

Supporting Information

Tuning the Dissociation of the Fe-PPh₂(OR) Bond in Chiral-at-Metal Complexes [CpFe(Prophos)PPh₂(OR)]PF₆, R = Me, Et, iPr, tBu. The Preparative Trick of N₂ Bubbling

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(R_{Fe},R_C)/(S_{Fe},R_C)-(η⁵-Cyclopentadienyl)[diphenylphosphane-κP][propane-1,2-diylbis(diphenylphosphane-κP)]iron hexafluorophosphate,
(R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂H]PF₆

A solution of (R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)NCMe]PF₆ 5:95 (132 mg, 0.18 mmol) and PHPh₂ (0.36 mL, 2.1 mmol) in THF (15 mL) was refluxed for 20 h. The reaction mixture was passed over a short Celite column. The orange filtrate was evaporated. The residue was washed with diethyl ether/hexane to remove excess ligand. The product (R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂H]PF₆ 15:85 was obtained in 57% (90 mg) yield. Recrystallization from dichloromethane/hexane afforded a crystalline fraction of diastereomer ratio 2:98. From this sample a crystal was used to determine structure.

¹H NMR (CDCl₃, 293 K, major (S_{Fe},R_C)-diastereomer, minor (R_{Fe},R_C)-diastereomer in brackets, if distinguishable): δ 7.80-6.84 (m, 28H, Ar-H), 6.45 (br t, 2H, Ar-H) 4.54 (br s, 5H, Cp-H) [4.18 (br s, 5H, Cp-H)], 3.94-2.14 (m, 3H, CH₂CH), 1.13 (br s, 3H, CH₃).

³¹P{¹H} NMR (CDCl₃, 293 K, major (S_{Fe},R_C)-diastereomer, minor (R_{Fe},R_C)-diastereomer in brackets): δ 100.36 (dd, 1P, ²J_{P-P} = 46.5 Hz, ²J_{P-P} = 37.6 Hz, PCH) [85.30 (br t, 1P, ²J_{P-P} = 41.1 Hz, PCH)], 67.31 (dd, 1P, ²J_{P-P} = 50.9 Hz, ²J_{P-P} = 37.6 Hz, PCH₂) [79.32 (br t, 1P, ²J_{P-P} = 43.0 Hz, PCH₂)], 47.97 (dd, 1P, ²J_{P-P} = 50.9 Hz, ²J_{P-P} = 46.5 Hz, PH) [51.14 (br t, 1P, ²J_{P-P} = 48.6 Hz, PH)], -144.08 (septet, 1P, ¹J_{P-F} = 713.5 Hz, PF₆).

ES-MS: *m/z* 719 (cation, 100).

Anal. Calcd for C₄₄H₄₂F₆FeP₄ (864.2): C, 61.13; H, 4.90. Found: C, 60.85; H, 4.75.

Decomposition of (R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂(O*i*Pr)]PF₆ and spectroscopic data of (R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂F]PF₆

In C₆D₅Cl: A solution of (R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂(O*i*Pr)]PF₆ (10 mg, 1:1 ratio) in C₆D₅Cl (0.4 mL) was heated at 80 °C. The reaction was monitored by ³¹P{¹H} NMR. The formation of the diastereomers (R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂F]PF₆ is obvious in the time-resolved spectrum of Fig. S8.

(R_{Fe},R_C)/(S_{Fe},R_C)-[CpFe(Prophos)PPh₂F]PF₆:

³¹P{¹H} NMR (C₆D₅Cl, 293 K, major (S_{Fe},R_C)-diastereomer, minor (R_{Fe},R_C)-diastereomer in brackets): δ 245.00 (dt, 1P, ¹J_{P-F} = 914.0 Hz, ²J_{P-P} = 67.1 Hz, PPh₂F) [245.56 (dt, 1P, ¹J_{P-F} = 900.8 Hz, ²J_{P-P} = 67.4 Hz, PPh₂F)], 112.28 (dd, 1P, ²J_{P-P} = 67.3 Hz, ²J_{P-P} = 37.2 Hz, PCH) [95.27 (dd, 1P,

$^2J_{P-P} = 63.4$ Hz, $^2J_{P-P} = 41.6$ Hz, PCH)], 73.44 (ddd, 1P, $^2J_{P-P} = 65.9$ Hz, $^2J_{P-P} = 37.2$ Hz, $^3J_{P-F} = 15.4$ Hz, PCH₂) [91.46 (m, 1P, PCH₂)], -142.78 (septet, 1P, $^1J_{P-F} = 712.4$ Hz, PF₆).

¹⁹F NMR (C₆D₅Cl, 293 K, major (*S*_{Fe},*R*_C)-diastereomer, minor (*R*_{Fe},*R*_C)-diastereomer in brackets): δ -71.90 (d, 1F, $^1J_{P-F} = 712.4$ Hz, PF₆), -136.21 (d, 1F, $^1J_{P-F} = 914.0$ Hz, PPh₂F) [-128.73 (d, 1F, $^1J_{P-F} = 900.8$ Hz, PPh₂F)].

MS (ESI): *m/z* 737.2 (cation, 100).

HRMS (ESI): calcd for the cation C₄₄H₄₁FFeP₃⁺ 737.1755; found *m/z* 737.1755.

In CDCl₃: A solution of (*R*_{Fe},*R*_C)/(*S*_{Fe},*R*_C)-[CpFe(Prophos)PPh₂(O*i*Pr)]PF₆ (10 mg, 1:1 ratio) in CDCl₃ (0.4 mL) was heated at 56 °C. The reaction was monitored by ³¹P{¹H} NMR and ¹⁹F NMR spectroscopy. The formation of F₂P(O)OH and F₂P(O)O*i*Pr was observed in the spectra of Figs. S9-1 and S9-2.

