Characterization and Reactivity Studies of Dinuclear Iridium Hydride Complexes Prepared from Iridium Catalysts with N,P and C,N Ligands Under Hydrogenation Conditions

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Supplementary Information

Characterization

A.	Characterization and labeling of complexes 7, 9a, 9b, 10 and 11	S2
B.	¹ H, ³¹ P and ¹³ C NMR spectra for complexes 5, 6,7, 8, 9a, 9b, 10 and 11	S 8
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Table S1. ¹H- and ¹³C-NMR data for the cation of salt **7** in CD_2Cl_2 at 233 K.

	Position	δ(¹ H) [ppm]	δ(¹³ C) [ppm]
H_{b} H_{t} H_{t	Ir-H _t	-20.24	-
8 6 N 3'4 H'µ	$Ir-H_{\mu}$	-17.94	-
	1	3.93	74.2
Ph - P' + H + H + O	2	2.37	29.6
$Ph = \prod_{10'} H - \frac{2}{H_b} H_b$	3	0.54	12.7
3 4	4	0.65	18.4
	5 _a	4.57	70.1
	5 _b	4.45	70.1
	6	-	169.1
	7	-	80.6
	8	1.45	26.4
	9	1.73	27.4
	10	-	121.0
	11	1.91	3.0
	Ir-H't	-20.04	-
	$Ir-H'_{\mu}$	-17.84	-
	1'	3.74	78.5
	2'	2.34	28.5
	3'	0.77	13.6
	4'	0.67	18.8
	5'a	4.47	70.0
	5' _b	4.40	70.0
	6'	-	169.8
	7'	-	80.7
	8'	1.37	26.3
	9'	1.76	27.6
	10'	-	120.8
	11'	1.78	2.3

Table S2.	¹ H- and	¹³ C-NMR	data for t	he cation	of salt	9a in	CD_2CI_2	at rt.

24 20 19	2+
23 13 14 18 H	
O 21 22 12 9 N R^1	6
	Ha
H H H H H H H H H H	$H_{\rm b}$
$ \begin{array}{cccc} H & H & H \\ & & & \\$	™ <mark>∖ ⊓a</mark> H _a H _b

Position	δ(¹ H) [ppm]	δ(¹³ C) [ppm]
Ir-H _t	-36.85	-
$Ir-H_{\mu}$	-18.40	-
1	-	181.6
2 _a	4.76	74.1
2 _b	4.30	74.1
3	4.35	66.9
4 _a	2.34	34.6
4 _b	2.08	34.6
5 _a	3.74	46.0
5 _b	4.11	46.0
6	6.95	126.7
7	6.96	124.6
8	-	133.4
9	-	137.6
10	-	147.5
11	7.14	124.4
12	7.38	130.3
13	7.28	124.3
14	-	145.5
15	1.50	29.1
16	0.92	25.8
17	1.32	22.3
18	2.70	29.9
19	1.08	25.9
20	1.30	23.8
21	-	37.2
22	1.95/1.10	36.3
23	2.00	27.7
24	1.83/1.62	35.8

	-		
²⁰ 19	Position	δ(¹ H) [ppm]	
13 14 18 H 2+	Ir-H _t	-34.45	
	$Ir-H_{\mu}$	-18.90	
$H_{\mathbb{A}} / \mathbb{A} $	1	-	
H H H_{μ} H_{μ	2 _a	4.82	
$H \rightarrow H \rightarrow$	2 _b	4.40	
$\begin{array}{cccc} H & \mathbf{N} & \mathbf{N} \\ & $	3	4.48	
~	4 _a	2.70	
	4 _b	2.06	
	5 _a	3.69	
	5 _b	4.20	
	6	6.97	
	7	6.97	
	8	-	
	9	-	
	10	-	
	11	7.17	
	12	7.37	
	13	7.23	
	14	-	
	15	1.75	
	16	0.88	
	17	1.22	
	18	2.34	
	19	1.01	
	20	1.19	
	21	-	
	22	0.80	

Table S3. ¹H- and ¹³C-NMR data for the cation (major isomer) of salt **9b** in CD_2CI_2 at 243 K .

