

Supporting Information

Complexes with Redox-Active Ligands: Synthesis, Structure, Electrochemical and Photophysical Behavior of Ru(II) Complex with TTF-Annulated Phenanthroline

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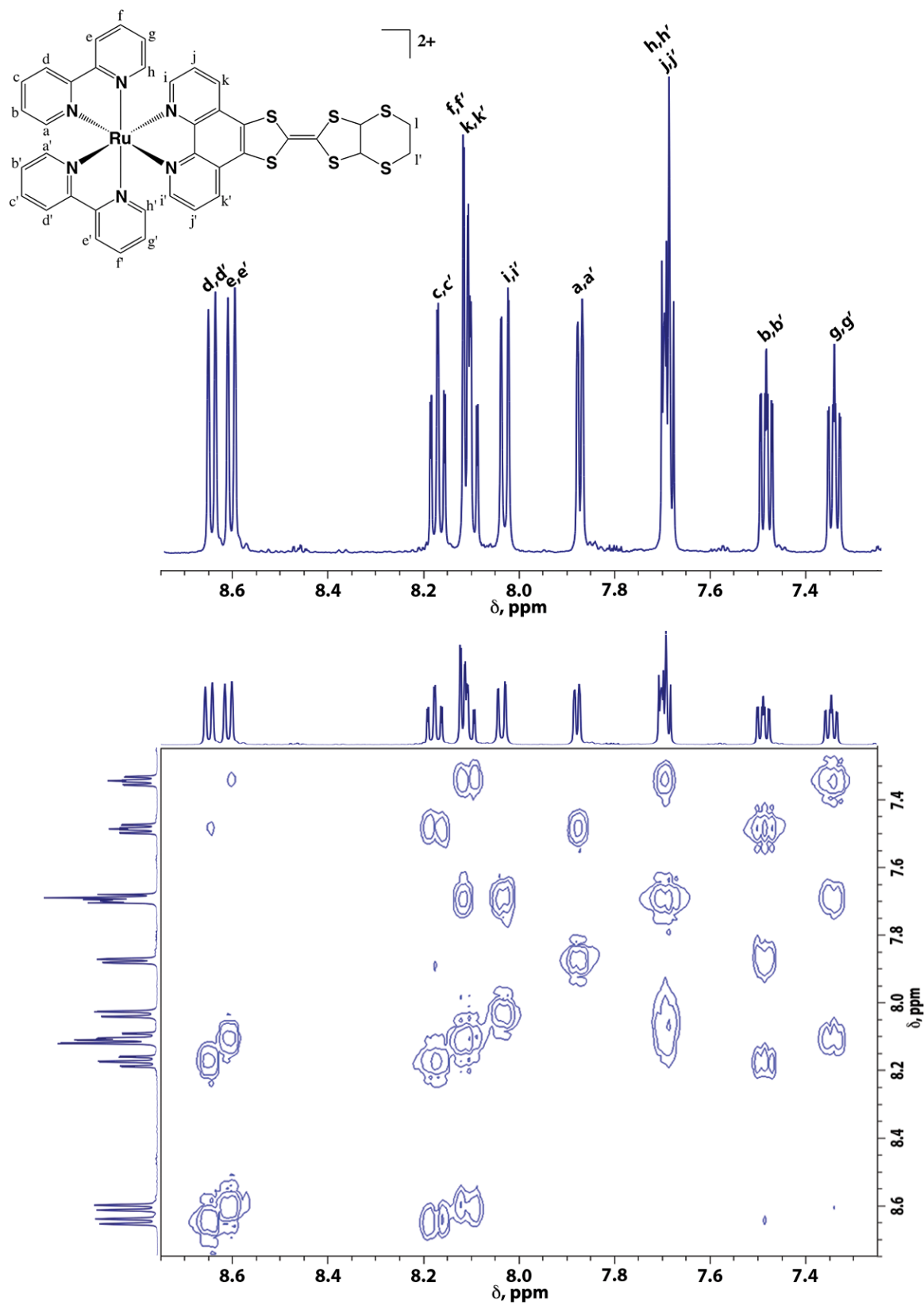


Figure S1. ^1H and 2D COSY NMR spectra of **1**, **2**, and **3** in CDCl_3 at room temperature.