F₂P(O)OH: ³¹P{¹H} NMR (CDCl₃): δ -18.10 (br t, 1P, $J_{P-F} = 976.7$ Hz). ¹⁹F NMR (CDCl₃): δ -83.41 (d, 2F, $^1J_{P-F} = 976.7$ Hz).

F₂P(O)O*i*Pr: ³¹P{¹H} NMR (CDCl₃): δ -20.58 (t, 1P, $J_{P-F} = 1011.8$ Hz). ¹⁹F NMR (CDCl₃): δ -84.97 (d, 2F, $^1J_{P-F} = 1011.8$ Hz).

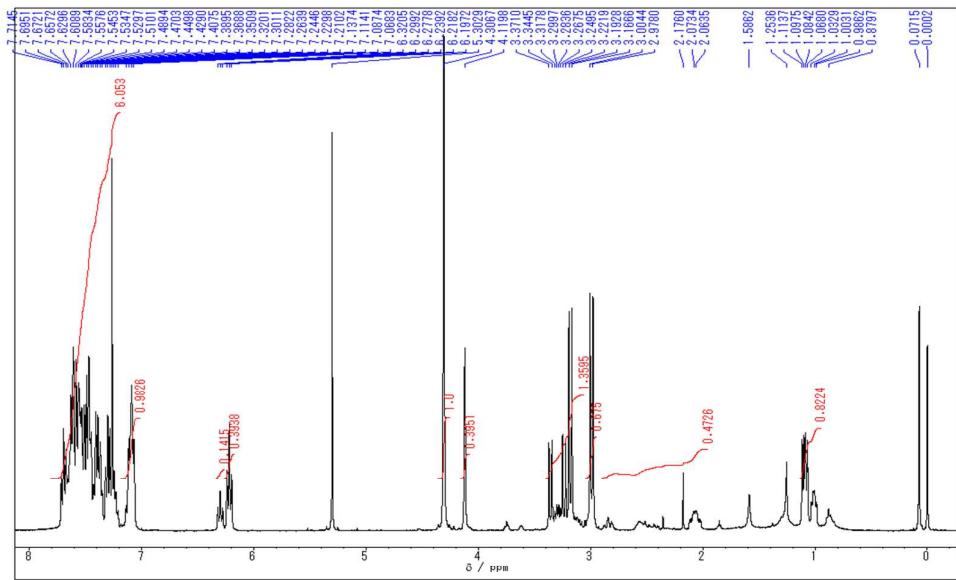


Figure S1-1. ^1H NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}}, R_{\text{C}})/(S_{\text{Fe}}, R_{\text{C}})$ -[CpFe(Prophos)-PPh(OMe)₂]PF₆ in CDCl₃ (32:68 ratio).

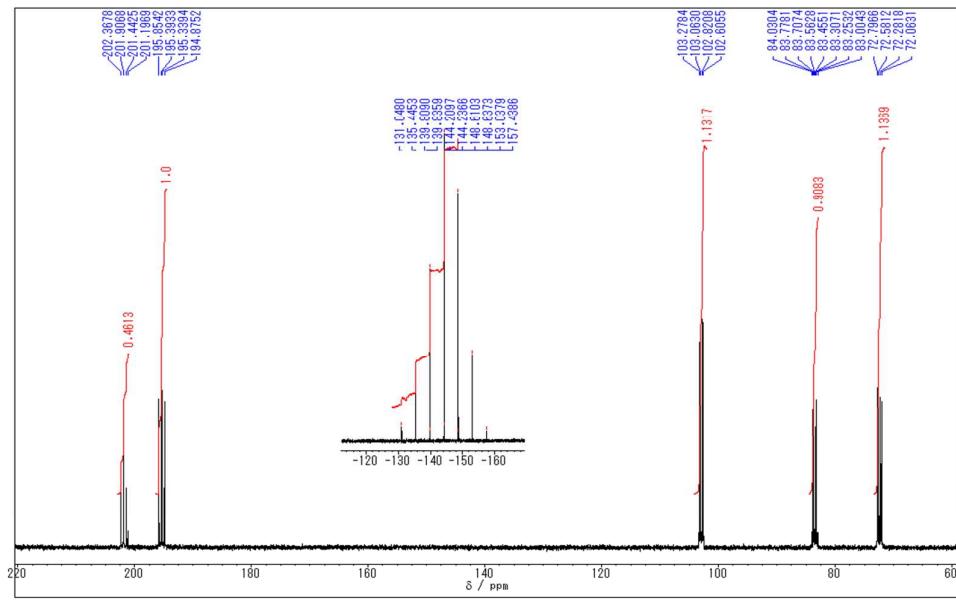


Figure S1-2. $^{31}\text{P}\{^1\text{H}\}$ NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}}, R_{\text{C}})/(S_{\text{Fe}}, R_{\text{C}})$ -[CpFe-(Prophos)PPh(OMe)₂]PF₆ in CDCl₃ (32:68 ratio).

