

Supporting Information

Cyclopentadienyl Iron(II) Complexes of N-heterocyclic Carbenes Bearing a Malonate or Imidate Backbone: Synthesis, Structure and Catalytic Potential in Hydrosilylation

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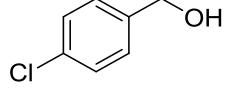
Contents

Characterization of the products of catalysis	S3
Normal resolution and high resolution XRD studies	S8
• Table S1. Crystal data and structures refinements for complexes 3_{tBu} , 3_{Me} , 4 , and [6^{Me}](OTf)	S8
• Table S2. Multipolar parameters for complex 3_{Me}	S9
• Table S3. Values of the experimental and theoretical (italic) orthodox parameters at bond critical points within complex 3_{Me}	S15
• Figure S1. Residual electron density maps for 3_{Me} after the multipolar refinement	S17
• Figure S2. Laplacian of the experimental (left) and theoretical (right, M06-2X/6-31G**) electron density for 3_{Me} in various planes	S18
Computational studies	S19
• Table S4. Main geometrical parameters (bond lengths in Å and bond angles in deg) for the complex 3_{Me} at different levels of theory	S21

<ul style="list-style-type: none"> • Table S5. Experimental X-Ray data for complexes 3_{Me}, 4, 6^{Me}(OTf). Calculated geometrical parameters for compounds 3_{Me}, 4, 6^{Me}(OTf), 7 and 8 at the M06-2X/6-31G** level of theory • Table S6. Experimental and theoretical topological analysis of complex 3_{Me} • Table S7. AIM analysis for complex 4 • Table S8. AIM analysis for complex 6^{Me}(OTf) • Table S9. AIM analysis for complex 7 • Table S10. AIM analysis for complex 8 • <i>Analysis of the influence of the electronic properties of the carbene and of the N-substituents on the interligand interaction</i> • <i>Analysis of the bonding situation of the Fe-C_{carbene} bond in 3_{Me}, 4 and 6^{Me}(OTf)</i> • Table S11. NBO analysis of the C→Fe interaction in the three complexes 3_{Me}, 4 and 6^{Me}(OTf) • Table S12. Topological analysis of the C(aryl)...C(=O) bond critical points • Table S13. NBO analysis of the C(aryl)...C(=O) non-covalent weak interactions for compounds 3_{Me}, 4, 6^{Me}(OTf), 7 and 8 at the M06-2X/6-31G** level of theory. • Figure S3. Contour plot of the Laplacien $\nabla^2 \rho$(BCP) with charge accumulation ($\nabla^2 \rho(r) < 0$) printed in blue and charge depletion ($\nabla^2 \rho(r) > 0$) printed in red for the complexes 3_{Me}, 4 and 6^{Me}(OTf) • Figure S4. Molekel plot (cutoff : 0.05) for the NLMO associated with the C→Fe interaction in complex 3_{Me} • Figure S5. Electronic delocalization and density $\rho(\mathbf{r}_b)$ in $e^{-\text{\AA}^{-3}}$ for the carbene skeleton in the complexes 3_{Me}, 4 and 6^{Me}(OTf) 	S22 S23 S25 S27 S29 S31 S33 S34 S35 S37 S38 S39 S39 S39 S39 S39 S45
Z-Matrices	S39
NMR Spectra	S45

Characterizations of the products of catalysis

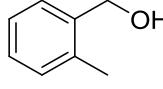
4-Chlorobenzyl alcohol

 According to the general procedure A, 4-chlorobenzaldehyde (141 mg, 1 mmol) gave 4-chlorophenylmethanol (118 mg, 83%) as a white solid.

¹H NMR (300 MHz, CDCl₃): δ 7.32 (d, 2H, *J* = 8.7 Hz), 7.26 (d, 2H, *J* = 8.7 Hz), 4.62 (s, 2H), 2.19 (br s, 1H).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 139.4, 133.5, 128.8, 128.4, 64.6.

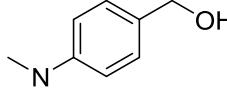
o-Tolylmethanol

 According to the general procedure A, *o*-tolualdehyde (116 μL, 1 mmol) gave *o*-tolylmethanol (88 mg, 72%) as a light yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 7.40-7.32 (m, 1H), 7.25-7.15 (m, 3H), 4.68 (s, 2H), 2.36 (s, 3H), 1.87 (br s, 1H).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 138.9, 136.3, 130.5, 127.8, 127.7, 126.2, 63.6, 18.8.

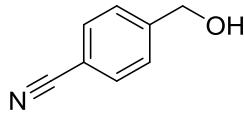
4-(*N,N*-dimethylaminophenyl)methanol

 A flame dried Schlenk flask was charged with the iron complex **3_{nBu}** (5.8 mg, 10 μmol, 1 mol%), 4-(*N,N*-dimethylamino)benzaldehyde (149 mg, 1.0 mmol), phenylsilane (113 μL, 1.2 mmol) and anhydrous toluene (0.2 mL). The reaction was stirred in a 30 °C pre-heated oil bath under light irradiation for 3 hours. The reaction was then hydrolyzed with methanol (2 mL) and a 2M aqueous solution of sodium hydroxide (2 mL). The product was extracted with diethylether (3×10 mL). The combined organic layers were washed with brine, dried with sodium sulfate and evaporated. The crude residue was purified by silica gel chromatography using a petroleum ether/diethyl ether mixture (7:3) as the eluent to give 4-(*N,N*-dimethylaminophenyl)methanol (124 mg, 82%) as a yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 7.25 (d, 2H, *J* = 8.4 Hz), 6.73 (d, 2H, *J* = 8.4 Hz), 4.57 (s, 2H), 2.95 (s, 6H), 1.61 (br s, 1H).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 150.5, 129.1, 128.8, 112.8, 65.6, 40.9.

4-Cyanobenzyl alcohol

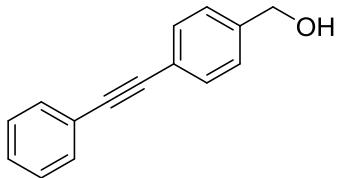


A flame-dried Schlenk tube was charged with the iron complex **3_tBu** (5.8 mg, 10 μmol, 1 mol%), 4-cyanobenzaldehyde (131.1 mg, 1 mmol), phenylsilane (113 μL, 1.2 mmol) and anhydrous toluene (0.2 mL). The reaction was stirred in a 30 °C pre-heated oil bath under light irradiation for 3 hours. The reaction was hydrolyzed with methanol (2 mL) and 2M aqueous solution of sodium hydroxide (2 mL). The product was extracted with diethylether (3×10 mL). The combined organic layer was washed with brine, dried with sodium sulfate and evaporated. The crude residue was purified by silica gel chromatography using a petroleum ether/diethyl ether mixture (6:4) as the eluent to give 4-cyanophenylmethanol (129 mg, 95%) as a yellow solid.

¹H NMR (300 MHz, CDCl₃): δ 7.53 (d, 2H, *J* = 8.3 Hz), 7.39 (d, 2H, *J* = 8.3 Hz), 4.66 (s, 2H), 3.48 (br s, 1H).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 146.7, 132.2, 127.0, 118.9, 110.5, 63.7.

[4-(Phenylethynyl)phenyl]methanol

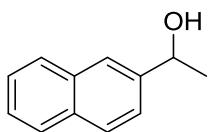


According to the general procedure A, 4-(phenylethynyl)benzaldehyde (103 mg, 0.5 mmol) gave [4-(phenylethynyl)phenyl]methanol (100 mg, 96%) as a white solid.

¹H NMR (300 MHz, CDCl₃): δ 7.64-7.45 (m, 4H), 7.42-7.28 (m, 5H), 4.68 (s, 2H), 2.13 (br s, 1H).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 141.2, 131.9, 131.8, 128.5, 128.4, 127.0, 123.4, 122.6, 89.6, 89.4, 65.1.

1-(Naphthalen-2-yl)ethanol

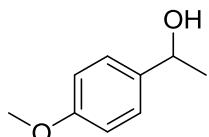


According to the general procedure B, 2-acetonaphthone (85 mg, 0.5 mmol) gave 1-(naphthalen-2-yl)ethanol (60 mg, 70%) as a white solid.

¹H NMR (300 MHz, CDCl₃): δ 7.90-7.73 (m, 4H), 7.54-7.43 (m, 3H), 5.05 (q, 1H, *J* = 6.5 Hz), 2.20 (br s, 1H), 1.58 (d, 3H, *J* = 6.5 Hz).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 143.2, 133.4, 132.9, 128.3, 128.0, 127.7, 126.2, 125.8, 123.9, 123.8, 70.5, 25.2.

1-(4-Methoxyphenyl)ethanol

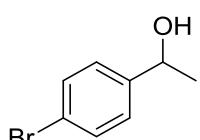


According to the general procedure B, 4-methoxyacetophenone (75 mg, 0.5 mmol) gave 1-(4-methoxyphenyl)ethanol (23 mg, 30%) as a yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 7.30 (d, 2H, *J* = 8.7 Hz), 6.88 (d, 2H, *J* = 8.7 Hz), 4.85 (q, 1H, *J* = 6.5 Hz), 3.80 (s, 3H), 1.88 (br s, 1H), 1.48 (d, 3H, *J* = 6.5 Hz).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 159.2, 138.2, 126.9, 114.0, 70.2, 55.5, 25.2.

1-(4-Bromophenyl)ethanol

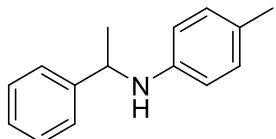


According to the general procedure B, 4-bromoacetophenone (199 mg, 1 mmol) gave 1-(4-bromophenyl)ethanol (94 mg, 47%) as a yellow oil.

¹H NMR (400 MHz, CDCl₃): δ 7.46 (d, 2H, *J* = 8.4Hz), 7.24 (d, 2H, *J* = 8.4Hz), 4.85 (q, 1H, *J* = 6.4 Hz), 1.97 (br s, 1H), 1.46 (d, 3H, *J* = 6.4 Hz).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 144.9, 131.7, 127.3, 121.3, 70.0, 25.4.

4-Methyl-N-(1-phenylethyl)aniline

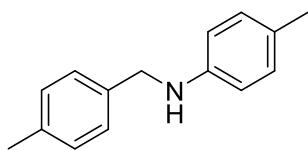


A flame dried Schlenk flask was charged with the iron complex $\mathbf{3}_{t\text{Bu}}$ (14.5 mg, 25 μmol , 5 mol%), 4-methyl-*N*-(1-phenylethylidene)aniline (105 mg, 0.5 mmol), phenylsilane (250 μL , 2 mmol) and anhydrous toluene (0.2 mL). The reaction was stirred in a 100 °C pre-heated oil bath under light irradiation for 24 hours. The reaction was hydrolyzed with methanol (2 mL) and a 2M aqueous solution of sodium hydroxide (2 mL). The product was extracted with diethylether (3×10 mL). The combined organic layer was washed with brine, dried with sodium sulfate and evaporated. The crude residue was purified by silica gel chromatography using a petroleum ether/diethyl ether mixture (95:5) as the eluent to give 4-methyl-*N*-(1-phenylethyl)aniline (98 mg, 93%) as a white solid.

^1H NMR (300 MHz, CDCl_3): δ 7.42-7.30 (m, 5H), 6.99 (d, 2H, $J = 8.3\text{Hz}$), 6.48 (d, 2H, $J = 8.7\text{Hz}$), 4.49 (q, 1H, $J = 6.8\text{ Hz}$), 3.90 (br s, 1H), 2.27 (s, 3H), 1.54 (d, 3H, $J = 6.8\text{ Hz}$).

$^{13}\text{C}\{\text{H}\}$ NMR (75 MHz, CDCl_3): δ 144.9, 144.2, 132.5, 129.8, 128.9, 127.4, 126.8, 113.6, 53.3, 25.2, 20.5.

4-Methyl-*N*-(4-methylbenzyl)aniline



A flame dried Schlenk flask was charged with the iron complex $\mathbf{3}_{t\text{Bu}}$ (14.5 mg, 25 μmol , 5 mol%), 4-methyl-*N*-(4-methylbenzylidene)aniline (105 mg, 0.5 mmol), phenylsilane (250 μL , 2 mmol) and anhydrous toluene (0.2 mL).

The reaction was stirred in a 30 °C pre-heated oil bath under light irradiation for 30 hours. The reaction was hydrolysed with methanol (2 mL) and 2M aqueous solution of sodium hydroxide (2 mL). The product was extracted with diethylether (3×10 mL). The combined organic layer was washed with brine, dried with sodium sulfate and evaporated. The crude residue was purified by silica gel chromatography using a petroleum ether/diethyl ether mixture (98:2) as the eluent to give 4-methyl-*N*-(4-methylbenzyl)aniline (55 mg, 52%) as a yellow solid.

¹H NMR (300MHz, CDCl₃): δ 7.32 (d, 2H, *J* = 7.9 Hz), 7.21 (d, 2H, *J* = 7.9 Hz), 7.05 (d, 2H, *J* = 8.3 Hz), 6.62 (d, 2H, *J* = 8.3 Hz), 4.32 (s, 2H), 3.93 (br s, 1H), 2.41 (s, 3H), 2.31 (s, 3H).

¹³C{¹H} NMR (75 MHz, CDCl₃): δ 146.0, 136.9, 136.7, 129.9, 129.4, 127.7, 126.9, 113.2, 48.6, 21.3, 20.6.

Normal resolution and high resolution XRD studies

Table S1. Crystal data and structure refinements for complexes **3_{tBu}**, **3_{Me}**, **4**, and **[6^{Me}](OTf)**

SHELX Refinement				
complex	3_{tBu}	3_{Me}	4	[6^{Me}](OTf)
empirical formula	C ₃₃ H ₃₆ FeN ₂ O ₄ CH ₂ Cl ₂ C ₃₀ H ₃₀ FeN ₃ O ₄		C ₂₈ H ₂₇ FeN ₃ O ₄	C ₂₉ H ₃₀ FeN ₃ O ₄ , CF ₃ O ₃ S
molecular weight (g)	612.42	538.39	525.38	689.48
temperature (K)	180	100	180	180
crystal system	orthorhombic	triclinic	monoclinic	monoclinic
space group	P2 ₁ 2 ₁ 2 ₁ (#19)	P 1 (bar) (#2)	P2 ₁ /n (#14)	P2 ₁ /n (#14)
a (Å)	13.079(3)	9.6895(8)	9.5746(3)	12.5292(5)
b (Å)	15.663(3)	9.9041(8)	15.8303(5)	15.1526(6)
c (Å)	15.731(3)	14.4926(12)	16.3931(6)	16.9812(6)
α (°)		109.600(3)		
β (°)		104.051(15)	93.601(1)	109.835(2)
γ (°)		107.503(2)		
volume (Å ³)	3222.6(11)	1259.82(18)	2479.78(14)	3032.6(2)
Z	4	2	4	4
D _{calcd.} (g.cm ⁻³)	1.372	1.419	1.407	1.510
μ (mm ⁻¹)	0.674	0.634	0.634	0.637
F ₀₀₀	1392	564	1096	1424
θ _{max} (°)	30.5	26.4	26.4	26.4
completeness to θ _{max}	99	99	99	99
index range	-18<h<18 -22<k<22 -22<l<22	-12<h<11 -12<l<11 -18<l<18	-11<h<11 -19<k<19 -20<l<20	-15<h<15 -18<k<18 -21<l<21
reflections collected	1013470	312535	97705	182350
independant reflections	9802	5144	6196	6203
data/restraints/parameters	9802 / 0 / 397	5144 / 0 / 341	6196 / 0 / 417	6203 / 0 / 363
g.o.f.	1.08	1.067	1.04	1.03
R, Rw [I>2σ(I)]	R1 = 0.0222 wR2 = 0.0588	R1 = 0.0255 wR2 = 0.0767	R1 = 0.0278 wR2 = 0.0741	R1 = 0.0278 wR2 = 0.0741
R , Rw (all data)	R1 = 0.0230 wR2 = 0.0594	R1 = 0.0257 wR2 = 0.0770	R1 = 0.0321 wR2 = 0.0780	R1 = 0.0332 wR2 = 0.0780
resid. electr. dens. (e.Å ⁻³)	0.44 / -0.31	0.35 / -0.39	0.35 / -0.33	0.35 / -0.31
MoPro Refinement (<i>S</i> < 1.1 Å ⁻¹)				
θ _{max}	53.2			
Completeness to θ _{max} (%)	99			
index range	-23<h<23 -23<l<23 -34<l<34			
Independent reflections	23392			
Parameters / restraints	965 / 84			
g.o.f.	1.05			
R [I>3σ(I)] (calculated against <i>F</i>)	0.0132			
R _w [I>3σ(I)] (calculated against <i>F</i>)	0.0169			

Table S2. Multipolar parameters for complex **3_{Me}**

atom: Fe1			atom: N1												
system coord	D0 X Fe1 C3 Y		system coord	C6 X N1 C3 Y											
κ, κ'	0.962746 0.798890		κ, κ'	1.001363	0.960001										
Pv	5.8753		Pv	5.0636											
P00	0.0000		P00	0.0000											
P1m \pm	0.0270	0.0630	-0.0840	P1m \pm	-0.0340	0.0290	0.0550								
P2m \pm	0.0170	0.0750	0.0110	-0.0050	-0.0040		P2m \pm	-0.0090	-0.0040	-0.0210	0.0050	0.0130			
P3m \pm	-0.0320	-0.0710	0.0390	-0.0370	-0.0730	0.0200	-0.0120	P3m \pm	0.0450	0.0030	0.0150	-0.0010	0.0190	0.0040	-0.2170
P4m \pm	-0.0310	-0.1150	-0.3060	-0.0180	0.0020	-0.0350	-0.1000	-0.0180	0.0870						
atom: O1			atom: N2												
system coord	C1 Z O1 Fe1 Y		system coord	C4 X N2 C3 Y											
κ, κ'	1.003132	0.998316	κ, κ'	1.001363	0.960001										
Pv	5.7827		Pv	5.0603											
P00	0.0000		P00	0.0000											
P1m \pm	-0.0120	-0.0260	-0.1160	P1m \pm	-0.0420	0.0100	0.0160								
P2m \pm	0.0110	-0.0090	0.0730	-0.0690	-0.0140		P2m \pm	0.0230	0.0060	-0.0240	-0.0040	-0.0130			
P3m \pm	0.0230	0.0100	0.0110	-0.0190	0.0130	0.0420	-0.0090	P3m \pm	0.0370	-0.0280	0.0270	0.0170	0.0100	-0.0230	-0.2220
atom: O2			atom: C1												
system coord	C2 Z O2 Fe1 Y		system coord	O1 Z C1 Fe1 Y											
κ, κ'	1.003132	0.998316	κ, κ'	0.980570	0.926660										
Pv	5.9365		Pv	4.2149											
P00	0.0000		P00	0.0000											
P1m \pm	0.0110	0.0530	-0.0420	P1m \pm	0.0400	-0.0020	0.0850								
P2m \pm	0.0180	0.0620	0.1060	0.0470	-0.0400		P2m \pm	-0.0280	-0.0250	0.4540	-0.0350	0.0210			
P3m \pm	0.0380	-0.0370	0.0000	0.0050	0.0390	0.0210	0.0210	P3m \pm	0.0360	0.0230	0.0180	0.1260	-0.0280	0.0220	0.0380
atom: O3			atom: C2												
system coord	C4 X O3 N1 Y		system coord	O2 Z C2 Fe1 Y											
κ, κ'	0.980041	0.963376	κ, κ'	0.980570	0.926660										
Pv	6.3211		Pv	4.2658											
P00	0.0000		P00	0.0000											
P1m \pm	-0.0200	0.0060	-0.0450	P1m \pm	-0.0150	-0.0790	0.0140								
P2m \pm	-0.0010	-0.0120	-0.0200	0.0040	-0.0640		P2m \pm	-0.0100	-0.0250	0.4000	0.0520	-0.0170			
P3m \pm	-0.0070	-0.0100	-0.0130	0.0030	-0.0010	0.0050	0.0380	P3m \pm	0.0230	-0.0420	-0.0190	0.0250	0.0500	0.0200	-0.0310
atom: O4			atom: C3												
system coord	C6 X O4 N2 Y		system coord	N1 X C3 N2 Y											
κ, κ'	0.980041	0.963376	κ, κ'	0.977123	0.931556										
Pv	6.2653		Pv	4.2024											
P00	0.0000		P00	0.0000											
P1m \pm	0.0410	-0.0070	-0.0700	P1m \pm	-0.0650	0.0310	-0.0680								
P2m \pm	0.0200	-0.0150	-0.0180	0.0090	-0.0510		P2m \pm	0.0360	0.0210	-0.2360	-0.0030	-0.0480			
P3m \pm	-0.0030	-0.0010	0.0320	-0.0170	0.0120	-0.0050	0.0350	P3m \pm	0.0390	0.0140	-0.0260	0.0130	-0.0210	0.0050	0.2660

Table S2 (continued)

atom: C4		atom: C12	
system coord	N2 X C4 C5 Y	system coord	D0 X C12 Fe1 Z
κ, κ'	1.016601 1.002564	κ, κ'	0.975475 0.916847
Pv	3.9252	Pv	4.2269
P00	0.0000	P00	0.0000
P1m \pm	0.1130 -0.0060 -0.0540	P1m \pm	0.0000 0.0410 -0.0410
P2m \pm	0.0990 -0.0080 -0.1970	P2m \pm	0.0000 0.0000 -0.2160 -0.0070 0.0530
P3m \pm	0.0380 0.0250 0.0050	P3m \pm	0.0150 0.0160 0.0090 0.0000 0.0330 0.0330 0.0360 -0.2790
atom: C5		atom: C13	
system coord	C3 X C5 C6 Y	system coord	D0 X C13 Fe1 Z
κ, κ'	1.008238 1.001011	κ, κ'	0.975475 0.916847
Pv	4.0630	Pv	4.1486
P00	0.0000	P00	0.0000
P1m \pm	0.0420 0.0240 0.0440	P1m \pm	0.0000 0.0360 -0.0570
P2m \pm	-0.0150 -0.0120 -0.0650	P2m \pm	0.0000 0.0000 -0.1890 -0.0050 0.0580
P3m \pm	0.0070 -0.0130 0.0280	P3m \pm	-0.0160 0.0470 -0.0290 -0.2600 0.0530 0.0640 0.0200 -0.2750
atom: C6		atom: C14	
system coord	N1 X C6 C5 Y	system coord	D0 X C14 Fe1 Z
κ, κ'	1.016601 1.002564	κ, κ'	0.975475 0.916847
Pv	3.8622	Pv	4.2428
P00	0.0000	P00	0.0000
P1m \pm	0.0840 0.0020 -0.0650	P1m \pm	0.0000 0.0280 0.0210
P2m \pm	0.1240 0.0010 -0.2270	P2m \pm	-0.0020 0.0900 0.0110 0.0500
P3m \pm	0.0700 -0.0060 -0.0060	P3m \pm	-0.0020 0.0300 -0.0310 -0.2560 0.0720 0.0580 -0.0060 -0.3340
atom: C7		atom: C15	
system coord	C5 Z C7 H71 X	system coord	D0 X C15 Fe1 Z
κ, κ'	0.973579 0.921152	κ, κ'	0.975475 0.916847
Pv	4.4084	Pv	4.3017
P00	0.0000	P00	0.0000
P1m \pm	0.0000 -0.0620 -0.0450	P1m \pm	0.0000 0.0310 0.0000
P2m \pm	0.0000 0.0000 -0.0370	P2m \pm	0.0000 0.0000 -0.1980 0.0440 0.1210
P3m \pm	-0.0540 0.0000 0.0000	P3m \pm	0.3500 0.0000 0.0000 0.0200 0.0370 0.0150 -0.2680
atom: C11		atom: C21	
system coord	D0 X C11 Fe1 Z	system coord	C24 X C21 C22 Y
κ, κ'	0.975475 0.916847	κ, κ'	1.002619 0.958249
Pv	4.2047	Pv	4.0339
P00	0.0000	P00	0.0000
P1m \pm	0.0000 0.0330 -0.0220	P1m \pm	0.0170 -0.0140 0.0880
P2m \pm	0.0000 0.0000 -0.2600	P2m \pm	-0.0030 -0.0280 -0.1400 -0.0210 -0.0900
P3m \pm	0.0000 0.0000 0.0000	P3m \pm	0.0090 0.0040 -0.0200 -0.0410 0.0330 0.0150 -0.2930
atom: C12		atom: C22	
system coord	D0 X C12 Fe1 Z	system coord	C25 X C22 C23 Y
κ, κ'	0.975475 0.916847	κ, κ'	1.002619 0.958249
Pv	4.2269	Pv	3.9882
P00	0.0000	P00	0.0000
P1m \pm	0.0000 0.0330 -0.0220	P1m \pm	0.0140 -0.0240 0.0370
P2m \pm	0.0000 0.0000 -0.2600	P2m \pm	-0.0180 0.0080 -0.1800 -0.0070 -0.0270
P3m \pm	0.0000 0.0000 0.0000	P3m \pm	0.0090 0.0440 0.0430 -0.3300 -0.0070 0.0010 -0.0040 -0.0200 0.0300 0.0220 -0.2270