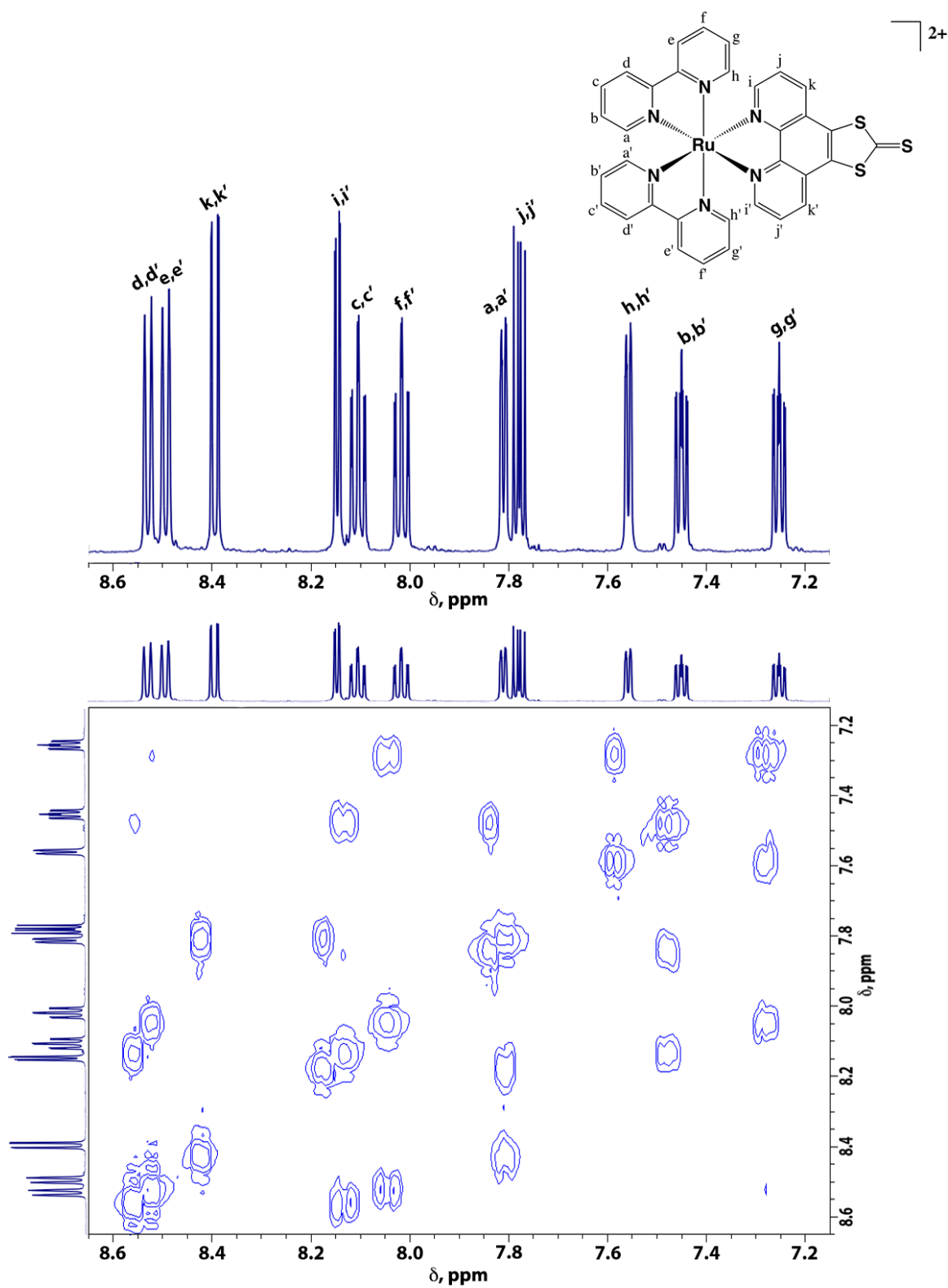


Figure S1 (continued). ^1H and 2D COSY NMR spectra of **1**, **2**, and **3** in CDCl_3 at room temperature.