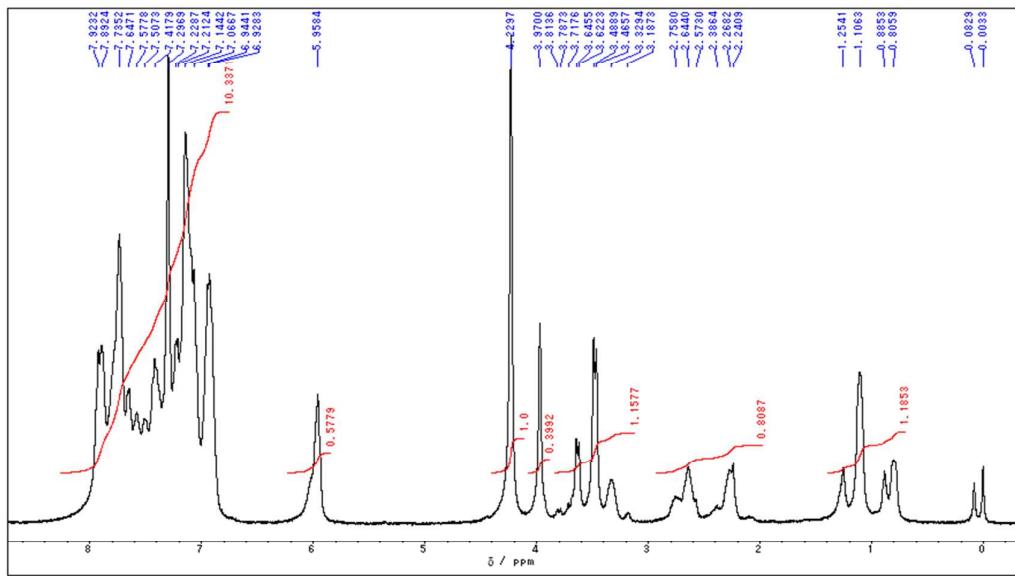


Figure S2-1. ^1H NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}}, R_{\text{C}})/(S_{\text{Fe}}, R_{\text{C}})$ -[CpFe(Prophos)-PPh₂(OMe)]PF₆ in CDCl₃ (23:77 ratio).

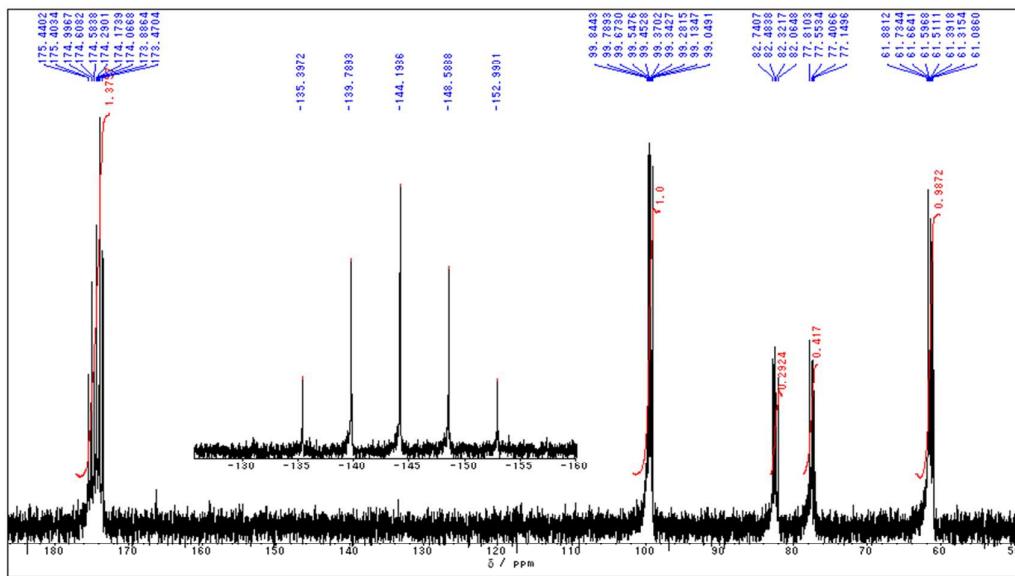


Figure S2-2. $^{31}\text{P}\{^1\text{H}\}$ NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}}, R_{\text{C}})/(S_{\text{Fe}}, R_{\text{C}})$ -[CpFe(Prophos)-PPh₂(OMe)] in CDCl₃ (23:77 ratio).

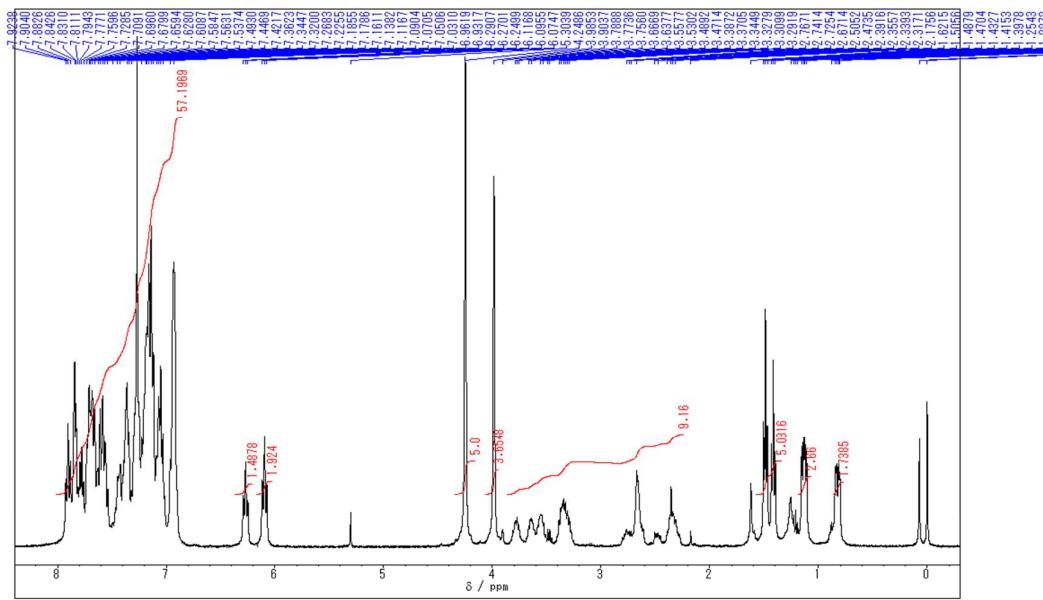


Figure S3-1. ^1H NMR Spectrum of a mixture of diastereomers $(R_{\text{Fe}}, R_C)/(S_{\text{Fe}}, R_C)$ -[CpFe(Prophos)-PPh₂(OEt)] in CDCl₃ (42:58 ratio).

