Supporting Information for:

Synthesis and Characterization of a Novel Symmetrical Sulfone-Substituted Polyphenylene Vinylene (SO₂EH-PPV) for Applications in Light Emitting Devices

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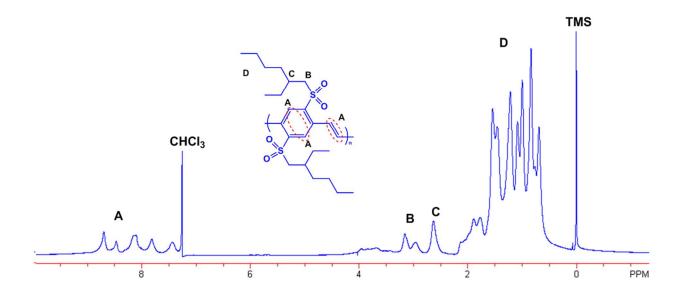


Figure S1. ¹H NMR of polymer (SO₂EH-PPV)

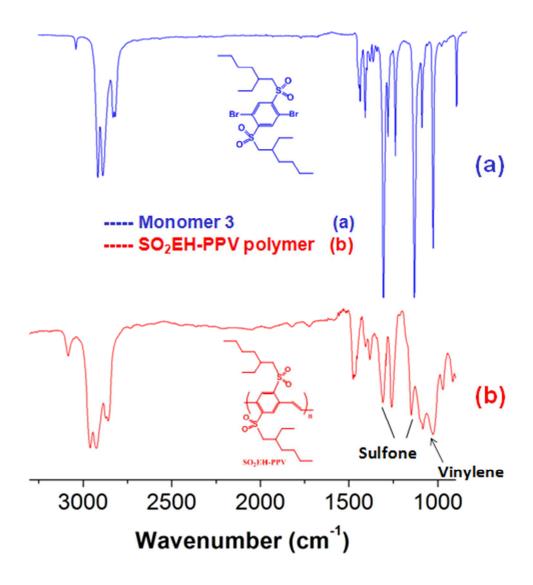


Figure S2. IR spectra of monomer 3 and a polymer (SO₂EH-PPV)

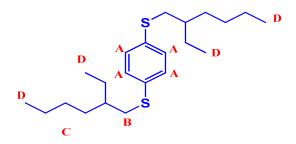
Quantum yield measurements: 45

The quantum yield of fluorescence for the polymer SO_2EH -PPE was determined using Rhodamine 6G (Φ =0.95) as a standard. Quantum yield was determined using the gradients (slope) determined for the sample and the standard. The following equation is used to determine the quantum yield:

$$Q_{S} = Q_{r} \left(\frac{Grad_{S}}{Grad_{r}} \right) \left(\frac{\eta_{S}^{2}}{\eta_{r}^{2}} \right)$$

where Grad represents the slope obtained from the plot of the integrated fluorescence intensity vs. absorbance. Subscripts r and s correspond to the reference and sample respectively. η represents the refractive index of the solvent. The fluorescence of the polymer was obtained using chloroform as a solvent and water was used as a solvent for the reference. Quantum yield was determined by preparing both reference and sample solutions with absorbancies between 0.02 to 0.1. The following table 4 below summarizes the results from quantum yield measurements.

Polymer	Excitation (nm)	Emission max (nm)	Quantum yield (Φ)
SO ₂ EH-PPV	442	532	0.95



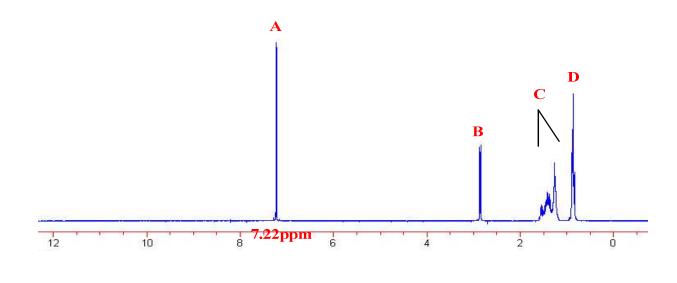
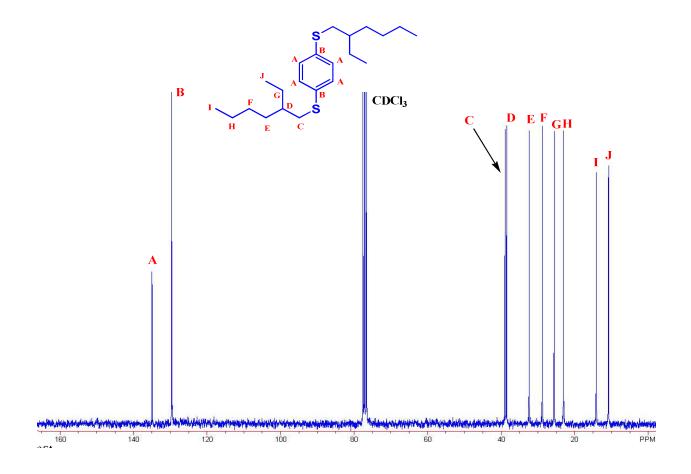


Figure S3. ¹H NMR spectrum of 1,4-bis-(2-ethylhexylthio)benzene (1).