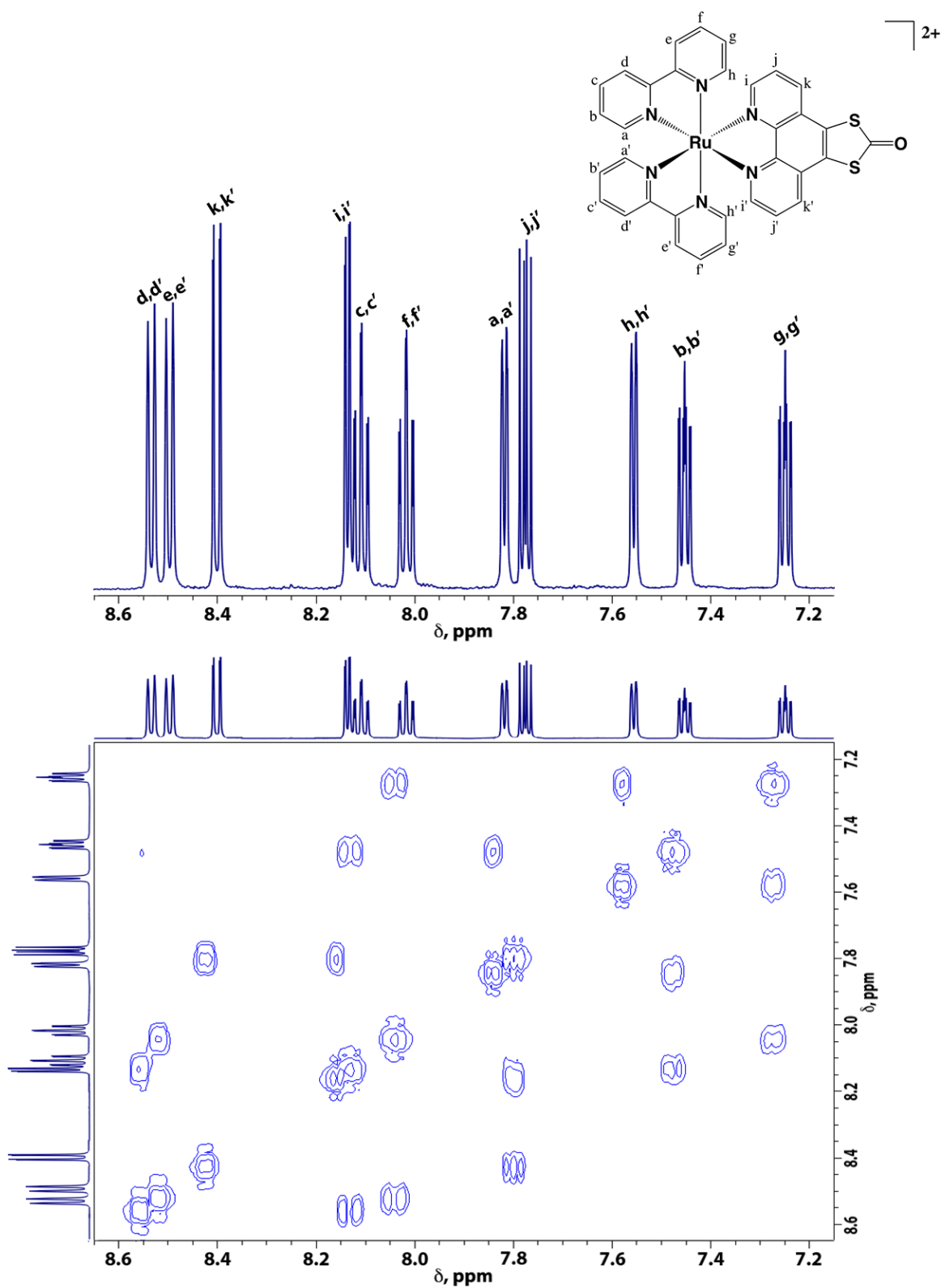


Figure S1 (continued). ^1H and 2D COSY NMR spectra of **1**, **2**, and **3** in CDCl_3 at room temperature.

