## Supporting Information for

# Diverse Structural Ag(I) Supramolecular Complexes <br> Constructed from Multidentate Dicyanoisophorone-Based 

 Ligands: Structure and Enhanced LuminescenceFeng Jin, ${ }^{\dagger}{ }^{\dagger}{ }^{\ddagger}$ Ying Zhang, ${ }^{\dagger}$ Hui-Zhen Wang, ${ }^{\dagger}$ Hui-Zhi Zhu, ${ }^{\dagger}$ Yan Yan, ${ }^{\dagger}$ Jun Zhang, ${ }^{\dagger}$ Jie-Ying Wu, ${ }^{\dagger}$ Yu-Peng Tian, ${ }^{\dagger}$ Hong-Ping Zhou ${ }^{*}{ }^{\dagger}$

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L2
Scheme S1 Synthetic route to $\mathbf{L} 1$ and $\mathbf{L 2}$

## Experimental section

## Synthesis of Ligands.

## 2-\{5,5-Dimethyl-3-[2-(pyridin-2-yl)-ethenyl]cyclohex-2-enylidene\}propanedinitril

e (L1) 2-(3, 5, 5-Trimethylcyclohex-2-enylidene)malononitrile ( $2.60 \mathrm{~g}, 14 \mathrm{mmol}$ ), $\mathrm{CH}_{3} \mathrm{CN}$ ( 100 mL ), pyridine-2-carboxaldehyde ( $1.50 \mathrm{~g}, 14 \mathrm{mmol}$ ) and piperidine ( 2 mL ) were added to a round-bottom flask. The reaction mixture was kept stirring for 12 h at $40{ }^{\circ} \mathrm{C}$. After cooled to room temperature, the yellow crystals formed slowly. The product was filtered and recrystallized from MeOH . Yield: $2.90 \mathrm{~g}(75 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 1.03(s, 6H), 2.57(s, 2H), 2.64(s, 2H), 6.95(s,1H), 7.28(d, 2H), 7.64(d, 2H), 7.81~7.87(m, 1H), 8.61(d, 1H); IR $v\left(\mathrm{~cm}^{-1}\right): 2596$ (s), 2931 (s), 2868 (s), 2219 (s), 1567 (s), 1526 (s), 1417 (s), 1401 (s), 1335 (s), 1312 (s), 1212 (s), 1162 (s), 1126 (s), 1024 (s), 965 (s); Anal. Calcd. (\%) for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{3}$ : C 78.45; H 6.17; N 15.25; Found (\%): C 78.48; H 6.21; N 15.28. MS (EI) (m/z): Calc. for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{3}$ : $276.15[\mathrm{HM}]^{+}$; Found: $276.15[\mathrm{HM}]^{+}$.

2-\{5,5-Dimethyl-3-[2-(pyridin-3-yl)-ethenyl]cyclohex-2-enylidene\}propanedinitril e (L2) 2-(3, 5, 5-Trimethylcyclohex-2-enylidene)malononitrile ( $2.60 \mathrm{~g}, 14 \mathrm{mmol}$ ), $\mathrm{CH}_{3} \mathrm{CN}(100 \mathrm{~mL})$, pyridine-3-carboxaldehyde ( 1.50 g , 14 mmol ) and piperidine (2
mL ) were added to a round-bottom flask. The reaction mixture was kept stirring for 12 h at $40^{\circ} \mathrm{C}$, then the solvent was removed with a rotary evaporator to give the black oil. EtOH ( 20 mL ) was added to the black oil, and the yellow solid product formed slowly. The product was filtered and recrystallized from EtOH. Yield: 2.63 g (68\%). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.09(\mathrm{~s}, 6 \mathrm{H}), 2.48(\mathrm{~s}, 2 \mathrm{H}), 2.62(\mathrm{~s}, 2 \mathrm{H}), 6.88(\mathrm{~s}, 1 \mathrm{H}), 7.03(\mathrm{~s}$, $2 H), 7.33 \sim 7.36(\mathrm{~m}, 1 \mathrm{H}), 7.85,7.87(\mathrm{~d}, 1 \mathrm{H}), 8.56 \sim 8.57(\mathrm{t}, 1 \mathrm{H}), 8.70(\mathrm{~d}, 1 \mathrm{H}) ;$ IR $v\left(\mathrm{~cm}^{-1}\right)$ : 2953 (s), 2914 (s), 2870 (s), 2218 (s), 1566 (s), 1530 (s), 1467 (s), 1427 (s), 1390 (s), 1330 (s), 1316 (s), 1156 (s), 972 (s); Anal. Calcd. (\%) for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{3}$ : C 78.45; H 6.17; N 15.25; Found (\%): C78.41; H 6.14; N 15.21. MS (EI) ( $\mathrm{m} / \mathrm{z}$ ): Calc. for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{3}: 276.15[\mathrm{HM}]^{+}$; Found: $276.15[\mathrm{HM}]^{+}$.