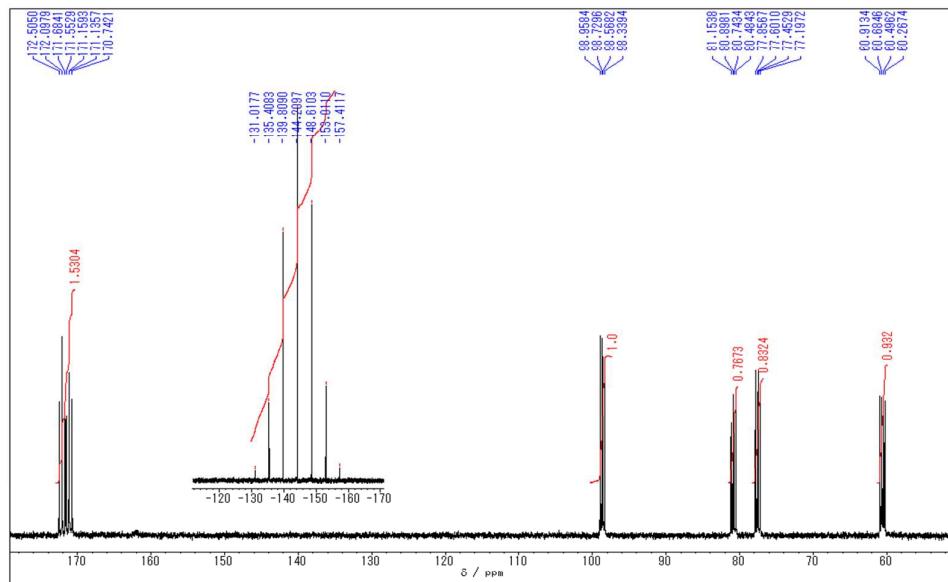
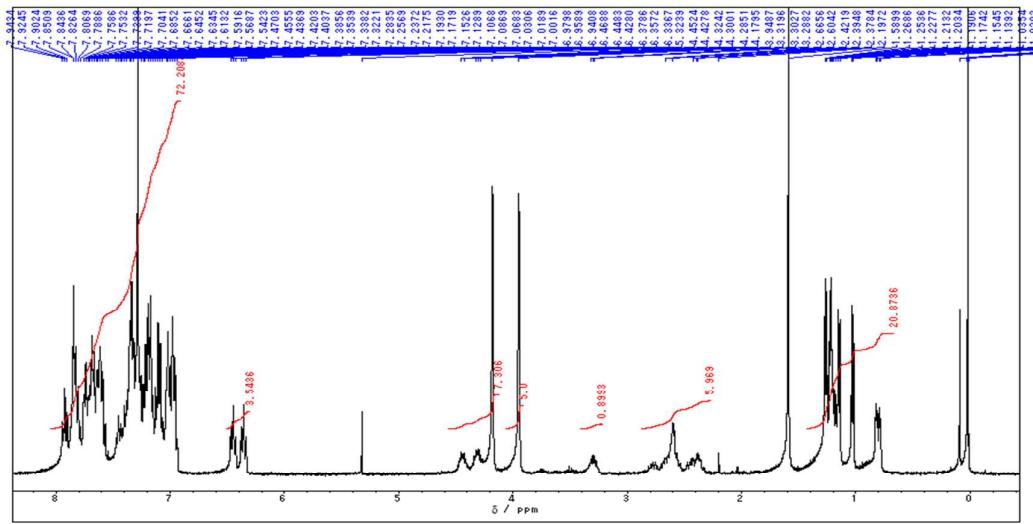


Figure S3-2. $^{31}\text{P}\{\text{H}\}$ NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}},R_{\text{C}}\right)/(S_{\text{Fe}},R_{\text{C}}\left)-[\text{CpFe-}(\text{Prophos})\text{PPh}_2(\text{OEt})]\text{PF}_6$ in CDCl_3 (42:58 ratio).



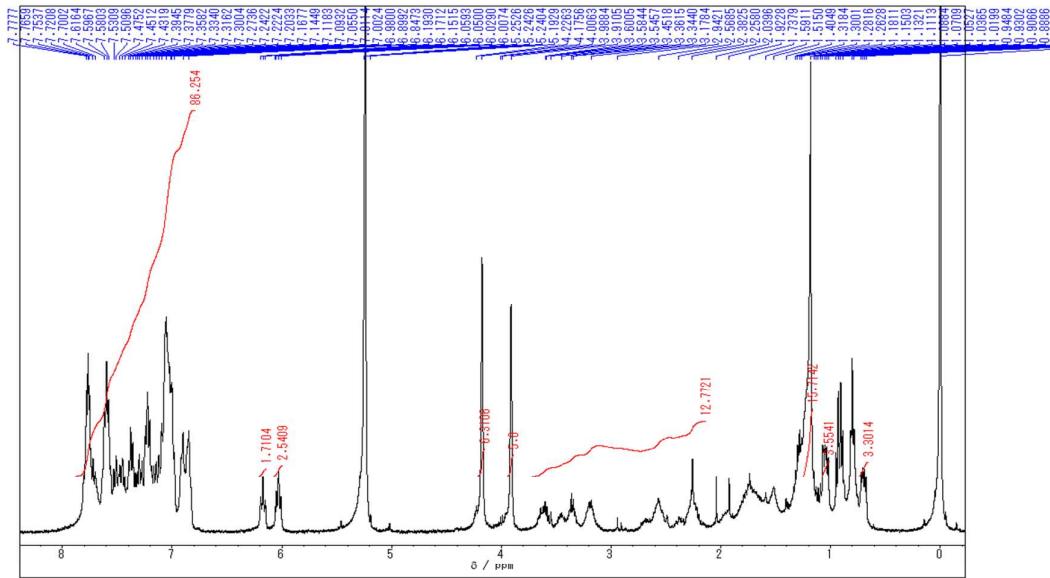


Figure S5-1. ^1H NMR Spectrum of a mixture of diastereomers $(R_{\text{Fe}}, R_C)/(S_{\text{Fe}}, R_C)$ -[CpFe(Prophos)-
PPh₂(OtBu)]PF₆ in CD₂Cl₂ (45:55 ratio).