δ(¹³C) [ppm]

-

181.4 73.9 73.9

66.1 32.5

32.5 45.7 45.7 126.2 123.8 132.2

136.9 147.3

123.8 130.3

124.3 145.0 28.4 25.5

22.3 28.9

26.0 22.0 34.2 23.4

13	20 19 14 18 H	+	Position	δ(¹ H) [ppm]	δ(¹³ C) [ppm]
1	9 4 7	22_{6}^{22}	Ir-H _a	-21.93	-
16 10 17	N 8	-N H	Ir-H _b	-26.12	-
H_{b}	`H∣ ∵Ir∽		1	-	179.4
N		$H_{\rm b}$	2 _a	4.12	71.9
2'	H _a	HaHa	2 _b	4.01	71.9
			3	4.09	70.7
			4 _a	2.00	36.2
	⊼(¹ ⊔)	$\mathbf{x}^{(13}\mathbf{C})$	4 _b	1.71	36.2
Position	0(H) [ppm]	[ppm]	5 _a	4.53	45.6
1'	2.21	3.4	5 _b	3.90	45.6
2'	-	118.1	6	6.98	120.6
3'	2.29	4.6	7	6.87	123.2
4'	-	119.8	8	-	154.2
			9	-	138.0
			10	-	145.6
			11	7.26	124.0
			12	7.46	129.7
			13	7.24	124.1
			14	-	146.0
			15	2.25	28.7
			16	1.03	25.3
			17	1.24	22.1
			18	2.24	28.6
			19	1.00	25.1
			20	1.22	21.9
			21	-	34.9
			22	1.35	27.9

Table S4. ¹H- and ¹³C-NMR data for the cation of salt **10** in CD_2CI_2 at 263 K.

	+
$\begin{array}{c} 11 \\ 16 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10 \\$	
	Ha K Hu
	H _b '' ^b 4 H _b
$3' H_a H_a H_a$	Ha

Table S5. ¹ H- ar	d ¹³ C-NMR o	data for the	cation of salt	11 in	CD ₂ Cl ₂ at 295 K
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Position	δ(¹H) [ppm]	δ(¹³ C) [ppm]
1'	2.76	30.8
2'	-	162.2
3'	7.43	126.9
4'	7.79	138.2
5'	7.93	121.5
6'	-	159.6
7'	-	158.4
8'	7.91	120.9
9'	7.76	137.7
10'	7.41	127.3
11'	-	161.5
12'	2.65	26.6

Position	δ(¹ H) [ppm]	δ(¹³ C) [ppm]
Ir-H _a	-21.92	-
Ir-H _b	-28.62	-
1	-	180.7
2 _a	4.19	72.9
2 _b	4.09	72.9
3	4.86	73.7
4 _a	2.26	36.3
4 _b	1.82	36.3
5 _a	4.81	46.6
5 _b	4.02	46.6
6	7.10	121.1
7	6.91	123.9
8	-	158.2
9	-	138.5
10	-	146.4
11	7.13	124.6
12	7.28	129.6
13	6.98	123.8
14	-	145.3
15	2.23	28.9
16	1.06	25.3
17	1.32	22.8
18	2.37	28.9
19	0.96	24.7
20	0.34	21.4
21	-	34.5
22	0.59	27.3

Ir-cat (2.5 mol%)							
	Ph12	Ph	50 bar H ₂ CH ₂ Cl ₂ , rt, 2 h	Ph	* Ph 13		
entry	cat.	cat. loading (mol%)	molarity (M)	pressure (bar)	13 (%) ^a	<i>ee</i> 13 (%) ^b	
1 ^c	9a	2.5	0.06	50	53	nd	
2^{c}	9b	2.5	0.06	50	5	nd	

^aYields were determined by GC analysis. ^bEnantioselectivities were determined by HPLC on chiral stationary phase ^cAverage values of at least two experiments.

Spectra of the stereoisomeric complexes 5a and 5b at 295 K

500 MHz 1H NMR of stereoisomeric complexes 5a and 5b (hydride region)



Section of the NOESY of stereoisomeric complexes 5a and 5b



Spectra for complex 6 at 295 K

500 MHz 1H NMR of a single crystal of complex 6 (hydride region)



1H-31P HMBC of complex 6.



H-H COSY of complex 6 (hydride region).



S11

Spectra for complex 7 at 233 K.