Table S2 (continued)

atom: C23												
system coord	C26 X C23 C24 Y											
κ, κ'	1.002619	0.958249										
Pv	4.2313											
P00	0.0000											
P1m \pm	-0.0500	-0.0080	-0.0220									
P2m \pm	-0.0410	-0.0120	-0.1930	0.0010	0.0210							
P3m \pm	-0.0270	-0.0210	0.0020	-0.0020	0.0430	-0.0030	-0.2610					
atom: C24												
system coord	C21 X C24 C25 Y											
κ, κ'	1.002619	0.958249										
Pv	3.9847											
P00	0.0000											
P1m \pm	-0.0420	-0.0250	0.0770									
P2m \pm	0.0240	0.0000	-0.2180	-0.0240	-0.0300							
P3m \pm	-0.0150	0.0040	0.0120	0.0150	0.0200	-0.0080	-0.2610					
atom: C25												
system coord	C22 X C25 C26 Y											
κ, κ'	1.002619	0.958249										
Pv	4.1730											
P00	0.0000											
P1m \pm	-0.0080	-0.0160	-0.0200									
P2m \pm	0.0070	0.0100	-0.1760	-0.0260	-0.0160							
P3m \pm	-0.0010	-0.0190	0.0280	-0.0120	0.0310	0.0260	-0.2850					
atom: C26												
system coord	C23 X C26 C25 Y											
κ, κ'	1.002619	0.958249										
Pv	3.9909											
P00	0.0000											
P1m \pm	-0.0170	0.0200	0.0640									
P2m \pm	-0.0110	0.0120	-0.1870	0.0110	-0.0280							
P3m \pm	0.0060	-0.0100	0.0190	-0.0070	0.0140	-0.0020	-0.2480					
atom: C27												
system coord	C22 Z C27 H271 X											
κ, κ'	0.973579	0.921152										
Pv	4.4549											
P00	0.0000											
P1m \pm	0.0000	-0.0540	0.0000									
P2m \pm	0.0000	0.0000	-0.0450	0.0000	0.0000							
P3m \pm	0.0060	0.0000	0.0000	0.3230	0.0000	0.0000	0.1990					

atom: C28												
system coord	C24 Z C28 H281 X											
κ, κ'	0.973579	0.921152										
Pv	4.4745											
P00	0.0000											
P1m \pm	0.0000	-0.0540	0.0000									
P2m \pm	0.0000	0.0000	-0.0450	0.0000	0.0000							
P3m \pm	0.0060	0.0000	0.0000	0.3230	0.0000	0.0000	0.1990					
atom: C29												
system coord	C26 Z C29 H291 X											
κ, κ'	0.973579	0.921152										
Pv	4.4109											
P00	0.0000											
P1m \pm	0.0000	-0.0540	0.0000									
P2m \pm	0.0000	0.0000	-0.0450	0.0000	0.0000							
P3m \pm	0.0060	0.0000	0.0000	0.3230	0.0000	0.0000	0.1990					
atom: C31												
system coord	C33 X C31 C32 Y											
κ, κ'	1.002619	0.958249										
Pv	4.0394											
P00	0.0000											
P1m \pm	-0.0570	-0.0080	0.1120									
P2m \pm	0.1180	0.0240	-0.1460	0.0020	-0.0220							
P3m \pm	-0.2840	-0.0020	-0.0120	-0.0120	0.0130	0.0070	0.0290					
atom: C32												
system coord	C35 X C32 C33 Y											
κ, κ'	1.002619	0.958249										
Pv	4.0102											
P00	0.0000											
P1m \pm	-0.0170	0.0160	0.0220									
P2m \pm	0.0260	0.0060	-0.1810	0.0220	0.0030							
P3m \pm	-0.0320	-0.0150	-0.0030	0.0280	0.0120	-0.0030	-0.2230					
atom: C33												
system coord	C36 X C33 C34 Y											
κ, κ'	1.002619	0.958249										
Pv	4.1531											
P00	0.0000											
P1m \pm	-0.0230	-0.0140	-0.0150									
P2m \pm	0.0240	-0.0100	-0.1800	-0.0240	-0.0050							
P3m \pm	-0.0350	-0.0230	-0.0330	-0.0010	0.0220	-0.0160	-0.2860					
atom: C34												
system coord	C31 X C34 C35 Y											
κ, κ'	1.002619	0.958249										
Pv	4.0465											
P00	0.0000											
P1m \pm	0.0190	-0.0040	0.0490									
P2m \pm	0.0210	-0.0170	-0.2110	0.0150	-0.0870							
P3m \pm	-0.0430	0.0030	0.0040	0.0090	0.0220	0.0070	-0.2660					

Table S2 (continued)

atom: C35
 system coord C32 X C35 C36 Y
 κ, κ' 1.002619 0.958249
 Pv 4.0275
 P00 0.0000
 P1m \pm 0.0000 -0.0060 -0.0340
 P2m \pm 0.0110 -0.0200 -0.1600 0.0000 0.0340
 P3m \pm -0.0240 -0.0070 0.0150 0.0100 0.0420 -0.0100 -0.2550

atom: C36
 system coord C33 X C36 C35 Y
 κ, κ' 1.002619 0.958249
 Pv 4.0154
 P00 0.0000
 P1m \pm 0.0060 -0.0010 0.0570
 P2m \pm -0.0040 0.0140 -0.1830 -0.0170 -0.0210
 P3m \pm 0.0430 -0.0080 -0.0120 0.0110 -0.0080 0.0230 -0.2880

atom: C37
 system coord C32 Z C37 H371 X
 κ, κ' 0.973579 0.921152
 Pv 4.4327
 P00 0.0000
 P1m \pm 0.0000 -0.0740 0.0000
 P2m \pm 0.0000 0.0000 -0.0570 0.0000 0.0000
 P3m \pm -0.0050 0.0000 0.0000 0.3210 0.0000 0.0000 0.1440

atom: C38
 system coord C34 Z C38 H381 X
 κ, κ' 0.973579 0.921152
 Pv 4.5581
 P00 0.0000
 P1m \pm 0.0000 -0.0740 0.0000
 P2m \pm 0.0000 0.0000 -0.0570 0.0000 0.0000
 P3m \pm -0.0050 0.0000 0.0000 0.3210 0.0000 0.0000 0.1440

atom: C39
 system coord C36 Z C39 H391 X
 κ, κ' 0.973579 0.921152
 Pv 4.3658
 P00 0.0000
 P1m \pm 0.0000 -0.0740 0.0000
 P2m \pm 0.0000 0.0000 -0.0570 0.0000 0.0000
 P3m \pm -0.0050 0.0000 0.0000 0.3210 0.0000 0.0000 0.1440

atom: H71
 system coord C7 Z H71 C5 X
 κ, κ' 1.142921 1.200000
 Pv 0.7630
 P00 0.0000
 P1m \pm 0.0000 0.1570 0.0000

atom: H72
 system coord C7 Z H72 C5
 κ, κ' 1.142921 1.200000
 Pv 0.7630
 P00 0.0000
 P1m \pm 0.0000 0.1970 0.0000

atom: H73
 system coord C7 Z H73 C5 X
 κ, κ' 1.142921 1.200000
 Pv 0.7630
 P00 0.0000
 P1m \pm 0.0000 0.1670 0.0000

atom: H11
 system coord C11 Z H11 C12 X
 κ, κ' 1.207180 1.200000
 Pv 0.7172
 P00 0.0000
 P1m \pm 0.0000 0.1020 0.0000

atom: H12
 system coord C12 Z H12 C11 X
 κ, κ' 1.207180 1.200000
 Pv 0.7172
 P00 0.0000
 P1m \pm 0.0000 0.1240 0.0000

atom: H13
 system coord C13 Z H13 C12 X
 κ, κ' 1.207180 1.200000
 Pv 0.7172
 P00 0.0000
 P1m \pm 0.0000 0.1400 0.0000

atom: H14
 system coord C14 Z H14 C13 X
 κ, κ' 1.207180 1.200000
 Pv 0.7172
 P00 0.0000
 P1m \pm 0.0000 0.1600 0.0000

atom: H15
 system coord C15 Z H15 C11 X
 κ, κ' 1.207180 1.200000
 Pv 0.7172
 P00 0.0000
 P1m \pm 0.0000 0.1170 0.0000

Table S2 (continued)

atom: H23					atom: H283				
system coord	C23 Z	H23	C22 X		system coord	C28 Z	H283	C24 X	
κ, κ'	1.265170	1.200000			κ, κ'	1.142921	1.200000		
Pv	0.7943				Pv	0.8437			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.0820	0.0000		P1m \pm	0.0000	0.1650	0.0000	
atom: H25					atom: H291				
system coord	C25 Z	H25	C24 X/		system coord	C29 Z	H291	C26 X	
κ, κ'	1.265170	1.200000			κ, κ'	1.142921	1.200000		
Pv	0.7943				Pv	0.8437			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.1160	0.0000		P1m \pm	0.0000	0.1410	0.0000	
atom: H271					atom: H292				
system coord	C27 Z	H271	C22 X		system coord	C29 Z	H292	C26 X	
κ, κ'	1.142921	1.200000			κ, κ'	1.142921	1.200000		
Pv	0.8437				Pv	0.8437			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.1640	0.0000		P1m \pm	0.0000	0.1230	0.0000	
atom: H272					atom: H293				
system coord	C27 Z	H272	C22 X		system coord	C29 Z	H293	C26 X	
κ, κ'	1.142921	1.200000			κ, κ'	1.142921	1.200000		
Pv	0.8437				Pv	0.8437			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.1840	0.0000		P1m \pm	0.0000	0.1470	0.0000	
atom: H273					atom: H33				
system coord	C27 Z	H273	C22 X		system coord	C33 Z	H33	C32 X	
κ, κ'	1.142921	1.200000			κ, κ'	1.265170	1.200000		
Pv	0.8437				Pv	0.7943			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.1320	0.0000		P1m \pm	0.0000	0.1440	0.0000	
atom: H281					atom: H35				
system coord	C28 Z	H281	C24 X		system coord	C35 Z	H35	C34 X	
κ, κ'	1.142921	1.200000			κ, κ'	1.265170	1.200000		
Pv	0.8437				Pv	0.7943			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.1730	0.0000		P1m \pm	0.0000	0.1310	0.0000	
atom: H282					atom: H371				
system coord	C28 Z	H282	C24 X		system coord	C37 Z	H371	C32 X	
κ, κ'	1.142921	1.200000			κ, κ'	1.142921	1.200000		
Pv	0.8437				Pv	0.8437			
P00	0.0000				P00	0.0000			
P1m \pm	0.0000	0.1040	0.0000		P1m \pm	0.0000	0.1620	0.0000	

Table S2 (continued)

atom: H372
 system coord C37 Z H372 C32 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1630 0.0000
atom: H373
 system coord C37 Z H373 C32 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1700 0.0000
atom: H381
 system coord C38 Z H381 C34 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1000 0.0000
atom: H382
 system coord C38 Z H382 C34 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1030 0.0000
atom: H383
 system coord C38 Z H383 C34 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1280 0.0000
atom: H391
 system coord C39 Z H391 C36 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1660 0.0000
atom: H392
 system coord C39 Z H392 C36 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m+ 0.0000 0.1440 0.0000

atom: H39
 system coord C39 Z H393 C36 X
 κ, κ' 1.142921 1.200000
 Pv 0.8437
 P00 0.0000
 P1m± 0.0000 0.1180 0.0000

Table S3. Values of experimental and theoretical (italic) orthodox parameters at bond critical points within complex **3_{Me}**^{a,b}

bond	R _b	d1	d2	ρ(r _b)	∇ ² ρ(r _b)	λ ₁	λ ₂	λ ₃	ε	Gρ(r _b)	Gρ(r _b)/ρ(r _b)	Vρ(r _b)	Hρ(r _b)	δ(A/B)	E _{int}
Interatomic bonds															
Fe1-C1	1.793	0.908	0.893	0.95	13.42	-4.29	-3.74	21.46	0.15	1.36	1.44	-1.79	-0.42		
	<i>1.889</i>	<i>0.971</i>	<i>0.926</i>	<i>0.75</i>	<i>11.89</i>	<i>-3.30</i>	<i>-2.31</i>	<i>17.48</i>	<i>0.44</i>	<i>1.05</i>	<i>1.40</i>	<i>-1.27</i>	<i>-0.27</i>	<i>0.74</i>	
Fe1-C2	1.782	0.907	0.875	1.09	13.48	-5.18	-4.91	23.56	0.06	1.55	1.43	-2.16	-0.61		
	<i>1.876</i>	<i>0.970</i>	<i>0.913</i>	<i>0.76</i>	<i>13.31</i>	<i>-2.75</i>	<i>-2.45</i>	<i>18.19</i>	<i>0.30</i>	<i>1.14</i>	<i>1.50</i>	<i>-1.35</i>	<i>-0.21</i>	<i>0.78</i>	
Fe1-C3	2.044	0.967	1.079	0.62	8.03	-2.09	-1.87	11.99	0.11	0.74	1.19	-0.91	-0.17		
	<i>2.075</i>	<i>0.966</i>	<i>1.111</i>	<i>0.60</i>	<i>6.72</i>	<i>-2.01</i>	<i>-1.58</i>	<i>10.52</i>	<i>0.12</i>	<i>0.64</i>	<i>1.07</i>	<i>-0.81</i>	<i>-0.17</i>	<i>0.61</i>	
Fe1-C11	2.141	1.097	1.055	0.53	6.14	-1.79	-0.19	8.12	8.41	0.57	1.07	-0.71	-0.14		
	<i>2.170</i>	<i>1.103</i>	<i>1.068</i>	<i>0.47</i>	<i>4.55</i>	<i>-1.54</i>	<i>-0.67</i>	<i>6.76</i>	<i>1.29</i>	<i>0.46</i>	<i>0.99</i>	<i>-0.60</i>	<i>-0.14</i>	<i>0.30</i>	
Fe1-C12						<i>not observed experimentally</i>									
	<i>2.159</i>	<i>1.123</i>	<i>1.057</i>	<i>0.46</i>	<i>5.06</i>	<i>-1.61</i>	<i>-0.28</i>	<i>6.95</i>	<i>4.74</i>	<i>0.48</i>	<i>1.04</i>	<i>-0.61</i>	<i>-0.13</i>	<i>0.33</i>	
Fe1-C13	2.103	1.056	1.07	0.53	6.10	-1.41	-0.41	7.93	2.40	0.57	1.06	-0.71	-0.14		
	<i>2.167</i>	<i>1.118</i>	<i>1.065</i>	<i>0.47</i>	<i>4.77</i>	<i>-1.53</i>	<i>-0.45</i>	<i>6.75</i>	<i>2.43</i>	<i>0.47</i>	<i>1.01</i>	<i>-0.61</i>	<i>-0.14</i>	<i>0.31</i>	
Fe1-C14	2.111	1.062	1.06	0.54	6.18	-1.59	-0.51	8.27	2.13	0.58	1.07	-0.73	-0.15		
	<i>2.177</i>	<i>1.110</i>	<i>1.072</i>	<i>0.46</i>	<i>4.53</i>	<i>-1.48</i>	<i>-0.62</i>	<i>6.62</i>	<i>1.39</i>	<i>0.46</i>	<i>0.99</i>	<i>-0.60</i>	<i>-0.14</i>	<i>0.30</i>	
Fe1-C15						<i>not observed experimentally</i>									
	<i>2.170</i>	<i>1.125</i>	<i>1.057</i>	<i>0.45</i>	<i>5.17</i>	<i>-1.74</i>	<i>-0.11</i>	<i>7.02</i>	<i>14.95</i>	<i>0.48</i>	<i>1.07</i>	<i>-0.60</i>	<i>-0.12</i>	<i>0.33</i>	
O1-C1	1.145	0.749	0.395	3.23	-11.26	-35.07	-32.74	56.55	0.07	5.16	1.60	-11.11	-5.95		
	<i>1.133</i>	<i>0.757</i>	<i>0.376</i>	<i>3.18</i>	<i>33.04</i>	<i>-34.91</i>	<i>-18.92</i>	<i>102.83</i>	<i>0.00</i>	<i>7.55</i>	<i>2.38</i>	<i>-12.79</i>	<i>-5.23</i>	<i>1.55</i>	
O2-C2	1.142	0.756	0.387	3.12	10.39	-32.41	-30.55	73.35	0.06	5.83	1.87	-10.93	-5.10		
	<i>1.133</i>	<i>0.756</i>	<i>0.376</i>	<i>3.18</i>	<i>32.82</i>	<i>-35.08</i>	<i>-18.93</i>	<i>102.90</i>	<i>0.00</i>	<i>7.55</i>	<i>2.37</i>	<i>-12.79</i>	<i>-5.24</i>	<i>1.55</i>	
O3-C4	1.241	0.767	0.474	2.76	-31.45	-25.25	-22.54	16.34	0.12	2.89	1.05	-7.99	-5.09		
	<i>1.227</i>	<i>0.821</i>	<i>0.406</i>	<i>2.66</i>	<i>2.33</i>	<i>-24.37</i>	<i>-13.35</i>	<i>49.09</i>	<i>0.09</i>	<i>4.65</i>	<i>1.75</i>	<i>-9.14</i>	<i>-4.49</i>	<i>1.18</i>	
O4-C6	1.236	0.747	0.489	2.90	-36.11	-27.56	-24.89	16.35	0.11	3.06	1.05	-8.64	-5.58	-	
	<i>1.227</i>	<i>0.822</i>	<i>0.406</i>	<i>2.66</i>	<i>2.04</i>	<i>-24.30</i>	<i>-13.34</i>	<i>48.78</i>	<i>0.08</i>	<i>4.63</i>	<i>1.74</i>	<i>-9.12</i>	<i>-4.49</i>	<i>1.17</i>	
N1-C3	1.355	0.823	0.532	2.18	-23.80	-16.84	-14.97	8.01	0.13	1.83	0.84	-5.34	-3.51		
	<i>1.349</i>	<i>0.891</i>	<i>0.458</i>	<i>2.16</i>	<i>-17.34</i>	<i>-15.62</i>	<i>-8.78</i>	<i>13.56</i>	<i>0.02</i>	<i>2.30</i>	<i>1.06</i>	<i>-5.81</i>	<i>-3.51</i>	<i>1.09</i>	
N1-C4	1.455	0.837	0.618	1.80	-11.31	-14.09	-11.88	14.66	0.19	1.61	0.90	-4.02	-2.41		
	<i>1.476</i>	<i>0.891</i>	<i>0.561</i>	<i>1.67</i>	<i>-15.24</i>	<i>-11.41</i>	<i>-5.19</i>	<i>7.05</i>	<i>0.05</i>	<i>0.92</i>	<i>0.55</i>	<i>-2.90</i>	<i>-1.99</i>	<i>0.79</i>	
N1-C21	1.446	0.822	0.624	1.87	-13.19	-13.88	-13.39	14.08	0.04	1.67	0.89	-4.26	-2.59		
	<i>1.439</i>	<i>0.911</i>	<i>0.529</i>	<i>1.82</i>	<i>-18.64</i>	<i>-12.40</i>	<i>-6.12</i>	<i>5.78</i>	<i>0.03</i>	<i>1.13</i>	<i>0.62</i>	<i>-3.56</i>	<i>-2.44</i>	<i>0.89</i>	
N2-C3	1.362	0.804	0.558	2.27	-23.58	-18.20	-15.54	10.16	0.17	2.04	0.90	-5.74	-3.70		
	<i>1.347</i>	<i>0.890</i>	<i>0.457</i>	<i>2.16</i>	<i>-17.29</i>	<i>-15.59</i>	<i>-8.85</i>	<i>13.73</i>	<i>0.01</i>	<i>2.31</i>	<i>1.07</i>	<i>-5.84</i>	<i>-3.53</i>	<i>1.10</i>	
N2-C6	1.462	0.843	0.620	1.75	-10.70	-13.54	-11.43	14.27	0.18	1.54	0.88	-3.82	-2.28		
	<i>1.480</i>	<i>0.915</i>	<i>0.565</i>	<i>1.66</i>	<i>-15.00</i>	<i>-11.33</i>	<i>-5.13</i>	<i>7.14</i>	<i>0.05</i>	<i>0.90</i>	<i>0.54</i>	<i>-2.84</i>	<i>-1.95</i>	<i>0.79</i>	
N2-C31	1.448	0.837	0.611	1.79	-12.29	-13.16	-12.33	13.20	0.07	1.55	0.87	-3.97	-2.42		
	<i>1.441</i>	<i>0.913</i>	<i>0.529</i>	<i>1.81</i>	<i>-18.41</i>	<i>-12.29</i>	<i>-11.93</i>	<i>5.81</i>	<i>0.03</i>	<i>1.12</i>	<i>0.62</i>	<i>-3.55</i>	<i>-2.42</i>	<i>0.08</i>	