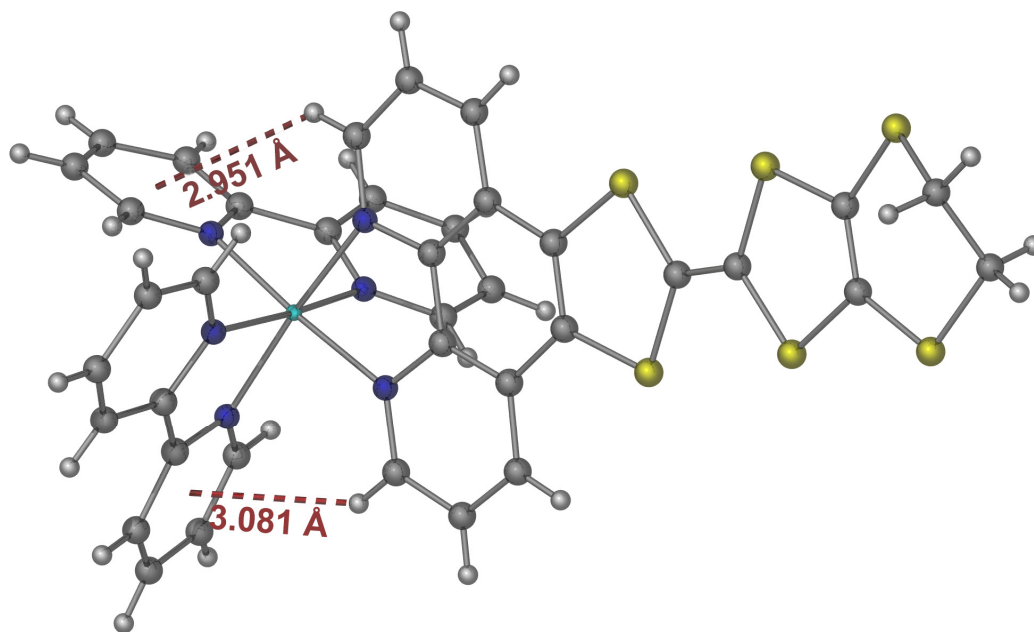


Figure S2. Molecular structure of [Ru(bpy)₂(L1)]²⁺ showing the proximity of the α-protons of coordinated L1 to the π-system of bpy ligands.

