Supporting Information

In-situ Aggregation of ZnSe Nanoparticles into Supraparticles: Shape Control and Doping Effects

Gaoling Yang,^{\dagger} Haizheng Zhong,^{* \dagger} Ruibin Liu,^{\dagger} Yongfang Li,^{\ddagger} and Bingsuo Zou^{\dagger}

[†] Beijing Key Laboratory of Nanophotonics and Ultrafine Optoelectronic Systems, School of Materials Science & Engineering, Beijing Institute of Technology, 5 Zhongguancun South Street, Haidian District, Beijing, 100081, China

[‡]CAS Key Laboratory of Organic Solids, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China

Sample	Zn Concentration (g/ml)	ZnSe/(ZnSe+Fe ₃ O ₄) molar ratio	Fe Concentration (g/ml)	Fe ₃ O ₄ /(ZnSe+Fe ₃ O ₄) molar ratio
1	2.26	87	0.681	13
2	1.10	73	0.594	27

Table S1. The result of ICP-AES measurements for magnetic Fe₃O₄@ZnSe SPs.

Figures:

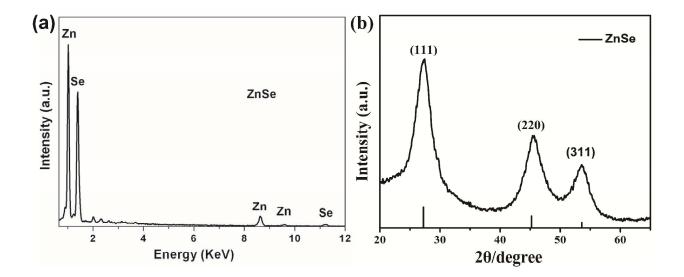


Figure S1. (a) EDS and (b) XRD spectrum of ZnSe SPs.

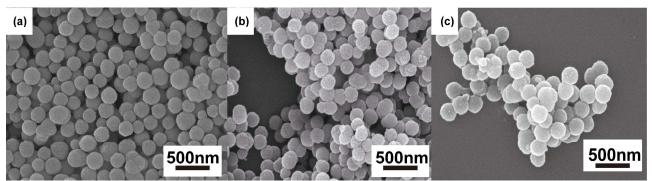


Figure S2. SEM images of magnetic Fe₃O₄@ZnSe SPs with Fe₃O₄/(ZnSe+Fe₃O₄) molar ratio of

(a) 0%, (b) 13% and (c) 27%.



Figure S3. Photograph of ZnSe SPs and SiO₂@ ZnSe SPs.

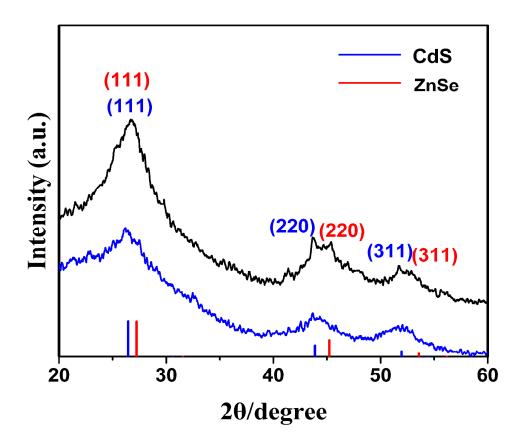


Figure S4. XRD patterns of CdS NPs (black line) and CdS@ZnSe SPs (blue line) and their corresponding standard patterns.