C4-C5	1.396	0.736	0.660	2.17	-19.65	-17.95	-13.84	12.14	0.30	2.02	0.93	-5.41	-5.01		
	<i>1.401</i>	<i>0.743</i>	<i>0.658</i>	<i>2.07</i>	<i>-20.12</i>	<i>-15.43</i>	<i>-5.49</i>	<i>7.43</i>	<i>0.27</i>	<i>0.70</i>	<i>0.34</i>	<i>-2.80</i>	<i>-2.11</i>	<i>1.20</i>	
C5-C6	1.401	0.673	0.728	2.14	-17.99	-17.54	-12.99	12.53	0.35	2.01	0.94	-5.28	-3.27		
	<i>1.401</i>	<i>0.659</i>	<i>0.743</i>	<i>2.06</i>	<i>-19.93</i>	<i>-15.37</i>	<i>-5.47</i>	<i>7.44</i>	<i>0.28</i>	<i>0.70</i>	<i>0.34</i>	<i>-2.79</i>	<i>-2.10</i>	<i>1.21</i>	
C5-C7	1.498	0.767	0.731	1.78	-13.47	-12.61	-11.44	10.59	0.10	1.46	0.82	-3.87	-2.41		
	<i>1.500</i>	<i>0.765</i>	<i>0.735</i>	<i>1.69</i>	<i>-13.77</i>	<i>-11.16</i>	<i>-4.04</i>	<i>8.02</i>	<i>0.05</i>	<i>0.42</i>	<i>0.25</i>	<i>-1.80</i>	<i>-1.39</i>	<i>1.01</i>	
C-C (<i>Cp</i>)	1.424	0.716	0.709	2.02	-16.64	-14.450	-11.65	9.50	0.24	1.82	0.90	-4.80	-2.98		
	<i>1.418</i>	<i>0.713</i>	<i>0.705</i>	<i>1.98</i>	<i>-17.49</i>	<i>-14.00</i>	<i>-11.35</i>	<i>7.86</i>	<i>0.23</i>	<i>0.67</i>	<i>0.34</i>	<i>-2.58</i>	<i>-1.90</i>	<i>1.27</i>	
C-H (<i>Cp</i>)	1.083	0.76	0.32	1.76	-16.64	-17.47	-16.62	17.46	0.05	1.28	0.73	-3.73	-2.45		
	<i>1.081</i>	<i>0.683</i>	<i>0.398</i>	<i>1.91</i>	<i>-24.91</i>	<i>-18.14</i>	<i>-17.85</i>	<i>11.07</i>	<i>0.02</i>	<i>0.26</i>	<i>0.14</i>	<i>-2.27</i>	<i>-2.01</i>	<i>0.95</i>	
C-C (<i>aryl</i>)	1.40	0.70	0.70	2.14	-18.69	-16.60	-13.43	11.33	0.23	1.99	0.93	-5.28	-3.29		
	<i>1.396</i>	<i>0.704</i>	<i>0.692</i>	<i>2.08</i>	<i>-19.94</i>	<i>-15.26</i>	<i>-12.56</i>	<i>7.88</i>	<i>0.22</i>	<i>0.67</i>	<i>0.32</i>	<i>-2.75</i>	<i>-2.07</i>	<i>1.33</i>	
C-H (<i>aryl</i>)	1.083	0.73	0.36	1.78	-14.10	-17.38	-16.40	19.71	0.06	1.45	0.81	-3.88	-2.43		
	<i>1.088</i>	<i>0.677</i>	<i>0.411</i>	<i>1.88</i>	<i>-23.50</i>	<i>-17.40</i>	<i>-17.06</i>	<i>10.96</i>	<i>0.02</i>	<i>0.29</i>	<i>0.15</i>	<i>-2.23</i>	<i>-1.94</i>	<i>0.95</i>	
C-C(<i>sp</i> ³)	1.502	0.76	0.74	1.71	-11.86	-11.54	-10.80	10.48	0.07	1.41	0.83	-3.65	-2.24		
	<i>1.505</i>	<i>0.765</i>	<i>0.740</i>	<i>1.70</i>	<i>-14.24</i>	<i>-11.34</i>	<i>-11.02</i>	<i>8.12</i>	<i>0.03</i>	<i>0.40</i>	<i>0.23</i>	<i>-1.80</i>	<i>-1.40</i>	<i>1.01</i>	
C(<i>sp</i> ³)-H	1.093	0.75	0.35	1.68	-14.11	-15.32	-14.16	15.37	0.08	1.24	0.74	-3.47	-2.23		
	<i>1.094</i>	<i>0.682</i>	<i>0.412</i>	<i>1.85</i>	<i>-22.52</i>	<i>-16.15</i>	<i>-16.65</i>	<i>10.93</i>	<i>0.01</i>	<i>0.30</i>	<i>0.16</i>	<i>-2.19</i>	<i>-1.88</i>	<i>0.94</i>	
Intramolecular interligand weak interactions															
C1...C21	2.720	1.337	1.388	0.13	1.15	-0.30	-0.24	1.69	0.28	0.08	0.61	-0.08	0.00		3.9
	<i>2.811</i>	<i>1.357</i>	<i>1.455</i>	<i>0.11</i>	<i>1.16</i>	<i>-0.28</i>	<i>-0.17</i>	<i>1.62</i>	<i>0.62</i>	<i>0.07</i>	<i>0.65</i>	<i>-0.06</i>	<i>0.01</i>	<i>0.05</i>	3.1
C2...C36	not observed experimentally														
	2.908	<i>1.516</i>	<i>1.409</i>	0.08	0.95	-0.19	-0.05	1.19	2.67	0.05	0.72	-0.04	0.01	0.02	2.0
C32...H15	2.39	1.44	0.999	0.09	1.08	-0.22	-0.18	1.48	0.25	0.06	0.67	-0.05	0.01		2.5
	<i>2.489</i>	<i>1.485</i>	<i>1.010</i>	<i>0.07</i>	<i>0.91</i>	<i>-0.19</i>	<i>-0.17</i>	<i>1.26</i>	<i>0.12</i>	<i>0.05</i>	<i>0.72</i>	<i>-0.04</i>	<i>0.01</i>	<i>0.03</i>	2.0
H11...H372	1.992	1.924	1.145	0.05	0.99	-0.18	-0.15	1.32	0.18	0.05	1.0	-0.04	0.02		1.7
	<i>2.155</i>	<i>1.101</i>	<i>1.072</i>	<i>0.05</i>	<i>0.73</i>	<i>-0.19</i>	<i>-0.15</i>	<i>1.07</i>	<i>0.22</i>	<i>0.04</i>	<i>0.73</i>	<i>-0.03</i>	<i>0.01</i>	<i>0.02</i>	1.4
O3...H48	not observed experimentally														
	2.525	<i>1.476</i>	<i>1.075</i>	0.07	0.92	-0.21	-0.15	1.28	0.46	0.06	0.82	-0.05	0.01	0.03	1.9
O4...H61	not observed experimentally														
	2.623	<i>1.522</i>	<i>1.151</i>	0.06	0.83	-0.15	-0.08	1.06	0.87	0.05	0.83	-0.04	0.01	0.02	1.6

^aTop line experimental values, second line (italic) theoretical values from isolated molecule DFT calculation. ^b R_b is the interatomic distance (Å); $d1$ and $d2$ are the distances of the bcp from the two atoms (Å); $\rho(\mathbf{r}_b)$ is the ED (e.Å⁻³); $\nabla^2\rho(\mathbf{r}_b)$ is the Laplacian of the ED (e.Å⁻⁵); λ_{1-3} are the curvatures of the bond (e.Å⁻⁵); ϵ is the ellipticity of the bond; $G\rho(\mathbf{r}_b)$, $V\rho(\mathbf{r}_b)$, and $H\rho(\mathbf{r}_b)$ are the kinetic, potential, and total electron energy densities estimated using the approximation of Abramov (hartree.Å⁻³).^f $\delta(C\dots C)$ are the delocalization indices; ^f E_{int} is the interaction energy estimated using Espinoza correlation scheme (kcal.mol⁻¹).

Figure S1. Residual electron density maps for **3_{Me}** after the multipolar refinement (0.1 e. \AA^{-3} isocontours; positive: full lines; negative: dashed lines; zero contour: yellow dashed line).

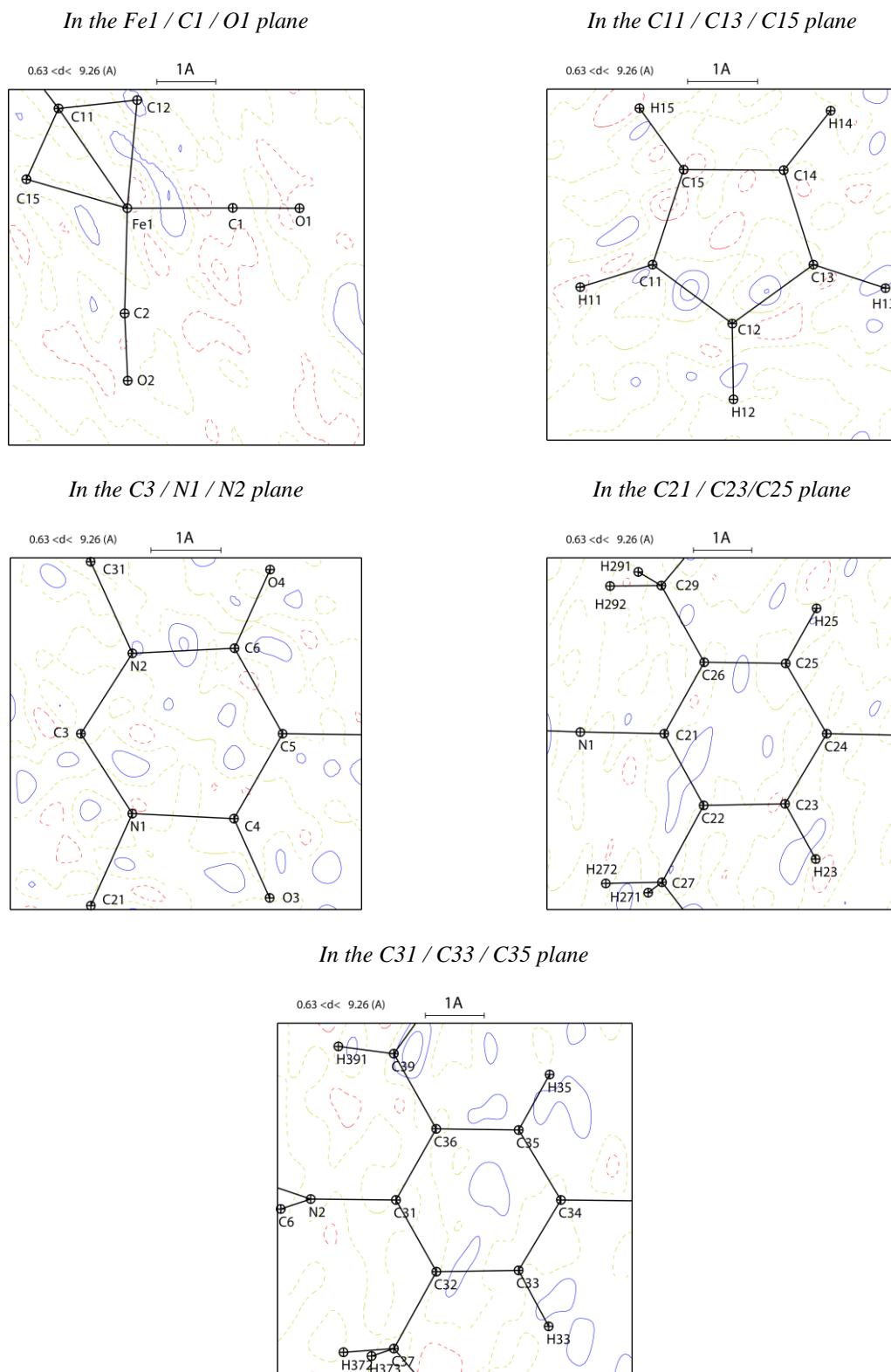


Figure S2. Laplacian of the experimental (left) and theoretical (right, M06-2X/6-31G**) electron density for $\mathbf{3}_{\text{Me}}$ in various planes. Contours are drawn at $0.000, \pm 2.0 \times 10n, \pm 4.0 \times 10n, \pm 8.0 \times 10n \text{ e}.\text{\AA}^{-5}$ levels, where $n = 0, -3, \pm 2, \pm 1$; positive: solid (red) lines, negative: dashed (blue) lines.

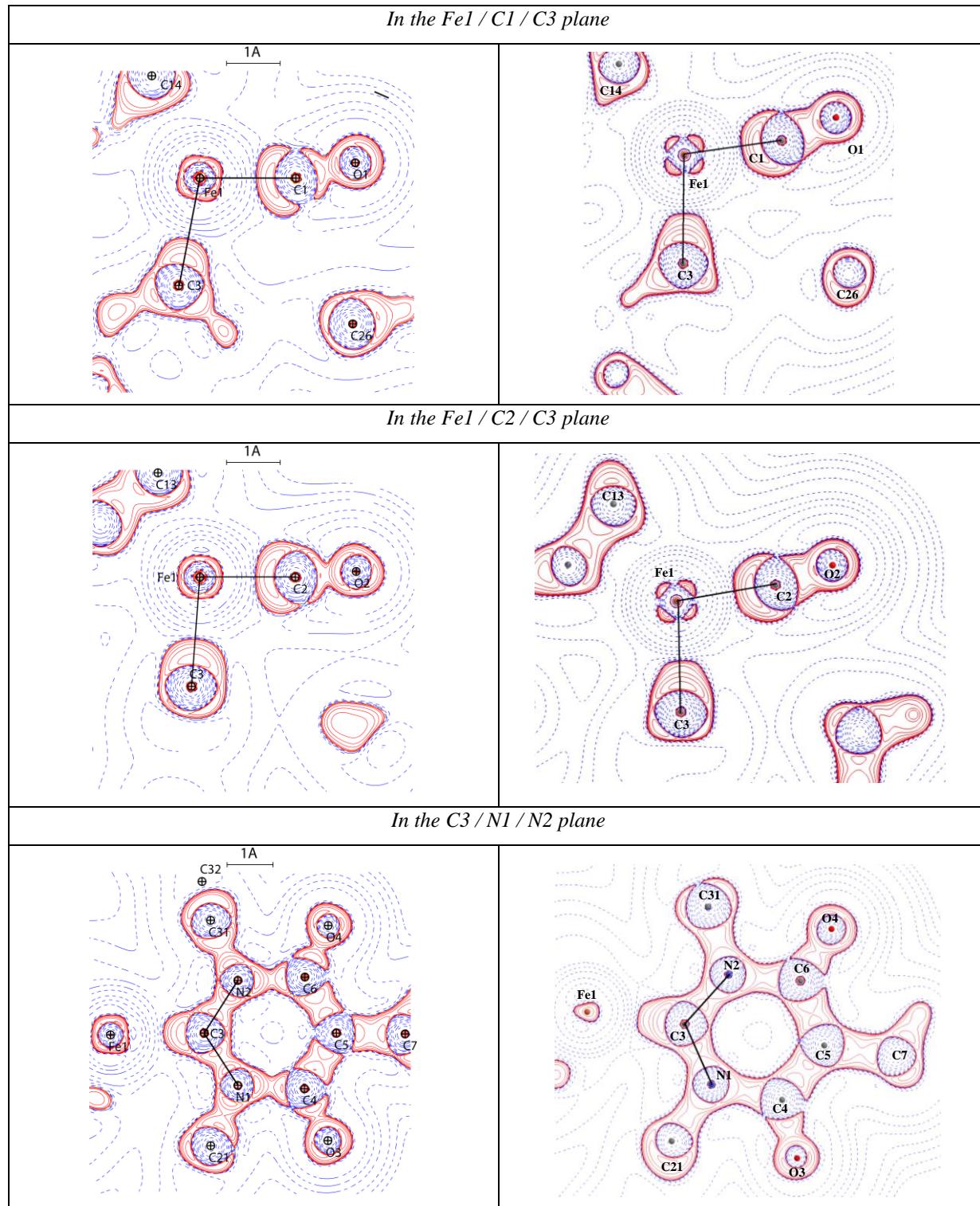
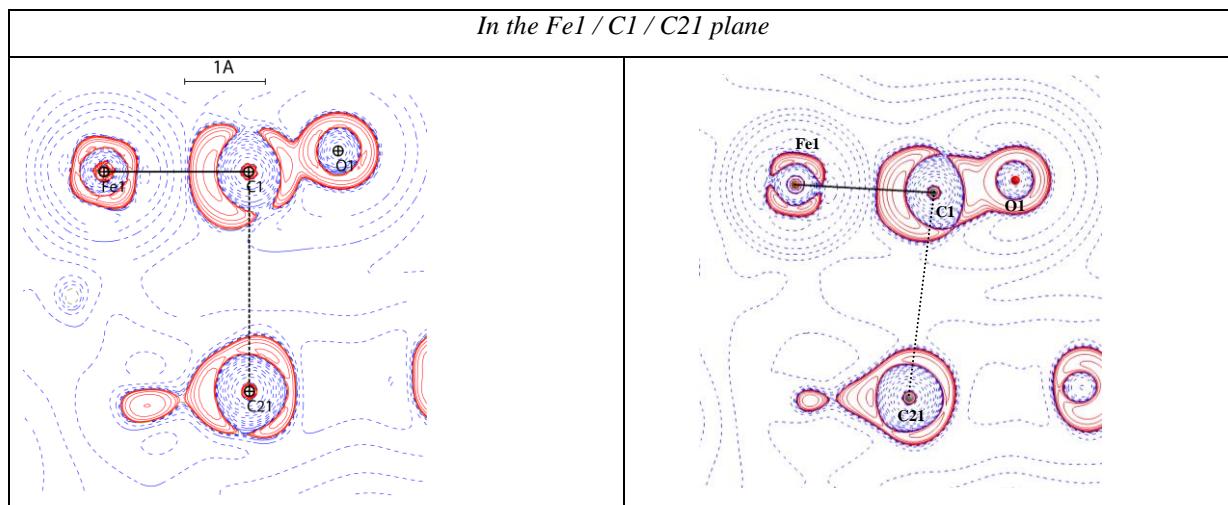


Figure S2 (continued)



Computational studies

Calculations were carried out with the Gaussian 09 program¹ at the DFT level of theory using the M06-2X functional.² All the atoms (C, N, H, O, Fe) were described with a 6-31G(d,p) double- ζ basis set.³ Geometry optimizations were carried out without any symmetry restrictions, the nature of the *minima* was verified with analytical frequency calculations. All total energies and Gibbs free energies were zero-point energy (ZPE) and temperature corrected using unscaled density functional frequencies.

The electronic structure of the different complexes was studied using Natural Bond Orbital analysis

¹ Gaussian 09. Revision C.01 Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, Jr., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford CT, 2009.

² Zhao, Y.; Truhlar, D. G. *Theor. Chem. Acc.* **2008**, *120*, 215.

³ Hariharan, P. C.; Pople, J. A. *Theor. Chim. Acta* **1973**, *28*, 213.

(NBO-5 program).⁴ The electron density of the optimized structures was subject to an Atoms-In Molecules analysis (QTAIM analysis).⁵ The Natural Localized Molecular Orbital associated with the Fe-C_{carbene} bond was plotted by using the molecular graphic package Molekel.⁶

In order to choose the better level of theory (M06-2X), geometry optimizations of **3_{Me}** were carried out with different functionals like B3LYP,⁷ BPW91,^{11a-b,8} M06,⁶ M06-L,⁹ B97D¹⁰ and WB97XD.¹¹ Diffuse functions were also been added in some cases on the different atoms except hydrogens and iron.

⁴ (a) *NBO 5.0*. Glendening, E. D.; Badenhoop, J. K.; Reed, A. E.; Carpenter, J. E.; Bohmann, J. A.; Morales, C. M.; Weinhold, F. Theoretical Chemistry Institute, University of Wisconsin, Madison, **2001**; (b) Reed, E.; Curtiss, L. A.; Weinhold, F. *Chem. Rev.* **1988**, *88*, 899; (c) Foster, J. P.; Weinhold, F. *J. Am. Chem. Soc.* **1980**, *102*, 7211; (d) Reed A. E.; Weinhold, F. *J. Chem. Phys.* **1985**, *83*, 1736.

⁵ (a) Bader, R. F. W. *Atoms in Molecules: A Quantum Theory*; Oxford University Press: New-York Ed., **1990**; (b) Bader, R. F. W. *Chem. Rev.* **1991**, *91*, 893; (c) AIMAll (Version 10.10.11), Keith, T. A. **2010** (aim.tkgristmill.com).

⁶ (a) Molekel 4.3, Flükiger, P.; Lüthi, H. P.; Portmann, S.; Weber, J. Swiss Center for Scientific Computing, Manno, Switzerland, **2000-2002**; (b) Portmann, S.; Lüthi, H. P. *Chimica* **2000**, *54*, 766.

⁷ (a) Becke, A. D. *J. Chem. Phys.* **1993**, *98*, 5648; (b) Becke, A. D. *Phys. Rev. A* **1988**, *38*, 3098; (c) Lee, C.; Yang, W.; Parr, R. G. *Phys. Rev.* **1988**, *B37*, 785.

⁸ Perdew, J. P. in *Electronic Structure of Solids '91*, Ed. P. Ziesche and H. Eschrig (Akademie Verlag, Berlin), **1991**, 11.

⁹ Zhao, Y.; Truhlar, D. G. *J. Chem. Phys.*, **2006**, *125*, 1.

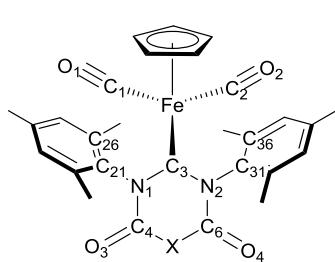
¹⁰ (a) Becke, A. D. *J. Chem. Phys.*, **1997**, *107*, 8554; (b) Schmider, H. L.; Becke, A. D. *J. Chem. Phys.*, **1998**, *108*, 9624; (c) Grimme, S. *J. Comp. Chem.*, **2006**, *27*, 1787.

¹¹ Chai, J.-D.; Head-Gordon, M. *Phys. Chem. Chem. Phys.*, **2008**, *10*, 6615.

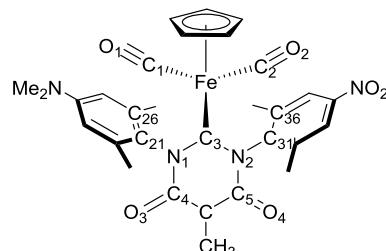
Table S4. Main geometrical parameters (Interatomic distances, in Å, and bond angles, in°) for the complex **3_{Me}** at different level of theory.

Bond or angle	B3LYP/6-31+G** SDD(Fe)	B3LYP/6-31G** SDD (Fe)	B3LYP/6-31+G**	B3LYP/6-31G**	B97D/6-31+G**	BPW91/6-31+G**	BPW91/6-31G**	WB97XD/6-31+G**	M06/6-31+G**	M06L/6-31+G**	M062X/6-31+G**	M062X/6-31G**	RX
<i>d</i> _(N1-C3)	1.357	1.356	1.357	1.356	1.356	1.364	1.361	1.350	1.350	1.351	1.350	1.349	1.363
<i>d</i> _(C1-N2)	1.357	1.355	1.357	1.355	1.356	1.362	1.362	1.350	1.350	1.351	1.349	1.349	1.356
$\angle_{\text{N1-C3-N2}}$ (deg)	115.09	114.92	115.15	114.83	115.70	115.21	114.93	114.89	115.25	115.57	114.79	114.59	114.49
<i>d</i> _(N1-C4)	1.491	1.499	1.495	1.500	1.513	1.510	1.517	1.474	1.485	1.495	1.471	1.476	1.462
<i>d</i> _(C4-C5)	1.402	1.401	1.403	1.401	1.410	1.409	1.408	1.399	1.400	1.400	1.402	1.401	1.400
<i>d</i> _(C5-C6)	1.403	1.401	1.402	1.401	1.410	1.409	1.408	1.399	1.400	1.400	1.403	1.401	1.396
<i>d</i> _(C6-N2)	1.495	1.497	1.492	1.498	1.511	1.510	1.516	1.474	1.490	1.495	1.476	1.480	1.454
$\angle_{\text{N1-C4-C5}}$ (deg)	115.96	115.63	115.96	115.58	115.436	115.758	115.686	116.088	116.030	115.528	115.974	115.477	116.409
$\angle_{\text{N1-C4-C5-C6}}$ (deg)	-2.38	-3.07	-3.02	-2.82	-1.62	-3.94	0.86	-1.46	4.69	2.04	1.62	-0.04	6.65
<i>d</i> _(Fe-C2)	1.789	1.785	1.780	1.773	1.756	1.753	1.748	1.784	1.782	1.767	1.894	1.876	1.782
<i>d</i> _(C2-O)	1.150	1.150	1.151	1.151	1.166	1.169	1.169	1.144	1.149	1.158	1.133	1.133	1.143
$\angle_{(\text{Fe-C2-O2})}$ (deg)	168.32	168.62	167.67	167.92	167.82	167.89	167.52	167.68	168.22	168.36	170.22	169.77	173.95
<i>d</i> _(Fe-C1)	1.79	1.79	1.780	1.775	1.756	1.755	1.749	1.784	1.782	1.767	1.90	1.89	1.793
<i>d</i> _(C1-O1)	1.150	1.149	1.151	1.151	1.166	1.169	1.169	1.144	1.149	1.158	1.133	1.133	1.146
$\angle_{(\text{Fe-C1-O1})}$ (deg)	168.33	168.91	167.68	167.18	167.77	166.57	166.70	167.72	168.21	168.35	167.03	165.49	165.45
<i>d</i> _(C3-Fe)	2.07	2.07	2.067	2.059	2.047	2.054	2.046	2.041	2.047	2.053	2.089	2.075	2.044
<i>d</i> _(N2-C31)	1.45	1.45	1.453	1.450	1.447	1.453	1.451	1.442	1.442	1.442	1.443	1.441	1.448
<i>d</i> _(C31-C2)	2.996	3.045	2.989	3.033	2.979	3.076	3.056	2.959	2.927	2.968	3.085	3.105	3.183
<i>d</i> _(C31-O2)	3.509	3.574	3.506	3.573	3.512	3.634	3.619	3.480	3.407	3.474	3.629	3.666	3.687
$\angle_{(\text{N2-C31-C2})}$ (deg)	90.54	88.84	90.40	88.64	91.33	87.03	86.12	91.54	92.14	91.77	92.11	90.32	84.03
<i>d</i> _(N1-C21)	1.45	1.45	1.453	1.450	1.447	1.453	1.451	1.442	1.441	1.442	1.442	1.439	1.446
<i>d</i> _(C21-C1)	3.00	2.96	2.991	2.964	2.968	2.946	2.980	2.954	2.925	2.955	2.863	2.811	2.720
<i>d</i> _(C21-O1)	3.51	3.47	3.508	3.483	3.497	3.467	3.520	3.474	3.404	3.453	3.352	3.296	3.178
$\angle_{(\text{N1-C21-C1})}$ (deg)	90.43	91.00	90.23	89.97	91.57	90.37	88.34	91.60	92.04	92.09	97.77	98.15	97.04

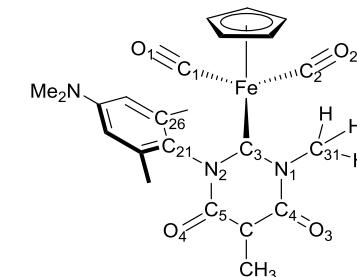
Table S5. Experimental and/or optimized (M06-2X/6-31G** level of theory) geometrical parameters for complexes **3_{Me}**, **4**, **6^{Me}(OTf)**, **7**, and **8**.