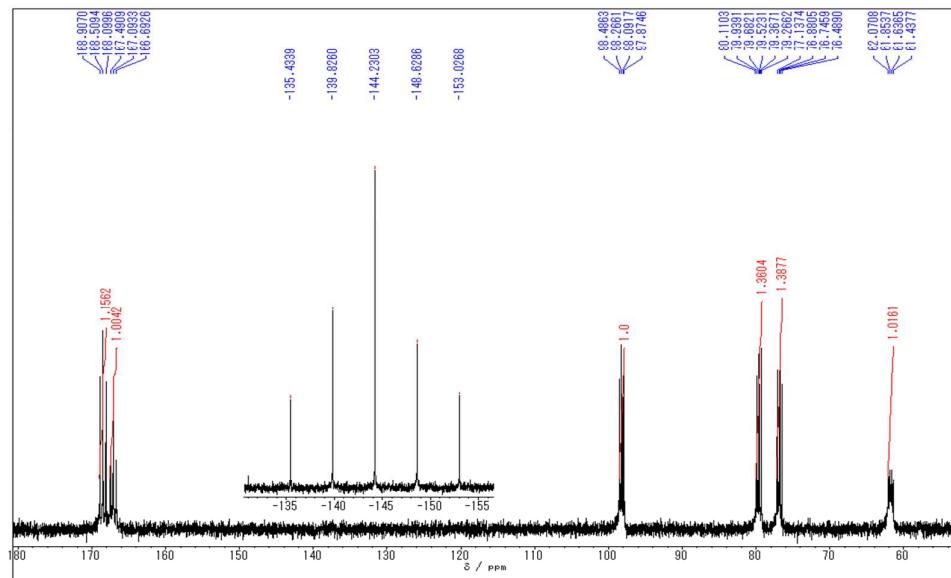


Figure S5-2. $^{31}\text{P}\{\text{H}\}$ NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}},R_{\text{C}}\right)/(S_{\text{Fe}},R_{\text{C}}\left)-[\text{CpFe-}(\text{Prophos})\text{PPh}_2(\text{OtBu})]\text{PF}_6$ in CDCl_3 (57:43 ratio).

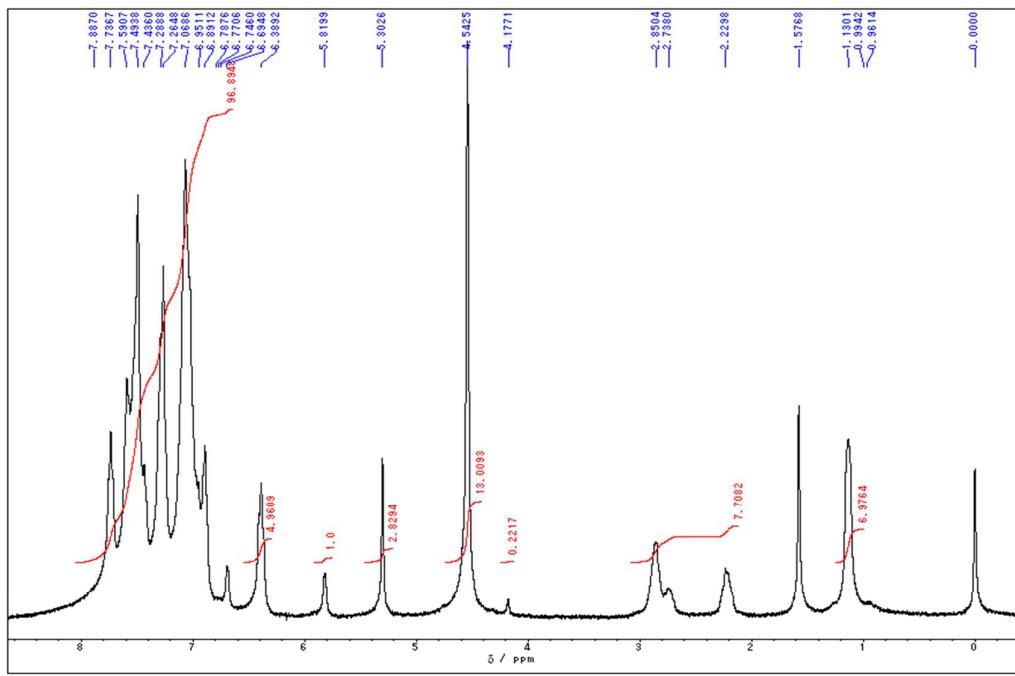


Figure S6-1. ^1H NMR Spectrum of a mixture of diastereomers (R_{Fe}, R_C)/(S_{Fe}, R_C)-[CpFe-(Prophos)PPh₂H]PF₆ in CDCl₃ (2:98 ratio).

