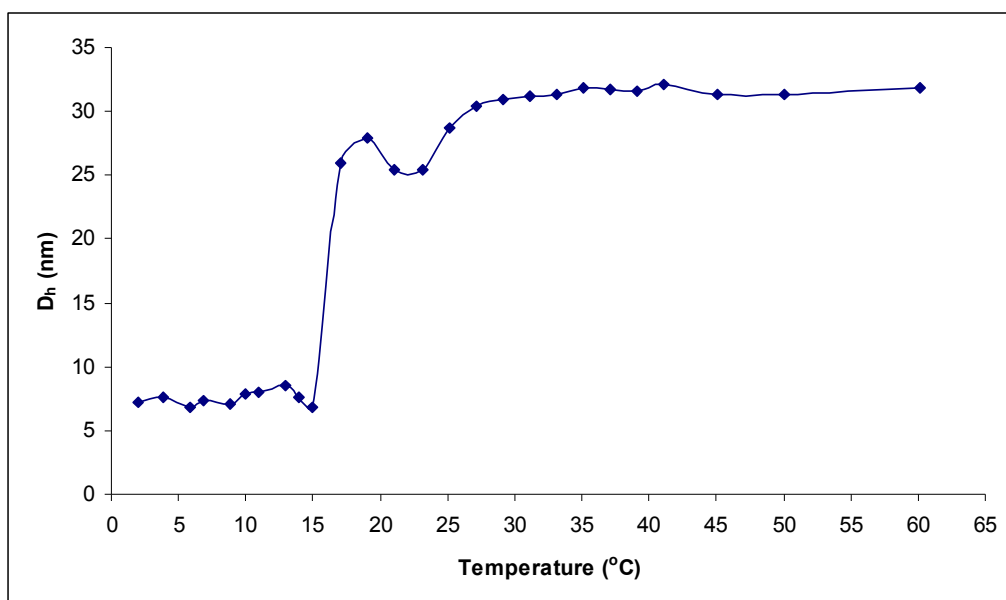
**Figure S27.** <sup>1</sup>H NMR spectrum of **D**: P(NIPAM<sub>97</sub>-co-BA<sub>13</sub>)) in CDCl<sub>3</sub>



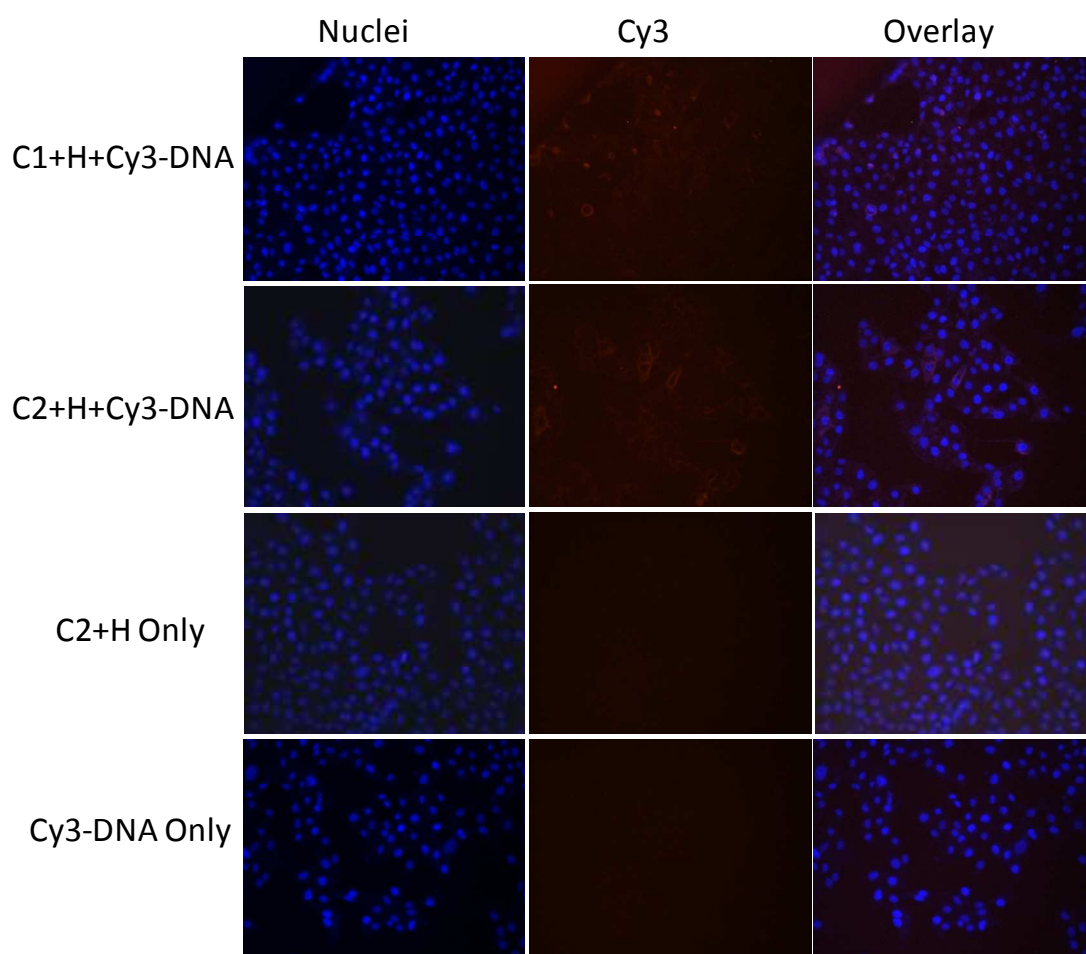
**Figure S28.** <sup>1</sup>H NMR spectrum of **E**: P(DMA<sub>99</sub>-b-(NIPAM<sub>97</sub>-co-BA<sub>13</sub>)) in CDCl<sub>3</sub>



**Figure S29** 1D DOSY NMR spectrum of folic acid conjugated copolymer E, PDMA-b-P(NIPAM-co-BA) in DMSO- $d_6$ .



**Figure S30.** LCST of folic acid conjugated copolymer E, PDMA-b-P(NIPAM-co-BA) in DMSO- $d_6$ . The polymer was dissolved in water. The data were reported as average numbers from five measurements on DLS machine at concentration 10 mg/mL. LCST = 15 – 19 °C.



**Figure S31:** Fluorescent microscopy photos of the osteosarcoma U2OS cells which were dosed with 50 nM Cy3 oligo DNA and copolymer polyplexes and cultured in 24-well plate ( $1 \times 10^5$ /well) in completed DMEM medium. The polyplexes were prepared in the ratio to siRNA as 50:1 in water. The mixtures were incubated in ice-bath for 10 minutes and then at 37°C for 30 minutes. They were then added to the cells and were incubated for 10 hours before washing with PBS buffer and fixation with 4% paraformaldehyde. The cell nuclei were stained with Hoechst 33341 and cell uptake was viewed under fluorescent microscope.

## Reference

1. Urbani, C. N.; Monteiro, M. J. *Macromolecules* **2009**, *42*, 3884-3886.