3_{Me}, 4, and [6^{Me}](OTf)



7



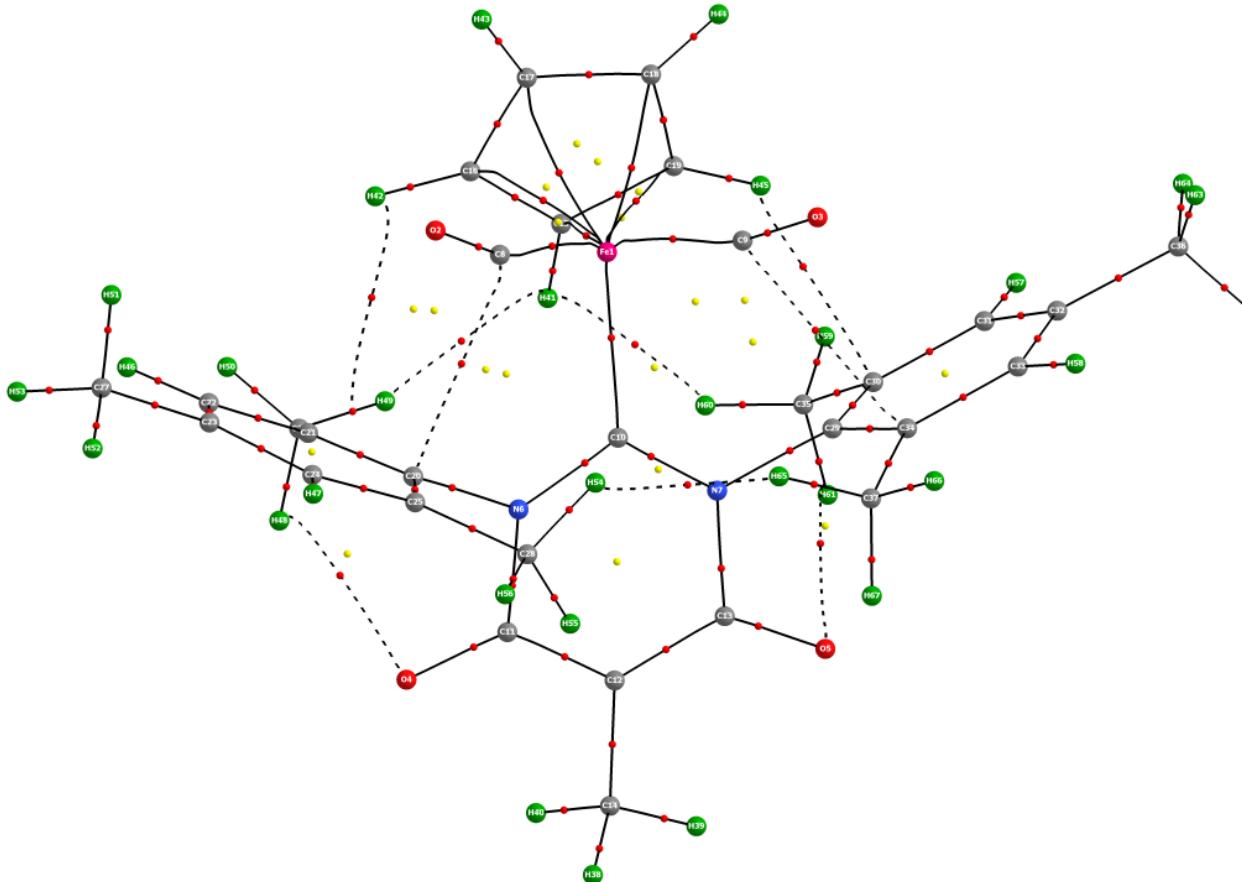
8

bond or angle ^a	3_{Me}(exp)	3_{Me}(theo)	4(exp)	4(theo)	[6^{Me}](OTf)(exp)	[6^{Me}](OTf)(theo)^b	7	8
Fe-C3	2.044	2.075	2.031	2.065	2.003	2.079	2.074	2.061
N1-C4	1.463	1.480	1.464	1.495	1.423	1.431	1.480	1.477
N2-C5	1.455	1.476	1.464	1.494	1.423	1.428	1.479	1.474
C4-O3	1.236	1.227	1.219	1.215	1.198	1.203	1.229	1.229
C5-O4	1.241	1.227	1.221	1.214	1.205	1.204	1.225	1.228
C4-X	1.401	1.401	1.333	1.331	1.373	1.374	1.399	1.401
C5-X	1.396	1.401	1.330	1.330	1.365	1.372	1.403	1.401
C1-C21	2.720	2.811	2.867	2.816	2.856	2.856	2.755	2.663
C1-C26	2.922	2.896	2.924	2.908	3.020	2.939	2.878	2.909
C2-C31	3.180	3.105	3.095	3.119	3.042	3.070	3.142	3.243
C2-C36	3.052	2.908	3.001	2.923	2.930	2.955	2.935	/
Fe-C1-O1	174.00	169.77	171.84	170.22	171.96	166.65	164.906	163.15
Fe-C2-O2	165.48	165.49	168.73	165.31	171.29	169.76	171.047	178.77

^a bond lengths in Å and bond angles in deg; ^b for **6^{Me}**, the geometry optimization was carried out by taking into account the counter-anion.

Table S6. Experimental and theoretical topological analysis of complex $\mathbf{3}_{\text{Me}}^a$

Molecular graph including bcp's (red dots), rcp's (yellow dots), and bp's (black lines) within or close to that plane.



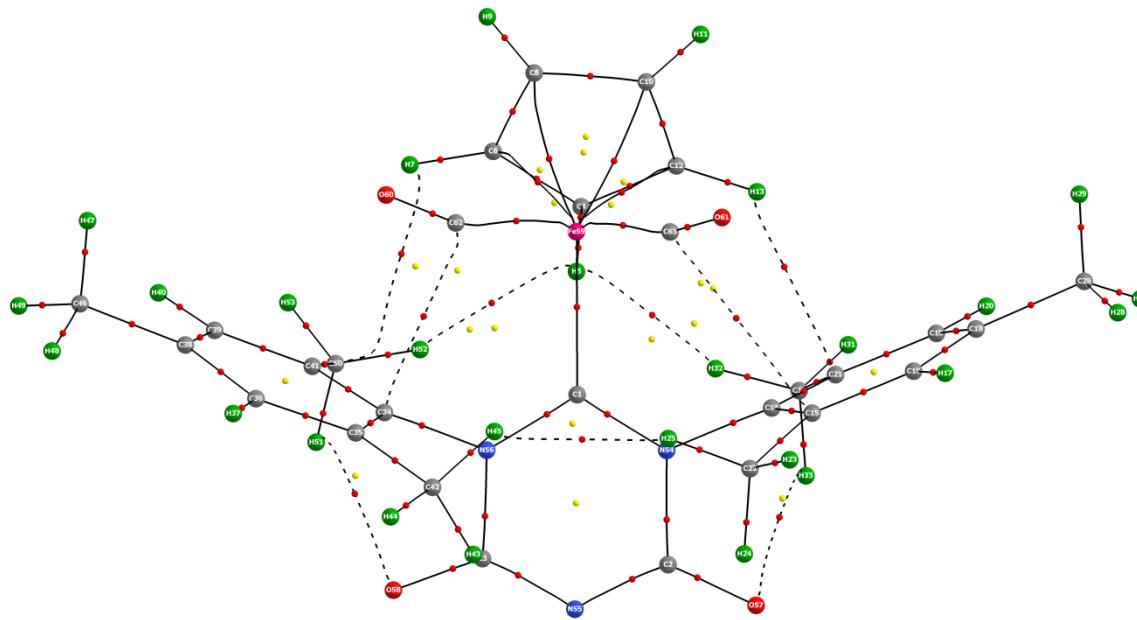
bond	R_b^b	$d1$	$d2$	$\rho(\mathbf{r}_b)^c$	$\nabla^2\rho(\mathbf{r}_b)^d$	λ_1^e	λ_2^e	λ_3^e	ε^f	$G\rho(\mathbf{r}_b)^g$	$G\rho(\mathbf{r}_b)/\rho(\mathbf{r}_b)$	$V\rho(\mathbf{r}_b)^g$	$H\rho(\mathbf{r}_b)^g$
Fe1-C1	1.793	0.908	0.893	0.95	13.42	-4.29	-3.74	21.46	0.15	1.36	1.44	-1.79	-0.43
	<i>1.889</i>	<i>0.971</i>	<i>0.926</i>	<i>0.75</i>	<i>11.89</i>	<i>-3.30</i>	<i>-2.31</i>	<i>17.48</i>	<i>0.44</i>	<i>1.05</i>	<i>1.40</i>	<i>-1.27</i>	<i>-0.22</i>
Fe1-C2	1.782	0.907	0.875	1.09	13.48	-5.18	-4.91	23.56	0.06	1.55	1.43	-2.16	-0.61
	<i>1.876</i>	<i>0.970</i>	<i>0.913</i>	<i>0.76</i>	<i>13.31</i>	<i>-2.75</i>	<i>-2.45</i>	<i>18.19</i>	<i>0.30</i>	<i>1.14</i>	<i>1.50</i>	<i>-1.35</i>	<i>-0.21</i>
Fe1-C3	2.044	0.967	1.079	0.62	8.03	-2.09	-1.87	11.99	0.11	0.74	1.19	-0.91	-0.17
	<i>2.075</i>	<i>0.966</i>	<i>1.111</i>	<i>0.60</i>	<i>6.72</i>	<i>-2.01</i>	<i>-1.58</i>	<i>10.52</i>	<i>0.12</i>	<i>0.64</i>	<i>1.07</i>	<i>-0.81</i>	<i>-0.17</i>
O1-C1	1.145	0.749	0.395	3.23	-11.26	-35.07	-32.74	56.55	0.07	5.16	1.60	-11.11	-5.95
	<i>1.133</i>	<i>0.757</i>	<i>0.376</i>	<i>3.18</i>	<i>33.04</i>	<i>-34.91</i>	<i>-18.92</i>	<i>102.83</i>	<i>0.00</i>	<i>7.55</i>	<i>2.38</i>	<i>-12.79</i>	<i>-5.23</i>
O2-C2	1.142	0.756	0.387	3.12	10.39	-32.41	-30.55	73.35	0.06	5.83	1.87	-10.93	-5.10
	<i>1.133</i>	<i>0.756</i>	<i>0.376</i>	<i>3.18</i>	<i>32.82</i>	<i>-35.08</i>	<i>-18.93</i>	<i>102.90</i>	<i>0.00</i>	<i>7.55</i>	<i>2.37</i>	<i>-12.79</i>	<i>-5.24</i>
N1-C3	1.355	0.823	0.532	2.18	-23.80	-16.84	-14.97	8.01	0.13	1.83	0.84	-5.34	-3.51
	<i>1.349</i>	<i>0.891</i>	<i>0.458</i>	<i>2.16</i>	<i>-17.34</i>	<i>-15.62</i>	<i>-8.78</i>	<i>13.56</i>	<i>0.02</i>	<i>2.30</i>	<i>1.06</i>	<i>-5.81</i>	<i>-3.51</i>
N1-C4	1.455	0.837	0.618	1.80	-11.31	-14.09	-11.88	14.66	0.19	1.61	0.90	-4.02	-2.41
	<i>1.476</i>	<i>0.891</i>	<i>0.561</i>	<i>1.67</i>	<i>-15.24</i>	<i>-11.41</i>	<i>-5.19</i>	<i>7.05</i>	<i>0.05</i>	<i>0.92</i>	<i>0.55</i>	<i>-2.90</i>	<i>-1.99</i>
N1-C21	1.446	0.822	0.624	1.87	-13.19	-13.88	-13.39	14.08	0.04	1.67	0.89	-4.26	-2.59
	<i>1.439</i>	<i>0.911</i>	<i>0.529</i>	<i>1.82</i>	<i>-18.64</i>	<i>-12.40</i>	<i>-6.12</i>	<i>5.78</i>	<i>0.03</i>	<i>1.13</i>	<i>0.62</i>	<i>-3.56</i>	<i>-2.44</i>
N2-C3	1.362	0.804	0.558	2.27	-23.58	-18.20	-15.54	10.16	0.17	2.04	0.90	-5.74	-3.70
	<i>1.347</i>	<i>0.890</i>	<i>0.457</i>	<i>2.16</i>	<i>-17.29</i>	<i>-15.59</i>	<i>-8.85</i>	<i>13.73</i>	<i>0.01</i>	<i>2.31</i>	<i>1.07</i>	<i>-5.84</i>	<i>-3.53</i>
N2-C6	1.462	0.843	0.620	1.75	-10.70	-13.54	-11.43	14.27	0.18	1.54	0.88	-3.82	-2.28
	<i>1.480</i>	<i>0.915</i>	<i>0.565</i>	<i>1.66</i>	<i>-15.00</i>	<i>-11.33</i>	<i>-5.13</i>	<i>7.14</i>	<i>0.05</i>	<i>0.90</i>	<i>0.54</i>	<i>-2.84</i>	<i>-1.95</i>
N2-C31	1.448	0.837	0.611	1.79	-12.29	-13.16	-12.33	13.20	0.07	1.55	0.87	-3.97	-2.42
	<i>1.441</i>	<i>0.913</i>	<i>0.529</i>	<i>1.81</i>	<i>-18.41</i>	<i>-12.29</i>	<i>-11.93</i>	<i>5.81</i>	<i>0.03</i>	<i>1.12</i>	<i>0.62</i>	<i>-3.55</i>	<i>-2.42</i>
C4-C5	1.396	0.736	0.660	2.17	-19.65	-17.95	-13.84	12.14	0.30	2.02	0.93	-5.41	-5.01
	<i>1.401</i>	<i>0.743</i>	<i>0.658</i>	<i>2.07</i>	<i>-20.12</i>	<i>-15.43</i>	<i>-5.49</i>	<i>7.43</i>	<i>0.27</i>	<i>0.70</i>	<i>0.34</i>	<i>-2.80</i>	<i>-2.11</i>
C5-C6	1.401	0.673	0.728	2.14	-17.99	-17.54	-12.99	12.53	0.35	2.01	0.94	-5.28	-3.27
	<i>1.401</i>	<i>0.659</i>	<i>0.743</i>	<i>2.06</i>	<i>-19.93</i>	<i>-15.37</i>	<i>-5.47</i>	<i>7.44</i>	<i>0.28</i>	<i>0.70</i>	<i>0.34</i>	<i>-2.79</i>	<i>-2.10</i>
C5-C7	1.498	0.767	0.731	1.78	-13.47	-12.61	-11.44	10.59	0.10	1.46	0.82	-3.87	-2.41
	<i>1.500</i>	<i>0.765</i>	<i>0.735</i>	<i>1.69</i>	<i>-13.77</i>	<i>-11.16</i>	<i>-4.04</i>	<i>8.02</i>	<i>0.05</i>	<i>0.42</i>	<i>0.25</i>	<i>-1.80</i>	<i>-1.39</i>

^aTop line experimental values, second line (italic) theoretical values from isolated molecule DFT calculation. ^b R_b is the interatomic distance (Å); ^c $d1$ and $d2$ are the distances of the bcp from the two atoms (Å); ^c $\rho(\mathbf{r}_b)$ is the ED (e.Å⁻³); ^d $\nabla^2\rho(\mathbf{r}_b)$ is the Laplacian of the ED (e.Å⁻⁵); ^e λ_{1-3} are the curvatures of the bond (e.Å⁻⁵); ^f ε is the ellipticity of the bond; ^g $G\rho(\mathbf{r}_b)$, $V\rho(\mathbf{r}_b)$, and $H\rho(\mathbf{r}_b)$ are the kinetic, potential, and total electron energy densities estimated using the approximation of Abramov (hartree.Å⁻³).¹²

¹² Abramov, Y. A. *Acta Crystallogr., Sect A*. **1997**, *53*, 264.

Table S7. AIM analysis for complex **4**.

Molecular graph including bcp's (red dots), rcp's (yellow dots), and bp's (black lines).



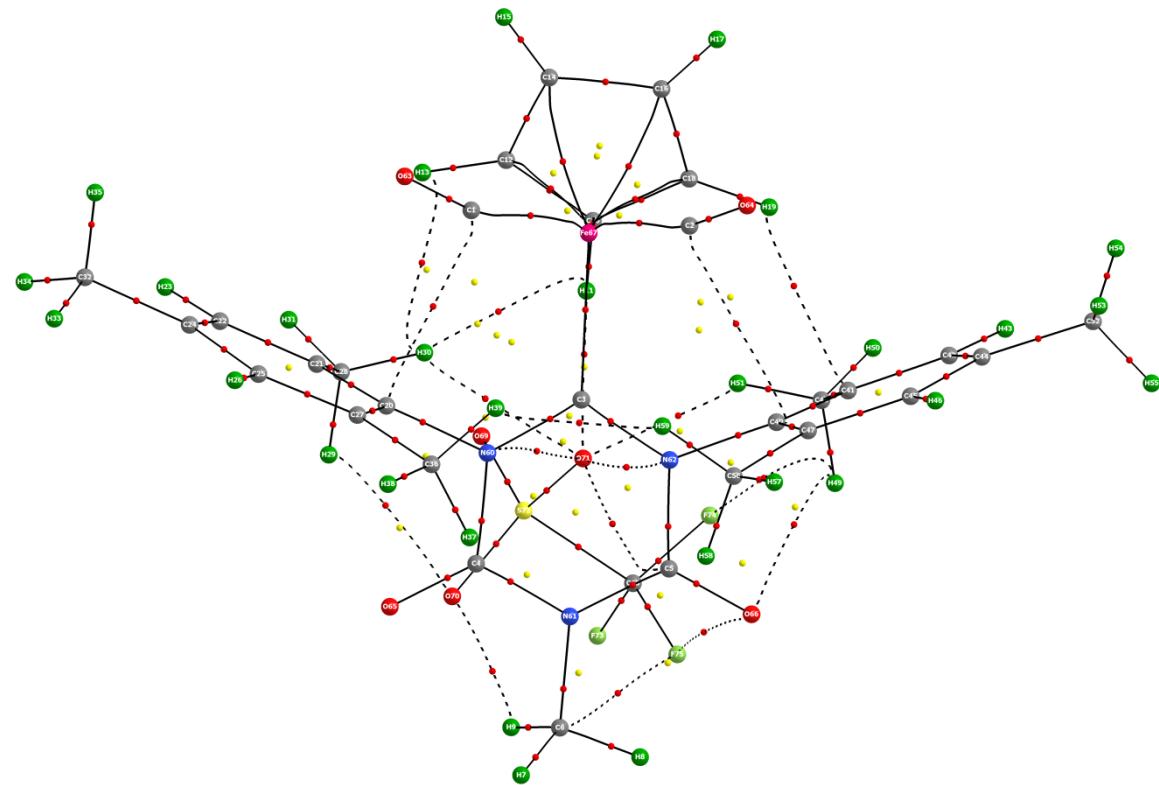
bond	$\rho(r_b)$ (eÅ ⁻³)	$\nabla^2\rho(r_b)$ (eÅ ⁻⁵)	ϵ	H	$\delta(A/B)$	bond	$\rho(r_b)$ (eÅ ⁻³)	$\nabla^2\rho(r_b)$ (eÅ ⁻⁵)	ϵ	H	$\delta(A/B)$
C1 - N54	2.19	-18.29	0.01	-3.58	1.10	H5 - H32	0.05	0.75	0.46	0.01	0.02
C2 - N55	2.36	-29.34	0.14	-3.70	1.08	C21 - C30	1.70	-14.26	0.03	-1.41	1.01
C1 - N56	2.19	-18.13	0.03	-3.58	1.09	C30 - H31	1.85	-22.57	0.01	-1.89	0.96
C3 - N55	2.36	-29.35	0.14	-3.71	1.08	C30 - H32	1.85	-22.34	0.01	-1.88	0.93
C1 - Fe59	0.61	6.93	0.12	-0.17	0.62	C30 - H33	1.85	-22.94	0.01	-1.89	0.93
C4 - H5	1.92	-25.09	0.01	-2.02	0.93	C34 - N56	1.83	-18.48	0.03	-2.48	0.89
C4 - C6	2.00	-17.76	0.25	-1.94	1.30	C3 - N56	1.62	-13.92	0.06	-1.77	0.74
C6 - H7	1.91	-24.88	0.02	-2.01	0.94	C34 - C35	2.07	-19.61	0.24	-2.05	1.29
C6 - Fe59	0.47	5.08	4.35	-0.13	0.33	C35 - C36	2.09	-20.00	0.21	-2.08	1.35
C4 - Fe59	0.47	4.57	1.29	-0.14	0.30	C36 - H37	1.88	-23.59	0.02	-1.94	0.95
C6 - C8	1.94	-16.88	0.21	-1.83	1.22	C36 - C38	2.09	-20.10	0.20	-2.09	1.36
C8 - Fe59	0.47	4.79	2.57	-0.14	0.32	C34 - C62	0.11	1.15	0.60	0.01	0.05
C8 - H9	1.90	-24.57	0.02	-1.99	0.96	C38 - C39	2.08	-19.90	0.20	-2.06	1.35
C12 - Fe59	0.45	5.22	12.40	-0.12	0.33	C39 - C41	2.09	-20.04	0.21	-2.09	1.37
C4 - C12	1.98	-17.45	0.23	-1.90	1.27	C39 - H40	1.88	-23.54	0.02	-1.94	0.95
C8 - C10	2.01	-18.02	0.25	-1.97	1.32	H7 - C50	0.05	0.69	1.44	0.01	0.02
C10 - C12	1.97	-17.31	0.22	-1.87	1.25	H5 - H52	0.05	0.71	0.50	0.01	0.02
C10 - Fe59	0.46	4.54	1.44	-0.14	0.30	C34 - C41	2.08	-19.87	0.22	-2.06	1.28
C10 - H11	1.91	-24.67	0.01	-1.99	0.96	H25 - H45	0.05	0.58	0.07	0.01	0.02
C12 - H13	1.92	-25.37	0.02	-2.04	0.93	C35 - C42	1.70	-14.15	0.03	-1.39	1.01
C14 - N54	1.81	-18.19	0.03	-2.45	0.89	C42 - H43	1.86	-22.95	0.01	-1.90	0.94
C2 - N54	1.63	-14.00	0.06	-1.78	0.74	C42 - H45	1.81	-21.33	0.01	-1.82	0.93
H13 - C21	0.07	0.86	0.12	0.01	0.03	C42 - H44	1.86	-22.90	0.01	-1.91	0.95
C14 - C15	2.09	-19.95	0.24	-2.08	1.30	C38 - C46	1.70	-14.30	0.03	-1.40	1.01
C15 - C16	2.08	-19.83	0.21	-2.07	1.35	C46 - H47	1.83	-22.08	0.01	-1.86	0.95
C16 - H17	1.88	-23.57	0.02	-1.94	0.95	C46 - H48	1.86	-22.66	0.01	-1.90	0.96
C16 - C18	2.09	-20.07	0.21	-2.09	1.37	C46 - H49	1.85	-22.63	0.01	-1.89	0.96
C18 - C19	2.08	-19.88	0.20	-2.06	1.35	H51 - O58	0.07	0.91	0.52	0.01	0.03
C19 - C21	2.09	-19.97	0.21	-2.08	1.36	C41 - C50	1.70	-14.29	0.03	-1.41	1.01
C19 - H20	1.88	-23.49	0.02	-1.94	0.95	C50 - H51	1.86	-23.20	0.01	-1.91	0.92
C14 - C21	0.06	0.77	0.25	0.01	0.02	C50 - H52	1.85	-22.50	0.01	-1.89	0.93
C15 - C22	2.08	-19.93	0.23	-2.07	1.28	C50 - H53	1.85	-22.36	0.01	-1.88	0.96
C22 - H25	1.70	-14.16	0.03	-1.40	1.00	C2 - O57	2.74	4.19	0.12	-4.66	1.13
C22 - H23	1.84	-22.11	0.01	-1.87	0.93	C3 - O58	2.75	4.40	0.13	-4.66	1.14
C22 - H24	1.86	-22.75	0.01	-1.90	0.95	C15 - C63	0.07	0.93	2.73	-0.08	0.02
C18 - C26	1.85	-22.70	0.01	-1.88	0.93	Fe59 - C62	0.75	11.83	0.44	-2.54	0.74
C26 - H27	1.70	-14.26	0.03	-1.39	1.01	Fe59 - C63	0.76	13.31	0.30	-2.70	0.78
C26 - H28	1.85	-22.60	0.01	-1.89	0.96	O60 - C62	3.18	33.13	0.00	-25.60	1.55
C26 - H29	1.86	-22.67	0.01	-1.90	0.96	O61 - C63	3.18	32.91	0.00	-25.61	1.56
H33 - O57	1.83	-22.07	0.01	-1.86	0.95						

Theoretical values from isolated molecule DFT calculation.