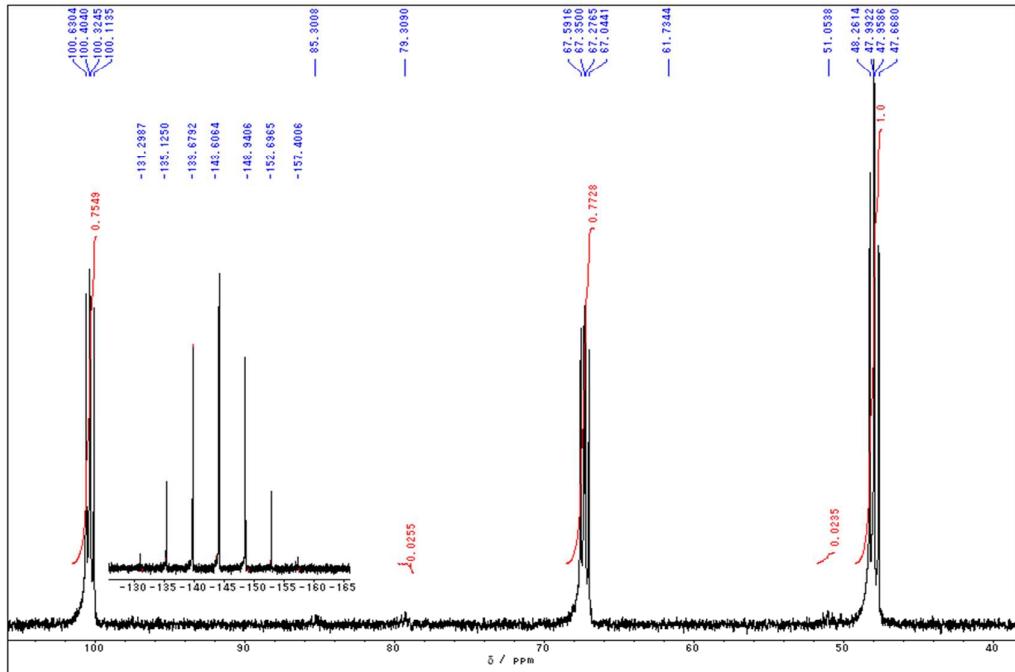


Figure S6-2. $^{31}\text{P}\{\text{H}\}$ NMR Spectrum of a mixture of diastereomers $(R_{\text{Fe}},R_{\text{C}})/(S_{\text{Fe}},R_{\text{C}})$ -[CpFe-(Prophos)PPh₂H]PF₆ in CDCl₃ (2:98 ratio).

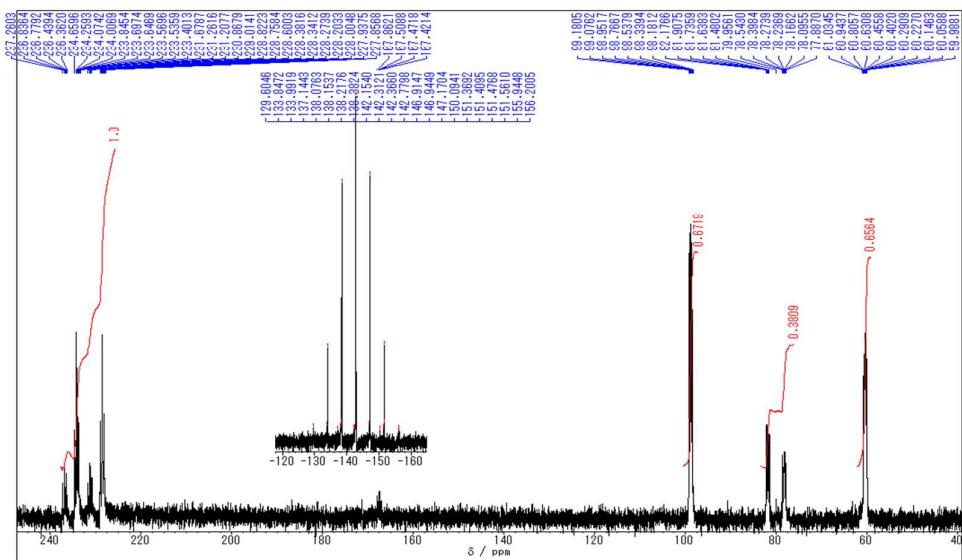


Figure S7. $^{31}\text{P}\{\text{H}\}$ NMR Spectrum of a mixture of diastereomers ($R_{\text{Fe}},R_{\text{C}}\right)/(S_{\text{Fe}},R_{\text{C}}\left)-[\text{CpFe-}(\text{Prophos})\text{PPh}_2\text{F}] \text{PF}_6$ in $\text{C}_6\text{D}_5\text{Cl}$ (15:85 ratio).