 $\delta\; ppm$



 $\delta\; ppm$

Figure S4. ¹³C NMR spectrum of 1,4-bis-(2-ethylhexylthio)benzene (1).

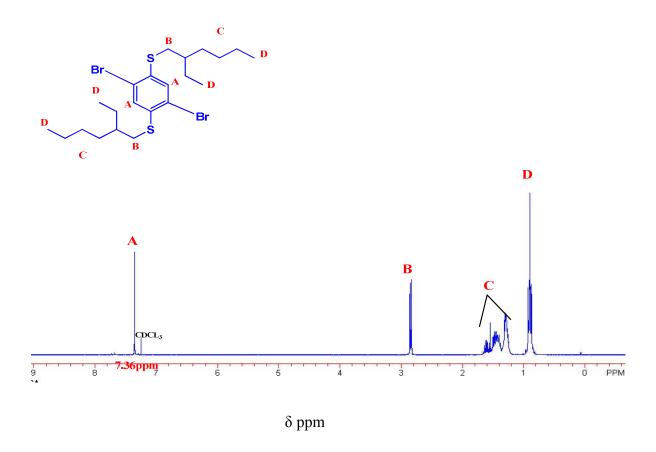


Figure S5. ¹H NMR spectrum of 2,5-dibromo-1,4-bis-(2-ethylhexylthio)benzene (2).

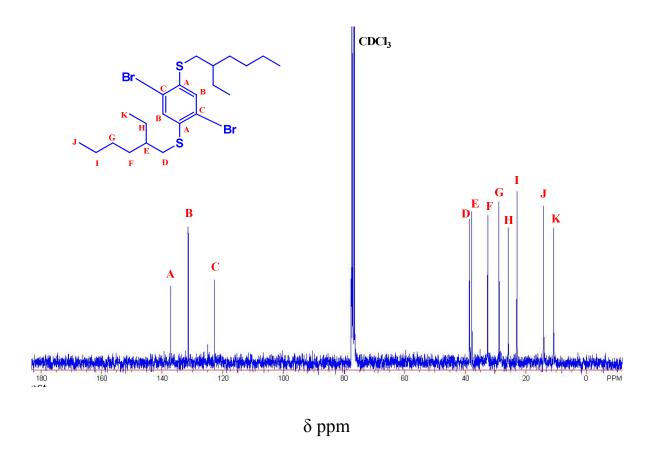


Figure S6. ¹³C NMR spectrum of 2,5-dibromo-1,4-bis-(2-ethylhexylthio)benzene (2).

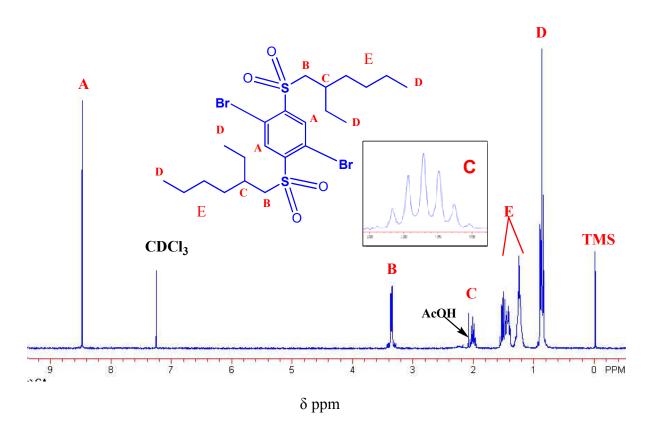


Figure S7. ¹H NMR spectrum of 1,4-dibromo-2,5-bis(2-ethylhexylsulfonyl)benzene (3).

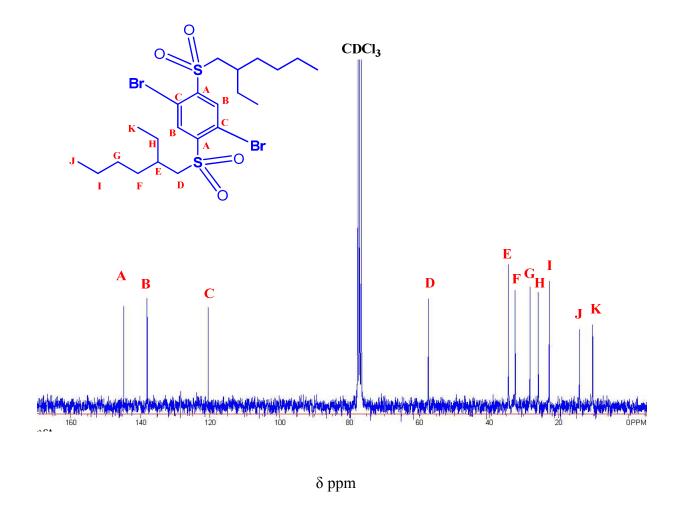


Figure S8. ¹³C NMR spectrum of 1,4-dibromo-2,5-bis(2-ethylhexylsulfonyl)benzene (3).