$\rho(r_b)$ is the ED (eÅ⁻³); $\nabla^2\rho(r_b)$ is the Laplacian of the ED (eÅ⁻⁵); ϵ is the ellipticity of the bond; $H\rho(r_b)$ is the total electron energy density. $\delta(A-B)$ is the delocalization index.

Table S8. AIM analysis for complex $\mathbf{6}^{\text{Me}}(\text{OTf})$.

Molecular graph including bcp's (red dots), rcp's (yellow dots), and bp's (black lines).



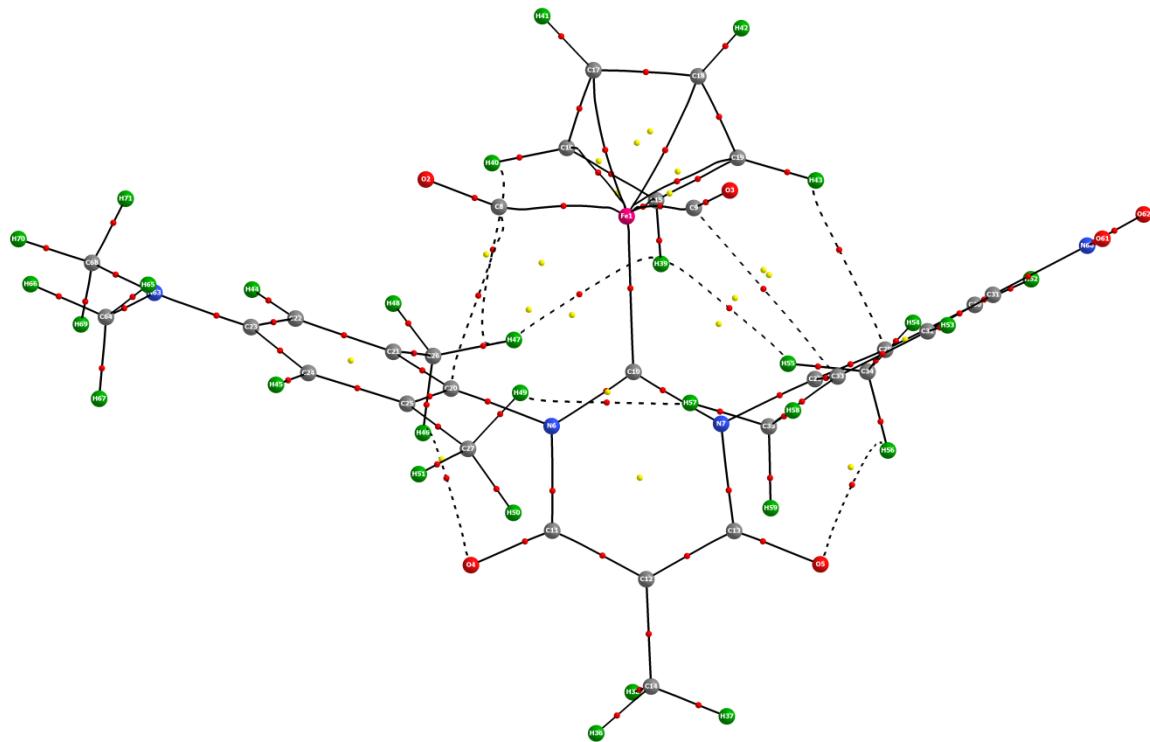
bond	$\rho(r_b)$ (e \AA^{-3})	$\nabla^2\rho(r_b)$ (e \AA^{-5})	ϵ	H	$\delta(A/B)$	bond	$\rho(r_b)$ (e \AA^{-3})	$\nabla^2\rho(r_b)$ (e \AA^{-5})	ϵ	H	$\delta(A/B)$
C3 - Fe67	0.56	7.13	0.19	-0.14	0.58	C36 - H37	1.85	-22.56	0.01	-1.88	0.94
C4 - N60	1.90	-20.73	0.10	-2.41	0.82	C36 - H39	1.81	-21.47	0.01	-1.82	0.93
C5 - N62	1.91	-21.00	0.09	-2.42	0.82	C27 - C36	1.70	-14.19	0.02	-1.40	1.01
C4 - N61	2.14	-24.59	0.13	-3.27	0.89	C36 - H37	1.85	-22.56	0.01	-1.88	0.94
C6 - N61	1.61	-12.15	0.03	-2.19	0.86	C36 - H38	1.86	-23.00	0.01	-1.91	0.95
C5 - N61	2.13	-24.63	0.13	-3.23	0.89	H39 - H59	0.06	0.70	0.10	0.01	0.02
C6 - H7	1.90	-24.07	0.04	-1.96	0.93	C2 - C40	0.07	0.84	6.20	0.01	0.03
C6 - H8	1.93	-25.42	0.03	-2.04	0.91	C3 - N62	2.14	-18.11	0.06	-3.47	1.06
C6 - H9	1.92	-25.82	0.03	-2.04	0.89	H19 - C41	0.06	0.82	1.65	0.01	0.02
C14 - Fe67	0.47	4.70	2.39	-0.14	0.32	C40 - C41	2.07	-19.79	0.22	-2.05	1.29
C12 - Fe67	0.47	5.03	3.79	-0.13	0.34	C42 - H43	1.88	-23.65	0.02	-1.94	0.95
C16 - Fe67	0.47	4.55	1.66	-0.14	0.31	C41 - C42	2.08	-19.87	0.20	-2.07	1.36
C18 - Fe67	0.46	5.16	6.73	-0.13	0.35	C42 - C44	2.08	-19.94	0.20	-2.07	1.36
C10 - Fe67	0.47	4.44	1.73	-0.14	0.30	C40 - C47	2.08	-19.64	0.25	-2.06	1.30
N62 - O71	0.09	1.19	2.15	0.01	0.05	C44 - C52	1.70	-14.26	0.03	-1.39	1.01
H51 - O71	0.12	1.38	0.10	0.00	0.06	C44 - C45	2.09	-20.06	0.20	-2.08	1.36
C10 - H11	1.93	-26.82	0.01	-2.09	0.88	C45 - H46	1.88	-23.61	0.02	-1.95	0.95
C10 - C12	1.99	-17.70	0.23	-1.93	1.29	C45 - C47	2.08	-19.85	0.21	-2.07	1.35
C12 - H13	1.91	-24.99	0.02	-2.01	0.94	C40 - N62	1.73	-15.92	0.02	-2.33	0.86
C12 - C14	1.94	-16.87	0.21	-1.82	1.22	H49 - O66	0.06	0.81	0.74	0.01	0.02
C14 - H15	1.90	-24.54	0.01	-1.99	0.96	C41 - C48	1.70	-14.29	0.03	-1.42	1.01
C10 - C18	1.98	-17.48	0.22	-1.90	1.27	C48 - H49	1.84	-22.58	0.01	-1.87	0.92
C14 - C16	2.02	-18.07	0.25	-1.97	1.32	C48 - H51	1.89	-24.09	0.01	-1.97	0.90
C16 - C18	1.96	-17.20	0.21	-1.86	1.24	C48 - H50	1.84	-22.29	0.01	-1.88	0.95
C16 - H17	1.90	-24.58	0.01	-1.99	0.96	H11 - O71	0.13	1.45	0.05	0.00	0.06
C18 - H19	1.92	-25.25	0.02	-2.03	0.94	C52 - H53	1.86	-22.66	0.01	-1.90	0.96
C1 - C20	0.10	1.07	0.62	0.01	0.05	C52 - H54	1.85	-22.50	0.01	-1.88	0.96
C3 - N60	2.15	-17.70	0.06	-3.50	1.07	C52 - H55	1.84	-22.29	0.01	-1.87	0.95
C20 - C21	2.08	-19.88	0.22	-2.06	1.29	C56 - H59	1.83	-21.97	0.01	-1.86	0.92
H13 - H30	0.05	0.79	1.36	0.01	0.01	C47 - C56	1.69	-14.13	0.03	-1.39	1.00
C22 - H23	1.89	-23.78	0.02	-1.95	0.95	C56 - H57	1.86	-22.96	0.01	-1.91	0.95
C21 - C22	2.08	-19.87	0.20	-2.06	1.36	C56 - H58	1.84	-22.31	0.01	-1.86	0.94
C22 - C24	2.08	-19.95	0.20	-2.07	1.36	O66 - F75	0.02	0.49	0.06	0.01	0.01
C20 - C27	2.07	-19.58	0.25	-2.05	1.29	C5 - O71	0.09	1.26	9.89	0.01	0.02
C24 - C32	1.70	-14.30	0.02	-1.40	1.01	C1 - O63	3.18	33.30	0.00	-5.24	1.56
C24 - C25	2.08	-20.03	0.20	-2.08	1.35	C2 - O64	3.18	33.18	0.00	-5.25	1.56
C25 - H26	1.88	-23.59	0.02	-1.94	0.95	C4 - O65	2.81	6.65	0.13	-4.80	1.17
C25 - C27	2.09	-19.95	0.21	-2.08	1.36	C5 - O66	2.80	6.03	0.12	-4.78	1.16
C20 - N60	1.74	-15.91	0.02	-2.35	0.86	C1 - Fe67	0.74	11.61	0.44	-0.21	0.73
H11 - H30	0.05	0.76	1.18	0.01	0.01	C2 - Fe67	0.74	12.49	0.37	-0.20	0.75
C21 - C28	1.70	-14.35	0.03	-1.43	1.01	C6 - F75	0.03	0.49	1.38	0.01	0.01
N60 - O71	0.07	0.89	1.13	0.00	0.04	H49 - F74	0.02	0.36	1.11	0.01	0.01
H30 - O71	0.08	0.94	0.11	0.00	0.04	H9 - O70	0.10	1.17	0.07	0.00	0.05
C28 - H29	1.84	-22.94	0.01	-1.88	0.90	H29 - O70	0.06	0.73	0.24	0.01	0.03
C28 - H30	1.87	-23.38	0.02	-1.94	0.90	O71 - S72	1.82	23.90	0.06	-2.03	0.89
C28 - H31	1.85	-22.52	0.01	-1.89	0.95	C68 - S72	1.35	-10.85	0.00	-1.04	0.65
C32 - H34	1.86	-22.70	0.01	-1.90	0.96	C68 - F73	1.88	-1.18	0.19	-2.78	0.67
C32 - H33	1.85	-22.63	0.01	-1.89	0.96	O69 - S72	1.91	28.77	0.06	-2.16	1.00
C32 - H35	1.83	-22.11	0.01	-1.86	0.95	O70 - S72	1.89	27.29	0.06	-2.12	0.97
C36 - H39	1.81	-21.47	0.01	-1.82	0.93	C68 - F74	1.85	-1.49	0.20	-2.72	0.66
C27 - C36	1.70	-14.19	0.02	-1.40	1.01	C68 - F75	1.83	0.00	0.21	2.58	0.66

Theoretical values from isolated molecule DFT calculation.

$\rho(r_b)$ is the ED (e. \AA^{-3}); $\nabla^2\rho(r_b)$ is the Laplacian of the ED (e. \AA^{-5}); ϵ is the ellipticity of the bond; $H\rho(r_b)$ is the total electron energy density. $\delta(A-B)$ is the delocalization index.

Table S9. AIM analysis for complex 7.

Molecular graph including bcp's (red dots), rcp's (yellow dots), and bp's (black lines).



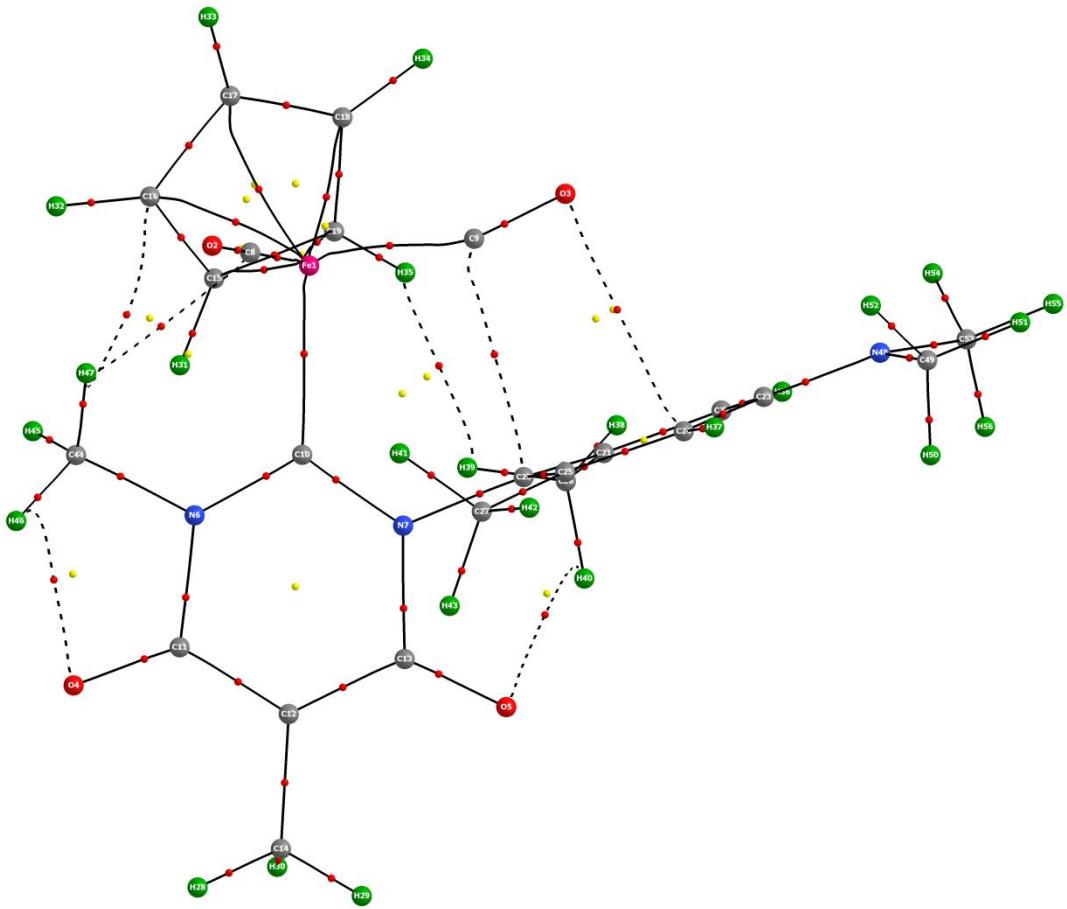
bond	$\rho(\mathbf{r}_b)$ (e \AA^{-3})	$\nabla^2\rho(\mathbf{r}_b)$ (e \AA^{-5})	ϵ	H	$\delta(A/B)$	bond	$\rho(\mathbf{r}_b)$ (e \AA^{-3})	$\nabla^2\rho(\mathbf{r}_b)$ (e \AA^{-5})	ϵ	H	$\delta(A/B)$
Fe1 - C10	0.6	6.9	0.1	-0.2	0.6	C30 - C31	2.1	-20.8	0.2	-2.2	1.3
Fe1 - C8	0.7	11.6	0.4	-0.2	0.7	C28 - C33	2.1	-19.9	0.2	-2.1	1.3
Fe1 - C9	0.8	13.6	0.3	-0.2	0.8	C31 - C32	2.1	-21.0	0.2	-2.2	1.3
C8 - C20	0.1	1.2	0.4	0.0	0.1	C32 - C33	2.1	-19.9	0.2	-2.1	1.3
N6 - C10	2.2	-17.6	0.0	-3.6	1.1	O5 - H56	0.1	0.9	0.7	0.0	0.0
N6 - C11	1.7	-15.1	0.1	-2.0	0.8	H39 - H55	0.1	0.7	0.2	0.0	0.0
O4 - C11	2.7	2.8	0.1	-4.5	1.2	C29 - C34	1.7	-14.3	0.0	-1.4	1.0
N7 - C10	2.1	-16.9	0.0	-3.5	1.1	H49 - H57	0.0	0.5	0.1	0.0	0.0
C12 - C13	2.1	-20.1	0.3	-2.1	1.2	C33 - C35	1.7	-14.2	0.0	-1.4	1.0
C9 - C33	0.1	0.9	2.4	0.0	0.0	C14 - H36	1.8	-22.3	0.0	-1.9	1.0
C11 - C12	2.1	-20.0	0.3	-2.1	1.2	C14 - H38	1.8	-22.1	0.0	-1.9	1.0
N7 - C13	1.7	-14.9	0.0	-2.0	0.8	C15 - H39	1.9	-25.1	0.0	-2.0	0.9
O5 - C13	2.7	1.6	0.1	-4.5	1.2	C16 - H40	1.9	-24.9	0.0	-2.0	0.9
O2 - C8	3.2	33.2	0.0	-5.2	1.6	C17 - H41	1.9	-24.6	0.0	-2.0	1.0
O3 - C9	3.2	32.7	0.0	-5.2	1.6	C18 - H42	1.9	-24.7	0.0	-2.0	1.0
C12 - C14	1.7	-13.8	0.0	-1.4	1.0	C19 - H43	1.9	-25.3	0.0	-2.0	0.9
C14 - H37	1.9	-23.3	0.0	-1.9	0.9	C22 - H44	1.9	-23.8	0.0	-2.0	0.9
Fe1 - C15	0.5	4.6	1.3	-0.1	0.3	C26 - H46	1.9	-23.1	0.0	-1.9	0.9
Fe1 - C16	0.5	5.0	4.4	-0.1	0.3	C26 - H47	1.9	-22.5	0.0	-1.9	0.9
Fe1 - C18	0.5	4.5	1.4	-0.1	0.3	C26 - H48	1.8	-22.4	0.0	-1.9	1.0
Fe1 - C19	0.4	5.2	13.5	-0.1	0.3	C27 - H49	1.8	-21.2	0.0	-1.8	0.9
C16 - C17	1.9	-16.9	0.2	-1.8	1.2	C27 - H50	1.9	-23.0	0.0	-1.9	0.9
C15 - C16	2.0	-17.8	0.2	-1.9	1.3	C27 - H51	1.9	-22.9	0.0	-1.9	1.0
Fe1 - C17	0.5	4.8	2.6	-0.1	0.3	C30 - H52	1.9	-25.1	0.0	-2.0	0.9
C17 - C18	2.0	-18.1	0.2	-2.0	1.3	C32 - H53	1.9	-25.2	0.0	-2.0	0.9
C18 - C19	2.0	-17.4	0.2	-1.9	1.3	C34 - H54	1.9	-22.6	0.0	-1.9	1.0
C15 - C19	2.0	-17.4	0.2	-1.9	1.3	C34 - H55	1.9	-22.6	0.0	-1.9	0.9
N6 - C20	1.8	-17.7	0.0	-2.5	0.9	C34 - H56	1.9	-23.1	0.0	-1.9	0.9
H40 - H47	0.0	0.7	1.8	0.0	0.0	C35 - H57	1.8	-22.4	0.0	-1.9	0.9
C20 - C21	2.1	-19.9	0.2	-2.1	1.3	C35 - H58	1.9	-22.9	0.0	-1.9	1.0
C21 - C22	2.1	-20.0	0.2	-2.1	1.4	C35 - H59	1.8	-22.7	0.0	-1.9	0.9
C22 - C23	2.0	-19.1	0.2	-2.0	1.3	C31 - N60	1.7	-17.3	0.1	-2.2	0.8
C20 - C25	2.1	-19.6	0.2	-2.0	1.3	N60 - O61	3.4	-25.1	0.1	-4.4	1.7
C23 - C24	2.0	-19.2	0.2	-2.0	1.3	N60 - O62	3.4	-25.1	0.1	-4.4	1.7
C24 - C25	2.1	-20.0	0.2	-2.1	1.4	N63 - C68	1.8	-18.3	0.1	-2.3	0.9
O4 - H46	0.1	0.9	0.5	0.0	0.0	C23 - N63	2.1	-21.1	0.1	-3.1	1.0
H39 - H47	0.1	0.7	0.5	0.0	0.0	N63 - C64	1.8	-18.3	0.1	-2.3	0.9
C21 - C26	1.7	-14.3	0.0	-1.4	1.0	C64 - H67	1.9	-22.9	0.0	-1.9	0.9
C25 - C27	1.7	-14.2	0.0	-1.4	1.0	C64 - H66	1.9	-23.9	0.0	-2.0	0.9
C24 - H45	1.9	-23.8	0.0	-2.0	0.9	C64 - H65	1.9	-23.4	0.0	-1.9	0.9
N7 - C28	1.8	-19.6	0.1	-2.4	0.9	C68 - H69	1.9	-22.9	0.0	-1.9	0.9
C29 - H43	0.1	0.9	0.1	0.0	0.0	C68 - H70	1.9	-23.9	0.0	-2.0	0.9
C28 - C29	2.1	-20.0	0.2	-2.1	1.3	C68 - H71	1.9	-23.3	0.0	-1.9	0.9
C29 - C30	2.1	-20.0	0.2	-2.1	1.4						

Theoretical values from isolated molecule DFT calculation.