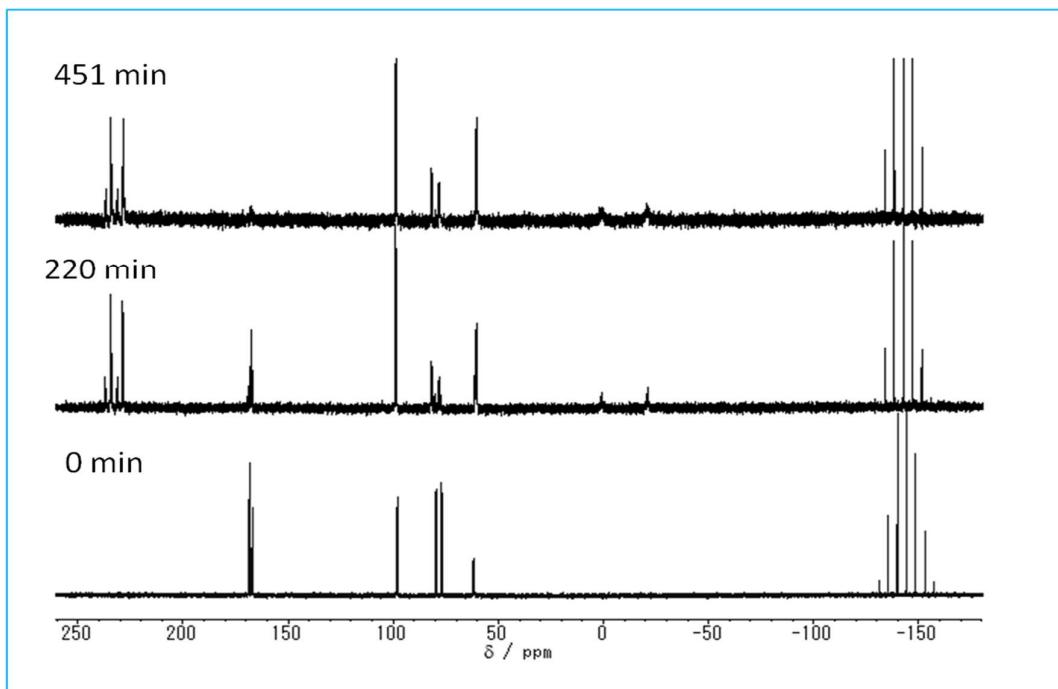


Figure S8. Time-resolved $^{31}\text{P}\{\text{H}\}$ NMR spectra of the epimerization of $[\text{CpFe}(\text{Prophos})\text{PPh}_2-(\text{O}i\text{Pr})]\text{PF}_6$ at 353 K in $\text{C}_6\text{D}_5\text{Cl}$.

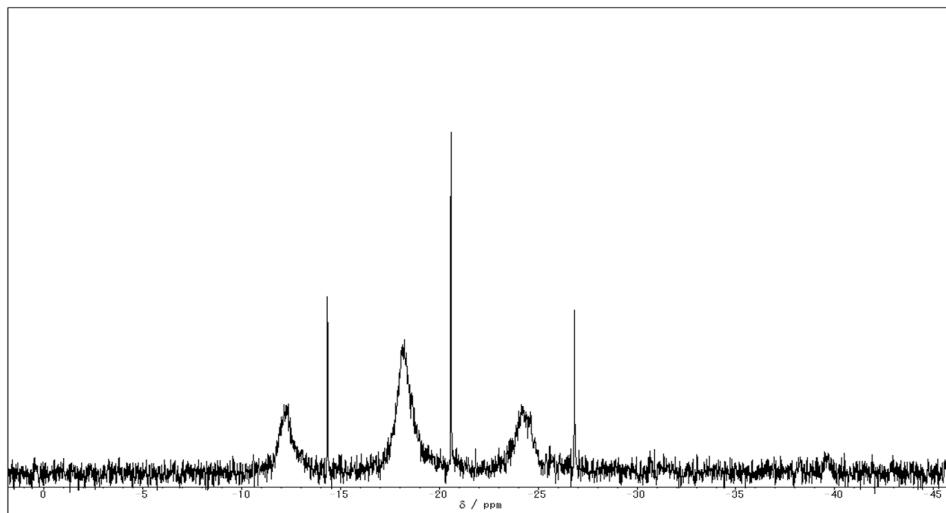


Figure S9-1. $^{31}\text{P}\{\text{H}\}$ NMR spectrum of $[\text{CpFe}(\text{Prophos})\text{PPh}_2(\text{O}i\text{Pr})]\text{PF}_6$, heated at 329 K in CDCl_3 .

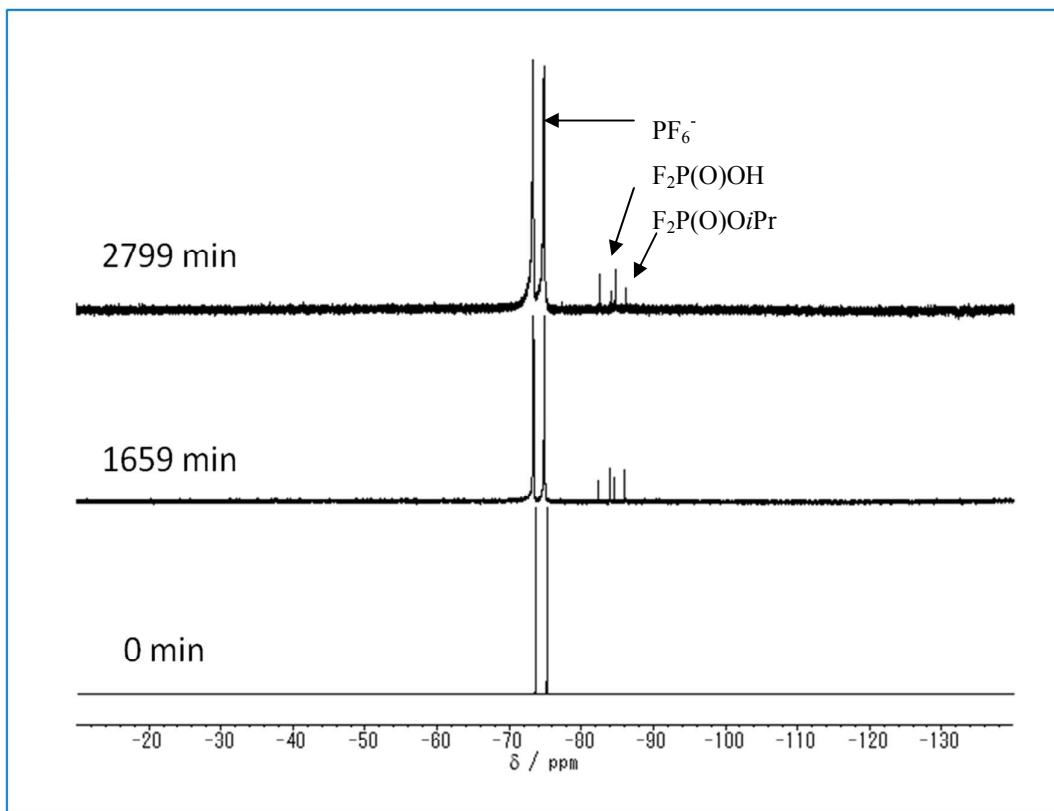


Figure S9-2. Time-resolved ^{19}F NMR spectra of $[\text{CpFe}(\text{Prophos})\text{PPh}_2(\text{O}i\text{Pr})]\text{PF}_6$, heated at 329 K in CDCl_3 .