500 MHz 1H NMR of complex 7



500 MHz 1H NMR of complex 7 (hydride region)





202 MHz 31P{1H} NMR of complex 7





Spectra for complex 8 at 295 K





500 MHz 1H NMR of a single crystal of complex 8 in CD2CL2 (region 8.2-0 ppm)



Spectra complex 9a at 295 K.

500 MHz 1H NMR of complex 9a









Spectra of complex 9b at 243 K.

500 MHz 1H NMR of complex 9b









S24

Phase sensitive 2D NOESY of complex 9b showing the selective exchange cross-peaks between the major compound and the minor species



Spectra of complex 10 at 263 K

500 MHz 1H NMR of complex 10



500 MHz 1H NMR of complex 10 (hydrides)

-26.11





S28

Spectra of complex 11 at 295 K

500 MHz 1H NMR of complex 11



500 MHz 1H NMR of complex 11 (hydrides)



-28.61





Identification of the byproducts *rel-(1S,3S)-1-benzyl-3-methyl-1,3-diphenyl-1,2,3,4-tetrahydronaphthalene, rel-(1R,3S)-1-benzyl-3-methyl-1,3-diphenyl-1,2,3,4-tetrahydronaphthalene and rel-(1S,2S,3S)-1-benzyl-1,3-dimethyl-2,3-diphenyl-2,3-dihydro-1H-indene*



Sulfuric acid (95%, 89 μ L, 1.29 mmol) was added to a CH₂Cl₂ solution (5 mL) of (*E*)-1,2-diphenyl-1-propene (500 mg, 2,57 mmol) and stirred at room temperature for 1 h. After that time the solvent was completely evaporated, the residue dissolved in hexane (15 mL) and filtered over a short column of silica gel (washed with 10 mL hexane). Evaporation of the solvent afforded the crude product as a mixture of **P1**, **P2** and **P3** in a ratio of ~ 1:0.7:0.1. The regioisomers could be partially separated by semipreparative HPLC on an OD-column (hexane, 25°C, 8.0 mL/min, 194 nm). The products were identified by ¹H, ¹³C NMR (see following spectra) and HR EI-MS and the

relative configuration was assigned on the basis of NOE spectroscopy.

HR EI-MS: M = C₃₀H₂₈

rel-(1S,3S)-1-benzyl-3-methyl-1,3-diphenyl-1,2,3,4-tetrahydronaphthalene: Calcd m/z for ($[M-C_7H_7]^{+}$: 297.1638; Found: 297.1644. **rel-(1R,3S)-1-benzyl-3-methyl-1,3-diphenyl-1,2,3,4-tetrahydronaphthalene:** Calcd m/z for ($[M-C_7H_7]^{+}$: 297.1638; Found: 297.1643. **rel-(1S,2S,3S)-1-benzyl-1,3-dimethyl-2,3-diphenyl-2,3-dihydro-1H-indene:** Calcd m/z for ($[M-C_7H_7]^{+}$: 297.1638; Found: 297.1635. 500 MHz 1H NMR of the crude containing P1, P2 and P3(reaction with sulfuric acid).



500 MHz 1H NMR of the crude of entry 7 in Table 1 showing the byproduct formation (crude contains(E)-1,2-diphenyl-1-propene, P1,P2 and P3).





S35

125 MHz 13C{1H} NMR of rel-(1S,3S)-1-benzyl-3-methyl-1,3-diphenyl-1,2,3,4-tetrahydronaphthalene





S37







500 MHz 1H NMR of rel-(1S,2S,3S)-1-benzyl-1,3-dimethyl-2,3-diphenyl-2,3-dihydro-1H-indene





Table S7. Crystal data for complex 6.