$\rho(\mathbf{r}_b)$ is the ED (e \AA^{-3}); $\nabla^2\rho(\mathbf{r}_b)$ is the Laplacian of the ED (e \AA^{-5}); ϵ is the ellipticity of the bond; $H\rho(\mathbf{r}_b)$ is the total electron energy density. $\delta(A-B)$ is the delocalization index.

Table S10. AIM analysis for complex **8**.

Molecular graph including bcp's (red dots), rcp's (yellow dots), and bp's (black lines)



bond	$\rho(r_b)$ (e \AA^{-3})	$\nabla^2\rho(r_b)$ (e \AA^{-5})	ϵ	H	$\delta(A/B)$
Fe1 - C8	0.79	15.50	0.04	-0.20	0.82
Fe1 - C9	0.74	11.46	0.40	-0.21	0.72
Fe1 - C10	0.62	6.84	0.12	-0.18	0.62
N7 - C10	2.17	-17.62	0.02	-3.53	1.10
N6 - C11	1.68	-15.62	0.05	-2.03	0.80
N7 - C13	1.66	-15.15	0.05	-1.98	0.79
C12 - C14	1.69	-13.78	0.05	-1.39	1.01
C9 - C20	0.14	1.40	0.16	0.01	0.08
C11 - C12	2.07	-20.03	0.28	-2.10	1.21
N7 - C20	1.83	-18.18	0.02	-2.49	0.89
O5 - C13	2.66	2.08	0.08	-4.48	1.17
O2 - C8	3.17	32.28	0.00	-5.23	1.55
C8 - H47	0.09	1.10	0.45	0.01	0.03
O3 - C9	3.17	32.88	0.00	-5.22	1.55
N6 - C10	2.18	-16.82	0.03	-3.58	1.11
C12 - C13	2.07	-20.08	0.28	-2.11	1.21
N6 - C44	1.73	-16.84	0.04	-2.22	0.91
O4 - C11	2.65	2.12	0.08	-4.46	1.17
C14 - H29	1.86	-22.88	0.01	-1.91	0.95
C14 - H28	1.86	-22.88	0.01	-1.91	0.95
Fe1 - C15	0.47	4.61	1.28	-0.14	0.30
Fe1 - C16	0.47	5.06	3.19	-0.13	0.33
Fe1 - C19	0.45	5.44	12.54	-0.12	0.33
C16 - H47	0.06	0.74	0.68	0.01	0.02
Fe1 - C18	0.46	4.52	1.35	-0.14	0.30
C15 - C16	2.00	-17.65	0.25	-1.94	1.30
Fe1 - C17	0.46	4.84	3.83	-0.13	0.31
C16 - C17	1.94	-16.91	0.21	-1.83	1.22
C17 - C18	2.02	-18.08	0.25	-1.98	1.32
C15 - C19	1.98	-17.49	0.24	-1.90	1.27
C18 - C19	1.96	-17.16	0.22	-1.86	1.24
O3 - C24	0.06	0.75	3.65	0.01	0.03
O5 - H40	0.06	0.85	0.86	0.01	0.03
H35 - H39	0.05	0.64	0.18	0.01	0.02

bond	$\rho(r_b)$ (e \AA^{-3})	$\nabla^2\rho(r_b)$ (e \AA^{-5})	ϵ	H	$\delta(A/B)$
C20 - C21	2.08	-19.86	0.23	-2.06	1.26
C21 - C22	2.09	-20.01	0.23	-2.10	1.38
C22 - C23	2.03	-19.15	0.20	-1.98	1.26
C20 - C25	2.07	-19.60	0.24	-2.04	1.26
C23 - C24	2.03	-19.13	0.20	-1.97	1.25
C24 - C25	2.10	-20.05	0.23	-2.11	1.38
C21 - C26	1.70	-14.30	0.03	-1.41	1.01
C25 - C27	1.70	-14.19	0.03	-1.39	1.01
C14 - H30	1.83	-21.79	0.01	-1.84	0.95
C15 - H31	1.91	-25.12	0.01	-2.01	0.94
C16 - H32	1.91	-24.68	0.02	-2.00	0.95
C17 - H33	1.90	-24.58	0.02	-1.99	0.96
C18 - H34	1.91	-24.69	0.01	-2.00	0.96
C19 - H35	1.91	-24.89	0.02	-2.01	0.95
C22 - H36	1.89	-23.76	0.03	-1.97	0.94
C53 - H56	1.86	-22.82	0.03	-1.88	0.93
C49 - H50	1.86	-22.82	0.03	-1.88	0.93
C24 - H37	1.90	-23.79	0.03	-1.98	0.94
C26 - H40	1.85	-23.03	0.01	-1.90	0.93
C26 - H38	1.84	-22.31	0.01	-1.88	0.96
C26 - H39	1.85	-22.50	0.01	-1.89	0.93
C27 - H42	1.86	-22.81	0.01	-1.91	0.96
C27 - H41	1.82	-21.70	0.01	-1.84	0.94
C27 - H43	1.85	-22.84	0.01	-1.89	0.94
O4 - H46	0.16	2.58	1.23	0.02	0.06
C44 - H46	1.94	-25.80	0.02	-2.06	0.90
C44 - H45	1.87	-23.23	0.03	-1.91	0.92
C44 - H47	1.91	-24.04	0.03	-1.99	0.91
C23 - N48	2.06	-21.13	0.08	-3.12	1.04
N48 - C49	1.79	-18.34	0.05	-2.26	0.94
N48 - C53	1.79	-18.34	0.05	-2.26	0.94
C49 - H51	1.89	-23.90	0.03	-1.96	0.93
C49 - H52	1.88	-23.35	0.03	-1.92	0.92

Theoretical values from isolated molecule DFT calculation.

$\rho(r_b)$ is the ED (e \AA^{-3}); $\nabla^2\rho(r_b)$ is the Laplacian of the ED (e \AA^{-5}); ϵ is the ellipticity of the bond; $H\rho(r_b)$ is the total electron energy density. $\delta(A-B)$ is the delocalization index.

Analysis of the influence of the electronic properties of the carbene and of the N-substituents on the interligand interaction

Both experimental and optimized geometries of complexes **4** and **6^{Me}** also exhibit a somewhat unsymmetrical geometry, one of the C_{ipso}(aryl)...C(=O) distances being slightly shorter than the other one (Table S5). As a matter of fact, the variation of σ-donor effects along the series of free carbenes **1_{Me}⁻.K⁺**, **2⁻.K⁺** and **2_{Me}** (n_C^σ : -6.40 eV for **1_{Me}⁻.K⁺**, -6.60 eV for **2⁻.K⁺** and -7.3 eV for **2_{Me}**), coherent with the IR observations (donor effect : **1_{Me}⁻ > 2⁻**), is not substantiated by important changes in the geometry of these complexes. As previously observed for **3^{Me}**, the AIM analysis of the theoretical ED within these complexes in their optimized geometry also reveals the recurrence of bcp's between the C_{ipso} and C_{ortho} of each aryl ring, and their proximal CO ligand (Tables S7 and S8). Topological characteristics ($\rho(\mathbf{r}_b)$, $\nabla^2\rho(\mathbf{r}_b)$) and derived values (including E_{int}) as well as delocalization indices at these pairs of bcp's appear to be very similar to those appearing in the molecular graph of **3_{Me}** and do not reveal any clear trend between the electron donating properties of the *malo-* or *imid*NHC rings and the magnitude of the weak non-covalent interactions (Tables S7 and S8). In the same way, the calculated energies E_{int} and the NBO stabilizing energies ΔE(2) are only weakly modified and do not exceed respectively 3 and 5 kcal/mol. The strongest interaction, even if it remains weak, corresponds to the shortest C_{Mes}...C_{C≡O} distance. As previously observed for **3_{Me}**, the NLMO associated with the π_{C=C^{Mes}} orbitals involved in the interaction with the π*_{C≡O} orbitals presents a weak CO participation. The latter is quasi similar for the three complexes **3_{Me}**, **4** and **6^{Me}(OTf)**, the highest participation corresponding to the shortest C_{Mes}...C_{CO} contact (Table S11). By contrast, the electron richness of the aryl ring appears to be more influential on these interligand interactions. In order to bring support to this hypothesis, DFT calculations were conducted on the unsymmetrical model complex **7** derived from **3_{Me}** by formal substitution of the 4-Me groups of the two Mesityl substituents by an NMe₂ group and an NO₂ group respectively. These changes do not affect the steric hindrance of the aryl groups, remaining *a priori* constant. Very characteristically, the DFT geometry optimization carried out on **7** leads to a slightly more unsymmetrical structure than that previously observed for **3_{Me}**. It is characterized by a shortening of the C_{ipso}(aryl)...C(=O) distance involving the aryl

group bearing the 4-NMe₂ donor substituent (C1...C21 = 2.755 Å, Table S5), and an increase in the C_{ortho}(aryl)...C(≡O) distance involving on the other side the aryl group bearing the 4-NO₂ donor substituent (C2...C36: 2.935 Å). Beyond distance variations, the AIM and NBO analyses conducted as above (Table S12 and S13) indicate a strengthening of the C1...C21 interaction, concomitant with a weakening of the C2...C26 one, corroborating our previous demonstration that the magnitude of such interactions is sensitive to the electron-richness of the aryl ring,¹³ favoring the π(C_{ipso}=C_{ortho}) to π*(C≡O) donation. For the shortest contact, the energy value E_{int} becomes higher than 3 kcal.mol⁻¹ and the ΔE(2) NBO stabilizing energy becomes higher than 5 kcal.mol⁻¹.

It is noteworthy that the magnitude of such an interaction is even more strengthened in the case of a totally unsymmetrical structure such as **8**, in which the nitrogen substituents are a para-NMe₂-2,6-dimethylphenyl substituent on one side and a methyl group on the other side. Here, the stabilizing interaction on the aryl group side is maximised up to 6.9 kcal.mol⁻¹, leading to a significant shortening of the C_{Mes}...C_{CO} bond (2.663 Å) and an even more bent carbonyl group, with a Fe-C-O bond angle of 163.15°. In the same way, the density ρ(r_b) is 0.14 Å.bohr⁻³ with a delocalization index δ(C_{Mes},C_{CO}) around 0.08 and the energy E_{int} is the highest calculated (3.8 kcal.mol⁻¹). All these results are corroborated by the NLMO analysis, associated with the π_{C=C} orbital of the aryl group since the small participation of the π*_{C=O} orbital slightly increases from **3_{Me}** to **8** (Table S13 : %C : from 0.5 to 1.4; %O : from 0.2 to 0.5).

Analysis of the bonding situation of the Fe-C_{carbene} bond in **3_{Me}, **4** and **6^{Me}(OTf)****

An analysis of the bonding situation of the Fe-C_{carbene} bond reveals the presence of a strong donor-acceptor interaction between the carbene unit and the FeCp(CO)₂ fragment. The sum of the stabilizing interaction [ΔE(2)] at the second order perturbation theory between the carbene lone pair (hybridization : ~40%*s*, ~60%*p*) and an orbital involving the mixing of the *s* and d_{z2} orbitals of iron is strong around 140-150 kcal.mol⁻¹, which is in the range of a covalent bond (Table S11). The corresponding NLMO (Plot see Figure S2) presents a

¹³ Valyaev, D. A.; Brousses, R.; Lugan, N.; Fernández, I.; Sierra, M. A. *Chem. Eur. J.* **2011**, *17*, 6602.

major contribution from the carbene center (~79-80 %) and a "delocalization tail" from the metal center (15-17%).

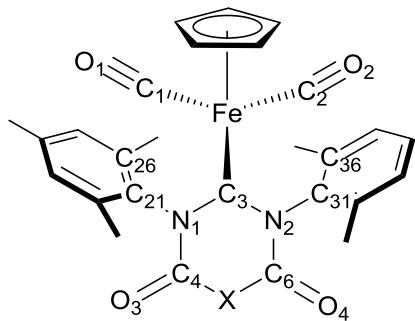
Table S11. NBO analysis of the C→Fe interaction in the three for the complexes **3_{Me}**, **4** and **6^{Me}(OTf)**.

NBO analysis		
	$\Delta E(2)^a$	NLMO ^b
	$n_C^\sigma \rightarrow Fe$	n_C^σ
3_{Me}	$n_{C3}^\sigma \rightarrow s_{(Fe)} : 74.45$	$C_3 : 79.2 \%$
	$n_{C3}^\sigma \rightarrow d_{(Fe)} : 70.97$	$Fe : 17.0 \%$
4	$n_{C3}^\sigma \rightarrow s_{(Fe)} : 74.78$	$C_3 : 79.2 \%$
	$n_{C3}^\sigma \rightarrow d_{(Fe)} : 71.32$	$Fe : 17.2 \%$
[6^{Me}](OTf)	$n_{C3}^\sigma \rightarrow s_{(Fe)} : 77.25$	$C_3 : 81.1 \%$
	$n_{C3}^\sigma \rightarrow d_{(Fe)} : 45.66$	$Fe : 15.2 \%$

^a Stabilizing energy in kcal/mol;

^b Natural Localized Molecular Orbital associated with the n_c^σ orbital.

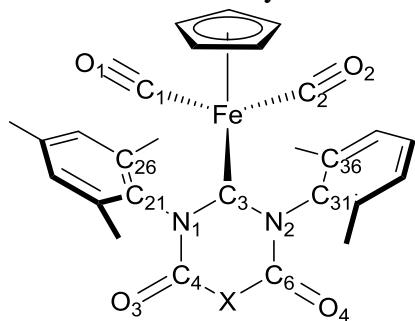
Table S12. Topological analysis of the C(aryl)...C(\equiv O) bond critical points^a



complex	contact	R_b^b	$\rho(\mathbf{r}_b)^c$	$\nabla^2\rho(\mathbf{r}_b)^d$	$G\rho(\mathbf{r}_b)^e$	$V\rho(\mathbf{r}_b)^e$	$\delta(\mathbf{C}...\mathbf{C})^f$	E_{int}^g
3_{Me} ^a	C1...C21	2.7197(4)	0.13	1.15	0.08	-0.08	n/a	3.9
		<i>2.811</i>	<i>0.11</i>	<i>1.16</i>			<i>0.05</i>	2.7
	C2...C36 ^h	3.0518(5)			bcp not observed experimentally			
4	C2...C36 ^h	2.908	0.08	0.95	0.05	-0.04	0.02	1.9
	C1...C21	<i>2.816</i>	<i>0.11</i>	<i>1.15</i>	<i>0.07</i>	<i>-0.06</i>	<i>0.05</i>	2.7
[6^{Me}]OTf	C2...C31 ^h	3.119	0.07	0.93	0.04	-0.04	0.02	2.0
	C1...C21	2.856	0.10	1.07	0.06	-0.05	0.05	2.4
7	C2...C31 ^h	3.070	0.07	0.84	0.05	-0.04	0.03	1.8
	C1...C21	2.755	0.12	1.25	0.08	-0.07	0.06	3.1
	C2...C36 ^h	2.935	0.07	0.91	0.05	-0.04	0.02	1.9
8	C2...C31	2.663	0.014	1.40	0.09	-0.08	0.08	3.8

^a For **3_{Me}**: top line experimental values, second line (italic) theoretical values from isolated molecule DFT calculation; for the other complexes: theoretical values from isolated molecule DFT calculation; ^b R_b is the interatomic distance (\AA); ^c $\rho(\mathbf{r}_b)$ is the ED ($e.\text{\AA}^{-3}$); ^d $\nabla^2\rho(\mathbf{r}_b)$ is the Laplacian of the ED ($e.\text{\AA}^{-5}$); ^e $G\rho(\mathbf{r}_b)$, and $V\rho(\mathbf{r}_b)$ are the kinetic and potential electron energy densities, respectively, estimated using the approximation of Abramov^{ref} (hartree. \AA^{-3}); ^f $\delta(\mathbf{C}...\mathbf{C})$ are the delocalization indices; ^g E_{int} is the interaction energy estimated using Espinoza correlation scheme (kcal.mol⁻¹). ^h for the weak interaction C(aryl)...C(\equiv O) either C_{ipso} (labeled C₂₁) or C_{ortho} (labeled C₃₆) carbon atom of the aryl group is involved.

Table S13. NBO analysis of the C(aryl)...C(≡O) non-covalent weak interactions for compounds **3_{Me}**, **4**, **6_{Me}(OTf)**, **7** and **8** at the M06-2X/6-31G** level of theory.



NBO Analysis				
	$\pi_{\text{C}=\text{C}} \rightarrow \pi^*_{\text{C}=\text{O}}$ interaction involved in the major $\text{C}_{\text{Mes}} \dots \text{C}_{\text{CO}}$ contact		$\pi_{\text{C}=\text{C}} \rightarrow \pi^*_{\text{C}=\text{O}}$ interaction involved in the minor $\text{C}_{\text{Mes}} \dots \text{C}_{\text{CO}}$ contact	
	$\Delta E(2)^{\text{a}}$	NLMO ^b	$\Delta E(2)^{\text{a}}$	NLMO ^b
	$\pi_{\text{C}=\text{C}} \rightarrow \pi^*_{\text{C}=\text{O}}$	$\pi_{\text{C}=\text{C}}$	$\pi_{\text{C}=\text{C}} \rightarrow \pi^*_{\text{C}=\text{O}}$	$\pi_{\text{C}=\text{C}}$
3_{Me}	4.4	C ₂₁ : 43.8 % C ₂₆ : 38.8 % C ₁ : 0.6% O ₁ : 0.2 %	3.4	C ₃₁ : 43.5 % C ₃₆ : 39.1 % C ₂ : 0.3% O ₂ : 0.1 %
4	4.9	C ₂₁ : 43.8 % C ₂₆ : 38.9 % C ₁ : 0.6% O ₁ : 0.2 %	2.6	C ₃₁ : 43.9 % C ₃₆ : 39.0 % C ₂ : 0.3% O ₂ : 0.1 %
6_{Me}(OTf)	3.7	C ₂₁ : 45.6 % C ₂₆ : 38.1 % C ₁ : 0.5% O ₁ : 0.2 %	3.4	C ₃₁ : 45.2 % C ₃₆ : 38.8 % C ₁ : 0.3% O ₁ : 0.1 %
7	5.2	C ₂₁ : 48.8 % C ₂₆ : 33.7 % C ₁ : 1.0% O ₁ : 0.4 %	2.8	C ₃₁ : 42.8 % C ₃₆ : 38.0 % C ₂ : 0.3% O ₂ : 0.1 %
8	6.9	C ₂₁ : 48.8 % C ₂₆ : 32.4 % C ₁ : 1.4% O ₁ : 0.5 %	/	/

^a Stabilizing energy in kcal.mol⁻¹; ^b Natural Localized Molecular Orbital associated with the $\pi_{\text{C}=\text{C}}^{\text{aryl}}$ orbital involved in the weak interaction $\pi_{\text{C}=\text{C}}^{\text{aryl}} \rightarrow \pi^*_{\text{C}\equiv\text{O}}$.

Figure S3. Contour plot of the Laplacien $\nabla^2 \rho(\text{BCP})$ with charge accumulation ($\nabla^2 \rho(r) < 0$) printed in blue and charge depletion ($\nabla^2 \rho(r) > 0$) printed in red for the complexes **3_{Me}**, **4** and **6^{Me}(OTf)**.

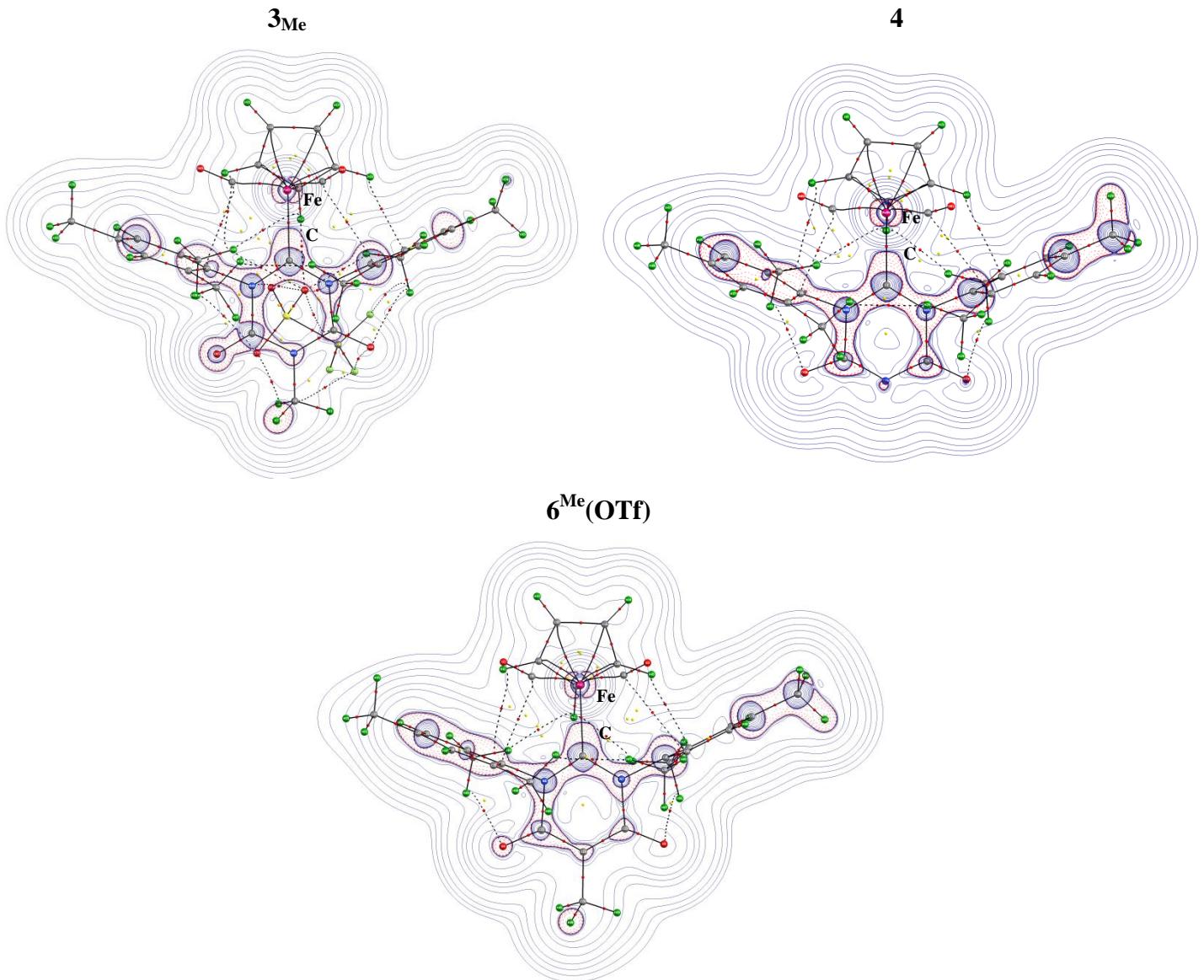
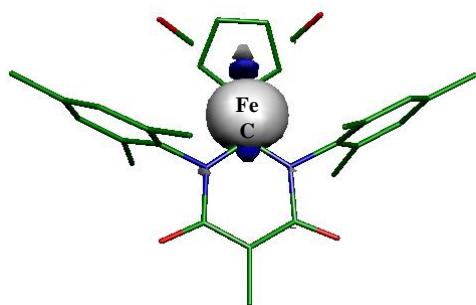


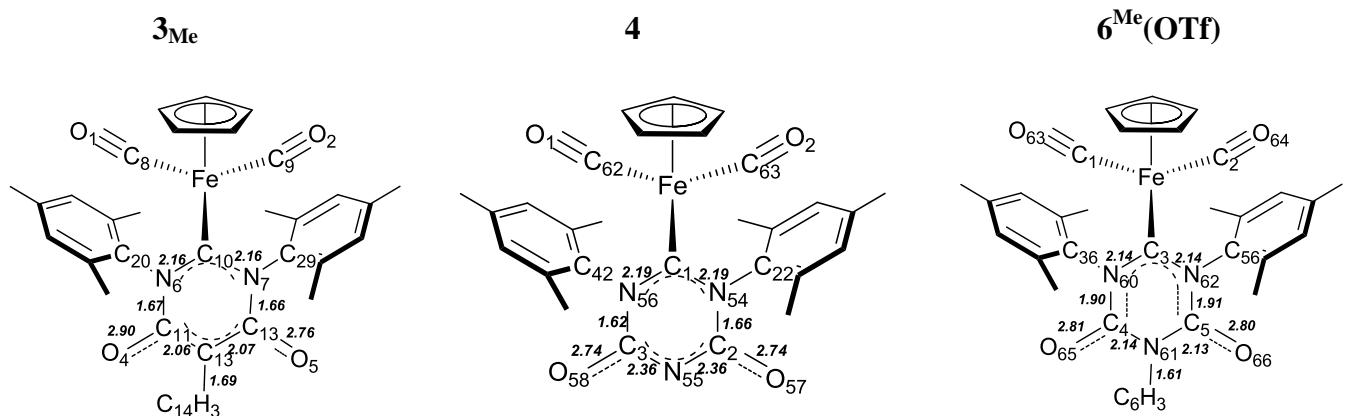
Figure S4. Molekel plot (cutoff : 0.05) for the NLMO associated with the C→Fe interaction in the complex **3_{Me}**.