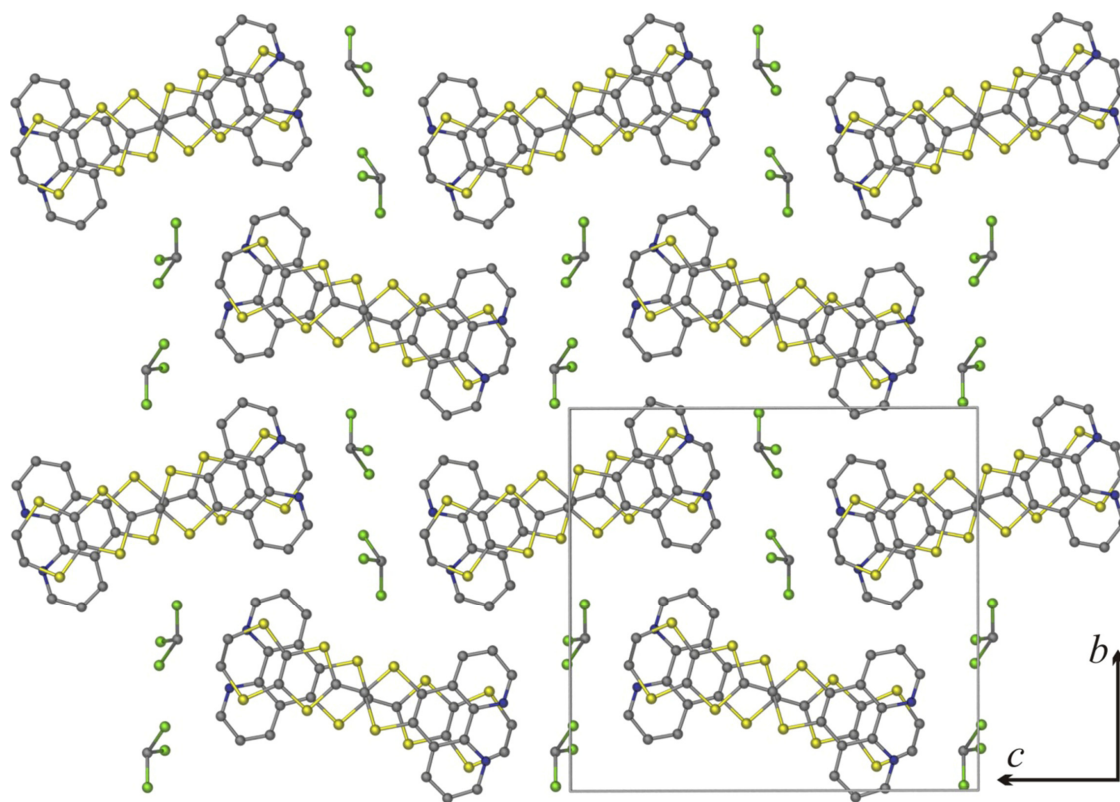


Figure S3. Crystal packing of L1 illustrating the columns of molecules stacked in a head-to-tail fashion. The columns run parallel to the *a* axis. H atoms are omitted for clarity.

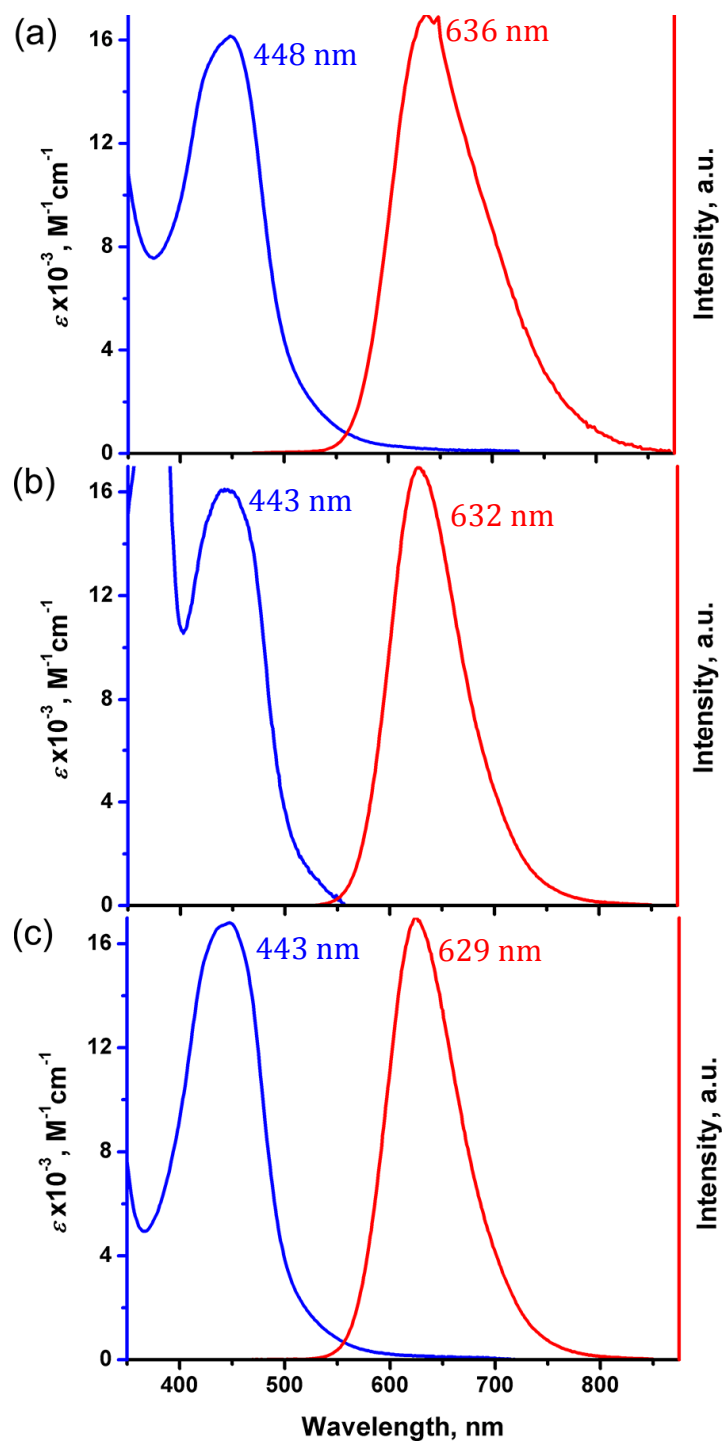


Figure S4. MLCT absorption bands and corresponding emission spectra of complexes **1-3** (a-c, respectively). The wavelengths of the absorption (λ_{abs}) and emission (λ_{em}) maxima are indicated.