Table S1 The data of IR spectra of all compounds

| compound | IR(assignment $)$ |
| :--- | :--- |
| $\mathbf{L 1}$ | $2219(\mathrm{C} \equiv \mathrm{N}$ stretching $)$ |
| $\mathbf{1}$ | $2216(\mathrm{C} \equiv \mathrm{N}$ stretching $) ; 1182 / 1030 / 1007\left(\mathrm{TsO}^{-}\right)$ |
| $\mathbf{2}$ | $2252(\mathrm{C} \equiv \mathrm{N}$ stretching $) ; 1092\left(\mathrm{ClO}_{4}{ }^{-}\right)$ |
| $\mathbf{3}$ | $2220(\mathrm{C} \equiv \mathrm{N}$ stretching $) ; 1337 / 1317\left(\mathrm{NO}_{3}{ }^{-}\right)$ |
| $\mathbf{L 2}$ | $2218(\mathrm{C} \equiv \mathrm{N}$ stretching $)$ |
| $\mathbf{4}$ | $2249 / 2219(\mathrm{C} \equiv \mathrm{N}$ stretching $) ; 1388 / 1300\left(\mathrm{NO}_{3}{ }^{-}\right)$ |
| $\mathbf{5}$ | $2254 / 2221(\mathrm{C} \equiv \mathrm{N}$ stretching $) ; 1090\left(\mathrm{ClO}_{4}{ }^{-}\right)$ |
| $\mathbf{6}$ | $2217(\mathrm{C} \equiv \mathrm{N}$ stretching $) ; 841\left(\mathrm{PF}_{6}{ }^{-}\right)$ |

Table S2 Selected Bond Lengths ( $(\AA)$ and Angles $\left({ }^{\circ}\right)$ of Complexes 1-6

| $\mathbf{1}$ |  |  |  |
| :--- | :--- | :--- | :--- |
| $\mathrm{N}(2)-\mathrm{Ag}(1) \# 1$ | $2.193(3)$ | $\mathrm{Ag}(1)-\mathrm{N}(1)$ | $2.190(3)$ |
| $\mathrm{Ag}(1)-\mathrm{N}(2) \# 1$ | $2.193(3)$ | $\mathrm{Ag}(1)-\mathrm{O}(4)$ | $2.465(3)$ |
| $\mathrm{Ag}(1)-\mathrm{O}(2)$ | $2.555(3)$ | $\mathrm{N}(2) \# 1-\mathrm{Ag}(1)-\mathrm{O}(4)$ | $95.3(1)$ |
| $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{N}(2) \# 1$ | $147.6(1)$ | $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{O}(4)$ | $114.0(1)$ |
| $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{O}(2)$ | $99.4(1)$ | $\mathrm{N}(2) \# 1-\mathrm{Ag}(1)-\mathrm{O}(2)$ | $97.9(1)$ |
| $\mathrm{O}(4)-\mathrm{Ag}(1)-\mathrm{O}(2)$ | $81.9(1)$ |  |  |
| $\mathbf{2}$ |  | $\mathrm{Ag}(2)-\mathrm{N}(3)$ | $2.189(4)$ |
| $\mathrm{Ag}(1)-\mathrm{N}(1)$ | $2.170(4)$ | $\mathrm{N}(1) \#-\mathrm{Ag}(1)-\mathrm{O}(1) \# 3$ | $88.6(2)$ |
| $\mathrm{Ag}(1)-\mathrm{O}(1) \# 2$ | $2.592(6)$ | $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{O}(1) \# 3$ | $88.6(2)$ |
| $\mathrm{Ag}(2)-\mathrm{N}(2) \# 4$ | $2.145(4)$ |  | $91.4(2)$ |
| $\mathrm{N}(1) \# 1-\mathrm{Ag}(1)-\mathrm{O}(1) \# 2$ | $91.4(2)$ |  |  |
| $\mathrm{N}(2) \# 4-\mathrm{Ag}(2)-\mathrm{N}(3)$ | $167.0(2)$ |  |  |