Electronic delocalization in complexes **3_{Me}**, **4** and **6^{Me}(OTf)**.

In agreement with the experimental description of the geometrical structure given in the main text, the calculated geometrical parameters and the AIM analysis carried out for the three complexes show that compounds **3_{Me}** and **4** can be described as the juxtaposition of two π -systems, namely, NCN NCN ($4\pi e^-$ system) and O=CXC=O ($6\pi e^-$ system) units whereas in **6^{Me}(OTf)** an electronic delocalization occurs between these two units. Indeed, a slight decrease of the central C-N bond lengths (1.46 to 1.48 Å *versus* 1.43 Å) and a slight increase of the density $\rho(r_b)$ from 1.6-1.7 Å·bohr⁻³ in **3_{Me}** and **4** to 1.9 Å·bohr⁻³ in **6^{Me}(OTf)** are theoretically observed (Figure S3).

Figure S5. Electronic delocalization and density $\rho(r_b)$ in $e^- \text{Å}^{-3}$ for the carbene skeleton in the complexes **3_{Me}**, **4** and **6^{Me}(OTf)**.



Z-matrices

3_{Me}	C	C	C	C
Fe	-0.00507100	-1.55535200	-0.28001400	1.56322900
O	2.46071500	-2.49974100	1.14385600	-1.08214100
O	-1.66354500	-2.43539800	2.05669900	-0.00503000
O	2.33530800	3.15164600	-0.48730500	1.21400800
O	-2.34992800	3.18835100	0.04478900	-0.00713300
N	1.12934800	1.22470800	-0.11219300	-1.22421400
N	-1.13519900	1.23502100	0.01662200	0.01291700

C	1.56322900	-1.98431300	0.68245700
C	-1.08214100	-1.98685300	1.19400100
C	-0.00503000	0.50247700	-0.01194700
C	1.21400800	2.68484100	-0.31427400
C	-0.00713300	3.37050400	-0.27439700
C	-1.22421400	2.71036500	-0.05815000
C	0.01291700	4.86047300	-0.44644700
C	-0.20378400	-1.24211600	-2.41857000
C	0.92636400	-2.05054900	-2.16400100
C	0.46336700	-3.25228600	-1.54416800
C	-0.93638900	-3.16364000	-1.41475100

C	-1.35864800	-1.91021500	-1.93964100	C	-0.21190900	-1.30841600	-2.32201700
C	2.42075800	0.63125800	0.11356200	H	-0.20234700	-0.37044200	-2.85921500
C	3.26690600	0.34154700	-0.96252200	C	0.92480400	-2.08670100	-2.01002200
C	4.50798500	-0.22814400	-0.68378800	H	1.94806900	-1.86417500	-2.27910000
C	4.94703800	-0.43170100	0.62585000	C	0.47299500	-3.23505500	-1.28831200
C	4.11751600	-0.03414300	1.67329900	H	1.09750600	-4.03383800	-0.91137100
C	2.85126100	0.49666700	1.43832800	C	-0.92630100	-3.14236200	-1.15462900
C	2.89509700	0.73814900	-2.36374900	H	-1.56203300	-3.85274200	-0.64243500
C	6.28574900	-1.06486400	0.90205400	C	-1.35950300	-1.93798800	-1.77713600
C	1.94827100	0.86888700	2.58428600	H	-2.37877300	-1.58814000	-1.84133900
C	-2.42898400	0.62194400	0.18342100	C	-2.43427100	0.77575800	0.09310300
C	-3.28332900	0.50338400	-0.91553400	C	-2.86968700	0.55912100	1.40247200
C	-4.52210700	-0.10898000	-0.72578400	H	-4.12654000	-0.01434700	1.59764500
C	-4.94156400	-0.54093100	0.53245700	C	-4.46486800	-0.20677400	2.61322400
C	-4.10488000	-0.30994600	1.62368700	C	-4.96453600	-0.31763800	0.52613100
C	-2.85919000	0.29814800	1.47293200	C	-4.53976200	0.01617500	-0.76141700
C	-2.91816000	1.10071800	-2.24581700	H	-5.20944100	-0.14721700	-1.60332000
C	-6.29466000	-1.17690100	0.72397200	C	-3.29135000	0.58929200	-0.99591400
C	-2.02223200	0.66279700	2.66973900	H	-2.03328600	1.01469800	2.56824300
H	0.51022700	5.35551400	0.39466000	H	-2.46737900	0.68491300	3.51418300
H	-1.00922200	5.23441900	-0.51758000	C	-1.98229900	2.10890600	2.56570700
H	0.56750000	5.13779000	-1.34803300	H	-1.00301600	0.65052400	2.51532800
H	-0.18446800	-0.26011800	-2.87060600	C	-6.30486900	-0.96919800	0.74915300
H	1.95309900	-1.81309500	-2.40539800	H	-7.04625300	-0.59849500	0.03715600
H	1.08064700	-4.08327700	-1.23020400	H	-6.67457600	-0.77772000	1.75898300
H	-1.58059100	-3.91049100	-0.96938300	C	-6.23851100	-2.05475600	0.62043700
H	-2.37615500	-1.55374000	-1.98438000	H	-2.91549700	1.08562200	-2.36437200
H	5.16835500	-0.47725700	-1.51169500	C	-3.48042700	0.57348400	-3.14670100
H	4.45922100	-0.15086800	2.69926000	H	-1.84790700	0.97259800	-2.56292400
H	3.10379200	1.80586500	-2.48138200	C	-3.13146600	2.15786900	-2.41705200
H	1.82895600	0.60248900	-2.55662000	H	2.41889100	0.80233200	0.03979900
H	3.46850500	0.17578500	-3.10502000	C	2.86813600	0.77945500	1.36563900
H	6.18735100	-2.15107300	1.00237000	C	4.14035600	0.27528100	1.62617500
H	6.71609300	-0.68564100	1.83191600	H	4.49642600	0.24570600	2.65335100
H	6.99216600	-0.86943900	0.09204900	C	4.95744300	-0.20442600	0.60389100
H	1.20693500	0.07941800	2.77002400	H	4.49999900	-0.11099000	-0.71220200
H	1.39560100	1.79058600	2.38186700	C	5.15072900	-0.42557100	-1.52538500
H	2.52429400	0.99842000	3.50236900	H	3.25194500	0.42630900	-1.02066000
H	-5.18674900	-0.22565800	-1.57940900	C	1.98091100	1.25025700	2.48766400
H	-4.43579200	-0.59226500	2.62081700	H	1.46890900	2.18294700	2.23468200
H	-3.45775100	0.61912200	-3.06499800	H	2.56253600	1.40796400	3.39763700
H	-1.84520200	1.04401900	-2.43996300	H	1.20527800	0.50710700	2.71729800
H	-3.17497500	2.16468800	-2.22453700	C	6.30301200	-0.80808200	0.91023000
H	-7.01932800	-0.44112700	1.08717500	H	6.22137900	-1.89325500	1.03394300
H	-6.25170400	-1.98509300	1.45868500	H	6.71691700	-0.40043900	1.83530600
H	-6.67804800	-1.58527100	-0.21374800	C	7.01512200	-0.62004300	0.10331500
H	-0.98878800	0.31618900	2.58084200	C	2.86088800	0.69756400	-2.44651800
H	-2.44881600	0.24929400	3.58577000	H	3.08124500	1.74559800	-2.67052100
H	-1.98087500	1.75354100	2.76117800	H	1.79032000	0.56027400	-2.61022500
4				H	3.41420900	0.06008800	-3.14085600
C	-0.00582600	0.62286800	-0.07797400	N	-1.13356500	1.35350700	-0.11638100
C	-1.16789500	2.84061400	-0.25862400	N	-0.01957400	3.45578300	-0.53031700
C	1.14140200	2.80686300	-0.55241100	N	1.11634500	1.35091200	-0.21361500
				O	1.26765400	3.34191900	-0.13720000
				O	2.23976100	3.25593500	-0.80990200
				Fe	0.00553100	-1.44087400	-0.16823100
				O	2.49655900	-2.24235000	1.30027300
				O	-1.65702900	-2.10871300	2.23638100
				C	1.58775100	-1.77376500	0.81234500
				C	-1.06708400	-1.74280800	1.34132600

6^{Me}(OTf).

C	-2.95699300	1.66027200	-0.64749000
C	-0.81039800	3.34307300	-0.36586200
C	-0.35723900	0.47141500	0.42894300
C	-0.46846000	-1.57234300	1.81447400
C	1.60353300	-0.32457600	1.70741100
C	1.46243300	-2.41015000	3.00317000
H	0.91593300	-2.47814700	3.94486500
H	2.49618100	-2.12025500	3.17169100
H	1.40370700	-3.35603800	2.46209800
C	-0.09938700	0.34320400	-2.54281400
H	0.45456700	-0.53409800	-2.22245200
C	-1.48168700	0.38617100	-2.84339600
H	-2.15990900	-0.45578900	-2.84680500
C	-1.81553200	1.73738900	-3.17597200
H	-2.79391900	2.10241800	-3.46027500
C	-0.64678200	2.51495100	-3.05764300
H	-0.57729800	3.58214400	-3.22594600
C	0.41763700	1.66088300	-2.64467700
H	1.44030500	1.96406700	-2.47887900
C	-2.43796900	-0.80774400	0.69362600
C	-2.86382100	-1.70060300	-0.29591600
C	-4.24046500	-1.87625100	-0.45176000
H	-4.59289100	-2.55387000	-1.22631600
C	-5.16738200	-1.25584500	0.38818000
C	-4.69172000	-0.44700900	1.42090000
H	-5.39757900	0.01584600	2.10698700
C	-3.32952100	-0.20847500	1.59045900
C	-1.88574300	-2.51423400	-1.09244800
H	-1.47860800	-3.31691600	-0.46464300
H	-1.02971600	-1.92633500	-1.42516200
H	-2.36527900	-2.97133600	-1.96092400
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H	3.74575800	1.03304200	-1.87363200
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C	3.84405200	5.53154800	0.33919200
H	3.32426400	6.36105200	0.82387200
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O	1.15778600	-3.68156200	-2.14557700
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O	1.27391700	-1.69825700	-0.67468000
S	1.44148200	-3.16761600	-0.80947500
F	3.65836700	-4.60780500	-0.70920300
F	3.92294900	-2.61694900	-1.51623800
F	3.63151000	-2.88947400	0.61061700

7			
Fe	-0.08574600	-1.42991500	-0.41103900
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O	-1.67848000	-2.36172900	1.94770300
O	2.27683100	3.26949600	-0.42909800
O	-2.40721700	3.30884300	0.12273800
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H	6.05911200	-1.58372300	2.63116200	H	-6.78582700	-0.19158500	1.21316500
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H	7.97930200	-1.53403500	-0.15242000	H	-5.79504800	0.89325400	-1.64208500
H	6.61781900	-1.72843400	-1.25681900				

8

Fe	1.11538200	-1.71552100	0.04690700
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O	-0.36324000	3.30303400	-0.58684600
N	2.63937600	0.83237000	0.38253300
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C	-0.76746600	-1.79457900	0.30033600
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C	2.26767300	4.54416300	-0.45947000
C	2.45239600	-1.69694800	-1.65518900
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C	-0.94562100	0.84211400	-0.03116200
C	-1.65656100	0.68663800	-1.22600100
C	-3.00995800	0.37909700	-1.16771400
C	-3.69291200	0.28428500	0.06194700
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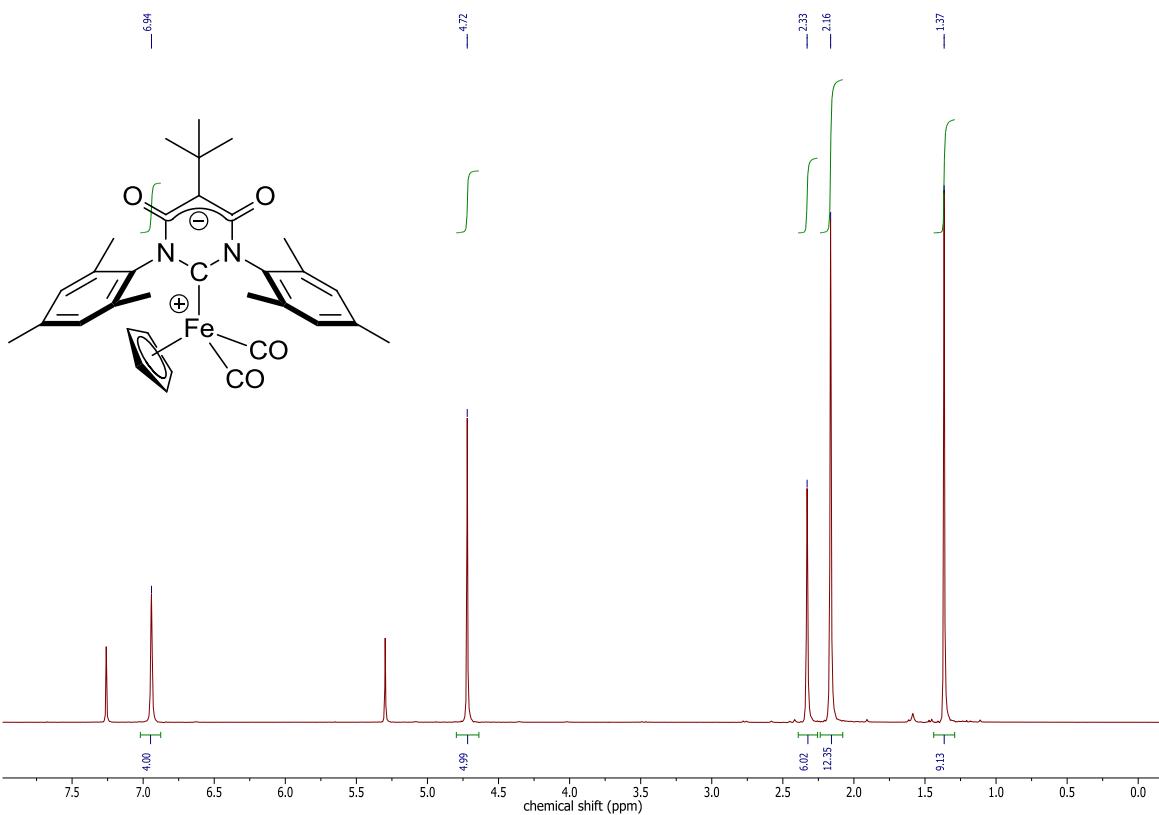
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C	1.20771800	1.41133000	-0.61850100
C	0.15572000	3.60921800	-1.30179800
H	-0.57670000	3.76372700	-2.09824900
H	1.15498200	3.80532300	-1.70351000
H	-0.09362600	4.37933800	-0.55015400
C	-2.57529100	-0.32945100	-0.07670400
C	-3.28964200	-0.02797400	1.08381600
C	-4.53215600	-0.63091000	1.27002300
H	-5.09869300	-0.40287200	2.17047700
C	-5.06373400	-1.51392700	0.33127400
C	-4.32466300	-1.78368200	-0.81950300
H	-4.72876400	-2.46459500	-1.56567900
C	-3.07875800	-1.20011000	-1.04478100
C	-2.72762500	0.93303800	2.09592100
H	-2.70499900	1.94230100	1.67240300
H	-1.70234700	0.65611900	2.36586500
H	-3.33512500	0.94324800	3.00350800
C	-6.39390300	-2.18398100	0.56572600
H	-6.92069700	-2.35566700	-0.37648900
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H	-6.26291000	-3.15704700	1.05107900
C	-2.28806800	-1.49859100	-2.28973400
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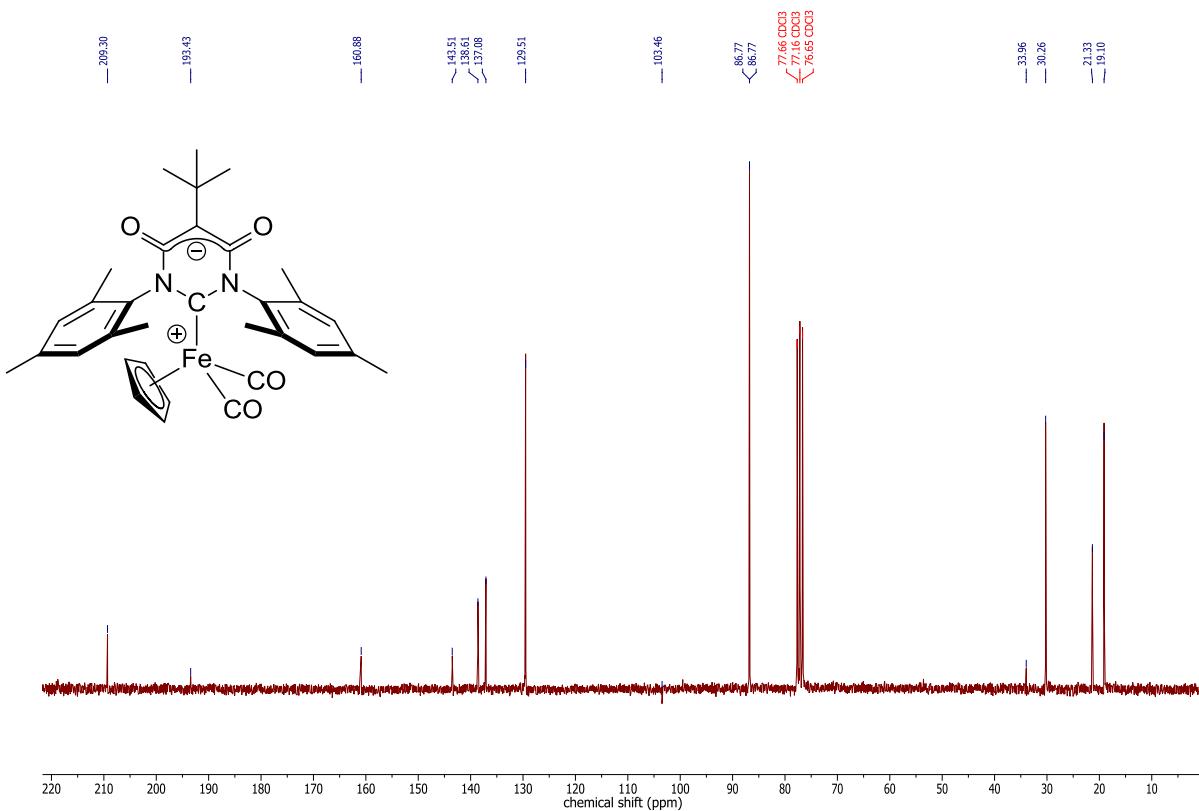
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C	2.45850900	-1.15754200	1.24830900	H	-1.36171300	-0.66133600	-2.58781200
C	3.57872700	-1.96068900	1.45557600	H	-2.14024700	-2.19844700	-3.01657900
H	3.82005700	-2.27907300	2.46783800	H	-0.71624400	-2.17129500	-1.95065800
C	4.38960400	-2.37029600	0.39769500	C	-5.74689100	-3.13459400	0.31345500
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H	4.66775800	-2.28197900	-1.73041300	H	-6.63561700	-2.54839000	0.05704000
C	2.93978200	-1.16092900	-1.14249200	H	-5.89079200	-3.52174800	1.32573300
C	1.57887200	-0.74558900	2.39938700	C	-2.65962700	0.30502100	2.32686900
H	1.41225300	0.33960900	2.40783800	H	-2.83121000	1.34475900	2.03143600
H	2.02490800	-1.03236700	3.35438400	H	-1.60241100	0.20488100	2.59147000
H	0.59169400	-1.20790600	2.30639500	H	-3.26021500	0.08775900	3.21256500
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H	5.44994500	-3.91410300	1.47247200	N	-1.03129000	0.11689700	-0.03004100
H	6.47497700	-2.59336700	0.90799200	O	2.18406600	2.39713800	-0.12799600
C	2.58642000	-0.72644800	-2.53805600	O	-2.33078900	1.98947800	-0.33399200
H	3.18756100	-1.26154300	-3.27608700	K	-1.50857100	4.40034100	-0.55368600
H	2.76206600	0.34787200	-2.65027100				
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N	-1.27468400	0.26278800	-0.27490300	2_{Me}			
N	0.98549500	0.04641800	-0.28588200	C	0.00258400	-0.25190200	-0.00199900
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C	0.06308800	2.19658000	-0.79096800	C	1.23154000	1.90429400	0.04013400
O	2.39730500	1.83431600	-0.65935000	C	-0.03833100	4.00819300	-0.00309100
K	1.93236300	3.51633900	1.09110600	H	-0.53865200	4.35786000	-0.90702000
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			H	-0.60194700	4.35555600	0.86376000	
			C	-2.38741200	-0.22786500	-0.05869500	
			C	-3.16924600	-0.27770600	1.09547700	
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			H	-4.98891500	-1.03942100	1.92979900	
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			H	-2.82597200	1.50996000	2.23589600	
			H	-1.69335100	0.19793200	2.59374400	
			H	-3.35671100	0.12066700	3.19900400	
			C	-6.05842800	-2.43551400	-0.14978500	
			H	-6.53669800	-2.37839400	-1.13072800	
			H	-6.76663500	-2.07085900	0.59777800	
			H	-5.86574500	-3.49224700	0.06242000	
			C	-1.88756600	-0.79084400	-2.46719600	
			H	-1.68063900	0.24793400	-2.74565600	
			H	-2.36898300	-1.28190700	-3.31469100	
			H	-0.92614400	-1.27814900	-2.27662700	
			C	2.39014800	-0.22356300	0.05669400	
			C	2.76495400	-0.85285300	1.24569600	
			C	3.96329400	-1.56185100	1.25496400	
			H	4.27066100	-2.06265200	2.17041500	
			C	4.77592600	-1.64427200	0.12327500	
			C	4.36177400	-1.00332900	-1.04196500	
			H	4.98036000	-1.06315500	-1.93431700	
			C	3.16785000	-0.28440900	-1.09847100	
			C	1.88927100	-0.77703200	2.46745300	
			H	1.68455600	0.26314300	2.74235000	
			H	2.36911200	-1.26630200	3.31687800	