• ·	
formula	$C_{112}H_{89}B_2CI_5F_{48}Ir_2N_4O_4P_2$
formula weight	3109.12
Z. calculated density	1. 1.718 Ma · m⁻³
F(000)	1529
description and size of crystal	vellow block $0.180 \cdot 0.311 \cdot 0.368 \text{ mm}^3$
absorption and size of crystal	2.477 mm^{-1}
	2.477 11111
min/max transmission	0.46 / 0.64
temperature	123K
radiation(wavelength)	Mo <i>K</i> _α (λ = 0.71073 Å)
Crystal system, space group	triclinic, P 1
a	14.2328(7) Å
b	14.5307(7) Å
C	15 9379(8) Å
a	92 544(3)°
ß	113 818(2)°
р У	113.010(2)
Ŷ	92.007(3)
V	3005.0(3) A°
min/max Θ	1.915° / 34.970°
number of collected reflections	172350
number of independent reflections	51163 (merging r = 0.034)
number of observed reflections	41743 (I>2.0σ(I))
number of refined parameters	1687
r	0.0267
rW	0.0497
and the second s	0.9621
goodhood of ht	

Table S8. Crystal data for complex 7.

formula formula weight	$C_{112.55}H_{87.10}B_2CI_{5.10}F_{48}Ir_2N_4O_4P_2$ 3120.38
	1, 1.669 Mg · m · 1535 100
description and size of crystal	vellow block $0.180 \cdot 0.250 \cdot 0.320 \text{ mm}^3$
absorption coefficient	2.401 mm ⁻¹
min/max transmission	0.55 / 0.65
temperature	123K
radiation(wavelength)	Mo <i>K</i> _α (λ = 0.71073 Å)
Crystal system, space group	triclinic, P 1
а	12.8192(3) Å
b	16.1838(4) Å
С	17.2644(4) A
α	67.9070(10)°
β	69.4010(10)°
Y	84.5930(10)°
V	3103.56(13) A ³
min/max Θ	1.773° / 30.073°
number of collected reflections	116614
number of independent reflections	35611 (merging r = 0.026)
number of observed reflections	31146 (I>2.0σ(I))
number of refined parameters	1770
r	0.0265
rvv	0.0342
goodness of fit	1.0761

Table S9. Crystal data for complex 8.

formula formula weight Z, calculated density F(000) description and size of crystal absorption coefficient min/max transmission temperature radiation(wavelength) Crystal system, space group a b c c α β γ γ γ γ γ γ γ γ γ γ	$\begin{array}{l} C_{109}H_{86}B_2Cl_6F_{48}Ir_2N_2O_4P_2\\ 3080.53\\ 1,\ 1.689\ Mg\cdot m^{-3}\\ 1514.000\\ yellow \ block,\ 0.050\cdot 0.130\cdot 0.170\ mm^{3}\\ 2.479\ mm^{-1}\\ 0.72\ /\ 0.88\\ 123K\\ Mo\ {\cal K}_{\alpha}\ (\lambda=0.71073\ Å)\\ triclinic,\ P\ 1\\ 13.3861(9)\ Å\\ 14.7080(10)\ Å\\ 16.6176(11)\ Å\\ 103.794(4)^\circ\\ 90.144(4)^\circ\\ 107.081(4)^\circ\\ 3027.9(4)\ Å^{3}\\ 1.596^\circ\ /\ 32.577^\circ\\ 72716\\ 42682\ (merging\ r=0.033)\\ 34145\ (I>2.0\sigma(I))\\ 1744\\ 0.0352 \end{array}$
number of refined parameters	1744
r	0.0352
rW	0.0548
goodness of fit	1.1125

Table S10. Crystal data for complex 9b.

formula formula weight Z, calculated density	C ₁₁₅ H ₁₀₄ B₂Cl ₆ F ₄₈ Ir₂N ₆ O₂ 3132.81 1, 1.582 Mg · m⁻³
F(000)	1546.000
description and size of crystal	yellow block, $0.050 \cdot 0.110 \cdot 0.230 \text{ mm}^3$
absorption coefficient	2.263 mm ⁻¹
min/max transmission	0.89 / 0.89
temperature	123K
radiation(wavelength)	Mo K_{α} (λ = 0.71073 Å)
Crystal system, space group	triclinic, P 1
а	13.0057(9) Å
b	16.3543(12) Å
С	17.8121(12) Å
α	70.889(3)°
β	85.622(4)°
γ	66.774(4)°
V	3283.6(4) Å ³
min/max Θ	1.547° / 32.576°
number of collected reflections	98169
number of independent reflections	46715 (merging r = 0.040)
number of observed reflections	35395 (I>2.0σ(I))
number of refined parameters	1860
r	0.0453
rW	0.0730
goodness of fit	1.1250