Table S1. Crystallographic Data for [CpFe(Prophos)L]PF₆ complexes

	(S _{Fe} ,R _C)-[CpFe(Prophos)PPh-(OMe) ₂]PF ₆	(R _{Fe} ,R _C)-[CpFe(Prophos)-PPh ₂ (OMe)]PF ₆ × EtOAc	(R _{Fe} ,R _C)-[CpFe(Prophos)-PPh ₂ (OEt)]PF ₆	(R _{Fe} ,R _C)-[CpFe(Prophos)-PPh ₂ (O <i>i</i> Pr)]PF ₆	(S _{Fe} ,R _C)-CpFe(Prophos)-PPh ₂ H]PF ₆
Empirical formula	C ₄₀ H ₄₂ F ₆ FeO ₂ P ₄	C ₄₉ H ₅₂ F ₆ FeO ₃ P ₄	C ₄₆ H ₄₆ F ₆ FeO ₃ P ₄	C ₄₇ H ₄₈ F ₆ FeOP ₄	C ₄₄ H ₄₂ F ₆ FeP ₄
Crystal size [mm ³]	0.19 × 0.04 × 0.01	0.27 × 0.21 × 0.07	0.39 × 0.32 × 0.19	0.10 × 0.05 × 0.03	0.31 × 0.09 × 0.05
Colour and shape	yellow rod	orange prism	orange block	orange rod	orange rod
Formula weight [g mol ⁻¹]	848.47	982.64	908.56	922.58	864.51
Crystal system	monoclinic	monoclinic	orthorhombic	orthorhombic	orthorhombic
Space group	<i>P</i> 2 ₁	<i>P</i> 2 ₁	<i>P</i> 2 ₁ 2 ₁ 2 ₁	<i>P</i> 2 ₁ 2 ₁ 2 ₁	<i>P</i> 2 ₁ 2 ₁ 2 ₁
Absorption correction	multi-scan	multi-scan	analytical	analytical	multi-scan
Transmission min/max	0.626 / 1.000	0.665 / 1.000	0.741 / 0.857	0.892 / 0.956	0.537 / 1.000
<i>a</i> [Å]	14.8881(4)	11.360(4)	9.71600	9.8728(2)	10.594(5)
<i>b</i> [Å]	12.2420(4)	18.091(5)	16.0141(1)	14.7452(3)	19.124(5)
<i>c</i> [Å]	20.6738(6)	11.767(3)	29.1646(2)	31.4174(5)	20.771(5)
α[°]	90	90	90	90	90
β[°]	94.951(2)	107.386(10)	90	90	90
γ[°]	90	90	90	90	90
<i>V</i> [Å ³]	3753.95(19)	2307.8(12)	4537.81(4)	4573.63(15)	4208(2)
<i>Z</i>	4	2	4	4	4
<i>T</i> [K]	123(1)	173(2)	123(1)	123(1)	173(2)
λ(Å)	1.54178	0.71073	1.54178	1.54178	0.71073
ρ _{calc} [[g/cm ³]]	1.501	1.414	1.330	1.340	1.365
μ (mm ⁻¹)	5.407	0.531	4.493	4.466	0.568
Theta range [°]	3.82/73.59	3.14 / 27.52	3.03 / 77.23	3.31 / 70.65	3.04 / 27.55
Reflections collected to θ _{full}	31163	17484	133311	18941	39858
Unique reflections (R _{int})	14673 (0.0536)	7241 (0.0633)	9607 (0.0372)	8225 (0.0316)	9502 (0.1704)
Refl. obs. [<i>I</i>]	13553	6331	9596	7103	5301

$>2\sigma(I)$	961	568	586	595	500
Parameters	961	568	586	595	500
Completeness to θ	0.984	0.981	0.999	0.974	0.991
R -values [$I >$ $2\sigma(I)$]	0.0487 / 0.1276	0.0476 / 0.0935	0.0260 / 0.0660	0.0356 / 0.0794	0.0769 / 0.1142
R -values (all data)	0.0527 / 0.1329	0.0600 / 0.1063	0.0260 / 0.0661	0.0449 / 0.0831	0.1508 / 0.1409
GOF on F^2	1.014	1.066	1.043	1.015	1.043
Residual density [e \AA^{-3}]	-0.611 / 0.708	-0.328 / 0.416	-0.270 / 0.253	-0.269 / 0.221	-0.712 / 0.562
Flack- Parameter	-0.012(3)	0.027(19)	0.006(2)	-0.004(4)	0.00(3)
CCDC-No.	946921	946922	946923	946924	946925

Table S2. Thermodynamic parameters of the complexes ($R_{\text{Fe}}, R_{\text{C}}$)- and ($S_{\text{Fe}}, R_{\text{C}}$)-[CpFe(Prophos)(L)]PF₆ (U = Total thermal energy)

Confi- guration	L	U (kcal·mol ⁻¹)	H (kcal·mol ⁻¹)	$T \cdot S$ (kcal·mol ⁻¹)
($R_{\text{Fe}}, R_{\text{C}}$)	MeCN	-2079631.29	-2079630.70	62.76
($R_{\text{Fe}}, R_{\text{C}}$)	PPh ₂ (OMe)	-2572910.72	-2572910.13	74.33
($S_{\text{Fe}}, R_{\text{C}}$)	MeCN	-2079629.09	-2079628.50	62.39
($S_{\text{Fe}}, R_{\text{C}}$)	PPh ₂ (OMe)	-2572910.33	-2572909.74	72.63