H	0.92687400	-1.26287400	2.27785400
C	6.07893700	-2.40004100	0.17479200
H	6.48819800	-2.55380300	-0.82598700
H	5.94857300	-3.37883500	0.64438600
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H	3.35166300	0.09823900	-3.20531900
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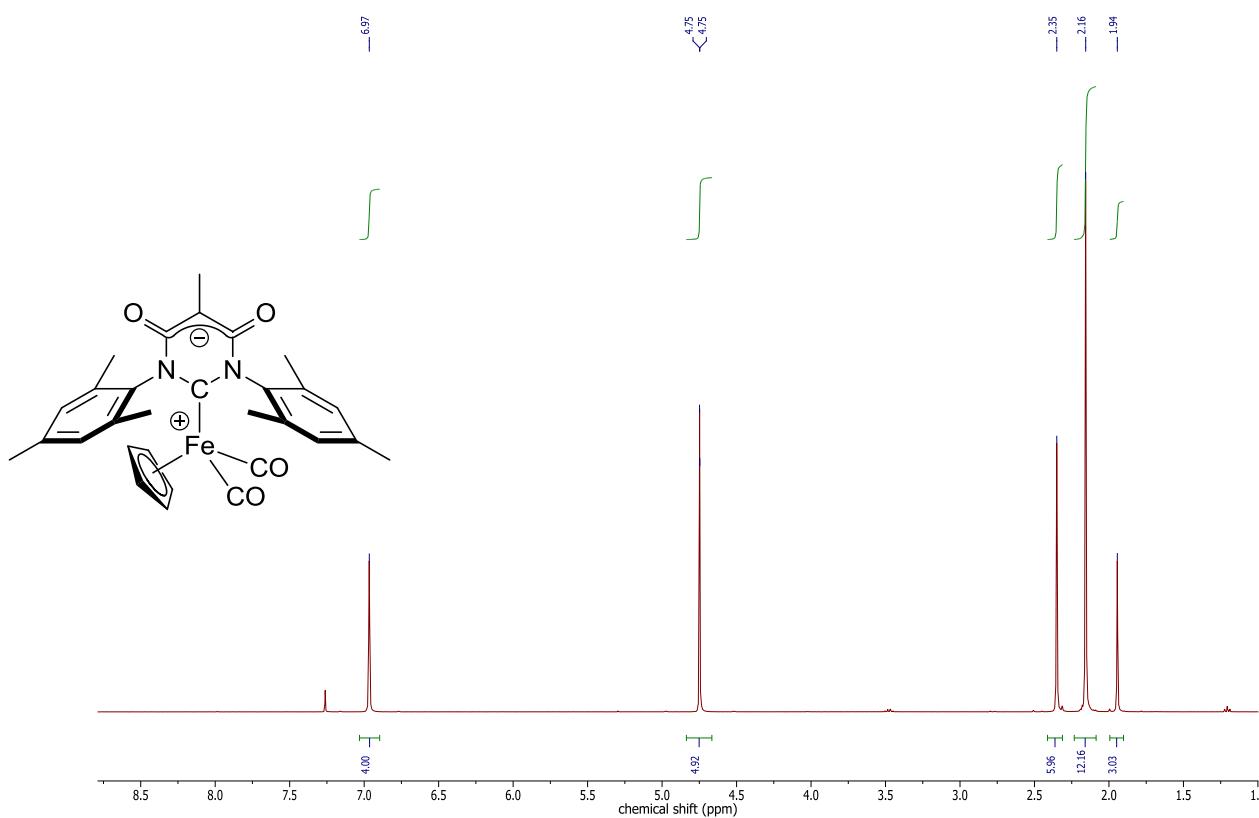
¹H NMR spectrum of 3_tBu (250 MHz, CDCl₃)



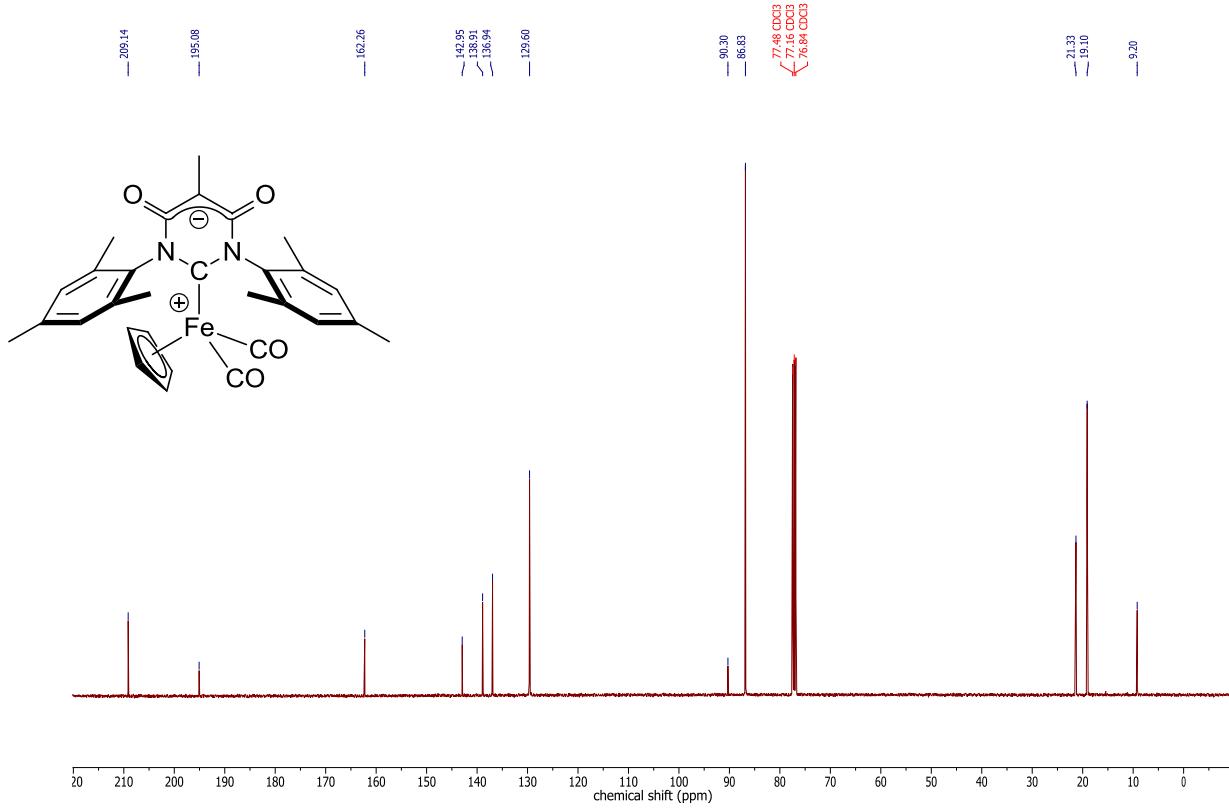
¹³C {¹H} NMR spectrum of 3_tBu (63 MHz, CDCl₃)



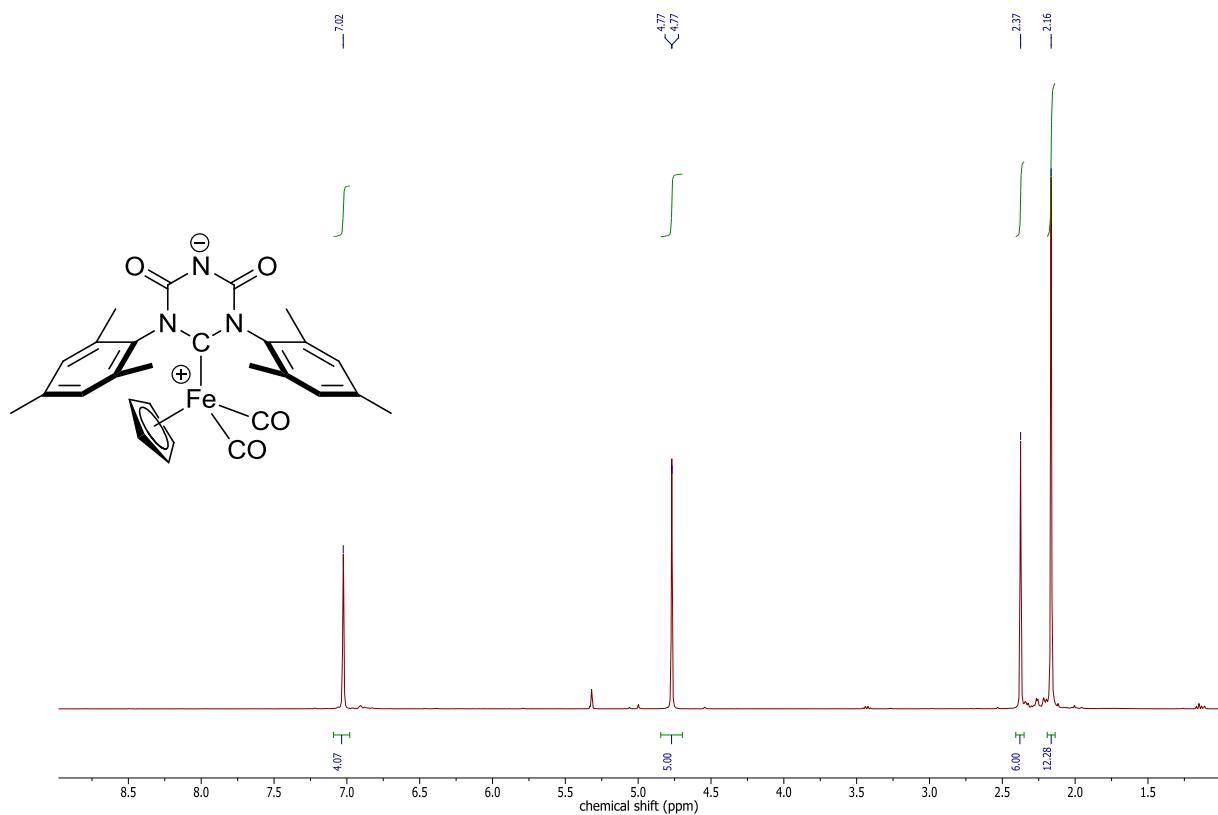
¹H NMR spectrum of 3_{Me} (400 MHz, CDCl₃)



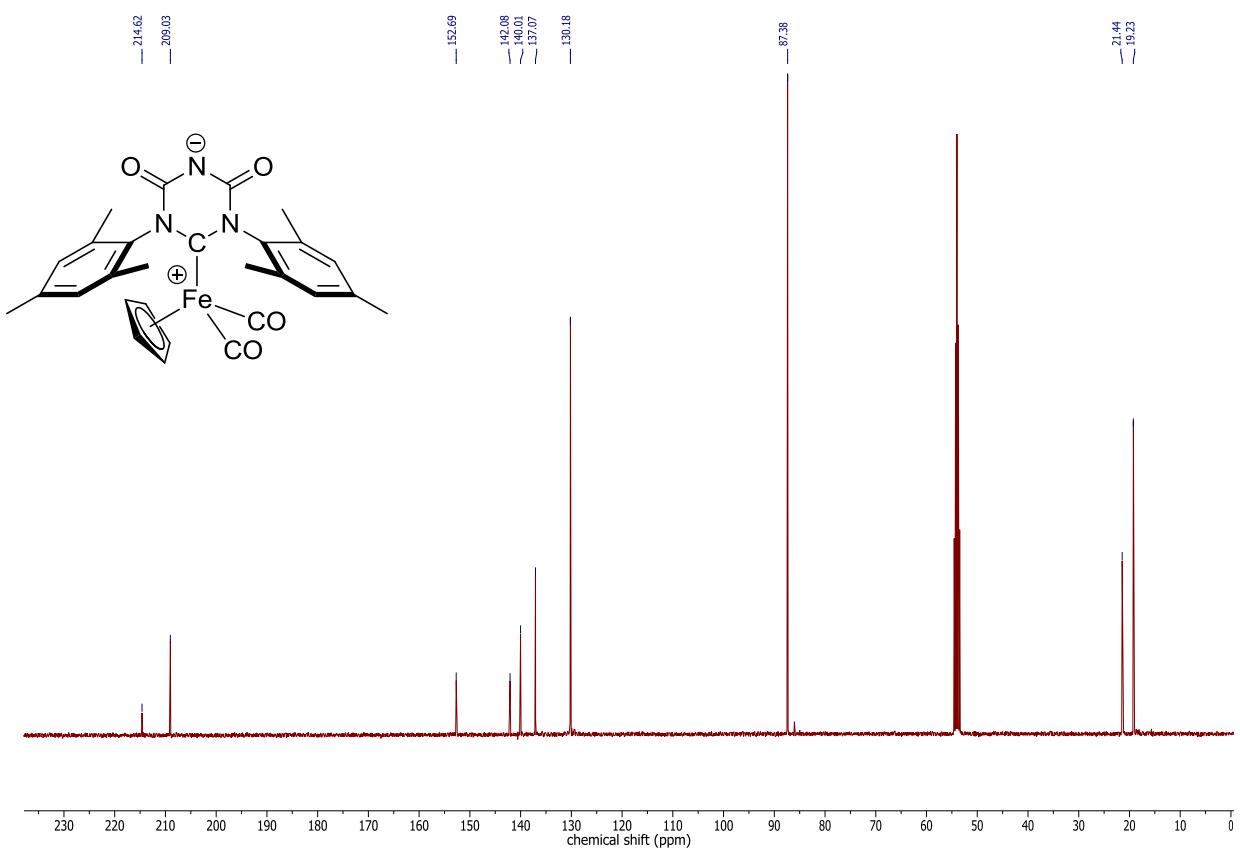
¹³C {¹H} NMR spectrum of 3_{Me} (100.5 MHz, CDCl₃)



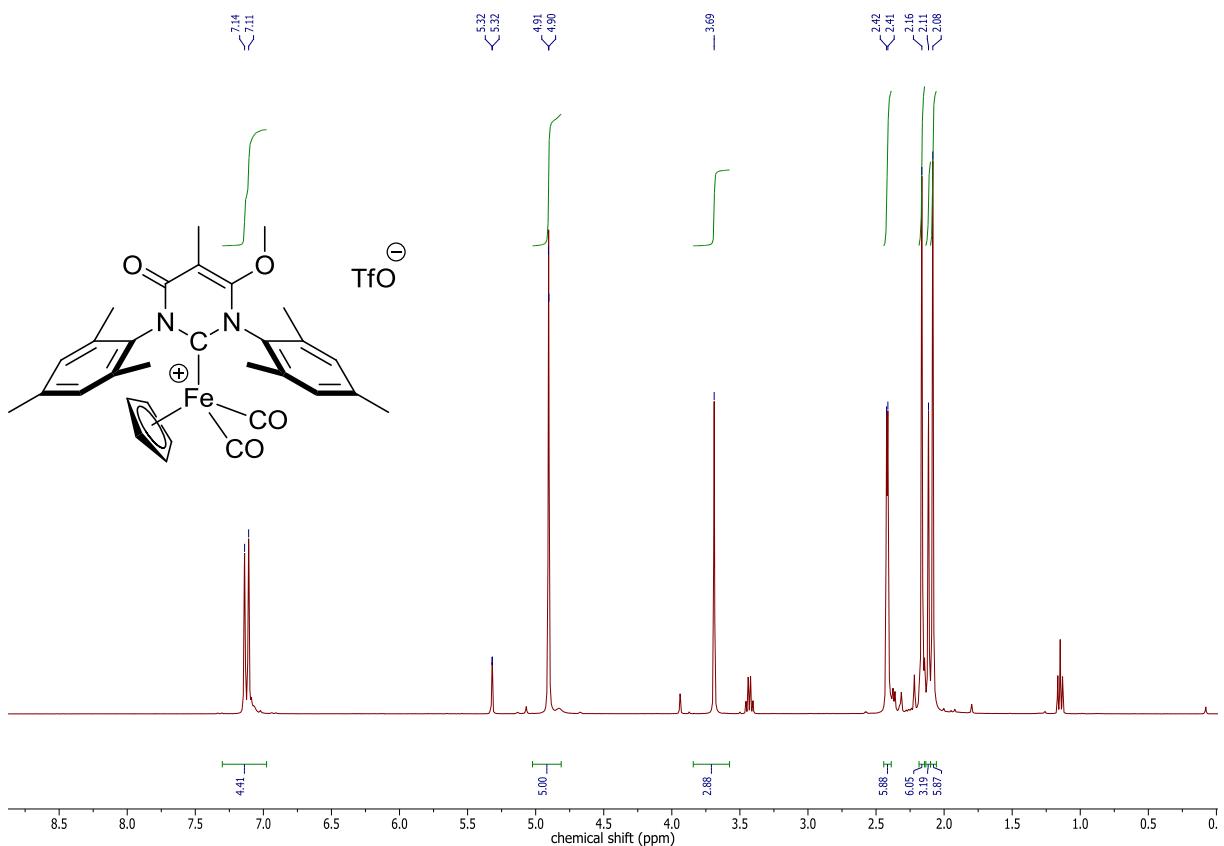
^1H NMR spectrum of 4 (400 MHz, CD_2Cl_2)



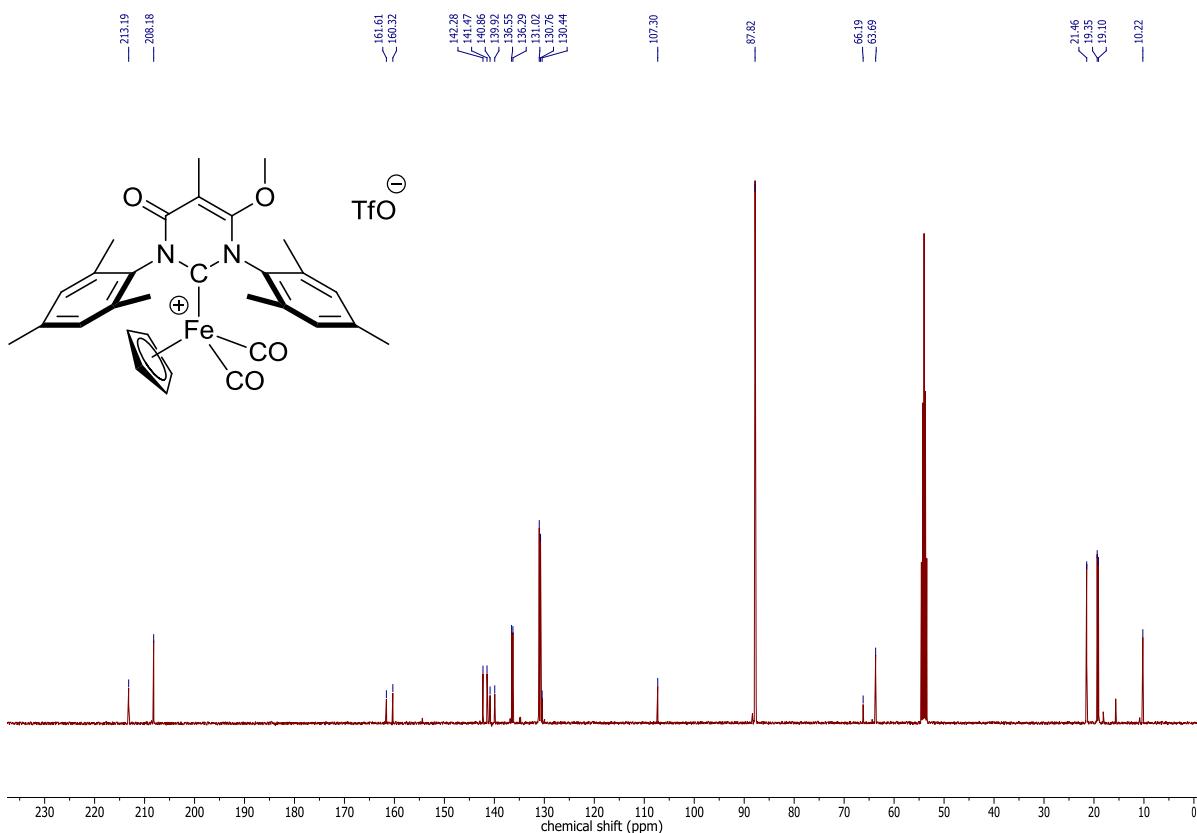
$^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of 4 (100.5 MHz, CD_2Cl_2)



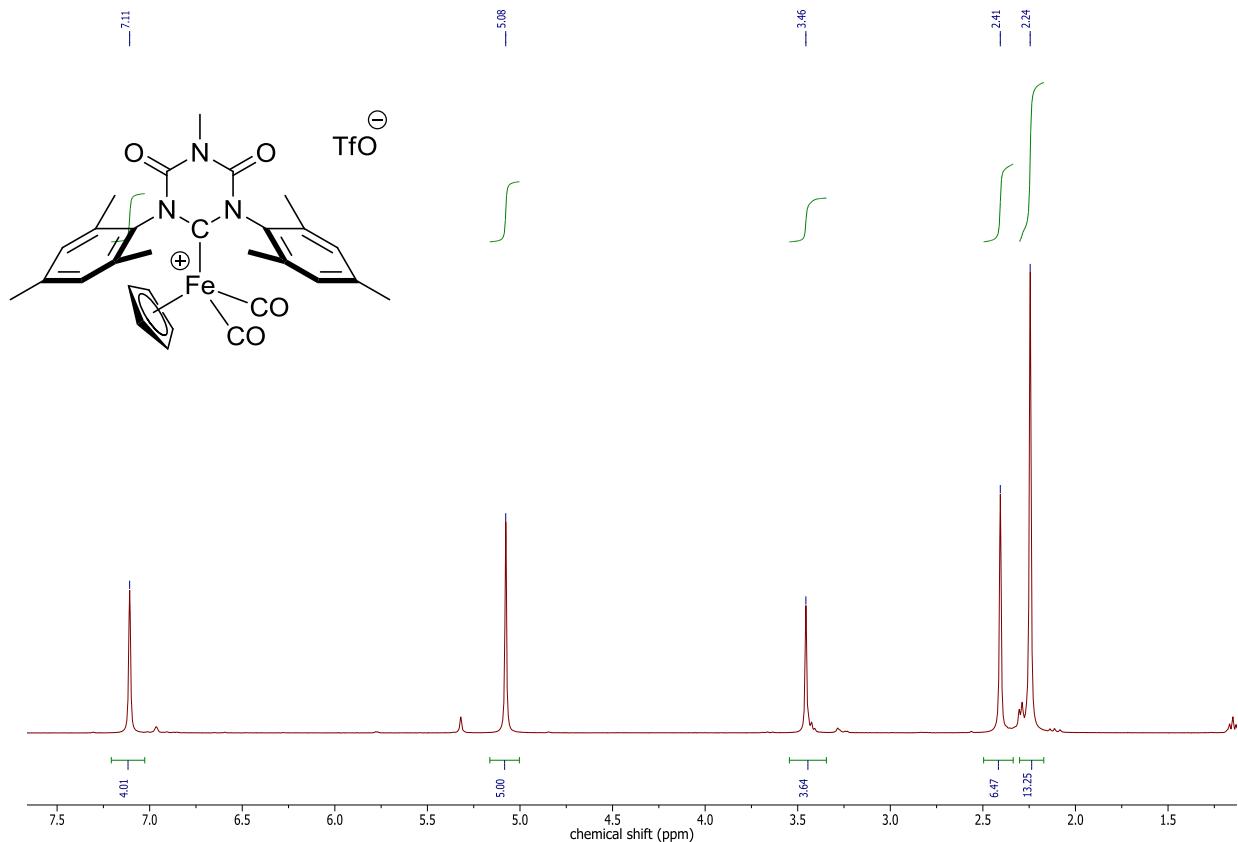
^1H NMR spectrum of $[5\text{Me}^{\text{Me}}](\text{TfO})$ (400 MHz, CD_2Cl_2)



$^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of $[5\text{Me}^{\text{Me}}](\text{TfO})$ (100.5 MHz, CD_2Cl_2)



¹H NMR spectrum of [6^{Me}](TfO) (400 MHz, CD₂Cl₂)



¹³C{¹H} NMR spectrum of [6^{Me}](TfO) (100.5 MHz, CD₂Cl₂)