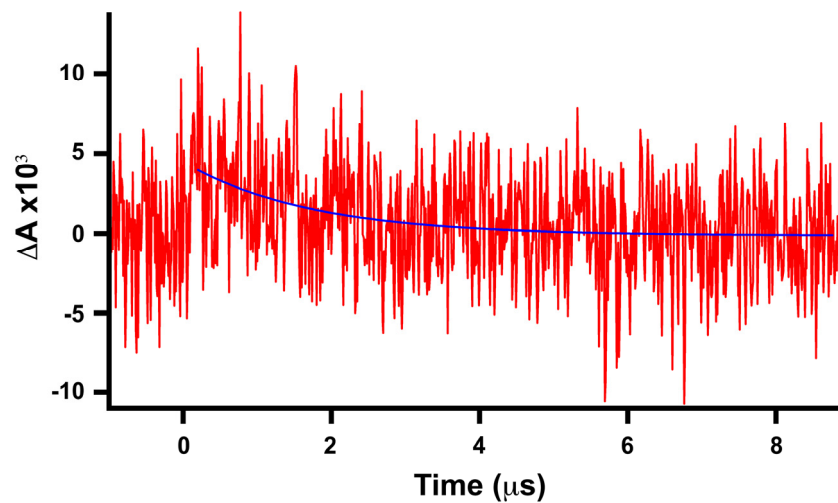


Figure S5. Kinetic profile of the difference absorbance spectrum of **1** in CH₃CN at room temperature. $\lambda_{\text{pump}}=488$ nm; $\lambda_{\text{probe}}=370$ nm. Red – experimental data; blue – single-exponential global fit.

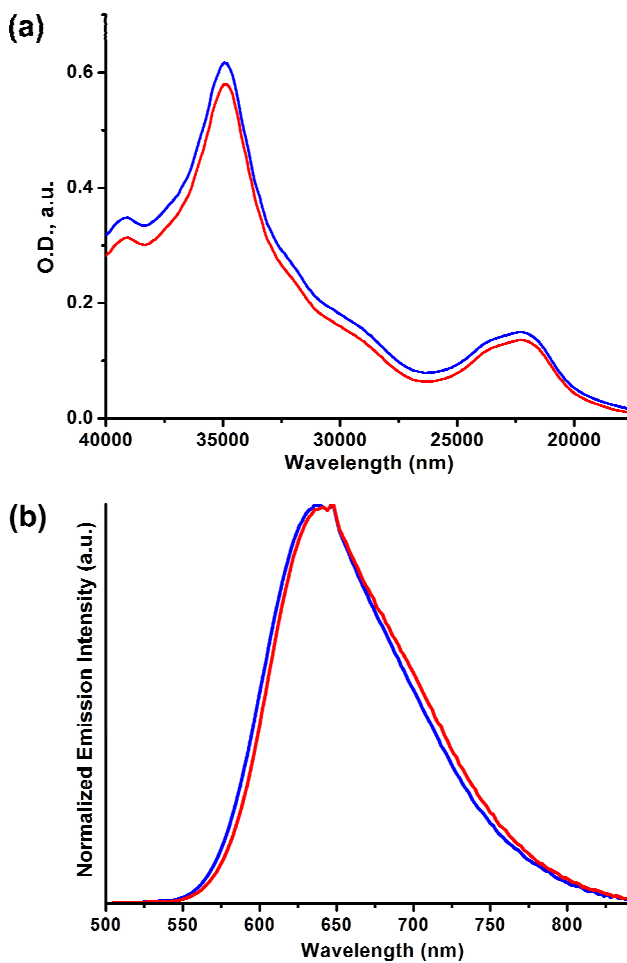


Figure S6. Absorption (a) and emission (b) spectra of **1** recorded on a solution prepared with freshly degassed CH₃CN (red) and on the same solution after keeping it in the dark for 4 hours (blue). Note that there is almost no difference between the absorption and emission bands of the two samples.