| $\mathbf{3}$ |  |  |  |
| :--- | :--- | :--- | :--- |
| $\mathrm{Ag}(1)-\mathrm{N}(1)$ | $2.159(3)$ | $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{N}(1) \# 1$ | $180.0(1)$ |
| $\mathrm{Ag}(1)-\mathrm{N}(2)$ | $2.398(3)$ | $\mathrm{N}(2) \# 2-\mathrm{Ag}(2)-\mathrm{N}(2)$ | $99.0(2)$ |
| $\mathrm{Ag}(2)-\mathrm{O}(4)$ | $\mathrm{N}(2)-\mathrm{Ag}(2)-\mathrm{O}(4)$ | $134.84(9)$ |  |
| $\mathrm{O}(4) \# 2-\mathrm{Ag}(2)-\mathrm{O}(4)$ | $105.3(1)$ | $\mathrm{N}(2) \# 2-\mathrm{Ag}(2)-\mathrm{O}(4)$ | $94.8(1)$ |
| $\mathbf{4}$ |  |  |  |
| $\mathrm{Ag}(1)-\mathrm{N}(3)$ | $\mathrm{Ag}(1)-\mathrm{N}(1)$ | $2.218(3)$ |  |
| $\mathrm{Ag}(1)-\mathrm{O}(1)$ | $\mathrm{N}(3)-\mathrm{Ag}(1)-\mathrm{O}(1)$ | $115.6(1)$ |  |
| $\mathrm{N}(3)-\mathrm{Ag}(1)-\mathrm{N}(1)$ | $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{O}(1)$ | $99.1(1)$ |  |
| $\mathbf{5}$ | $2.530(3)$ |  |  |
| $\mathrm{Ag}-\mathrm{N}(1)$ | $\mathrm{N}(1) \# 1-\mathrm{Ag}-\mathrm{N}(1)$ | $180.000(1)$ |  |
| $\mathbf{6}$ |  |  |  |
| $\mathrm{Ag}(1)-\mathrm{N}(1)$ | $2.154(3)$ | $\mathrm{Ag}(1)-\mathrm{N}(2)$ | $2.182(4)$ |
| $\mathrm{N}(1)-\mathrm{Ag}(1)-\mathrm{N}(2)$ |  |  |  |



Figure S1 The crossed double-chain of $\mathbf{2}$ in the 2D structure.


Figure S2 The double-sheet of complex $\mathbf{6}$ formed by C-H $\cdots \mathrm{N}$ interactions. The red dotted lines represent the weak interactions. The hydrogen atoms are omitted for clarity.


Figure S3 (a) Time-resolved fluorescence curves of L1 and complexes 1-3 in the solid state at room temperature; (b) Time-resolved fluorescence curves of $\mathbf{L} 2$ and complexes 4-6 in the solid state at room temperature.

